

1 **BACTERIAL RECYCLING OF ARCHAEOAL BIOMASS AS A NEW STRATEGY**
2 **FOR EXTREME LIFE IN THE DEAD SEA DEEP SEDIMENT**

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11
12 **Abstract**

13 Archaea and Bacteria that inhabit the deep subsurface (known as the deep biosphere) play a
14 prevalent role in the recycling of sedimentary organic carbon. In such extreme environment,
15 this process can occur over millions of years¹ and requires microbial communities to cope
16 with limited sources of energy. Because of this scarcity, metabolic processes come at a high
17 energetic cost, but the ways heterotrophic microbial communities develop to enable the least
18 energy expenses for a maximized yield remain unclear. Here, we report molecular biomarker
19 evidence for the recycling of archaeal cell wall constituents by bacteria in extreme evaporitic
20 facies of the Dead Sea deep sediments. Isoprenoid wax esters (WE) derived from the
21 recombination of hydrolyzed products of archaeal membrane lipids were retrieved in gypsum
22 and/or halite sedimentary deposits down to 243 meters below the lake floor (mblf), implying
23 the reutilization of archaeal necromass by deep subsurface bacteria. By recycling the building
24 blocks of allegedly better adapted archaea, heterotrophic bacteria build up intracellular carbon
25 stocks and gain access to free water in this deprived environment. This strategy illustrates a
26 new pathway of carbon transformation in the subsurface and how life is maintained in
27 extreme environments experiencing long-term isolation and minimal energetic resources.

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31 In extreme environments, any metabolic process that significantly lowers the energetic cost of
32 life is favored². In such settings, growth rates decrease, and most energy is diverted to
33 maintenance functions³ such as osmotic equilibration, O₂ stress defense, motility or shifts to
34 sustainable metabolic pathways. These selective conditions of life promote the dominance of
35 prokaryotes and generally favor Archaea relative to Bacteria². This is mostly due to the
36 reduced membrane permeability of Archaea, which requires less maintenance energy with
37 respect to bacterial membranes⁴. This advantage is particularly striking in environments
38 characterized by high osmotic stress such as hypersaline environments. There, Bacteria may
39 use alternative strategies that allow competition with allegedly better adapted Archaea, for
40 example by recycling available organic molecules as osmotic solutes⁵. The intracellular
41 accumulation of available organic carbon is a common way of economizing energy in harsh
42 environments. Under stressing conditions of growth, some bacteria are known to accumulate
43 intracellular lipid droplets^{6,7} in the form of polyhydroxyalkanoates, triglycerides or wax
44 esters. The latter can be biosynthesized by condensation of long-chain alkyl lipids present in
45 the sediment or formed during organic matter degradation⁸. However, in environments such
46 as the deep subsurface that are deprived of labile organic carbon⁹, this mechanism may not be
47 a satisfactory option and Bacteria must resort to other strategies to survive.

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49 We here provide unprecedented evidence for an original strategy developed by Bacteria to
50 survive in the deep hypersaline sediment of the Dead Sea. The lipid composition of different
51 sedimentary facies obtained through the Dead Sea Deep Drilling Project showed the presence
52 of significant amounts of isoprenoid WE (up to 0.2 $\mu\text{g}\cdot\text{g}^{-1}$ TOC) in halite and gypsum samples
53 retrieved between 90 and 250 mblf (Fig. 1). Wax esters are a type of energy storage molecules
54 that can be formed and accumulated by Eukarya and Bacteria, particularly in conditions of
55 stress, but that have not been reported in the archaeal domain¹⁰. The formation of WE arises

56 from the condensation of linear or isoprenoid alkyl chains available in the environment and
57 provides easily accessible (intracellular) sources of carbon⁷. The WE detected in the Dead Sea
58 halite/gypsum samples mostly consist of lipid subunits derived from the cell walls of Archaea,
59 the dominant organisms of the lake^{11,12}. The major WE detected was phytanyl phytanate
60 (3,7,11,15-tetramethylhexadecyl-3,7,11,15-tetramethylhexadecanoate; iC_{20} - iC_{20} , Fig. 2)
61 which was accompanied by series of other isoprenoid WE composed of a C_{20} or C_{25}
62 isoprenoid alcohol esterified to a linear, methyl-branched or isoprenoid acyl chain (Fig. 2a
63 and Table S2). The isoprenoid WE were systematically observed with significant amounts of
64 membrane lipids of halophilic archaea -archaeol (2,3-*O*-diphytanyl-*sn*-glycerol) and extended
65 archaeol (2,3-*O*-phytanyl-*O*-sesterterpanyl-*sn*-glycerol)^{13,14}- along with several of their
66 hydrolyzed (and eventually oxidized) products (Figs 1; 2b). Lipids of halophilic archaea are
67 specifically composed of C_{20} and C_{25} isoprenoid alkyl chains which allow for a better control
68 on membrane permeability under strong osmotic conditions compared to bacterial fatty acid
69 membranes^{4,15}. Archaeol was the most abundant isoprenoid alcohol in most of the
70 investigated sediment samples (Fig. 1). Extended archaeol, a membrane lipid specific of
71 halophilic archaea of the Halobacteria class¹³ that dominate the Dead Sea halite and gypsum
72 sediment¹⁶, was also found enriched in halite and gypsum facies. In most of the sedimentary
73 intervals where WE occurred, the core lipids of halophilic archaea (archaeol and extended
74 archaeol) were found in lower abundance than in the other halite/gypsum samples (Fig. 1).
75 Conversely, the degradation products of these archaeal membrane lipids were preferentially
76 observed in the WE-rich intervals (Fig. 1), and mainly consisted of C_{20} and C_{25} isoprenoid
77 alcohols and acids (phytanol, phytanic acid and C_{25} homologues), and of *sn*-1 and *sn*-2
78 isoprenoid C_{20} and C_{25} monoalkyl glycerols (Fig 2b and supplementary data).
79 The report of WE with isoprenoid C_{25} carbon chains (iC_{20} - iC_{25} and iC_{25} - iC_{20}) in natural
80 samples (Fig. 2a and Table S2) is unprecedented and testifies for the use of subunits of core

81 lipids from halophilic archaea as a source for the isoprenoid alkyl and acyl chains present in
82 the WE. The formation of isoprenoid WE has been previously demonstrated during growth of
83 bacteria on free isoprenoid compounds such as phytane, phytol or squalene^{8,17,18}. However,
84 such isoprenoid compounds were not observed together with the isoprenoid WE from the
85 Dead Sea. Together, our results constitute an indisputable proof for the reutilization of
86 archaeal membrane lipids to form isoprenoid WE.

87 The characterization of isoprenoid WE has been shown to constitute a useful tool to decipher
88 the metabolic pathways employed by microorganisms for the degradation of isoprenoid
89 substrates^{8,17,18}. The condensation of isoprenoid compounds with themselves or with their
90 degradation products to form isoprenoid WE indeed allows the trapping of transient
91 metabolites which, under their free form, would not necessarily be preserved. The structural
92 analysis of the different WE present in the Dead Sea halite samples allowed to decipher the
93 metabolic pathways involved in the degradation of archaeal core lipids. Following the
94 hydrolysis of the ether linkages present in archaeol and extended-archaeol, part of the released
95 isoprenoid C₂₀ and C₂₅ alcohol moieties is being oxidized to the corresponding isoprenoid
96 fatty acids, providing the building blocks for the biosynthesis of iC₂₀-iC₂₀, iC₂₅-iC₂₀ and iC₂₀-
97 iC₂₅ WE. The isoprenoid fatty acids may then be further degraded to shorter chain acids via
98 successive α -oxidation, β -oxidation and β -decarboxymethylation sequences^{8,18} (Fig. S2). The
99 identification of iC₅-iC₂₀ WE in some samples demonstrates the almost complete degradation
100 of the isoprenoid alkyl chains released from the hydrolysis of archaeol and extended archaeol
101 to provide carbon and energy to the deep biosphere.

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103 Current knowledge suggests that the ability to form and accumulate WE has only arisen in
104 bacterial and eukaryal domains¹⁰. Eukaryotic life in the Dead Sea has been constrained to
105 humid intervals of the glacial periods and occasional blooms triggered by high precipitation

106 during interglacial stages^{16,19}. Hence, isoprenoid WE present in the most arid intervals of the
107 Dead Sea sediment (halite/gypsum) cannot originate from eukaryotes. The occurrence of
108 specific bacterial methyl-branched C₁₇ fatty acids in the sedimentary intervals where WE
109 were preferentially produced (Fig. 1) rather supports a bacterial origin for the isoprenoid
110 WE²⁰. Such methyl-branched C-odd fatty acids are common in sulfate-reducing bacteria²¹.
111 Our data therefore support the development of bacteria upon archaeal necromass in the
112 extreme environment of the Dead Sea subsurface.

113 The concentration of intracellular WE in Bacteria has been shown to rise particularly in
114 situations of nutrient starvation, especially when nitrogen is limiting²². Additionally, the
115 recycling of archaeal biomass by Archaea themselves was suggested to minimize the energy
116 cost of life in nutrient- and carbon-limited environments²³. We here suggest that the bacterial
117 recycling of archaeal necromass as storage lipids like WE constitutes a way to save and store
118 energy in nutrient- and energy-demanding hypersaline environment of the Dead Sea. The
119 presence of WE with mixed linear and isoprenoid chains alongside the major isoprenoid WE
120 indicates that the WE-forming bacterial population also recycled bacterial and/or eukaryal
121 lipids, in addition to the archaeal ones (Figs. 1, 2 and 3). As a result, bacteria create easily
122 accessible carbon stocks from the necromass, in a lake with very little allochthonous inputs
123 and rare primary production²⁴.

124 The accumulation of WE has been previously evidenced in temporarily emerged microbial
125 mats from hypersaline environments²⁵, and suggested to allow for a better survival of
126 microbial cells during periods of desiccation, due to the release of H₂O from esterification.
127 The Dead Sea sediments that bear WE originate from the deepest part of the lake, where no
128 trace of desiccation has been observed. These sedimentary levels correspond, however, to
129 periods where salinity was the highest in the deep brine, as supported by the bromide
130 concentration curve (Fig. 1). The latter is interpreted as a good estimate of the degree of

131 dilution/concentration of the Dead Sea water column during the Quaternary²⁶. The WE levels
132 generally correspond to the highest concentrations of this conservative element (Fig. 1; Table
133 S1) observed during the driest climatic intervals (Marine Isotopic Stages 5E, early 5A and 2/1
134 transition). Water availability is a major issue not only in dry environments, but also in
135 hypersaline settings due to high concentrations of salts. By recombining hydrolyzed moieties
136 of core lipids originating from the buried archaeal necromass, the WE-forming bacterial
137 population creates accessible water molecules that favor their survival.

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139 The conditions under which the early Holocene or Pleistocene halite have deposited are
140 expected to be similar to the present-day environment that allows halite precipitation from the
141 Dead Sea water column²⁶. If such conditions were favorable to the formation of WE, the
142 occurrence of these lipids would also be expected in the most surficial halite sediment. Since
143 this is not the case, the WE biosynthesis likely occurred deeper in the sedimentary column.
144 The formation of WE has been essentially documented for aerobic microorganisms and
145 environments^{8,27,28}, but their biosynthesis by anaerobic microorganisms has already been
146 recognized^{29,30}. Hence, the unique occurrence of isoprenoid WE in deep halite/gypsum
147 deposits (dated from the late Marine Isotopic Stage 5 to the early Holocene) is suggestive of
148 their production within the anoxic sediment.

149 Wax esters are generally poorly preserved in the sedimentary record²⁷, although some of these
150 compounds have been reported in ca. 40 ka-old lacustrine sediments³¹. The age of the Dead
151 Sea sediments containing the isoprenoid WE has been estimated between 120 and 11.4 ka
152 (Table S1) based on U-Th on aragonite³² and ¹⁴C dating³³. This supports either the
153 extraordinary preservation of labile organic matter in the Dead Sea subsurface, or the post-
154 depositional biosynthesis of these compounds by active bacterial communities until 243 m
155 below lake floor.

156 The present investigation illustrates the high plasticity of Bacteria and their ability to use
157 varied strategies for energy production and preservation under harsh conditions. By studying
158 an environment that pushes life to its limits, we catch a glimpse of the processes that fuel life
159 in the deep subsurface, and add a new loop to the sedimentary carbon cycle.

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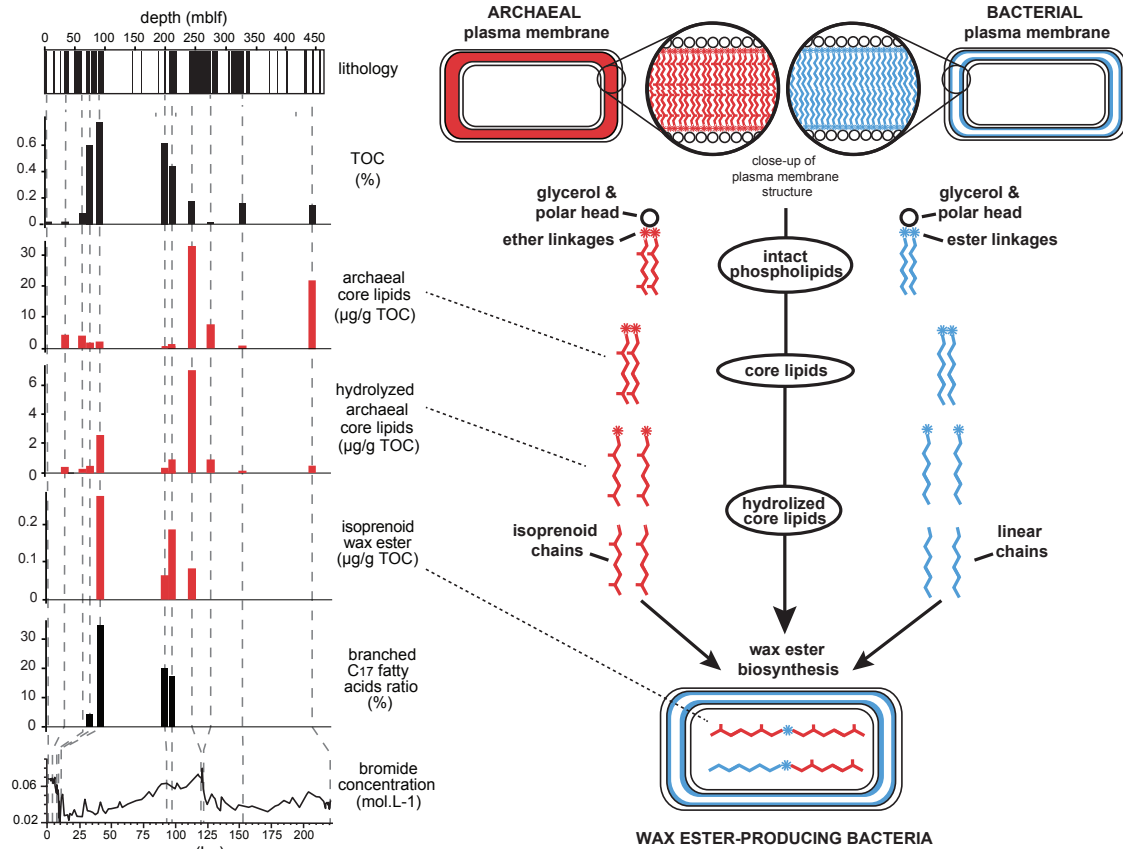
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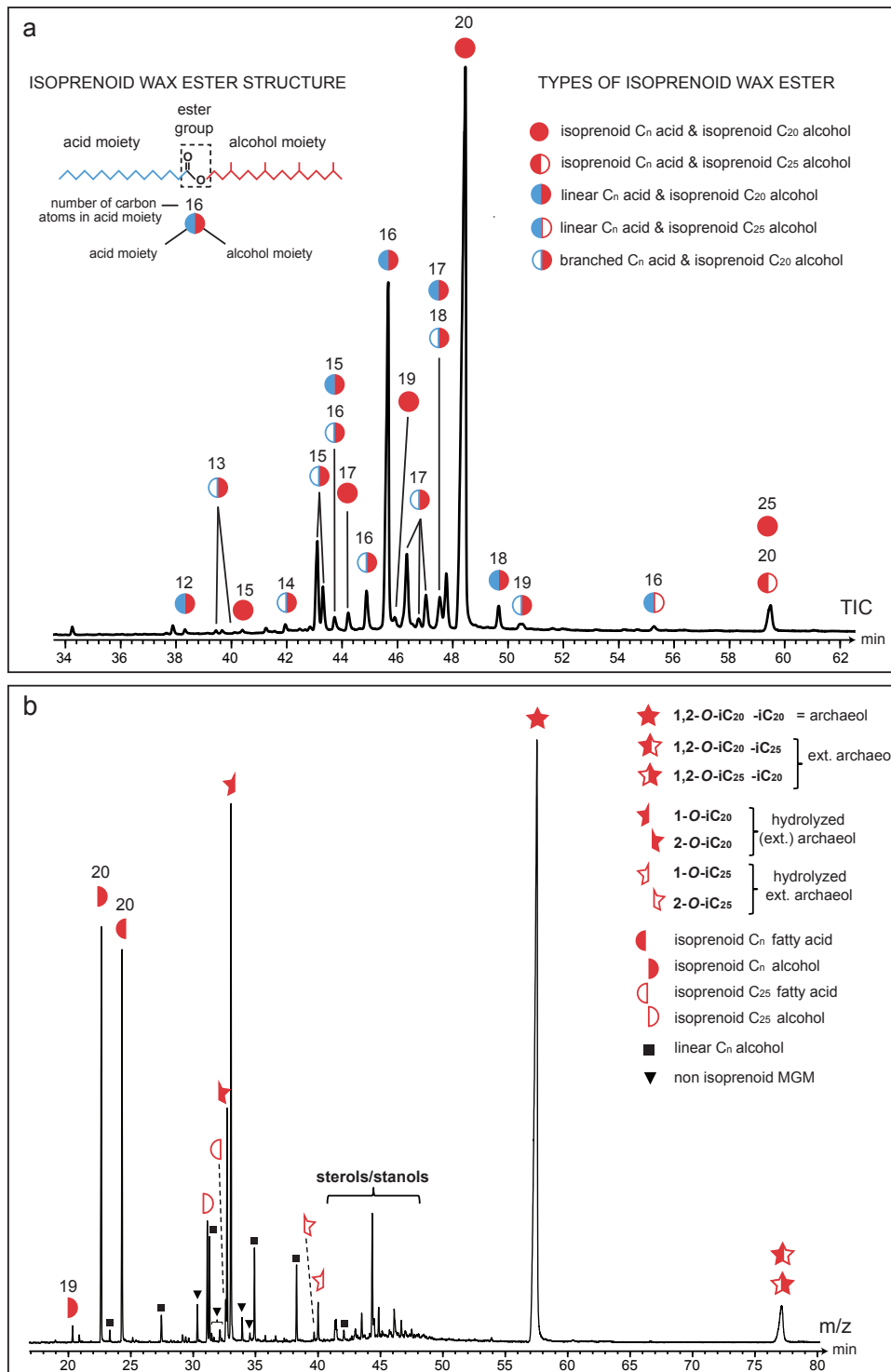
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256 The authors declare no conflict of interest.

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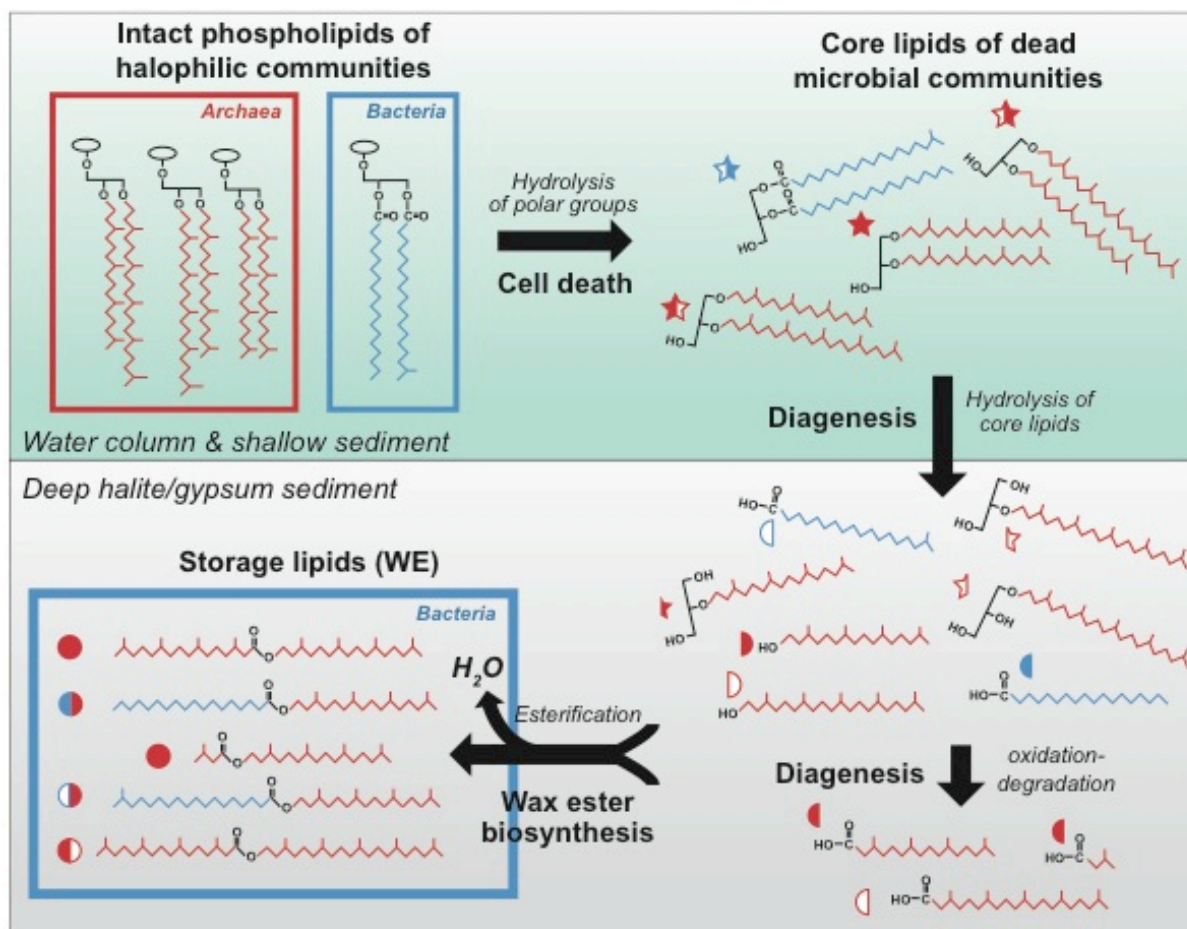
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Fig. 1: Occurrence of wax esters (WE) and other lipid biomarkers along the Dead Sea core (gypsum and halite intervals are blacked, broken lines show the measured samples). The formation of the building blocks of WE are schematized on the right hand side of the panel and linked to their quantitative profiles. The branched C₁₇ fatty acids ratio indicates the presence of specific bacteria likely involved in WE-production. The sampled layers are compared to bromide concentrations in the pore water of the Dead Sea [used as a proxy for lake water dilution/concentration²⁶] and fitted to ¹⁴C ages³⁴ and U-ages³². Red items indicate an archaeal origin, and blue items a bacterial origin.



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Fig. 2: (a) Total ion chromatogram (TIC) of the wax ester fraction from the gypsum sample at 91.04 mblf and (b) selected ions chromatogram of a polar lipid fraction from the gypsum sample at 243.22 mblf. These intervals were selected as they show the highest concentrations of isoprenoid WE and hydrolyzed core lipids, respectively. MGM: monoalkyl glycerol monoethers. Red symbols indicate an archaeal origin and blue symbols a bacterial and/or eukaryal origin. The numbers on top of symbols shown in (b) correspond to the number of carbon atoms in the chain.



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Fig. 3: Schematic pathway for the formation of isoprenoid wax esters by bacteria recycling membrane lipids of halophilic organisms in extreme sediments from the Dead Sea. Intact phospholipids from archaeal and bacterial communities are degraded and some of the degradation products are transformed into WE by bacterial communities of the deep halite/gypsum sediment. Symbols refer to molecules identified in Fig. 2. For the degradation products, moieties in red have an archaeal origin and blue moieties have a bacterial origin.

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300 **Methods**

301 During the drilling campaign, sediments were sampled from core catchers using sterile tools,
302 and kept in the freezer until further processing. The main characteristics of samples used for
303 lipid biomarkers are given in Table S1. Samples were freeze-dried, ground and extracted
304 through sonication cycles (methanol x2, methanol/dichloromethane (1:1, x2) and
305 dichloromethane x3). Elemental sulfur was removed with activated copper. Lipid extracts
306 were filtered out and separated over an inactivated column of silica gel (5% H₂O) into five
307 fractions of increasing polarity. Fraction F1 was eluted with hexane/dichloromethane (9:1),
308 fraction F2 with hexane/dichloromethane (1:1), fraction F3 with dichloromethane, fraction F4
309 with ethyl acetate and fraction F5 with methanol. Fractions F3 and F4 were silylated with
310 pyridine/BSTFA 2:1 (v/v). Fraction F5 was trans-esterified by incubating at 60°C overnight
311 with toluene (0.5 ml) and 2% H₂SO₄ in methanol (2 ml). NaCl 5% and
312 hexane:dichloromethane were then added and the organic upper phase was extracted three
313 times, washed with NaHCO₃ (2%) and dried with sulfate. The dry extract was then silylated
314 with pyridine/BSTFA 2:1 (v/v) before analysis by gas chromatography-mass spectrometry
315 (GC-MS). GC-MS analyses were performed on a HP 6890 Series Plus gas chromatograph
316 equipped with a cool on-column injector and coupled to an Agilent 5975C (VL MSD) mass
317 spectrometer. The samples were injected at 60°C and the oven temperature was programmed
318 as follow: 60 °C isotherm for 30 sec, 20 °C /min to 130 °C, 5 °C /min to 250 °C and 3 °C/min
319 to 300 °C. Identification of compounds was based on interpretation of their mass spectral
320 characteristics and comparison of their mass spectra and retention times with those of
321 authentic standards or literature data. Quantification was realized using stepwise dilution of
322 external standards of alcohol, ester and alkanes. Peak areas were integrated manually using
323 the ChemStation software. Isoprenoid fatty acids are less polar than linear acids and partly
324 eluted in the alcohol fraction.

325 Analyses of C organic content were determined using an Elementar Vario Micro Cube
326 coupled to a Thermal Conductivity Detector. Between 1 and 8 mg of bulk samples were
327 weighted into tin capsules (Elemental Microanalyses, 11.5*7mm) with a CPA26P Sartorius
328 microbalance (2.10^{-6} g). The capsules were individually introduced into a combustion furnace
329 (950°C) with an excess of oxygen. Copper oxide was used as oxidation catalyst and He as
330 carrier gas. Reduction of N_xO_y to N_2 and removal of excess O_2 was achieved with reduced
331 copper at 550°C. Water was removed with a phosphorous pentoxide chemical trap. N_2 and
332 CO_2 were separated with a purge and trap desorption column. The working standard, IVA
333 sediment, was measured every ten samples. Normalization was made using a working
334 standard of IVA sediment containing 9.15% of C.

335 Calculation of the branched C_{17} fatty acids ratio (Fig.1) was realized using the ratio of 10-
336 methyl C_{17} fatty acid over the sum of 10-methyl, iso and anteiso C_{17} fatty acids. To enhance
337 the visualization of the alcohol and acid fragments, selected ions of m/z
338 103+133+159+205+218 were added on a single chromatogram (Fig. 2b).

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