



Peer review status:

This is a non-peer-reviewed preprint submitted to EarthArXiv.

1 **ATMOSPHERIC DEPOSITION AND SOURCE APPORTIONMENT OF MERCURY AND LEAD**
2 **OVER THE LAST 150 YEARS IN HIGHLAND PEATLANDS OF SOUTHEASTERN BRAZIL**

3
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17 **ABSTRACT**

18 Trace elements like mercury (Hg) and lead (Pb) are major global pollutants subject to
19 long-range atmospheric transport, posing a threat for human and environmental health at
20 a global scale. Both have a great affinity with organic matter and show limited mobility
21 in soils under stable environmental conditions. Therefore, peatlands efficiently trap and
22 act as reliable archives of changes in anthropogenic Hg and Pb fluxes over time. This
23 study assessed the Pb and Hg accumulation in high mountain peatlands from two
24 Conservation Units of the Rio de Janeiro State, SE-Brazil: Itatiaia National Park - INP
25 and Serra dos Órgãos National Park – SONP to reconstruct the history of atmospheric
26 trace metals pollution over the last 150 years. Results indicated that in both Parks, Hg and
27 Pb concentrations were higher from 1950s, reflecting the beginning of the industrial
28 development in Brazil. Whereas in INP, there was a decreasing accumulation of Hg and
29 Pb, probably due to the influence of hydrological changes since 1950s, especially. On the
30 other hand, in SONP, both metals' accumulation rates were higher on current times (2013
31 to 1973) coincident with the increase in these elements' emissions to the atmosphere in
32 South America. The Hg enrichment factors (EF) were 215 and 130 for INP and SONP
33 respectively, whereas the Pb EF values were 47.4 in INP and 24.7 in SONP. Such
34 evidence indicates a current need for more effective environmental policies and
35 guidelines to control emissions to the atmosphere of these two environmentally
36 significant pollutants.

37 **Keywords:** Accumulation rate; anthropogenic emissions; Enrichment factor; Peat core;
38 Trace elements.

39 1. INTRODUCTION

40 Human activities have significantly increased metal use in various industries
41 leading to their release into the environment and causing toxic effects (Watari et al.,
42 2021). Among these metals, mercury (Hg) and lead (Pb) are of particular concern due to
43 their high toxicity (WHO, 2021; Taylor et al., 2019; Zhang et al., 2019). Metals enter the
44 atmosphere from both natural sources, such as volcanic eruptions, geothermal springs,
45 geologic deposits, the weathering of rocks and soils, forest fires, degassing from oceans,
46 and human activities, including fossil fuel combustion and industrial processes. When
47 these anthropogenic metallic aerosols mix with natural aerosols, they can become more
48 readily deposited in soils (Moteki et al., 2017). While volcanic eruptions and natural
49 processes still contribute significantly to Hg and Pb total emission to the environment,
50 human activities like fossil fuel combustion and industrial processes became their major
51 sources after industrialization. These emissions are transported globally through the
52 atmosphere, with projections suggesting a decrease in deposition in the Northern
53 Hemisphere but an increase in the Southern Hemisphere. This trend may be linked to the
54 observed four-fold rise in Hg concentration within the surface ocean over the past 600
55 years, for example (Pacyna et al., 2009; Sigel et al., 2017; Zhang et al., 2014). Studies on
56 sediments, ice cores and peatlands reveal a concerning trend of increasing Hg deposition
57 that has tripled since pre-industrial times (Lindeberg et al., 2007). In the case of Pb, the
58 main sources to the atmosphere have been industry, mostly coal burning, ferrous and
59 nonferrous smelting, and open waste incineration until ~1950. After, most significant Pb
60 source was *leaded* motor vehicle gasoline use since World War II, resulting in the largest
61 Pb emissions to the atmosphere between 1950 and 1980, related to traffic exhaust. From
62 the 1980's thereafter Pb fluxes to the atmosphere decreased sharply due to the phasing
63 out of leaded motor vehicle gasoline (Weiss et al., 1999), and this trend has occurred in
64 both hemispheres (Lacerda and Ribeiro, 2004). Climate change is expected to exacerbate
65 this issue by further increasing the remobilization of deposited metals from terrestrial and
66 marine surfaces (Li et al., 2020). This is particularly worrisome in the case of Hg, because
67 methyl-Hg, its most toxic form, can bioaccumulate in the food chain, posing a significant
68 threat to ecosystems and human health (Buch et al., 2018).

69 Despite government regulations, environmental pollution from Hg and Pb, like
70 other heavy metals, remains a global concern. Estimates of global anthropogenic Hg
71 emissions vary widely, ranging from 2,700 to 7,500 tons per year (Lacerda, 2003; De
72 Jesus et al., 2018). The US Environmental Protection Agency (USEPA) estimates annual

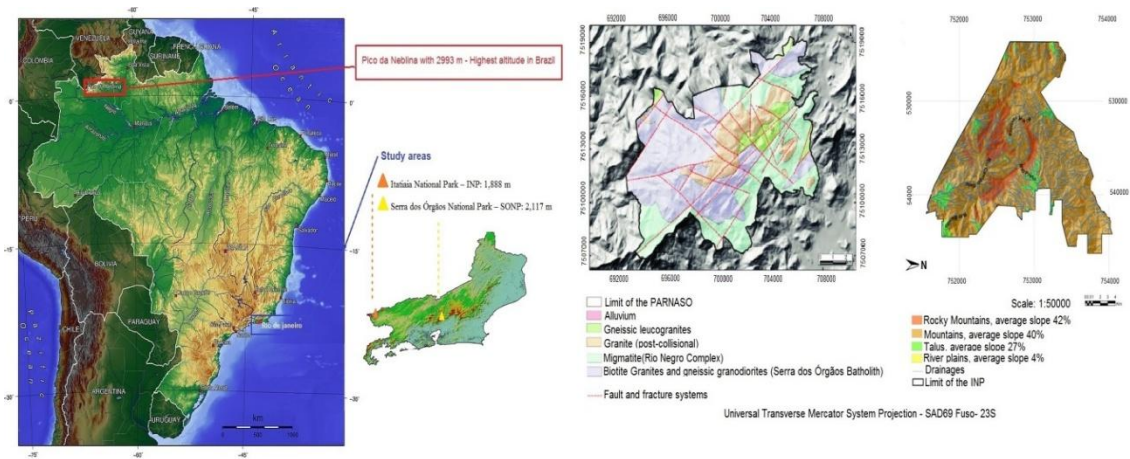
73 global anthropogenic emissions at approximately 2,220 metric tons per year, with South
74 America contributing approximately 18.4% (AMAP, 2018). Despite recent
75 advancements, significant uncertainties remain in these estimates, with a range of 1,010-
76 4,070 tons per year (UNEP, 2013). Globally, Pb emissions decreased sharply after the
77 phasing out of Pb in motor vehicle gasoline. However, recent trends in Pb usage and
78 recycling suggest an estimated emission of $2.84 \cdot 10^6$ t of Pb were discharged into the
79 environment in 2000, in 2022 this emission nearly doubled reaching about $5 \cdot 10^6$ t. (Luby
80 et al., 2024).

81 Understanding the history of Hg and Pb deposition in remote areas like Southeast
82 Brazil is crucial to assess its long-term environmental impact. Peat cores, which act as
83 natural archives of atmospheric deposition, can provide valuable insights into historical
84 trends. Although peatlands have been used extensively to reconstruct atmospheric metal
85 deposition in boreal and temperate regions (Shotyk et al., 2002 and 2016; Martínez-
86 Cortizas et al., 1999), research using tropical peat records remains scarce (Page et al.,
87 1999; Anshari et al., 2001; Wüst and Bustin, 2004; Lacerda and Ribeiro, 2004; Cooke et
88 al., 2021). In this sense, this study aimed to reconstruct the historical atmospheric
89 deposition of Hg and Pb for the last 150 years, archived in two tropical peat cores in
90 Southeast Brazil.

91 **2. MATERIAL AND METHODS**

92 **2.1 Sampling area**

93 The study was conducted in two Conservation Units of Rio de Janeiro State,
94 southeastern Brazil: Itatiaia National Park – INP (22°17'57" S; 44°37'15" W; 1,888 m
95 a.s.l.) and Serra dos Órgãos National Park – SONP (22°27'32" S; 43°01'37" W; 2,117 m
96 a.s.l.) (Figure 1). The parks were selected due to their elevation range (resulting in cool,
97 humid climates with mean annual temperatures of 10–15°C) and differing precipitation
98 regimes (2,500 mm yr⁻¹ at INP; 3,600 mm yr⁻¹ at SONP) (Nimer, 1979). Both areas have
99 low anthropogenic impact and preserve ombrotrophic peatlands fed exclusively by
100 atmospheric inputs, making them ideal for establishing background levels of potentially
101 toxic elements and reconstructing historical atmospheric deposition.



102

103 Figure 1. Location map of the study areas in southeastern Brazil. (a) Geographic position of Rio
 104 de Janeiro State within Brazil; (b) detailed view of the two Conservation Units: Itatiaia National
 105 Park (INP) and Serra dos Órgãos National Park (SONP).

106 2.2 Peat Core Collection Sample Handling and Storage

107 Peat cores were collected using a 100 cm long and 10 cm diameter core sampler.
 108 The sampler was manually introduced from the soil surface into the basal peat layer
 109 (Figure 2). The core collection depth varied between the two sites: Itatiaia National Park
 110 (INP) reached up to 41 cm, while Serra dos Órgãos National Park (SONP) had a
 111 maximum depth of 30 cm. The depth of peat core collection was determined by several
 112 factors, including logistical considerations, research objectives, and anticipated peat
 113 depth. A ~50 cm depth was sufficient to capture recent historical trends in metal
 114 deposition and assess anthropogenic impacts. Once collected, each peat core was
 115 sectioned at a resolution of 1.5 cm. To preserve the integrity of the samples, the 1.5 cm
 116 peat slices were placed in a thermal box and kept refrigerated during transport to the
 117 laboratory. Upon arrival, the samples were stored in a freezer until further analysis.



118

119 Figure 2. Ombrotrophic peatland sampling sites. (a) Itatiaia National Park (INP, 22°17'57" S;
 120 44°37'15" W; 1,888 m a.s.l.); (b) Serra dos Órgãos National Park (SONP, 22°27'32" S; 43°01'37"
 121 W; 2,117 m a.s.l.). The peatlands are located within high-altitude grasslands (*campo de altitude*)
 122 and receive water and nutrients exclusively from atmospheric deposition, making them suitable
 123 natural archives for reconstructing historical inputs of Hg, Pb, and other potentially toxic
 124 elements.

125 2.3 Chemical analysis

126 2.3.1 Peat characterization parameters

127 Prior to mercury (Hg), lead (Pb), and titanium (Ti) analysis, the peat samples were
 128 characterized for unrubbed fiber (UF), rubbed fiber (RF), solubility in sodium
 129 pyrophosphate, pH in CaCl₂, soil bulk density (SBD), organic matter density (OMD),
 130 mineral material (MM), organic matter (OM), and gravimetric moisture (GM), according
 131 to Lynn et al. (1974). These parameters provided fundamental information on peat
 132 decomposition degree, organic matter provenance, and mineral input, which are essential
 133 for interpreting trace element concentrations in the context of atmospheric deposition.
 134 Total organic carbon (TOC) and total nitrogen (N) were determined by dry combustion
 135 using a Perkin-Elmer CHNS/O Analyzer 2400, with acetanilide as the calibration
 136 standard.

137 2.3.2 Mercury analysis

138 For the determination of total Hg concentration, approximately 1.0 g of freeze-dried
 139 samples were weighed and subjected to acidic extraction using aqua regia (3:1 mixture of
 140 HCl:HNO₃ (v/v), both suprapure) at 70°C in a closed cold-finger system (Malm et al.,

141 1990). Mercury (Hg) determination was performed using Cold Vapor Atomic Absorption
 142 Spectrophotometry (CVAAS) with a Bacharach Coleman Model 50d mercury analyzer.
 143 In this method, Hg²⁺ ions are reduced to volatile Hg⁰ using SnCl₂ (Marins et al., 1998),
 144 which is then quantified by the CVAAS instrument.

145 2.3.3 Lead and titanium analysis

146 Freeze-dried peat samples (0.2-0.3 g dry weight) were processed for Pb and Ti
 147 analysis using a microwave digester (speedwave™ MWS-3+; Bergof). Each sample was
 148 digested with 6 ml of 65% HNO₃ (Merck suprapure) following a food-grade temperature
 149 program (100-170 °C). After cooling, the digested samples were diluted with deionized
 150 water to a final volume of 25 ml. Concentrations of Pb and Ti were then determined by
 151 Inductively Coupled Plasma Optical Emission Spectrometry (ICP-OES).

152 2.3.4 Quality control

153 To ensure the accuracy of the analysis, blanks and replicates were included in each
 154 analytical batch. Additionally, a certified reference material (SRM 2709a San Joaquin
 155 Soil, National Institute of Standards and Technology-NIST) was incorporated to assess
 156 the digestion process effectiveness. The recovery for this reference material ranged from
 157 82% to 92% (Table 1). The detection limit was calculated as 3 times the standard
 158 deviation of the blank measurements.

159 **Table 1.** Analytical quality control data for mercury (Hg), lead (Pb), and titanium (Ti)
 160 measurements, including reference values (SRM 2709a - NIST), mean measured values (n = 8),
 161 reproducibility (%), and detection limits.

Element	Reference values	Mean measured values (n = 8)	Recovery (%)	Detection limit (n = 8)
<i>Hg</i> (mg kg ⁻¹)	0.9 ± 0.2	0.73 ± 0.1	82	0.004
<i>Pb</i> (mg kg ⁻¹)	17.3 ± 0.1	15.2 ± 13.5	88	0.01
<i>Ti</i> (%)	0.336 ± 0.007	0.309 ± 0.614	92	0.001

162

163

164 2.4 Data analysis

165 2.4.1 Trace metal accumulation rates

166 Trace metal accumulation rates (AR) were calculated for Hg and Pb using the
 167 equation (Eq. 1) established by Roos-Barraclough et al. (2002, 2006) and Givelet et al.
 168 (2003).

169
$$AR \text{ (mg m}^{-2} \text{ y}^{-1}) = 10 \times [E] \text{ (}\mu\text{g g}^{-1}) \times D \text{ (g cm}^{-3}) \times SR \text{ (cm y}^{-1}) \quad \text{Eq. 1}$$

170 Where, AR represents the accumulation rate, [E] is the measured metal
171 concentration ($\mu\text{g g}^{-1}$), D is the peat density (g cm^{-3}), and SR is the sedimentation rate
172 (cm y^{-1}).

173 **2.4.2 Enrichment factors**

174 Enrichment factors (EF) for Hg and Pb were determined using the formula from
175 Shotyk et al. (2002):

$$176 \quad \text{EF} = ([\text{E}]_{\text{sample}} / [\text{Ti}]_{\text{sample}}) / ([\text{E}]_{\text{crust}} / [\text{Ti}]_{\text{crust}})$$

177 Here, EF represents the enrichment factor, [E]_{sample} is the concentration of the
178 trace element in the sample ($\mu\text{g g}^{-1}$), [Ti]_{sample} is the titanium concentration in the
179 sample ($\mu\text{g g}^{-1}$), [E]_{crust} is the reference concentration of the trace element in the Earth's
180 crust ($\mu\text{g g}^{-1}$), and [Ti]_{crust} is the reference concentration of titanium in the Earth's crust
181 ($\mu\text{g g}^{-1}$).

182 Since local reference concentrations for Hg and Pb were unavailable, crustal
183 reference values from Wedepohl (1995) were employed.

184 **2.4.3 Lithogenic background estimation**

185 To estimate the lithogenic contribution (natural background) to the total Hg and Pb
186 concentration, the following equation based on Shotyk et al. (2000) (Eq. 2)

$$187 \quad [\text{E lithogenic}] = [\text{Ti}]_{\text{sample}} \times ([\text{E}]_{\text{crust}} / [\text{Ti}]_{\text{crust}}) \quad \text{Eq. 2}$$

188 In this equation, [E lithogenic] represents the estimated lithogenic Hg or Pb
189 concentration ($\mu\text{g g}^{-1}$), [Ti]_{sample} is the titanium concentration in the sample ($\mu\text{g g}^{-1}$),
190 [E]_{crust} is the reference concentration of Hg or Pb in the Earth's crust ($\mu\text{g g}^{-1}$), and [Ti]
191 crust is the reference concentration of titanium in the Earth's crust ($\mu\text{g g}^{-1}$).

192 **2.4.4 Statistical Analyses**

193 All statistical analyses were performed in the R environment (R Core Team, 2024)
194 using the RStudio interface (Posit Team, 2024). Descriptive statistics (mean, median,
195 standard deviation, minimum, maximum, and coefficient of variation) were calculated for
196 each variable (peat characterization parameters, Hg, Pb, Ti, TOC, and N) per study area.

197 A Principal Component Analysis (PCA) was conducted to explore the relationships
198 among variables and to assess differences in geochemical signatures between Itatiaia
199 National Park (INP) and Serra dos Órgãos National Park (SONP). Prior to PCA, the data
200 were normalized (mean-centered and scaled to unit variance) using the prcomp function
201 from the stats package (R Core Team, 2024). The results were visualized using the first

202 two principal components, which together explained 69.4% of the total variance (to be
203 filled in Results), employing the ggbiplot function (ggbiplot package, Vu, 2011). To
204 evaluate whether the two areas occupy distinct multivariate spaces, 95% confidence
205 ellipses were drawn around the centroid of each study area.

206 **2.5. Sedimentation rates and chronology**

207 **2.5.1 ²¹⁰Pb Dating**

208 Lead-210 (²¹⁰Pb) activity was measured to establish sedimentation rates and
209 develop a chronology for the peat cores. This method is widely used for peat studies
210 because it assumes minimal post-depositional remobilization of trace metals once
211 deposited (Mróz et al., 2017). ²¹⁰Pb activity was determined using gamma spectrometry
212 with a high-purity germanium well detector (Canberra) coupled to a multichannel
213 analyzer (Cutshall et al., 1983). Refer to Lourençato et al. (2017) for further details on the
214 radioisotope dating procedure employed.

215 **2.5.2. Sedimentation rate calculation**

216 The Constant Initial Concentration (CIC) model developed by Appleby and
217 Oldfield (1983) was used to calculate accumulation rates (AR) based on the ²¹⁰Pb activity
218 profiles. Sedimentation rate (SR) was then determined using the following equation:

$$219 \quad SR \text{ (cm y}^{-1}\text{)} = \lambda / S$$

220 Where:

221 SR is the sedimentation rate (cm y⁻¹), λ is the decay constant of ²¹⁰Pb (0.0307 y⁻¹),
222 S is the slope of the linear regression of depth plotted against the logarithm of unsupported
223 ²¹⁰Pb activity (Bq g⁻¹).

224 **2.5.3. Peat accumulation rate**

225 The peat accumulation rate (PAR) was calculated using the following equation:

$$226 \quad PAR \text{ (g cm}^{-2} \text{ y}^{-1}\text{)} = SR \text{ (cm y}^{-1}\text{)} \times BD \text{ (g cm}^{-3}\text{)}$$

227 where:

228 PAR is the peat accumulation rate (g cm⁻² y⁻¹), SR is the sedimentation rate (cm
229 y⁻¹), BD is the bulk density of the peat (g cm⁻³).

230

231 **3. RESULTS**

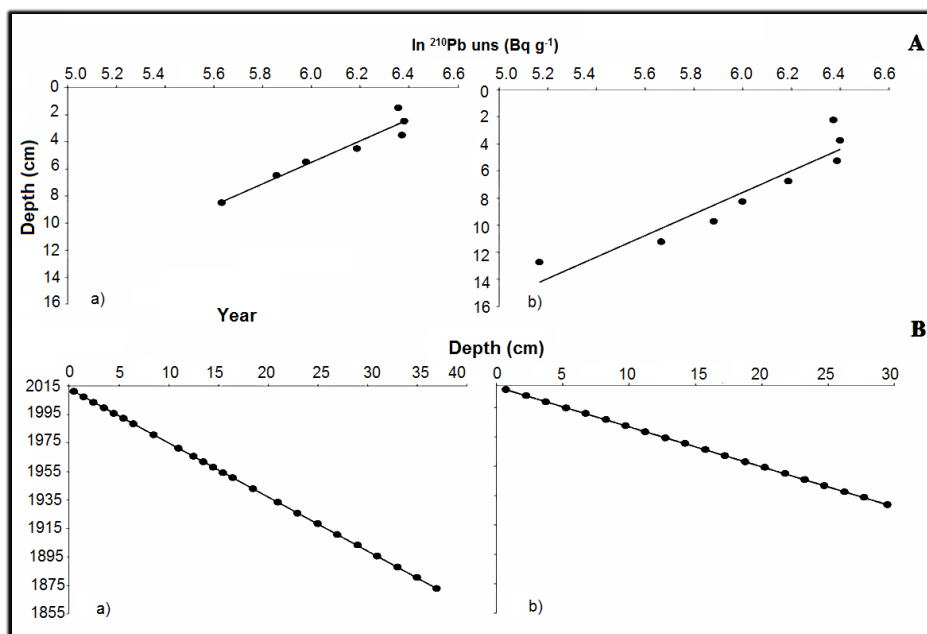
232 The peat soils from both INP and SONP were classified as Typical Haplosaprist
233 Histosols according to the Embrapa classification system (2013) (Figure 2). The dominant

234 plant family identified at the peat surface was Poaceae, with additional presence of
235 families such as Myrtaceae, Asteraceae, Euphorbiaceae, Rubiaceae, Leguminosae,
236 Cyperaceae, and Bromeliaceae (identification source: ICMBio, 2013a, 2013b). The peat
237 samples displayed a dark color, with an average Munsell Soil Color Chart value of 3 for
238 brightness and 2 for Chroma.

239 The pH measured in CaCl₂ solution were extremely acidic, ranging from an
240 average of 2.43 to 3.78. The average unrubbed and rubbed fiber contents were 60% and
241 19%, respectively. Gravimetric moisture content exceeded 200% in most peat samples.
242 The average nitrogen content was 0.44 ± 0.26% in INP and 0.45 ± 0.14% in SONP,
243 whereas the average total organic carbon content was 12.69% in INP and 23.34% in
244 SONP.

245 The vertical distribution of ²¹⁰Pb activity in both INP and SONP peat profiles
246 exhibited a relatively well-defined exponential decrease with depth (Figure 3A). This
247 pattern is consistent with the distribution typically observed in natural sediments
248 (Appleby and Oldfield, 1983). In simpler terms, the ²¹⁰Pb activity generally decreased
249 with deeper peat layers. However, it's important to note that subsurface layer (2-4 cm) of
250 both cores showed slightly higher ²¹⁰Pb values compared to the top layers (0-2 cm).

251



252

253 **Figure 3. A)** Excess ²¹⁰Pb activity (calculated using the constant flux model) as a function of depth
254 (cm) for peat cores from (a) Itatiaia National Park (INP) and (b) Serra dos Órgãos National Park
255 (SONP). **B)** Age-depth models derived from linear interpolation between ¹⁴C and ²¹⁰Pb dating
256 points for (a) INP and (b) SONP.

257

258 The chronology reconstructed using the ^{210}Pb profile from INP (41 cm depth)
259 indicated a peat accumulation rate of $0.18 \text{ g m}^{-2} \text{ y}^{-1}$ starting from 1861 (Figure 3B). This
260 rate decreased to an average of $0.08 \text{ g m}^{-2} \text{ y}^{-1}$ around 1950. In contrast, the SONP peat
261 profile (30 cm depth) displayed a more constant peat accumulation rate of $0.10 \text{ g m}^{-2} \text{ y}^{-1}$
262 throughout the profile, with deposition starting around 1934 (Figure 3B). The average
263 sedimentation rates determined using ^{210}Pb were 0.26 cm y^{-1} for INP and 0.37 cm y^{-1} for
264 SONP (Figure 3B).

265 Table 2 provides a statistical summary of the measured parameters in peat profiles
266 from INP and SONP. As a lithogenic element derived from bedrock weathering, titanium
267 (Ti) concentrations varied considerably between the two parks, reflecting their distinct
268 geological substrates. INP, underlain by Precambrian alkaline intrusive rocks (Silva Rosa
269 & Ruberti, 2018), exhibited Ti concentrations ranging from 24.4 to $62.1 \mu\text{g g}^{-1}$ (mean:
270 $45.7 \pm 10.8 \mu\text{g g}^{-1}$). In contrast, SONP, situated on Precambrian granitoid gneisses
271 (Fernandes et al., 2022), displayed a wider range of Ti concentrations (17.9 – $132.0 \mu\text{g g}^{-1}$;
272 mean: $82.1 \pm 36.6 \mu\text{g g}^{-1}$). Interestingly, despite its lithogenic origin, Ti concentrations
273 decreased in the surface layers of both peatlands, suggesting either dilution by organic
274 matter accumulation or post-depositional mobilization. In INP, this decrease began
275 around the 1990s (depth $< 5 \text{ cm}$), while in SONP, it started earlier, around the 1970s
276 (depth $< 15.75 \text{ cm}$). These divergent temporal trends may reflect differences in peat
277 accumulation rates, organic matter inputs, or hydrological regimes between the two sites.
278 Notably, while Ti concentrations decreased toward the surface in both peatlands, Hg
279 concentrations showed the opposite trend, increasing significantly in recent decades
280 (Table 2; Figures 4 and 5). This inverse relationship further supports the interpretation
281 that Hg is derived primarily from atmospheric deposition rather than lithogenic sources
282 or soil erosion. Having established the contrasting behavior of Ti between the two parks,
283 we now examine the temporal trends and enrichment patterns of Hg and Pb, the primary
284 pollutants of interest in this study.

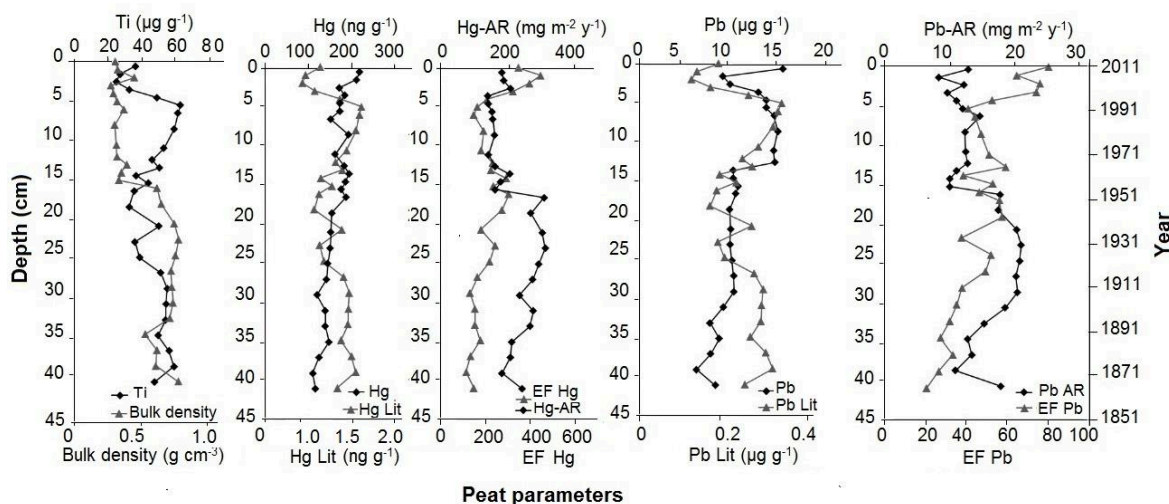
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295 **Table 2.** Statistical summary (median, mean, standard deviation, minimum, and maximum) of the
 296 measured parameters in peat profiles from Itatiaia National Park (INP) and Serra dos Órgãos
 297 National Park (SONP), southeastern Brazil. Parameters include bulk density, total mercury (Hg)
 298 and lead (Pb) concentrations, accumulation rates (AR) for Hg and Pb, titanium (Ti)
 299 concentrations, lithogenic fractions (Hg Lit and Pb Lit), and enrichment factors (EF Hg and EF
 300 Pb). Hg AR is expressed in $\mu\text{g m}^{-2} \text{yr}^{-1}$; Pb AR is expressed in $\text{mg m}^{-2} \text{yr}^{-1}$. Lithogenic fractions
 301 represent the estimated contribution from local rock weathering. Enrichment factors were
 302 calculated using Ti as the reference element, following [referência utilizada, ex.: Shotyk et al.,
 303 1998].

INP	Median	Mean	Std. Dev.	Minimum	Maximum
Bulk Density (g cm^{-3})	0.49	0.52	0.19	0.27	0.79
Hg (ng g^{-1})	158.4	162.2	28.7	110.5	219.4
Hg AR ($\mu\text{g m}^{-2} \text{yr}^{-1}$)	203.3	210.2	59.2	133.4	311.4
Pb ($\mu\text{g g}^{-1}$)	12.2	11.1	2.5	6.6	15.6
Pb AR ($\text{mg m}^{-2} \text{yr}^{-1}$)	14.8	14.8	4.9	6.3	24.9
Ti ($\mu\text{g g}^{-1}$)	48.5	45.7	10.8	24.4	62.1
Hg Lit (ng g^{-1})	1.2	1.1	0.3	0.6	1.5
EF Hg	191.5	215.0	86.5	95.3	429.4
Pb Lit ($\mu\text{g g}^{-1}$)	0.2	0.3	0.1	0.2	0.3
EF Pb	42.5	47.4	13.4	27.7	68.7
SONP	Median	Mean	Std. Dev.	Minimum	Maximum
Bulk Density (g cm^{-3})	0.5	0.5	0.2	0.3	0.8
Hg (ng g^{-1})	156.9	155.6	30.1	102.5	202.7
Hg AR ($\mu\text{g m}^{-2} \text{yr}^{-1}$)	189.5	186.0	40.3	103.8	250.9
Pb ($\mu\text{g g}^{-1}$)	9.3	9.6	2.0	5.0	13.1
Pb AR ($\text{mg m}^{-2} \text{yr}^{-1}$)	10.0	10.1	2.1	5.3	13.9
Ti ($\mu\text{g g}^{-1}$)	85.7	82.1	36.6	17.9	132.0
Hg Lit (ng g^{-1})	1.5	1.5	0.7	0.3	2.4
EF Hg	114.6	130.0	66.2	50.3	321.4
Pb Lit ($\mu\text{g g}^{-1}$)	0.5	0.5	0.2	0.1	0.7
EF Pb	23.9	24.7	14.3	10.2	64.3

304 **Notes:** AR = accumulation rate. Hg AR is presented in $\mu\text{g m}^{-2} \text{yr}^{-1}$; Pb AR is presented in mg m^{-2}
 305 yr^{-1} . EF = enrichment factor (calculated using Ti as the reference element). Lit = lithogenic
 306 fraction.
 307

308 The vertical distribution of Ti, Hg, Pb, and derived parameters in the INP peat core
 309 is presented in Figure 4, revealing distinct trends that reflect both anthropogenic inputs
 310 and natural processes. Concentrations of Hg in the INP peat profile ranged from 110.5 ng
 311 g^{-1} to 219.4 ng g^{-1} , with a mean of $162.2 \pm 28.7 \text{ ng g}^{-1}$ (Figure 4).

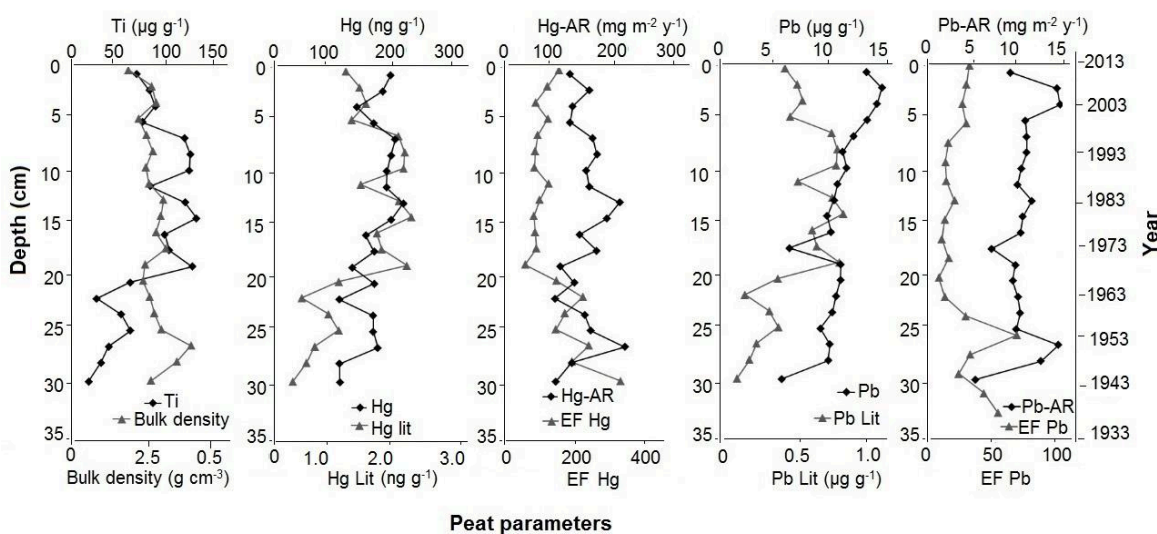


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313 **Figure 4.** Depth profiles of bulk density, Ti, HgLit, HgT, EF Hg, Hg AR, PbLit, PbT, EF Pb, and
 314 Pb AR in the INP peat core. Depths (cm) and corresponding ages (^{210}Pb dating) are shown on the
 315 left axis. Units: bulk density (g cm^{-3}); Ti ($\mu\text{g g}^{-1}$); Hg (ng g^{-1}); Hg AR ($\mu\text{g m}^{-2} \text{yr}^{-1}$); Pb ($\mu\text{g g}^{-1}$);
 316 Pb AR ($\text{mg m}^{-2} \text{yr}^{-1}$).

317

318 Similarly, Hg concentrations in the SONP profile varied between 102.5 ng g^{-1} and
 319 202.7 ng g^{-1} , with a mean value of $155.6 \pm 30.1 \text{ ng g}^{-1}$ (Figure 5). Lead (Pb)
 320 concentrations ranged from 5.0 to $13.1 \mu\text{g g}^{-1}$ (mean: $9.6 \pm 2.0 \mu\text{g g}^{-1}$), while titanium
 321 (Ti) concentrations varied between 17.9 and $132.0 \mu\text{g g}^{-1}$ (mean: $82.1 \pm 36.6 \mu\text{g g}^{-1}$).
 322 Enrichment factors confirmed severe anthropogenic enrichment for both metals, with
 323 mean EF Hg of 130.0 ± 66.2 and mean EF Pb of 24.7 ± 14.3 . Lithogenic fractions were
 324 negligible for both elements (Hg Lit: $0.3\text{--}2.4 \text{ ng g}^{-1}$; Pb Lit: $0.1\text{--}0.7 \mu\text{g g}^{-1}$), further
 325 supporting an atmospheric origin for the observed enrichment.



326

327 **Figure 5.** Depth profiles of bulk density, Ti, HgLit, HgT, EF Hg, Hg AR, PbLit, PbT, EF Pb, and
 328 Pb AR in the SONP peat core. Depths (cm) and corresponding ages (^{210}Pb dating) are shown on
 329 the left axis. Units: bulk density (g cm^{-3}); Ti ($\mu\text{g g}^{-1}$); Hg (ng g^{-1}); Hg AR ($\mu\text{g m}^{-2} \text{yr}^{-1}$); Pb ($\mu\text{g g}^{-1}$);
 330 Pb AR ($\text{mg m}^{-2} \text{yr}^{-1}$).

331 Both INP and SONP profiles exhibited higher Hg concentrations in the top 18.5
332 cm and 15.5 cm layers, respectively. These upper layers, corresponding to the post-1950s
333 period, showed average Hg concentrations of $179.9 \pm 19.5 \text{ ng g}^{-1}$ (INP) and 175.2 ± 20.1
334 ng g^{-1} (SONP). In contrast, deeper layers (pre-1950s) had lower average Hg
335 concentrations of $137.2 \pm 15.2 \text{ ng g}^{-1}$ (INP) and $136 \pm 25 \text{ ng g}^{-1}$ (SONP). This trend likely
336 reflects increased anthropogenic Hg emissions and deposition during the post-industrial
337 era, highlighting the potential ecological and human health risks associated with Hg
338 pollution.

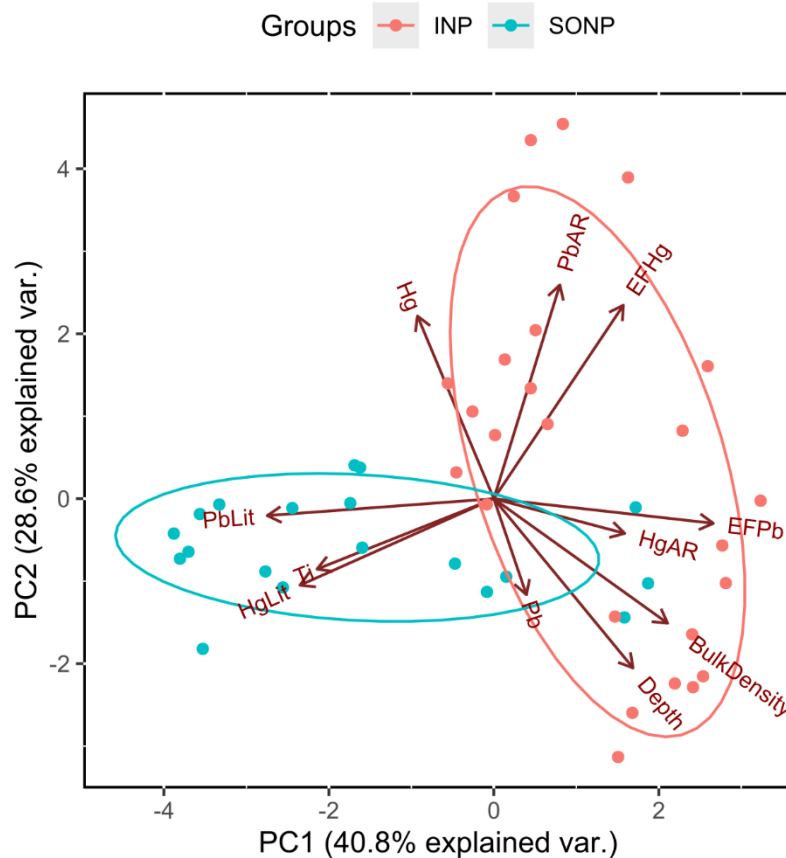
339 The average Hg accumulation rate in the INP peat profile was $210.2 \pm 59.2 \text{ mg}$
340 $\text{m}^{-2} \text{ y}^{-1}$, with a range of 133.4 to $311.4 \text{ mg m}^{-2} \text{ y}^{-1}$ (Figure 4). Notably, despite higher Hg
341 concentrations in surface layers, the accumulation rate was lower at the surface and
342 increased with depth. The SONP peat profile displayed a similar trend. The average Hg
343 accumulation rate was $186 \pm 40.3 \text{ mg m}^{-2} \text{ y}^{-1}$, with a range of 103.8 to $250.9 \text{ mg m}^{-2} \text{ y}^{-1}$
344 (Figure 5). In addition, the Hg accumulation rate was higher in the surface layers (176.3
345 $\pm 33.9 \text{ mg m}^{-2} \text{ y}^{-1}$) compared to deeper sections ($155.4 \pm 45.0 \text{ mg m}^{-2} \text{ y}^{-1}$).

346 Concentrations of Pb in the INP peat profile displayed a similar trend to Hg, with
347 higher values observed in surface layers compared to deeper sections (Figure 4). Lead
348 concentrations ranged from 6.6 to $15.6 \text{ } \mu\text{g g}^{-1}$, with an average of $11.1 \pm 2.5 \text{ } \mu\text{g g}^{-1}$. The
349 highest Pb concentration ($13.6 \pm 2.3 \text{ } \mu\text{g g}^{-1}$) was found in the surface layer (around 12.5
350 cm depth), corresponding to the mid-1960s. In contrast, deeper sections (below 12.5 cm)
351 exhibited a lower average Pb concentration of $9.6 \pm 1.2 \text{ } \mu\text{g g}^{-1}$. The SONP peat profile
352 showed a decreasing trend in Pb concentration with depth (Figure 5). Lead concentrations
353 ranged from 5.0 to $13.1 \text{ } \mu\text{g g}^{-1}$, with an average of $9.6 \pm 2.0 \text{ } \mu\text{g g}^{-1}$. Interestingly, the Pb
354 accumulation rate (AR) in the INP peat profile was lower in the surface layers compared
355 to deeper sections. Surface layers (above 12.5 cm depth) had an average Pb AR of $12.6 \pm$
356 $3.3 \text{ mg m}^{-2} \text{ y}^{-1}$, while deeper layers displayed a higher average Pb AR of $17.3 \pm 3.5 \text{ mg}$
357 $\text{m}^{-2} \text{ y}^{-1}$. The Pb AR in the SONP peat profile remained relatively constant (average of
358 $10.1 \pm 2.1 \text{ mg m}^{-2} \text{ y}^{-1}$) throughout the profile, except for three specific depths (2.25, 3.75,
359 and 26.75 cm) where higher Pb AR values (13.6 , 13.9 , and $13.7 \text{ mg m}^{-2} \text{ y}^{-1}$, respectively)
360 were observed.

361 The Hg enrichment factor (EF) was higher than the Pb EF in both INP (Hg EF =
362 215, Pb EF = 47) and SONP (Hg EF = 130, Pb EF = 25). Both Hg and Pb exhibited a
363 decrease in the lithogenic fraction with increasing depth in both peat profiles (Figures 4

364 and 5). This trend was disrupted only in the uppermost layers (less than 5 cm), where a
365 decrease in the overall concentration of both elements was observed.

366 Despite the similar behavior patterns of Hg and Pb observed in the PCA (Figure 6:
367 variables HgLit and PbLit), the contribution of lithogenic Pb to total Pb concentration in
368 the peat was more significant than that of lithogenic Hg. The mean percentage of
369 lithogenic Pb relative to total Pb was 2.4% in INP and 4.8% in SONP. In contrast, the
370 mean percentage of lithogenic Hg relative to total Hg was only 0.5% in INP and 1.0% in
371 SONP. These results indicate that atmospheric deposition is the dominant source of both
372 elements, with a negligible lithogenic influence for Hg and a minor contribution for Pb,
373 particularly in SONP.



374

375 Figure 6. Principal Component Analysis (PCA) biplot of the geochemical variables measured in
376 peat samples from Itatiaia National Park (INP, red) and Serra dos Órgãos National Park (SONP,
377 blue). The first two principal components (PC1 and PC2) explain 69.4% of the total variance.
378 Ellipses represent 95% confidence intervals around the centroid of each study area. Vectors
379 indicate the contribution of each variable (Hg, Pb, Ti, TOC, N, OM, SBD, pH in CaCl₂, MM,
380 GM, UF, RF, and pyrophosphate solubility) to the principal components. The angle and length of
381 each vector indicate the correlation and contribution strength, respectively.
382

383 Given the low lithogenic contributions observed, the following section discusses
384 the historical trends of atmospheric Hg and Pb deposition recorded in the peat profiles of
385 INP and SONP

386 **4. DISCUSSION**

387 The significant increase in Hg and Pb accumulation rates observed in the INP peat
388 profile after the 1950s warrants further investigation (Figure 4). This pattern might be
389 linked to climate changes influencing biogeochemical processes in the peatland, as
390 previously documented for this region (Groissman et al., 2005; Marengo et al., 2009).
391 Hydrological conditions play a crucial role in peatland development and metal
392 accumulation. Changes in precipitation patterns, evapotranspiration rates, and water table
393 levels can significantly influence the rate of peat accumulation, decomposition rates, and
394 the transport of metals within the peatland system (Taminskas et al., 2018). As reported
395 by Lourençato et al. (2017, 2019), a decrease in peat accumulation rates in this region has
396 been linked to a reduction in bulk density, associated with increased frequency and
397 intensity of extreme rainfall events since the 1950s.

398 Increased precipitation can lead to higher water table levels, promoting peat
399 accumulation and reducing decomposition rates. However, extreme rainfall events can
400 also cause erosion and transport of sediments and associated metals into the peatland.

401 Decreased precipitation or increased evapotranspiration can lower the water table,
402 leading to increased decomposition rates and reduced peat accumulation. This may also
403 result in the oxidation of previously reduced metals, making them more mobile and
404 susceptible to leaching.

405 Alterations in hydrological conditions can affect the redox conditions within the
406 peatland, influencing the speciation and mobility of metals. For example, fluctuating
407 water levels can lead to cycles of oxidation and reduction, which can affect the
408 bioavailability and toxicity of metals. Mesocosm experiments have demonstrated that
409 hydrological changes, such as water table fluctuations, can significantly impact peatland
410 dynamics, including decomposition rates and carbon accumulation (Barel et al., 2021).
411 These findings are consistent with observations from other tropical peatland systems,
412 including the work of Lourençato et al. (2017). While climate change is likely a major
413 driver of these changes, human activities, such as deforestation and pollution, may also
414 contribute to altered hydrological regimes and increased metal deposition. Further

415 research is needed to fully elucidate the complex interplay between climate change,
416 human activities, and peatland ecosystem processes.

417 Titanium (Ti) is widely used as a geochemical normalizer due to its minimal
418 response to water table fluctuations and biological processes (Nieboer, 1982; Hausmann
419 et al., 2018). Accordingly, our data show no correlation between Ti and either Hg or Pb
420 (Figure 6). In peatlands, increasing Ti content often reflects broader landscape changes,
421 such as soil erosion linked to land-use practices or climatic variability (Küttner et al.,
422 2014). The observed increase in Ti concentration within the INP peat profile from the
423 1950s onwards (Figure 4) coincides with the reported intensification of extreme rainfall
424 events in southeastern Brazil (Groisman et al., 2005; Marengo et al., 2009). A similar
425 trend was reported by Shotytk et al. (2002) in a study of Holocene climate change, where
426 elevated Ti concentrations indicated increased mineral input to the peatland. Importantly,
427 the absence of correlation between Ti and Hg or Pb suggests that this Ti increase does not
428 represent an additional atmospheric source of these contaminants but rather enhanced
429 mineral particle transport driven by hydrological changes.

430 We propose that the increase in heavy precipitation events in recent decades has
431 enhanced surface runoff from the slopes surrounding the INP peatland, transporting Ti-
432 rich particulate matter into the site. This influx of mineral matter may also explain the
433 decrease in bulk density observed in the INP profile. In contrast, the SONP peat profile
434 did not exhibit a similar increase in Ti concentration, and our data indicate that Ti was
435 among the main variables distinguishing the two areas (Figure 6). The influence of runoff
436 events on peatland sediment input is strongly dependent on local topography and
437 catchment area size (Price, 2011). Therefore, it is likely that the specific
438 geomorphological and hydrological characteristics surrounding the SONP peatland
439 rendered it less susceptible to the climate-driven changes observed at INP.

440 In contrast to the divergent behavior of Ti, our study revealed a significant increase
441 in Hg concentrations within both INP and SONP peat profiles, particularly in the surface
442 layers (Figures 4 and 5, Table 3). This similar increasing trend for Hg across both parks,
443 despite their contrasting Ti dynamics, strongly suggests a regional-scale atmospheric
444 source for Hg, rather than local runoff or geomorphological controls. These findings are
445 consistent with documented increases in anthropogenic Hg emissions from industrial and
446 urban centers in southeastern Brazil over recent decades, as well as long-range
447 atmospheric transport of Hg from global sources.

448

449 Table 3. Comparison of mercury (Hg) and lead (Pb) concentrations and accumulation rates (AR)
 450 in peatlands from different regions of the world, including values from the present study (INP and
 451 SONP, southeastern Brazil).

REGION / PEATLAND	Hg (ng g ⁻¹)	Hg AR (µg m ⁻² yr ⁻¹)	Pb (µg g ⁻¹)	Pb AR (mg m ⁻² yr ⁻¹)	REFERENCE
Ontario, Canada	189	–	90	–	Givelet et al., 2003
Manitoba, Canada	120–187	405–832	–	–	Outridge et al., 2011
Xiaoxing'an, China	126–275	–	–	–	Liu et al., 2003
Switzerland	–	–	0.8–89.7*	–	Zaccone et al., 2009
Scotland	100–450	–	30–400	–	Yang et al., 2001
Guanabara Bay, Brazil	100–322	–	–	1–18	Covelli et al., 2012
–	38–177*	6–13	26–59	0.98	Coggins et al., 2006
Sweden	10–115	3.9	–	–	Osterwalder et al., 2017
Poland	–	–	91–190	–	Fiałkiewicz-Kozieł et al., 2011
INP (Brazil) – Lacerda & Ribeiro (2004)	20–420	36–120	40–70	8–40*	Lacerda & Ribeiro, 2004
INP (Brazil) – This study	111–219*	133–311*	7–16*	9.4–21*	This study
SONP (Brazil) – This study	103–203*	104–251*	5–13*	8.0–14*	This study

452 **Notes:** AR = accumulation rate. Values marked with an asterisk (*) have been rounded to the
 453 nearest unit. Dashes (–) indicate that the parameter was not reported in the original study. Hg
 454 accumulation rates (AR) are presented in µg m⁻² yr⁻¹ to ensure consistency with the majority of
 455 published studies; original values from references were converted where necessary. See original
 456 references for complete methodological details.

457

458 Overall, the Hg concentrations and accumulation rates measured in INP and SONP
 459 fall within the range of values reported for other ombrotrophic peatlands worldwide
 460 (Table 3). However, the Hg AR values in the present study (133–311 µg m⁻² yr⁻¹ at INP)
 461 are among the highest recorded, particularly when compared to European and North
 462 American sites. Conversely, Pb concentrations (5–16 µg g⁻¹) are relatively low, reflecting
 463 the absence of major local industrial sources within these protected Conservation Units.

464 These findings align with a well-documented global trend of rising Hg
 465 accumulation in peatlands since the early 20th century (Roos-Barraclough et al., 2002;
 466 Küttner et al., 2014; Grimaldi et al., 2008), which is widely attributed to the intensification
 467 of anthropogenic activities, including industrial emissions, fossil fuel combustion, and
 468 agricultural practices. Furthermore, sediment cores from Guanabara Bay (Silva et al.,
 469 2002), located in close proximity to our study sites, have revealed similar increasing

470 trends in Hg accumulation, suggesting a potential influence of human activities from the
471 Rio de Janeiro metropolitan region.

472 While humification processes in peatlands are known to be influenced by
473 atmospheric Hg deposition (Biester et al., 2003; Franzen et al., 2004; Perez-Rodríguez et
474 al., 2015), our findings suggest that this factor may not be the primary driver of Hg
475 accumulation in tropical peatlands. Notably, the Hg concentrations we observed in
476 surface layers (140–200 ng g⁻¹) are comparable to those reported in temperate peatlands
477 (Givelet et al., 2003; Outridge et al., 2011; Liu et al., 2003; Zaccone et al., 2009; Yang et
478 al., 2001) (Table 3). This suggests that other factors, potentially related to regional Hg
479 sources or biogeochemical processes, may play a more significant role in Hg
480 accumulation within these tropical peatlands.

481 The observed increase in Hg AR within the INP peat profile after the 1950s (Figure
482 4) is likely linked to the intensification of human activities, particularly the increased use
483 of fossil fuels and the expansion of industrial activities (Gębka et al., 2016; Yang et al.,
484 2017; EEA, 2018). The Second World War and subsequent post-war industrialization
485 period contributed to elevated atmospheric Hg levels due to the widespread use of
486 mercury in various applications, including explosives and industrial processes.
487 Additionally, regional factors such as gold mining in Brazil, especially the use of mercury
488 amalgamation, have significantly contributed to Hg emissions, particularly in the Amazon
489 region (Lacerda & Marins, 1997). These factors, combined with global atmospheric
490 circulation patterns, have likely resulted in increased Hg deposition in our study areas.

491 Our findings are further supported by research on Hg accumulation rates in
492 Guanabara Bay sediments (Covelli et al., 2012), which documented a significant increase
493 in Hg AR after the 1960s, reflecting intensifying human activities in the Rio de Janeiro
494 metropolitan region. Lacerda et al. (2004) reported a similar trend in Hg AR for lake
495 sediments within the INP region, with a maximum value observed in the 1960s. This
496 concurrence strengthens the evidence for a historical increase in Hg deposition within the
497 area.

498 The results obtained in this study support the hypothesis that the INP region has
499 been significantly impacted by climate change, particularly since the 1950s (Groisman et
500 al., 2005; Marengo et al., 2009). The observed changes in hydrological processes likely
501 played a key role in altering trace element accumulation patterns within the INP peatland.
502 In contrast, the SONP profile did not exhibit a similar response, suggesting that climatic
503 variations in this region may exert a weaker influence on metal accumulation. Indeed, our

504 PCA results show that Hg AR, Pb AR, Hg EF, and Pb EF were important factors
505 distinguishing the two study sites (Figure 6).

506 The percentage of lithogenic Hg in the INP and SONP peat profiles (averaging 0.5%
507 and 1.0%, respectively) falls within the lower range of values reported by Guedron et al.
508 (2006) for French Guiana oxisols (10–25%). Although our lithogenic Hg percentages are
509 lower than those reported for tropical oxisols, this difference is expected given that
510 peatlands are dominated by organic matter accumulation rather than mineral weathering
511 products. Lithogenic metals originate from the weathering of bedrock minerals, and their
512 concentration is typically higher near the surface due to enhanced weathering processes
513 (Grimaldi et al., 2008). This explains the observed decrease in the lithogenic fraction with
514 depth in both peat profiles.

515 The Pb concentrations measured in this study (6.6–15.6 $\mu\text{g g}^{-1}$) are comparable to
516 those reported for peatlands in temperate and boreal regions (1–10 $\mu\text{g g}^{-1}$) (Coggins et
517 al., 2006; Givelet et al., 2003; Jackson et al., 2004; Pratte et al., 2013; Ukonmaanaho et
518 al., 2004; Weiss et al., 2002; Zaccone et al., 2007). The increase in Pb concentrations
519 observed in the SONP profile is likely linked to the historical use of leaded gasoline. Zhan
520 et al. (2020) reported that anthropogenic Pb emissions from fossil fuel combustion
521 contributed up to 75% of global atmospheric Pb contamination during the latter half of
522 the 20th century. This timeframe coincides with the increased Pb levels observed in
523 SONP, suggesting a potential influence of past gasoline usage on Pb accumulation in this
524 peatland. Consistent with this interpretation, SONP samples were mostly located in the
525 negative portion of PC2, a region strongly influenced by Pb (Figure 6). The phase-out of
526 leaded gasoline in Brazil between 1989 and 1992 likely contributed to the subsequent
527 decline in Pb emissions observed globally since the early 2000s (WHO, 2002; Li et al.,
528 2012).

529 Our findings are consistent with Pb accumulation rates reported for peatlands in
530 other regions (Coggins et al., 2006). Additionally, the presence of Pb enrichments in
531 nearby surface sediments of the Guarapina Lagoon (Patchineelam et al., 1988) suggests
532 a potential link between atmospheric Pb deposition and Pb accumulation in these tropical
533 peatlands. Monteiro et al. (2012) reported an increase in Pb AR within Guanabara Bay
534 sediments over the past century, further supporting the historical trend of rising Pb
535 emissions.

536 The EF values for Pb in this study were considerably higher than those reported by
537 Baptista Neto et al. (2000) for Guanabara Bay surface sediments. Based on the

538 classification system proposed by Yongming et al. (2006), our findings suggest "very
539 severe enrichment" of Pb in the studied peatlands, particularly within the INP profile
540 where EF values increased from the 1950s onwards (Figures 4 and 5). This contrasts with
541 the SONP profile, where Pb EF remained relatively constant.

542 It is important to acknowledge limitations associated with EF calculations. The use
543 of continental crustal averages in the absence of local background values can potentially
544 lead to overestimations or underestimations of enrichment (Moura et al., 2019; De
545 Vleeschouwer et al., 2007). Ideally, regional background values specific to the study area
546 would provide a more accurate assessment of anthropogenic Pb influence. However, as
547 highlighted by Arhin et al. (2017), rapid urbanization and industrialization processes
548 make obtaining pristine background values increasingly difficult. Future research efforts
549 should focus on establishing reliable regional background values for Pb in these tropical
550 peatlands, which would allow for a more precise evaluation of anthropogenic Pb
551 enrichment and its impact on the environment.

552 Despite these limitations, the consistent trends observed across multiple
553 independent records (peatlands, lake sediments, and bay sediments) provide compelling
554 evidence for increasing atmospheric deposition of Hg and Pb in southeastern Brazil over
555 the past century, with important implications for the conservation of high-altitude
556 peatlands in the Atlantic Forest biome.

557

558 **CONCLUSION**

559 This study analyzed geochemical records from two high-altitude ombrotrophic
560 peatlands in southeastern Brazil—Itatiaia National Park (INP) and Serra dos Órgãos
561 National Park (SONP) —to investigate the influence of anthropogenic activities and
562 climate change on trace element accumulation (Hg, Pb, and Ti).

563 Both peatlands exhibited a significant increase in Hg concentrations and
564 accumulation rates (AR) in surface layers, with INP showing some of the highest Hg AR
565 values recorded globally ($133\text{--}311\ \mu\text{g m}^{-2}\ \text{yr}^{-1}$). These increases coincide with the rise of
566 industrial activity in Brazil from the 1950s onwards.

567 Low lithogenic contributions (Hg: 0.5–1.0%; Pb: 2.4–4.8%) and enrichment factors
568 indicating "very severe enrichment" confirm that Hg and Pb are primarily of
569 anthropogenic origin, linked to fossil fuel combustion, industrial emissions, and past use
570 of leaded gasoline.

571 While INP showed increasing Pb concentrations, the SONP profile did not exhibit
572 a similar increase after the phase-out of leaded gasoline in Brazil (1989–1992). This
573 discrepancy may reflect regional variations in industrial development, local atmospheric
574 circulation patterns, or site-specific geomorphological characteristics.

575 Ti concentrations increased in INP from the 1950s onwards, coinciding with
576 intensified extreme rainfall events in southeastern Brazil. This suggests that climate-
577 driven hydrological changes have enhanced mineral particle transport into the INP
578 peatland, a trend not observed at SONP due to differences in topography and catchment
579 area.

580 These findings are consistent with historical Hg and Pb emissions associated with
581 post-World War II industrialization and demonstrate that even protected Conservation
582 Units in Brazil are not immune to regional anthropogenic impacts. Future research should
583 focus on: (i) establishing regional background values for trace elements to improve
584 enrichment factor accuracy; (ii) investigating the potential ecological effects of metal
585 deposition on local flora and fauna; and (iii) assessing the long-term impacts of climate
586 change on peatland integrity and carbon storage in tropical high-altitude ecosystems.

587

588 **ACKNOWLEDGEMENTS**

589 We wish to thank to the personnel of the “Parque Nacional de Itatiaia (PNI) e Parque
590 Nacional da Serra dos Órgãos (PARNASO)” of the [Instituto Chico Mendes de](#)
591 [Conservação da Biodiversidade](#) (ICMBio) for the support to develop these studies. E.
592 Silva-Filho, would like to thank the Conselho Nacional de Desenvolvimento Científico e
593 Tecnológico – CNPq (Research Productivity Fellowship, CNPq N ° 06/2019) and
594 FAPERJ research fellows: E. Silva-Filho (State Scientist Program – E-26/203.037/2007);
595 A. Buch (Post-Doctoral Program Grade 10 – E-26/10204.459/2021)

596 **Author contribution** All authors (LFL, EDM, ACB, VTK, RADR, MCB and EVS-F)
597 contributed to the study conception and design. The first draft of the manuscript was
598 written by LFL and all authors commented on previous versions of the manuscript. All
599 authors read and approved the final manuscript.

600 **Funding** This study was funded in part by the Coordenação de Aperfeiçoamento de
601 Pessoal de Nível Superior – Brasil (CAPES) – Finance Code 001 and is an output of the
602 INCT Continent-Ocean Materials Transfer (INCT-TMCOcean supported by Conselho
603 Nacional de Desenvolvimento Científico e Tecnológico - CNPq Proc. No. 573.601/2008-
604 2).

605 **Data Availability** The datasets generated during and/or analysed during the current study
606 are available from the corresponding author on reasonable request.

607 **Declarations**

608 **Competing Interests** The authors declare that they have no known competing financial
609 interests or personal relationships that could have appeared to influence the work
610 reported in this paper. **Ethical approval** Not applicable.

611 **Consent to participate** Not Applicable.

612 **Consent to publish** This manuscript is for exclusive publication in Environmental
613 Pollution.

614

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