1	Variety of the drift pumice clasts from the 2021 Fukutoku-Oka-no-Ba eruption, Japan.
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#### 22 Abstract

23 Pumice rafts that arrived at the Nansei Islands, Japan, provided a unique opportunity to investigate the 24 Fukutoku-Oka-no-Ba (FOB) eruption of August 2021. Despite drifting for two months for ~1300 km, 25 the drift pumice raft had a large volume and contained a variety of pumice clasts, some of which were 26 deposited during a high tide in a typhoon, while others were washed up on a sandy beach. Most of the 27 drift pumice clasts are gray in color, vesicular, and have a groundmass containing black enclaves. Rare 28 black pumice and the main gray pumice components have similar trachytic compositions, with SiO2 = 61-62 mass% and total alkalis = 8.6-10 mass% (on an anhydrous basis). Both pumice types contain 29 30 clinopyroxene, plagioclase, and rare olivine phenocrysts. Thin-section observations show that the gray 31 pumice has more elongated vesicles as compared with the black pumice that has spherical vesicles, 32 even where the two types of pumice are in the same clast. The glass in the black pumice is transparent 33 and brown in color, while that in the gray pumice is colorless. No micro or nano-crystals were observed 34 during electron and optical microscopy. Raman spectra of the brown-colored glass exhibit a clear 35 magnetite peak, suggesting magnetite nanolites cause the brown color. High-Mg olivine in the black 36 pumice has an equilibrium temperature of 1240 °C and a rim diffusion profile indicative of reequilibration with the surrounding melt over a period of hours to days. 37

The textural relationships between the gray and black pumice suggest that the black pumice had become black and viscous before the two types of pumice mixed. Therefore, crystallization of magnetite nanolites and a corresponding increase in melt viscosity were important in the eruption preparation process, which then resulted in a large-scale Plinian eruption.

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Keywords: Fukutoku-Oka-no-Ba, Izu-Bonin-Mariana arc, Plinian eruption, drift pumice, nanolite
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#### 45 1. Introduction

Fukutoku-Oka-no-Ba (FOB) volcano is located at 24°17.1'N/141°28.9'E, ~5 km north-east to 46 47 Minami-Ioto Island, south of mainland Japan (Fig. 1a). Five eruptions have been recorded in 1904-05, 48 1914, 1986, 2005, and 2010 (https://www1.kaiho.mlit.go.jp/GIJUTSUKOKUSAI/kaiikiDB/kaiyo24-49 2.htm), and discoloration of the sea surface has been occasionally observed (e.g., Furukawa, 1995). 50 Geochemical analyses and petrographical observations have been conducted on pumice erupted in 1904, 1914, and 1986 (Tsuya, 1937; Yoshida et al., 1987; Kato, 1988; Nakano & Kawanabe, 1992). 51 52 Major element composition the FOB pumices are trachyte similar to the nearby loto volcano, which 53 magma type is rare in the Izu-Ogasawara arc. In contrast, FOB pumices are isotopically distinct from 54 the products of the loto, but similar to the Hiyoshi Volcanic Complex to the south of the FOB (Sun et

55 al., 1998).

56 Detailed topography and preliminary geophysical observations of the relevant area show that there 57 exists a large volcanic complex, which has a size of 15 and 30 km for EW and NW direction,

- respectively, and rises 2000–2200 m above the surrounding ocean floor (Fig. 1b). This complex
- 59 consists of Kita-Fukutoku-Tai, Kita-Fukutoku caldera, and Minami-Ioto volcanoes, from north to
- 60 south (Ito et al., 2011). The Kita-Fukutoku Caldera has east-west and north-south length of 10 and 16
- 61 km, respectively (Fig. 1b). The seismic basement of the caldera has a mortar shape, which is filled by
- 62 low velocity and low density materials (Onodera et al., 2003). Nishizawa et al. (2002) suggested the
- existence of partial melts below the caldera at >1.5-2 km beneath the sea level. FOB is a central cone
- of the Kita-Fukutoku caldera, which have  $\sim 2$  km in diameter at the bottom and the height is  $\sim 200$  m.
- 65 The summit of FOB had an oval shape elongated NE-SW, with the length of 1.5 km and 1 km,
- 66 respectively, and was flat at the depth of ~30 m below sea level before the eruption (Ito et al., 2011).
- 67 The 2021 FOB Plinian eruption occurred from 04:30 (JST) on 13 August, reported by a local fisherman,
- to the morning of 16 August (Japan Meteorological Agency, 2021). The eruption column reached 16
- 69 km in height, the tropopause of the relevant area. The total volume of the erupted pumices was

row estimated to be  $100-500 \times 10^6$  m<sup>3</sup> (Oikawa et al., 2021). These pumices were ejected high in the air,

- fell on the ocean surface and started floating. A large pumice raft was observed by satellite images at 08:00, 3–4 h after the eruption started (Ikegami, 2021). After the eruption, two small islands were observed as "()" shape, then eastern one disappeared within a month (Geospatial Information
- 74 Authority of Japan, 2021).
- 75 Pumice rafting is typically observed once a decade worldwide (e.g., Bryan et al., 2012), where silicic magma erupted explosively beneath the ocean. The 1986 FOB eruption also generated pumice rafts, 76 77 and large amounts of drift pumice clasts arrived at numerous locations, including the Nansei Islands, 78 to where the pumice clasts were transported for ~1300 km by the Kuroshio Counter-current after a 79 duration of >4 months (Fig. 1a; Yoshida et al., 1987; Kato, 1988; Mori et al., 1992). Ocean bottom 80 observatory instruments also drift from the Izu-Bonin arc towards the Nansei Islands, which are ~ 81 1700 km apart (Tada et al., 2021). Ocean current simulations indicate that the drift from the Izu-Bonin 82 arc to the Nansei Islands takes <6 months, but depends on the seasonal current and wind conditions
- 83 (Tada et al., 2021).
- 84 The 2021 FOB pumice rafts travelled westward after the eruption. The RV Keifu-Maru of the Japan
- 85 Meteorological Agency collected samples of floating pumices at 25°30.3'N/138°53.3'E on 22 August
- 86 2021 (Japan Meteorological Agency, 2021). Subsequently, the drift pumice rafts arrived at the Nansei
- 87 Islands in early October, ~2 months after the eruption. The 2021 pumice raft drifted twice as fast as
- the 1986 FOB pumice raft, possibly due to the seasonal change of the Kuroshio Counter-current that
- is thought to be weakened in the winter season (Uchiyama et al., 2016). We undertook comprehensive
- 90 analyses of the *Keifu-Maru* samples and drift pumice clasts collected from several locations on the
- 91 Nansei Islands. Petrographic observations revealed a variety of pumice types originated from the 2021

92 eruption. In the present paper, we describe the textural and geochemical characteristics of the collected

- 93 pumice clasts, and discuss the mechanisms of the 2021 FOB eruption.
- 94

95 2. Methods

96 Mineral compositions were determined with a field emission gun electron microprobe (EPM) analyzer 97 equipped with five wavelength-dispersive X-ray detectors (JXA-8500F; JEOL) at Japan Agency for 98 Marine-Earth Science and Technology (JAMSTEC; Yokosuka, Japan). Natural and synthetic standards 99 were used to calibrate the quantitative analyses. The analytical conditions were 15 kV and 10 nA for 100 the accelerating voltage and beam current, respectively, except for the olivine analyses. For olivine, 101 we used an accelerating voltage of 20 kV and beam current of 25 nA.

102 Raman spectra were obtained with a Raman spectrophotometer (RAMANtouch VIS-HP-MAST; 103 Nanophoton) equipped with a 532 nm semiconductor green laser at JAMSTEC. The laser power on 104 the sample surface was  $\sim$ 2 mW, and data were acquired in 2 × 20 s cycles. The spectrometer was 105 calibrated to the Raman peak of a Si wafer (520.7 cm<sup>-1</sup>).

- 106 Whole-rock major element compositions of the pumice clasts were determined by X-ray fluorescence 107 (XRF) spectrometry (Rigaku ZSX Primus II) following the analytical procedure of Tani et al. (2006) 108 and sample preparation methods of Sato et al. (2020). Prior to analysis, the pumice samples were 109 crushed to pebble size (5-10 mm) and soaked in hot water ( $\sim$ 40 °C) for 0–3 days. The clasts were then 110 repeatedly boiled in Milli-Q water in a microwave oven until addition of a AgNO<sub>3</sub> solution showed 111 that precipitation of AgCl did not occur. After desalinization, all samples were washed with Milli-O water and acetone in an ultrasonic bath, and powdered in an agate mortar or with a Multi-beads 112 113 Shocker pulverizer. Finally, a mixture of 0.4 g of sample powder and 4 g of  $Li_2B_4O_7$  was fused and 114 made into a glass bead for XRF analysis. Accuracy and reproducibility of the major element data are 115 better than  $\pm 1\%$  and  $\pm 2\%$  (relative standard deviations), respectively. We also analyzed trace element 116 composition of whole-rock using solution mode ICP-MS (iCAP Qc, ThermoFisher Scientific). Rock
- 117 powder was digested by acids of HF, HClO4, and HNO3.
- 118 Trace elements of selected melt inclusions and glass in vesiculated groundmass were determined by
- 119 LA-ICP-MS which is a sector-field type inductively coupled plasma-mass spectrometer (Element XR,

120 ThermoFisher Scientific) combined with femto-second laser ablation (FsLA: OK-Fs2000K, OK Lab.)

- 121 installed at JAMSTEC (Kimura and Chang, 2012). Ablated spot is 30 μm in diameter and ~20 μm in
- depth. BCR-2G (basalt standard glass issued by the United States Geological Survey) was used as
- 123 external calibration standard. Any contaminations from surface and proximal phases were checked by
- 124 the time-resolving profiles of the signal and turned out to be negligible. During the analysis, 100%
- 125 normalized major element compositions are also obtained.
- 126 The mass-normalized susceptibility of the pumice clasts was measured with a kappabridge (KLY-4;

- 127 AGICO).
- 128

### 129 3. Field occurrence of the drift pumice clasts

130 Pumice clasts that had drifted to the Nansei Islands were first reported from Kita-daito Island by local 131 residents via Twitter (https://twitter.com/ufuagari jima/status/1445317054043602945) on 5 October 132 2021. The drift pumice clasts were reported on Kikai Island on 10 October, and they continued to other 133 islands located farther west, finally the arrival were reported from Izu Islands and Boso Peninsula on 134 Mid November (Fig. 1a). The first identification of the drift pumice clasts on Kita-daito Island was on 135 5 October, because it was the first day that a ban on coastal access due to high waves caused by a 136 typhoon was lifted. An interview with the local residents suggests that the pumice raft was offshore 137 on 30 September, the day of the typhoon attack (Fig. 1a). On Minami-daido Island, a large amount of 138 drift pumice clasts was also deposited in a coastal area (i.e., Kaigunbo pool). At Kaigunbo pool, some pumice clasts became trapped in crevices up to 1 m above sea level during a normal high tide (Fig. 139 140 2a-b). Other occurrences of pumice clasts include those collected on a rocky beach a short distance 141 from the shoreline which was not tide-related (Fig. 2c). These occurrences suggest that the pumice on 142 Minami-daito Island was washed onshore by storm waves (Goto et al., 2011) and were then protected 143 from the rising tide. The samples from Kita- and Minami-daito islands are relatively small in size (up 144 to 5-10 cm).

In contrast, the pumice clasts on Kikai Island and islands farther west were deposited as "morainelike" features on the shorelines of sandy beaches at high tide (Fig. 2d). At low tide, there were rocks and mudflats on the seaward side of the pumice moraines, but almost no pumice. The amount of pumice deposited varied greatly from beach to beach, possibly due to the orientation of the beach and the direction of waves and winds on the days around when the pumice was deposited. The pumice clasts deposited on the sandy beach are occasionally large (>10 cm).

- The differences in pumice depositional patterns reflect variations in coastal topography. Given that the coasts of Kita- and Minami-daito islands have steep cliffs and no sandy beaches, almost no pumice was deposited these islands. Kikai Island and islands farther west generally have sandy beaches, and drifting materials are easily beached (and subsequently carried away) depending on the direction of
- 155 the wind and tide.
- 156 The drift pumice clasts described below were collected from Kita- and Minami-daito islands and sandy
- 157 beaches of Kikai Island, Amami Oshima, and Okinawa Island (Fig. 1a).
- 158

### 159 4. Pumice classification

160 The pumice clasts collected from the drifting pumice raft by the RV Keifu-Maru (samples 15, 18, and

- 161 19 provided by the JMA, herein referred to as FOB-JMA-15, -18, and -19, respectively) have similar
- 162 characteristics to the drift pumice clasts collected from the Nansei Islands. Notably, the large FOB-
- 163 JMA-18 sample has a highly vesiculated interior (Fig. 3a-b), whereas such highly vesiculated pumice
- 164 was rarely observed in the drift pumice clasts collected from the Nansei Islands. Regardless of the
- 165 deposited locations, the characteristics of drift pumice clasts are similar and they can be classified into
- six types, based on color and texture: gray, black, brown, pale gray, amber, and streaky (Fig. 3c). The
- 167 details of each type are described below.
- **Gray type**: This is the most abundant pumice type (>90%). The drift pumice clasts collected by the RV *Keifu-Maru* (samples FOB-JMA-15, 18, and 19) are also classified as this type. The pumice consists mainly of gray-colored vesicular glass, containing dark-colored fragments (Fig. 3a and c) that
- are termed as black xenoliths (Kato, 1988) or mafic inclusions (Sun et al., 1998), whose appearance
- is sometimes compared to "chocolate-chip cookie." The fragments are a few millimeters to 1 cm in
- size. Hereafter, we refer to these as black enclaves. The vesicles in the groundmass are occasionally
- 174 elongate.
- 175 Black type: This type occurs as either independent black pumice clasts or sub-clasts within the gray
- 176 pumice (Fig. 3c). The independent black pumice clasts are not common, and the sub-clasts in the gray
- pumice are more common. Most black pumice clasts do not contain elongate vesicles or evidence forductile deformation.
- Brown type: This pumice type occurs occasionally as a transitional form of the gray pumice. Thebrown pumice bands are parallel to the elongate groundmass texture.
- Pale gray type: The pale gray pumice has a groundmass that is a much darker gray color as compared
  with the gray type. This type is transitional with the gray pumice.
- 183 Amber type: The amber type pumice has an amber-colored vesicular groundmass that contains coarse
  184 bubbles (up to several millimeters) and is harder than the other types of pumice.
- 185 **Streaky type**: This type of pumice consists of banded gray and black pumice. The bands of black
- 186 pumice (up to 5 mm wide) are generally thinner than those of the gray pumice.
- 187 In addition to the above six pumice types, some pumice clasts have blocky and glassy surfaces that
- 188 possibly formed by quenching.
- 189

### 190 5. Petrography and Geochemistry

191 5.1 Petrography and mineral chemistry

The pumice clasts consist mainly of clinopyroxene (Cpx), plagioclase (Pl), rare olivine (Ol), and a groundmass of vesiculated glass and minor amounts of apatite and opaque minerals (Fig. 4a). Representative mineral and glass analyses are listed in Tables 1–3, and whole-rock compositions

195 determined by XRF spectrometry are listed in Table 4.

196 Two generations of Ol, Cpx, and Pl were recognized based on optical and electron microscopic197 observations.

198 Olivine occurs in two populations. One is phenocrysts or inclusions in Pl phenocrysts in the 199 vesiculated groundmass of the gray pumice, with Mg# values (Mg/[Mg+Fe]  $\times$  100) of ~65 and almost

200 free of NiO. The other type occurs as euhedral phenocrysts in the vesiculated groundmass of black,

- and has Mg# = 92 (Fig. 4c and g), NiO (up to 0.17 mass%), and  $Al_2O_3$  (~0.019 mass%). This high-Mg
- 202 Ol has a low-Mg rim with Mg# =  $\sim$ 80 and, towards the rim, a clear diffusion profile is  $\sim$ 20 µm thick
- 203 (Fig. 4g). High-Mg Ol also occurs in the pale gray pumice, with lower Mg# values of up to 87. A few
- 204 Ol micro-crystals were observed in the glassy groundmass of the black pumice, which are up to  $\sim 5$
- $\mu$ m in diameter and have a similar composition as the low-Mg Ol (Mg# = ~65).

Clinopyroxene occurs as phenocrysts in all pumice types. The Cpx has a diopside (Di) to augite (Aug)
composition (Fig. 4e), with Di cores with higher Mg# values (~95) and Aug rims with lower Mg#
values (~75). Cpx also occurs as micro-crystals in the glassy groundmass of the black and gray pumice,

- and is Aug with Mg# values of  $\sim$ 75.
- 210 Plagioclase occurs as phenocrysts (up to 5 mm in size) in the groundmass of gray pumice, and is
- and esine with Ca-rich cores  $(An_{45})$  and Na-rich rims  $(An_{33})$  (Fig. 4d). Pl in the black pumice is fine-
- grained (<300  $\mu m$ ) and homogeneous, with An\_{40-45}. Some Pl that occurs as white-colored enclaves
- and micro-crystals in the black enclaves has an anorthite composition with An<sub>89–95</sub>. Rare amphibole
- coexists with the anorthitic plagioclase (Fig. 4f), and dendritic crystals within melt inclusions in high-Mg Ol.

216 The opaque minerals are generally Ti-bearing magnetite (Mag), while those occurring as inclusions in

217 Pl and Cpx are rarely Fe-sulfide. High-Mg type Ol occasionally contained chromian spinel (Cr-Spl).

Plagioclase occasionally contains abundant brown-colored melt inclusions (Fig. 4a). Melt inclusions
in low-Mg Ol are brown-colored, while those in high-Mg Ol are colorless (Fig. 4b).

- 220 Black enclaves have two types of occurrences, called type-1 and type-2. Type-1 black enclaves have
- 221 a dense and fine-grained groundmass consisting mainly of Pl, Cpx, and intergranular glass with

222 phenocrysts of Cpx and Ol (Fig. 5a). Type-2 black enclave have an equigranular texture and consist

- 223 of Cpx, Ol, and Pl, with minor amounts of magnetite and intergranular glass (Fig. 5b). Ol phenocrysts
- in type-1 black enclaves are high-Mg and have NiO contents up to 0.12 mass%. Cpx phenocryst in
- type-1 black enclaves is diopside composition with Mg# values of 92-95 that decrease to 83 in the

rims. Pl occurs as fine-grained crystals in the groundmass of type-1 black enclaves, and has high

- anorthite contents of up to An<sub>87</sub>. In contrast, type-2 black enclaves contain Cpx, Ol, and Pl that have
- similar compositions as those in the vesicular groundmass of the gray pumice, except for the cores of
- 229 zoned Pl. Pl in type-2 black enclaves exhibited decrease in anorthite content from core (An<sub>82</sub>) to rim

230 (An<sub>32</sub>) (Fig. 5d and e).

231

#### 232 5.2 Glass and whole-rock geochemical compositions

The texture of vesicles vary in the different types of pumice. Gray pumice comprises colorless glass with the elongate vesicles, whereas black pumice has a relatively undeformed vesicle texture and brown-colored glass (Fig. 6a–c). Amber pumice has a completely different texture, comprising large bubbles and relatively high glass connectivity (Fig. 6d), resulting in its relatively high hardness.

- Vesicular glass in the groundmass is transparent and brown in the black and brown pumices, whereas that in the gray pumice is colorless (Fig. 6a–b). Glass in the pale gray pumice is also colorless, but contains abundant micro-crystals (nanolites) visible under microscope, which are either rectangular or circular in shape (Fig. e–f). These black nanolites were identified as magnetite by Raman microscopy
- circular in shape (Fig. e–f). These black nanolites were identified as magnetite by Raman microscopy
   (Fig. 6g). Brown-colored glass in the pale gray pumice only occurs around or inclusions in phenocrysts.
- 242 Textural characteristics also vary amongst the different types of pumice. Figure 6c shows a scanning
- 243 electron microscopy (SEM) image of the contact between black and gray pumice. The gray pumice
- 244 domain contains highly elongate vesicles as compared with the adjacent black pumice that has a bubble
- 245 aspect ratio of  $\sim 1$ .
- 246 Differences in the brown-colored and colorless glass were further investigated by Raman spectroscopy.
- The Raman spectrum of the brown-colored glass shows a clear peak at 663 cm<sup>-1</sup> that is attributed to magnetite (Fig. 6g), even though no micro-crystals were visible under the microscope. In contrast, the 663 cm<sup>-1</sup> peak did not appear in the spectra of the colorless glass in gray and pale gray pumices.
- 24) 605 cm peak and not appear in the spectra of the colories glass in gray and pare gray particles.
- The glass compositions determined by EMP analyses exhibit trachytic compositions, regardless of the pumice type (Fig. 7a). In particular, the black and gray pumices have very similar compositions, while the amber pumice is relatively enriched in CaO, MgO, and total FeO (Fig. 7b).
- 253 Intragranular melt in type-1 black enclaves has lower SiO<sub>2</sub> contents as compared with the vesiculated
- glass, whereas intragranular melt in type-2 black enclaves has similar compositions to vesiculatedglass.
- 256 Melt inclusions in Pl, augitic Cpx, and low-Mg Ol have similar compositions as the groundmass glass,
- whereas those in diopsidic Cpx and high-Mg Ol have low  $SiO_2$  contents ( $SiO_2 = 50-55$  mass%) and
- are basaltic to basaltic-andesitic in composition (Fig. 7a). Basaltic melt inclusions occasionally contain
- 259 dendritic minerals, which are difficult to analyze. The most SiO<sub>2</sub>-rich melt inclusion was discovered
- in a Aug-Cpx enclosed in Pl (Fig. 4a), and had a rhyolitic composition with  $SiO_2 = 68-69$  mass% (Fig.
- 261 7a).
- 262 Whole-rock compositions of the gray, black, and amber pumice clasts were determined by XRF
- spectrometry, and all had trachytic compositions regardless of pumice type (Fig. 7a). Pale gray and
- 264 brown pumice grade into gray pumice, and we did not undertake separate whole-rock analyses of these
- types. 265
- 266 Selected mafic melt inclusions in type-1 back enclaves and vesicular (trachytic) glass of gray pumice
- are analyzed and shown in spider diagram (Fig. 7c). Groundmass of type-1 black enclave was also

- 268 measured as the mixture of intergrain melt and groundmass minerals. Figure 7c also shows whole-
- 269 rock trace element compositions of gray pumice (FOB-JMA-15, 18, 19). The whole-rock composition
- and the trachytic glass of gray pumice showed similar trace element patterns to that of the 1986
- 271 eruption. In contrast, mafic melt inclusions exhibited different patterns such as positive anomaly of Sr,
- 272 negative anomaly of Pr, Zr, and Hr. Although measured as mixtures, groundmass of type-1 BI
- 273 exhibited intermediate compositions between trachytic glass and mafic inclusions.
- Mass-normalized magnetic susceptibility was determined on the black and gray pumices (Table 5).
  The black pumice had a higher magnetic susceptibility than the gray pumice.
- 276
- 277 6. *P*–*T* calculation
- Table 6 summarizes the coexisting mineral and melt assemblages observed in the drift pumice clasts.
- 279 At least two generations can be clearly identified: (1) those associated with mafic melt, including high-
- 280 Mg Ol and diopsidic Cpx; and (2) those associated with trachytic melt, including low-Mg Ol, augitic
- 281 Cpx, Mag, and Pl (An<sub>44–33</sub>).
- Given that the high-Mg Ol in the black pumice contains Cr-Spl, we applied the Al-in-Ol thermometer (Coogan, 2014). The compositional pair of an Ol core (Mg# = 92) and Cr-Spl of the black pumice (KGB-1) yielded a temperature of 1242 °C.
- Growth conditions of the augitic Cpx were estimated using the magnetite geothermometer (Canil and Lacourse, 2020) for Mag inclusions in Cpx (Fig. 4e) and the Cpx single mineral geobarometer (Petrelli et al., 2020) for associated Cpx. Mag in augitic Cpx (FSD-1) yielded ~930°C while pressures for associated Cpx is ~250 MPa.
- The Cpx single mineral geobarometer applied to Cpx containing a rhyolitic melt inclusion (Fig. 4a) yielded a pressure of 409 MPa, which is slightly higher than that obtained from the augitic Cpx containing trachytic melts.
- 292

### 293 7. Discussion

- 294 7.1 Different color types of the drift pumice clasts
- Despite the different colors of the pumice clasts, whole-rock compositions of the 2021 FOB pumice are similar. The drift pumice clasts from the 1986 eruption also included gray and black (described as
- 297 dark-gray in the literature) pumice with similar whole-rock compositions including the Fe<sub>2</sub>O<sub>3</sub>/FeO
- ratio (Kato, 1988). Raman spectroscopy revealed that the brown-colored glass contained magnetite,
- although no micro-crystals were observed under the microscope (Fig. 6g). Such a Raman signature is
- 300 known to originate from sub-microscopic magnetite nanoparticles (Di Genova et al., 2017; Lerner et
- 301 al., 2021). Based on transmitted electron microscope (TEM) observations of volcanic glass that

- 302 revealed a crystal size gap of crystalline nanoparticles between <30 nm and >1 µm, Mujin et al. (2017)
- 303 defined the term "ultrananolite" for grains smaller than <30 nm and the term "nanolite" for a grain
- 304 size of 30–1000 nm. We did not perform TEM observations and only detected Mag nanoparticles based
- 305 on Raman spectroscopy. As such, we here use the term nanolite to describe the sub-microscopic Mag 306 in our glass samples. The precipitation of Mag nanolites is consistent with the higher magnetic 307 susceptibility of the black pumice as compared with the gray pumice.
- 308 Paulick and Franz (1997) documented very similar characteristics for trachytic pumice in the Meidob 309 volcanic field, Sudan. They measured the Fe<sub>2</sub>O<sub>3</sub>/FeO ratios and magnetic susceptibility for the pumice 310 erupted at 5 ka, which revealed a weak positive correlation between whole-rock Fe<sub>2</sub>O<sub>3</sub> contents and 311 magnetic susceptibility. In their study, glass in dark gray pumice was also transparent and brown, and 312 they concluded that the brown color was caused by sub-microscopic Mag precipitation in the glass. 313 Schlinger et al. (1986, 1988) identified nano-crystals of Fe oxides in volcanic glasses by TEM 314 observations. In samples from southern Nevada, the size of the Fe oxide grains was up to 140 nm for 315 a quenched sample, and greater in the sample that was more slowly cooled (up to 800 nm; Schlinger 316 et al., 1988). Schlinger et al. (1986) performed heating experiments on colorless, precipitate-free, glass 317 shards at 950°C for 5 min, which resulted in darkening of the glass and precipitates forming on the 318 scale of TEM analysis.
- A recent experimental study by Di Genova et al. (2020) revealed that nanolite precipitation is a transient phenomenon that is preserved at a high cooling rate of  $10-20^{\circ}$ C/s, whereas slow cooling allowed microlites to form. Di Genova et al. (2020) suggested that a small amount (~ 4 vol.%) of nanoparticles and a shear rate of  $3.5 \text{ s}^{-1}$  would increase the viscosity by a factor of  $10^2$  within 100 s of nanolite formation. Given that the brown-colored glass of the present study contains nanolites and that this increased its viscosity, this can explain the less deformed texture of the black pumice as compared with the gray pumice (Fig. 6c).
- 326
- 327 7.2 Timescales of magma mixing
- High-Mg Ol found in type-1 black enclaves and black pumice indicated that the mafic magma injection involved in the 2021 FOB explosive eruption. Black pumice clasts are the evidence of heating by the mafic magma of >1200°C and the clear diffusion profile at the rim recorded the timescales of the magma mixing.
- 332 To assess the timescales of the mafic magma injection, diffusion modeling of Fe-Mg zoning in Ol was
- undertaken following the methods of Costa and Dungan (2005) and Viccaro et al. (2016). Diffusion
- 334 coefficients for Fe–Mg in Ol along *c*-axis were calculated following Costa and Chakraborty (2004):

335 
$$D_c^{\text{Fe-Mg}} = 5380 \times \left(\frac{f_{02}}{10^{-12}}\right)^{1/6} \times 10^{3\left(\frac{86-\text{Fo}}{100}\right)} exp\left(\frac{-226000}{8.314 \times T(\text{K})}\right) (1)$$

336 where the diffusion coefficients for the other axis are assumed to be:

$$D_c^{\rm Fe-Mg} \sim 6D_a^{\rm Fe-Mg} \sim 6D_b^{\rm Fe-Mg}$$

Therefore, the pressure dependence of the diffusion coefficients was ignored. The following form of
Fick's second law (in one dimension) with concentration-dependent diffusion coefficients was used
for the diffusion modelling:

(2)

341 
$$\frac{\partial C}{\partial t} = \frac{\partial}{\partial x} \left( D \frac{\partial C}{\partial x} \right)$$
(3)

Given that the high-Mg Ol had a plateau core composition of Mg# = 92 (Fig. 4g), we used this as the initial value. The rim composition of Mg# = 81 (point A in Fig. 4b) was regarded as the boundary condition at the rim for the calculation (i.e., we used a time-invariant constant composition at the rim). Given that the diffusion rim is very narrow, point B yielded a higher Mg# due to a small mis location of the analytical site. We fitted between point A and corresponding side.

In the present study, the crystal orientation was not determined, and thus diffusional anisotropy was not strictly evaluated. The diffusion modeling was performed assuming a direction parallel to the *c*axis and the calculated time could be up to 36 times larger than that determined. Diffusion coefficients were calculated at T = 1242°C (derived from Ol–Cr-Spl thermometry) and  $f_{O2}$  values varying from QFM+1 to +3 following the calibration of Myers and Eugster (1983), although 1242 °C is slightly out of the calibration range.

Figure 4g shows the best-fit model and Figure 7d shows the relationship between the calculated time and  $f_{02}$  values. Assuming that diffusion occurred parallel to the *c*-axis, the calculated time varies from 11 to 24 hours for the  $f_{02}$  range from QFM+1 to +3. Depending on the crystal orientation, the estimated time becomes as long as 50 d.

357

337

#### 358 7.3 Variable pumice types and their role during the explosive eruption

Based on textural and geochemical characteristics, different types of pumice had different roles in the explosive eruption of FOB in 2021. Mitchell et al. (2021) suggested that the pumice clasts form a floating raft and those that suddenly sink to the seafloor have distinct micro-textures (i.e., the floating pumice has a higher vesicle number density and lower pore space connectivity). This could bias the pumice clasts that were sampled. However, the range of drift pumice clasts sampled does partly represent the nature of the 2021 FOB eruption.

- The main gray pumice represents the magma reservoir of the FOB. The texture of type-2 black enclaves suggests an origin from a highly crystalized part of the magma reservoir, such as crystal mush.
- 367 In contrast, type-1 black enclaves and high-Mg olivine in black and pale gray pumice record mafic
- 368 magma involvement. As groundmass of type-1 black enclaves shows intermediate composition
- 369 between trachytic and mafic melts (Fig. 7c), type-1 black enclaves represent the mixing nature of the
- 370 ascending mafic magma and trachytic magma reservoir (Fig. 8). Mafic melt inclusions in both
- abeending mane magina and saving to magina reserven (rig. o). Mane met metasions in cour
- 371 diopsidic Cpx and high-Mg Ol indicate that mafic magma triggered the eruption. Explosive eruption

of silicic magma can be triggered by the cryptic mafic magma injection (e.g., Tamura et al., 2003;
Shukuno et al., 2006; Tamura et al., 2009). Such involvement are sometimes recognized as cooccurrence of bimodal mafic and silicic clastic materials, although drift pumice clasts in this study all
yielded trachytic compositions.

376 In the present case, black pumice could have been heated by injected mafic magma; however, the 377 whole-rock composition does not change and the high-Mg Ol and Di-Cpx phenocrysts (should be 378 called as xenocrysts) only recorded it. Transport process of xenocrysts from the mafic magma to 379 trachytic magma without changing whole-rock compositions remains unclear. When hydrous mafic 380 magma injected into resident felsic crystal-rich mushes, mafic magma dramatically crystallizes due to 381 the water escape into the felsic magma and corresponding change in liquidus temperature (Pistone et 382 al., 2017). The solidification of hot mafic magma essentially releases the latent heat that can enhance 383 the rejuvenation of the crystal mush. Injected mafic magma would either (1) get highly solidified and 384 could not ejected by the eruption, or (2) sink suddenly around the FOB and we cannot obtain such 385 samples from the drift pumice raft.

386 Despite the similar whole-rock geochemical compositions of the gray and black pumice, we rarely 387 observed a gradual transition between them. Adjacent gray and black pumice generally have clear boundaries and distinct textures (Fig. 6c), indicating the black pumice magma was highly viscous prior 388 389 to mingling and that the eruption occurred soon after mingling. The diffusion modeling of Ol also 390 showed a short timescale of black pumice activity, from hours to days (Fig. 7d). Fe oxide nanolites are 391 considered to form due to cooling and/or diffusive H<sub>2</sub>O loss (Danyushevsky et al., 2002; Di Genova 392 et al., 2017, 2018), also indicating short timescales. The presence of magmatic nanolites can enhance 393 heterogeneous bubble nucleation and lead to an explosive eruption of silicic magma (Ceseres et al., 394 2020).

Although the detailed mechanisms of the 2021 FOB Plinian eruption remain unclear, the common and co-occurrence of the nanolite-bearing black pumice within gray pumice might record an important process involved in the explosive eruption. To sum up, ascending mafic magma evidenced by the melt inclusions in high-Mg Ol and diopsidic Cpx heated certain amount of trachytic magma reservoir under

FOB, triggering the 2021 explosive eruption (Fig. 8). More detailed micro-textural observations of the

400 mingled black and gray pumice clasts and/or streaky pumice might provide further insights into the

401 magmatic systems of the FOB and neighboring volcanoes in the Mariana arc.

402

403 Supplementary Materials

404 Trace element analysis data of vesiculated glass, mafic melt inclusions, and whole-rock are available

405 from the online depository materials.

406

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416

#### 417 References

- Bryan, S.E., Cook, A.G., Evans, J.P., Hebden, K., Hurrey, L., Colls, P., Jell, J.S., Weatherly, D., and
  Firn, J. (2012) Rapid, long-distance dispersal by pumice rafting. (2012) Plos one, 7, e40583.
- 420 Canil, D. & Lacourse, T. (2020) Geothermometry using minor and trace elements in igneous and
  421 hydrothermal magnetite. Chemical Geology, 541, 119576.
- 422 Ceseres, F., Wadsworth, F.B., Scheu, B., Colombier, M., Madonna, C., Cimarelli, C., Hess, K.-U.,
  423 Kaliwoda, M., Ruthensteiner, B., & Dingwell, D.B. (2020) Can nanolites enhance eruption
  424 explosivity? Geology, 48, 997-1001.
- 425 Coogan, L.A., Saunders, A.D., & Wilson, R.N. Aluminum-in-olivine thermometry of primitive
  426 basalts: Evidence of an anormalously hot mantle source for large igneous provinces. Chemical
  427 Geology, 368, 1-10.
- 428 Costa, F. & Chakraborty, S. (2004) Decadal time gaps between mafic intrusion and silicic eruption
  429 obtained from chemical zoning patterns in olivine: Earth and Planetary Science Letters, 227, 517–
  430 530.
- 431 Costa, F. & Dungan, M. (2005) Short time scales of magmatic assimilation from diffusion modeling
  432 of multiple elements in olivine. Geology, 33, 837-840.
- Danyushevsky, L.V., McNeill, A.W., and Sobolev, A.V. (2002) Experimental and petrological studies
   of melt inclusions in phenocrysts from mantle-derived magmas: An overview of techniques,
- 435 advantages and complications. Chemical Geology, 183, 5-24.
- 436 Di Genova, D., Sicola, S., Romano, C., Vona, A., Fanara, S., and Spina, L. (2017) Effect of iron and
  437 nanolites on Raman spectra of volcanic glasses: A reassessment of existing strategies to estimate
  438 the water content. Chemical Geology, 475, 76-86.
- Di Genova, D., Caracciolo, A., and Kolzenburg, S. (2018) Measureing the degree of "nanolilization"
   of volcanic glasses: Understanding syn-eruptive processes recorded in melt inclusions. Lithos,

- 441 318, 209-218.
- 442 Di Genova, D., Brooker, R.A., Mader, H.M., Drewitt, J.W.E., Longo, A., Deubener, J., Neuvillem D.R.,
- Fanara, S., Shebanova, O., Anzellini, S., Arzilli, F., Bamber, E.C., Hennet, L., La Spina, G.,
  Miyajima, N. (2020) In situ observation of nanolite growth in volcanic melt: A driving force for
  exprosive eruptions. Science Advances, 6, eabb0413.
- Furukawa, H. (1995) Annual report of world volcanic eruptions in 1992, Fukutoku-oka-no-ba. Bulletin
  of Volcanology, 57, 81-82.
- 448GeospatialInformationAuthorityofJapan(2021)449https://www.gsi.go.jp/uchusokuchi/20210820fukutokuokanoba.html
- Goto, K., Miyagi, K., Kawana, T., Takahashi, J., & Imamura, F. (2011) Emplacement and movement
  of boulders by known storm waves Field evidence from the Okinawa Islands, Japan. Marine
  Geology, 283, 66-78.
- Ikegami, F. (2021) Pumice raft dispersion of Fukutoku-oka-no-ba 2021 eruption. The abstract volume
  of the annual meeting of the Volcanological Society of Japan, A2-02.
- Ito, K., Kato, S., Takahashi, M., Saito, A. (2011) Volcanic topography of Fukutoku-Oka-no-ba volcano
  in Izu-Ogasawara arc after 2010 eruption. Report of Hydrographic and Oceanographic
  Researches, 47, 9-13.
- Japan Meteoric Agency (2021) Monthly volcanic activity reports of Fukutoku-oka-no-ba, 2021 August.
   https://www.data.jma.go.jp/svd/vois/data/tokyo/STOCK/monthly\_v-

460 act\_doc/tokyo/21m08/331\_21m08.pdf (Webpage title was translated by the authors)

- Kato, Y. (1988) Gray pumice drifted from Fukutoku-oka-no-ba to Ryukyu Islands. Bulletin of the
  Volcanological Society of Japan Series 2, 33, 21-30 (in Japanese).
- Kimura, J.-I. & Chang, Q. (2012) Origin of suppressed matrix effect for improved analytical
  performance in determination of major and trace elements in anhydrous silicate samples using
  200 nm femtosecond laser ablation sector-field inductively coupled plasma mass spectrometry.
  Journal of Analytical Atomic Spectrometry, 27, 1549-1559.
- Mitchell, S.J., Fauria, K., Houghton, B.F., & Carey, R.J. (2021) Sink or float; microtextural controls
  on the fate of pumice deposition during the 2012 submarine Harve eruption. Bulletin of
  Volcanology, 83, 80.
- 470 Mori, S., Yamashita, H., & Goto, M. (1992) Drifted pumices from Ogasawara Arc to the coast of
  471 Sagami Bay. Bulletin of the Hiratsuka City Museum, 15, 1-14.
- 472 Mujin, M., Nakamura, M., and Miyake, A. (2017) Eruption style and crystal size distibutions:
  473 Crystallization of groundmass nanolites in the 2011 Shinmoedake eruption. American
  474 Mineralogist, 102, 2367-2380.
- 475 Myers, J. & Eugster, H.P. (1983) The system Fe-Si-O: Oxygen buffer calibrations to 1,500 K.
  476 Contributions to Mineralogy and Petrology, 82, 75-90.

- 477 Nakano, S. & Kawanabe, Y. (1992) Pumices drifted to Iriomote Island in 1991. Bulletin of the
  478 Volcanological Society of Japan, 37, 95-98 (in Japanese).
- Nishizawa, A., Ono, T., Sakamoto, H., Matsumoto, Y., Otani, Y. (2002) Ocean bottom seismographic
  observation at Fukutoku-okanoba submarine volcano. Report of Hydrographic Researches, 38,
  101-123.
- Oikawa, T., Yanagisawa, H., Ikegami, F., Ishizuka, O., Mizuochi, H., Tomiya, A., Morita, M., Nakano,
  S., Kawaguchi, R., & Nakamura, M. (2021) The August 2021 Fukutoku Okanoba Eruption in
  Ogasawara Islands, Japan. The abstract volume of the annual meeting of the Volcanological
  Society of Japan, P1-34.
- 486 Onodera, K., Kato, T., Seo, N. (2003) Crustal structure in the vicinities of Fukutoku-okanoba
  487 submarine volcano estimated from gravity and magnetic anomalies. Report of Hydrographic and
  488 Oceanographic Researches, 39, 23-31.
- Paulick, H. & Franz, G. (1997) The color of pumice: case study on a trachytic fall deposit, Meidob
  volcanic field, Sudan. Bulltine of Volcanology, 59, 171-185.
- 491 Petrelli, M., Caricchi, L., & Perugini, D. (2020) Machine learning thermo-barometry: Application to
  492 clinopyroxene-bearing magmas. Journal of Geophysical Research: Solid Earth, 125,
  493 e2020JB020130.
- 494 Pistone, M., Blundy, J., Brooker, R.A., & EIMF (2017) Water transfer during magma mixing events:
  495 Insights into crystal mush rejuvenation and melt extraction process. American Mineralogist, 102,
  496 766-776.
- 497 Sato, T., Miyazaki, T., Tamura, Y., Gill, J.B., Jutzeler, M., Senda, R., & Kimura, J.-I. The earliest stage
  498 of Izu rear-arc volcanism revealed by drilling at Site U1437, International Ocean Discovery
  499 Program Expedition 350. Island Arc, 2020, e12340
- Schlinger, C.M., Smith, R.M., Veblen, D.R. (1986) Geologic origin of magnetic volcanic glasses in
  the KBS tuff. Geology, 14, 959-962.
- Schlinger, C.M., Rosenbaum, J.G., Veblen, D.R. (1988) Fe-oxide microcrystals in welded tuff from
  southern Nevada: Origin of remanence carriers by precipitation in volcanic glass. Geology, 16,
  556-559.
- 505 Shukuno, H., Tamura, Y., Tani, K., Chang, Q., Suzuki, T. & Fiske, R. S. (2006). Origin of silicic magmas
- and the compositional gap at Sumisu submarine caldera, Izu-Bonin arc, Japan. *Journal of Volcanology*
- 507 and Geothermal Research, 156, 187-216.
- Sun, C.-H. Stern, R., Yoshida, T., & Kimura, J.-I. (1998) Fukutoku-oka-no-ba Volcano: A new
  perspective on the Alkalic Volcano Province in the Izu-Bonin-Mariana arc. The Island Arc, 7,
  432-442.
- Sun, W. & McDonough, W.F. (1989) Chemical and isotopic systematics of oceanic basalts:
   Implications for mantle composition and processes. Geological Society London Special

- 513 Publications, 42, 313-345.
- Tada, N., Nishikawa, H., Ichihara, H., Watanabe Kayama, H., Kuwatani, T. (2021) Ocean Bottom
  Electromagnetometer Carried From Bonin to Ryukyu Islands by Sea Currents.
  doi:10.21203/rs.3.rs-966838/v1
- 517 Tamura, Y., Yuhara, M., Ishii, T., Irino, N. & Shukuno, H. (2003). Andesites and dacites from Daisen
- volcano, Japan: partial-to-total remelting of an andesite magma body. Journal of Petrology, 44, 22432260.
- Tamura, Y., Gill, J. B., Tollstrup, D., Kawabata, H., Shukuno, H., Chang, Q., Miyazaki, T., Takahashi, T.,
  Hirahara, Y., Kodaira, S., Ishizuka, O., Suzuki, T., Kido, Y., Fiske, R. S. & Tatsumi, Y. (2009). Silicic
  magmas in the Izu-Bonin oceanic arc and implications for crustal evolution. Journal of Petrology, 50,
  685-723.
- Tani, K., Kawabata, H., Chang, Q., Sato, K., & Tatsumi, Y. (2006) Quantitative analyses of silicate
   rock major and trace elements by X-ray fluorescence spectrometer: Evaluation of analytical
   precision and sample preparation. Frontier Research on Earth Evolution, 2, 1-8.
- Tsuya, H. (1937) On the volcanics of the Huzi Volcanic zone, with special reference to the geology
  and petrology of the Idu and Southern Islands. Bulletin of Earthquake Research Institute, 15, 215357.
- Uchiyama, Y., Odani, S., Yamanishi, T., Kamidaira, Y., & Mitarai, S. (2016) Impact of mesoscale
  recirculation of the Kuroshio on asymmetric oceanic structure around Okinawa Island. Journal
  of Japan Society of Civil Engineers, Ser. B2 (Coastal Engineering), 72, I 481-I 486.
- Viccaro, M., Giuffrida, M., Nicotra, E., & Cristofolini, R. (2016) Timescales of magma storage and
   migration recorded by olivine crystals in basalts of the March-April 2010 eruption at
   Eyjafjallajökull volcano, Iceland. American Mineralogist, 101, 222-230.
- Yoshida, T., Fujiwara, S., & Aoki, K. (1987) Geochemistry of Fukutoku-oka-no-ba submarine volcano,
  Izu-Ogasawara arc. Research Report of Laboratory of Nuclear Science, Tohoku University, 20,
  202-215 (in Japanese).
- 539

540 Figure Caption

541

542 Figure 1.

543 (a) Index map of the Fukutoku-Oka-no-Ba (FOB) located near the loto Island, south of mainland Japan. 544 Summary of the arrival reports of drift pumice are shown. The track of the pumice raft originated from 545 1986's eruption is also shown. Arrival date are based on Twitter posts for 2021 eruption summarized 546 in https://togetter.com/li/1762225, while those of 1986 eruption are taken from Yoshida et al. (1987), 547 Kato (1988), and Mori et al. (1992). Specific events arranged in chronological order is also 548 summarized. The course of typhoon 16 passed along Japanese Islands on the end of September and 549 should have affected on pumice drifting is also shown (sourced from Japan Meteorological Agency 550 database, https://www.data.jma.go.jp/yoho/typhoon/route map/bstv2021.html). (b) Topographic map 551 around FOB after Ito et al. (2011).

552

553 Figure 2.

554 Field occurrence of the unloaded drift pumice. (a) Kaigunbo artificial pool at the Minami-daito Island. 555 Small drift pumice clasts were observed on the rugged rock on the road, up to approximately 1m height. 556 Pumice clasts on the road were almost removed. (b) Closed view of the orange rectangle of (a), 557 showing remained pumice clasts in the rock crevices. (c) Rocky beach near the Kaigunbo pool. Large 558 number of pumice clasts were collected in the topographic low. The pumice-pooled area was isolated 559 from the sea by a wall-like topographic high of the coastal rocks. (d) Drift pumice clasts unloaded on 560 the sandy beach of the Amami-Oshima Island. Pumice clasts comprised a "pumice moraine" at the 561 high tide shoreline, while clasts did not remain at more seaward rocky area.

562

563 Figure 3

564 Photo of the typical drift pumice clasts. (a) The large sample collected by JMA RV Keifu-Maru (FOB-

565 JMA-18). Compared with the drift sample from the Nansei Islands, rugged surface was preserved. (b)

566 Inside of FOB-JMA-18. Central part is highly vesicular compared to the outer part. (c) Variety of

567 pumice clasts collected from the Nansei Islands. Details are described in the text.

568

569 Figure 4.

570 Photomicrograph and SEM images of the phenocryst minerals in the studied pumice. (a) Cpx and

571 brown-colored melt in Pl phenocryst in brown pumice. Brown-colored melt also surrounded the Pl 572 grain. (b) Colorless mafic melt inclusion in high-Mg Ol occurred in pale-gray pumice. Ol was

- 573 surrounded by trachytic melt. (c) Euhedral high-Mg Ol occurred in black pumice. Outer-most part
- 574 show decrease in Mg#. Cr-Spl is included. A-B line indicates the line-profile measurement shown in

575 Figure 4g. (d) Pl + Cpx phenocryst clot occurred in gray pumice. Pl exhibited  $An_{-44}$  core and  $An_{-35}$ 

576 rim. (e) Cpx phenocryst showing a clear zonation from the diopsidic core to augitic rim. Abundant

577 Mag and melt inclusions were observed in the rim. (f) Amphibole associated with anorthite rich Pl

578  $(An_{-95})$  occurring together with Cpx phenocryst in the gray pumice. (g) Mg# line-profile of the high-

- 579 Mg Ol. (h) Representative Cpx compositions in Ca-Mg-Fe ternary diagram.
- 580

581 Figure 5.

Back-scattered electron images of two-types of black enclave observed in the studied pumice. (a)
Type-1 black enclave exhibiting Cpx phenocrysts embedded in a dense groundmass composed of Pl,
Cpx, opaque minerals, and intergranular melt. (b) Type-2 black enclave exhibiting equigranular texture
of Cpx, Pl, and Ol, filled with intergranular melt. (c) Closed photo of (a) showing fine grains of Cpx,
Pl, and opaque minerals. (d) Zoned Pl occurred in type-2 black enclave, showing rim-ward decrease
in An content. A-B line indicates the position of the line profile shown in (e). (e) A-B line profile of
An content.

- 589
- 590 Figure 6

(a) Photomicrograph of the vesicular glass in gray pumice, almost free from microcrystals. (b) 591 592 Vesicular glass of black pumice exhibits brown color with fine sticky microcrystals of Cpx and rare 593 Ol. (c) Boundary of the black and gray pumice in BSE image. Gray pumice exhibits extensive 594 elongation. (d) Vesicular glass of amber pumice, exhibiting large size bubbles (>500 μm in diameter), 595 and almost colorless. (e) Vesicular glass of pale gray pumice. Abundant nanolites of sticky-shape and small spot are observed (arrowed). (f) BSE image of the glass of pale gray pumice, showing abundant 596 597 nanolites in the glass. (g) Typical Raman spectra of the black nanoliter in the pale gray pumice and 598 other glasses. The brown-colored glass of black pumice shows the magnetite peak at 663 cm<sup>-1</sup> although 599 no crystal is visible under microscope observation.

- 600
- 601 Figure 7.

(a) Total alkali vs SiO<sub>2</sub> diagram of the vesiculated glass and whole-rock of the studied pumice clasts.
Also whole-rock analyses of previous studies are shown. (b) MgO-CaO binary diagram of the vesiculated glass of the studied sample. Legends are similar to those of (a). (c) Spider diagram showing trace element compositions of the whole-rock gray pumice (FOB-JMA-15, 18, and 19) determined by solution ICP-MS, and vesiculated glass and selected mafic melt inclusions in FOB-JMA-18 determined by LA-ICP-MS. N-MORB data are sourced from Sun and McDonough (1989). (d)

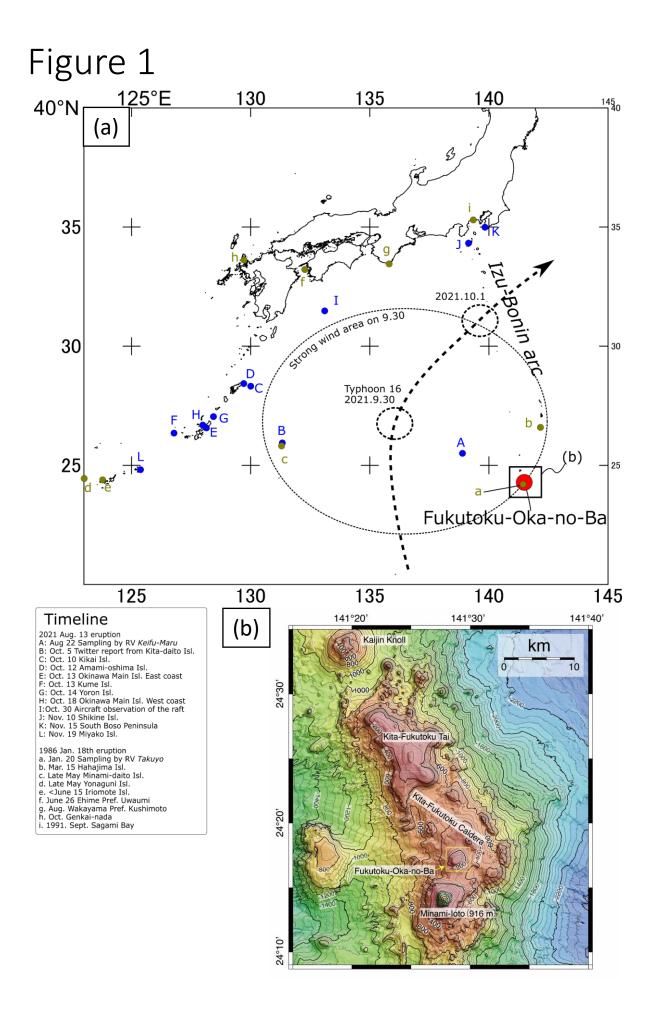
- Relationship between the time calculated from the diffusion modelling and applied  $f_{O2}$  relative to QFM
- 609 buffer.
- 610

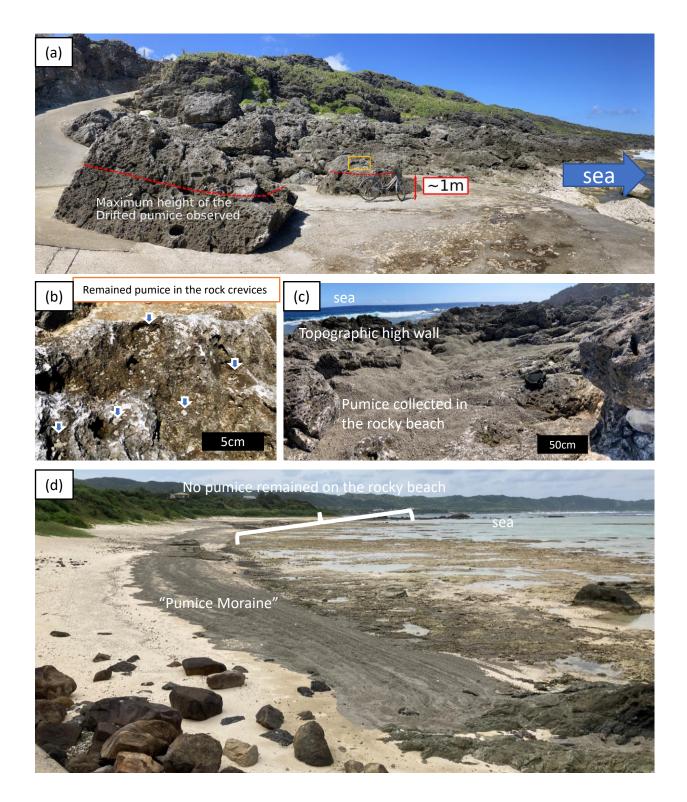
- 611 Figure 8.
- 612 Schematic picture of the FOB 2021 eruption from the preparing stage (A) to the explosive eruption
- 613 stage (C). Details are in text.

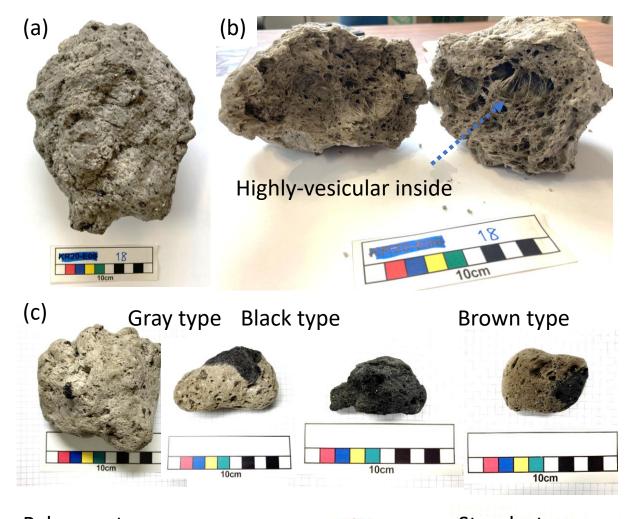
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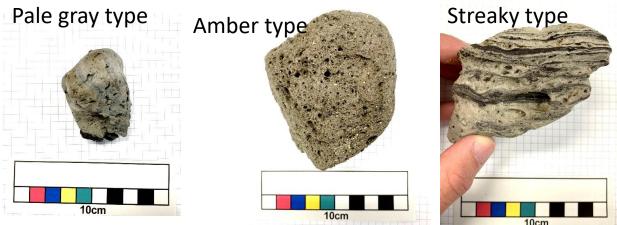
### 615 Tables

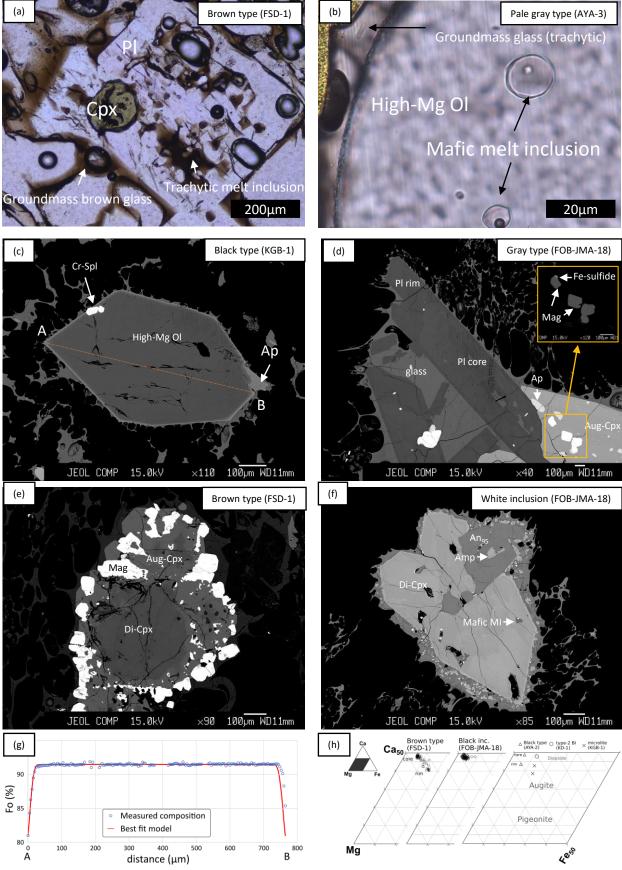
616	Table 1. Representative chemical composition of olivine and clinopyroxene.
617	Footnote
618	FeO*: total iron as FeO. $Fe^{3+}/Fe^{2+}$ for clinopyroxene were calculated so that total cation =4 (O=6
619	basis). B.E.: Black enclave
620	
621	Table 2. Representative chemical composition of plagioclase and opaque minerals.
622	Footnote
623	FeO*: total iron as FeO. $Fe^{3+}/Fe^{2+}$ for opaque minerals were calculated so that sum of $Fe^{2+}$ , Mn, and
624	Mg to be 1 (O=4 basis). Calculations of the Pl endmember were as follows: An = $Ca/(Ca+Na+K) \times 100$ ,
625	$Ab = Na/(Ca+Na+K) \times 100$ , $Or = K/(Ca+Na+K) \times 100$ . B.E.: Black enclave
626	
627	Table 3. Representative chemical composition of volcanic glass determined by EMP analyses.
628	Footnote
629	FeO*: total iron as FeO. B.E.: Black enclave
630	
631	Table 4. XRF whole-rock analyses of the selected pumice samples.
632	
633	Table 5. Mass-normalized susceptibility of gray and black pumice samples.
634	
635	Table 6. Summary of the coexisting mineral and melt assemblages.
636	
637	Table S1. Trace element analyses of glass of inclusions and groundmass, in addition to XRF whole-
638	rock analyses.

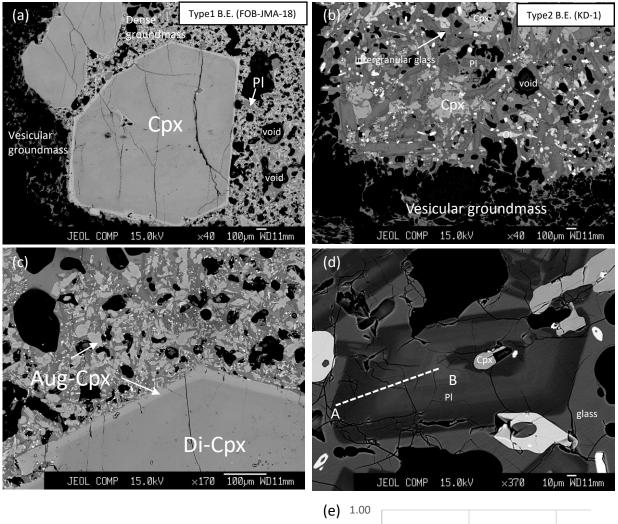


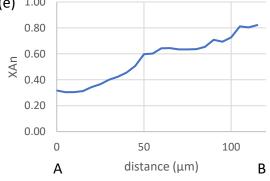


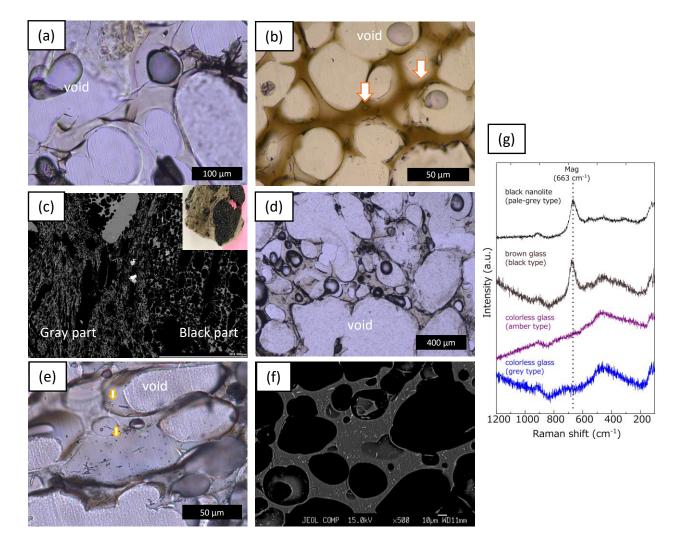


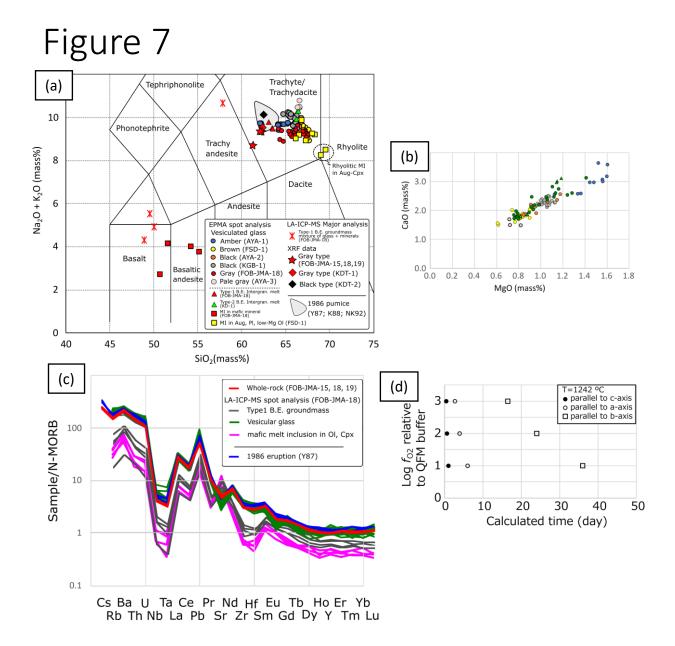












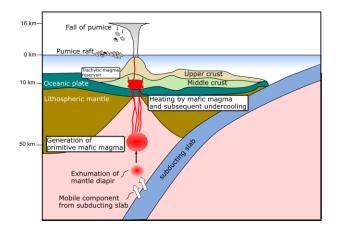


Table 1						
Mineral	01					
pumice type	Gray	Black			Brown	Type-1 B.E.
site	Pacific ocean	Minami-da	aito		Amami	Pacific ocean
sample No.	FOB-JMA-18	KGB-1			FSD-1	FOB-JMA-18
note		core	rim	microlite		
SiO2	35.79	40.22	40.25	34.96	36.37	38.964
TiO2	0.01	0.00	0.01	0.00	0.00	0
AI2O3	0.01	0.019	0.040	0.07	0.00	0.029
Cr2O3	0.00	0.01	0.00	0.00	0.01	0.016
FeO*	30.31	8.18	17.47	29.23	29.66	14.226
MnO	1.56	0.15	0.47	1.26	1.52	0.254
MgO	32.39	50.42	41.79	33.79	32.61	46.032
CaO	0.14	0.34	0.18	0.20	0.15	0.336
Na2O	0.00	0.01	0.00	0.00	0.00	0
K2O	0.00	0.00	0.01	0.05	0.00	0
NiO	0.00	0.17	0.11	0.02	0.00	0.097
Total	100.21	99.53	100.32	99.58	100.32	99.95
0=	4	4	4	4	4	4
Si	0.98		1.02	0.96	0.99	0.98
Ti	0.00	0.00	0.00	0.00	0.00	0.00
AI	0.00		0.00	0.00	0.00	0.00
Cr	0.00	0.00	0.00	0.00	0.00	0.00
Fe3+						
Fe2+	0.69		0.37	0.67	0.67	0.30
Mn	0.04			0.03	0.03	0.01
Mg	1.32		1.58	1.38		
Са	0.00	0.01	0.00	0.01	0.00	0.01
Na	0	0.00	0.00	0.00	0.00	0.00
K	0			0.00		
Ni	0.00	0.002	0.002	0.00	0.00	0.00
	0.00	0.01	0.00	0.04	0.01	0.00
total cation	3.02		2.98	3.04		
Mg#	0.66	0.92	0.81	0.67	0.67	0.67

	Срх						
Type-2 B.E.	Grey		Black		Brown		
Kita-daito	Amami		Minami-da	aito	Amami		
KD-1	AYA-2		KGB-1		FSD-1		
	core	rim	core	microlite	core	rim	host of rhyolitic MI
36.737	50.49	49.30	52.12	50.52	51.45	52.61	53.43
0.035	0.30	0.72	0.27	0.75	0.27	0.39	0.32
0.012	5.10	5.67	3.38	4.26	3.89	1.48	1.13
0	0.00	0.00	0.39	0.02	0.12	0.01	0.00
30.282	6.32	8.92	4.57	8.76	4.28	9.35	9.91
1.498	0.10	0.19	0.00	0.45	0.11	0.83	0.86
32.21	14.62	13.30	15.56	14.52	16.06	15.35	15.26
0.17	23.21	21.34	23.15	19.63	23.76	19.71	19.50
0.011	0.22	0.33	0.10	0.41	0.07	0.45	0.40
0.012	0.02	0.01	0.02	0.03	0.02	0.05	0.00
0.005							
100.97	100.38	99.79	99.57	99.36	100.03	100.21	100.82
4	6	6	6	6	6	6	6
0.99	1.85	1.83	1.92	1.88	1.88	1.95	1.97
0.00	0.01	0.02	0.01	0.02	0.01	0.01	0.01
0.00	0.22	0.25	0.15	0.19	0.17	0.06	0.05
0.00	0.00	0.00	0.01	0.00	0.00	0.00	0.00
	0.08	0.06	0.00	0.04	0.07	0.06	0.02
0.68	0.11	0.21	0.14	0.24	0.07	0.23	0.29
0.03	0.00	0.01	0.00	0.01	0.00	0.03	0.03
1.29	0.80	0.74	0.85	0.81	0.87	0.85	0.84
0.00	0.91	0.85	0.91	0.78	0.93	0.78	0.77
0.00	0.02	0.02	0.01	0.03	0.01	0.03	0.03
0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00
0.00							
3.01	4.00	4.00	4.00	4.00	4.00		
0.67	0.88	0.78	0.86	0.77	0.93	0.79	0.75

Туре-1 В	.E.		Type-2 B.E.	
Pacific of	cean		Kita-daito	
FOB-JMA	\-18		KD-1	
core	rim			
53.60	C	49.23	49.53	
0.20	C	0.43	0.76	
1.63	3	4.94	5.49	
0.24	4	0.07	0.08	
3.2	1	8.46	9.45	
0.0	ô	0.07	0.42	
17.54	4	14.63	13.04	
24.24	4	22.06	21.55	
0.12	2	0.16	0.39	
0.02	2	0.01	0.00	
100.8 <sup>-</sup>	7	100.05	100.71	
(	ô	6	6	
1.93	3	1.82	1.83	
0.0	1	0.01	0.02	
0.0	7	0.22	0.24	
0.0	1	0.00	0.00	
0.0	ō	0.13	0.09	
0.0	ō	0.13	0.20	
0.0	C	0.00	0.01	
0.94	4	0.81	0.72	
0.94	4	0.87	0.85	
0.0	1	0.01	0.03	
0.0	C	0.00	0.00	
4.00	C	4.00	4.00	
0.9	5	0.86	0.78	

TiO2         0.05         0.00         0.05         0.00         0.05         0.03         0           Al2O3         26.60         24.72         26.54         24.95         33.78         32.77         24.           Cr2O3         0.12         0.00         0.00         0.01         0.00         0.03         0           FeO*         0.56         0.46         0.55         0.58         1.05         0.81         0           MnO         0.06         0.00         0.12         0.00         0.01         0.05         0           MgO         0.02         0.05         0.04         0.04         0.14         0.03         0           CaO         9.00         7.30         9.06         7.09         17.35         16.24         6           Na2O         5.81         6.54         5.78         6.57         1.39         1.89         6           K2O         0.65         1.06         0.68         1.11         0.03         0.10         1           Total         100.18         99.98         100.73         100.52         100.07         100.24         100           Cr         0.00         0.00         0.00	Table 2							
Site         Pacific ocean         Minami-di Amami         Pacific ocean         Kita-daito           sample No.         FOB-JMA-18         KGB-1         FSD-1         FOB-JMA-18         KD-1           note         core         rim         core         rim           SiO2         57.33         59.86         57.91         60.18         46.27         48.30         61           TiO2         0.05         0.00         0.05         0.00         0.05         0.03         0           Al2O3         26.60         24.72         26.54         24.95         33.78         32.77         24           Cr2O3         0.12         0.00         0.00         0.01         0.00         0.03         0           MnO         0.06         0.00         0.12         0.00         0.01         0.05         0           MgO         0.02         0.05         0.04         0.04         0.14         0.03         0           CaO         9.00         7.30         9.06         7.09         17.35         16.24         6           Na2O         5.81         6.54         5.78         6.57         1.39         1.89         6           K2O		PI						
Sample No.         FOB-JMA-18         KGB-1         FSD-1         FOB-JMA-18 KD-1           note         core         rim         core         rim           SiO2         57.33         59.86         57.91         60.18         46.27         48.30         61.           TiO2         0.05         0.00         0.05         0.00         0.05         0.03         0           Al2O3         26.60         24.72         26.54         24.95         33.78         32.77         24           Cr2O3         0.12         0.00         0.01         0.00         0.03         0           MnO         0.06         0.00         0.12         0.00         0.01         0.05         0           MgO         0.02         0.05         0.04         0.04         0.14         0.03         0           CaO         9.00         7.30         9.06         7.09         17.35         16.24         6           Na2O         5.81         6.54         5.78         6.57         1.39         1.89         6           K2O         0.65         1.06         0.68         1.11         0.03         0.0         0         0           Y </td <td>pumice type</td> <td>Gray</td> <td></td> <td>Black</td> <td>Brown</td> <td>Type-1 B.E.</td> <td>Type-2 B.I</td> <td>Ξ.</td>	pumice type	Gray		Black	Brown	Type-1 B.E.	Type-2 B.I	Ξ.
note         core         rim         core         rim           SiO2         57.33         59.86         57.91         60.18         46.27         48.30         61           TiO2         0.05         0.00         0.05         0.00         0.05         0.03         0           Al2O3         26.60         24.72         26.54         24.95         33.78         32.77         24           Cr2O3         0.12         0.00         0.00         0.01         0.00         0.03         0           FeO*         0.56         0.46         0.55         0.58         1.05         0.81         0           MnO         0.06         0.00         0.12         0.00         0.01         0.05         0           MgO         0.02         0.05         0.04         0.04         0.14         0.03         0           CaO         9.00         7.30         9.06         7.09         17.35         16.24         6           Na2O         5.81         6.54         5.78         6.57         1.39         1.89         6           K2O         0.65         1.06         0.68         1.11         0.03         0.00	site	Pacific ocear	1	Minami-d	Amami	Pacific ocean	Kita-daito	
SiO2 $57.33$ $59.86$ $57.91$ $60.18$ $46.27$ $48.30$ $61.7$ TiO2 $0.05$ $0.00$ $0.05$ $0.00$ $0.05$ $0.03$ $0.65$ Al2O3 $26.60$ $24.72$ $26.54$ $24.95$ $33.78$ $32.77$ $24.72$ Cr2O3 $0.12$ $0.00$ $0.00$ $0.01$ $0.00$ $0.03$ $0.72$ FeO* $0.56$ $0.46$ $0.55$ $0.58$ $1.05$ $0.81$ $0.72$ MnO $0.06$ $0.00$ $0.12$ $0.00$ $0.01$ $0.05$ $0.91$ MgO $0.02$ $0.05$ $0.04$ $0.04$ $0.14$ $0.03$ $0.73$ CaO $9.00$ $7.30$ $9.06$ $7.09$ $17.35$ $16.24$ $6.73$ Na2O $5.81$ $6.54$ $5.78$ $6.57$ $1.39$ $1.89$ $6.73$ K2O $0.65$ $1.06$ $0.68$ $1.11$ $0.03$ $0.10$ $1.12$ Total $100.13$ $99.98$ $100.73$ $100.52$ $100.07$ $100.24$ $100$ C= $8$ $8$ $8.00$ $8.00$ $8.00$ $8.00$ $8.00$ $8.00$ Si $2.57$ $2.68$ $2.59$ $2.68$ $2.13$ $2.21$ $2.72$ Ti $0.00$ $0.00$ $0.00$ $0.00$ $0.00$ $0.00$ $0.00$ Al $1.41$ $1.30$ $1.40$ $1.31$ $1.84$ $1.77$ $1.77$ Cr $0.00$ $0.00$ $0.00$ $0.00$ $0.00$ $0.00$ <t< td=""><td>sample No.</td><td>FOB-JMA-18</td><td>8</td><td>KGB-1</td><td>FSD-1</td><td>FOB-JMA-18</td><td>KD-1</td><td></td></t<>	sample No.	FOB-JMA-18	8	KGB-1	FSD-1	FOB-JMA-18	KD-1	
TiO2         0.05         0.00         0.05         0.00         0.05         0.03         0           Al2O3         26.60         24.72         26.54         24.95         33.78         32.77         24           Cr2O3         0.12         0.00         0.00         0.01         0.00         0.03         0           FeO*         0.56         0.46         0.55         0.58         1.05         0.81         0           MnO         0.06         0.00         0.12         0.00         0.01         0.05         0           MgO         0.02         0.05         0.04         0.04         0.14         0.03         0           CaO         9.00         7.30         9.06         7.09         17.35         16.24         6           Na2O         5.81         6.54         5.78         6.57         1.39         1.89         6           K2O         0.65         1.06         0.68         1.11         0.03         0.10         1           Total         100.18         99.98         100.73         100.52         100.07         100.24         100           Cr         0.00         0.00         0.00	note	core rir	n				core	rim
TiO2         0.05         0.00         0.05         0.00         0.05         0.03         0           Al2O3         26.60         24.72         26.54         24.95         33.78         32.77         24           Cr2O3         0.12         0.00         0.00         0.01         0.00         0.03         0           FeO*         0.56         0.46         0.55         0.58         1.05         0.81         0           MnO         0.06         0.00         0.12         0.00         0.01         0.05         0           MgO         0.02         0.05         0.04         0.04         0.14         0.03         0           CaO         9.00         7.30         9.06         7.09         17.35         16.24         6           Na2O         5.81         6.54         5.78         6.57         1.39         1.89         6           K2O         0.65         1.06         0.68         1.11         0.03         0.10         1           Total         100.18         99.98         100.73         100.52         100.07         100.24         100           Cr         0.00         0.00         0.00	SiO2	57.33	59 86	57 91	60.18	46 27	48.30	61.17
Al2O3       26.60       24.72       26.54       24.95       33.78       32.77       24.         Cr2O3       0.12       0.00       0.00       0.01       0.00       0.03       0         FeO*       0.56       0.46       0.55       0.58       1.05       0.81       0         MnO       0.06       0.00       0.12       0.00       0.01       0.05       0         MgO       0.02       0.05       0.04       0.04       0.14       0.03       0         CaO       9.00       7.30       9.06       7.09       17.35       16.24       6         Na2O       5.81       6.54       5.78       6.57       1.39       1.89       6         K2O       0.65       1.06       0.68       1.11       0.03       0.10       1         Total       100.18       99.98       100.73       100.52       100.07       100.24       100         O=       8       8       8.00       8.00       8.00       8.00       8         Si       2.57       2.68       2.59       2.68       2.13       2.21       2         Ti       0.00       0.00       0.00								0.03
Cr2O3         0.12         0.00         0.00         0.01         0.00         0.03         0           FeO*         0.56         0.46         0.55         0.58         1.05         0.81         0           MnO         0.06         0.00         0.12         0.00         0.01         0.05         0           MgO         0.02         0.05         0.04         0.04         0.14         0.03         0           CaO         9.00         7.30         9.06         7.09         17.35         16.24         6           Na2O         5.81         6.54         5.78         6.57         1.39         1.89         6           K2O         0.65         1.06         0.68         1.11         0.03         0.10         1           Total         100.18         99.98         100.73         100.52         100.07         100.24         100           C         0         0.00         0.00         0.00         0.00         0.00         0           C         0         0.00         0.00         0.00         0.00         0.00         0           C         0         0.00         0.00         0.00 <t< td=""><td></td><td></td><td></td><td></td><td></td><td></td><td></td><td>24.29</td></t<>								24.29
$\begin{array}{cccccccccccccccccccccccccccccccccccc$								0.03
MnO         0.06         0.00         0.12         0.00         0.01         0.05         0           MgO         0.02         0.05         0.04         0.04         0.14         0.03         0           CaO         9.00         7.30         9.06         7.09         17.35         16.24         6           Na2O         5.81         6.54         5.78         6.57         1.39         1.89         6           K2O         0.65         1.06         0.68         1.11         0.03         0.10         1           Total         100.18         99.98         100.73         100.52         100.07         100.24         100           O=         8         8         8.00         8.00         8.00         8.00         8           Si         2.57         2.68         2.59         2.68         2.13         2.21         2           Ti         0.00         0.00         0.00         0.00         0.00         0.00         0.00           Al         1.41         1.30         1.40         1.31         1.84         1.77         1           Cr         0.00         0.00         0.00         0.00								0.56
MgO         0.02         0.05         0.04         0.04         0.14         0.03         0           CaO         9.00         7.30         9.06         7.09         17.35         16.24         6           Na2O         5.81         6.54         5.78         6.57         1.39         1.89         6           K2O         0.65         1.06         0.68         1.11         0.03         0.10         1           Total         100.18         99.98         100.73         100.52         100.07         100.24         100           O=         8         8         8.00         8.00         8.00         8.00         8           Si         2.57         2.68         2.59         2.68         2.13         2.21         2           Ti         0.00         0.00         0.00         0.00         0.00         0.0         0           AI         1.41         1.30         1.40         1.31         1.84         1.77         1           Cr         0.00         0.00         0.00         0.00         0.00         0.0         0           Fe3+         0.00         0.02         0.02         0.02 <t< td=""><td></td><td></td><td></td><td></td><td></td><td></td><td></td><td>0.00</td></t<>								0.00
CaO         9.00         7.30         9.06         7.09         17.35         16.24         6           Na2O         5.81         6.54         5.78         6.57         1.39         1.89         6           K2O         0.65         1.06         0.68         1.11         0.03         0.10         1           Total         100.18         99.98         100.73         100.52         100.07         100.24         100           O=         8         8         8.00         8.00         8.00         8.00         8           Si         2.57         2.68         2.59         2.68         2.13         2.21         2           Ti         0.00         0.00         0.00         0.00         0.00         0.00         0.00           Al         1.41         1.30         1.40         1.31         1.84         1.77         1           Cr         0.00         0.00         0.00         0.00         0.00         0.00         0.00           Fe2+         0.02         0.02         0.02         0.02         0.04         0.03         0           Mn         0.00         0.00         0.00         0.00								0.03
K2O         0.65         1.06         0.68         1.11         0.03         0.10         1           Total         100.18         99.98         100.73         100.52         100.07         100.24         100           O=         8         8         8.00         8.00         8.00         8.00         8.00         8           Si         2.57         2.68         2.59         2.68         2.13         2.21         2           Ti         0.00         0.00         0.00         0.00         0.00         0.00         0.00           Al         1.41         1.30         1.40         1.31         1.84         1.77         1           Cr         0.00         0.00         0.00         0.00         0.00         0.00         0.00           Fe3+         0.00         0.00         0.00         0.00         0.00         0.00         0.00         0.00           Mn         0.00         0.00         0.00         0.00         0.00         0.00         0.00           Ma         0.51         0.57         0.50         0.57         0.12         0.17         0	-							6.17
Total         100.18         99.98         100.73         100.52         100.07         100.24         100           O=         8         8         8.00         8.00         8.00         8.00         8.00         8           Si         2.57         2.68         2.59         2.68         2.13         2.21         2           Ti         0.00         0.00         0.00         0.00         0.00         0.00         0.00           Al         1.41         1.30         1.40         1.31         1.84         1.77         1           Cr         0.00         0.00         0.00         0.00         0.00         0.00         0.00           Fe3+         0.00         0.02         0.02         0.02         0.04         0.03         0           Mn         0.00         0.00         0.00         0.00         0.00         0.00         0.00           Mg         0.00         0.00         0.00         0.00         0.01         0.00         0           Ma         0.51         0.57         0.50         0.57         0.12         0.17         0	Na2O	5.81	6.54	5.78	6.57	1.39	1.89	6.50
O=         8         8         8.00         9.00 <td>K20</td> <td>0.65</td> <td>1.06</td> <td>0.68</td> <td>1.11</td> <td>0.03</td> <td>0.10</td> <td>1.29</td>	K20	0.65	1.06	0.68	1.11	0.03	0.10	1.29
Si2.572.682.592.682.132.212Ti0.000.000.000.000.000.000.000.00Al1.411.301.401.311.841.771Cr0.000.000.000.000.000.000.00Fe3+0.000.020.020.020.020.000.00Fe2+0.020.020.020.020.040.030Mn0.000.000.000.000.000.000Mg0.000.000.000.340.860.800Na0.510.570.500.570.120.170	Total	100.18	99.98	100.73	100.52	100.07	100.24	100.06
Si2.572.682.592.682.132.212Ti0.000.000.000.000.000.000.000.00Al1.411.301.401.311.841.771Cr0.000.000.000.000.000.000.00Fe3+0.000.020.020.020.020.000.00Fe2+0.020.020.020.020.040.030Mn0.000.000.000.000.000.000Mg0.000.000.000.340.860.800Na0.510.570.500.570.120.170								
Ti0.000.000.000.000.000.000.000.00Al1.411.301.401.311.841.771Cr0.000.000.000.000.000.000.000Fe3+0.000.000.000.000.000.000.000Fe2+0.020.020.020.020.040.030Mn0.000.000.000.000.000.000Mg0.000.000.000.340.860.800Na0.510.570.500.570.120.170	0=	8	8	8.00	8.00	8.00	8.00	8.00
Al1.411.301.401.311.841.771Cr0.000.000.000.000.000.000.000.00Fe3+0.000.000.000.000.000.000.000.00Fe2+0.020.020.020.020.020.000.000.000.00Mn0.000.000.000.000.000.000.000.000.000.00Mg0.000.000.000.000.010.000.000.00Na0.510.570.500.570.120.170.50	Si	2.57	2.68	2.59	2.68	2.13	2.21	2.72
Cr0.000.000.000.000.000.000.000.00Fe3+0.000.000.000.000.000.000.000.000.00Fe2+0.020.020.020.020.040.030Mn0.000.000.000.000.000.000.000Mg0.000.000.000.000.010.000Ca0.430.350.430.340.860.800Na0.510.570.500.570.120.170	Ti	0.00	0.00	0.00	0.00	0.00	0.00	0.00
Fe3+0.000.000.000.000.000.000.00Fe2+0.020.020.020.020.020.040.030Mn0.000.000.000.000.000.000.000.000Mg0.000.000.000.000.000.010.000Ca0.430.350.430.340.860.800Na0.510.570.500.570.120.170	AI	1.41	1.30	1.40	1.31	1.84	1.77	1.28
Fe2+0.020.020.020.020.030Mn0.000.000.000.000.000.000.00Mg0.000.000.000.000.010.000Ca0.430.350.430.340.860.800Na0.510.570.500.570.120.170	Cr	0.00	0.00	0.00	0.00	0.00	0.00	0.00
Mn0.000.000.000.000.000.000.00Mg0.000.000.000.000.010.000.00Ca0.430.350.430.340.860.800.00Na0.510.570.500.570.120.170.00	Fe3+	0.00	0.00	0.00	0.00	0.00	0.00	0.00
Mg0.000.000.000.000.010.000Ca0.430.350.430.340.860.800Na0.510.570.500.570.120.170	Fe2+	0.02	0.02	0.02	0.02	0.04	0.03	0.02
Ca0.430.350.430.340.860.800.43Na0.510.570.500.570.120.170.43	Mn	0.00	0.00	0.00	0.00	0.00	0.00	0.00
Na 0.51 0.57 0.50 0.57 0.12 0.17 0.	Mg	0.00	0.00	0.00	0.00	0.01	0.00	0.00
	Са	0.43	0.35	0.43	0.34	0.86	0.80	0.29
K 0.04 0.06 0.04 0.06 0.00 0.01 0	Na	0.51	0.57	0.50	0.57	0.12	0.17	0.56
	К	0.04	0.06	0.04	0.06	0.00	0.01	0.07
total cation 4.99 4.98 4.98 4.98 5.01 4.99 4	total cation	4.99	4.98	4.98	4.98	5.01	4.99	4.95
An 44 36 45 35 87 82								32
Ab 52 58 51 59 13 17		52						60
Or 4 6 4 7 0 1							1	8

Opaque	Cr-Spl	Mag
	Black	Brown
	Minami-daito	Amami
	KGB-1	FSD-1
	in high-Mg Ol	in Cpx
	0.09	0.14
	0.25	10.46
	8.34	2.91
	58.20	0.11
	21.82	77.73
	0.31	0.87
	12.54	2.92
	0.02	0.00
	0.00	0.00
	0.00	0.00
	101.57	95.14
	4	4
	0.00	0.00
	0.01	0.28
	0.32	0.12
	1 /0	0.00

	0.01	0.28
	0.32	0.12
	1.48	0.00
	0.19	1.49
	0.39	0.82
	0.01	0.03
	0.60	0.15
	0.00	0.00
	0	0.00
	0	0.00
Cr#	0.74	
Mg#	0.60	0.15

Table 3							
pumice type Gray Black Amber Pale gray Brown							
site	Amami	Minami-daito	Amami	Amami	Amami		
sample No.	AYA2	KGB1	AYA1	АҮАЗ	FSD-1		
note					MI in Cpx	MI in Iow- Mg OI	MI in PI
SiO2	65.63	65.22	64.69	65.56	67.67	65.89	66.75
TiO2	0.52	0.49	0.46	0.53	0.55	0.50	0.37
AI2O3	16.10	16.12	16.05	15.89	16.05	16.15	15.78
Cr2O3	0.00	0.04	0.01	0.00	0.03	0.00	0.02
FeO*	3.77	3.83	4.37	3.72	3.47	3.71	3.34
MnO	0.05	0.09	0.33	0.18	0.02	0.14	0.17
MgO	0.90	1.11	1.56	0.81	0.34	0.56	0.72
CaO	2.15	2.24	3.07	1.46	0.95	1.94	1.57
Na2O	4.70	4.85	4.82	4.91	4.30	4.67	4.57
K2O	4.74	4.75	4.90	5.37	3.98	5.09	4.43
P2O5	0.11	0.14	0.17	0.19	0.07	0.18	0.07
F	0.09	0.09	0.14	0.12	0.14	0.13	0.16
CI	0.31	0.30	0.29	0.36	0.35	0.39	0.33
Total	98.65	98.86	100.43	98.61	97.42	99.33	97.77

Type-1 B.E. Type-2 B.E.							
Pacific oc	ean		Kita-daito				
FOB-JMA-	-18		KD-1				
MI in Cpx	MI in high- Mg Ol	intergranular	intergranular				
51.92	48.93	62.63	65.53				
0.53	0.57	0.33	0.41				
20.91	16.39	16.12	16.29				
0.08	0.00	0.00	0.00				
5.59	9.04	5.83	3.95				
0.25	0.18	0.22	0.21				
3.43	3.96	1.13	0.87				
9.17	14.87	2.97	1.85				
2.91	1.78	4.54	4.74				
0.91	0.82	4.84	5.05				
n.a.	n.a.	0.51	0.20				
0.04	0.02	0.21	0.11				
0.14	0.09	0.44	0.32				
95.68	96.54	99.11	99.11				

Table 4						
Pumice type	Gray	Gray	Gray	Gray	Black	Amber
site	Pacific ocean	Pacific ocean	Pacific ocean	Kita-daito	Kita-daito	Amami
Sample No.	FOB-JMA-15	FOB-JMA-18	FOB-JMA-19	KD-FOB1	KD-FOB2	AYA-01
SiO <sub>2</sub>	60.24	60.50	59.60	60.81	60.85	60.61
TiO <sub>2</sub>	0.58	0.58	0.56	0.57	0.61	0.57
$AI_2O_3$	16.28	15.94	15.35	16.31	16.28	15.52
$Fe_2O_3$	5.33	5.37	5.55	5.20	5.48	5.66
MnO	0.17	0.17	0.17	0.17	0.18	0.17
MgO	1.93	2.09	3.03	1.82	1.42	3.04
CaO	3.92	3.79	5.03	3.59	3.03	4.49
Na <sub>2</sub> O	4.70	4.70	4.33	4.82	5.14	4.55
K <sub>2</sub> O	4.34	4.42	4.10	4.43	4.65	4.27
$P_{2}O_{5}$	0.23	0.23	0.21	0.23	0.28	0.22
Total	97.70	97.78	97.93	97.94	97.91	99.10
LOI	1.38	1.47	1.58	0.55	0.78	0.04

Table 5				
	weight (g)	bulk mag. Sus. (cr	m3) M	IS (cm3/g)
Gray type	2.996		0.022	7.44E-03
Black type	4.857		0.045	9.23E-03

Tabl	le 6
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	Mafic member	Trachytic member
	Type-1 black inclusion	major member of pumice
Occurrence	xenocryst in black-/ pale gray-	
	type	clasts
melt SiO2	48-55 mass%	62-70 mass%
OI Mg#	85-92	~65
Cpx composition	diopside	augite
PI composition	An80-95	An33-44
Other minerals	amphibole	magnetite