

1 **Mass independent fractionation of even and odd mercury isotopes during free tropospheric mercury**
2 **oxidation**

3

4 Xuewu Fu^{1,2}, Martin Jiskra^{1,3}, Xu Yang^{1,2}, Nicolas Maruszczak¹, Maxime Enrico^{1,§}, Jerome Chemleff¹, Lars-
5 Eric Heimbürger^{1,¶} François Gheusi⁴, Jeroen E. Sonke^{1,*}

6

7 ¹ Géosciences Environnement Toulouse, CNRS/IRD/CNRS/Université de Toulouse, France

8 ² State Key Laboratory of Environmental Geochemistry, Institute of Geochemistry, Chinese Academy of
9 Sciences, Guiyang, 550081, China.

10 ³ Environmental Geosciences, University of Basel, Switzerland

11 ⁴ Laboratoire d'Aérologie, CNRS/IRD/Université de Toulouse, France

12 [§] present address: Harvard John A. Paulson School of Engineering & Applied Sciences, Harvard University,
13 Cambridge, MA, USA

14 [¶] present address: Mediterranean Institute of Oceanography, Aix Marseille Université, CNRS/IRD, Université
15 de Toulon, IRD, Marseille, France

16

17 * Corresponding Author: jeroen.sonke@get.omp.eu

18

19

20

21

22

23

24

25

26

27 **Abstract.** Even mercury (Hg) isotope mass independent fractionation (MIF), observed in rainfall globally, is
28 used to quantify atmospheric Hg deposition pathways. The underlying reaction and MIF mechanism are
29 unknown however. Here we investigate the Hg isotope composition of free tropospheric gaseous elemental Hg⁰
30 and Hg^{II} forms. We find that gaseous oxidized Hg^{II} has positive $\Delta^{199}\text{Hg}$, $\Delta^{201}\text{Hg}$, $\Delta^{200}\text{Hg}$, and negative $\Delta^{204}\text{Hg}$
31 signature, similar to rainfall Hg^{II}, and we document rainfall Hg^{II} $\Delta^{196}\text{Hg}$ to be near-zero. Cloud water and rainfall
32 Hg^{II} show enhanced odd MIF compared to gaseous Hg^{II}, indicating in-cloud Hg^{II} photoreduction. Hg MIF
33 observations of free tropospheric Hg⁰ dynamics show how net Hg⁰ oxidation leads to opposite MIF in Hg⁰ and
34 Hg^{II}. A $\Delta^{200}\text{Hg}$ mass balance for Hg⁰ and Hg^{II} forms suggests that measurements and models underestimate the
35 tropospheric Hg^{II} pool.

36

37 **Introduction**

38 Modern anthropogenic Hg emissions (2400 Mg a⁻¹) outweigh natural Hg emissions by an order of magnitude
39 (Bagnato et al., 2014; Streets et al., 2019). Hg emissions are mostly in the form of gaseous elemental Hg (GEM),
40 which has a long atmospheric life-time against oxidation to divalent reactive Hg (RM) forms that include
41 gaseous oxidized Hg (GOM) and particulate bound Hg (PBM) (Amos et al., 2012). RM is rapidly deposited to
42 continental and marine ecosystems where microbial activity transforms inorganic Hg into the bioaccumulating
43 and toxic methylmercury form (Fitzgerald and Lamborg, 2004). Predicting the impact of anthropogenic Hg
44 emission scenarios on global Hg deposition requires detailed understanding of atmospheric Hg transformations
45 and of GEM oxidation in particular. Two decades of atmospheric Hg research using mountain-top and in-flight
46 observations, experimental and theoretical reaction kinetics, and coupled 3D atmosphere-ocean chemistry and
47 transport models, have given broad understanding of atmospheric Hg dynamics (Ariya et al., 2015; Dibble et
48 al., 2020; Donohoue et al., 2006; Holmes et al., 2010; Seigneur et al., 2004). Experimental, theoretical and
49 modeling studies suggest Hg oxidation initiated by bromine (Br) radicals to be faster than by ozone or OH
50 (Horowitz et al., 2017; Wang et al., 2014), yet major uncertainties in RM concentrations, oxidation and reduction
51 mechanisms and rates remain (Jaffe et al., 2014; Saiz-Lopez et al., 2018; Weiss-Penzias et al., 2015).

52 Hg stable isotope analysis of atmospheric Hg compounds may provide new information on Hg emission
53 sources and complex atmospheric redox cycling. During the last decade, significant advances have been made
54 in the measurement of Hg isotope compositions of biogeochemical reservoirs (Kwon et al., 2020). Laboratory
55 and field evidence suggests that processes controlling Hg emission such as reduction/oxidation,
56 adsorption/desorption, or volatilization can induce significant mass dependent Hg isotope fractionation (MDF,

57 $\delta^{202}\text{Hg}$ signature)(Jiskra et al., 2015; Kritee et al., 2012; Wiederhold et al., 2010). Large mass independent Hg
58 isotope fractionation (MIF) of odd Hg isotopes by the magnetic isotope effect has been observed in aqueous
59 photochemical Hg reduction experiments ($\Delta^{199}\text{Hg}$, $\Delta^{201}\text{Hg}$ signatures) (Bergquist and Blum, 2007; Zheng and
60 Hintelmann, 2009). Odd-MIF signatures imparted on inorganic Hg and methyl-Hg by photochemistry can be
61 traced into all Earth surface environments, i.e sediments, biota, atmosphere, soils etc. Unusual even isotope MIF
62 ($\Delta^{200}\text{Hg}$, $\Delta^{204}\text{Hg}$) has been observed in rainfall and snowfall globally (Chen et al., 2012; Demers et al., 2013;
63 Gratz et al., 2010), and has been speculated to originate in GEM photo-oxidation at and above the tropopause
64 (Chen et al., 2012). The exact even-MIF mechanism and even-MIF inducing Hg transformation(s) remain
65 essentially unknown.

66 In this study we investigate Hg isotope composition of GEM, RM, and cloud water at the high altitude Pic
67 du Midi Observatory (PDM, 2877m, France). The PDM regularly receives free tropospheric air masses that are
68 enriched in RM and depleted in GEM (Fu et al., 2016a). High frequency, 12h and 24h GEM isotope observations
69 of a free tropospheric air mass rich in GEM and depleted in GEM are investigated to derive MDF and MIF
70 factors for net oxidation. Photoreduction experiments on rainfall are made to constrain atmospheric aqueous
71 MDF and MIF factors. Finally, we investigated the even-MIF on the low abundance ^{196}Hg isotope to help
72 understand potential even-MIF mechanisms.

73

74 **Materials and methods**

75 **Site description and ancillary parameters**

76 The Pic du Midi Observatory (PDM, 0.14° E, 42.94° N, 2877 m a.s.l) is a high-altitude monitoring station
77 situated on top of an isolated peak on the north edge of the central Pyrenees Mountains, Southwest France. The
78 PDM receives free tropospheric air from the Northeast Atlantic Ocean and Europe and boundary layer air from
79 upslope winds in Spain and southwest France (Gheusi et al., 2011; Hulin et al., 2019). The station receives
80 predominantly southerly and southwesterly winds (> 70 %) across the main range of Pyrenees (Hulin et al.,
81 2019). There are no major anthropogenic Hg emission point sources around the station or in the surrounding
82 areas.

83

84 **Atmospheric sampling and processing**

85 Short-term (12/24h) variations in atmospheric GEM isotopic composition were investigated from 19 to 28 in
86 Dec 2012 at PDM using large chlorine-impregnated activated carbon (CLC) traps. Traps with 600 mg CLC were

87 utilized for 24 hour sampling at a flow rate of $\sim 10 \text{ L min}^{-1}$; whereas traps with 1000 mg of CLC were utilized
88 for 12 hour sampling at a flow rate of $\sim 20 \text{ LPM}$ (Fu et al., 2014). The sampling inlets of the CLC traps (heated
89 at 70°C to avoid water condensation), preceded by a pre-burned 47mm quartz fiber filter were connected to a
90 Tekran 1104 Teflon coated sampling manifold, operating at 100 LPM and 70°C , and installed into the prevailing
91 wind direction. Sampling flow rates of CLC traps were regulated via ball flow meters installed at the outlets of
92 the vacuum pumps, and the total sampling volumes of the traps were recorded using gas meters. Similar to
93 Arctic studies (Berg et al., 2013; Steffen et al., 2005) we find that the Hg collected on CLC traps reflects the
94 GEM fraction, and is not confounded by the GOM fraction. Both GOM and PBM fractions are retained on the
95 quartz filter preceding the CLC trap. Quartz has been previously shown to efficiently scavenge GOM over time
96 periods shorter than 24h (Lyman and Jaffe, 2011; Lynam and Keeler, 2002). CLC traps were thermally
97 decomposed (25°C to 1000°C over 8h) in a Hg-free oxygen flow (25 ml min^{-1}) in a tube furnace. Combustion
98 products were further decomposed in a 2nd pyrolyzer tube furnace at 1000°C , and purged into a 10 ml 40 vol%
99 $2\text{HNO}_3/1\text{HCl}$ oxidizing solution (Sun et al., 2013). CLC sampling efficiencies were better than 95%, mean
100 recovery efficiencies were 87%, and full procedural blanks $<2\%$ of Hg concentrations in trapping solutions (Fu
101 et al., 2014).

102 Twelve samples of the atmospheric RM (RM=GOM+PBM) were obtained using Millipore
103 polyethersulfone cation exchange membranes (CEMs, Millipore, $0.45 \mu\text{m}$ pore size, 47 mm and 90 mm diameter
104 (Maruszczak et al., 2017)) at the PDM from June to October 2014, with sampled air volumes between 25 and 77
105 m^3 . Cloud water samples were collected from July to November 2014 with a CASCC-2 sampler at the PDM
106 (Demoz et al., 1996). Rainfall was sampled from July to August 2014 150km downwind from the PDM at 800m
107 altitude in the Pyrenees foothills using acid-washed polypropylene buckets and reported elsewhere (Enrico et
108 al., 2016). Snowfall and rainfall was collected at the PDM and the Geosciences Environnement Toulouse (GET,
109 Toulouse, France, 43.54°N , 1.48°E) laboratory between 2016 and 2017 using the same methods. Cloud water,
110 rainfall and snowfall samples collected after each event were transferred into pre-cleaned Pyrex glass bottles (1
111 and 2 L volume), then oxidized by adding 0.1 – 1 % BrCl , and kept refrigerated before Hg concentration analysis
112 by CVAFS and further pre-concentration for Hg isotope analysis. Hg from rain and cloud water samples was
113 pre-concentrated by reducing and purging Hg with SnCl_2 and a clean argon flow and subsequent trapping in an
114 oxidizing solution, following Sherman, et al. (Sherman et al., 2012). Recoveries of procedural standards, rainfall
115 and cloud water samples was $>95\%$. RM on CEMs was leached in 20 vol% $2\text{HNO}_3/1\text{HCl}$ with blanks $<4\%$ of
116 sampled RM.

117 Hg isotope fractionation factors during aqueous photochemical reduction of rainwater was studied using a
118 solar simulator (see (Yang et al., 2019) for details). A low Hg level (2.6 ng L⁻¹) rainfall sample ‘GET rain 4’
119 obtained at GET was spiked with 103 ng L⁻¹ NIST 3133 and equilibrated for 48 in the dark. A 450 mL aliquot
120 was then transferred into an acid-cleaned quartz reactor, and illuminated for 24h with the solar spectrum from
121 a Xe lamp. After 24, the 78 ng L⁻¹ dissolved Hg(II) remaining in the 450mL reactor volume was transferred to
122 a pre-cleaned 1L Pyrex bottle, and pre-concentrated for Hg isotope analysis.

123

124 **Mercury isotopic analysis**

125 Hg isotope ratios were measured using cold vapor multi-collector inductively coupled plasma mass
126 spectrometry (CV-MC-ICPMS, Thermo-Finnigan Neptune, and Neptune Plus) at the Midi-Pyrenees
127 Observatory, Toulouse, France following the methods described in previous studies (Jiskra et al., 2019; Obrist
128 et al., 2017). Instrumental mass bias of MC-ICPMS was corrected by standard-sample-standard bracketing using
129 NIST3133 Hg at matching concentrations. Hg isotopic composition is reported in delta notation (δ) in units of
130 per mil (‰) referenced to the bracketed NIST 3133 Hg standard (Blum and Bergquist, 2007): $\delta^{xxx}\text{Hg} =$
131 $[(^{xxx}\text{Hg}/^{198}\text{Hg})_{\text{sample}}/(^{xxx}\text{Hg}/^{198}\text{Hg})_{\text{NIST3133}} - 1]$, where xxx represent Hg isotope masses 199, 200, 201, 202, 204.
132 MIF values are expressed by “capital delta (Δ)” notation (‰), which is the difference between the measured
133 values of $\delta^{199}\text{Hg}$, $\delta^{200}\text{Hg}$, $\delta^{201}\text{Hg}$ and those predicted from $\delta^{202}\text{Hg}$ using the kinetic MDF law: $\Delta^{xxx}\text{Hg} (\text{‰}) =$
134 $\delta^{xxx}\text{Hg} - (\beta^{xxx} \times \delta^{202}\text{Hg})$, where β^{xxx} is -0.5074, 0.2520, 0.5024, 0.7520, 1.493 for the 196, 199, 200, 201, and 204
135 Hg isotopes respectively. The analytical uncertainty of isotopic analysis was assessed by repeated analysis of
136 the UM-Almaden and ETH-Fluka standards over different analysis sessions and at Hg concentrations
137 corresponding to the samples. The overall mean values of $\delta^{202}\text{Hg}$, $\Delta^{199}\text{Hg}$, $\Delta^{200}\text{Hg}$, $\Delta^{201}\text{Hg}$, and $\Delta^{204}\text{Hg}$ for all
138 the UM-Almaden standards in the 0.5 to 1.5 ng g⁻¹ range were $-0.56 \pm 0.07 \text{‰}$, $-0.04 \pm 0.08 \text{‰}$, $-0.01 \pm 0.09 \text{‰}$
139 and $-0.05 \pm 0.09 \text{‰}$, and $0.04 \pm 0.19 \text{‰}$ (2SD, n = 11) respectively, which agree well with previously reported
140 values (Blum and Bergquist, 2007). The procedural standard for rainfall pre-concentration, i.e. rainfall sample
141 ‘GET rain 4’ spiked with 103 ng g⁻¹ NIST 3133 showed good recovery and no isotopic bias (Supporting Data).
142 In the present study, the analytical uncertainty of CV-MC-ICPMS isotope analysis of RM, GEM, rainfall, and
143 snowfall is the 2SD uncertainty of the UM-Almaden standard, unless the 2SD uncertainty on repeated analysis
144 of the same sample over different analytical sessions was larger.

145 To attempt rainfall $\Delta^{196}\text{Hg}$ analysis, we screened rainfall samples obtained at GET in 2016 for both high
146 Hg concentration levels and high volume single events. Rainfall sample ‘GET rain 6’ was the only sample

147 retained as it consisted of a single thunderstorm event of 20.5 L, collected in the 8 large buckets, and had a Hg
148 concentration of 6.0 ng L⁻¹. The entire 20.5L volume was pre-concentrated 1200x in a single large Pyrex glass
149 bottle to a final concentration of 8.0 ng g⁻¹ in 15.3 g of 20vol% iAR. The sample was run in duplicate during
150 two different days using dedicated UM-Almaden and ETH-Fluka standards, also at 8.0 ng g⁻¹. All individual
151 sample and standard results are given in the Supporting Data. The signals obtained on the ¹⁹⁶Hg and ²⁰²Hg
152 isotopes during these sessions were 15mV and 3 Volt. We recommend the use of 10¹² or 10¹³ Ω resistors on the
153 signal amplifier of the ¹⁹⁶Hg Faraday cup. The 2σ uncertainty, based on sample, UM-Almaden, and ETH-Fluka
154 replicates, on Δ¹⁹⁶Hg during these analysis sessions were 0.19‰ (Supplementary Data).

155

156 **Ancillary parameters**

157 Atmospheric Hg speciation at the PDM, including GEM, GOM and PBM, was continuously measured off
158 the same sampling manifold with a Tekran 2537/1130/1135 system (Tekran Inc., Canada) (Fu et al., 2016a;
159 Maruszczak et al., 2017). Automated GOM and PBM analyses are subject to mounting criticism as the techniques
160 lack calibration, collection efficiency control and interference correction (Jaffe et al., 2014). We partly address
161 these issues by parallel manual sampling of RM (GOM + PBM) on CEMs (Huang et al., 2013). We find that
162 denuder-based GOM at the PDM is systematically underestimated by a factor of 1.56 for which we correct
163 (indicated by an asterisk hereafter, i.e. GOM*)(Maruszczak et al., 2017). Atmospheric CO and O₃ concentrations
164 were continuously measured using a TEI (Thermo Environmental Instruments Inc. USA) 48CTL gas filter
165 correlation analyzer and 49C UV-absorption ozone analyzer, respectively, following published procedures
166 (Gheusi et al., 2011). Meteorology, CO and O₃ data at the PDM were provided through the Pyrenean platform
167 for observation of the atmosphere (P2OA, <http://p2oa.aero.obs-mip.fr>).

168

169 **Air mass backward trajectories**

170 Seven day backward trajectories ending at PDM at a height of 3000 m above sea level were calculated
171 every 2 hours with simulated gridded meteorological data from the Weather Research and Forecasting (WRF)
172 model Version 3.4 with the Noah Land Surface Model, Thompson microphysics scheme, Grell-3 cumulus
173 scheme, Dudhia radiation physics scheme, YSU scheme for PBL. The modeled domain of meteorological data
174 mainly covered the eastern North Atlantic Ocean and European area, with a time resolution of 1 h, 400 × 800
175 grid cells at a 12-km horizontal resolution, and vertical levels of 50 from 1000 hPa to 20 hPa. All the trajectories
176 during the 12h, 24h of individual Hg isotope samples were grouped into an average backward trajectory to show

177 the mean air mass source region of samples.

178

179 **Results and discussion**

180 **GEM isotope compositions**

181 The nine day sampling period from 19 to 28 Dec 2012 was characterized by a shift from cloudy conditions (days
182 1-2, samples D1-D3) to clear skies (days 3-5, samples D4-D9), and back to clouds (days 7-9, samples D10-
183 D14). The wind direction was predominantly southwest bringing North-Atlantic Ocean air masses, over
184 northwest Spain, to the PDM. The continuous Hg speciation observations show a pronounced increase in GOM*
185 to 252 pg m⁻³ (48h mean), and concurrent decrease in GEM during samples D4-D6 (clear skies, Figure 1, Table
186 S1). Back trajectory analysis, and low observed relative humidity and CO, suggest that the high GOM event
187 represents the intrusion of a free tropospheric air mass that originated at 5000m (550hPa, Figures 1, 2). Such
188 intrusions are typical at PDM, e.g. 61 events were registered over a one year period from Nov 2011 to Nov 2012
189 (Fu et al., 2016a). The intrusions are of particular interest as they offer a snapshot of the free troposphere where
190 GEM oxidation is thought to take place (Fu et al., 2016a; Lyman and Jaffe, 2011; Swartzendruber et al., 2006;
191 Timonen et al., 2013). $\delta^{202}\text{Hg}$, $\Delta^{199}\text{Hg}$ and $\Delta^{200}\text{Hg}$ of 12h and 24h GEM samples collected from 19 to 28 Dec
192 2012 were in the range of 0.16 to 0.51‰ ($\delta^{202}\text{Hg}$, mean = 0.34‰), -0.31 to -0.11‰ ($\Delta^{199}\text{Hg}$, mean = -0.23‰),
193 and -0.17 to 0.04‰ ($\Delta^{200}\text{Hg}$, mean = -0.07‰) (Table S1). These observations are consistent with a single two-
194 week integrated GEM isotope sample B19, collected over the same period 19 to 28 Dec 2012 at a lower flow
195 rate of 2 L min⁻¹. with $\delta^{202}\text{Hg}$ of 0.39‰, $\Delta^{199}\text{Hg}$ of -0.27‰, $\Delta^{200}\text{Hg}$ of -0.10 ‰ (Fu et al., 2016b).

196

197 **RM, cloud water and rainfall Hg isotope compositions**

198 In a previous study we found that rainfall Hg isotope signatures 150km east from the PDM are typical of rainfall
199 observations globally, with a mean $\delta^{202}\text{Hg}$ 0.17 ± 0.33 ‰ and significantly positive odd MIF ($\Delta^{199}\text{Hg} = 0.72 \pm$
200 0.15 ‰; $\Delta^{201}\text{Hg} = 0.76 \pm 0.14$ ‰) and even-MIF ($\Delta^{200}\text{Hg} = 0.21 \pm 0.04$ ‰; $\Delta^{204}\text{Hg} = -0.33 \pm 0.10$ ‰, all 1σ , n
201 $= 9$) (Figure 3, Table S1, Enrico et al., 2016). Here we find that PDM cloud water THg concentrations and MDF
202 and MIF signatures are not significantly different from PDM and regional rainfall ($p=0.26$ to 0.79). For the first
203 time we observe that RM, i.e. GOM+PBM, at PDM also has positive odd MIF ($\Delta^{199}\text{Hg} = 0.44 \pm 0.17$ ‰; $\Delta^{201}\text{Hg}$
204 $= 0.44 \pm 0.10$ ‰) and even-MIF ($\Delta^{200}\text{Hg} = 0.15 \pm 0.06$ ‰; $\Delta^{204}\text{Hg} = -0.22 \pm 0.17$ ‰, all 1σ , $n = 12$), similar to
205 PDM cloud water and rainfall (Figure 3) and similar to global rainfall ($\Delta^{199}\text{Hg} = 0.35 \pm 0.19$ ‰; $\Delta^{200}\text{Hg} = 0.14$
206 ± 0.06 ‰; $\Delta^{201}\text{Hg} = 0.35 \pm 0.20$ ‰ (Kwon et al., 2020). Demers et al. (2013) were the first to point out a negative

207 correlation between $\Delta^{200}\text{Hg}$ and $\Delta^{204}\text{Hg}$ in rainfall (Demers et al., 2013). Our work at the PDM extends this
208 relationship to RM, cloud water, and GEM phases (Figure 3). The presence of even- and odd-MIF in RM
209 suggests that globally observed Hg MIF signatures in rainfall are inherited from RM. Equally important, the
210 isotopic composition of RM is in stark contrast with that of GEM (Figure 3), even though the only
211 transformation that separates the two forms of Hg is net oxidation. This corroborates the suggestion that net
212 photochemical oxidation of GEM is responsible for even MIF (Chen et al., 2012), but it does not identify the
213 oxidative or reductive nature of the MIF inducing step among the myriad of possible reactions that drive
214 atmospheric net GEM oxidation.

215 Cloud water and rainfall $\delta^{202}\text{Hg}$ ($0.17 \pm 0.34 \text{ ‰}$) and $\Delta^{199}\text{Hg}$ ($0.21 \pm 0.04 \text{ ‰}$) show significant enrichment
216 in the heavier and in the odd Hg isotopes relative to mean RM $\delta^{202}\text{Hg}$ ($-0.73 \pm 0.63 \text{ ‰}$) and $\Delta^{199}\text{Hg}$ ($0.44 \pm$
217 0.18 ‰)(both $p < 0.001$, all mean, 1σ). We investigated if this is caused by in-cloud Hg^{II} photoreduction, by
218 incubating a rainfall sample spiked with 103 ng L^{-1} of NIST 3133 Hg. After 24h exposure to artificial sunlight
219 from a Xe lamp, the Hg concentration had dropped to 78 ng L^{-1} and small positive shifts in $\delta^{202}\text{Hg}$ (0.15 ‰),
220 $\Delta^{199}\text{Hg}$ (0.10‰), and $\Delta^{201}\text{Hg}$ (0.11‰), but not $\Delta^{200}\text{Hg}$ (-0.03‰) were detected. These observations are
221 consistent with limited, slow in-cloud Hg^{II} photoreduction affecting cloud water and rainfall MDF and odd-MIF.
222 Finally, we observe for the first time the $\Delta^{196}\text{Hg}$ signature to be near-zero ($-0.06 \pm 0.19\text{‰}$, 2σ , $n=2$) in the ‘GET
223 rain 6’ sample (Figure 4), which has pronounced $\Delta^{199}\text{Hg}$ (0.70‰), $\Delta^{200}\text{Hg}$ (0.25‰), $\Delta^{201}\text{Hg}$ (0.82‰), and $\Delta^{204}\text{Hg}$
224 (-0.45‰). The observation that the least abundant ^{196}Hg , and most abundant ^{202}Hg isotope, both referenced to
225 ^{198}Hg , show near-zero MIF, further suggests that nuclear self-shielding is unlikely to be the mechanism
226 responsible for atmospheric Hg MIF.

227

228 $\delta^{202}\text{Hg}$ and $\Delta^{199}\text{Hg}$ variation during the high GOM event

229 We obtained additional evidence for odd and even MIF during net GEM oxidation from the high frequency (12-
230 24h) GEM isotope observations during the free tropospheric air mass intrusion, enriched in GOM and depleted
231 in GEM (Figure 1, days 4-6). The high GOM event showed GOM* concentrations of 252 pg m^{-3} from 23-24
232 Dec (48h mean). The GOM event was accompanied by low CO concentrations, depleted GEM concentrations,
233 and low relative humidity (Figure 1,2), which is consistent with previous studies and indicative of GOM
234 production by oxidation of GEM in a free tropospheric air mass (Gratz et al., 2015; Shah et al., 2016). Seven-
235 day back-trajectories of the GOM event and of the pre- (days 1-3) and post-event (days 7-9) air masses illustrate
236 the important control of cloudiness along the trajectory on detected GOM levels at the PDM (Figure 2). Pre/post-

237 GOM-event air masses generally travelled at 700-1000 hPa (3000m – 0m asl) over the N-Atlantic Ocean (35-
238 50°N) with RH from 60-100% and daytime solar radiation $<200 \text{ w m}^{-2}$ suggestive of GOM scavenging by clouds
239 (Figure 1,2). GOM-event air masses also travelled over the subtropical N-Atlantic Ocean, though at lower
240 latitudes (20-35°N), and higher altitudes of 500-900 hPa (5500m – 1500m asl) with RH from 0-40% and daytime
241 solar radiation from 200-500 w m^{-2} suggestive of clear skies. GOM-event mean ozone (47 ± 2 ppbv, sd), CO
242 (83 ± 3 ppbv, sd) and relative humidity ($34 \pm 10\%$, sd) are typical of lower to middle free tropospheric air. Both
243 the back trajectories and ozone, CO, RH characteristics suggest a negligible influence of stratospheric air on the
244 observed GOM-event.

245 A negative correlation between GEM and GOM* was observed during the 8-day study period of Dec 2012
246 with GOM*/GEM slope of -0.44 ± 0.10 ($r^2 = 0.66$; Figure 5). It has been previously suggested that GOM*/GEM
247 slopes higher than -1 reflect the loss of newly produced GOM by scavenging processes, or the mixing with other
248 air masses with different GEM levels (Lyman and Jaffe, 2011; Swartzendruber et al., 2006). PBM concentrations
249 during the high-GOM event are uniformly low (mean, 4 pg m^{-3}) and RH and solar radiation levels along the
250 back trajectories do not support an important scavenging sink (Figure 2). A prerequisite for the GOM*/GEM
251 slope analysis is that the ensemble of observations has a common air mass origin, which is not the case here.
252 The back trajectory analysis of our 8-day observations suggests overlapping altitudes for pre- and post-event
253 (3000m – 0m asl) and GOM-event (5500m – 1500m asl) air masses. Published reviews of GEM altitude profiles
254 suggest little variation up to 5km (Holmes et al., 2010). However, observed and modeled boundary layer GEM
255 levels suggest maximum levels between 30-60°N, corresponding to mid-latitude emission sources, and declining
256 levels from 30°N to the equator (1% decrease in GEM per degree latitude, Holmes et al., 2010). As mentioned
257 above pre/post-event and GOM-event back trajectories point out different latitudinal origins of 20-35°N and 35-
258 50°N respectively. It seems therefore plausible that the original GEM levels in pre/post-event and GOM-event
259 levels air masses were different due to different latitudinal air mass origins. A correction for latitudinal
260 confounding allows estimating original GEM levels to be 15% lower in the GOM-event air masses, based on
261 the 1% drop in GEM per degree latitude, and yields a corrected GOM*/GEM slope of -0.88 (Figure 5, gray
262 line). The corrected slope is close to -1 and therefore suggests that all nearly GOM produced has been retained
263 within the free tropospheric air masses observed during the high GOM-event at the PDM.

264 $\delta^{202}\text{Hg}_{\text{GEM}}$ and $\Delta^{204}\text{Hg}_{\text{GEM}}$ during the high-GOM event were slightly but significantly higher, and $\Delta^{199}\text{Hg}_{\text{GEM}}$,
265 $\Delta^{200}\text{Hg}_{\text{GEM}}$ and $\Delta^{201}\text{Hg}_{\text{GEM}}$ slightly but significantly lower than pre- and post-event samples during 20 - 28 Dec
266 2012 (Figure 1; t-test all $p < 0.05$). All five GEM isotope signatures are significantly correlated with GEM (Figure

267 6) and GOM concentration. Net GEM oxidation therefore seems to have caused significant MDF, odd-MIF and
268 even-MIF. The above analysis of the GOM event at the PDM suggests that the observed air masses represent
269 the reactants, products and isotope fractionation characteristics of net GEM oxidation. Because the GOM-event
270 was long-lived, displaying elevated GOM levels over 48 hours, we suggest that the GEM and GOM
271 concentration, and GEM isotope observations represent a snapshot of net GEM oxidation that has been little
272 influenced by further mixing with boundary layer air. This implies that the observed MDF, odd-MIF and even-
273 MIF has accompanied net GEM oxidation in the lower troposphere. Previous observations on even-MIF in
274 precipitation ($\Delta^{200}\text{Hg}$ only; $\Delta^{204}\text{Hg}$ not observed; Chen et al. 2012) led to the suggestion that even-MIF
275 originates at the tropopause or stratosphere. Our results suggest that the even-MIF mechanism also operates
276 within the free troposphere.

277 In Figure 6 we plot the GEM and RM (\sim GOM) isotope observations in Rayleigh diagrams, and fit MDF,
278 odd-MIF and even-MIF factors for net GEM oxidation. Assuming pre-/post GOM event isotope observations
279 as the starting point for GEM reactant in the free troposphere, we applied Rayleigh distillation equations for
280 residual $\delta^{202}\text{Hg}_{\text{GEM}}$, $\Delta^{199}\text{Hg}_{\text{GEM}}$, $\Delta^{200}\text{Hg}_{\text{GEM}}$, $\Delta^{201}\text{Hg}_{\text{GEM}}$, and $\Delta^{204}\text{Hg}_{\text{GEM}}$ and for cumulative product GOM to
281 estimate corresponding isotope fractionation factors $\epsilon_{\text{MDF}(\text{GOM}-\text{GEM})}^{202/198}$ of $-1.1\pm 0.2\%$, $\epsilon_{\text{MIF}(\text{GOM}-\text{GEM})}^{199/198}$ of
282 $0.65\pm 0.11\%$, $\epsilon_{\text{MIF}(\text{GOM}-\text{GEM})}^{200/198}$ of $0.22\pm 0.05\%$, $\epsilon_{\text{MIF}(\text{GOM}-\text{GEM})}^{201/198}$ of $0.65\pm 0.10\%$, and $\epsilon_{\text{MIF}(\text{GOM}-\text{GEM})}^{204/198}$ of $-0.39\pm 0.12\%$
283 for the net GEM to GOM conversion observed (Figure 6). It should be kept in mind that these fractionation
284 factor do not inform on whether oxidation or reduction is the underlying cause, i.e. they represent net oxidation.

285

286 **Atmospheric GEM and RM pool size**

287 GEM and RM represent the two dominant reduced and oxidized atmospheric Hg pools. The relative size of
288 these two pools is not very well constrained due to limited mountain-top and in-flight measurements and bias
289 in denuder based RM measurements (Jaffe et al., 2014). Recent global Hg chemistry and transport models
290 indicate that RM (500 Mg) represents 12% and GEM 88% (3600 Mg) of atmospheric Hg, most of this residing
291 in the free troposphere (Horowitz et al., 2017). Based on the newly observed free tropospheric RM and GEM
292 isotope compositions (this study), and previously observed rainfall Hg, and modeled Hg emission fluxes and
293 isotopic composition, we can estimate by isotopic mass balance the proportions of RM and GEM (Table 1). The
294 $\Delta^{200}\text{Hg}$ signature, is best suited for this, since it is not confounded by additional even-MIF in cloud water, ocean
295 or soil environment. Annual mean GEM $\Delta^{200}\text{Hg}$ at the PDM of $-0.03\pm 0.03\%$ (Fu et al., 2016b) is identical to

296 GEM $\Delta^{200}\text{Hg}$ of -0.04‰ of the pre-/post GOM intrusion air masses, representing the northern hemisphere free
297 tropospheric GEM pool. RM $\Delta^{200}\text{Hg}$ at PDM (0.15‰) is similar to mean northern hemispheric rainfall $\Delta^{200}\text{Hg}$
298 ($0.13\pm 0.06\text{‰}$), suggesting that both values represent the free tropospheric RM pool. We use a model estimate
299 for $\Delta^{200}\text{Hg}$ ($0.02\pm 0.02\text{‰}$) of global Hg emissions (Sun et al., 2019) to constrain the mass balance, and calculate
300 that $30\pm 17\%$ (1σ) of free tropospheric Hg should be present as RM in order to explain the rainfall, RM and
301 GEM $\Delta^{200}\text{Hg}$ observations. This would suggest that current RM observations and models underestimate the size
302 of the atmospheric RM pool by a factor of two. Our assessment is potentially biased due to the lack of
303 observations in the southern hemisphere.

304

305 **Acknowledgements**

306 This work was supported by research grants ERC-2010-StG-20091028, ERC-2015-PoC-665482 from the
307 European Research Council and the H2020 ERA-PLANET (689443) iGOSP programme to JES. We
308 acknowledge technical support from the UMS 831 Pic du Midi Observatory team. This observatory is part of
309 the P2OA (<http://p2oa.aero.obs-mip.fr>). P2OA facilities and staff are funded and supported by the University
310 Paul Sabatier Toulouse 3, France, and CNRS (Centre National de la Recherche Scientifique). XY thanks the
311 Chinese Scholarship Council for his PhD fellowship.

312

313 **Author contributions**

314 J.E.S conceived the study. X.F, N.M., Y.X, M.J. and J.E.S performed field sampling. X.F, N.M., Y.X, M.J, LEHB,
315 JC. and J.E.S performed laboratory measurements. F.G. managed ancillary data collection at the PDM. X.F and
316 J.E.S wrote the draft manuscript, which was improved by contributions from all authors.

317

318 **Additional information**

319 All Hg stable isotope data measured and discussed in this study is reported in the Supplementary Data.

320

321 **References**

- 322 Amos, H. M., Jacob, D. J., Holmes, C. D., Fisher, J. A., Wang, Q., Yantosca, R. M., Corbitt, E. S., Galarneau, E.,
323 Rutter, A. P., Gustin, M. S., Steffen, A., Schauer, J. J., Graydon, J. A., St Louis, V. L., Talbot, R. W., Edgerton, E. S.,
324 Zhang, Y. and Sunderland, E. M.: Gas-particle partitioning of atmospheric Hg(II) and its effect on global mercury
325 deposition, *Atmospheric Chemistry and Physics*, 12(1), 591–603, doi:10.5194/acp-12-591-2012, 2012.
- 326 Ariya, P. A., Amyot, M., Dastoor, A., Deeds, D., Feinberg, A., Kos, G., Poulain, A., Ryjkov, A., Semeniuk, K., Subir,
327 M. and Toyota, K.: Mercury Physicochemical and Biogeochemical Transformation in the Atmosphere and at
328 Atmospheric Interfaces: A Review and Future Directions, *Chemical Reviews*, 115(10), 3760–3802,
329 doi:10.1021/cr500667e, 2015.
- 330 Bagnato, E., Tamburello, G., Avaró, G., Martínez-Cruz, M., Enrico, M., Fu, X., Sprovieri, M. and Sonke, J. E.:
331 Mercury fluxes from volcanic and geothermal sources: an update., *Geological Society, London, Special Publications*,
332 doi 10.1144/SP410.2, 2014.
- 333 Berg, T., Pfaffhuber, K. A., Cole, A. S., Engelsen, O. and Steffen, A.: Ten-year trends in atmospheric mercury
334 concentrations, meteorological effects and climate variables at Zeppelin, Ny-Ålesund, *Atmospheric Chemistry and*
335 *Physics*, 13(13), 6575–6586, doi:10.5194/acp-13-6575-2013, 2013.
- 336 Bergquist, B. A. and Blum, J. D.: Mass-Dependent and -Independent Fractionation of Hg Isotopes by Photoreduction
337 in Aquatic Systems., *Science*, 318, 417–420, 2007.
- 338 Blum, J. D. and Bergquist, B. A.: Reporting of variations in the natural isotopic composition of mercury., *Analytical*
339 *and Bioanalytical Chemistry*, 388, 353–359, 2007.
- 340 Blum, J. D., Sherman, L. S. and Johnson, M. W.: Mercury Isotopes in Earth and Environmental Sciences, in *Annual*
341 *Review of Earth and Planetary Sciences*, Vol 42, vol. 42, edited by R. Jeanloz, pp. 249–269., 2014.
- 342 Buchachenko, A. L., Ivanov, V. L., Roznyatovskii, V. A., Artamkina, G. A., Vorob'ev, A. K. and Ustynyuk, Y. A.:
343 Magnetic isotope effect for mercury nuclei in photolysis of Bis(p-trifluoromethylbenzyl)mercury., *Doklady Physical*
344 *Chemistry*, 413, 39–41, 2007.
- 345 Chen, J., Hintelmann, H., Feng, X. and Dimock, B.: Unusual fractionation of both odd and even mercury isotopes in
346 precipitation from Peterborough, ON, Canada, *Geochimica et Cosmochimica Acta*, 90, 33–46,
347 doi:10.1016/j.gca.2012.05.005, 2012.
- 348 Demers, J. D., Blum, J. D. and Zak, D. R.: Mercury isotopes in a forested ecosystem: Implications for air-surface
349 exchange dynamics and the global mercury cycle, *Global Biogeochemical Cycles*, 27(1), 222–238,
350 doi:10.1002/gbc.20021, 2013.
- 351 Demoz, B., Collett, J., J. L. and Daube, J., B. C.: On the Caltech Active Strand Cloudwater Collectors., *Atmospheric*
352 *Research*, 41, 47–62, 1996.
- 353 Dibble, T. S., Tetu, H. L., Jiao, Y., Thackray, C. P. and Jacob, D. J.: Modeling the OH-Initiated Oxidation of Mercury
354 in the Global Atmosphere without Violating Physical Laws, *Journal of Physical Chemistry A*, 124(2), 444–453,
355 doi:10.1021/acs.jpca.9b10121, 2020.

356 Donohoue, D. L., Bauer, D., Cossairt, B. and Hynes, A.: Temperature and pressure dependent rate coefficients for the
357 reaction of Hg with Br and the reaction of Br with Br: A pulsed laser photolysis-pulsed laser induced fluorescence
358 study, *Journal of Physical Chemistry A*, 110, 6623–6632, 2006.

359 Enrico, M., Le Roux, G., Maruszczak, N., Heimbürger, L.-E., Claustres, A., Fu, X., Sun, R. and Sonke, J. E.:
360 Atmospheric mercury transfer to peat bogs dominated by gaseous elemental mercury dry deposition., *Environmental
361 Science & Technology*, doi:10.1021/acs.est.5b06058, 2016.

362 Fitzgerald, W. F. and Lamborg, C. H.: *Geochemistry of Mercury in the Environment*, in *Treatise on Geochemistry*,
363 vol. 9, edited by E. D. Holland and K. K. Turekian, pp. 107–148, Elsevier-Pergamon, Oxford., 2004.

364 Fu, X., Heimbürger, L. E. and Sonke, J. E.: Collection of atmospheric gaseous mercury for stable isotope analysis
365 using iodine- and chlorine-impregnated activated carbon traps, *Journal of Analytical Atomic Spectrometry*, DOI:
366 10.1039/C3JA50356A, 2014.

367 Fu, X. W., Maruszczak, N., Heimbürger, L. E., Sauvage, B., Gheusi, F. and Sonke, J. E.: Atmospheric mercury
368 speciation dynamics at the high-altitude Pic du Midi Observatory, southern France, *Atmospheric Chemistry and
369 Physics*, 16, 5623–5639, doi:10.5194/acp-16-5623-2016, 2016a.

370 Fu, X. W., Maruszczak, N., Wang, X., Gheusi, F. and Sonke, J. E.: The isotopic composition of total gaseous mercury
371 in the free troposphere of the Pic du Midi Observatory (2877 m a.s.l, France), *Environmental Science & Technology*,
372 submitted, 2016b.

373 Gheusi, F., Ravetta, F., Delbarre, H., Tsamalis, C., Chevalier-Rosso, A., Leroy, C., Augustin, P., Delmas, R., Ancellet,
374 G., Athier, G., Bouchou, P., Campistron, B., Cousin, J. M., Fourmentin, M. and Meyerfeld, Y.: Pic 2005, a field
375 campaign to investigate low-tropospheric ozone variability in the Pyrenees, *Atmospheric Research*, 101(3), 640–665,
376 doi:10.1016/j.atmosres.2011.04.014, 2011.

377 Gratz, L., Keeler, G., Blum, J. and Sherman, L. S.: Isotopic composition and fractionation of mercury in Great Lakes
378 precipitation and ambient air, *Environmental Science and Technology*, 44, 7764–7770, 2010.

379 Gratz, L. E., Ambrose, J. L., Jaffe, D. A., Shah, V., Jaegle, L., Stutz, J., Festa, J., Spolaor, M., Tsai, C., Selin, N. E.,
380 Song, S., Zhou, X., Weinheimer, A. J., Knapp, D. J., Montzka, D. D., Flocke, F. M., Campos, T. L., Apel, E.,
381 Hornbrook, R., Blake, N. J., Hall, S., Tyndall, G. S., Reeves, M., Stechman, D. and Stell, M.: Oxidation of mercury
382 by bromine in the subtropical Pacific free troposphere, *Geophysical Research Letters*, 42(23),
383 doi:10.1002/2015gl066645, 2015.

384 Grissom, C. B.: Magnetic field effects in biology: A survey of possible mechanisms with emphasis on radical-pair
385 recombination, *Chemical Reviews*, 95, 3–24, 1995.

386 Holmes, C. D., Jacob, D. J., Corbitt, E. S., Mao, J., Yang, X., Talbot, R. and Slemr, F.: Global atmospheric model for
387 mercury including oxidation by bromine atoms, *Atmospheric Chemistry and Physics*, 10, 12037–12057, 2010.

388 Horowitz, H. M., Jacob, D. J., Zhang, Y., Dibble, T. S., Slemr, F., Amos, H. M., Schmidt, J. A., Corbitt, E. S., Marais,
389 E. A. and Sunderland, E. M.: A new mechanism for atmospheric mercury redox chemistry: implications for the global
390 mercury budget, *Atmospheric Chemistry and Physics*, 17(10), 6353–6371, doi:10.5194/acp-17-6353-2017, 2017.

391 Hulin, M., Gheusi, F., Lothon, M., Pont, V., Lohou, F., Ramonet, M., Delmotte, M., Derrien, S., Athier, G., Meyerfeld,
392 Y., Bezombes, Y., Augustin, P. and Ravetta, F.: Observations of Thermally Driven Circulations in the Pyrenees:
393 Comparison of Detection Methods and Impact on Atmospheric Composition Measured at a Mountaintop, *Journal of*
394 *Applied Meteorology and Climatology*, 58(4), 717–740, doi:10.1175/JAMC-D-17-0268.1, 2019.

395 Jaffe, D. A., Lyman, S., Amos, H. M., Gustin, M. S., Huang, J., Selin, N. E., Levin, L., ter Schure, A., Mason, R. P.,
396 Talbot, R., Rutter, A., Finley, B., Jaegle, L., Shah, V., McClure, C., Ambrose, J., Gratz, L., Lindberg, S., Weiss-
397 Penzias, P., Sheu, G.-R., Feddersen, D., Horvat, M., Dastoor, A., Hynes, A. J., Mao, H., Sonke, J. E., Slemr, F., Fisher,
398 J. A., Ebinghaus, R., Zhang, Y. and Edwards, G.: Progress on Understanding Atmospheric Mercury Hampered by
399 Uncertain Measurements, *Environmental Science & Technology*, 48(13), 7204–7206, doi:10.1021/es5026432, 2014.

400 Jiskra, M., Wiederhold, J. G., Skyllberg, U., Kronberg, R.-M., Hajdas, I. and Kretzschmar, R.: Mercury Deposition
401 and Re-emission Pathways in Boreal Forest Soils Investigated with Hg Isotope Signatures, *Environmental Science*
402 *& Technology*, 49(12), 7188–7196, doi:10.1021/acs.est.5b00742, 2015.

403 Jiskra, M., Sonke, J. E., Agnan, Y., Helmig, D. and Obrist, D.: Insights from mercury stable isotopes on terrestrial-
404 atmosphere exchange of Hg(0) in the Arctic tundra, *Biogeosciences*, 16(20), 4051–4064, doi:10.5194/bg-16-4051-
405 2019, 2019.

406 Kritee, K., Blum, J. D., Reinfelder, J. R. and Barkay, T.: Microbial stable isotope fractionation of mercury: A synthesis
407 of present understanding and future directions, *Chemical Geology*, this issue, doi:10.1016/j.chemgeo.2012.08.017,
408 2012.

409 Kwon, S. Y., Blum, J. D., Yin, R., Tsui, M. T.-K., Yang, Y. H. and Choi, J. W.: Mercury stable isotopes for monitoring
410 the effectiveness of the Minamata Convention on Mercury, *Earth-Science Reviews*, 203, 103111,
411 doi:https://doi.org/10.1016/j.earscirev.2020.103111, 2020.

412 Lyman, S. N. and Jaffe, D. A.: Formation and fate of oxidized mercury in the upper troposphere and lower
413 stratosphere, *Nature Geoscience*, 5(2), 114–117, doi:10.1038/ngeo1353, 2011.

414 Lynam, M. M. and Keeler, G. J.: Comparison of methods for particulate phase mercury analysis: sampling and
415 analysis, *Analytical and Bioanalytical Chemistry*, 374(6), 1009–1014, doi:10.1007/s00216-002-1584-4, 2002.

416 Maruszczak, N., Sonke, J. E., Fu, X. and Jiskra, M.: Tropospheric GOM at the Pic du Midi Observatory-Correcting
417 Bias in Denuder Based Observations, *Environmental Science & Technology*, 51(2), 863–869,
418 doi:10.1021/acs.est.6b04999, 2017.

419 Motta, L. C., Blum, J. D., Johnson, M. W., Umhau, B. P., Popp, B. N., Washburn, S. J., Drazen, J. C., Benitez-Nelson,
420 C. R., Hannides, C. C. S., Close, H. G. and Lamborg, C. H.: Mercury Cycling in the North Pacific Subtropical Gyre
421 as Revealed by Mercury Stable Isotope Ratios, *Global Biogeochemical Cycles*, 33(6), 777–794,
422 doi:10.1029/2018GB006057, 2019.

423 Motta, L. C., Kritee, K., Blum, J. D., Tsz-Ki Tsui, M. and Reinfelder, J. R.: Mercury Isotope Fractionation during the
424 Photochemical Reduction of Hg(II) Coordinated with Organic Ligands., *The journal of physical chemistry. A*,
425 doi:10.1021/acs.jpca.9b06308, 2020a.

426 Motta, L. C., Chien, A. D., Rask, A. E. and Zimmerman, P. M.: Mercury Magnetic Isotope Effect: A Plausible

427 Photochemical Mechanism, *J. Phys. Chem. A*, 124(19), 3711–3719, doi:10.1021/acs.jpca.0c00661, 2020b.

428 Obrist, D., Agnan, Y., Jiskra, M., Olson, C. L., Colegrove, D. P., Hueber, J., Moore, C. W., Sonke, J. E. and Helmig,
429 D.: Tundra uptake of atmospheric elemental mercury drives Arctic mercury pollution, *Nature*, 547(7662), 201–+,
430 doi:10.1038/nature22997, 2017.

431 Saiz-Lopez, A., Sitkiewicz, S. P., Roca-Sanjuán, D., Oliva-Enrich, J. M., Dávalos, J. Z., Notario, R., Jiskra, M., Xu,
432 Y., Wang, F., Thackray, C. P., Sunderland, E. M., Jacob, D. J., Travnikov, O., Cuevas, C. A., Acuña, A. U., Rivero,
433 D., Plane, J. M. C., Kinnison, D. E. and Sonke, J. E.: Photoreduction of gaseous oxidized mercury changes global
434 atmospheric mercury speciation, transport and deposition, *Nature Communications*, 9(1), 4796, doi:10.1038/s41467-
435 018-07075-3, 2018.

436 Saiz-Lopez, A., Acuña, A. U., Trabelsi, T., Carmona-García, J., Dávalos, J. Z., Rivero, D., Cuevas, C. A., Kinnison,
437 D. E., Sitkiewicz, S. P., Roca-Sanjuán, D. and Francisco, J. S.: Gas-Phase Photolysis of Hg(I) Radical Species: A
438 New Atmospheric Mercury Reduction Process, *J. Am. Chem. Soc.*, 141(22), 8698–8702, doi:10.1021/jacs.9b02890,
439 2019.

440 Schauble, E. A.: Role of nuclear volume in driving equilibrium isotope fractionation of mercury, thallium, and other
441 very heavy elements., *Geochimica et Cosmochimica Acta*, 71, 2170–2189, 2007.

442 Seigneur, C. and Lohman, K.: Effect of bromine chemistry on the atmospheric mercury cycle, *Journal of Geophysical*
443 *Research - Atmosphere*, 113, D23309, 2008.

444 Seigneur, C., Vijayaraghavan, K., Lohman, K., Karamchandani, P. and Scott, C.: Global Source Attribution for
445 Mercury Deposition in the United States, *Environmental Science and Technology*, 38, 555–569, 2004.

446 Shah, V., Jaegle, L., Gratz, L. E., Ambrose, J. L., Jaffe, D. A., Selin, N. E., Song, S., Campos, T. L., Flocke, F. M.,
447 Reeves, M., Stechman, D., Stell, M., Festa, J., Stutz, J., Weinheimer, A. J., Knapp, D. J., Montzka, D. D., Tyndall, G.
448 S., Apel, E. C., Hornbrook, R. S., Hills, A. J., Riemer, D. D., Blake, N. J., Cantrell, C. A. and Mauldin, R. L.: Origin
449 of oxidized mercury in the summertime free troposphere over the southeastern US, *Atmospheric Chemistry and*
450 *Physics*, 16(3), 1511–1530, doi:10.5194/acp-16-1511-2016, 2016.

451 Sherman, L. S., Blum, J. D., Keeler, G. J., Demers, J. D. and Dvornch, J. T.: Investigation of Local Mercury Deposition
452 from a Coal-Fired Power Plant Using Mercury Isotopes, *Environmental Science & Technology*, 46(1), 382–390,
453 doi:10.1021/es202793c, 2012.

454 Steffen, A., Schroeder, W., Macdonald, R., Poissant, L. and Konoplev, A.: Mercury in the Arctic atmosphere: An
455 analysis of eight years of measurements of GEM at Alert (Canada) and a comparison with observations at Amderma
456 (Russia) and Kuujuarapik (Canada), *Science of the Total Environment*, 342(1–3), 185–198,
457 doi:10.1016/j.scitotenv.2004.12.048, 2005.

458 Streets, D. G., Horowitz, H. M., Lu, Z., Levin, L., Thackray, C. P. and Sunderland, E. M.: Global and regional trends
459 in mercury emissions and concentrations, 2010-2015, *Atmospheric Environment*, 201, 417–427,
460 doi:10.1016/j.atmosenv.2018.12.031, 2019.

461 Sun, G., Sommar, J., Feng, X., Lin, C.-J., Ge, M., Wang, W., Yin, R., Fu, X. and Shang, L.: Mass-Dependent and -
462 Independent Fractionation of Mercury Isotope during Gas-Phase Oxidation of Elemental Mercury Vapor by Atomic

463 Cl and Br, *Environ. Sci. Technol.*, 50(17), 9232–9241, doi:10.1021/acs.est.6b01668, 2016.

464 Sun, R., Enrico, M., Heimbürger, L.-E., Scott, C. and Sonke, J. E.: A double-stage tube furnace-acid-trapping protocol
465 for the pre-concentration of mercury from solid samples for isotopic analysis, *Analytical and Bioanalytical Chemistry*,
466 405(21), 6771–6781, doi:10.1007/s00216-013-7152-2, 2013.

467 Sun, R., Jiskra, M., Amos, H. M., Zhang, Y., Sunderland, E. M. and Sonke, J. E.: Modelling the mercury stable isotope
468 distribution of Earth surface reservoirs: Implications for global Hg cycling, *Geochimica et Cosmochimica Acta*, 246,
469 156–173, doi:https://doi.org/10.1016/j.gca.2018.11.036, 2019.

470 Swartzendruber, P. C., Jaffe, J. A., Prestbo, E. M., Weiss-Penzias, P., Selin, N. E., Park, R., Jacob, D. J., Strode, S.
471 and Jaegle, L.: Observations of reactive gaseous mercury in the free troposphere at the Mount Bachelor Observatory,
472 *Journal of Geophysical Research*, 111, D24301, 2006.

473 Timonen, H., Ambrose, J. L. and Jaffe, D. A.: Oxidation of elemental Hg in anthropogenic and marine air masses,
474 *Atmospheric Chemistry and Physics*, 13(5), 2827–2836, doi:10.5194/acp-13-2827-2013, 2013.

475 Turro, N. J.: Influence of nuclear spin on chemical, reactions: Magnetic isotope and magnetic field effects (A Review),
476 *Proceedings of the National Academy of Science USA*, 80, 609–621, 1983.

477 Wang, F., Saiz-Lopez, A., Mahajan, A. S., Gomez Martin, J. C., Armstrong, D., Lemes, M., Hay, T. and Prados-
478 Roman, C.: Enhanced production of oxidised mercury over the tropical Pacific Ocean: a key missing oxidation
479 pathway, *Atmospheric Chemistry and Physics*, 14(3), 1323–1335, doi:10.5194/acp-14-1323-2014, 2014.

480 Wang, Z., Chen, J., Feng, X., Hintelmann, H., Yuan, S., Cai, H., Huang, Q., Wang, S. and Wang, F.: Mass-dependent
481 and mass-independent fractionation of mercury isotopes in precipitation from Guiyang, SW China, *Comptes Rendus*
482 *Geoscience*, 347(7), 358–367, doi:https://doi.org/10.1016/j.crte.2015.02.006, 2015.

483 Weiss-Penzias, P., Amos, H. M., Selin, N. E., Gustin, M. S., Jaffe, D. A., Obrist, D., Sheu, G. R. and Giang, A.: Use
484 of a global model to understand speciated atmospheric mercury observations at five high-elevation sites, *Atmospheric*
485 *Chemistry and Physics*, 15(3), 1161–1173, doi:10.5194/acp-15-1161-2015, 2015.

486 Wiederhold, J. G., Cramer, C. J., Daniel, K., Infante, I., Bourdon, B. and Kretzschmar, R.: Equilibrium mercury
487 isotope fractionation between dissolved Hg(II) species and thiol-bound Hg, *Environmental Science and Technology*,
488 44, 4191–4197, 2010.

489 Yang, X., Jiskra, M. and Sonke, J. E.: Experimental rainwater divalent mercury speciation and photoreduction rates
490 in the presence of halides and organic carbon, *Science of The Total Environment*, 697, 133821,
491 doi:https://doi.org/10.1016/j.scitotenv.2019.133821, 2019.

492 Zheng, W. and Hintelmann, H.: Mercury isotope fractionation during photoreduction in natural water is controlled
493 by its Hg/DOC ratio., *Geochimica et Cosmochimica Acta*, 73, 6704–6715, 2009.

494 Zheng, W. and Hintelmann, H.: Isotope Fractionation of Mercury during Its Photochemical Reduction by Low-
495 Molecular-Weight Organic Compounds, *Journal of Physical Chemistry A*, 114, 4246–4253, 2010.

496

497 Ariya, P. A., Amyot, M., Dastoor, A., Deeds, D., Feinberg, A., Kos, G., Poulain, A., Ryjkov, A., Semeniuk, K., Subir,
498 M. and Toyota, K.: Mercury Physicochemical and Biogeochemical Transformation in the Atmosphere and at
499 Atmospheric Interfaces: A Review and Future Directions, *Chemical Reviews*, 115(10), 3760–3802,
500 doi:10.1021/cr500667e, 2015.

501 Bagnato, E., Tamburello, G., Avarid, G., Martinez-Cruz, M., Enrico, M., Fu, X., Sprovieri, M. and Sonke, J. E.:
502 Mercury fluxes from volcanic and geothermal sources: an update., *Geological Society, London, Special Publications*,
503 doi 10.1144/SP410.2, 2014.

504 Berg, T., Pfaffhuber, K. A., Cole, A. S., Engelsens, O. and Steffen, A.: Ten-year trends in atmospheric mercury
505 concentrations, meteorological effects and climate variables at Zeppelin, Ny-Ålesund, *Atmospheric Chemistry and
506 Physics*, 13(13), 6575–6586, doi:10.5194/acp-13-6575-2013, 2013.

507 Bergquist, B. A. and Blum, J. D.: Mass-Dependent and -Independent Fractionation of Hg Isotopes by Photoreduction
508 in Aquatic Systems., *Science*, 318, 417–420, 2007.

509 Blum, J. D. and Bergquist, B. A.: Reporting of variations in the natural isotopic composition of mercury., *Analytical
510 and Bioanalytical Chemistry*, 388, 353–359, 2007.

511 Blum, J. D., Sherman, L. S. and Johnson, M. W.: Mercury Isotopes in Earth and Environmental Sciences, in *Annual
512 Review of Earth and Planetary Sciences*, Vol 42, vol. 42, edited by R. Jeanloz, pp. 249–269., 2014.

513 Buchachenko, A. L., Ivanov, V. L., Roznyatovskii, V. A., Artamkina, G. A., Vorob'ev, A. K. and Ustynyuk, Y. A.:
514 Magnetic isotope effect for mercury nuclei in photolysis of Bis(p-trifluoromethylbenzyl)mercury., *Doklady Physical
515 Chemistry*, 413, 39–41, 2007.

516 Chen, J., Hintelmann, H., Feng, X. and Dimock, B.: Unusual fractionation of both odd and even mercury isotopes in
517 precipitation from Peterborough, ON, Canada, *Geochimica et Cosmochimica Acta*, 90, 33–46,
518 doi:10.1016/j.gca.2012.05.005, 2012.

519 Demers, J. D., Blum, J. D. and Zak, D. R.: Mercury isotopes in a forested ecosystem: Implications for air-surface
520 exchange dynamics and the global mercury cycle, *Global Biogeochemical Cycles*, 27(1), 222–238,
521 doi:10.1002/gbc.20021, 2013.

522 Demoz, B., Collett, J., J. L. and Daube, J., B. C.: On the Caltech Active Strand Cloudwater Collectors., *Atmospheric
523 Research*, 41, 47–62, 1996.

524 Dibble, T. S., Tetu, H. L., Jiao, Y., Thackray, C. P. and Jacob, D. J.: Modeling the OH-Initiated Oxidation of Mercury
525 in the Global Atmosphere without Violating Physical Laws, *Journal of Physical Chemistry A*, 124(2), 444–453,
526 doi:10.1021/acs.jpca.9b10121, 2020.

527 Donohoue, D. L., Bauer, D., Cossairt, B. and Hynes, A.: Temperature and pressure dependent rate coefficients for the
528 reaction of Hg with Br and the reaction of Br with Br: A pulsed laser photolysis-pulsed laser induced fluorescence
529 study, *Journal of Physical Chemistry A*, 110, 6623–6632, 2006.

530 Enrico, M., Le Roux, G., Maruszczak, N., Heimbürger, L.-E., Claustres, A., Fu, X., Sun, R. and Sonke, J. E.:
531 Atmospheric mercury transfer to peat bogs dominated by gaseous elemental mercury dry deposition., *Environmental*

532 Science & Technology, doi:10.1021/acs.est.5b06058, 2016.

533 Fitzgerald, W. F. and Lamborg, C. H.: Geochemistry of Mercury in the Environment, in *Treatise on Geochemistry*,
534 vol. 9, edited by E. D. Holland and K. K. Turekian, pp. 107–148, Elsevier-Pergamon, Oxford., 2004.

535 Fu, X., Heimbürger, L. E. and Sonke, J. E.: Collection of atmospheric gaseous mercury for stable isotope analysis
536 using iodine- and chlorine-impregnated activated carbon traps, *Journal of Analytical Atomic Spectrometry*, DOI:
537 10.1039/C3JA50356A, 2014.

538 Fu, X. W., Maruszczak, N., Heimbürger, L. E., Sauvage, B., Gheusi, F. and Sonke, J. E.: Atmospheric mercury
539 speciation dynamics at the high-altitude Pic du Midi Observatory, southern France, *Atmospheric Chemistry and
540 Physics*, 16, 5623–5639, doi:10.5194/acp-16-5623-2016, 2016a.

541 Fu, X. W., Maruszczak, N., Wang, X., Gheusi, F. and Sonke, J. E.: The isotopic composition of total gaseous mercury
542 in the free troposphere of the Pic du Midi Observatory (2877 m a.s.l, France), *Environmental Science & Technology*,
543 submitted, 2016b.

544 Gheusi, F., Ravetta, F., Delbarre, H., Tsamalis, C., Chevalier-Rosso, A., Leroy, C., Augustin, P., Delmas, R., Ancellet,
545 G., Athier, G., Bouchou, P., Campistron, B., Cousin, J. M., Fourmentin, M. and Meyerfeld, Y.: Pic 2005, a field
546 campaign to investigate low-tropospheric ozone variability in the Pyrenees, *Atmospheric Research*, 101(3), 640–665,
547 doi:10.1016/j.atmosres.2011.04.014, 2011.

548 Gratz, L., Keeler, G., Blum, J. and Sherman, L. S.: Isotopic composition and fractionation of mercury in Great Lakes
549 precipitation and ambient air, *Environmental Science and Technology*, 44, 7764–7770, 2010.

550 Gratz, L. E., Ambrose, J. L., Jaffe, D. A., Shah, V., Jaegle, L., Stutz, J., Festa, J., Spolaor, M., Tsai, C., Selin, N. E.,
551 Song, S., Zhou, X., Weinheimer, A. J., Knapp, D. J., Montzka, D. D., Flocke, F. M., Campos, T. L., Apel, E.,
552 Hornbrook, R., Blake, N. J., Hall, S., Tyndall, G. S., Reeves, M., Stechman, D. and Stell, M.: Oxidation of mercury
553 by bromine in the subtropical Pacific free troposphere, *Geophysical Research Letters*, 42(23),
554 doi:10.1002/2015gl066645, 2015.

555 Grissom, C. B.: Magnetic field effects in biology: A survey of possible mechanisms with emphasis on radical-pair
556 recombination, *Chemical Reviews*, 95, 3–24, 1995.

557 Holmes, C. D., Jacob, D. J., Corbitt, E. S., Mao, J., Yang, X., Talbot, R. and Slemr, F.: Global atmospheric model for
558 mercury including oxidation by bromine atoms, *Atmospheric Chemistry and Physics*, 10, 12037–12057, 2010.

559 Horowitz, H. M., Jacob, D. J., Zhang, Y., Dibble, T. S., Slemr, F., Amos, H. M., Schmidt, J. A., Corbitt, E. S., Marais,
560 E. A. and Sunderland, E. M.: A new mechanism for atmospheric mercury redox chemistry: implications for the global
561 mercury budget, *Atmospheric Chemistry and Physics*, 17(10), 6353–6371, doi:10.5194/acp-17-6353-2017, 2017.

562 Hulin, M., Gheusi, F., Lothon, M., Pont, V., Lohou, F., Ramonet, M., Delmotte, M., Derrien, S., Athier, G., Meyerfeld,
563 Y.: Observations of plain-mountain circulation in the Pyrenees : detection methods and impact on atmospheric
564 composition measured at a mountain top, *J. Applied. Meteor. Climatol.*, 2019, 58, 717–740,
565 <https://doi.org/10.1175/JAMC-D-17-0268.1>.

566 Jaffe, D. A., Lyman, S., Amos, H. M., Gustin, M. S., Huang, J., Selin, N. E., Levin, L., ter Schure, A., Mason, R. P.,

567 Talbot, R., Rutter, A., Finley, B., Jaegle, L., Shah, V., McClure, C., Ambrose, J., Gratz, L., Lindberg, S., Weiss-
568 Penzias, P., Sheu, G.-R., Feddersen, D., Horvat, M., Dastoor, A., Hynes, A. J., Mao, H., Sonke, J. E., Slemr, F., Fisher,
569 J. A., Ebinghaus, R., Zhang, Y. and Edwards, G.: Progress on Understanding Atmospheric Mercury Hampered by
570 Uncertain Measurements, *Environmental Science & Technology*, 48(13), 7204–7206, doi:10.1021/es5026432, 2014.

571 Jiskra, M., Wiederhold, J. G., Skyllberg, U., Kronberg, R.-M., Hajdas, I. and Kretzschmar, R.: Mercury Deposition
572 and Re-emission Pathways in Boreal Forest Soils Investigated with Hg Isotope Signatures, *Environmental Science
573 & Technology*, 49(12), 7188–7196, doi:10.1021/acs.est.5b00742, 2015.

574 Jiskra, M., Sonke, J. E., Agnan, Y., Helmig, D. and Obrist, D.: Insights from mercury stable isotopes on terrestrial-
575 atmosphere exchange of Hg(0) in the Arctic tundra, *Biogeosciences*, 16(20), 4051–4064, doi:10.5194/bg-16-4051-
576 2019, 2019.

577 Kritee, K., Blum, J. D., Reinfelder, J. R. and Barkay, T.: Microbial stable isotope fractionation of mercury: A synthesis
578 of present understanding and future directions, *Chemical Geology*, this issue, doi:10.1016/j.chemgeo.2012.08.017,
579 2012.

580 Kwon, S. Y., Blum, J. D., Yin, R., Tsui, M. T.-K., Yang, Y. H. and Choi, J. W.: Mercury stable isotopes for monitoring
581 the effectiveness of the Minamata Convention on Mercury, *Earth-Science Reviews*, 203, 103111,
582 doi:https://doi.org/10.1016/j.earscirev.2020.103111, 2020.

583 Lyman, S. N. and Jaffe, D. A.: Formation and fate of oxidized mercury in the upper troposphere and lower
584 stratosphere, *Nature Geoscience*, 5(2), 114–117, doi:10.1038/ngeo1353, 2011.

585 Lynam, M. M. and Keeler, G. J.: Comparison of methods for particulate phase mercury analysis: sampling and
586 analysis, *Analytical and Bioanalytical Chemistry*, 374(6), 1009–1014, doi:10.1007/s00216-002-1584-4, 2002.

587 Maruszczak, N., Sonke, J. E., Fu, X. and Jiskra, M.: Tropospheric GOM at the Pic du Midi Observatory-Correcting
588 Bias in Denuder Based Observations, *Environmental Science & Technology*, 51(2), 863–869,
589 doi:10.1021/acs.est.6b04999, 2017.

590 Motta, L. C., Blum, J. D., Johnson, M. W., Umhau, B. P., Popp, B. N., Washburn, S. J., Drazen, J. C., Benitez-Nelson,
591 C. R., Hannides, C. C. S., Close, H. G. and Lamborg, C. H.: Mercury Cycling in the North Pacific Subtropical Gyre
592 as Revealed by Mercury Stable Isotope Ratios, *Global Biogeochemical Cycles*, 33(6), 777–794,
593 doi:10.1029/2018GB006057, 2019.

594 Motta, L. C., Kritee, K., Blum, J. D., Tsz-Ki Tsui, M. and Reinfelder, J. R.: Mercury Isotope Fractionation during the
595 Photochemical Reduction of Hg(II) Coordinated with Organic Ligands., *The journal of physical chemistry. A*,
596 doi:10.1021/acs.jpca.9b06308, 2020a.

597 Motta, L. C., Chien, A. D., Rask, A. E. and Zimmerman, P. M.: Mercury Magnetic Isotope Effect: A Plausible
598 Photochemical Mechanism, *J. Phys. Chem. A*, 124(19), 3711–3719, doi:10.1021/acs.jpca.0c00661, 2020b.

599 Obrist, D., Agnan, Y., Jiskra, M., Olson, C. L., Colegrove, D. P., Hueber, J., Moore, C. W., Sonke, J. E. and Helmig,
600 D.: Tundra uptake of atmospheric elemental mercury drives Arctic mercury pollution, *Nature*, 547(7662), 201–+,
601 doi:10.1038/nature22997, 2017.

602 Saiz-Lopez, A., Sitkiewicz, S. P., Roca-Sanjuán, D., Oliva-Enrich, J. M., Dávalos, J. Z., Notario, R., Jiskra, M., Xu,
603 Y., Wang, F., Thackray, C. P., Sunderland, E. M., Jacob, D. J., Travníkov, O., Cuevas, C. A., Acuña, A. U., Rivero,
604 D., Plane, J. M. C., Kinnison, D. E. and Sonke, J. E.: Photoreduction of gaseous oxidized mercury changes global
605 atmospheric mercury speciation, transport and deposition, *Nature Communications*, 9(1), 4796, doi:10.1038/s41467-
606 018-07075-3, 2018.

607 Saiz-Lopez, A., Acuña, A. U., Trabelsi, T., Carmona-García, J., Dávalos, J. Z., Rivero, D., Cuevas, C. A., Kinnison,
608 D. E., Sitkiewicz, S. P., Roca-Sanjuán, D. and Francisco, J. S.: Gas-Phase Photolysis of Hg(I) Radical Species: A
609 New Atmospheric Mercury Reduction Process, *J. Am. Chem. Soc.*, 141(22), 8698–8702, doi:10.1021/jacs.9b02890,
610 2019.

611 Schauble, E. A.: Role of nuclear volume in driving equilibrium isotope fractionation of mercury, thallium, and other
612 very heavy elements., *Geochimica et Cosmochimica Acta*, 71, 2170–2189, 2007.

613 Seigneur, C. and Lohman, K.: Effect of bromine chemistry on the atmospheric mercury cycle, *Journal of Geophysical*
614 *Research - Atmosphere*, 113, D23309, 2008.

615 Seigneur, C., Vijayaraghavan, K., Lohman, K., Karamchandani, P. and Scott, C.: Global Source Attribution for
616 Mercury Deposition in the United States, *Environmental Science and Technology*, 38, 555–569, 2004.

617 Shah, V., Jaegle, L., Gratz, L. E., Ambrose, J. L., Jaffe, D. A., Selin, N. E., Song, S., Campos, T. L., Flocke, F. M.,
618 Reeves, M., Stechman, D., Stell, M., Festa, J., Stutz, J., Weinheimer, A. J., Knapp, D. J., Montzka, D. D., Tyndall, G.
619 S., Apel, E. C., Hornbrook, R. S., Hills, A. J., Riemer, D. D., Blake, N. J., Cantrell, C. A. and Mauldin, R. L.: Origin
620 of oxidized mercury in the summertime free troposphere over the southeastern US, *Atmospheric Chemistry and*
621 *Physics*, 16(3), 1511–1530, doi:10.5194/acp-16-1511-2016, 2016.

622 Sherman, L. S., Blum, J. D., Keeler, G. J., Demers, J. D. and Dvonch, J. T.: Investigation of Local Mercury Deposition
623 from a Coal-Fired Power Plant Using Mercury Isotopes, *Environmental Science & Technology*, 46(1), 382–390,
624 doi:10.1021/es202793c, 2012.

625 Steffen, A., Schroeder, W., Macdonald, R., Poissant, L. and Konoplev, A.: Mercury in the Arctic atmosphere: An
626 analysis of eight years of measurements of GEM at Alert (Canada) and a comparison with observations at Amderma
627 (Russia) and Kuujuarapik (Canada), *Science of the Total Environment*, 342(1–3), 185–198,
628 doi:10.1016/j.scitotenv.2004.12.048, 2005.

629 Streets, D. G., Horowitz, H. M., Lu, Z., Levin, L., Thackray, C. P. and Sunderland, E. M.: Global and regional trends
630 in mercury emissions and concentrations, 2010-2015, *Atmospheric Environment*, 201, 417–427,
631 doi:10.1016/j.atmosenv.2018.12.031, 2019.

632 Sun, G., Sommar, J., Feng, X., Lin, C.-J., Ge, M., Wang, W., Yin, R., Fu, X. and Shang, L.: Mass-Dependent and -
633 Independent Fractionation of Mercury Isotope during Gas-Phase Oxidation of Elemental Mercury Vapor by Atomic
634 Cl and Br, *Environ. Sci. Technol.*, 50(17), 9232–9241, doi:10.1021/acs.est.6b01668, 2016.

635 Sun, R., Enrico, M., Heimbürger, L.-E., Scott, C. and Sonke, J. E.: A double-stage tube furnace-acid-trapping protocol
636 for the pre-concentration of mercury from solid samples for isotopic analysis, *Analytical and Bioanalytical Chemistry*,
637 405(21), 6771–6781, doi:10.1007/s00216-013-7152-2, 2013.

638 Sun, R., Jiskra, M., Amos, H. M., Zhang, Y., Sunderland, E. M. and Sonke, J. E.: Modelling the mercury stable isotope
639 distribution of Earth surface reservoirs: Implications for global Hg cycling, *Geochimica et Cosmochimica Acta*, 246,
640 156–173, doi:<https://doi.org/10.1016/j.gca.2018.11.036>, 2019.

641 Swartzendruber, P. C., Jaffe, J. A., Prestbo, E. M., Weiss-Penzias, P., Selin, N. E., Park, R., Jacob, D. J., Strode, S.
642 and Jaegle, L.: Observations of reactive gaseous mercury in the free troposphere at the Mount Bachelor Observatory,
643 *Journal of Geophysical Research*, 111, D24301, 2006.

644 Timonen, H., Ambrose, J. L. and Jaffe, D. A.: Oxidation of elemental Hg in anthropogenic and marine airmasses,
645 *Atmospheric Chemistry and Physics*, 13(5), 2827–2836, doi:10.5194/acp-13-2827-2013, 2013.

646 Turro, N. J.: Influence of nuclear spin on chemical, reactions: Magnetic isotope and magnetic field effects (A Review),
647 *Proceedings of the National Academy of Science USA*, 80, 609–621, 1983.

648 Wang, F., Saiz-Lopez, A., Mahajan, A. S., Gomez Martin, J. C., Armstrong, D., Lemes, M., Hay, T. and Prados-
649 Roman, C.: Enhanced production of oxidised mercury over the tropical Pacific Ocean: a key missing oxidation
650 pathway, *Atmospheric Chemistry and Physics*, 14(3), 1323–1335, doi:10.5194/acp-14-1323-2014, 2014.

651 Wang, Z., Chen, J., Feng, X., Hintelmann, H., Yuan, S., Cai, H., Huang, Q., Wang, S. and Wang, F.: Mass-dependent
652 and mass-independent fractionation of mercury isotopes in precipitation from Guiyang, SW China, *Comptes Rendus*
653 *Geoscience*, 347(7), 358–367, doi:<https://doi.org/10.1016/j.crte.2015.02.006>, 2015.

654 Weiss-Penzias, P., Amos, H. M., Selin, N. E., Gustin, M. S., Jaffe, D. A., Obrist, D., Sheu, G. R. and Giang, A.: Use
655 of a global model to understand speciated atmospheric mercury observations at five high-elevation sites, *Atmospheric*
656 *Chemistry and Physics*, 15(3), 1161–1173, doi:10.5194/acp-15-1161-2015, 2015.

657 Wiederhold, J. G., Cramer, C. J., Daniel, K., Infante, I., Bourdon, B. and Kretzschmar, R.: Equilibrium mercury
658 isotope fractionation between dissolved Hg(II) species and thiol-bound Hg, *Environmental Science and Technology*,
659 44, 4191–4197, 2010.

660 Yang, X., Jiskra, M. and Sonke, J. E.: Experimental rainwater divalent mercury speciation and photoreduction rates
661 in the presence of halides and organic carbon, *Science of The Total Environment*, 697, 133821,
662 doi:<https://doi.org/10.1016/j.scitotenv.2019.133821>, 2019.

663 Zheng, W. and Hintelmann, H.: Mercury isotope fractionation during photoreduction in natural water is controlled
664 by its Hg/DOC ratio., *Geochimica et Cosmochimica Acta*, 73, 6704–6715, 2009.

665 Zheng, W. and Hintelmann, H.: Isotope Fractionation of Mercury during Its Photochemical Reduction by Low-
666 Molecular-Weight Organic Compounds, *Journal of Physical Chemistry A*, 114, 4246–4253, 2010.

667
668
669
670

671 **Table 1. Mean $\Delta^{200}\text{Hg}$ signatures for Hg emissions and atmospheric reservoirs.** Mean and 1σ for
672 published rainfall data represents the mean of means of remote site data points for each study in order to avoid
673 site type and sample number bias. Global GEM emission $\Delta^{200}\text{Hg}$ was calculated from the listed Hg fluxes and
674 modeled $\Delta^{200}\text{Hg}$. The combined uncertainty on calculated atmospheric GEM and RM fractions was
675 propagated from uncertainties on observational data for emissions and RM and GEM $\Delta^{200}\text{Hg}$.

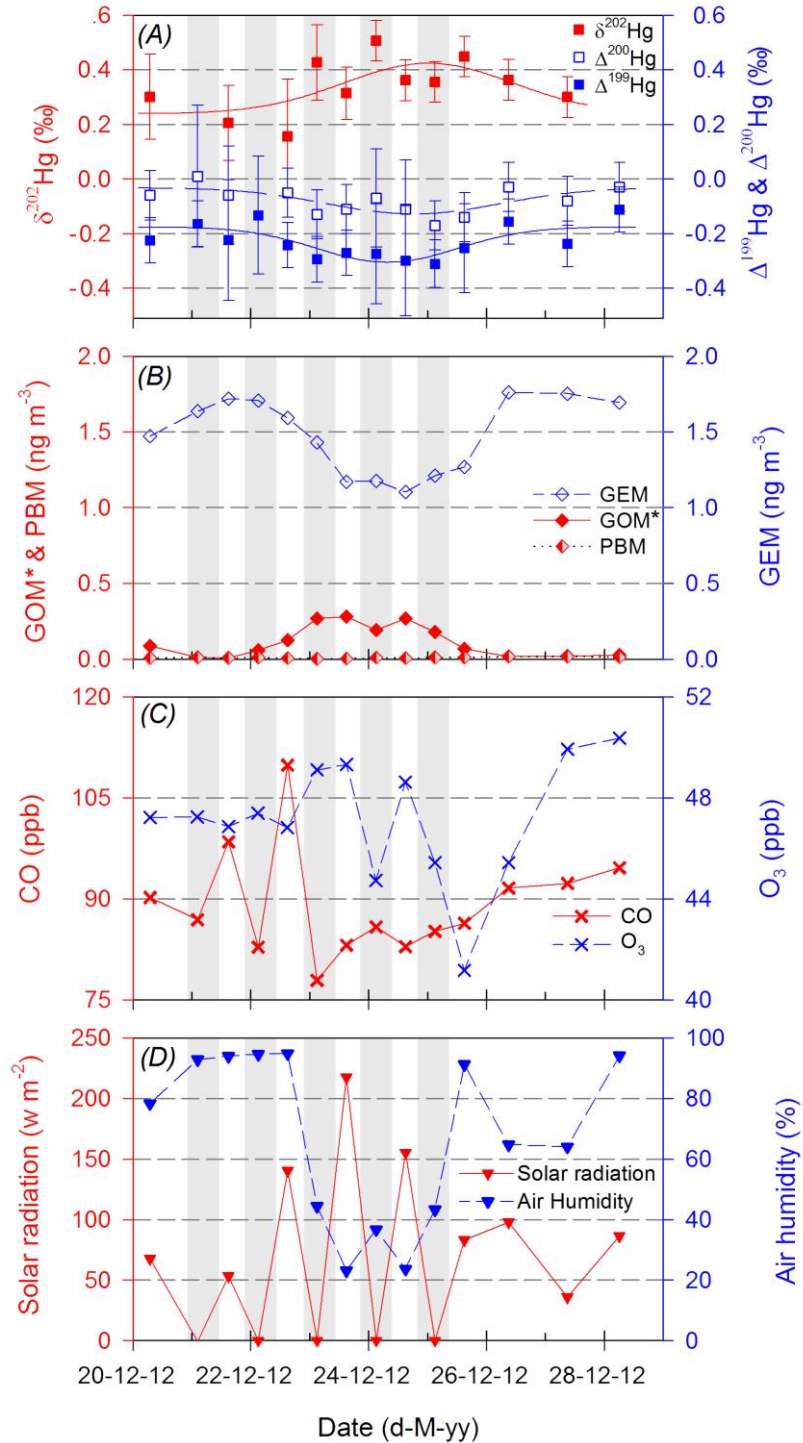
Sample type	Hg flux	$\Delta^{200}\text{Hg}$	1σ	
	Mg y^{-1}	‰	‰	
Pic du Midi RM		0.15	0.12	This study
Rainfall Hg		0.13	0.06	(Blum et al., 2014; Demers et al., 2013; Enrico et al., 2016; Gratz et al., 2010; Motta et al., 2019; Obrist et al., 2017; Sherman et al., 2012; Wang et al., 2015)
Pic du Midi GEM		-0.03	0.02	This study, (Fu et al., 2016b)
Ocean GEM emission	4600	0.03		(Horowitz et al., 2017; Sun et al., 2019)
Anthropogenic GEM	1470	0		(Horowitz et al., 2017; Sun et al., 2019)
Soil GEM emission	1420	-0.02		(Horowitz et al., 2017; Sun et al., 2019)
Volcanic GEM emission	250	0		(Horowitz et al., 2017; Sun et al., 2019)
Global GEM emission	7740	0.02	0.02	(Horowitz et al., 2017; Sun et al., 2019)
f(RM)		0.30 ^a	0.17	
f(GEM)		0.70 ^a	0.17	

676 ^a calculated from mass balance: $\Delta^{200}\text{Hg}(\text{global GEM emitted}) = f(\text{RM}) \times \Delta^{200}\text{Hg}(\text{rainfall}) + f(\text{GEM}) \times$
677 $\Delta^{200}\text{Hg}(\text{PDM GEM})$, and $f(\text{RM}) + f(\text{GEM}) = 1$

678

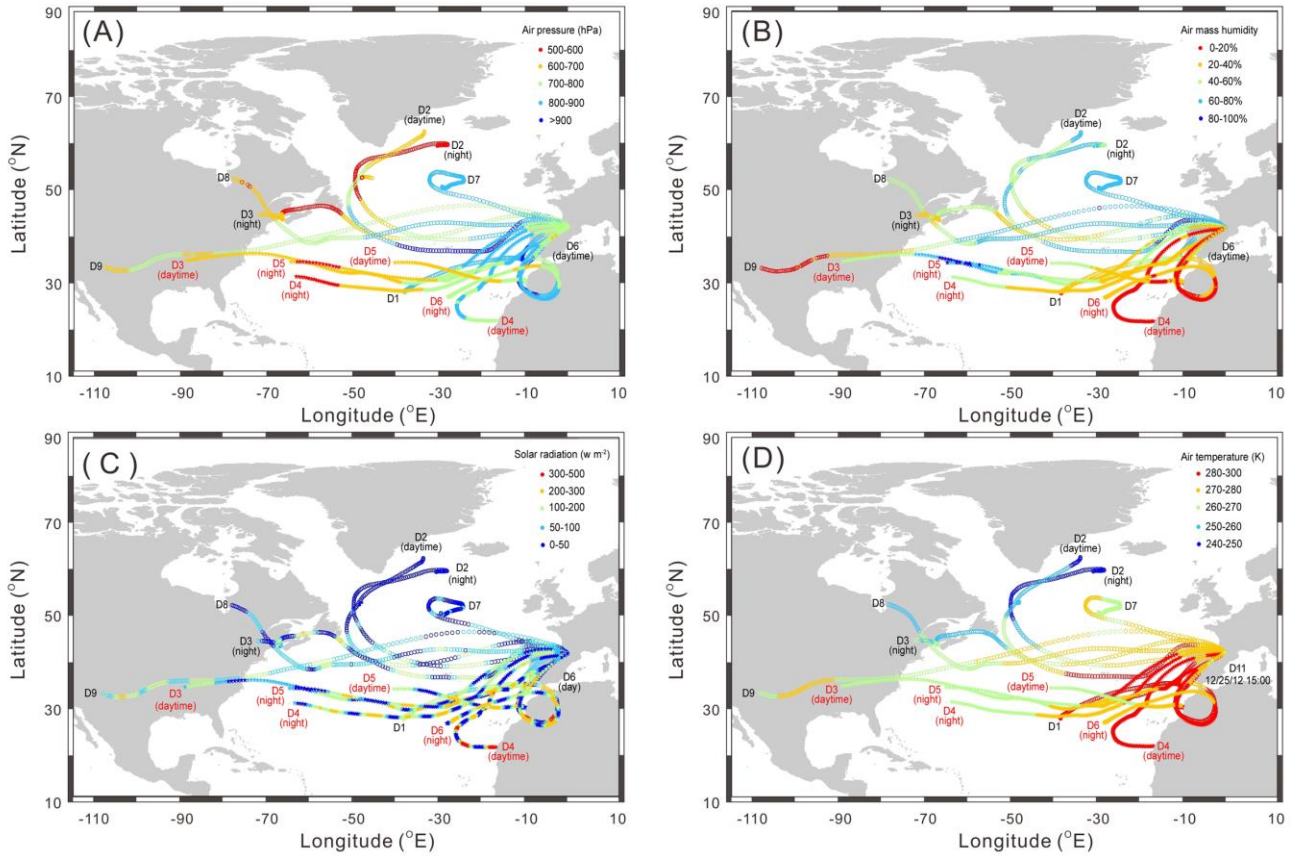
679

680 **Figure 1. Hg speciation and isotope variability during a free tropospheric air mass intrusion.** From 20 to
 681 28 Dec 2012, high frequency (12h, 24h) gaseous elemental Hg⁰ (GEM) isotope signatures $\delta^{202}\text{Hg}$, $\Delta^{199}\text{Hg}$,
 682 $\Delta^{200}\text{Hg}$ (A), CO and O₃ concentrations (B), atmospheric GOM, PBM and GEM concentrations (C), and solar
 683 radiation and relative air humidity (D) showed a 48h long intrusion of a free tropospheric air mass, high in
 684 GOM, and low in GEM and humidity.



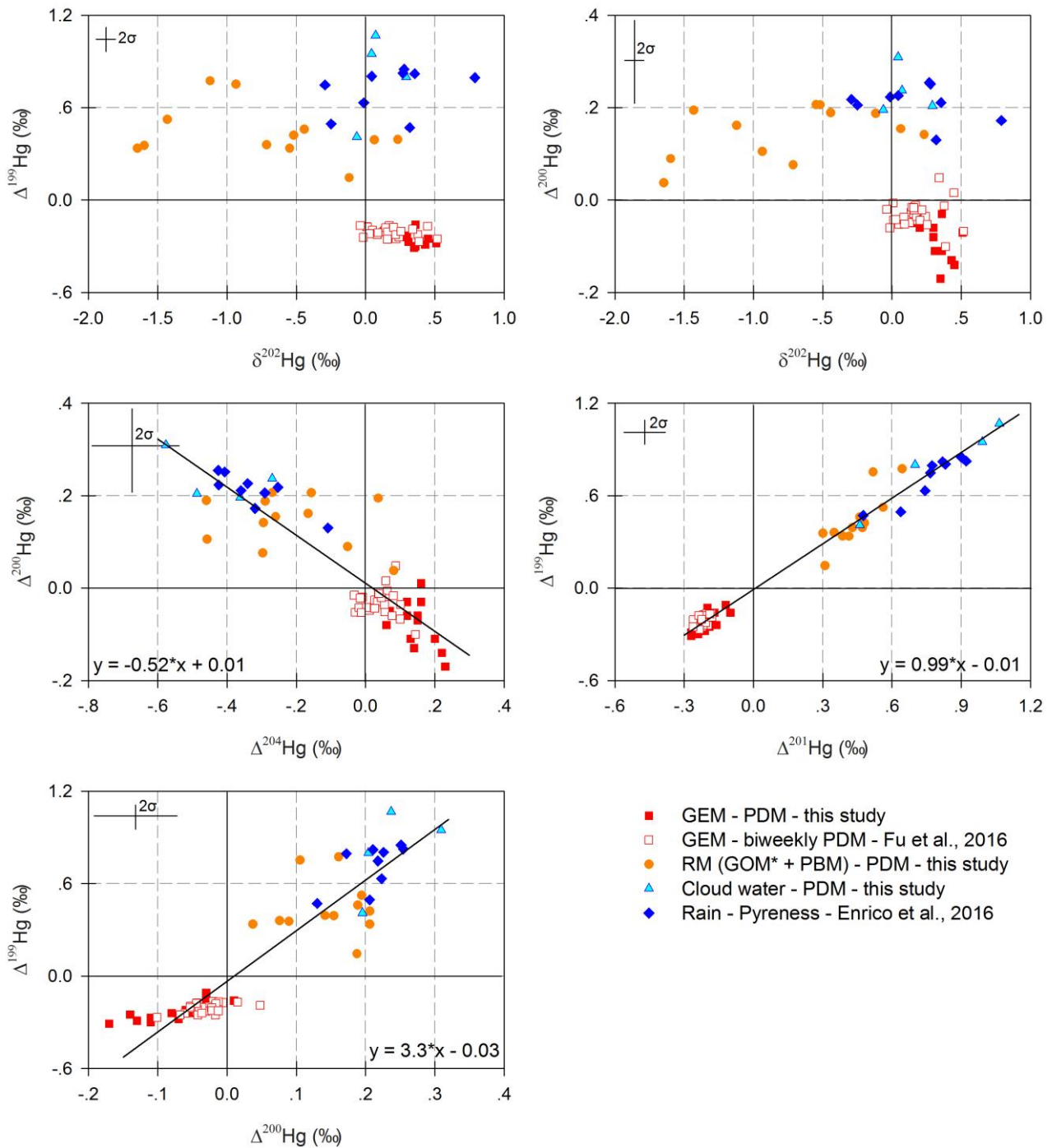
685
 686

687 **Figure 2. Air mass back-trajectories of the free tropospheric air mass intrusion.** 7-day backward
 688 trajectories of the high frequency (12h, 24h) gaseous elemental Hg⁰ (GEM) isotope samples obtained from 20
 689 to 28 Dec 2012: Travelling height (A, indicated by air pressure), air mass humidity (B), solar radiation (C),
 690 and air temperature (D). Date and time labels are mid-points of 12h and 24h Hg isotope sampling windows.



691

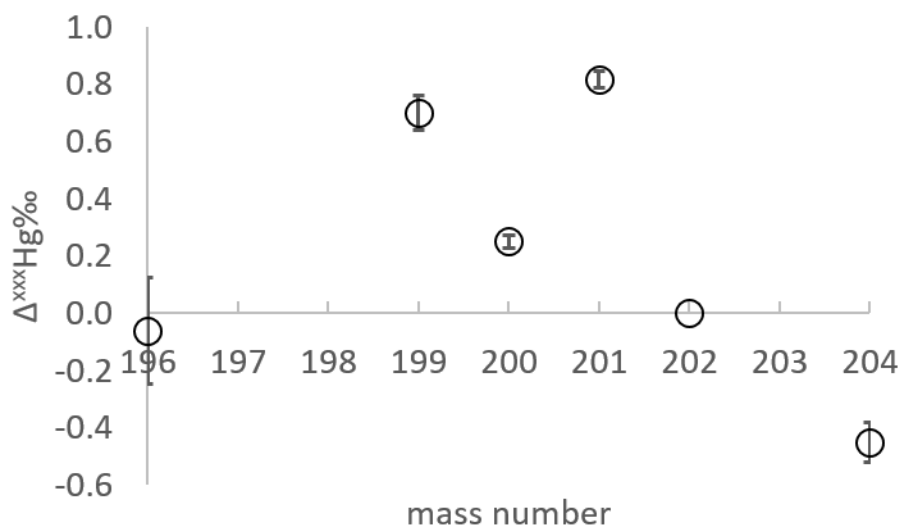
692 **Figure 3. Atmospheric Hg stable isotope variation at the Pic du Midi.** Hg mass dependent ($\delta^{202}\text{Hg}$) and mass
 693 independent ($\Delta^{199}\text{Hg}$, $\Delta^{200}\text{Hg}$, $\Delta^{201}\text{Hg}$, $\Delta^{204}\text{Hg}$) isotope signatures in reactive Hg (RM=GOM+PBM), cloud
 694 water, 12h/24h GEM from this study, and previously published GEM (Fu et al., 2014) and rainfall (Enrico et
 695 al., 2016).



696

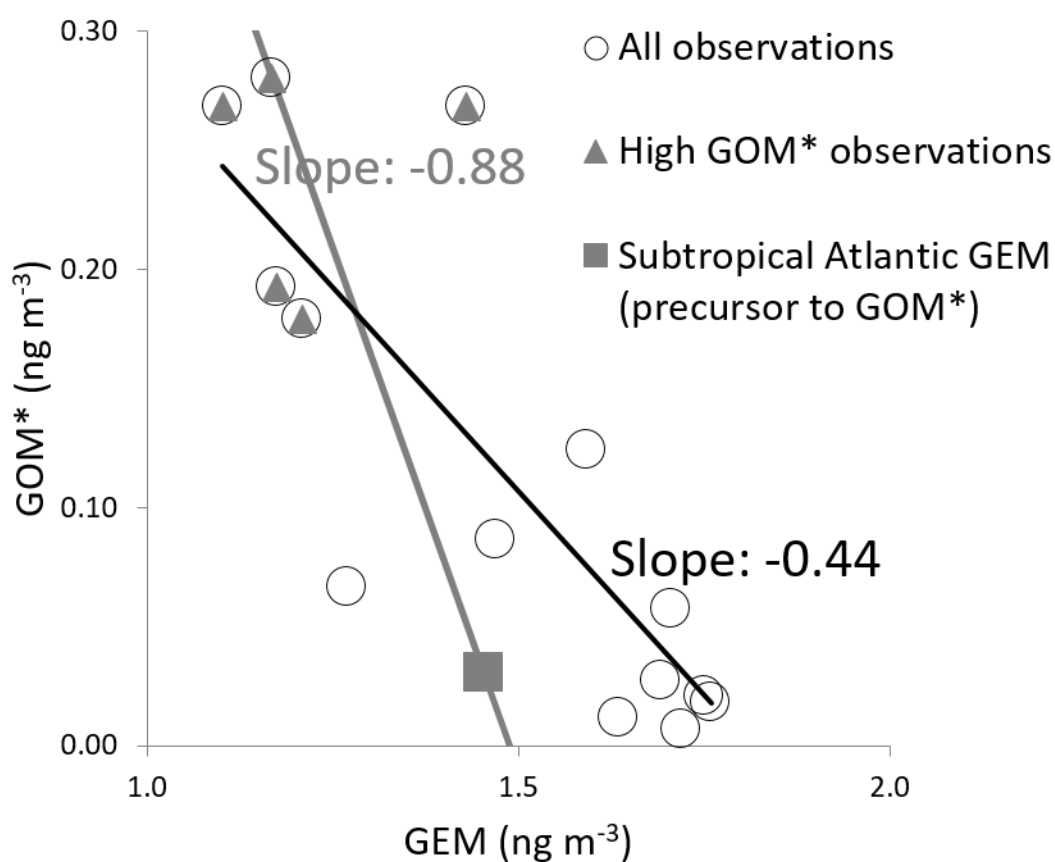
697

698 **Figure 4. $\Delta^{196}\text{Hg}$ measurement of rainfall.** A 20.5L rainfall sample containing 6.0 ng L^{-1} of Hg was pre-
699 concentrated 1200x to deliver a 0.015V signal on ^{196}Hg . The resulting $\Delta^{196}\text{Hg}$ signature is $-0.06 \pm 0.19\text{‰}$ (2σ ,
700 $n=2$) suggesting that nuclear self-shielding is not the underlying cause of atmospheric even-Hg MIF. A $\Delta^{202}\text{Hg}$
701 of 0 is indicated in the graph, since the $\Delta^{\text{xxx}}\text{Hg}$ (where xxx are Hg isotopes masses 196 to 204) are calculated
702 relative to $\delta^{202/198}\text{Hg}$ ratios, themselves anchored to the ^{198}Hg reference isotope.



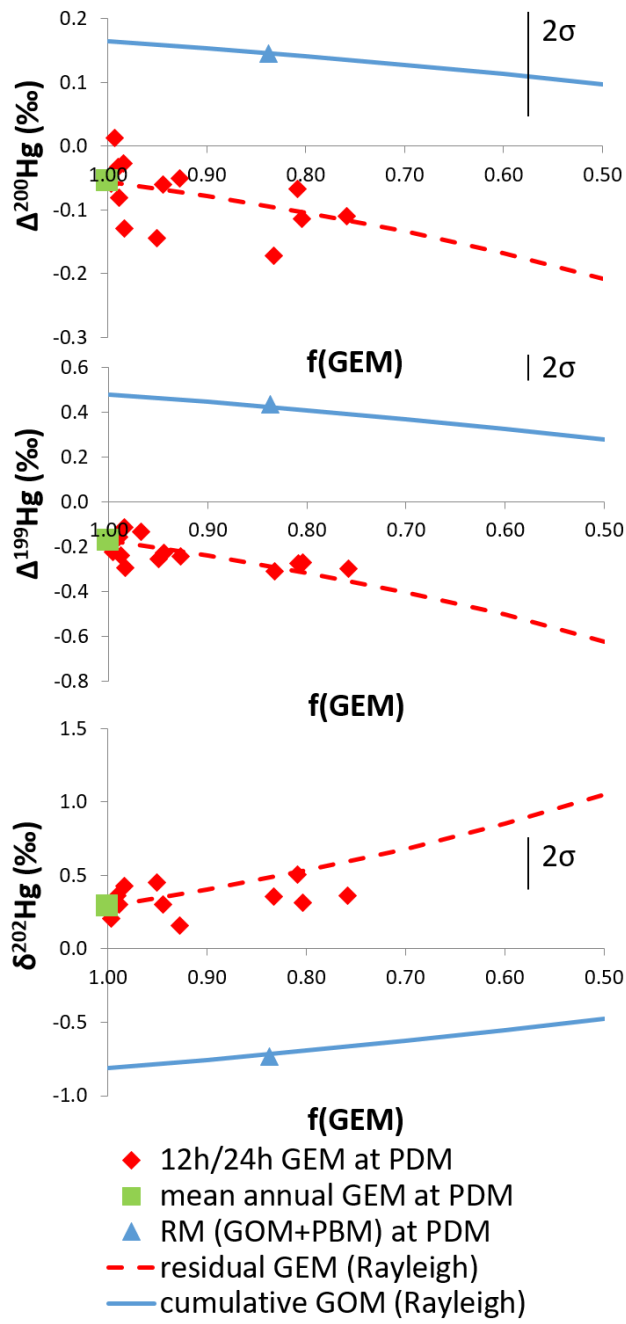
703
704

705 **Figure 5. Hg oxidation mass balance in free tropospheric air at the Pic du Midi.** Relationship between
706 gaseous elemental Hg⁰ (GEM, ng m⁻³) and gaseous oxidized Hg^{II} (GOM*, ng m⁻³) observed during the nine day
707 sampling campaign at PDM. Black circles and grey diamonds are Tekran GEM and GOM* data (Extended
708 Data), where grey diamonds represent the high GOM event only. A slope of -1 is indicative of Hg mass
709 conservation during GEM oxidation to GOM in the observed air mass. The uncorrected GOM*/GEM slope of
710 -0.44 is biased low due to latitudinal bias in the GEM precursor to GOM (see main text). Latitude corrected
711 subtropical Atlantic Ocean GEM levels (grey square) indicate a slope of -0.88, suggesting that Hg mass has
712 been conserved in the high GOM air mass detected at PDM.



713
714
715

716 **Figure 6. Rayleigh diagrams for Hg isotope fractionation during Hg⁰ net oxidation.** This dataset is a
717 composite of GEM isotope composition (red diamonds) recorded in the Atlantic Ocean free tropospheric air
718 mass intrusion at PDM in Dec 2012, and mean RM (=GOM*+PBM ~ GOM*) isotope composition (blue
719 triangles) at PDM recorded from June – October 2014. Isotope fractionation factors $\epsilon_{MDF(GOM-GEM)}^{202}$ of $-1.1 \pm$
720 0.2% , $\epsilon_{MIF(GOM-GEM)}^{4199}$ of $0.65 \pm 0.11\%$, $\epsilon_{MIF(GOM-GEM)}^{4200}$ of $0.22 \pm 0.05\%$, $\epsilon_{MIF(GOM-GEM)}^{4201}$ of $0.65 \pm 0.10\%$, and
721 $\epsilon_{MIF(GOM-GEM)}^{4204}$ of $-0.39 \pm 0.12\%$ (all 1SE) were fitted by log-linear regression of Rayleigh equations for residual
722 GEM and cumulative GOM*. All 12h and 24h observations (blue diamonds) over the nine day period were
723 included in the fitting. The starting isotopic composition (green square) is the mean annual GEM composition
724 at PDM (Fu et al., 2016b).



725
726