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3 **Reconstructing Magma Storage Depths for the 2018**
4 **Kīlauean Eruption from Melt inclusion CO₂ Contents:**
5 **The Importance of Vapor Bubbles**

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17 **Key Points:**

- 18 • Petrological, gaseous and geophysical observations can be reconciled by a
19 model where Fissure 8 was supplied from two storage reservoirs (~1–2 and
20 3–5 km depth)
- 21 • Extensive post-entrapment crystallization of melt inclusions within High-Fo
22 olivines (Fo>81.5) caused ~90% of the CO₂ to enter the vapor bubble.
- 23 • Raman analyses of vapor bubbles combined with choice of a suitable H₂O-
24 CO₂ solubility model is required to accurately determine magma storage
25 depths.

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Abstract

The 2018 lower East Rift Zone (LERZ) eruption and the accompanying collapse of the summit caldera marked the most destructive episode of activity at Kīlauea Volcano in the last 200 years. The eruption was extremely well-monitored, with extensive real-time lava sampling as well as continuous geodetic data capturing the caldera collapse. This multi-parameter dataset provides an exceptional opportunity to determine the reservoir geometry and magma transport paths supplying Kīlauea’s LERZ. The forsterite contents of olivine crystals, together with the degree of major element disequilibrium with carrier melts, indicates that two distinct crystal populations were erupted from Fissure 8 (termed High- and Low-Fo). Melt inclusion entrapment pressures reveal that Low-Fo olivines (close to equilibrium with their carrier melts) crystallized within the Halema’uma’u reservoir (~ 2 km depth), while many High-Fo olivines ($> \text{Fo}_{81.5}$; far from equilibrium with their carrier melts) crystallized within the South Caldera reservoir (~ 3 – 5 km depth). Melt inclusions in High-Fo olivines experienced extensive post-entrapment crystallization following their incorporation into cooler, more evolved melts. This favoured the growth of a CO_2 -rich vapor bubble, containing up to 99% of the total melt inclusion CO_2 budget (median=93%). If this CO_2 -rich bubble is not accounted for, entrapment depths are significantly underestimated. Conversely, reconstructions using equation of state methods rather than direct measurements of vapor bubbles overestimate entrapment depths. Overall, we show that direct measurements of melts and vapor bubbles by SIMS and Raman Spectroscopy, combined with a suitable H_2O - CO_2 solubility model, is a powerful tool to identify the magma storage reservoirs supplying volcanic eruptions.

Plain Language Summary

Pockets of frozen magma trapped within olivine crystals, termed “melt inclusions”, can provide information about the depths at which magma is stored beneath the surface prior to a volcanic eruption. This is because the amount of CO_2 and H_2O that can be dissolved in a melt is dependent on the pressure, and therefore the depth. We examine melt inclusions from lava flows produced during the 2018 eruption of Kīlauea Volcano. Previous geophysical work has shown that magma is stored in two main reservoirs at Kīlauea, located at ~ 1 – 2 km and ~ 3 – 5 km depth.

58 However, because many melt inclusions host almost all of their CO₂ within a vapor
59 bubble, which is rarely measured, previous petrological estimates of magma storage
60 depths at Kīlauea do not align with the depths of these reservoirs identified by geo-
61 physics. In this study, we measure the amount of CO₂ in the glass and the bubble
62 using Secondary Ion Mass Spectrometry (SIMS) and Raman Spectroscopy respec-
63 tively. By adding these two measurements together, we can reconstruct the amount
64 of CO₂ that was present when melt inclusions were trapped. Calculated depths align
65 remarkably well with geophysical estimates, and demonstrate that the 2018 eruption
66 was supplied by both magma storage reservoirs.

67 1 Introduction

68 The 2018 lower East Rift Zone (LERZ) eruption was the largest and most
69 destructive in the last 200 years of activity at Kīlauea Volcano, Hawai'i (Neal et
70 al., 2019), accompanied by the highest co-eruptive fluxes of SO₂ ever measured at
71 Kīlauea (up to 200 kt a day; Kern et al., 2020; Whitty et al., 2020), and very high
72 lava effusion rates (100-300 m³/s; Neal et al., 2019; Patrick, Orr, et al., 2019). Be-
73 fore the onset of this new eruptive episode in May 2018, Kīlauea had been erupting
74 near-continuously for 35 years on the middle East Rift Zone (ERZ) at Pu'u Ō'ō
75 cone and surrounding vents, located approximately ~20 km east of Kīlauea's summit
76 (1983–2018), and ~24 km uprift of the 2018 eruption site (Fig. 1b). From 2008 to
77 2018, a persistently active lava lake was also present within Halema'uma'u (HMM)
78 pit crater, located in the south west area of Kīlauea's summit caldera (Fig. 1b).

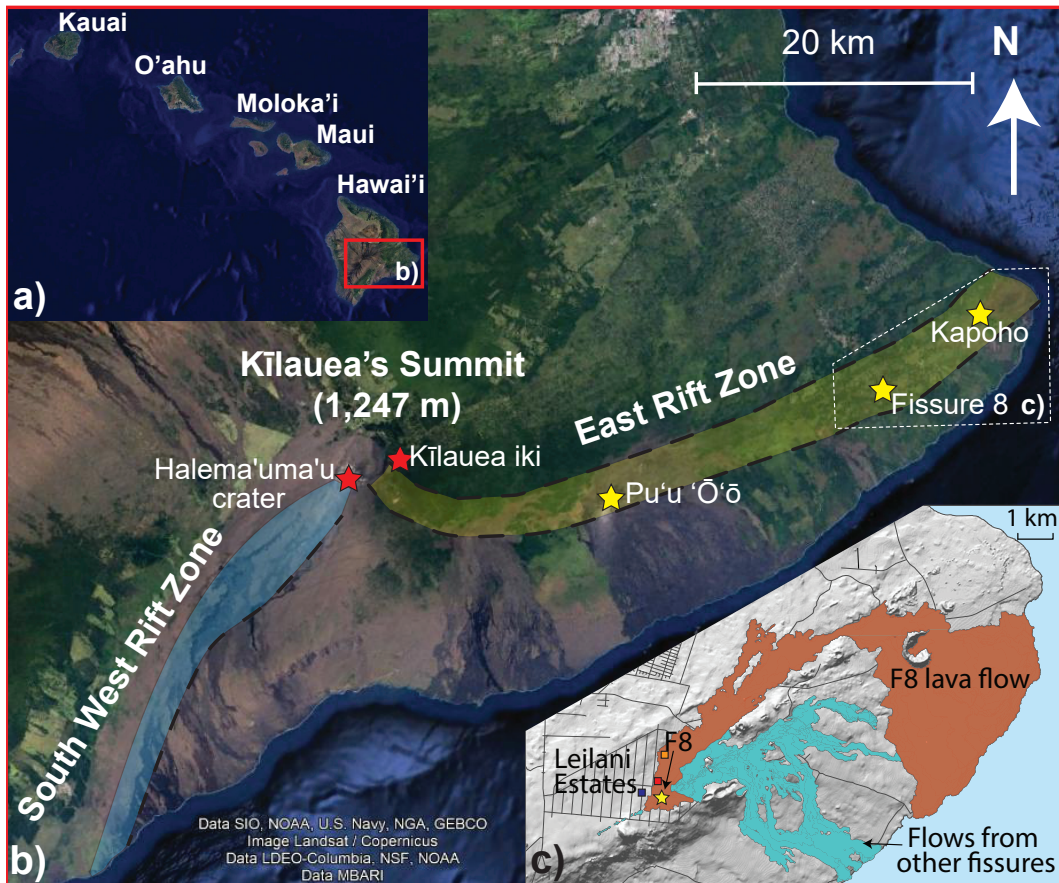


Figure 1. Map of Kīlauea Volcano (b), located on the southwest of the island of Hawai'i (a). Two prominent rift zones radiate from Kīlauea's summit caldera (b). The 2018 eruption occurred within the Leilani Estates subdivision on the lower East Rift Zone (LERZ; expanded region in c). The lava flows from Fissure 8 (marked with a yellow star) are colored deep orange, while flows from Fissures 1–7, and 9–24 are colored light blue. Sample locations are marked with squares (blue=May, 2018, red=July, 2018, orange=Aug, 2018). Base maps for a) and b) are from Google Earth, and the map in c) is adapted from Patrick, Orr, et al. (2019).

79 The 2018 eruption was preceded by swarms of lower-crustal earthquakes at
 80 ~6–12 km depth beneath Kīlauea's summit area on March 7th, April 11th, and
 81 April 18th, 2018 (Flinders et al., 2020). This inflation has been variably interpreted
 82 to result from a short-term increase in magma supply (Flinders et al., 2020), or a
 83 decrease in the output of magma along the ERZ to Pu'u 'Ō'ō, leading to magma
 84 backing up within the summit reservoir (Patrick et al., 2020). On March 13th,
 85 2018, inflation was recorded by tiltmeters located at Kīlauea's summit. Inflationary

86 ground deformation also began at Pu‘u ‘Ō‘ō, suggesting that excess magma was
87 accumulating beneath this vent (Neal et al., 2019). The pressurization at these two
88 locations continued throughout March and April, demonstrated by the rise of the
89 lava pond at Pu‘u ‘Ō‘ō, and overflows of the summit lava lake in mid-late April. On
90 April 30th, the crater floor at Pu‘u ‘Ō‘ō collapsed, followed by an eastward migra-
91 tion of seismicity along the rift zone, consistent with the propagation of a dyke (Neal
92 et al., 2019). A hazard notice released early in the morning of May 1st warned the
93 residents of Lower Puna to be alert, as a large area along the ERZ east of Pu‘u ‘Ō‘ō
94 was at risk from a new outbreak of lava. Following the appearance of ground cracks
95 in the Leilani Estates subdivision (Fig. 1c) on May 2nd, lava reached the surface
96 just before 5 pm on May 3rd (Neal et al., 2019). Over the next few days, multiple
97 fissures opened, preceded by gas emissions and ground cracking. In all, 24 fissures
98 opened between the 3rd and 27th of May 2018.

99 Activity between the 3rd and 9th of May, classified as Early Phase 1 by
100 Gansecki et al. (2019), was characterized by the eruption of spatter mounds and
101 sluggish, slow-moving lava flows. This relatively evolved magma (mean $\text{SiO}_2=51$
102 wt% and $\text{MgO}=4$ wt%; Lee et al., 2019; Gansecki et al., 2019) is thought to have
103 formed by differentiation within LERZ storage reservoirs over decades to centuries
104 (Neal et al., 2019). Throughout May, the compositions of erupted melts and crys-
105 tals became increasingly primitive as summit-derived magma flushed out the LERZ
106 storage reservoirs, with the exception of the involvement of an andesitic composition
107 erupted in mid to late May (Gansecki et al., 2019). The eruption of hotter, less vis-
108 cous lava led to the generation of fast-moving lava flows on May 18th, which reached
109 the coast five days later (Neal et al., 2019, Fig. 1c). By May 28th, activity had lo-
110 calized at Fissure 8 (F8), with the effusion of fast-flowing magma as a channelized
111 flow (Patrick, Dietterich, et al., 2019). Activity ended abruptly on August 4th, by
112 which time F8 had erupted $\sim 1.5 \text{ km}^3$ of lava (Kauahikaua & Trusdell, 2020).

113 Despite the abundant geophysical and geochemical observations made during
114 the LERZ eruption, the source of the magma erupted at F8 from late May-August
115 2018 has not yet been established. It is generally accepted that two main reservoirs
116 are located beneath Kilauea’s summit. The shallower Halema’uma’u (HMM) reser-
117 voir is recognised as an inflation source located beneath the eastern rim of the HMM
118 crater, and is thought to be centred at $\sim 0.5\text{--}2$ km depth (Anderson et al., 2019;

119 Cervelli & Miklius, 2003; Baker & Amelung, 2012; Fiske & Kinoshita, 1969), while
120 the deeper South Caldera (SC) reservoir manifests as an inflation source located
121 beneath the southern portion of the caldera, at $\sim 3\text{--}5$ km depth (Baker & Amelung,
122 2012; Poland et al., 2015). The 2018 LERZ eruption was accompanied by large-scale
123 subsidence of the caldera floor centred around the HMM crater (500 m in certain
124 locations; Neal et al., 2019), which has been attributed to magma withdrawal from
125 the underlying HMM reservoir to feed the effusion of lava from F8 (Anderson et al.,
126 2019). However, recent estimates of the total SO_2 emissions requires the erupted vol-
127 ume to be approximately twice the modelled volume loss from the HMM reservoir,
128 suggesting that a second magma source was involved (Kern et al., 2020).

129 Additionally, the erupted crystal cargo from F8 contained some of the most
130 forsteritic olivines ($\text{Fo}_{88\text{--}89}$) erupted at Kīlauea since 1974, which must have grown
131 in melts with 13–14 wt% MgO (Gansecki et al., 2019). Some of these crystals also
132 contain prominent kink bands (Gansecki et al., 2019), indicating that their crystal
133 lattices have been deformed (Wieser, Edmonds, et al., 2020). Previous work has
134 suggested that highly forsteritic, deformed olivines are derived from the deeper,
135 SC reservoir at 3–5 km depth (Helz et al., 2014, 2015; Wieser et al., 2019; Wieser,
136 Edmonds, et al., 2020), or Kīlauea’s deep rift zones at 6–9 km depth (Clague & Den-
137 linger, 1994; Vinet & Higgins, 2010). Alternatively, Lynn et al. (2017) suggest that
138 highly forsteritic olivines from the Keanakāko’i Tephra may originate from deeper
139 crustal storage reservoirs, perhaps located near the base of the volcanic pile at $\sim 8\text{--}10$
140 km depth.

141 Our study utilizes the strong pressure dependence of the solubility of CO_2 (and
142 H_2O) in silicate melts to determine the pressures at which pockets of melt, termed
143 melt inclusions, were trapped within olivine crystals. Through prior constraints
144 on the density profile of the crust, entrapment pressures from F8 melt inclusions
145 erupted in late May, mid-July and early August 2018 can be converted into entrap-
146 ment depths. In turn, these depths can be compared to geophysical estimates of the
147 depths of the main magma storage regions at Kīlauea to determine the source(s) of
148 magma erupted at F8.

2 Melt Inclusion Entrapment Pressures

2.1 The Importance of Vapor Bubbles

The solubility of pure CO₂ and H₂O in silicate melts is dependent on the pressure, the major element content of the melt, and the melt temperature. Assuming that a melt was saturated in a CO₂-H₂O fluid phase at the time of melt inclusion formation, the pressure at which a melt inclusion was trapped can be calculated by reconstructing its initial volatile and major element composition. In relatively water-poor systems like Kilauea, where melts contain <1 wt% H₂O (Dixon et al., 1991; Clague et al., 1995; Sides, Edmonds, Maclennan, Swanson, & Houghton, 2014; Sides, Edmonds, Maclennan, Houghton, et al., 2014; Tucker et al., 2019; Wallace & Anderson, 1998), the entrapment pressure is most sensitive to the CO₂ content of the melt, and its major element composition. Variations in melt H₂O content between 0–1 wt% have a relatively small effect on the entrapment pressure (except at very low CO₂ contents; see Supporting Information Fig. S1; Newman & Lowenstern, 2002).

However, estimating the CO₂ content of a melt inclusion at the point of entrapment is not straightforward. The host crystal may experience a period of cooling after the melt inclusion was trapped, leading to the growth of olivine on the walls of the inclusion (termed post-entrapment crystallization, or PEC; Roedder, 1984; Danyushevsky et al., 2000; Anderson & Brown, 1993). The precipitation of denser olivine from the silicate melt, combined with the differential thermal contraction of the melt phase and the host olivine, causes the internal pressure of the melt inclusion to drop, driving the growth of a vapor bubble (Roedder, 1979; Anderson, 1974; Anderson & Brown, 1993). Combined with a reduction in the solubility of CO₂ associated with major element changes during PEC, these processes cause CO₂ to migrate from the melt phase into the bubble (Steele-Macinnis et al., 2011; Sides, Edmonds, Maclennan, Houghton, et al., 2014; Maclennan, 2017; Aster et al., 2016). An additional phase of bubble growth is caused by the differential thermal contraction of the melt inclusion and the host olivine during syn-eruptive cooling from high magmatic temperatures (~1150° C at F8; Helz & Thornber, 1987; Gansecki et al., 2019) to the glass transition temperature (~725°C; Ryan & Sammis, 1981).

180 Unfortunately, the vast majority of published volatile contents in melt in-
181 clusions globally, and at Kilauea, only measured CO₂ in the glass phase, using
182 techniques such as secondary-ion mass spectrometry (SIMS), or Fourier transform
183 infrared spectroscopy (FTIR; Bennett et al., 2019; Ruth et al., 2018; Sides, Ed-
184 monds, Maclennan, Houghton, et al., 2014; Sides, Edmonds, Maclennan, Swanson,
185 & Houghton, 2014). Given that recent work has shown that ~40–90% of the total
186 CO₂ budget of melt inclusions may be held within the vapor bubble (Hartley et al.,
187 2014; Wallace et al., 2015; Moore et al., 2015; Rasmussen et al., 2020), entrapment
188 pressures from studies neglecting vapor bubble carbon must be viewed as minimum
189 estimates (Anderson & Brown, 1993; Ruth et al., 2018).

190 **2.2 Reconstructing Vapor Bubble CO₂**

191 Several approaches have been used to explore the contribution of vapor bubbles
192 to the CO₂ budget of Hawaiian melt inclusions. Anderson and Brown (1993) theo-
193 retically reconstruct vapor bubble CO₂ by assuming that the melt and vapor bubble
194 were in chemical equilibrium at high magmatic temperatures prior to syn-eruptive
195 quenching. Specifically, they calculated melt inclusion internal pressures from glass
196 CO₂ contents, and used these pressures to determine the molar volume of CO₂ in
197 vapor bubbles using the CO₂ equation of state (EOS). They converted their molar
198 volumes into CO₂ concentrations assuming that bubbles occupied 0.5 vol% of the
199 melt inclusion prior to quenching, and added these values to measurements of glass
200 CO₂ concentrations. Riker (2005) used a similar method to reconstruct bubble car-
201 bon for melt inclusions from the 1859 eruption of Mauna Loa. However, instead of
202 using a fixed bubble volume, they account for the differential amounts of cooling
203 and PEC experienced by erupted crystals, and calculate the bubble volumes prior to
204 quench-induced expansion as a function of the drop in temperature (ΔT) between
205 the melt inclusion at the point of entrapment and eruption ($\text{VB vol\%} = 0.0162 \Delta T$
206 $- 0.0016$). More recently, Tucker et al. (2019) theoretically reconstructed bubble
207 carbon contents for a large suite of melt inclusions from several Hawaiian volcanoes,
208 including 167 from Kilauea. However, instead of estimating the size of the vapor
209 bubble prior to syn-eruptive quenching as in Anderson and Brown (1993) and Riker
210 (2005), they used observed bubble volumes to convert CO₂ densities obtained from
211 the EOS into bubble CO₂ concentrations. This approach is problematic because ex-

212 pansion of the bubble during syn-eruptive cooling and quenching continues until the
213 glass transition temperature, while CO₂ diffusion through the melt into the bubble
214 may effectively cease at a higher temperature. Thus, the final stages of bubble ex-
215 pansion will occur without concurrent CO₂ diffusion from the glass into the bubble,
216 meaning that the EOS method will overpredict the amount of CO₂ in the bubble
217 (Anderson & Brown, 1993; Maclennan, 2017; Rasmussen et al., 2020).

218 The total amount of CO₂ within melt inclusions can also be determined using
219 experimental homogenization techniques, where crystals containing melt inclusions
220 are heated to magmatic temperatures. This drives the dissolution of the olivine
221 rim precipitated during PEC, which changes the chemistry and volume of the melt
222 inclusion so that CO₂ held within the vapor bubble dissolves back into the melt.
223 Following rapid quenching, the glass phase of these rehomogenized melt inclusions
224 can be analyzed by SIMS or FTIR (Esposito et al., 2012; Rasmussen et al., 2020;
225 Skirius et al., 1990; Tuohy et al., 2016; Wallace et al., 2015). However, experimental
226 homogenization can lead to H₂O loss, excess dissolution of olivine on the walls of
227 the melt inclusion, and loss of mineral and melt inclusion zoning, which degrades
228 the overall utility of the melt inclusion record (Rasmussen et al., 2020; Tuohy et al.,
229 2016). Additionally, it is not always possible to fully dissolve the original bubbles,
230 and new bubbles containing CO₂ may nucleate upon quench (Wallace et al., 2015;
231 Tuohy et al., 2016; Skirius et al., 1990; Rasmussen et al., 2020).

232 Most recently, the density of CO₂ in vapor bubbles has been measured di-
233 rectly using Raman Spectroscopy (Esposito et al., 2011; Steele-Macinnis et al.,
234 2011; Hartley et al., 2014; Moore et al., 2015, 2018; Aster et al., 2016; Taracsák et
235 al., 2019). The Raman spectrum of CO₂ consists of two peaks nominally at 1285
236 cm⁻¹ and 1388 cm⁻¹ at 1 bar (see Supporting Information Fig. S2), resulting from
237 the interaction of a symmetrical stretching mode and an active bending mode in
238 the CO₂ molecule by a process known as Fermi resonance (Rosso & Bodnar, 1995;
239 Lamadrid et al., 2017; Fermi, 1931). Hence, collectively, these peaks are referred to
240 as the Fermi diad (FD), and the distance between the peak centres is the Fermi diad
241 splitting (Δ). However, while it is well accepted that Δ correlates with CO₂ den-
242 sity (ρ_{CO_2}), there are a number of different parameterizations for this relationship
243 in the literature (Wang et al., 2019; Rosso & Bodnar, 1995; Lamadrid et al., 2017;
244 Kawakami et al., 2003, and refs. within). The diversity of published densimeters

245 reflects different instrument hardware, as well as the choice of analytical conditions
 246 (Lamadrid et al., 2017). Thus, the approach used by a number of studies where a
 247 densimeter is chosen from the literature to convert measurements of Δ to ρ_{CO_2} on a
 248 different Raman instrument from the one used to calibrate the densimeter results in
 249 large systematic uncertainties in the absolute density of CO_2 (e.g., Venugopal et al.,
 250 2020; Taracsák et al., 2019; Hartley et al., 2014). For example, $\Delta=102.8 \text{ cm}^{-1}$ yields
 251 $\rho_{CO_2}=0.0281 \text{ g/cm}^3$ using the densimeter of Wang et al. (2019), but $\rho_{CO_2}=0.1397$
 252 g/cm^3 using the densimeter of Kawakami et al. (2003). For a bubble volume of 5%
 253 (the 80th percentile of bubble volume proportions at Kilauea from Tucker et al.,
 254 2019) and a melt density of 2.75 g/cm^3 , these different densimeters predict a con-
 255 tribution of 538 ppm vs. ~ 2674 ppm CO_2 to the reconstructed total CO_2 budget
 256 of the melt inclusion. For a melt inclusion with $SiO_2=49 \text{ wt\%}$, and $H_2O=0.5 \text{ wt\%}$,
 257 these CO_2 contents correspond to entrapment pressures of $\sim 1.2 \text{ kbar}$ vs. 4.8 kbar
 258 (at 1200°C ; Newman & Lowenstern, 2002), and entrapment depths of $\sim 4 \text{ km}$ vs.
 259 $\sim 18 \text{ km}$ respectively for a crustal density of 2700 kg/m^3 . This demonstrates that
 260 the development of an instrument-specific calibration is essential to be able to dif-
 261 ferentiate between lower and upper crustal storage at ocean island volcanoes, let
 262 alone fingerprinting the involvement of different reservoirs identified by geophysical
 263 techniques.

264 An additional source of error affecting both Raman measurements and EOS
 265 methods arises during the conversion of ρ_{CO_2} into the equivalent amount of CO_2 in
 266 ppm held within the vapor bubble ($[CO_2]^{VB}$):

$$[CO_2]^{VB} = 10^6 \times \frac{\rho_{CO_2} V_{VB}}{\rho_{Melt} V_{Melt}} \quad (1)$$

267 Where V_{VB} and V_{Melt} are the volume of the vapor bubble and the melt phase of
 268 the inclusion respectively, and ρ_{Melt} is the density of the silicate melt calculated
 269 here using DensityX (Iacovino & Till, 2019). Total CO_2 contents are obtained by
 270 summing the equivalent amount of CO_2 in the vapor bubble with the concentration
 271 of CO_2 measured in the melt phase ($[CO_2]^{Melt}$) by SIMS or FTIR:

$$[CO_2]^{Tot} = [CO_2]^{VB} + [CO_2]^{Melt} \quad (2)$$

272 The volumes of the vapor bubble and melt inclusion are typically determined
 273 from 2D transmitted light images, estimating the length of the third, unmeasurable

274 dimension from the major and minor axes of the plan view of the inclusion. Tucker
 275 et al. (2019) simulate this process by randomly intersecting ellipses and show that
 276 the smallest errors are achieved by calculating the third dimension as the arithmetic
 277 mean of the two measured axes. However, this approach is still associated with a 1σ
 278 error of -47 to +37% (Tucker et al., 2019). Although important, we note that this
 279 random error is entirely overwhelmed by the systematic error of up to a factor of 4
 280 in literature datasets which have arbitrarily chosen a literature densimeter.

281 To mitigate the systematic error associated with Raman calibration, we de-
 282 termine the relationship between Δ and ρ_{CO_2} for the specific instrument and ac-
 283 quisition conditions used in this study through the analysis of synthetic fluid melt
 284 inclusions with known CO_2 densities. Analysis of both the melt phase (using SIMS)
 285 and the vapor bubble (using a calibrated Raman system) yields the first extensive
 286 dataset critically evaluating the contribution of vapor bubbles to the total CO_2 bud-
 287 get of specific melt inclusions at Kīlauea. Combined with a rigorous examination of
 288 the suitability of different CO_2 - H_2O solubility models, these measurements place
 289 accurate constraints on entrapment depths of olivine-hosted melt inclusions from the
 290 2018 LERZ eruption. This dataset, combined with quantitative models of bubble
 291 growth, also allows assessment of the relative importance of post-entrapment crys-
 292 tallization and syn-eruptive quenching on the partitioning of CO_2 between the melt
 293 and vapor phase. In turn, this allows the accuracy of EOS methods as an alternative
 294 to direct measurements of ρ_{CO_2} using Raman Spectroscopy to be evaluated.

295 **3 Materials and Methods**

296 **3.1 Sample Details, Preparation and Analytical Methods**

297 We examine three samples erupted at F8 (square symbols; Fig. 1c):

- 298 1. May-18 (erupted May 30th, 2018; USGS code KE62-3293; blue symbols),
 299 comprising vesicular reticulite and scoria which landed in a bucket placed near
 300 the F8 vent ($19^\circ 27.7486'$ N, $154^\circ 54.8636'$ W).
- 301 2. July-18 (erupted Mid-July 2018; red symbols), from the selvages of a
 302 naturally-quenched, and highly vesicular proximal overflow from the F8 chan-
 303 nel (<50 m from the vent; $19^\circ 27.879'$ N, $154^\circ 54.645'$ W).

304 3. Aug-18 (erupted Aug 1st; USGS code KE62-3321F; orange symbols), which
305 was sampled directly from the F8 channel using a metal rod and chain, and
306 rapidly quenched in water. Direct lava sampling took place on a stable chan-
307 nel levee ($19^{\circ} 28.31508' \text{ N}$, $154^{\circ} 54.51426' \text{ W}$), $\sim 700 \text{ m}$ downstream of the
308 position of the July-18 overflow.

309 Samples were jaw crushed and sieved into three size fractions (250–840, 840–
310 1000 and $>1000 \mu\text{m}$). Olivines were picked under a binocular microscope, and in-
311 dividually mounted in CrystalBondTM on glass slides. Care was taken to prepare
312 melt inclusions hosted within olivine crystals from all three size fractions. Melt in-
313 clusions were exposed by grinding with 250–3000 grade wet and dry paper, allowing
314 embayments to be avoided, and melt inclusions containing vapor bubbles to be iden-
315 tified. Melt inclusions without vapor bubbles were ground down with progressively
316 finer wet and dry paper until the center of the inclusion was exposed. Melt inclu-
317 sions containing vapor bubbles were ground down to just above the top of the melt
318 inclusion of interest (to avoid intersecting the bubble, and releasing the trapped
319 CO_2). A photo was taken of the melt inclusion and vapor bubble using a transmit-
320 ted light microscope to allow estimation of melt inclusion and bubble volumes. For
321 larger melt inclusions, two images were acquired: one where the bubble was in focus,
322 and one where the melt inclusion outline was in focus. The outline of the bubble
323 and melt inclusion were traced using ImageJ (Schneider et al., 2012), and a best
324 fit ellipse was fitted to each. Volumes were calculated by assuming that the third
325 (non-measurable dimension) was equal to the arithmetic mean of the two measured
326 dimensions (Tucker et al., 2019). Several melt inclusions contained large spinel crys-
327 tals that were likely co-entrapped. The volume of these spinels (assuming a cuboid
328 shape, with the third dimension also equal to the arithmetic mean of the visible
329 dimensions) was subtracted from the volume of the melt inclusion.

330 Following optical measurements, crystals were ground down until the vapor
331 bubble was within $\sim 30 \mu\text{m}$ of the surface. Depending on the optical quality after
332 fine grinding (using 2000-7000 grade wet and dry paper), melt inclusions were vari-
333 ably polished using $9 \mu\text{m}$ diamond pastes prior to Raman analysis. Raman spectra
334 of vapor bubbles were collected using a confocal LabRAM 300 (Horiba Jobin Yvon)
335 Raman spectrometer in the Department of Earth Sciences at the University of Cam-

336 bridge. The two CO₂ Fermi Diads were fitted with Gaussians (see Supporting Infor-
 337 mation Fig. S4). The relationship between the Δ and ρ_{CO_2} for the specific Raman
 338 acquisition condition used in this study was determined by analyzing 16 synthetic
 339 CO₂ – H₂O fluid melt inclusions with a range of densities (~ 0.04 g/cm³, ~ 0.08
 340 g/cm³ and ~ 0.14 g/cm³) hosted in quartz, as well as three Kīlauean melt inclusion
 341 vapor bubbles. The densities of all 19 of these primary standards were measured
 342 using a JY Horiba LabRam HR in the Fluids Research Laboratory at Virginia Tech
 343 Raman, which has been specifically calibrated for low CO₂ densities using a high-
 344 pressure optical cell (Lamadrid et al., 2017). A linear regression through repeated
 345 measurements of standards yielded the following relationship with 95% confidence
 346 intervals on the regression (see Supporting Information Fig. S3):

$$\rho_{CO_2}(\text{g/cm}^3) = 0.3217 \pm 0.026 \Delta (\text{cm}^{-1}) - 32.995 \pm 2.7 \quad (3)$$

347 Further analytical details are presented in the Supporting Information (Text
 348 S1). Following Raman analyses, individual crystals were ground down to expose the
 349 center of each melt inclusion to maximize the available analyzable area. The bubble
 350 was exposed in approximately half of bubble-bearing inclusions. Following sonication
 351 to remove polishing residue, exposed bubble walls were examined on the FEI Quanta
 352 650FEG SEM at the University of Cambridge in low vacuum mode prior to the ap-
 353 plication of any coatings. Crystals were then mounted in epoxy in groups of 20–40,
 354 and polished with progressively finer diamond pastes (9, 6, 3, 1, 0.25 μm).

355 Following the application of a gold coat, the concentrations of H₂O and CO₂
 356 (as well as MgO and SiO₂ for normalization) in melt inclusions and co-erupted
 357 matrix glasses were determined using the Cameca IMS-7f GEO at the NERC Ion
 358 Microprobe Facility, University of Edinburgh. SIMS analysis was performed prior to
 359 EPMA analysis to avoid volatile migration under the electron beam, and to avoid
 360 contamination of measured carbon concentrations by a carbon coat. Epoxy stubs
 361 were placed in the sample chamber at vacuum for a minimum of 6 hours before
 362 analysis to allow them to outgas. A wide variety of standards were analyzed to cre-
 363 ate calibration curves for H₂O and CO₂ (N71, M10, 519-4-1, M5, M40, M36, M21,
 364 M47, M36; see Supporting Information S5; Shishkina et al., 2010; Hauri, 2002).
 365 Additional information regarding calibration, background and drift corrections are
 366 provided in the Supporting Information (Text S2).

367 Following SIMS analyses, the Au coat was removed by polishing on a 0.25 μm
 368 diamond polishing pad, and a carbon coat was applied for electron microprobe an-
 369 alyzer (EPMA) analyses. Spot analyses of melt inclusions, matrix glasses and host
 370 olivines were obtained using a Cameca SX100 EPMA in the Department of Earth
 371 Sciences, University of Cambridge following the two-condition analytical set up de-
 372 scribed in Wieser et al. (2019). Spectrometer configurations, count times, calibration
 373 materials, and estimates of precision and accuracy calculated from repeated analyses
 374 of secondary standards (San Carlos Olivine, VG2 and A99; Jarosewich, 2002) are
 375 presented in the Supporting Information (Text S3, Tables S2-4).

376 Melt inclusions were corrected for the effects of post-entrapment crystalliza-
 377 tion using the Olivine MI tool in Petrolog3 (Danyushevsky & Plechov, 2011). This
 378 requires the user to specify the initial FeO_T and the host Fo content of each inclu-
 379 sion. FeO_T was set at 11.33 wt% for melt inclusions hosted in olivines with forsterite
 380 contents ($[\text{Fo}=\text{Mg}^{2+}/(\text{Mg}^{2+}+\text{Fe}^{2+}) \text{ atomic}]>79 \text{ mol\%}$ based on the liquid line of
 381 descent at Kilauea, and for consistency with previous studies (Wieser et al., 2019;
 382 Sides, Edmonds, Maclennan, Swanson, & Houghton, 2014). For olivine crystals with
 383 $\text{Fo}<79 \text{ mol\%}$, the initial FeO content was estimated from the relationship between
 384 the equilibrium olivine forsterite content and melt FeO_T contents in a fractional
 385 crystallization model computed in MELTS for MATLAB (Supporting Information
 386 Fig. S5 Antoshechkina & Ghiorso, 2018).

387 4 Results

388 F8 melt inclusions are hosted in olivine crystals with a wide range of core com-
 389 positions (Fo_{77-89} ; Fig. 2a). Core compositions in all three samples show a peak
 390 at $\sim\text{Fo}_{88-89}$ (Fig. 2b-d), which lies significantly above the equilibrium field cal-
 391 culated from the Mg# of co-erupted matrix glasses [$\text{Mg\#}=\text{Mg}^{2+}/(\text{Mg}^{2+}+\text{Fe}^{2+})$,
 392 atomic], even considering a wide range of experimentally-determined values for
 393 $K_{D_{\text{Fe}^{2+}}^{\text{ol-melt}}}$ (black lines, Fig. 2a; 0.270–0.352; Roeder & Emslie, 1970; Matzen et
 394 al., 2011). Fourteen melt inclusions from May-18, but only six melt inclusions from
 395 July-18 and one from Aug-18 are hosted in olivines which lie within the equilibrium
 396 field. F8 olivines have some of the highest Fo contents ever reported at Kilauea (Fig.
 397 2a-d vs. Fig. 2e-f; Sides, Edmonds, Maclennan, Swanson, & Houghton, 2014; Wieser
 398 et al., 2019), but relatively low carrier melt Mg#s (51–57 mol%; assuming $\text{Fe}^{3+}/\text{Fe}_T$

399 = 0.15). In turn, this juxtaposition produces some of the most extreme degrees of
400 olivine-carrier melt Fe-Mg disequilibrium seen at Kīlauea (Fig. 2a). Crystals with
401 high forsterite cores show strong normal zoning, while those with core compositions
402 plotting closer to the equilibrium field on Fig. 2a are not visibly zoned in rapid EDS
403 acquisitions (see Supporting Information Figs. S7-9).

404 The majority of F8 melt inclusions exhibit lower measured FeO_T contents
405 than co-erupted matrix glasses and the composition of Kīlauean melt inclusions
406 from the literature (grey dots; Wieser et al., 2019; Tucker et al., 2019; Sides, Ed-
407 monds, MacLennan, Houghton, et al., 2014; Sides, Edmonds, MacLennan, Swan-
408 son, & Houghton, 2014). Melt inclusion MgO contents are more similar to those of
409 co-erupted matrix glasses (Fig. 3a). Following a correction for the effects of post-
410 entrapment crystallization, F8 melt inclusions have MgO contents between 6.4 and
411 13.7 wt%, and FeO_T contents between 11.3 and 12 wt% (Fig. 3a, Supporting In-
412 formation Fig. S5). Despite the high degree of Mg# disequilibrium between olivine
413 crystals and their carrier melts (Fig. 2a), measured melt inclusion Mg#s (uncor-
414 rected for the effects of PEC) mostly lie within, or close to the equilibrium field
415 calculated from the core compositions of their host olivines (Fig. 3b). The distance
416 from the equilibrium field degree is largest in the July-18 sample, but still smaller
417 than the vast majority of melt inclusions data from other Kīlauean eruptions, par-
418 ticularly those hosted in olivines with higher Fo contents (Fig. 3b). Melt inclusions
419 hosted in olivine crystals which have the highest degree of disequilibrium with their
420 carrier melts (calculated by subtracting the equilibrium Fo content of the co-erupted
421 matrix glass from the Fo content of each olivine) have experienced the most PEC
422 (Fig. 3c) and have the lowest measured FeO_T contents (Fig. 3d).

423 To encapsulate the variable degrees of olivine-melt disequilibrium, and to aid
424 comparisons between different crystal populations, we subdivide F8 olivines into two
425 groups. The first group contains olivines which lie within, or close to the equilib-
426 rium field calculated from the Mg# of the co-erupted matrix glass (Fig. 2a). For
427 the May-18 sample, the division was placed at $\text{Fo}_{81.5}$, based on the near continuous
428 distribution of olivines from slightly above to within the equilibrium field (which can
429 easily be generated by slight cooling between crystallization and eruption), and the
430 slight gap between these olivines and those with higher Fo contents (Fig 2b). The
431 second group contains olivines which lie outside the equilibrium field. For brevity,

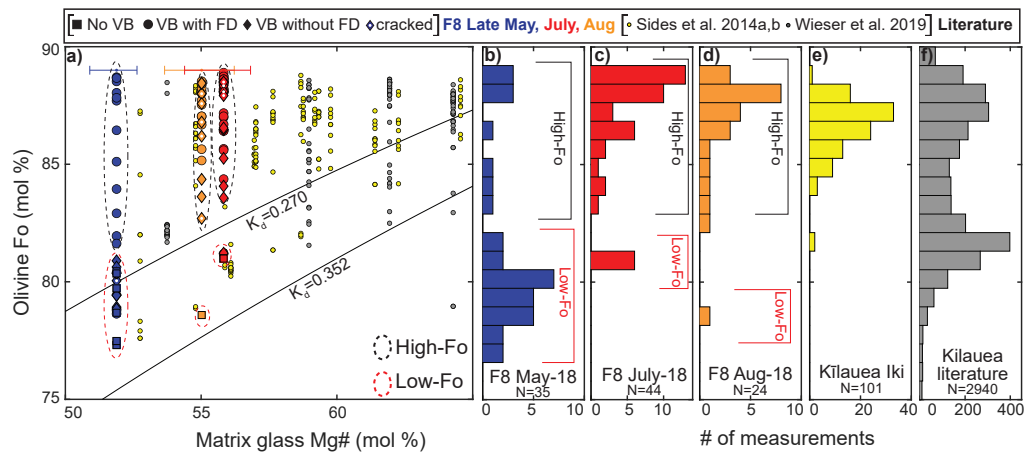


Figure 2. Olivine populations and olivine-melt relationships at F8 compared to literature data. a) Core olivine forsterite content versus matrix glass Mg# for $\text{Fe}^{3+}/\text{Fe}_T=0.15$ (Moussallam et al., 2016; Helz et al., 2017). Olivines lying between the black lines ($K_D=0.270-0.352$) are in equilibrium with their carrier melts considering the range of experimentally-determined Fe-Mg partition coefficients (Roeder & Emslie, 1970; Matzen et al., 2011). F8 olivines have some of the highest Fo contents observed at Kilauea, yet are hosted in carrier liquids with some of the lowest Mg#s. Literature data from Wieser et al. (2019), Sides, Edmonds, Maclennan, Houghton, et al. (2014), Sides, Edmonds, Maclennan, Swanson, and Houghton (2014). b-d) Histograms of olivine Fo contents from this study, e) Kilauea Iki (Sides, Edmonds, Maclennan, Houghton, et al., 2014; Sides, Edmonds, Maclennan, Swanson, & Houghton, 2014), and f) the compilation of literature analyses presented in Wieser et al. (2019) combined with new measurements from Tucker et al. (2019). The strong bimodality in F8 forsterite contents, along with the degree of olivine-melt disequilibrium was used to subdivide melt inclusions into those hosted within High-Fo olivines (black dotted outline) and Low-Fo olivines (red dotted outline). Olivines are further subdivided into those hosting a melt inclusion without a vapor bubble (no VB), with a vapor bubble which produces a Fermi diad (VB with FD), those with a vapor bubble that does not produce a Fermi diad (VB without FD). Melt inclusions which are cracked, and have a vapor bubble without a FD, are indicated with a white dot.

432 these groups are referred to as Low-Fo and High-Fo olivines, although this classi-
 433 fication evaluates the forsterite content of the olivine relative to the Mg# of the
 434 co-erupted matrix glass, rather than the absolute Fo content (see Fig. 3c). A sim-

435 ilar classification for the eruptions on Fig. 2 with higher glass Mg#s would place
436 the boundary between groups at higher Fo contents (e.g., the Fo₈₄ division used by
437 Wieser et al., 2019).

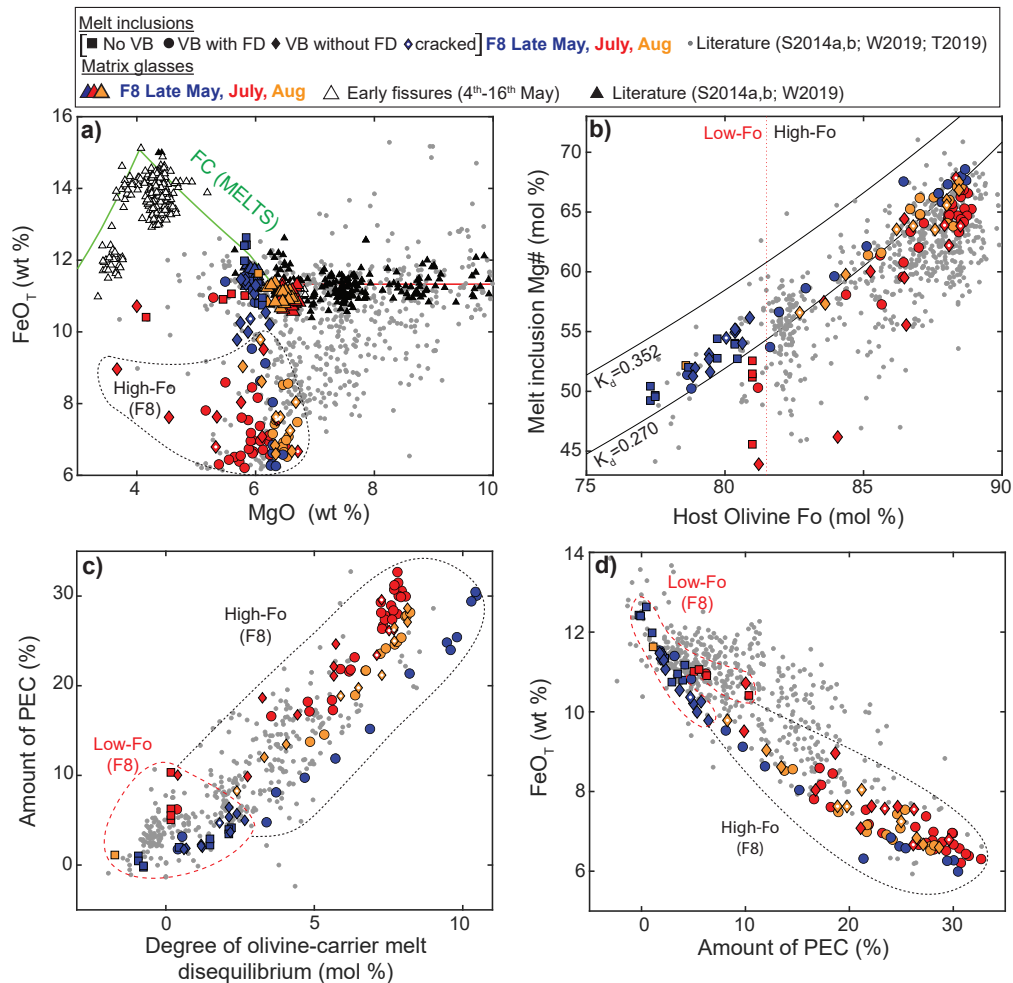


Figure 3. Measured major element systematics for F8 melt inclusions (uncorrected for the effects of PEC). a) High-Fo F8 melt inclusions have significantly lower FeO_T contents than liquid line of descent defined by Kilauean matrix glasses from (this study, Wieser et al., 2019; Sides, Edmonds, MacLennan, Houghton, et al., 2014), and a MELTS for MATLAB (Antoshechkina & Ghiorso, 2018) fractionation path following the onset of clinopyroxene and plagioclase fractionation (green line) which recreates glass compositions erupted from earlier, more evolved fissures during the 2018 eruption (4-5 wt% MgO, white triangles). Despite highly variable FeO_T contents, the MgO contents of melt inclusions mostly align with those of their co-erupted matrix glasses. b) In contrast to the prominent disequilibrium between High-Fo olivine compositions and co-erupted matrix glasses (Fig. 1a), melt inclusion Mg#s uncorrected for the effects of PEC (for Fe³⁺/Fe_T=0.15) plot close to the equilibrium field with their host olivines (particularly melt inclusions from the May-18 and Aug-18 samples). Melt inclusions from previous Kilauean eruptions (Wieser et al., 2019; Tucker et al., 2019; Sides, Edmonds, MacLennan, Houghton, et al., 2014; Sides, Edmonds, MacLennan, Swanson, & Houghton, 2014, grey dots) lie much further below the equilibrium field. c) The amount of PEC (calculated in Petrolog3; Danyushevsky & Plechov, 2011) is strongly correlated with the degree of ol-melt disequilibrium, calculated by subtracting the equilibrium olivine composition of the co-erupted matrix glass (for K_D=0.3) from the measured Fo content. d) The FeO_T contents of F8 melt inclusions also shows a strong negative correlation with the amount of PEC, extending to lower values than the vast majority of

438 All High-Fo melt inclusions contain a vapor bubble (Fig. 3c), 73% (N=53) of
439 which produce a Fermi diad (FD) during Raman analysis. Vapor bubbles which do
440 not produce a FD may contain no CO₂, or CO₂ densities below the detection limit
441 of Raman spectroscopy. While the detection limit will depend on the exact depth
442 of the bubble below the surface, as well as the transparency of the host crystal, the
443 distribution of densities in vapor bubbles which produced a FD indicates that the
444 detection limit lies between 0–0.02 g/cm³ (light green bar in Fig. 4c). Nine of the
445 bubbles without a FD are hosted within cracked melt inclusions, which may have
446 resulted in CO₂ loss from the bubble (diamonds with white dots; Fig. 3 and 4, see
447 Supporting Information Fig. S10 Aster et al., 2016). In contrast, only 50% (N=15)
448 of Low-Fo melt inclusions contain a vapor bubble, and only 20% (N=3) of these pro-
449 duce a FD (Fig. 3c). Only 1 of the bubbles without a FD is hosted within a cracked
450 melt inclusion.

451 Bubble-bearing melt inclusions show a correlation between the volume % of
452 the bubble and the amount of PEC, despite the large random errors associated with
453 measuring bubble proportions from 2D images (grey error bars; Fig. 4a). There is
454 a substantial drop in glass CO₂ contents with increasing PEC, and melt inclusions
455 containing vapor bubbles with a FD show significantly lower glass CO₂ contents
456 than bubble-free melt inclusions (Fig. 4b, $p=10^{-7}$; Kolmogorov Smirnov test).
457 There is no obvious correlation between the CO₂ density in vapor bubbles and the
458 amount of PEC (Fig. 4c, $R^2=10^{-5}$), the CO₂ density and the glass CO₂ content
459 ($R^2=0.1$) or the CO₂ density and the volume of the bubble ($R^2=0.0004$). The me-
460 dian and mean proportion of the total melt inclusion CO₂ budget hosted within the
461 bubble is 93% and 87% respectively (black histogram; Fig. 4d). This exceeds the
462 proportions calculated by Moore et al. (2015) for melt inclusions from the 1959 and
463 1960 eruptions of Kīlauea (median=67%, mean=65%; blue histogram). This dis-
464 crepancy reflects the fact that Moore et al. (2015) did not measure the CO₂ content
465 of the glass in each melt inclusion, so they calculated proportions assuming a glass
466 CO₂ content of 300 ppm (the maximum measured in the same suite of samples by
467 Tuohy et al., 2016). Our new data shows the importance of measuring CO₂ in the
468 glass and bubble of a specific melt inclusion; while bubble-free melt inclusions have
469 CO₂ contents up to 417 ppm in the glass phase, those with vapor bubbles produc-
470 ing a FD have median CO₂ contents of only 45 ppm (mean=54 ppm; Fig. 4b). In

471 contrast to the highly variable CO₂ contents in melt inclusion glasses, H₂O contents
472 are remarkably constant within a given eruption, despite significant variation in the
473 contents of incompatible elements such as Na₂O and K₂O (Fig. 5a). Excluding two
474 degassed melt inclusions (~ 0.09 wt% H₂O), F8 melt inclusions have between 0.19–
475 0.33 wt% H₂O, which is lower than most of the Kīlauean melt inclusions measured
476 by Sides, Edmonds, MacLennan, Houghton, et al. (2014); Sides, Edmonds, MacLen-
477 nan, Swanson, and Houghton (2014) and almost all of those measured by Tucker et
478 al. (2019) (Fig. 5b).

479 5 Discussion

480 5.1 Mineral-melt disequilibrium drives the growth of a CO₂-rich 481 bubble

482 The prominent Mg# disequilibrium between the core compositions of High-Fo
483 olivines from F8 and their carrier melts has been observed in a number of historic
484 eruptions at Kīlauea (Fig. 2; Tuohy et al., 2016; Wieser et al., 2019; Sides, Ed-
485 monds, MacLennan, Houghton, et al., 2014; Sides, Edmonds, MacLennan, Swanson,
486 & Houghton, 2014). Based on major and trace element disequilibrium between melt
487 inclusions and their carrier melts (e.g., Nb/Y ratios), as well as microstructures
488 consistent with deformation of the crystal lattice (also observed in some High-Fo
489 olivines from F8 by Gansecki et al., 2019), Wieser, Edmonds, et al. (2020) and
490 Wieser et al. (2019) suggested that highly forsteritic olivines are scavenged from
491 long-lived plastically-deforming mush piles at the base of the SC reservoir, and
492 incorporated into cooler, lower Mg# carrier melts with different trace element sig-
493 natures just prior to eruption. In contrast, these studies suggest that olivines with
494 lower forsterite contents exhibiting small amounts of olivine-melt disequilibrium
495 (similar to the Low-Fo olivines in this study), no lattice distortions, and a high de-
496 gree of trace element equilibrium may have crystallized from their carrier melts as
497 true phenocrysts.

498 Kīlauean melts with greater than ~ 6.8 wt% MgO are saturated in only olivine
499 and minor chrome-spinel (Wright & Fiske, 1971), so show a strong correlation be-
500 tween temperature and the MgO content of the melt (Helz & Thornber, 1987). The
501 remarkably constant FeO contents of these high MgO melts (Fig. 3a) means that

502 glass Mg# is strongly correlated with MgO, and therefore temperature. As glass
503 Mg# is closely related to the olivine forsterite content through the Fe-Mg olivine-
504 liquid exchange coefficient, equilibrium olivine forsterite contents are also strongly
505 correlated with temperature. Thus, the difference in Mg# between the measured
506 olivine core composition, and the equilibrium olivine forsterite content calculated
507 from the composition of co-erupted matrix glasses (termed the degree of olivine-melt
508 disequilibrium) is proportional to the amount of cooling experienced by the inclu-
509 sion prior to syn-eruptive quenching (Wieser et al., 2019). The close relationship
510 between the amount of cooling experienced by an inclusion, and the amount of PEC
511 (Danyushevsky et al., 2000) accounts for the excellent correlation between the degree
512 of olivine-melt disequilibrium and the amount of PEC (Fig. 3c).

513 F8 melt inclusions are hosted in some of the most forsteritic olivines erupted
514 at Kilauea, yet were erupted in carrier melts with some of the lowest Mg#s (Fig.
515 2a). Consequently, they have experienced some of the largest amounts of cooling fol-
516 lowing entrapment, and, by extension, some of the largest amounts of PEC ever re-
517 ported at Kilauea (up to $\sim 33\%$; Fig. 3c), see also A. Lerner (2020) and A. H. Lerner
518 et al. (2020). These PEC extents are also significantly larger than those reported
519 from other volcanic systems; olivine-hosted melt inclusions from Holuhraun (Ice-
520 land), Piton de la Fournaise (Réunion) and Erebus (Antarctica) have experienced
521 $\sim 5\%$, $< 12\%$ and 0–4.2% PEC respectively (Hartley et al., 2015; Collins et al., 2012;
522 Moussallam et al., 2014). The small amounts of cooling (and therefore PEC) expe-
523 rienced by Low-Fo olivines, which are close to equilibrium with their carrier melts,
524 likely occurred during fractionation between the formation and eruption of these
525 crystals (Fig. 3c). However, progressive fractionation and cooling of a batch of melt
526 cannot account for the peak at $\sim \text{Fo}_{88-89}$ in F8 samples (Wieser et al., 2019; Maaløe
527 et al., 1988), nor the paucity of olivines with Fo contents in equilibrium with the
528 co-erupted matrix glasses (particularly in the July and Aug samples; Fig. 2a). Based
529 on the similarities between the High-Fo olivines from F8 and previous studies (large
530 amounts of olivine-melt disequilibrium, presence of lattice distortions; Gansecki et
531 al., 2019), we appeal to the process proposed by Wieser et al. (2019), where cooling
532 is not a gradual process during progressive differentiation of a given magma batch
533 (Maaløe et al., 1988), but occurs over short timescales, when High-Fo olivine crys-
534 tals residing in hot mush piles are mixed into significantly cooler, lower Mg# melts

535 (Wieser et al., 2019; Sides, Edmonds, MacLennan, Houghton, et al., 2014), see also
536 Shea et al. (2019).

537 Melt inclusion MgO and FeO_T contents are strongly affected by the crystal-
538 lization of olivine on the walls of the melt inclusion (PEC), and subsequent diffusive
539 re-equilibration. Based on the strong coupling between MgO content and tempera-
540 ture in olivine-saturated liquids (Helz & Thornber, 1987), thermal equilibration of
541 a hot olivine crystal with a cooler carrier melt drives the crystallization of a zoned
542 olivine rim from the melt inclusion, causing the MgO content of the melt inclusion
543 to drop to match that of the carrier melt (Fig. 3a). This zoned olivine rim begins to
544 re-equilibrate with the host crystal, and, in turn, the melt inclusion re-equilibrates
545 with the changing rim composition (Danyushevsky et al., 2000). The melt inclusion
546 loses FeO by diffusion to achieve Mg# equilibration with the host olivine follow-
547 ing the large initial drop in MgO during cooling. As the MgO content of the melt
548 inclusion is a function of the temperature, FeO diffusion is countered by MgO dif-
549 fusion in the opposite direction, which is sequestered by further post-entrapment
550 crystallization of olivine on the wall of the melt inclusion.

551 This FeO-loss process accounts for the negative correlation between melt in-
552 clusion FeO_T contents and the amount of PEC (Fig. 3d). For a given amount of
553 PEC, F8 melt inclusions have lower FeO_T contents and display a smaller degree
554 of Mg# disequilibrium with their olivine host than the vast majority of literature
555 data (Fig. 3b, d). It is important to note that methods calculating the amount of
556 PEC based on the degree of Mg# disequilibrium between the melt inclusion and the
557 host crystal (e.g., Tucker et al., 2019; Neave et al., 2017) will significantly under-
558 estimate the true amount of PEC in melt inclusions where extensive FeO-loss has
559 occurred compared to the Petrolog3 method used here where the user specifies an
560 initial FeO_T content. For example, the May-18 melt inclusions with $\text{Fo} > 85$ have lost
561 sufficient quantities of FeO by diffusive re-equilibration such that their Mg#s are in
562 equilibrium with the composition of the host olivine. Thus, methods based on Mg#
563 comparisons would indicate that these melt inclusions have experienced very minor
564 amounts of PEC. However, their FeO contents lie ~ 4 wt% below the composition
565 of co-erupted matrix glasses, indicating that their compositions have been heavily
566 altered by the PEC process (Fig. 3a).

567 The higher degrees of diffusive FeO-loss for a given amount of PEC for F8
 568 melt inclusions compared to literature data (Fig. 3d) indicates that there was a
 569 longer time lag between the entrainment of crystals into cooler melts and their
 570 eventual eruption. Danyushevsky et al. (2002) quantitatively model Fe-Mg re-
 571 equilibration to estimate this time lag: their Fig. 4c shows that a melt inclusion
 572 with a $\sim 50 \mu\text{m}$ radius that has experienced $\Delta T=100\text{--}150^\circ\text{C}$ and undergone FeO loss
 573 at $T=1150\text{--}1200^\circ\text{C}$ achieves 98% equilibrium in ~ 2 years. These extents of cooling
 574 and temperatures of re-equilibration are representative of F8 inclusions. However,
 575 Danyushevsky et al. (2002) assume isotropic diffusion of Fe through the host olivine
 576 crystal with $D_{\text{Fe, Mg}} \sim 3\text{--}6 \times 10^{-17} \text{ m}^2/\text{s}$ at $1150\text{--}1200^\circ\text{C}$. In reality, FeO loss will be
 577 dominated by diffusion along the fast c-direction in olivine ($D_{\text{Fe, Mg}} \sim 1\text{--}4 \times 10^{-16}$
 578 m^2/s for $\text{Fo}_{80\text{--}89}$, $T=1150\text{--}1200^\circ\text{C}$, and QFM to QFM+0.3; Chakraborty, 2010;
 579 Barth et al., 2019). Thus, complete re-equilibration could be achieved almost an
 580 order of magnitude faster, in a matter of months. Considering the substantial un-
 581 certainties in this method associated with the fact the model of Danyushevsky et
 582 al. (2002) does not account for diffusional anisotropy, and the fact the degree of
 583 re-equilibration is very sensitive to the choice of K_D (Fig. 3b), the FeO_T system-
 584 atics of melt inclusions within High-Fo olivines erupted on May 28th ($\sim 70\text{--}100\%$
 585 re-equilibration) indicate that entrainment into cooler carrier melts occurred approx-
 586 imately a month to a year prior to eruption.

587 5.2 Diffusive H₂O-loss

588 Given that H₂O in melt inclusions diffusively re-equilibrates over hours to days
 589 (Hartley et al., 2015; Le Voyer et al., 2014; Gaetani et al., 2012), the timescales in-
 590 ferred from Fe-Mg disequilibrium are more than sufficient for H₂O contents within
 591 F8 melt inclusions to be fully reset to the H₂O content of the melt which carried
 592 them to the site of the eruption. This re-equilibration accounts for the remarkably
 593 uniform H₂O contents of F8 melt inclusions in each sample, despite substantial
 594 variation in the concentration of other incompatible elements (e.g., Na₂O; Fig. 5a).
 595 The approximately constant H₂O contents in melt inclusions from each sample in-
 596 dicates that F8 carrier melts erupted in late May had H₂O contents of 0.29 wt%,
 597 while those erupted in July and August had slightly lower H₂O contents ($\sim 0.22\text{--}0.23$
 598 wt%). These carrier melts are relatively H₂O-poor compared to the composition

599 of previously-erupted Kilauean melts (inferred from published melt inclusion data;
600 Fig. 5). The presence of more H₂O-poor carrier melts in 2018 likely results from the
601 extensive mixing of magmas which had partially degassed their H₂O at the sum-
602 mit lava lake with undegassed melts within the plumbing system between 2008 and
603 2018 (see also A. Lerner, 2020). This is similar to the mechanism proposed for the
604 variable volatile contents of Puna Ridge magmas by Dixon (1991) degassing.

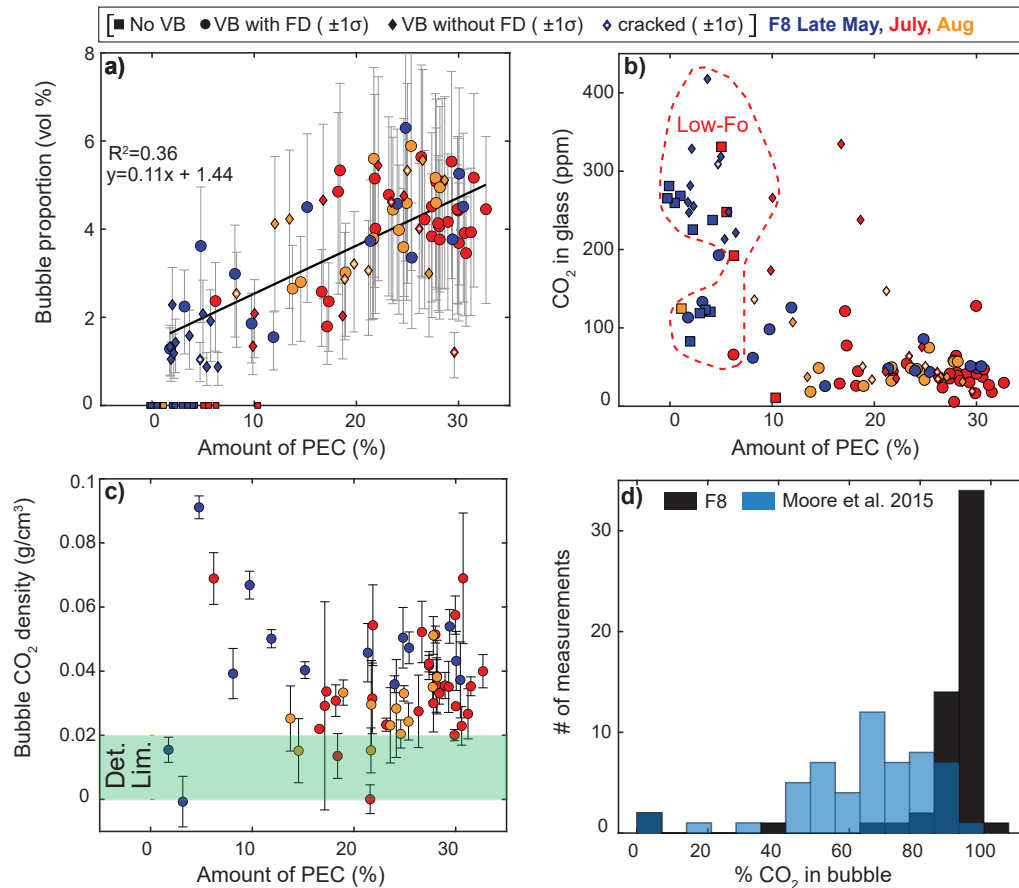


Figure 4. Vapor bubble and melt inclusion CO₂ systematics. a) There is a positive correlation between the volume proportion of the vapor bubble (VB) and the amount of PEC. Only melt inclusions which have experienced <10% PEC are bubble-free. Error bars show the 1σ errors associated with estimating bubble volume proportions from 2D images (-45% and +37% Tucker et al., 2019). b) With increasing amounts of PEC, the amount of CO₂ within the glass phase of the melt inclusion declines. The highest glass CO₂ contents are observed in melt inclusions with no vapor bubbles (squares), and melt inclusion with bubbles that did not produce a FD (diamonds). In contrast, the vast majority of melt inclusions with low glass CO₂ contents have vapor bubbles which produced a FD (circles), or vapor bubbles without a FD that were hosted within cracked melt inclusions (diamonds with white dots). c) There is no correlation between the CO₂ density in vapor bubble measured using Raman Spectroscopy and the amount of PEC. Error bars show the $\pm 1\sigma$ deviation of three repeated measurements of each vapor bubble. The green bar shows our estimate of the detection limit (Det. Lim.) of Raman analyses based on the distribution of measured bubble densities. d) The black histogram shows the proportion of CO₂ held within the vapor bubble for F8 melt inclusions that produced a FD (mean=87%, median=93%). Estimates by Moore et al. (2015) for Kīlauean melt inclusions from the 1959 and 1960 eruptions are also shown.

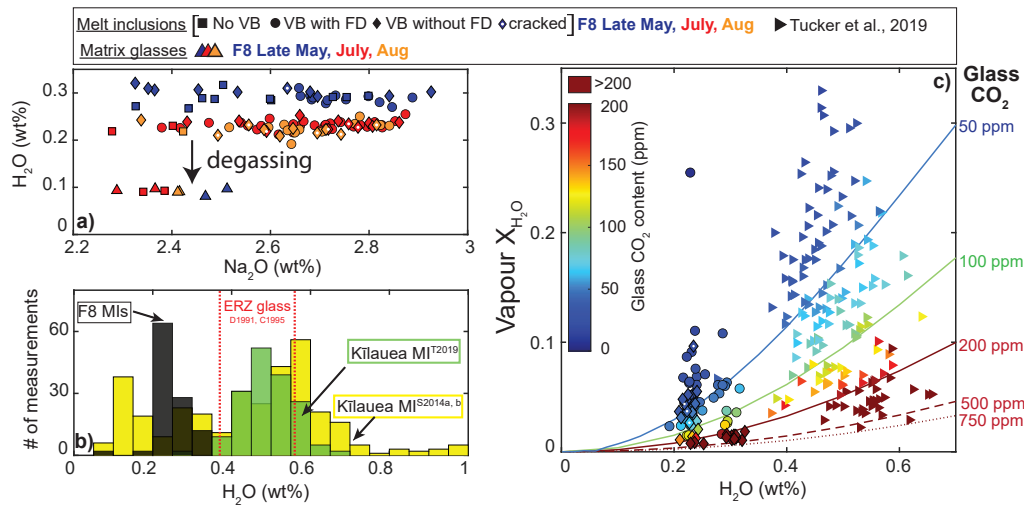


Figure 5. H₂O systematics of F8 melt inclusions relative to literature data from Kilauea. a) F8 melt inclusion H₂O contents are remarkably constant within each sample, despite substantial variations in Na₂O. This indicates that melt inclusion H₂O contents were reset by diffusive re-equilibration with their carrier liquid. The precision of SIMS measurements ($\pm 1.5\%$) is smaller than the symbol size, so error bars are not shown. b) F8 melt inclusions have lower H₂O contents than the majority of Kilauean melt inclusions measured by Sides, Edmonds, Maclennan, Swanson, and Houghton (2014); Sides, Edmonds, Maclennan, Houghton, et al. (2014) (yellow histogram) and almost all of the melt inclusions measured by Tucker et al. (2019). H₂O contents from submarine ERZ glasses with 7–16 wt% H₂O from Dixon et al. (1991); Clague et al. (1995) are shown with red dashed lines. c) Relationship between the molar fraction of H₂O in the vapor phase (X_{H_2O}) and the melt H₂O content for five different melt CO₂ contents (50, 100, 200, 500 and 750 ppm; using VolatileCalc-Basalt; Newman and Lowenstern, 2002). X_{H_2O} ratios for the co-existing vapor in equilibrium with the measured concentration of CO₂ and H₂O in the melt phase of the bubble-bearing inclusions from this study and Tucker et al. (2019) (triangles) are overlain, with symbols colored by the CO₂ content of the glass phase. The relatively low H₂O contents of F8 melt inclusions mean that X_{H_2O} is generally < 0.1 . However, a number of inclusions from Tucker et al. (2019) with glass CO₂ contents < 100 ppm have much higher X_{H_2O} ratios. This causes the CO₂ densities predicted using the EOS method to fall below the trend line defined by F8 melt inclusions on Fig. 8a.

5.3 PEC and melt-vapor CO₂ partitioning

It is well recognized that extensive PEC drives the growth of a CO₂-rich vapor bubble (Steele-Macinnis et al., 2011; Sides, Edmonds, MacLennan, Houghton, et al., 2014; Sides, Edmonds, MacLennan, Swanson, & Houghton, 2014; Aster et al., 2016; MacLennan, 2017). Thus, studies measuring only the CO₂ in the melt phase using SIMS or FTIR will yield spuriously low entrapment depths for melt inclusions which have undergone extensive PEC (e.g., Sides, Edmonds, MacLennan, Houghton, et al., 2014). Our concurrent measurements of CO₂ in the melt and bubble phase of a large number of melt inclusions which have experienced a wide range of PEC amounts (Fig. 3c-d) provides a unique opportunity to interrogate the various processes causing CO₂ to partition into the vapor bubble.

To investigate the effects of compositional changes in the melt inclusion associated with PEC, we use the CO₂ solubility model of Shishkina et al. (2014):

$$\ln[\text{CO}_2] = 1.15\ln(P) + 6.71\Pi^* - 1.345 \quad (4)$$

Where [CO₂] is the concentration of CO₂ in ppm, and P is the pressure in MPa. The Π^* term accounts for the compositional dependence on CO₂ solubility, expressed in terms of the cation fractions of 7 major element species:

$$\Pi^* = \frac{\text{Ca}^{2+} + 0.8\text{K}^+ + 0.7\text{Na}^+ + 0.4\text{Mg}^{2+} + 0.4\text{Fe}^{2+}}{\text{Si}^{4+} + \text{Al}^{3+}} \quad (5)$$

We calculate the change in Π^* during PEC, $\Delta \Pi^*$, by subtracting the Π^* value of the PEC-corrected major element composition of each melt inclusion from the Π^* value of the measured composition. $\Delta \Pi^*$ becomes progressively more negative with increasing amounts of PEC, showing that CO₂ becomes progressively less soluble (red dots; Fig. 6b, see also MacLennan, 2017). Changes in Π^* are dominated by a decrease in X_{Mg} , and increase in X_{Si} and X_{Al} resulting from the crystallization of olivine on the walls of the inclusion. These changes are partially counteracted by an increase in X_{Ca} (as Ca is incompatible in olivine). To quantify the magnitude of this drop in Π^* in terms of CO₂ partitioning between the melt and bubble, we consider the 8 melt inclusions which have experienced >30% PEC (all of which contain bubbles which produce a FD). The mean Π^* value of the measured compositions of

632 these melt inclusions is 0.33, while the mean Π^* of their PEC-corrected compositions
633 is 0.39 ($\Delta \Pi^* = -0.068$). For $P = 0.76$ kbar, which is the average entrapment pressure
634 for the PEC-corrected compositions of these melt inclusions calculated using equa-
635 tion 4, CO_2 solubility drops by ~ 192 ppm. As melts at Kīlauea are CO_2 saturated
636 at crustal storage depths (Gerlach et al., 2002), this extra CO_2 will partition into
637 the vapor bubble.

638 However, the mean amount of CO_2 sequestered within the vapor bubbles of
639 these 8 melt inclusions is 657 ± 231 ppm (calculated using equation 1). This reflects
640 three additional processes which enhance CO_2 partitioning into the bubble during
641 PEC. Firstly, the crystallization of olivine, which contains negligible quantities of
642 CO_2 , drives up the total concentration of the CO_2 in the remaining melt by a factor
643 of 1 plus the amount of PEC (1.3 to $1.33\times$ for these 8 melt inclusions). As men-
644 tioned above, because Kīlauea melt inclusions are CO_2 saturated (Gerlach et al.,
645 2002), this excess partitions into the bubble (mean 145 ppm, up to 230 ppm CO_2).
646 Secondly, the preferential contraction of the melt phase relative to the olivine during
647 thermal re-equilibration leads to a reduction in the volume of the melt phase. This
648 is enhanced by the third process; the crystallization of denser olivine on the rim of
649 the melt inclusion. A drop in the internal pressure of the melt inclusion causes the
650 CO_2 solubility to decrease further, driving more CO_2 into the vapor bubble (equa-
651 tion 5). Evidence for these volume changes is provided by the correlation between
652 the amount of PEC and the volume of the vapor bubble (Fig. 4a), as well as the
653 observation that all melt inclusions without a vapor bubble have experienced $< 10\%$
654 PEC (Fig. 4a), while all melt inclusions that have experienced $> 10\%$ PEC have a
655 vapor bubble.

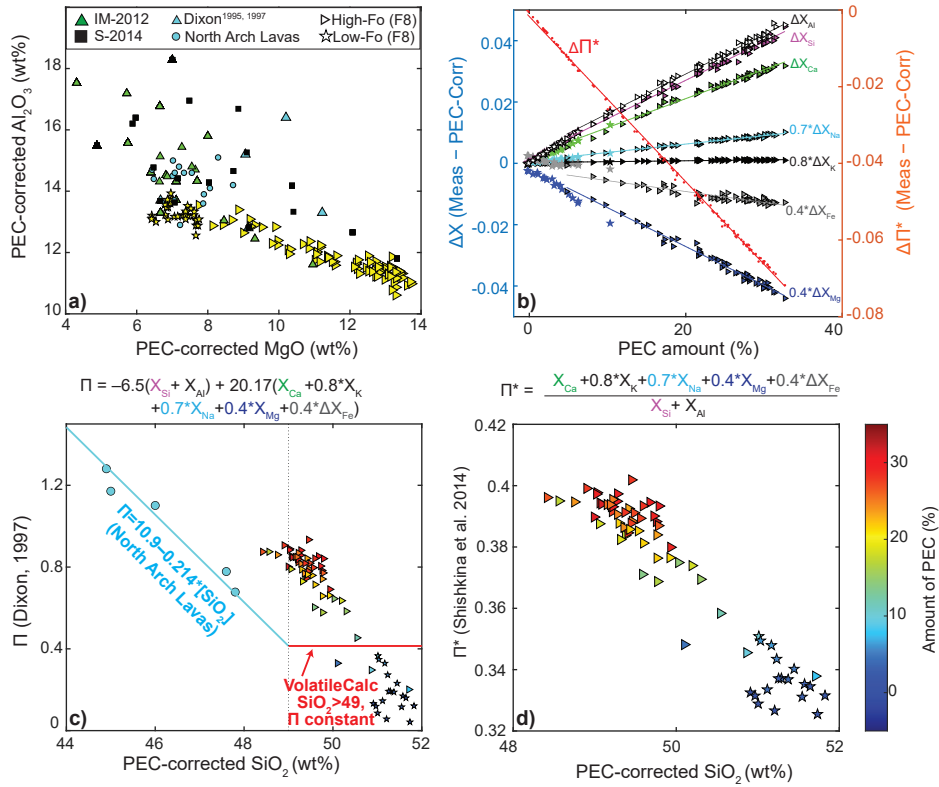


Figure 6. Evaluating the compositional sensitivity of CO₂ solubility. a) Comparison of the MgO vs. Al₂O₃ systematics of PEC-corrected F8 inclusions to the glass compositions used to calibrate each solubility model. The North Arch lavas which define the simplified Π vs. SiO₂ relationship presented in Dixon (1997) and implemented in VolatileCalc-Basalt (Newman & Lowenstern, 2002) are also shown (blue circles). The MagmaSat dataset (Ghiorso & Gualda, 2015) includes the experiments in the calibration datasets of Shishkina et al. (2014), Iacono-Marziano et al. (2012) and Dixon et al. (1995) (so is not shown, as it would cover all these symbols). b) ΔX (triangle and star symbols) and ΔΠ* (red dots Shishkina et al., 2014) for F8 melt inclusions plotted against the amount of PEC. ΔX and ΔΠ* were calculated by subtracting the values of X and Π* for PEC-corrected melt inclusions from the values of X and Π* for measured compositions. For example, inclusion LL8.156 has experienced 33% PEC, and has a PEC-corrected MgO content of 13.5 wt% and a measured MgO content of 5.4 wt%. Thus, ΔX_{MgO} is strongly negative. c) The compositional parameter Π of Dixon (1997) calculated for PEC-corrected F8 melt inclusion compositions varies substantially with SiO₂, following an offset trend to that defined by North Arch Glasses (Dixon et al., 1997, blue dots and linear regression). VolatileCalc-Basalt effectively treats all melt inclusions with >49 wt% SiO₂ as if Π is constant (red line). d) The compositional parameter Π* from Shishkina et al. (2014), and therefore the solubility of CO₂, is significantly higher for High-Fo melt inclusions (which have the highest PEC-corrected MgO, and lowest SiO₂ and Al₂O₃ contents). The color of the symbols for F8 melt inclusions in c) and d) represents the amount of PEC.

656 Overall, changes in melt chemistry, the incompatible behaviour of CO₂, and a
657 drop in the internal pressure of the melt inclusion accounts for the rapid decrease in
658 glass CO₂ contents with increasing PEC (Fig. 4b). Our concurrent measurements of
659 glass and bubble CO₂ provide the first opportunity to see through these convoluting
660 effects of PEC to robustly determine total CO₂ contents, and therefore entrapment
661 depths of Kīlauean melt inclusions. To account for the uncertainty regarding the
662 amount of CO₂ held within bubbles that did not produce a FD (diamond symbols),
663 particularly those hosted within cracked olivines (diamond symbols with white dot),
664 we only calculate total CO₂ contents and entrapment depths for melt inclusions
665 which had no bubble, or a bubble that produced a FD. These total CO₂ were cor-
666 rected for the incompatible behaviour of CO₂ during PEC to determine the total
667 CO₂ content at the point of melt inclusion entrapment.

668 Total PEC-corrected CO₂ contents in melt inclusions hosted within High-Fo
669 olivines are offset to significantly higher values compared to those hosted within
670 Low-Fo olivines (Fig. 7a), indicating that these two olivine populations crystal-
671 lized at distinct depths within Kīlauea’s plumbing system. It is also interesting to
672 compare our total CO₂ contents to previously published data on Kīlauean melt
673 inclusions. Although these studies investigate products from different eruptions,
674 the apparent stability in the geometry of Kīlauea’s plumbing system since at least
675 the 1950s (Helz et al., 2014; Poland et al., 2015; Eaton & Murata, 1960) means
676 such comparisons are still useful (and particularly relevant for studies of the 1959–
677 1960 eruptive period, where activity at the summit was followed by a large LERZ
678 eruption; e.g., Tuohy et al., 2016; Moore et al., 2015; Sides, Edmonds, Maclen-
679 nan, Houghton, et al., 2014; Sides, Edmonds, Maclennan, Swanson, & Houghton,
680 2014) . Unsurprisingly given our findings that ~90% of CO₂ is held within the va-
681 por bubble (Fig. 4d), CO₂ contents in F8 melt inclusions are significantly higher
682 than measurements of just the glass phase by Sides, Edmonds, Maclennan, Swanson,
683 and Houghton (2014); Sides, Edmonds, Maclennan, Houghton, et al. (2014) (Fig.
684 7c). F8 melt inclusions are also offset to higher CO₂ contents than experimentally-
685 rehomogenized melt inclusions (Tuohy et al., 2016, Fig. 7d). Tuohy et al. (2016)
686 note similar offsets between their measurements and Raman reconstructions of bub-
687 ble CO₂ by Moore et al. (2015) in the same sample set. They suggest that their
688 analyses may have been biased towards melt inclusions with smaller bubbles that

689 fully disappear upon heating, lower pressure inclusions that do not fracture during
690 heating, and larger inclusions that can be analysed by FTIR.

691 Interestingly, our distribution of total CO₂ contents for melt inclusions which
692 possessed bubbles are indistinguishable using the Kolmogorov-Smirnov (KS) test
693 (p=0.1) from the CO₂ contribution of just the vapor bubbles in melt inclusions
694 from the 1959 and 1960 eruptions of Kilauea (Moore et al., 2015, Fig. 7e). This
695 demonstrates that in olivine populations which have experienced extensive PEC,
696 measurements of glass CO₂ contents are of subordinate importance to measurements
697 of bubble CO₂. Furthermore, the contribution of CO₂ from the melt phase for the
698 majority of High-Fo melt inclusions from F8 is entirely overwhelmed by the errors
699 on the amount of CO₂ in the bubble associated with estimating bubble volume pro-
700 portions from 2D images. However, it is worth noting that only measuring CO₂ in
701 vapor bubble would have failed to identify the population of Low-Fo olivines which
702 host almost all of their CO₂ within the glass phase. Thus, we suggest that future
703 studies use a small number of SIMS or FTIR analyses of melt inclusions, combined
704 with EPMA analyses of host crystals and melt inclusions, to determine the relation-
705 ship between glass and bubble CO₂ contents and the amount of PEC in different
706 subpopulations of melt inclusions. If the vast majority of CO₂ in a given population
707 is held in the vapor bubble, a limited analytical budget would be better spent accu-
708 rately measuring bubble volumes (using MicroCT or 3D Raman mapping; Pamukcu
709 et al., 2013; Venugopal et al., 2020) to combine with Raman measurements of CO₂
710 density in the rest of the sample set, instead of precisely quantifying the insignificant
711 amount of CO₂ held within the glass phase using SIMS or FTIR.

712 Importantly, we also observe that the distribution of total CO₂ contents in
713 bubble-bearing melt inclusions is significantly higher than bubble-free melt inclu-
714 sions (Fig. 7b). This result invalidates the approach of preferentially targeting
715 bubble-free melt inclusions to avoid having to account for CO₂ within the vapor
716 bubbles (e.g., Helo et al., 2011; Esposito et al., 2011) in systems where erupted crys-
717 tals have experienced extensive PEC prior to eruption. Crucially, analysis of only
718 bubble-free melt inclusions by SIMS or FTIR, or analyses of just vapor bubbles us-
719 ing Raman, would have failed to identify that crystals are supplied from two distinct
720 storage regions within Kilauea's plumbing system.

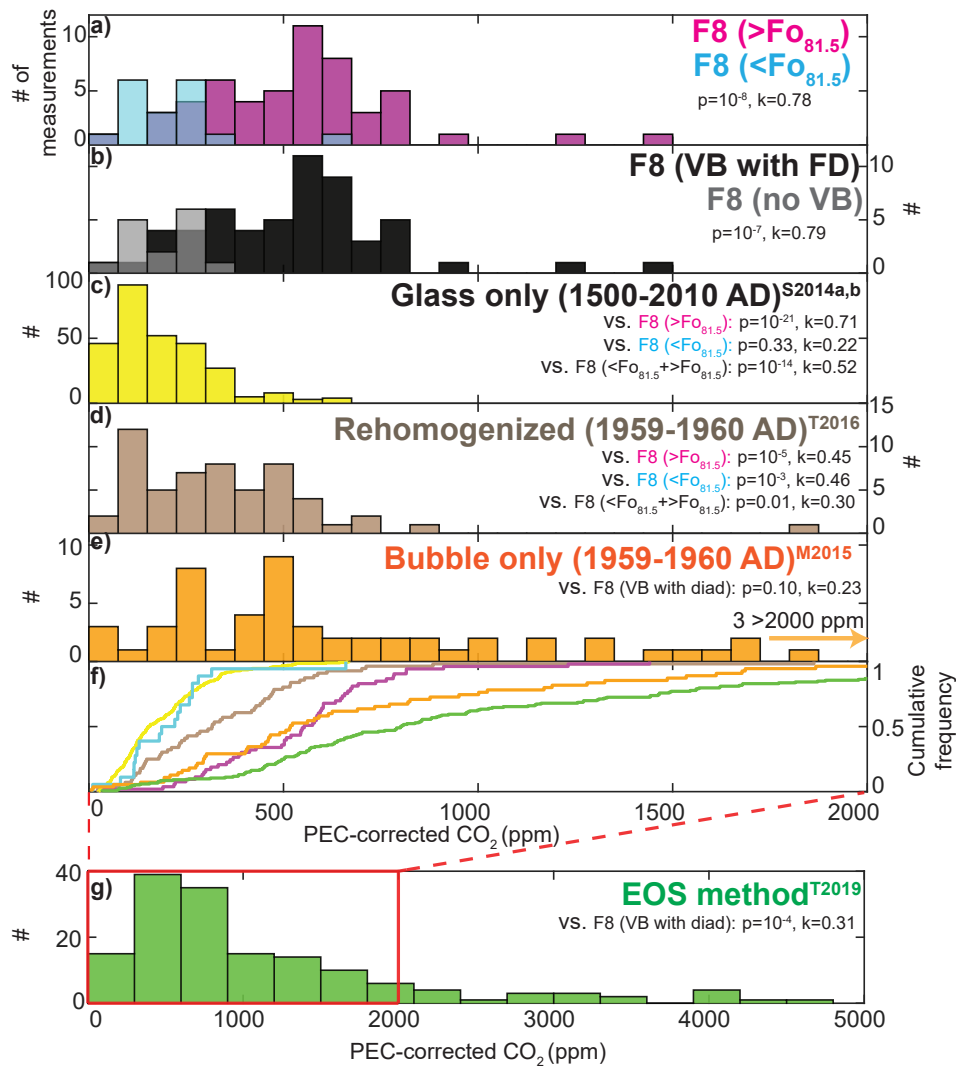


Figure 7.

721 **Caption Fig. 7** Histograms of melt inclusion CO₂ contents from this study and
 722 the literature (all corrected for the effects of PEC). a) Total CO₂ contents (bub-
 723 ble+glass) for High and Low-Fo melt inclusions are statistically distinguishable at
 724 $p=0.05$ using the Kolmogorov Smirnov (KS) test (p value and test statistic k shown
 725 on the figure). b) Similarly, melt inclusions which contain a vapor bubble (VB) with
 726 a FD have significantly higher total CO₂ contents than bubble-free melt inclusions.
 727 c) Melt inclusion CO₂ contents from a suite of eruptions at Kilauea between 1500
 728 and 2008 AD where only the glass phase was measured (Sides, Edmonds, Maclennan,
 729 Swanson, & Houghton, 2014; Sides, Edmonds, Maclennan, Houghton, et al.,
 730 2014). d) CO₂ contents of experimentally-homogenized melt inclusions from the

731 1959 Kīlauea Iki and 1960 Kapoho eruptions (Tuohy et al., 2016). e) Bubble CO₂
732 contents from Moore et al. (2015) in the same suite of samples as in d). For consis-
733 tency, these bubble CO₂ contents were corrected for PEC using the average amount
734 of PEC reported by Tuohy et al. (2016) (13%). f) Cumulative distribution plots
735 for these datasets. g) Total inclusion CO₂ contents from Tucker et al. (2019) where
736 the contribution from bubble CO₂ was estimated using the EOS method (excluding
737 inclusions with bubble volumes >8% that the authors suggest were co-entrapped).
738 35 melt inclusions have CO₂ >1500 ppm. Note the change in x axis scale from plots
739 a-f). For literature data, all melt inclusions are shown, as Fo contents were not re-
740 ported by Moore et al. (2015), and matrix glass Mg#s were not reported in Tucker
741 et al. (2019), so it was not possible to classify data based on the degree of olivine-
742 melt disequilibrium as for F8 samples.

743 **5.4 Analytical versus theoretical constructions of vapor bubble CO₂**

744 In contrast to the good agreement between our estimates of total CO₂ con-
745 tents from combined SIMS and Raman measurements from F8 and the bubble-only
746 measurements of Moore et al. (2015), the total CO₂ contents estimated by Tucker
747 et al. (2019) for a range of Kīlauean eruptions using the EOS method are displaced
748 to significantly higher values (Fig. 7g). To assess the cause of this discrepancy, we
749 follow the EOS method they describe to calculate CO₂ bubble densities for F8 melt
750 inclusions to compare to our Raman measurements. The simplification of the Dixon
751 (1997) solubility model implemented in the excel workbook VolatileCalc (hereafter
752 VolatileCalc-Basalt Newman & Lowenstern, 2002) was used to calculate the internal
753 pressure of the melt inclusion based on the measured SiO₂, CO₂ and H₂O contents
754 of the glass phase. The pure CO₂ EOS of Span and Wagner (1996) implemented in
755 Python3 through CoolProp (Bell et al., 2014) was used to calculate the CO₂ den-
756 sity at this internal pressure and 725 °C, which was the presumed glass transition
757 temperature of Tucker et al. (2019) based on Ryan and Sammis (1981). The Duan
758 and Zhang (2006) EOS utilized by Tucker et al. (2019) yields identical densities to
759 the fourth decimal place (see Supporting Information Fig. S11). The more signifi-
760 cant source of error involves the choice of the glass transition temperature. This is
761 fixed at 725 °C in Tucker et al. (2019) and 825 °C in Moore et al. (2015)(dashed
762 and solid magenta line; Fig. 8a) for simplicity, but in reality, varies as a function of

763 cooling rate and melt viscosity (and, by extension, melt composition; Giordano et
 764 al., 2005; MacLennan, 2017). The average glass transition temperatures predicted by
 765 the bubble-growth python code MIMiC (which uses the model of Giordano et al.,
 766 2005; Rasmussen et al., 2020) for bubble-bearing F8 melt inclusions for cooling rates
 767 of 10 °C/s is 680 °C (dotted magenta line; Fig. 8a). Following Tucker et al. (2019),
 768 we multiply the density obtained from the pure-CO₂ EOS by the mole fraction of
 769 CO₂ (X_{CO_2}) in the vapor phase determined in VolatileCalc (Newman & Lowenstern,
 770 2002). This correction neglects the non-ideal mixing of H₂O and CO₂ at magmatic
 771 temperatures compared to the use of a mixed H₂O-CO₂ EOS (e.g., Moore et al.,
 772 2015) but is probably a reasonable approximation for relatively dry systems such as
 773 Kilauea (Fig. 5a-b).

774 The dominant control of the glass CO₂ content on the internal pressure of the
 775 inclusion in relatively anhydrous melts, and the positive relationship between the in-
 776 ternal pressure and ρ_{CO_2} from the EOS evaluated at a constant temperature, means
 777 that predicted ρ_{CO_2} values increase with increasing glass CO₂ contents (Fig. 8a).
 778 Predicted CO₂ densities from Tucker et al. (2019) plot on or below the quadratic fit
 779 through the EOS predictions for F8 melt inclusions at 725 °C (magenta solid line),
 780 because of the higher values of X_{H_2O} (and thus lower X_{CO_2}) for a number of melt
 781 inclusions which possess high glass H₂O, but low glass CO₂ contents (Fig. 5c). How-
 782 ever, unlike the predictions from the EOS method, there is no correlation between
 783 ρ_{CO_2} measured using Raman spectroscopy and glass CO₂ contents ($R^2=0.11$). In-
 784 terestingly, all melt inclusions with >200 ppm CO₂ in the glass have vapor bubbles
 785 which did not produce a FD (diamond symbols; Fig. 8a), indicating that their CO₂
 786 densities were below the detection limit of Raman Spectroscopy ($\sim 0-0.02$ g/cm³;
 787 green bar in Fig. 4c). It seems implausible that these bubbles could possess the high
 788 CO₂ densities predicted by the EOS ($\rho_{CO_2} > 0.2$ g/cm³) and fail to produce a FD.
 789 Furthermore, melt inclusions with $\rho_{CO_2} > 0.2$ g/cm³ will consist of an outer shell of
 790 liquid CO₂, and an inner sphere of vapor CO₂ at room temperature ($\sim 21-22$ °C).
 791 For $\rho_{CO_2}=0.4$ g/cm³, this liquid phase will comprise 26% of the radius of the bub-
 792 ble, and the motion of the inner sphere of vapor because of Brownian motion would
 793 be readily observable under an optical microscope. Yet, we observe no two-phase
 794 bubbles, and there are no reports of two-phase bubbles in the Kilauean literature.

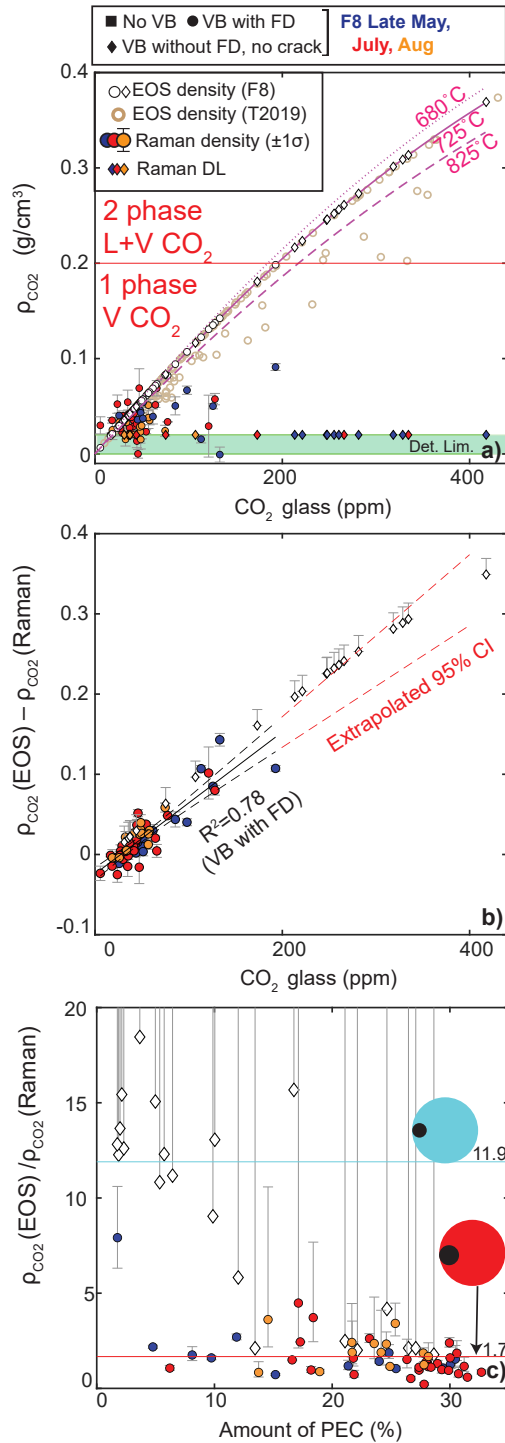


Figure 8. Comparisons of bubble CO₂ densities calculated using the EOS with those measured by Raman Spectroscopy. a) Calculated ρ_{CO_2} correlates strongly with glass CO₂. Bubbles within F8 melt inclusions are shown as white circles and diamonds (FD and no FD), bubbles within melt inclusions from Tucker et al. (2019) are shown as beige hollow circles. Magenta lines shows quadratic fits through calculated bubble densities for F8 melt inclusions for the EOS evaluated at 680°C, 725°C and 825°C. A number of inclusions with low inclusion CO₂ contents and high H₂O contents from Tucker et al. (2019) lie below this line, because of their higher X_{H_2O} values (Fig. 5c). Measured ρ_{CO_2} in this study are shown as colored circles, with error bars showing the 1σ of repeated acquisitions of each bubble. Colored diamonds (no FD, not cracked) are plotted at 0.02 g/cm³ (the presumed detection limit of Raman Spectroscopy; see Fig. 4c). b) The absolute discrepancy between predicted and measured ρ_{CO_2} , $\Delta \rho_{CO_2}$, correlates strongly with glass CO₂ content. The 95% confidence interval on a linear regression for measured bubble densities is shown with red dotted lines. Bubbles which did not produce a FD lie within error of the extrapolated confidence interval (assuming $\rho_{CO_2} = 0.02$ g/cm³). c) To allow comparison with bubble growth models in Fig. 9, the discrepancy between EOS methods and Raman measurements are shown as a factor (as above, VB without a FD assumed to contain 0.02 g/cm³). The proportion of the total bubble volume grown during quench for the High- and Low-Fo models shown in Fig. 9 are shown with red and cyan lines respectively. Error bars in b) and c) for VB with FD show the 1σ uncertainty of repeated Raman measurements, and those for VB without FD are calculated for DL between 0–0.02 g/cm⁻³ (hence they extend to infinity in c).

795 The fundamental tenet of the EOS method used by Tucker et al. (2019) is that
796 CO₂ continues to partition between the vapor bubble and the melt until the bubble
797 stops growing at the glass transition temperature. However, during syn-eruptive
798 quenching, the strong temperature dependence of CO₂ diffusivity means that the
799 diffusion of CO₂ from the melt into the bubble may cease before the bubble reaches
800 its final volume (Anderson and Brown, 1993). Continued bubble growth without
801 concurrent diffusion causes the density of CO₂ within the bubble to drop below that
802 predicted from the EOS (Aster et al., 2016; Moore et al., 2015; Maclennan, 2017).
803 Non-equilibrium bubble expansion has been proposed to account for the presence
804 of vapor bubbles in Icelandic melt inclusions with CO₂ concentrations below the
805 detection limit of Raman Spectroscopy (Neave et al., 2014).

806 The discrepancy between EOS predictions and Raman measurements ($\Delta \rho_{CO_2}$)
807 increases linearly with glass CO₂ content ($R^2=0.75$; shown as an absolute discrep-
808 ancy, Fig. 8b) and decreases with the amount of PEC (shown as a factor, Fig. 8c).
809 Melt inclusions containing bubbles without a FD lie within the confidence interval
810 of the regression through bubbles which produced a FD if the Raman detection
811 limit (0.02 g/cm³) is subtracted from CO₂ densities calculated from the EOS (Fig.
812 8b). To investigate these correlations, we assess the relative contribution of bubble
813 growth at high magmatic temperatures during PEC and ascent (where CO₂ diffusion
814 and bubble growth are coupled) compared to bubble growth during quench (where
815 CO₂ diffusion is temperature-limited, and therefore decoupled from the mechanical
816 expansion of the bubble).

817 We model melt inclusions from the point of entrapment to the glass transition
818 temperature using the model of Maclennan (2017; Fig. 9). Quench rates of 10°C/s
819 were used based on video footage of the sampling and quenching of the Aug-18 sam-
820 ple; ~40 s elapsed between the sample being pulled from the channel (~1150°C)
821 and becoming brittle at the glass transition temperature (~725 °C Tucker et al.,
822 2019). At these cooling rates, there is negligible transfer of CO₂ from the melt to
823 the bubble during syn-eruptive quenching. Two end-member cooling histories were
824 modelled. The red melt inclusion in Figure 9a experienced large amounts of cooling
825 ($\Delta T=150^\circ\text{C}$) and PEC at high magmatic temperatures and pressures, representa-
826 tive of the PT path followed by melt inclusions hosted within the most forsteritic
827 olivines. The blue melt inclusion in Figure 9b experiences no cooling and post-

828 entrapment crystallization prior to ascent and syn-eruptive quenching, representative
829 of Low-Fo melt inclusions which form in carrier melts with similar temperatures to
830 the ones in which they were erupted.

831 The High-Fo melt inclusion (red) grows a considerable proportion of its final
832 bubble volume (58%) during PEC at high magmatic temperatures (square to star
833 symbol; Fig. 9a). The diffusion of CO₂ into this growing bubble causes the CO₂
834 content of the melt phase to drop rapidly (Fig. 9c). During syn-eruptive quenching,
835 there is no further CO₂ diffusion between the melt and bubble (Fig. 9c). This stage
836 of bubble growth accounts for 42% of the final volume, with ρ_{CO_2} decreasing from
837 0.10 to 0.06 g/cm³ (Fig. 9a, d). As the EOS method effectively predicts the density
838 of CO₂ in the vapor bubble prior to the final, quench-induced stage of bubble ex-
839 pansion, the EOS method overpredicts the CO₂ density by a factor of 1.7× in this
840 example. This lies well within the deviation between measured and predicted CO₂
841 contents for High-Fo F8 melt inclusions which have experienced >10% PEC (red line
842 on Fig. 8c). In this case, the proportion of the bubble grown at high temperatures
843 will be substantially greater, as the model of Maclennan (2017) does not account for
844 the FeO-loss process, which greatly increases the amount of PEC for a given ΔT .
845 The volume of the bubble grown during syn-eruptive quench is determined by the
846 difference between the temperature at the initiation of syn-eruptive quenching, and
847 the glass transition temperature, so is almost constant for different PT paths. In
848 contrast, with increasing amounts of PEC, the volume of the bubble grown at high
849 temperatures gets progressively larger, so the relative expansion of the bubble during
850 quench (and therefore the change in CO₂ density) gets progressively smaller. For
851 example, in models with $\Delta T=200^\circ\text{C}$ instead of $\Delta T=150^\circ\text{C}$, the amount of PEC
852 increases from 18% to 25%, and the proportion of the bubble grown at high temper-
853 ature increases from 58% to 68%. In turn the bubble density drops from only 0.073
854 to 0.052 g/cm³ during syn-eruptive quenching (so the EOS method would only over
855 predict by a factor of $\sim 1.4\times$).

856 In contrast, the Low-Fo melt inclusion (blue) grows a very small proportion of
857 its total bubble volume at high temperatures (10%), with 90% of the final bubble
858 volume growing upon quench (Fig. 9b). Substantial bubble expansion upon quench
859 without concurrent CO₂ diffusion causes ρ_{CO_2} to drop substantially (Fig. 9d). Ef-
860 fectively, the EOS method calculates the density of the bubble at the initiation of

861 the quench stage ($\rho_{CO_2}=0.205$ g/cm³; star symbol), while the true bubble density
862 is $11.9\times$ lower ($\rho_{CO_2}=0.021$ g/cm³; circle symbol), close to the detection limit of
863 Raman spectroscopy. This calculated discrepancy is very similar to that for vapor
864 bubbles in Low-Fo inclusions which do not have Fermi diads (assuming the detection
865 limit= 0.02 g/cm³, cyan line, Fig. 8c).

866 In summary, the EOS substantially overestimates ρ_{CO_2} for melt inclusions
867 which have experienced small amounts of PEC and retain high CO₂ contents (Fig.
868 8b,c), because bubble growth in these melt inclusions is dominated by the quench-
869 ing process where there is no diffusion of CO₂ into the bubble. In contrast to these
870 very large discrepancies (factors of ~ 10), bubble densities in melt inclusions which
871 have experienced extensive PEC are broadly matched by the EOS method (within a
872 factor of ~ 2 ; Fig. 8c).

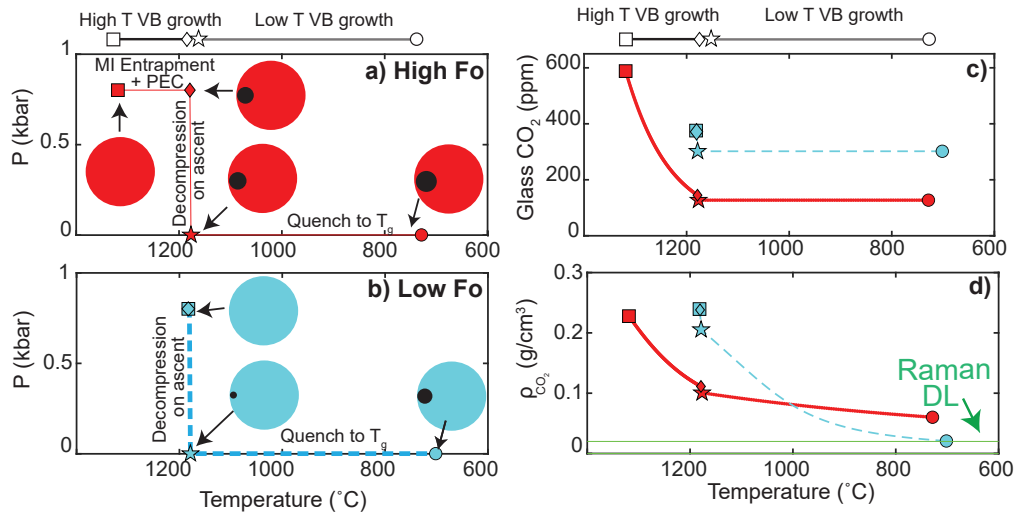


Figure 9. Model of CO₂ partitioning between the melt and bubble for PT scenarios representative of inclusions hosted within High and Low-Fo olivines (red and blue colors, respectively). a) The red melt inclusion experiences considerable cooling ($\Delta T=150^\circ\text{C}$) and post-entrapment crystallization at high temperatures and pressures (square to diamond symbol), driving the growth of a vapor bubble. This high temperature phase of bubble growth is accompanied by CO₂ diffusion from the melt to the bubble, causing the glass CO₂ content to drop substantially (c). This inclusion then ascends to the surface (diamond to star symbol), and experiences a second stage of vapor bubble growth during syn-eruptive quenching (star to circle symbol). b) The blue melt inclusion follows an end-member PT path representative of an inclusion hosted within a Low-Fo olivine. It experiences no cooling and post-entrapment crystallization at high temperature. A bubble only begins to grow during ascent to the surface, with 90% of the total bubble volume of this inclusion occurs during syn-eruptive quenching (star to circle). At the quenching rates of 10°C/s used in this model, there is negligible CO₂ transfer from the glass to the bubble during this low temperature phase of bubble growth. The large amount of bubble expansion without concurrent CO₂ diffusion causes the density of CO₂ in the vapor bubble to drop close to the detection limit of Raman Spectroscopy (green line, d), while the CO₂ of the glass phase remains unchanged (c).

873 These bubble-growth models show that the magnitude of the discrepancy be-
 874 tween measured bubble densities and those predicted by the EOS relates to the
 875 proportion of the bubble grown during syn-eruptive quenching. In contrast, Tucker
 876 et al. (2019) suggest that Raman measurements may underestimate ρ_{CO_2} relative to

EOS methods because of the sequestration of significant quantities of CO₂ as thin
films of solid carbonate on bubble walls. Carbonate phases have been identified in a
number of melt inclusion vapor bubbles from subduction zone settings based on the
presence of a distinctive peak in the Raman spectra at $\sim 1090\text{ cm}^{-1}$ (Venugopal et
al., 2020; Moore et al., 2015). However, while Moore et al. (2015) report relatively
abundant carbonate phases in vapor bubbles from Seguam and Fuego, only four of
the 142 Kilauean vapor bubbles they examined contained carbonates, all of which
were hosted within a single olivine crystal. This suggests that vapor bubble carbon-
ates are significantly less common in H₂O-poor ocean island systems. We observe no
carbonate peaks in Raman spectra from F8 bubbles, nor during optical observations
made prior to the exposure of bubbles during polishing. Additionally, no carbonate
phases were identified during detailed examination of exposed bubble walls using
backscatter and secondary electron imaging, and Energy-Dispersive Spectroscopy
(EDS) maps on a FEG-SEM. These EDS maps reveal that bubble wall coatings with
a “dotted” appearance identified by Tucker et al. (2019) (see their Fig. 2F) consist
of Fe-Cu sulfides, rather than carbonates (see also Venugopal et al., 2020; Moore et
al., 2015; Wieser, Jenner, et al., 2020). Finally, even if carbonates in bubble walls
remained undetected, our observations regarding the systematic relationship between
PEC amounts, CO₂ contents, and the discrepancy between Raman measurements
and the EOS would necessitate that only bubbles hosted in melt inclusions which
had undergone negligible PEC contain carbonate phases.

5.5 Reconstructing Magma Storage Depths

Under the assumption that any reservoir from which a substantial proportion
of the crystal cargo was derived must also have supplied melt (in order to entrain
these crystals, and carry them to the surface), the depths of the main magma reser-
voirs supplying F8 can be estimated from melt inclusion entrapment pressures (for a
known crustal density). Entrapment pressures were calculated from PEC-corrected
total CO₂ and major element contents, and temperatures calculated using the MgO-
liquid thermometer of Helz and Thornber (1987) for PEC-corrected MgO contents.
As melt inclusion H₂O contents have been reset by diffusive re-equilibration, satu-
ration pressures were calculated assuming H₂O=0.5 wt%, based on the distribution
of measured H₂O contents in literature studies of Kilauean melt inclusions and un-

909 degassed submarine glasses from the ERZ (Fig. 5b; Sides, Edmonds, Maclennan,
 910 Houghton, et al., 2014; Sides, Edmonds, Maclennan, Swanson, & Houghton, 2014;
 911 Clague et al., 1995; Dixon et al., 1991; Tucker et al., 2019). Entrapment pressures
 912 for measured water contents are also shown in the Supplementary Information. En-
 913 trapment pressures were converted into magma storage depths assuming $\rho=2400$
 914 kg/m^{-3} (for consistency with modelling of the geodetic signals from the 2018 sum-
 915 mit collapse by Anderson et al., 2019). Initially, we consider melt inclusions with no
 916 vapor bubble, or a vapor bubble which produced a FD, due to the uncertainty in the
 917 CO_2 density of vapor bubbles which do not contain a FD.

918 Literature studies of Kīlauean melt inclusions have mostly calculated satura-
 919 tion pressures using the $\text{CO}_2\text{-H}_2\text{O}$ solubility model of Dixon et al. (1995) and Dixon
 920 (1997), implemented in the excel workbook VolatileCalc (Newman & Lowenstern,
 921 2002, e.g., Tuohy et al. 2016; Sides et al. 2014a, b; Moore et al., 2015; Tucker et al.,
 922 2019). VolatileCalc-Basalt uses a simplified relationship for the compositional de-
 923 pendence of CO_2 solubility expressed in terms of just the melt SiO_2 content, rather
 924 than the full compositional parameter Π which accounts for the abundance of seven
 925 cations (Dixon, 1997, Fig. 6c). In this simplification the parameter $X_{\text{CO}_3^{2-}}(P_0, T_0)$,
 926 which representing the solubility of CO_2 at 1200 °C and 1 bar for a specified fluid
 927 CO_2 fugacity in the thermodynamic expression of Dixon et al. (1995), is expressed
 928 as:

$$X_{\text{CO}_3^{2-}}(P_0, T_0) = 8.7 \times 10^{-6} - 1.7 \times 10^{-7}[\text{SiO}_2] \quad (6)$$

929 This relationship derives from the excellent linear correlation between Π and
 930 SiO_2 in a suite of lavas with 40–49 wt% from the North Arch Volcanic field (blue
 931 regression line; Fig. 6c; Dixon et al., 1997). However, extrapolation of Equation 6
 932 beyond 51.2 wt% SiO_2 returns a negative value for $X_{\text{CO}_3^{2-}}(P_0, T_0)$, which, in turn,
 933 predicts that the solubility of CO_2 is negative at all pressures. To avoid these ex-
 934 trapolation issues, VolatileCalc-Basalt does not let users enter a SiO_2 content >49
 935 wt%, so most studies simply calculate the CO_2 solubility for melts with >49 wt%
 936 SiO_2 using the expression for $\text{SiO}_2=49$ wt% (e.g., Tucker et al., 2019; Sides, Ed-
 937 monds, Maclennan, Houghton, et al., 2014; Sides, Edmonds, Maclennan, Swanson,
 938 & Houghton, 2014). Newman and Lowenstern (2002) suggest that this approxima-

939 tion should return accurate entrapment pressures for basaltic compositions with up
940 to 52 wt% SiO₂ contents. However, the simplified compositional parameter used in
941 VolatileCalc-Basalt is only valid for melt compositions which define the same tra-
942 jectories in Π vs. SiO₂ space as the North Arch Lavas. F8 melt inclusions which
943 have undergone >10% PEC are offset to substantially higher Π values at a given
944 SiO₂ (Fig. 6c), so VolatileCalc-Basalt underestimates the solubility of CO₂. Addi-
945 tionally, while F8 melt inclusions show a large drop in Π with increasing SiO₂, all
946 but four melt inclusions have SiO₂ >49 wt%, so are treated as if they had the same
947 composition in VolatileCalc-Basalt (red line; Fig. 6c). Thus, VolatileCalc-Basalt not
948 only underestimates CO₂ solubility, and therefore overestimates entrapment pres-
949 sures for F8 melt inclusions hosted in High-Fo olivines, it also neglects compositional
950 variations in CO₂ solubility within this suite (Fig. 6c).

951 To demonstrate the importance of evaluating the suitability of different solu-
952 bility models, we compare entrapment pressures from VolatileCalc-Basalt with the
953 models of Ghiorso and Gualda (2015), hereafter MagmaSat, Iacono-Marziano et
954 al. (2012) with hydrous coefficients, hereafter IM-2012, and Shishkina et al. (2014),
955 hereafter S-2014, using the open-source python tool VESIcal (Iacovino et al., 2020).
956 These three models utilize more than a decade of additional experiments on basaltic
957 compositions compared to the expressions implemented in VolatileCalc-Basalt. By
958 extension, these models are calibrated on a significantly larger compositional range
959 (Fig. 6a), so more effectively encapsulate variability in CO₂ solubility as a function
960 of melt composition.

961 Entrapment pressures for melt inclusions hosted in Low-Fo olivines from F8
962 calculated using VolatileCalc-Basalt, S-2014, and IM-2012 are statistically indistin-
963 guishable using the KS test at $p=0.05$ (Fig. 10a), likely because the major element
964 compositions of these melt inclusions lie within the calibration range of all four sol-
965 ubility models (Fig. 6a). MagmaSat returns slightly lower pressures, although these
966 are not statistically distinguishable ($p=0.1$ vs. S-2014). These slight discrepancies
967 likely reflect the differential treatment of mixing between H₂O and CO₂ fluids in
968 these different models (e.g., non-ideal mixing in MagmaSat and IM-2012 vs. ideal
969 mixing in S-2014 and VolatileCalc-Basalt; see Supporting Information Fig. S1).
970 As only 2 Low-Fo melt inclusions have vapor bubbles producing a FD ($N=2$), the
971 distribution of entrapment pressures calculated using just glass CO₂ contents are

972 indistinguishable from those using total CO₂ contents (dotted magenta vs. solid red
973 lines; Fig. 10a).

974 In contrast, there are substantial differences between the entrapment pressures
975 obtained from different solubility models for High-Fo melt inclusions ($>F_{O_{81.5}}$), with
976 MagmaSat and S-2014 plotting to significantly lower pressures than IM-2012 and
977 VolatileCalc-Basalt (both pairs are statistically indistinguishable from one another
978 at $p=0.05$; Fig. 10b). As discussed above, the simplification of the compositional
979 dependence in VolatileCalc-Basalt means that this model underestimates CO₂ solu-
980 bility, and therefore overestimates entrapment pressures for High-Fo melt inclusions
981 (Fig. 6c). Similarly, Iacono-Marziano et al. (2012) warn that their semi-empirical
982 model poorly incorporates the compositional effect of melt MgO contents on CO₂
983 solubility, as the vast majority of melts in their calibration dataset have $\sim 6\text{--}8$ wt%
984 MgO. In contrast, High-Fo PEC-corrected melt inclusions have MgO contents rang-
985 ing from 7.8–13.7 wt% (Fig. 6a). The calibration dataset for the S-2014 model
986 incorporates a significantly broader range of basaltic compositions, including melts
987 with MgO contents similar to PEC-corrected High-Fo melt inclusions (Fig. 6a). The
988 MagmaSat calibration dataset is similarly extensive (including the experiments used
989 to calibrate S-2014, IM-2012 and VolatileCalcBasalt). As for Low-Fo melt inclusions,
990 MagmaSat is offset to slightly lower pressures than S-2014 (median offset of 0.1
991 kbar).

992 Overall, we favour entrapment pressures from MagmaSat (Fig. 11, as it has
993 the largest calibration dataset, and is a full thermodynamic model (whereas S-2014
994 is purely empirical). Additionally, the S-2014 model predicts ~ 1 wt% H₂O at 0
995 bar, meaning that it is effectively evaluating the solubility of pure CO₂ for the H₂O
996 contents considered here (so shows no change in saturation pressure with variation
997 in H₂O contents between 0–1 wt% H₂O, see Supporting Information Fig. S1). As
998 shown in Fig. 10, differences between Shishkina and MagmaSat are relatively small.
999 For High-Fo inclusions, the differences between these models are statistically in-
1000 significant, and easily overwhelmed with the errors associated with bubble volumes
1001 (error bars on Fig. 11a). For completeness, Supporting Information Fig. S12 shows
1002 forsterite vs. depth plots similar to those shown in Fig. 11 for reconstructions using
1003 Shishkina, and for measured and fixed H₂O contents.

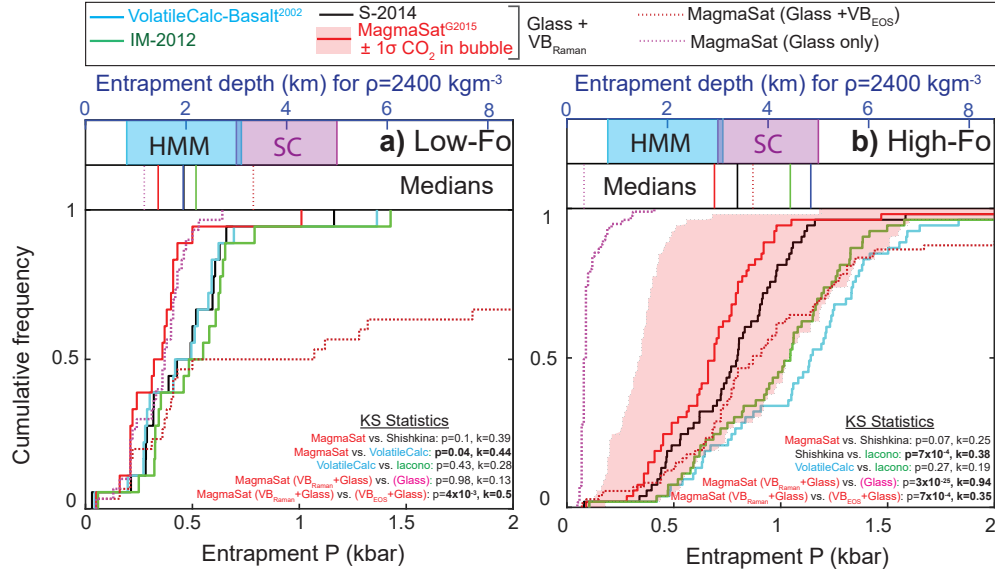
1004 Using MagmaSat, Low-Fo melt inclusions yield median entrapment depths
1005 (assuming $\rho=2400 \text{ kg/m}^{-3}$) of 1.44 km (lower and upper 68%=0.89–1.74 km). The
1006 median centroid depth, aspect ratio and reservoir volume derived from modelling of
1007 the first stage of the 2018 caldera collapse by Anderson et al. (2019) suggests that
1008 the HMM reservoir spans depths of 0.82–3.1 km, which aligns well with our entrap-
1009 ment depths, which mainly cluster in the top half of that range (perhaps suggesting
1010 melt inclusion formation was favoured in the upper half of the reservoir). The low
1011 PEC amounts experienced by these melt inclusions, the absence of cracks, and the
1012 fact that the two Low-Fo inclusions which did yield a diad had very low CO_2 den-
1013 sities (Fig. 4c), suggests that melt inclusions with a vapor bubble which did not
1014 produce a FD likely contained very small quantities of CO_2 (because the bubble
1015 predominantly forming during syn-eruptive quench; Fig. 9). Thus, we also consider
1016 entrapment depths from these melt inclusions (diamond shapes on Fig. 11a). This
1017 extends the distribution of entrapment depths to slightly deeper depths, which show
1018 an even better overlap with the depths of the HMM reservoir suggested by Anderson
1019 et al. (2019).

1020 Considering only High-Fo melt inclusions with a measurable Fermi diad (due to
1021 the uncertainty in the amount of CO_2 held within vapor bubbles which did not pro-
1022 duce a FD in melt inclusions which have undergone extensive PEC), the distribution
1023 of entrapment depths (KS test, $p=1.6 \times 10^{-7}$) and means (ANOVA, $p=2.5 \times 10^{-6}$) are
1024 offset to significantly higher pressures than Low-Fo melt inclusions (Fig. 11a). Con-
1025 sidering the error associated with reconstructing bubble CO_2 contents from bubble
1026 volumes estimated from 2D images (shown in pink on Fig. 10b), the distribution of
1027 entrapment depths for High-Fo olivines overlaps remarkably well with geophysical
1028 estimates of the depth of the SC reservoir (3–5 km; Poland et al., 2015). In detail,
1029 High-Fo olivines seem to form two main groups, one located at ~ 2 km depth, and a
1030 second located at 3–5 km depth (Fig. 11a).

1031 The quench-dominated mechanism of bubble growth in Low-Fo olivines means
1032 that very little CO_2 is held within the vapor bubble. Thus, entrapment depths
1033 calculated using glass-only measurements are statistically indistinguishable from
1034 those combining bubble and glass measurements (Fig. 10a). In contrast, entrapment
1035 depths calculated using just glass CO_2 contents in High-Fo olivines are anomalously
1036 shallow (median=0.38 km, lower and upper 68%=0.3–0.51 km; Fig. 11b), because

1037 bubble growth at high temperatures during PEC has resulted in the vast majority of
1038 the CO₂ entering the vapor bubble (Fig. 9).

1039 Use of EOS techniques to reconstruct CO₂ contents of vapor bubbles yields
1040 very high entrapment depths for Low-Fo olivines (median=3.3 km, lower and upper
1041 68%=0.89–10.8 km). Crucially, 13 inclusions yield entrapment depths >5 km (the
1042 inferred base of the SC reservoir), because the EOS method drastically overestimates
1043 bubble CO₂ densities in inclusions which have experienced minimal PEC (Fig. 8b-c).
1044 For High-Fo olivines, there is a better overlap between entrapment depths calculated
1045 using EOS methods, and Raman measurements, and EOS methods get closer to the
1046 true distribution of entrapment pressures than measurements of only the glass phase
1047 (Fig. 10b). However, EOS methods still predict that 23 melt inclusions crystallized
1048 at >5 km depth, with one forming at 26.4 km, compared to only two entrapment
1049 depths at 6.3 and 8.8 km using Raman reconstructions of bubble CO₂.



5.6 Summit-Rift Connectivity

Melt inclusion entrapment depths indicate that olivine crystals erupted at F8 crystallized within both the shallower HMM reservoir (Low-Fo olivines) and the deeper, SC reservoir (High-Fo olivines, see also A. Lerner, 2020). The low degrees of olivine-melt disequilibrium and limited amounts of PEC experienced by melt inclusions hosted within Low-Fo olivines implies that these crystals grew in a melt with a similar Mg#, and therefore temperature, to the carrier melt in which they were erupted. In contrast, the high degrees of olivine-melt disequilibrium and large amounts of PEC indicates that High-Fo crystals were mixed into a significantly lower Mg# (and therefore cooler) carrier liquid than the liquid in which they crystallized. Based on reports of lattice distortions (Gansecki et al., 2019) in some F8 olivines, high core forsterite contents, and the clustering of entrapment pressures between 3–5 km (Fig. 11), we suggest that these olivines grew in the SC reservoir, and then settled into mush piles at the base of this reservoir where they were stored for prolonged periods (perhaps as long as centuries to millenia; Wieser, Edmonds, et al., 2020).

Seismic swarms and the initiation of inflationary tilt in March to April 2018 have been interpreted to record the injection of new melts into the South Caldera reservoir (Neal et al., 2019; Flinders et al., 2020), which may have disturbed the olivine mush pile. These new melts (along with the High-Fo olivines they scavenged) would then have mixed into the cooler, lower Mg# melts present within the middle to upper parts of the SC reservoir. Alternatively, if inflationary signals were generated by a reduction in the amount of magma flowing along the ERZ to Pu‘u ‘Ō‘ō (Patrick et al., 2020), progressive internal pressurization of the SC reservoir could also disturb piles of settled crystals. Rapid cooling of mush-derived olivines following their mixing into more evolved melts would have initiated large amounts of PEC. Using the method of Danyushevsky et al. (2002, 2000), the degrees of Mg# re-equilibration between melt inclusions and host olivine crystals (~70-100%) indicate that crystals were resident in these cooler melts for timescales of approximately a month to a year prior to their eruption at Fissure 8. This is consistent with the time lag between geophysical signals indicating increasing pressurization of the magmatic system in March, and the eruption of crystals between late May and August.

1082 The fact that only two melt inclusions record entrapment depths >5 km rules
1083 out models where high forsterite olivines grew in deeper magma storage reservoirs
1084 near the base of the volcanic pile (as suggested for Kīlauea's prehistoric explosive
1085 period by Lynn et al., 2017), or within Kīlauea's deep rift zones at ~ 6 –9 km (Fig 11
1086 Clague & Denlinger, 1994).

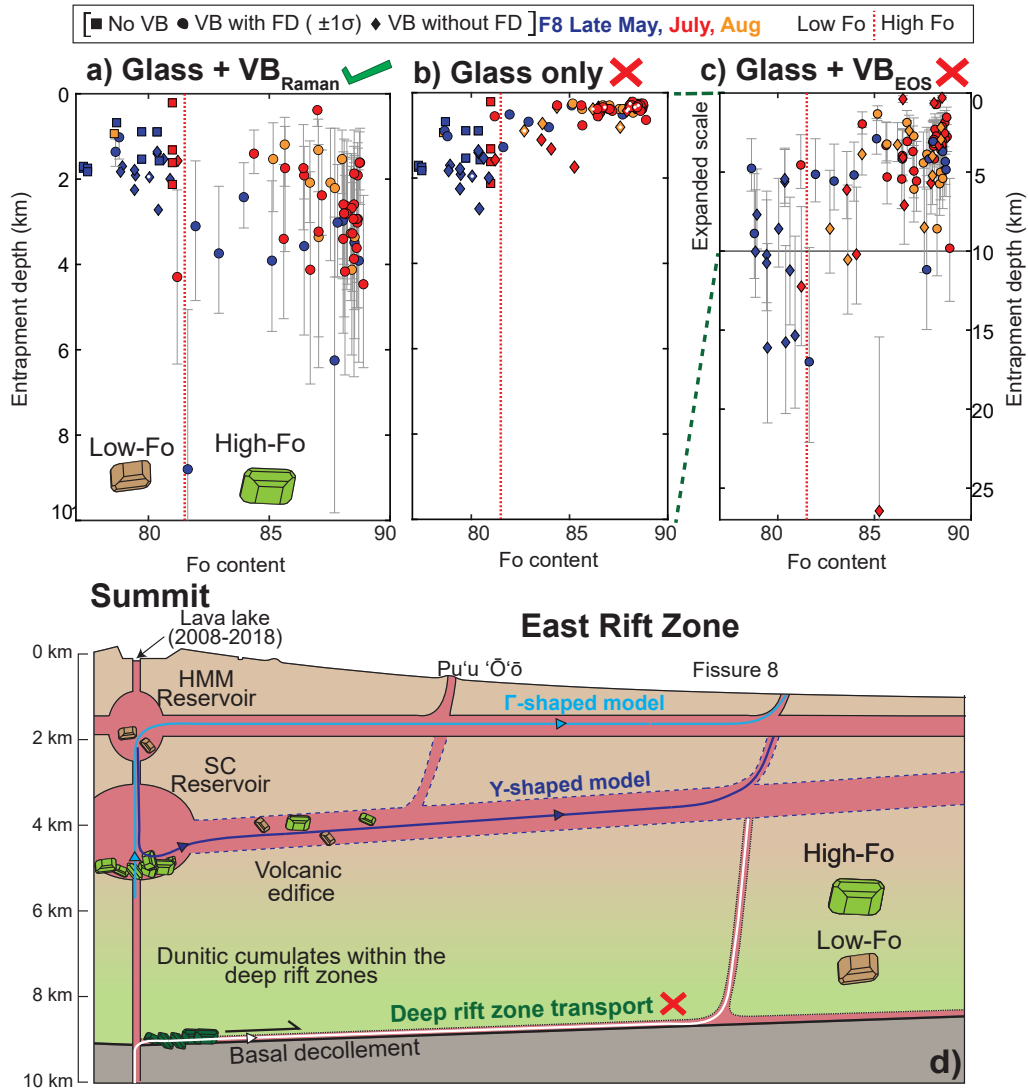


Figure 11.

1087 **Caption Fig. 11.** Schematic diagram of Kilauea's plumbing system, informed
 1088 by entrapment depths from MagmaSat for PEC-corrected melt inclusion compo-
 1089 sitions (assuming $\rho=2400 \text{ kg/m}^3$ following Anderson et al. (2019)). a) Preferred
 1090 entrapment depths from this study (all melt inclusions for Low-Fo olivines, only
 1091 those with a FD for High-Fo olivines). Error bars on bubble-free melt inclusions
 1092 from SIMS analyses are smaller than the symbol size. Error bars for bubble-bearing
 1093 melt inclusions were calculated from the minimum and maximum possible total CO_2
 1094 content using the 1σ error calculated from repeated Raman analyses of each bubble,
 1095 and the 1σ estimated by Tucker et al. (2019) associated with calculating 3D bubble

1096 volume proportions from 2D images (-48 to 37%). b) Entrapment depths estimated
1097 from analyses of only the glass phase are anomalously shallow for High-Fo olivines.
1098 c) Entrapment depths using the EOS method to reconstruct bubble CO₂ contents
1099 are anomalously deep, with large numbers of inclusions plotting at >5 km depth
1100 (note change in scale). Error bar reflects the uncertainty associated with calculating
1101 3D bubble volume proportions from 2D images. d) Cross section showing the three
1102 hypothesized magma transport paths supplying rift zone eruptions.

1103 The mechanism by which crystal populations grown in the HMM and SC
1104 reservoirs were mixed into a single carrier melt encapsulates an ongoing debate at
1105 Kīlauea regarding the geometry of the connection between the rift zone conduit and
1106 the summit reservoir system. This connection has been variably described as a Y-
1107 shaped feeder system with the SC reservoir feeding both the HMM reservoir and the
1108 ERZ conduit with two discrete conduits (Pietruszka et al., 2018; Poland et al., 2015,
1109 Model 2, Fig. 11d), or a Γ -shaped feeder system with a vertical conduit between the
1110 HMM and the SC reservoir, and a single, near-horizontal conduit from the HMM
1111 reservoir into the ERZ (Cervelli & Miklius, 2003, Model 3, Fig. 11d). Cervelli and
1112 Miklius (2003) suggest that the Γ -shaped model is more plausible because a shal-
1113 low conduit (which is subject to less lithostatic pressure) is more likely to remain
1114 open during pauses in eruptive activity than a deep conduit, and because shallow
1115 intrusions into the upper ERZ influence both the HMM reservoir and activity at
1116 Pu‘u ‘Ō‘ō. However, Poland et al. (2015) favour the Y-shaped model based on earth-
1117 quake and InSAR observations that dyke intrusions into the ERZ in 2007 and 2011
1118 ascended from a depth of \sim 2–3 km.

1119 For both reservoir geometries, the olivine mush pile at the base of the SC
1120 reservoir may have been disturbed by the input of new magma into Kīlauea’s sum-
1121 mit inferred from geophysical signals (Neal et al., 2019; Flinders et al., 2020), or
1122 progressive internal pressurization due to a drop in magma output to Pu‘u ‘Ō‘ō.
1123 In the Γ -shaped model, High-Fo crystals sourced from the SC mush pile may have
1124 ascended into the HMM reservoir, and then been transported along a shallow rift
1125 zone conduit to the site of the eruption along with Low-Fo olivines. However, the
1126 Y-shaped model provides an additional mechanism by which to disturb the SC mush
1127 pile. In this geometry, melts from the HMM reservoir carrying Low-Fo olivine crys-
1128 tals would have drained down through the SC reservoir before passing out onto

1129 the rift zone, with significant potential for this downward flow, aided by the large
1130 scale collapse of Kīlauea’s caldera, to erode the SC mush pile. Interestingly, the pro-
1131 portion of crystals which are out of equilibrium with their carrier melts increases
1132 substantially between May-August 2018 (Fig. 2a), and the degree of re-equilibration
1133 between melt inclusions and host crystals decreases (Fig. 3b).

1134 If the disturbance to the mush pile was solely the result of pressurization of
1135 the volcanic plumbing system, it might be expected that the majority of High-Fo
1136 olivines were disturbed from their mush piles in mid-March to April 2019, when in-
1137 flationary signals were the strongest (Patrick et al., 2020; Neal et al., 2019). In this
1138 scenario, High-Fo olivines might be expected to be more dominant in the May-18 vs.
1139 July and Aug-18 samples. In contrast, increasing erosion and scavenging of High-
1140 Fo olivines during the downdraining of melts from the HMM reservoir into the SC
1141 reservoir during the summit collapse could account for the increase in the proportion
1142 of High-Fo olivines with time, similar to the mechanism suggested by Teasdale et al.
1143 (2005) for the 1998 eruption of Cerro Azul, Galápagos. Erosion of the mush pile by
1144 down-draining from the shallower HMM reservoir, into which the summit caldera
1145 was collapsing, also accounts for the fact that High-Fo olivines were extremely rare
1146 during the 35 year Pu‘u ‘Ō‘ō eruption.

1147 Another possibility is that some melt inclusions were trapped during the 40
1148 km of transport down the ERZ to the site of the eruption (Patrick, Dietterich, et al.,
1149 2019). Assessing this hypothesis requires assumptions regarding the depth of magma
1150 transport. Given that the dyke to the LERZ propagated downrift from Pu‘u ‘Ō‘ō,
1151 we assume that the dyke had a similar depth to intrusions within the proximity of
1152 Pu‘u ‘Ō‘ō between 1997–2007, which have been studied in detail, and shown to rise
1153 from the ERZ conduit at depths of ~ 2 –2.4 km (Owen et al., 2000; Montgomery-
1154 Brown et al., 2011, and refs within). Thus, it is plausible that some of the Low-Fo
1155 olivines with entrapment depths near ~ 2 km may have growth in the rift zone. How-
1156 ever, crystallization within the ERZ conduit and dyke would likely occur throughout
1157 the eruption, yet the abundance of Low-Fo olivine crystals declines as the eruption
1158 proceeds

1159 The cluster of High-Fo olivines at ~ 2 km could also represent crystallization
1160 during down-rift transport. These olivine crystals have Fo contents between 84 and

1161 89, which must have grown from melts with MgO contents between 8.5–13.1 wt%
1162 (for $K_D=0.3$, $\text{FeO}_T=11.33$ wt%, with $\text{Fe}^{3+}/\text{Fe}_T=0.15$). Yet, the highest erupted
1163 glass MgO content during the 2018 LERZ eruption is 6.74 wt% MgO (Fig. 3a and
1164 Gansecki et al., 2019). Moreover, glass MgO contents during the 35-year Pu‘u ‘Ō‘ō
1165 eruption did not exceed 8 wt% MgO (see Fig. 8.2 Thornber et al., 2015), suggesting
1166 that high MgO melts may not have been present in the rift zone conduit since the
1167 early phases of the Mauna Ulu eruption in 1969 (Wieser et al., 2019). In contrast,
1168 based on the occurrence of high MgO glass shards in a number of eruptions around
1169 the summit caldera, Helz et al. (2015) suggest that melts with 6.5–11 wt% MgO are
1170 present in the summit reservoir over many centuries. This supports our inference
1171 that the High-Fo olivines erupted at F8 crystallized from high MgO melts supplied
1172 from the Hawaiian mantle plume within the SC reservoir. These high MgO melts are
1173 very rarely erupted at the surface as they rapidly mix with more evolved, resident
1174 melts within the reservoir, so the only record of their existence are the olivines they
1175 crystallize. Given the rarity of these high MgO melts at the surface, it is difficult to
1176 imagine a situation where these melts would avoid mixing with resident magmas in
1177 the summit reservoir, and manage to ascend prolonged distances along the ERZ con-
1178 duct (which must be dominated by low MgO melts based on the composition of the
1179 co-erupted carrier liquid at F8). Finally, if these High-Fo olivines crystallized in the
1180 rift zone, they must have been resident for between a month and a year before they
1181 erupted at F8 (based on the degree of Mg# re-equilibration between melt inclusions
1182 and host olivine crystals).

1183 Interestingly, the May-18 sample does not show the distinctive clustering of
1184 High-Fo entrapment depths at ~ 2 km seen in the July and Aug-18 sample. This
1185 may result from the relatively small number of measurements of High-Fo olivines
1186 in this sample ($N=12$). Alternatively, it may suggest that the two reservoirs be-
1187 came increasingly connected during the collapse of the summit caldera, allowing
1188 remobilized High-Fo crystals from the SC mush pile to be transported up into the
1189 shallower HMM reservoir. The juxtaposition of these hot crystals with cooler melts
1190 within this reservoir may have led to dissolution or rapid growth (Shea et al., 2019;
1191 Mourey et al., 2020), favouring the formation of embayments. Perhaps due to the
1192 mixing with a hotter, and higher Mg# melt, growth may have resumed, sealing off
1193 melt inclusions recording shallower entrapment depths, before the crystal cargo was

1194 drained back down in the SC reservoir, and out along the ERZ conduit. It is also
1195 possible that the two reservoir systems always have a higher degree of connectivity
1196 than indicated by schematic diagrams such as Fig. 11, with frequent cycling of melt
1197 and crystals between the two reservoirs (and it is simply chance that these lower P
1198 inclusions were not seen in the May-18 sample). Further investigation of geophysical
1199 datasets from the 2018 eruption should provide tighter constraints on the depth of
1200 rift zone transport and dike propagation, allowing more rigorous assessments of the
1201 magma transport geometries indicated by our barometric estimates. Additionally,
1202 more detailed work on timescales from diffusive re-equilibration of Fe-Mg in both
1203 melt inclusions and host crystals will help evaluate differences between the High-Fo
1204 crystal cargo erupted at F8 between May and August.

1205 **6 Conclusion**

1206 Detailed investigations of melt inclusion volatile systematics from the 2018
1207 eruption of Kīlauea reveal that the erupted crystal cargo originated from both the
1208 Halema'uma'u reservoir (Low-Fo olivines; $\sim 1\text{--}2$ km depth) and the South Caldera
1209 reservoir (High-Fo olivines, $\sim 3\text{--}5$ km depth). This demonstrates that in addition to
1210 the supply of magma from the HMM reservoir inferred from geophysical modelling
1211 of the summit collapse (Anderson et al., 2019), a substantial volume of magma must
1212 also have been derived from the SC reservoir in order to transport these High-Fo
1213 crystals to the surface. This supports recent estimates of the total amount of SO_2
1214 emitted from F8 (Kern et al., 2020), which requires the erupted volume to have
1215 been approximately twice that inferred to have drained from the HMM reservoir by
1216 Anderson et al. (2019).

1217 High-Fo Melt inclusions, which mostly yield entrapment depths aligned with
1218 geophysical estimates of the depth of the SC reservoir ($\sim 3\text{--}5$ km), host the vast
1219 majority of their CO_2 budget in the vapor bubble ($\sim 90\%$). This is a consequence
1220 of the large amounts of PEC experienced by these melt inclusions following their
1221 entrainment into cooler, lower Mg# melts. Based on the textural and chemical
1222 similarities of these High-Fo crystals and those observed at previous eruptions at
1223 Kīlauea (Wieser, Edmonds, et al., 2020; Wieser et al., 2019), we suggest that these
1224 olivines grew from high MgO melts present at the base of the SC reservoir (Helz
1225 et al., 2015), and settled into mush piles for prolonged time periods. Based on the

1226 degree of Mg# re-equilibration between melt inclusions and host olivines, we sug-
1227 gest that these olivines were mobilized from mush piles and mixed into lower Mg#
1228 carrier melts approximately a month to a year before they erupted at Fissure 8.
1229 This disturbance may correspond with the onset of geophysical signals of inflation
1230 in March-April, 2018, interpreted to represent the injection of new melts into the
1231 plumbing system, or a reduction in output from the summit reservoir (Flinders et
1232 al., 2020; Patrick et al., 2020). Because of the large amount of CO₂ in the vapour
1233 bubbles of these inclusions, entrapment depths calculated using only glass CO₂ con-
1234 tents would yield anomalously low entrapment depths (~0.3–0.5 km), and fail to
1235 recognise that the SC reservoir supplied significant volumes of magma to Fissure 8.

1236 In contrast, Low-Fo melt inclusions are closer to equilibrium with their carrier
1237 melts, so have experienced smaller amounts of PEC. Where present, the vapor bub-
1238 ble in these melt inclusions is very CO₂-poor, and grew most of its volume during
1239 during syn-eruptive quenching (~90%). As the quench rates of these samples mean
1240 that there was almost no diffusion of CO₂ between the melt and bubble during this
1241 growth phase, reconstructions of bubble CO₂ using equation of state methods yield
1242 anomalously high entrapment depths (4.5–16.1 km; Fig. 11c).

1243 Careful choice of a CO₂-H₂O solubility model is also vital to obtain accurate
1244 entrapment pressures, and therefore depths. Importantly, the basaltic functions of
1245 VolatileCalc, which has been used the majority of previous Kīlauean melt inclusion
1246 studies, overpredict entrapment pressures for High-Fo melt inclusions, due to the
1247 simplified relationship between CO₂ solubility and melt composition in this model.
1248 Like EOS methods, use of this model would indicate that ~50% of melt inclusions
1249 crystallized deeper than the base of the SC reservoir at >5 km (requiring the pres-
1250 ence of a previously unrecognised storage reservoir; Fig. 10).

1251 Overall, our study highlights the importance of measuring bubble densities using
1252 Raman Spectroscopy in addition to measurements of the melt phase by SIMS or
1253 FTIR. We also emphasize the importance of carefully evaluating the compositional
1254 range of different solubility models relative to the melt composition of interest. The
1255 strong agreement between our entrapment depths and models of magma storage
1256 inferred from geophysical datasets at Kīlauea shows that melt inclusion records are

1257 a powerful tool to accurately constrain the location of magma storage reservoirs
1258 supplying volcanic eruptions.

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1273 **Data Availability** The melt inclusion and glass compositions presented in
1274 this paper are provided as an excel spreadsheet. This data has been uploaded to
1275 the Cambridge University Repository <https://doi.org/10.17863/CAM.60202>, and
1276 is also available on Github <https://github.com/PennyWieser/G3-2018-MI>. This
1277 spreadsheet also contains the results of the bubble growth models shown in Fig. 9.

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