

This manuscript has been submitted for publication in AMERICAN MINERALOGIST. Please note that, despite having undergone peer-review (this is the first revised version) the manuscript has yet to be formally accepted for publication. Because of that, the final version of the paper may have slightly different content. If accepted, the final version of this manuscript will be available via the "Peer-reviewed Publication DOI" that appears on the right-hand side of this webpage. Please feel free to contact me if you have questions, if you want to use the script I have used for the calculations presented here, and with any feedback you may want to give.

Title: Interphase misorientation as a tool to study metamorphic reactions and crystallization in geological materials
Manuscript Number: 7902R
Authors: Luiz Morales, ETH Zürich

3
4 **Interphase misorientation as a tool to study metamorphic reactions and crystallization in geological**
5 **materials**

6
7 Luiz F. G. Morales ^{a,b}

8
9 ^a Scientific Center for Optical and Electron Microscopy (ScopeM), ETH Zürich, Otto-Stern-Weg 3, 8093,
10 Zürich.

11 ^b Geological Institute, Department of Earth Sciences, ETH Zürich, Sonneggstrasse 5, 8092 Zurich

12
13 Email: luiz.morales@scopem.ethz.ch

14
15 **Abstract**

16
17 Interphase boundaries are planar defects that separate two different minerals, which in general have
18 different compositions and/or crystalline structures and may play an important role as a pathway for fluids
19 in rocks and affect their physical properties. For the proper characterization of interphase boundaries, one
20 needs to define the misorientation between adjacent grains and the orientation of the grain boundary plane,
21 but the analysis performed here are only limited to the misorientation characterization and the trace of the
22 interphase boundary. Although the determination of possible orientation relationships between the two
23 adjacent phases is routinely performed by selected area electron diffraction in the transmission electron
24 microscope, this method lacks statistical representativeness. With the advent of techniques like electron
25 backscatter diffraction (EBSD), it is possible to calculate orientation relationships not only in single pairs
26 of crystals, but in full thin sections, and not limited to single phases, but also between different minerals.

27 The interphase misorientation is calculated from two orientations of two adjacent crystals of different
28 phases. A set of single misorientations is then used to calculate the misorientation distribution function
29 (MDF), from where it is possible to identify a maximum, and their crystallographic interpretation. If we
30 then know the misorientation and the unit cell parameters of the individual phases, the crystallographic
31 relationships between them can be described with the pairs of parallel crystallographic planes and the pairs
32 of crystallographic directions. We present examples of the use of interphase misorientation analysis on the
33 transformation of calcite-aragonite, olivine-antigorite, magnetite-hematite, and on the study of orientation
34 relationships between plagioclase-olivine-ilmenite in mid-ocean ridges gabbros (ODP Hole 735).

35

36 **Keywords**

37 Interphase misorientation; EBSD; orientation relationships; phase transformation

38

39 **INTRODUCTION**

40

41 Polycrystalline material such as rocks, metals and ceramics are made of different grains with different
42 crystal orientations. Where those grains meet each other, there is normally an interatomic mismatch
43 between the lattices of the two adjacent crystals, which are called grain or phase boundaries. Grain and
44 phase boundaries are a very important part of the microstructure of rocks because they are high diffusivity
45 paths for both fluids and solid-state diffusion. They play an important role on the rheology of rocks and
46 influence petrophysical properties, such as strength of rocks, resistance to cracking and permeability. As
47 grain and phase boundaries are relatively disordered structures, they can accommodate large amounts of
48 impurities, which may influence, for example, boundary migration

49

50 Grain boundaries are interfaces that separate grains of the same phase (e.g. quartz grain boundaries),
51 whereas phase boundaries separate grains of different compositions and/or structure (e.g. quartz-feldspar
52 or calcite-aragonite phase boundaries). The structure and energy of grain boundaries are defined by the

53 misorientation of the two neighboring grains and the orientation of the boundary plane itself. To fully
54 define a grain boundary, one must define 5 variables related to the grain boundary misorientation angle,
55 misorientation axes and the orientation of the interface itself. Standard classification of grain boundaries
56 distinguishes two end members, named tilt and twist boundaries. In tilt boundaries, the rotation axis to
57 bring the lattices of two neighboring crystals into overlap is parallel to the interface plane, whereas in
58 twist boundaries, the rotation axis is normal to the interface, and pretty much all grain boundaries lie in
59 between these two end-members. Grain boundaries are also classified according to the misorientation
60 angle. Adjacent grain boundaries with less than 10° misorientation between neighbor grains are said to be
61 low angle grain boundaries, while neighboring grain boundaries with higher degrees of misorientations re
62 classified as high angle grain boundaries. In deformed materials, low angle grain boundaries are
63 commonly associated with array of aligned dislocations along a certain plane (forming in this case a
64 “subgrain”), but it is important to note that low angle grain boundaries are very important in
65 biomineralization and have nothing to do with intracrystalline defects.

66
67 Phase boundaries on the other hand are classified according to the degree of mismatch between the lattices
68 of the two phases along the interface (e.g. Sutton and Baluffi, 1995; Porter and Easterling, 1992). In
69 coherent (or commensurate) phase boundaries (Fig. 1), the two neighbor crystals match perfectly along the
70 interface, and the interfacial plane has the same atomic configuration in both phases. An important aspect
71 of the coherency definition is that the crystallographic plane comprising the phase boundary does not need
72 to be the same in the different phases, but an orientation relationship between the two grains needs to be
73 satisfied. This can be described by a Burgers orientation relationship that specifies a pair of parallel
74 planes and directions, i.e., $(hkl)_{\text{phase A}} \parallel (hkl)_{\text{phase B}}$ and $[uvw]_{\text{phase A}} \parallel [uvw]_{\text{phase B}}$ for the general case of single
75 planes and directions. In order to keep the interface coherent, small differences in the lattice parameters
76 between the two phases can be accommodated by elastic strains on both sides separated by the interface.
77 Such an effect reduces the interface energy at the expense of increasing the internal energy of both phases
78 adjacent to the interface. As such, elastic straining in order to force a perfect coherency only occurs if the

79 interface is strong, if the mismatch is small and if the grain size is small. Although the structure at the
80 interface is perfect, the interfacial energy at the boundary is quite variable, but depends essentially on the
81 composition of the phases separated by the interface. In semi-coherent (or discommensurate) phase
82 boundaries, the elastic strain described previously becomes too large and the lattice mismatch is
83 accommodated by an array of misfit dislocations that do not create long-range strain fields. The total
84 energy of semi-coherent phase boundaries results from both compositional and structural components. In
85 incoherent (or incommensurate) phase boundaries, the different lattices of the two phases at the interface
86 prevents the good matching at the phase boundary, leading to a more disorder structure, similar to those
87 observed in high-angle grain boundaries. The surface energy of both grain and phase boundaries are
88 function of the misorientation angle, and it is known from the material science literature that during
89 deformation, grain boundaries in metals tend to rotate to minimize such energy, leading to the
90 development of grain boundaries with specific misorientation angles.

91
92 Grain boundaries on common minerals such as quartz, olivine and calcite are relatively well understood in
93 various scales (e.g. McLaren 1986; Faul and Fitz-Gerald 1999; Fliervoet et al., 1999; Wheeler et al. 2001;
94 Bestmann et al. 2004; Lloyd et al. 2004, Hiraga et al. 2004; Wenk et al. 2011, Menegon et al. 2011; Kilian
95 et al. 2011; Krühl et al. 2013 and others). In olivine, for example, special grain boundaries have even been
96 characterized in terms of grain boundary character distribution (Marquardt et al. 2015) and in terms of
97 grain boundary energies (Duyster and Stöckert, 2001). However, although "real" rocks are mostly
98 composed of two or more phases, our understanding of phase boundaries between different minerals is
99 very scarce and predominantly studied at the nanometric scale, with most observations performed in
100 transmission electron microscopes (TEM - e.g. Boudier et al. 2010; Plümper et al. 2011; Obata and Osawa
101 2011; Marti et al. 2018). Although this is a very powerful tool to investigate interphase boundaries
102 particularly due to its angular resolution (interphase orientations can be determined up to a precision of
103 0.1°), it has a very limited statistical representativeness, where an individual TEM lamella will show at
104 most a few interfaces. However, the TEM is not the only technique available to study orientation

105 relationships between different phases, and any technique involving local diffraction of an electron beam
106 by the crystalline structure of the phases separated by the interface will allow us to calculate the
107 orientation relationship. Here it is proposed to use orientation maps generated via electron backscatter
108 diffraction (EBSD) in a scanning electron microscope (SEM) to calculate the interphase misorientation
109 angle/axis and from these data to calculate potential orientation relationships. . It is important to clarify
110 here that the “orientation relationships” described here have a different meaning to that described in
111 crystallography, where there is partial/total coincidence between the lattices of the two materials separated
112 by an interface. Here I use the term in a more “loose” way, to describe that, in certain cases, two different
113 minerals may show some degree of parallelism along certain interfaces without necessarily sharing the
114 same lattice along the interface. Advantages of this method include (i) the higher statistical
115 representativeness of the data, (ii) the potential to calculate the misorientation angle-axis pair of any pair
116 of mineral phases, (iii) the potential to plot phase boundaries of different misorientation angles in the map,
117 and highlight those which have special misorientation angles/axis and even possible orientation
118 relationships and (iv) to use EBSD results and subsequent calculations to tackle more detailed studies of
119 specific interphase boundaries on the TEM scale, preparing TEM lamellae using target preparation with
120 focused ion beam methods for example. Here I present 4 examples of interphase misorientations and
121 interphase orientation relationships between olivine-antigorite from a serpentinite, calcite-aragonite from
122 high P-T experiments, magnetite-hematite from an undeformed iron ore, and plagioclase-ilmenite
123 relationships from a mid-ocean ridge gabbro. The results are presented in terms of interphase
124 misorientation angles, axes, and plotted directly in the EBSD maps. In order to facilitate the use of this
125 method, a MATLAB script to be used in the MTEX toolbox from Hielscher and Schaeben (2008) is
126 provided as supplementary material.

127

128 **INTERPHASE BOUNDARY MISORIENTATIONS**

129

130 Interphase boundary misorientations calculations are relatively new in Earth and Material Sciences. For
131 instance, McNamara et al. (2012) explored interphase misorientation relationships on the study of
132 crystallographic preferred orientation (CPO) development of barroisite due to mimetic growth at the
133 expense of glaucophane and omphacite. Later, Morales et al. (2018) use the interphase angle/axis pair on
134 the study of olivine-antigorite transformation relationships, where they reported two new orientation
135 relationships never described in the literature. We should briefly step back and look the concept of
136 intraphase misorientation, where a complete overview of this topic is given by Wheeler et al. (2001).

137
138 Crystal orientations \mathbf{g} as determined via EBSD and consistent with the definitions of MTEX can be
139 described as passive rotations \mathbf{r} to bring the coordinates from the crystal reference into coordinates in the
140 sample reference frame. Both reference frames are right-handed, the three axes are orthogonal to each
141 other, and both share a common origin for the sake of simplicity (note however that in some cases, the
142 crystal reference frames might not be orthogonal). The rotation angle is always positive if the rotation is
143 counterclockwise when viewed along the rotation axis towards the origin. Using the definitions of Krakow
144 et al. (2017), if we define the specimen reference frame as s and the crystal reference frame as e , the
145 rotation has to satisfy

$$r = \mathbf{G}.e \quad (1)$$

146
147
148
149 where $r=(x,y,z)$ in specimen coordinates and $e=(e_1,e_2,e_3)$, in crystal coordinates. This equation states that a
150 vector r can be transformed into a vector e by an operation \mathbf{G} , which is a matrix operation.

151
152 The misorientation angle in an EBSD map is the angle necessary to bring two adjacent objects of this map
153 (i.g. two neighbor pixels, or two neighbor grains) into overlap, whereas the misorientation axis is the axis
154 in which this rotation needs to occur to bring the lattices of the two objects into parallelism. These
155 misorientations \mathbf{M} are also passive rotations, but in this case between two crystals reference frames, where

156 the two crystal normally have two different orientations (g_1 & g_2). The misorientation M between these
157 two crystals transforms the crystal coordinates c_1 into crystal coordinates c_2 , as follows (e.g. Krakow et al.,
158 2017)

$$159$$
$$160 \quad M = g_2^{-1} \cdot g_1 \quad (2)$$

161
162 and

$$163$$
$$164 \quad M c_1 = g_2^{-1} g_1 c_1 = g_2^{-1} r = c_2 \quad (3)$$

165
166 When dealing with one single phase, the lattice parameters for the adjacent objects are the same, and so
167 the misorientation axis can be defined by a number of symmetrically equivalent axis/angle pair
168 combinations that are directly dependent on the symmetry of the studied phase. In the case of two
169 neighbor plagioclase (triclinic) grains, there is only one set of misorientation angle/axis capable to bring
170 the lattices of these two grains into overlap, while in the case of magnetite or garnet (cubic minerals),
171 there are 24 different possibilities (Mainprice et al. 1993; Lloyd et al. 1997; Wheeler et al. 2001). In the
172 case of olivine (orthorhombic) there will be less than 24 possibilities Due to this large number of
173 possibilities in non-triclinic phases, the current convention is to adopt the pair with the minimum
174 misorientation angle of the equivalent rotations (e.g. Morawiec et al., 1995; Wheeler et al. 2001).

175
176 On the other hand, when dealing with the misorientation between two phases that belong to different
177 crystallographic systems, the reference misorientation is not unique. Because of that, and due to the fact,
178 that in most cases we will deal with phases of different symmetries, the asymmetric domain for the
179 misorientation axis plots is not 'fundamental'. That implies that the shape and form of misorientation
180 angle/axis distribution depends on the choice of reference misorientation. If we consider the example of
181 olivine (orthorhombic) and antigorite (monoclinic), the fundamental region requires the entire hemisphere

182 due to the combination of these two different symmetries. In this case, we must define in relation to which
183 crystal reference frame we will plot the misorientation axes. In the case of phase transformations, it makes
184 sense to use the parent phase reference frame, but if one suspects that a different interphase orientation
185 relationship can be deduced from the interphase misorientation data, one has to examine the
186 misorientation axes plot using also the daughter phase reference frame.

187
188 In most of the studies dealing with misorientations, fundamental zones are based on Rodrigues-Frank
189 parameters (e.g. Morawiec, 1997), but in MTEX this construction is based on quaternion geometry (see
190 Krakow et al., 2017 for details). The definition of a particular fundamental zone depends on the alignment
191 of crystal axes and the order in which the symmetry operators are combined for the misorientation
192 calculations. These fundamental zones are calculated by selecting, in a family of symmetrically related
193 equivalent points, the ones with the smallest angle of rotation. If then multiple points have the same
194 distance from the origin, the choice is made based on the direction of the rotation axis.

195
196 Crystallographic point groups control the symmetry operations related to misorientations. If we have
197 misorientation M relating point groups S_1 and S_2 with crystal coordinate systems c_1 and c_2 , and using the
198 equation (3), we can write the following expression

199
200
$$M = s_2 M s_1, s_1 \in S_1, s_2 \in S_2 \quad (4)$$

201
202 Here we use the smallest unique rotation angle (disorientation angle) and the axis of rotation within the
203 inverse pole figure sector that correspond to the point group common to both symmetries, fulfilling the
204 relation $SC = S_1 \cap S_2$. The fundamental zones for all possible combinations of proper point group
205 symmetry operations are presented in Krakow et al. (2017), but for the sake of simplicity, all the
206 misorientation axes plots are presented in inverse pole figures resulting from those combinations.

207

208

209 **DESCRIPTION OF THE SAMPLES**

210

211 To test the potential of misorientation analysis on the study of metamorphic reactions and magmatic
212 processes, we have studied four different samples (one experimental and three natural). Three of these
213 samples are known examples of phase transformations or crystallographically controlled metamorphic
214 reactions and include (i) calcite-aragonite; (ii) olivine-antigorite and (iii) magnetite-hematite. In all those
215 cases, the neighbouring phases that share a common boundary do not belong to the same crystallographic
216 system. The fourth example is an oxide-rich gabbro coming from the Atlantis Bank (Indian Ocean), where
217 I have tested all the possible pairs of 6 different minerals that compose this rock (plagioclase, diopside,
218 olivine, ilmenite, magnetite and pargasite) and I found that plagioclase-olivine and plagioclase-ilmenite
219 have specific misorientation angle/axis pairs that suggests that both olivine and ilmenite crystallized using
220 specific plagioclase crystal planes as substrate..

221

222 **Calcite - Aragonite phase transformation produced experimentally**

223

224 The calcite-aragonite aggregated studied here (sample 010-SC-3-X) belong to the collection of
225 deformation experiments performed by Sebastian Cionoiu during his PhD thesis at ETH Zürich, whose
226 objective was to understand the effect of stress on mineral reactions (see thesis volume for sample
227 preparation details). The calcite-aragonite phase transformation observed in this sample occurred under
228 hydrostatic conditions (i.e. no axial load applied) under 1.6 GPa pressure and temperatures of 600°C for
229 about 12 hours. In the studied sample, aragonite comprises about 15 % of the total aggregate and occurs
230 predominantly wrapped around large calcite crystals, as if forming tails of recrystallized material (Fig. 4a).
231 Calcite is heavily twinned and has some undulose extinction, but no clear subgrains or recrystallized grains,
232 and no noticeable grain growth has been observed. Aragonite grain sizes vary from 3 μm to 12 μm and
233 there is no clear evidence of crystal plasticity in this phase such as undulose extinction or subgrain walls.

234

235 **Tremolite-chlorite-antigorite schist from Moses Rock (USA)**

236

237 The sample of tremolite-chlorite-antigorite schist studied here (MR-1) was collected from the Moses Rock
238 dike and was previously studied by Boudier et al. (2010) and Morales et al. (2013). The Moses rock dike
239 belongs to the Navajo Volcanic Field (NVF) that is exposed in the central part of the Colorado Plateau, in
240 the Four-Corners region, SW of USA. The Moses Rock dike belongs to kimberlitic and lamprophyric
241 breccias that contain a large variety of mantle and lower crustal xenoliths of a variety of compositions
242 (Smith, 1995, 2010). The mantle fragments, associated with the subduction of the Farallon slab, include
243 metaperidotites rich in hydrous phases, jadeite clinopyroxenites and eclogites, spinel websterite and spinel
244 lherzolites and were formed at depths between 50 and 150 km and temperatures ~900 °C according to
245 clinopyroxene thermometry (Hunter and Smith, 1981),

246

247 **Magnetite – hematite phase transformation**

248

249 The magnetite-hematite studied here was collected in the Corrego do Feijão mine in the western part of
250 the Quadrilátero Ferrífero, southeastern Brazil. This “iron quadrangle” lies within the Cauê formation of
251 the Itabira group, a metasedimentary sequence of Archean/Paleoproterozoic ages that lies on the southern
252 boundary of the São Francisco craton (Alkmin and Marshak, 1998). This sequence hosts large iron ore
253 deposits in the form of itabirites and polycrystalline hematite with variable contents of magnetite.

254 Deformation intensity and metamorphic conditions increases from east to west (Rosière et al., 2001), and
255 the iron ores present a progressive enrichment of hematite in relation to magnetite towards the east. The
256 studied sample is still relatively rich in magnetite and is composed approximately by 60 % magnetite and
257 40% hematite, with goethite occurring as alteration material.

258

259 **Plagioclase-olivine and plagioclase-ilmenite from oxide gabbro from the Atlantis Bank**

260
261 The Atlantis bank is an oceanic core complex that was exhumed by a large-scale detachment fault on the
262 Southwest Indian Ridge (Karson and Lawrance 1997, Kelemen et al. 2007). Deformation in these rocks is
263 localized along hypersolidus and HT shear zones, later overprinted by brittle faults of different scales, as
264 detailed described by Miranda et al. 2010 and Allard et al. (2021). Here we studied oxide-rich gabbros that
265 occur with higher frequency toward the top of the borehole and are interpreted to have intruded the more
266 primitive gabbros at a later stage. Composition of the studied samples varies considerably from sample to
267 sample, between clinopyroxene and plagioclase-rich gabbros, but the content of oxides (mostly magnetite
268 and ilmenite) is between 5-10%. The studied sample have well-developed foliation, but lineation is not
269 well marked. In general, plagioclase occurs as equigranular aggregates, whereas clinopyroxene occurs
270 predominantly as porphyroclasts. Olivine occurs as equigranular, idiomorphic grains within the gabbros,
271 but also as "pods" rich in olivine in pressure shadows of clinopyroxenes. Ilmenite and magnetite occur
272 along bands parallel or oblique to the foliation.

273

274 **METHODS**

275

276 The crystallographic preferred orientations of the studied samples were determined by electron backscatter
277 diffraction technique (EBSD) in a scanning electron microscope. All the samples were measured in a FEI
278 Quanta 200F with EDAX Hikari EBSD camera operating at the Scientific Centre for Optical and Electron
279 Microscopy (ScopeM) of ETH Zürich. All the samples were mechanically polished to a 0.25 μm diamond
280 suspension and chemically-mechanically polished with an alkaline solution of colloidal silica (0.025 μm)
281 for 3-10 minutes on a neoprene polishing cloth. All the EBSD maps were acquired using an accelerating
282 voltage of 20 kV, beam current of 8 nA, working distance of 17 mm, and variable stepsizes, from 0.5 to 1
283 μm , depending on the scale of the map. With the exception of the olivine-antigorite sample, all the EBSD
284 maps used 4 x 4 binning of the patterns. The antigorite was indexed using the structure determination of
285 Capitani and Mellini (2006), using an a-axis of 35 \AA , typical of high-temperature antigorite. As antigorite

286 generates poor patterns in the above standard EBSD mapping conditions, a binning of 2 x 2 combined
287 with a pixel Hough binning of 160 and about 120 reflectors were used for the correct indexation. Post-
288 acquisition processing in the EDAX-OIM 8 software included the standardization of the confidence index
289 (CI) using a minimum grain tolerance angle of 10°, and minimum 10 indexed pixels per grain, followed
290 by a CI correlation between neighbor points, where pixels with low CI (<0.1) are reassigned to the
291 orientation and the CI of the neighbor data point with highest CI in the individual grains. All the EBSD
292 maps and calculations and plots were carried out with the MTEX toolbox for MATLAB (Hielscher and
293 Schaeben, 2008). The orientation distribution functions (ODFs) were calculated using the complete
294 datasets with the de La Vallee Poussin kernel with a half-width of 10°, which is equivalent of a bandwidth
295 of 28 in the spherical harmonic coefficients. All the data is plotted in equal-area, upper hemisphere pole
296 figures, with a resolution of 5°. Interphase misorientation pairs between phase A and phase B were
297 calculated for the entire maps using a common misorientation threshold of 10°, and therefore low angle
298 interphase misorientations, if present, are not studied here. Detected grains with less than 10 pixels were
299 also not considered in the calculations.

300

301 **RESULTS**

302

303 **Calcite – Aragonite**

304

305 In the studied sample, calcite is the dominant phase (~75% sample) while aragonite is secondary (making
306 up the other ~25% - Fig. 2a). Calcite grain size varies between 25 µm -120 µm, some of the large grains
307 are heavily twinned, and most of the large calcite grains have large internal misorientations, up to angles
308 of 10° (Fig. 2b). The aragonite occurs predominantly wrapped around the calcite large grains along
309 discontinuous, anastomosing bands (Fig. 2a, c), except for the small aggregate of coarser grain aragonite
310 that occurs in the bottom of the Fig. 2a. It has grain sizes varying from 4 µm to 12 µm, and from the
311 orientation maps, it seems that the crystallographic preferred orientation of both calcite and aragonite is

312 very weak (Fig. 2a, d). This is confirmed on the pole figures (Fig. 3), where both minerals show a weak
313 CPO, with calcite being slightly stronger than aragonite (4.3 multiples of uniform distribution of calcite
314 against 2.6 from aragonite). The pole figures are also used here to look for possible similarities between
315 different poles of the studied mineral pairs, which may indicate that the development of a CPO of phase 2
316 is crystallographically controlled by phase 1, but do not fully prove that both phases have any orientation
317 relationship along the interfaces. A careful comparison between the pole figures of both phases shows
318 similarities. For instance, the [100] axes of aragonite are distributed in a broad girdle that is parallel to a
319 similar girdle of poles to $(2 \bar{1} \bar{1} 0)$ of calcite, whereas the distribution of [001] of aragonite is similar to
320 the distribution of $(0 1 \bar{1} 8)$ of calcite. In addition, the strongest concentration of [010] and [110] in
321 aragonite are subparallel to the maximum concentrations of (0001) and $(1 0 \bar{1} 4)$ of calcite, respectively.
322

323 The uniform misorientation angle distribution (i.e. the misorientation distribution expected in the case of a
324 uniform, “random” ODF) shows a progressive increase in frequency to a maximum of 90° (Fig. 4a) and a
325 sudden drop to a maximum misorientation angle $\sim 92^\circ$ (for the trigonal-orthorhombic relationship). The
326 uniform misorientation axis distribution (Fig. 4b) has a weak maximum parallel to the poles of $(\bar{6} 2 4 5)$
327 and symmetrically related, which are about 10° from the poles to $(1 \bar{2} 1 0)$, with a maximum of uniform
328 distribution (MUD) around 2.5. The misorientation angle distribution for correlated (neighbor) interphase
329 boundaries tend to follow the distribution expected in the case of a uniform distribution (Fig. 4c). In detail
330 however, the red bars that represent the misorientation angle between neighbor phase boundaries show
331 that the peaks between 25° to 50° occur in higher frequencies than the ones expected in the case of non-
332 neighbor pairs (red line) or uniform distribution (orange line). This suggests that phase boundaries within
333 this angle range show some sort of physical relationship that the uncorrelated distribution does not
334 present, which might indicate a special orientation relationship between calcite and aragonite.
335

336 For the orientation relationship $(11-20)\text{calcite} \parallel (100)\text{aragonite}$ and $[0001]\text{calcite} \parallel [110]\text{aragonite}$, the
337 misorientation axis is 54.2° , and the misorientation axis is parallel to $(2 \bar{1} \bar{1} 0)$ in the calcite reference

338 frame and to (100) in the aragonite reference frame. If we now plot the misorientation axes for the whole
339 range of misorientation angles for the calcite-aragonite phase boundaries, there is a vast distribution
340 around the primitive circle of the stereonet and a weak maximum parallel to the pole of $(2 \bar{1} \bar{1} 0)$, in
341 agreement with a dominant topotactic relationship transformation between calcite and aragonite (Fig. 4d).
342 Note that due to the symmetry, only two out of four symmetrically equivalent pole to planes of aragonite
343 are parallel to two out of three pole to planes in calcite, which are $(1 \bar{1} 0 8)_{\text{cal}} \parallel (\bar{1} 1 \bar{3})_{\text{ar}}$ and $(\bar{1} 0 1 \bar{8})_{\text{cal}}$
344 $\parallel (\bar{1} \bar{1} 3)_{\text{ar}}$. The interface trace map in the Figure 4e shows a broad range of distribution of
345 misorientation angles for the interface traces, but about 41% of the phase boundaries between calcite and
346 aragonite have a range of misorientations between 40-70°.

347

348 **Olivine – Antigorite**

349

350 The sample MR-1 contains about 60% of hydrous phases (chlorite, tremolite and antigorite), and about
351 40% of the lithospheric mantle olivine framework is still preserved (e.g. Fig. 5a), and its microstructure is
352 detailed described in Morales et al. (2013). The thin section studied here was cut normal to the antigorite
353 foliation and parallel to the dominant lineation. The phase boundaries between olivine and the hydrous
354 phases in this sample are predominantly straight and sharp, while the contacts between hydrous phases
355 might be blurred and difficult to identify visually.

356

357 Olivine and antigorite bulk crystallographic preferred orientation is given in the Figure 6. Due to the
358 relatively coarse grains and the dismembering due to the antigorite, the olivine CPO is relatively complex
359 but not very strong (maxima of 3.9 MUD), with [100] orientated around 20° from the antigorite lineation
360 and [010] and [001] distributed along incomplete girdles, normal and parallel to the XZ plane
361 respectively.. Antigorite on the other hand has a strong texture (8.5 MUD for (001) and has a typical (001)
362 fiber-texture, with the poles of (001) strongly orientated at a small angle to the pole of the foliation and the
363 poles of (100) and (010) distributed along girdles parallel to the reference foliation. Although it is not

364 possible to establish clear orientation relationships between olivine and antigorite primary maxima from
365 these figures, in detail the secondary maxima show some similarities. For instance, the secondary maxima
366 of olivine [010] and [001] close to Z is parallel to the maxima of poles to (010) of antigorite.

367
368 The uniform misorientation angle distribution for olivine-antigorite (Fig. 7a) shows a progressive increase
369 up to a maximum peak at 90° , when it drops down substantially, reaching a maximum misorientation
370 angle $\sim 120^\circ$ for the combination of orthorhombic-monoclinic. The uniform misorientation axis
371 distribution (Fig. 7b) shows a maximum peak on $\langle 947 \rangle$, with a MUD 1.6. The correlated misorientation
372 angle distribution shows some peaks that are much higher than the expected distribution of non-correlated
373 phase boundaries or the misorientation expected in the case of uniform ODF. In particular, the
374 misorientation angles between 70° - 100° are considerably higher. If the misorientation angle is now limited
375 to this angular range, the resulting misorientation axis plot show a maximum parallel to the poles (094),
376 which lies about 30° from [010]. This is almost equivalent to the $[5 \bar{9} 4]$ axis described in Morales et al.
377 (2018). The interphase boundary map presented in the Figure 7e shows that about 48% of the interfaces
378 between olivine and antigorite have a range of misorientations between 80° - 100° , from which 27% have a
379 10° misorientation range of 90° - 100° .

380

381 **Magnetite – hematite**

382

383 In the studied sample, magnetite comprises $\sim 60\%$ and hematite $\sim 40\%$ and present a massive structure.
384 Hematite nevertheless seem to occur in “patches” on the EBSD map (Fig. 8a). Magnetite shows almost an
385 uniform distribution of its main crystal directions [100], [110] and [111], with only [100] showing some
386 degree of orientation subparallel to Z in the sample reference frame (Fig. 9a). Hematite CPO on the other
387 hand is better developed (Fig. 9b) and shows a maximum concentration of poles to (0001) parallel to Y
388 and a more complex distribution of poles of $(1 0 \bar{1} 0)$ and $(1 0 \bar{1} 1)$. At first it seems that is no clear
389 relationships between the pole figures of magnetite and hematite. However, as magnetite is cubic, a

390 rotation of 90° around x in the pole figure is allowed and would bring this maximum into parallelism with
391 hematite (0001) pole figure.

392
393 The uniform misorientation angle distribution of magnetite-hematite shows a peak around 42° (Fig. 10 a),
394 followed by a fast drop to the maximum possible interphase misorientation angle between these two
395 phases, around 60°. The uniform misorientation axis distribution (Fig. 10b) shows a maximum peak on
396 <231>, with a MUD 3.5. The interphase misorientation angle distribution for neighbor crystals follows the
397 same trend for the uncorrelated grains, or the distribution expected in the case of uniform distribution (Fig.
398 10c). Nevertheless, it shows frequencies around 5% higher than the frequency expected in the case of
399 uniform distribution. The correlated interphase misorientation axis (Fig. 10d) show a similar distribution
400 as the uniform one, but only one stronger concentration parallel to [231] with MUD of ~8, with the other
401 symmetrically equivalents ($[\bar{3} 2 1]$, $[\bar{2} \bar{3} 1]$, $[3 \bar{2} 1]$) showing weaker concentrations. The interphase
402 boundary map (Fig. 10e) shows that 73% of the interphase boundaries have misorientations between 30-
403 50°, and 41% lie in a range of 40-50°.

404

405 **Plagioclase – Olivine – Ilmenite**

406

407 In the studied sample of gabbro from the Atlantis bank, plagioclase is the dominant phase, and together
408 with clinopyroxene, are the primary phases to crystallize in this sample. Plagioclase are predominantly
409 subhedral, grain size varies from 100 to 600 μm (Fig. 11a) and a considerable number of grains still
410 preserve magmatic twinning. The smaller plagioclase grains are interpreted as plagioclase subgrain
411 rotation recrystallization (Allard et al., 2021). Plagioclase CPO is relatively weak, as seen in the orientation
412 map with a large variety of colors (Fig. 11c). Olivine on the other hand occurs predominantly as anhedral
413 grains that are much smaller in grain sizes (50-100 μm). In this sample, olivine seems to be (at least in
414 part) a late phase, as it commonly occurs along interstitial spaces between plagioclase and diopside
415 grain/phase boundaries, and along pressure shadows in clinopyroxene porphyroclasts. Ilmenite grains are

416 mostly anhedral in shape, grain sizes are between 20-70 μm , and occur predominantly in contact with
417 plagioclase (Fig. 11a).

418
419 The pole figures for plagioclase, olivine and ilmenite are plotted in a sample reference frame where the
420 foliation is vertical N-S (so the pole of the foliation is at E in the pole figure), and the lineation also N-S,
421 but horizontal (Fig. 12). All the pole figures show a weak crystallographic preferred orientation.
422 Plagioclase (100) poles are distributed along a broad, asymmetrical girdle cross-cutting Y, while the poles
423 to the (010) are mainly concentrated at an angle of $\sim 15^\circ$ with the pole of foliation. Olivine [100] axes are
424 predominantly concentrated parallel to Y, [010] and [001] broadly distributed with weak maxima parallel
425 to the pole of the foliation (the former) and parallel to the lineation (the later). The poles to the (0001) of
426 ilmenite are preferentially aligned with Y, while the poles to the $(1\ 0\ \bar{1}\ 1)$ rhombs are at low angle with
427 the pole of the foliation Z and the poles to the $(1\ 0\ \bar{1}\ 0)$ prisms subparallel to the lineation, but showing
428 secondary, symmetrically related maxima every $\sim 60^\circ$. More important than the CPO strength in this case
429 though are similarities in the pole figures. For instance, the distribution of [100] and [010] of olivine
430 shows similarities with the poles to the (100) and (010) of plagioclase, while the poles to the (0001) and
431 $(10\text{-}10)$ of ilmenite shows similarities with the poles to the (100) of plagioclase, and the distribution of
432 poles to the $(10\text{-}11)$ ilmenite are comparable to the poles to the (010) planes of plagioclase (Fig. 12).

433
434 The misorientation angle distribution assuming a uniform distribution show, for the plagioclase-olivine, a
435 peak at misorientation angles $\sim 83^\circ$ and a sudden drop to a maximum misorientation of $\sim 119^\circ$ (Fig. 13a).
436 Due to the triclinic-orthorhombic combined symmetry (plagioclase-olivine), the misorientation axes have
437 to be presented in both upper and lower hemisphere, where the maximum misorientation angle expected in
438 the case of a uniform distribution is subparallel to the $\langle 253 \rangle$ of olivine, with a MUD of 1.7 (Fig. 13b). In
439 the case of plagioclase-ilmenite, the maximum misorientation angle expected in the uniform case is $\sim 90^\circ$,
440 with frequencies dropping progressively to a maximum angle of 180° (Fig. 13c). The maximum
441 misorientation angle in the combination of plagioclase-ilmenite symmetries seems to be distributed all

442 over along the primitive circle of the stereonet, and only statistical calculations allows to pinpoint a
443 maximum parallel to the poles to the $\{3\ 2\ \bar{1}\ 0\}$ planes of ilmenite and subparallel to $[100]$ of plagioclase,
444 with a MUD of 3 (Fig. 13d).

445
446 The distribution of interphase misorientation angles for neighbor (correlated) plagioclase-olivine grains
447 show abnormal frequencies for misorientation angles between 65 and 90° ($>2\%$ of the
448 uncorrelated/uniform distribution – Fig. 14a). If we limit the range of misorientation angles to the one
449 above and plot the misorientation axes, the orientation is not very strong (2.5 MUD), but it shows a
450 maxima parallel to the pole to the (100) of plagioclase, 2 other maxima with similar MUD, and a
451 secondary maxima parallel to $[010]$ of plagioclase (parallel to the pole to the (010) plane – Fig. 14b). The
452 misorientation angle distribution for the pair plagioclase-ilmenite shows a variety of higher frequency
453 peaks intercalated with lower frequency peaks (Fig. 14c), when compared with the uncorrelated/uniform
454 distributions. Also, the studied sample does not show misorientation peaks larger than 165° . For
455 simplicity, we focus on the misorientation peaks between 85 - 130° . When limited to this range, most of the
456 misorientation axes are parallel to the $[100]$ of plagioclase, with a MUD of 9. The interphase boundary
457 map (Fig. 15a) shows that about 70% of the boundaries between plagioclase and olivine have a narrow
458 range of misorientations between 60 and 100° , while in the case of plagioclase-ilmenite the range is more
459 variable (Fig. 15b), and about 50% of the interphase boundaries between these two phases have a range of
460 misorientations between 60 and 120° .

461

462 **IMPLICATIONS**

463

464 **Easier assessment of interphase boundary orientation relationships**

465

466 Differently from grain boundaries separating grains of same composition and structure, interphase
467 boundaries are interfaces that normally separate grains with different structures and / or compositions.

468 Because of this complexity, only individual interfaces are normally studied at the time, in most cases via
469 selected area diffraction (SAD) in a transmission electron microscope (TEM). SAD-TEM provides very
470 detailed information in terms of angular and spatial resolution of orientation relationships between the two
471 phases separated by the interface (atomic scale resolution with angular resolutions $<0.1^\circ$) at the expenses
472 of statistical representation. Electron-transparent sample preparation for TEM is not trivial and for certain
473 geological materials can be challenging. As an example, to prepare standard in-situ TEM lamellae via
474 focused ion beam techniques, one needs around two hours. However, one has to consider that the
475 dimension of such a sample is around $15 \times 10 \mu\text{m}$ (length x deep), and $\sim 100 \text{ nm}$ thick, so in the case of a
476 very fine-grained material, one may observe a few interphase boundaries, but in the case of a more coarse-
477 grained rock, one might be able to see only one interface. In addition, TEM operation and correct
478 indexing of diffraction patterns from SAD (particularly for low-symmetry phases) is more complex and
479 time consuming than SEM operation and EBSD mapping and require extensive training to be done
480 effectively. As demonstrated here with EBSD maps, one can see a statistically representative number of
481 interfaces on samples that have different phases and different grain sizes. As we know the orientation of
482 the grains separated by the interface, we can then calculate the misorientation angle/axis between these
483 two grains, which can be used to infer possible orientation relationships between the two phases. As an
484 example, we know that $(11\bar{2}0)_{\text{cal}} \parallel (100)_{\text{ara}}$ and $[0001]_{\text{cal}} \parallel [011]_{\text{ara}}$. If we know now the orientation of the
485 daughter phase (in this case, aragonite), we can compute the orientation of the parent phase following the
486 MTEX script presented in the supplementary material.

487
488 The intention here is not to say that EBSD-derived interphase misorientation is a substitute for TEM
489 analysis. In fact, the idea is to use EBSD mapping to precisely locate the orientation and misorientation of
490 specific interphase boundaries and use this data to select the interfaces one wants to study in more detail in
491 the TEM. The interphase misorientation analysis does not give us the degree of coherency between the
492 phases separated by the interface, nor any atomic resolution along the interface separating two minerals.
493 On the other hand, once you have an EBSD map, you can test possible misorientation relationships

494 between any phase present in the map. That was the case of the gabbro sample whose results are presented
495 in the Figs. 11-15, we have tested all the possible pairs of phases within the map and found that
496 plagioclase-olivine and plagioclase-ilmenite have an orientation relationship that can further be analyzed
497 in the TEM.

498

499 **Interphase orientation relationships**

500

501 Phase transformations in geological materials leading to interphase orientation relationships between
502 parent→daughter phases occur predominantly by two major mechanisms: martensitic and
503 nucleation/growth mechanisms. From the four different systems studied in this paper, two belong to the
504 nucleation/growth mechanism (olivine→antigorite and magnetite→hematite), one can be either
505 interpreted as a martensitic or nucleation/growth (calcite→aragonite), and are discussed in a bit more
506 detailed below. In fact, martensitic transformations have been referred in the material sciences literature as
507 a type of nucleation/growth type of transformation that occurs in much shorter time scales (e.g. Olson and
508 Morris-Cohen, 1972; Guimarães and Rios, 2008). Plagioclase→olivine and plagioclase→ilmenite on the
509 other hand do not represent phase transformation of any sort, but because the misorientation angle/axis
510 pair show what looks like special relationships, these results are discussed in terms of orientated growth
511 below.

512

513 Nucleation and growth mechanisms are normally associated with diffusion and therefore tend to be
514 thermally activated or enhanced. The growth of one phase into another also depends on the interfacial free
515 energy, the Gibbs free energy of the reaction and the strain free energy (e.g. Mainprice et al., 1990, Porter
516 and Easterling, 1992). Phase transformation from parent phase to daughter phase may occur by only
517 changing the structure from the parent phase (e.g. calcite-aragonite) or by changing both the composition
518 structure between the two phases (e.g. magnetite-hematite or olivine-antigorite). This process starts with
519 the homogenous (or heterogeneous) nucleation of the daughter phase. The homogeneous nucleation is

520 controlled by the energy to form the nucleus of the daughter phase, and because of that, it requires
521 activation energies much higher than those required for heterogeneous nucleation (e.g. Porter and
522 Easterling, 1992; Sunagawa 1994). Heterogeneous nucleation refers to the nucleation of the daughter
523 phase along the parent phase substrate and requires much less activation energy because the interface
524 between the old and new phases reduces considerably the surface energy value. Heterogeneous nucleation
525 normally occurs along intracrystalline defects (dislocations, grain/interphase boundaries) or due to the
526 presence of inclusions, where the initial nuclei of the daughter phases growth normally by diffusion
527 processes.

528
529 Classical examples of this mechanism include the hydration of olivine and formation of antigorite and the
530 transformation of magnetite into hematite. Plümper et al. (2012) showed that the initial exchange of Fe-
531 Mg along (100) dislocation walls in olivine lead to the first steps of topotactic formation of antigorite
532 along this plane in olivine, something also observed in Boudier et al. 2010. As demonstrated in the Fig. 7c,
533 the higher frequency interphase misorientation angle are between 80-100°, with a dominant misorientation
534 axis parallel to the poles to the (094), which departs 15° from the interphase misorientation axes parallel to
535 the poles to the (5-94) determined in Morales et al. (2018). Considering that the sample studied here is less
536 deformed and has the olivine grain network much better preserved than the antigorite schist studied in the
537 aforementioned study, we conclude that the interphase misorientation relationships are compatible with
538 the type 4 phase transformation of $(010)_{ol} \parallel (210)_{atg}$ and $[100]_{ol} \parallel [001]_{atg}$, determined by Morales et al.
539 (2018) using the interphase misorientation angle/axes.

540
541 Calcite-aragonite transition and the orientation relationships between the two phases has been extensively
542 studied in the past (e.g. Carlson and Rosenfeld, 1981; McTigue and Wenk, 1985; Gillet et al., 1987) and
543 two main mechanisms for the phase transformation has been proposed. While the experimental results
544 suggest that the heterogeneous nucleation and topotactic growth is the dominant mechanism under a
545 variety of conditions, in agreement with the model from Carlson and Rosenfeld (1981), Gillet and Madon

546 (1982) proposed a martensitic mechanism for the calcite-aragonite transition. In this model, stacking faults
547 dragging by partial dislocation movement is responsible for the phase transformation. In contrast to
548 (normally) slow nucleation and growth mechanisms, martensitic phase transition can be a very fast
549 mechanism, normally producing a metastable phase. Martensitic transformation occurs by the progressive
550 and systematic shearing of the lattice of the parent phase in a way that the distance in which any atom
551 moves is less than one atomic spacing, which implies that the lattice is distorted, but the atoms retain the
552 same neighbors. Because of that, martensitic transformation only leads to change in the structure of the
553 phases, and not in composition, in order to accommodate the shearing described above.

554
555 Gillet et al. (1987) described the dominant phase transition orientation relationship for calcite→aragonite
556 as $(11\bar{2}0)_{\text{cal}} \parallel (100)_{\text{ara}}$ and $[0001]_{\text{cal}} \parallel [011]_{\text{ara}}$. McTigue and Wenk (1985) on the other hand described
557 $(11\bar{2}0)_{\text{cal}} \parallel (010)_{\text{ara}}$, $(10\bar{1}4)_{\text{cal}} \parallel (011)_{\text{ara}}$ and $[r2:f1]_{\text{cal}} \parallel [001]_{\text{ara}}$. The predominant misorientation angle of
558 54.2° with a dominant axis parallel $(2\bar{1}\bar{1}0)$ of calcite reference frame of the studied sample indicate that
559 in general, the transformation orientation relationship in this case is the one described by Gillet et al.
560 (1987). However, when the misorientation angle is limited to a narrow range between 25° and 50° , the
561 misorientation axis change to parallel to the pole to the $(01\bar{1}8)$ which is the twin plane of e-twins in
562 calcite. Although we don't have access to the orientation of the interface along twin planes, it seems from
563 the EBSD map that the red interphase boundary traces are visually subparallel to the twin planes, for
564 example, of the grains on top-left of the map (Fig. 4e). This observation is in agreement with the Fig. 2a-b,
565 where a number of aragonite grains occur wrapped around a heavily twinned calcite grain, and those
566 grains in contact with that specific calcite grain are the ones that produce the misorientation axis
567 subparallel to $(01\bar{1}8)$.

568
569 Transformation from magnetite to hematite is another example of nucleation and growth mechanism,
570 where hematite grows topotactically on magnetite, following the main orientation relationship $(111)_{\text{mag}} \parallel$
571 $(001)_{\text{hem}}$ and $(-101)_{\text{mag}} \parallel (100)_{\text{hem}}$ (Heizmann et al., 1981, Lagoeiro, 1998, Barbosa and Lagoeiro, 2010).

572 These authors described other orientation relationships, and the transformation magnetite-hematite-
573 magnetite is, in their case, reversible and always topotactically controlled. The misorientation angle for the
574 topotactic orientation described above is $\sim 56^\circ$, and the misorientation axis is parallel to $\langle 793 \rangle$ of
575 magnetite, which is about 4° from the direction $\langle 231 \rangle$ expected in the case of a uniform distribution (Fig.
576 10b) As we see in the histogram from Figure 10c, all the misorientation angles above 40° occur in higher
577 frequencies than the frequencies expected in the case of a uniform distribution, with the highest bin
578 showing misorientations between $45\text{-}50^\circ$. However, the misorientation axis is very close to the one
579 expected in the case of a random orientation of both magnetite and hematite. As demonstrated in the pole
580 figures (Fig. 9), hematite CPO is stronger than magnetite, and no clear relationship between the pole
581 figures can be made, unless we consider a rotation of 90° of magnetite around x, which would bring the
582 maximum now at the S to parallelism with (0001) hematite. That would imply however that, in this
583 specific sample, we have a $(100)_{\text{mag}} \parallel (001)_{\text{hem}}$ sort of relationship. Although this may represent a new
584 possible topotactic orientation relationship, one must consider that the studied EBSD map has a very large
585 number of grains and already around 40% of hematite. At these conditions, it is difficult to imagine that
586 every single crystal of hematite resulted from the transformation of magnetite on one of the symmetrically
587 related $(111)_{\text{mag}}$ planes. I believe that the transformation magnetite-hematite is indeed initiated along
588 $(111)_{\text{mag}}$ as clearly shown by Barbosa and Lagoeiro (2010), but once the reaction “kicks in”, hematite
589 started to be dissolved along other low index interfaces, and the initial orientation relationship is
590 progressively lost. Deformation localization in iron ores like the studied sample normally leads to the
591 oxidation of magnetite and consequent transformation to hematite, and deformation is predominantly
592 accommodated by hematite. In this case, hematite can develop stronger CPOs than magnetite (e.g.
593 Morales et al., 2008) and that may potentially affect the interphase misorientation distribution angles. It
594 seems that in all studied cases here, transformation from parent to daughter phase occurs preferentially
595 (but certainly not exclusively) following certain orientation relationships.

596

597

598 **Exploring unknown orientation relationships**

599

600 So far, we have explored the interphase misorientation analysis in examples where we know that the
601 orientation of a daughter phase is partially/completely controlled by the orientation of the parent phase.
602 However, if one has EBSD orientation maps of “normal” rocks (polymineralic), one can explore potential
603 interphase misorientation angle/axis relationships and find possible “orientation relationships” between
604 the phases presented in this map. Naturally we shall not expect that all the phases will have any sort of
605 relationship with the other phases in the aggregate, but let’s take the example of the oxide-rich gabbro
606 studied here. The EBSD map of this sample has 6 different minerals (plagioclase, clinopyroxene, olivine,
607 pargasite, ilmenite and magnetite – Fig. 11 only shows the three where orientation relationships were
608 found). Tests conducted in all possible pairs of minerals within this list resulted in 15 different interphase
609 misorientation angle/axis pairs. From these 15 pairs, only 2 (plagioclase-olivine and plagioclase-ilmenite)
610 have shown some sort of orientation relationships between the two phases. In both cases, the orientation
611 relationships have nothing to do with nucleation/growth nor martensitic transformations, as the minerals
612 have different compositions and structures. The orientation relationships between plagioclase-olivine and
613 plagioclase-ilmenite seem in the studied sample seem to be related to some sort of epitaxial growth of
614 olivine and ilmenite on plagioclase, possibly controlled by the surface energy of plagioclase and wetting
615 surfaces on this phase.

616

617 In the studied sample, plagioclase (together with diopside) are the primary phases and crystallize first in
618 their magmatic history, while both olivine and (later) ilmenite seem to be residual melts that crystallized
619 later in the gabbro history. The surface energy along a solid-liquid interface is mainly controlled by the
620 atomic structure on the interface, which in turn is dependent to its crystal orientation (e.g. Laporte and
621 Watson, 1995). In addition, the crystallization of second phases using a host crystal as “substrate” depends
622 on the crystal structure between the two phases, and any possible relation orientation between the host and
623 the precipitate depends on the fit of lattice space between these two phases (e.g. Sutton and Balluffi,

1994). In the case of the studied gabbros, the interphase misorientation axis for the pair plagioclase-ilmenite is subparallel to plagioclase [100]. If we assume that the interphase misorientation axis is contained along the interface, in a similar fashion as tilt grain boundaries (e.g. Lloyd. 2004), and if we assume that the interface has a high tilt angle, this interface of plagioclase can anything between (010) or (001). Although this is the topic of another publication in preparation, we know the distance between the oxygen atoms connecting the SiO₄ – AlO₄ tetrahedra in labradorite is between 4.23 and 4.26 Å, based on the determinations of Wenk et al (1980). This distance is very close to the distance between O-Fe-O along the long axis of FeO₆ octahedra of ilmenite, which is 4.224Å, determined by Wechsler and Prewitt (1984) and visualized with CrystalMaker ®. Although plagioclase is either monoclinic or triclinic and ilmenite is trigonal, is it not uncommon to find higher symmetry in certain plagioclase crystal direction. For example, plagioclase is surprisingly symmetrical along [001] as noticed by Wenk et al. (2011) and Ageeva et al. (2020), Along this direction, there are six-component tetrahedra “rings” that can easily accommodate the FeO₆ octahedra from ilmenite. If ilmenite is then the late phase, it can crystallize using the pre-existent plagioclase as substrate, where the FeO₆ octahedra is e accommodated along plagioclase [001] axis by sharing some of the oxygen atoms in the crystalline structure of plagioclase (substrate) with those with similar distances (precipitate). Although more detail is needed and it is out of scopus of this paper, it is possible that ilmenite (0001) plane and [1 0 $\bar{1}$ 0] direction of ilmenite are parallel to (120)/ ($\bar{1}$ 20) planes and [001] direction of plagioclase, respectively, previously reported in Wenk et al., (2011), or to one of the parallelism relationships described in Ageeva et al., (2016).

As a final remark, the calculation of interphase misorientations from EBSD maps is a potential tool to, for example (i) study topotactic relationships between minerals, (ii) explore potential “orientation relationships” of different phases in a rock, (iii) to plot interphase boundaries with variable misorientation in a map, and highlight those which have special misorientation angles/axis and even possible orientation relationships and (iv) to use those maps to precisely pinpoint specific interfaces that can be further analysed with TEM, using target preparation with focused ion beam methods for example. Steps for the

650 calculations include (i) calculate the interphase misorientation angle distribution assuming uniform ODFs
651 for both phases, (ii) calculate the actual correlated and uncorrelated misorientation angle distribution for
652 the two studied phases, (iii) compare the correlated misorientation with the uncorrelated/uniform
653 distributions in a histogram and find correlated misorientation peaks that are more frequent than those
654 calculated for a uniform/uncorrelated distribution. After that, one can (iv) limit the misorientation angle to
655 those higher frequent peaks of any range and then plot those in inverse pole figures of combined crystal
656 symmetries and (v) plot the different range of interphase misorientations directly on EBSD maps. If for
657 example topotactic relationships exist between the two phases, one can calculate what is the pair
658 misorientation angle/pair for such a relation, and look into an EBSD map for the presence of these
659 relationships, based on a misorientation angle range that includes the specific misorientation angle for the
660 specific topotactic relationship.

661

662 **ACKNOWLEDGEMENTS**

663

664 I would like to thanks Sebastian Cionoiu and Lucie Tajcmanova for the calcite-aragonite sample, and
665 Jessica Till for the gabbro sample. I am also grateful for the detailed reviews by Luca Menegon and the
666 anonymous reviewer, which improved considerably the quality of the paper, and by the editorial handling
667 of Dr. Sergio Speziale.

668

669 **REFERENCES CITED**

670

671 Ageeva O, Habler G, Topa D, Waitz T, Li C, Pertsev A, Zhilicheva O, Abart R (2016) Plagioclase hosted
672 Fe-Ti-oxide micro-inclusions in an oceanic gabbro-plagiogranite association from the Mid Atlantic Ridge
673 at 13° 34' N. American Journal of Science 316:85–109

674

675 Ageeva, O., Bian, G., Habler, G., Pertsev, A., Abart, R. (2020). Crystallographic and shape orientations
676 of magnetite micro-inclusions in plagioclase, *Contributions to Mineralogy and Petrology* 175: 95, 020)
677 175:95, doi.org/10.1007/s00410-020-01735-8.
678

679 Alkmim, F., Marshak, S., (1998) Transamazonian orogeny in the southern São Francisco craton region,
680 Minas Gerais, Brazil: evidence for Paleoproterozoic collision and collapse in the Quadrilátero Ferrífero.
681 *Precambrian Research* 90, 29–58.
682

683 Barbosa, P. F., Lagoeiro, L.E. (2010). Crystallographic texture of the magnetite-hematite transformation:
684 Evidence for topotactic relationships in natural samples from Quadrilátero Ferrífero, Brazil. *American*
685 *Mineralogist* 95, 118-125.
686

687 Bestmann, M.; Prior, D.J. (2003). Intragranular dynamic recrystallization in naturally deformed calcite
688 marble: diffusion accommodated grain boundary sliding as a result of subgrain rotation recrystallization.
689 *Journal of Structural Geology* 25, 1597-1613
690

691 Boudier, F., Baronnet, A., Mainprice, D. (2010). Serpentine mineral replacements of natural olivine and
692 their seismic implications: oceanic lizardite versus subduction-related antigorite. *Journal of Petrology* 51
693 (1–2), 495–512, <http://dx.doi.org/10.1093/petrology/egp049>.
694

695 Carlson, W.D., Rosenfeld, J.L. (1981) Optical determination of topotactic aragonite-calcite growth
696 kinetics: metamorphic implications. *Journal of Geology* 89, 615-638
697

698 Cionoiu, S. (2019). *Experimental and Numerical Predictions of Phase Transitions in Solids under*
699 *Differential Stress*. PhD Thesis, ETH Zürich. <https://doi.org/10.3929/ethz-b-000353445>
700

701 Duyster, J., Stöckhert, B. (2001) Grain boundary energies in olivine derived from natural microstructures.
702 Contributions to Mineralogy and Petrology 140 (5), 567-576.
703

704 Faul, U.H., Fitz-Gerald, J. (1999) Grain misorientations in partially molten olivine aggregates: an electron
705 backscatter diffraction study. Contributions to Mineralogy and Petrology 26, 187-197
706

707 Gillet, P., Gerard, Y., Williame, C. (1987) The calcite-aragonite transition: mechanism and microstructure
708 induced by transformation stresses and strain. Bulletin de Minéralogie 110, 481-496.
709

710 Guillet, P., Madon, M., (1982) Un modèle de dislocations pour la transition aragonite-calcite. Bulletin de
711 Minéralogie 105, 590-597.
712

713 Hielscher, R., Schaeben, H. (2008) A novel pole figure inversion method: specification of the MTEX
714 algorithm. Journal of Applied Crystallography 41, 1024–1037. [http://](http://dx.doi.org/10.1107/S0021889808030112)
715 dx.doi.org/10.1107/S0021889808030112.
716

717 Hiraga, T., Anderson, I. M., Kohlstedt, D. L. (2004) Grain boundaries as reservoirs of incompatible
718 elements in the Earth's mantle. Nature 427, 699-703
719

720 Hunter, W.C., Smith, D. (1981) Garnet peridotite from Colorado Plateau ultramafic diatremes: hydrates,
721 carbonates and comparative geothermometry. Contributions to Mineralogy and Petrology 76, 312–320.
722

723 Karson, J. A., Lawrence, R. M. (1997) Tectonic window into gabbroic rocks of the middle oceanic crust in
724 the MARK area near Sites 921–924, Proceeding of the Ocean Drilling Program Scientific Results, 153,
725 61–76, doi:10.2973/odp.proc.sr.153.005.1997.
726

727 Kelemen, P. B., Kikawa, E., Miller, D. J. and the Shipboard Scientific Party (2007), Leg 209 summary:
728 Processes in a 20- km-thick conductive boundary layer beneath the Mid-Atlantic Ridge, 14_–16_N, Proc.
729 Ocean Drill. Program Sci. Results, 209, 1–33, doi:10.2973/odp.proc.sr.209.001.2007.
730
731 Kilian, R., Heilbronner, R., and Stunitz, H. (2011). Quartz grain size reduction in a granitoid rock and the
732 transition from dislocation to diffusion creep. *Journal of Structural Geology* 33, 1265–1284,
733 <https://doi.org/10.1016/j.jsg.2011.05.004>
734
735 Krakow, R., Bennet, R.B., Johnstone, D.N., Vukmanovic, Z., Solano-Alvares, W., Laine, S.J., Einsle, J.F.,
736 Midgley, P., Rae, C., Hielscher, R. (2017) On three-dimensional misorientation spaces. *Proceedings of the*
737 *Royal Society A* 473: 20170274. <http://dx.doi.org/10.1098/rspa.2017.0274>
738
739 Kruhl, J., Wirth, R., Morales, L.F.G. (2013) Quartz grain boundaries as fluid pathways in metamorphic
740 rocks. *Journal of Geophysical Research* Vol. 118(5), 1957–1967. doi:10.1029/2012JB009644
741
742 Lagoeiro, L.E. 1998. Transformation of magnetite to hematite and its influence on the dissolution of iron
743 oxide minerals. *Journal of Metamorphic Petrology* 16, 415-423
744
745 Laporte, D., Watson, E. B. (1995) Experimental and theoretical constraints on melt distribution in crustal
746 sources: The effect of crystalline anisotropy on melt interconnectivity, *Chemical Geology* 124, 161–184,
747 doi:10.1016/0009-2541(95)00052-N.
748
749 Lloyd G.E., Farmer A.B., Mainprice, D. (1997) Misorientation analysis and the formation and
750 orientation of subgrain and grain boundaries. *Tectonophysics* 279, 55-78.
751
752 Lloyd, G. E. (2004) Microstructural evolution in a mylonitic quartz simple shear zone: the significant
753 roles of dauphine twinning and misorientation. In: Alsop, G. I., Holdsworth, R. E., McCaffrey, K. &

754 Handy, M. (eds) *Transports and Flow Processes in Shear Zones*. Geological Society, London, Special
755 Publications, 224, 39–61.

756

757 Mainprice, D., Humbert, M., Wagner, F. 1990. Phase transformations and inherited lattice preferred
758 orientations: implications for seismic properties. *Tectonophysics* 180, 213-218.

759

760 Mainprice, D., Lloyd, G.E., Casey, M. (1993) Individual orientation measurements in quartz
761 polycrystals - advantages and limitations for texture and petrophysical property determinations.
762 *Journal of Structural Geology* 15, 1169-1187

763

764 Marquardt, K., Rohrer, G.S., Morales, L., Rybacki, E., Marquardt, H.; Lin, B. (2015). The most frequent
765 interfaces in olivine aggregates: The GBCD and its importance for grain boundary related processes.
766 *Contributions to Mineralogy and Petrology* 170:40, doi: 10.1007/s00410-015-1193-9.

767

768 Marti, S., Stünitz, H., Heilbronner, R., Plümper, O., Kilian, R. (2018) Syn-kinematic hydration reactions,
769 grain size reduction, and dissolution–precipitation creep in experimentally deformed plagioclase–
770 pyroxene mixtures. *Solid Earth*, 9, 985–1009 <https://doi.org/10.5194/se-9-985-2018>

771

772 McGetchin, T.R., Silver, L.T. (1970) Compositional relations in minerals from kimberlitic and related
773 rocks in the Moses Rock Dike, San Juan County, Utah. *American Mineralogist* 55, 1738–1771.

774

775 McLaren, A. C. (1986) Some speculations on the nature of high angle grain boundaries in quartz rocks, in
776 *Mineral and Rock Deformation: Laboratory Studies: The Paterson Volume*, pp. 233–245, AGU,
777 Washington, D. C.

778

779 McNamara, D. D., Wheeler, J., Pearce, M., & Prior, D. J. (2012). Fabrics produced mimetically during
780 static metamorphism in retrogressed eclogites from the Zermatt-Saas zone, Western Italian Alps. *Journal*
781 *of Structural Geology* 44, 167-178. <http://dx.doi.org/10.1016/j.jsg.2012.08.006>

782
783 McTigue Jr., J.W., Wenk, H.-R. (1985) Microstructure and orientation relationships in the dry-state
784 aragonite-calcite and calcite-lime phase transformations. *American Mineralogist* 70, 1253-1261

785 Menegon, L., Nasipuri, P., Stünitz, H., Behrens, H., Ravna, E. (2011). Dry and strong quartz during
786 deformation of the lower crust in the presence of melt. *Journal of Geophysical Research* 116, B10410.
787 <http://dx.doi.org/10.1029/2011JB008371>.

788
789 Menegon, L., Piazzolo, S., Pennacchioni, G. (2010). The effect of Dauphiné twinning on plastic strain in
790 quartz. *Contributions to Mineralogy and Petrology*. 161, 635–652.

791
792 Miranda, E.A., John, B.E. (2010) Strain localization along the Atlantis Bank oceanic detachment fault
793 system, Southwest Indian Ridge. *Geochemistry, Geophysics, Geosystems* 11(4),
794 doi:10.1029/2009GC002646

795
796 Morales, L.F.G., Mainprice, D., Boudier, F. (2013) The influence of hydrous phase on the microstructure
797 and seismic properties of a hydrated mantle rock. *Tectonophysics* 594, 103–117.

798
799 Morales, L.F.G., Mainprice, D., Kern, H. (2018) Olivine-antigorite orientation relationships:
800 microstructures, phase boundary misorientation and seismic properties. *Tectonophysics* 724-725, 93-115.

801
802 Morawiec A (1997) Distributions of misorientation angles and misorientation axes for crystallites with
803 different symmetries. *Acta Crystallographica A* 53, 273-285

804

805 Morawiec, A. (1995) Misorientation angle distribution of randomly oriented symmetric objects. *Journal of*
806 *Applied Crystallography* 28, 289-293

807

808 Obata, M, Ozawa, K. (2011) Topotaxial relationships between spinel and pyroxene in kelyphite after
809 garnet in mantle-derived peridotites and their implications to reaction mechanism and kinetics.
810 *Mineralogy and Petrology* 101, 217-224, doi:10.1007/s00710-011-0145-y

811

812 Plümper, O., King, H.E., Vollmer, C., Ramasse, Q., Jung, H., Austrheim, H. (2012) The legacy of crystal-
813 plastic deformation in olivine - high-diffusivity pathways during serpentinization. *Contribution to*
814 *Mineralogy and Petrology* 163 (4), 701–724.

815

816 Porter, D.A., Easterling, K.E. (1992). *Phase transformations in metals and alloys*. Chapman and Hall

817

818 Rosière, C.A., Siemes, H., Quade, H., Brokmeier, H.G., Jansen, E.M. (2001) Microstructures, textures and
819 deformation mechanisms in hematite. *Journal of Structural Geology* 23, 1429–1440.

820

821 Smith, D. (1995) Chlorite-rich ultramafic reaction zones in Colorado Plateau xenoliths: recorders of sub-
822 Moho hydration. *Contributions to Mineralogy and Petrology* 121, 185–200.

823

824 Smith, D. (2010). Antigorite peridotite, metaserpentinite, and other inclusions within diatremes on the
825 Colorado Plateau, SW USA: implications for the mantle wedge during low-angle subduction. *Journal of*
826 *Petrology* 51 (6), 1355–1379.

827

828 Sutton A.P., Balluffi, R.W. (1995) *Interfaces in Crystalline Materials*. Clarendon Press, Oxford, UK

829

830 Wenk, H.-R., Janssen, C., Kenkmann, T., Dresen, G. (2011) Mechanical twinning in quartz: Shock
831 experiments, impact, pseudotachylites and fault breccias. Tectonophysics 510, 59-79

832
833 Wenk HR, Chen K, Smith R (2011) Morphology and microstructure of magnetite and ilmenite inclusions
834 in plagioclase from Adirondack anorthositic gneiss. American Mineralogist 96(8–9):1316–1324

835
836 Wheeler, J., Prior, D.J., Jiang, Z., Speiss, R., & Trimby, P.W. (2001). The petrological significance of
837 misorientations between grains. Contributions to Mineralogy and Petrology 141, 109-124.

838

839 **FIGURE CAPTIONS**

840

841 **Figure 1** – Schematic drawing showing the misorientation between 2 different phases, one “hexagonal”
842 (yellow grain) and a second, “cubic” (purple and green grains). The orientation of the individual
843 crystallites is described by an orientation \mathbf{g}_1 , \mathbf{g}_2 or \mathbf{g}_3 , which describes the transformation from crystal
844 reference frame (h_i) and sample reference frame (r), here denoted as x,y and z . The misorientation (M)
845 between grains of the same phase can be described as $M=\mathbf{g}_2^{-1}\mathbf{g}_3$, where g are their respective orientations.
846 The misorientation along different phases nevertheless ($M(ip)$) needs to consider the crystallographic
847 system of the individual phases (cs), and can be described like $M(ip)=\mathbf{g}_{1(cs1)}^{-1}\mathbf{g}_{3(cs2)}$, where cs_1 is the crystal
848 symmetry of phase 1 (e.g. hexagonal) and cs_2 the crystal symmetry of phase 2 (e.g. cubic). The combined
849 symmetries of the different cs is what defines the dimensions of the fundamental zones for the plots
850 presented here.

851

852 **Figure 2** – (a) Combined phase and image quality map, showing the distribution of calcite and aragonite.
853 Orientation maps for calcite and aragonite, respectively, inverse pole figure color-coded.

854

855 **Figure 3** – Pole figures for calcite and aragonite considering all points in the orientation maps from Fig. 2.
856 Scale is given in multiples of uniform distribution. ODFs calculated assuming a halfwidth of 10° .

857
858 **Figure 4** – Interphase misorientation angle (a) and axis (b) distribution for calcite-aragonite, assuming a
859 uniform distribution function, with calcite reference frame plotted in (b). Interphase misorientation angle
860 distribution (c) as calculated from the EBSD orientation map. Red bars on the histogram show the
861 correlated misorientation (between neighbor grains), the orange line shows the uncorrelated distribution
862 calculated from the orientation distribution function, and the yellow line shows the uniform misorientation
863 angle distribution as in (a). Correlated interphase misorientation axis distribution plotted against the
864 calcite reference frame, for all range of misorientation axes from (c). (e) Phase map showing calcite
865 (purple) and aragonite (green), and the more common interphase boundaries traces, with misorientations
866 between adjacent crystals ranging from 30 to 70° . Scales in the pole figures b and d are given in multiples
867 of uniform distribution.

868
869 **Figure 5** – (a) Combined phase and image quality map, showing the distribution of olivine (purple) and
870 antigorite (green), whereas the red phase at the bottom is an isolated grain of magnetite.

871
872 **Figure 6** - Pole figures for olivine and antigorite, considering all points in the orientation maps from Fig.
873 5. Scale is given in multiples of uniform distribution. ODFs calculated assuming a halfwidth of 10° .

874
875 **Figure 7** - Interphase misorientation angle (a) and axis (b) distribution for olivine-antigorite, assuming a
876 uniform distribution function, with olivine reference frame plotted in (b). Interphase misorientation angle
877 distribution (c) as calculated from the EBSD orientation map. Red bars on the histogram show the
878 correlated misorientation (between neighbor grains), the orange line shows the uncorrelated distribution
879 calculated from the misorientation distribution function, and the yellow line shows the uniform
880 misorientation angle distribution as in (a). Correlated interphase misorientation axis distribution (d)

881 plotted against the olivine reference frame, for misorientations ranging from 70-100°. (e) Image quality
882 map showing the more common interphase boundaries, with misorientations ranging from 70 to 100°.
883 Scales in the pole figures b and d are given in multiples of uniform distribution, phase map is not
884 presented here for better visualization of the interfaces.

885
886 **Figure 8** - (a) Combined phase and image quality map, showing the distribution of magnetite (purple) and
887 hematite (green).

888
889 **Figure 9** - Pole figures for magnetite and hematite, considering all points in the orientation maps from
890 Fig. 8. Scale is given in multiples of uniform distribution. ODFs calculated assuming a halfwidth of 10°.

891
892 **Figure 10** - Interphase misorientation angle (a) and axis (b) distribution for magnetite and hematite,
893 assuming a uniform distribution function, with magnetite reference frame plotted in (b). Interphase
894 misorientation angle distribution (c) as calculated from the EBSD orientation map. Red bars on the
895 histogram show the correlated misorientation (between neighbor grains), the orange line shows the
896 uncorrelated distribution calculated from the misorientation distribution function, and the yellow line
897 shows the uniform misorientation angle distribution as in (a). Correlated interphase misorientation axis
898 distribution (d) plotted against the magnetite reference frame, for misorientations ranging from 20-50°. (e)
899 Image quality (IQ) map showing the most common interphase boundaries traces with misorientations
900 between adjacent grains ranging from 20 to 50°. Scales in the pole figures b and d are given in multiples
901 of uniform distribution, phase map is not presented here for better visualization of the interfaces.

902
903 **Figure 11** - (a) Combined phase and image quality map, showing the distribution of plagioclase (green),
904 olivine (red) and ilmenite (blue) in the studied gabbro sample. (b) Combined orientation map for
905 plagioclase (IPF color-coded) and image quality for the rest of the phases.

906

907 **Figure 12** - Pole figures for plagioclase, olivine and ilmenite, considering all points in the orientation
908 maps from Fig. 8. Scale is given in multiples of uniform distribution. ODFs calculated assuming a
909 halfwidth of 10° .

910
911 **Figure 13** - Interphase misorientation angle (a, c) and axis (b, d) distribution for the pairs plagioclase-
912 olivine and plagioclase-ilmenite, respectively, assuming a uniform orientation distribution function. On
913 stereogram (b) plagioclase reference frame is given as black letters in white background, whereas certain
914 olivine directions are given in black background and white letters. The same is valid for the plot (d), but
915 for ilmenite. Scale bars in b and d are given in multiples of uniform distribution.

916
917 **Figure 14** - Interphase misorientation angle distribution (a, c) and axes (b, d) as calculated from the EBSD
918 orientation map, for the pairs plagioclase-olivine and plagioclase-ilmenite, respectively. Red bars on the
919 histogram show the correlated misorientation (between neighbor grains), the orange line shows the
920 uncorrelated distribution calculated from the misorientation distribution function, and the yellow line
921 shows the uniform misorientation angle distribution as in Fig. 13 a,c. On the plot (b) misorientation angles
922 are limited to a range between $65-90^\circ$ for plagioclase-olivine, whereas in (d) the misorientation range is
923 from $85-130^\circ$, for plagioclase-ilmenite pair. On stereogram (b) plagioclase reference frame is given as
924 black letters in white background, whereas certain olivine directions are given in black background and
925 white letters. The same is valid for the plot (d), but for ilmenite. Scale bars in b and d are given in
926 multiples of uniform distribution.

927
928 **Figure 15** – Image quality maps showing plagioclase-olivine interphase boundaries traces with adjacent
929 interphase misorientations ranging from $60-100^\circ$ misorientation (a) and plagioclase-ilmenite interface
930 traces with misorientation ranging from $60-120^\circ$.

931

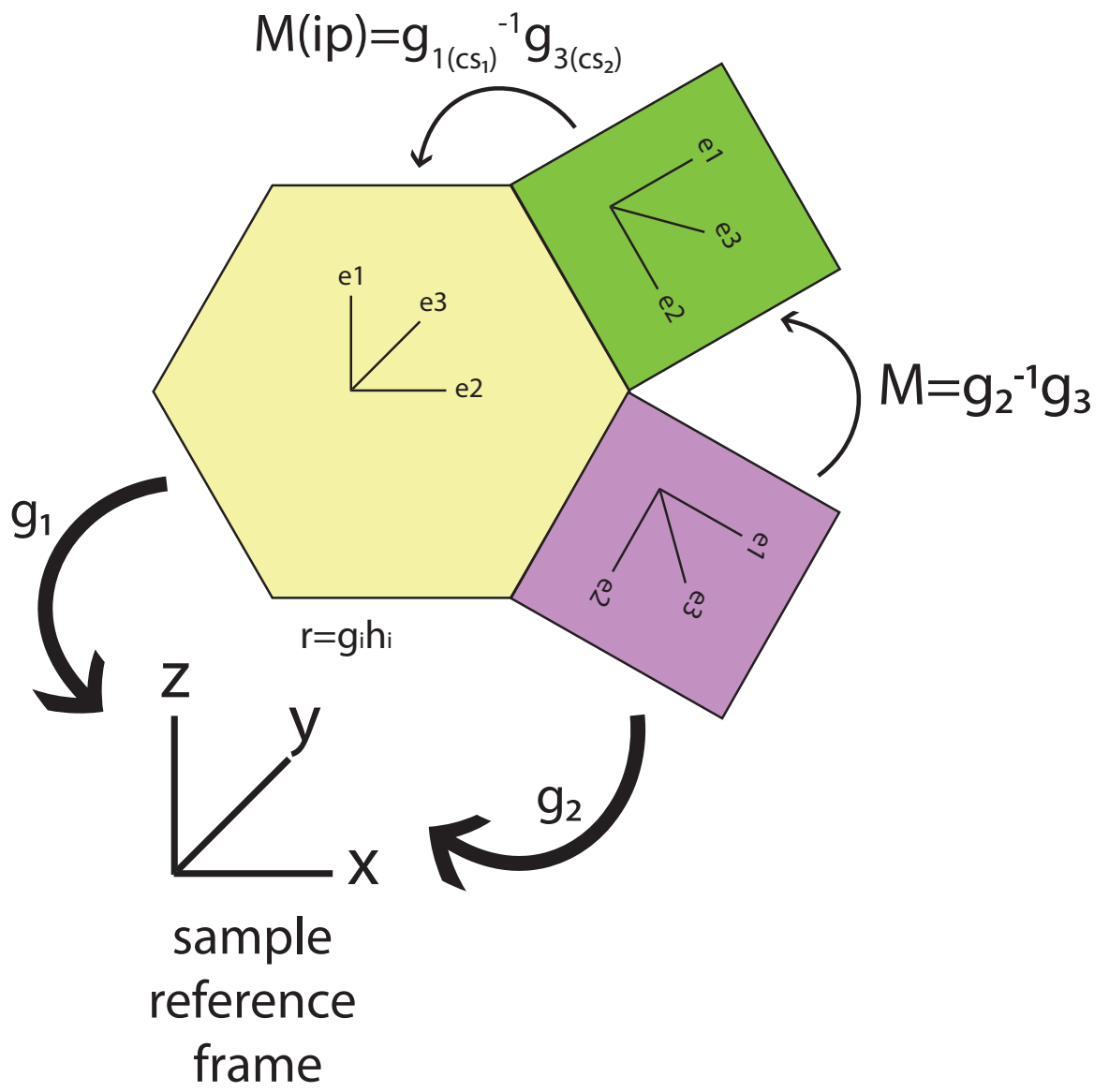


Figure 1

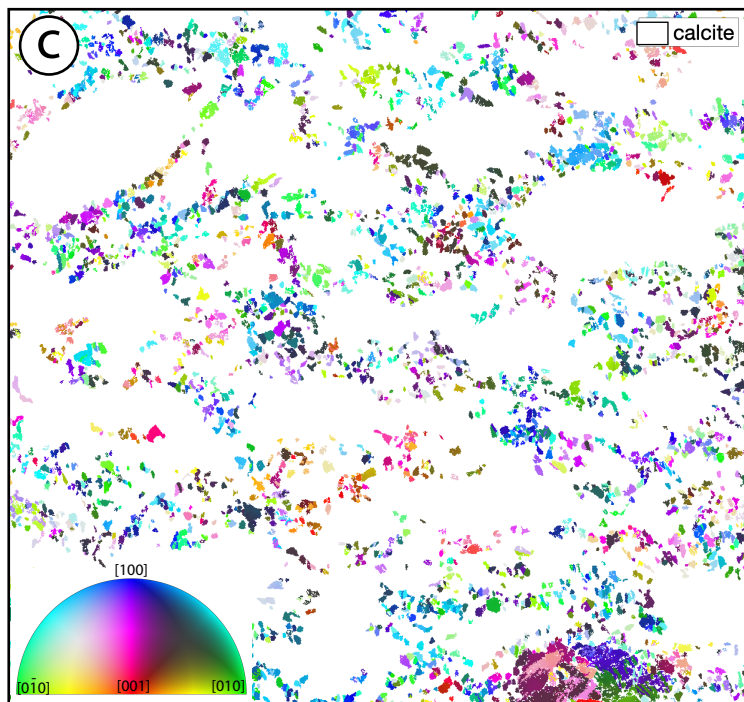
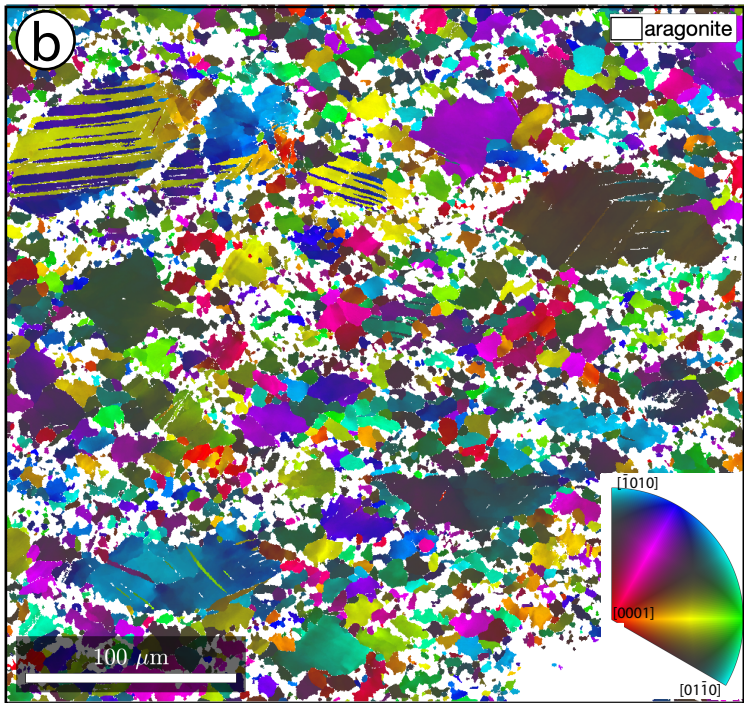
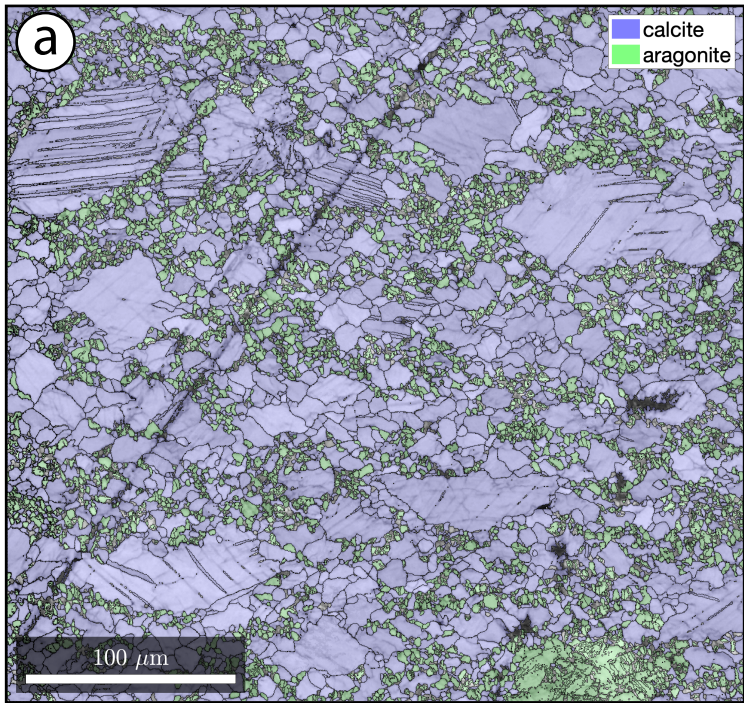
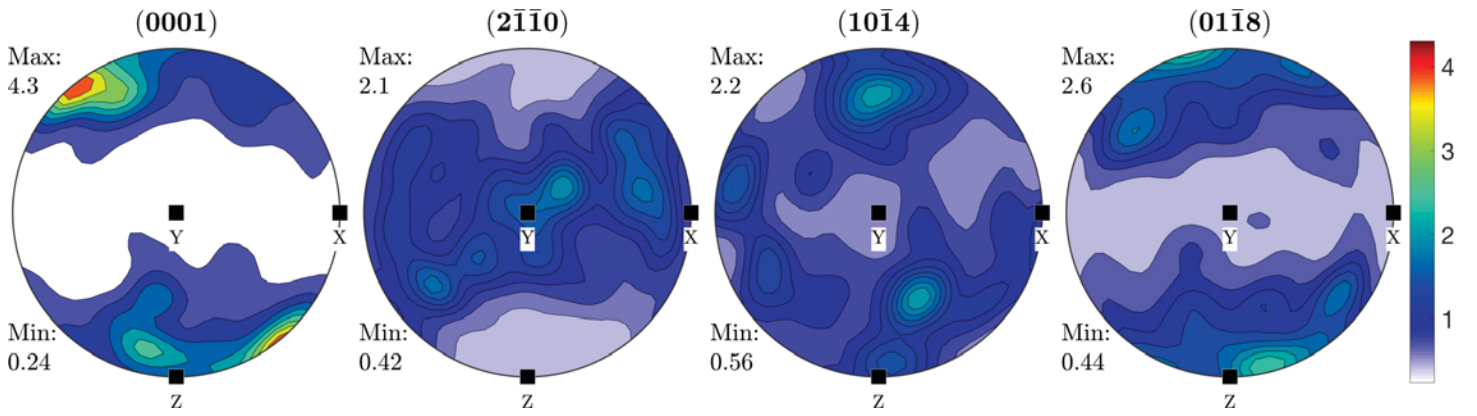


Figure 2

Calcite pole figures



Aragonite pole figures

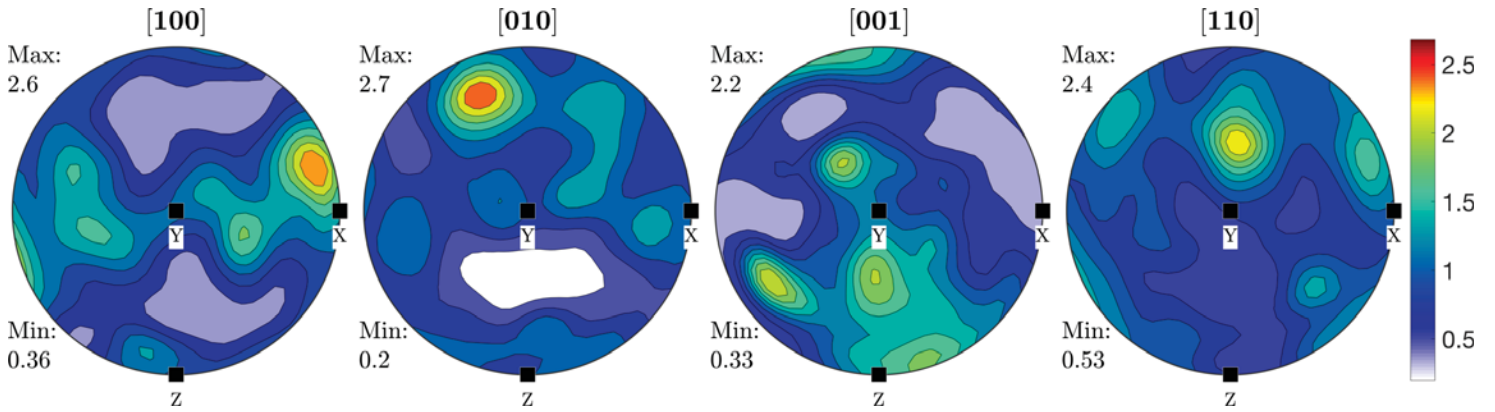
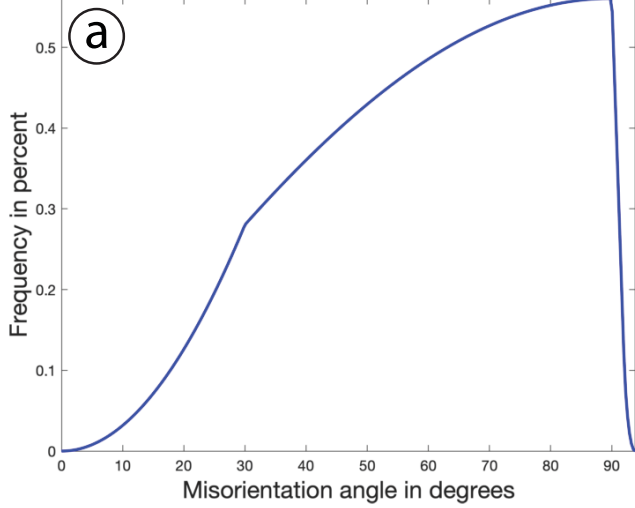
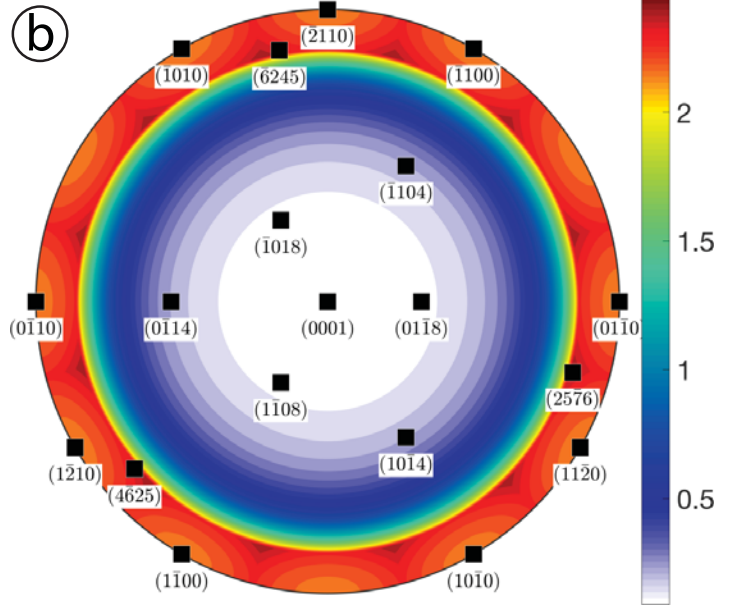


Figure 3

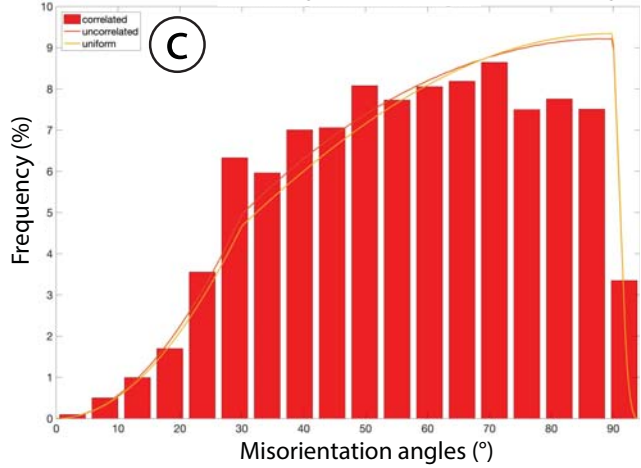
Uniform misorientation angle distribution: Calcite - Aragonite



Uniform misorientation axis distribution



Correlated misorientation angle distribution - Calcite - Aragonite



Correlated misorientation axis distribution

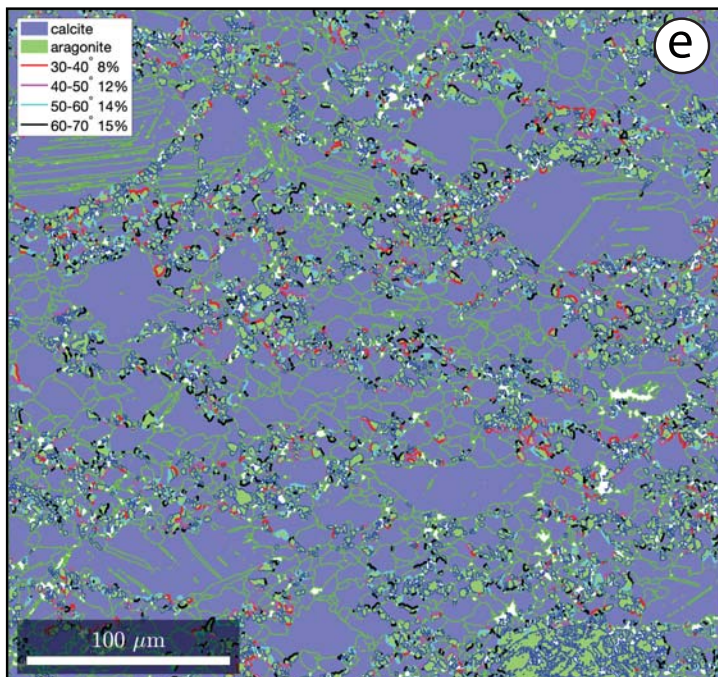
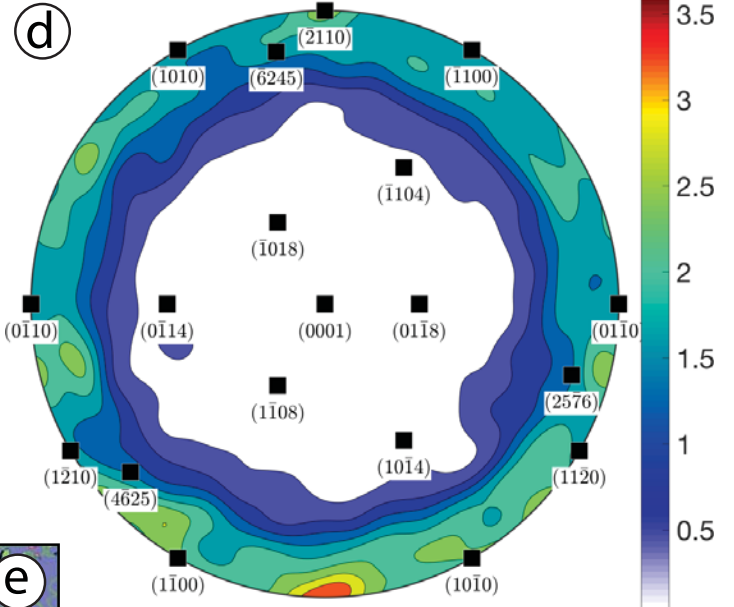


Figure 4

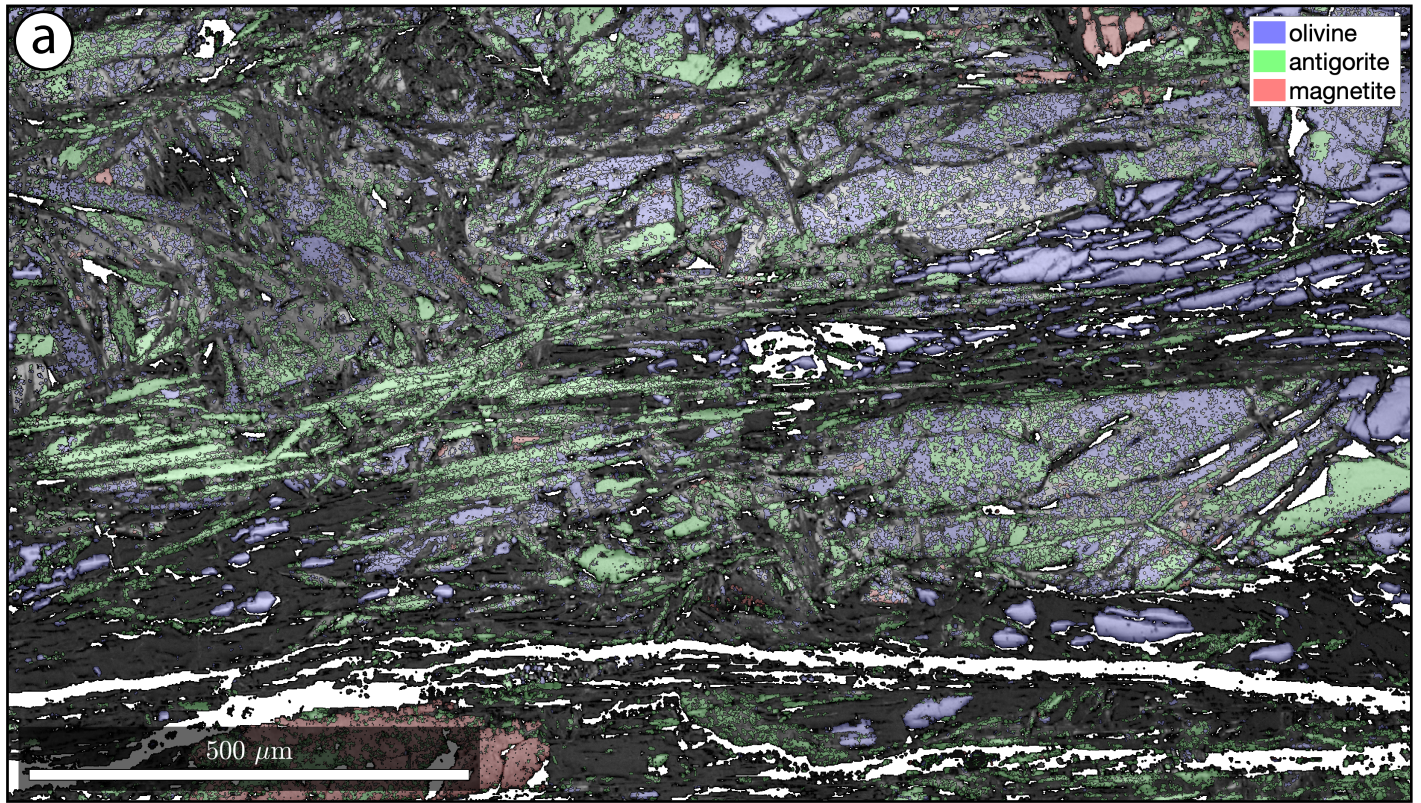
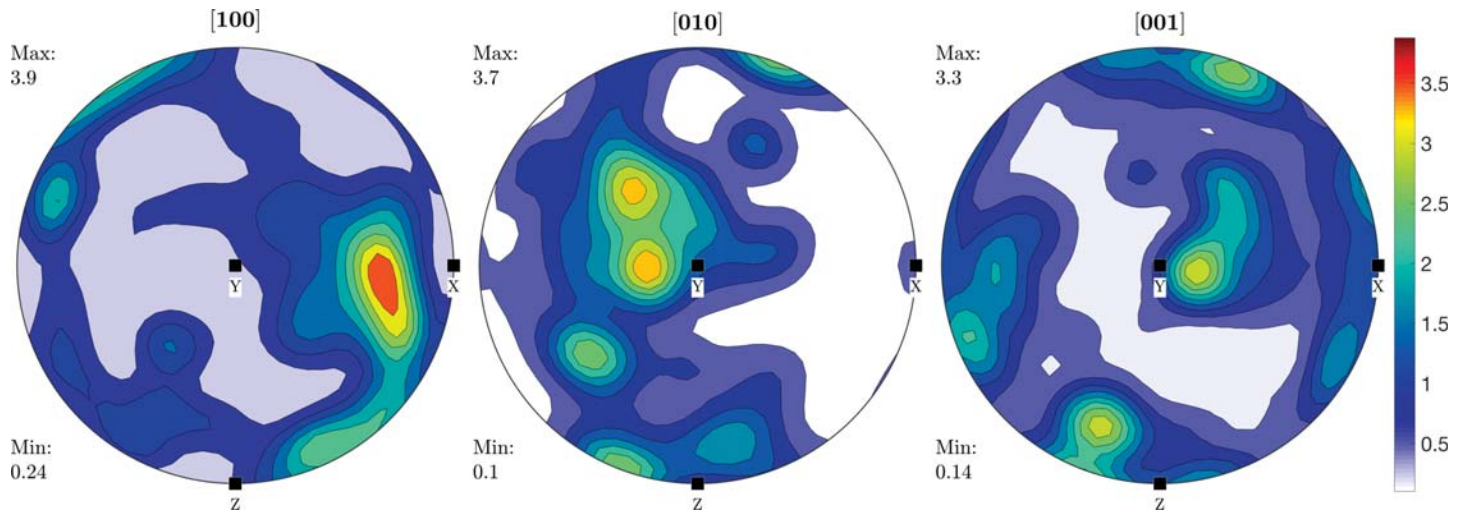


Figure 5

Olivine pole figures



Antigorite pole figures

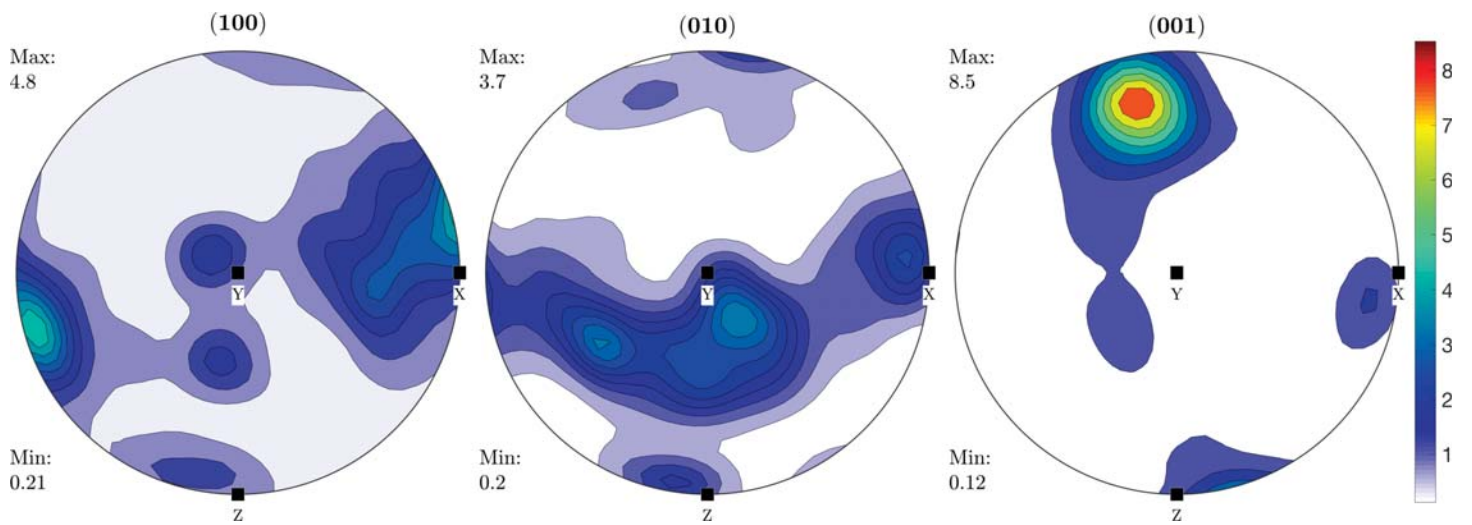


Figure 6

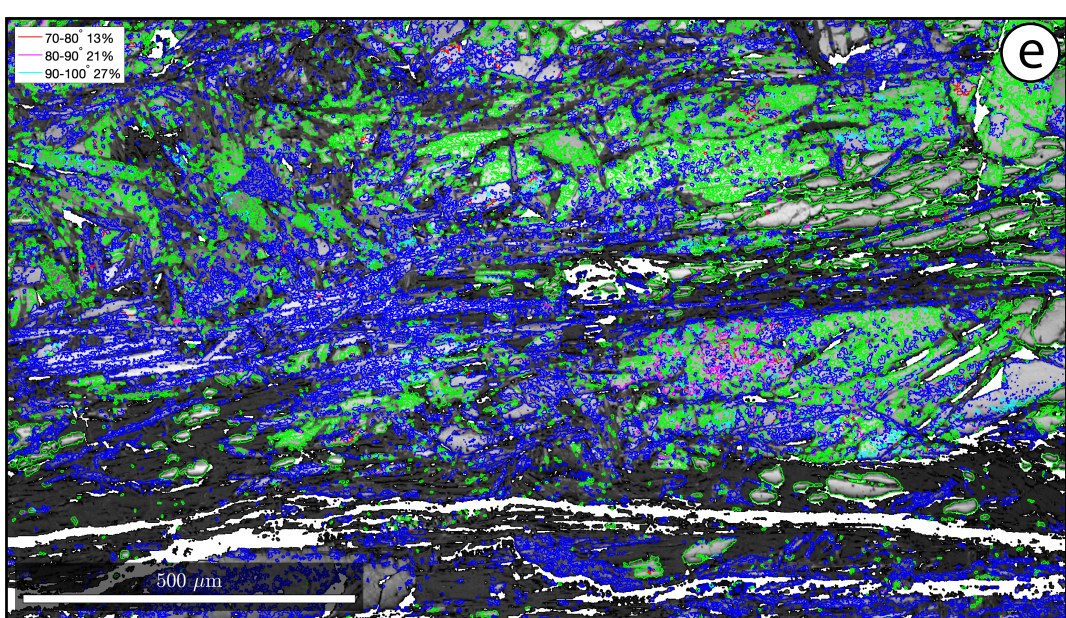
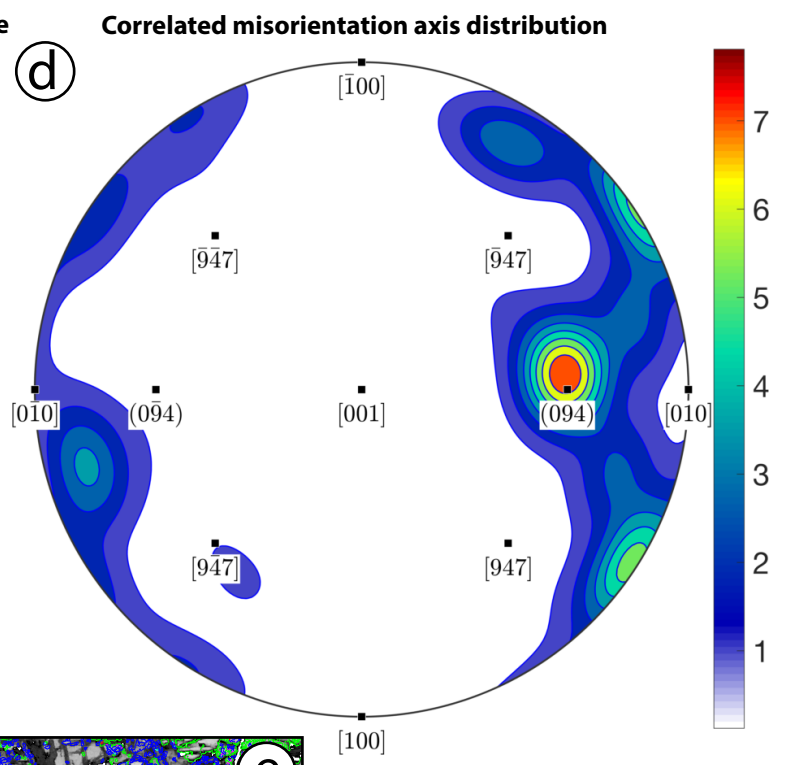
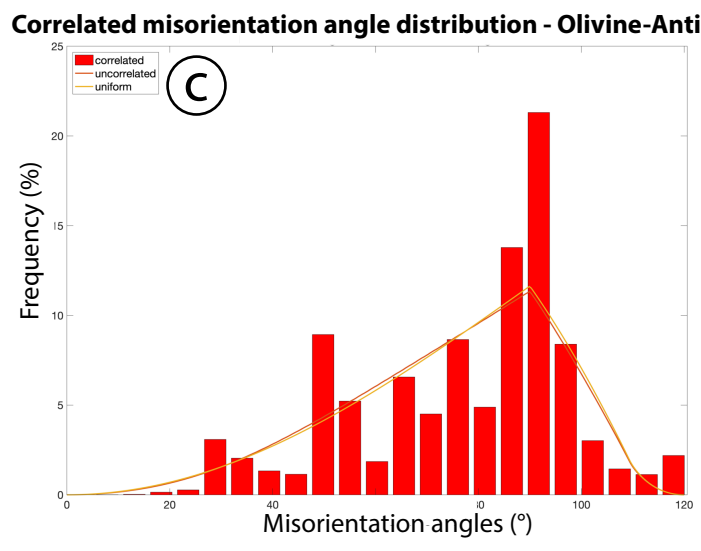
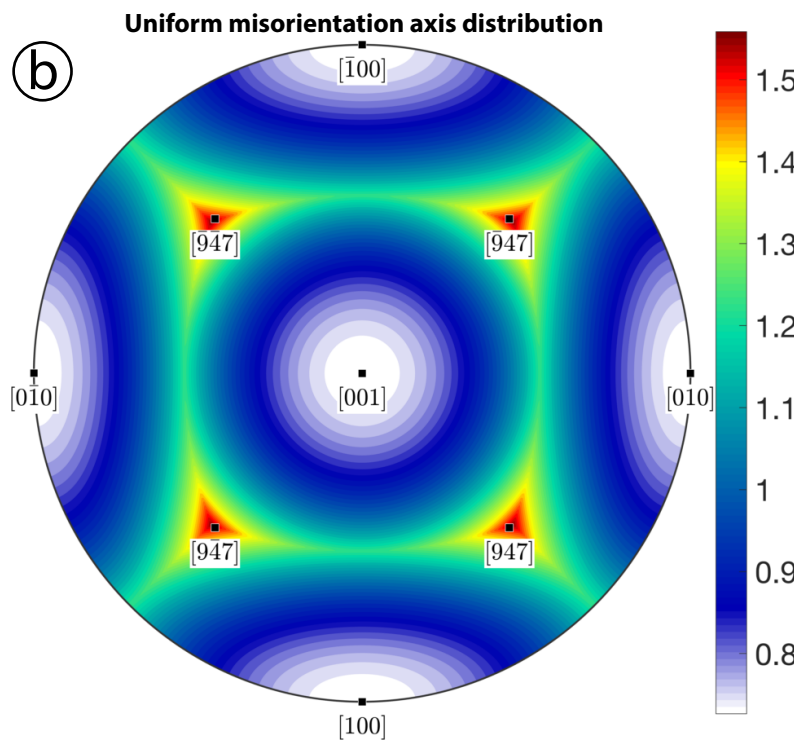
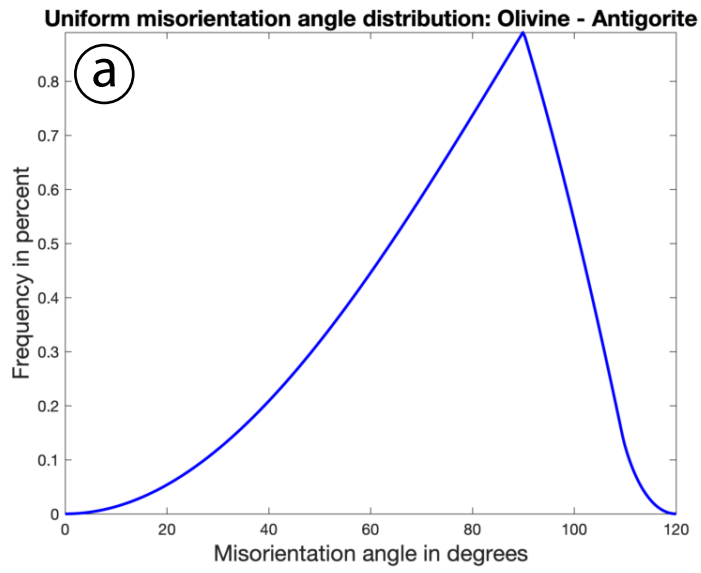


Figure 7

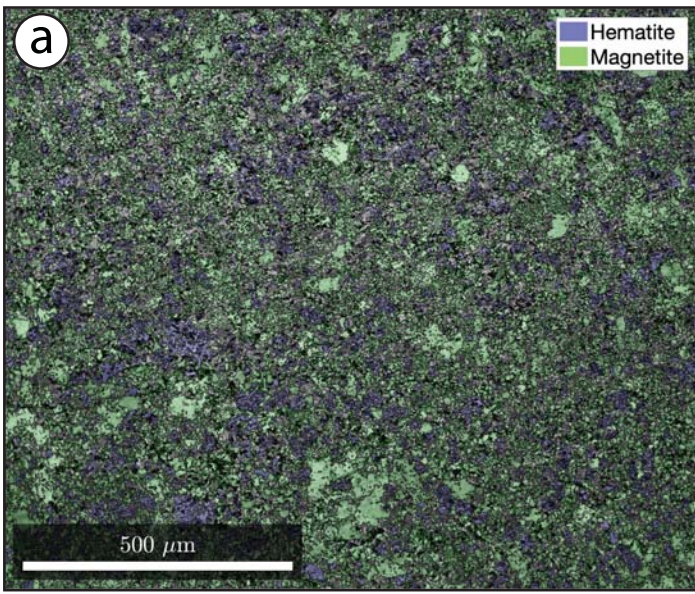
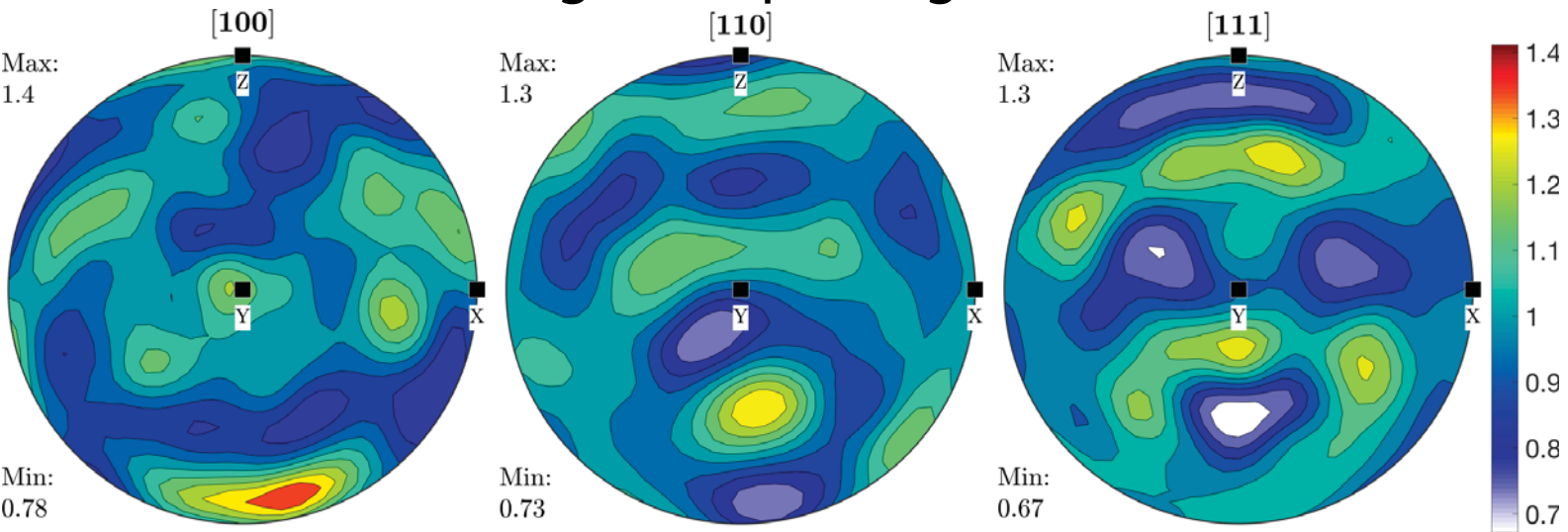


Figure 8

Magnetite pole figures



Hematite pole figures

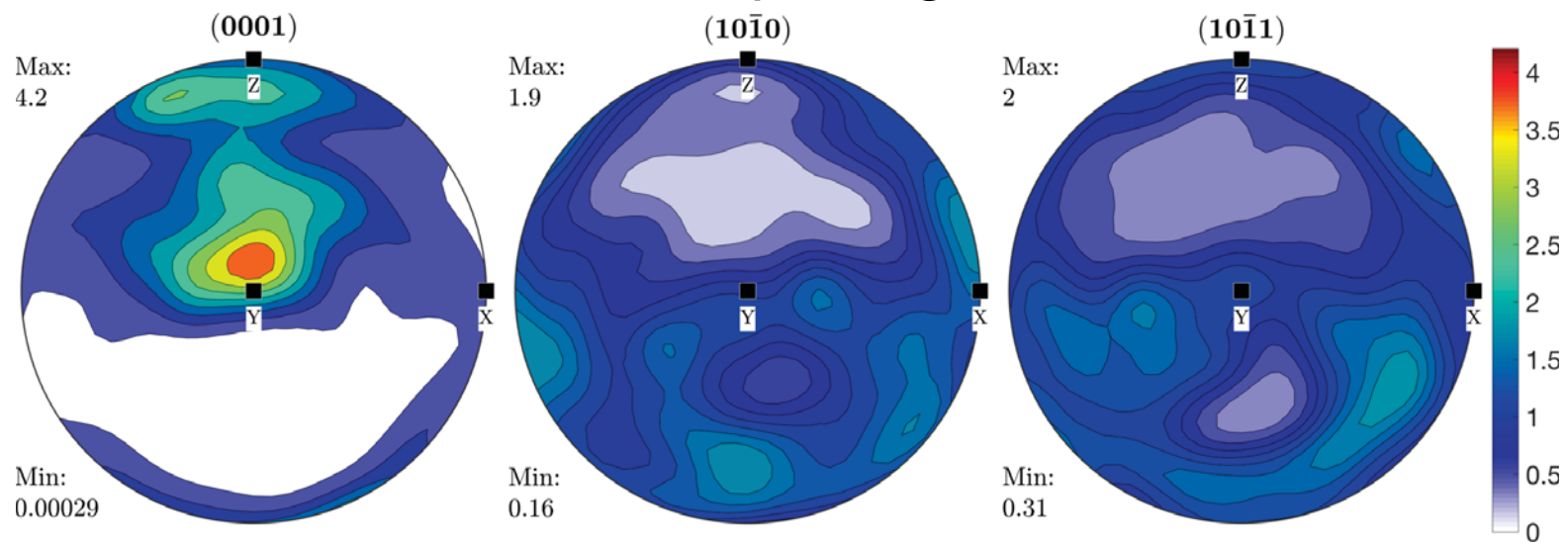


Figure 9

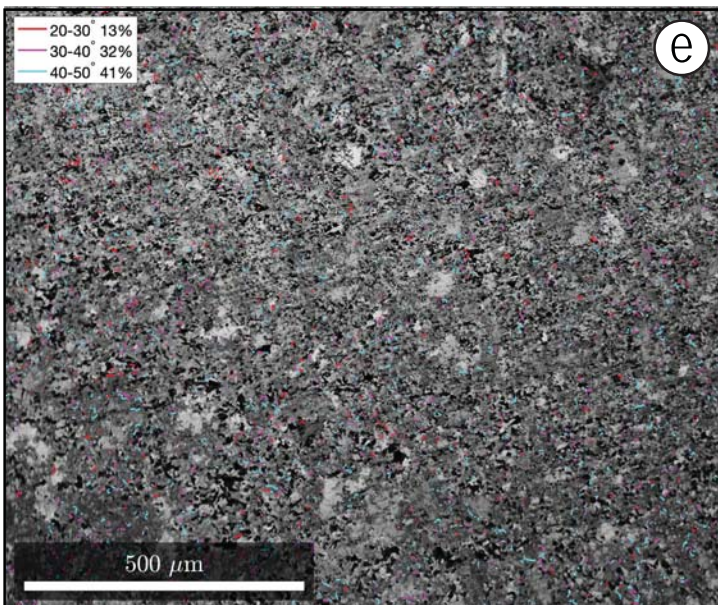
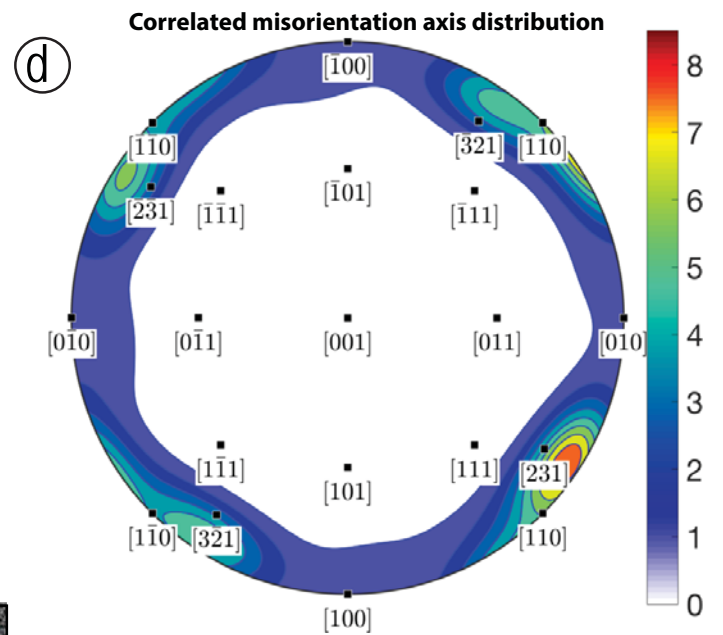
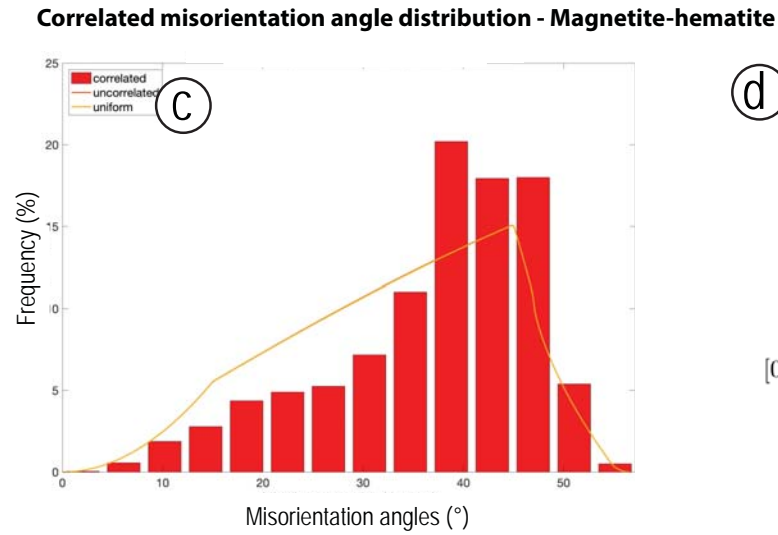
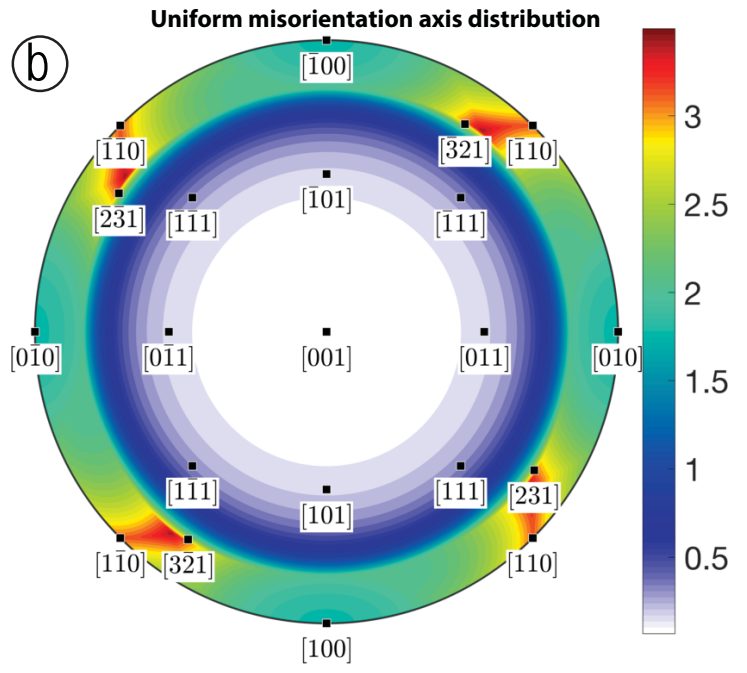
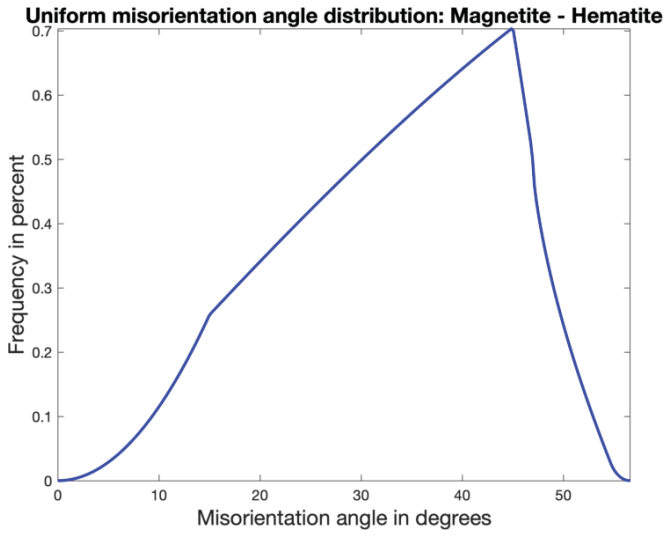


Figure 10

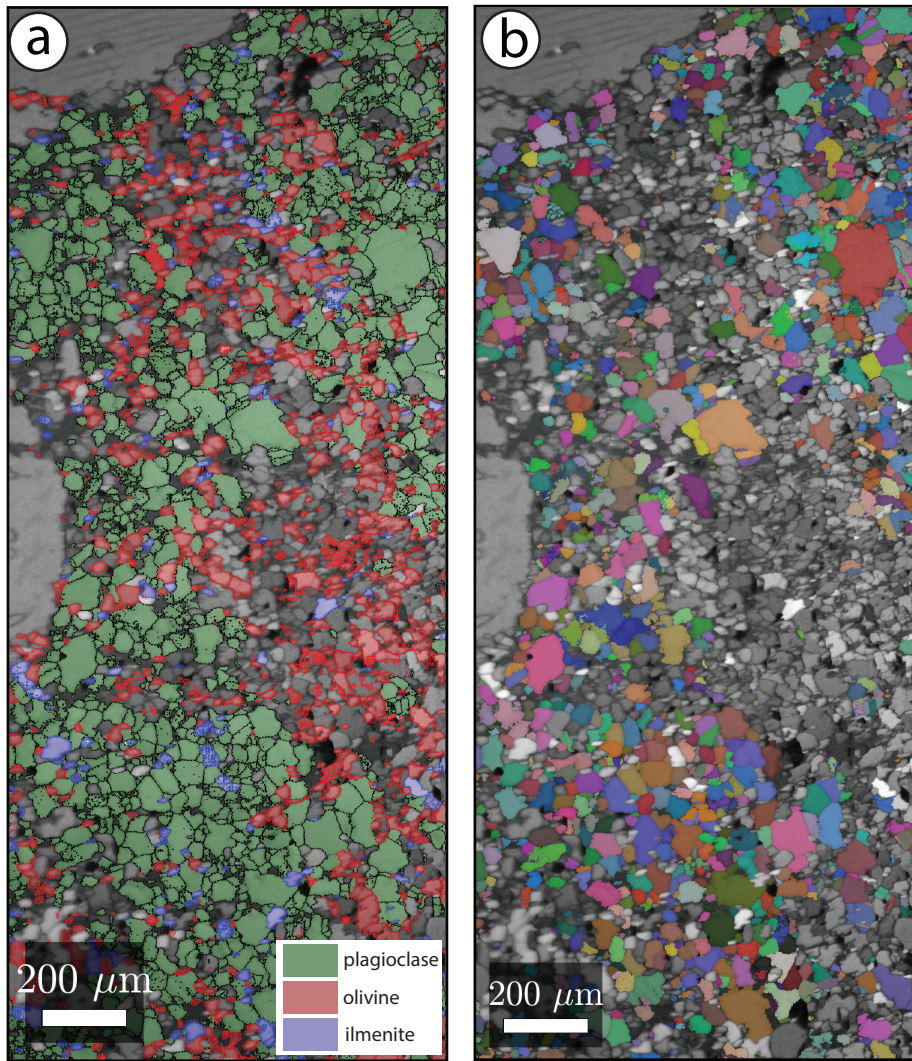
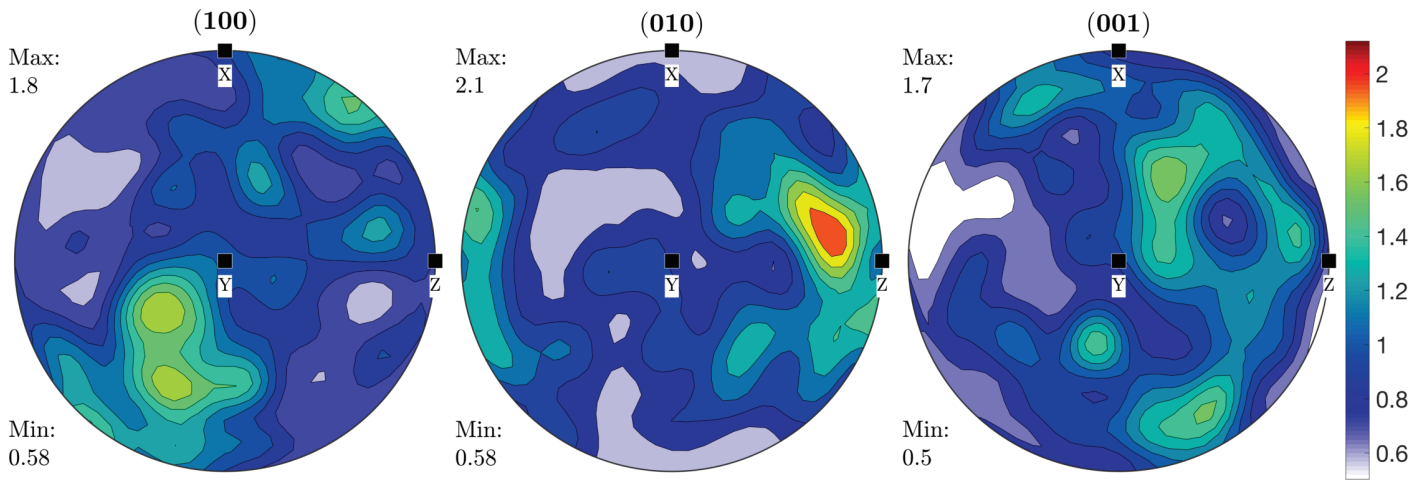
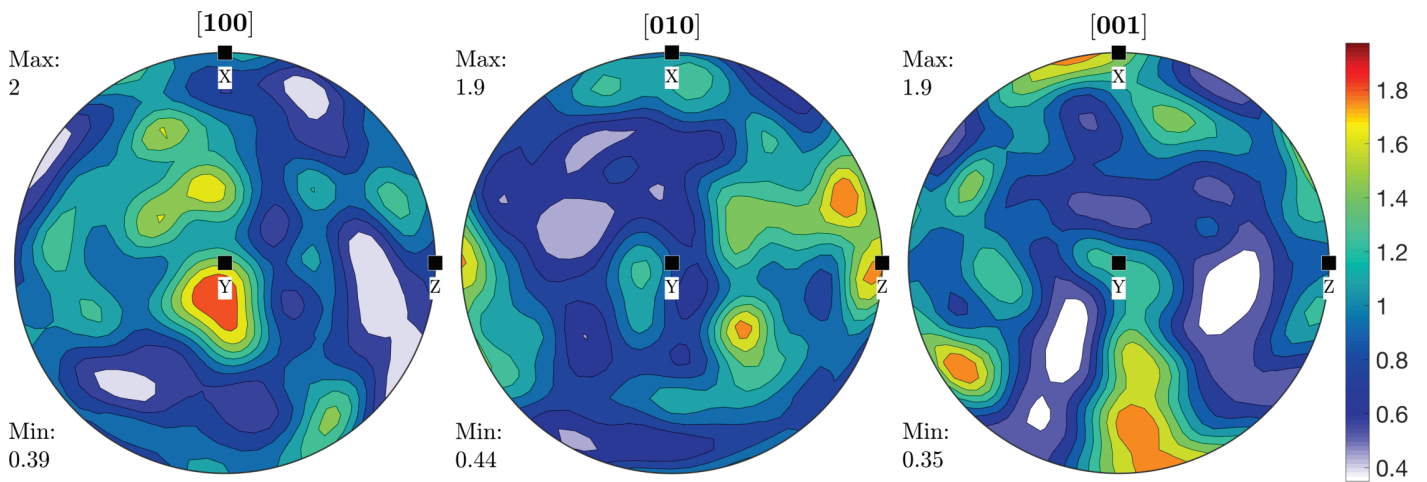


Figure 11

Plagioclase pole figures



Olivine pole figures



Ilmenite pole figures

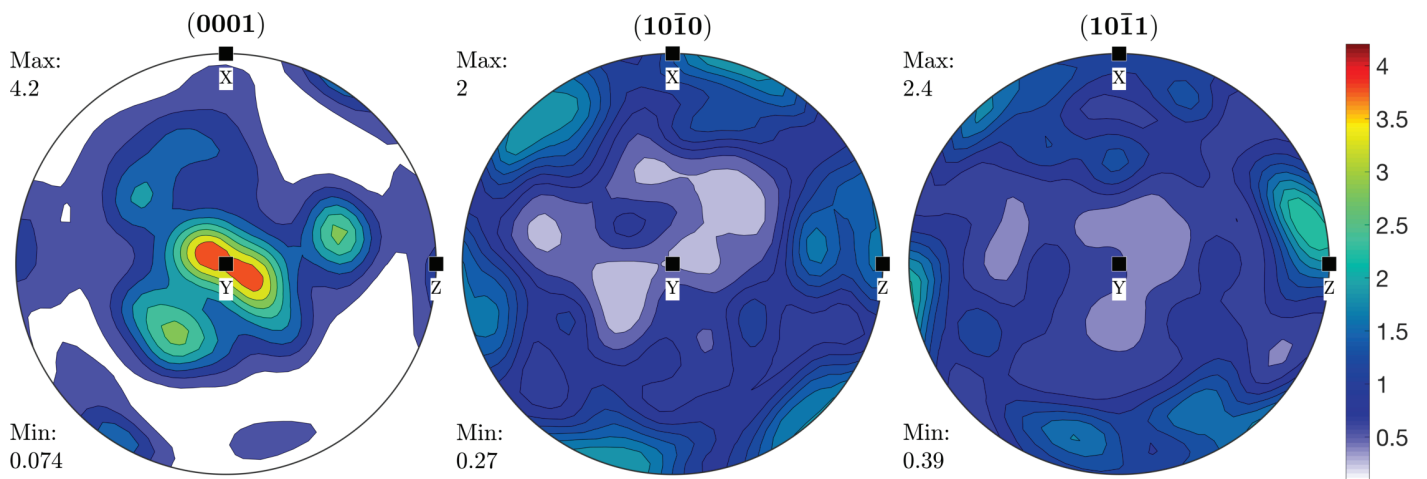


Figure 12

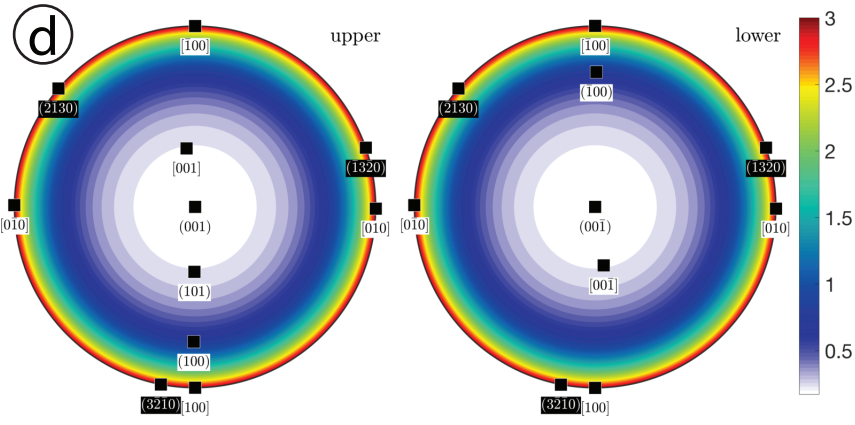
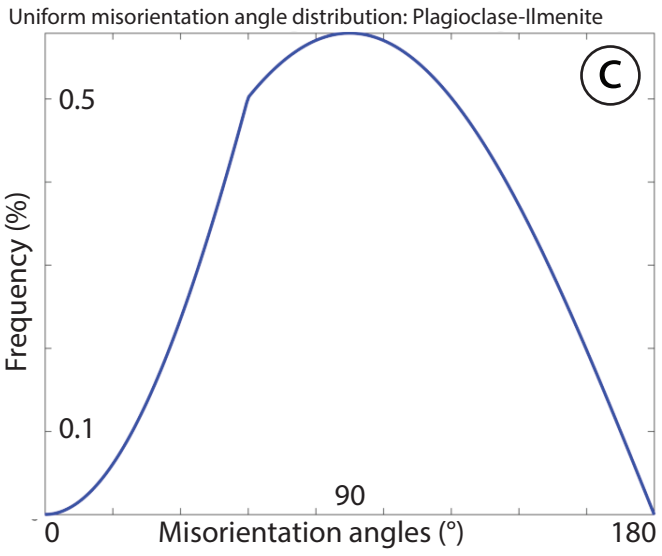
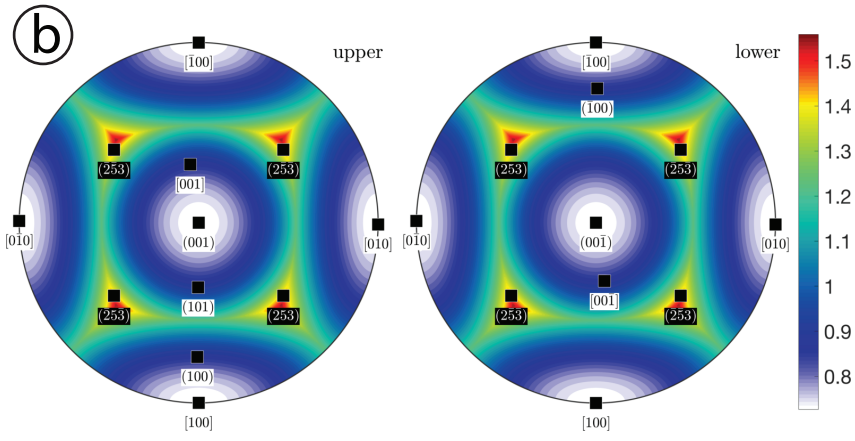
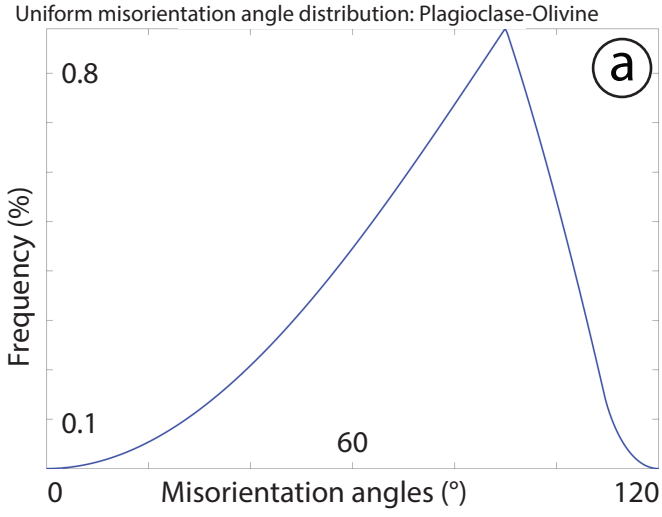


Figure 13

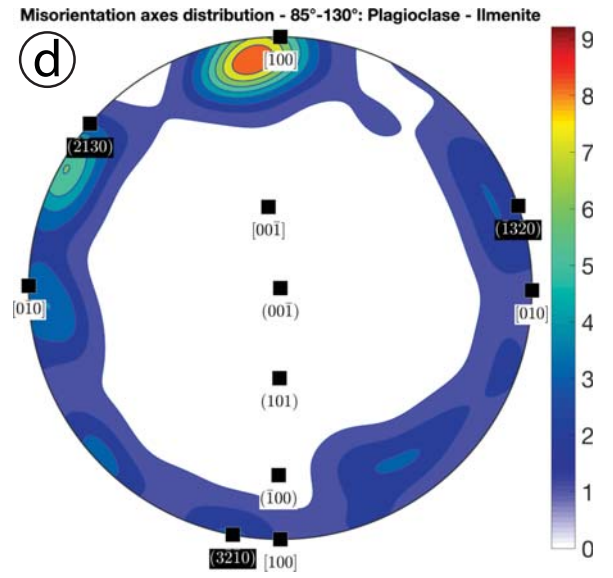
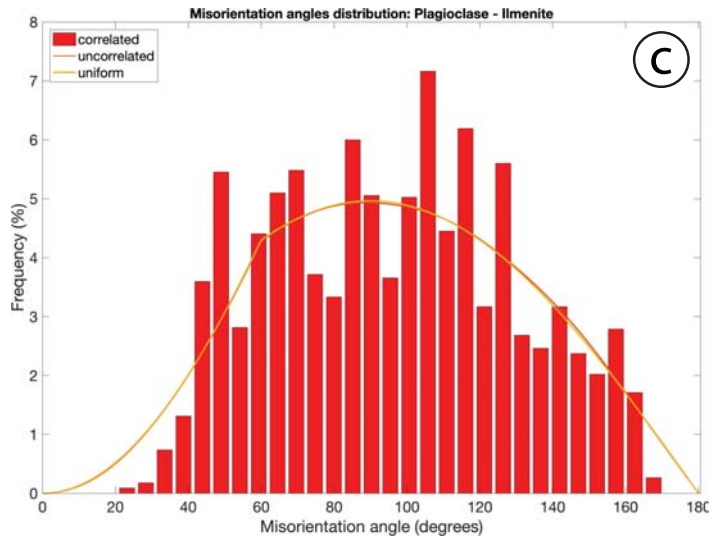
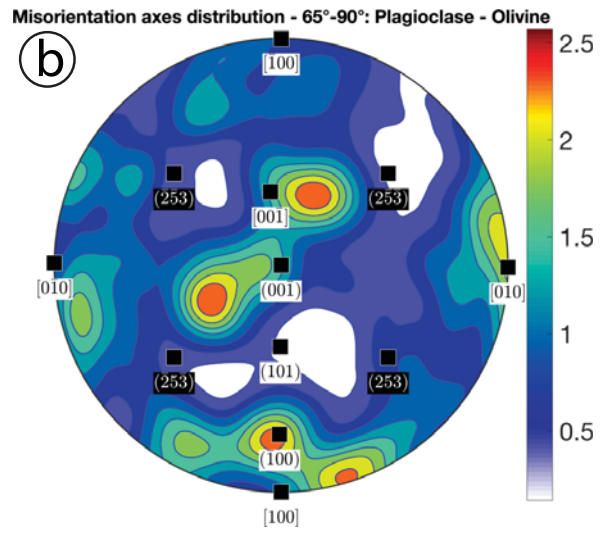
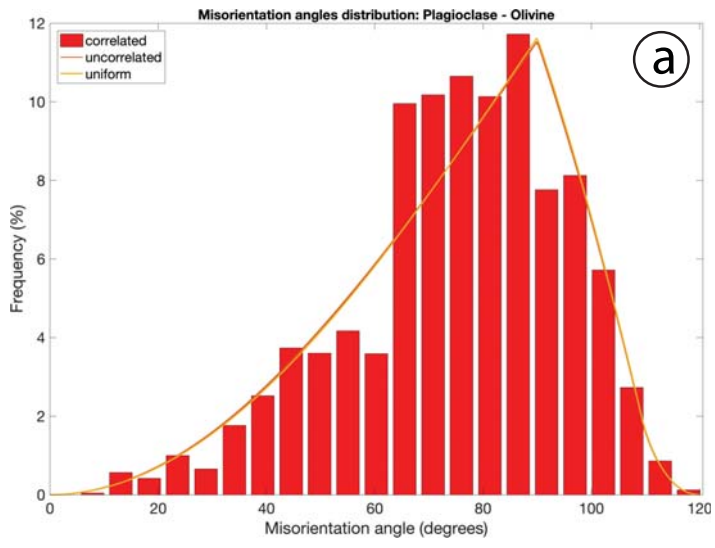
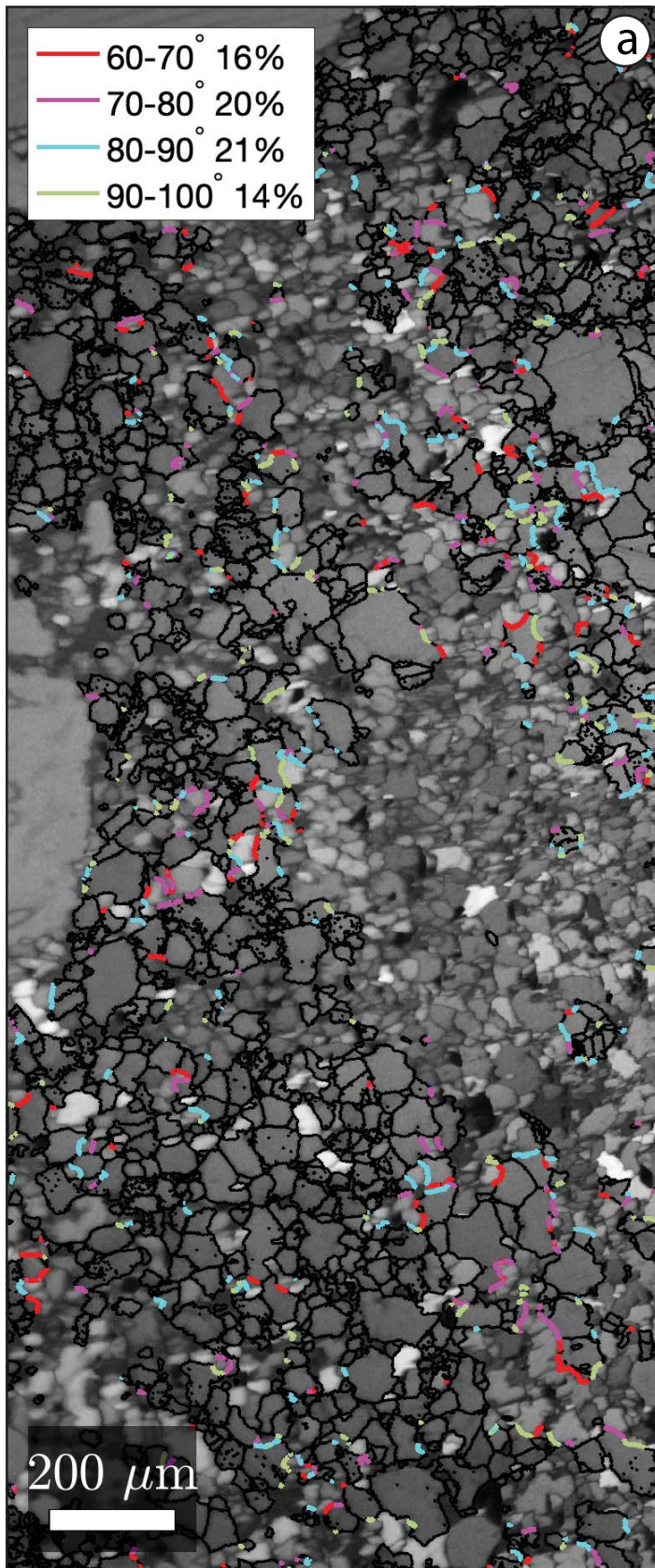


Figure 14

Plagioclase - Olivine interphase boundaries



Plagioclase - Ilmenite interphase boundaries

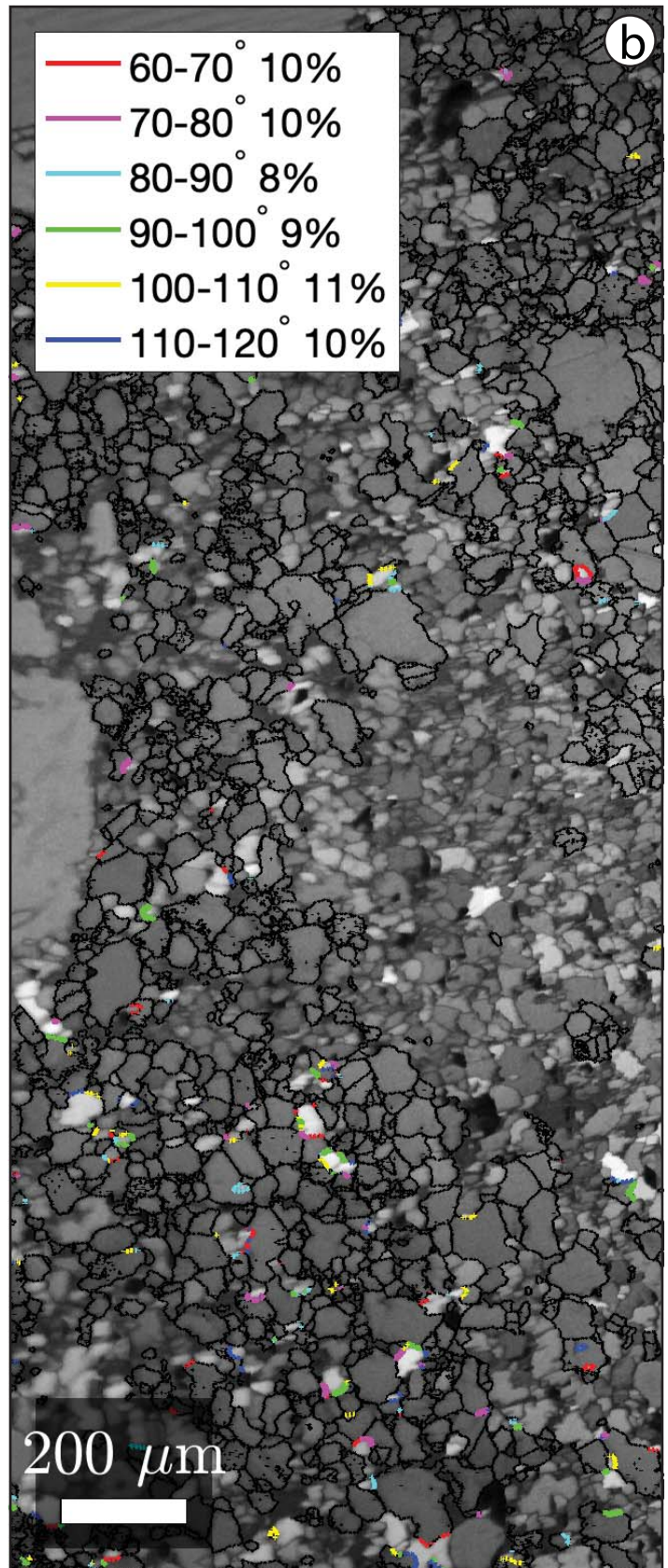


Figure 15