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2 ABSTRACT

3

4 Explosive volcanic eruptions can inject sulfur dioxide (SO₂) into the stratosphere to form aerosol
5 particles that modify Earth's radiation balance and drive surface cooling. Eruptions involving
6 interactions with shallow layers (≤ 500 m) of surface water and ice modify the eruption dynamics
7 that govern the delivery of SO₂ to the stratosphere. External surface water controls the evolution
8 of explosive eruptions in two ways that are poorly understood: (1) by modulating the hydrostatic
9 pressure within the conduit and at the vent, and (2) through the ingestion and mixing of external
10 water, which governs fine ash production and eruption column buoyancy flux. To make progress,
11 we couple one-dimensional models of conduit flow and atmospheric column rise through a novel
12 "magma-water interaction" model that simulates the occurrence, extent and consequences of
13 water entrainment depending on the depth of a surface water layer. We explore the effects of
14 hydrostatic pressure on magma ascent in the conduit and gas decompression at the vent, and
15 the conditions for which water entrainment drives fine ash production by quench fragmentation,
16 eruption column collapse, or outright failure of the jet to breach the water surface. We show
17 that the efficiency of water entrainment into the jet is the predominant control on jet behavior.
18 For an increase in water depth of 50 to 100 m, the critical magma mass eruption rate required
19 for eruption columns to reach the tropopause increases by an order of magnitude. Finally, we
20 estimate that enhanced emission of fine ash leads to up to a 2-fold increase in the mass flux of
21 particles $< 125 \mu\text{m}$ to spreading umbrella clouds, together with up to a 10-fold increase in water
22 mass flux, conditions that can enhance the removal of SO₂ via chemical scavenging and ash
23 sedimentation. Overall, compared to purely magmatic eruptions, we suggest that hydrovolcanic
24 eruptions will be characterized by a reduced delivery of SO₂ to the stratosphere. Our results
25 thus suggest the possibility of an unrecognized volcano-climate feedback mechanism arising
26 from modification of volcanic climate forcing by direct interaction of erupting magma with varying
27 distributions of water and ice at the Earth's surface.

28 **Keywords:** External forcing, Hydrovolcanism, magma-water interactions, Explosive eruptions, 1D plume model, 1D conduit flow
29 model, Stratospheric sulfur input, Climate feedback

1 INTRODUCTION

30 Volcanic SO₂ injected into the stratosphere forms sulfate aerosols that persist for 1-3 years, affect Earth's
31 radiation balance and produce one of the strongest natural surface climate cooling mechanisms (Timmreck,
32 2012; Sigl et al., 2015; Kremser et al., 2016). Although the direct radiative forcing from volcanic aerosols
33 typically acts over annual to decadal timescales (Robock, 2000), the last decade of research has shown
34 that the climate impacts of eruptions are not restricted to discrete and intermittent cooling events with
35 durations of a few years. For example, volcanic emission from small to moderate eruptions and passive
36 degassing provide background concentrations of sulfate aerosols, resulting in a near-continuous negative
37 (cooling) forcing to the planetary surface (Solomon et al., 2011; Schmidt et al., 2012; Santer et al., 2014).
38 Furthermore, a growing body of evidence suggests that volcanic forcing from aerosols can also drive
39 non-linear climate responses on multidecadal to millennial timescales (Zhong et al., 2011; Schleussner
40 and Feulner, 2013; Zanchettin et al., 2013; Santer et al., 2014; Baldini et al., 2015; Toohey et al., 2016;
41 Soreghan et al., 2019; Mann et al., 2021). The strength of aerosol climate forcing depends strongly on the
42 SO₂ mass flux to the stratosphere (e.g. Marshall et al. (2019)), which is governed by the eruption magnitude
43 and eruption column height (the altitude at which gas and ash are dispersed as a neutrally bouyant cloud)
44 relative to the tropopause (Aubry et al., 2019; Marshall et al., 2019; Krishnamohan et al., 2019; Aubry
45 et al., 2021b). In addition to the injection height of SO₂, the chemistry and microphysics governing aerosol
46 formation and stratospheric residence time are also critical controls on the climate effects of eruptions
47 (Timmreck, 2012; Kremser et al., 2016; LeGrande et al., 2016; Zhu et al., 2020; Staunton-Sykes et al.,
48 2021). SO₂ is frequently transported together with fine ash and water from the eruption column (e.g. Rose
49 et al., 2001; Ansmann et al., 2011), where chemical scavenging of SO₂ onto ash surfaces (Rose, 1977;
50 Schmauss and Keppler, 2014) and physical incorporation into hydrometeors (Rose et al., 1995; Textor
51 et al., 2003) can scrub SO₂ from the eruption column. Water transported by the eruption cloud can enhance
52 nucleation and growth rates of aerosol particles (LeGrande et al., 2016), and ash particles provide sites
53 for aerosol nucleation or direct uptake of SO₂ (Zhu et al., 2020). Consequently, the presence of water
54 and fine ash influences resulting aerosol formation rates, particle sizes, optical properties, and residence
55 times, which are key parameters governing climate forcing (Kremser et al., 2016). Constraining the climate
56 impacts of volcanic eruptions therefore requires understanding of eruption transport processes governing
57 injection height, as well as the quantities of fine ash and water in eruption columns and clouds.

58 Climate-forcing related to eruptions is sensitive to the environmental conditions of eruptions as well as
59 global eruption frequency-magnitude distributions, both of which can evolve with global climate warming
60 or cooling. For example, sustained anthropogenic climate change will drive an increase in the strength
61 of tropospheric density stratification and tropopause height, and alter stratospheric circulation. These
62 atmospheric changes are expected to reduce the stratospheric delivery of SO₂ in moderate-magnitude
63 eruptions (Aubry et al., 2016, 2019), while exacerbating the radiative effects of relatively rare, large-
64 magnitude eruptions (e.g. Pinatubo 1991) (Aubry et al., 2021b). Other potential mechanisms for climatic
65 influence on volcanism include eruption triggering by extreme rainfall events (e.g. Elsworth et al., 2004;
66 Capra, 2006; Farquharson and Amelung, 2020) or changes to ocean stratification (Fasullo et al., 2017).
67 Climate cycles also influence rates and locations of global volcanism; growth or loss of ice sheets and
68 associated changes to hydrostatic pressure on the crust inhibit or enhance, respectively, melt generation,
69 ascent rates, and eruption rates (Jull and McKenzie, 1996; Huybers and Langmuir, 2009; Watt et al.,

2013; Baldini et al., 2015; Cooper et al., 2018). For example, Huybers and Langmuir (2009) correlated observed spikes in atmospheric CO₂ with increased rates of volcanism following the Last Glacial Maximum, and proposed a glaciovolcanic-CO₂ feedback, where enhanced rates of volcanism and CO₂ outgassing contribute to additional warming and ice sheet loss. Importantly, deglaciation and sea level changes are also likely to influence the frequency of direct interaction of erupting magma with surface water and ice, and the implications of magma-water interaction (MWI) for volcano-climate forcing remain largely unexplored.

Explosive volcanic eruptions involving interactions of magma with external surface water or ice (termed hereafter *hydrovolcanic eruptions*) evolve as a result of thermophysical and chemical processes that are wholly distinct from those of “dry” magmatic eruptions (those in which the main component of water present is that exsolved from the melt) (Self and Sparks, 1978; Houghton et al., 2015). Figure 1 shows a summary of hydrovolcanic eruption processes affecting the transport and stratospheric delivery of SO₂ as compared with purely magmatic eruptions. The presence of external surface water influences eruption evolution via two primary controls: (1) hydrostatic pressure, and (2) exchange of mass, momentum, and heat via direct interaction between water and the erupting gas-pyroclast mixture (Wohletz et al., 2013; Smellie and Edwards, 2016; Cas and Simmons, 2018). Increased hydrostatic pressure can reduce eruption explosivity by suppressing bubble nucleation and growth, reducing magma decompression and ascent rates, and potentially preventing magmatic fragmentation (Smellie and Edwards, 2016; Cas and Simmons, 2018; Manga et al., 2018). In contrast, secondary fragmentation and ash production can be relatively enhanced as a result of the actions of large thermal stresses arising through the rapid transfer of heat from hot pyroclasts to entrained surface water (Gonnermann, 2015; van Otterloo et al., 2015; Zimanowski et al., 2015). Heat consumption by the vaporization of entrained external water results in a loss (or redistribution) of the thermal buoyancy delivered by the eruption at the vent, which may be recovered via condensation higher in the plume where temperatures are colder (Koyaguchi and Woods, 1996). The efficiency of mixing and heat transfer between pyroclasts and external water therefore controls the eruption column source parameters (e.g. bulk temperature, density, velocity, and column radius) (Koyaguchi and Woods, 1996; Mastin, 2007b), as well as the intensity of secondary fragmentation and the resulting particle size distribution (PSD) (Mastin, 2007a; van Otterloo et al., 2015). The PSD influences rates of particle aggregation and fallout, and available particle surface area (Bonadonna et al., 1998; Brown et al., 2012; Girault et al., 2014). In turn, increased water content, ash surface area, and colder temperatures in the rising eruption column provide conditions likely to enhance chemical scavenging of SO₂ during transport and dispersal relative to dry eruptions (Schmauss and Keppler, 2014). For example, Textor et al. (2003) simulate dynamical, chemical, and microphysical processes occurring in a dry Plinian eruption and estimate that the percent of SO₂ erupted at the vent that is ultimately injected into the stratosphere was $\gtrsim 80\%$. However, in marked contrast, for the glaciovolcanic eruption of Grimsvöfn in 2011, Sigmarsson et al. (2013) estimate that approximately 50% of the exsolved sulfur gas was dispersed to the atmosphere, with much of the remainder lost to scavenging by ash particles or external surface water.

Magma-water interactions (MWI) and their effects throughout an eruptive phase are maximized in persistent deep layers of water where significant entrainment can occur over the time of column rise. In subglacial or subaqueous environments where water availability is limited by, say, ice melting and melt-water drainage (e.g. Magnússon et al., 2012), build-up of insulating volcanic tephra (e.g. Fee et al., 2020), or by simply the finite volume of a reservoir (e.g. Gudmundsson et al., 2014), water access to the volcanic vent can decline during an eruption, causing the extent of MWI to evolve in turn. With declining water layer depths, eruption styles may progress from an initial suppression of explosive behavior, to collapsing jets, to buoyant plumes of increasing height (Koyaguchi and Woods, 1996; Mastin, 2007b; Van Eaton et al., 2012; Wohletz et al., 2013; Manga et al., 2018). This evolution is important to recognize:

115 The degree to which an erupting magma interacts with surface water can exert critical control over the
116 ultimate delivery of ash, water, and SO₂ into the troposphere and stratosphere (Rose et al., 1995). Although
117 observational, experimental, and numerical studies have individually investigated processes relevant to
118 hydrovolcanic eruptions, it is critical to assess their behavior as a system to reveal controls on the ultimate
119 fate of erupted ash and gas.

120 To make critical progress in understanding the extent to which surface water governs the character and
121 magnitude of volcano-climate forcing, it is necessary to examine syn-eruptive processes that determine the
122 transport and ultimate fate of volcanic SO₂. In particular:

- 123 1. How do hydrostatic pressure, water entrainment, and MWI affect the coupled dynamics of gas
124 exsolution and magma fragmentation in the subterranean conduit, heat transfer from pyroclasts to
125 external water, secondary production of fine ash, and transport of ash, water, and SO₂ in the eruption
126 column?
- 127 2. To what extent can MWI processes and their control on eruption source conditions be quantitatively
128 linked to the observable thickness or abundance of a surface water layer?
- 129 3. What are the critical relationships among water mass fraction at the eruption column source and mass
130 fluxes of SO₂, fine ash, and water to the stratosphere?

131 In this study, we address these questions using coupled conduit-plume 1D numerical simulations of
132 sustained, sub-Plinian to Plinian hydrovolcanic eruptions. We estimate the sensitivity of the efficiency of
133 stratospheric SO₂ injection to the presence of water layers up to 500 m deep. The model approach consists
134 of three coupled components (see Figures 1 and 2): (1) a 1D conduit model simulating magma ascent and
135 fragmentation (Hajimirza et al., 2019), which we modify with an arbitrary hydrostatic pressure boundary
136 condition applied at the vent; (2) a novel near-field “vent” model simulating decompression of the initial
137 gas-pyroclast mixture, water entrainment, and quench fragmentation as a function of surface water depth
138 Z_e ; and (3) a modified version of the 1D eruption column model from Degruyter and Bonadonna (2012),
139 incorporating a particle size distribution with sedimentation following Girault et al. (2014). We focus our
140 analysis on the main factors affecting overall column rise and gravitational stability (e.g. magma ascent and
141 fragmentation, MWI and eruption column source parameters, and resulting column gravitational stability,
142 height, and sedimentation) and environmental conditions for vertical SO₂ transport (e.g. temperature, water
143 mass fluxes, and mass and surface area of ash particles). In considering only column height, entrainment
144 of water mass, and particle loss, we neglect a number of issues that will enter into more complete future
145 treatments of an SO₂ delivery efficiency: 1) a thermodynamic control in the conduit on the SO₂ solubility
146 behaviour below the fragmentation depth; 2) the coupled microphysics and kinetics of SO₂ scavenging
147 by ash particles sedimenting from the column and overlying umbrella cloud through various mechanisms
148 (Rose, 1977; Bursik et al., 1992; Durant et al., 2009; Niemeier et al., 2009; Carazzo and Jellinek, 2012;
149 Manzella et al., 2015); and 3) the kinetics of sulfur aerosol nucleation and growth (Kremser et al., 2016)
150 with or without ash (Zhu et al., 2020). As a consequence of ignoring the above effects, our study does
151 not address: (1) effects on the amount of sulfur gas exsolved from the melt (e.g. possibly reduced SO₂
152 exsolution due to hydrostatic pressure); (2) scavenging and sedimentation of sulfur species during eruption
153 and column ascent (i.e. we assume 100% of exsolved sulfur is transported along with the column and is
154 delivered to the final buoyancy level or is carried downwards with column collapse); (3) the formation,
155 dispersal, atmospheric lifetime, and radiative effects of sulfate aerosols following co-injection of SO₂,
156 ash, and water into the spreading eruption cloud. However, we discuss the implications of injection of
157 co-injection of SO₂ with enhanced quantities of fine ash and water in Section 4.

2 METHODS

158 2.1 A Model of Sustained, Explosive Hydrovolcanism

159 Our focus is on sustained eruptions with sufficient momentum and buoyancy fluxes at the source to inject
160 SO₂ into the stratosphere. Consequently we restrict our analysis and modelling efforts to a class of powerful
161 and sustained eruptions driven by initial magmatic vesiculation and fragmentation in the conduit, where
162 the gas-pyroclast mixture is modified by the entrainment and mixing of external water that is primarily
163 confined to the surface environment. This approach is motivated by observations of pyroclast textures
164 and particle size distributions from several hydrovolcanic eruptions, including the 25 ka Oruanui and 1.8
165 ka Taupo eruptions, New Zealand (Self and Sparks, 1978; Wilson and Walker, 1985), the 1875 eruption
166 of Askja Volcano, Iceland (Self and Sparks, 1978; Carey et al., 2009), the 2011 eruption of Grímsvötn
167 (Liu et al., 2015), the 2500 BP Hverfjall Fires eruption (Liu et al., 2017), and the 10th century eruption
168 of Eldgjá Volcano, Iceland (Moreland, 2017; Moreland et al., 2019). Whereas airfall deposits from dry
169 phases of each of these eruptions have total particle size distributions (PSD - we refer to total particle
170 size distributions throughout unless otherwise stated) and porosities typical of Plinian events (Cas and
171 Wright, 1987; Fisher and Schmincke, 2012), PSDs from wet eruption phases are relatively fines-enriched.
172 Observations of PSDs, pyroclast textures and vesicularities from these events lead to the interpretation
173 that melts fragmenting inside the conduit produce approximately similar PSDs that are modified, in turn,
174 through a “secondary” episode of fragmentation related to the quenching of the gas-pyroclast mixture
175 within overlying surface water layers (Liu, 2016; Aravena et al., 2018; Moreland et al., 2019; Houghton and
176 Carey, 2019). In principle, PSDs can also be modified through effects of groundwater infiltration through
177 the conduit walls, which can be enhanced with an overlying water layer as has been suggested on the basis
178 of field observations (Barberi et al., 1989; Houghton and Carey, 2019). However, numerical simulations of
179 Aravena et al. (2018) demonstrate that the extent of groundwater infiltration from 100-300 m-thick aquifers
180 perched at or above the fragmentation depth depends on the magma mass eruption rate (MER). Crucially,
181 for $MER \gtrsim 5 \times 10^6$ [kg / s], which is typical of the sustained explosive eruptions on which we focus, water
182 infiltration into the overpressured conduit flow is largely inhibited or impossible. Aravena et al. (2018)
183 further suggest this condition may be an explanation for why phreatomagmatic activity associated with
184 direct interaction of un-fragmented melt with external water is more commonly associated with eruptions
185 with relatively low MER, a result consistent with field observations (Walker, 1981; Houghton and Wilson,
186 1989; Cole et al., 1995; Moreland et al., 2019; Houghton and Carey, 2019).

187 Taking these observations and inferences into consideration in our modelling approach, we assume that
188 secondary fragmentation from MWI is driven by quench fragmentation (a.k.a. thermal granulation) (van
189 Otterloo et al., 2015), as opposed to phreatomagmatic fragmentation by molten-fuel-coolant interaction
190 (Büttner et al., 2002). Following Jones et al. (2019), the MWI model is based on the physics of water
191 entrainment for a subaqueous jet as well as the energetics of quench fragmentation. We do not consider
192 classes of hydrovolcanic events such as Surtseyan-type, or those driven by episodic molten-fuel coolant
193 interactions with $MER \ll 10^6$ [kg/s] (Wohletz et al., 2013; Houghton et al., 2015). Thus, we focus on
194 eruptive phases in a sub-Plinian to Plinian to Phreatoplinian continuum under established classification
195 schemes (Walker, 1973; Self and Sparks, 1978). Furthermore, we model only the sustained, steady-state
196 phases of these events. Figure 2 shows a conceptual overview of the problem definition and model setup in
197 the near-vent region where an erupting jet emerges from the volcanic vent and encounters a shallow (≤ 500
198 m) water layer. On the basis of arguments from Aravena et al. (2018) for eruptions with $MER \gtrsim 10^6$ kg/s,
199 we do not consider water infiltration into the shallow conduit and assume MWI occurs only within the
200 overlying water layer. This study is not an exhaustive coverage over the full range of hydrovolcanic events,

201 but rather is a first attempt at characterizing the broad behavior of an important sub-class of sustained
 202 hydrovolcanic eruption for which substantial stratospheric injection of SO₂ is a likely outcome.

203 2.2 1D Conduit Model

204 We use the one dimensional conduit model of Hajimirza et al. (2021) and integrate flow properties over
 205 the cross-sectional area of the conduit. We assume a vertical cylindrical conduit with radius a_c and depth z
 206 (for a complete description of mathematical symbols and nomenclature, see Table 1). The conduit radius is
 207 fixed except near the surface, where flaring near the vent is possible to enforce mass conservation for a
 208 choked flow at the vent (Gonnermann and Manga, 2013). We assume the flow is steady - i.e. the duration
 209 of magma ascent is much shorter than the duration of Plinian eruptions (Mastin and Ghiorso, 2000). The
 210 magma is a mixture of rhyolitic melt (76% SiO₂) and H₂O bubbles that exsolve continuously during ascent
 211 because H₂O solubility is proportional to the square root of pressure. We assume crystals are only present
 212 at the nano-scale to enable heterogeneous bubble nucleation (Shea, 2017) and that their effect on magma
 213 rheology is negligible. Below the level of fragmentation we define magma as the mixture of silicate melt
 214 and H₂O bubbles, and we assume the melt phase is incompressible (Massol and Koyaguchi, 2005). The
 215 flow transitions discontinuously above the level of fragmentation to a dilute mixture of continuous H₂O
 216 vapor with suspended fragments of vesicular pyroclasts. For model purposes, we treat water as the only
 217 magmatic volatile, assuming SO₂ and other gases are carried passively by the flow, and use the term “gas”
 218 interchangeably with water vapor throughout unless otherwise stated.

219 We assume the relative velocity between the two phases (melt and H₂O vapor/fluid) to be negligible
 220 below and above the fragmentation level. Below fragmentation, bubbles are entrained in the very viscous
 221 melt and the magma rises as a foam (e.g. Mastin and Ghiorso, 2000; Gonnermann and Manga, 2007).
 222 Above fragmentation, a real volcanic flow will experience complex phenomena including solid/gas phase
 223 separation and sound wave dispersion, as well as buoyancy effects including the excitation of compaction
 224 and porosity waves (e.g. Bercovici and Michaut, 2010; Michaut et al., 2013). Such dynamics are important
 225 for degassing and can modify fragmentation processes in one-dimensional conduit models. However,
 226 their inclusion is practically challenging and the effect of resulting fluctuations in MER on the height
 227 and gravitational stability of steady-state plumes is ultimately small in comparison to controls arising
 228 through parameterizations for water and air entrainment. For simplicity and to retain a focus on the effect
 229 of entrainment and MWI on plume height and SO₂ delivery to the stratosphere, we neglect these dynamics
 230 and apply the common pseudo-gas approximation for fully-coupled gas and particle flow (Wilson et al.,
 231 1980; Mastin and Ghiorso, 2000). The properties of the magma mixture (melt and bubbles or gas and
 232 pyroclasts) are, consequently, the volumetric average of the two phases. We also assume the conduit flow
 233 to be isothermal (Colucci et al., 2014) because heat transfer across conduit walls is negligible over the time
 234 scale of rise through the depth z (Mastin and Ghiorso, 2000). The latent heat flux consumed through the
 235 exsolution of a H₂O with magma ascent helps to enforce this condition, although the effect is very small.

236 With these assumptions and simplifications, conservation of mass and momentum for the ascending
 237 magma are (Wilson et al., 1980; Mastin and Ghiorso, 2000)

$$\frac{\partial(\rho u A)}{\partial z} = 0, \quad (1)$$

238 and

$$\rho u \frac{\partial u}{\partial z} = -\frac{\partial p}{\partial z} - \rho g - F_{\text{fric}}, \quad (2)$$

239 respectively. Here u is magma ascent rate, $A = \pi a_c^2$ is the conduit cross sectional area, and ρ is bulk
240 magma density, averaged over liquid and gas phases,

$$\rho = \chi_v \rho_v + (1 - \chi_v) \rho_m, \quad (3)$$

241 where χ_v is the volume fraction of bubbles, and ρ_v and $\rho_m = 2400 \text{ kg/m}^3$ are gas and melt densities
242 respectively. The frictional pressure loss $F_{\text{fric}} = \rho u^2 f / a_c$ where f is a friction factor. Below the
243 fragmentation depth $f = 16/Re + f_0$ and above the fragmentation depth $f = f_0$. Here, the Reynolds
244 number $Re = 2\rho u a / \eta$, where η is the viscosity of the mixture. The reference friction factor $f_0 = 0.0025$
245 depends on the conduit wall roughness (Mastin and Ghiorso, 2000). By substituting equation (1) into (2)
246 and defining the isothermal mixture sound speed,

$$\left(\frac{\partial \rho}{\partial p} \right)_{T_0} = c^2, \quad (4)$$

247 we obtain (Gonnermann and Manga, 2013; Hajimirza et al., 2021)

$$-\frac{\partial p}{\partial z} = \frac{\rho g + F_{\text{fric}} - \frac{\rho u^2}{A} \frac{\partial A}{\partial z}}{1 - M^2}, \quad (5)$$

248 where $M = u/c$ is the Mach number of the mixture. Below the fragmentation depth $c^2 = K/\rho$, where K
249 is the bulk modulus of the mixture

$$\frac{1}{K} = \frac{\chi_v}{K_v} + \frac{1 - \chi_v}{K_m}. \quad (6)$$

250 Above the fragmentation depth, the bulk modulus of the gas phase K_v is calculated from the equation of
251 state for water (Holloway, 1977) at constant temperature.

252 The conduit model includes treatments for water vapor exsolution from the melt and subsequent bubble
253 growth; details are in Hajimirza et al. (2021). At a given depth below fragmentation, heterogeneous bubble
254 nucleation on crystal nanolites occurs with a critical supersaturation, and growth is by the diffusion of water
255 from the melt. Above the fragmentation depth the bubble volume and number density are fixed, although
256 vapor can continue to exsolve and escape from pyroclasts into the surrounding free vapor by permeable
257 flow. We employ a fixed porosity threshold of 75% as a fragmentation condition, which is consistent with
258 measurements and analyses of pumice permeabilities and vesicle size distributions that show that PSDs
259 follow power laws comparable to those of pore-scale microstructures in erupted pumice (Kaminski and
260 Jaupart, 1998; Rust and Cashman, 2011). We consequently do not fix a PSD in the conduit and assume
261 only that fragmentation proceeds to small enough length scales such that permeable gas escape from the
262 pyroclasts is sufficient to ensure that pore-scale pressures equilibrate to the free gas in the conduit at the
263 vent height (Rust and Cashman, 2011).

264 Assuming negligible gas escape or water infiltration through conduit walls, the primary effect of overlying
265 surface water or ice is to modify the pressure boundary condition at the volcanic vent. Above magmatic
266 fragmentation, the gas-pyroclast mixture fluidizes, accelerates, and decompresses towards the conduit exit.
267 If the flow speed remains below the mixture sound speed, c , then the vent exit pressure, p_e , must balance
268 the ambient pressure above the vent, p_e , which is determined by water depth:

$$p_e = \rho_e g Z_e + p_{\text{atmo}}, \quad (7)$$

269 where ρ_e is the density of external water and p_{atmo} is the atmospheric pressure at the water surface. If
 270 however, the speed approaches $M = 1$, then flow becomes choked and the flow at vent can become
 271 overpressured relative to ambient (Gonnermann and Manga, 2013). As a metric for vent overpressure, we
 272 introduce the vent overpressure ratio $\beta = p_c/p_e$. To enforce mass conservation for choked flow, either
 273 choking must occur at the vent exit of a fixed radius conduit or the conduit radius must flare accordingly
 274 (Gonnermann and Manga, 2013). The conduit modelling approach is therefore to seek solutions where the
 275 pressure in the conduit flow matches the surface pressure boundary condition (i.e. $\beta \approx 1$), or for which the
 276 conduit is choked at (no flaring) or near (with flaring) the vent (i.e. $\beta \gtrsim 1$, $M \approx 1$).

277 To gain insight into how an ascending magma responds to changes in hydrostatic pressure related to
 278 loading by overlying layers of water or ice, it is instructive to compare solutions for eruptions with and
 279 without external water, with other independent parameters fixed. To this end we choose a fixed conduit
 280 depth $z = 6$ km, an initial magmatic temperature $T_0 = 1123.15$ K and a maximal (unexsolved) magmatic
 281 water content corresponding to saturation as determined with the method of Liu et al. (2005). We then
 282 use an iterative search to find conduit parameters that satisfy the pressure-balanced or choked conditions.
 283 We first allow conduit radius to vary to obtain solutions for a “dry” or subaerial vent where no external
 284 water is present and the ambient pressure above the vent is equal to atmospheric ($Z_e = 0$). Subaerial vent
 285 simulations were run and suitable conduit radii obtained for a range of “control” MER $10^{5.5} \leq Q_0 \leq 10^9$
 286 kg/s, and we refer to these subaerial vent scenarios as “control” simulations hereafter. For control scenarios,
 287 we seek specifically solutions where choking occurs at the vent exit and thus no conduit flaring is required.
 288 This calculation provides a reference conduit radius to use in scenarios with a water layer present above
 289 the vent, with water depths $0 < Z_e \leq 500$ m. For these hydrovolcanic cases, we then fix the conduit
 290 radius to that of the control scenario and find an adjusted conduit MER q_c such that the surface pressure
 291 and/or choking boundary conditions are again satisfied. All values of MER referred to herein (i.e. Q_0 , q_c)
 292 indicate magmatic mass fluxes in the conduit (i.e. excluding external water). See Supplemental Figure 1
 293 for a visualization of the search process for conduit radius and MER in control and hydrovolcanic cases,
 294 respectively. Although we choose MER as our adjusted parameter, other choices are possible, such as the
 295 excess pressure of the magma reservoir at the base of the conduit or modification of the vent geometry. To
 296 make clear our approach and the consequences of our approximations and simplifications, see Section 3.1
 297 for example conduit model results.

298 2.3 Vent and MWI Model

299 2.3.1 Initial Particle Size Distribution

300 The model PSD is first defined explicitly at the vent ($z = 0$) as a function of the output from the conduit
 301 model. We define an initial power-law PSD following Kaminski and Jaupart (1998) and Girault et al.
 302 (2014), over the particle size range $-10 \leq \phi_i \leq 8$. The number of particles N_i at size ϕ_i is given by

$$N_i = 2^{\log_2(N_0) + D_0 \phi_i} \quad (8)$$

303 where D_0 is the power-law exponent, N_0 is an arbitrary normalization constant, and subscript i indicates a
 304 particle size bin. We choose a default value of $D_0 = 2.9$. Each size class is assigned an effective porosity
 305 value χ_i on the basis of an effective particle radius:

$$\begin{aligned} \chi_i &= \chi_0, \quad r_i \geq r_{c1} \\ \chi_i &= \chi_0(1 - r_{c2}/r_i), \quad r_{c2} \leq r_i \leq r_{c1} \\ \chi_i &= 0, \quad r < r_{c2}. \end{aligned} \quad (9)$$

306 Here, $\chi_0 = 0.75$ is the porosity threshold for fragmentation, r_i is the particle radius for bin i , $r_{c1} = 10^{-2}$ m
 307 and $r_{c2} = 10^{-4}$ m. Particles of sufficiently small size have, thus, no effective porosity and densities equal
 308 to that of the pure melt phase ($\rho_{s,i} = \rho_m$). By contrast, the density of larger particles is a strong function
 309 of porosity and bubble gas density (Kaminski and Jaupart, 1998). This approach leads to expressions for
 310 particle mass fraction in each size bin, $n_{s,i}$, and the bubble gas mass fraction of each size bin, $n_{b,i}$:

$$\rho_{s,i} = (1 - \chi_i)\rho_m + \chi_i\rho_v \quad (10)$$

311

$$n_{s,i} = \frac{N_i r_i^3 \rho_{s,i}}{N_\phi \sum_{i=1} (N_i r_i^3 \rho_{s,i})} \quad (11)$$

312

$$n_{b,i} = \frac{\frac{\rho_v \chi_i}{\rho_m(1-\chi_i)}}{\left(1 + \frac{\rho_v \chi_i}{\rho_m(1-\chi_i)}\right)}, \quad (12)$$

313 where subscript s denotes the bulk “solids” phase (melt plus bubbles). Figure 3d shows the initial PSD for
 314 $D = 2.9$, accounting for particle density as a function of porosity (light gray line and square symbols).

315 2.3.2 Vent Decompression

316 Figure 2 highlights the geometry and relevant length scales for the MWI model. For an overpressured
 317 steam jet in the near-vent region involving flows with $M \geq 1$ (e.g Ogden et al., 2008), mixing of the
 318 gas-pyroclast mixture with external water is negligible over a “decompression length scale” L_d where
 319 expanding gas prevents pyroclasts inside the jet from interacting with external water (e.g. Kokelaar, 1986).
 320 Our decompression model therefore assumes that turbulent entrainment and mixing of external water
 321 begins at heights above L_d . For L_d , we use a modified form of the free decompression condition of Woods
 322 and Bower (1995) to find the height at which the jet gas pressure plus dynamic pressure is equivalent to
 323 external water hydrostatic pressure:

$$p_d + \frac{u_d^2 \rho_d}{2} = p_e(L_d) = \rho_e g(Z_e - L_d) + p_{atmo}, \quad (13)$$

324 where p is pressure, u_d is the speed after decompression, and ρ is density. Subscripts d and e denote
 325 properties of the jet mixture after “decompression” and of “external” water, respectively. Assuming the
 326 decompression speed is approximately the mixture sound speed (Ogden et al., 2008), using the dusty-gas
 327 approximation (Woods and Bower, 1995),

$$\begin{aligned} u_d \approx c_d \approx c_{v,d} \sqrt{\frac{\rho_{v,d}}{\rho_d \chi_{v,d}}} &= \sqrt{\frac{\rho_v}{\rho_d \chi_{v,d}} \frac{\gamma p_d}{\rho_v}} \\ &\approx \sqrt{\frac{\gamma p_d}{\rho_d}} \end{aligned} \quad (14)$$

328 where subscript v denotes the “vapor” phase, the free gas volume fraction $\chi_v \approx 1$, and γ is the ratio of
 329 specific heats for the vapor phase. Substituting Equation 14 into Equation 13 gives

$$p_d = \frac{p_e(L_d)}{1 + \frac{\gamma}{2}}. \quad (15)$$

330 We approximate decompression length L_d as proportional to the change in jet radius with decompression:

$$L_d = 2\Delta a = 2(a_d - a_c), \quad (16)$$

331 where

$$a_d = \left(\frac{\rho_c u_c a_c^2}{\rho_d c_d} \right)^{1/2}, \quad (17)$$

332 and

$$\rho_d = \left(\frac{1 - n_v}{\rho_s} + \frac{n_v}{\rho_{v,d}} \right)^{-1}. \quad (18)$$

333 Here n_v is the jet gas mass fraction, and the subscript c indicates properties in the “conduit” prior to
 334 decompression. Momentum and energy are not perfectly conserved after decompression in this formulation
 335 as they are in Woods and Bower (1995), because the radially averaged decompression velocity is taken to
 336 be the mixture sound speed. However, this approach is consistent with the results of numerical simulations
 337 (e.g. Ogden et al., 2008), where excess energy is dissipated via shock formation and related effects of
 338 supersonic flow, and radially average velocities after decompression are close to sonic. These equations
 339 give a decompression length approximately similar to the Mach disk height relation of Ogden et al. (2008),
 340 (see Supplemental Figure 2 for a comparison), but with the difference that $L_d \rightarrow 0$ for $\beta \lesssim 1$. This
 341 is an important distinction since the formal definition of L_d in our model is the height at which the jet
 342 overpressure is sufficiently small that turbulent mixing and entrainment can begin. For a pressure-balanced
 343 jet ($\beta = 1$), this critical height should be immediately above the vent. We note, however, that due to the
 344 rapid pressure change with height in the water column, the mixture will continue to expand and decompress,
 345 such that the static estimate of L_d used here is likely a lower bound.

346 2.3.3 Water Entrainment and MWI Model

347 The mixing of water, steam, pyroclasts, and lithic debris in the vent region in explosive hydrovolcanic
 348 eruptions is complex and may involve effects of shocks, supersonic flow, film boiling, and multiple
 349 fragmentation mechanisms (Wohletz et al., 2013; Houghton and Carey, 2015; van Otterloo et al., 2015) that
 350 introduce inherently time-dependent and three-dimensional mechanisms for entrainment and mechanical
 351 stirring that are not captured in a one-dimensional steady-state integral model. However, following extensive
 352 studies of entrainment and mixing into turbulent plumes (Morton et al., 1956; Linden, 1979; Turner, 1986),
 353 a recent complementary analysis of water entrainment into supersonic, submerged gas jets (Zhang et al.,
 354 2020) and studies of the bulk energetics of interactions between hot pyroclasts and water (Dufek et al., 2007;
 355 Mastin, 2007a; Schmid et al., 2010; Sonder et al., 2011; Dürig et al., 2012; Woodcock et al., 2012) we can
 356 parameterize these processes to explore effects on total budgets for mass, energy, and buoyancy. Following
 357 Morton et al. (1956); Kaminski et al. (2005); Carazzo et al. (2008); Zhang et al. (2020), we will relate
 358 the radial entrainment speed of water or atmosphere to the local rise speed of a jet and prescribe resulting
 359 velocity, pressure and temperature fields. We assume the rate of mixing and heat transfer between solid
 360 pyroclasts and entrained water to be sufficiently fast that all phases maintain thermal equilibrium inside
 361 the jet over the timescale of rise through the water column. We discuss consequences of this assumption
 362 further in Section 4.

363 We initialize the water entrainment model at height L_d above the vent. Initial conditions for jet velocity,
 364 radius, and density are determined after decompression by balancing jet gas pressure with hydrostatic
 365 pressure at L_d . Other parameters such as gas mass fraction and temperature are obtained from values at
 366 the top of the conduit model, while the PSD and pyroclast porosity and density are determined according

367 to Section 2.3.1 above. An iterative MATLAB solver integrates solutions to the differential equations for
 368 water and particle mass, bulk momentum and energy, and PSD mass fractions from the decompression
 369 height to the water surface. The physical properties of entrained water are calculated using the International
 370 Association for the Properties of Water and Steam 1995 formulation (Junglas, 2009). To capture the
 371 evolutions with height of the energy and enthalpy of the mixture, we follow a similar approach to Mastin
 372 (2007b). The initial enthalpy of the solid phase at the vent surface h_{s0} is determined from a weighted
 373 combination of the enthalpy of exsolved gas bubbles and the specific heat of the melt phase:

$$h_{s0} = h_b(p_b, T_0) \sum_{i=1}^{N_\phi} n_{s,i} n_{b,i} + C_m (T_0 - T_{ref}) \sum_{i=1}^{N_\phi} (1 - n_{b,i}) n_{s,i}. \quad (19)$$

374 Here $h_b(p_b, T_0)$ is bubble gas enthalpy as a function of pressure and temperature, $C_m = 1250 \text{ J/(kg K)}$
 375 is the melt heat capacity (assumed constant), and $T_{ref} = 274.15 \text{ K}$ is a reference temperature. The total
 376 mixture enthalpy, h is then:

$$h = n_w h_w(p, T) + (1 - n_w) h_s \quad (20)$$

377 where n_w and h_w are the mass fraction and enthalpy of water (gas and liquid) within the jet mixture. At the
 378 decompression length, the total power supplied by the jet is:

$$\dot{E}_0 = q_c (h_0 + g' L_d + \frac{u_d^2}{2}) \quad (21)$$

379 Where q_c is the conduit MER and $g' = g(\rho - \rho_e)/\rho_e$ is the reduced gravity, and the dot notation over E
 380 indicates the rate of energy delivered (i.e. power).

381 From an initial value T_0 , the bulk temperature of the jet mixture T is calculated at each solver step
 382 following Mastin (2007b). Specifically, the enthalpy at each step is compared with two values: the enthalpy
 383 h_{vap} that the mixture would have at the water saturation temperature assuming 100% steam (dryness
 384 fraction $x_v = 1$), and h_{liq} , where the water phase is 100% liquid ($x_v = 0$). For $h > h_{vap}$, the mixture
 385 temperature is found using an iterative approach to match the known enthalpy value h . For $h_{liq} < h < h_{vap}$,
 386 $T = T_{sat}$ and $x_v = (h - h_{liq})/(h_{vap} - h_{liq})$. We employ a stop condition as dryness fraction reaches
 387 $x_{v,crit} = 0.02$. This condition is justified physically because as the jet water fraction becomes mostly liquid
 388 with $x_v \rightarrow 0$, the resulting high-density jets always collapse almost immediately after breaching the water
 389 surface and are therefore ineffectual at injecting SO_2 into the stratosphere. Conceptually, this condition is
 390 equivalent to the case where *at most* only minor quantities of steam breach the water surface, potentially
 391 generating steam plumes but carrying negligible quantities of volcanic ash or other volatiles (e.g. Cahalan
 392 and Dufek, 2021). We refer to the above ultra-high water fraction scenarios as the ‘‘steam plume’’ regime
 393 hereafter. For greater water depths still, the gas jet would entirely condense and fail to breach the water
 394 surface (Cahalan and Dufek, 2021). Furthermore, as the vapor fraction approaches zero, steep gradients in
 395 density significantly increase problem stiffness and computation time, and we thus discard these results
 396 and do not integrate further.

397 Entrainment of ambient fluid into a jet or plume is driven by both radial pressure variations arising
 398 from the relatively fast rise of the jet and local shear at the jet boundary (see Figure 1). Entrainment
 399 parameterizations in integral plume models typically assume that the rate of radial inflow of ambient fluid
 400 v_ε at any height is proportional to the upflow speed (Morton et al., 1956):

$$v_\varepsilon = \alpha u, \quad (22)$$

401 where α is an entrainment coefficient of order 0.1. Here we employ a variable entrainment coefficient
 402 following Kaminski et al. (2005); Carazzo et al. (2008):

$$\alpha = 0.0675 + \left(1 - \frac{1}{A}\right) Ri + \frac{a}{2} \frac{d}{dz} \ln(A) \quad (23)$$

403 where

$$Ri = \frac{g'a}{u^2} \quad (24)$$

404 is the local Richardson number that expresses the balance between the momentum and stabilizing buoyancy
 405 fluxes at a given height. The shape function $A = A(z)$ depends on the diameter of the jet and Ri at
 406 $z = 0$. This well-established hypothesis for ambient fluid entrainment is, however, strictly valid only where
 407 turbulence is fully developed. This picture assumes that there is a direct momentum exchange between
 408 large entraining eddies that form plume edges and a full spectrum of turbulent overturning motions that
 409 mix momentum, heat and mass across the plume radius down to spatial scales limited by either molecular
 410 diffusion or dissipation by very fine ash (Lherm and Jellinek, 2019). In general, this condition is established
 411 over heights of roughly 5 to 10 vent diameters (i.e. the vent near-field, see also Figure 2) and corresponds to
 412 a transition from flow as a jet governed by the momentum flux delivered at L_d to flow as a buoyant plume
 413 driven by a balance between buoyancy and inertial forces (Carazzo et al., 2006; Saffaraval and Solovitz,
 414 2012). A key issue for the character and magnitude of effects related to MWI is whether and where in the
 415 water layer this transition occurs such that water entrainment is fully established.

416 To constrain this transition height relative to L_D we follow an approach developed in Kotsovinos (2000)
 417 to identify the dynamical ‘‘crossover height’’ L_X at which fully turbulent plume rise starts and above which
 418 Equation 22 holds. Below L_X , the flow evolves predominantly in response to the momentum flux supplied.
 419 In this regime, drag related to turbulent instabilities, accelerations, overturning motions and mixing is not
 420 established and on dimensional grounds the evolving height of the jet

$$h_{jet} \sim (\pi a_d^2 u_d)^{1/4} t_{jet}^{1/2}, \quad (25)$$

421 Above L_X , plume height predominantly governed by a balance between buoyancy and inertial forces is, by
 422 contrast,

$$h_{BI} \sim \left(\frac{g'q}{\pi a_d^2}\right)^{1/2} t_{BI}^{3/2}, \quad (26)$$

423 The transition height L_X occurs where $h_{jet} = h_{BI}$, which corresponds to where the characteristic time
 424 scale $t_{jet} = t_{BI}$. After algebra we obtain

$$L_X = \pi^{5/8} u_d^{3/4} \left(\frac{a_d^5 \rho_d}{g'q}\right)^{1/4}. \quad (27)$$

425 Starting from height $z = L_D$, we assume the thickness a_{mix} of a turbulent mixing layer at the jet boundary
 426 develops monotonically over distance L_X :

$$a_{mix} = a \frac{z - L_d}{L_X}; \quad a_{mix} \leq a, \quad (28)$$

427 above which the radial turbulent mixing is complete and the velocity profile is top-hat or Gaussian,
 428 consistent with the assumption of self-similar flow (Morton et al., 1956; Turner, 1986). We then obtain

429 an effective entrainment coefficient, α_{eff} , by scaling the entrainment coefficient based on the volumetric
430 growth of the mixing layer:

$$\alpha_{eff} = \alpha \frac{2aa_{mix} - a_{mix}^2}{a^2}. \quad (29)$$

431 Using a similar entrainment parameterization to Mastin (2007b) which accounts for the relative density
432 difference of the ambient and entraining fluid, the rate of water entrainment into the jet is

$$\frac{dq_{w,e}}{dz} = 2\pi a \alpha_{eff} u \sqrt{\rho \rho_e}. \quad (30)$$

433 In a recent study of supersonic air jets intruding 1-400 m deep layers of water from below (Zhang
434 et al., 2020) shows that entrainment and mixing is significantly augmented by buoyancy effects related
435 to the rise of air through layers of relatively dense water. Their results suggest that this mechanism will
436 dominate the mechanics of entrainment for water layer depths exceeding a few hundred meters. This
437 condition is presumably set by the height in the water column at which the overturn time of large entraining
438 eddies related to the rise of buoyant air becomes less than the time scale for water ingestion through
439 shear-induced turbulence (Equation 23). The extent to which this mechanism governs the evolution of
440 rapidly expanding hot volcanic jets erupting through comparably thick layers of water is, however, unclear
441 and particularly so where L_d is of the same order of magnitude as the water depth. For completeness, we
442 compare results obtained from Equations 23 to 29 with complementary calculations assuming entrainment is
443 partially governed through the buoyancy-driven ‘‘Rayleigh-Taylor’’ entrainment mode of Zhang et al. (2020).
444 Specifically, we define an alternative α_{eff} as a weighted average of the shear-driven and Rayleigh-Taylor
445 entrainment modes:

$$\alpha_{eff} = B\alpha + (1 - B)\alpha_{RT}, \quad (31)$$

446 where

$$\alpha_{RT} = 4\pi \frac{a_d}{q_c} a \sqrt{\frac{2\rho}{3} (3\sigma \rho_e \omega)^{1/2}}. \quad (32)$$

447 Here, α_{RT} is the Rayleigh-Taylor coefficient for buoyancy driven entrainment, B is the weight determining
448 the relative balance between entrainment driven by buoyancy and that driven by shear-induced turbulence, σ
449 is the surface tension at the water-steam interface, and $\omega \approx (0.3u)^2 / (2\pi a)$ is the average radial acceleration
450 of the interface (Zhang et al., 2020). The geometric constant of 0.3 is an approximate scaling for the
451 magnitude of turbulent velocity fluctuations (Cerminara et al., 2016) and ensures that the radial momentum
452 flux carried by the inflow is an order of magnitude smaller than the vertical momentum flux carried by
453 the jet itself. This condition is required for the jet to remain intact and approximately conical, consistent
454 with the results of (Zhang et al., 2020), and for the equations underlying the 1D plume model to hold
455 (Morton et al., 1956). We compare the consequences of different entrainment modes for eruption behavior
456 in Sections 3.2 and 4.1.

457 2.3.4 Quench Fragmentation Model

458 The process of quench fragmentation of pyroclastic particles of various size during MWI is complex.
459 Driving thermal stresses and stress concentrations arising through interactions with cold water depend on
460 the curvatures of the outer surfaces of pyroclasts, their porosity and surface area-to-volume ratio, and on
461 the spatial distributions and rates of both surface cooling and film boiling. How to capture thoroughly these
462 particle-scale effects and their consequences for the mean particle size distribution in an evolving volcanic
463 jet mixture is unclear and remains a subject of vigorous research (e.g. Wohletz, 1983; Büttner et al., 2002,
464 2006; Mastin, 2007a; Woodcock et al., 2012; Patel et al., 2013; Liu et al., 2015; van Otterloo et al., 2015;

465 Fitch and Fagents, 2020; Dürig et al., 2020b; Moitra et al., 2020). However, with a specified magmatic heat
 466 flow at the vent, considerations of the surface energy consumed to generate fine ash fragments (Sonder
 467 et al., 2011), guided by published experiments along with observational constraints on the hydromagmatic
 468 evolution of particle sizes (Costa et al., 2016), provide a way forward that is appropriate for a 1D integral
 469 model. Figure 3 highlights the salient features of the fragmentation model, using the example of a single
 470 simulation with $q_c = 1.03 \times 10^8$ kg/s and $Z_e = 120$ m. Sonder et al. (2011) performed lab experiments
 471 submerging molten basalt into a fresh water tank to constrain the partitioning of thermal energy lost from
 472 the melt between that which is transferred from melt to heat external water and that which is consumed
 473 irreversibly through fracturing of the melt to generate new surface area and fine ash. At any height above
 474 the vent, the total power delivered to entrained external water from the melt is:

$$\Delta \dot{E}_e = (1 - \zeta) \Delta \dot{E}_m \quad (33)$$

475 and $\Delta \dot{E}_m$ is the rate of heat loss from the melt phase. The remaining heat loss from the melt i.e. $\zeta \Delta \dot{E}_m$ is
 476 the energy consumed by fragmentation. Note that we define fragmentation energy efficiency in the opposite
 477 sense to Sonder et al. (2011) such that $\zeta = 1 - \eta$, where η is as defined in that work. The parameter ζ is
 478 an empirical fragmentation energy efficiency that gives the fraction of thermal energy lost irreversibly to
 479 fragmenting pyroclasts to generate fine ash. Where thermal stresses related to cooling produce no fine ash,
 480 $\zeta = 0$ and $\Delta \dot{E}_e = \Delta \dot{E}_m$. Experimentally, Sonder et al. (2011) find $0.05 \lesssim \zeta \lesssim 0.2$ for thermal granulation
 481 processes, with typical values of ~ 0.1 .

482 Below, we use Equation 33 to define power transfer during each step of the MWI model. In more detail,
 483 entrained water must thermally equilibrate with both pyroclasts and internal water already in the volcanic
 484 jet. With both sinks for thermal energy included, we recast Equation 33 to be the total power transferred to
 485 entrained water at each height step:

$$\Delta \dot{E}_e = (1 - \zeta) \Delta \dot{E}_m + \Delta \dot{E}_w \quad (34)$$

486 where $\Delta \dot{E}_w$ is the power supplied for heating external water by heated water already in the volcanic jet.
 487 Although this energy sink is very small for typical magma water mass fractions of $\lesssim 5\%$ at the vent height,
 488 this contribution to the energy balance in Equation 34 evolves to be significant with height in the jet as a
 489 result of progressive water entrainment.

490 Neglecting a comparatively very small contribution from the specific heat of water trapped within the
 491 pores of pyroclasts, Equation 34 can be recast as an enthalpy change with water entrainment over a height
 492 step.

$$- \Delta q_{w,e} (h_{w,f} - h_e) = (1 - \zeta) q_s C_m (T_f - T) + q_w (h_{w,f} - h_w) \quad (35)$$

493 where $\Delta q_{w,e}$ is the mass flux of entrained water, $h_{w,f}$ is the final enthalpy of the water phase after thermal
 494 equilibration (i.e. where the jet gas and particles are well-mixed and at the same temperature), h_e is
 495 the external water enthalpy, q_w and h_w are the mass fluxes and enthalpy, respectively, of water already
 496 equilibrated thermally within the jet. In Equation 35, T_f and T are the unknown final mixture temperature
 497 and known initial mixture temperature for the current step, respectively. To estimate heat transfer to the
 498 entrained water phase, we assume that the change in temperature after equilibration $T_f - T$, is sufficiently
 499 small at each step that the jet water heat capacity can be approximated as constant for the current step, such
 500 that

$$h_{w,f} = h_w + C_w (T_f - T) \quad (36)$$

501 where C_w is the water heat capacity at temperature T . Substituting 36 into 35 leads to

$$T_f = \frac{(1 - \zeta)q_s C_m T + q_w C_w T - \Delta q_{w,e}(h_w - C_w T - h_e)}{(\Delta q_{w,e} + q_w)C_w + (1 - \zeta)q_s C_m} \quad (37)$$

502 T_f can then be used to estimate heat transfer to entrained water $\Delta h_w = h_{w,f} - h_e$, which is used along
503 with ζ and the PSD to later calculate the specific fragmentation energy, ΔE_{ss} .

504 Since we assume that the energy consumption during quench fragmentation results from the generation
505 of new surface area (Sonder et al., 2011; Dürig et al., 2012; Fitch and Fagents, 2020), we calculate the
506 specific surface area at each particle bin size assuming spherical particle geometry,

$$S_i = \frac{3\Lambda}{\rho_i r_i} \quad (38)$$

507 where Λ is a scaling parameter accounting for particle roughness, as true particle surface area can potentially
508 exceed that of ideal spherical particles by up to two orders of magnitude (Fitch and Fagents, 2020). We
509 take a default value $\Lambda = 10$, and discuss the effects of different choices for Λ in Sections 3.2 and 4. The
510 total surface specific surface area for a given PSD is

$$S = \sum_{i=1}^{N_\phi} S_i n_{s,i} \quad (39)$$

511 To simulate the evolution of the PSD by quench fragmentation, we prescribe a representative range
512 of particle sizes produced by thermal granulation based on the fine mode of particle sizes for the
513 phreatomagmatic phase C of the 1875 Askja eruption, as reported in Costa et al. (2016). The resulting
514 “output” PSD, $n_{si,f}$, is a normal probability density function, in ϕ size units, with mean $\phi_\mu = 3.43$
515 ($\sim 100 \mu\text{m}$) and standard deviation $\phi_\sigma = 1.46$, and is shown in Figure 3a (blue line).

516 The “input” particle sizes (i.e. particles that fragment to produce the fine fraction) are defined according
517 to the available surface area in the coarse fraction ($\phi < \phi_\mu$). We use the output mean, ϕ_μ as a fragmentation
518 cutoff - particles of this size and smaller are assumed to not participate in quench fragmentation, but
519 can participate in heat transfer to water. This allows the definition of an effective fragmentation energy
520 efficiency as a function of particle size (see Figure 3a, black line),

$$\zeta_i = \begin{cases} \zeta \frac{1 - n_{si,f}}{n_{s\phi_\mu,f}} & \phi_i < \phi_\mu \\ 0 & \phi_i \geq \phi_\mu \end{cases} \quad (40)$$

521 where $n_{s\phi_\mu,f}$ is the mass fraction of the mean size bin in the output PSD. Fragmentation efficiency thus
522 quickly reduces to zero as particle sizes approach the mean output size. In addition to the above particle
523 size limitation on fragmentation, we also halt fragmentation once the bulk mixture passes below the glass
524 transition temperature. We define the glass transition lower bound for a hydrous rhyolitic melt using an
525 empirical fit to data from Dingwell (1998) (note that Equation 41 is a distinct equation from the empirical
526 fit provided in that work):

$$T_g = 785.5 - 83.48 \log(c_{H_2O}) \quad (41)$$

527 where c_{H_2O} is the residual concentration (in wt.%) of H_2O still dissolved in the melt and obtained from the
528 conduit model (see Figure 3b). Since the glass transition occurs over a range of temperatures (Giordano

529 et al., 2005; van Otterloo et al., 2015), we apply the glass transition limit using a smooth-heaviside step
530 function of temperature,

$$h_{s_{sm}} = \left\{ 1 + \exp \left[\frac{-6}{\Delta T_g} \left(T - \left(T_g + \frac{\Delta T_g}{2} \right) \right) \right] \right\}^{-1} \quad (42)$$

531 where ΔT_g is the glass transition temperature range, with typical values of ~ 50 K (Giordano et al., 2005).
532 Using $h_{s_{sm}}$ to scale ζ with temperature (Figure 3c), Equation 40 becomes:

$$\zeta_i = h_{s_{sm}} \begin{cases} \zeta \frac{1-n_{si,f}}{n_{s\phi_\mu,f}} & \phi_i < \phi_\mu \\ 0 & \phi_i \geq \phi_\mu \end{cases} \quad (43)$$

533 and the effective fragmentation energy efficiency for determining total fragmentation energy from the PSD
534 is

$$\zeta_{eff} = \sum_{i=1}^{N_\phi} (1 - n_{bi}) n_{s,i} \zeta_i \quad (44)$$

535 The PSD of the coarse particle fraction (i.e. particle sizes that experience mass loss due to quench
536 fragmentation), $n_{si,0}$, is calculated as proportional to available particle surface area in each size bin,
537 modified by the fragmentation efficiency (Figure 3a, red lines):

$$n_{si,0} = \frac{\zeta_i S_i n_{s,i} (1 - n_{bi})}{N_\phi \sum_{i=1} \zeta_i S_i n_{s,i} (1 - n_{bi})} \quad (45)$$

538 Finally, we define the specific fragmentation energy (per mass of pyroclasts in the jet)

$$\Delta E_{ss} = \frac{\zeta_{eff}}{1 - \zeta_{eff}} \frac{\Delta h_w}{q_s} \frac{dq_{w,e}}{dz} \quad (46)$$

539 and the change in mass of the pyroclast fraction due to gas release from vesicles on fragmentation:

$$\frac{dm_{w,fr}}{dz} = m_s \frac{\Delta E_{ss}}{S_f E_s} \left[\sum_{i=1}^{N_\phi} \frac{n_{bi} n_{s,i}}{1 - n_{bi}} - \sum_{i=1}^{N_\phi} \frac{n_{bi} \left(n_{s,i} + \frac{dw_{s,i}}{dz} \right)}{1 - n_{bi}} \right] \quad (47)$$

540 where we choose $E_s = 100$ J/m² for the particle surface energy for fragmentation (Dürig et al., 2012).
541 The final differential equations for evolution of the PSD, and conservation of water mass, pyroclast mass,
542 momentum, and energy, are respectively

$$\frac{dn_{s,i}}{dz} = \frac{\Delta E_{ss}}{S_f E_s} (-n_{si,0} + n_{si,f}) \quad (48)$$

543

$$\frac{dq_w}{dz} = \frac{dq_{w,e}}{dz} + \frac{dq_{w,fr}}{dz} \quad (49)$$

544

$$\frac{dq_s}{dz} = -\frac{dq_{w,fr}}{dz} \quad (50)$$

$$\frac{d\varphi}{dz} = g(\rho_w - \rho)r^2 \quad (51)$$

545

$$\frac{d\dot{E}}{dz} = \frac{dq_{w,e}}{dz}(g'z + h_w) - q_s \Delta E_{ss} \quad (52)$$

546 Figure 3d shows the evolution of the total PSD during water entrainment and quench fragmentation
 547 in the MWI stage of the model according to Equation 48. The coarse to mid-size fraction of particles
 548 ($-3 \lesssim \phi \lesssim 2$) of particles deplete fastest owing to the surface area dependence in Equation 45. For example
 549 results of the MWI model, see Section 3.2.

550 2.4 1D Plume Model

551 For jets that breach the water surface, conditions at $z = Z_e$ are taken as the source parameters for the
 552 integral plume model. We use the integral plume model of Degruyter and Bonadonna (2012), modified
 553 with the particle fallout parameterization of Girault et al. (2014) to simulate differences in sedimentation
 554 in the eruption column as a function of fine ash production. Figure 3e shows the total PSD evolution
 555 due to particle fallout in the eruption column for a PSD that has been fines-enriched during MWI. The
 556 conservation equations for mass of dry air, water vapor, liquid water, and particles are, respectively:

$$\frac{d}{dz}(\rho_a u a^2 \chi_a) = 2v_\varepsilon a \rho_{a,e} \chi_{a,e} \quad (53)$$

557

$$\frac{d}{dz}(\rho_v u a^2 \chi_v) = 2v_\varepsilon a \rho_{v,e} \chi_{v,e} - \lambda \rho_v a^2 \chi_v \quad (54)$$

558

$$\frac{d}{dz}(\rho_l u a^2 \chi_l) = \lambda \rho_v a^2 \chi_v \quad (55)$$

559

$$\frac{d}{ds}(\rho_{s,i} u a^2 \chi_{si}) = -\xi \frac{q_s n_{s,i} u_{\phi,i}}{au} \quad (56)$$

560 where v_ε is the entrainment velocity, subscript a denotes properties for dry air, $\lambda = 10^{-2} \text{ s}^{-1}$ is a constant
 561 condensation rate (Glaze et al., 1997), $u_{\phi,i}$ are particle settling velocities following Bonadonna et al. (1998),
 562 and $\xi = 0.27$ is the particle fallout probability. The equations for vertical momentum and energy are,
 563 respectively:

$$\frac{d}{dz}(\rho u^2 a^2) = g(\rho_e - \rho)a^2 - w \frac{d(\rho u r^2)}{dz} + u \sum_{i=1}^{N_\phi} \frac{dq_{s,i}}{dz} \quad (57)$$

564

$$\frac{d}{dz}(\rho C T u a^2) = C_e T_e \rho_e a u_\varepsilon - \rho u a^2 g \sin \varphi + L \frac{d}{ds}(\rho_l u a^2 \phi_l) + C_s T \sum_{i=1}^{N_\phi} \frac{dq_{s,i}}{dz} \quad (58)$$

565 where C_s and C_e are the heat capacities of particles and air, respectively, T_e is the ambient air temperature,
 566 and L is the latent heat of condensation of water vapor. Note that the plume model retains the capability
 567 for simulating cross-winds as in Degruyter and Bonadonna (2012), but we show here only the vertical
 568 component of the momentum equation as we do not consider wind effects (wind fields are set to zero
 569 in atmospheric profiles). For further details on the plume model, we refer the reader to Degruyter and
 570 Bonadonna (2012, 2013), and to Girault et al. (2014) for the particle fallout details.

571 2.5 Simulation Scenarios

572 As described above, our model approach is to simulate eruptions across a parameter space with $10^{5.5} \leq$
 573 $Q_0 \leq 10^9 \text{ kg/s}$ and $0 \leq Z_e \leq 500 \text{ m}$. In Table 2 we define the *Reference* scenario which employs default

574 values as described above for the various model parameters. Specifically, the *Reference* scenario uses a
 575 water entrainment scheme that includes both decompression and cross-over length scalings, and default
 576 fragmentation parameters $\Lambda = 10$, $\zeta = 0.1$, $D = 2.9$. The atmospheric profile used in the *Reference*
 577 scenario is obtained from ERA reanalysis data for the 2011 eruption of Grímsvötn Volcano, with a
 578 corresponding vent altitude of 1750 m a.s.l (Hersbach et al., 2020; Aubry et al., 2021a). Note that we are
 579 not attempting to reproduce precise conditions for that eruption, but rather use this as a representative
 580 environmental condition for a high-latitude subglacial or sublacustrine eruption. To explore the effects of
 581 various model assumptions and parameter choices, we carried out nine additional simulation scenarios in
 582 addition to the *Reference* scenario, with each varying a single model parameter and performed over the
 583 same parameter space for MER and water depth. The second scenario we define, *Low-Lat*, uses an ERA
 584 reanalysis atmospheric profile for the 2014 eruption of Tungarahua Volcano with vent altitude 0 m a.s.l. as
 585 a representative atmosphere for a low-latitude submarine setting, keeping other parameters the same as
 586 the *Reference* scenario (see Supplemental Figure 3 for a comparison of atmospheric profiles used in the
 587 *Reference* and *Low-Lat* scenarios). Additional scenarios are broadly categorized into those with differing
 588 water entrainment assumptions and those with different fragmentation parameters relative to the *Reference*
 589 scenario. Entrainment scenarios include those without one or both of the decompression and crossover
 590 length scalings (*No- L_d* , *No- L_X* , and *No- L_d -No- L_X*), and a scenario with the Rayleigh-Taylor entrainment
 591 scheme of Equation 31 (αRT). Additional fragmentation scenarios include one with a higher particle
 592 roughness (*High- Λ*), higher and lower fragmentation energy efficiencies (*High- ζ* and *Low- ζ*), and a higher
 593 initial PSD power-law exponent (*High- D*). We highlight the effects of different entrainment scenarios in
 594 Section 3.2, and discuss the consequences of different parameter choices for these scenarios in Section 4.

3 RESULTS

595 3.1 Conduit Flow: Effects of an External Water Layer

596 An external water layer modifies hydrostatic pressure in the conduit, which affects bubble nucleation and
 597 growth by diffusion of water vapor, decompression rate and fragmentation conditions (Cas and Simmons,
 598 2018). In Figure 4, we compare conduit model output for control ($Z_e = 0$ m, red lines) and hydrovolcanic
 599 ($Z_e = 400$ m, blue lines) simulations for $Q_0 \sim 1.6 \times 10^8$ kg/s. In the dry scenario, gas exsolution begins
 600 with an initial bubble nucleation event at a depth of 5.5 km below the vent (panel (e)). Above the first
 601 nucleation event, gas exsolution continues, driving increasing magma buoyancy, ascent and decompression
 602 rates. A sharp increase in exsolution and bubble growth near $z = 1.3$ km drives the gas volume fraction
 603 above the fragmentation threshold of 75% (panel (d)). At this depth, fragmentation occurs and the flow
 604 becomes a fluidized mixture of pyroclasts suspended within a flow of free gas, which continues to expand
 605 and accelerate towards the vent. As the flow nears the vent, it accelerates to the mixture sound speed,
 606 becomes choked (panel (b)), overpressured relative to the hydrostatic pressure condition at the vent ($\beta \approx 11$,
 607 panel (a) inset), and erupts as an explosively decompressing subaerial jet.

608 Consistent with previous studies of subaqueous eruptions, the higher hydrostatic pressure at the vent
 609 in the hydrovolcanic case results in slower gas exsolution and bubble growth, and consequently a slower
 610 decompression rate in the ascending magma (Cas and Simmons, 2018). Slower exsolution also results in
 611 lower total gas exsolution from the magma, and lower gas volume fraction above fragmentation (panel (d)).
 612 Above the fragmentation depth, both the lower fraction of free gas and the higher hydrostatic pressure in
 613 the wet scenario result in less acceleration of the mixture, and the flow is subsonic ($M \approx 0.5$, panel (b))
 614 and pressure-balanced ($\beta \approx 1$, panel (a) inset) at the vent. For this water depth and MER, we find no viable
 615 conduit solution where the vent is choked [see also Supplemental Figure 1 for conduit solution search
 616 details]. Across all model scenarios (see Table 2), water depths sufficient to cause this pressure-balanced

617 condition usually lead to a weak jet that does not breach the water surface and/or to a steam plume condition
 618 (see Section 3.3 and Figure 9 below).

619 Figure 5 shows select parameters of the conduit model output as a function of MER and water depth,
 620 including vent overpressure ratio (panel (a), color field and contours), Mach number at the vent (b), MER
 621 adjustment relative to control runs (c), magma decompression rate at fragmentation depth (d), fragmentation
 622 depth (e), and the weight percent of residual water content dissolved in the pyroclasts at vent level (f). For
 623 the control runs ($Z_e = 0$), the vent is always overpressured and choked, with $\beta \rightarrow 45$ for the largest values
 624 of MER. Overpressure declines rapidly with increasing water depth until choking at the vent is impossible
 625 and the gas-pyroclast mixture enters the water layer as a pressure balanced, subsonic jet (solid blue line in
 626 panels (a),(c),(d)). We find that the largest water depth for which choking is possible is typically equal to
 627 about 5 vent radii. For example, for $Q_0 = 10^7$ kg/s, conduit radius $a_c = 20$ m, and the choking threshold
 628 depth occurs at ~ 100 m, whereas this threshold increases to ~ 220 m for $Q_0 = 10^8$ kg/s and $a_c = 45.5$
 629 m. For depths greater than the choking limit, the Mach number falls off rapidly to values of 0.5 and 0.1
 630 for depths equal to about 10 and 30 vent radii, respectively. For sufficiently large water depths and small
 631 MER, we find no conduit solutions in which fragmentation occurs (blue region, panel (a) top-left). As
 632 introduced in Section 2.2, for hydrovolcanic runs we adjust the MER relative to control runs to match the
 633 vent boundary condition. Figure 5c shows the ratio of adjusted MER to control MER, q_c/Q_0 , which for
 634 control simulations is always equal to 1 by definition. The adjustment is minor (no more than about 10%)
 635 and positive in most cases where vent choking is maintained. For water depths greater than the choking
 636 threshold, q_c begins to decrease, reaching values as low as 20-30% of Q_0 for low MER and large water
 637 depths. This trend is, however, not universal: for low MER, a strong second nucleation event occurs near
 638 the fragmentation depth and leads to relatively larger values of released gas and consequently greater MER
 639 until water depths of about 150 to 200 m (panels (c) and (f), lower-left corner).

640 Figure 5d shows the peak magma decompression rate \dot{p} at the fragmentation depth. Where the choking
 641 condition holds, peak decompression rate ranges between about 4 and 7 MPa/s and varies with MER, but
 642 for all depths greater than about 5 vent radii, decompression rate decreases, falling to values well below
 643 3 MPa/s for depths greater than about 15 to 20 vent radii. The blue dashed line in panel (d) shows the
 644 maximum water depth for which peak bubble overpressure $\Delta p_b = p_b - p_m$ (i.e. the difference between
 645 the gas pressure inside bubbles and pressure in the ascending magma *at the fragmentation depth*) is equal
 646 to 5 MPa, which is an approximate low bound for the tensile strength of the magma (Cas and Simmons,
 647 2018). Our fragmentation criterion allows fragmentation regardless of peak decompression rate or bubble
 648 overpressures, so long as sufficient vapor exsolution occurs to reach a porosity of 75%. However, the
 649 decrease in both maximum decompression rate and maximum bubble overpressure with increasing water
 650 depth has important implications if alternative criteria for magma fragmentation are considered, which
 651 we discuss further in Section 4.3. Fragmentation depth (panel (e)) is governed by decompression and gas
 652 exsolution rates and decreases with both increasing MER and increasing hydrostatic pressure, reaching
 653 about 500 m at its shallowest for the largest values of MER and water depth. As shown in Figure 5f, we
 654 find that for $Q_0 \lesssim 3 \times 10^6$ kg/s and $Z_e \lesssim 150$ to 200 m, a second nucleation event in the conduit near
 655 fragmentation results in a notably higher total gas exsolution from pyroclasts (a difference of up to about
 656 0.5 wt%). Higher total gas exsolution increases the free gas mass fraction at the vent, which in turn slightly
 657 boosts vent overpressure and adjusted MER. Importantly for our results, enhanced gas exsolution alters the
 658 glass transition temperature according to Equation 41, with consequences for quench fragmentation during
 659 MWI which we discuss below.

660 3.2 MWI model and the effects of water entrainment

661 Figure 6 shows MWI model results for four simulation scenarios with different water entrainment
 662 parameterizations: the *Reference* scenario (blue) with scalings for both decompression length (L_d , equations
 663 13 to 16) and mixing length (L_X , equations 25 through 29), no mixing length scaling (*No- L_X* , red), no
 664 decompression length (*No- L_d* , purple), and with the weighted Rayleigh-Taylor entrainment coefficient in
 665 Equations 31 and 32 (α_{RT} , light blue). In the simulation shown ($q_c = 1.03 \times 10^8$ kg/s, and $Z_e = 120$ m),
 666 the jet in the *Reference* scenario begins entraining water after decompression at a height of about 55 m
 667 above the vent. In contrast to a sub-aerial jet, the gas jet is buoyant in sub-aqueous settings and accelerates
 668 towards the water surface (panel (a)). Bulk temperature (panel (b)) decreases with water entrainment,
 669 and bulk density (panel (c)) decreases from both an increase in the vapor mass fraction (panel (d), solid
 670 lines) and decompression as the jet moves upwards in the water column. New ash surface area is produced
 671 through quench fragmentation (panel (e)), proportional to the mass of water ingested. This process proceeds
 672 until the mixture cools below the glass transition at a height of about 105 m above the vent (marked with
 673 circle symbols in panels (b) and (e)), after which no additional ash surface area is generated. The effective
 674 entrainment coefficient (panel (f)), scaled by L_X (Equation 27), grows approximately linearly from an
 675 initial value of zero according to Equation 27, resulting in a continuous increase in the rate of water
 676 ingestion. In the *No- L_X* scenario, the entrainment coefficient is equal to that given by Equation 23. Here,
 677 the entrained mass of water rises much more sharply with height and causes the mixture to reach the glass
 678 transition by around 10 m of above the decompression length L_D . Furthermore, in these calculations water
 679 vapor saturation is reached after only 25 m of rise. Above water saturation, the liquid water fraction in the
 680 jet increases rapidly with height (panel (d), dashed lines). The concomitant increase in density reduces
 681 jet acceleration relative to the *Reference*, until breach of the water surface occurs. In the *No- L_d* scenario,
 682 the entrainment coefficient initiates at a value of zero as in the *Reference*, but entrainment begins from
 683 $z = 0$ rather than $z = L_d$. The crossover length $L_T = 230$ m is greater than water depth for this event, and
 684 consequently the entrainment rate increases over the full height of the water layer (see Equations 28, 29),
 685 reaching a larger maximum value at the water surface ($\alpha = 0.76$ versus $\alpha = 0.4$ in the *Reference*). The
 686 bulk mixture temperature for the *No- L_d* scenario reaches the saturation temperature at a height of 80 m,
 687 and ultimately a similar total mass of entrained water to the *No- L_X* scenario on reaching the water surface
 688 (about 45 wt.%). The α_{RT} scenario uses a weighted combination of entrainment coefficients driven by
 689 buoyancy and turbulent shear. Buoyancy-driven entrainment in Equation 32 is approximately proportional
 690 to the surface area to volume ratio of the plume, i.e. $\alpha_{RT} \propto a^2/q_c$. For the relatively large MER shown
 691 here, q_c dominates in the above ratio resulting in a low value of α_{RT} , and the weighted α_{eff} is consequently
 692 a middle value between the *Reference* and *No- L_X* scenarios. We further discuss the consequences of these
 693 water entrainment scenarios in Sections 3.3 and 4.1.

694 For a specified fragmentation efficiency ζ , the production of ash surface area from quench fragmentation
 695 increases with the extent of water entrainment, which increases with water depth (see Equation 34). Quench
 696 fragmentation proceeds rapidly compared with the timescale for the jet to cross the water layer (Figures
 697 3d and 6e). In the model, the primary limit for fine ash production is, thus, the height at which water
 698 entrainment causes the mixture temperature to become less than the glass transition temperature. For
 699 $C_m = 1250$ J/(kg K) and $T_0 = 1123$ K, this condition is met where $n_e \gtrsim 0.12$. However, even with
 700 this imposed temperature limit for quench fragmentation, Figure 3d shows that the PSD is substantially
 701 enriched in fine ash for this mass fraction of entrained water. For an initial PSD exponent of $D = 2.9$
 702 (Figure 3d, light grey line), the mass fraction of ash particles less than $120 \mu\text{m}$ ($\phi \leq 3$) is about 45%, while
 703 it is 80% after the glass transition is passed (Figure 3d, black line). Therefore in the absence of the glass
 704 transition limit, coarse particles could be fully depleted. In Section 4 we further discuss the consequences

705 of our choice of fragmentation model and the associated key parameters: initial PSD, particle roughness,
706 fragmentation energy efficiency, and glass transition temperature.

707 3.3 Effects of the water layer on column rise

708 Figure 7 compares eruption column model results for example control and hydrovolcanic simulations.
709 Dashed grey lines show parameters of the ambient atmosphere. The control scenario (in red) inherits
710 conditions directly from sub-aerial vent decompression: bulk density (panel (a)) is determined by the mass
711 fractions of pyroclasts and magmatic vapor (shown in panels (e) and (f), respectively), velocity (panel (b))
712 is equal to the mixture sound speed, and the bulk temperature is equal to the initial value in the conduit
713 (panel (d)). The jet cools rapidly with entrainment of ambient air and condensation of water vapor begins
714 shortly above the vent, though the liquid mass fraction remains below 1% (panel (f), dashed lines). The jet
715 becomes buoyant (density less than ambient atmosphere) within a few hundred meters of the vent, becomes
716 negatively buoyant above the neutral buoyancy height of about 9 kilometers above the vent (Z_{nbl}), and rises
717 to a maximum overshoot height Z_{max} of over 12 km. In contrast, the hydrovolcanic simulation emerges at
718 the water vapor saturation temperature, $T_{sat} = 367K$, with a total water mass fraction of 46% (near the
719 threshold for gravitational collapse). Acceleration through the water layer results in a higher initial velocity
720 relative to the control simulation (see Figure 6a), and the high mass fraction of water vapor gives the initial
721 jet a relatively low density. However, due to the low temperature and increasing density from condensation,
722 the hydrovolcanic jet generates buoyancy much more slowly than in the control case, becoming buoyant
723 relative to ambient 3 km above the vent. The reduction in total buoyancy flux results in maximum height
724 and neutral buoyancy level approximately 1.5 km and 700 m less than the control case, respectively.

725 To demonstrate behavior of the coupled system, Figure 8 shows values of controlling parameters in
726 the conduit, vent, and column model components for *Reference* simulations with $Q_0 = 10^8$ kg/s and
727 varying water depths $0 \leq Z_e \leq 300$ m. Figure 8a compares the eruption column maximum height and
728 level of neutral buoyancy (in km above sea level) against tropopause and vent altitudes. Panels (b) through
729 (e) highlight parameters of the conduit including adjusted MER q_c , fragmentation depth Z_{frag} , vent
730 overpressure β , and vent Mach number M . Panels (f) through (i) show output of the MWI model. Panel (f)
731 shows the scalings for decompression L_d and crossover length L_X , and panel (g) shows the maximum value
732 of the effective entrainment coefficient over the height of the water layer (as determined by equations 23
733 and 29, see Figure 6f). Panels (h) and (i) show jet radius and velocity, respectively, at two different heights:
734 after decompression $z = L_d$ and at the water surface level $z = Z_e$ (water surface level also corresponds to
735 the eruption column source height as shown in Figure 2). Finally, panels (j) and (k) show the water mass
736 fractions (vapor and liquid) and temperature for the eruption column source (i.e. $z = Z_e$). In all panels
737 in Figure 8, vertical dashed lines show the threshold water depths for four important behavior regimes:
738 (1) the height at which water depth and decompression length are equivalent $L_d = Z_e$, (2) the water
739 depth above which the subaerial eruption column collapses before reaching a level of neutral buoyancy,
740 (3) transition at the vent between a pressure balanced jet at high Z_e and one that is overpressured and
741 choked ($\beta \gtrsim 1.05$, $M \gtrsim 0.95$) at lower Z_e , and (4) the depth above which the water dryness fraction
742 $x_v \lesssim .05$, where at most minor quantities of steam breach the water surface (the “steam plume” condition
743 as introduced in Section 2.3.3). The decompression length L_d defines the lower limit for water entrainment
744 to start, and decreases with increasing hydrostatic pressure: For water depths in excess of L_D (panel (f)),
745 water begins to entrain and mix into the jet, whereas our decompression length scaling prevents water
746 ingestion for shallower depths (panel (g)). As the water mass fraction increases above about 30%, the
747 water saturation temperature is reached and the column source includes liquid water (panel (j)), increasing
748 its density. Consequently, jet velocity (panel (i)) decreases for greater water depths, and combined with
749 reduced heat content in the particle fraction to generate buoyancy (panel (k)), it becomes impossible

750 for the jet to undergo a buoyancy reversal, and gravitational collapse occurs (panel (a)). Since the vent
 751 maintains the choked and overpressured condition until depths greater than the collapse threshold, the
 752 collapse condition for the subaerial column is not significantly influenced by changes in conduit conditions
 753 with increasing water depth, and is primarily determined by the mass fraction of entrained external water.
 754 At the upper limit for water entrainment, once the water mass fraction reaches ~ 0.7 , the heat budget of the
 755 pyroclasts is largely exhausted and most of the plume water ($\gtrsim 95\%$ by mass) is in liquid form, resulting in
 756 steam plume conditions where the a dense pyroclast jet collapses within at most ~ 1 km above the water
 757 surface.

758 Figure 9a shows total plume water mass fraction at the base of the subaerial eruption column as a function
 759 of MER and water depth for the *Reference* scenario. For comparison, the vent radius is marked in purple.
 760 The shaded light gray region highlights conditions for which stable buoyant plumes form, whereas collapse
 761 occurs for all simulations outside this region. At slightly lower water depths than the collapse threshold
 762 and for $\text{MER} \gtrsim 10^6$ kg/s, buoyant plumes breach the tropopause (tropopause height $Z_{tp} \approx 8.6$ km a.s.l. for
 763 the high latitude atmosphere used in the *Reference* scenario). The critical conduit MER for stratospheric
 764 injection, Q_{crit} , is highly sensitive to water depth. For example, the MER required for a buoyant column to
 765 reach the tropopause for a water depth of 150 m is over 10 times that for a water depth of 50 m, and nearly
 766 100 times that for a subaerial vent. This is driven primarily by the shift of the column collapse condition
 767 with increasing water depth (see also Figure 10). A notable feature is that for $\text{MER} \gtrsim 10^{8.3}$, the column
 768 collapses for the control case with no external water, but becomes a buoyant column for entrained water
 769 mass fractions up to $\sim 30\%$. In addition, low MER eruptions are able to support higher mass fractions of
 770 external water without collapse (e.g. $n_w \approx 45\%$ for $q_c = 10^7$ kg/s versus $n_w \approx 35\%$ for $q_c = 10^8$ kg/s).
 771 The relative buoyancy of low MER columns is caused by more efficient entrainment of air at smaller jet
 772 radii, as well as entrainment of atmospheric humidity and condensation and latent heat release in the plume.
 773 We note that condensation of atmospheric moisture has a more significant impact on buoyancy for smaller
 774 MER in the condensation parameterization used here (Glaze et al., 1997; Aubry and Jellinek, 2018). The
 775 solid blue line in Figure 9a marks the threshold where weak steam plumes may form, or fail to breach
 776 the water surface entirely for greater depths still. In the *Reference* scenario, the steam plume threshold is
 777 approximately coincident with the water depth limit for choked and overpressured vents. This limiting
 778 condition is a consequence of greater entrainment efficiency near the choking limit; Since $L_d \rightarrow 0$ as
 779 $\beta \rightarrow 1$, and entrainment rate grows over the height of the water column until $z = L_d + L_X$, maximum
 780 water entrainment rates are favored for pressure-balanced jets. However, the choking and steam plume
 781 limits need not be coincident, as shown in Figure 9b.

782 Figure 9b shows the threshold water depths for failed plumes (dashed lines) and stratospheric injections,
 783 (solid lines), for a subset of the simulation scenarios (see Table 2). The black lines in panel (b) are for
 784 the *Reference* scenario with high latitude (Iceland) atmosphere, (corresponding to the solid blue line for
 785 steam plumes and solid black line for stratospheric injection in panel (a)). Blue lines show the scenario
 786 for low latitude (Equador) atmosphere (*Low-Lat*). Neglecting the effects of wind, atmospheric humidity,
 787 stratification, and tropopause height are the primary drivers of differences between these two scenarios,
 788 particularly affecting the low values of Q_{crit} for water depths less than about 60 m. The remaining lines in
 789 Figure 9b show the results of the different entrainment scenarios in the MWI model as shown in Figure
 790 6 and Table 2. With the exception of the αRT scenario, these alternative scenarios for water entrainment
 791 lead to more rapid mixing of the jet with external water, thereby reducing the maximum depth of water
 792 through which the jet can penetrate and increasing the critical MER required to reach the tropopause. For
 793 the αRT scenario, the dependence of the entrainment coefficient on jet surface area to volume ratio (see
 794 Equation 32) causes the collapse and steam plume conditions to occur at shallow water depths compared to

795 *Reference* scenario for $Q_0 \lesssim 10^7$. In contrast as $Q_0 \rightarrow 10^8$, collapse conditions still occur for shallower
 796 water depths than the *Reference*, but the steam plume condition occurs at greater depths. For large MER,
 797 jet radius expands rapidly as the jet rises in the water column due to both decompression and an increase in
 798 steam volume fraction. As a consequence, α_{RT} decreases with height in the water column, reducing water
 799 entrainment rate and delaying the point at which the steam plume condition is reached. Critically, for all
 800 entrainment scenarios considered here, and regardless of the choice of atmospheric profile, we find that
 801 only the largest eruptions with $Q_0 \sim 10^9$ kg/s breach the tropopause for water depths greater than about
 802 200 m.

803 Figure 10 shows example results of eruption column height at both high latitude (*Reference* scenario, left
 804 column) and low latitude (*Low-Lat*, right column). Panels (a), (b) show column heights at varying water
 805 depth for three control values of MER, and (c), (d) show heights for varying MER at three fixed values of
 806 water depth. Solid lines show maximum column height, dashed lines show neutral buoyancy height, open
 807 circles show thresholds for column collapse, and closed circles show the threshold for steam plumes. The
 808 dominant effect of added external water on column height is to drive column collapse, which is consistent
 809 with the results of previous integral models of hydrovolcanic columns (e.g. Koyaguchi and Woods, 1996;
 810 Mastin, 2007b). Panels (a) and (b) show that for buoyant plumes, column height is essentially unchanged
 811 for water depth below decompression length, while for greater depths there is a 10 to 25% decrease in
 812 column height. For relatively low water depths and low MER, the release of latent heat drives increased
 813 column height, particularly from entrained atmospheric moisture in a humid atmosphere (e.g. panel (b)
 814 for $Z_e = 20$ m and $Q_0 = 10^6$ kg/s). However, for the high latitude atmosphere this is largely offset by the
 815 decreases in total height resulting from changes to column source parameters (e.g. panel (a) for $Z_e = 70$ m
 816 and $Q_0 = 10^7$ kg/s, see Figure 8). Therefore in most cases, we find that both both maximum height and
 817 neutral buoyancy levels of plumes decrease relative to the control simulations for increasing water depth.
 818 For buoyant plume scenarios with non-zero mass fraction of external water ($Z_e > L_d$), neutral buoyancy
 819 levels are typically reduced by 10 to 25%. Panels (c) and (d) show that increasing water depth narrows the
 820 range of MER for which buoyant columns may form. For example, at only 100 m of water depth, buoyant
 821 columns are restricted to MER between about 3×10^7 and 2×10^8 kg/s for the reference scenario, and an
 822 even narrower range for the low latitude atmosphere. Water depths greater than about 200 to 250 m result
 823 in either column collapse or failed plume conditions in our *Reference* our simulations, except for very large
 824 MER $\sim 10^9$ kg/s.

825 3.4 Evolution of Particle Surface Area With Fragmentation and Sedimentation

826 Figure 11a shows particle specific surface area S (surface area per unit mass of particles) at the water
 827 surface after MWI, as a function of the concentration of residual water dissolved in the melt, c_{H_2O} , and
 828 is a metric for fine ash production. Symbol size represents MER for all panels in Figure 11 and colors
 829 denote the mass fraction of entrained external water. The upper limit of S following quench fragmentation
 830 is determined in the model primarily by the glass transition temperature, T_g . Simulations with high rates
 831 of exsolution in the conduit (particularly those with strong second bubble nucleation events near the
 832 fragmentation depth, see Figure 5f) result in lower c_{H_2O} and higher T_g (see Equation 41 and Figure 3b)
 833 upon entering the water layer. Higher T_g in turn reduces the total thermal energy available for production
 834 of fine ash during quench fragmentation, and these events have PSD's with consequently lower particle
 835 surface area. Since total gas exsolution is inversely correlated with Q_0 in our conduit model, values of S
 836 after quench fragmentation increase with increasing Q_0 , as shown by symbol size in Figure 11a.

837 Figure 11b shows S at both column source (i.e. water surface $z = Z_e$, grey symbols) and at maximum
 838 column height ($z = Z_{max}$, blue symbols) as a function of the water mass fraction at the plume source.

839 In both panels (b) and (c), circles show buoyant plumes that do not breach the tropopause, 'x' symbols
840 show collapsing columns, and diamonds show plumes that are both buoyant and of sufficient magnitude to
841 breach the tropopause at the height of neutral buoyancy Z_{nbl} . Considering first values of S at the eruption
842 column source (grey symbols, panel (b)), the sharp plateau in S above $n_w \approx 0.15$ in panel (b) is a result of
843 cooling below the glass transition temperature, marked with a vertical blue bar (see also Figure 6e). For
844 entrained water mass fractions greater than this, quench fragmentation halts and S remains approximately
845 constant at a value determined primarily by the glass transition and the size of particles produced by quench
846 fragmentation (see Section 2.3.4 and Figure 3).

847 Blue symbols in panel (b) highlight the effects of sedimentation on ash surface area over the rise of the
848 subaerial eruption column. The PSD is further enriched in fine ash following fallout of coarse particles, and
849 S consequently increases with height of the eruption column. Furthermore, because the local rate of particle
850 loss from the edges of entraining eddies is proportional to the ratio of particle fall speeds to the mixture
851 rise speed according to Equation 56, buoyant plumes with low MER, rise velocities, and radii have the
852 largest increase in S during column rise. For collapsing columns ('x' symbols), S increases proportional to
853 maximum height prior to collapse. Owing to a combination of fines enrichment from quench fragmentation
854 and enhanced sedimentation due to reduced column rise speeds, all buoyant hydrovolcanic plumes (circle
855 and diamond symbols) increase in particle specific surface area at their maximum height with increasing
856 mass fraction of water.

857 The combined effects of quench fragmentation followed by sedimentation in the rising column influences
858 both total retained mass of ash in the eruption cloud and the surface area per unit mass of particles. Figure
859 11c shows the fraction of total erupted particle mass remaining in the column at its maximum rise height,
860 again as a function of water mass at the column source; symbols are as in panel (b), with colors showing S
861 at maximum column height. Small eruptions that do not reach the tropopause (circle symbols) lose the
862 greatest portion of their particle mass to sedimentation, while collapsing columns retain mass up to their
863 (relatively much lower) maximum height before collapsing entirely. Of note, however, are the subset of
864 eruptions that are both buoyant and of sufficiently high magnitude to breach the tropopause (highlighted
865 with an arrow in panel (c)). With increasing water mass fractions, such events not only retain a greater
866 portion of their initial pyroclast mass relative to control runs, but also have a more fines-enriched PSD in
867 the spreading cloud as measured by the S parameter. Provided they generate buoyant eruption columns, the
868 above results highlight the greater total flux of ash surface area to the spreading cloud for hydrovolcanic
869 scenarios, with important implications for chemical and microphysical interactions with SO_2 .

4 DISCUSSION

870 Here for the first time, we link coupled dynamics of flow in a volcanic conduit, vent, and eruption column
871 for hydrovolcanic eruptions. In marked contrast to previous studies which parameterize the mass fraction
872 of external water ingested into the subaerial eruption column source (e.g. Koyaguchi and Woods, 1996;
873 Mastin, 2007b; Van Eaton et al., 2012), we interrogate eruption dynamics that evolve with magma-water
874 interactions that depend explicitly on the depth of an external water layer. The dynamics of integral conduit
875 and eruption column models are well established (Gonnermann and Manga, 2007; de' Michieli Vitturi and
876 Aravena, 2021; Woods, 2010). Consequently, here we focus on effects of a water layer on the couplings
877 among the conduit, vent and eruption column model components and their consequences for column rise
878 and gravitational stability. We identify critical water depth conditions where column heights exceed the
879 tropopause, explore sensitivities of these results to parameterizations for water entrainment and quench
880 fragmentation, and compare results to observations of hydrovolcanic eruptions. We address, in particular,
881 how key parameters in the fragmentation model influence the fragmentation energy budget and govern the

882 production of particle surface area (ash). In addition to modulating the rise of a hydrovolcanic eruption
883 column, the extent of ash production potentially affects also the SO₂ absorption and the heterogeneous
884 nucleation and growth of sulfur aerosols. Thus, we conclude by discussing the co-injection across the
885 tropopause of ash, SO₂, and water in hydrovolcanic eruption clouds and implications for chemistry,
886 microphysics, and associated climate impacts.

887 **4.1 Water Entrainment and Mixing Efficiency Governs Eruption Column Buoyancy**

888 For a given MER, the model parameter that exerts the greatest control on injection height and mass of
889 fine ash and water is the effective water entrainment coefficient α_{eff} . For a given water depth, the height
890 above the vent at which water entrainment effectively begins and the rate at which water ingestion occurs
891 govern the total mass of external water introduced into the column. The resulting water budget controls,
892 in turn, the total thermal energy transfer from the melt to heat external water and supply the irreversible
893 work to fragment pyroclasts to produce ash. The extent and rate of water entrainment therefore governs the
894 conditions for column collapse or buoyant rise, the extent of fine ash production by quench fragmentation,
895 and the depth at which water vapor is largely exhausted and the pyroclastic jet transitions to a weak steam
896 plume. To make clear the insight gained through our considering the controls on the entrainment mechanics
897 that govern column evolution, we will discuss in detail the behavior of our different entrainment scenarios.
898 For comparison, we introduce natural examples of eruptive phases that involve interactions with water
899 layers of various depth.

900 Except in the special case where the column does not decompress on exiting the vent, the decompression
901 length L_D acts to reduce the fraction of the water column height where entrainment can occur. Over the
902 height to the crossover length L_X , where turbulent buoyant plume rise starts, the evolving local rate of
903 entrainment is less than the steady-state value above L_X . These expectations are broadly consistent with
904 Saffaraval et al. (2012) who demonstrate that for overpressured jets, entrainment was 30 to 60% less efficient
905 at axial distances less than about 5 vent diameters and vent overpressures up to about 3 atmospheres. In more
906 detail, over the decompression length L_D water entrainment is impossible by definition and none occurs
907 where $L_D > Z_e$. In contrast, for $L_D \leq Z_e$ water ingestion is possible and enhanced for (shallow) water
908 depths greater than around 2 vent radii because increases in hydrostatic pressure suppress decompression
909 (Figure 8f). Consequently, with no decompression scaling (*No- L_d* scenario), whereas the threshold depth
910 for steam plumes is, for example, not significantly affected because the decompression length is very small
911 at these depths (see Figure 8f), the threshold water depth for column collapse and stratospheric injection
912 decreases by ~ 20 to 30% (see Figure 9b).

913 The mechanism of decompression length inhibiting water entrainment in our model can be related to
914 observations of real eruptions in shallow water layers. For example, the 2016-2017 eruption of Bogoslof
915 volcano featured both transient explosions and sustained plumes emerging from vents typically in water
916 depths of 5 to 100 m (Lyons et al., 2019). Lyons et al. (2019) interpreted acoustic signals of transient
917 events at Bogoslof to result from explosive expansion of large bubbles of magmatic gas, which limited
918 the direct interaction of external water with the erupting fragmented mixture. Deposits from these events
919 in the near-vent region suggested that little or no condensed water was present during emplacement of
920 pyroclastic surges, and Waythomas et al. (2020) interpreted this to mean that any water present was entirely
921 in vapor form, further suggesting that these explosive events were drier than is typical of “Surtseyan”-type
922 activity. The requirement for low liquid water content in pyroclastic surges at Bogoslof, combined with
923 the observations of Lyons et al. (2019) suggests either a highly efficient mixing process and complete
924 vaporization (possibly driven by molten-fuel-coolant explosions (Wohletz et al., 2013)), or limited ingestion
925 of external water by explosive expansion of magmatic gas in a shallow water setting. Whereas events in

926 our model with water depths less than L_d result in no incorporation of external water, we suggest this
927 regime is analogous to real events similar to those of Bogoslof where water depths are comparable to or
928 less than length scales for gas decompression, resulting in *limited* (though likely non-zero) amounts of
929 external water incorporated into the eruption column. An overpressured vent is required for this event to
930 occur, which is possible for either a steady eruption with choked vent flow, or for transient explosions
931 originating in the shallow conduit. In our simulations, pyroclasts cool to the water saturation temperature
932 around water mass fractions of 30-35% assuming that mixing and heat transfer are complete, at which point
933 the liquid water content rises dramatically. This is therefore a likely upper bound for the mass fraction of
934 external water in these relatively dry events at Bogoslof.

935 The crossover length scale L_X governs where in the water layer column rise transitions from that of a
936 pure jet to a turbulent buoyant plume. At and above this transition, entrainment by turbulent motions is
937 fully developed (see Equation 23). The crossover length is most sensitive to jet radius and velocity after
938 decompression (see Equation 27). The column rise speed changes little over L_D so long as the conduit
939 remains choked. However, the jet radius after decompression decreases rapidly with increasing hydrostatic
940 pressure and decreasing vent overpressure, and for deep water L_X approaches a value less than half of
941 that for a subaerial jet (see Figure 8 panels (d), (f), and (h)). As L_X decreases with increasing water depth,
942 α_{eff} increases more rapidly with height above the vent (see Equations 28, 29) and the jet entrains external
943 water at slightly greater rates for deeper water layers. However, more important remains the total height
944 over which water entrainment occurs. Without considering the crossover length scale (*No- L_X* scenario),
945 entrainment sufficient to cause column collapse or steam plumes occurs within only a few tens of meters of
946 where entrainment starts, even for very large MER (see Figure 9b). Because of the progressive increase of
947 α_{eff} with height in scenarios that include the L_X scaling, removing it in the *No- L_X* scenario has a greater
948 impact on the threshold for steam plumes than for the column collapse condition, relative to the *No- L_d*
949 scenario.

950 By definition, the *No- L_d -No- L_X* scenario has entrainment at rates corresponding to those for fully
951 developed turbulence in subaerial jets (e.g. Morton et al., 1956; Carazzo et al., 2008), and even for the
952 largest MER leads to ingestion of water masses sufficient to overwhelm jets that would otherwise lead to
953 stratospheric injections. For example at $Q_0 \approx 10^9$ kg/s stratospheric injection is prevented at water depths
954 greater than about 60 m, compared to a limit of 250 m in the *Reference* scenario (see Figure 9b). The
955 entrainment rates and collapse conditions in the *No- L_d -No- L_X* scenario are therefore likely inconsistent
956 with real hydrovolcanic eruptions. For example the $\sim 24,000$ BP Oruanui hydrovolcanic eruption in New
957 Zealand, had estimated magma mass fluxes of 10^8 to 10^9 kg/s and is recognized for its remarkably wide
958 dispersal of airfall deposits (Wilson, 2001). This eruption emerged through Lake Taupo, which in modern
959 times has water depths averaging about 150 m, and is believed to have had depths of at least 100 m at the
960 time of the eruption (Nelson and Lister, 1995). These inferences are consistent with little water entrainment
961 and mixing in the near-field and reinforce the importance of considering L_d and L_X in the evolution of
962 buoyant subaerial columns from submerged volcanic jets.

963 The isothermal, single-phase experiments of Zhang et al. (2020) show that fully developed turbulence
964 with steady-state entrainment in subaqueous, supersonic jets occurs at a distance from the vent greater than
965 about ten vent diameters, with comparatively inefficient and transient entrainment modes dominating closer
966 to the jet source. For such subaqueous jets, both turbulent shear and buoyancy effects contribute to the
967 development of large turbulent eddies that inject surrounding water. For comparison with the typical shear-
968 driven entrainment condition used in our *Reference* scenario and to highlight potential variability in the
969 entrainment mechanisms of real sub-aqueous volcanic jets, we parameterize buoyancy-driven entrainment

970 in the αRT scenario using a slightly modified form of the “Rayleigh-Taylor” entrainment coefficient of
971 Zhang et al. (2020) in Equations 31 and 32. Differences between the αRT and *Reference* scenarios (see light
972 blue and black lines in Figure 9b, respectively) are governed by the $\alpha \propto a^2/q_c$ dependence of Equation
973 32. For $Q_0 \lesssim 10^7$, the ratio of jet cross-sectional area to mass flux a^2/q_c is relatively large, resulting in
974 large entrainment rates comparable to those for fully developed plumes (i.e. *No- L_d -no L_T* scenario) and
975 consequently shallow water depths for the column collapse and steam-plume conditions. For $Q_0 \gg 10^7$
976 kg/s, as entrained water is vaporized jet density initially decreases, resulting in enhanced Rayleigh-Taylor
977 entrainment and column collapse for slightly shallower depths than the *Reference* scenario. However, for
978 larger water depths where the jet cools to the water saturation temperature, entrained water remains liquid,
979 jet density increases and radius decreases (see Figure 8, panels (h) and (j)). As a result, q_c dominates
980 in Equation 32 for water depths much greater than the threshold for collapse, and entrainment rates are
981 suppressed. The reduced entrainment rates for large MER and deep water layers, in turn, prevent total
982 exhaustion of the particle heat budget such that, in contrast to other scenarios, the steam plume condition
983 occurs for pressure-balanced jets much deeper than the limit for vent choking (c.f. Figure 9b)). As a final
984 remark here, we reiterate that the mechanics of water entrainment exert the greatest control over column
985 rise. Our results underscore, however, that this process is poorly understood and is a key avenue for future
986 work on hydrovolcanism. As implemented, the shear-driven and buoyancy-driven modes govern water
987 ingestion for very different MER-water depth conditions. Whereas it is straightforward to embrace both
988 contributions parametrically through the effective entrainment coefficient given by Equation 31, there are
989 no observational or experimental constraints on how best to characterize the relative contributions of each
990 mode. Furthermore, how the underlying dynamics and their couplings are modified by local MFCI as well
991 as particle inertial and buoyancy effects, as well as the character and thermal mixing properties of MWI,
992 are unknown.

993 Conditions leading to gravitational collapse in our model (water mass fractions $\gtrsim 30$ -40 wt%) are
994 consistent with those in previous integral plume models of wet eruption columns (Koyaguchi and Woods,
995 1996; Mastin, 2007b). Our results are further consistent with observations that buoyant, ash-laden subaerial
996 eruption columns are rarely observed for water depths greater than about 100 m (Mastin and Witter, 2000).
997 However, a challenge with interpretation of integral plume models is that they predict sharp boundaries
998 between behavioral regimes (i.e. collapse or no collapse), whereas real eruptions have gradual transitions
999 between behaviors. Columns that are either fully buoyant or completely collapsing are now understood to
1000 be end member behaviors, with eruption columns undergoing partial collapse and simultaneous rise of
1001 buoyant central columns and secondary plumes from pyroclastic density currents being commonplace (Neri
1002 et al., 2002; Gilchrist and Jellinek, 2021). Indeed, hydrovolcanic eruptions are noted for highly dispersive
1003 eruption columns with multiple spreading levels (Carazzo and Jellinek, 2013; Houghton and Carey, 2015),
1004 owing to complex cloud microphysical processes including latent heat exchange and hydrometeor formation
1005 (Van Eaton et al., 2012, 2015), wet particle aggregation (Brown et al., 2012; Telling et al., 2013; Van Eaton
1006 et al., 2015), or collective settling and diffusive convection (Carazzo and Jellinek, 2012, 2013). The
1007 thresholds shown in Figure 9, including for column collapse, stratospheric injection, vent choking, and
1008 plume failure are best interpreted as gradual transitions between likely behavioral regimes. Similarly, the
1009 condition for steam plumes represents a transitional regime where jets of liquid water, ash and steam can
1010 still breach the water surface and may produce water-rich plumes driven by moist convection, but the vast
1011 majority of water and particle mass collapses immediately at the surface or does not breach it at all (see
1012 Figure 8a). As an example of this regime, the eruption of South Sarigan Volcano in 2010 occurred in
1013 water depths of 180-350 m, and produced a column up to 12 km in height during its peak phase. However,
1014 satellite observations showed that the plume was very short-lived and consisted primarily of water, with

1015 only minimal ash fallout or aerosols detected (McGimsey et al., 2010; Global Volcanism Program, 2013;
1016 Green et al., 2013)).

1017 A final consideration for the development of buoyancy in the subaerial eruption column is the effect of
1018 thermal disequilibrium. To validate the assumption of thermal equilibrium in an integral model, Koyaguchi
1019 and Woods (1996) assumed timescales for heat transfer between particles and entrained water of order
1020 1 second or less, which is reasonable for particle diameters less than about 1 mm, and also requires the
1021 column to be well-mixed. For the range of water depths considered here, typical timescales for the jet to
1022 penetrate the water surface are about 0.1 to 5 seconds (assuming choked flow at the vent). Our MWI model
1023 therefore assumes entrainment and heat transfer occur on timescales < 0.1 seconds, and further assumes
1024 that internal turbulent mixing of the jet mixture with entrained water is complete on these timescales. If
1025 disequilibrium heat transfer or incomplete mixing are considered, entrained water may not vaporize fully
1026 over the timescale of rise through the water column, even for jets with bulk pyroclast temperatures well
1027 above the water saturation temperature. In turn, the subaerial jet would host domains of varying fractions of
1028 liquid water and vapor, resulting in heterogeneous density distributions in the early stages of the eruption
1029 column. Such effects are beyond the capability of a 1D integral model and could further contribute to
1030 partial column collapse or particle shedding events, with consequently reduced mass flux of particles and
1031 gas in the rising column. An additional consequence of incomplete mechanical and thermal mixing is that
1032 the column may retain a hot core of particles that do not supply thermal energy to entrained external water
1033 to drive quench fragmentation, which is consistent with observations of pyroclast textures and particle sizes
1034 (e.g. Moreland, 2017). Our assumed complete mixing and parameterized fragmentation efficiency thus
1035 probably provides an upper bound to the extent of quench fragmentation and ash production.

1036 **4.2 Trade-offs Among Thermal Energy Budget, Particle Loss, Particle Surface** 1037 **Roughness, and Fragmentation Efficiency**

1038 Our fragmentation model aims to capture the essential energy and mass budget characteristics of quench
1039 fragmentation derived from observational and experimental constraints on the glass transition temperature
1040 T_g (Dingwell, 1998), the fragmentation energy efficiency ζ (Sonder et al., 2011), particle roughness Λ
1041 (Zimanowski and Büttner, 2003; Fitch and Fagents, 2020), the initial PSD power-law exponent D (e.g.
1042 Girault et al., 2014), and measured hydrovolcanic particle sizes (Costa et al., 2016). Here we focus on the
1043 consequences of varying Λ , ζ , and D for production of fine ash. For reference, we refer to Section 3.4 and
1044 Figure 11b, which plots *Reference* scenario particle specific surface area at two heights - column source
1045 and maximum column height - as a function of column water mass fraction at the water surface. These
1046 same data for the *Reference* scenario (i.e. gray and blue diamond symbols in Figure 11b) are again plotted
1047 in Figure 12 in blue (now circles and diamonds for values at the column source and maximum height,
1048 respectively), together with results of scenarios with alternative fragmentation model parameters (see Table
1049 2). As in Figure 11b, MER is represented by symbol size. As described in Section 3.4, cooling below the
1050 glass transition temperature limits the generation of additional ash surface area for total mass fractions of
1051 water $n_w \gtrsim 0.15$. First examining the *Reference* scenario ($\zeta = 0.1$, $\Lambda = 10$, $D = 2.9$, and mean output
1052 particle size, $\phi_\mu = 3.4$; blue symbols in Figure 12), this mechanical limit results in approximately a 20%
1053 increase in ash specific surface area S at the base of the eruption column, and a 10-15% increase in S
1054 at the spreading height, relative to control scenarios. As discussed in Section 3.4, coarse particle fallout
1055 is relatively enhanced for low-MER events which have small radii and lower column rise speeds when
1056 compared with larger MER. As a consequence, sedimentation in low-MER ($\ll 10^7$ kg/s) columns exerts a
1057 stronger control on particle surface area than does quench fragmentation in our simulations, whereas the
1058 two mechanisms are comparable in magnitude for larger eruptions.

1059 Red symbols in Figure 12 show the *High- Λ* scenario, where the particle roughness scale Λ is increased
1060 from 10 to 25 and other input parameters are held constant. Similar to Fitch and Fagents (2020), Λ has
1061 the largest influence on total ash surface area. Increasing Λ to 25 results in a proportional increase in
1062 initial surface area; the minimum value of S for the *Reference* scenario with no entrained external water
1063 is 860 m²/kg, and is 2160 m²/kg for the *High- Λ* scenario. However, the energy requirement to generate
1064 particles of a given size also increases proportionally. Since the fragmentation energy budget per unit mass
1065 of pyroclasts is approximately the same as in the *Reference* scenario (determined by magma heat capacity,
1066 fragmentation energy efficiency, and the glass transition temperature), the amount of total surface area
1067 generated during MWI is similar to the *Reference* scenario, but the proportional increase in S resulting
1068 from MWI is less than 10% relative to the control simulations. Comparing change in surface area resulting
1069 from water entrainment and quench fragmentation (red circles) with that resulting from sedimentation
1070 (difference between circles and diamonds), the effects of sedimentation in this case exert a much stronger
1071 control on ash surface area in the eruption cloud than does MWI. High particle roughness scenarios thus
1072 have the greatest total ash surface area in the eruption cloud, but a relatively modest change compared to
1073 control simulations with no external water.

1074 The fragmentation energy efficiency ζ governs the relative partitioning of thermal energy loss from the
1075 melt between that used to heat and vaporize water and that consumed by fragmentation and production
1076 of particle surface area. Choosing a low value for the fragmentation energy efficiency, $\zeta = 0.05$, (*Low- ζ*
1077 scenario, yellow symbols in Figure 12) reduces the energy consumed by fragmentation per unit mass of
1078 entrained water, resulting in overall less ash production before the glass transition limit is reached. This
1079 scenario has both the lowest total particle surface area after quench fragmentation and a modest change
1080 relative to control scenarios of 5 to 10%. The high fragmentation energy efficiency scenario with $\zeta = 0.15$,
1081 (*High- ζ* scenario, data not shown) has an effect of similar magnitude but opposite sign on SSA compared
1082 with the *Low- ζ* scenario. S after sedimentation in the eruption column, however, is very similar to that for
1083 the *Reference* scenario, and we consequently do not show those results in Figure 12.

1084 The initial PSD, governed by D , determines the relative weight of particles towards fine or coarse
1085 fractions prior to MWI. Since we fix the particle sizes produced by quench fragmentation to values based
1086 on the phreatomagmatic Phase C of the Askja 1875 eruption (see Section 2.3.4 and Figure 3), an initial PSD
1087 already enriched in these particle sizes will not change significantly in our MWI model, and consequently
1088 little fragmentation energy will be consumed. The *High- D* scenario with $D = 3.2$, (purple symbols in
1089 Figure 12) results in very high initial particle surface area (~ 2050 m²/kg) but only minor changes to
1090 the PSD and S from MWI and sedimentation (the highest values of S at the maximum plume height are
1091 ~ 2200 m²/kg). Consequently, the strongest control on production of ash surface in this scenario is the
1092 minimum particle size that can be produced during quench fragmentation.

1093 The results of the various fragmentation scenarios above reveal an important trade among PSD, particle
1094 roughness, and the consumption of fracture surface energy during quench fragmentation. The primary
1095 effect of the glass transition limit and fragmentation energy efficiency is to determine the energy budget
1096 for fragmentation, whereas particle roughness and surface energy limit the mass of fine particles than
1097 can be produced within a given energy budget. The initial PSD, in turn, determines the mass of “coarse”
1098 particles available with which to generate new fine ash. The mass in this coarse fraction is dependent on the
1099 choice of particle sizes that fragment during quenching, and the preferred sizes of particles produced. Our
1100 simple mechanical energy balance model relies on a prescribed initial PSD and on a perfect conversion of
1101 fragmentation energy to the plastic work of brittle fragmentation. For a given ζ , the approach provides
1102 a crude bound that should be applied cautiously. Whereas we fix the particle sizes generated by quench

1103 fragmentation to those of a known deposit, modal particle sizes from quench fragmentation vary as a
1104 function of melt properties and cooling rates (van Otterloo et al., 2015), as well as bubble size distributions
1105 (Liu et al., 2015). Our model further assumes that quench fragmentation is a brittle failure process and
1106 requires the outer surface of pyroclasts to rapidly cool past the glass transition temperature (e.g. Mastin,
1107 2007a; van Otterloo et al., 2015). In reality, quench fragmentation in pyroclasts is a complex function
1108 of temperature contrast between melt and coolant and can continue smoothly below this point, albeit at
1109 progressively smaller rates (Woodcock et al., 2012; van Otterloo et al., 2015). Presumably the evolution
1110 depends on the rate and anisotropy at which thermal stresses and stress concentrations accumulate in
1111 response to cooling with the main consequence being that the cessation of quench fragmentation with
1112 decreasing particle temperature is probably more gradual in real eruptions than in our model. Despite
1113 these complexities, together with consideration of the entrained masses of water in hydrovolcanic eruption
1114 columns, these constraints allow initial estimation of the total mass and surface area of fine ash delivered to
1115 the spreading levels of buoyant hydrovolcanic eruption clouds.

1116 4.3 Water Layer Depth, Volatile Saturation and Fragmentation in the Conduit, and Vent 1117 Choking

1118 The additional hydrostatic pressure with a water layer overlying the vent influences the results of our
1119 coupled model in two primary ways: (1) it modulates the extent to which a vent is choked and overpressured,
1120 and (2) it controls the total amount of gas exsolved from the melt (Smellie and Edwards, 2016; Cas and
1121 Simmons, 2018; Manga et al., 2018), which, in turn, influences both the magma ascent rate and the quench
1122 fragmentation process. For water depths near the collapse threshold, magma flow at the vent is choked
1123 and overpressured (see Figure 8 panels (a),(d), and (e), and Figure 9a). Therefor the column collapse
1124 condition is not heavily influenced by changes in conduit conditions with increasing water depth, and is
1125 primarily determined by the mass fraction of entrained external water. However, for water depths sufficient
1126 to suppress vent overpressure, $L_d \rightarrow 0$ and L_X approaches its minimum value. Entrainment consequently
1127 starts near the vent and ingestion rates are typically faster overall for pressure-balanced jets, which is
1128 broadly consistent with experimental comparisons of overpressured and pressure-balanced jets (Saffaraval
1129 and Solovitz, 2012). This condition leads to the tendency for the steam plume regime to coincide with the
1130 water depth limit for choking (Figure 9a). However, as discussed in Section 4.1, the choking and steam
1131 plume conditions need not coincide if entrainment rates are either very high (e.g. *No- L_d -no L_X* scenario) or
1132 very low (e.g. *αRT* scenario for $Q_0 \gtrsim 10^8$ kg/s). Therefore buoyant columns are most likely for subaqueous
1133 eruptions that are choked and overpressured at the vent as opposed to pressure-balanced, but this is not a
1134 strict requirement and depends on the dynamics of decompression and water entrainment near the vent, as
1135 well as the conditions for choking (for example the mixture sound speed).

1136 Comparing total exsolution for small and large water depths (Figure 5f), differences in vapor exsolution
1137 in the conduit model control the glass transition temperature (Figure 3b), which, in turn, governs the heat
1138 budget available for ash production during the quench fragmentation (Figure 11a). This effect is most
1139 apparent when considering events with a second nucleation event occurring in the conduit model for low
1140 MER (Figure 5f). Specifically, diffusion rate of vapor leaving the melt is sensitive to bubble number density,
1141 so a second nucleation event near fragmentation enhances exsolution rate above fragmentation, leading
1142 to the sharp change in total exsolution shown in Figure 5f. Simulations with a strong second nucleation
1143 in the conduit result in distinctly different production of ash surface area during quench fragmentation
1144 (Figure 11a for $c_{H_2O} < 0.6$ wt.%). As we will show in Section 4.4 below, the influence of this process on
1145 the dispersed mass of fine ash is apparent in our model even at the spreading height of the eruption cloud.

1146 For primary brittle fragmentation and explosive volcanism to occur during magma ascent in the conduit
1147 (i.e. without the influence of external water), either gas overpressure in bubbles must exceed the tensile
1148 strength of the melt, or the rate of magma ascent must be sufficiently high to exceed the critical strain
1149 rate for brittle failure of the melt (Papale, 1999; Gonnermann, 2015). As described in Section 3.1, both
1150 maximum decompression rate and maximum bubble overpressure (as recorded at the fragmentation depth)
1151 decrease with increasing hydrostatic pressure in our conduit model. In Figure 5d, we show that for water
1152 depths of about 200 m or greater, the maximum bubble overpressure Δp_b in our model falls below values
1153 likely to cause rupture of bubble walls. Were bubble overpressure used as the fragmentation criteria in our
1154 conduit model, fragmentation could in principle still occur, albeit at shallower depths in the conduit, but
1155 becomes increasingly less likely with increasing water depth (Campagnola et al., 2016; Cas and Simmons,
1156 2018). For example, Manga et al. (2018) used a strain-rate fragmentation criterion to estimate that for the
1157 2012 submarine eruption of Havre volcano, magmatically-driven brittle fragmentation in the conduit could
1158 only have occurred if the vent were shallower than about 290 m. It is worth noting that brittle fragmentation
1159 mechanisms in general, particularly those driven by water interaction, are not precluded at such depths,
1160 though explosive expansion of steam is suppressed (Murch et al., 2019; Dürig et al., 2020a). Critically,
1161 increasing thicknesses of water or ice will increasingly suppress the conditions for which sustained brittle
1162 or explosive fragmentation may drive gas jets or plumes, particularly those capable of reaching tens of
1163 kilometers into the atmosphere.

1164 4.4 Stratospheric Injection in Hydrovolcanic Eruptions and Implications for Sulfate 1165 Aerosol Lifecycle

1166 Radiative forcing by sulphate aerosols is governed by the total mass of injected sulfur dioxide, the height,
1167 season, and latitude of injection, and the chemical and microphysical processes that determine the resulting
1168 aerosol particle size distribution (Timmreck, 2012; Lacis, 2015; Kremser et al., 2016; Marshall et al.,
1169 2019; Toohey et al., 2019). The injection height relative to tropopause height is critical for determining
1170 the mass of stratospheric sulfur burden. However, the total mass and size distribution characteristics of
1171 fine ash as well as high water content in hydrovolcanic eruptions is also likely to play a role in the life
1172 cycle of sulfur aerosols. For example, LeGrande et al. (2016) showed that the coincident injection of
1173 SO₂ with high concentrations of water can shorten the characteristic timescale for conversion of SO₂ to
1174 aerosol from weeks to days, enhancing aerosol radiative forcing in the earliest weeks after an eruption.
1175 Chemical scavenging of SO₂ onto ash surfaces is a potentially important source of SO₂ removal both
1176 during eruption column rise and in the days and weeks following an eruption (Rose, 1977; Schmauss and
1177 Keppler, 2014; Zhu et al., 2020). Experimental results from Schmauss and Keppler (2014) demonstrated
1178 that SO₂ absorption onto ash particle surfaces is most efficient where volcanic plumes are cool, SO₂ is
1179 dilute, and ash surface areas are high - all conditions that are likely to be enhanced in hydrovolcanic
1180 eruption columns relative to purely magmatic cases. Zhu et al. (2020) reported that persistent fine ash
1181 particles dispersed along with SO₂ from the 2014 eruption of Kelut Volcano contributed to enhanced
1182 nucleation of aerosol particles onto ash surfaces and aerosol particles sizes up to 10 times that of typical
1183 background stratospheric aerosol. Critically, chemical uptake of SO₂ onto ash surfaces increased the rate
1184 of sulfur removal by sedimentation by 43% in the first two months following the eruption.

1185 Figure 13 shows estimates for the flux of SO₂, fine ash, and water to the tropopause for simulations
1186 with two different atmospheric profiles (*Reference*, top row of panels and *Low-Lat*, bottom row). Panels (a)
1187 and (b) show the estimated fraction of SO₂ delivered to or above the tropopause, where we approximate
1188 the vertical distribution of the SO₂ cloud $\psi_{SO_2}(z)$ as a gaussian profile of thickness proportional to (and

1189 centered on) injection height Z_{nbl} (Aubry et al., 2019):

$$\psi_{SO_2} = \exp\left(\frac{-(z - Z_{nbl})^2}{(0.108(Z_{nbl} - Z_e))^2}\right) \quad (59)$$

1190 The estimated fraction of SO_2 delivered to the stratosphere is the fraction of the integrated area of Equation
 1191 59 that lies above the tropopause. Events with injection heights close to the tropopause ($Q_0 \approx 3 \times 10^6$ kg/s
 1192 and $Q_0 \approx 3 \times 10^7$ kg/s in the high and low latitude atmospheres, respectively) show reduced efficiency
 1193 of stratospheric delivery of SO_2 for water depths that surpass the decompression length (and therefore
 1194 non-zero quantities of external water are entrained). The exceptions are columns in the low-latitude
 1195 atmosphere with minor quantities of entrained water ($n_w \approx 0.15$), which have increased column heights
 1196 relative to control scenarios (see Figure 10b). Panels (b) and (c) show the ratio of fine ash mass flux (particle
 1197 diameter $< 125 \mu\text{m}$) at the maximum plume height relative to control simulations. We find that events
 1198 with sufficient entrained water to pass the glass transition (and thus maximize production of fine ash in our
 1199 model) deliver a fine ash mass flux approximately 2-fold that of the control simulations. For low MER
 1200 simulations with a second nucleation event in the conduit ($Q_0 \lesssim 4 \times 10^6$ kg/s), and consequently relatively
 1201 less fine ash production, the mass flux of fine ash delivered is approximately 1.5 times that of the control
 1202 cases. Finally, panels (e) and (f) show the ratio of water mass flux at maximum plume height compared to
 1203 control scenarios. Buoyant hydrovolcanic plumes that breach the tropopause carry water mass fluxes of up
 1204 to 10 times that of control simulations. Low-latitude eruption columns in humid atmospheres entrain a
 1205 greater mass of atmospheric moisture, such that this ratio is somewhat less for the *Low-Lat* scenario, with
 1206 typical values of 2 to 7 times that of control simulations.

1207 In summary, incorporation of high mass fractions of external water in eruption columns acts to reduce
 1208 eruption column height or induce gravitational collapse, while also enhancing conditions for chemical
 1209 scavenging of SO_2 into ash and hydrometeors, including initially colder temperatures, high available ash
 1210 surface area, and abundant water. For SO_2 that does reach the stratosphere, results of LeGrande et al.
 1211 (2016) and Zhu et al. (2020) suggest that the presence of water and fine ash enhance aerosol reaction
 1212 rates and sedimentation. Our results imply that in the absence of an explicit functional dependence on
 1213 the change in PSD related to MWI, the SO_2 delivery efficiency given by Equation 59 is at best an upper
 1214 bound where eruptions interact with water layers deeper than about 50 m. On the basis of results presented
 1215 here, we suggest the hypothesis that hydrovolcanic eruption processes will on average act to reduce the
 1216 climate impacts of volcanic aerosols. However, the evaluation of stratospheric sulfur loading in volcanic
 1217 eruptions requires further analysis, particularly of microphysical processes not included in our model.
 1218 For example, moist convection in water saturated air can lead to lofting of secondary plumes even with
 1219 the occurrence of column collapse, potentially delivering SO_2 to the lower-most stratosphere following
 1220 dynamics similar to thunderstorms (Van Eaton et al., 2012; Houghton and Carey, 2015). Alternatively,
 1221 formation of hydrometeors (graupel, hail, or liquid water droplets) and aggregation of ash particles can
 1222 lead to sedimentation of fine ash and water at much higher rates than predicted by particle settling time
 1223 alone (Brown et al., 2012; Van Eaton et al., 2015), and column buoyancy and sedimentation processes can
 1224 be further modified by interaction with atmospheric cross-winds (Girault et al., 2016). If sedimentation
 1225 occurs faster than the timescales for chemical scavenging of SO_2 onto ash surfaces, this can lead to early
 1226 separation of ash and gas phases, as was observed for the 2011 eruption of Grímsvötn Volcano (Prata
 1227 et al., 2017). However, if the timescale for SO_2 scavenging is fast relative to particle fallout time as a result
 1228 of say, high particle surface area and cold column temperature (Schmauss and Keppler, 2014), then
 1229 aggregation-enhanced particle settling could act to efficiently remove scavenged SO_2 from the eruption

1230 column . For example, despite the observed separation of ash and gas clouds in the Grímsvötn eruption,
1231 Sigmarsson et al. (2013) estimated that approximately 30% of outgassed SO₂ was scavenged by ash
1232 particles and subsequently removed from the eruption cloud, with an additional 10% lost directly to the
1233 subglacial lake (16% and 5% of the total magmatic sulfur budget, respectively).

1234 4.5 Implications of Hydrovolcanism for Volcano-Climate Feedback

1235 We have discussed coupled processes in hydrovolcanic eruptions which suggest that the stratospheric
1236 sulfate aerosol climate impacts of hydrovolcanic eruptions are likely to be reduced relative to dry eruptions.
1237 This hypothesis, in turn, suggests the potential for a previously unrecognized mechanism for volcano-
1238 climate feedback, where changes to the relative extent or frequency of hydrovolcanism resulting from
1239 evolving climatic conditions in turn modulate volcanic aerosol forcing. Huybers and Langmuir (2009)
1240 suggest that globally enhanced rates of volcanism would lead to an amplifying feedback where outgassing
1241 of volcanic carbon contributed to additional warming. This hypothesis was based on the assumption that
1242 time-averaged radiative forcing of volcanic CO₂ is stronger (over century to millennial timescales) than
1243 that of short-lived aerosol cooling events. However, the potential for climate impacts on multi-decadal to
1244 millennial timescales (Zhong et al., 2011; Baldini et al., 2015; Soreghan et al., 2019; Mann et al., 2021)
1245 challenges this view, and there is open debate on whether (or under what climate conditions and/or
1246 timescales) the effects of global volcanism drive net climate cooling or warming (Baldini et al., 2015;
1247 Lee and Dee, 2019; Soreghan et al., 2019). For example, Baldini et al. (2015) suggest that large volcanic
1248 sulphate injections during the Last Glacial Maximum drove hemispherically asymmetric temperature shifts
1249 and millennial-scale cooling feedbacks. The relative global frequency of hydrovolcanism is one potential
1250 mechanism for steering the volcanic climate control in one direction or another. In particular, the outgassing
1251 of volcanic CO₂ is likely less affected by surface MWI than is SO₂, since CO₂ exsolves at initially greater
1252 crustal depths (Wallace et al., 2015) than SO₂ and its climate impacts are insensitive to injection height.
1253 Regional to global-scale changes in the occurrence of hydrovolcanism resulting from growth and decay
1254 of ice sheets or sea level rise, on timescales of centuries to millenia, could in principle modulate the
1255 global importance for climate forcing of volcanic sulfate aerosols relative to volcanic carbon and therefore
1256 alter the character (e.g. amplifying or stabilizing) of volcano-climate feedbacks. The extent to which
1257 hydrovolcanism modulates global volcano-climate forcing remains an open question, and likely depends
1258 critically on both eruption rates and the surface distribution and thickness of ice sheets overlying volcanic
1259 regions, and the resulting frequency and intensity of MWI.

1260 4.6 Summary

1261 We present a novel coupled integral model of conduit and eruption column dynamics for hydrovolcanic
1262 eruptions. We have simulated steady phases of explosive eruptions through a shallow water layer ($Z_e \lesssim 500$
1263 m) overlying the volcanic vent, including the effects of gas exsolution and magma ascent in the conduit,
1264 water entrainment and quench fragmentation, and eruption column rise and particle fallout. Based on our
1265 model results and arguments in Sections 4.1 to 4.4, we summarize key effects of changes in hydrostatic
1266 pressure and direct MWI on steady explosive eruption processes:

- 1267 1. Increasing hydrostatic pressure with water depth reduces vent overpressure and the tendency for
1268 choking in the conduit, limiting explosive decompression and reducing vent velocities. Choked vents
1269 do not occur in our simulations for water depths greater than about 5 vent diameters.
- 1270 2. Increasing hydrostatic pressure with water depth reduces gas exsolution and decompression rates in
1271 the conduit, decreasing the total fraction of gas that is exsolved on eruption at the vent, and potentially
1272 driving the eruption towards more effusive behavior.

- 1273 3. Total mass of entrained water increases with water depth, driving a decrease in eruption column heights
1274 and inducing column collapse for water mass fractions greater than about 30%.
- 1275 4. There is a range of water mass fractions (10-15%) in the starting subaerial jet in which plumes heights
1276 are increased relative to dry control scenarios as a result of high vapor mass fractions and the release of
1277 latent height with condensation. However, this window occurs only for moist, low-latitude atmospheres
1278 and for a very narrow range of water depths in our simulations.
- 1279 5. The critical mass eruption rate required for eruption columns to reach the tropopause is sensitive
1280 to increasing water depth, and is primarily governed by the column collapse condition. For water
1281 depths greater than about 200 m, only the largest eruptions ($\text{MER} \sim 10^9 \text{ kg/s}$) reach the tropopause,
1282 independent of the eruption latitude.
- 1283 6. As water depth increases well past the limit for vent overpressure ($Z_e \gtrsim 5$ vent diameters in our
1284 *Reference* scenario), the magmatic heat budget becomes exhausted, gas phases condense, and water in
1285 the jet approaches 100% liquid. Such events may still generate subaerial jets and steam plumes, but are
1286 unlikely to inject significant quantities of SO_2 or ash into the stratosphere. We find that hydrostatic
1287 pressures sufficient to suppress choking in the vent are similar to those for which minimal steam ($\lesssim 5$
1288 wt.% of the jet water phase) breaches the surface of the external water layer.
- 1289 7. Fine ash production by quench fragmentation leads in our *Reference* scenario to an approximately
1290 2-fold increase in the mass flux of fine ash ($< 125\mu\text{m}$) delivered to buoyant eruption clouds, and
1291 entrained external water increases mass flux of water to the spreading cloud by up to 10-fold.
- 1292 8. The total ash surface area available for chemical absorption of SO_2 systematically increases in
1293 hydrovolcanic scenarios relative to control cases. However, the total surface area generated is sensitive
1294 to processes governing particle fallout and to the physics of quench fragmentation (e.g. particle
1295 roughness and surface fracture energy, and the fraction of thermal energy consumed for fragmenting
1296 particles). We suggest that the high water and fine ash content and colder temperature of hydrovolcanic
1297 columns provide conditions for more efficient scavenging of SO_2 by ash and hydrometeors relative to
1298 subaerial eruptions (Schmauss and Keppler, 2014).

1299 The above results are consistent with expectations for conduit ascent in submarine and subglacial eruptions
1300 (Smellie and Edwards, 2016; Wallace et al., 2015), and for the rise of hydrovolcanic eruption columns in
1301 the atmosphere (Koyaguchi and Woods, 1996; Mastin, 2007b). Increasing water depths or ice thicknesses
1302 will furthermore drive processes not included in our model that act to reduce or prevent stratospheric
1303 dispersal, for example by restricting eruptions to effusive behavior (Manga et al., 2018) or through englacial
1304 confinement of erupted material in the case of subglacial eruptions (e.g. Gudmundsson et al., 2004). On the
1305 basis of these arguments, we hypothesize that hydromagmatic eruptions will, *on average*, tend towards
1306 reduced stratospheric loading and residence times of sulfate aerosols relative to purely magmatic eruptions,
1307 with concomitantly reduced aerosol radiative forcing. Depending on the distributions of water and ice sheets
1308 on the Earth's surface, hydrovolcanism could, in principle, modulate known volcano-climate feedbacks
1309 associated with deglaciation (Cooper et al., 2018) by limiting the radiative forcing associated with volcanic
1310 sulfur aerosols. Evaluating the climate impacts of hydrovolcanic eruptions relative to purely magmatic
1311 eruptions requires further detailed analysis of the interplay between the coupled processes of conduit ascent
1312 and gas exsolution, fragmentation mechanisms, and the fluid dynamics, microphysics, and chemistry of
1313 transport and dispersal of SO_2 , ash, and water in eruption columns.

1314 4.7 Tables

Table 1 List of variables and subscript nomenclature.

Variable	Description	Units
a	Radius of conduit or plume	m
A	Cross-sectional area of conduit or plume	m ²
C	Heat capacity	J/(kg K)
c	Sound speed	m/s
c_{H_2O}	Concentration of water dissolved in melt	wt. %
D_0	Power law exponent for initial particle size distribution	-
E_s	Particle fracture surface energy	J/m ²
\dot{E}	Energy flux	J/s
ΔE_{ss}	Specific fragmentation energy (per mass of melt)	J/(kg m)
F_{fric}	Frictional pressure loss	Pa/m
f	Friction factor	-
g	Gravitational acceleration	m/s ²
h	Enthalpy	J/kg
h_{vap}	Bulk mixture enthalpy at $T = T_{sat}$ and $x_v = 1$	J/kg
h_{vap}	Bulk mixture enthalpy at $T = T_{sat}$ and $x_v = 0$	J/kg
K	Bulk modulus	Pa
L_d	Decompression length scale	m
L_X	Crossover length scale	m
M	Mach number	-
N_i	Number of particles in size bin i	-
n	Mass fraction	-
r	Particle radius	m
r_{c1}	Critical particle radius for maximum effective porosity	m
r_{c2}	Critical particle radius for zero effective porosity	m
p	Pressure	Pa
\dot{p}	Magma decompression rate	MPa/s
Δp_b	Bubble overpressure	MPa
q	Mass flux	kg/s
q_c	Adjusted conduit mass flux (MER) for hydrovolcanic simulations	kg/s
Q_0	Reference conduit MER for control ($Z_e = 0$) simulations	kg/s
Q_{crit}	Critical MER to reach the tropopause	kg/s
S	Specific surface area of particles	m ² /kg
T	Temperature	K
T_g	Glass transition temperature lower bound	K
ΔT_g	Temperature range for glass transition	K
T_0	Initial magma temperature	K
T_{ref}	Reference temperature for enthalpy calculations	K
T_{sat}	Water saturation temperature	K
u	Vertical velocity (radially averaged)	m/s ²
x_v	Water phase dryness fraction	-
z	Vertical coordinate	m
Z_e	External surface water depth	m

Continuation of Table 1

Variable	Description	Units
Z_{tp}	Height of tropopause	m
Z_{max}	Maximum height of eruption column	m a.v.l.
Z_{nbl}	Neutral buoyancy (spreading) height of eruption column	m a.v.l.
α	Entrainment coefficient	-
α_{RT}	Rayleigh-Taylor entrainment coefficient	-
β	Vent overpressure ratio	-
ζ	Fragmentation energy efficiency	-
η	Magma mixture dynamic viscosity	Pa s
Λ	Particle roughness scaling parameter	-
λ	Water vapor condensation rate	s^{-1}
ρ	Density	kg/m^3
ϕ	Particle sieve size	-
ϕ_{μ}	Mean ϕ size of quench fragmented particles	-
ϕ_{σ}	Standard deviation ϕ size of quench fragmented particles	-
χ	Volume fraction	-
χ_i	Porosity of particle size bin i	-
χ_0	Threshold porosity for conduit fragmentation	-
ψ_{SO_2}	Gaussian profile for vertical distribution of SO_2 injection	-
ω	Jet-water interface acceleration for Rayleigh-Taylor entrainment	-
Subscripts:		
–	Bulk mixture (no subscript for material property)	
a	Dry air phase	
b	Bubble gas properties in pyroclasts	
c	Property of mixture in the conduit or vent	
d	Property after vent decompression	
e	Property of external water (MWI model) or air (plume model)	
f	“Final” value, or next iteration step	
i	Particle size bin i	
l	Liquid water phase	
m	Magma phase (excluding bubbles)	
s	“Solids” phase (melt + bubbles)	
v	Water vapor phase	
w	Total water phase in conduit or plume (liquid + vapor)	
0	Initial value	

Table 2. List of simulations sets highlighting varied model parameters: Atmospheric profile, external water temperature T_e , decompression length switch, crossover length switch, entrainment equation, PSD power-law exponent D , particle roughness scale Λ , and fragmentation energy efficiency ζ .

Name	Atmosphere	T_e (K)	use L_d ?	use L_X ?	α Equation	D	Λ	ζ
<i>Reference</i>	Iceland	274	Yes	Yes	23	2.9	10	0.1
<i>Low-Lat</i>	Equador	294	Yes	Yes	23	2.9	10	0.1
<i>No-L_d</i>	Iceland	274	No	Yes	23	2.9	10	0.1
<i>No-L_X</i>	Iceland	274	Yes	No	23	2.9	10	0.1
<i>No-L_d-No-L_X</i>	Iceland	274	No	No	23	2.9	10	0.1
<i>αRT</i>	Iceland	274	Yes	No	31	2.9	10	0.1
<i>High-Λ</i>	Iceland	274	Yes	Yes	23	2.9	25	0.1
<i>High-ζ</i>	Iceland	274	Yes	Yes	23	2.9	10	0.2
<i>Low-ζ</i>	Iceland	274	Yes	Yes	23	2.9	10	0.05
<i>High-D</i>	Iceland	274	Yes	Yes	23	3.2	10	0.1

CONFLICT OF INTEREST STATEMENT

1316 The authors declare that the research was conducted in the absence of any commercial or financial
1317 relationships that could be construed as a potential conflict of interest.

AUTHOR CONTRIBUTIONS

1318 C.R.R. was the primary study author, and performed the bulk of code development for the coupled
1319 model, including novel components. C.R.R. performed data analysis and the bulk of manuscript writing.
1320 A.M.J. was the primary investigator and holder of funding sources, and provided critical physical
1321 insight for the development of model equations and extensive discussion and review of the study results,
1322 interpretations, and manuscript writing. S.H. is the author of the conduit model and provided relevant
1323 code with modifications necessary for this study, provided code examples and critical physical insight
1324 for the development of the water-entrainment model, and authored the conduit model components of the
1325 manuscript methods section. T.J.A. provided code for the eruption column model and analysis for estimates
1326 of stratospheric injection of sulfur dioxide, and provided and advice and oversight of data analysis and
1327 interpretation related to eruption column components of the study.

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REFERENCES

1335 Ansmann, A., Tesche, M., Seifert, P., Groß, S., Freudenthaler, V., Apituley, A., et al. (2011). Ash and
1336 fine-mode particle mass profiles from EARLINET-AERONET observations over central Europe after

- 1337 the eruptions of the Eyjafjallajökull volcano in 2010. *Journal of Geophysical Research: Atmospheres*
1338 116, D00U02. doi:10.1029/2010JD015567
- 1339 Aravena, A., Vitturi, M. d. M., Cioni, R., and Neri, A. (2018). Physical constraints for effective magma-
1340 water interaction along volcanic conduits during silicic explosive eruptions. *Geology* 46, 867–870.
1341 doi:10.1130/G45065.1
- 1342 Aubry, T. J., Cerminara, M., and Jellinek, A. M. (2019). Impacts of Climate Change on Volcanic
1343 Stratospheric Injections: Comparison of 1-D and 3-D Plume Model Projections. *Geophysical Research*
1344 *Letters* 46. doi:10.1029/2019GL083975
- 1345 Aubry, T. J., Engwell, S., Bonadonna, C., Carazzo, G., Scollo, S., Van Eaton, A. R., et al. (2021a). The
1346 Independent Volcanic Eruption Source Parameter Archive (IVESPA, version 1.0): A new observational
1347 database to support explosive eruptive column model validation and development. *Journal of Volcanology*
1348 *and Geothermal Research*, 107295doi:10.1016/j.jvolgeores.2021.107295
- 1349 Aubry, T. J. and Jellinek, A. M. (2018). New insights on entrainment and condensation in volcanic plumes:
1350 Constraints from independent observations of explosive eruptions and implications for assessing their
1351 impacts. *Earth and Planetary Science Letters* 490, 132–142. doi:10.1016/j.epsl.2018.03.028
- 1352 Aubry, T. J., Jellinek, A. M., Degruyter, W., Bonadonna, C., Radić, V., Clyne, M., et al. (2016). Impact
1353 of global warming on the rise of volcanic plumes and implications for future volcanic aerosol forcing.
1354 *Journal of Geophysical Research: Atmospheres* 121, 2016JD025405. doi:10.1002/2016JD025405
- 1355 Aubry, T. J., Staunton-Sykes, J., Marshall, L. R., Haywood, J., Abraham, N. L., and Schmidt, A. (2021b).
1356 Climate change modulates the stratospheric volcanic sulfate aerosol lifecycle and radiative forcing from
1357 tropical eruptions. *Nature Communications* 12, 4708. doi:10.1038/s41467-021-24943-7
- 1358 Baldini, J. U. L., Brown, R. J., and McElwaine, J. N. (2015). Was millennial scale climate change during
1359 the Last Glacial triggered by explosive volcanism? *Scientific Reports* 5, 17442. doi:10.1038/srep17442
- 1360 Barberi, F., Cioni, R., Rosi, M., Santacroce, R., Sbrana, A., and Vecci, R. (1989). Magmatic and
1361 phreatomagmatic phases in explosive eruptions of Vesuvius as deduced by grain-size and component
1362 analysis of the pyroclastic deposits. *Journal of Volcanology and Geothermal Research* 38, 287–307.
1363 doi:10.1016/0377-0273(89)90044-9
- 1364 Bercovici, D. and Michaut, C. (2010). Two-phase dynamics of volcanic eruptions: Compaction,
1365 compression and the conditions for choking. *Geophysical Journal International* 182, 843–864.
1366 doi:10.1111/j.1365-246X.2010.04674.x
- 1367 Bonadonna, C., Ernst, G. G. J., and Sparks, R. S. J. (1998). Thickness variations and volume estimates
1368 of tephra fall deposits: The importance of particle Reynolds number. *Journal of Volcanology and*
1369 *Geothermal Research* 81, 173–187. doi:10.1016/S0377-0273(98)00007-9
- 1370 Brown, R. J., Bonadonna, C., and Durant, A. J. (2012). A review of volcanic ash aggregation. *Physics and*
1371 *Chemistry of the Earth, Parts A/B/C* 45–46, 65–78. doi:10.1016/j.pce.2011.11.001
- 1372 Bursik, M. I., Sparks, R. S. J., Gilbert, J. S., and Carey, S. N. (1992). Sedimentation of tephra by volcanic
1373 plumes: I. Theory and its comparison with a study of the Fogo A plinian deposit, Sao Miguel (Azores).
1374 *Bulletin of Volcanology* 54, 329–344. doi:10.1007/BF00301486
- 1375 Büttner, R., Dellino, P., Raue, H., Sonder, I., and Zimanowski, B. (2006). Stress-induced brittle
1376 fragmentation of magmatic melts: Theory and experiments. *Journal of Geophysical Research: Solid*
1377 *Earth* 111. doi:10.1029/2005JB003958
- 1378 Büttner, R., Dellino, P., Volpe, L. L., Lorenz, V., and Zimanowski, B. (2002). Thermohydraulic explosions
1379 in phreatomagmatic eruptions as evidenced by the comparison between pyroclasts and products from
1380 Molten Fuel Coolant Interaction experiments. *Journal of Geophysical Research: Solid Earth* 107, ECV
1381 5–1–ECV 5–14. doi:10.1029/2001JB000511

- 1382 Cahalan, R. C. and Dufek, J. (2021). Explosive Submarine Eruptions: The Role of Condensable Gas
1383 Jets in Underwater Eruptions. *Journal of Geophysical Research: Solid Earth* 126, e2020JB020969.
1384 doi:10.1029/2020JB020969
- 1385 Campagnola, S., Romano, C., Mastin, L. G., and Vona, A. (2016). Confort 15 model of conduit dynamics:
1386 Applications to Pantelleria Green Tuff and Etna 122 BC eruptions. *Contributions to Mineralogy and*
1387 *Petrology* 171, 60. doi:10.1007/s00410-016-1265-5
- 1388 Capra, L. (2006). Abrupt climatic changes as triggering mechanisms of massive volcanic collapses. *Journal*
1389 *of Volcanology and Geothermal Research* 155, 329–333. doi:10.1016/j.jvolgeores.2006.04.009
- 1390 Carazzo, G. and Jellinek, A. M. (2012). A new view of the dynamics, stability and longevity of volcanic
1391 clouds. *Earth and Planetary Science Letters* 325–326, 39–51. doi:10.1016/j.epsl.2012.01.025
- 1392 Carazzo, G. and Jellinek, A. M. (2013). Particle sedimentation and diffusive convection in volcanic
1393 ash-clouds. *Journal of Geophysical Research: Solid Earth* 118, 1420–1437. doi:10.1002/jgrb.50155
- 1394 Carazzo, G., Kaminski, E., and Tait, S. (2006). The route to self-similarity in turbulent jets and plumes.
1395 *Journal of Fluid Mechanics* 547, 137–148. doi:10.1017/S002211200500683X
- 1396 Carazzo, G., Kaminski, E., and Tait, S. (2008). On the rise of turbulent plumes: Quantitative effects
1397 of variable entrainment for submarine hydrothermal vents, terrestrial and extra terrestrial explosive
1398 volcanism
- 1399 Carey, R. J., Houghton, B. F., and Thordarson, T. (2009). Abrupt shifts between wet and dry phases
1400 of the 1875 eruption of Askja Volcano: Microscopic evidence for macroscopic dynamics. *Journal of*
1401 *Volcanology and Geothermal Research* 184, 256–270. doi:10.1016/j.jvolgeores.2009.04.003
- 1402 Cas, R. and Wright, J. (1987). *Volcanic Successions Modern and Ancient: A Geological Approach to*
1403 *Processes, Products and Successions* (Springer Science & Business Media)
- 1404 Cas, R. A. F. and Simmons, J. M. (2018). Why Deep-Water Eruptions Are So Different From Subaerial
1405 Eruptions. *Frontiers in Earth Science* 6. doi:10.3389/feart.2018.00198
- 1406 Cerminara, M., Esposti Ongaro, T., and Berselli, L. C. (2016). ASHEE-1.0: A compressible,
1407 equilibrium–Eulerian model for volcanic ash plumes. *Geoscientific Model Development* 9,
1408 697–730. doi:10.5194/gmd-9-697-2016
- 1409 Cole, P. D., Queiroz, G., Wallenstein, N., Gaspar, J. L., Duncan, A. M., and Guest, J. E. (1995). An historic
1410 subplinian/phreatomagmatic eruption: The 1630 AD eruption of Furnas volcano, Sa~o Miguel, Azores.
1411 *Journal of Volcanology and Geothermal Research* 69, 117–135. doi:10.1016/0377-0273(95)00033-X
- 1412 Colucci, S., de' Michieli Vitturi, M., Neri, A., and Palladino, D. M. (2014). An integrated model of magma
1413 chamber, conduit and column for the analysis of sustained explosive eruptions. *Earth and Planetary*
1414 *Science Letters* 404, 98–110. doi:10.1016/j.epsl.2014.07.034
- 1415 Cooper, C. L., Swindles, G. T., Savov, I. P., Schmidt, A., and Bacon, K. L. (2018). Evaluating the
1416 relationship between climate change and volcanism. *Earth-Science Reviews* 177, 238–247. doi:10.1016/
1417 j.earscirev.2017.11.009
- 1418 Costa, A., Pioli, L., and Bonadonna, C. (2016). Assessing tephra total grain-size distribution: Insights from
1419 field data analysis. *Earth and Planetary Science Letters* 443, 90–107. doi:10.1016/j.epsl.2016.02.040
- 1420 de' Michieli Vitturi, M. and Aravena, Á. (2021). Chapter 6 - Numerical modeling of magma ascent
1421 dynamics. In *Forecasting and Planning for Volcanic Hazards, Risks, and Disasters*, ed. P. Papale
1422 (Elsevier), vol. 2 of *Hazards and Disasters*. 239–284. doi:10.1016/B978-0-12-818082-2.00006-8
- 1423 Degruyter, W. and Bonadonna, C. (2012). Improving on mass flow rate estimates of volcanic eruptions.
1424 *Geophysical Research Letters* 39. doi:10.1029/2012GL052566
- 1425 Degruyter, W. and Bonadonna, C. (2013). Impact of wind on the condition for column collapse of volcanic
1426 plumes. *Earth and Planetary Science Letters* 377–378, 218–226. doi:10.1016/j.epsl.2013.06.041

- 1427 Dingwell, D. B. (1998). The glass transition in hydrous granitic melts. *Physics of the Earth and Planetary*
1428 *Interiors* 107, 1–8. doi:10.1016/S0031-9201(97)00119-2
- 1429 Dufek, J., Manga, M., and Staedter, M. (2007). Littoral blasts: Pumice-water heat transfer and the conditions
1430 for steam explosions when pyroclastic flows enter the ocean. *Journal of Geophysical Research: Solid*
1431 *Earth* 112, B11201. doi:10.1029/2006JB004910
- 1432 Durant, A. J., Rose, W. I., Sarna-Wojcicki, A. M., Carey, S., and Volentik, A. C. M. (2009). Hydrometeor-
1433 enhanced tephra sedimentation: Constraints from the 18 May 1980 eruption of Mount St. Helens. *Journal*
1434 *of Geophysical Research: Solid Earth* 114. doi:10.1029/2008JB005756
- 1435 Dürig, T., Sonder, I., Zimanowski, B., Beyrichen, H., and Büttner, R. (2012). Generation of volcanic ash
1436 by basaltic volcanism. *Journal of Geophysical Research: Solid Earth* 117. doi:10.1029/2011JB008628
- 1437 Dürig, T., White, J. D. L., Murch, A. P., Zimanowski, B., Büttner, R., Mele, D., et al. (2020a). Deep-sea
1438 eruptions boosted by induced fuel–coolant explosions. *Nature Geoscience* 13, 498–503. doi:10.1038/
1439 s41561-020-0603-4
- 1440 Dürig, T., White, J. D. L., Zimanowski, B., Büttner, R., Murch, A., and Carey, R. J. (2020b). Deep-sea
1441 fragmentation style of Havre revealed by dendrogrammatic analyses of particle morphometry. *Bulletin*
1442 *of Volcanology* 82, 67. doi:10.1007/s00445-020-01408-1
- 1443 Elsworth, D., Voight, B., Thompson, G., and Young, S. (2004). Thermal-hydrologic mechanism for
1444 rainfall-triggered collapse of lava domes. *Geology* 32, 969–972. doi:10.1130/G20730.1
- 1445 Farquharson, J. I. and Amelung, F. (2020). Extreme rainfall triggered the 2018 rift eruption at Kīlauea
1446 Volcano. *Nature* 580, 491–495. doi:10.1038/s41586-020-2172-5
- 1447 Fasullo, J. T., Tomas, R., Stevenson, S., Otto-Bliesner, B., Brady, E., and Wahl, E. (2017). The amplifying
1448 influence of increased ocean stratification on a future year without a summer. *Nature Communications* 8,
1449 1236. doi:10.1038/s41467-017-01302-z
- 1450 Fee, D., Lyons, J., Haney, M., Wech, A., Waythomas, C., Diefenbach, A. K., et al. (2020). Seismo-acoustic
1451 evidence for vent drying during shallow submarine eruptions at Bogoslof volcano, Alaska. *Bulletin of*
1452 *Volcanology* 82, 2. doi:10.1007/s00445-019-1326-5
- 1453 Fisher, R. V. and Schmincke, H.-U. (2012). *Pyroclastic Rocks* (Springer Science & Business Media)
- 1454 Fitch, E. P. and Fagents, S. A. (2020). Using the characteristics of rootless cone deposits to estimate
1455 the energetics of explosive lava–water interactions. *Bulletin of Volcanology* 82, 83. doi:10.1007/
1456 s00445-020-01422-3
- 1457 Gilchrist, J. T. and Jellinek, A. M. (2021). Sediment waves and the gravitational stability of volcanic jets.
1458 *Bulletin of Volcanology* 83, 64. doi:10.1007/s00445-021-01472-1
- 1459 Giordano, D., Nichols, A. R. L., and Dingwell, D. B. (2005). Glass transition temperatures of natural
1460 hydrous melts: A relationship with shear viscosity and implications for the welding process. *Journal of*
1461 *Volcanology and Geothermal Research* 142, 105–118. doi:10.1016/j.jvolgeores.2004.10.015
- 1462 Girault, F., Carazzo, G., Tait, S., Ferrucci, F., and Kaminski, É. (2014). The effect of total grain-size
1463 distribution on the dynamics of turbulent volcanic plumes. *Earth and Planetary Science Letters* 394,
1464 124–134. doi:10.1016/j.epsl.2014.03.021
- 1465 Girault, F., Carazzo, G., Tait, S., and Kaminski, E. (2016). Combined effects of total grain-size distribution
1466 and crosswind on the rise of eruptive volcanic columns. *Journal of Volcanology and Geothermal*
1467 *Research* 326, 103–113. doi:10.1016/j.jvolgeores.2015.11.007
- 1468 Glaze, L. S., Baloga, S. M., and Wilson, L. (1997). Transport of atmospheric water vapor by volcanic
1469 eruption columns. *Journal of Geophysical Research: Atmospheres* 102, 6099–6108. doi:10.1029/
1470 96JD03125
- 1471 [Dataset] Global Volcanism Program (2013). South Sarigan Seamount (284193)

- 1472 Gonnermann, H. M. (2015). Magma Fragmentation. *Annual Review of Earth and Planetary Sciences* 43,
1473 431–458. doi:10.1146/annurev-earth-060614-105206
- 1474 Gonnermann, H. M. and Manga, M. (2007). The Fluid Mechanics Inside a Volcano. *Annual Review of*
1475 *Fluid Mechanics* 39, 321–356. doi:10.1146/annurev.fluid.39.050905.110207
- 1476 Gonnermann, H. M. and Manga, M. (2013). Dynamics of magma ascent in the volcanic conduit. In
1477 *Modeling Volcanic Processes: The Physics and Mathematics of Volcanism*, eds. R. M. C. Lopes,
1478 S. A. Fagents, and T. K. P. Gregg (Cambridge: Cambridge University Press). 55–84. doi:10.1017/
1479 CBO9781139021562.004
- 1480 Green, D. N., Evers, L. G., Fee, D., Matoza, R. S., Snellen, M., Smets, P., et al. (2013). Hydroacoustic,
1481 infrasonic and seismic monitoring of the submarine eruptive activity and sub-aerial plume generation at
1482 South Sarigan, May 2010. *Journal of Volcanology and Geothermal Research* 257, 31–43. doi:10.1016/j.
1483 jvolgeores.2013.03.006
- 1484 Gudmundsson, M. T., Pálsson, F., Thordarson, T., Hoskuldsson, A., Larsen, G., Hognadóttir, T., et al.
1485 (2014). Water/magma mass fractions in phreatomagmatic eruption plumes - constraints from the
1486 Grímsvötn 2011 eruption. *AGU Fall Meeting Abstracts* 11, V11B–4718
- 1487 Gudmundsson, M. T., Sigmundsson, F., Björnsson, H., and Högnadóttir, T. (2004). The 1996 eruption at
1488 Gjalp, Vatnajökull ice cap, Iceland: Efficiency of heat transfer, ice deformation and subglacial water
1489 pressure. *Bulletin of Volcanology* 66, 46–65. doi:10.1007/s00445-003-0295-9
- 1490 Hajimirza, S., Gonnermann, H. M., and Gardner, J. E. (2021). Reconciling bubble nucleation in explosive
1491 eruptions with geospeedometers. *Nature Communications* 12, 283. doi:10.1038/s41467-020-20541-1
- 1492 Hajimirza, S., Gonnermann, H. M., Gardner, J. E., and Giachetti, T. (2019). Predicting Homogeneous
1493 Bubble Nucleation in Rhyolite. *Journal of Geophysical Research: Solid Earth* 124, 2395–2416.
1494 doi:10.1029/2018JB015891
- 1495 Hersbach, H., Bell, B., Berrisford, P., Hirahara, S., Horányi, A., Muñoz-Sabater, J., et al. (2020). The
1496 ERA5 global reanalysis. *Quarterly Journal of the Royal Meteorological Society* 146, 1999–2049.
1497 doi:10.1002/qj.3803
- 1498 Holloway, J. R. (1977). Fugacity and Activity of Molecular Species in Supercritical Fluids. In
1499 *Thermodynamics in Geology*, ed. D. G. Fraser (Dordrecht: Springer Netherlands), NATO Advanced
1500 Study Institutes Series, 161–181. doi:10.1007/978-94-010-1252-2_9
- 1501 Houghton, B. and Carey, R. J. (2015). Chapter 34 - Pyroclastic Fall Deposits. In *The Encyclopedia of*
1502 *Volcanoes (Second Edition)*, ed. H. Sigurdsson (Amsterdam: Academic Press). 599–616. doi:10.1016/
1503 B978-0-12-385938-9.00034-1
- 1504 Houghton, B., White, J. D. L., and Van Eaton, A. R. (2015). Chapter 30 - Phreatomagmatic and Related
1505 Eruption Styles. In *The Encyclopedia of Volcanoes (Second Edition)*, ed. H. Sigurdsson (Amsterdam:
1506 Academic Press). 537–552. doi:10.1016/B978-0-12-385938-9.00030-4
- 1507 Houghton, B. F. and Carey, R. J. (2019). Physical constraints for effective magma-water interaction along
1508 volcanic conduits during silicic explosive eruptions: COMMENT. *Geology* 47, e461–e461. doi:10/c84b
- 1509 Houghton, B. F. and Wilson, C. J. N. (1989). A vesicularity index for pyroclastic deposits. *Bulletin of*
1510 *Volcanology* 51, 451–462. doi:10.1007/BF01078811
- 1511 Huybers, P. and Langmuir, C. (2009). Feedback between deglaciation, volcanism, and atmospheric CO₂.
1512 *Earth and Planetary Science Letters* 286, 479–491. doi:10.1016/j.epsl.2009.07.014
- 1513 Jones, T., Gonnermann, H., Moreland, W., and Thordarson, T. (2019). Quantifying water entrainment in
1514 volcanic jets. In *Geophysical Research Abstracts*. vol. 21, 1
- 1515 Jull, M. and McKenzie, D. (1996). The effect of deglaciation on mantle melting beneath Iceland. *Journal*
1516 *of Geophysical Research: Solid Earth* 101, 21815–21828. doi:10.1029/96JB01308

- 1517 Junglas, P. (2009). Implementation of the IAPWS-95 Standard for Use in Thermodynamics Lectures 25,
1518 3–10
- 1519 Kaminski, E. and Jaupart, C. (1998). The size distribution of pyroclasts and the fragmentation sequence
1520 in explosive volcanic eruptions. *Journal of Geophysical Research: Solid Earth* 103, 29759–29779.
1521 doi:10.1029/98JB02795
- 1522 Kaminski, E., Tait, S., and Carazzo, G. (2005). Turbulent entrainment in jets with arbitrary buoyancy.
1523 *Journal of Fluid Mechanics* 526, 361–376. doi:10.1017/S0022112004003209
- 1524 Kokelaar, P. (1986). Magma-water interactions in subaqueous and emergent basaltic. *Bulletin of*
1525 *Volcanology* 48, 275–289. doi:10.1007/BF01081756
- 1526 Kotsovinos, N. E. (2000). Axisymmetric Submerged Intrusion in Stratified Fluid. *Journal of Hydraulic*
1527 *Engineering* 126, 446–456. doi:10.1061/(ASCE)0733-9429(2000)126:6(446)
- 1528 Koyaguchi, T. and Woods, A. W. (1996). On the formation of eruption columns following explosive
1529 mixing of magma and surface-water. *Journal of Geophysical Research: Solid Earth* 101, 5561–5574.
1530 doi:10.1029/95JB01687
- 1531 Kremser, S., Thomason, L. W., von Hobe, M., Hermann, M., Deshler, T., Timmreck, C., et al. (2016).
1532 Stratospheric aerosol-Observations, processes, and impact on climate: Stratospheric Aerosol. *Reviews of*
1533 *Geophysics* 54, 278–335. doi:10.1002/2015RG000511
- 1534 Krishnamohan, K.-P. S.-P., Bala, G., Cao, L., Duan, L., and Caldeira, K. (2019). Climate system response
1535 to stratospheric sulfate aerosols: Sensitivity to altitude of aerosol layer. *Earth System Dynamics* 10,
1536 885–900. doi:10.5194/esd-10-885-2019
- 1537 Lacis, A. (2015). Volcanic aerosol radiative properties. *Past Global Change Magazine* 23, 50–51.
1538 doi:10.22498/pages.23.2.50
- 1539 Lee, C.-T. and Dee, S. (2019). Does volcanism cause warming or cooling? *Geology* 47, 687–688.
1540 doi:10.1130/focus072019.1
- 1541 LeGrande, A. N., Tsigaridis, K., and Bauer, S. E. (2016). Role of atmospheric chemistry in the climate
1542 impacts of stratospheric volcanic injections. *Nature Geoscience* 9, 652–655. doi:10.1038/ngeo2771
- 1543 Lherm, V. and Jellinek, A. M. (2019). Experimental constraints on the distinct effects of ash, lapilli,
1544 and larger pyroclasts on entrainment and mixing in volcanic plumes. *Bulletin of Volcanology* 81, 73.
1545 doi:10.1007/s00445-019-1329-2
- 1546 Linden, P. F. (1979). Mixing in stratified fluids. *Geophysical & Astrophysical Fluid Dynamics* 13, 3–23.
1547 doi:10.1080/03091927908243758
- 1548 Liu, E. J. (2016). *The Generation of Volcanic Ash during Basaltic Hydromagmatic Eruptions : From*
1549 *Fragmentation to Resuspension*. Ph.D. thesis, University of Bristol
- 1550 Liu, E. J., Cashman, K. V., Rust, A. C., and Gislason, S. R. (2015). The role of bubbles in generating fine
1551 ash during hydromagmatic eruptions. *Geology* 43, 239–242. doi:10.1130/G36336.1
- 1552 Liu, E. J., Cashman, K. V., Rust, A. C., and Höskuldsson, A. (2017). Contrasting mechanisms of magma
1553 fragmentation during coeval magmatic and hydromagmatic activity: The Hverfjall Fires fissure eruption,
1554 Iceland. *Bulletin of Volcanology* 79, 68. doi:10.1007/s00445-017-1150-8
- 1555 Liu, Y., Zhang, Y., and Behrens, H. (2005). Solubility of H₂O in rhyolitic melts at low pressures and a
1556 new empirical model for mixed H₂O–CO₂ solubility in rhyolitic melts. *Journal of Volcanology and*
1557 *Geothermal Research* 143, 219–235. doi:10.1016/j.jvolgeores.2004.09.019
- 1558 Lyons, J. J., Haney, M. M., Fee, D., Wech, A. G., and Waythomas, C. F. (2019). Infrasound from
1559 giant bubbles during explosive submarine eruptions. *Nature Geoscience* 12, 952–958. doi:10.1038/
1560 s41561-019-0461-0

- 1561 Magnússon, E., Gudmundsson, M. T., Roberts, M. J., Sigursson, G., Höskuldsson, F., and Oddsson, B.
1562 (2012). Ice-volcano interactions during the 2010 Eyjafjallajökull eruption, as revealed by airborne
1563 imaging radar. *Journal of Geophysical Research: Solid Earth* 117. doi:10.1029/2012JB009250
- 1564 Manga, M., Fauria, K. E., Lin, C., Mitchell, S. J., Jones, M., Conway, C. E., et al. (2018). The pumice
1565 raft-forming 2012 Havre submarine eruption was effusive. *Earth and Planetary Science Letters* 489,
1566 49–58. doi:10.1016/j.epsl.2018.02.025
- 1567 Mann, M. E., Steinman, B. A., Brouillette, D. J., and Miller, S. K. (2021). Multidecadal climate oscillations
1568 during the past millennium driven by volcanic forcing. *Science* 371, 1014–1019. doi:10.1126/science.
1569 abc5810
- 1570 Manzella, I., Bonadonna, C., Phillips, J. C., and Monnard, H. (2015). The role of gravitational instabilities
1571 in deposition of volcanic ash. *Geology* 43, 211–214. doi:10.1130/G36252.1
- 1572 Marshall, L., Johnson, J. S., Mann, G. W., Lee, L., Dhomse, S. S., Regayre, L., et al. (2019). Exploring
1573 How Eruption Source Parameters Affect Volcanic Radiative Forcing Using Statistical Emulation. *Journal*
1574 *of Geophysical Research: Atmospheres* 124, 964–985. doi:10.1029/2018JD028675
- 1575 Massol, H. and Koyaguchi, T. (2005). The effect of magma flow on nucleation of gas bubbles in a volcanic
1576 conduit. *Journal of Volcanology and Geothermal Research* 143, 69–88. doi:10.1016/j.jvolgeores.2004.
1577 09.011
- 1578 Mastin, L. G. (2007a). Generation of fine hydromagmatic ash by growth and disintegration of glassy rinds.
1579 *Journal of Geophysical Research: Solid Earth* 112. doi:10.1029/2005JB003883
- 1580 Mastin, L. G. (2007b). A user-friendly one-dimensional model for wet volcanic plumes. *Geochemistry,*
1581 *Geophysics, Geosystems* 8. doi:10.1029/2006GC001455
- 1582 Mastin, L. G. and Ghiorso, M. S. (2000). *A Numerical Program for Steady-State Flow of Magma-*
1583 *Gas Mixtures Through Vertical Eruptive Conduits*. Tech. rep., DEPARTMENT OF THE INTERIOR
1584 WASHINGTON DC
- 1585 Mastin, L. G. and Witter, J. B. (2000). The hazards of eruptions through lakes and seawater. *Journal of*
1586 *Volcanology and Geothermal Research* 97, 195–214. doi:10.1016/S0377-0273(99)00174-2
- 1587 McGimsey, R. G., Neal, C. A., Searcy, C. K., Camacho, J. T., Aydlett, W. B., Embley, R. W., et al. (2010).
1588 The May 2010 submarine eruption from South Sarigan seamount, Northern Mariana Islands 2010,
1589 T11E–07
- 1590 Michaut, C., Ricard, Y., Bercovici, D., and Sparks, R. S. J. (2013). Eruption cyclicity at silicic volcanoes
1591 potentially caused by magmatic gas waves. *Nature Geoscience* 6, 856–860. doi:10.1038/ngeo1928
- 1592 Moitra, P., Sonder, I., and Valentine, G. A. (2020). The role of external water on rapid cooling and
1593 fragmentation of magma. *Earth and Planetary Science Letters* 537, 116194. doi:10.1016/j.epsl.2020.
1594 116194
- 1595 Moreland, W. (2017). *Explosive Activity in Flood Lava Eruptions: A Case Study of the 10th Century Eldgjá*
1596 *Eruption, Iceland*. Ph.D. thesis, University of Iceland, School of Engineering and Natural Sciences,
1597 Faculty of Earth Sciences
- 1598 Moreland, W. M., Thordarson, T., Houghton, B. F., and Larsen, G. (2019). Driving mechanisms of
1599 subaerial and subglacial explosive episodes during the 10th century Eldgjá fissure eruption, southern
1600 Iceland. *Volcanica* 2, 129–150. doi:10.30909/vol.02.02.129150
- 1601 Morton, B. R., Taylor, G. I., and Turner, J. S. (1956). Turbulent gravitational convection from maintained
1602 and instantaneous sources. *Proceedings of the Royal Society of London. Series A. Mathematical and*
1603 *Physical Sciences* 234, 1–23. doi:10.1098/rspa.1956.0011
- 1604 Murch, A. P., White, J. D. L., and Carey, R. J. (2019). Characteristics and Deposit Stratigraphy of
1605 Submarine-Erupted Silicic Ash, Havre Volcano, Kermadec Arc, New Zealand. *Frontiers in Earth*

- 1606 *Science* 7, 1. doi:10.3389/feart.2019.00001
- 1607 Nelson, C. S. and Lister, G. S. (1995). Surficial bottom sediments of Lake Taupo, New Zealand: Texture,
1608 composition, provenance, and sedimentation rates. *New Zealand Journal of Geology and Geophysics* 38,
1609 61–79. doi:10.1080/00288306.1995.9514639
- 1610 Neri, A., Di Muro, A., and Rosi, M. (2002). Mass partition during collapsing and transitional columns by
1611 using numerical simulations. *Journal of Volcanology and Geothermal Research* 115, 1–18. doi:10.1016/
1612 S0377-0273(01)00304-3
- 1613 Niemeier, U., Timmreck, C., Graf, H.-F., Kinne, S., Rast, S., and Self, S. (2009). Initial fate of fine
1614 ash and sulfur from large volcanic eruptions. *Atmospheric Chemistry and Physics* 9, 9043–9057.
1615 doi:10.5194/acp-9-9043-2009
- 1616 Ogden, D. E., Wohletz, K. H., Glatzmaier, G. A., and Brodsky, E. E. (2008). Numerical simulations of
1617 volcanic jets: Importance of vent overpressure. *Journal of Geophysical Research: Solid Earth* 113,
1618 B02204. doi:10.1029/2007JB005133
- 1619 Papale, P. (1999). Strain-induced magma fragmentation in explosive eruptions. *Nature* 397, 425–428.
1620 doi:10.1038/17109
- 1621 Patel, A., Manga, M., Carey, R. J., and Degruyter, W. (2013). Effects of thermal quenching on mechanical
1622 properties of pyroclasts. *Journal of Volcanology and Geothermal Research* 258, 24–30. doi:10.1016/j.
1623 jvolgeores.2013.04.001
- 1624 Prata, F., Woodhouse, M., Huppert, H. E., Prata, A., Thordarson, T., and Carn, S. (2017). Atmospheric
1625 processes affecting the separation of volcanic ash and SO₂ in volcanic eruptions: Inferences from the
1626 May 2011 Grímsvötn eruption. *Atmospheric Chemistry and Physics* 17, 10709–10732
- 1627 Robock, A. (2000). Volcanic eruptions and climate. *Reviews of Geophysics* 38, 191–219. doi:10.1029/
1628 1998RG000054
- 1629 Rose, W. I. (1977). Scavenging of volcanic aerosol by ash: Atmospheric and volcanologic implications.
1630 *Geology* 5, 621–624. doi:10.1130/0091-7613(1977)5<621:SOVABA>2.0.CO;2
- 1631 Rose, W. I., Bluth, G. J. S., Schneider, D. J., Ernst, G. G. J., Riley, C. M., Henderson, L. J., et al.
1632 (2001). Observations of Volcanic Clouds in Their First Few Days of Atmospheric Residence: The
1633 1992 Eruptions of Crater Peak, Mount Spurr Volcano, Alaska. *The Journal of Geology* 109, 677–694.
1634 doi:10.1086/323189
- 1635 Rose, W. I., Delene, D. J., Schneider, D. J., Bluth, G. J. S., Krueger, A. J., Sprod, I., et al. (1995). Ice in
1636 the 1994 Rabaul eruption cloud: Implications for volcano hazard and atmospheric effects. *Nature* 375,
1637 477–479. doi:10.1038/375477a0
- 1638 Rust, A. C. and Cashman, K. V. (2011). Permeability controls on expansion and size distributions of
1639 pyroclasts. *Journal of Geophysical Research: Solid Earth* 116. doi:10.1029/2011JB008494
- 1640 Saffaraval, F. and Solovitz, S. A. (2012). Near-exit flow physics of a moderately overpressured jet. *Physics*
1641 *of Fluids* 24, 086101. doi:10.1063/1.4745005
- 1642 Saffaraval, F., Solovitz, S. A., Ogden, D. E., and Mastin, L. G. (2012). Impact of reduced near-field
1643 entrainment of overpressured volcanic jets on plume development. *Journal of Geophysical Research:*
1644 *Solid Earth* 117. doi:10.1029/2011JB008862
- 1645 Santer, B. D., Bonfils, C., Painter, J. F., Zelinka, M. D., Mears, C., Solomon, S., et al. (2014). Volcanic
1646 contribution to decadal changes in tropospheric temperature. *Nature Geoscience* 7, 185–189. doi:10.
1647 1038/ngeo2098
- 1648 Schleussner, C. F. and Feulner, G. (2013). A volcanically triggered regime shift in the subpolar North
1649 Atlantic Ocean as a possible origin of the Little Ice Age. *Climate of the Past* 9, 1321–1330. doi:10.5194/
1650 cp-9-1321-2013

- 1651 Schmauss, D. and Keppler, H. (2014). Adsorption of sulfur dioxide on volcanic ashes. *American*
1652 *Mineralogist* 99, 1085–1094. doi:10.2138/am.2014.4656
- 1653 Schmid, A., Sonder, I., Seegelken, R., Zimanowski, B., Büttner, R., Gudmundsson, M. T., et al. (2010).
1654 Experiments on the heat discharge at the dynamic magma-water-interface. *Geophysical Research Letters*
1655 37. doi:10.1029/2010GL044963
- 1656 Schmidt, A., Carslaw, K. S., Mann, G. W., Rap, A., Pringle, K. J., Spracklen, D. V., et al. (2012).
1657 Importance of tropospheric volcanic aerosol for indirect radiative forcing of climate. *Atmospheric*
1658 *Chemistry and Physics* 12, 7321–7339. doi:10.5194/acp-12-7321-2012
- 1659 Self, S. and Sparks, R. S. J. (1978). Characteristics of widespread pyroclastic deposits formed by the
1660 interaction of silicic magma and water. *Bulletin Volcanologique* 41, 196. doi:10.1007/BF02597223
- 1661 Shea, T. (2017). Bubble nucleation in magmas: A dominantly heterogeneous process? *Journal of*
1662 *Volcanology and Geothermal Research* 343, 155–170. doi:10.1016/j.jvolgeores.2017.06.025
- 1663 Sigl, M., Winstrup, M., McConnell, J. R., Welten, K. C., Plunkett, G., Ludlow, F., et al. (2015). Timing
1664 and climate forcing of volcanic eruptions for the past 2,500 years. *Nature* 523, 543–549. doi:10.1038/
1665 nature14565
- 1666 Sigmarsson, O., Haddadi, B., Carn, S., Moune, S., Gudnason, J., Yang, K., et al. (2013). The sulfur budget
1667 of the 2011 Grímsvötn eruption, Iceland. *Geophysical Research Letters* 40, 6095–6100. doi:10.1002/
1668 2013GL057760
- 1669 Smellie, J. L. and Edwards, B. R. (2016). *Glaciovolcanism on Earth and Mars* (Cambridge University
1670 Press)
- 1671 Solomon, S., Daniel, J. S., Neely, R. R., Vernier, J.-P., Dutton, E. G., and Thomason, L. W. (2011). The
1672 Persistently Variable “Background” Stratospheric Aerosol Layer and Global Climate Change. *Science*
1673 333, 866–870. doi:10.1126/science.1206027
- 1674 Sonder, I., Schmid, A., Seegelken, R., Zimanowski, B., and Büttner, R. (2011). Heat source or heat sink:
1675 What dominates behavior of non-explosive magma-water interaction? *Journal of Geophysical Research:*
1676 *Solid Earth* 116. doi:10.1029/2011JB008280
- 1677 Soreghan, G. S., Soreghan, M. J., and Heavens, N. G. (2019). Explosive volcanism as a key driver of the
1678 late Paleozoic ice age. *Geology* 47, 600–604. doi:10.1130/G46349.1
- 1679 Staunton-Sykes, J., Aubry, T. J., Shin, Y. M., Weber, J., Marshall, L. R., Luke Abraham, N., et al. (2021).
1680 Co-emission of volcanic sulfur and halogens amplifies volcanic effective radiative forcing. *Atmospheric*
1681 *Chemistry and Physics* 21, 9009–9029. doi:10.5194/acp-21-9009-2021
- 1682 Telling, J., Dufek, J., and Shaikh, A. (2013). Ash aggregation in explosive volcanic eruptions. *Geophysical*
1683 *Research Letters* 40, 2355–2360. doi:10.1002/grl.50376
- 1684 Textor, C., Graf, H.-F., Herzog, M., and Oberhuber, J. M. (2003). Injection of gases into the stratosphere
1685 by explosive volcanic eruptions. *Journal of Geophysical Research: Atmospheres* 108, 4606. doi:10.
1686 1029/2002JD002987
- 1687 Timmreck, C. (2012). Modeling the climatic effects of large explosive volcanic eruptions. *Wiley*
1688 *Interdisciplinary Reviews: Climate Change* 3, 545–564. doi:10.1002/wcc.192
- 1689 Toohey, M., Krüger, K., Schmidt, H., Timmreck, C., Sigl, M., Stoffel, M., et al. (2019). Disproportionately
1690 strong climate forcing from extratropical explosive volcanic eruptions. *Nature Geoscience* 12, 100.
1691 doi:10.1038/s41561-018-0286-2
- 1692 Toohey, M., Krüger, K., Sigl, M., Stordal, F., and Svensen, H. (2016). Climatic and societal impacts of a
1693 volcanic double event at the dawn of the Middle Ages. *Climatic Change* 136, 401–412. doi:10.1007/
1694 s10584-016-1648-7

- 1695 Turner, J. S. (1986). Turbulent entrainment: The development of the entrainment assumption, and
1696 its application to geophysical flows. *Journal of Fluid Mechanics* 173, 431–471. doi:10.1017/
1697 S0022112086001222
- 1698 Van Eaton, A. R., Herzog, M., Wilson, C. J. N., and McGregor, J. (2012). Ascent dynamics of large
1699 phreatomagmatic eruption clouds: The role of microphysics. *Journal of Geophysical Research: Solid*
1700 *Earth* 117, B03203. doi:10.1029/2011JB008892
- 1701 Van Eaton, A. R., Mastin, L. G., Herzog, M., Schwaiger, H. F., Schneider, D. J., Wallace, K. L., et al.
1702 (2015). Hail formation triggers rapid ash aggregation in volcanic plumes. *Nature Communications* 6,
1703 7860. doi:10.1038/ncomms8860
- 1704 van Otterloo, J., Cas, R. A. F., and Scutter, C. R. (2015). The fracture behaviour of volcanic glass and
1705 relevance to quench fragmentation during formation of hyaloclastite and phreatomagmatism. *Earth-*
1706 *Science Reviews* 151, 79–116. doi:10.1016/j.earscirev.2015.10.003
- 1707 Walker, G. P. L. (1973). Explosive volcanic eruptions — a new classification scheme. *Geologische*
1708 *Rundschau* 62, 431–446. doi:10.1007/BF01840108
- 1709 Walker, G. P. L. (1981). Characteristics of two phreatoplinian ashes, and their water-flushed origin. *Journal*
1710 *of Volcanology and Geothermal Research* 9, 395–407. doi:10.1016/0377-0273(81)90046-9
- 1711 Wallace, P. J., Plank, T., Edmonds, M., and Hauri, E. H. (2015). Chapter 7 - Volatiles in Magmas. In *The*
1712 *Encyclopedia of Volcanoes (Second Edition)*, ed. H. Sigurdsson (Amsterdam: Academic Press). 163–183.
1713 doi:10.1016/B978-0-12-385938-9.00007-9
- 1714 Watt, S. F. L., Pyle, D. M., and Mather, T. A. (2013). The volcanic response to deglaciation: Evidence
1715 from glaciated arcs and a reassessment of global eruption records. *Earth-Science Reviews* 122, 77–102.
1716 doi:10.1016/j.earscirev.2013.03.007
- 1717 Waythomas, C. F., Angeli, K., and Wessels, R. L. (2020). Evolution of the submarine–subaerial edifice of
1718 Bogoslof volcano, Alaska, during its 2016–2017 eruption based on analysis of satellite imagery. *Bulletin*
1719 *of Volcanology* 82, 21. doi:10.1007/s00445-020-1363-0
- 1720 Wilson, C. J. N. (2001). The 26.5ka Oruanui eruption, New Zealand: An introduction and overview.
1721 *Journal of Volcanology and Geothermal Research* 112, 133–174. doi:10.1016/S0377-0273(01)00239-6
- 1722 Wilson, C. J. N. and Walker, G. P. L. (1985). The Taupo eruption, New Zealand I. General aspects.
1723 *Philosophical Transactions of the Royal Society of London. Series A, Mathematical and Physical*
1724 *Sciences* 314, 199–228. doi:10.1098/rsta.1985.0019
- 1725 Wilson, L., Sparks, R. S. J., and Walker, G. P. L. (1980). Explosive volcanic eruptions — IV. The
1726 control of magma properties and conduit geometry on eruption column behaviour. *Geophysical Journal*
1727 *International* 63, 117–148. doi:10.1111/j.1365-246X.1980.tb02613.x
- 1728 Wohletz, K. (1983). Mechanisms of hydrovolcanic pyroclast formation: Grain-size, scanning electron
1729 microscopy, and experimental studies. *Journal of Volcanology and Geothermal Research* 17, 31–63.
1730 doi:10.1016/0377-0273(83)90061-6
- 1731 Wohletz, K., Zimanowski, B., and Büttner, R. (2013). Magma–water interactions. In *Modeling Volcanic*
1732 *Processes: The Physics and Mathematics of Volcanism* (Cambridge: Cambridge University Press).
1733 doi:10.1017/CBO9781139021562.011
- 1734 Woodcock, D. C., Gilbert, J. S., and Lane, S. J. (2012). Particle-water heat transfer during explosive
1735 volcanic eruptions. *Journal of Geophysical Research. Solid Earth* 117. doi:http://dx.doi.org/10.1029/
1736 2012JB009240
- 1737 Woods, A. W. (2010). Turbulent Plumes in Nature. *Annual Review of Fluid Mechanics* 42, 391–412.
1738 doi:10.1146/annurev-fluid-121108-145430

- 1739 Woods, A. W. and Bower, S. M. (1995). The decompression of volcanic jets in a crater during explosive
1740 volcanic eruptions. *Earth and Planetary Science Letters* 131, 189–205. doi:10.1016/0012-821X(95)
1741 00012-2
- 1742 Zanchettin, D., Timmreck, C., Bothe, O., Lorenz, S. J., Hegerl, G., Graf, H.-F., et al. (2013). Delayed
1743 winter warming: A robust decadal response to strong tropical volcanic eruptions? *Geophysical Research*
1744 *Letters* 40, 204–209. doi:10.1029/2012GL054403
- 1745 Zhang, X., Li, S., Yu, D., Yang, B., and Wang, N. (2020). The Evolution of Interfaces for Underwater
1746 Supersonic Gas Jets. *Water* 12, 488. doi:10.3390/w12020488
- 1747 Zhong, Y., Miller, G. H., Otto-Bliesner, B. L., Holland, M. M., Bailey, D. A., Schneider, D. P., et al. (2011).
1748 Centennial-scale climate change from decadal-paced explosive volcanism: A coupled sea ice-ocean
1749 mechanism. *Climate Dynamics* 37, 2373–2387. doi:10.1007/s00382-010-0967-z
- 1750 Zhu, Y., Toon, O. B., Jensen, E. J., Bardeen, C. G., Mills, M. J., Tolbert, M. A., et al. (2020). Persisting
1751 volcanic ash particles impact stratospheric SO₂ lifetime and aerosol optical properties. *Nature*
1752 *Communications* 11, 4526. doi:10.1038/s41467-020-18352-5
- 1753 Zimanowski, B. and Büttner, R. (2003). Phreatomagmatic explosions in subaqueous volcanism. In
1754 *Geophysical Monograph Series*, eds J. D. L. White, J. L. Smellie, and D. A. Clague (Washington, D. C.:
1755 American Geophysical Union), vol. 140. 51–60. doi:10.1029/140GM03
- 1756 Zimanowski, B., Büttner, R., Dellino, P., White, J. D. L., and Wohletz, K. H. (2015). Chapter
1757 26 - Magma–Water Interaction and Phreatomagmatic Fragmentation. In *The Encyclopedia of*
1758 *Volcanoes (Second Edition)*, ed. H. Sigurdsson (Amsterdam: Academic Press). 473–484. doi:10.
1759 1016/B978-0-12-385938-9.00026-2

FIGURE CAPTIONS

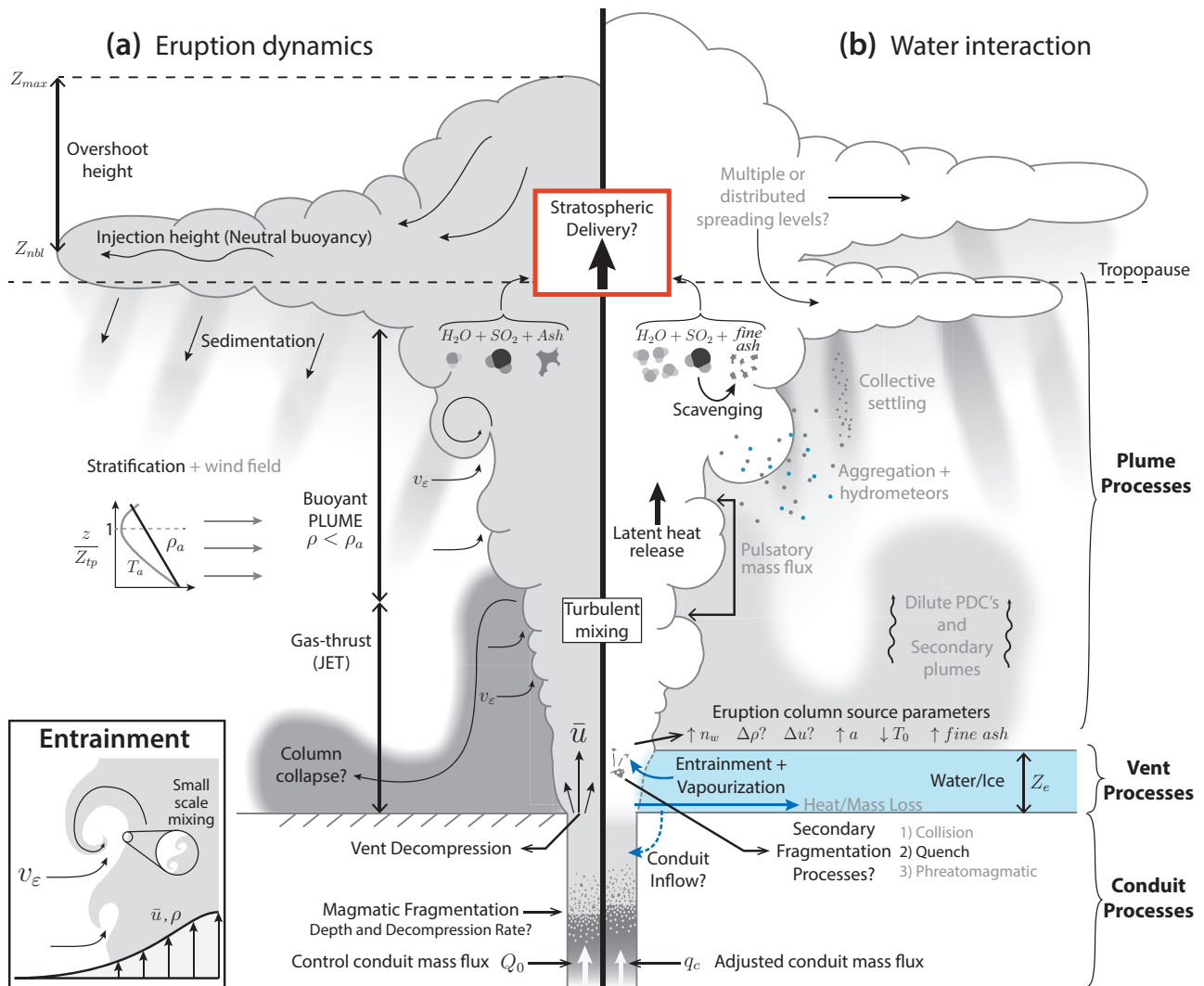


Figure 1. Summary of eruption processes from conduit to atmospheric dispersal. See text for a description of processes and their relevance for SO_2 transport. See Table 1 for a complete description of symbols. (a) Dynamical processes during a sustained, "dry" Plinian eruption. Inset: illustration of the entrainment process. (b) Summary of processes influenced by surface water interaction during a hydrovolcanic eruption. Processes in lighter gray text are those not considered in this study, but which are relevant to hydrovolcanic eruptions processes and may play a role in stratospheric delivery of SO_2 .

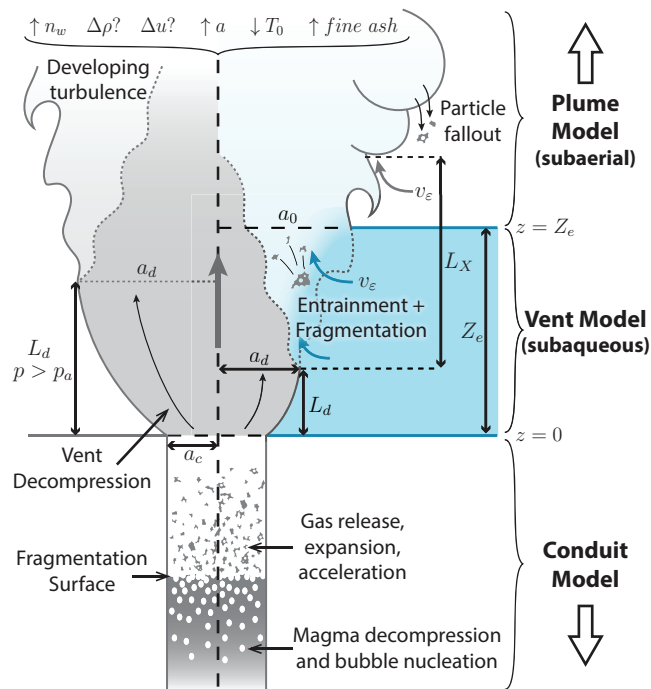


Figure 2. Schematic summary of coupled model, highlighting geometry of the vent and MWI region. The left and right sides are divided between a control scenario with no external water and a scenario with a shallow water layer, respectively. In the hydrovolcanic case, decompression of the erupting jet of gas and pyroclasts is suppressed relative to the dry control scenario (indicated by decompression length L_d and radius a_d), and initiation of turbulent mixing with external water results in water entrainment and quench fragmentation. In the water layer scenario shown here, water depth Z_e is greater than the decompression length L_d but less than the height at which large entraining eddies are fully developed, $L_d + L_X$. See Table 1 and Sections 2.2, 2.3.3, and 2.4 for a complete description of symbols and processes.

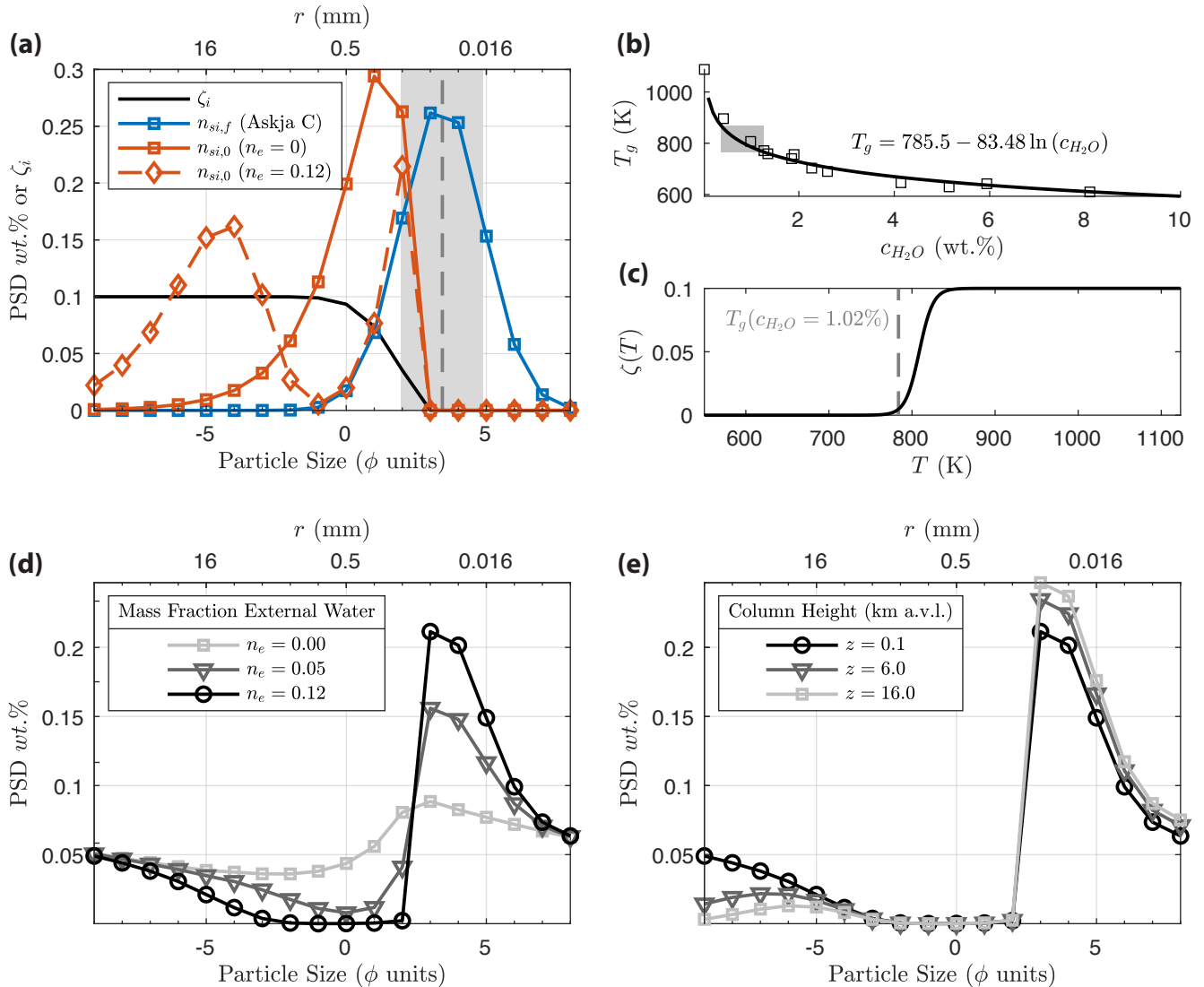


Figure 3. PSD and quench fragmentation model for rhyolitic melt, using the example of a single simulation with $q = 1.03 \times 10^8$ kg/s, $Z_e = 120$ m, and $\zeta = 0.1$. (a) Input (red lines, Equation 45) and output (blue line) PSDs for quench fragmentation. The output PSD is defined from the mean and standard deviation (in ϕ units, shown as the vertical grey dashed line and shaded region, respectively) of the Askja phase C deposit, as reported in Costa et al. (2016). The “input” PSD, (i.e. those particle sizes from which mass is removed to generate the products of quench fragmentation), is defined on the basis of available surface area in the total PSD coarse fraction. The input PSD therefore evolves from an initial value (solid red line) to a final value (dashed red line) as the total PSD coarse fraction is progressively depleted (see panel (d)). The solid black line shows the fragmentation energy efficiency as a function of particle size, ζ_i (Equation 43), which defines the size bins for the “coarse” fraction. (b) Glass transition temperature data from Dingwell (1998) (squares) and curve fit (black line) as a function of concentration of dissolved water in the melt. The grey shaded rectangle shows the range of values in the *Reference* set of simulations after exit from the vent. (c) Fragmentation efficiency as a function of temperature (equations 41, 42) for $T_g = 784$ K. (d) Evolution of the total PSD during quench fragmentation, from initial power law with no external water ($n_e = 0$, light grey line) to a coarse-depleted PSD after sufficient external water is entrained ($n_e \approx 0.12$, black line) to cross the glass transition temperature. Note the preferential depletion of particles in the mid-range ($-3 \lesssim \phi \lesssim 2$) driven by particle surface area. The reduced mass fraction of coarse particles ($\phi \lesssim 2$) in the initial PSD is due to the low density of these particles owing to their large porosity (equations 9-11). (e) Further evolution of PSD due to particle fallout, after water breach and during column rise, with preferential fallout of the coarsest fraction ($\phi \lesssim -3$) and additional enriching of fines.

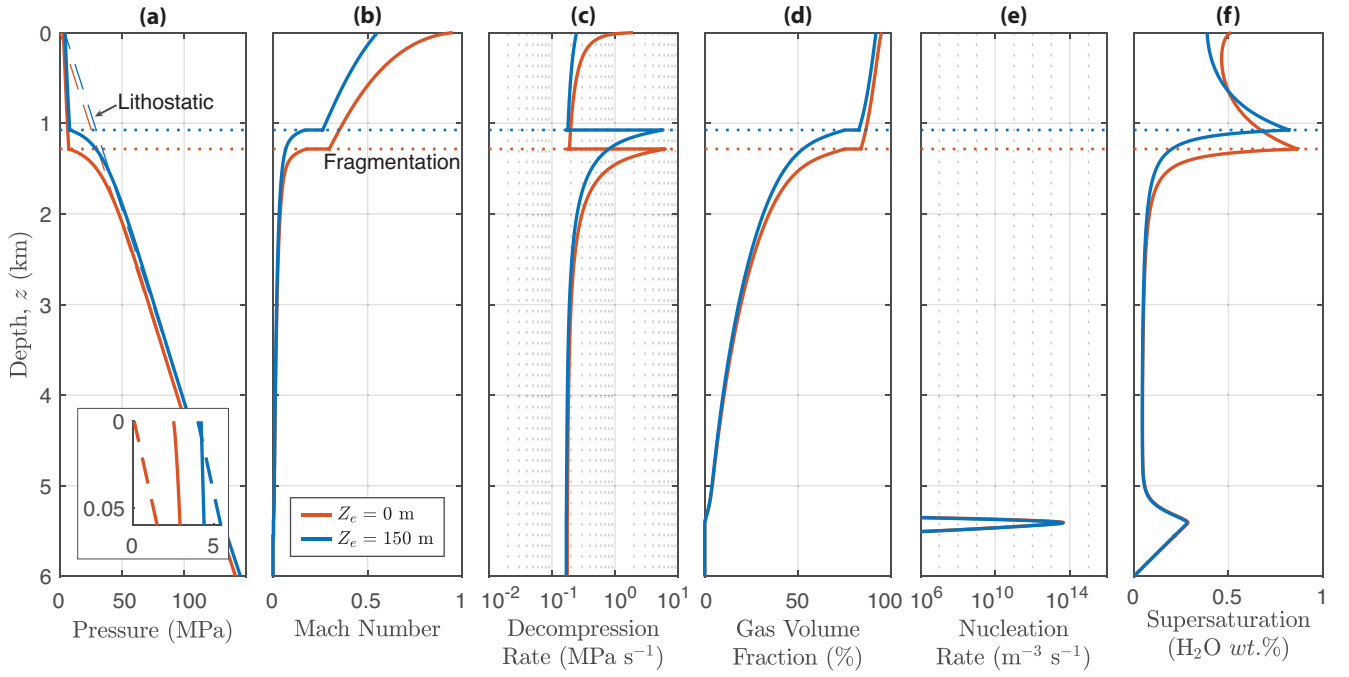


Figure 4. Example conduit model output from the *Reference* set (see Table 2) versus depth below the vent for a pair of simulations: red lines show a “dry” control run with $a_c = 53.8$ m, $Z_e = 0$, and $Q_0 = 1.6 \times 10^8$ kg/s. Blue lines show a hydrovolcanic scenario with $a_c = 53.8$ m, $Z_e = 400$ m, and $q_c = 1.53 \times 10^8$ kg/s (blue lines). (a) Magma pressure. Inset: pressure in the top 60 m of the conduit (same units as panel (a) axes), highlighting the vent overpressure of the control run versus the pressure-balanced vent of the hydrovolcanic run. (b) Mach number. (c) Decompression rate. (d) Gas Volume Fraction. (e) Bubble Nucleation Rate. (f) Supersaturation of dissolved water (i.e. difference between dissolved water c_{H_2O} and water solubility).

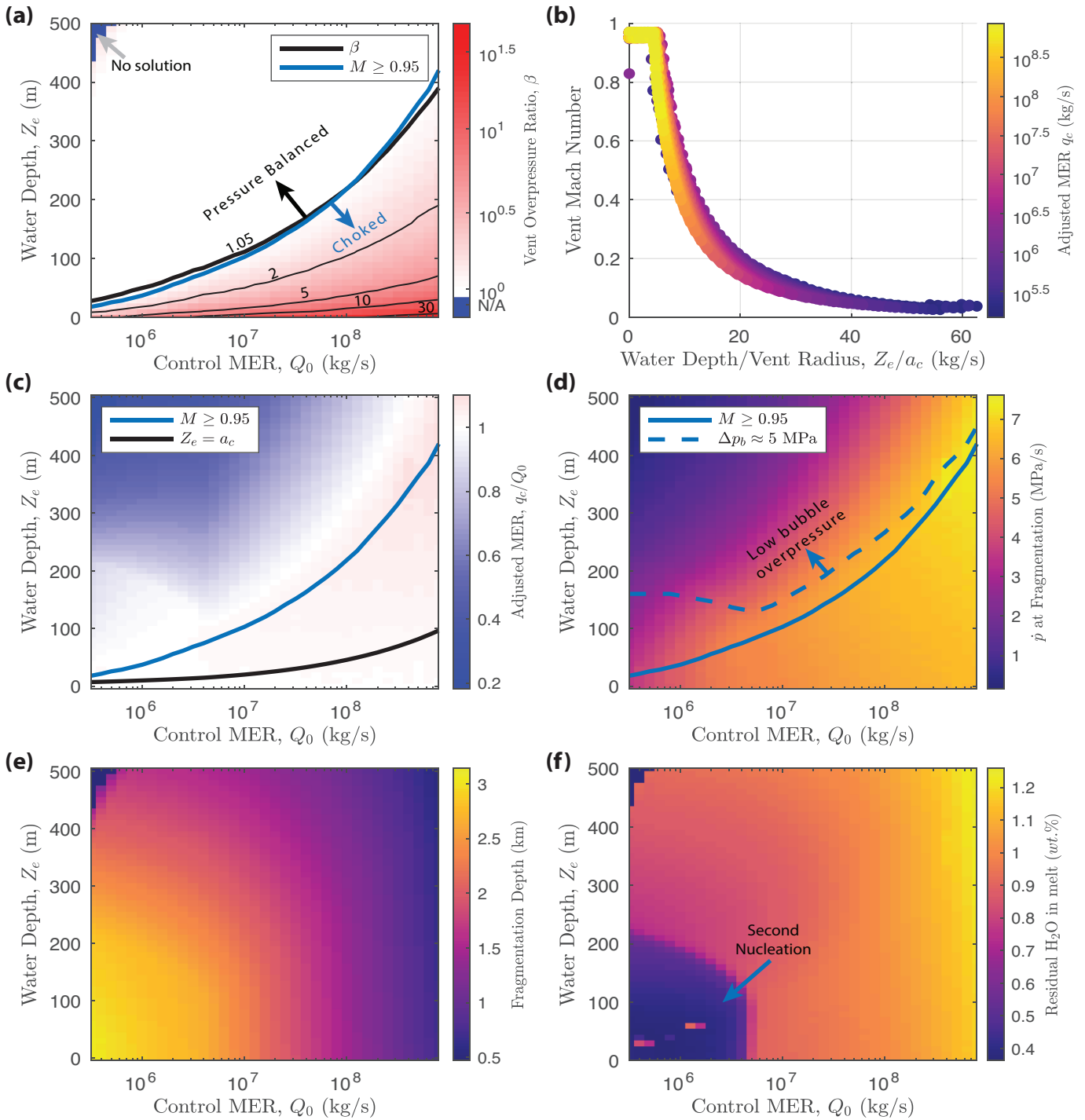


Figure 5. Conduit model output as a function of control MER, Q_0 , and external water depth, Z_e . (a) Vent overpressure β . The blue line in panels (a), (c), and (d) denotes the tolerance threshold for Mach number ($M = 0.95$), and the red line is the (approximately coincident) vent overpressure threshold, $\beta = 1.05$. The vent is choked and overpressured for water depths less than this. The blue region in the top left (high Z_e and low Q_0) are failed simulations - no viable conduit solutions were found in this region. (b) Vent Mach number. (c) Mass eruption rate adjustment for fixed conduit radius, relative to the control case for $Z_e = 0$. (d) Maximum decompression rate recorded at fragmentation ($\chi_0 = 0.75$). The dashed blue line highlights the maximum water depth for which peak bubble overpressure is at least 5 MPa, which is an approximate low bound for bubble wall rupture (Cas and Simmons, 2018). (e) Fragmentation depth. (f) Residual dissolved water in pyroclasts at the vent, highlighting a strong second nucleation event for low MER and water depths less than about 200 m.

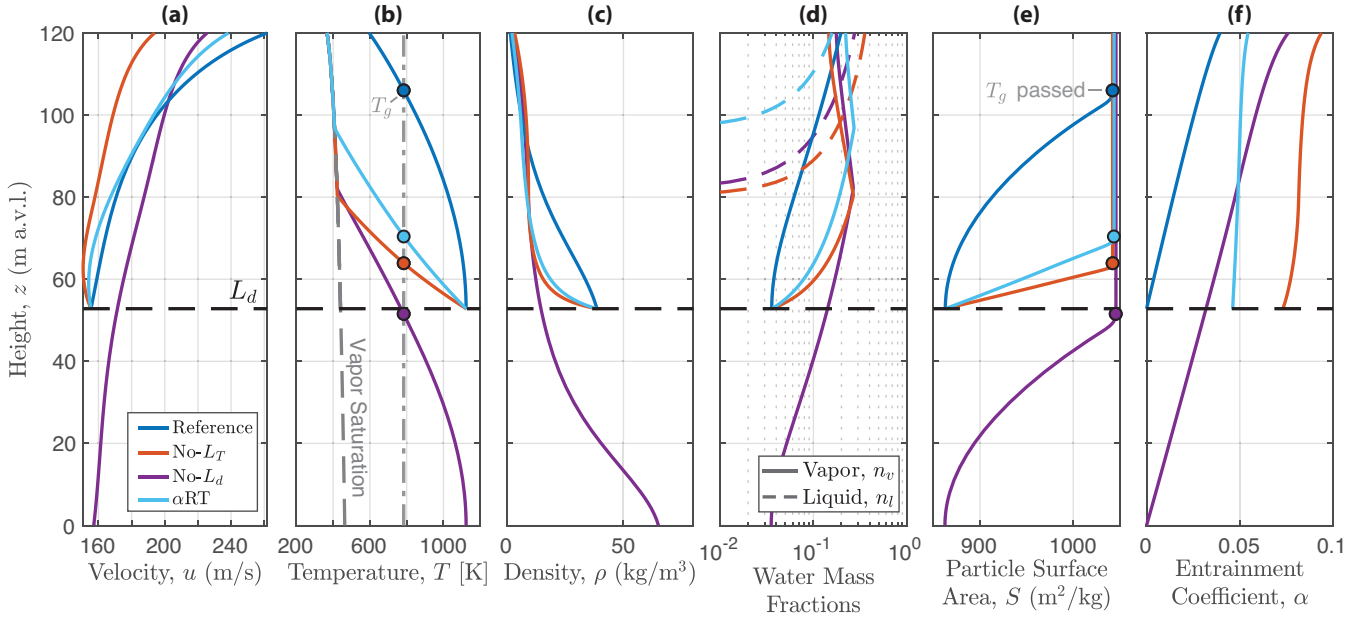


Figure 6. Example MWI model parameters versus position in the water layer above the vent for a single simulation at $q_c = 1.03 \times 10^8$ kg/s, and $Z_e = 120$ m. Four different water entrainment scenarios are shown: the *Reference* scenario using an entrainment condition modified by both decompression and crossover length scales (blue), a scenario with no scaling for turbulent mixing length (*no- L_X* , red), a scenario with no decompression length scale, where entrainment initiates immediately at the vent (*no- L_d* , purple), and a scenario using the weight Rayleigh-Taylor entrainment mode of Equation 31 (αRT scenario, light blue).

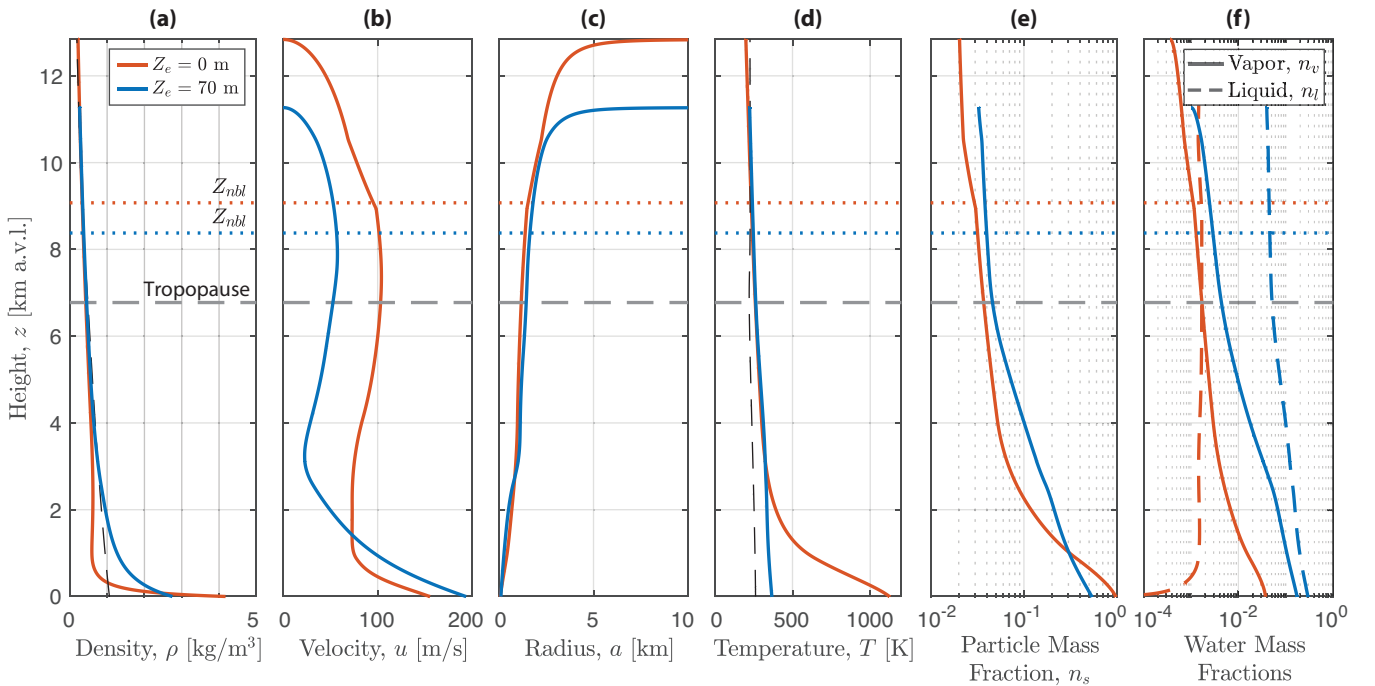


Figure 7. Example plume model output from the *Reference* set (see Table 2) versus height above the vent for a pair of simulations: red lines show a “dry” control run with $a_c = 20.0$ m, $Z_e = 0$, and $Q_0 = 1.00 \times 10^7$ kg/s. Blue lines show a hydrovolcanic scenario with $a_c = 20.0$ m, $Z_e = 70$ m, and $q_c = 1.01 \times 10^7$ kg/s (blue lines).

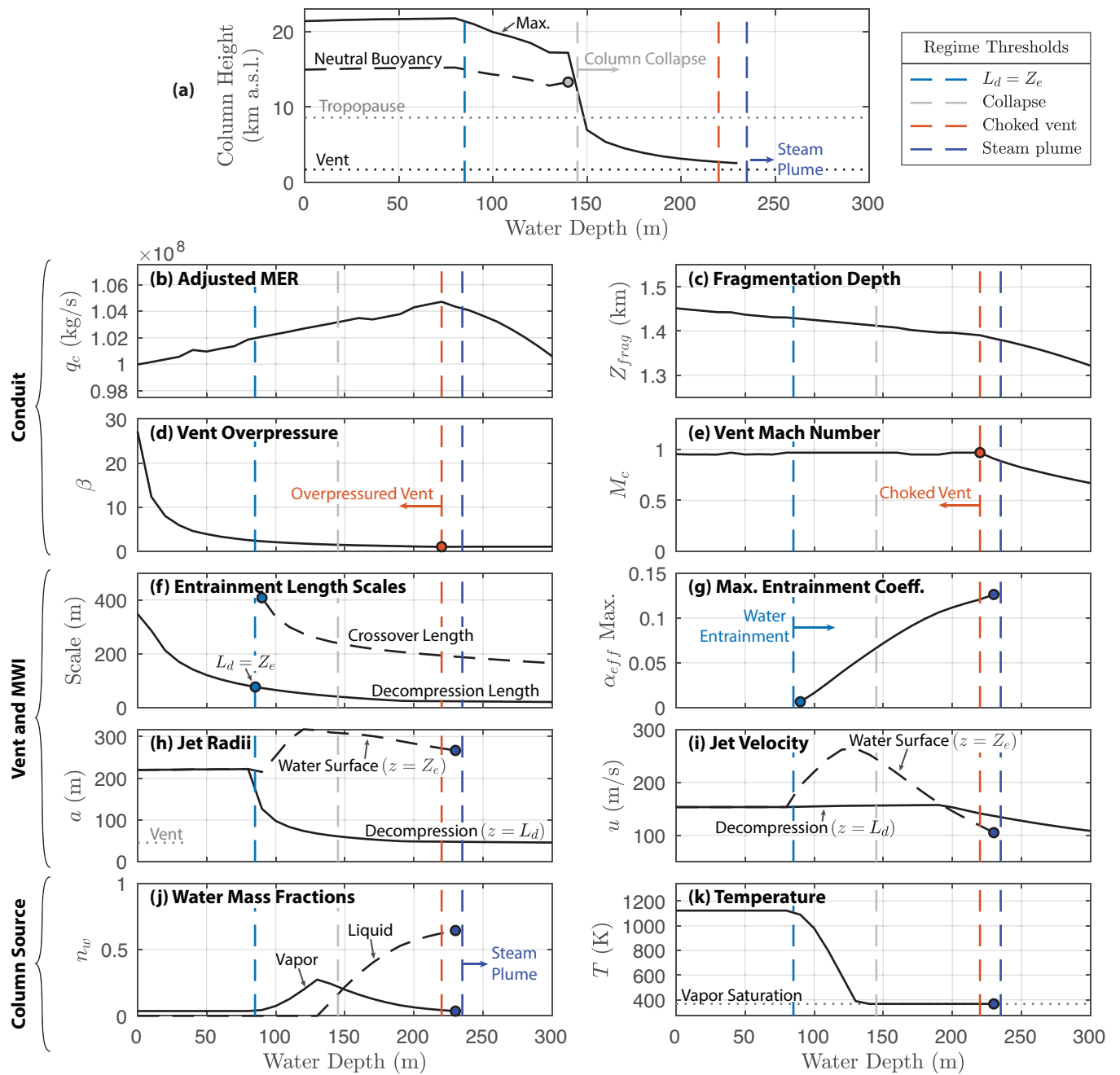


Figure 8. Output of the coupled model (conduit, vent, and column) *Reference* scenario for $Q_0 = 10^8$ kg/s and a range of water depths. Behavior thresholds for decompression length, column collapse, vent choking, and steam plumes corresponding to regimes in Figure 9a are marked with vertical dashed lines. (a) Eruption column maximum height and neutral buoyancy height above sea level, shown with vent and tropopause altitude. Conduit results: (b) adjusted conduit MER q_c ; (c) depth of fragmentation surface; vent (d) overpressure β and (e) Mach number M_c . MWI model results: (f) decompression L_d and crossover L_X length scales; (g) maximum value of the entrainment coefficient in the water layer; (h) radius of the vent and jet after initial decompression (at $z = L_d$) and at the water surface ($z = Z_e$); (i) velocity of the jet after initial decompression (at $z = L_d$) and at the water surface ($z = Z_e$). Column source conditions: (j) vapor and liquid water mass fractions; (k) bulk mixture temperature.

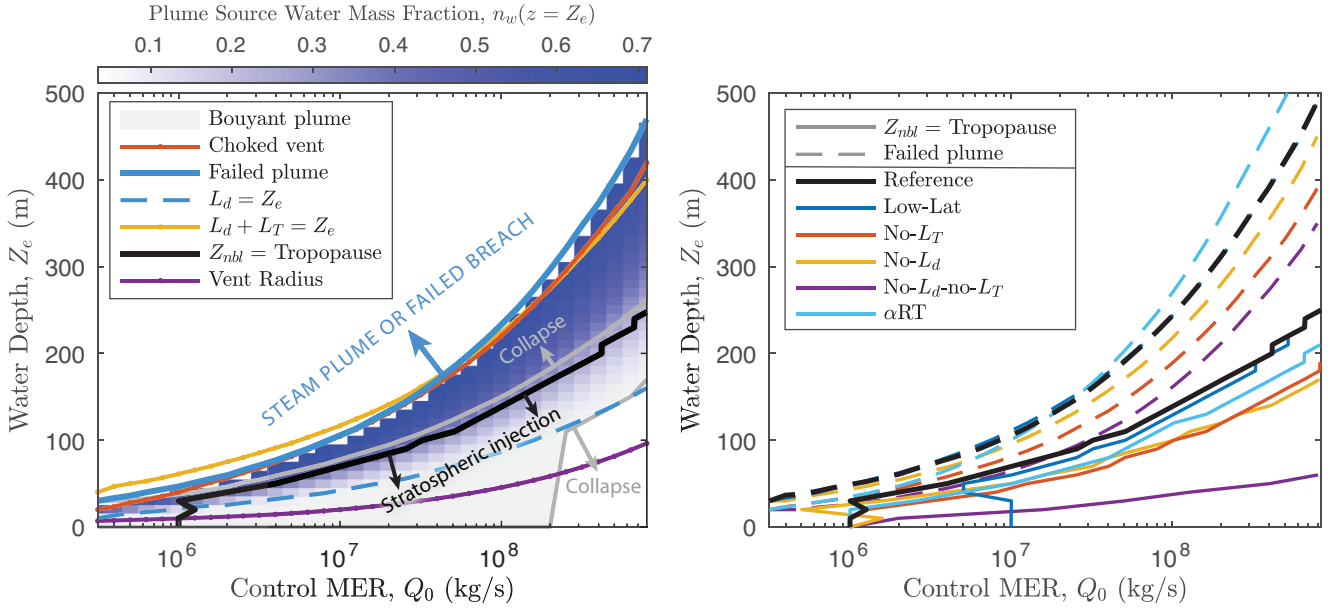


Figure 9. (a) Plume source water mass fraction as a function of MER and water depth, with overlaid thresholds for behavior of the coupled conduit-plume system. The red line marks the threshold for which the vent is choked and overpressured, with pressure-balanced, subsonic jets occurring at deeper depths. The decompression length is equal to water depth at the blue dashed line, which is the depth above which water entrainment begins. Buoyant columns occur within the grey shaded region, with column collapse elsewhere. The steam plume threshold is marked by the solid blue line - failed plumes with only negligibly small amounts of steam reach the water surface for depths greater than this (indicated by the blue arrow). Finally, the solid black line marks the water depth above which decompression length is zero. (b) Variation in the critical MER to reach the tropopause (solid lines) and maximum water depth before plume failure (i.e. only minor steam breach of the water surface, dashed lines) for different simulation scenarios (see Table 2). Black lines are for the *Reference* scenario (high latitude atmosphere), while blue lines are for the low latitude atmosphere. The remaining colors are for the four scenarios with different water entrainment parameterizations: no mixing length (*No- L_X* , red), no decompression length (*No- L_d* , yellow), neither mixing length nor decompression length (*No- L_X -no- L_X* , purple), and the weighted Rayleigh-Taylor entrainment mode (αRT , light blue).

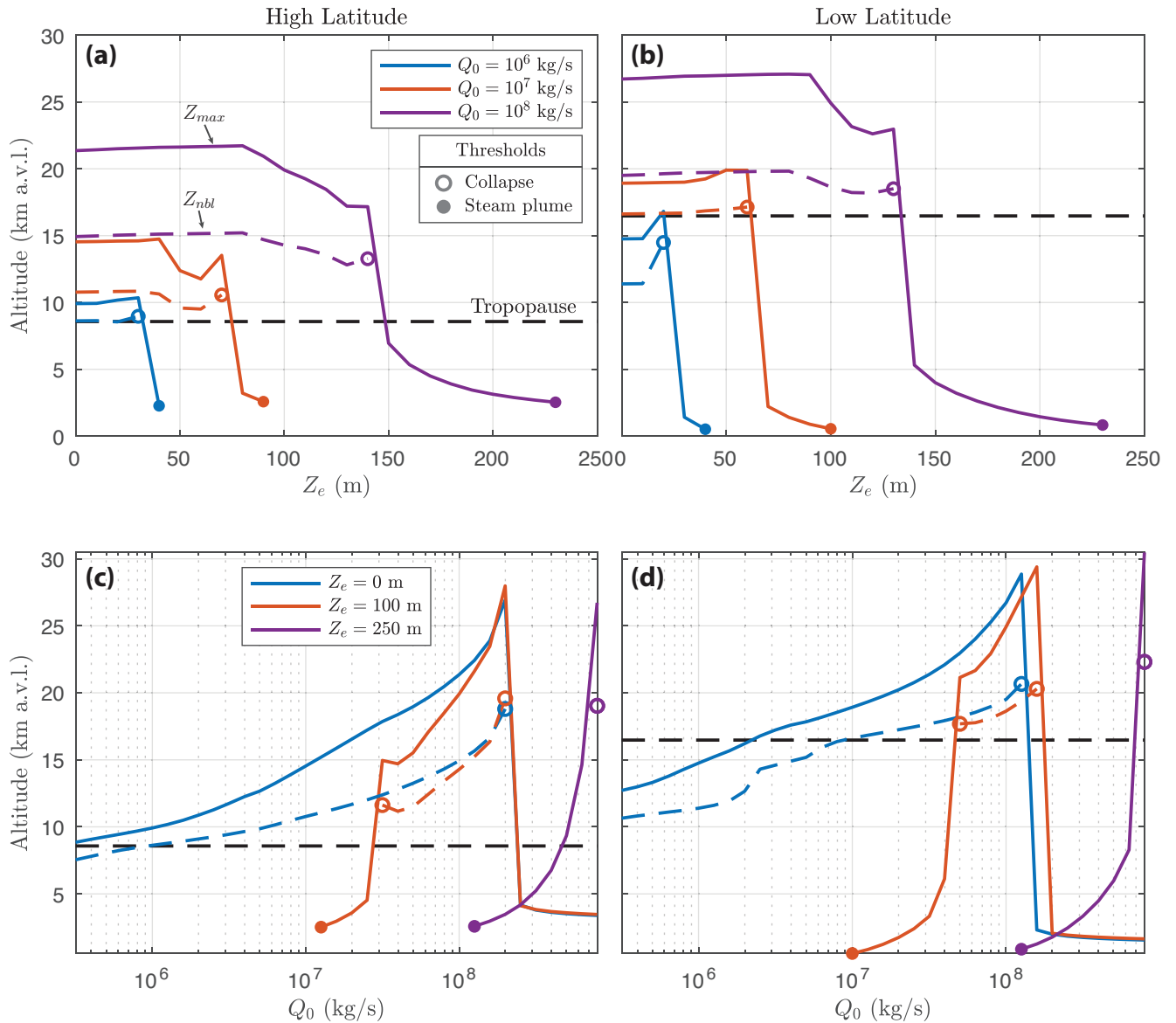


Figure 10. Eruption column height (above vent level) versus (a,b) surface water depth for three control values of MER and (c,d) MER for three fixed values of water depth. Left column plots (a,c) are for high latitude and right column (b,d) for low latitude atmospheres. For all plots, solid lines denote maximum column height, Z_{max} , dashed lines are height of neutral buoyancy, Z_{nbl} , open circles indicate threshold values for column collapse, and closed circles indicate threshold values for steam plumes at the water surface.

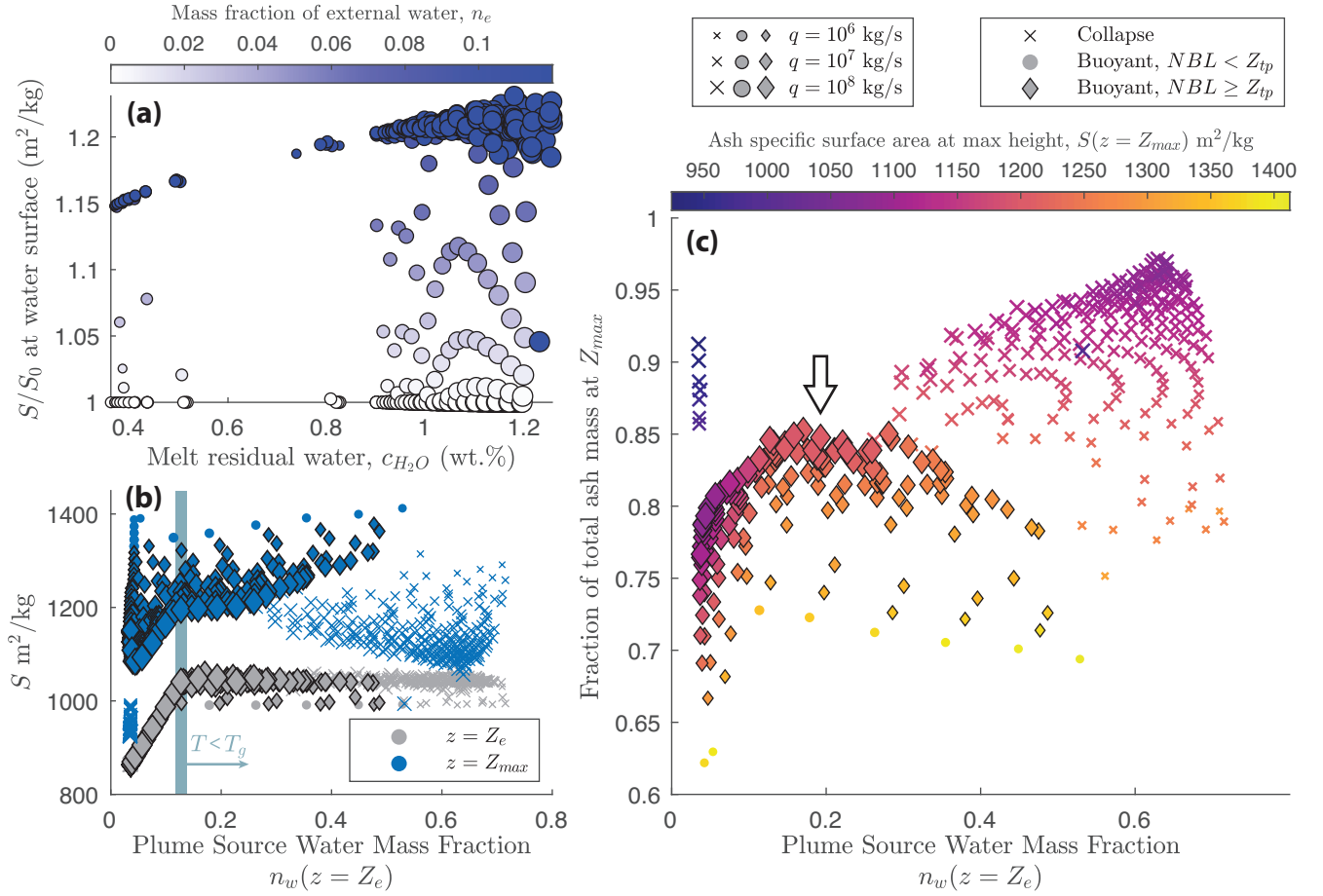


Figure 11. Effects of MWI and sedimentation on particle specific surface area S . (a) Specific surface area, S , immediately after the jet breaches the water surface ($Z = Z_e$), as a function of $c_{\text{H}_2\text{O}}$, the water mass fraction still dissolved in the melt after conduit exit. Symbols are sized according to MER and colored according to the mass fraction of entrained external water. The dissolved water content controls the glass transition temperature, T_g , which in turn is the primary limiting factor in the model for how much surface area can be generated during quench fragmentation. (b) S at two different heights in the eruption column: at column source, immediately after MWI ($Z = Z_e$, grey symbols), and at the column maximum height ($z = Z_{max}$, blue symbols) as a function of water mass fraction at column source. Symbol sizes as in (a). An 'x' denotes a collapsing column, a filled circle denotes a column that is buoyant but with Neutral Buoyancy Level (NBL) below the tropopause, and diamonds are columns that are buoyant with NBL at or above the tropopause. Evolution from grey to blue symbols is a result of sedimentation over the rise height of the column. The approximate water mass fraction above which the pyroclasts cool below the glass transition temperature T_g is marked with a vertical blue bar. (c) Fraction of particle mass remaining in the column at its maximum rise height as a function of column source water mass fraction. Symbols are sized by MER as in (a) and (b), and colored according to the value of S at maximum column height. Symbol shapes as in (b). The arrow highlights the subset of simulations with NBL above the tropopause and where the column retains increased (relative to “dry” runs) particle mass and specific surface area.

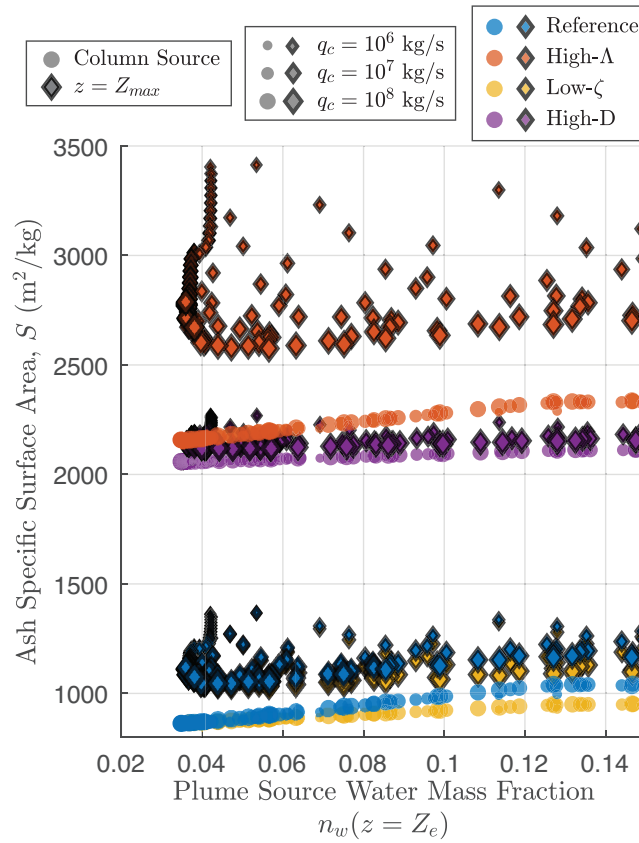


Figure 12. Specific surface area as a function of water mass fraction at the water surface (circles) and height of neutral buoyancy (diamonds) for scenarios with different fragmentation properties. The *Reference* scenario is shown in blue. Reducing the fragmentation energy efficiency to $\zeta = 0.05$ (*Low- ζ* scenario, yellow symbols) reduces the amount of energy consumed to generate surface area per unit mass of entrained water, resulting in a smaller increase in S during MWI relative to the *Reference* scenario. Conversely, a high initial value of the PSD power-law exponent, $D = 3.2$ (*High- D* scenario, purple symbols), concentrates initial particle mass in the fine fraction. Because of the fixed particle sizes for output from quench fragmentation used here (see Figure 3), there is relatively little particle mass available to fragment for the creation of new surface area and the relative change in S with water entrainment is small. Finally, increasing the particle roughness scale, $\Lambda = 25$ (*High- Λ* scenario, red symbols), results in initially high particle surface area, but also a greater energy requirement to generate new particles of a given size. This scenario results in the highest absolute changes in particle surface area after quench fragmentation and sedimentation, but a smaller relative change than for the *Reference* scenario.

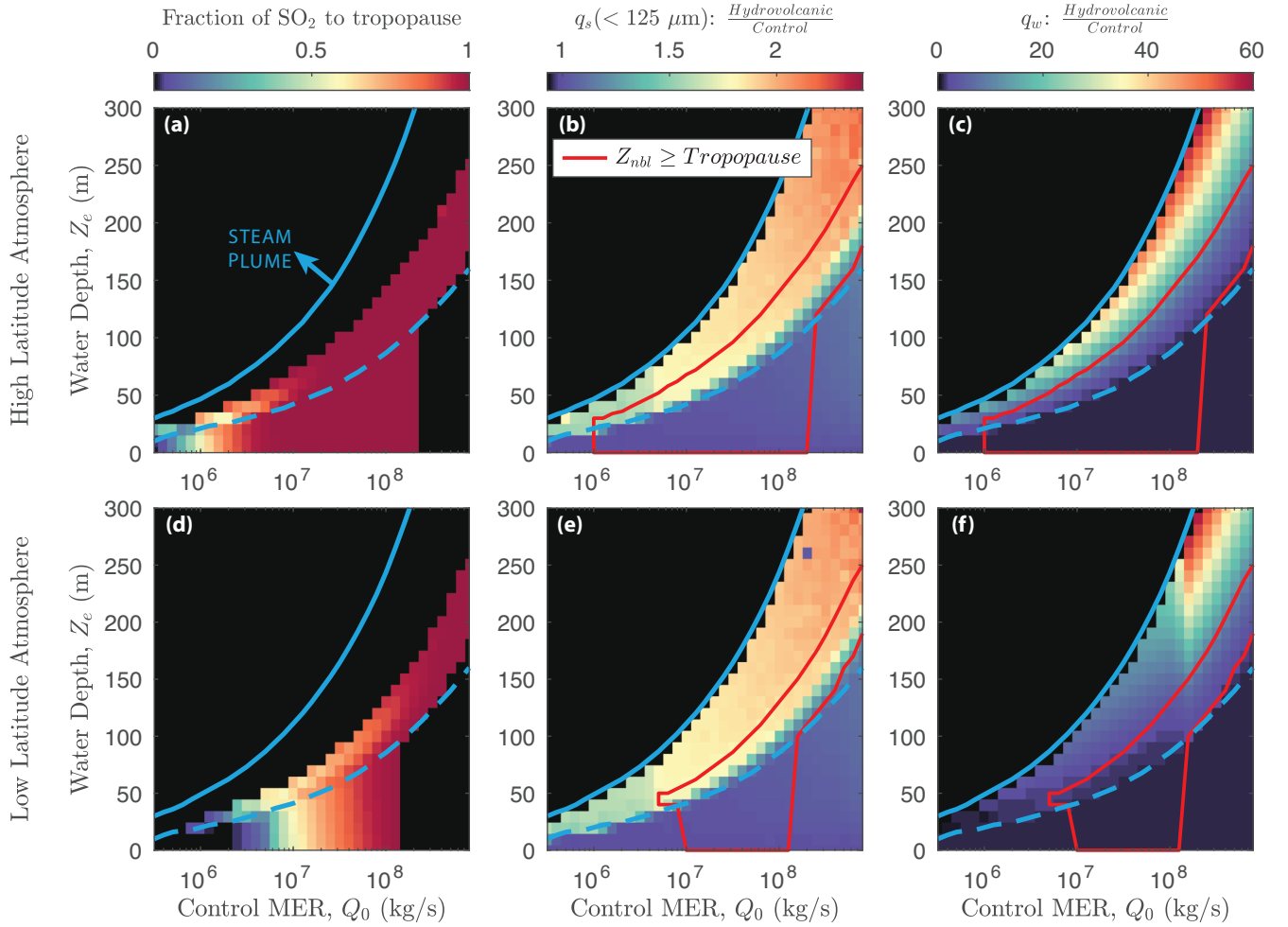


Figure 13. Estimated fraction of SO_2 , fine ash mass flux, and water mass flux to the stratosphere. (a) Estimated fraction of outgassed SO_2 injected above the tropopause assuming a gaussian injection profile centered about the height of neutral buoyancy (Equation 59), as a function of control MER Q_0 and water depth Z_e . In all panels, the dashed blue line is threshold water depth for water entrainment (decompression length equal to water depth, $L_d = Z_e$), and the solid blue line is the threshold depth for steam plumes (see Figure 9). Black regions indicate column collapse. (b) Fine ash mass flux to the eruption column maximum height as a ratio of hydrovolcanic ($Z_e > 0$) to control ($Z_e = 0$) simulations, for particle diameters less than $125 \mu\text{m}$. Red line outlines simulations with buoyant plumes at spreading heights at or above the tropopause. (c) Water mass flux to the eruption column maximum height as a ratio of hydrovolcanic ($Z_e > 0$) to control ($Z_e = 0$) simulations. Black regions indicate the steam plume regime in panels (b), (c), (e), (f). Panels (a)-(c) are for with a high latitude (Iceland) atmospheric profile (*Reference* scenario). Panels (d)-(f) are the same as (a)-(c), respectively, but for the low latitude (Equador) atmosphere (*Low-lat* scenario).