

# Detection of organic carbon in Mars-analog paleosols with thermal and evolved gas analysis

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## Key points

- Trace amounts of organic carbon (OC) and organic fragments in Mars-analog paleosols were detected with thermal and evolved gas analysis
- The near-surface horizons of 33-million-year-old paleosols had significantly higher OC content relative to deeper layers
- Radiocarbon dates of approximately 6-14 thousand years were consistent with small amounts of modern carbon in paleosol samples

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**36 Abstract**

37 Ancient, buried soils, or paleosols, may have been preserved in the geological record on Mars, and are  
38 considered high-priority targets for biosignature investigation. Studies of paleosols on Earth that are  
39 similar in composition to putative martian paleosols can provide a reference frame for constraining their  
40 organic preservation potential on Mars. However, terrestrial paleosols typically preserve only trace  
41 amounts of organic carbon, and determining what carbon is original is complicated by diagenesis and  
42 additions of modern carbon. The objectives of this study were a) to determine whether organic carbon in  
43 Mars-analog paleosols can be detected with thermal and evolved gas analysis, and b) constrain the age of  
44 organic carbon using radiocarbon ( $^{14}\text{C}$ ) dating. Oligocene (33 Ma) paleosols from Oregon were examined  
45 with an instrument similar to the Sample Analysis at Mars Evolved Gas Analysis (SAM-EGA) instrument  
46 onboard the Mars Science Laboratory *Curiosity* rover. Trace amounts of organic carbon and fragments of  
47 organic molecules were observed in all samples. Total organic carbon (TOC) ranged from 0.002 - 0.032  $\pm$   
48 0.006 wt. %. The near-surface horizons of paleosols had significantly higher TOC relative to subsurface  
49 layers. Radiocarbon dating of four samples revealed an organic carbon age of  $\sim$ 6,200 – 14,500 years  
50 before present and a fraction modern ranging from 0.16-0.46. Modeled abundances of modern carbon in  
51 bulk samples ranged from 0.41 – 3.1 %  $\pm$  0.11%, which were consistent with additions of small amounts  
52 of modern organic carbon. This work demonstrates that martian paleosols are a potential high priority  
53 location for *in-situ* biosignature investigation.

**54 Plain language summary**

55 Ancient, buried soils, or paleosols, may have been preserved in the geological record on Mars. On Earth,  
56 paleosols that are billions of years old contain signs of ancient life (biosignatures) and therefore paleosols  
57 on Mars are considered high-priority locations for biosignature investigation. One way to determine the  
58 biosignature preservation potential of possible martian paleosols is to examine organic carbon  
59 preservation in paleosols from Earth that resemble martian paleosols. The objectives of this work were to  
60 determine if carbon in Mars-analog paleosols can be detected by an instrument configured to operate like  
61 the Sample Analysis at Mars (SAM) Evolved Gas Analyzer (SAM-EGA) onboard *Curiosity* rover, and to  
62 use radiocarbon dating to determine if samples contain modern carbon. Trace amounts of organic carbon  
63 and fragments of organic molecules were observed. Like modern soils, the near-surface horizons of all  
64 paleosols had significantly higher amounts of carbon relative to subsurface layers. Radiocarbon dating  
65 revealed an organic carbon age of  $\sim$ 6,200 – 14,500 years before present. Application of an isotopic  
66 mixing model suggested these dates could be explained by slight additions of modern carbon to ancient  
67 samples. These results demonstrate that putative martian paleosols are a potential high priority location  
68 for *in-situ* biosignature investigation.

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## 72 **Introduction**

73 Paleosols are ancient, buried soils that are commonly lithified into sedimentary rocks. Terrestrial  
74 paleosols are a geological record of the atmospheric composition, climate, topography and organisms  
75 present before soil burial (Retallack, 2019). On Mars, paleosols, also known as weathering profiles, may  
76 have formed in sediments such as basaltic sand or volcanic ash that were subject to subaerial weathering  
77 by surface waters (Retallack, 2014; Amundson, 2018; Liu et al., 2021b; Ye and Michalski, 2021) and  
78 were subsequently buried and preserved in the geological record. Orbital remote sensing of the global  
79 martian surface has detected dioctahedral clay minerals within Noachian (4.1-3.7 Ga) layered sedimentary  
80 rocks, which are consistent with precipitation-driven pedogenic weathering of mafic sediments (Carter et  
81 al., 2015; Bishop et al., 2018b; Loizeau et al., 2018). Noachian sedimentary rocks with spectral signatures  
82 of subaerial weathering have been detected in thousands of locations across the surface of Mars (Bishop  
83 et al., 2018b). One hypothesis is that these deposits are paleosols (Carter et al., 2015), which are the  
84 common products of pedogenic alteration followed by burial. Mounting evidence of global-scale aqueous  
85 alteration of the Martian surface during the Noachian (Carter et al., 2015; Liu et al., 2021a) suggests that  
86 pedogenesis could have been a critical process early in the planet's history. As such, paleosols have been  
87 recently named a high priority location for biosignature investigation (Bishop et al., 2018a) and Mars  
88 Sample Return (Beaty et al., 2019), but the biosignature preservation potential of paleosols with Mars-like  
89 mineralogy remains poorly constrained (Horgan, 2016).

90 On Earth, soils are highly habitable environments. Modern soils are teeming with microbial  
91 biomass, often averaging  $10^{10}$ – $10^{11}$  bacterial cells and  $10^3$  and  $10^4$  species per gram of soil (Raynaud and  
92 Nunan, 2014). Modern soils also contain more organic carbon than global vegetation and the atmosphere  
93 combined (Lehmann and Kleber, 2015; Dynarski et al., 2020). Similarly, Earth's oldest soils also appear  
94 to have been highly habitable environments. Many Precambrian ( $> 541$  Ma) paleosols contain organic  
95 carbon and other chemical biosignatures that are thought to be remnants of surface biomass (Matthewman  
96 et al., 2012; Kremer et al., 2017; Liivamägi et al., 2018; Broz, 2020). Furthermore, Archean ( $> 2$  Ga)  
97 paleosols contain filamentous organic carbon and organo-mineral complexes possibly derived from  
98 cyanobacterial mats on the soil surface (Rye and Holland, 2000; Watanabe et al., 2000).

99 Many terrestrial paleosols preserve only trace amounts of organic carbon, especially compared to  
100 modern soils, marine shales, and lacustrine rocks (Retallack, 2019). Organic carbon losses during  
101 diagenesis can reduce the organic carbon content of paleosols by up to two orders of magnitude relative to  
102 their modern soil counterparts (Broz, 2020). Severe losses of organic carbon are most common in  
103 paleosols that originally formed under oxidizing, well-drained conditions (Retallack and Mao, 2019). In  
104 general, oxidized paleosols typically contain only low amounts ( $< 0.1$  wt. %) of organic carbon (Broz,  
105 2020). Diagenetic losses of organic carbon in oxidized terrestrial paleosols poses major challenges for

106 detection of chemical and isotopic biosignatures preserved within the organic matter fraction. However,  
107 this is not the case for all paleosols. Those that originally formed under reducing conditions, such as  
108 Permian (~250 Ma) Histosols (poorly drained organic soils), preserve organic carbon with abundances >  
109 25 wt. % (Retallack and Krull, 1999), implying that redox state before burial may provide a first-order  
110 control on the preservation of organic carbon in ancient soils (Krull and Retallack, 2000).

111 An additional concern for the study of terrestrial paleosol organic matter is that diagenetic  
112 alterations ranging from groundwater alteration to precipitation-driven leaching of dissolved organic  
113 carbon can result in the addition of exogenous (e.g., allochthonous) organic molecules, so caution is  
114 necessary for interpreting whether the organic fraction has indeed been “preserved” over geological time  
115 scales. In other words, terrestrial paleosols can be contaminated by organic molecules that were not  
116 original to the soil, thereby complicating efforts to interpret their biosignature preservation potential.

117 One way to constrain the biosignature preservation potential of putative paleosols on Mars is to  
118 examine the organic fraction of paleosols from Earth that share compositional and morphological  
119 similarities to layered sedimentary rocks on Mars. Examination of oxidized paleosols with Mars-like  
120 mineralogy can help determine if low amounts of organic molecules within natural pedogenic mineral  
121 matrices can be detected with Mars flight-analog instruments. Furthermore, cosmogenic nuclide dating of  
122 these analog paleosols can also identify potential diagenetic additions of organic carbon.

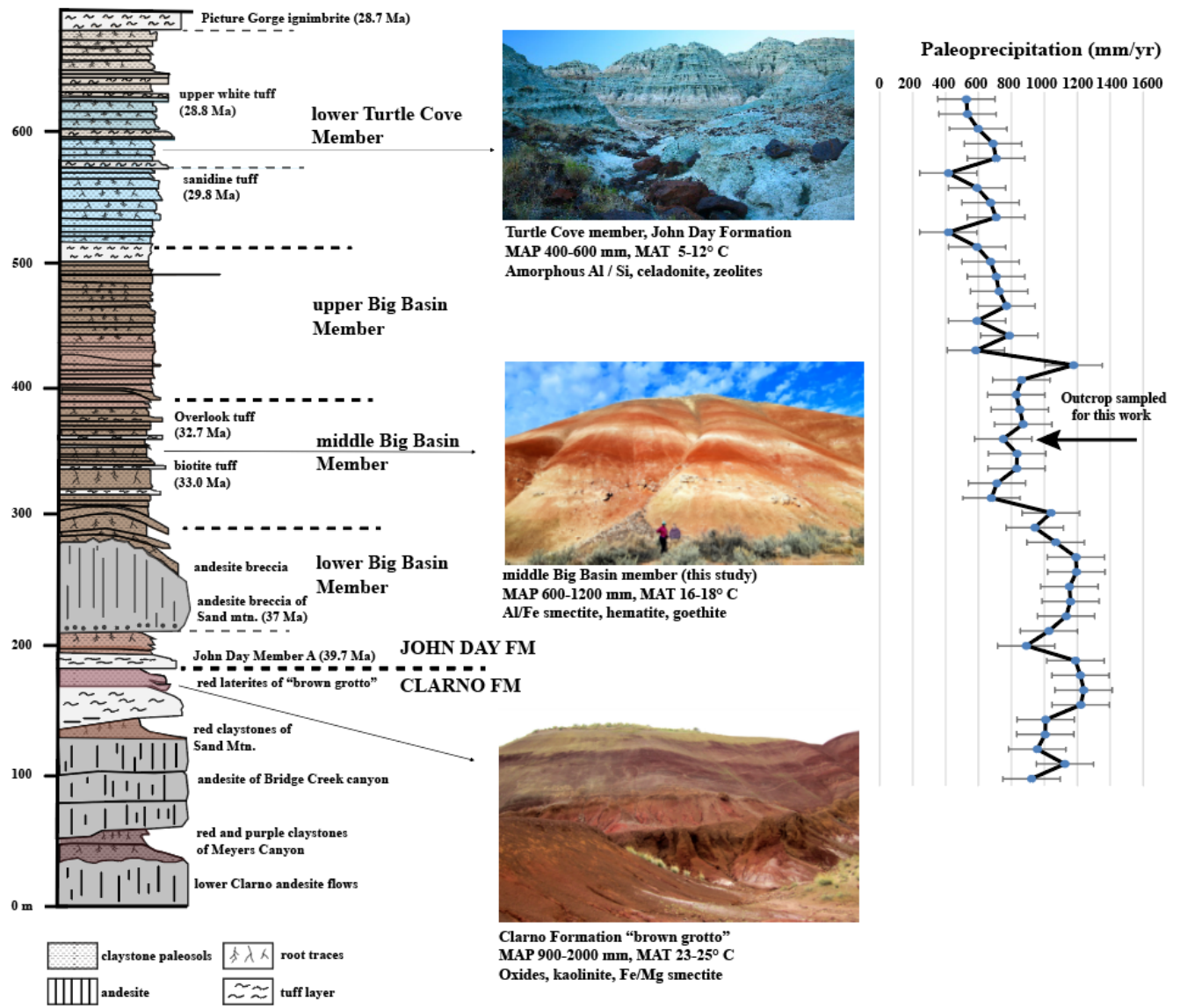
123 This study builds on and combines previous results from an established Mars-analog paleosol  
124 sequence (Broz et al., 2021, 2022). The focus here is to determine if organic molecules in oxidized  
125 paleosols are detectable with Mars flight-analog instrumentation, and to determine if there have been  
126 post-depositional additions of modern organic carbon. The objectives of this study were a) to determine  
127 whether organic carbon in ~33-million-year-old Mars-analog paleosols can be detected with thermal and  
128 evolved gas analysis, and b) constrain the age of organic carbon using radiocarbon ( $^{14}\text{C}$ ) dating.

129

### 130 **1.1 Paleosols at John Day Fossil Beds National Monument in eastern Oregon, USA**

131 Eocene and Oligocene (42 to 28 Ma) sedimentary rocks at John Day Fossil Beds National  
132 Monument in eastern Oregon are a thick sequence of volcanoclastic paleosols, which altogether spans  
133 over 400 meters of vertical stratigraphy (Bestland, 1997; Retallack et al., 2000; Horgan et al., 2012; Smith  
134 et al., 2018a). The paleosol sequence contains over 500 clay-rich (30-95 wt. %) paleosols formed over  
135 ~15 Myr through the Eocene-Oligocene boundary (Figure 1) (Retallack et al., 2000; Horgan et al., 2018).  
136 Each of the individual profiles formed from pedogenic alteration of andesitic to rhyodacitic volcanic ash  
137 and/or tuff, followed by rapid burial via emplacement of an additional layer of tephra onto the soil  
138 surface. This process of soil formation and burial repeated for nearly 15 million years, and as such, the  
139 paleosol sequence provides a unique record of Cenozoic climate change in eastern Oregon (Fremd, 1996).

140 Changes in mineralogy throughout the paleosol sequence reflect the Eocene-Oligocene cooling and  
 141 drying of the climate (Bestland, 2002; Retallack et al., 2004).



142

143 **Figure 1. A sequence of Eocene and Oligocene (42-28 Ma) volcanoclastic paleosols in the Clarno and John Day**  
 144 **Formations, John Day Fossil Beds National Monument (after Bestland (1997) and Sheldon et al. (2002).**  
 145 Declining precipitation through the Eocene/Oligocene boundary is inferred from the mineralogy of paleosols from  
 146 the Clarno and John Day Formations. Paleoprecipitation estimates are from Sheldon et al. (2002) who used an  
 147 equation relating molecular weathering ratios (Bases/alumina [ $\text{CaO} + \text{MgO} + \text{Na}_2\text{O} + \text{K}_2\text{O} / \text{Al}_2\text{O}_3$ ] to mean annual  
 148 precipitation from a database of modern U.S. soils ( $r=0.79$ , standard error = 179 mm). The stratigraphic level of  
 149 paleosols analyzed in this work are indicated (black arrow, far right).  
 150

151 The Eocene (42-39 Ma) Clarno Formation represents the lowest and oldest unit of the Oregon  
 152 paleosol sequence. The basal Clarno Formation is characterized by andesite flows interspaced with

153 severely weathered paleosols with accumulations of kaolinite, Fe/Mg smectite and oxides (Oxisols and  
154 Ultisols in US soil taxonomy), which indicate tropical weathering conditions in the middle to late Eocene  
155 (Retallack et al., 2000). A particularly striking exposure of the late Eocene Clarno Formation at the  
156 “Brown Grotto” area of the Painted Hills is characterized by thick (~2-6 meter), repeated profiles of  
157 deeply weathered lateritic paleosols (Figure 1, bottom) that are similar to modern soils from Southern  
158 Mexico and Central America in climates that are subtropical and humid (Retallack et al., 2000).  
159 Stratigraphically above the Clarno Formation, the early Oligocene (~33 Ma) Big Basin Member of the  
160 John Day Formation is characterized by less intensely weathered paleosols (Alfisols and Inceptisols) that  
161 are rich in Al and Fe smectites such as nontronite and montmorillonite (Figure 1, middle). This middle  
162 unit represents a dramatic cooling and drying of the climate through the Eocene-Oligocene boundary.  
163 Overlying this unit of the paleosol sequence is the mid-late Oligocene (~28 Ma) Turtle Cove Member of  
164 the John Day Formation (Figure 1, top). The brown, green and celadon-colored paleosols of this unit are  
165 minimally weathered Aridisols (desert soils) and Andisols (amorphous-rich volcanic soils). Paleosols in  
166 this unit are characterized by accumulations of amorphous and nanocrystalline materials (e.g., allophane  
167 and imogolite), pedogenic calcite, and diagenetic celadonite (Horgan et al., 2012). There is a dramatic  
168 reduction in smectite content and absence of kaolinite, which is consistent with weathering under a semi-  
169 arid to arid climate regime (Bestland, 2002). Finally, the entire paleosol sequence is capped by  
170 approximately 300 vertical meters of flood basalts from the Miocene (~16 Ma) Columbia River Basalt  
171 Group. Based on these mineralogical transitions, the Eocene paleosols with accumulations of kaolinite  
172 and oxides appear to reflect a subtropical and humid climate, while accumulations of nanophase  
173 aluminosilicates and amorphous phases in overlying paleosols indicate stepwise cooling and drying  
174 during the late Oligocene (Retallack et al., 2000).

175

## 176 **1.2 Pedogenic weathering on early Mars and comparisons to terrestrial paleosols**

177 From a global perspective, orbital remote sensing from Observatoire pour la Minéralogie, l'Eau,  
178 les Glaces, et l'Activité (OMEGA) and the Compact Reconnaissance Spectrometer at Mars (CRISM)  
179 instrument onboard the Mars Reconnaissance Orbiter provide the most compelling evidence of extensive  
180 pedogenic-like alteration across the surface of Mars. Potential weathering sequences have been detected  
181 in hundreds of locations across Noachian terrains, wherever these ancient rocks are not obscured by dust,  
182 sand, or overlying strata (Carter et al., 2015; Loizeau et al., 2018). OMEGA and CRISM have collectively  
183 gathered visible and near-infrared (VNIR) reflectance spectra of sedimentary deposits at Mawrth Vallis,  
184 Oxia Planum, Nili Fossae and other altered Noachian terrains. (Horgan et al., 2012, 2018; Hays et al.,  
185 2017; Smith et al., 2018b; Poulet et al., 2020). Spectral refinements for CRISM images (e.g., Viviano-  
186 Beck et al., 2014) now allow for identification of smaller-scale hydrated mineral deposits at Mawrth

187 Vallis that facilitate a detailed reconstruction of possible geochemical environments on early Mars  
188 (Bishop et al., 2020). Some of these deposits, such as layered outcrops at Mawrth Vallis, have spectral  
189 and stratigraphic similarities to terrestrial paleosol sequences (Horgan et al., 2012; Broz et al., 2022).

190

### 191 **Terrestrial paleosols: Mineralogy and diagenesis**

192 The present study focuses on three paleosol profiles from the early Oligocene (33 Ma) Big Basin  
193 Member of the John Day Formation (Figure 1, middle). A detailed analysis of mineralogy and diagenetic  
194 alteration of these samples was previously performed (Broz et al., 2022). X-ray diffraction, evolved gas  
195 analysis and visible-near-infrared spectroscopy revealed high abundances (> 80 wt. %) of  
196 montmorillonite and nontronite with lesser amounts of hematite, zeolites, gypsum, and hydrated silica  
197 (Tables S4-S6). Diagenetic alterations previously observed in these samples included a) "burial  
198 gleization" of near-surface horizons, which is attributed to microbial reduction of  $\text{Fe}^{3+}$  in near-surface  
199 horizons of paleosols resulting from anaerobic decay of organic matter; b) dehydration of ferrous  
200 oxyhydroxides (goethite) to form fine-grained hematite; c) zeolitization to form clinoptilolite, possibly  
201 resulting from diagenetic recrystallization of a poorly crystalline smectite; and d) significant mechanical  
202 compaction to approximately 70% of the original soil thickness (Retallack et al., 2000).

203 Previously, there was no effort to examine the organic component of these paleosols, and the  
204 resulting influence of diagenesis on the organic fraction of Mars-analog paleosols is poorly understood.  
205 Examination of terrestrial paleosols with Mars flight-analog instruments such as evolved gas analysis  
206 allows for a detailed characterization of the organic fraction (discussed below). This can help ascertain if  
207 diagenesis has resulted in severe losses of the organic carbon in Mars-analog paleosols and determine if  
208 organic carbon that remains is detectable with analytical techniques relevant to Mars exploration.

209

### 210 **1.3 Sample Analysis at Mars (SAM) instrument onboard *Curiosity* Mars Rover**

211 The overall goal of the SAM instrument is to assess the potential for past habitability by  
212 characterizing the martian chemical and isotopic composition of the atmosphere and volatile-bearing  
213 surface materials (Mahaffy et al., 2012). The SAM instrument is integral in providing an understanding of  
214 organic materials and phases undetectable by CheMin (e.g., amorphous phases, low abundance phases).  
215 SAM heats scooped or drilled rock samples from  $\sim 30 - 870^\circ \text{C}$  at  $35^\circ \text{C min}^{-1}$ . Evolved gases produced  
216 from the thermal decomposition of volatile-bearing phases are analyzed by a quadrupole mass  
217 spectrometer (QMS), gas chromatograph columns for GCMS, or a tunable laser spectrometer (TLS)  
218 (Mahaffy et al., 2012). In evolved gas analysis (SAM-EGA) mode, SAM detects bulk gas evolution,  
219 whereas in gas chromatography-mass spectroscopy (SAM-GCMS) mode, SAM performs molecular  
220 separation and identification of organic molecules (Mahaffy et al., 2012; Eigenbrode et al., 2018). This

221 study focused on SAM-EGA, so SAM-GCMS will not be further discussed. A comprehensive discussion  
222 on how organic molecules are detected with SAM- GCMS mode can be found in Mahaffy et al. (2012)  
223 and Szopa et al. (2020).

224 Evolved gases and their release temperatures detected by SAM-EGA provide constraints on the  
225 mineralogy and organic content of samples in Gale Crater (Archer et al., 2014; Ming et al., 2014;  
226 Freissinet et al., 2015; Eigenbrode et al., 2018). The thermal decomposition of solid samples during  
227 SAM-EGA occurs during ramped heating, which releases volatile gases including CO<sub>2</sub>, CO,  
228 hydrocarbons, and organic fragments (i.e., CH<sub>2</sub>, CH<sub>3</sub>, C<sub>2</sub>H<sub>2</sub> and others) that are detected by the QMS. The  
229 intensity (relative abundance) of volatile release is plotted as a function of the release temperature,  
230 generating a time and temperature series of data for each volatile gas release. The volatile release peak  
231 temperature during sample decomposition depends on the thermodynamics of the reaction and can be  
232 used to constrain the composition of minerals and organic carbon in the sample (Archer et al., 2014), as  
233 well as to identify possible associations between minerals and organics (McAdam et al., 2020b).

234 Additional future missions to Mars will also employ EGA-like analysis to search for organic  
235 molecules. The Mars Organic Molecule Analyzer (MOMA) onboard European Space Agency's ExoMars  
236 2022 *Rosalind Franklin* rover will use pyrolysis gas chromatography-mass spectrometry (GCMS) and  
237 laser desorption spectroscopy to search for biosignatures on Mars. The *Rosalind Franklin* rover will land  
238 at Noachian (3.9 Ga) Oxia Planum, which appears to be a westward extension of the lower parts of the  
239 stratigraphy observed at Mawrth Vallis (Ivanov et al., 2020; Loizeau et al., 2020). Like Mawrth Vallis,  
240 strong and ubiquitous spectral signatures of dioctahedral Al-rich clay minerals overlying Fe/ Mg clay  
241 minerals suggests Oxia Planum may host remnants of a thick (~200 m) deep weathering profile or  
242 paleosol sequence that the rover could encounter during its primary mission.

243

#### 244 **1.4 Previous detections of organic carbon with SAM-EGA**

245 Organic carbon has been detected in sedimentary rocks at Gale Crater with the SAM instrument  
246 using both QMS and GCMS (Ming et al., 2014; Rampe et al., 2014; Freissinet et al., 2015; Szopa et al.,  
247 2020). Abundances of reduced carbon were very low (< 1 wt. %) and restricted to three samples  
248 (Cumberland [CB], Confidence Hills [CH], and Mojave [MJ]). Though all CO<sub>2</sub> and CO detected to date is  
249 consistent with oxidized organic compounds, it is possible that contamination by the SAM-GCMS  
250 derivatization agent MTBSTFA could have resulted in these peaks. However, high-temperature (> 600°  
251 C) CO releases were consistent with indigenous oxidized martian organics. Additionally, chlorinated  
252 hydrocarbons (chlorobenzene [*m/z* 112], ~30 pmol) in the Cumberland drill sample and organo-sulfur  
253 compounds including thiophenes and thiols (~90 nmol) in the Mojave and Confidence Hills drill samples  
254 were identified in ~3.5 Ga mudstone, but the sources of these organic molecules was not constrained

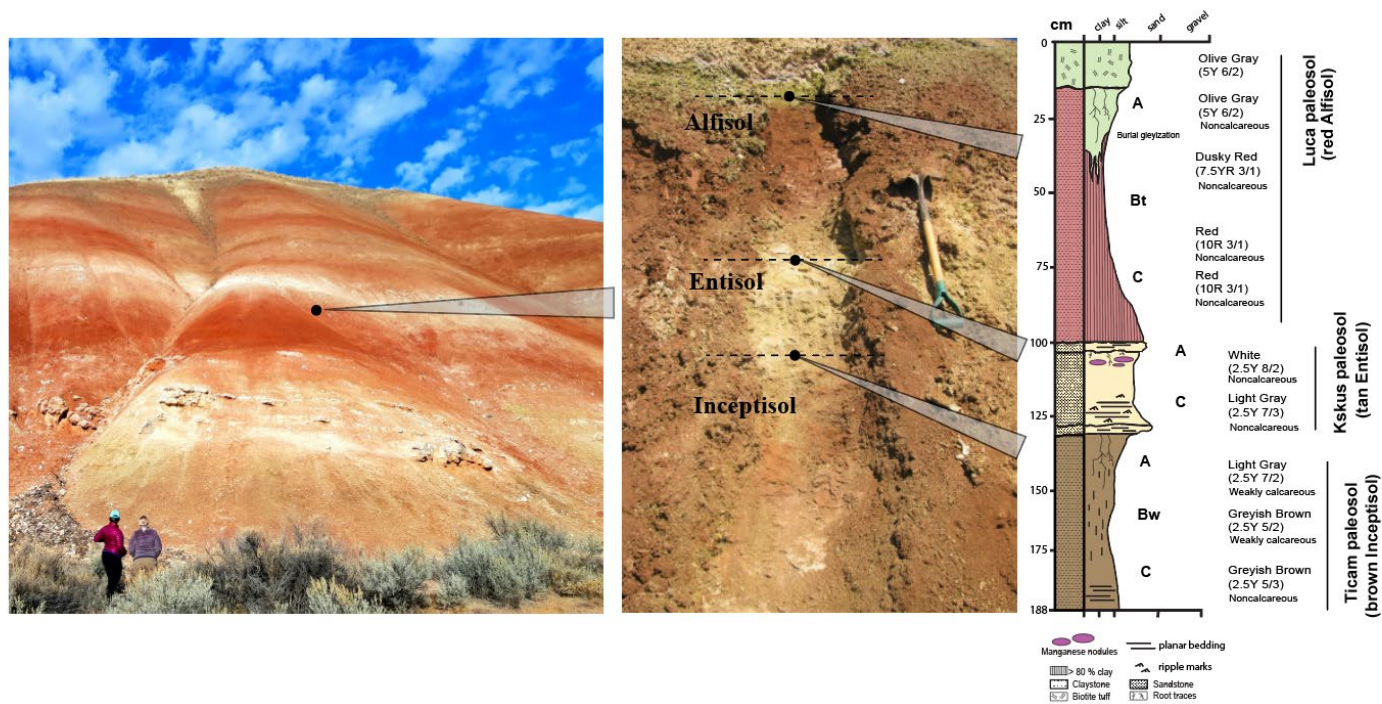


255 (Freissinet et al., 2015; Eigenbrode et al., 2018). Observation of dichlorobenzene and  
256 trichloromethylpropane in the CB sample at Yellowknife Bay could have been produced by chemical  
257 reactions between organic molecules and oxychlorines occurring in the SAM ovens. These chlorinated  
258 hydrocarbons could have been derived from organic carbon, either from an indigenous martian source  
259 and/or from meteoritic infall (Fornaro et al., 2018; Szopa et al., 2020). Though organic carbon has been  
260 detected by SAM in numerous sedimentary rocks at Gale Crater, the source(s) of the organic molecules  
261 are not yet fully understood. Recent work has suggested that organic salts such as Ca/Mg oxalates and/or  
262 Ca/Mg acetates may be present in abundances of 1-2 wt. % in modern eolian sediments (Rocknest  
263 sample) as well as in ancient sedimentary rocks at Gale Crater (JK and CB samples) (Lewis et al., 2021).  
264 The accumulation of organic salts in eolian deposits suggests that they may be a component of regional or  
265 global dust on Mars (Lewis et al., 2021).  
266

## 267 **2. Methods**

### 268 **2.1 Sample collection**

269 The paleosols examined in this study were collected from the Painted Hills Unit of the John Day  
270 Fossil Beds National Monument in eastern Oregon, USA. A previous study (Broz et al., 2022) determined  
271 the mineralogy and diagenetic alteration of the same set of samples examined in the present study.  
272 Samples from three individual paleosols in vertical succession were described and sampled approximately  
273 7 km SW of the entrance to the Painted Hills unit (Lat: 44.631105, Long: -120.213107), in the Middle  
274 Big Basin Member of the John Day Formation, approximately 6 m above the local Eocene-Oligocene  
275 boundary (Figure 1, middle). Samples were chosen from this location because they were previously  
276 examined for mineralogy and diagenesis, and because  $^{40}\text{Ar}/^{39}\text{Ar}$  dating of volcanic tuffs at this  
277 stratigraphic level in the section allow for a constrained age of 33.0 +/- 0.10 to 32.7 +/- 0.03 Ma (Biotite  
278 Tuff and Overlook Tuff, respectively) (Bestland, 1997).



280 **Figure 2. Morphology of three successive paleosols** from the early Oligocene (33 Ma) middle Big Basin Member  
 281 of the John Day Formation in eastern Oregon, USA showing lithology, grain size, horizon designations, and Munsell  
 282 color. The upper paleosol (red with drab green top) is a moderately weathered red Alfisol (Hapludalf in USDA  
 283 Taxonomy); stratigraphically below is a minimally weathered and weakly developed Entisol (Fluvent; tan color); the  
 284 lowest soil (brown color) is an Inceptisol (Andic Eutrochrept, brown color).

285

286 To minimize contamination from modern organic carbon during sampling, all loose surface soil  
 287 and saprolite was removed until the lithified, brick-like unweathered paleosol surface was exposed.  
 288 Below the saprolite of the thin (~30 cm) modern soil, all three paleosols were lithified claystone. Before  
 289 collection, all sampling and storage materials were ashed at 550° C to remove organic contaminants.  
 290 Sample collection for evolved gas analysis (EGA) began by trenching with a rock hammer to a ~30 cm  
 291 depth into (perpendicular to) the lithified paleosol sequence. Next, a set of samples were gathered down a  
 292 vertical transect at approximately 10 cm intervals, similar to sampling the horizons of a modern soil.  
 293 Large (~1 kg) lithified blocks of claystone were broken out of the outcrop and placed into aluminum foil  
 294 to ensure subsampling for thermal analysis had sufficient volume to expose fresh sample surface (e.g., no  
 295 weathered surfaces were selected for analysis). Small (5-8 g) subsamples were acquired from the inside of  
 296 bulk lithified samples with steel chisels. These subsamples were then ground to < 0.02 mm using an agate  
 297 mortar and pestle and stored in glass vials prior to thermal analysis and radiocarbon dating. No vegetation  
 298 was present within ~30 m of the sampled paleosol sequence (Figure 2), likely because the “popcorn”  
 299 weathering of the smectite-rich modern soil appears to inhibit plant germination and growth. The Munsell

300 color and qualitative calcareousness of samples were described during collection. The three paleosols  
301 sampled were a red Alfisol (“Luca” pedotype from (Retallack et al., 2000), a tan Entisol (“Kskus”  
302 pedotype), and a brown Inceptisol (“Ticam” pedotype, Figure 2).

## 303 **2.2 Radiocarbon dating of organic carbon in paleosols**

304 The purpose of radiocarbon ( $^{14}\text{C}$ ) dating was to constrain the age of organic carbon in paleosol  
305 samples. Specifically, radiocarbon dating was used to determine if the organic carbon fraction of  
306 paleosols was entirely endogenous (e.g., autochthonous, deposited during soil formation) and had been  
307 preserved for millions of years, or if there had been additions of exogenous (allochthonous) organic  
308 carbon to paleosols in the last ~45,000 years (a typical method-level detection limit for radiocarbon dating  
309 techniques).

310 For radiocarbon analysis, samples were gathered by trenching at a horizontal angle into the  
311 outcrop, whereas sampling for EGA was performed vertically (down section) once the surface soil had  
312 been removed from the outcrop (Figure 2). In other words, EGA samples were collected down section  
313 (e.g., vertical depth in the section) whereas radiocarbon samples were obtained by trenching further into  
314 the outcrop (e.g., horizontal depth into the outcrop) at the vertical intervals where EGA samples were  
315 collected. The purpose of this sampling difference was to help determine if the horizontal sampling depth  
316 into the outcrop was related to the amount of radiocarbon accumulation in the brick-like paleosol samples  
317 collected within ~0.5 meters of the modern weathering zone.

318 A radiocarbon age of organic carbon was obtained from four samples, two from the exposed  
319 surface and near-surface horizons (A and Bt horizons) of the stratigraphically highest soil (Alfisol) and  
320 two from the surface and near-surface horizons of the stratigraphically lowest profile (Inceptisol). All  
321 samples for radiocarbon dating were acid-washed to remove inorganic carbonates. Ground paleosol  
322 samples (~5 g) were treated with approximately 20 mL of 0.1 M HCl at room temperature for 1 hour, then  
323 washed three times with ~30 ml of deionized water and dried at 60° C for 24 hr. Radiocarbon dating of  
324 acid-washed paleosol samples was performed at the W.M. Keck Carbon Cycle Accelerator Mass  
325 Spectrometer at the University of California, Irvine. Additional TOC determination of these samples was  
326 performed at UC Irvine and are reported in Table 1. The accuracy and precision ( $1\ \sigma$ ) of this analysis on  
327 modern carbon ( $\Delta^{14}\text{C} > 0\text{‰}$ ) was better than 9‰. Laboratory blanks yielded a  $\Delta^{14}\text{C}$  value of -996.2 ‰.

328

## 329 **Quantifying additions of modern carbon to bulk paleosol samples**

330 Calibrated radiocarbon dates, taken at face value, may represent a mixture of radiocarbon-free  
331 Oligocene carbon and some amount of post-bomb modern organic carbon. To test the hypothesis that

332 paleosols contained modern organic carbon, we used a two-endmember mixing model to estimate the  
 333 relative proportion of modern carbon in bulk paleosol samples. Based on the distinct isotopic composition  
 334 of modern organic carbon and radiocarbon-free Oligocene carbon, a two-endmember mixing model  
 335 (Sickman et al., 2010; Silva et al., 2013) was used to quantify the relative proportions of modern and  
 336 ancient carbon as distinct sources of the paleosol organic carbon pool (Zech et al., 2017). We used the  
 337 following equation to partition modern organic carbon sources ( $D^{14}C = 0 \text{ ‰}$ ) from radiocarbon-dead  
 338 Oligocene organic carbon ( $D^{14}C \sim -999\text{‰}$ ):

339

$$340 \quad C_{\text{modern}} = C_t(D^{14}C_{\text{bulk}} - D^{14}C_{\text{Oligocene}})/(D^{14}C_{\text{Modern}} - D^{14}C_{\text{Oligocene}}) \quad (1)$$

341 Where  $C_t$  is the total amount of organic carbon (TOC) measured in bulk samples,  $D^{14}C_{\text{bulk}}$  is the measured  
 342  $D^{14}C$  value of bulk samples,  $D^{14}C_{\text{Modern}}$  is a typical value for a modern organic carbon endmember ( $D^{14}C$   
 343  $= 0 \text{ ‰}$ ),  $D^{14}C_{\text{Oligocene}}$  is a typical  $D^{14}C$  value for a radiocarbon-free organic carbon endmember ( $D^{14}C = -$   
 344  $999\text{‰}$ ), and  $C_{\text{modern}}$  is the modelled fraction of modern organic carbon in bulk samples. Errors were  
 345 propagated to estimate uncertainty associated with modelled values. The sources of uncertainty  
 346 considered in the model were a) the uncertainty of the measured TOC values and b) uncertainty of the  
 347 measured  $D^{14}C$  values, both from analysis of duplicate samples.

### 348 **2.3 Thermal and evolved gas analysis of paleosol samples**

349 The purpose of this work was to use SAM-EGA-like conditions to characterize bulk gas evolution  
 350 and to measure abundance of organic and inorganic carbon in Mars-analog paleosol samples. A Setaram  
 351 Labsys Evo differential scanning calorimeter (DSC) / thermal gravimeter (TG) connected to a Pfeiffer  
 352 Omnistar QMS was configured to operate similarly to the SAM evolved gas analyzer. The SAM  
 353 instrument does not have TG/DSC capabilities, but these components permit a better understanding of  
 354 phase transitions and chemical reactions in laboratory experiments. Approximately  $50 \text{ mg} \pm 3 \text{ mg}$  of  
 355 ground paleosol sample (previously stored in glass vials ashed at  $550^\circ \text{C}$  to minimize organic  
 356 contamination) were placed in an  $\text{Al}_2\text{O}_3$  sample crucible (also previously ashed at  $550^\circ \text{C}$ ). The sample  
 357 crucible and an identical empty reference crucible were placed in the furnace and then the system was  
 358 purged twice with helium gas and set to a pressure of 3 kPa. Helium was chosen as a carrier gas because it  
 359 is inert and because it used as a carrier gas in the SAM instrument. The crucibles were heated from  
 360 approximately  $35^\circ \text{C}$  to  $1000^\circ \text{C}$  at a heating rate of  $35^\circ \text{C}/\text{min}$  and at a flow rate of  $10 \text{ cm}^3/\text{s}$ . Volatiles  
 361 ranging from mass/charge ( $m/z$ ) 1 - 100 were measured. All analyses were performed in duplicate.

362 Total organic carbon (TOC) content was determined using a Netzsch TG/DSC coupled to a  
363 Pfeiffer QMS using a modified method of Fernández et al., (2012). An Al<sub>2</sub>O<sub>3</sub> sample crucible and an  
364 identical reference crucible were placed in the furnace. The instrument was purged twice with ultra-high  
365 purity O<sub>2</sub> and set to a pressure of 1000 mbar prior to sample analyses to remove any contamination in the  
366 system. Oxygen was chosen as a carrier gas because it encourages complete combustion of all organic  
367 and inorganic carbon in samples. The crucibles containing samples were heated from approximately 35  
368 °C to 1000 °C at a heating rate of 35°C/min and at a flow rate of 19 ml O<sub>2</sub>/min. A series of three blanks  
369 were analyzed before and after each group (n=10) of samples. A calibration curve for CO<sub>2</sub> was created by  
370 analyzing a calcite standard (Iceland sparry calcite 40µM) at eight sample masses ranging from 0.01 – 4  
371 mg (Table S1). This calibration curve was used to calculate the amount of CO<sub>2</sub> evolved from each  
372 sample, and these values were used to calculate total carbon in each sample (Table S2). Errors ranged  
373 from 0.01 – 0.003 wt. % carbon.

374 Thermal techniques including TG-DSC-EGA allow for quantitative estimates of organic and  
375 inorganic carbon without sample pretreatment, in part due to the large differences in thermodynamics and  
376 peak CO<sub>2</sub> release temperatures. By contrast, TOC determinations via elemental analysis involve an acid  
377 pretreatment step to remove carbonates. Paleosol samples were not acid-pretreated for thermal and  
378 evolved gas analyses because some organic carbon can be oxidized during acid-washing (e.g., Apesteguia  
379 et al., 2018). In the present study, carbon was considered organic between 150-550° C and inorganic from  
380 ~700-900° C. Total organic carbon was quantified by deconvolving CO<sub>2</sub> peaks if a carbonate-C peak was  
381 present. This was done by determining the relative percentage of peak area from inorganic carbon-  
382 evolved CO<sub>2</sub> (~700-900° C) then subtracting this value from total carbon-evolved CO<sub>2</sub> peak area to solve  
383 for TOC. All evolved gas plots were background-corrected to account for possible atmospheric  
384 contamination.

## 285 **3. Results/Discussion**

### 386 **3.1 Radiocarbon dating of paleosol organic carbon**

387  
388 Radiocarbon dating of four samples from two different paleosol profiles showed raw D<sup>14</sup>C values  
389 ranging from -539.1‰ ± 1.3‰ to -836.7‰ ± 3.4‰ and calibrated ages between 6,265 ± 25 years BP and  
390 14,560 ± 170 years BP (Table 1). The fraction of modern carbon (FM) ranged from 0.469 ± 0.039 to  
391 0.1633 ± 0.034 (Table 1) and was highest in the surface horizon of the Alfisol and lowest in the  
392 subsurface horizon, suggesting this paleosol contained a mixture of ancient and modern organic carbon.  
393 These samples all showed a distinct signature of exogenous organic carbon because the samples were not  
394 radiocarbon dead (<sup>14</sup>C-free). Two hypotheses to explain the radiocarbon dates are A) additions of modern

395 organic carbon to bulk samples (e.g., a  $D^{14}C \sim 0\%$  modern carbon pool mixing with an ancient,  
 396 radiocarbon-free pool) or B) a Holocene (6-14 Ka) productivity event introduced carbon into the  
 397 paleosols (e.g., the carbon is indeed tens of thousands of years old).

398 One possible source of exogenous organic carbon may derive from precipitation-driven leaching of  
 399 dissolved organic carbon from modern biota living in the current weathering zone above the paleosol  
 400 outcrop. As such, it is possible that small amounts of exogenous modern organic carbon from the  
 401 weathered zone above paleosol outcrops have mixed with larger amounts of  $^{14}C$ -free organic carbon  
 402 endogenous to paleosols. In this way, a radiocarbon date of  $\sim 6$ -14 Ka BP could represent a mixing of  
 403 modern organic carbon and  $\sim 33$  Ma organic carbon. This hypothesis is supported by the erosion rate for  
 404 the site, which was previously determined to be approximately  $4.94 \pm 0.05$  mm/yr. (Sweeney et al., 2015).  
 405 Using this erosion rate, the 20 cm-thick soils that formed on top of the paleosol outcrops are only about  
 406 40 years old and could have leached modern organics into the underlying paleosols during this time.

407

408 Table 1. Radiocarbon dates and application of a two-endmember isotopic mixing model to measured  $D^{14}C$   
 409 values from four bulk paleosol samples. Depth (cm) represents the horizontal depth into the outcrop from  
 410 where samples were gathered. TOC, total organic carbon; FM, fraction of modern.

Sample ID	Depth (cm)	TOC (%)	TOC $\pm$	FM	$D^{14}C$	$\pm$	$^{14}C$ age (years BP)	$\pm$	Modern carbon (%)	$\pm$
Alfisol	32	0.015	0.001	0.163	-836.7	3.4	<b>14560</b>	170	0.487	0.066
Alfisol	14	0.010	0.001	0.461	-539.1	3.9	<b>6220</b>	70	3.141	0.105
Inceptisol	20	0.031	0.001	0.459	-541.5	1.3	<b>6265</b>	25	0.916	0.032
Inceptisol	22	0.018	0.001	0.437	-563.3	1.8	<b>6655</b>	35	0.872	0.056

411

412 Application of a two-endmember isotopic mixing model to raw  $D^{14}C$  values (Equation 1) for  
 413 estimation of modern organic carbon abundances in bulk paleosol samples is shown in Table 1. The  
 414 modelled abundances of modern carbon ranged from  $0.487 \pm 0.066$  % and  $3.141 \pm 0.105$  % of the total  
 415 organic carbon in each sample. These results support the hypothesis that the measured  $D^{14}C$  values  
 416 represent the mixing of small amounts of modern organic carbon with larger amounts of presumably  
 417 radiocarbon-free Oligocene carbon. There was also a significant power law relationship between the  
 418 modeled organic carbon abundance and the horizontal depth into the profile from where the sample was  
 419 collected ( $y = 21.518x^{-0.418}$ ,  $R^2 = 0.9194$ , **Figure S1**). Modelled abundances of modern carbon in samples  
 420 from shallower horizontal depths in the outcrop were significantly ( $P < 0.053$ ) greater compared to  
 421 samples from deeper depths (Figure S1), which was consistent with precipitation-driven leaching of  
 422 modern organic carbon into paleosol outcrops. Due to the limited sample size ( $n=4$ ), additional efforts are  
 423 needed to determine if this relationship exists across larger samples sizes and within deeper transects

424 (e.g., > 100 cm). In summary, additions of exogenous carbon were most likely modern (last 40 years),  
425 rather than from the Holocene (6-14 Ka). However, results from this work alone cannot definitively  
426 exclude the possibility of a Holocene productivity event which could have led to leaching of carbon into  
427 ancient samples during that time. These results demonstrate that a radiocarbon approach can be a useful  
428 technique for constraining the age and sources of organic carbon in ancient pedogenic samples.

429

### 430 **3.2 Thermal and Evolved Gas Analysis**

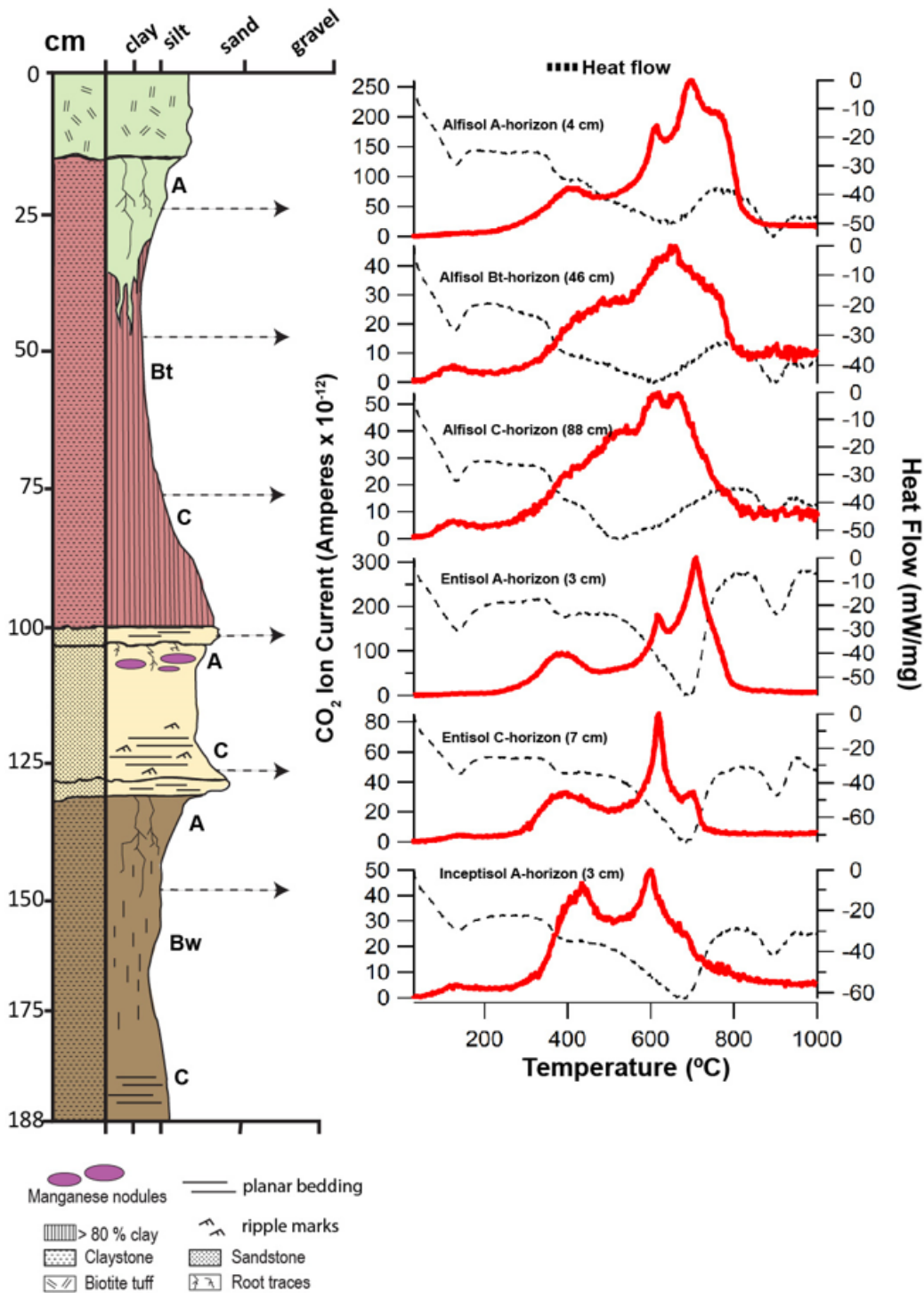
#### 431 **3.2.1 CO<sub>2</sub> and CO evolutions**

432 When subject to thermal and evolved gas analysis, all samples evolved CO<sub>2</sub> at temperatures  
433 ranging from ~150 – 800° C and had two to three distinctive CO<sub>2</sub> peaks (Figure 3). In several of the  
434 samples, a broad low temperature ~200 - 500° C peak was followed by a well-resolved (e.g., sharp) high-  
435 temperature peak at ~650-750° C. Four of the six samples also showed a well-resolved CO<sub>2</sub> peak at 600°  
436 C which was consistent with the decomposition of inorganic carbon. In general, CO<sub>2</sub> evolved at 150-500°  
437 C is primarily from organic carbon decomposition (Sutter et al., 2017; Apesteguia et al., 2018) but also  
438 possible contributions are from CO<sub>2</sub> inclusions in minerals or amorphous phases. The organic C  
439 contributing to evolved CO<sub>2</sub> is most likely from simple organic compounds (~350° C) or refractory  
440 macromolecular organic compounds (300-600° C) (Eigenbrode et al., 2018). Table 2 shows the calculated  
441 organic carbon content of EGA samples.

442

443

444



445  
 446 **Figure 3. Evolutions of CO<sub>2</sub> (red traces) and heat flow (dashed traces) from early Oligocene (33 Ma) Al/Fe**  
 447 **smectite-rich paleosols from the John Day Fossil Beds National Monument, Oregon. Red trace is CO<sub>2</sub> (*m/z* 44)**  
 448 **and dashed trace is heat flow from differential scanning calorimetry (DSC). Helium was used as a carrier gas for all**  
 449 **analyses.**

450

451



452 Table 2. Total organic carbon (TOC) and total inorganic carbon (TIC) of paleosols examined in this study. Depth  
 453 (cm) represents the vertical depth of each paleosol profile within the measured stratigraphic section

Paleosol	Horizon <sup>¶</sup>	Depth (cm)	Total C (wt. %)	TOC (wt. %) <sup>†</sup>	±σ TOC <sup>§</sup>	TIC (wt. %)
Alfisol	A	4	0.073	0.031	0.0062	0.042
Alfisol	A	14	0.094	0.026	0.0097	0.068
Alfisol	Bt <sub>1</sub>	46	0.021	0.018	0.0016	0.003
Alfisol	Bt <sub>2</sub>	63	0.067	ND <sup>∪</sup>	-	0.067
Alfisol	C	88	0.033	0.002	0.007	0.031
Entisol	A	100	0.046	0.021	0.0068	0.024
Entisol	C	120	0.036	0.013	0.0037	0.024
Inceptisol	A	125	0.024	0.018	0.007	0.006
Inceptisol	Bw <sub>1</sub>	131	0.027	0.008	0.0013	0.019
Inceptisol	Bw <sub>2</sub>	160	0.020	0.011	0.0023	0.010
Inceptisol	C	175	0.026	0.001	0.0013	0.026

454 <sup>¶</sup> Horizons follow USDA Soil Survey Staff (2014) major horizon designations

455 <sup>†</sup> Determined by thermal analysis (TG-DSC-EGA) without acid pre-treatment of samples; average of two duplicates.

<sup>∪</sup> ND = No detection; below limit of quantification

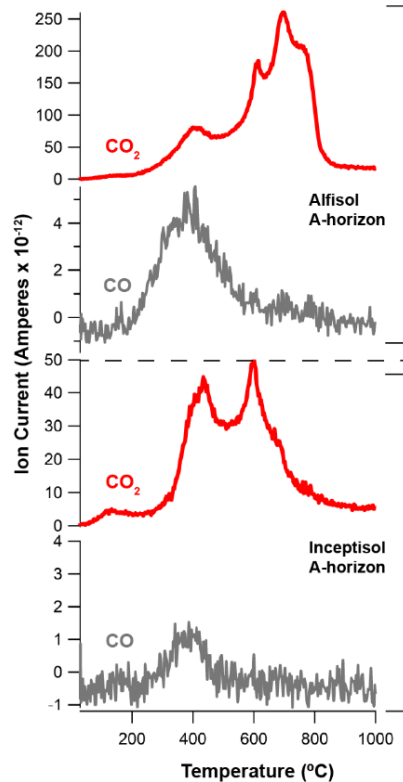
<sup>§</sup> Standard error for TOC determination from duplicate analysis

456 The endothermic thermal decomposition of Ca carbonate was a probable source of CO<sub>2</sub> release  
 457 from approximately 550 – 800° C (Figure 3) (Cannon et al., 2012; Sutter et al., 2012). High temperature  
 458 (> 650° C) CO<sub>2</sub> evolutions generally exceeded the low temperature (150-550° C) CO<sub>2</sub> evolutions in peak  
 459 area across all samples. Total inorganic carbon (TIC) values varied from 0.006 to 0.068 wt. % (Table 2)  
 460 and the ratio of inorganic carbon to total carbon ranged from 0.14 to 1, consistent with variable mixtures  
 461 of Ca carbonate and organic carbon in each sample. There are well-defined Ca carbonate endotherms in  
 462 the heat flow data from the Entisol and Inceptisol, but this trend was not so well-defined in the Alfisol  
 463 (Figure 3) and there also appears to be “doublet” high-temperature CO<sub>2</sub> peaks for the Entisol, suggesting a  
 464 combination of Ca-carbonate and perhaps dolomite or ankerite, though these phases were not previously  
 465 observed with XRD (Table S3 and Figure S2). DSC-EGA analysis of modern soils containing various  
 466 amounts of calcite have a similar sharp endothermic CO<sub>2</sub> peak release temperature at ~700° C that was  
 467 attributed to the thermal decomposition of Ca carbonate (Apesteguia et al., 2018). An additional ~900° C  
 468 endotherm observed in all samples was unrelated to carbonate decomposition and instead was attributed  
 469 to thermal decomposition of sulfate minerals such as jarosite, which was previously detected in trace  
 470 amounts with x-ray diffraction (Tables S3 and S4). Interestingly, Ca-carbonate was not previously  
 471 identified from x-ray diffraction patterns (Table S3 and Figure S2), and thus it is likely that these samples  
 472 contain inorganic carbon below detection limit of XRD (~1 wt. %), but not SAM-EGA (0.01 wt. %). This  
 473 agrees with estimated abundances of TIC, which were below 1 wt. % (Table 1) and demonstrates the

474 ability of SAM-EGA-like analyses to detect trace amounts of inorganic carbon in complex pedogenic  
 475 mineral matrices.

476 Other sources of high temperature CO<sub>2</sub> release could have result from the decarboxylation of  
 477 organic compounds in refractory or thermally mature organic matter that occur over a broad range of  
 478 temperatures (150 - 800° C). Previous investigation showed no coalification of organic matter or  
 479 development of secondary porosity in any paleosols from the Painted Hills, which were buried by an  
 480 estimated 1.5 - 2 km of overburden (Retallack et al., 2000; Horgan, 2016), suggesting paleosol samples  
 481 here contain refractory but not thermally mature organic compounds.

482 All samples evolved CO with a peak release temperature of ~350° C (Figure 4). Evolution of CO  
 483 was consistent with incomplete combustion of organics and/ or the presence of oxygen-bearing organics  
 484 (Sutter et al., 2017). We did not observe clear co-evolutions of CO and CO<sub>2</sub> because the peak release  
 485 temperature for CO was ~50° C lower than for CO<sub>2</sub> (Figure 4), but there was a distinct overlap in the peak  
 486 release temperature (e.g., they have overlapping elution times, but they are not coeluting) suggesting that  
 487 a portion of the CO could have been released from incomplete combustion of organic carbon. The co-  
 488 occurrence of CO and CO<sub>2</sub> was not observed at high (~700° C) temperatures (Figure 4) because the  
 489 thermal decomposition of Ca carbonate does not produce CO. Thus, evolved CO detections near 400° C  
 490 were consistent with the decomposition of organic compounds.

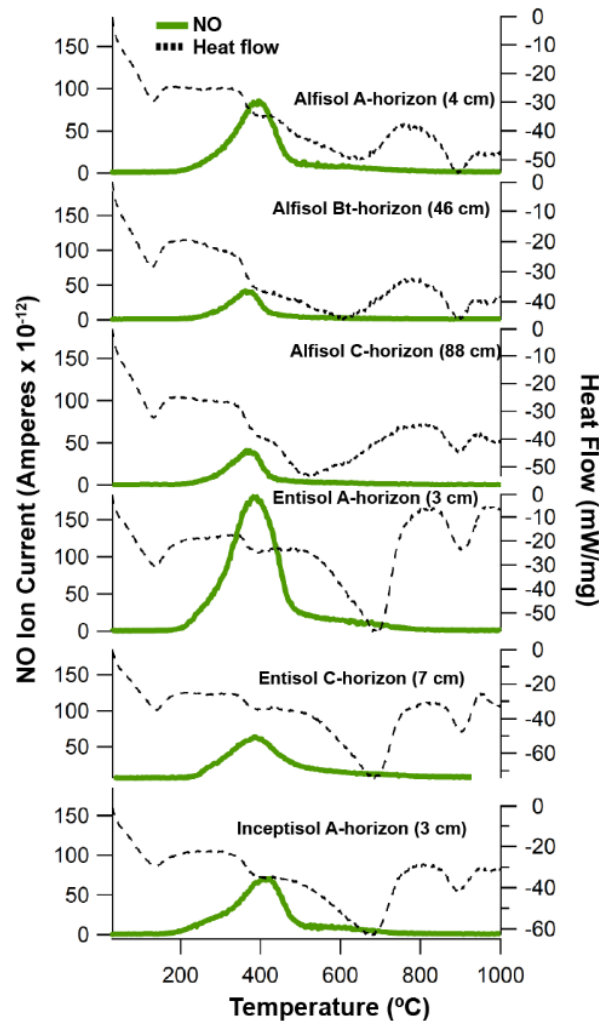


491  
 492

493 **Figure 4. CO<sub>2</sub> (red trace) and CO (*m/z* 28) (grey trace) evolutions from the surface horizons of paleosols at the**  
 494 **John Day Fossil Beds National Monument, Oregon.** Top panel is surface (4 cm) horizon of an Alfisol; bottom  
 495 panel is surface (3 cm) horizon of an Inceptisol.  
 496

497 **3.2.2 NO evolutions**

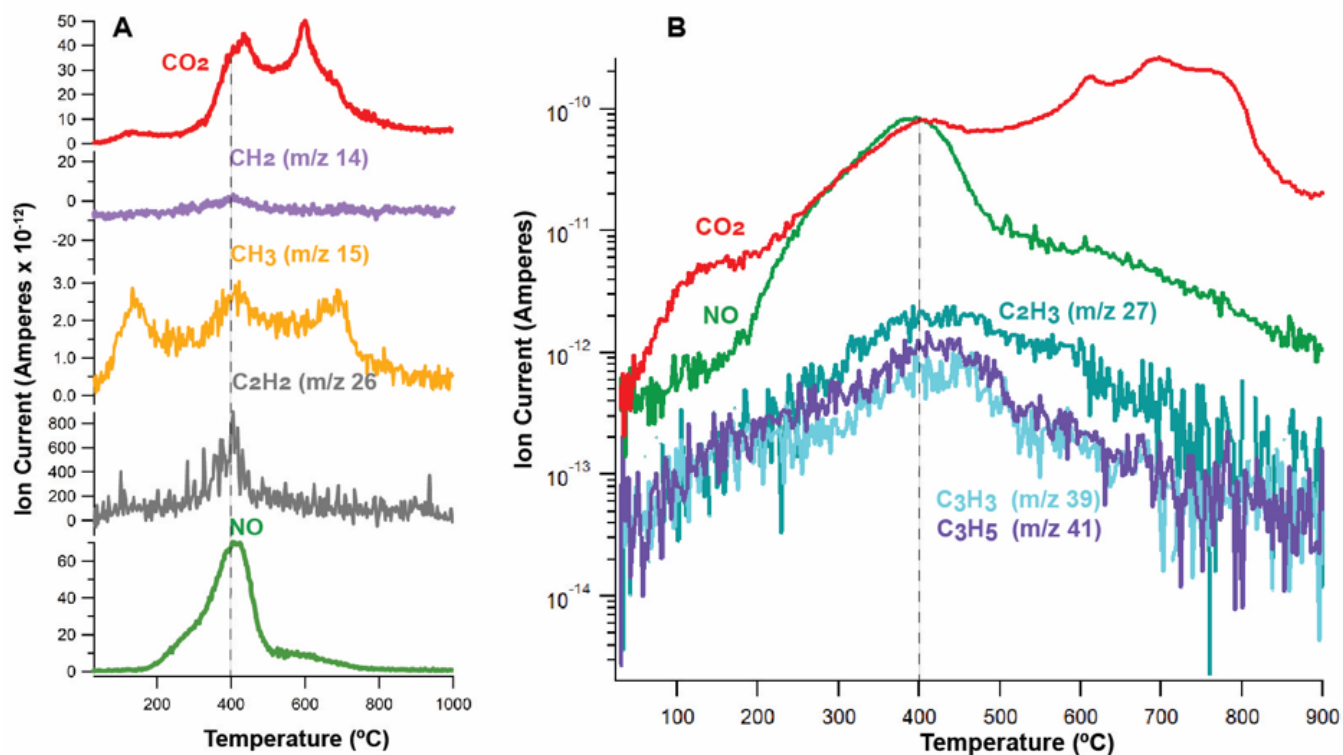
498 Nitric oxide (NO; *m/z* 30) was detected in all paleosol samples (Figure 5) and was consistent with  
 499 the oxidation of nitrogen-bearing organics. In the sample with the greatest amount of organic carbon  
 500 (Alfisol A-horizon), evolutions of NO start at ~300° C and show peak release temperature of ~ 400° C as  
 501 well as a small shoulder at ~ 600° C (Figure 5). Additionally, organic fragments observed in this sample  
 502 included CH<sub>2</sub>(*m/z* 14), CH<sub>3</sub> (*m/z* 15), and C<sub>2</sub>H<sub>2</sub> (*m/z* 26) (Figure 6). Evolutions of NO were simultaneous  
 503 with evolutions of these organic fragments at ~400° C, suggesting the oxidation of a nitrogen-bearing  
 504 organic compound was a significant contributor to evolved NO (Navarro-González et al., 2009).  
 505



506 **Figure 5. Evolutions of nitric oxide (NO) from paleosols at the John Day Fossil Beds National Monument,**  
 507 **Oregon.** Green trace is NO (*m/z* 30) and dashed trace is heat flow from differential scanning calorimetry (DSC).  
 508

509

510 The thermal decomposition of nitrates can also release NO, though at elevated temperatures (>  
 511 500° C) relative to nitrogenated organics (Stern et al., 2015), and therefore the NO release temperate can  
 512 constrain the origin of NO. For example Alkali (Na, K) and alkaline earth (Mg, Ca) metal nitrates  
 513 decompose to NO at temperatures > 560° C (Stern et al., 2015). In laboratory experiments under SAM-  
 514 like conditions,  $\text{Fe}(\text{NO}_3)_3$  begins to decompose to NO at < 200° C and exhibits two distinct releases of  
 515 NO at ~300 and ~450° C, which has been attributed to dehydration and hydrolysis of  $\text{Fe}(\text{NO}_3)_3$  hydrates,  
 516 respectively (Stern et al., 2015). Instead, NO release in paleosol samples begins at ~250° C and exhibited  
 517 a single peak at ~400° C across all samples (Figure 5), unlike the dual high-temperature NO peaks from  
 518 decomposition of  $\text{Fe}(\text{NO}_3)_3$  hydrates. The simultaneous evolution of NO,  $\text{CO}_2$ , and organic fragments in  
 519 paleosol samples analyzed here (Figure 6) are an additional line of evidence suggesting NO releases could  
 520 have resulted from oxidation and/or decarboxylation of nitrogen-bearing organic compounds. We also  
 521 observed evolutions of small nitrogen-bearing organics (HCN,  $m/z$  27, 26) that co-occurred with  
 522 evolutions of NO ( $m/z$  30) in several samples, including the surface horizon of the Alfisol (Figure S5).



523 **Figure 6. Simultaneous evolutions of  $\text{CO}_2$ , NO ( $m/z$  30) and organic fragments in paleosol surface horizons.**  
 524 **(A)** Evolutions of  $\text{CO}_2$ , NO ( $m/z$  30), and organic fragments ( $m/z$  14, 15, 26) from the surface horizon of the  
 525 Inceptisol (3 cm). **(B)** Semi-log scale plot of  $\text{CO}_2$ , NO and additional organic fragments ( $m/z$  27, 39, 41) from the  
 526 surface horizon (4 cm) of the Alfisol  
 527

528

529

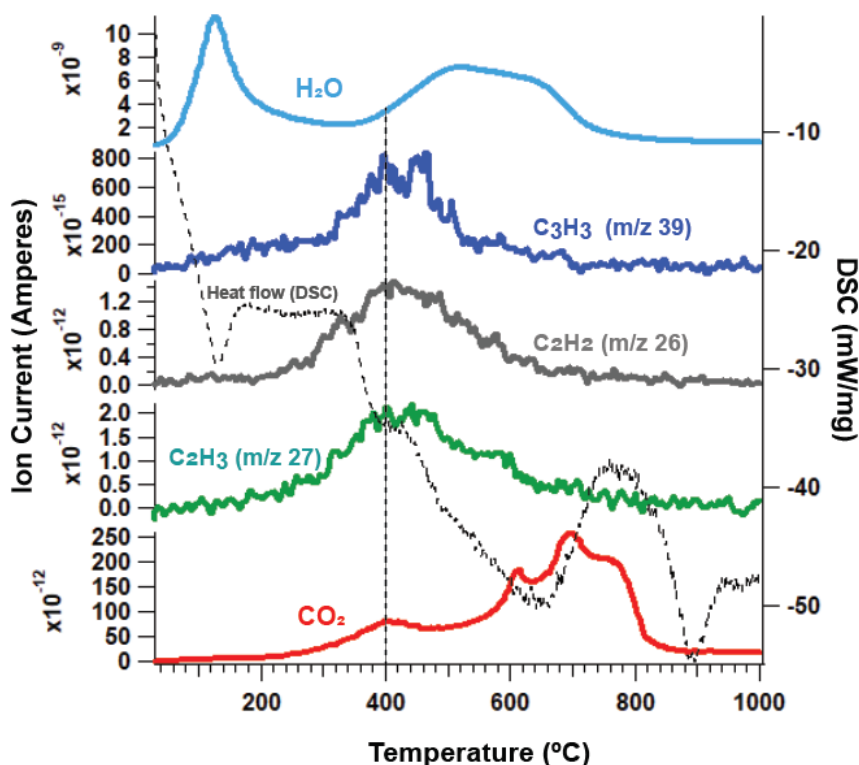
530 One possible source of nitrogen-bearing organic carbon compounds is pyrogenic carbon (char) in  
531 paleosol samples that formed as a result of wildfires across the original soil landscape (Kurth et al., 2006).  
532 Nitrogen-bearing organic compounds such as nitriles, pyridine and pyrrole-derive compounds have been  
533 observed with pyrolysis GC/MS in modern fire-affected soils (De la Rosa et al., 2008). These modern  
534 soils were also observed to have a broad  $\sim 400^\circ$  C peak release temperature of NO ( $m/z$  30) during TG-  
535 DSC-EGA, suggesting that pyrogenic carbon may be the source of low-temperature NO release in soils.  
536 However, the fire history of paleosols in the present study has not yet been investigated. Although it is  
537 impossible to determine the original inputs of nitrogen-bearing organics in paleosols, nitrogen may have  
538 been incorporated into increasingly stable organic matter as a consequence of forest fires, and may persist  
539 in paleosols today as char. Previous authors have reported the occurrence of preserved char in paleosols  
540 of late Permian (Miller et al., 1996) and late Jurassic age (Matthewman et al., 2012), so it is plausible that  
541 nitrogen-bearing char may be responsible for the overlapping  $\sim 400^\circ$  C NO and CO<sub>2</sub> peaks observed here.  
542

### 543 **3.3 Possible mechanisms of organic carbon preservation in paleosols**

544 There are many competing factors that control the preservation and degradation of organic carbon  
545 in terrestrial environments. Minerals themselves may facilitate oxidation of organic carbon compounds  
546 (Kleber et al., 2021), as is the case with certain Mn-bearing minerals (Reardon et al., 2016). Paleosols  
547 examined here contained between  $\sim 70$ -90 wt. % smectite, primarily as mixtures of montmorillonite and  
548 nontronite (Tables S3 and S5). Possible mechanisms of organic carbon preservation include the formation  
549 of organo-mineral complexes, and/or the formation of microaggregates that may have increased the  
550 biochemical stability of organic molecules, as observed in modern soils (Plante et al., 2011) and  
551 Quaternary paleosols (Marin-Spiotta et al., 2014). Smectite clay minerals in particular preserve organic  
552 molecules due to their high specific surface area, negatively charged interlayers, and cations that inhibit  
553 water flow, thus making them favorable locations for the preservation of organic carbon over millions or  
554 possibly billions of years (Bishop et al., 2013; Noe Dobrea et al., 2016; Szopa et al., 2020).

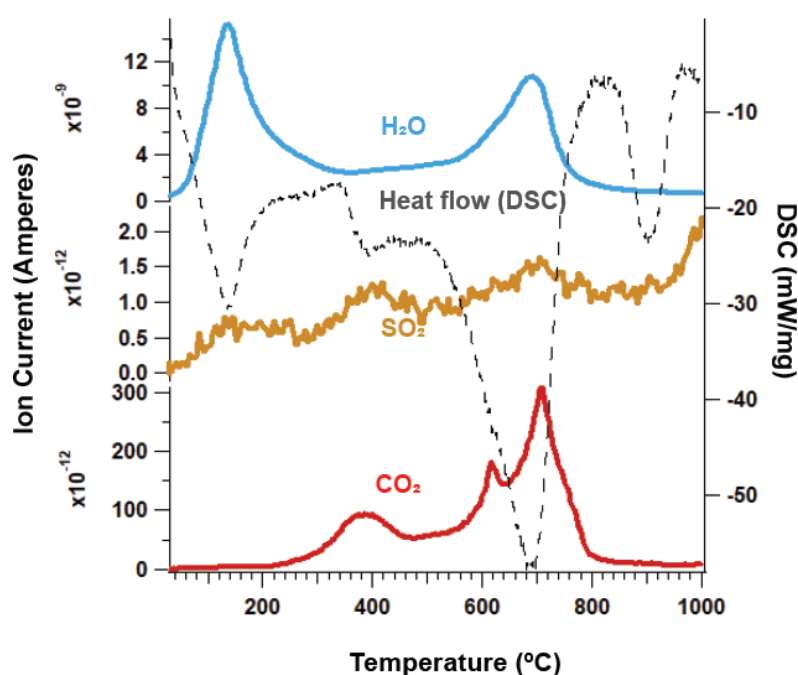
555 We observed evolutions of volatile gases from clay-rich paleosols that may have resulted from  
556 the preservation of organic carbon over geological time scales, as well as possible contributions from  
557 small amounts of modern organic carbon detected with radiocarbon analysis. The fragmentation of  
558 organic molecules during pyrolysis EGA provides limited constraints on the types of molecules present,  
559 but it can help determine if there are any associations between minerals and organic molecules. These  
560 associations can include physical occlusion, chemisorption and/or adsorption to mineral surfaces, or  
561 intercalation in clay minerals (Kleber et al., 2005, 2021; Schmidt et al., 2011; François et al., 2015). A  
562 strong correlation between the peak release temperature of organic fragments and the release of H<sub>2</sub>O or  
563 other volatiles would suggest that organic matter could have been associated with minerals.

564 In all samples, there were no apparent co-occurring evolutions of organic fragments, CO<sub>2</sub>, and  
 565 water releases from clay dehydroxylation (Figure 7). The Al-smectite in the surface horizon of the Alfisol  
 566 began dehydroxylation near 400° C. Figure 7 shows the peak release temperature of organics and CO<sub>2</sub>  
 567 was ~100° C lower relative to peak H<sub>2</sub>O release from Al-smectite dehydroxylation. This sample had the  
 568 highest TOC (~0.03 wt. %), FM of ~0.46 (Table 1), and a phyllosilicate content of ~85 wt. % (Table S3).  
 569 Interestingly, samples from other depths had higher temperatures of clay dehydroxylation due to  
 570 differences in clay mineralogy and/or abundance, such as the Entisol and Alfisol (Figures S3 and S4), but  
 571 these samples were depleted in organic carbon relative to the Al smectite-rich surface horizon of the  
 572 Alfisol (Table 1). Though our EGA results do not provide evidence that phyllosilicates and organics were  
 573 strongly associated with one another, it is possible that the high clay mineral abundances provided other  
 574 means of organic preservation, including physical occlusion, the formation of organic-mineral aggregates,  
 575 or other weaker types of sorption to phyllosilicate surfaces, such as outer-sphere complexation  
 576 (Guggenberger and Kaiser, 2003; Schmidt et al., 2011).  
 577



578  
 579 **Figure 7. Evolutions of CO<sub>2</sub>, organic fragments, and H<sub>2</sub>O from the surface horizon of the Alfisol (4 cm).** Co-  
 580 occurring releases of CO<sub>2</sub> and organic fragments at the onset of Al-smectite dehydroxylation (H<sub>2</sub>O release at ~400°  
 581 C, vertical dotted line) suggest organic molecules may have been associated with clay minerals. Dashed trace is heat  
 582 flow from differential scanning calorimetry (DSC).  
 583

584 Interactions with sulfur can also aid in the preservation of organic molecules in soils and  
 585 sediments over geological time scales (Matthewman et al., 2012; François et al., 2015; Eigenbrode et al.,  
 586 2018; Alekseeva et al., 2019). The incorporation of organic C into the crystal lattice of sulfate minerals  
 587 can increase thermodynamic stability and therefore increase the temperature of organic carbon  
 588 decomposition during EGA (François et al., 2015). Minor detections of jarosite and gypsum in XRD  
 589 patterns (< 5 wt. %) were observed, and these minerals could have also contributed to the preservation of  
 590 organics (Tables S5 and S6). Minor SO<sub>2</sub> evolutions at ~800° C across all samples (Figure S3) were  
 591 consistent with the decomposition of sulfate minerals. A single sample (Entisol 7 cm) had a low-  
 592 temperature SO<sub>2</sub> peak that co-occurred with the CO<sub>2</sub> release at ~400° C (Figure 8).



593  
 594 **Figure 8. Evolutions of H<sub>2</sub>O (blue trace), SO<sub>2</sub> (yellow trace), CO<sub>2</sub> (red trace), and heat flow (dashed trace)**  
 595 **from the surface horizon of the Entisol (3 cm). DSC – differential scanning calorimetry (heat flow), H<sub>2</sub>O – *m/z***  
 596 **18, SO<sub>2</sub> – *m/z* 64, CO<sub>2</sub> – *m/z* 44.**  
 597

598 Trace amounts of Mg sulfates in the Alfisol could account for minor SO<sub>2</sub> releases > 700° C  
 599 including the ~790° C SO<sub>2</sub> peaks (François et al., 2015; Mcdam et al., 2020a). At higher temperatures, all  
 600 soils showed a major release of SO<sub>2</sub> beginning at 900° C that co-occurred with an endotherm, both of  
 601 which are consistent with the thermal decomposition of crystalline sulfates (Ming et al., 2014; François et  
 602 al., 2015). Since the samples were only heated to ~1000° C for this work, the maximum peak height of  
 603 this release cannot be ascertained. Peak SO<sub>2</sub> release temperatures generally did not co-occur with low-  
 604 temperature (~400° C) CO<sub>2</sub> evolutions across the rest of the samples (Figure S3), so it is unlikely that  
 605 sulfate minerals played a significant role in organic preservation in samples examined here. However, it

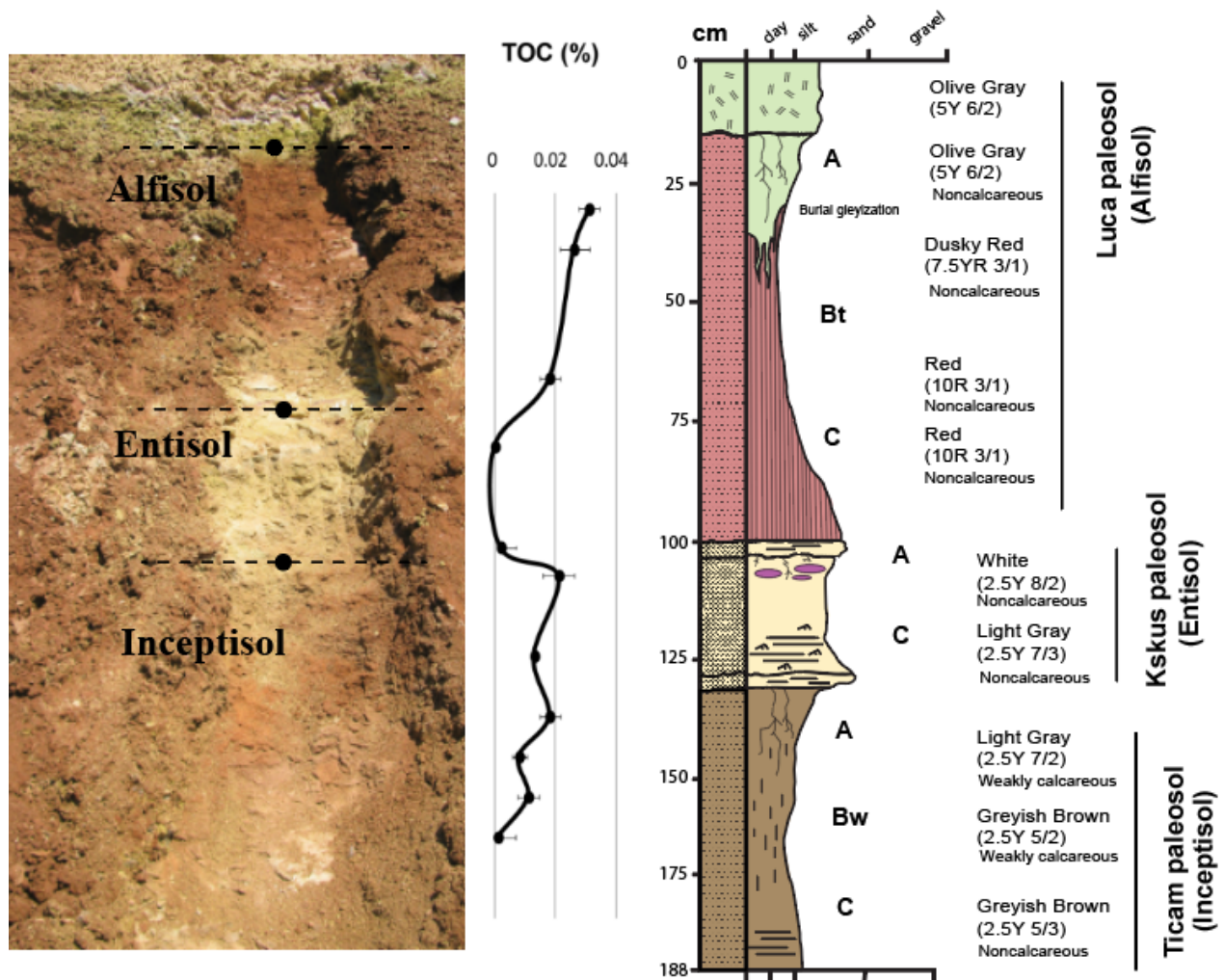


606 has been hypothesized that sulfate minerals were most likely diagenetic and not original to the paleosols,  
 607 but instead were inherited from the modern weathering zone (Broz et al., 2022). Thus, if modern sulfate  
 608 minerals were associated with small amounts of modern organic carbon, the sulfate minerals could have  
 609 been a source of radiocarbon in samples with co-occurring SO<sub>2</sub> and CO<sub>2</sub> evolutions.

610  
 611

612 **3.4 Enrichment of organic carbon in surface layers of paleosols**

613 The near-surface horizons of all paleosols examined here had greater amounts of total organic  
 614 carbon (TOC) relative to deeper horizons (Figure 9, Table 1). The highest amounts of CO<sub>2</sub> released from  
 615 decomposition of organic carbon (150-500° C) were in the near surface (A) horizons of all three paleosol  
 616 types while subsurface layers of paleosols generally had lower quantities of evolved CO<sub>2</sub> from organic  
 617 carbon (Figure 3, Table 1). The A horizon of the Alfisol had TOC of 0.031 ± 0.006 wt. % and  
 618 progressively decreased to 0.002 ± 0.007 wt. % in the C horizon.



619



620 **Figure 9. Trends of organic carbon enrichment in the near-surface horizons of three paleosols** from the early  
621 Oligocene (33 Ma) middle Big Basin Member of the John Day Formation in eastern Oregon, USA. The surface (A)  
622 horizons of all three profiles had significantly ( $P > 0.05$ ) higher total organic carbon (TOC) content relative to  
623 subsurface horizons (Bt, Bw and C horizons, respectively). Average TOC content of samples ( $n=2$ ) was determined  
624 by thermal and evolved gas analysis (oxygen as a carrier gas).  
625

626 Samples from the A-horizon of the Alfisol had been affected by burial gleization (Broz et al.,  
627 2022). Burial gleization is an early diagenetic process in paleosols that involves the reduction of  $\text{Fe}^{3+}$  to  
628  $\text{Fe}^{2+}$  by anaerobic microbes as a result of rapid burial (PiPujol and Buurman, 1994; Retallack, 2019). This  
629 is also thought to promote anaerobic decay of organic matter (Retallack, 2019), even in soils that  
630 originally formed under oxidizing conditions before burial. Typical burial gleization is closed system  
631 alteration, without depletion of total iron, and is usually limited to the surface horizons where organic  
632 matter is most concentrated. The surface horizon of the Alfisol examined in this work showed classic  
633 evidence of burial gleization with drab-colored mottles and tubular features predominantly in the A-  
634 horizon with minor radiation downward into the subsurface (Bt) horizon (Figure 9). Rapid emplacement  
635 of a biotite-bearing tuff on the paleosurface of the Alfisol may have led to gleization. Similar trends of  
636 near-surface TOC enrichment were noted in the Entisol and Alfisol (Table 1), but these samples lacked  
637 chemical ( $\text{FeO}/\text{Fe}_2\text{O}_3 > 0$ ) and morphological evidence (drab green color) of burial gleization (Tables S5  
638 and S6). Previous work on terrestrial paleosols has shown positive and significant correlations between  
639  $\text{Fe}^{2+}$  and TOC (Broz, 2020). This supports the interpretation that burial gleization, which increases  $\text{Fe}^{2+}$  in  
640 bulk samples, may have been associated with the preservation of organic carbon in the surface horizon of  
641 the Alfisol.

642 Organic carbon from organisms living in surface horizons of soils may have been preserved upon  
643 rapid burial of the paleosurface, and therefore the trend of surface enrichment may represent the  
644 preservation of endogenous organic carbon. Major losses of endogenous organic carbon from burial  
645 decomposition of organic matter are common in paleosols that originally formed under oxidizing  
646 conditions such as those examined here (Retallack, 2019). Despite diagenetic additions and losses of  
647 organic carbon, the trend of surface enrichment of organic carbon remains apparent, even in soils that  
648 formed under strongly oxidizing conditions prior to burial. These results are consistent with other studies  
649 of paleosols where the surface enrichment of organic carbon was attributed to preservation of carbon from  
650 organisms living above and within the soil profile (Rye and Holland, 2000; Watanabe et al., 2004;  
651 Liivamägi et al., 2018; Liu et al., 2020). However, it is possible that exogenous addition of organic carbon  
652 caused this enrichment, for example, by preferential groundwater flow and accumulation in the  
653 paleosurface of each profile. Results from this work cannot definitively rule out groundwater as a source  
654 of modern organic carbon, though it is unlikely that groundwater additions of organic carbon would have  
655 preferentially accumulated in the discreet, thin (10-20 cm) surface layers of the three buried soils.

### 656 **3.5 Implications for Mars**

657           Recent work has considered paleosols on Mars as potential high priority environments for *in-situ*  
658 biosignature investigation (Hays et al., 2017) and Mars Sample Return (Bishop et al., 2018a). This is  
659 because the composition and properties of paleosols preserve evidence of paleoclimate, aqueous  
660 conditions, and life (Bishop et al., 2018a). A major finding of this study is that near-surface horizons of  
661 terrestrial paleosols appear to be a favorable location for SAM-EGA detection of organic carbon. Like  
662 modern soils, the surface layers of Mars-analog paleosols examined here show evidence of surface  
663 enrichment of organic carbon. Although the early diagenetic process of burial decomposition of organic  
664 matter has likely reduced the organic carbon content of these ancient soils by at least two orders of  
665 magnitude relative to modern soils (Broz, 2020), the enrichment of organic carbon in surface horizons  
666 (~0.03 wt. %) and subsequent depletion in deeper layers (<0.01 wt. %) was readily observable with  
667 SAM-EGA analog instrument.

668           It should be noted that the Eocene-Oligocene (42-28 Ma) eastern Oregon paleosols are an  
669 incomplete Mars analog due to several fundamental differences. Most importantly, the source of organic  
670 carbon within the paleosols is primarily from microbial and/or plant biomass and represents a complex  
671 consortium of life above and within the soil profile. These irrevocable differences preclude direct  
672 comparisons between Earth and a presumably lifeless Mars. As such, the organic carbon component of  
673 these paleosols and the resulting preservation and degradation mechanisms may not be the same on Mars,  
674 but it is possible that putative paleosols at Mawrth Vallis have mineralogy analogous to terrestrial  
675 paleosols and thus have a similar mineralogical control on the fate of organic carbon. Large differences in  
676 age (Oligocene [33 Ma] versus Noachian [4.1-3.7 Ga]) also have implications for diagenesis. The Oregon  
677 paleosols have experienced a range of minor to moderate diagenetic alteration including illitization of  
678 smectite, zeolitization and celadonization all resulting from alteration to clinoptilolite facies (Retallack et  
679 al., 2000; Horgan, 2016), but it is currently unclear if similar diagenetic alteration has affected potential  
680 paleosol sequences on Mars. Also worth mentioning are differences in the oxidation state of the  
681 atmosphere during subaerial weathering. The Oregon paleosols formed under a thoroughly oxidizing  
682 atmosphere but at present the oxidation state of an early Mars atmosphere is not well constrained  
683 (Ramirez et al., 2014). Leaching of  $\text{Fe}^{2+}$  in putative martian paleosols at Mawrth Vallis has been inferred  
684 from spectral weathering indices, which are consistent with an anoxic, reducing atmosphere during the  
685 early Noachian (Liu et al., 2021a). Lastly, there are most likely differences in the pH of fluids  
686 participating in hydrolytic weathering of volcanoclastic sediments. The Oregon paleosols formed by  
687 pedogenic weathering with circumneutral-pH fluids, but subaerial alteration on Mars could have preceded  
688 with acidic,  $\text{H}_2\text{SO}_4$  and HCl-rich surface waters as a result of volcanic outgassing of  $\text{H}_2$  and  $\text{SO}_2$  (Liu et  
689 al., 2021a, 2021b). Despite these significant differences, the paleosol sequence from eastern Oregon

690 offers a natural example to determine if organic carbon can be detected with thermal and evolved gas  
691 analysis techniques similar to SAM-EGA.

692 Results from this study have implications for interpreting the chemical and isotopic biosignature  
693 preservation potential of Al and Fe smectite-bearing weathering profiles on Mars. Observations of  
694 refractory organic compounds that persist in terrestrial paleosols with Mars-like mineralogy provide a  
695 reference frame for interpreting future observations of putative weathering profiles on Mars. It is possible  
696 that clay minerals and/or amorphous phases in martian weathering profiles have imparted a similar  
697 control on the fate of organic carbon. However, putative martian weathering profiles have likely been  
698 subject to both galactic cosmic ray bombardment and the accumulation of oxychlorine salts that, together,  
699 have likely contributed to the degradation of organic matter in the top 1-2 m of the martian surface  
700 (Dartnell et al., 2007; Freissinet et al., 2015). This poses major challenges for the future detection of  
701 organic carbon in martian weathering profiles using current drilling techniques. Trace amounts of organic  
702 carbon may have been preserved in the Oregon paleosols (Table 2), but if putative martian paleosols have  
703 a Cretaceous exposure age comparable with Gale crater sedimentary rocks (Farley et al., 2013), millions  
704 of years of cosmic ray bombardment and interactions with oxychlorine salts may have led to the oxidation  
705 of a significant fraction of the organic carbon pool.

706 This work also provides an initial framework for investigation and sampling of martian  
707 weathering profiles should they be encountered by current or future landed missions. If possible, future  
708 *in-situ* analysis of putative weathering profiles should begin at the apparent ancient surface horizon (e.g.,  
709 the buried topsoil layer) and sample down section into the unaltered protolith, similar to the sampling of a  
710 terrestrial soil profile. If the entire profile is not accessible for investigation (e.g., outcrop is at a  
711 topographic position inaccessible to the rover), the near-surface horizons of the profile, just below the  
712 overlying burial layer, should be considered the highest priority target for remote sensing, contact science,  
713 and collection of a drilled sample for sample return to Earth. Two locations on Mars where current and  
714 future landed missions could encounter a paleosol profile are discussed below.

715

### 716 **Mawrth Vallis**

717 Previous work identified compositional and morphological similarities between the eastern  
718 Oregon paleosol sequence and layered outcrops at Mawrth Vallis (Horgan et al., 2012; Hays et al., 2017;  
719 Lantz et al., 2020; Smith and Horgan, 2021). The compositional stratigraphy at Mawrth Vallis has been  
720 interpreted as a paleosol sequence (Horgan et al., 2012) or a deep weathering profile (Liu et al., 2021a).  
721 Here, a ~200 m stack of layered sedimentary rocks has spectral signatures consistent with pedogenic-like  
722 alteration of mafic sediments (Horgan, 2013). The basal Fe/Mg smectite unit at Mawrth Vallis is  
723 dominated by spectral signatures of Fe/Mg smectite and is consistent with subaerial alteration under a

724 warm and wet climate (Loizeau et al., 2015; Poulet et al., 2020). This basal unit transitions upward into a  
725 thin, ferrous-bearing clay unit that could have resulted from reducing conditions during subaerial  
726 weathering. This unit is overlain by layered rocks containing minerals consistent with formation in acidic  
727 and evaporitic geochemical settings, inferred from spectral signatures consistent with mixtures of smectite  
728 and jarosite (the “doublet” spectral feature) (Noe Dobrea et al., 2016; Danielson et al., 2019; Bishop et al.,  
729 2020) . Stratigraphically higher, layered rocks with signatures of Al and Fe smectite are consistent with  
730 pedogenic weathering of volcanoclastic sediments under a semi-arid climate. Finally, the uppermost layers  
731 are characterized by accumulations of poorly crystalline aluminosilicates, suggesting minimal or cool-  
732 climate alteration, which may represent the terminus of warm and wet conditions on early Mars (Bishop  
733 et al., 2020). The entire sequence is then capped by dark-toned sands and/or lava flows. One hypothesis  
734 to explain the stratigraphy at Mawrth Vallis is that it may represent the cooling and drying of the Martian  
735 climate during the mid to late Noachian (Bishop and Rampe, 2016; Bishop et al., 2020). Similarly, the  
736 Oregon paleosol sequence represents a transition from warm and wet conditions. The intensely weathered  
737 basal Clarno Formation paleosols transition upward into less-weathered semi-arid paleosols of the middle  
738 Big Basin Member, which contains both oxidized (red, brown) and reduced (yellow, black) paleosols  
739 (Figure 1). These are subsequently overlain by minimally weathered paleosols of the Oligocene Turtle  
740 Cove Member with accumulations of unweathered volcanic glass, amorphous colloids, calcite and  
741 hydrated silica, which are thought to represent the terminus of warm and wet climate conditions in eastern  
742 Oregon. Terrestrial paleosols at the analog site have therefore been considered analogous to dioctahedral  
743 clay sequences on Mars. The layered stratigraphy at Mawrth Vallis currently stand as the best example of  
744 a putative paleosol sequence on Mars, but other locations, such as Jezero Crater, also show VNIR spectra  
745 consistent with subaerial alteration of mafic sediments.

746

#### 747 **Jezero Crater**

748 Spectral signatures of Al-bearing clay minerals and/or silica deposits that could have formed in  
749 subaerial environments were detected approximately 3 km from the Perseverance rover landing  
750 site at Jezero Crater. Across Jezero’s western delta and northern fans, there are strong and  
751 ubiquitous orbital detections of Al-bearing clay minerals and/or silica that could be either detrital  
752 or authigenic in origin (Horgan et al., 2020). The strongest signatures across the western delta are  
753 associated with features that resemble point bar deposits and are consistent with formation in  
754 subaerial and/or seasonally waterlogged paleoenvironments (Horgan et al., 2020). If authigenic  
755 in origin, these deposits could be subaerial paleoenvironments and include individual paleosol  
756 profiles. If targeted for *in-situ* examination by *Perseverance* rover, these deposits could provide

757 critical information about the climate and duration of delta activity at Jezero Crater (Horgan et  
758 al., 2020).

## 759 **Conclusions**

760 The objectives of this study were a) to determine whether the organic carbon content of ~33-  
761 million-year-old paleosols can be detected with a thermal and evolved gas analyzer configured to operate  
762 like the SAM-EGA instrument onboard *Curiosity* Mars rover, and b) use radiocarbon ( $^{14}\text{C}$ ) dating to  
763 constrain the age of organic carbon in bulk paleosol samples. Radiocarbon dating of organic carbon in  
764 four paleosol samples revealed the presence of recent and/or modern exogenous organic carbon. Samples  
765 from 10 - 20 cm were dated to ~6,200 years BP and had a fraction modern (FM) value of ~0.4, while a  
766 single deeper sample collected from 43 cm into the outcrop had a radiocarbon age of ~14,600 years BP  
767 and ~0.16 fM. The presence of radiocarbon in paleosols could have resulted from the addition of small  
768 amounts of modern organic carbon that mixed with  $^{14}\text{C}$ -free endogenous organic carbon. Alternatively, a  
769 diagenetic event between 6-14 Ka could have introduced exogenous organics, possibly through  
770 groundwater alteration and/or precipitation-driven leaching of dissolved organic carbon. Application of a  
771 two-endmember mixing model (Equation 1) to the  $\text{D}^{14}\text{C}$  values of bulk samples values provided  
772 supporting evidence for the hypothesis that paleosols from the site contain small amounts of modern  
773 organic carbon. The amount of modern carbon ( $\text{D}^{14}\text{C} = 0\text{‰}$ ) that could have been added to a pool of  
774 Oligocene-age carbon ( $\text{D}^{14}\text{C} = -999\text{‰}$ ) was determined to range from ~0.5 – 3% of the total carbon in  
775 each sample. These results highlight a new approach for constraining the age and sources of organic  
776 matter in terrestrial paleosols.

777 SAM-EGA-like characterization of paleosols showed evolutions of  $\text{CO}$ ,  $\text{NO}$ ,  $\text{CO}_2$ , and organic  
778 fragments. Coevolutions of  $\text{CO}_2$  and organic fragments at ~400° C suggested the presence of refractory  
779 organic carbon. However, like other oxidized terrestrial paleosols of Cenozoic age and older, only trace  
780 amounts (<0.1 wt. %) of organic carbon was detected, which most likely was a result of diagenetic  
781 decomposition of organic matter over geological time scales. Many samples examined in this work  
782 typically contained very low amounts (~0.01 wt. %) of organic carbon, but these low values were  
783 nevertheless detectable by SAM-like evolved gas analysis of bulk samples. These results suggest the  
784 organic fraction of potential martian weathering profiles may be detectable with evolved gas analysis,  
785 even if organic concentrations are low.

786 A major result of this work was that organic carbon was concentrated in near-surface horizons of  
787 paleosols while deeper horizons were depleted in organic carbon. Like modern soils, these ancient soils  
788 were enriched in organic carbon in near-surface horizons, and it appears that this trend persisted in these  
789 samples despite burial decomposition of organic carbon over geological time scales. Alternatively,

790 modern carbon could have leached into the outcrop, but it is unlikely such organic contamination would  
791 preferentially accumulate in the surface layers of each successive buried soil profile. Enrichment of  
792 organic carbon in the original surface horizons of buried weathering profiles on Mars may therefore  
793 constitute a putative chemical biosignature. This work demonstrates that analytical techniques similar to  
794 SAM-EGA can detect trace amounts of organic carbon in complex pedogenic mineral matrices. The  
795 search for past life on ancient land surfaces of Mars should include targeting Martian weathering profiles  
796 for *in-situ* biosignature investigation and Mars Sample Return.

797

## 798 **Additional Information**

799

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812

## 813 **Author Contribution Statement**

814 A.P.B and J.V.C designed the study, performed all laboratory analyses and drafted the manuscript. J.V.C,  
815 D.W.M, P.D.A, and B.S contributed to data analysis and interpretation. B.H.H identified similarities  
816 between Mars and Oregon paleosols and assisted with fieldwork and data interpretation. L.C.R.S  
817 provided radiocarbon analyses and interpreted the data. J.V.C, P.D.A, B.S, D.W.M, and L.C.R.S  
818 supervised the project. A.P.B drafted all figures. All authors contributed to the manuscript.

819

## 820 **Author Disclosure Statement**

821 No competing financial interests exist.

## 822 **Data Availability Statement**

823 All data supporting the conclusions can be found within the article and in the following repository: Broz,  
824 2022. All raw data to reproduce EGA traces are included in the supplementary dataset (Broz, 2022).

825  
826

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