1	Mass independent fractionation of even and odd mercury isotopes during free tropospheric mercury
2	oxidation
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27 Abstract. Even mercury (Hg) isotope mass independent fractionation (MIF), observed in rainfall globally, is used to quantify atmospheric Hg deposition pathways. The underlying reaction and MIF mechanism are 28 unknown however. Here we investigate the Hg isotope composition of free tropospheric gaseous elemental Hg<sup>0</sup> 29 and Hg<sup>II</sup> forms. We find that gaseous oxidized Hg<sup>II</sup> has positive  $\Delta^{199}$ Hg,  $\Delta^{201}$ Hg,  $\Delta^{200}$ Hg, and negative  $\Delta^{204}$ Hg 30 signature, similar to rainfall Hg<sup>II</sup>, and we document rainfall Hg<sup>II</sup>  $\Delta^{196}$ Hg to be near-zero. Cloud water and rainfall 31 Hg<sup>II</sup> show enhanced odd MIF compared to gaseous Hg<sup>II</sup>, indicating in-cloud Hg<sup>II</sup> photoreduction. Hg MIF 32 observations of free tropospheric Hg<sup>0</sup> dynamics show how net Hg<sup>0</sup> oxidation leads to opposite MIF in Hg<sup>0</sup> and 33 Hg<sup>II</sup>. A  $\Delta^{200}$ Hg mass balance for Hg<sup>0</sup> and Hg<sup>II</sup> forms suggests that measurements and models underestimate the 34 tropospheric Hg<sup>II</sup> pool. 35

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### 37 Introduction

Modern anthropogenic Hg emissions (2400 Mg a<sup>-1</sup>) outweigh natural Hg emissions by an order of magnitude 38 (Bagnato et al., 2014; Streets et al., 2019). Hg emissions are mostly in the form of gaseous elemental Hg (GEM), 39 which has a long atmospheric life-time against oxidation to divalent reactive Hg (RM) forms that include 40 gaseous oxidized Hg (GOM) and particulate bound Hg (PBM) (Amos et al., 2012). RM is rapidly deposited to 41 42 continental and marine ecosystems where microbial activity transforms inorganic Hg into the bioaccumulating 43 and toxic methylmercury form (Fitzgerald and Lamborg, 2004). Predicting the impact of anthropogenic Hg emission scenarios on global Hg deposition requires detailed understanding of atmospheric Hg transformations 44 and of GEM oxidation in particular. Two decades of atmospheric Hg research using mountain-top and in-flight 45 46 observations, experimental and theoretical reaction kinetics, and coupled 3D atmosphere-ocean chemistry and 47 transport models, have given broad understanding of atmospheric Hg dynamics (Ariya et al., 2015; Dibble et al., 2020; Donohoue et al., 2006; Holmes et al., 2010; Seigneur et al., 2004). Experimental, theoretical and 48 modeling studies suggest Hg oxidation initiated by bromine (Br) radicals to be faster than by ozone or OH 49 (Horowitz et al., 2017; Wang et al., 2014), yet major uncertainties in RM concentrations, oxidation and reduction 50 mechanisms and rates remain (Jaffe et al., 2014; Saiz-Lopez et al., 2018; Weiss-Penzias et al., 2015). 51

Hg stable isotope analysis of atmospheric Hg compounds may provide new information on Hg emission sources and complex atmospheric redox cycling. During the last decade, significant advances have been made in the measurement of Hg isotope compositions of biogeochemical reservoirs (Kwon et al., 2020). Laboratory and field evidence suggests that processes controlling Hg emission such as reduction/oxidation, adsorption/desorption, or volatilization can induce significant mass dependent Hg isotope fractionation (MDF,

 $\delta^{202}$ Hg signature)(Jiskra et al., 2015; Kritee et al., 2012; Wiederhold et al., 2010). Large mass independent Hg 57 isotope fractionation (MIF) of odd Hg isotopes by the magnetic isotope effect has been observed in aqueous 58 photochemical Hg reduction experiments ( $\Delta^{199}$ Hg,  $\Delta^{201}$ Hg signatures) (Bergquist and Blum, 2007; Zheng and 59 Hintelmann, 2009). Odd-MIF signatures imparted on inorganic Hg and methyl-Hg by photochemistry can be 60 traced into all Earth surface environments, i.e sediments, biota, atmosphere, soils etc. Unusual even isotope MIF 61  $(\Delta^{200}$ Hg,  $\Delta^{204}$ Hg) has been observed in rainfall and snowfall globally (Chen et al., 2012; Demers et al., 2013; 62 Gratz et al., 2010), and has been speculated to originate in GEM photo-oxidation at and above the tropopause 63 64 (Chen et al., 2012). The exact even-MIF mechanism and even-MIF inducing Hg transformation(s) remain essentially unknown. 65

In this study we investigate Hg isotope composition of GEM, RM, and cloud water at the high altitude Pic du Midi Observatory (PDM, 2877m, France). The PDM regularly receives free tropospheric air masses that are enriched in RM and depleted in GEM (Fu et al., 2016a). High frequency, 12h and 24h GEM isotope observations of a free tropospheric air mass rich in GOM and depleted in GEM are investigated to derive MDF and MIF factors for net oxidation. Photoreduction experiments on rainfall are made to constrain atmospheric aqueous MDF and MIF factors. Finally, we investigated the even-MIF on the low abundance <sup>196</sup>Hg isotope to help understand potential even-MIF mechanisms.

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#### 74 Materials and methods

### 75 Site description and ancillary parameters

The Pic du Midi Observatory (PDM, 0.14° E, 42.94° N, 2877 m a.s.l) is a high-altitude monitoring station situated on top of an isolated peak on the north edge of the central Pyrenees Mountains, Southwest France. The PDM receives free tropospheric air from the Northeast Atlantic Ocean and Europe and boundary layer air from upslope winds in Spain and southwest France (Gheusi et al., 2011; Hulin et al., 2019). The station receives predominantly southerly and southwesterly winds (> 70 %) across the main range of Pyrenees (Hulin et al., 2019). There are no major anthropogenic Hg emission point sources around the station or in the surrounding areas.

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# 84 Atmospheric sampling and processing

Short-term (12/24h) variations in atmospheric GEM isotopic composition were investigated from 19 to 28 in
 Dec 2012 at PDM using large chlorine-impregnated activated carbon (CLC) traps. Traps with 600 mg CLC were

utilized for 24 hour sampling at a flow rate of ~ 10 L min<sup>-1</sup>; whereas traps with 1000 mg of CLC were utilized 87 88 for 12 hour sampling at a flow rate of ~ 20 LPM (Fu et al., 2014). The sampling inlets of the CLC traps (heated 89 at 70°C to avoid water condensation), preceded by a pre-burned 47mm quartz fiber filter were connected to a 90 Tekran 1104 Teflon coated sampling manifold, operating at 100 LPM and 70°C, and installed into the prevailing 91 wind direction. Sampling flow rates of CLC traps were regulated via ball flow meters installed at the outlets of 92 the vacuum pumps, and the total sampling volumes of the traps were recorded using gas meters. Similar to 93 Arctic studies (Berg et al., 2013; Steffen et al., 2005) we find that the Hg collected on CLC traps reflects the 94 GEM fraction, and is not confounded by the GOM fraction. Both GOM and PBM fractions are retained on the 95 quartz filter preceding the CLC trap. Quartz has been previously shown to efficiently scavenge GOM over time periods shorter than 24h (Lyman and Jaffe, 2011; Lynam and Keeler, 2002). CLC traps were thermally 96 97 decomposed (25°C to 1000°C over 8h) in a Hg-free oxygen flow (25 ml min<sup>-1</sup>) in a tube furnace. Combustion products were further decomposed in a 2<sup>nd</sup> pyrolizer tube furnace at 1000°C, and purged into a 10 ml 40 vol% 98 2HNO<sub>3</sub>/1HCl oxidizing solution (Sun et al., 2013). CLC sampling efficiencies were better than 95%, mean 99 recovery efficiencies were 87%, and full procedural blanks <2% of Hg concentrations in trapping solutions (Fu 100 101 et al., 2014).

102 Twelve samples of the atmospheric RM (RM=GOM+PBM) were obtained using Millipore 103 polyethersulfone cation exchange membranes (CEMs, Millipore, 0.45 µm pore size, 47 mm and 90 mm diameter (Marusczak et al., 2017)) at the PDM from June to October 2014, with sampled air volumes between 25 and 77 104 105 m<sup>3</sup>. Cloud water samples were collected from July to November 2014 with a CASCC-2 sampler at the PDM 106 (Demoz et al., 1996). Rainfall was sampled from July to August 2014 150km downwind from the PDM at 800m 107 altitude in the Pyrenees foothills using acid-washed polypropylene buckets and reported elsewhere (Enrico et al., 2016). Snowfall and rainfall was collected at the PDM and the Geosciences Environnement Toulouse (GET, 108 109 Toulouse, France, 43.54° N, 1.48° E) laboratory between 2016 and 2017 using the same methods. Cloud water, 110 rainfall and snowfall samples collected after each event were transferred into pre-cleaned Pyrex glass bottles (1 and 2 L volume), then oxidized by adding 0.1 - 1 % BrCl, and kept refrigerated before Hg concentration analysis 111 by CVAFS and further pre-concentration for Hg isotope analysis. Hg from rain and cloud water samples was 112 pre-concentrated by reducing and purging Hg with SnCl<sub>2</sub> and a clean argon flow and subsequent trapping in an 113 114 oxidizing solution, following Sherman, et al. (Sherman et al., 2012). Recoveries of procedural standards, rainfall 115 and cloud water samples was >95%. RM on CEMs was leached in 20 vol% 2HNO<sub>3</sub>/1HCl with blanks <4% of 116 sampled RM.

Hg isotope fractionation factors during aqueous photochemical reduction of rainwater was studied using a solar simulator (see (Yang et al., 2019) for details). A low Hg level (2.6 ng L<sup>-1</sup>) rainfall sample 'GET rain 4' obtained at GET was spiked with 103 ng L<sup>-1</sup> NIST 3133 and equilibrated for 48 in the dark. A 450 mL aliquot was then transferred into an acid-cleaned quartz reactor, and illuminated for 24h with the solar spectrum from a Xe lamp. After 24, the 78 ng L<sup>-1</sup> dissolved Hg(II) remaining in the 450mL reactor volume was transferred to a pre-cleaned 1L Pyrex bottle, and pre-concentrated for Hg isotope analysis.

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# 124 Mercury isotopic analysis

Hg isotope ratios were measured using cold vapor multi-collector inductively coupled plasma mass 125 spectrometry (CV-MC-ICPMS, Thermo-Finnigan Neptune, and Neptune Plus) at the Midi-Pyrenees 126 Observatory, Toulouse, France following the methods described in previous studies (Jiskra et al., 2019; Obrist 127 128 et al., 2017). Instrumental mass bias of MC-ICPMS was corrected by standard-sample-standard bracketing using NIST3133 Hg at matching concentrations. Hg isotopic composition is reported in delta notation ( $\delta$ ) in units of 129 130 per mil (‰) referenced to the bracketed NIST 3133 Hg standard (Blum and Bergquist, 2007):  $\delta^{xxx}$ Hg = [(xxxHg/<sup>198</sup>Hg)<sub>sample</sub>/(xxxHg/<sup>198</sup>Hg)<sub>NIST3133</sub> -1], where xxx represent Hg isotope masses 199, 200, 201, 202, 204. 131 132 MIF values are expressed by "capital delta ( $\Delta$ )" notation (‰), which is the difference between the measured values of  $\delta^{199}$ Hg,  $\delta^{200}$ Hg,  $\delta^{201}$ Hg and those predicted from  $\delta^{202}$ Hg using the kinetic MDF law:  $\Delta^{xxx}$ Hg (‰) = 133  $\delta^{xxx}$ Hg - ( $\beta^{xxx} \times \delta^{202}$ Hg), where  $\beta^{xxx}$  is -0.5074, 0.2520, 0.5024, 0.7520, 1.493 for the 196, 199, 200, 201, and 204 134 Hg isotopes respectively. The analytical uncertainty of isotopic analysis was assessed by repeated analysis of 135 136 the UM-Almaden and ETH-Fluka standards over different analysis sessions and at Hg concentrations 137 corresponding to the samples. The overall mean values of  $\delta^{202}$ Hg,  $\Delta^{199}$ Hg,  $\Delta^{200}$ Hg,  $\Delta^{201}$ Hg, and  $\Delta^{204}$ Hg for all the UM-Almaden standards in the 0.5 to 1.5 ng g<sup>-1</sup> range were  $-0.56 \pm 0.07$  ‰,  $-0.04 \pm 0.08$  ‰,  $-0.01 \pm 0.09$  ‰ 138 and  $-0.05 \pm 0.09$  ‰, and  $0.04 \pm 0.19$  ‰ (2SD, n = 11) respectively, which agree well with previously reported 139 values (Blum and Bergquist, 2007). The procedural standard for rainfall pre-concentration, i.e. rainfall sample 140 141 'GET rain 4' spiked with 103 ng g<sup>-1</sup> NIST 3133 showed good recovery and no isotopic bias (Supporting Data). In the present study, the analytical uncertainty of CV-MC-ICPMS isotope analysis of RM, GEM, rainfall, and 142 snowfall is the 2SD uncertainty of the UM-Almaden standard, unless the 2SD uncertainty on repeated analysis 143 144 of the same sample over different analytical sessions was larger.

145 To attempt rainfall  $\Delta^{196}$ Hg analysis, we screened rainfall samples obtained at GET in 2016 for both high 146 Hg concentration levels and high volume single events. Rainfall sample 'GET rain 6' was the only sample 147 retained as it consisted of a single thunderstorm event of 20.5 L, collected in the 8 large buckets, and had a Hg concentration of 6.0 ng L<sup>-1</sup>. The entire 20.5L volume was pre-concentrated 1200x in a single large Pyrex glass 148 bottle to a final concentration of 8.0 ng g<sup>-1</sup> in 15.3 g of 20vol% iAR. The sample was run in duplicate during 149 150 two different days using dedicated UM-Almaden and ETH-Fluka standards, also at 8.0 ng g<sup>-1</sup>. All individual sample and standard results are given in the Supporting Data. The signals obtained on the <sup>196</sup>Hg and <sup>202</sup>Hg 151 isotopes during these sessions were 15mV and 3 Volt. We recommend the use of  $10^{12}$  or  $10^{13} \Omega$  resistors on the 152 signal amplifier of the <sup>196</sup>Hg Faraday cup. The 2<sub>o</sub> uncertainty, based on sample, UM-Almaden, and ETH-Fluka 153 replicates, on  $\Delta^{196}$ Hg during these analysis sessions were 0.19‰ (Supplementary Data). 154

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#### 156 Ancillary parameters

Atmospheric Hg speciation at the PDM, including GEM, GOM and PBM, was continuously measured off 157 158 the same sampling manifold with a Tekran 2537/1130/1135 system (Tekran Inc., Canada) (Fu et al., 2016a; Marusczak et al., 2017). Automated GOM and PBM analyses are subject to mounting criticism as the techniques 159 160 lack calibration, collection efficiency control and interference correction (Jaffe et al., 2014). We partly address 161 these issues by parallel manual sampling of RM (GOM + PBM) on CEMs (Huang et al., 2013). We find that 162 denuder-based GOM at the PDM is systematically underestimated by a factor of 1.56 for which we correct (indicated by an asterisk hereafter, i.e. GOM\*)(Marusczak et al., 2017). Atmospheric CO and O<sub>3</sub> concentrations 163 were continuously measured using a TEI (Thermo Environmental Instruments Inc. USA) 48CTL gas filter 164 correlation analyzer and 49C UV-absorption ozone analyzer, respectively, following published procedures 165 166 (Gheusi et al., 2011). Meteorology, CO and O3 data at the PDM were provided through the Pyrenean platform for observation of the atmosphere (P2OA, http://p2oa.aero.obs-mip.fr). 167

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## 169 Air mass backward trajectories

Seven day backward trajectories ending at PDM at a height of 3000 m above sea level were calculated every 2 hours with simulated gridded meteorological data from the Weather Research and Forecasting (WRF) model Version 3.4 with the Noah Land Surface Model, Thompson microphysics scheme, Grell-3 cumulus scheme, Dudhia radiation physics scheme, YSU scheme for PBL. The modeled domain of meteorological data mainly covered the eastern North Atlantic Ocean and European area, with a time resolution of 1 h, 400 × 800 grid cells at a 12-km horizontal resolution, and vertical levels of 50 from 1000 hPa to 20 hPa. All the trajectories during the 12h, 24h of individual Hg isotope samples were grouped into an average backward trajectory to show 177 the mean air mass source region of samples.

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179 **Results and discussion** 

## 180 **GEM isotope compositions**

181 The nine day sampling period from 19 to 28 Dec 2012 was characterized by a shift from cloudy conditions (days 1-2, samples D1-D3) to clear skies (days 3-5, samples D4-D9), and back to clouds (days 7-9, samples D10-182 183 D14). The wind direction was predominantly southwest bringing North-Atlantic Ocean air masses, over 184 northwest Spain, to the PDM. The continuous Hg speciation observations show a pronounced increase in GOM\* to 252 pg m<sup>-3</sup> (48h mean), and concurrent decrease in GEM during samples D4-D6 (clear skies, Figure 1, Table 185 186 S1). Back trajectory analysis, and low observed relative humidity and CO, suggest that the high GOM event represents the intrusion of a free tropospheric air mass that originated at 5000m (550hPa, Figures 1, 2). Such 187 188 intrusions are typical at PDM, e.g. 61 events were registered over a one year period from Nov 2011 to Nov 2012 (Fu et al., 2016a). The intrusions are of particular interest as they offer a snapshot of the free troposphere where 189 190 GEM oxidation is thought to take place (Fu et al., 2016a; Lyman and Jaffe, 2011; Swartzendruber et al., 2006; Timonen et al., 2013).  $\delta^{202}$ Hg,  $\Delta^{199}$ Hg and  $\Delta^{200}$ Hg of 12h and 24h GEM samples collected from 19 to 28 Dec 191 2012 were in the range of 0.16 to 0.51‰ ( $\delta^{202}$ Hg, mean = 0.34‰), -0.31 to -0.11‰ ( $\Delta^{199}$ Hg, mean = -0.23‰), 192 193 and -0.17 to 0.04‰ ( $\Delta^{200}$ Hg, mean = -0.07‰) (Table S1). These observations are consistent with a single two-194 week integrated GEM isotope sample B19, collected over the same period 19 to 28 Dec 2012 at a lower flow rate of 2 L min<sup>-1</sup>. with  $\delta^{202}$ Hg of 0.39‰,  $\Delta^{199}$ Hg of -0.27‰,  $\Delta^{200}$ Hg of -0.10 ‰ (Fu et al., 2016b). 195

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## 197 RM, cloud water and rainfall Hg isotope compositions

In a previous study we found that rainfall Hg isotope signatures 150km east from the PDM are typical of rainfall 198 observations globally, with a mean  $\delta^{202}$ Hg 0.17 ± 0.33 ‰ and significantly positive odd MIF ( $\Delta^{199}$ Hg = 0.72 ± 199 0.15 %;  $\Delta^{201}$ Hg =  $0.76 \pm 0.14 \%$ ) and even-MIF ( $\Delta^{200}$ Hg =  $0.21 \pm 0.04 \%$ ;  $\Delta^{204}$ Hg =  $-0.33 \pm 0.10 \%$ , all 1 $\sigma$ , n 200 = 9) (Figure 3, Table S1, Enrico et al., 2016). Here we find that PDM cloud water THg concentrations and MDF 201 and MIF signatures are not significantly different from PDM and regional rainfall (p=0.26 to 0.79). For the first 202 time we observe that RM, i.e. GOM+PBM, at PDM also has positive odd MIF ( $\Delta^{199}$ Hg = 0.44 ± 0.17 ‰;  $\Delta^{201}$ Hg 203 = 0.44 ± 0.10 ‰) and even-MIF ( $\Delta^{200}$ Hg = 0.15 ± 0.06 ‰;  $\Delta^{204}$ Hg = -0.22 ± 0.17 ‰, all 1 $\sigma$ , n = 12), similar to 204 205 PDM cloud water and rainfall (Figure 3) and similar to global rainfall ( $\Delta^{199}$ Hg = 0.35 ± 0.19 ‰;  $\Delta^{200}$ Hg = 0.14  $\pm 0.06$  %;  $\Delta^{201}$ Hg = 0.35  $\pm 0.20$  % (Kwon et al., 2020). Demers et al. (2013) were the first to point out a negative 206

correlation between  $\Delta^{200}$ Hg and  $\Delta^{204}$ Hg in rainfall (Demers et al., 2013). Our work at the PDM extends this 207 208 relationship to RM, cloud water, and GEM phases (Figure 3). The presence of even- and odd-MIF in RM 209 suggests that globally observed Hg MIF signatures in rainfall are inherited from RM. Equally important, the 210 isotopic composition of RM is in stark contrast with that of GEM (Figure 3), even though the only 211 transformation that separates the two forms of Hg is net oxidation. This corroborates the suggestion that net photochemical oxidation of GEM is responsible for even MIF (Chen et al., 2012), but it does not identify the 212 213 oxidative or reductive nature of the MIF inducing step among the myriad of possible reactions that drive 214 atmospheric net GEM oxidation.

Cloud water and rainfall  $\delta^{202}$ Hg (0.17 ± 0.34 ‰) and  $\Delta^{199}$ Hg (0.21 ± 0.04 ‰) show significant enrichment 215 in the heavier and in the odd Hg isotopes relative to mean RM  $\delta^{202}$ Hg (-0.73 ± 0.63 ‰) and  $\Delta^{199}$ Hg (0.44 ± 216 0.18 %)(both p<0.001, all mean,  $1\sigma$ ). We investigated if this is caused by in-cloud Hg<sup>II</sup> photoreduction, by 217 incubating a rainfall sample spiked with 103 ng L<sup>-1</sup> of NIST 3133 Hg. After 24h exposure to artificial sunlight 218 from a Xe lamp, the Hg concentration had dropped to 78 ng L<sup>-1</sup> and small positive shifts in  $\delta^{202}$ Hg (0.15 ‰), 219  $\Delta^{199}$ Hg (0.10‰), and  $\Delta^{201}$ Hg (0.11‰), but not  $\Delta^{200}$ Hg (-0.03‰) were detected. These observations are 220 consistent with limited, slow in-cloud Hg<sup>II</sup> photoreduction affecting cloud water and rainfall MDF and odd-MIF. 221 Finally, we observe for the first time the  $\Delta^{196}$ Hg signature to be near-zero (-0.06 ± 0.19‰, 2 $\sigma$ , n=2) in the 'GET 222 223 rain 6' sample (Figure 4), which has pronounced  $\Delta^{199}$ Hg (0.70‰),  $\Delta^{200}$ Hg (0.25‰),  $\Delta^{201}$ Hg (0.82‰), and  $\Delta^{204}$ Hg (-0.45‰). The observation that the least abundant <sup>196</sup>Hg, and most abundant <sup>202</sup>Hg isotope, both referenced to 224 <sup>198</sup>Hg, show near-zero MIF, further suggests that nuclear self-shielding is unlikely to be the mechanism 225 226 responsible for atmospheric Hg MIF.

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# 228 $\delta^{202}$ Hg and $\Delta^{199}$ Hg variation during the high GOM event

We obtained additional evidence for odd and even MIF during net GEM oxidation from the high frequency (12-229 24h) GEM isotope observations during the free tropospheric air mass intrusion, enriched in GOM and depleted 230 in GEM (Figure 1, days 4-6). The high GOM event showed GOM\* concentrations of 252 pg m<sup>-3</sup> from 23-24 231 Dec (48h mean). The GOM event was accompanied by low CO concentrations, depleted GEM concentrations, 232 and low relative humidity (Figure 1,2), which is consistent with previous studies and indicative of GOM 233 production by oxidation of GEM in a free tropospheric air mass (Gratz et al., 2015; Shah et al., 2016). Seven-234 235 day back-trajectories of the GOM event and of the pre- (days 1-3) and post-event (days 7-9) air masses illustrate the important control of cloudiness along the trajectory on detected GOM levels at the PDM (Figure 2). Pre/post-236

GOM-event air masses generally travelled at 700-1000 hPa (3000m - 0m asl) over the N-Atlantic Ocean (35-237 50°N) with RH from 60-100% and daytime solar radiation  $<200 \text{ wm}^{-2}$  suggestive of GOM scavenging by clouds 238 239 (Figure 1,2). GOM-event air masses also travelled over the subtropical N-Atlantic Ocean, though at lower 240 latitudes (20-35°N), and higher altitudes of 500-900 hPa (5500m - 1500m asl) with RH from 0-40% and daytime solar radiation from 200-500 w m<sup>-2</sup> suggestive of clear skies. GOM-event mean ozone (47  $\pm$  2 ppbv, sd), CO 241  $(83 \pm 3 \text{ ppbv, sd})$  and relative humidity  $(34 \pm 10\%, \text{ sd})$  are typical of lower to middle free tropospheric air. Both 242 243 the back trajectories and ozone, CO, RH characteristics suggest a negligible influence of stratospheric air on the 244 observed GOM-event.

A negative correlation between GEM and GOM\* was observed during the 8-day study period of Dec 2012 245 with GOM\*/GEM slope of -0.44  $\pm$  0.10 (r<sup>2</sup> = 0.66; Figure 5). It has been previously suggested that GOM/GEM 246 slopes higher than -1 reflect the loss of newly produced GOM by scavenging processes, or the mixing with other 247 248 air masses with different GEM levels (Lyman and Jaffe, 2011; Swartzendruber et al., 2006). PBM concentrations during the high-GOM event are uniformly low (mean, 4 pg m<sup>-3</sup>) and RH and solar radiation levels along the 249 back trajectories do not support an important scavenging sink (Figure 2). A prerequisite for the GOM\*/GEM 250 slope analysis is that the ensemble of observations has a common air mass origin, which is not the case here. 251 252 The back trajectory analysis of our 8-day observations suggests overlapping altitudes for pre- and post-event (3000m – 0m asl) and GOM-event (5500m – 1500m asl) air masses. Published reviews of GEM altitude profiles 253 suggest little variation up to 5km (Holmes et al., 2010). However, observed and modeled boundary layer GEM 254 levels suggest maximum levels between 30-60°N, corresponding to mid-latitude emission sources, and declining 255 256 levels from 30°N to the equator (1% decrease in GEM per degree latitude, Holmes et al., 2010). As mentioned above pre/post-event and GOM-event back trajectories point out different latitudinal origins of 20-35°N and 35-257 50°N respectively. It seems therefore plausible that the original GEM levels in pre/post-event and GOM-event 258 levels air masses were different due to different latitudinal air mass origins. A correction for latitudinal 259 confounding allows estimating original GEM levels to be 15% lower in the GOM-event air masses, based on 260 the 1% drop in GEM per degree latitude, and yields a corrected GOM\*/GEM slope of -0.88 (Figure 5, gray 261 line). The corrected slope is close to -1 and therefore suggests that all nearly GOM produced has been retained 262 within the free tropospheric air masses observed during the high GOM-event at the PDM. 263

 $\delta^{202}\text{Hg}_{\text{GEM}} \text{ and } \Delta^{204}\text{Hg}_{\text{GEM}} \text{ during the high-GOM event were slightly but significantly higher, and } \Delta^{199}\text{Hg}_{\text{GEM}},$  $\Delta^{200}\text{Hg}_{\text{GEM}} \text{ and } \Delta^{201}\text{Hg}_{\text{GEM}} \text{ slightly but significantly lower than pre- and post-event samples during 20 - 28 Dec}$ 2012 (Figure 1; t-test all p<0.05). All five GEM isotope signatures are significantly correlated with GEM (Figure 1) 267 6) and GOM concentration. Net GEM oxidation therefore seems to have caused significant MDF, odd-MIF and even-MIF. The above analysis of the GOM event at the PDM suggests that the observed air masses represent 268 the reactants, products and isotope fractionation characteristics of net GEM oxidation. Because the GOM-event 269 was long-lived, displaying elevated GOM levels over 48 hours, we suggest that the GEM and GOM 270 271 concentration, and GEM isotope observations represent a snapshot of net GEM oxidation that has been little influenced by further mixing with boundary layer air. This implies that the observed MDF, odd-MIF and even-272 273 MIF has accompanied net GEM oxidation in the lower troposphere. Previous observations on even-MIF in precipitation ( $\Delta^{200}$ Hg only;  $\Delta^{204}$ Hg not observed; Chen et al. 2012) led to the suggestion that even-MIF 274 originates at the tropopause or stratosphere. Our results suggest that the even-MIF mechanism also operates 275 276 within the free troposphere.

In Figure 6 we plot the GEM and RM (~GOM) isotope observations in Rayleigh diagrams, and fit MDF, 277 278 odd-MIF and even-MIF factors for net GEM oxidation. Assuming pre-/post GOM event isotope observations as the starting point for GEM reactant in the free troposphere, we applied Rayleigh distillation equations for 279 residual  $\delta^{202}$ Hg<sub>GEM</sub>,  $\Delta^{199}$ Hg<sub>GEM</sub>,  $\Delta^{200}$ Hg<sub>GEM</sub>,  $\Delta^{201}$ Hg<sub>GEM</sub>, and  $\Delta^{204}$ Hg<sub>GEM</sub> and for cumulative product GOM to 280 estimate corresponding isotope fractionation factors  $\varepsilon_{MDF(GOM-GEM)}^{202/198}$  of -1.1±0.2‰,  $\varepsilon_{MIF(GOM-GEM)}^{199/198}$  of 281  $0.65 \pm 0.11\%$ ,  $\varepsilon_{MIF(GOM-GEM)}^{200/198}$  of  $0.22 \pm 0.05\%$ ,  $\varepsilon_{MIF(GOM-GEM)}^{201/198}$  of  $0.65 \pm 0.10\%$ , and  $\varepsilon_{MIF(GOM-GEM)}^{204/198}$  of  $-0.39 \pm 0.12\%$ 282 for the net GEM to GOM conversion observed (Figure 6). It should be kept in mind that these fractionation 283 284 factor do not inform on whether oxidation or reduction is the underlying cause, i.e. they represent net oxidation.

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## 286 Atmospheric GEM and RM pool size

GEM and RM represent the two dominant reduced and oxidized atmospheric Hg pools. The relative size of 287 these two pools is not very well constrained due to limited mountain-top and in-flight measurements and bias 288 289 in denuder based RM measurements (Jaffe et al., 2014). Recent global Hg chemistry and transport models indicate that RM (500 Mg) represents 12% and GEM 88% (3600 Mg) of atmospheric Hg, most of this residing 290 291 in the free troposphere (Horowitz et al., 2017). Based on the newly observed free tropospheric RM and GEM isotope compositions (this study), and previously observed rainfall Hg, and modeled Hg emission fluxes and 292 isotopic composition, we can estimate by isotopic mass balance the proportions of RM and GEM (Table 1). The 293 294  $\Delta^{200}$ Hg signature, is best suited for this, since it is not confounded by additional even-MIF in cloud water, ocean or soil environment. Annual mean GEM  $\Delta^{200}$ Hg at the PDM of -0.03±0.03‰ (Fu et al., 2016b) is identical to 295

GEM  $\Delta^{200}$ Hg of -0.04‰ of the pre-/post GOM intrusion air masses, representing the northern hemisphere free 296 tropospheric GEM pool. RM  $\Delta^{200}$ Hg at PDM (0.15‰) is similar to mean northern hemispheric rainfall  $\Delta^{200}$ Hg 297 (0.13±0.06‰), suggesting that both values represent the free tropospheric RM pool. We use a model estimate 298 299 for  $\Delta^{200}$ Hg (0.02±0.02‰) of global Hg emissions (Sun et al., 2019) to constrain the mass balance, and calculate that  $30\pm17\%$  (1 $\sigma$ ) of free tropospheric Hg should be present as RM in order to explain the rainfall, RM and 300 GEM  $\Delta^{200}$ Hg observations. This would suggest that current RM observations and models underestimate the size 301 of the atmospheric RM pool by a factor of two. Our assessment is potentially biased due to the lack of 302 303 observations in the southern hemisphere.

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#### 313 Author contributions

J.E.S conceived the study. X.F, N.M., Y.X, M.J. and J.E.S performed field sampling. X.F, N.M., Y.X, M.J, LEHB,

315 JC. and J.E.S performed laboratory measurements. F.G. managed ancillary data collection at the PDM. X.F and

316 J.E.S wrote the draft manuscript, which was improved by contributions from all authors.

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## 318 Additional information

319 All Hg stable isotope data measured and discussed in this study is reported in the Supplementary Data.

## 321 References

- 322 Amos, H. M., Jacob, D. J., Holmes, C. D., Fisher, J. A., Wang, Q., Yantosca, R. M., Corbitt, E. S., Galarneau, E.,
- Rutter, A. P., Gustin, M. S., Steffen, A., Schauer, J. J., Graydon, J. A., St Louis, V. L., Talbot, R. W., Edgerton, E. S.,
  Zhang, Y. and Sunderland, E. M.: Gas-particle partitioning of atmospheric Hg(II) and its effect on global mercury
  deposition, Atmospheric Chemistry and Physics, 12(1), 591–603, doi:10.5194/acp-12-591-2012, 2012.
- Ariya, P. A., Amyot, M., Dastoor, A., Deeds, D., Feinberg, A., Kos, G., Poulain, A., Ryjkov, A., Semeniuk, K., Subir,
  M. and Toyota, K.: Mercury Physicochemical and Biogeochemical Transformation in the Atmosphere and at
- 328 Atmospheric Interfaces: A Review and Future Directions, Chemical Reviews, 115(10), 3760-3802,
- 329 doi:10.1021/cr500667e, 2015.
- Bagnato, E., Tamburello, G., Avard, G., Martinez-Cruz, M., Enrico, M., Fu, X., Sprovieri, M. and Sonke, J. E.:
  Mercury fluxes from volcanic and geothermal sources: an update., Geological Society, London, Special Publications,
  doi 10.1144/SP410.2, 2014.
- Berg, T., Pfaffhuber, K. A., Cole, A. S., Engelsen, O. and Steffen, A.: Ten-year trends in atmospheric mercury
   concentrations, meteorological effects and climate variables at Zeppelin, Ny-Ålesund, Atmospheric Chemistry and
   Physics, 13(13), 6575–6586, doi:10.5194/acp-13-6575-2013, 2013.
- Bergquist, B. A. and Blum, J. D.: Mass-Dependent and -Independent Fractionation of Hg Isotopes by Photoreduction
   in Aquatic Systems., Science, 318, 417–420, 2007.
- Blum, J. D. and Bergquist, B. A.: Reporting of variations in the natural isotopic composition of mercury., Analytical
   and Bioanalytical Chemistry, 388, 353–359, 2007.
- Blum, J. D., Sherman, L. S. and Johnson, M. W.: Mercury Isotopes in Earth and Environmental Sciences, in Annual
  Review of Earth and Planetary Sciences, Vol 42, vol. 42, edited by R. Jeanloz, pp. 249–269., 2014.
- Buchachenko, A. L., Ivanov, V. L., Roznyatovskii, V. A., Artamkina, G. A., Vorob'ev, A. K. and Ustynyuk, Y. A.:
  Magnetic isotope effect for mercury nuclei in photolysis of Bis(p-trifluoromethylbenzyl)mercury., Doklady Physical
  Chemistry, 413, 39–41, 2007.
- Chen, J., Hintelmann, H., Feng, X. and Dimock, B.: Unusual fractionation of both odd and even mercury isotopes in
  precipitation from Peterborough, ON, Canada, Geochimica et Cosmochimica Acta, 90, 33–46,
  doi:10.1016/j.gca.2012.05.005, 2012.
- Demers, J. D., Blum, J. D. and Zak, D. R.: Mercury isotopes in a forested ecosystem: Implications for air-surface
  exchange dynamics and the global mercury cycle, Global Biogeochemical Cycles, 27(1), 222–238,
  doi:10.1002/gbc.20021, 2013.
- Demoz, B., Collett, J., J. L. and Daube, J., B. C.: On the Caltech Active Strand Cloudwater Collectors., Atmospheric
   Research, 41, 47–62, 1996.
- Dibble, T. S., Tetu, H. L., Jiao, Y., Thackray, C. P. and Jacob, D. J.: Modeling the OH-Initiated Oxidation of Mercury
  in the Global Atmosphere without Violating Physical Laws, Journal of Physical Chemistry A, 124(2), 444–453,
  doi:10.1021/acs.jpca.9b10121, 2020.

- Donohoue, D. L., Bauer, D., Cossairt, B. and Hynes, A.: Temperature and pressure dependent rate coefficients for the reaction of Hg with Br and the reaction of Br with Br: A pulsed laser photolysis-pulsed laser induced fluorescence
- 358 study, Journal of Physical Chemistry A, 110, 6623–6632, 2006.
- 359 Enrico, M., Le Roux, G., Marusczak, N., Heimbürger, L.-E., Claustres, A., Fu, X., Sun, R. and Sonke, J. E.:
- 360 Atmospheric mercury transfer to peat bogs dominated by gaseous elemental mercury dry deposition., Environmental
- 361 Science & Technology, doi:10.1021/acs.est.5b06058, 2016.
- Fitzgerald, W. F. and Lamborg, C. H.: Geochemistry of Mercury in the Environment, in Treatise on Geochemistry,
  vol. 9, edited by E. D. Holland and K. K. Turekian, pp. 107–148, Elsevier-Pergamon, Oxford., 2004.
- Fu, X., Heimburger, L. E. and Sonke, J. E.: Collection of atmospheric gaseous mercury for stable isotope analysis
   using iodine- and chlorine-impregnated activated carbon traps, Journal of Analytical Atomic Spectrometry, DOI:
   10.1039/C3JA50356A, 2014.
- Fu, X. W., Marusczak, N., Heimbürger, L. E., Sauvage, B., Gheusi, F. and Sonke, J. E.: Atmospheric mercury
  speciation dynamics at the high-altitude Pic du Midi Observatory, southern France, Atmospheric Chemistry and
  Physics, 16, 5623–5639, doi:10.5194/acp-16-5623-2016, 2016a.
- Fu, X. W., Marusczak, N., Wang, X., Gheusi, F. and Sonke, J. E.: The isotopic composition of total gaseous mercury
  in the free troposphere of the Pic du Midi Observatory (2877 m a.s.l, France), Environmental Science & Technology,
  submitted, 2016b.
- Gheusi, F., Ravetta, F., Delbarre, H., Tsamalis, C., Chevalier-Rosso, A., Leroy, C., Augustin, P., Delmas, R., Ancellet,
  G., Athier, G., Bouchou, P., Campistron, B., Cousin, J. M., Fourmentin, M. and Meyerfeld, Y.: Pic 2005, a field
  campaign to investigate low-tropospheric ozone variability in the Pyrenees, Atmospheric Research, 101(3), 640–665,
  doi:10.1016/j.atmosres.2011.04.014, 2011.
- Gratz, L., Keeler, G., Blum, J. and Sherman, L. S.: Isotopic composition and fractionation of mercury in Great Lakes
   precipitation and ambient air, Environmental Science and Technology, 44, 7764–7770, 2010.
- Gratz, L. E., Ambrose, J. L., Jaffe, D. A., Shah, V., Jaegle, L., Stutz, J., Festa, J., Spolaor, M., Tsai, C., Selin, N. E.,
  Song, S., Zhou, X., Weinheimer, A. J., Knapp, D. J., Montzka, D. D., Flocke, F. M., Campos, T. L., Apel, E.,
  Hornbrook, R., Blake, N. J., Hall, S., Tyndall, G. S., Reeves, M., Stechman, D. and Stell, M.: Oxidation of mercury
  by bromine in the subtropical Pacific free troposphere, Geophysical Research Letters, 42(23),
  doi:10.1002/2015gl066645, 2015.
- Grissom, C. B.: Magnetic field effects in biology: A survey of possible mechanisms with emphasis on radical-pair
   recombination, Chemical Reviews, 95, 3–24, 1995.
- Holmes, C. D., Jacob, D. J., Corbitt, E. S., Mao, J., Yang, X., Talbot, R. and Slemr, F.: Global atmospheric model for
  mercury including oxidation by bromine atoms, Atmospheric Chemistry and Physics, 10, 12037–12057, 2010.
- 388 Horowitz, H. M., Jacob, D. J., Zhang, Y., Dibble, T. S., Slemr, F., Amos, H. M., Schmidt, J. A., Corbitt, E. S., Marais,
- 389 E. A. and Sunderland, E. M.: A new mechanism for atmospheric mercury redox chemistry: implications for the global
- 390 mercury budget, Atmospheric Chemistry and Physics, 17(10), 6353–6371, doi:10.5194/acp-17-6353-2017, 2017.

- Hulin, M., Gheusi, F., Lothon, M., Pont, V., Lohou, F., Ramonet, M., Delmotte, M., Derrien, S., Athier, G., Meyerfeld,
- 392 Y., Bezombes, Y., Augustin, P. and Ravetta, F.: Observations of Thermally Driven Circulations in the Pyrenees:
- 393 Comparison of Detection Methods and Impact on Atmospheric Composition Measured at a Mountaintop, Journal of
- 394 Applied Meteorology and Climatology, 58(4), 717–740, doi:10.1175/JAMC-D-17-0268.1, 2019.
- Jaffe, D. A., Lyman, S., Amos, H. M., Gustin, M. S., Huang, J., Selin, N. E., Levin, L., ter Schure, A., Mason, R. P.,
- 396 Talbot, R., Rutter, A., Finley, B., Jaegle, L., Shah, V., McClure, C., Arnbrose, J., Gratz, L., Lindberg, S., Weiss-
- 397 Penzias, P., Sheu, G.-R., Feddersen, D., Horvat, M., Dastoor, A., Hynes, A. J., Mao, H., Sonke, J. E., Slemr, F., Fisher,
- 398 J. A., Ebinghaus, R., Zhang, Y. and Edwards, G.: Progress on Understanding Atmospheric Mercury Hampered by
- 399 Uncertain Measurements, Environmental Science & Technology, 48(13), 7204–7206, doi:10.1021/es5026432, 2014.
- Jiskra, M., Wiederhold, J. G., Skyllberg, U., Kronberg, R.-M., Hajdas, I. and Kretzschmar, R.: Mercury Deposition
   and Re-emission Pathways in Boreal Forest Soils Investigated with Hg Isotope Signatures, Environmental Science
- 402 & Technology, 49(12), 7188–7196, doi:10.1021/acs.est.5b00742, 2015.
- Jiskra, M., Sonke, J. E., Agnan, Y., Helmig, D. and Obrist, D.: Insights from mercury stable isotopes on terrestrialatmosphere exchange of Hg(0) in the Arctic tundra, Biogeosciences, 16(20), 4051–4064, doi:10.5194/bg-16-40512019, 2019.
- Kritee, K., Blum, J. D., Reinfelder, J. R. and Barkay, T.: Microbial stable isotope fractionation of mercury: A synthesis
  of present understanding and future directions, Chemical Geology, this issue, doi:10.1016/j.chemgeo.2012.08.017,
  2012.
- Kwon, S. Y., Blum, J. D., Yin, R., Tsui, M. T.-K., Yang, Y. H. and Choi, J. W.: Mercury stable isotopes for monitoring
  the effectiveness of the Minamata Convention on Mercury, Earth-Science Reviews, 203, 103111,
  doi:https://doi.org/10.1016/j.earscirev.2020.103111, 2020.
- Lyman, S. N. and Jaffe, D. A.: Formation and fate of oxidized mercury in the upper troposphere and lower stratosphere, Nature Geoscience, 5(2), 114–117, doi:10.1038/ngeo1353, 2011.
- Lynam, M. M. and Keeler, G. J.: Comparison of methods for particulate phase mercury analysis: sampling and analysis, Analytical and Bioanalytical Chemistry, 374(6), 1009–1014, doi:10.1007/s00216-002-1584-4, 2002.
- Marusczak, N., Sonke, J. E., Fu, X. and Jiskra, M.: Tropospheric GOM at the Pic du Midi Observatory-Correcting
  Bias in Denuder Based Observations, Environmental Science & Technology, 51(2), 863–869,
  doi:10.1021/acs.est.6b04999, 2017.
- Motta, L. C., Blum, J. D., Johnson, M. W., Umhau, B. P., Popp, B. N., Washburn, S. J., Drazen, J. C., Benitez-Nelson,
  C. R., Hannides, C. C. S., Close, H. G. and Lamborg, C. H.: Mercury Cycling in the North Pacific Subtropical Gyre
  as Revealed by Mercury Stable Isotope Ratios, Global Biogeochemical Cycles, 33(6), 777–794,
  doi:10.1029/2018GB006057, 2019.
- Motta, L. C., Kritee, K., Blum, J. D., Tsz-Ki Tsui, M. and Reinfelder, J. R.: Mercury Isotope Fractionation during the
  Photochemical Reduction of Hg(II) Coordinated with Organic Ligands., The journal of physical chemistry. A,
  doi:10.1021/acs.jpca.9b06308, 2020a.
- 426 Motta, L. C., Chien, A. D., Rask, A. E. and Zimmerman, P. M.: Mercury Magnetic Isotope Effect: A Plausible

- 427 Photochemical Mechanism, J. Phys. Chem. A, 124(19), 3711–3719, doi:10.1021/acs.jpca.0c00661, 2020b.
- 428 Obrist, D., Agnan, Y., Jiskra, M., Olson, C. L., Colegrove, D. P., Hueber, J., Moore, C. W., Sonke, J. E. and Helmig,
- 429 D.: Tundra uptake of atmospheric elemental mercury drives Arctic mercury pollution, Nature, 547(7662), 201-+,
- 430 doi:10.1038/nature22997, 2017.
- Saiz-Lopez, A., Sitkiewicz, S. P., Roca-Sanjuán, D., Oliva-Enrich, J. M., Dávalos, J. Z., Notario, R., Jiskra, M., Xu,
  Y., Wang, F., Thackray, C. P., Sunderland, E. M., Jacob, D. J., Travnikov, O., Cuevas, C. A., Acuña, A. U., Rivero,
  D., Plane, J. M. C., Kinnison, D. E. and Sonke, J. E.: Photoreduction of gaseous oxidized mercury changes global
  atmospheric mercury speciation, transport and deposition, Nature Communications, 9(1), 4796, doi:10.1038/s41467018-07075-3, 2018.
- Saiz-Lopez, A., Acuña, A. U., Trabelsi, T., Carmona-García, J., Dávalos, J. Z., Rivero, D., Cuevas, C. A., Kinnison,
  D. E., Sitkiewicz, S. P., Roca-Sanjuán, D. and Francisco, J. S.: Gas-Phase Photolysis of Hg(I) Radical Species: A
  New Atmospheric Mercury Reduction Process, J. Am. Chem. Soc., 141(22), 8698–8702, doi:10.1021/jacs.9b02890,
  2019.
- Schauble, E. A.: Role of nuclear volume in driving equilibrium isotope fractionation of mercury, thallium, and other
  very heavy elements., Geochimica et Cosmochimca Acta, 71, 2170–2189, 2007.
- Seigneur, C. and Lohman, K.: Effect of bromine chemistry on the atmospheric mercury cycle, Journal of Geophysical
   Research Atmosphere, 113, D23309, 2008.
- Seigneur, C., Vijayaraghavan, K., Lohman, K., Karamchandani, P. and Scott, C.: Global Source Attribution for
  Mercury Deposition in the United States, Environmental Science and Technology, 38, 555–569, 2004.
- 446 Shah, V., Jaegle, L., Gratz, L. E., Ambrose, J. L., Jaffe, D. A., Selin, N. E., Song, S., Campos, T. L., Flocke, F. M.,
- 447 Reeves, M., Stechman, D., Stell, M., Festa, J., Stutz, J., Weinheimer, A. J., Knapp, D. J., Montzka, D. D., Tyndall, G.
- 448 S., Apel, E. C., Hornbrook, R. S., Hills, A. J., Riemer, D. D., Blake, N. J., Cantrell, C. A. and Mauldin, R. L.: Origin
- 449 of oxidized mercury in the summertime free troposphere over the southeastern US, Atmospheric Chemistry and
- 450 Physics, 16(3), 1511–1530, doi:10.5194/acp-16-1511-2016, 2016.
- Sherman, L. S., Blum, J. D., Keeler, G. J., Demers, J. D. and Dvonch, J. T.: Investigation of Local Mercury Deposition
  from a Coal-Fired Power Plant Using Mercury Isotopes, Environmental Science & Technology, 46(1), 382–390,
  doi:10.1021/es202793c, 2012.
- Steffen, A., Schroeder, W., Macdonald, R., Poissant, L. and Konoplev, A.: Mercury in the Arctic atmosphere: An
  analysis of eight years of measurements of GEM at Alert (Canada) and a comparison with observations at Amderma
  (Russia) and Kuujjuarapik (Canada), Science of the Total Environment, 342(1–3), 185–198,
  doi:10.1016/j.scitotenv.2004.12.048, 2005.
- Streets, D. G., Horowitz, H. M., Lu, Z., Levin, L., Thackray, C. P. and Sunderland, E. M.: Global and regional trends
  in mercury emissions and concentrations, 2010-2015, Atmospheric Environment, 201, 417–427,
  doi:10.1016/j.atmosenv.2018.12.031, 2019.
- Sun, G., Sommar, J., Feng, X., Lin, C.-J., Ge, M., Wang, W., Yin, R., Fu, X. and Shang, L.: Mass-Dependent and Independent Fractionation of Mercury Isotope during Gas-Phase Oxidation of Elemental Mercury Vapor by Atomic

- 463 Cl and Br, Environ. Sci. Technol., 50(17), 9232–9241, doi:10.1021/acs.est.6b01668, 2016.
- Sun, R., Enrico, M., Heimbürger, L.-E., Scott, C. and Sonke, J. E.: A double-stage tube furnace-acid-trapping protocol
  for the pre-concentration of mercury from solid samples for isotopic analysis, Analytical and Bioanalytical Chemistry,
  466 405(21), 6771–6781, doi:10.1007/s00216-013-7152-2, 2013.
- Sun, R., Jiskra, M., Amos, H. M., Zhang, Y., Sunderland, E. M. and Sonke, J. E.: Modelling the mercury stable isotope
  distribution of Earth surface reservoirs: Implications for global Hg cycling, Geochimica et Cosmochimica Acta, 246,
  156–173, doi:https://doi.org/10.1016/j.gca.2018.11.036, 2019.
- 470 Swartzendruber, P. C., Jaffe, J. A., Prestbo, E. M., Weiss-Penzias, P., Selin, N. E., Park, R., Jacob, D. J., Strode, S.
- and Jaegle, L.: Observations of reactive gaseous mercury in the free troposphere at the Mount Bachelor Observatory,
  Journal of Geophysical Research, 111, D24301, 2006.
- Timonen, H., Ambrose, J. L. and Jaffe, D. A.: Oxidation of elemental Hg in anthropogenic and marine airmasses,
  Atmospheric Chemistry and Physics, 13(5), 2827–2836, doi:10.5194/acp-13-2827-2013, 2013.
- Turro, N. J.: Influence of nuclear spin on chemical, reactions: Magnetic isotope and magnetic field effects (A Review),
  Proceedings of the National Academy of Science USA, 80, 609–621, 1983.
- Wang, F., Saiz-Lopez, A., Mahajan, A. S., Gomez Martin, J. C., Armstrong, D., Lemes, M., Hay, T. and PradosRoman, C.: Enhanced production of oxidised mercury over the tropical Pacific Ocean: a key missing oxidation
- 479 pathway, Atmospheric Chemistry and Physics, 14(3), 1323–1335, doi:10.5194/acp-14-1323-2014, 2014.
- Wang, Z., Chen, J., Feng, X., Hintelmann, H., Yuan, S., Cai, H., Huang, Q., Wang, S. and Wang, F.: Mass-dependent
  and mass-independent fractionation of mercury isotopes in precipitation from Guiyang, SW China, Comptes Rendus
  Geoscience, 347(7), 358–367, doi:https://doi.org/10.1016/j.crte.2015.02.006, 2015.
- 483 Weiss-Penzias, P., Amos, H. M., Selin, N. E., Gustin, M. S., Jaffe, D. A., Obrist, D., Sheu, G. R. and Giang, A.: Use
- 484 of a global model to understand speciated atmospheric mercury observations at five high-elevation sites, Atmospheric
- 485 Chemistry and Physics, 15(3), 1161–1173, doi:10.5194/acp-15-1161-2015, 2015.
- Wiederhold, J. G., Cramer, C. J., Daniel, K., Infante, I., Bourdon, B. and Kretzschmar, R.: Equilibrium mercury
  isotope fractionation between dissolved Hg(II) species and thiol-bound Hg, Environmental Science and Technology,
  44, 4191–4197, 2010.
- Yang, X., Jiskra, M. and Sonke, J. E.: Experimental rainwater divalent mercury speciation and photoreduction rates
  in the presence of halides and organic carbon, Science of The Total Environment, 697, 133821,
  doi:https://doi.org/10.1016/j.scitotenv.2019.133821, 2019.
- Zheng, W. and Hintelmann, H.: Mercury isotope fractionation during photoreduction in natural water is controlled
  by its Hg/DOC ratio., Geochimica et Cosmochimca Acta, 73, 6704–6715, 2009.
- Zheng, W. and Hintelmann, H.: Isotope Fractionation of Mercury during Its Photochemical Reduction by LowMolecular-Weight Organic Compounds, Journal of Physical Chemistry A, 114, 4246–4253, 2010.
- 496

- Ariya, P. A., Amyot, M., Dastoor, A., Deeds, D., Feinberg, A., Kos, G., Poulain, A., Ryjkov, A., Semeniuk, K., Subir,
  M. and Toyota, K.: Mercury Physicochemical and Biogeochemical Transformation in the Atmosphere and at
  Atmospheric Interfaces: A Review and Future Directions, Chemical Reviews, 115(10), 3760–3802,
  doi:10.1021/cr500667e, 2015.
- Bagnato, E., Tamburello, G., Avard, G., Martinez-Cruz, M., Enrico, M., Fu, X., Sprovieri, M. and Sonke, J. E.:
  Mercury fluxes from volcanic and geothermal sources: an update., Geological Society, London, Special Publications,
  doi 10.1144/SP410.2, 2014.
- 504 Berg, T., Pfaffhuber, K. A., Cole, A. S., Engelsen, O. and Steffen, A.: Ten-year trends in atmospheric mercury 505 concentrations, meteorological effects and climate variables at Zeppelin, Ny-Ålesund, Atmospheric Chemistry and 506 Physics, 13(13), 6575–6586, doi:10.5194/acp-13-6575-2013, 2013.
- Bergquist, B. A. and Blum, J. D.: Mass-Dependent and -Independent Fractionation of Hg Isotopes by Photoreduction
   in Aquatic Systems., Science, 318, 417–420, 2007.
- Blum, J. D. and Bergquist, B. A.: Reporting of variations in the natural isotopic composition of mercury., Analytical
  and Bioanalytical Chemistry, 388, 353–359, 2007.
- Blum, J. D., Sherman, L. S. and Johnson, M. W.: Mercury Isotopes in Earth and Environmental Sciences, in Annual
  Review of Earth and Planetary Sciences, Vol 42, vol. 42, edited by R. Jeanloz, pp. 249–269., 2014.
- Buchachenko, A. L., Ivanov, V. L., Roznyatovskii, V. A., Artamkina, G. A., Vorob'ev, A. K. and Ustynyuk, Y. A.:
  Magnetic isotope effect for mercury nuclei in photolysis of Bis(p-trifluoromethylbenzyl)mercury., Doklady Physical
  Chemistry, 413, 39–41, 2007.
- Chen, J., Hintelmann, H., Feng, X. and Dimock, B.: Unusual fractionation of both odd and even mercury isotopes in
  precipitation from Peterborough, ON, Canada, Geochimica et Cosmochimica Acta, 90, 33–46,
  doi:10.1016/j.gca.2012.05.005, 2012.
- 519 Demers, J. D., Blum, J. D. and Zak, D. R.: Mercury isotopes in a forested ecosystem: Implications for air-surface 520 exchange dynamics and the global mercury cycle, Global Biogeochemical Cycles, 27(1), 222–238, 521 doi:10.1002/gbc.20021, 2013.
- Demoz, B., Collett, J., J. L. and Daube, J., B. C.: On the Caltech Active Strand Cloudwater Collectors., Atmospheric
   Research, 41, 47–62, 1996.
- Dibble, T. S., Tetu, H. L., Jiao, Y., Thackray, C. P. and Jacob, D. J.: Modeling the OH-Initiated Oxidation of Mercury
  in the Global Atmosphere without Violating Physical Laws, Journal of Physical Chemistry A, 124(2), 444–453,
  doi:10.1021/acs.jpca.9b10121, 2020.
- 527 Donohoue, D. L., Bauer, D., Cossairt, B. and Hynes, A.: Temperature and pressure dependent rate coefficients for the 528 reaction of Hg with Br and the reaction of Br with Br: A pulsed laser photolysis-pulsed laser induced fluorescence 529 study, Journal of Physical Chemistry A, 110, 6623–6632, 2006.
- Enrico, M., Le Roux, G., Marusczak, N., Heimbürger, L.-E., Claustres, A., Fu, X., Sun, R. and Sonke, J. E.:
  Atmospheric mercury transfer to peat bogs dominated by gaseous elemental mercury dry deposition., Environmental

- 532 Science & Technology, doi:10.1021/acs.est.5b06058, 2016.
- Fitzgerald, W. F. and Lamborg, C. H.: Geochemistry of Mercury in the Environment, in Treatise on Geochemistry,
  vol. 9, edited by E. D. Holland and K. K. Turekian, pp. 107–148, Elsevier-Pergamon, Oxford., 2004.
- Fu, X., Heimburger, L. E. and Sonke, J. E.: Collection of atmospheric gaseous mercury for stable isotope analysis
   using iodine- and chlorine-impregnated activated carbon traps, Journal of Analytical Atomic Spectrometry, DOI:
   10.1039/C3JA50356A, 2014.
- Fu, X. W., Marusczak, N., Heimbürger, L. E., Sauvage, B., Gheusi, F. and Sonke, J. E.: Atmospheric mercury
  speciation dynamics at the high-altitude Pic du Midi Observatory, southern France, Atmospheric Chemistry and
  Physics, 16, 5623–5639, doi:10.5194/acp-16-5623-2016, 2016a.
- Fu, X. W., Marusczak, N., Wang, X., Gheusi, F. and Sonke, J. E.: The isotopic composition of total gaseous mercury
  in the free troposphere of the Pic du Midi Observatory (2877 m a.s.l, France), Environmental Science & Technology,
  submitted, 2016b.
- 544 Gheusi, F., Ravetta, F., Delbarre, H., Tsamalis, C., Chevalier-Rosso, A., Leroy, C., Augustin, P., Delmas, R., Ancellet,
- 545 G., Athier, G., Bouchou, P., Campistron, B., Cousin, J. M., Fourmentin, M. and Meyerfeld, Y.: Pic 2005, a field 546 campaign to investigate low-tropospheric ozone variability in the Pyrenees, Atmospheric Research, 101(3), 640–665,
- 547 doi:10.1016/j.atmosres.2011.04.014, 2011.
- Gratz, L., Keeler, G., Blum, J. and Sherman, L. S.: Isotopic composition and fractionation of mercury in Great Lakes
   precipitation and ambient air, Environmental Science and Technology, 44, 7764–7770, 2010.
- Gratz, L. E., Ambrose, J. L., Jaffe, D. A., Shah, V., Jaegle, L., Stutz, J., Festa, J., Spolaor, M., Tsai, C., Selin, N. E.,
  Song, S., Zhou, X., Weinheimer, A. J., Knapp, D. J., Montzka, D. D., Flocke, F. M., Campos, T. L., Apel, E.,
  Hornbrook, R., Blake, N. J., Hall, S., Tyndall, G. S., Reeves, M., Stechman, D. and Stell, M.: Oxidation of mercury
  by bromine in the subtropical Pacific free troposphere, Geophysical Research Letters, 42(23),
  doi:10.1002/2015gl066645, 2015.
- 555 Grissom, C. B.: Magnetic field effects in biology: A survey of possible mechanisms with emphasis on radical-pair 556 recombination, Chemical Reviews, 95, 3–24, 1995.
- Holmes, C. D., Jacob, D. J., Corbitt, E. S., Mao, J., Yang, X., Talbot, R. and Slemr, F.: Global atmospheric model for
  mercury including oxidation by bromine atoms, Atmospheric Chemistry and Physics, 10, 12037–12057, 2010.
- 559 Horowitz, H. M., Jacob, D. J., Zhang, Y., Dibble, T. S., Slemr, F., Amos, H. M., Schmidt, J. A., Corbitt, E. S., Marais,
- 560 E. A. and Sunderland, E. M.: A new mechanism for atmospheric mercury redox chemistry: implications for the global
- 561 mercury budget, Atmospheric Chemistry and Physics, 17(10), 6353–6371, doi:10.5194/acp-17-6353-2017, 2017.
- Hulin, M., Gheusi, F., Lothon, M., Pont, V., Lohou, F., Ramonet, M., Delmotte, M., Derrien, S., Athier, G., Meyerfeld,
  Y.: Observations of plain-mountain circulation in the Pyrenees : detection methods and impact on atmospheric
  composition measured at a mountain top, J. Applied. Meteor. Climatol, 2019, 58, 717–740,
  <a href="https://doi.org/10.1175/JAMC-D-17-0268.1">https://doi.org/10.1175/JAMC-D-17-0268.1</a>.
- Jaffe, D. A., Lyman, S., Amos, H. M., Gustin, M. S., Huang, J., Selin, N. E., Levin, L., ter Schure, A., Mason, R. P.,

- 567 Talbot, R., Rutter, A., Finley, B., Jaegle, L., Shah, V., McClure, C., Arnbrose, J., Gratz, L., Lindberg, S., Weiss-
- Penzias, P., Sheu, G.-R., Feddersen, D., Horvat, M., Dastoor, A., Hynes, A. J., Mao, H., Sonke, J. E., Slemr, F., Fisher,
  J. A., Ebinghaus, R., Zhang, Y. and Edwards, G.: Progress on Understanding Atmospheric Mercury Hampered by
- 570 Uncertain Measurements, Environmental Science & Technology, 48(13), 7204–7206, doi:10.1021/es5026432, 2014.
- 570 Oncertain weasurements, Environmental Science & Technology, +0(15), 720+7200, uoi.1021/053020+52, 2014.
- Jiskra, M., Wiederhold, J. G., Skyllberg, U., Kronberg, R.-M., Hajdas, I. and Kretzschmar, R.: Mercury Deposition
   and Re-emission Pathways in Boreal Forest Soils Investigated with Hg Isotope Signatures, Environmental Science
- 573 & Technology, 49(12), 7188–7196, doi:10.1021/acs.est.5b00742, 2015.
- Jiskra, M., Sonke, J. E., Agnan, Y., Helmig, D. and Obrist, D.: Insights from mercury stable isotopes on terrestrialatmosphere exchange of Hg(0) in the Arctic tundra, Biogeosciences, 16(20), 4051–4064, doi:10.5194/bg-16-40512019, 2019.
- Kritee, K., Blum, J. D., Reinfelder, J. R. and Barkay, T.: Microbial stable isotope fractionation of mercury: A synthesis
  of present understanding and future directions, Chemical Geology, this issue, doi:10.1016/j.chemgeo.2012.08.017,
  2012.
- Kwon, S. Y., Blum, J. D., Yin, R., Tsui, M. T.-K., Yang, Y. H. and Choi, J. W.: Mercury stable isotopes for monitoring
  the effectiveness of the Minamata Convention on Mercury, Earth-Science Reviews, 203, 103111,
  doi:https://doi.org/10.1016/j.earscirev.2020.103111, 2020.
- 583 Lyman, S. N. and Jaffe, D. A.: Formation and fate of oxidized mercury in the upper troposphere and lower 584 stratosphere, Nature Geoscience, 5(2), 114–117, doi:10.1038/ngeo1353, 2011.
- Lynam, M. M. and Keeler, G. J.: Comparison of methods for particulate phase mercury analysis: sampling and analysis, Analytical and Bioanalytical Chemistry, 374(6), 1009–1014, doi:10.1007/s00216-002-1584-4, 2002.
- Marusczak, N., Sonke, J. E., Fu, X. and Jiskra, M.: Tropospheric GOM at the Pic du Midi Observatory-Correcting
  Bias in Denuder Based Observations, Environmental Science & Technology, 51(2), 863–869,
  doi:10.1021/acs.est.6b04999, 2017.
- Motta, L. C., Blum, J. D., Johnson, M. W., Umhau, B. P., Popp, B. N., Washburn, S. J., Drazen, J. C., Benitez-Nelson,
  C. R., Hannides, C. C. S., Close, H. G. and Lamborg, C. H.: Mercury Cycling in the North Pacific Subtropical Gyre
  as Revealed by Mercury Stable Isotope Ratios, Global Biogeochemical Cycles, 33(6), 777–794,
  doi:10.1029/2018GB006057, 2019.
- Motta, L. C., Kritee, K., Blum, J. D., Tsz-Ki Tsui, M. and Reinfelder, J. R.: Mercury Isotope Fractionation during the
   Photochemical Reduction of Hg(II) Coordinated with Organic Ligands., The journal of physical chemistry. A,
   doi:10.1021/acs.jpca.9b06308, 2020a.
- Motta, L. C., Chien, A. D., Rask, A. E. and Zimmerman, P. M.: Mercury Magnetic Isotope Effect: A Plausible
  Photochemical Mechanism, J. Phys. Chem. A, 124(19), 3711–3719, doi:10.1021/acs.jpca.0c00661, 2020b.
- 599 Obrist, D., Agnan, Y., Jiskra, M., Olson, C. L., Colegrove, D. P., Hueber, J., Moore, C. W., Sonke, J. E. and Helmig,
- D.: Tundra uptake of atmospheric elemental mercury drives Arctic mercury pollution, Nature, 547(7662), 201-+,
  doi:10.1038/nature22997, 2017.

- 602 Saiz-Lopez, A., Sitkiewicz, S. P., Roca-Sanjuán, D., Oliva-Enrich, J. M., Dávalos, J. Z., Notario, R., Jiskra, M., Xu,
- Y., Wang, F., Thackray, C. P., Sunderland, E. M., Jacob, D. J., Travnikov, O., Cuevas, C. A., Acuña, A. U., Rivero,
  D., Plane, J. M. C., Kinnison, D. E. and Sonke, J. E.: Photoreduction of gaseous oxidized mercury changes global
  atmospheric mercury speciation, transport and deposition, Nature Communications, 9(1), 4796, doi:10.1038/s41467018-07075-3, 2018.
- 607 Saiz-Lopez, A., Acuña, A. U., Trabelsi, T., Carmona-García, J., Dávalos, J. Z., Rivero, D., Cuevas, C. A., Kinnison,
- 608 D. E., Sitkiewicz, S. P., Roca-Sanjuán, D. and Francisco, J. S.: Gas-Phase Photolysis of Hg(I) Radical Species: A
- 609 New Atmospheric Mercury Reduction Process, J. Am. Chem. Soc., 141(22), 8698–8702, doi:10.1021/jacs.9b02890,
- 610 2019.
- Schauble, E. A.: Role of nuclear volume in driving equilibrium isotope fractionation of mercury, thallium, and other
  very heavy elements., Geochimica et Cosmochimca Acta, 71, 2170–2189, 2007.
- 613 Seigneur, C. and Lohman, K.: Effect of bromine chemistry on the atmospheric mercury cycle, Journal of Geophysical
- 614 Research Atmosphere, 113, D23309, 2008.
- Seigneur, C., Vijayaraghavan, K., Lohman, K., Karamchandani, P. and Scott, C.: Global Source Attribution for
  Mercury Deposition in the United States, Environmental Science and Technology, 38, 555–569, 2004.
- 617 Shah, V., Jaegle, L., Gratz, L. E., Ambrose, J. L., Jaffe, D. A., Selin, N. E., Song, S., Campos, T. L., Flocke, F. M.,
- 618 Reeves, M., Stechman, D., Stell, M., Festa, J., Stutz, J., Weinheimer, A. J., Knapp, D. J., Montzka, D. D., Tyndall, G.
- 619 S., Apel, E. C., Hornbrook, R. S., Hills, A. J., Riemer, D. D., Blake, N. J., Cantrell, C. A. and Mauldin, R. L.: Origin
- 620 of oxidized mercury in the summertime free troposphere over the southeastern US, Atmospheric Chemistry and
- 621 Physics, 16(3), 1511–1530, doi:10.5194/acp-16-1511-2016, 2016.
- Sherman, L. S., Blum, J. D., Keeler, G. J., Demers, J. D. and Dvonch, J. T.: Investigation of Local Mercury Deposition
  from a Coal-Fired Power Plant Using Mercury Isotopes, Environmental Science & Technology, 46(1), 382–390,
  doi:10.1021/es202793c, 2012.
- Steffen, A., Schroeder, W., Macdonald, R., Poissant, L. and Konoplev, A.: Mercury in the Arctic atmosphere: An
  analysis of eight years of measurements of GEM at Alert (Canada) and a comparison with observations at Amderma
  (Russia) and Kuujjuarapik (Canada), Science of the Total Environment, 342(1–3), 185–198,
  doi:10.1016/j.scitotenv.2004.12.048, 2005.
- Streets, D. G., Horowitz, H. M., Lu, Z., Levin, L., Thackray, C. P. and Sunderland, E. M.: Global and regional trends
  in mercury emissions and concentrations, 2010-2015, Atmospheric Environment, 201, 417–427,
  doi:10.1016/j.atmosenv.2018.12.031, 2019.
- Sun, G., Sommar, J., Feng, X., Lin, C.-J., Ge, M., Wang, W., Yin, R., Fu, X. and Shang, L.: Mass-Dependent and
  Independent Fractionation of Mercury Isotope during Gas-Phase Oxidation of Elemental Mercury Vapor by Atomic
  Cl and Br, Environ. Sci. Technol., 50(17), 9232–9241, doi:10.1021/acs.est.6b01668, 2016.
- Sun, R., Enrico, M., Heimbürger, L.-E., Scott, C. and Sonke, J. E.: A double-stage tube furnace-acid-trapping protocol
   for the pre-concentration of mercury from solid samples for isotopic analysis, Analytical and Bioanalytical Chemistry,
- 637 405(21), 6771–6781, doi:10.1007/s00216-013-7152-2, 2013.

- 638 Sun, R., Jiskra, M., Amos, H. M., Zhang, Y., Sunderland, E. M. and Sonke, J. E.: Modelling the mercury stable isotope
- distribution of Earth surface reservoirs: Implications for global Hg cycling, Geochimica et Cosmochimica Acta, 246,
- 640 156–173, doi:https://doi.org/10.1016/j.gca.2018.11.036, 2019.
- 641 Swartzendruber, P. C., Jaffe, J. A., Prestbo, E. M., Weiss-Penzias, P., Selin, N. E., Park, R., Jacob, D. J., Strode, S.
- and Jaegle, L.: Observations of reactive gaseous mercury in the free troposphere at the Mount Bachelor Observatory,
   Journal of Geophysical Research, 111, D24301, 2006.
- Timonen, H., Ambrose, J. L. and Jaffe, D. A.: Oxidation of elemental Hg in anthropogenic and marine airmasses,
  Atmospheric Chemistry and Physics, 13(5), 2827–2836, doi:10.5194/acp-13-2827-2013, 2013.
- Turro, N. J.: Influence of nuclear spin on chemical, reactions: Magnetic isotope and magnetic field effects (A Review),
  Proceedings of the National Academy of Science USA, 80, 609–621, 1983.
- Wang, F., Saiz-Lopez, A., Mahajan, A. S., Gomez Martin, J. C., Armstrong, D., Lemes, M., Hay, T. and PradosRoman, C.: Enhanced production of oxidised mercury over the tropical Pacific Ocean: a key missing oxidation
  pathway, Atmospheric Chemistry and Physics, 14(3), 1323–1335, doi:10.5194/acp-14-1323-2014, 2014.
- Wang, Z., Chen, J., Feng, X., Hintelmann, H., Yuan, S., Cai, H., Huang, Q., Wang, S. and Wang, F.: Mass-dependent
- and mass-independent fractionation of mercury isotopes in precipitation from Guiyang, SW China, Comptes Rendus
- 653 Geoscience, 347(7), 358–367, doi:https://doi.org/10.1016/j.crte.2015.02.006, 2015.
- Weiss-Penzias, P., Amos, H. M., Selin, N. E., Gustin, M. S., Jaffe, D. A., Obrist, D., Sheu, G. R. and Giang, A.: Use
- of a global model to understand speciated atmospheric mercury observations at five high-elevation sites, Atmospheric
   Chemistry and Physics, 15(3), 1161–1173, doi:10.5194/acp-15-1161-2015, 2015.
- Wiederhold, J. G., Cramer, C. J., Daniel, K., Infante, I., Bourdon, B. and Kretzschmar, R.: Equilibrium mercury
  isotope fractionation between dissolved Hg(II) species and thiol-bound Hg, Environmental Science and Technology,
  44, 4191–4197, 2010.
- Yang, X., Jiskra, M. and Sonke, J. E.: Experimental rainwater divalent mercury speciation and photoreduction rates
  in the presence of halides and organic carbon, Science of The Total Environment, 697, 133821,
  doi:https://doi.org/10.1016/j.scitotenv.2019.133821, 2019.
- Zheng, W. and Hintelmann, H.: Mercury isotope fractionation during photoreduction in natural water is controlled
  by its Hg/DOC ratio., Geochimica et Cosmochimea Acta, 73, 6704–6715, 2009.
- Zheng, W. and Hintelmann, H.: Isotope Fractionation of Mercury during Its Photochemical Reduction by Low Molecular-Weight Organic Compounds, Journal of Physical Chemistry A, 114, 4246–4253, 2010.
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- 671 Table 1. Mean  $\Delta^{200}$ Hg signatures for Hg emissions and atmospheric reservoirs. Mean and 1σ for
- published rainfall data represents the mean of means of remote site data points for each study in order to avoid
- 673 site type and sample number bias. Global GEM emission  $\Delta^{200}$ Hg was calculated from the listed Hg fluxes and
- 674 modeled  $\Delta^{200}$ Hg. The combined uncertainty on calculated atmospheric GEM and RM fractions was
- 675 propagated from uncertainties on observational data for emissions and RM and GEM  $\Delta^{200}$ Hg.

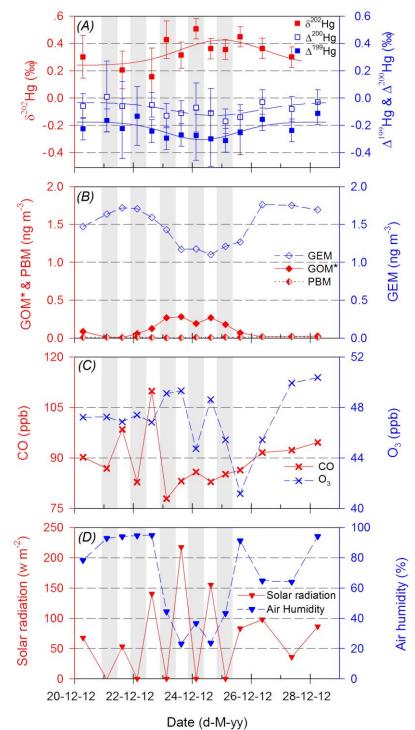
Sample type	Hg flux	Δ <sup>200</sup> Hg	1σ	
	Mg y <sup>-1</sup>	‰	‰	
Pic du Midi RM		0.15	0.12	This study
Rainfall Hg		0.13	0.06	(Blum et al., 2014; Demers et al., 2013; Enrico et al.,
				2016; Gratz et al., 2010; Motta et al., 2019; Obrist et
				al., 2017; Sherman et al., 2012; Wang et al., 2015)
Pic du Midi GEM		-0.03	0.02	This study, (Fu et al., 2016b)
Ocean GEM emission	4600	0.03		(Horowitz et al., 2017; Sun et al., 2019)
Anthropogenic GEM	1470	0		(Horowitz et al., 2017; Sun et al., 2019)
Soil GEM emission	1420	-0.02		(Horowitz et al., 2017; Sun et al., 2019)
Volcanic GEM	250	0		(Horowitz et al., 2017; Sun et al., 2019)
emission				
Global GEM emission	7740	0.02	0.02	(Horowitz et al., 2017; Sun et al., 2019)
f(RM)		0.30 <sup>a</sup>	0.17	
f(GEM)		0.70ª	0.17	

<sup>a</sup> calculated from mass balance:  $\Delta^{200}$ Hg(global GEM emitted) = f(RM) ×  $\Delta^{200}$ Hg(rainfall) + f(GEM) ×

677  $\Delta^{200}$ Hg(PDM GEM), and f(RM) + f(GEM) = 1

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Figure 1. Hg speciation and isotope variability during a free tropospheric air mass intrusion. From 20 to 28 Dec 2012, high frequency (12h, 24h) gaseous elemental Hg<sup>0</sup> (GEM) isotope signatures  $\delta^{202}$ Hg,  $\Delta^{199}$ Hg,  $\Delta^{200}$ Hg (A), CO and O<sub>3</sub> concentrations (B), atmospheric GOM, PBM and GEM concentrations (C), and solar radiation and relative air humidity (D) showed a 48h long intrusion of a free tropospheric air mass, high in GOM, and low in GEM and humidity.



# 687 Figure 2. Air mass back-trajectories of the free tropospheric air mass intrusion. 7-day backward

 $figure{1}$  trajectories of the high frequency (12h, 24h) gaseous elemental Hg<sup>0</sup> (GEM) isotope samples obtained from 20

- 689 to 28 Dec 2012: Travelling height (A, indicated by air pressure), air mass humidity (B), solar radiation (C),
- and air temperature (D). Date and time labels are mid-points of 12h and 24h Hg isotope sampling windows.

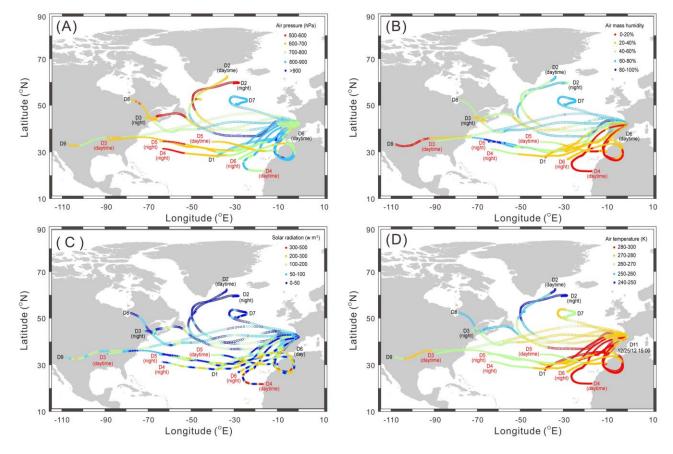


Figure 3. Atmospheric Hg stable isotope variation at the Pic du Midi. Hg mass dependent ( $\delta^{202}$ Hg) and mass independent ( $\Delta^{199}$ Hg,  $\Delta^{200}$ Hg,  $\Delta^{201}$ Hg,  $\Delta^{204}$ Hg) isotope signatures in reactive Hg (RM=GOM+PBM), cloud water, 12h/24h GEM from this study, and previously published GEM (Fu et al., 2014) and rainfall (Enrico et al., 2016).

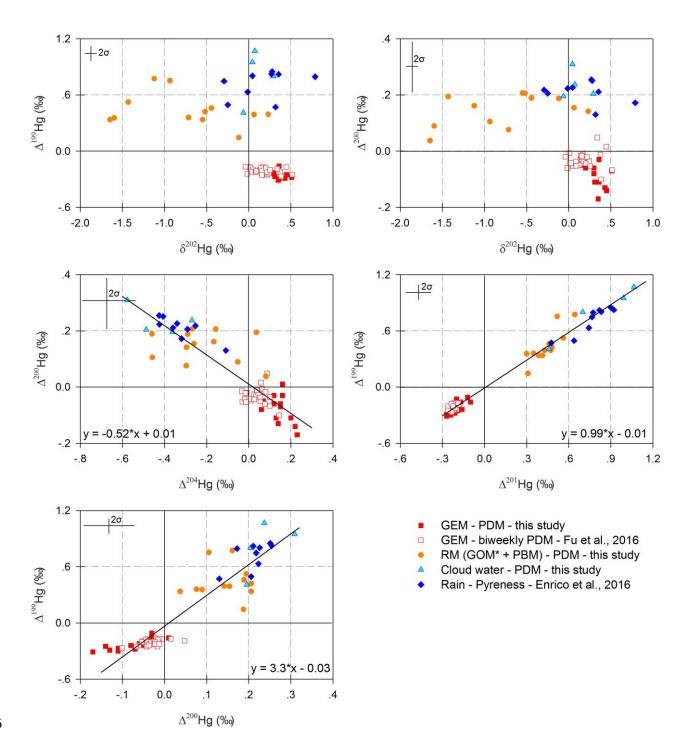
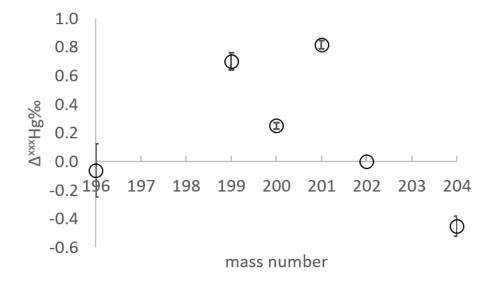
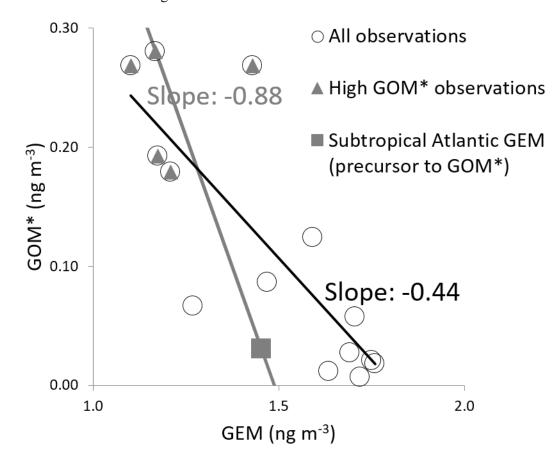




Figure 4. Δ<sup>196</sup>Hg measurement of rainfall. A 20.5L rainfall sample containing 6.0 ng L<sup>-1</sup> of Hg was preconcentrated 1200x to deliver a 0.015V signal on <sup>196</sup>Hg. The resulting Δ<sup>196</sup>Hg signature is -0.06 ± 0.19‰ (2σ, n=2) suggesting that nuclear self-shielding is not the underlying cause of atmospheric even-Hg MIF. A Δ<sup>202</sup>Hg of 0 is indicated in the graph, since the Δ<sup>xxx</sup>Hg (where xxx are Hg isotopes masses 196 to 204) are calculated relative to δ<sup>202/198</sup>Hg ratios, themselves anchored to the <sup>198</sup>Hg reference isotope.



705 Figure 5. Hg oxidation mass balance in free tropospheric air at the Pic du Midi. Relationship between gaseous elemental Hg<sup>0</sup> (GEM, ng m<sup>-3</sup>) and gaseous oxidized Hg<sup>II</sup> (GOM\*, ng m<sup>-3</sup>) observed during the nine day 706 sampling campaign at PDM. Black circles and grey diamonds are Tekran GEM and GOM\* data (Extended 707 708 Data), where grey diamonds represent the high GOM event only. A slope of -1 is indicative of Hg mass conservation during GEM oxidation to GOM in the observed air mass. The uncorrected GOM\*/GEM slope of 709 710 -0.44 is biased low due to latitudinal bias in the GEM precursor to GOM (see main text). Latitude corrected subtropical Atlantic Ocean GEM levels (grey square) indicate a slope of -0.88, suggesting that Hg mass has 711 712 been conserved in the high GOM air mass detected at PDM.



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Figure 6. Rayleigh diagrams for Hg isotope fractionation during Hg<sup>0</sup> net oxidation. This dataset is a 716 717 composite of GEM isotope composition (red diamonds) recorded in the Atlantic Ocean free tropospheric air mass intrusion at PDM in Dec 2012, and mean RM (=GOM\*+PBM ~ GOM\*) isotope composition (blue 718 triangles) at PDM recorded from June – October 2014. Isotope fractionation factors  $\varepsilon_{MDF(GOM-GEM)}^{\delta 202}$  of -1.1 ± 719  $0.2\%, \varepsilon_{MIF(GOM-GEM)}^{\Delta 199}$  of  $0.65 \pm 0.11\%, \varepsilon_{MIF(GOM-GEM)}^{\Delta 200}$  of  $0.22 \pm 0.05\%, \varepsilon_{MIF(GOM-GEM)}^{\Delta 201}$  of  $0.65 \pm 0.10\%$ , and 720  $\varepsilon_{MIF(GOM-GEM)}^{4204}$  of -0.39 ± 0.12‰ (all 1SE) were fitted by log-linear regression of Rayleigh equations for residual 721 GEM and cumulative GOM\*. All 12h and 24h observations (blue diamonds) over the nine day period were 722 723 included in the fitting. The starting isotopic composition (green square) is the mean annual GEM composition at PDM (Fu et al., 2016b). 724

