

1 Explosive activity on Kīlauea's Lower East Rift Zone fuelled by a volatile-rich, dacitic
2 melt

3 Penny. E. Wieser*^{1,2}, Marie Edmonds¹, Cheryl Gansecki³, John Maclennan¹, Frances E. Jenner⁴,
4 Barbara Kunz⁴, Paula Antoshechkina⁵, Frank Trusdell⁶, R. L. Lee⁶, EIMF⁷.

5 ¹Department of Earth Sciences, University of Cambridge, UK

6 ²Now at College of Earth, Ocean and Atmospheric sciences, Oregon State University, USA

7 ³Department of Geology, University of Hawai'i at Hilo, Hilo, HI 96720, USA

8 ⁴School of Environment, Earth and Ecosystem Sciences, The Open University, MK7 6AA, UK

9 ⁵Division of Geological and Planetary Sciences, California Institute of Technology, Pasadena, CA
10 91125, USA

11 ⁶USGS Hawaiian Volcano Observatory, Hilo, HI 96720, USA

12 ⁷Edinburgh Ion Microprobe Facility, School of Geosciences, University of Edinburgh, EH9 3JW, UK

13 Abstract (250 words for G3)

14 Magmas with matrix glass compositions ranging from basalt to dacite erupted from a series of 24
15 fissures in the first two weeks of the 2018 Lower East Rift Zone (LERZ) eruption of Kīlauea Volcano.
16 Eruption styles ranged from low spattering and fountaining to strombolian activity. Major element
17 trajectories in matrix glasses and melt inclusions hosted by olivine, pyroxene and plagioclase are
18 consistent with variable amounts of fractional crystallization, with incompatible elements (e.g., Cl, F,
19 H₂O) becoming enriched by 4-5 times as melt MgO contents evolve from 6 to 0.5 wt%. The high
20 viscosity and high H₂O contents (~2 wt%) of the dacitic melts erupting at Fissure 17 account for the
21 explosive Strombolian behavior exhibited by this fissure, in contrast to the low fountaining and
22 spattering observed at fissures erupting basaltic to basaltic-andesite melts. Saturation pressures
23 calculated from melt inclusions CO₂-H₂O contents indicate that the magma reservoir(s) supplying
24 these fissures was located at ~2-3 km depth, which is in agreement with the depth of a dacitic
25 magma body intercepted during drilling in 2005 (~2.5 km) and a seismically-imaged low Vp/Vs
26 anomaly (~2 km depth). Nb/Y ratios in erupted products are similar to lavas erupted between 1955-
27 1960, indicating that melts were stored and underwent variable amounts of crystallization in the
28 LERZ for >60 years before being remobilized by a dike intrusion in 2018. We demonstrate that
29 extensive fractional crystallization generates viscous and volatile-rich magma with potential for
30 hazardous explosive eruptions, which may be lurking undetected at many ocean island volcanoes.

31 Plain Language Summary

32 During the first two weeks of the 2018 eruption of Kīlauea Volcano, Hawai'i, a wide variety of
33 magma compositions were erupted from 24 separate fissures. Of particular interest is the magma
34 erupted at Fissure 17, which was much more explosive than nearby fissures, throwing large pieces of
35 molten rock several hundred metres into the air, causing a serious injury. The explosivity of volcanic
36 eruptions is partially controlled by the viscosity of the magma (how easily it can flow), and partially
37 by the amount of dissolved H₂O and CO₂ (which exsolve like bubbles in a carbonated beverage as the
38 melt ascends to the surface). We show that melts erupted at Fissure 17 were significantly more
39 viscous and H₂O-rich than the melts erupted more passively at nearby fissures. These two factors
40 account for the explosive eruptive style of this fissure. Using the relationship between the amount of

41 dissolved H₂O and CO₂ and pressure, we show that erupted magmas were stored at ~2km depth
42 below the surface since at least 1955-1960 A.D.

43 Key points (140 characters)

- 44 1. The first 2 weeks of the 2018 Kīlauea eruption tapped melts undergoing variable amounts of
45 fractionation at ~2-3 km depth over >60 yrs.
- 46 2. Saturation pressures calculated from melt inclusion volatile contents align with the depth of
47 a seismically imaged Vp/Vs anomaly.
- 48 3. The strombolian explosions at Fissure 17 result from the high viscosity and the high H₂O
49 contents of these dacitic melts

50 1. Introduction

51 The main shield-building stage of volcanism at ocean island volcanoes fed by high melt fluxes from
52 mantle plume melting (e.g., Hawai'i, Galápagos, Réunion) is characterized by the eruption of basaltic
53 lava flows, spatter, and occasional energetic lava fountains (Macdonald, 1962; Swanson et al., 1979).
54 On a number of occasions at Kīlauea Volcano, HI, erupted basaltic lava has pooled within existing pit
55 craters, undergoing extensive fractional crystallization at near atmospheric pressures (e.g., 1959
56 Kīlauea Iki lava lake, 1963 Alae lava lake, 1965 Makaopuhi Crater; Helz, 1980; Wright et al., 1976). It
57 is becoming increasingly apparent that andesitic to rhyolitic melts analogous to those formed in
58 surface lava lakes may also co-exist with basaltic melts at depth within a range of volcanic plumbing
59 systems, even if these more silicic compositions are poorly represented in erupted rock
60 compositions (Stock et al., 2020). The first detailed description of such lavas from the shield stage of
61 a Hawaiian volcano was provided by Bauer et al., (1973), who investigated a rhyodacite from the late
62 shield stage from the Kaua'ōpu'u ridge of Oahu. In 2005, a hydrothermal injection well on Kīlauea's
63 Lower East Rift Zone (LERZ) intercepted a stored body of dacitic magma (with a bulk composition of
64 67 wt% SiO₂; Teplow et al., 2009). More recently, examination of lava and tephra samples from the
65 2015 eruption of Wolf Volcano and the 1968 eruption of Fernandina in the Galapagos identified
66 plagioclase crystals whose compositions indicate that they grew from basaltic trachy-andesite and
67 trachy-andesitic melts (Stock et al., 2020).

68 In contrast to these tantalizing glimpses of more silicic melts in basalt-dominated ocean island
69 volcanic settings, the first two weeks of the 2018 Lower East Rift Zone (LERZ) eruption of Kīlauea
70 Volcano produced significant volumes of lava spanning basaltic to andesitic bulk compositions from
71 24 different fissures (Fig. 1b, Gansecki et al., 2019). Importantly, while lava lakes act as excellent
72 natural laboratories to investigate major element evolution of Kīlauea magmas during differentiation,
73 because the pooling magmas have already degassed the majority of their volatiles, they do not
74 provide representative insights into volatile element evolution within the plumbing system (Wieser
75 et al., 2020). The eruption of melts with a wide range of bulk compositions at the surface, combined
76 with rapid-response sampling of quenched spatter and tephra before it was covered by later lava
77 flows (Gansecki et al., 2019), provides an unprecedented opportunity to investigate the pre-eruptive
78 processes producing SiO₂-rich magmas from ocean island basalts, such as the depth and longevity of
79 magma storage and differentiation, and the effect of melt composition on eruption style.

80 One fissure, named Fissure 17 (F17) produced highly explosive Strombolian style activity, launching
81 2-3 m pieces of spatter ~125 m in the air (see
82 <https://www.youtube.com/watch?v=WXcV6WqvDYs&t=108s>; Supporting Video 1). These explosive
83 outbursts caused a severe injury when spatter was thrown 150 m from the vent and shattered the
84 leg of a resident on a third floor balcony. F17 also exhibited gas venting and the generation of shock

85 waves, audible in the town of Mountain View >20 km to the NW. This explosive eruption style at F17
86 contrasted with the low-moderate fountaining (heights 30 to 80 m) and less energetic spattering
87 observed at nearby fissures, and even the eastern end of F17 (Fig. 1d-e). F17 is offset ~220 m to the
88 north-east of the linear trend defined by the other early eruptive fissures (Fig. 1b), and rapid-
89 response petrological monitoring during the eruption indicated that the western, more explosive
90 end of F17 was tapping a more SiO₂-rich composition than the eastern, less explosive end (Gansecki
91 et al., 2019). However, the cause of the higher explosivity remains uncertain; it has been suggested
92 that explosivity might result from the involvement of groundwater, or the higher viscosity of these
93 more SiO₂-rich melts (Gansecki et al., 2019; Soldati et al., 2021).

94 Here we utilize the major, trace element and volatile element contents of olivine, pyroxene and
95 plagioclase-hosted melt inclusions to investigate three questions:

- 96 1) What was the cause of the high explosivity observed in the eruptions at the western end of
97 F17?
- 98 2) What was the most probable parent magma(s) of the variable melt compositions erupted in
99 the first few weeks of the 2018 eruption?
- 100 3) At what depth were these magmas stored in the LERZ?

101 We show that fissures erupting in the first few weeks of the 2018 eruption tapped a stored magma
102 body which experienced variable amounts of fractional crystallization at 2-3 km depth. Based on a
103 comparison of incompatible trace element ratios from 2018 samples and historic activity in this
104 section of the LERZ, we show that the 2018 eruption tapped a magma body with very similar
105 chemistry to that supplying the 1955 and 1960 eruption. The large amounts of fractional
106 crystallization experienced by dacitic melts erupted at F17 drove up melt H₂O contents to
107 unprecedented levels at Kīlauea (~2 wt%) which, combined with an increase in melt viscosity,
108 accounts for the explosive activity observed at this fissure.

109 1.1 Chronology of the 2018 LERZ eruption

110 The 2018 LERZ eruption marked the end of a 35-year-long period at Kīlauea Volcano during which
111 activity was focused predominantly at the Pu'ū'ō'ō vent, on the Middle East Rift Zone (the name
112 change from Pu'ū'ō'ō signifies that the eruption has ended; Hawai'i Board on Geographic Names).
113 Following inflation of the summit and Pu'ū'ō'ō during March and April 2018 (Neal et al., 2018;
114 Patrick et al., 2020), Pu'ū'ō'ō vent collapsed on April 30th, accompanied by the propagation of a dike
115 downrift into the Leilani Estates subdivision on the Lower East Rift Zone (LERZ). Following the
116 formation of ground cracks, the first eruptive fissures opened on May 3rd. Fourteen additional
117 fissures opened along a linear trend in the first week (F1 to F15; Fig. 1b). These first fissures,
118 classified as Early Phase 1 by Gansecki et al., (2019), were characterized by whole-rock compositions
119 with ~4-5 wt% MgO, erupted as relatively small-volume spatter mounds and lava flows. These lavas
120 are thought to represent material which was intruded during previous eruptive episodes in the
121 Lower East Rift Zone (e.g., 1790, 1840, 1955 or 1960 A.D; Gansecki et al., 2019; Fig. 1c). The injection
122 of the dike beneath the Leilani Estates subdivision at the start of May, 2018, is thought to have
123 disturbed these resident melts and forced them to the surface (e.g., by increasing overpressure,
124 changing the crustal stress state, or thermal rejuvenation).

125 Following an eruptive pause, activity recommenced on May 12th-18th (termed Late Phase 1), with
126 more fissures opening up along the same linear trend (Fissures 16, 18, 20, 22). These later fissures
127 erupted higher magma compositions with higher bulk MgO contents (5-6 wt% vs. 4-5 wt% in Early
128 Phase 1), which have been interpreted to represent mixing between the early rift-stored component,
129 and the more MgO-rich magma supplied by the dike (Gansecki et al., 2019). Between the 17-27th of

130 May (Phase 2) effusion rates increased and whole-rock MgO contents increased again to become
131 very similar to those erupted at Pu'u'ō'ō vent (6-7 wt%), indicating that the vast majority of melts
132 stored in the LERZ had been flushed from the system (Gansecki et al., 2019; Neal et al., 2018; Patrick
133 et al., 2019). In addition to this general progression from relatively MgO-poor to MgO-rich melt
134 compositions, F17 (active 13-25th of May, offset to the NE of the other fissures) erupted magma with
135 significantly lower MgO and higher SiO₂ contents (including andesitic bulk compositions), which is
136 thought to represent a third, stored magmatic component (Gansecki et al., 2019).

137 By the 28th of May (Phase 3), activity had focused at a single fissure (F8, recently renamed
138 Ahu'ailā'au) which developed a fast-flowing lava channel, and erupted material with even higher
139 bulk MgO contents resulting from the incorporation of antecrystic olivine crystals (Fig. 2a; Gansecki
140 et al., 2019; Wieser et al., 2021b). For brevity, to distinguish between this main phase of activity at
141 Ahu'ailā'au (F8), and the earlier, more MgO-poor material, we refer to activity in early Phase 1, Late
142 Phase 1 and Phase 2 as "early fissures", and activity at Ahu'ailā'au after May 28th as Phase 3.

143 It has been suggested, based on the presence of high forsterite olivines incorporated within Phase 3
144 lavas, and the bimodal distribution of melt inclusion saturation pressures at 1-2 and 3-5 km, that this
145 magma originated from the two magma reservoirs identified by geophysical imaging beneath
146 Kīlauea's summit (Wieser et al., 2021, Lerner et al., 2021), with a hydraulic connection linking these
147 reservoirs to Ahu'ailā'au (Gansecki et al., 2019; Neal et al., 2018; Patrick et al., 2019). Conversely,
148 Pietruszka et al. (2021) suggest that major and trace element differences between Phase 3 lavas and
149 those erupted at Kīlauea's summit and Pu'u'ō'ō prior to the onset of the 2018 eruption rules out a
150 summit origin, and instead suggest that this MgO-rich magma had accumulated in the Middle East
151 Rift Zone (MERZ) downrift of Pu'u'ō'ō over ~10 years prior to 2018. The exact source of Phase 3
152 magma is beyond the scope of this study, which focuses on the early phase lavas erupted between
153 May 3rd to May 16th, so we refer to Phase 3 magma as "dike-supplied" to avoid ambiguity.

154 2. Methods

155 2.1 Analytical methods

156 Spatter, reticulite and lava samples were collected from 9 separate fissures (Fissures 4, 5, 8, 10, 11,
157 12, 13, 17, and 20) on Kīlauea's LERZ during the 2018 eruption and a follow-up field campaign in
158 2019 (Supporting Table S1). Care was taken to select the most rapidly-quenched water and air-
159 quenched material to minimise H⁺ diffusion out of melt inclusions (Gaetani et al., 2012). Specifically,
160 we selected the smallest and most visibly glassy clasts available in each location. Clast size was 0.5-8
161 cm and water-quenched material was favoured where available. Samples were jaw-crushed, and
162 olivine, orthopyroxene, clinopyroxene, and plagioclase crystals were picked under a binocular
163 microscope. Crystals were individually mounted on glass slides using Crystalbond and ground down
164 to the level of target melt inclusions. Care was taken to prioritize melt inclusions not visibly
165 connected to the outside of the crystal, but particularly in plagioclase-hosted melt inclusions, it was
166 often difficult to determine the degree of connectivity due to complex melt networks. An indication
167 of whether each melt inclusion was fully isolated or connected to the carrier melt is provided in
168 Supporting Dataset S1.

169 All vapour bubbles within melt inclusions were analysed for CO₂ using Raman spectroscopy following
170 the methods described by Wieser et al., (2021). Unlike the olivine-hosted melt inclusions erupted in
171 the Phase 3 between 28th May and Aug 1st, none of the vapour bubbles in melt inclusions from the
172 early fissures (3rd-16th May) investigated in this study produced measurable CO₂ peaks, indicating
173 that the concentration of CO₂ in their vapour bubbles were below the detection limit of Raman

174 Spectroscopy ($\sim 0.02 \text{ g/cm}^3$; Wieser et al., 2021). Following exposure of a target melt inclusion at the
175 surface, crystals were mounted in epoxy stubs in groups of 30-40 inclusions, and polished using
176 progressively finer diamond pastes.

177 Samples were analysed for H_2O , CO_2 , F (as well as MgO and SiO_2 for normalization) using the Cameca
178 IMS-7f GEO at the NERC Ion Microprobe Facility at the University of Edinburgh in two analytical
179 sessions (July 2019, January 2020). A variety of glass standards were analysed with a range of Si
180 contents to convert counts to element concentrations, and to account for matrix effects on ion
181 yields as a function of melt SiO_2 content. Detailed analytical information is available in the
182 Supporting information (Supporting Fig. S1-S2, Supporting Table S2).

183 Following SIMS analysis, the Au coat was removed by polishing on a $1 \mu\text{m}$ pad (which also helped to
184 reduce the size of SIMS pits and remove SIMS sputter) and a C coat was applied. Matrix glasses, melt
185 inclusions, and the composition of the mineral host $\sim 30\text{-}100 \mu\text{m}$ away from each melt inclusion were
186 analysed using a Cameca SX100 EPMA at the Department of Earth Sciences, University of Cambridge
187 following the analytical techniques described in Wieser et al., (2021) (also see Supporting Text 1,
188 Table S3-S9). Care was taken to analyse away from the SIMS pit where possible.

189 Trace elements were measured in melt inclusions and matrix glasses using a Photon Machines G2
190 193 nm excimer laser system equipped with a HelEx II 2-volume cell coupled to an Agilent 8800 ICP-
191 MS/MS) at the School of Environment, Earth and Ecosystem Sciences, The Open University.
192 Depending on the size of the melt inclusion and the number of microlites in matrix glasses, ablation
193 spots were between 20 and $50 \mu\text{m}$. Analyses were conducted following the techniques described by
194 Jenner et al., (2015) and Wieser et al., (2020). Comparison of measurements of BCR-2G run as
195 secondary standard to long-term laboratory preferred values is shown in Supporting Dataset S2.

196 2.2 Correction for Post-Entrapment Crystallization

197 Olivine-hosted melt inclusions were corrected for the effects of post-entrapment crystallization
198 (PEC) using the "Olivine MI" tool in Petrolog3.1.1.3 (Danyushevsky and Plechov, 2011), which
199 requires users to enter the measured major element composition of the melt inclusion, the host Fo
200 content (both taken from EPMA analyses), and an estimate of the initial FeO_t content of the melt
201 inclusion prior to post-entrapment crystallization. The initial FeO_t content was estimated by
202 determining a relationship between olivine Fo content and melt FeO_t contents by calculating the
203 equilibrium olivine Fo content for co-erupted matrix glasses (see Supporting text 2, Fig. S3). While
204 olivine-hosted melt inclusions erupted in the later stages of F8 (28th May-1st Aug) have experienced
205 up to 35% PEC (Wieser et al., 2021), the maximum amount of PEC experienced by olivines erupted at
206 the earlier fissures examined in this study is 10% (median= 4%).

207 Unlike for olivine, there is no clear consensus as to the best way to reconstruct orthopyroxene,
208 clinopyroxene, and plagioclase-hosted melt inclusion compositions for the effects of PEC. We discuss
209 the approach taken in this study in detail in Supporting Text S2. Briefly, for clinopyroxene we
210 examine the degree of disequilibrium between each melt inclusion and its host clinopyroxene crystal
211 using the four equilibrium tests described by Neave et al. (2017): $K_D^{\text{Fe-Mg}}$, Di-Hd, En-Fs and CaTs.
212 Melt inclusions lie within the equilibrium field for all parameters except $K_D^{\text{Fe-Mg}}$ (Fig. S4, likely
213 resulting from the fact that the concentration of FeO and MgO are far more sensitive than other
214 major elements to the crystallization of small amounts of clinopyroxene on the wall of the melt
215 inclusion). If a melt inclusion has crystallized clinopyroxene on its walls, the composition of that
216 clinopyroxene must be added back into the measured melt inclusion composition to restore the
217 composition at the time of entrapment. Similarly, if the inclusion was heated up and clinopyroxene

218 dissolved following entrapment, that clinopyroxene must be removed from the measured inclusion
219 composition. We add/subtract the composition of the host clinopyroxene until the melt inclusion
220 and host crystal are in K_D^{Fe-Mg} equilibrium (using Equation 35 of Wood and Blundy, 1997). The
221 median amount of clinopyroxene addition needed to obtain K_D^{Fe-Mg} equilibrium was 4 wt% ($1\sigma=5\%$;
222 Supporting Fig. S5). Corrected melt inclusion compositions meet all four equilibrium tests, and lie
223 closer to major element trajectories defined by co-erupted matrix glasses than uncorrected
224 compositions (Supporting Fig. S4-S5).

225 Many orthopyroxene-hosted melt inclusions also lie out of K_D^{Fe-Mg} equilibrium with the
226 composition of the host crystal (Supporting Fig. S6). However, unlike for clinopyroxene,
227 adding/subtracting the composition of the host crystal to reach K_D^{Fe-Mg} equilibrium results in a
228 worse fit to the major element trajectories defined by matrix glasses in CaO, and TiO₂ vs. MgO space
229 (Supporting Fig. S7). Given that most orthopyroxene-hosted melt inclusions lie close to the major
230 element trajectory of matrix glasses, we leave their compositions uncorrected.

231 The vast majority of plagioclase-hosted melt inclusions lie very close to the compositional trend
232 defined by matrix glasses, with the exception of 6 melt inclusions erupted at F17, which are offset to
233 significantly higher FeO (Fig. 2b) and lower CaO contents at ~6 wt% MgO (Supporting Fig. S8). These
234 offsets are indicative of crystallization of ~15 wt% plagioclase on the walls of the melt inclusion
235 following entrapment, likely due to extensive cooling between the formation of these relatively high
236 anorthite crystals (~An₆₀) and their incorporation in the significantly cooler F17 melt composition (in
237 equilibrium with ~An₄₅₋₅₀). These six melt inclusion compositions were corrected by adding the
238 composition of the host crystal back into the measured melt inclusion composition to obtain the
239 best fit to matrix glass data in MgO vs. Al₂O₃ space (Supporting Fig. S9). All other plagioclase-hosted
240 melt inclusions were left uncorrected.

241 3. Results

242 3.1 Major and Volatile Elements

243 The composition of melt inclusions and matrix glasses erupted between May 3rd and 16th define
244 coherent trends in major element space, spanning basaltic to dacitic compositions (~7 to 0.5 wt%
245 MgO, 48 to 68 wt% SiO₂; Fig. 2a). The compositions of matrix glasses are distinctly clustered, with
246 the largest group at ~4 wt% MgO (erupted at F4, F5, F8, F10-12 during Early Phase 1 on May 3-9th), a
247 second cluster at ~1 wt% MgO (erupted at F17 on May 14th), and a third cluster at ~6-7 wt% MgO
248 erupted in Phase 3 at Ahu'ailā'au (F8) after May 28th. Lavas erupted from F13 following its re-
249 activation (F13-react) plot between the Early Phase 1 and F17 cluster, while lavas erupted at F20 on
250 May 16th plot between the Early Phase 1 cluster and the higher MgO Phase 3 cluster. As above, we
251 refer to early Phase 1 fissures (F4, F5, F8, F10-12), F13-react, F17 and F20 as early fissures (3rd-16th
252 May) to distinguish them from the more MgO-rich material erupted at Ahu'ailā'au (F8) after May
253 28th in Phase 3.

254 Melt inclusion and matrix glass Cl and F concentrations increase with decreasing melt MgO contents,
255 with the most volatile-rich dacitic melt inclusions erupted at F17 having Cl and F concentrations of
256 ~1000 ppm (3-5 times higher than observed in melt inclusions erupted in phase 3; Fig. 3a-b). F
257 concentrations show significantly more scatter than Cl concentrations. The upper limit of melt
258 inclusion H₂O contents also increases with fractional crystallization, reaching 1 wt% in the early
259 Phase 1 lavas with ~4 wt% MgO, and up to 2 wt% in melt inclusions erupted at F17 with 0.5-1 wt%
260 MgO (Fig. 3d). These F17 measurements greatly exceed the highest H₂O content measured in melt
261 inclusions from historic eruptions at Kīlauea (~1 wt% H₂O, Sides et al. 2014). Unlike Cl, F, H₂O and Zr

262 (Fig. 2d , Fig. 3a-c), melt inclusion CO₂ concentrations are approximately constant between 4 and 6
263 wt% MgO (with a maximum concentration of ~300-400 ppm), then decline slightly with decreasing
264 MgO to ~200 ppm at 0.5-1 wt% MgO (Fig. 3c).

265 3.2 Trace elements

266 As discussed further in section 4.4, Nb/Y is not extensively fractionated during crystallization at
267 Kīlauea, so can be used to identify distinct magma batches (Pietruszka et al., 2021). Early Phase 1
268 matrix glasses from F4, F5, F8, F10-12 (green symbols) have Nb/Y ratios of ~0.6-0.8, which overlap
269 with the composition of matrix glasses erupted at F17 (Fig. 4a). In contrast, matrix glasses erupted at
270 F13 following its reactivation and F20 have lower Nb/Y ratios, plotting closer to the composition of
271 Phase 3 glasses. Overall, the trends defined by matrix glasses measured by LA-ICP-MS in this study
272 are very similar to those determined by whole-rock XRF measurements (Pietruszka et al., 2021,
273 magenta dots, Fig. 4a).

274 Unlike the temporal variations seen in matrix glass and whole-rock trace element compositions,
275 there are no apparent inter-sample differences in the Nb/Y ratios of melt inclusions erupted from
276 early fissures, or relationships between Nb/Y ratios and host crystal An# or Mg# in these samples
277 (Fig. 4c-d). As for matrix glasses, the vast majority of melt inclusions from Phase 3 (orange colors;
278 Fig. 4c-d) lie to distinctly lower Nb/Y ratios, with only a small number of melt inclusions erupted on
279 May 28th overlapping with the Nb/Y ratios of melt inclusions from early fissures (particularly those
280 hosted in lower forsterite olivines).

281 4. Discussion

282 4.1 Modelling Fractional Crystallization

283 The coherent trends in major and volatile element trajectories defined by melt inclusions and matrix
284 glasses indicate that the dominant cause of chemical variability in erupted melts during the first two
285 weeks of the 2018 eruption was fractional crystallization. Previous studies have shown that Kīlauea
286 melts fractionate only olivine (+minor Cr-spinel) at >6.8 wt% MgO, defining a relatively flat trajectory
287 for all elements vs. MgO (Wright and Fiske, 1971). Next, clinopyroxene joins the liquidus, followed by
288 plagioclase after only 5-15 °C cooling, as well as minor orthopyroxene (Helz and Wright, 1992). The
289 crystallization of plagioclase accounts for the rapid increase in melt FeO_T as MgO contents drop from
290 6 to 4 wt% (Fig. 2b). The appearance of plagioclase and clinopyroxene on the liquidus also causes the
291 slope of relatively incompatible elements like TiO₂ and Zr vs. MgO to greatly increase, because a
292 larger mass of solid is fractionated for a given decrease in melt MgO content. At ~4 wt% MgO, Fe-Ti
293 oxides saturate (Helz and Wright, 1992), driving the concentration of FeO_T and TiO₂ down, and SiO₂
294 up. Concurrently, P/Nd begins to decline (along with P₂O₅; Supporting Fig. S10), indicating the onset
295 of apatite crystallization (visible as microphenocrysts in matrix glasses from F17).

296 Fractional crystallization models were run using alphaMELTS for MATLAB (Rhyolite melts V1.2.0,
297 Antoshechkina and Ghiorso, 2018; Ghiorso and Gualda, 2015; Gualda et al., 2012) to recreate
298 observed major element trends. Although it is likely that the melt fractionating within the LERZ was
299 a previously stored component, the starting composition of models was set as the average major-
300 element composition of glasses erupted in Phase 3. This is because the vast majority of literature
301 data for previous eruptions comprises whole-rock compositions, which may be affected by crystal
302 addition. The starting composition used for the modelling lies within the range of whole-rock
303 compositions from various early fissures from Gansecki et al. (2019), justifying this choice. The initial
304 Fe³⁺/Fe_T ratio was set at 0.15 (Lerner et al., 2021; Moussallam et al., 2016), and models were run
305 without a buffer, allowing for changes in oxidation state following Fe-Ti oxide saturation. The

306 pressure was set at 650 bars, based on the upper range of saturation pressures measured in Early
307 Phase 1 melt inclusions (see section 4.5). The trajectory of trace and volatile elements during
308 fractional crystallization was modelled using the mass of solid phases produced, and the fractional
309 crystallization equation ($C_{\text{final}}=C_{\text{initial}}*F^{D-1}$, where F is the amount of melt remaining, and D is the bulk
310 partition coefficient).

311 In detail, we consider two separate model trajectories in MELTS. Model A (red line) was run with an
312 initial H₂O concentration of 0.1 wt%. Model A is a good match to the trajectory of major elements
313 from ~6.8 to 4 wt% MgO, capturing the sharp increase in FeO_t and TiO₂ contents with decreasing
314 MgO. At lower MgO contents, Model A is a good fit to FeO_t, TiO₂ and SiO₂ (Fig. 2a-c) but
315 overestimates CaO, and underestimates Al₂O₃ (Supporting Fig. S11) indicating that the ratio of
316 clinopyroxene vs. plagioclase fractionating in this model is incorrect. Similarly, Model A does not
317 recreate the prominent downtick in P₂O₅ (and P/Nd) caused by apatite crystallization, because
318 MELTS currently only accounts for a hydroxy-apatite phase (rather than the more stable fluorapatite
319 observed in natural magmas; Rooney et al., 2012). The failure to correctly model apatite saturation
320 may also account for some of the overestimation of melt CaO contents at <4 wt% MgO (Supporting
321 Fig. S11). However, despite these major-element discrepancies, the concentration of Zr and Cl
322 (assuming complete incompatibility) in Model A provides a good fit to the composition of melt
323 inclusions and matrix glasses. This indicates that MELTS model results are accurately predicting the
324 amount of solid being fractionated, even if predictions of relative phase proportions are slightly
325 inaccurate.

326 If MELTS models are initialized with water contents similar to those in melt inclusions with ~6 wt%
327 MgO (H₂O_i=0.5 wt%, Model B, Fig. 2, dashed black lines), MELTS saturates Fe-Ti oxides significantly
328 earlier and plagioclase later than Model A, such that Model B fails to achieve the prominent TiO₂ and
329 FeO enrichment seen in erupted glass compositions. This has been noted previously for Kīlauea by
330 Garcia (2003). Fortunately, the total mass of solids predicted by model A and model B are reasonably
331 similar, so these models predict similar trajectories for incompatible elements such as Zr (Fig. 2d).
332 We also generate a 3rd major element path for fractional crystallization, using MELTS model A at >4
333 wt% MgO, and a best fit through measured glass and melt inclusion data for <4 wt% (Supporting Fig.
334 S11). The major element differences between these three models have a relatively small effect on
335 volatile solubility (Supporting Fig. S12) and calculations of melt viscosity (Fig. 5b). This justifies the
336 use of Model B to investigate changes in volatile systematics and viscosity during fractional
337 crystallization, even though this model doesn't fully recreate the observed major element
338 systematics (Fig. 2).

339 Rhyolite-MELTS V1.2.0 fails to reproduce olivine-only fractionation at >6.8 wt% MgO, so this section
340 of the fractionation path was modelled using the “reverse crystallization” (RFC) tool in Petrolog3
341 (Danyushevsky and Plechov, 2011) from the specified starting composition.

342 4.2 Volatile Element Evolution during Crystallization

343 MELTS modelling demonstrates that the prominent enrichment in melt inclusion Cl concentrations
344 with decreasing MgO results from extensive fractional crystallization (Fig. 3a). F data shows
345 significantly more scatter than Cl (Fig. 3b), but approximately follows the fractional crystallization
346 trend. Notably, six plagioclase-hosted F17 melt inclusions with higher MgO contents (~4-6 wt%) lie
347 significantly above the fractional crystallization line for F, indicating that these inclusions may have
348 diffusively re-equilibrated with the significantly more F-rich carrier melts in which they were erupted
349 (Fig. 3b). Apparent open-system behavior of F in olivine-hosted melt inclusions has been noted
350 previously by Koleszar et al., (2009) and Portnyagin et al. (2008), while Neave et al. (2017) attribute F

351 enrichment in plagioclase-hosted melt inclusions to diffusive pile up during rapid growth. We favour
352 diffusive re-equilibration here, because many F17 melt inclusions have lower F contents than co-
353 erupted matrix glasses, and it is only the most MgO-rich melt inclusions which are expected to have
354 equilibrated with more F-rich, MgO-poor melts that show notable enrichment above the fractional
355 crystallization trend. Two of these melt inclusions also show elevated H₂O contents (Fig. 3d), which is
356 consistent with these more primitive melt inclusions equilibrating with the more MgO-poor, F- and
357 H₂O- rich F17 carrier melt prior to eruption (with F re-equilibration appearing to occur faster than
358 H₂O re-equilibration in plagioclase).

359 Accounting for these complexities with F, the relative similarity of the trajectories of matrix glasses
360 and melt inclusions indicates that the amount of F and Cl degassed upon eruption is sufficiently
361 small that it cannot be resolved petrologically by comparing melt inclusions and matrix glasses. In
362 contrast, H₂O and CO₂ degas strongly upon eruption, shown by the significantly lower concentrations
363 in matrix glasses relative to melt inclusions (Fig. 3c-d).

364 The upper limit of melt inclusion H₂O contents is bracketed by treating H₂O as an incompatible
365 element in MELTS model A, with a number of melt inclusions plotting to lower H₂O contents. In
366 contrast, melt inclusion CO₂ contents show a slight decline with decreasing MgO contents, again
367 with a lot of scatter. The evolution of H₂O-CO₂ is significantly more complicated than elements like Zr
368 and Cl, because Kīlauea melts are saturated in a mixed H₂O-CO₂ fluid at crustal pressures (Gerlach et
369 al., 2002). This means there is a competing influence between their incompatible behavior in solid
370 phases and their loss to an exsolving fluid phase. To demonstrate this trade-off, we use the solubility
371 model MagmaSat (Ghiorso and Gualda, 2015) implemented in the Python3 tool VESical (Iacovino et
372 al. 2021) to calculate the solubility of pure-CO₂ fluids and pure-H₂O fluids at 650 bars for the
373 synthetic major element path which provides the best fit to measured glass and melt inclusion data
374 (Supporting Fig. S11), using temperatures from the MgO geothermometer of Helz and Thornber,
375 (1987). At low pressures where non-ideality between H₂O-CO₂ is negligible, the concentration of
376 either volatile species will not exceed the pure solubility limit. The solubility of pure CO₂ is relatively
377 low (~400 ppm), and plots close to the measured CO₂ contents of melt inclusions (dashed blue line,
378 Fig. 3c). In contrast, the solubility of pure H₂O is much higher, plotting significantly above the
379 concentration of melt inclusions (dashed blue line, Fig. 3d).

380 In reality, H₂O-CO₂ exsolves as a mixed fluid in magmatic systems. At the low pressures considered
381 here, where mixing between H₂O and CO₂ is close to ideal, the behavior of mixed fluids is best
382 understood considering Henry's Law (Lowenstern, 2001). Namely, the addition of H₂O to a pure CO₂
383 fluid lowers the partial pressure of CO₂, and therefore lowers its solubility in the co-existing melt
384 phase. Similarly, adding CO₂ to a pure H₂O fluid lowers the solubility of H₂O in the melt. The solubility
385 of H₂O-CO₂ in equilibrium with a mixed fluid during fractionation can be simulated in alphaMELTS for
386 MATLAB along the fractionation path of MELTS model B (which has realistic initial H₂O contents).
387 From 6 to 4 wt% MgO, when the melt is relatively H₂O -poor, the composition of the exsolved vapour
388 phase (X_{CO_2}) is >90 wt% CO₂. With increasing fractionation, the progressive rise in the H₂O content of
389 the melt causes X_{CO_2} to steadily decrease, reaching ~65 wt% (40 mol%) at 0.5 wt% MgO (Fig. 5a). This
390 decrease in the proportion of CO₂ in the fluid causes the CO₂ solubility predicted by MELTS model B
391 (black dotted line, Fig. 3c) to decrease more rapidly than the model for pure CO₂ solubility. MELTS
392 model B tracking mixed volatile solubility during fractional crystallization effectively brackets the
393 upper limit of the majority of melt inclusion data, providing strong evidence that melts were vapour-
394 saturated at depth, and that the solubility model MagmaSat used in Rhyolite-MELTS is effectively
395 capturing changes in volatile solubility during fractional crystallization.

396 A significant proportion of the total amount of CO₂ in the system is lost to the vapour phase during
397 fractional crystallization, partially because $X_{\text{CO}_2} > 65$ wt% across most of the fractionation interval, and
398 because there was so little CO₂ in the system to start with (~385 ppm at 6.7 wt% MgO). For example,
399 fractionation of just 0.02 wt% fluid with $X_{\text{CO}_2} = 80$ wt% removes 160 ppm of CO₂ from the system. In
400 contrast, the higher initial mass of H₂O in the system combined with high X_{CO_2} ratios mean that very
401 little H₂O is lost to the fluid; in the example above only 40 ppm of H₂O is lost to the vapour compared
402 to 160 ppm of CO₂. Notably, this is only ~1% of the initial amount of H₂O in the system, which is well
403 within analytical error of SIMS measurements (Fig. 5b). Thus, the trajectory of melt H₂O during
404 extensive fractional crystallization at shallow crustal pressures in the presence of an exsolved H₂O-
405 CO₂ fluid is very similar to models of H₂O enrichment assuming no volatile loss. Minimal H₂O-loss to
406 the exsolving fluid accounts for the high H₂O contents of the dacitic melts erupted at F17. We
407 caution that the apparent incompatible behavior of H₂O should not be interpreted as evidence for
408 “H₂O-undersaturation”; it may simply indicate that the magma was saturated in a mixed fluid that
409 was relatively H₂O-poor.

410 While the upper limit of melt inclusion volatile contents follow MELTS model B, a number of melt
411 inclusions plot to lower CO₂ and H₂O concentrations. The cause of this spread is best evaluated using
412 plots of H₂O vs. CO₂, with degassing paths for specific melt inclusions overlain (calculated using the
413 solubility model MagmaSat implemented in VESICAL; Ghiorso and Gualda, 2015; Iacovino et al., 2021;
414 Fig. 6). Some samples (e.g., F10; Fig. 6b) show tightly clustered volatile contents, which plot very
415 close to fractional crystallization trends (Fig. 3c-d). Others (e.g. F13-react, Fig. 6a) show considerable
416 spread to lower CO₂ contents, approximately following the trajectory of modelled degassing paths.
417 This indicates that many melt inclusions were formed or sealed during ascent, trapping a
418 progressively degassing magma. Horizontal spread to variable H₂O contents may reflect diffusive re-
419 equilibration of melt inclusions with degassing carrier melts, or crystals being entrained into melts
420 with higher H₂O than those they grew from (e.g., 6 plag-hosted MI from F17). F17 melt inclusions
421 show the most scatter, which can be recreated by a combination of the trajectory shown by MELTS
422 model B, degassing, and H⁺ re-equilibrium (Fig. 6d). However, without measurements of D/H ratios
423 in melt inclusions, the exact contribution of degassing, fractional crystallization, and re-equilibration
424 of H₂O is difficult to determine. In general, we note that H⁺ loss through plagioclase is likely slower
425 than in olivine (Johnson and Rossman, 2013). Additionally, we observe that multiple melt inclusions
426 hosted within single crystals erupted at F17 have a wide range of H₂O-CO₂ contents, and there is no
427 relationship between H₂O contents and the position of the melt inclusion within the crystal, or the
428 size of the inclusion (inconsistent with H-loss by diffusion, Supporting Fig. S13-15). The observed
429 variability within individual crystals is more consistent with melt inclusions being trapped at various
430 points along a degassing path. The influence of degassing is considered further in section 4.5.

431 4.3 Implications for Eruption Style Variations

432 The higher water contents of F17 melts relative to basaltic and basaltic-andesite melts (resulting
433 from larger amounts of fractional crystallization) means that a significantly larger volume of volatiles
434 is exsolved as these melts ascend to the surface (Fig. 7c). We demonstrate this by tracking the ascent
435 of magma along a closed-system degassing path in alphaMELTS for MATLAB for a composition
436 representative of the basaltic material erupted at the early fissures (MgO=4.3 wt%, SiO₂=50.8 wt%,
437 H₂O=0.95 wt%) and a composition representative of the dacitic melt erupted at F17 (MgO=1 wt%,
438 H₂O=1.78 wt%; Fig. 7). For both melt compositions, these models predict that the volume of
439 exsolved volatiles relative to the total volume of the system (fluid+melt+crystals) is very low at 650
440 to 400 bars (Fig. 7a), and degassing only produces measurable changes in the dissolved
441 concentration of CO₂ in the melt (Fig. 7b). This is also shown on the degassing paths in Fig. 6a-d. At

442 <300 bars for the dacitic melt, H₂O begins to exsolve in larger quantities, causing the volume of
443 exsolved volatiles to increase rapidly with decreasing pressure. The onset of significant quantities of
444 H₂O exsolution occurs at a pressure ~100-200 bars lower for basaltic melts compared to dacitic melt.

445 To compare these ascent paths, we divide the volume of volatiles exsolved during ascent of the
446 dacite by the volume of volatiles exsolved during ascent of the basaltic (Fig. 7a). The higher initial
447 H₂O content of the dacite, combined with the fact that H₂O begins to exsolve at higher pressures
448 means that at ~200 bars, the volume of volatiles in the dacitic melt is nine times higher than in the
449 basaltic melt (Fig. 7c). The larger volatile volume at depth, combined with higher melt viscosity, will
450 promote bubble nucleation and coalescence, which is a key driver of Strombolian activity (Jaupart
451 and Vergnolle, 1988). By the time these melts reach the surface, where almost all of the initial H₂O
452 in both melts has exsolved, the volume of volatiles exsolved in the dacitic melt is approximately
453 twice that for the rift-stored basalts.

454 Eruption style is also influenced by melt viscosity. First, we model changes in melt viscosity during
455 fractional crystallization (Fig. 5c), and then assess changes in viscosity during ascent using the
456 viscosity model of Giordano et al. (2008) (Fig. 7d). As viscosity is highly sensitive to melt H₂O content,
457 we calculate viscosity for the major and volatile-element trajectories from MELTS Model B, using
458 MELTS temperatures and temperatures from Helz and Thornber (1987). Because Model B is a
459 relatively poor fit to major element data at <4 wt% MgO, we also calculate the viscosity for the best
460 fit to measured major and volatile element data (see Supporting Fig. S11) using the MgO
461 geothermometer of Helz and Thornber, (1987) for melt temperature, and H₂O contents from a best
462 fit to MELTS model B. The two viscosity models using Helz and Thornber (1987) are very similar,
463 predicting a rapid increase in melt viscosity below 4 wt% MgO (Fig. 5b) as a result of SiO₂ enrichment
464 in the melt following the onset of Fe-Ti oxide fractionation (Fig. 2a) and a steady decline in the
465 temperature of the melt. The model using MELTS temperatures is similar until ~2 wt% MgO, after
466 which it rises to higher values because MELTS predicts significantly lower temperatures than Helz
467 and Thornber (1987).

468 Prior to the onset of degassing upon ascent, the dacitic melt is ~10-20 times more viscous than the
469 basaltic melt (Fig. 7d). During ascent along a closed-system degassing path, the discrepancies in
470 viscosity between the basaltic and dacitic melt are enhanced, so at the surface, the dacitic melt is
471 70-130 times more viscous (Fig. 7d, Soldati and Dingwell, 2021). The relative viscosities calculated in
472 our models are very similar to viscosities obtained from direct measurements of 2018 lavas at
473 atmospheric pressure (yellow and blue stars on Fig. 7d, Soldati and Dingwell, 2021) particularly when
474 the Helz and Thornber, (1987) MgO thermometer is used for both comparisons. The increasing
475 contrast in viscosity between the basalt and dacite during ascent to the surface results from the fact
476 that the dacitic melt has more H₂O to lose (which drives a rapid increase in viscosity). It is also worth
477 noting that these calculations only track the viscosity of the melt phase. While the proportion of
478 crystals relative to melt is similar for F17 vs. early phase 1 lavas (~39% vs. 42%; Gansecki et al.,
479 2019), F17 samples have completely different crystal shapes, with a larger number of smaller,
480 elongated plagioclase crystals which may further amplify these differences in viscosity (see BSE
481 images in Supplement Dataset S3, p78-80 vs. p82-84). The larger volume of bubbles in F17 melts will
482 also affect viscosity, although this is difficult to quantify because it depends greatly on the amount of
483 shear rate and strain of bubbles (Llewellyn et al., 2002; Manga and Loewenberg, 2001).

484 It is well recognised that differences between Hawaiian style lava fountaining (exhibited weakly by
485 the early fissures) and Strombolian activity (exhibited strongly by the F17 andesite) are controlled by
486 magma rise speed, viscosity, and volatile content (Houghton et al., 2016; Houghton and

487 Gonnermann, 2008; Wilson and Head, 1981). Strombolian eruptions occur when bubbles can
488 coalesce and rise faster than the surrounding melt, resulting in the bursting of large bubbles at the
489 surface along with relatively minimal volumes of melt (see Supporting Video 1 and 2, Wilson and
490 Head, 1981). Separated two phase flow is promoted at higher gas/magma ratios and for higher melt
491 viscosities, which tends to reduce magma rise rates (Gonnermann and Manga, 2013). The 10-100
492 higher viscosities of F17 melts compared to more primitive basaltic melts at <300 bars (Fig. 7d),
493 combined with the higher exsolved volatile fraction (Fig. 7c), mean that melt rise speeds would have
494 been significantly slower for a given conduit width and pressure gradient, favouring bubble rise and
495 coalescence (e.g., Parfitt and Wilson, 1995). Thus, the high explosivity observed at F17 is likely a
496 direct consequence of extensive fractional crystallization driving up melt viscosity and H₂O contents,
497 causing a larger volume of bubbles to exsolve (and coalesce) at greater depths in the conduit at a
498 deeper depth, along with likely feedbacks between a higher viscosity and lower ascent rates in the
499 dacitic melt (further favouring bubble coalescence).

500 4.4 Identifying Parent Magmas

501 Major and volatile element systematics indicate that melts erupted in the first two weeks of the
502 2018 eruption underwent extensive (and variable amounts) of fractional crystallization during rift-
503 zone storage. However, using major elements systematics alone it is difficult to determine whether
504 these melts were derived from fractionation of a single parent magma, or a number of different
505 magma bodies, because the major element contents of different Kīlauea eruptions are relatively
506 similar at a given MgO content (Helz and Wright, 1992). In contrast, through time, Kīlauea eruptions
507 show clear variations in incompatible trace element ratios such as Nb/Y, La/Yb, and Zr/Y (Fig. 4b),
508 interpreted to represent heterogeneity in the Hawaiian mantle plume, as well as variations in the
509 degree of mantle melting (Pietruszka and Garcia, 1999a, 1999b). It is probable that the parent
510 magma(s) for the variably-evolved melts erupted in 2018 were intruded into the rift zone and
511 erupted at the surface during a previous eruptive episode (the most recent eruptive events occurred
512 in 1790, 1823, 1840, 1955 and 1960 A.D). The 1790 and early 1955 fissures are of particular interest,
513 because these fissures were located in the same area of the LERZ as the 2018 eruption (Fig. 1c; Helz,
514 2008; Moore, 1992; Wright and Fiske, 1971). Similarly, the 1960 eruption occurred a short distance
515 downrift, so presumably its magma supply passed through the section of the rift zone where the 2018
516 eruption occurred. Here, we use trace element ratios to determine the composition of historic
517 eruptions on the LERZ and compare these to our measurements from 2018 eruptive products to
518 deduce the most probable parent magma(s).

519 Unlike previous work using trace elements to track magma batches at Kīlauea in lavas which have
520 undergone mostly olivine-saturation (Pietruszka and Garcia, 1999b; Wieser et al., 2019), the melts
521 analysed here have experienced fractionation of a number of phases (olivine, clinopyroxene,
522 orthopyroxene, plagioclase, \pm ilmenite, magnetite, and apatite). The presence of these other phases
523 in addition to olivine provides a potential mechanism for these trace element ratios to be
524 fractionated during crystallization, which might obscure attempts to compare rift-stored melt
525 compositions to magma batches defined by analyses in the literature of mostly olivine-saturated
526 samples from previous LERZ eruptions and Kīlauea's summit. To assess which element ratios are
527 least fractionated by extensive crystallization at Kīlauea, we use the trace element systematics of
528 glasses from drill cores taken through the Kīlauea Iki lava lake (Greaney et al., 2017). This lava lake
529 experienced extensive fractionation of a single magma batch across a range of MgO contents very
530 similar to the lavas examined here, with a very similar mineral assemblage (Helz, 1980; Helz and
531 Thornber, 1987). Thus, evaluation of changes in trace element contents with decreasing MgO in
532 these lava lake samples providing a unique opportunity to establish whether trace element ratios

533 are fractionated across this differentiation interval. Between 6 and ~0 wt% MgO, Kīlauea Iki matrix
534 glass Nb, Yb and Y all increase by a factor of 2-3, while La and Zr behave more incompatibly;
535 increasing by a factor of 3-4 times (Supporting Fig. S16). This causes the Nb/Y ratio to remain almost
536 constant during extensive fractionation, while La/Yb increases slightly (~20%), and Zr/Y more than
537 doubles (Supporting Fig. S17). Pietruszka et al., (2021) also demonstrate that Nb/Y is resistant to
538 fractionation using available partitioning data for relevant phases. In addition to being the least
539 fractionated ratio, it is also advantageous to use Nb/Y for comparisons because it is the only trace
540 element ratio which has been reported by a number of different studies investigating historic lavas
541 on the LERZ (because both Nb and Y can be measured accurately and precisely by XRF, unlike La and
542 Yb).

543 Previous work has shown that Nb/Y ratios in historic summit and rift eruptions follow a prominent
544 trough-peak-trough shape between 1800 to present (Garcia et al., 2021; Marske, 2010; Pietruszka
545 and Garcia, 1999b, 1999a). The highest Nb/Y ratios are observed in lavas from the mid 1900s,
546 declining towards the values measured during the final stages of the Pu'ū'ō'ō eruption (Fig. 4b;
547 Gansecki et al., 2019). Nb/Y ratios from whole rock analyses of previous LERZ eruptions follow a
548 similar trend to summit eruptions (Helz and Wright, 1992; Marske, 2010; Norman, 2005). Matrix
549 glasses erupted in Early Phase 1 in 2018 have much more similar Nb/Y ratios to the 1955-1960
550 eruptions than melts erupted in the period between 1790-1840 AD (characterized by lower Nb/Y
551 ratios; ~0.5-0.6, see also Pietruszka et al., 2021). This indicates that melts erupted in 1955 or 1960
552 are a more probable parent than those erupted in 1790 and 1840.

553 Two main phases of the 1955 eruption have been identified, with an early phase (1955E) thought to
554 tap more MgO-poor magmas already present within the rift zone (5-5.7 wt % MgO_{Whole-rock}), while
555 later phase (1955L) tapped more MgO-rich magmas (6.2-6.8 wt% MgO_{Whole-rock}) interpreted to reflect
556 mixing of the 1955E lava with a more MgO-rich, summit-derived component (Helz and Wright,
557 1992). This is analogous to the temporal evolution from MgO-poor to MgO-rich observed in the 2018
558 eruption. Using the analyses of Marske (2010), along with additional analyses of 1955 lavas from
559 Helz and Wright, (1992) and Norman, (2005), we show that the mean and range of Nb/Y ratios in
560 lavas erupted in the early and late phase of 1955 are a good match to the matrix glass compositions
561 erupted in Early Phase 1 in 2018 (Fig. 4a). Using Nb/Y ratios alone, it is not possible to definitively
562 distinguish between the 1955E, 1955L and 1960 lavas, but it is notable that the 1955E vents are
563 geographically extremely close to the 2018 vents (Fig. 1c).

564 Whole-rock Nb/Y ratios (pink dots on Fig. 4a, Pietruszka et al., 2021) of lavas erupted in the first 20
565 days of the 2018 eruption form a mixing trend between the "1955-1960 like" Nb/Y ratios exhibited
566 by Early Phase 1 lavas, and the Nb/Y ratios measured in Phase 3. This trend is also seen in our matrix
567 glass analyses of F13-react and F20. These mixing trends indicate that LERZ-stored melts were
568 progressively flushed out by the more MgO-rich, lower Nb/Y component supplying phase 3
569 (Gansecki et al., 2019). Interestingly, while some of the more MgO-rich samples erupted at F17 (WR
570 MgO=3-4 wt%, Gansecki et al., 2019; Pietruszka et al., 2021) show evidence for mixing with the
571 lower Nb/Y component from Phase 3 (glass MgO=~0.5 wt%, WR MgO=2.3-2.6 wt%, Supporting Fig.
572 S18), our sample has glass Nb/Y ratios which entirely overlap with these Early Phase 1 glasses (Fig.
573 4a). This overlap indicates that melts erupted in Early Phase 1 and at F17 were likely derived from a
574 common parent magma, and that our F17 sample experienced minimal mixing with magma erupted
575 in Phase 3. Thus, it seems likely that the initial activity at F17 was triggered by an increase in
576 overpressure, change in crustal stress, or thermal rejuvenation resulting from the intrusion of a dike
577 containing Phase 3 magma into this section of the rift zone, rather than as a direct consequence of
578 mixing between different magma compositions. Subsequent mixing of magmas indicated by whole-

579 rock Nb/Y data from Gansecki et al., (2019) and Pietruszka et al., (2021; Supporting Fig. S18) likely
580 helped to liberate more of this stored material, fuelling the large lava flows with more MgO-rich
581 compositions that erupted from F17 until May 25th (not sampled in this study).

582 Melt inclusion Nb/Y ratios in plagioclase crystals show no correlation with the anorthite (An) content
583 of the host (Fig. 4c), which is consistent with a common parental magma for all rift-stored melts.
584 Similarly, melt inclusion Nb/Y ratios in olivine, clinopyroxene and orthopyroxene from early fissures
585 show no correlation with host Mg# [Mg/(Mg+Fe) atomic; Fig. 4d)], and are clearly distinct from the
586 significantly lower Nb/Y ratios in matrix glasses and the majority of melt inclusions in Phase 3. This
587 indicates that very few inclusions trapped melts that had undergone significant mixing with Phase 3
588 melts. Interestingly, 3 plagioclase-hosted melt inclusions from early F8, F5 and F20 plot with F17
589 melt inclusions in Nb/Y-An space, and 6 inclusions from F17 have higher MgO contents and major
590 element contents indicating post-entrapment cooling. This supports the idea that more MgO-poor
591 pockets of melt exist in close proximity with more MgO-rich bodies in the rift zone magma body, in
592 order for more MgO-rich fissures to scavenge more An-poor crystals, and for more MgO-poor
593 fissures to scavenge more An-rich crystals. The five olivine-hosted melt inclusions erupted on May
594 28th in Phase 3 (F8) with higher Nb/Y ratios (~0.7; Fig. 4d) are hosted in olivines with some of the
595 lowest forsterite contents of phase 3, and have melt inclusion TiO₂ contents overlapping with Early
596 Phase 1 melt inclusions and glasses (Supporting Fig. S19). This indicates that they may have also
597 grown from rift-stored melts with higher Nb/Y ratios, and these crystals were recycled into the lower
598 Nb/Y Phase 3 melts.

599 Overall, melt inclusion Nb/Y ratios indicate that melt inclusions erupted in the first 2 weeks of 2018
600 likely crystallized from a single parent magma body (or a series of magma bodies with similar
601 parental magmas), and that very few melt inclusions crystallized following mixing between rift-
602 stored and Phase 3 melts (although crystal recycling between different stored magma batches did
603 occur).

604 4.5 Pre-eruptive magma storage depths

605 Melt inclusion H₂O and CO₂ contents provide further constraints on the nature of magma storage
606 within the LERZ. The increase in H₂O and decrease in CO₂ with decreasing MgO content (Fig. 3c-d)
607 provide strong evidence that melts were volatile-saturated throughout the fractionation interval
608 considered here. Thus, estimates of the melt inclusion temperature, major elements, H₂O and CO₂
609 contents at the time of entrapment may be used to calculate the pressure at which those dissolved
610 volatile contents would be saturated, and, by extension, the pressure at which the melt inclusion
611 was trapped within its host crystal. This is termed the “saturation pressure”, or “entrapment
612 pressure”. By estimating the crustal density profile, these saturation pressures can be converted into
613 depths within the volcanic edifice.

614 As discussed in section 4.2, H₂O-CO₂ systematics indicate that a number of melt inclusions were
615 likely trapped during ascent, after degassing had begun. While these still yield useful saturation
616 pressures (indicating the depths at which inclusions are forming on the way to the surface), we also
617 filter the dataset to only consider melt inclusions trapped from magmas prior to the onset of
618 degassing on ascent to identify magma storage reservoirs. To remove melt inclusions trapped during
619 degassing, we exclude inclusions with CO₂ contents lying more than 30% below the CO₂ content
620 predicted by MELTS model B at the MgO content of each melt inclusion. To remove inclusions which
621 have undergone H₂O degassing or diffusive re-equilibration, we filter inclusions more than ±30%
622 from MELTS model B. An example of this filtering process is shown in Fig. 6d-f for F17 melt

623 inclusions. Melt inclusions meeting both criteria are marked with red crosses (see also Supporting
624 Fig. S20).

625 We use the solubility model MagmaSat (Ghiorso and Gualda, 2015) implemented in the Python3 tool
626 VESIcal (v.0.01; Iacovino et al., 2021) to calculate entrapment pressures from PEC-corrected major
627 element, H₂O, and CO₂ contents of each melt inclusion, and a temperature estimated using the Helz
628 and Thornber (1987) MgO thermometer. We choose MagmaSat because this model provides the
629 best fit of available models to experimental data with basaltic to dacitic compositions, as a result of
630 its thermodynamic nature and extensive calibration dataset (P. Wieser et al., 2021). Additionally, use
631 of this model provides consistency with our models of volatile solubility during fractionation and
632 ascent, as MagmaSat is the solubility model used in Rhyolite-MELTS v.1.2.0.

633 Filtered early phase 1 melt inclusions show a distinct clustering of saturation pressures at ~0.48-0.8
634 kbar, with 4 inclusions spanning pressures up to 1.15 kbars (Fig. 8a). Filtered F17 melt inclusions
635 show a remarkable overlap with Early Phase 1, mostly spanning 0.55 – 0.8 kbar, with 2 higher
636 pressure inclusions (Fig. 8b). Only two melt inclusions from F13-react and five from F20 pass our
637 filters (Fig. 6a, c, Fig. 8c-d, Supporting Fig. S20), but these also yield very similar pressure distributions.
638 Unfiltered data for all samples stretches to significantly lower pressures, indicating that a number of
639 melt inclusions were likely trapped during ascent (Fig. 8e-h). It is worth noting that melt inclusions
640 from F13-react and F20 samples show clusters at 0.01-0.03 wt% CO₂, which could indicate that a
641 large number of inclusions were trapped during degassing, or that these fissures were tapping a
642 slightly shallower magma reservoir.

643 Using a crustal density of $\rho=2600 \text{ kg/m}^3$ based on the relationship between drill depth and pressure
644 presented from borehole drilling results at Puna Geothermal Venture from Teplow et al. (2009), melt
645 inclusion saturation pressures can be converted into magma storage depths. Filtered melt inclusions
646 cluster between ~2-3 km (Fig. 8), which aligns remarkably well with the depth of a V_p/V_s anomaly
647 imaged in this area centred at 2 km depth between Pu'uhonua'ula and Pu'ulena Crater (Fig. 1b) with
648 a vertical extent of 1.5 km (Cooper and Dustman, 1995). The lateral dimensions of this body (2.5 km
649 parallel to the rift zone, 2 km perpendicular to the rift zone) means that it directly underlies the
650 fissures erupted in the first two weeks of the 2018 eruption. Thus, it is highly probable that the
651 magma body being tapped by the 2018 eruption is the source of this seismic anomaly. Similarly, F17
652 saturation pressures align remarkably well with the depth at which an even more silicic dacitic
653 magma was drilled at the PGV injection well KS-13 in 2005 (depth of 2488m, ~650 bars pressure,
654 yellow star on Fig. 8).

655 4.6 Temporal evolution of Kīlauea's LERZ

656 We present a schematic model summarizing the evolution of Kīlauea's LERZ, based on our findings
657 and previous work on historic lava samples (Fig. 9). The early fissures of the 1955 eruption, which
658 are located very close to the 2018 eruption site (Fig. 1c), tapped melts with whole-rock contents of ~
659 5-5.7 wt% MgO. These melts were likely resident in the rift zone since at least since 1924 (when a
660 dike swarm propagated past the 2018 eruption site into Kapoho) or 1790 (the last eruption where
661 magma breached the surface in this section of the LERZ; Fig. 1c). Alternatively, ²³⁰Th-²²⁶Ra dating of
662 1955E lavas suggests that clinopyroxenes and plagioclases were resident for at least 550 years prior
663 to eruption. The later phases of the 1955 eruption focused at fissures uprift of the 2018 eruption site
664 tapped more MgO-rich magmas (6.2-6.8 wt% MgO_{Whole-rock}) which have been interpreted to reflect
665 renewed supply from the summit reservoir (Helz and Wright, 1992; Fig. 9a). This resupply may have
666 propagated downrift, helping to replenish reservoirs depleted during the early phases of the 1955
667 eruption. In 1960, activity commenced downrift of the 1955E fissures, first trapping stored magma

668 with chemical similarities to the 1955 lavas (6.1-6.6 wt% MgO_{Whole-rock}), followed by the appearance
669 of progressively higher MgO contents associated with supply of material from Kīlauea's summit
670 reservoir (7-13 wt% MgO_{Whole-rock}, Russell and Stanley, 1990). As in the later phases of the 1955
671 eruption, the 1960 summit resupply may also have replenished the larger reservoir complex residing
672 beneath the 2018 eruption site (as the 1960 magmas are very similar in terms of Nb/Y and major
673 elements; Russell and Stanley, 1990, Fig. 9b). This pulse of activity in 1955-1960 left substantial
674 amounts of magma at depth, which continued to cool and fractionate over the next 60 years as
675 activity shifted to Kīlauea's summit, South West Rift Zone and Middle East Rift Zone. In 2005, a
676 dacitic pocket of magma was encountered during geothermal drilling (Fig. 9c).

677 The propagation of a dike downrift from Pu'u'ō'ō on April 30th, 2018, disturbed these rift-stored
678 melts, with magma erupting at the surface beginning on May 3rd (Fig. 9d). After a short pause after
679 May 9th, erupted melt compositions between the 12-18th of May begin to show mixing between
680 LERZ-stored and dike-supplied melts (e.g. in matrix glass Nb/Y ratios, Fig. 4a, Gansecki et al., 2019,
681 Supporting Fig. S18), although erupted crystals are still dominated by geochemical signatures
682 indicative of the LERZ-stored component (Fig. 4c-d). Dacitic melts also began erupting at F17, and
683 Fissure 13 shows mixing between these dacitic melts and the more MgO-rich LERZ-stored
684 component. By the time activity focused at Ahu'ailā'au (F8) on May 28th, almost all LERZ-stored
685 crystals (and melts) had been flushed out by dike stored material (Fig. 9e, Fig. 4d, Supporting Fig.
686 S19).

687 It is noteworthy that Early Phase 1 lavas have similar MgO contents of early 1955 lavas, so if these
688 are the parent, the portion of the reservoir tapped by Early Phase 1 fissures must have been
689 sufficiently thermally buffered that minimal fractionation took place over ~60 years. Relatively slow
690 cooling rates within a thermally buffered magma body are consistent with ~500 year residence times
691 inferred for the early 1955 lavas by Cooper et al. (2001). Alternatively, the 1955E reservoir may have
692 been topped up by more MgO-rich magma in the later phases of the 1955 eruption or the 1960
693 eruption (Fig. 9b).

694 The larger amount of fractional crystallization experienced by F17 melts erupting on May 14th and
695 the dacitic magma encountered during drilling in 2005 compared to melts erupted between May 3-
696 9th is intriguing. The similarity of melt inclusion saturation pressures and drilling depths between all
697 three samples, and similarity of Nb/Y ratios between Early Phase 1 and F17 from these samples
698 indicate that the melts supplying all three bodies had a similar parent magma, and were stored at a
699 similar depth. We suggest that the F17 dacitic melt may formed on the periphery of the larger
700 magma body located in this region, allowing more cooling to occur in a specific amount of time (Fig.
701 9d). This interpretation is supported by the fact that F17 is offset ~220 m NE from the trend defined
702 by other fissures. Pockets of dacitic melt may also have formed within regions of enhanced
703 hydrothermal cooling. Deconvolving the nature of the evolution of this complex LERZ magma body
704 in space and time will likely require detailed thermal modelling, and isotopic data to provide a more
705 precise way to identify different magma batches with more resolution than can be achieved using
706 incompatible trace element ratios.

707 5. Global occurrence of more silicic melts in basalt-dominated 708 volcanic region

709 Analyses of erupted matrix glasses and melt inclusions from the 2018 eruption demonstrate that
710 extensive crystallization of an ocean island basalt causes substantial enrichment in incompatible
711 volatile and trace element species (Cl, F, H₂O). In particular, the increase in melt H₂O contents

712 combined with a rapid increase in melt viscosity following Fe-Ti oxide fractionation means that these
713 silicic magmas have the potential to display more explosive eruption styles relative to the more
714 typical Hawaiian style exhibited by basaltic to basaltic-andesite magmas.

715 The eruption of dacitic melt in 2018 at Kīlauea provides support to the growing volume of literature
716 indicating that andesitic-rhyolitic magmas are more common than we thought in predominantly
717 basalt-dominated volcanic regions. For example, Stock et al. (2020) report low An plagioclase crystals
718 (down to An₄₈) in basaltic lava and reticulite dominated by a higher An plagioclase population (An₇₈₋
719 ₈₂) from the 2015 eruption of Volcan Wolf in the Galapagos Archipelego. They also identify low An
720 plagioclase crystals in gabbroic nodules from the 1968 Fernandina eruption. Fractionation models
721 indicate that these plagioclase crystals must have grown from basaltic trachy-andesite and trachy-
722 andesitic melts. Additionally, resorbed quartz has been identified in tephra from the 2015 eruption
723 of Wolf. This indicates that the subvolcanic plumbing system at these two volcanoes which erupt
724 monotonous basaltic compositions must contain highly fractionated, silicic melts at depth not yet
725 observed at the surface.

726 Rhyolites comprise approximately 10% of erupted lavas in Iceland (Jónasson, 2007), and have been
727 noted in a variety of settings among mid-oceanic ridges (Wanless et al., 2010). While the continuum
728 of melt inclusion major element contents from basalt to dacite, and good fit of major, volatile and
729 trace element contents to MELTS fractional crystallization models indicate that Kīlauea dacites result
730 from extensive fractional crystallization, silicic magma occurrences in Iceland and MOR have been
731 variably attributed to fractional crystallization, crystal-melt segregation, and partial melting of
732 hydrothermally altered crust based on isotopic, volatile, major and trace element data (Elders et al.,
733 2011; Geist et al., 2021; Masotta et al., 2018; Rooyakkers et al., 2021a; Wanless et al., 2010).

734 Interestingly, Rooyakkers et al. (2021b) highlight the fact that silicic magmas have been encountered
735 4 times in the last two decades years during borehole drilling: dacite at Kīlauea (Teplow et al., 2009),
736 trachyte in Menegai, Kenya (Mbia et al., 2015), and rhyolite on two occasions at Krafla (Elders et al.,
737 2011; Mortensen et al., 2010). While the Mengeai and Krafla drilling encounters were a little less
738 surprising in terms of the composition encountered compared to Kīlauea (because rhyolitic and
739 syenetic magmas are better represented in surface deposits at these locations), It is noteworthy that
740 all of these encounters intercepted silicic magmas at very similar depths (2488 m at Kīlauea, ~2000
741 m in Kenya, 2571 m and 2104 m at Krafla), all in areas of active hydrothermal exploration. Although
742 these silicic magmas may be generated by different mechanisms, all require high heat fluxes at
743 shallow levels, either to melt/rejuvenate existing material, or to allow fractional crystallization to
744 proceed without the magma simply freezing in the crust. At Kīlauea, this high heat flow likely results
745 from the focus of repeated intrusion over hundreds of years in this particular section of the rift zone,
746 perhaps resulting from a “jog” in the rift zone, with two offset branches creating a local extension
747 stress field (Kenedi et al., 2010). This is somewhat analogous to the higher focusing of melt, and
748 therefore higher occurrence of silicic magma in overlapping spreading centers of mid-oceanic ridges
749 (Kent et al., 2000).

750 While 2018 marked the first documented occurrence of dacitic melt erupting at the surface, given
751 that 70% of Kīlauea’s surface is <500 yrs old, and 90% is <1100 yrs old (Holcomb, 1987), these melts
752 could erupt on a centurial-millennial basis, and still be absent in surface exposures. This is particularly
753 true given that dacitic melts are most likely to form in parts of the volcano characterised by high
754 intrusion rates, to maintain the high heat fluxes at shallow levels required for melts to fractionate to
755 dacitic compositions without freezing in the crust. In turn, high intrusion rates mean these sites are
756 most likely to be rapidly resurfaced by volcanic activity. For example, 75% of the surface of the

757 lower-east rift zone is <500 yrs old (Moore, 1992) . If magmas with silicic compositions were able to
758 evade sampling at Kīlauea, one of the world’s best-studied ocean island volcanoes, prior to drilling in
759 2005, it seems highly likely that these compositions are present at other ocean island volcanoes.
760 Specifically, based on the need for high heat flow to generate silicic magma bodies, and the
761 increasing number of drilling encounters, the anomalously hot areas of ocean islands are not only
762 the most likely to contain more silicic melts at shallow depths, but will also be the preferred targets
763 for hydrothermal exploration.

764 When the dacitic body was drilled at Kīlauea in 2005 at 2488 m depth, it was noted that the glass
765 did not show vesiculation (Teplow et al 2009), leading to inferences that the magma was volatile-
766 poor. Given that our melt inclusion measurements demonstrate that F17 melts were H₂O-rich, we
767 instead hypothesize that the low vesicularity resulted from the fact that a relatively small mass of
768 volatiles is exsolved during fractionation from a basalt to dacite (~0.04 wt%) at this depth, meaning
769 there is limited potential for a large, gas-rich cap to develop on top of these bodies (Fig. 5a). Upon
770 ascent, significant quantities of H₂O only begin to exsolve from a dacitic melt composition at ~300
771 bars. Thus, as long as the drilling fluid quenches these silicic melts at higher confining pressures,
772 there would be limited potential for large amounts of volatile exsolution and catastrophic well
773 blowout (Teplow et al., 2009). However, if allowed to rise to shallow levels prior to quenching, silicic
774 melts could present a hazard to geothermal drilling in basalt-dominated settings.

775 In addition to being encountered during drilling, it appears that the eruption of silicic melts at the
776 surface may be closely coupled to the injection of basaltic dikes into the region. At Kīlauea, based on
777 the limited evidence for mixing between the dacite and the dike-supplied magma in the initial
778 products of F17, we suggest that dike injection increased overpressure or changed the stress state in
779 the magma body, causing the eruption. Similarly, Rooyakkers et al., (2021b) examine the rhyolitic
780 magma producing the Víti maar at Krafla, Iceland, and conclude that these melts show very little
781 textural or chemical evidence for mingling/mixing with the basaltic “trigger”. This contrasts with
782 explosive eruptions like Askja, 1875 which show clear evidence textural evidence for pre-eruptive
783 magma mixing (Sparks et al., 1977). Clearly, the potential of silicic magma bodies to generate
784 explosive eruptions should be considered during hazard assessments in a wide variety of ocean
785 islands and other basalt-dominated settings (Rooyakkers et al., 2021b), both in terms of natural
786 eruptions initiated by a dike, and drilling encounters.

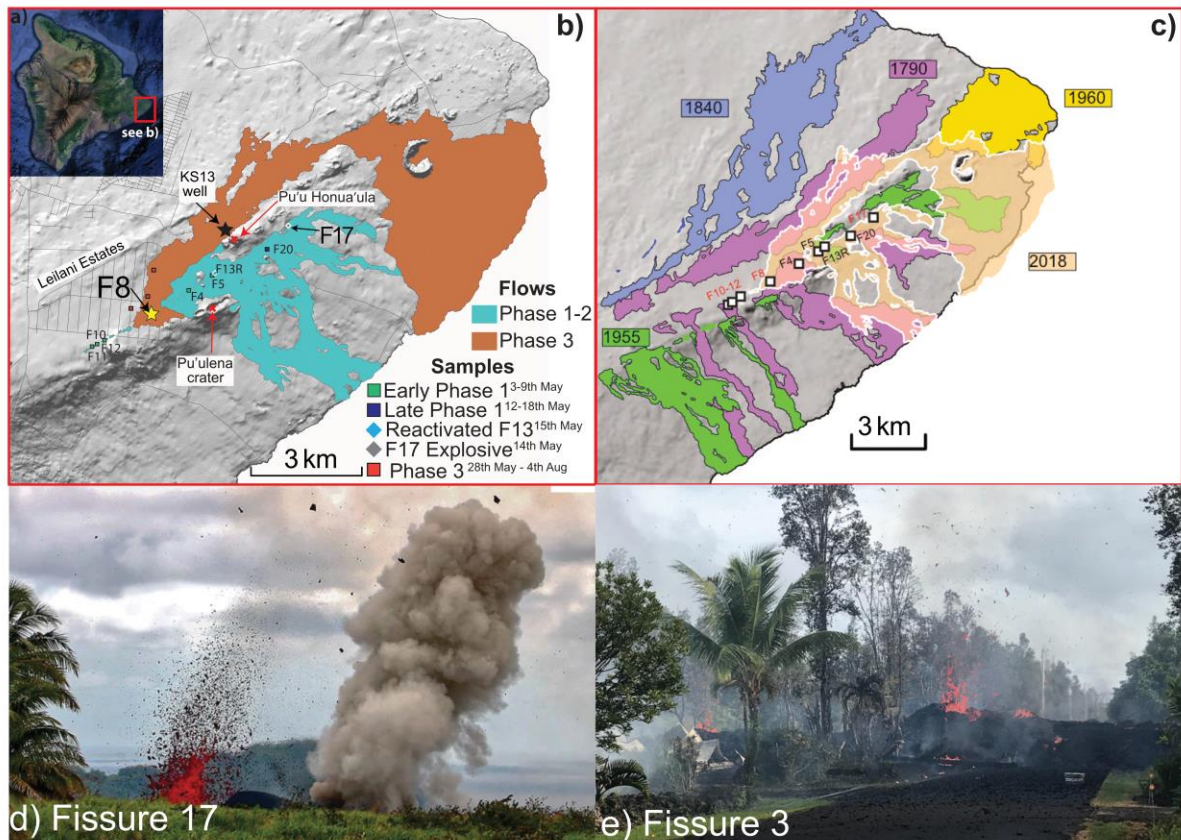
787 6. Conclusions

788 Basaltic to andesitic magmas erupted during the first two weeks of the 2018 eruption (3rd – 16th
789 May) were likely formed from variable amounts of crystallization of a magma body at ~2-3 km depth
790 within the LERZ. The similarity in Nb/Y ratios in melt inclusions from these early eruptive fissures and
791 a range of host crystal chemistry implies that crystallization proceeded from a single magma parent
792 body (or a series of bodies with near-identical trace element chemistry). Comparison of Nb/Y ratios
793 to historic LERZ lavas indicates that melts erupted in 1955-1960 are the most likely parent to the
794 early 2018 lavas (see also Helz, 2008; Pietruszka et al., 2021; Teplow et al., 2009). Extensive
795 crystallization of a section of this larger magma body (perhaps on the periphery or in a region with
796 enhanced hydrothermal cooling) produced a dacitic melt composition highly enriched in
797 incompatible elements such as Cl, F, Zr, and H₂O. Combined with an increase in magma viscosity with
798 increasing SiO₂ content and dropping temperatures, this H₂O-enrichment accounts for the explosive
799 strombolian behavior exhibited by the eruptive fissure tapping this melt (F17) without requiring
800 external sources of volatiles such as groundwater. Although the high viscosities of these dacitic melts
801 mean they are unlikely to erupt spontaneously, the 2018 eruption shows that they may be triggered

802 by the injection of basaltic intrusion. The volatile-rich nature of these bodies should also be
803 considered when prospecting for geothermal wells; if the melt erupted at F17 had been drilled and
804 allowed to depressurize to ~300 bars prior to quenching, large amounts of vesiculation may occur.

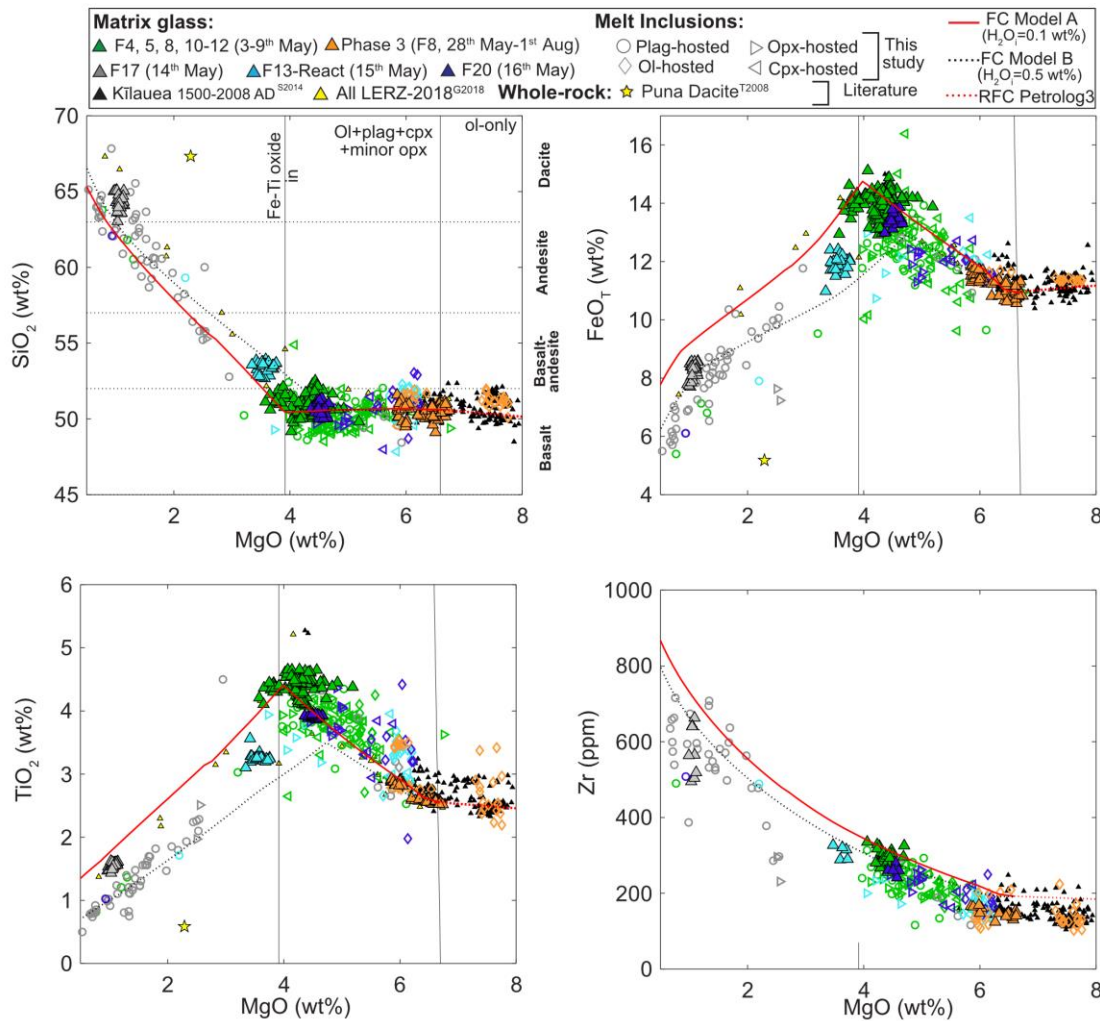
805 Figures

806 (see attached pdfs for higher resolution figs)



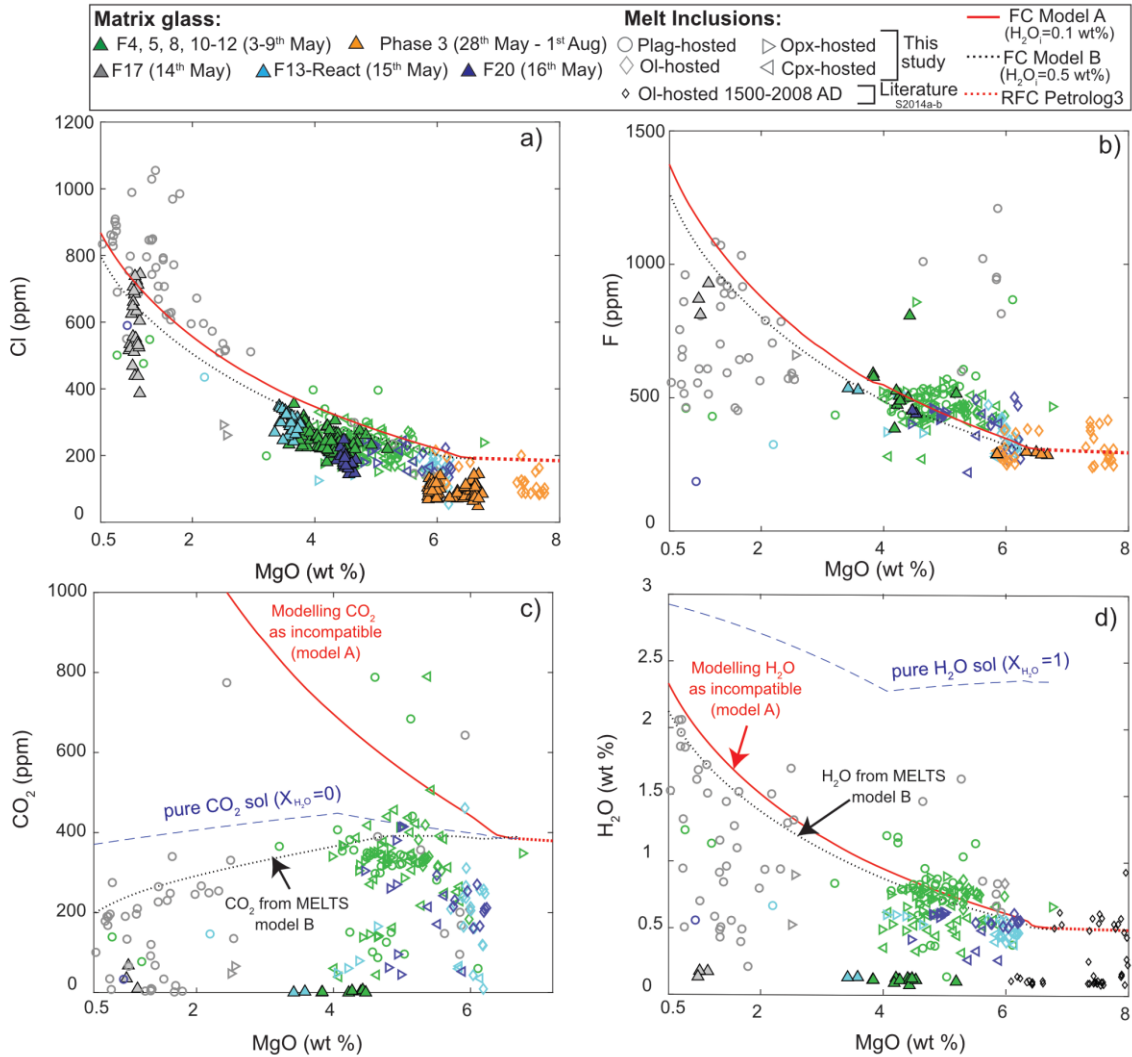
807

808 Figure 1 – a) Map of the island of Hawai'i, with the area inside the red box expanded in b). b)
809 Hillshade map from DEM of the Lower East Rift Zone showing the fissures and lava flows produced
810 during the 2018 eruption. Locations of samples analysed in this study are shown with colored
811 symbols. Ahu'ailā'au (F8) is marked with a yellow star, and the location of borehole KS13 where a
812 dacitic magma was drilled in 2005 is shown with a black star. Pu'u honua'ula and Pu'ulena Crater are
813 marked with red crosses (reference points for the position of the seismic anomaly). c) Map showing
814 the location of previous eruptive episodes on the LERZ, with the 2018 flows overlain in pale pink. d)
815 F17 showing the difference in eruption style between the basaltic-andesite eastern end (low
816 fountaining and spattering, LHS), and the more silicic andesitic western end (strombolian explosions
817 and gas jetting, RHS). e) Low spatter mounds produced during the first day of the eruption at F3.
818 Photos in d and e) from USGS. Part b) was adapted from Patrick et al., (2019), and c) from Pietruszka
819 et al., (2021), using geospatial data from Trusdell et al., 2006; Zoeller et al., 2020



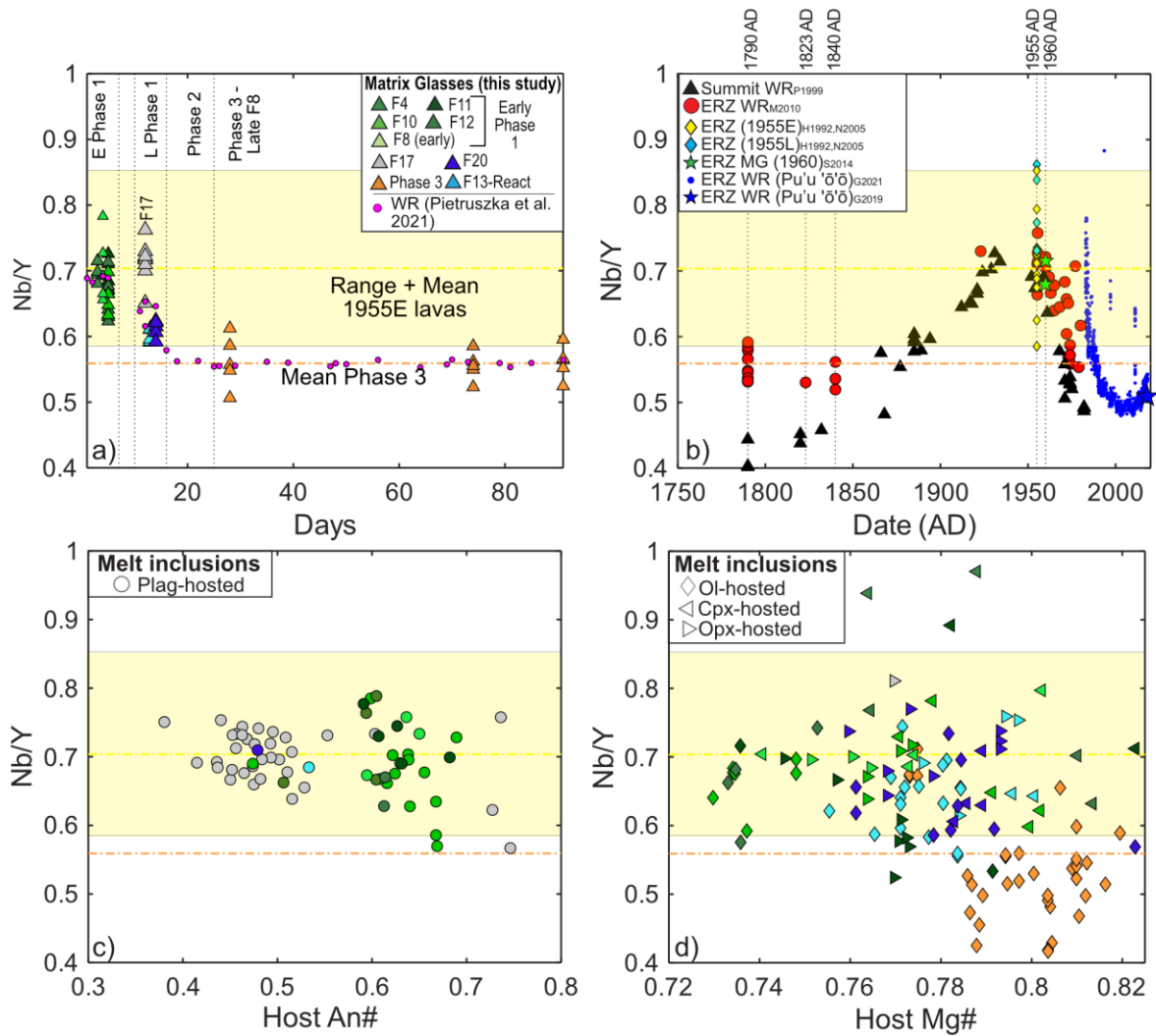
820

821 Figure 2- Major element systematics of matrix glasses and melt inclusions. The composition of
 822 olivine, clinopyroxene and 6 plagioclase melt inclusions are corrected for the effects of PEC.
 823 Literature data for matrix glasses erupted during the 2018 eruption (yellow triangles, Gansecki et al.,
 824 2019) and previous Kilauea eruptions (black triangles, Sides et al., 2014a, 2014b; Wieser et al., 2019)
 825 are shown, along with the whole-rock composition of the dacitic magma drilled in 2005 (Teplow et
 826 al., 2009). Two fractional crystallization models run in alphaMELTS for MATLAB (Rhyolite-MELTS
 827 v1.2.0, Antoshechkina and Ghiorso, 2018; Ghiorso and Gualda, 2015; Gualda et al., 2012) are shown,
 828 with $Fe^{3+}/Fe_{T, initial}=0.15$ (unbuffered), $P=650$ bars. Model A uses an initial H_2O content of 0.1 wt%
 829 (red line, best recreates the onset of Fe-Ti oxide saturation), while Model B uses a more realistic
 830 initial H_2O content of 0.5 wt% (black dashed line). As the rhyolite-MELTS model fails to recreate the
 831 olivine-only fractionation assemblage at >6.8 wt% (Wright and Fiske, 1971), the red dashed line
 832 shows a reverse fractionation crystallization (RFC) model run from the starting composition in
 833 Petrolog3.1.3 (Danyushevsky and Plechov, 2011). The fractionation trajectory for Zr is calculated
 834 assuming incompatible behavior.



835

836 Figure 3– Volatile element systematics, with fractionation trajectories assuming perfect
 837 incompatibility for all elements from MELTS model A shown in red. Incompatible behavior is
 838 assumed for Cl and F in MELTS model B, while CO₂ and H₂O solubility in equilibrium with a mixed
 839 fluid are tracked in the solubility model MagmaSat integrated into MELTS (dashed black line). The
 840 dashed red section of the line shows reverse crystallization of olivine from the starting composition
 841 in Petrolog3. For Cl and F, models A and B produce very similar trajectories, as these models predict
 842 similar amounts of solid fractionated. The solubility of pure CO₂ and H₂O calculated using the
 843 MagmaSat solubility model (Ghiorso and Gualda, 2015) implemented in VESlcal (Iacovino et al.,
 844 2021) for a best-fit to the observed major element path (see Supporting Fig. S11) is shown with a
 845 blue dashed line.

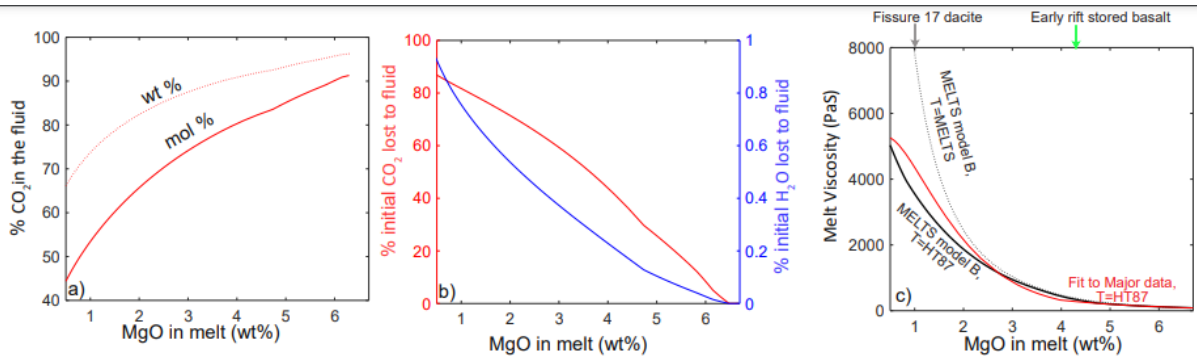


846

847

848 Figure 4– Ratios of incompatible trace elements in 2018 eruption products and previous LERZ
 849 eruptions. **a)** variations in Nb/Y in co-erupted matrix glasses (this study) with time throughout the
 850 eruption (May 3rd = Day 1). Whole-rock measurements from Pietruszka et al. (2021) are overlain. **b)**
 851 Compilation of Nb/Y ratios versus time for whole-rock measurements of summit lavas (black
 852 triangles, Pietruszka and Garcia, 1999), LERZ lavas from 1790, 1823, 1840, 1955, 1960 (red dots,
 853 Marske, 2010), early phases of the 1955 LERZ eruption (yellow colors), and later phases (cyan colors,
 854 Helz and Wright, 1992; Norman, 2005). Analyses of matrix glass from the 1960 Kapoho eruption
 855 from Sides et al., (2014b) and whole-rock measurements from Pu‘u‘ō‘ō are also shown (Gansecki et
 856 al., 2019; Garcia et al., 2021). The mean composition of the early 1955 lavas and Phase 3 matrix
 857 glasses are shown with a yellow and orange dashed line on all plots, with the range for 1955 shown
 858 in light yellow. **c)** Nb/Y ratios in plagioclase-hosted melt inclusions vs. host anorthite content. **d)**
 859 Nb/Y ratios vs. host Mg# for olivine, orthopyroxene and clinopyroxene-hosted melt inclusions.
 860 Symbols in part c-d are colored by fissure as in part b.

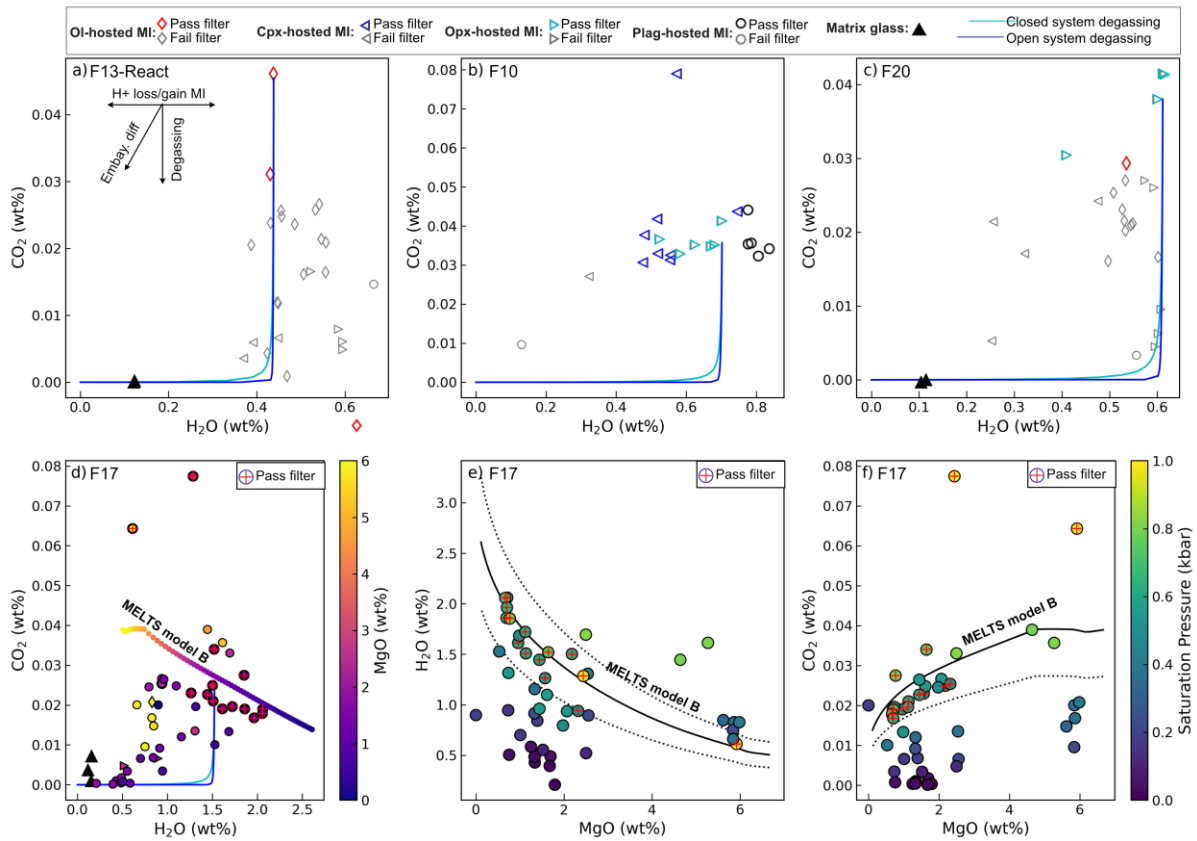
861



862

863

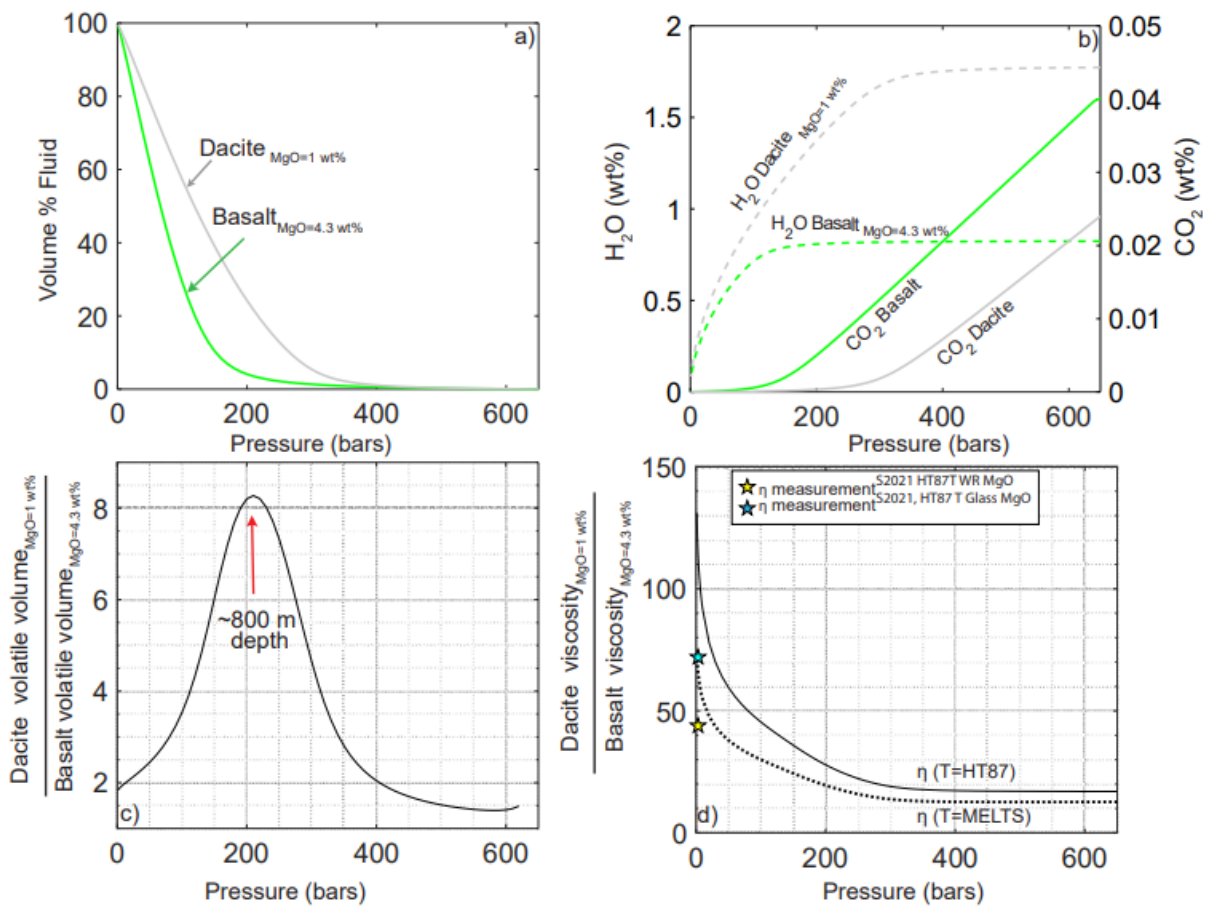
864 Figure 5- Volatile exsolution and changes in viscosity during fractional crystallization tracked in
 865 MELTS model B. **a)** % of CO₂ in the exsolved fluid (by wt% and mol%) decreases with progressive
 866 fractionation. **b)** % of the initial amount of CO₂ (red solid line) and H₂O (blue solid line) lost to the
 867 fluid phase during fractionation. **c)** Changes in viscosity during fractionation. The black dashed line
 868 shows viscosity from MELTS model B using Giordano et al., (2008) and the MELTS temperature, and
 869 the black solid line shows MELTS model B with temperatures calculated from the MELTS MgO
 870 content using the Helz and Thornber (1987) MgO thermometer. The red line shows the viscosity
 871 calculated using Giordano et al., (2008) for the best-fit model to measured major elements
 872 (Supporting Fig. S11), with temperatures calculated using Helz and Thornber (1987), and H₂O
 873 contents estimated for a parameterization of the relationship between H₂O and MgO shown as a red
 874 line in Fig. 3d.



875

876 Fig. 6. a-c) Example degassing paths for three fissures, with open and closed system degassing paths
 877 calculated using MagmaSat (Ghiorso and Gualda, 2015) in VESlcal (Iacovino et al., 2021) for specific
 878 melt inclusions overlain. Vectors for different processes are shown in a). d) Degassing path for F17
 879 with melt inclusions colored by their MgO content. The trajectories caused by degassing (blue and
 880 cyan lines) and fractional crystallization (yellow-purple line) are overlain. e-f) Example of the filtering
 881 process used to calculate saturation pressures for F17. Only melt inclusions with CO₂ contents less
 882 than 30% below the MELTS fractionation line (filtering out degassing), and only H₂O contents within
 883 +30% are used (filtering out degassing and H⁺ re-equilibration to higher and lower values). Symbols
 884 are colored by their saturation pressure, and are marked with a red cross if they pass both filters.

885



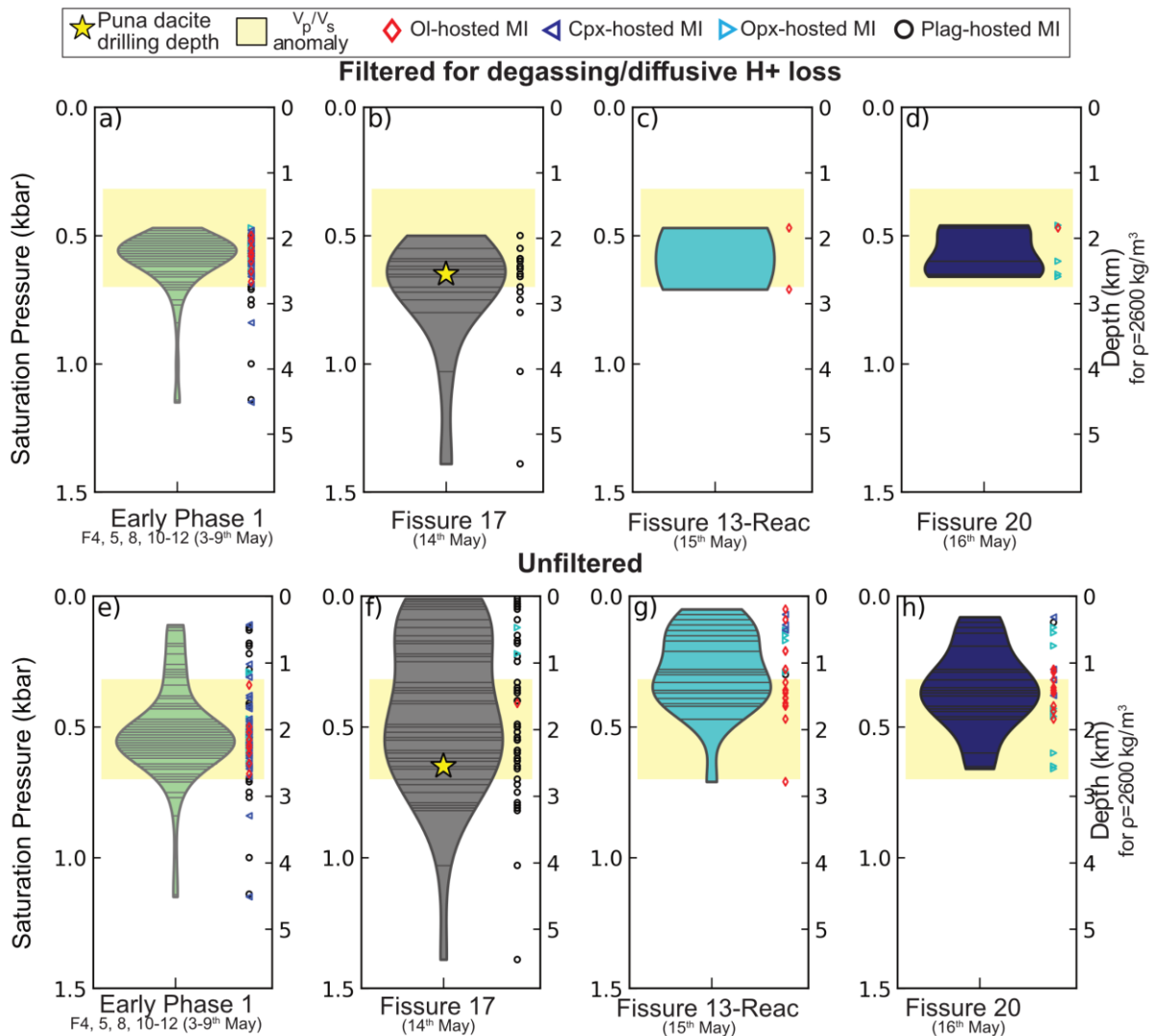
886

887 Figure 7. Volatile exsolution and changes in viscosity during ascent to the surface along a closed
 888 system degassing path for a basaltic andesite (BA) melt (~4.3 wt% MgO, average composition of
 889 Early Phase 1 matrix glasses), and a dacitic melt composition (MgO=1 wt%, average composition of
 890 F17 matrix glasses). **a)** Volume of the exsolved fluid phase relative to the total volume of the system
 891 for both melt compositions with decreasing pressure. **b)** Concentration of H₂O (LH axis, dashed
 892 lines), and CO₂ in the melt (RH- solid lines) during ascent. **c-d)** Comparison of the volume of volatiles
 893 and viscosity during ascent of the dacitic and basaltic andesite melt. **c).** The volume of exsolved
 894 volatiles is ~8-9 times higher for the dacite at ~200 bars (800 m depth), because of the higher H₂O
 895 content of this melt, and the earlier exsolution of H₂O from this melt relative to the basalt (**b**). Even
 896 at the surface, the volume of exsolved volatiles is ~2 times higher for the dacite. **d)** The viscosity of
 897 the dacite divided by the viscosity of the basalt at each point on the ascent path calculated using the
 898 Giordano et al., 2008 model. The dashed line shows viscosity using the temperature from MELTS,

899 and the solid line shows temperature calculated from the MELTS MgO content using Helz and
 900 Thornber (1987). The yellow star shows the viscosity ratio between the dacitic F17 lavas and the
 901 basaltic/basaltic-andesite Early phase 1 lavas reported by Soldati and Dingwell, (2021), using
 902 temperatures they calculate from whole-rock MgO contents using Helz and Thornber (1987). The
 903 cyan star uses their viscosity parameters (A, B, C) to calculate viscosity at the temperatures for the
 904 MgO contents considered here for the dacitic melt (1 wt% MgO) and basaltic melt (4.3 wt% MgO)
 905 using Helz and Thornber (1987). The lower MgO content of our F17 glass relative to their whole-rock
 906 measurement result in a lower temperature, so a higher viscosity.

907

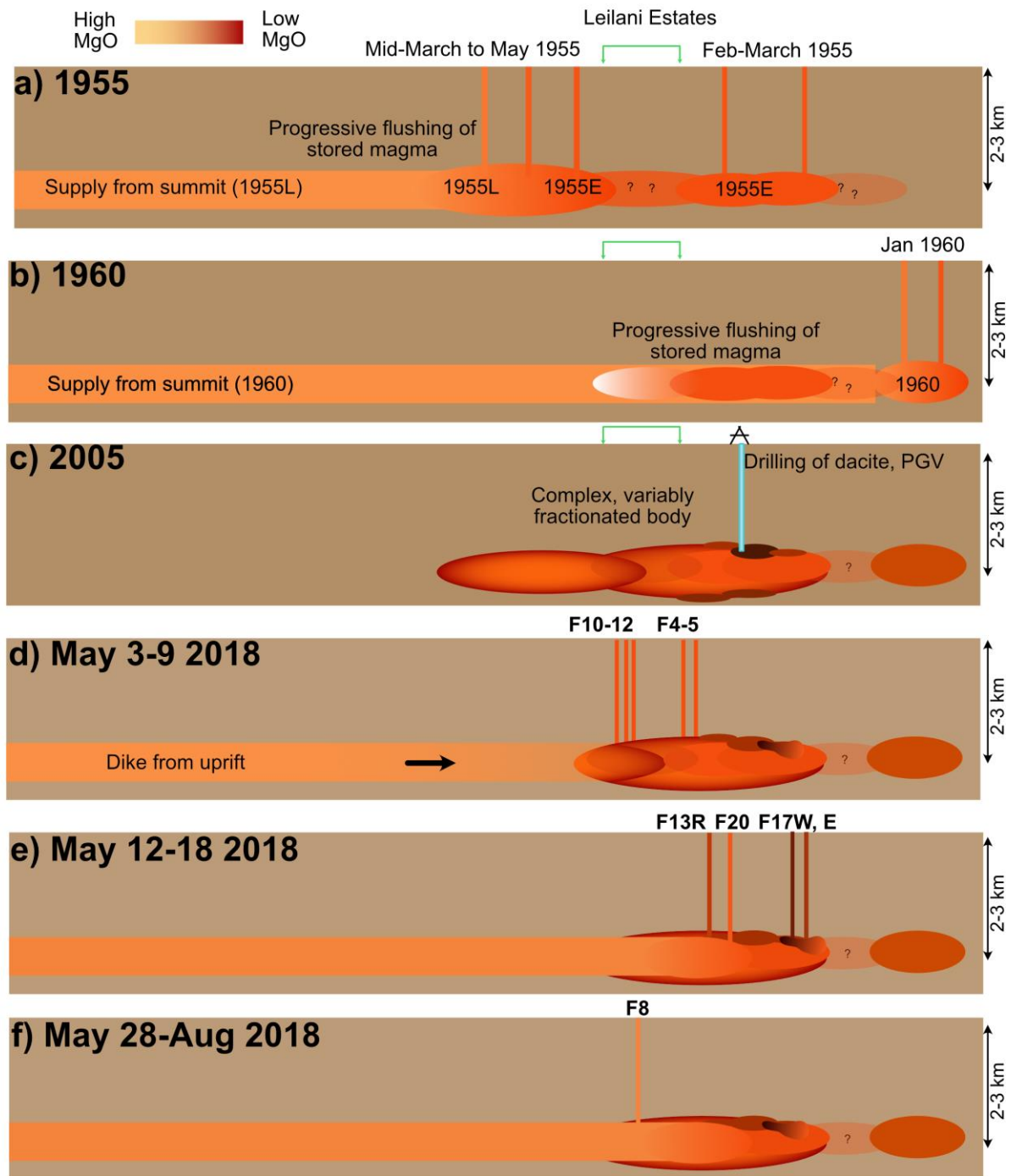
908



909

910 Figure 8. Magma storage depths for different early fissures from the 2018 eruption. A-h) Violin plots
 911 showing saturation pressures calculated using the solubility model MagmaSat (Ghiorso and Gualda,
 912 2015) implemented through VESlcal (Iacovino et al. 2021). A-d) shows filtered melt inclusions (within
 913 $\pm 30\%$ of MELTS model B for H₂O, and -30% for CO₂, see Fig. 6e-f). e-h) shows all melt inclusions. Each
 914 grey line on the violin plot represents a melt inclusion, and the width of the “violin” represents the
 915 clustering of data at each pressure. On the RHS, the individual saturation depths for each inclusion
 916 are shown with symbol shapes and colors corresponding to the host phase. The depth at which the

917 dacitic magma was drilled at PGV is shown as a yellow star, and the depth of the low Vp/Vs region
 918 (low of Vp/Vs=1.66 vs. 1.78 background) imaged by Cooper and Dustman (1995) is shown in yellow.



919
 920 Figure 9 – Schematic diagram showing evolution of the LERZ in the vicinity of the 2018 eruption site.
 921 A) More MgO-poor stored melts (1955E) erupt between Feb-March 1955 very close to the 2018
 922 eruption sites (Fig. 1c). More MgO-rich melts (1955L) erupt slightly uprift in mid-March to May. B)
 923 Activity farther down the rift zone in 1960 initiates with the eruption of more MgO-poor (1955-like)
 924 melts, followed by flushing by more MgO-rich summit magma. C) Hydrothermal drilling in 2005 taps
 925 a dacitic body, likely located on the periphery of a complex, variably fractionated magma body. D)
 926 Hydraulic pressure from a dike propagating downrift from Pu‘u‘ō‘ō forces stored melts to the

927 surface. E) Progressively, this dike-supplied material begins to mix with stored melts, until by May
928 28th, this new component dominates.

929 Acknowledgements

930 PW acknowledges funding from a NERC DTP studentship (NE/ L002507/1), a NERC Edinburgh Ion
931 Microprobe Facility grant (IMF675/1118), and a Cambridge University Leave to Work away grant. PA
932 acknowledges funding for development of alphaMELTS for MATLAB by NSF through grant EAR-
933 1947616. We also acknowledge support from the National Science Foundation ICER-20-26904,
934 granted to OFM Research (PI. Ghiorso) for maintaining the ENKI server used to perform VESlcal
935 calculations. PW thanks Matthew Gleeson for help running MELTS for MATLAB, including providing
936 access to his high-powered desktop. PW thanks Iris Buisman and Guilio Lampronti for help collecting
937 EPMA and SEM data for this study, Rod Kindel for the intrepid hike to Fissures 10, 11 and 12 in his
938 backyard, Sarah Allen for help sampling these fissures, and Evgenia Ilyinskaya for coordinating the
939 2018 and 2019 field campaign. We thank Emily Johnson and two anonymous reviewers, along with
940 associate editor Ken Rubin for very helpful and detailed comments which greatly improved this
941 manuscript.

942 Data has been archived on Zenodo (<https://zenodo.org/badge/latestdoi/417367063>) and on Github
943 (where example MELTS for Matlab codes are also provided- [https://github.com/PennyWieser/G3-
944 Kil2018-Evolved-MIs](https://github.com/PennyWieser/G3-Kil2018-Evolved-MIs)).

945 Bibliography

- 946 Antoshechkina, P., Ghiorso, M.S., 2018. MELTS for MATLAB: A new educational and research tool for
947 computational thermodynamics. AGU.
- 948 Bauer, G.R., Fodor, R.V., Husler, J.W., Keil, K., 1973. Contributions to the mineral chemistry of
949 Hawaiian rocks: III. Composition and mineralogy of a new rhyodacite occurrence on Oahu, Hawaii.
950 *Contr. Mineral. and Petrol.* 40, 183–194. <https://doi.org/10.1007/BF00373783>
- 951 Cooper, K.M., Reid, M.R., Murrell, M.T., Clague, D.A., 2001. Crystal and magma residence at Kilauea
952 Volcano, Hawaii: 230Th–226Ra dating of the 1955 east rift eruption. *Earth and Planetary Science
953 Letters* 184, 703–718. [https://doi.org/10.1016/S0012-821X\(00\)00341-1](https://doi.org/10.1016/S0012-821X(00)00341-1)
- 954 Cooper, P., Dustman, M., 1995. Geothermal research, monitoring and testing: final report,
955 geophysics subtask, microseismicity study. *Seismological Research Letters*.
- 956 Danyushevsky, L.V., Plechov, P., 2011. Petrolog3: Integrated software for modeling crystallization
957 processes. *Geochemistry, Geophysics, Geosystems* 12, n/a-n/a.
958 <https://doi.org/10.1029/2011GC003516>
- 959 Elders, W.A., Friðleifsson, G.Ó., Zierenberg, R.A., Pope, E.C., Mortensen, A.K., Guðmundsson, Á.,
960 Lowenstern, J.B., Marks, N.E., Owens, L., Bird, D.K., Reed, M., Olsen, N.J., Schiffman, P., 2011. Origin
961 of a rhyolite that intruded a geothermal well while drilling at the Krafla volcano, Iceland. *Geology* 39,
962 231–234. <https://doi.org/10.1130/G31393.1>
- 963 Gaetani, G.A., O’Leary, J.A., Shimizu, N., Bucholz, C.E., Newville, M., 2012. Rapid reequilibration of
964 H₂O and oxygen fugacity in olivine-hosted melt inclusions. *Geology* 40, 915–918.
965 <https://doi.org/10.1130/G32992.1>

**This is a preprint submitted to EarthArxiv.
It has been accepted in Geochemistry, Geophysics and Geosystems (Jan, 2022)**

- 966 Gansecki, C., Lee, R.L., Shea, T., Lundblad, S.P., Hon, K., Parcheta, C., 2019. The tangled tale of
967 Kīlauea's 2018 eruption as told by geochemical monitoring. *Science* 366, eaaz0147.
968 <https://doi.org/10.1126/science.aaz0147>
- 969 Garcia, M.O., Pietruszka, A.J., Norman, M.D., Rhodes, J.M., 2021. Kīlauea's Pu'u Ō'ō Eruption (1983–
970 2018): A synthesis of magmatic processes during a prolonged basaltic event. *Chemical Geology* 581,
971 120391. <https://doi.org/10.1016/j.chemgeo.2021.120391>
- 972 Geist, D., Harpp, K., Oswald, P., Wallace, P., Bindeman, I., Christensen, B., 2021. Hekla Revisited:
973 Fractionation of a Magma Body at Historical Timescales. *Journal of Petrology* 62, egab001.
974 <https://doi.org/10.1093/petrology/egab001>
- 975 Gerlach, T.M., McGee, K.A., Elias, T., Sutton, A.J., Doukas, M.P., 2002. Carbon dioxide emission rate
976 of Kīlauea Volcano: Implications for primary magma and the summit reservoir. *Journal of*
977 *Geophysical Research: Solid Earth* 107, ECV 3-1-ECV 3-15. <https://doi.org/10.1029/2001JB000407>
- 978 Ghiorso, M.S., Gualda, G.A.R., 2015. An H₂O–CO₂ mixed fluid saturation model compatible with
979 rhyolite-MELTS. *Contrib Mineral Petrol* 169, 53. <https://doi.org/10.1007/s00410-015-1141-8>
- 980 Giordano, D., Russell, J.K., Dingwell, D.B., 2008. Viscosity of magmatic liquids: A model. *Earth and*
981 *Planetary Science Letters* 271, 123–134. <https://doi.org/10.1016/j.epsl.2008.03.038>
- 982 Gonnermann, H.M., Manga, M., 2013. Chapter 4 - Dynamics of magma ascent in the volcanic
983 conduit, in: *Modelling Volcanic Processes. The Physics and Mathematics of Volcanism*.
- 984 Greaney, A.T., Rudnick, R.L., Helz, R.T., Gaschnig, R.M., Piccoli, P.M., Ash, R.D., 2017. The behavior of
985 chalcophile elements during magmatic differentiation as observed in Kilauea Iki lava lake, Hawaii.
986 *Geochimica et Cosmochimica Acta* 210, 71–96. <https://doi.org/10.1016/j.gca.2017.04.033>
- 987 Gualda, G.A.R., Ghiorso, M.S., Lemons, R.V., Carley, T.L., 2012. Rhyolite-MELTS: a Modified
988 Calibration of MELTS Optimized for Silica-rich, Fluid-bearing Magmatic Systems. *Journal of Petrology*
989 53, 875–890. <https://doi.org/10.1093/petrology/egr080>
- 990 Helz, R.T., 2008. How to Produce Dacitic Melt at Kilauea: Evidence from Historic Kilauea Lava.
991 Presented at the AGU.
- 992 Helz, R.T., 1980. Crystallization history of Kilauea Iki lava lake as seen in drill core recovered in 1967–
993 1979. *Bull Volcanol* 43, 675–701. <https://doi.org/10.1007/BF02600365>
- 994 Helz, R.T., Thornber, C.R., 1987. Geothermometry of Kilauea Iki lava lake, Hawaii. *Bulletin of*
995 *Volcanology* 49, 651–668. <https://doi.org/10.1007/BF01080357>
- 996 Helz, R.T., Wright, T.L., 1992. Differentiation and magma mixing on Kilauea's east rift zone: A further
997 look at the eruptions of 1955 and 1960. Part I. The late 1955 lavas. *Bulletin of Volcanology* 54, 361–
998 384. <https://doi.org/10.1007/BF00312319>
- 999 Holcomb, R.T., 1987. Eruptive History and Long-Term Behaviour of Kilauea Volcano.
- 1000 Houghton, B.F., Gonnermann, H.M., 2008. Basaltic explosive volcanism: Constraints from deposits
1001 and models. *Geochemistry* 68, 117–140. <https://doi.org/10.1016/j.chemer.2008.04.002>
- 1002 Houghton, B.F., Taddeucci, J., Andronico, D., Gonnermann, H.M., Pistolesi, M., Patrick, M.R., Orr,
1003 T.R., Swanson, D.A., Edmonds, M., Gaudin, D., Carey, R.J., Scarlato, P., 2016. Stronger or longer:

- 1004 Discriminating between Hawaiian and Strombolian eruption styles. *Geology* 44, 163–166.
1005 <https://doi.org/10.1130/G37423.1>
- 1006 Iacovino, K., Matthews, S., Wieser, P.E., Moore, G., Begue, F., 2021. VESical Part I: An open-source
1007 thermodynamic model engine for mixed volatile (H₂O-CO₂) solubility in silicate melt. *Earth and*
1008 *Space Science*. <https://doi.org/10.1029/2020EA001584>
- 1009 Jaupart, C., Vergnolle, S., 1988. Laboratory models of Hawaiian and Strombolian eruptions. *Nature*
1010 333, 58–60.
- 1011 Jenner, F.E., Hauri, E.H., Bullock, E.S., König, S., Arculus, R.J., Mavrogenes, J.A., Mikkelsen, N.,
1012 Goddard, C., 2015. The competing effects of sulfide saturation versus degassing on the behavior of
1013 the chalcophile elements during the differentiation of hydrous melts. *Geochemistry, Geophysics,*
1014 *Geosystems* 16, 1490–1507. <https://doi.org/10.1002/2014GC005670>
- 1015 Johnson, E.A., Rossman, G.R., 2013. The diffusion behavior of hydrogen in plagioclase feldspar at
1016 800–1000 C: Implications for re-equilibration of hydroxyl in volcanic phenocrysts. *American*
1017 *Mineralogist* 98, 1779–1787. <https://doi.org/10.2138/am.2013.4521>
- 1018 Jónasson, K., 2007. Silicic volcanism in Iceland: Composition and distribution within the active
1019 volcanic zones. *Journal of Geodynamics* 43, 101–117. <https://doi.org/10.1016/j.jog.2006.09.004>
- 1020 Kenedi, C., Shalev, E., Lucas, A., Malin, P., 2010. Microseismicity and 3-D Mapping of an Active
1021 Geothermal Field, Kilauea Lower East Rift Zone, Puna, Hawaii. Presented at the Proceedings World
1022 Geothermal Congress 2010.
- 1023 Kent, G.M., Singh, S.C., Harding, A.J., Sinha, M.C., Orcutt, J.A., Barton, P.J., White, R.S., Bazin, S.,
1024 Hobbs, R.W., Tong, C.H., Pye, J.W., 2000. Evidence from three-dimensional seismic re⁻activity images
1025 for enhanced melt supply beneath mid-ocean -ridge discontinuities 406, 5.
- 1026 Koleszar, A.M., Saal, A.E., Hauri, E.H., Nagle, A.N., Liang, Y., Kurz, M.D., 2009. The volatile contents of
1027 the Galapagos plume; evidence for H₂O and F open system behavior in melt inclusions. *Earth and*
1028 *Planetary Science Letters* 287, 442–452. <https://doi.org/10.1016/j.epsl.2009.08.029>
- 1029 Lerner, A.H., Wallace, P., Shea, T., 2021. The petrologic and degassing behavior of sulfur and other
1030 magmatic volatiles from the 2018 eruption of Kīlauea, Hawai'i: melt concentrations, magma storage
1031 depths, and magma recycling. *Bulletin Volcanology* 83:43, 1–32.
- 1032 Llewellyn, E.W., Mader, H.M., Wilson, S.D.R., 2002. The rheology of a bubbly liquid. *Proc. R. Soc.*
1033 *Lond. A* 458, 987–1016. <https://doi.org/10.1098/rspa.2001.0924>
- 1034 Lowenstern, J., 2001. Carbon dioxide in magmas and implications for hydrothermal systems.
1035 *Mineralium Deposita* 36, 490–502. <https://doi.org/10.1007/s001260100185>
- 1036 Macdonald, G.A., 1962. The 1959 and 1960 eruptions of Kilauea volcano, Hawaii, and the
1037 construction of walls to restrict the spread of the lava flows. *Bull Volcanol* 24, 249–294.
1038 <https://doi.org/10.1007/BF02599351>
- 1039 Manga, M., Loewenberg, M., 2001. Viscosity of magmas containing highly deformable bubbles.
1040 *Journal of Volcanology and Geothermal Research* 105, 19–24. [https://doi.org/10.1016/S0377-0273\(00\)00239-0](https://doi.org/10.1016/S0377-0273(00)00239-0)
- 1042 Marske, J.P., 2010. Magmatic History Of Lavas from Kīlauea Volcano, Hawai'i and South Pagan
1043 Volcano, Northern Mariana Islands. PhD Thesis, UH Manoa.

- 1044 Masotta, M., Mollo, S., Nazzari, M., Tecchiato, V., Scarlato, P., Papale, P., Bachmann, O., 2018.
1045 Crystallization and partial melting of rhyolite and felsite rocks at Krafla volcano: A comparative
1046 approach based on mineral and glass chemistry of natural and experimental products. *Chemical*
1047 *Geology* 483, 603–618. <https://doi.org/10.1016/j.chemgeo.2018.03.031>
- 1048 Mbia, P., Mortensen, A., Oskarsson, N., Hardarson, B., 2015. Sub-surface geology, petrology and
1049 hydrothermal alteration of the Menengai geothermal field, Kenya: Case study of wells MW-02, MW-
1050 04, MW-06 and MW-07. Presented at the World Geothermal Congress 2015.
1051 <https://www.geothermal-energy.org/pdf/IGStandard/WGC/2015/12071.pdf>.
- 1052 Moore, R.B., 1992. Volcanic geology and eruption frequency, lower east rift zone of Kilauea volcano,
1053 Hawaii. *Bulletin of Volcanology* 54, 475–483. <https://doi.org/10.1007/BF00301393>
- 1054 Mortensen, A.K., Gronvold, K., Gudmundsson, A., Steingrímsson, B., Egilson, T., 2010. Quenched
1055 Silicic Glass from Well KJ-39 in Krafla, North-Eastern Iceland, in: *Proceedings World Geothermal*
1056 *Congress 2010, Bali, Indonesia, 25-29 April 2010*.
- 1057 Moussallam, Y., Edmonds, M., Scaillet, B., Peters, N., Gennaro, E., Sides, I., Oppenheimer, C., 2016.
1058 The impact of degassing on the oxidation state of basaltic magmas: A case study of Kilauea volcano.
1059 *Earth and Planetary Science Letters* 450, 317–325. <https://doi.org/10.1016/j.epsl.2016.06.031>
- 1060 Neal, C.A., Brantley, S.R., Antolik, L., Babb, J.L., Burgess, M., Calles, K., Cappos, M., Chang, J.C.,
1061 Conway, S., Desmither, L., Dotray, P., Elias, T., Fukunaga, P., Fuke, S., Johanson, I.A., Kamibayashi, K.,
1062 Kauahikaua, J., Lee, R.L., Pekalib, S., Miklius, A., Million, W., Moniz, C.J., Nadeau, P.A., Okubo, P.,
1063 Parcheta, C., Patrick, M.R., Shiro, B., Swanson, D.A., Tollett, W., Trusdell, F., Younger, E.F., Zoeller,
1064 M.H., Montgomery-Brown, E.K., Anderson, K.R., Poland, M.P., Ball, J.L., Bard, J., Coombs, M.,
1065 Dietterich, H.R., Kern, C., Thelen, W.A., Cervelli, P.F., Orr, T., Houghton, B.F., Gansecki, C., Hazlett, R.,
1066 Lundgren, P., Diefenbach, A.K., Lerner, A.H., Waite, G., Kelly, P., Clor, L., Werner, C., Mulliken, K.,
1067 Fisher, G., Damby, D., 2018. The 2018 rift eruption and summit collapse of Kilauea Volcano. *Science*
1068 eaav7046. <https://doi.org/10.1126/science.aav7046>
- 1069 Neave, D.A., Hartley, M.E., Maclennan, J., Edmonds, M., Thordarson, T., 2017. Volatile and light
1070 lithophile elements in high-anorthite plagioclase-hosted melt inclusions from Iceland. *Geochimica et*
1071 *Cosmochimica Acta* 205, 100–118. <https://doi.org/10.1016/j.gca.2017.02.009>
- 1072 Norman, M., 2005. Trace-element distribution coefficients for pyroxenes, plagioclase, and olivine in
1073 evolved tholeiites from the 1955 eruption of Kilauea Volcano, Hawai'i, and petrogenesis of
1074 differentiated rift-zone lavas. *American Mineralogist* 90, 888–899.
1075 <https://doi.org/10.2138/am.2005.1780>
- 1076 Parfitt, E.A., Wilson, L., 1995. Explosive volcanic eruptions-IX. The transition between Hawaiian-style
1077 lava fountaining and Strombolian explosive activity. *Geophysical Journal International* 121, 226–232.
1078 <https://doi.org/10.1111/j.1365-246X.1995.tb03523.x>
- 1079 Patrick, M.R., Dietterich, H.R., Lyons, J.J., Diefenbach, A.K., Parcheta, C., Anderson, K.R., Namiki, A.,
1080 Sumita, I., Shiro, B., Kauahikaua, J.P., 2019. Cyclic lava effusion during the 2018 eruption of Kilauea
1081 Volcano. *Science* 366, eaay9070. <https://doi.org/10.1126/science.aay9070>
- 1082 Patrick, M.R., Houghton, B.F., Anderson, K.R., Poland, M.P., Montgomery-Brown, E., Johanson, I.,
1083 Thelen, W., Elias, T., 2020. The cascading origin of the 2018 Kilauea eruption and implications for
1084 future forecasting. *Nat Commun* 11, 5646. <https://doi.org/10.1038/s41467-020-19190-1>

- 1085 Pietruszka, A.J., Garcia, M.O., 1999a. A Rapid Fluctuation in the Mantle Source and Melting History
1086 of Kilauea Volcano Inferred from the Geochemistry of its Historical Summit Lavas (1790–1982).
1087 *Journal of Petrology* 40, 1321–1342.
- 1088 Pietruszka, A.J., Garcia, M.O., 1999b. The size and shape of Kilauea Volcano’s summit magma storage
1089 reservoir: a geochemical probe. *Earth and Planetary Science Letters* 167, 311–320.
1090 [https://doi.org/10.1016/S0012-821X\(99\)00036-9](https://doi.org/10.1016/S0012-821X(99)00036-9)
- 1091 Pietruszka, A.J., Garcia, M.O., Rhodes, J.M., 2021. Accumulated Pu’u ‘Ō’ō magma fed the voluminous
1092 2018 rift eruption of Kilauea Volcano: evidence from lava chemistry. *Bull Volcanol* 83, 59.
1093 <https://doi.org/10.1007/s00445-021-01470-3>
- 1094 Pietruszka, A.J., Heaton, D.E., Marske, J.P., Garcia, M.O., 2015. Two magma bodies beneath the
1095 summit of Kilauea Volcano unveiled by isotopically distinct melt deliveries from the mantle. *Earth
1096 and Planetary Science Letters* 413, 90–100. <https://doi.org/10.1016/j.epsl.2014.12.040>
- 1097 Portnyagin, M., Almeev, R., Matveev, S., Holtz, F., 2008. Experimental evidence for rapid water
1098 exchange between melt inclusions in olivine and host magma. *Earth and Planetary Science Letters*
1099 272, 541–552. <https://doi.org/10.1016/j.epsl.2008.05.020>
- 1100 Rooney, T.O., Hart, W.K., Hall, C.M., Ayalew, D., Ghiorso, M.S., Hidalgo, P., Yirgu, G., 2012.
1101 Peralkaline magma evolution and the tephra record in the Ethiopian Rift. *Contrib Mineral Petrol* 164,
1102 407–426. <https://doi.org/10.1007/s00410-012-0744-6>
- 1103 Rooyackers, Shane M, Stix, J., Berlo, K., Petrelli, M., Hampton, R.L., Barker, S.J., Morgavi, D., 2021.
1104 The Origin of Rhyolitic Magmas at Krafla Central Volcano (Iceland). *Journal of Petrology* 62, egab064.
1105 <https://doi.org/10.1093/petrology/egab064>
- 1106 Rooyackers, Shane M., Stix, J., Berlo, K., Petrelli, M., Sigmundsson, F., 2021. Eruption risks from
1107 covert silicic magma bodies. *Geology*. <https://doi.org/10.1130/G48697.1>
- 1108 Russell, J.K., Stanley, C., 1990. Origins of the 1954-1960 Lavas, Kilauea Volcano, Hawaii: Major
1109 Element Constraints on Shallow Reservoir Magmatic Processes. *Journal of Geophysical Research* 95,
1110 5021–5047.
- 1111 Sides, Edmonds, M., Maclennan, J., Houghton, B.F., Swanson, D.A., Steele-MacInnis, M.J., 2014a.
1112 Magma mixing and high fountaining during the 1959 Kilauea Iki eruption, Hawai’i. *Earth and
1113 Planetary Science Letters* 400, 102–112. <https://doi.org/10.1016/j.epsl.2014.05.024>
- 1114 Sides, Edmonds, M., Maclennan, J., Swanson, D.A., Houghton, B.F., 2014b. Eruption style at Kilauea
1115 Volcano in Hawai’i linked to primary melt composition. *Nature Geoscience* 7, 464–469.
1116 <https://doi.org/10.1038/ngeo2140>
- 1117 Soldati, A., Houghton, B.F., Dingwell, D.B., 2021. A lower bound on the rheological evolution of
1118 magmatic liquids during the 2018 Kilauea eruption. *Chemical Geology* 576, 120272.
1119 <https://doi.org/10.1016/j.chemgeo.2021.120272>
- 1120 Sparks, S.R.J., Sigurdsson, H., Wilson, L., 1977. Magma mixing: a mechanism for triggering acid
1121 explosive eruptions. *Nature* 267, 315–318. <https://doi.org/10.1038/267315a0>
- 1122 Stock, M.J., Geist, D., Neave, D.A., Gleeson, M.L.M., Bernard, B., Howard, K.A., Buisman, I.,
1123 Maclennan, J., 2020. Cryptic evolved melts beneath monotonous basaltic shield volcanoes in the
1124 Galápagos Archipelago. *Nat Commun* 11, 3767. <https://doi.org/10.1038/s41467-020-17590-x>

**This is a preprint submitted to EarthArxiv.
It has been accepted in Geochemistry, Geophysics and Geosystems (Jan, 2022)**

- 1125 Swanson, D.A., Duffield, W.A., Jackson, D.B., Peterson, D.W., 1979. Chronological narrative of the
1126 1969-71 Mauna Ulu eruption of Kilauea Volcano, Hawaii. Geological Survey Professional Paper 1056.
- 1127 Teplow, W., Marsh, B., Hulen, 2009. Dacite Melt at the Puna Geothermal Venture Wellfield, Big
1128 Island of Hawaii. GRC Transactions 33.
- 1129 Trusdell, F., Wolfe, E., Morris, J., 2006. Digital Database of the Geologic Map of the Island of Hawai'i.
1130 USGS Data Series 144.
- 1131 Wanless, V.D., Perfit, M.R., Ridley, W.I., Klein, E., 2010. Dacite Petrogenesis on Mid-Ocean Ridges:
1132 Evidence for Oceanic Crustal Melting and Assimilation. *Journal of Petrology* 51, 2377–2410.
1133 <https://doi.org/10.1093/petrology/egq056>
- 1134 Wieser, P., Iacovino, K., Matthews, S., Moore, G., Allison, C., 2021. VESlcal II - A Critical Approach to
1135 Volatile Solubility Modelling using an Open-Source Python3 Engine. EarthArxiv.
1136 <https://doi.org/10.31223/X5K03T>
- 1137 Wieser, P.E., Edmonds, M., Maclennan, J., Jenner, F.E., Kunz, B.E., 2019. Crystal Scavenging from
1138 Mush Piles Recorded by Melt Inclusions. *Nature Communications*.
- 1139 Wieser, P.E., Iacovino, K., Matthews, S., 2021a. VESlcal part II.
- 1140 Wieser, P.E., Jenner, F., Edmonds, M., Maclennan, J., Kunz, B.E., 2020. Chalcophile elements track
1141 the fate of sulfur at Kīlauea Volcano, Hawai'i. *Geochimica et Cosmochimica Acta*
1142 S0016703720303239. <https://doi.org/10.1016/j.gca.2020.05.018>
- 1143 Wieser, P.E., Lamadrid, H., Maclennan, J., Edmonds, M., Matthews, S., Iacovino, K., Jenner, F.E.,
1144 Gansecki, C., Trusdell, F., Lee, R.L., Ilyinskaya, E., 2021b. Reconstructing Magma Storage Depths for
1145 the 2018 Kīlauean Eruption From Melt Inclusion CO₂ Contents: The Importance of Vapor Bubbles.
1146 *Geochem Geophys Geosyst* 22. <https://doi.org/10.1029/2020GC009364>
- 1147 Wilson, L., Head, J.W., 1981. Ascent and eruption of basaltic magma on the Earth and Moon. *J.*
1148 *Geophys. Res.* 86, 2971–3001. <https://doi.org/10.1029/JB086iB04p02971>
- 1149 Wright, T.L., Fiske, R.S., 1971. Origin of the Differentiated and Hybrid Lavas of Kilauea Volcano,
1150 Hawaii. *Journal of Petrology* 12, 1–65. <https://doi.org/10.1093/petrology/12.1.1>
- 1151 Wright, T.L., Peck, D.L., Shaw, H.R., 1976. Kilauea Lava Lakes: Natural Laboratories for Study of
1152 Cooling, Crystallization, and Differentiation of Basaltic Magma, in: Sutton, G.H., Manghnani, M.H.,
1153 Moberly, R., Mcafee, E.U. (Eds.), *Geophysical Monograph Series*. American Geophysical Union,
1154 Washington, D. C., pp. 375–390. <https://doi.org/10.1029/GM019p0375>
- 1155 Zoeller, M.H., Perroy, R.L., Wessels, R., Fisher, G.B., Robinson, J.E., Bard, J.A., Peters, J., Mosbrucker,
1156 A., Parcheta, C.E., 2020. Geospatial database of the 2018 lower East Rift Zone eruption of Kilauea
1157 Volcano, Hawaii. <https://doi.org/10.5066/P9S7UQKQ>
- 1158