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7 **Pore-scale imaging of hydrogen** 8 **displacement and trapping in porous media**

9 *Eike M. Thaysen^{1*}, Ian B. Butler¹, Aliakbar Hassanpouryouzband¹, Damien Freitas¹,*
10 *Fernando Alvarez-Borges², Samuel C. Krevor³, Niklas Heinemann¹, Robert Atwood²,*
11 *Katriona Edlmann¹*

12 ¹School of Geoscience, Grant Institute, The King's Buildings, The University of Edinburgh,
13 James Hutton Road, Edinburgh, EH9 3FE, United Kingdom

14 ² Diamond Light Source Ltd, Harwell Campus, Didcot OX11 0DE, United Kingdom

15 ³ Department of Earth Science and Engineering, Imperial College London, London SW7
16 2AZ, United Kingdom

* = corresponding author email, phone number: eike.thaysen@ed.ac.uk, +34 682079236

18 **Abstract**

19 Hydrogen can act as an energy store to balance supply and demand in the renewable energy
20 sector. Hydrogen storage in subsurface porous media could deliver high storage capacities but
21 the volume of recoverable hydrogen is unknown. We imaged the displacement and capillary
22 trapping of hydrogen by brine in a Clashach sandstone cylinder at 2-7 MPa pore fluid pressure
23 using X-ray computed microtomography. Hydrogen saturation obtained during drainage at
24 capillary numbers of $<10^{-7}$ was $\sim 50\%$ of the pore volume and independent of the injection
25 pressure. Fluid configuration in the pore space was indicative of a water wetting system at all
26 conditions. Capillary trapping of hydrogen via snap-off during secondary imbibition at a
27 capillary number of 2.4×10^{-6} systematically increased with pressure, 20%, 24% and 43% of the
28 initial hydrogen trapped at 2, 5 and 7 MPa, respectively. Injection of brine at increasing
29 capillary numbers up to 9.4×10^{-6} reduced capillary trapping. Based on these results, we
30 recommend more shallow, lower pressure sites for future hydrogen storage operations in
31 porous media.

32

33 Keywords: geological hydrogen storage, μ CT, porous media, recovery, drainage, imbibition

34

35 Highlights:

- 36 • Hydrogen injectivity and recovery in rock is imaged with x-ray computed micro-CT
- 37 • Hydrogen recovery decreases with increasing reservoir depth
- 38 • Hydrogen recovery increases with increasing brine flow rate
- 39 • Hydrogen trapping occurs via snap-off processes
- 40 • Nitrogen is a poor proxy for hydrogen

41

42 **1. Introduction**

43 The storage of surplus electrical energy could increase the utility of renewable energy sources
44 thereby reducing the environmental impact of energy generation^{1, 2}. Storage mediums like
45 batteries, compressed air, capacitors, or flywheels are only applicable for the short-term and
46 small-scale electricity storage³. The use of hydrogen (H₂) as an energy vector however could
47 provide long-term storage to balance the intermittent demand and supply challenges affecting
48 renewable methods³. Considering the low energy density of H₂⁴, storage of H₂ in large storage
49 sites is the only solution for economical, gigawatt to terawatt-scale H₂ storage. Underground
50 storage of H₂ in salt caverns is an established technology⁵, yet, only subsurface porous media
51 (saline aquifers and depleted oil and gas fields), can provide TWh storage capacities that could
52 balance seasonal demands⁴. Relative to cavern storage, inter-seasonal storage in porous media
53 is poorly understood and introduces a range of other complexities², including potential pore-
54 clogging by microbial growth⁶, gas loss out of diffuse lateral boundaries, viscous fingering of
55 H₂ into reservoir brines, residual trapping and possible reactions with the primary formation
56 fluid and reservoir minerals⁷⁻¹⁰. Each of these requires consideration of multiphase flow,
57 transport and reaction^{2, 8}.

58 Recent years have seen several studies on H₂ geochemistry¹¹⁻¹³ and the elucidation of many
59 important flow characteristics of H₂ including; Hydrogen relative permeability^{14, 15}, interfacial
60 tension (IFT)^{16, 17}, and H₂/brine/rock contact angles^{14, 16-24}. Yet, additional data describing the
61 H₂ fluid flow in different porous formations and under varying conditions are vital to make
62 accurate predictions of the H₂ plume development and to define optimum production
63 strategies⁸. Of particular relevance is the degree of capillary trapping, which leads to a reduced
64 volume of recoverable H₂, in this way impacting the economic feasibility of the operation⁸. The
65 capillary forces that control capillary trapping also control the relative permeability⁸. As such

66 studies on residual trapping give insight to a range of crucial input parameters for pore-scale
67 and reservoir H₂ storage models.

68 The effects of the brine pore fluid pressures and of brine flow rate on capillary trapping of H₂
69 are hitherto unknown. The brine pore fluid pressure is linked to the reservoir depth via the
70 hydrostatic gradient, hence an investigation of the latter could facilitate the definition of an
71 optimal storage depth for gas recovery. The flowrate affects the capillary number, N_c , via
72 equation 1:

$$73 \quad N_c = \frac{q \cdot \mu}{\sigma} \quad (1)$$

74 where q is the flow, μ is the viscosity and σ is the interfacial tension (IFT). As N_c increases,
75 viscous forces dominate capillary forces and capillary trapping of the nonwetting phase
76 decreases²⁵. This capillary desaturation with increasing N_c , depends on the rock type²⁶ and on
77 the kind of nonwetting phase²⁵ and is observed around critical capillary numbers of 10^{-5} to 10^{-8} ²⁵.

79 A non-destructive standard technique for visualization and analysis of wetting and non-wetting
80 phase displacement processes in porous media is x-ray micro-computed tomography (μ CT)²⁷.
81 Pore-scale models are directly reliant on μ CT images to build up the basic models for
82 appropriate understanding of gas storage operations²⁸. Previously published μ CT studies on
83 two-phase flow in porous media involving gas have largely focused on supercritical carbon
84 dioxide (sCO₂) in sandstones and limestones, e.g.^{27, 29-34}. Reported saturations of sCO₂ in non-
85 aged (i.e. not wettability altered by exposure to oil or organic acids) water-wet rocks range
86 from 30-60% during drainage and 15-30% during imbibition, at N_{Ca} between 10^{-8} to 10^{-6} and
87 10^{-8} to 2×10^{-5} , respectively^{27, 30, 32-34}. Studies using subcritical, gaseous carbon dioxide (CO₂),
88 which could be more readily compared to H₂, are scarce. Yet, the characteristic trapping curves
89 for sCO₂ and CO₂, which show the residual saturation (S_{nmwr}) as a function of initial saturation
90 (S_{nwi}), are not significantly different³⁵. Saturations of nitrogen (N₂) during drainage and

91 imbibition in sandstones with 6-22% porosity is 43-64% and 43%, respectively, at N_C between
92 $1-7 \times 10^{-8}$ ^{36, 37}. Carbon dioxide trapping increases at lower pressures due to decreasing water
93 contact angles³⁸. Unlike in CO₂ storage, trapping is not desirable in geological H₂ storage as it
94 leads to unrecoverable H₂.

95 Some observational studies have suggested that H₂ is less wetting on rocks than CO₂^{19, 21}. At
96 the time of writing two μ CT studies, one micro-model study and one nuclear magnetic
97 resonance (NMR) study on H₂ exist: Al-Yaseri et al. (2022)³⁹ used NMR to find H₂ S_{nwi} and
98 S_{nwr} of 4% and <2%, respectively, in a Fontainebleau sandstone at 0.4 MPa and ambient
99 temperature. Higgs et al. (2021)¹⁶ μ CT-imaged H₂ injection into a 5 mm diameter and 7.6 mm
100 length core of Bentheimer sandstone at pressures of 6.8 to 20.8 MPa and documented
101 decreasing IFT between H₂ and water with increasing pressure (72.5 mN/m at 6.9 MPa to 69.4
102 mN/m at 20.7 MPa). Jha et al. (2021)⁴⁰ conducted a single cycle H₂-brine displacement
103 sequence in a 5 mm diameter and 15 mm length Gosford sandstone, and used μ CT to calculate
104 an S_{nwi} of 65% during drainage and a S_{nwr} of 41% after brine imbibition. Rock samples in Jha
105 et al. (2021)⁴⁰ and Higgs et al. (2021)¹⁶ were sufficiently short for capillary end effects to
106 dominate the flow behaviour, where the wetting phase accumulates close to the production face
107 of the core⁴¹. The experiment by Jha et al. (2021)⁴⁰ was performed at ambient pressure and
108 leaves open the question of variations in H₂ wetting and flow behaviour at reservoir
109 conditions¹⁹. Increasing pore fluid pressures have been reported to increase the gas saturation
110 and pore network connectivity for nitrogen (N₂) in a brine wet, aged Bashijiqike tight sandstone
111 (0.6% gas saturation at 2 MPa versus 43% at 8 MPa)³⁶. The effect of N_C increases from 7.7×10^{-7}
112 to 3.8×10^{-4} on the H₂ flow through a sandstone micro-model at 0.5 MPa was investigated by
113 Lysy et al. (2022)²⁰ who found that H₂ saturation increased correspondingly from 18% to
114 79%²⁰.

115 In this work, we used μ CT to investigate the displacement and capillary trapping of H_2 by brine
116 in 4.7 mm diameter and 54-57 mm length Clashach sandstone cores as a function of gas and
117 brine pore fluid pressure (2-7 MPa) and flow rate (20 to 80 μ l min^{-1} , corresponding to H_2 and
118 brine bulk N_C of $1.7\text{-}6.8 \times 10^{-8}$ and $1.2\text{-}9.4 \times 10^{-6}$, respectively). We show that the H_2 S_{nwi} is
119 independent of pore fluid pressure but that higher brine pore fluid pressures during secondary
120 imbibition increase capillary trapping of H_2 , indicating increased trapping at elevated reservoir
121 depths. Higher flowrates during secondary imbibition on the other hand decrease capillary
122 trapping. Our work advances the current understanding of H_2 flow in porous media by
123 delivering the first realistic estimates on how much H_2 can be recovered from the subsurface
124 after injection, disregarding any H_2 loss by microbial or geochemical reactions, as a function
125 of reservoir depth and flow rate.

126

127 **2. Materials and Methods**

128 *2.1 Core-flooding experiments*

129 We performed non-steady and steady state, two-phase core flooding experiments with H_2
130 (purity 99.9995 vol.%, BOC Ltd.) and brine (0.5 M CsCl or 2 M KI, Sigma-Aldrich) in a
131 Clashach sandstone (Permian, Aeolian sandstone from near Elgin in Scotland. Clashach
132 composition: \sim 96 wt.% quartz, 2% K-feldspar, 1% calcite, 1% ankerite⁴², porosity 11.1-
133 14.4%⁴²⁻⁴⁴) at ambient temperature (\sim 293 K). The Clashach sandstone, with its low
134 mineralogical variability, yet still comparably big pore space, was chosen in order to facilitate
135 the visualization of the injected H_2 and the interpretation of the results.

136 Two types of experiments were carried out: The first set of experiments (from now on referred
137 to as UoE (Univeristy of Edinburgh) experiments) was carried out using the μ CT instrument
138 at the University of Edinburgh and were aimed at imaging the displacement and capillary

139 trapping of H₂ by brine as a function of saturation after drainage and imbibition under different
140 experimental conditions. Three UoE experiments were carried out that investigated the effect
141 of injection pressure (2-7 MPa), whereof each was repeated once (UoE exp. 1-3, Table 1). The
142 standard error on the H₂ saturation in the repeated experiments was calculated as the standard
143 deviation divided by the square root of the number of repeated experiments. One UoE
144 experiment looked at the effect of the H₂/brine injection ratio on the H₂ saturation (UoE exp.
145 4, Table 1). This experiment was carried out to evaluate the effect of H₂ injection into aquifers
146 with different magnitudes of brine flow (hydrodynamic aquifers), and to simulate the far field
147 situation, where H₂ and brine move together. Another UoE experiment investigated the effect
148 of secondary drainage and imbibition (UoE exp. 5, Table 1). UoE exp. 6 used N₂ instead of H₂.
149 This experiment was undertaken in order to compare the flow behaviour of the two gases,
150 which is of relevance because N₂ is sometimes used as an analogue for H₂, e.g. in permeability
151 measurements^{11, 15}. Because rearrangement processes in the pore volume were noted previously
152 for N₂³⁶, UoE exp. 7 (Table 1) was undertaken to examine the stability of H₂ in the rock volume
153 over an experimental duration (10 hours): Ten PV of H₂ were injected into a brine-saturated
154 rock and keeping the pressure constant inside the pressure vessel, with imaging undertaken at
155 the start and after 10 hours. The effect of bulk capillary number on drainage and imbibition
156 was deduced by a comparison of the results from UoE exp. 2 and 5 which used flowrates of 20
157 and 80 $\mu\text{l min}^{-1}$, respectively, and 5 MPa injection pressure (Table 1).

158 The second type of experiment (from now on referred to as the ‘dry rock experiment’; Table
159 1) used synchrotron radiation (Diamond Light Source, I12-JEEP tomography beamline) to
160 capture the time-resolved displacement of H₂ by brine in a dry, H₂-saturated rock. This
161 experiment served as a base of comparison to the displacements of H₂ in wet, brine-saturated
162 rock (UoE experiments).

163 All experiments used a bespoke x-ray transparent core holder for a 5 mm diameter rock core,

164 **Table 1: Overview over experiments**

Experiment	Type of x-ray source	Description	Sample	H ₂ /brine injection ratio(s)	Flowrate (μl min ⁻¹)	Injection Pressure (MPa)	Confining Pressure (MPa)	Pore volumes injected	Repetitions of experiment
UoE exp. 1	laboratory	Primary drainage and imbibition in non-steady state displacements of H ₂ and brine	1	-	20	7	9	10	1 entire repetition, 1 repetition on drainage only
UoE exp. 2	laboratory	Primary drainage and imbibition in non-steady state displacements of H ₂ and brine	1	-	20	5	9	10	1
UoE exp. 3	laboratory	Primary drainage and imbibition in non-steady state displacements of H ₂ and brine	1	-	20	2	9	10	1
UoE exp. 4	laboratory	Simultaneous, steady state injections of H ₂ and brine	1	1:4, 1:1, 4:1	20	5	9	10	-
UoE exp. 5	laboratory	Primary and secondary drainage and imbibition in non-steady state displacements of H ₂ and brine	1	-	80	5	9	10	-
UoE exp.6	laboratory	Non-steady state displacements of N ₂ and brine	1	-	20	5	9	10	-
UoE exp.7	laboratory	H ₂ stability	1	-	80	5	9	10	-
Dry rock	synchrotron	Non-steady state, brine injection into dry, H ₂ saturated rock	2	-	5	5	10	10	-

166 which was a scaled-up version of the cell described by Fuisseis et al. (2014)⁴⁵. In UoE
167 experiments a carbon-fibre reinforced PEEK pressure vessel was used to permit good x-ray
168 transparency. The dry rock experiment used an aluminium pressure vessel to comply with the
169 health and safety requirements at Diamond Light Source. The specifications of the pressure
170 vessels of PEEK and aluminium were engineered with safety factors of 2 times or more over
171 and above the maximum applied confining pressure (10 MPa; Table 1).

172 Rock cores for the experiments were obtained by diamond drill coring with a water-flushed
173 chuck, followed by preparation of the core ends by grinding on a lathe. Experiments used a set
174 of four high-pressure pumps (Cetoni NemesysTM, flowrate range 0.072 nl s⁻¹ to 13.76 ml s⁻¹):
175 One for the injection of H₂, one for the injection of brine, one to hold the backpressure and one
176 for the confining pressure (Fig. 1). A bespoke manifold system composed of high-pressure 1/8”
177 and 1/16” 316 stainless steel and 1/16” PEEK tubing (Swagelok, Top Industrie and Cole
178 Parmer, respectively) connected the pumps to the core-flood cell (Fig. 1). Additional pressure
179 transducers (ESI Technology; accuracy 0.1% full-scale) were coupled to the flow system at the
180 inlet and outlet to allow for higher precision pressure monitoring than was possible using the
181 integral pressure gauges in the syringe pumps. Cyclic H₂ and brine injections used a Clashach
182 outcrop sample without further cleaning of 4.7 mm diameter and a relatively long length of 54-
183 57 mm to avoid the influences of capillary end effects^{46, 47}. To prevent leakage of H₂ into the
184 confining fluid, the rocks were jacketed in aluminium foil and polyolefin heatshrink tubing and
185 sealed with silicone adhesive between the conical-ended pistons within the pressure vessel. In
186 UoE experiments, a water-wet rock was first saturated with brine (0.5 M CsCl) at a flow rate
187 of 70 µl min⁻¹. Afterwards, H₂ was injected (drainage) into the brine-saturated rock at flow
188 rates of 20-80 µl min⁻¹, based on desired capillary-regime N_C of 1.7-6.8x10⁻⁸ (The viscosity of
189 H₂ is 9.01 µPa s at 298K and 4.7 MPa⁴⁸ and the IFT between H₂ and water is 72.6 mN m⁻¹ at
190 298K and 5 MPa^{49, 50}). Subsequently, the brine was reinjected (imbibition) at flow rates of 20-

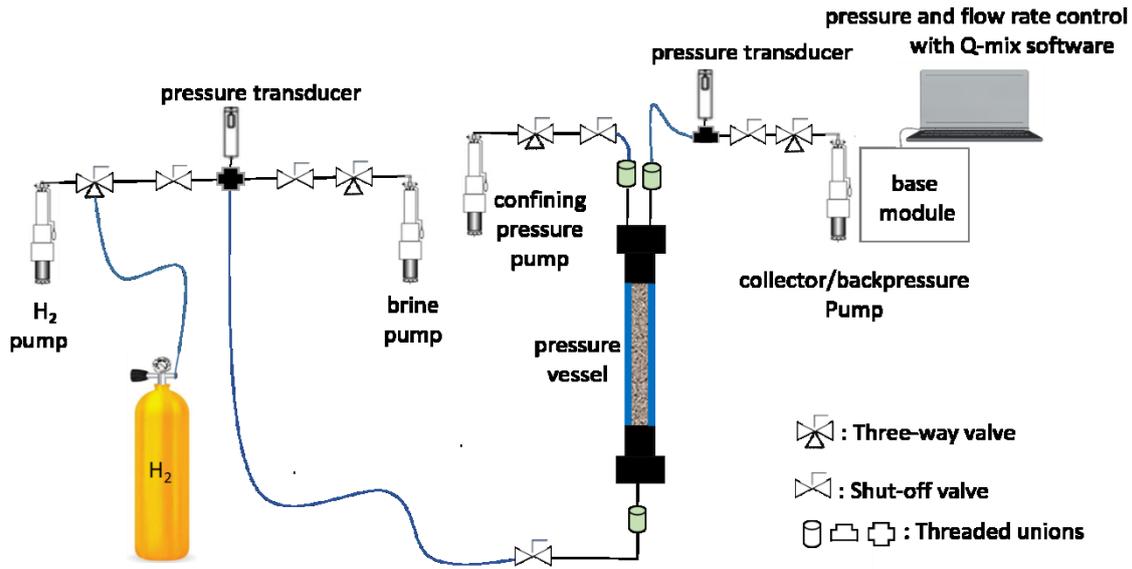


Figure 1: Experimental setup showing the manifold system that connected the X-ray transparent pressure vessel to a set of four high-pressure Cetoni Nemesys™ pumps: one to inject H₂, one inject brine, one to maintain backpressure and one to maintain confining pressure. The materials for the connections were 316 stainless steel (black), HPLC (green) and PEEK or carbon fibre reinforced PEEK (blue). The pressure vessel consisted of carbon fibre reinforced PEEK in UoE experiments and of aluminium in the dry rock experiment (see text). Pressure and flow rate control was achieved with the Q-mix software.

191

192 80 $\mu\text{l min}^{-1}$, resulting in N_C of $2.35\text{-}9.45 \times 10^{-6}$ (using the same IFT between H₂ and water of
 193 72.6 mN m^{-1} at 5 MPa and 298 K^{49, 50} and a viscosity of 1.247 Pa s at 5 MPa that was estimated
 194 from the reported 1.2503 Pa s and 1.233×10^{-3} Pa s at 0.1 MPa and 25 MPa, respectively, and
 195 298 K⁵¹). The N_C the N₂ experiment was 3.5×10^{-8} (using an IFT of 73 mN m^{-1} between N₂ and
 196 water 1t 298K and 10 MPa³⁵ and a viscosity of 1.89×10^{-5} Pa s at 5 MPa and 295K⁴⁸). Each
 197 injection used ten pore volumes to ensure completely flushing of the sample cores with the
 198 injected fluid.

199 In the dry rock experiment, H₂ was directly injected into a dry rock at a flow rate of 70 $\mu\text{l min}^{-1}$
 200 ¹. Subsequently, the brine (2 M KI) was injected at a flow rate of 5 $\mu\text{l min}^{-1}$, resulting in N_C of
 201 5×10^{-7} (using the same IFT between H₂ and water of 72.6 mN m^{-1} at 5 MPa and 298 K^{49, 50} and
 202 a viscosity of 1.07×10^{-3} Pa s for 0.6 M KI and 293 K⁴²).

203 The combined application of an x-ray transparent core holder and μ CT allowed the
204 visualization of the fluid saturation distributions at pore scale at each injection step. The
205 difference in the x-ray attenuation coefficient of the fluids (H_2 and 0.5 M CsCl/ 2 M KI)
206 provided an excellent contrast between the two fluid phases and the rock on the acquired μ CT
207 images, combined with the respective radiation energy in the two different laboratories.
208 3D volumes were acquired from the lower central portion of the sample to avoid the impact of
209 capillary end effects on fluid saturation^{46, 47}. For the UoE experiments, image acquisition used
210 a μ CT instrument built in-house at the University of Edinburgh, comprising a Feinfocus 10-
211 160 kV reflection source, a Micos UPR-160-air rotary table and a Perkin-Elmer XRD 0822 1
212 MP amorphous silicon flat panel detector with a terbium doped gadolinium oxysulfide
213 scintillator. Data acquisition software was developed in-house. The following settings were
214 used for UoE experiments: 120 keV, 16 W, 2 seconds exposure time, 1200 projections and 2
215 frames per stop. The voxel size was $5.4 \mu m^3$. In the dry rock experiment, time-resolved imaging
216 of the H_2 and brine displacement processes was achieved by means of a 65 keV monochromatic
217 beam detected by a high-resolution imaging camera with optical module 2 (PCO.edge 5.5, 7.91
218 x 7.9 μm /pixel with FoV 20 mm x 12 mm) using 17-25 milliseconds exposure time and 900
219 projections. The voxel size was $7.9 \mu m^3$.

220 *2.2 Image analysis*

221 Tomographic reconstructions were undertaken by filtered back projection using Octopus 8.9⁵²
222 on a GPU accelerated workstation. All subsequent image processing and analysis of
223 tomographic data was performed using Avizo Version 9.1.1 (FEI, Oregon, USA). Data from
224 UoE experiments were processed using a non-local means filter⁵³. Processing of the dry rock
225 experiment used a combination of median filter and unsharp mask to reduce image
226 noise. Segmentation of UoE experiment data used a global threshold on the 2D greyscale image
227 histogram, and encompassed two phases. In the water-saturated scans, water and rock were

228 treated as two discrete phases. In scans after brine and H₂ injections, the H₂ was treated as one
229 phase and the brine and rock as a single separate phase, following protocols of Andrew et al.
230 (2014)³². Holes and spots which were at the resolution limit of the data were removed from all
231 datasets (applied thresholds corresponded to 3³ and 5³ voxels, respectively). Based on the
232 segmented image of the water-saturated scan in UoE experiments, a pore size distribution was
233 calculated. The 3D image was separated into individual pores and throats using Avizo's
234 'separate objects' module, which calculates a chamfer distance map of the pore-space and then
235 applies a marker based watershed algorithm to the distance map to define discrete pore bodies
236 as catchment basins separated by the watershed which marks the location of pore throats (SI
237 Figure S1d). Supporting information Figure S1 shows the work flow for the water-saturated
238 scan.

239 In scans following brine and H₂ injections in UoE experiments, the segmented image was
240 analysed in 3D using the 'labeling' and 'label analysis' modules to identify, label and measure
241 the volume of each H₂ cluster. Hydrogen cluster size distributions were compared to the pore
242 size distribution to evaluate the H₂ connectivity and identify trapping mechanisms during brine
243 imbibition.

244 *2.3 Capillary pressure*

245 Recovered Clashach cores were submerged in 25% w/v NaOH solution (Fisher Scientific) for
246 2 hours to remove the aluminium foil from the core surface, and rinsed in successive milli-Q
247 water, acetone and ethanol ultrasonic baths. Subsequently, the cores were cut and squared to
248 the dimensions of 25 mm, overlapping the μ CT visualized rock volumes, and cleaned
249 ultrasonically with milli-Q water to remove grinding products. Mercury injection capillary
250 pressure (MICP) was performed on the cleaned cores, using a micromeritics automated
251 mercury injection equipment (Autopore IV 9500) to estimate the capillary pressure-saturation
252 relationship and pore size distribution. The pressure range tested was from vacuum to 379 MPa.

253 3. Results

254 3.1 Characterization of the pore space

255 The μ CT-evaluated porosity of the
256 Clashach sandstone from the segmented
257 volume of the water-saturated rock was
258 12.5% to 13.5%, depending on the imaged
259 region of the rock core. The MICP-
260 evaluated pore throat size distribution
261 showed a large number of very small pore
262 throats with $<5 \mu\text{m}$ radius and a small
263 number of small to intermediate size pore
264 throats (>5 to $90 \mu\text{m}$) (Fig. 2). The largest pore throat had a radius of $195 \mu\text{m}$ (Fig. 2). The
265 μ CT-evaluated size distribution of the pores evidenced a narrow pore size distribution with
266 comparably small pores with radii $<50 \mu\text{m}$ (Fig. 2, 4a).

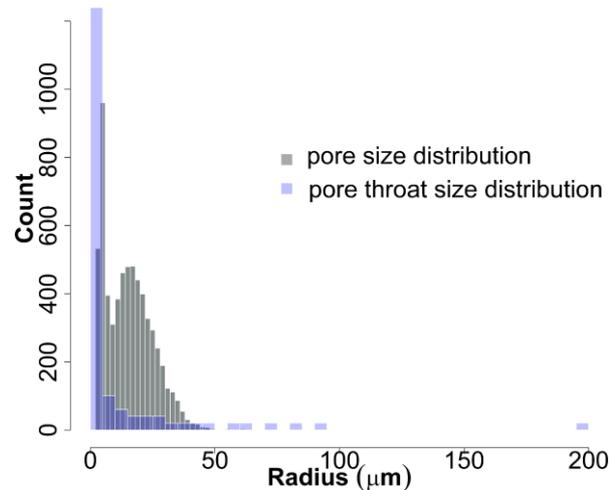


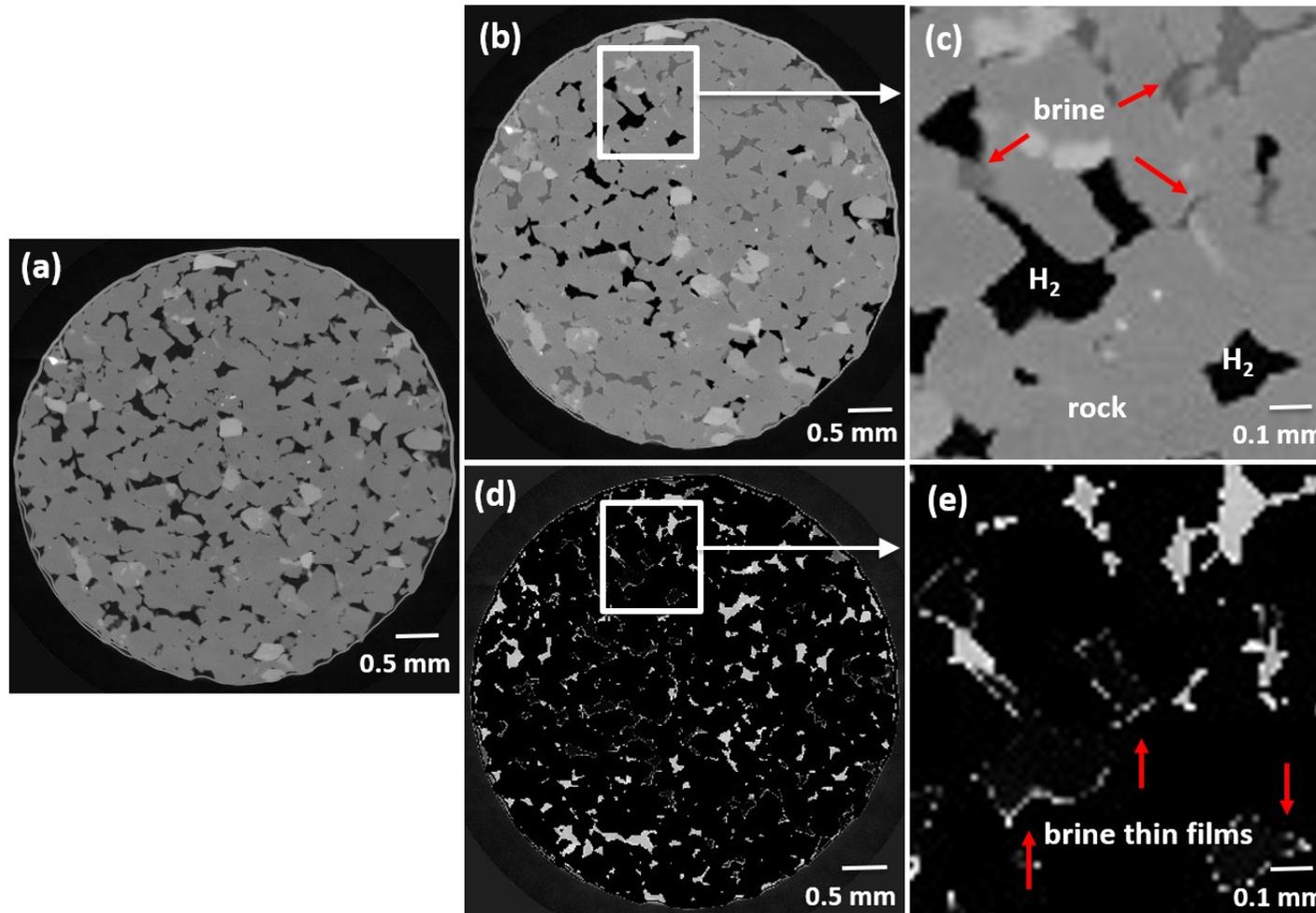
Fig. 2. μ CT-derived pore size distribution (grey) and MICP-derived pore throat size distribution (transparent blue). Note that MICP derived counts were multiplied by a factor of 20 to facilitate visualization of the results. All plots used a bin size of 30.

267 3.2 Hydrogen wetting behaviour and stability in UoE experiments

268 Hydrogen sat in the centre of the pore bodies. Residual brine sat in corners, pore throats (Fig.
269 3b and c) and, as a subtraction of the water saturated scan from the H_2 -and brine filled rock
270 revealed, in thin films around the grains (Fig. 3d). The injected H_2 remained stable within the
271 pore volume under no-flow conditions and at constant pore fluid pressure over a time period
272 of 10 hours which was the maximum experimental duration (SI Fig. S2).

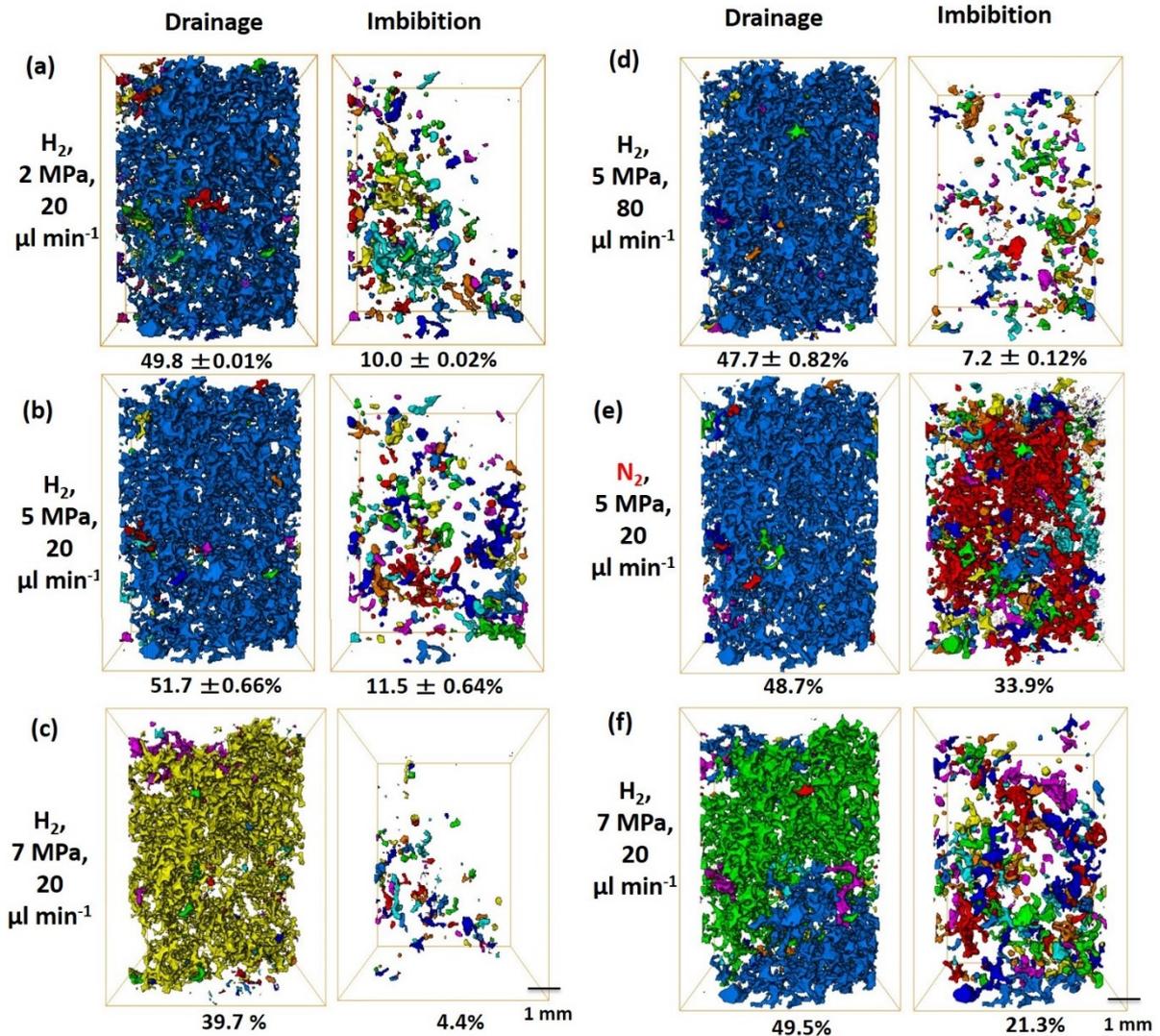
273 3.2 Effect of pore fluid pressure on hydrogen connectivity, saturation and recovery

274 Hydrogen saturation during drainage was independent of the pore fluid pressure with 49.8%,
275 51.7% and 39.7%-52.6% saturation at pore fluid pressures of 2, 5 and 7 MPa, respectively (Fig.
276 4a-c, f, SI Fig. S3). Hydrogen connectivity during drainage generally showed one large,



277

278 Fig. 3: (a) Water-saturated Clashach sandstone with the water shown in black and the rock in different shades of grey. (b) and (c) Brine-saturated Clashach sandstone after
 279 injection of H₂. H₂ (black) fills the centre of the pores while the brine (dark grey) remains in corners and small pore throats around grains. (d) and (e) Subtraction of the water-
 280 wet scan from the brine-saturated scan after H₂ injection, following registration of the brine-saturated scan after H₂ injection to the water-saturated scan, revealing discontinuous
 281 brine thin films around grains. The rim around the Al foil in (d) is caused by continued shrinkage of the Al foil onto the rock during the experiments.



282

283 Fig. 4: 3D rendering of H₂ and N₂ clusters with saturation percentages in UoE experiments. Discrete clusters were
 284 rendered in colours, where mainly one color marks one large, connected cluster and different colors indicate
 285 several, not connected clusters. (a-c) Effect of pore fluid pressure on H₂ clusters and saturation after drainage and
 286 after primary imbibition. (a) 2 MPa, (b) 5 MPa and (c, f) 7 MPa, all at a constant flow rate of 20 $\mu\text{l min}^{-1}$
 287 corresponding to capillary numbers of 1.7×10^{-8} and 2.4×10^{-6} , respectively. Large, connected clusters that existed
 288 after drainage were broken down to numerous smaller clusters after imbibition, with apparently no clear
 289 relationship between H₂ saturation and pore fluid pressure. Experiments were repeated once at 2 and 5 MPa, and at
 290 7 MPa twice for drainage runs and once for imbibition. For experiments at 2 MPa and 5 MPa averages and standard
 291 errors for the H₂ saturation are reported. For experiments at 7 MPa, due to the discrepancy in the results, both of
 292 the full primary drainage and imbibition experiments are visualized in (c) and (f). (d) Effect of cyclic injections
 293 on H₂ clusters and saturation: Averages and standard errors of the H₂ saturation after primary and secondary
 294 drainage, and after primary and secondary imbibition, all at 5 MPa pore fluid pressure and a flowrate of 80 $\mu\text{l min}^{-1}$
 295 corresponding to a capillary number of 9.4×10^{-6} . (e) Nitrogen clusters and saturations during drainage and
 296 imbibition at 5 MPa pore fluid pressure and a flowrate of 20 $\mu\text{l min}^{-1}$. For the full display of the results see SI Fig.
 297 S3 and SI Fig S4.

298 connected cluster at all pore fluid pressures except for one out of three images at 7 MPa which
 299 showed three large disconnected clusters (Fig. 5, SI Fig. S3). During drainage, the largest H₂
 300 cluster had a volume of $1 \times 10^8 \mu\text{m}^3$ at all pore fluid pressures except for the one run at 7 MPa
 301 with the disconnected clusters were the largest volume was $7 \times 10^7 \mu\text{m}^3$ (Fig. 5e).
 302

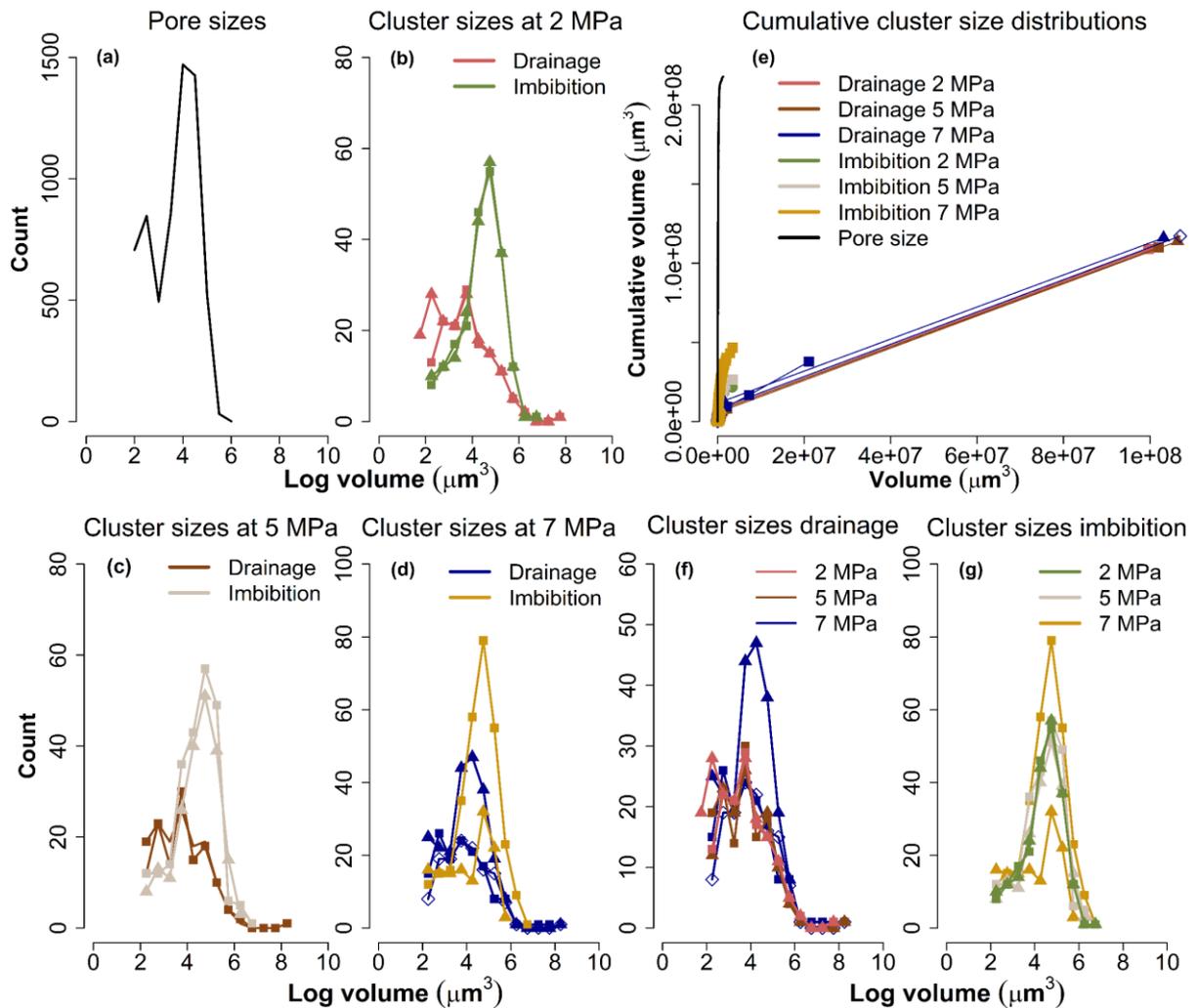


Fig. 5: (a) Pore size distribution as derived from the micro-CT image of the water-wet rock. Hydrogen cluster size distributions after drainage and imbibition in experiments at $20 \mu\text{l min}^{-1}$ flowrate and pore fluid pressures of 2 MPa (b), 5 MPa (c) and 7 MPa (d), and cumulative pore size and H₂ cluster size (CS) distributions at different pore fluid pressures (e), where squares, triangles and rhombi mark the distinct repeat experiments. (f) Hydrogen cluster size distribution after drainage for all experiments and (g) Hydrogen cluster size distribution after imbibition for all experiments. Note the large H₂ clusters of $\sim 10^8 \mu\text{m}^3$ that exist after drainage in (b)-(d). A decrease in the biggest cluster volume after imbibition in (b)-(d) along with an increase in the number of small clusters marks the change in H₂ structure during the drainage and imbibition processes. Histogram plots in (a)-(d), (f) and (g) used a bin size of 10.

303

304 Hydrogen clusters during drainage were at all pore fluid pressures much larger than discrete
305 pores with a maximum volume of $1.3 \times 10^6 \mu\text{m}^3$ (Fig. 5e, a). Comparing all H₂ cluster size
306 distributions during drainage (Figure 5f) reveals that all drainage curves, including two of the
307 distributions at 7 MPa (squares and rhombi), have largely the same distribution, however one
308 of the three distributions at 7 MPa (triangles) is distinct. This outlier experiment corresponds
309 to the experiment showing a lower S_{nwi} (Fig. 4c).

310 Capillary trapping of H₂ during imbibition seemed independent of the pore fluid pressure with
311 10%, 12% and 4-21% of trapped H₂ at 2, 5 and 7 MPa, respectively (Fig. 4a-c,f, SI Fig. S3),
312 corresponding to 20%, 22% and 11-43% of the initially injected H₂. During imbibition, large
313 H₂ clusters were broken down into smaller clusters (Fig. 5a-d), in line with the visual changes
314 of the H₂ clusters (Fig. 4a-c and f). The largest H₂ clusters after imbibition remained above the
315 maximum pore size during all experiments except for one experiment at 7 MPa (triangles in
316 Fig. 5g, Fig. 5a), showing that not only was H₂ trapped in discrete pore bodies but also as larger
317 H₂ ganglia. The break-down of the largest H₂ clusters during imbibition caused the number of
318 clusters in the size range $\log 4$ to $\log 6 \mu\text{m}^3$ to increase while the number of very small clusters
319 of $\log 2-4 \mu\text{m}^3$ typically decreased (Figs. 5b-d). Comparing all H₂ cluster size distributions
320 during imbibition (Figure 5g) shows that the distributions at 2 and 5 MPa are largely the same
321 while the imbibition distributions at 7 MPa are distinct.

322 Injections of H₂ and brine into the same rock volume and at the same flow rates and pore fluid
323 pressures of 2-5 MPa were repeatable with small standard errors between 0.01-0.66% (Fig. 4a
324 and b, Fig. 5b and c). At 7 MPa very distinct S_{nwi} and S_{nwr} were measured (Fig. 4c and f); During
325 drainage the standard error was 4.8% at an average H₂ saturation of 47.4%. The standard error
326 during imbibition was 8.5% at an average saturation of 12.9%. The pressure differences
327 between inlet and outlet during these experiments were within the error of the pressure sensors

328 of 0.1% full-scale.

329 *3.3 Effect of hydrogen/brine injection ratio on*
 330 *hydrogen connectivity and saturation*

331 We studied the effect of the H₂ /brine injection
 332 ratio on the H₂ saturation in order to evaluate
 333 the effect of H₂ injection into hydrodynamic
 334 aquifers, and to evaluate what happens in the
 335 far field, where fluids will be moving together.
 336 The results showed that the H₂ saturation and
 337 H₂ interconnected pore volume increased with
 338 increasing H₂/brine injection ratio from 32.6%
 339 at 4 μl min⁻¹ H₂ plus 16 μl min⁻¹ brine to 43.2%
 340 at 16 μl min⁻¹ H₂ plus 4 μl min⁻¹ brine (Fig. 6).

341 The H₂ clusters in simultaneous injection
 342 experiments occupied many of the same pore
 343 spaces as the clusters after H₂ in the non-
 344 steady state experiments at the same pressure
 345 and total flowrate (Fig. 4b, Fig. 6, SI Fig. S5),
 346 and even at the lowest H₂ brine injection ratio
 347 of 4 μl min⁻¹ H₂ plus 16 μl min⁻¹ brine, the H₂
 348 clusters were large, and spanning multiple
 349 pores. The percolation threshold, i.e. one
 350 connected path from inlet to outlet, was
 351 apparently only reached at 100% H₂ injection
 352 (Fig. 4b vs. Fig. 6, SI Fig. S5). The pressure

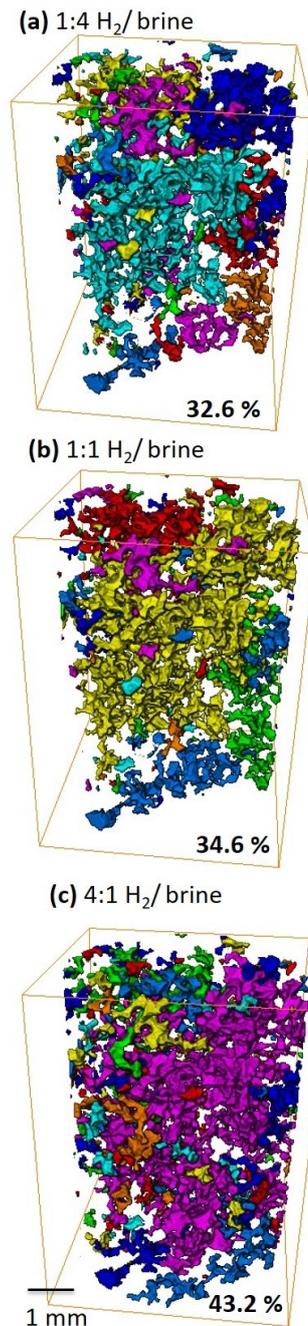


Fig. 6: Effect of H₂/brine injection ratio on H₂ saturation and H₂ connectivity during simultaneous H₂ and brine injection at 5 MPa (a) 4 μl min⁻¹ H₂ plus 16 μl min⁻¹ brine, (b) 10 μl min⁻¹ H₂ plus 10 μl min⁻¹ brine, (c) 16 μl min⁻¹ H₂ plus 4 μl min⁻¹ brine. Discrete H₂ clusters were rendered in colours, where mainly one color marks one large, connected cluster and different colors indicate several, not connected clusters. With increasing injection ratio H₂ saturation and H₂ connectivity increase.

353 differences between inlet and outlet
 354 during the simultaneous injection
 355 experiments were up to 0.05 MPa.
 356 The H₂ cluster volume distributions
 357 were similar at different H₂:brine
 358 injection ratios (Fig. 7a). However, with
 359 increasing ratio the smallest H₂ clusters
 360 of volume $\sim \log 2 \mu\text{m}^3$ decreased in
 361 number while the number of
 362 intermediate size (log 2.5 to log 6.25
 363 μm^3) H₂ clusters and the volume of the
 364 biggest cluster increased (Fig. 7a),
 365 confirming observations of increasing
 366 H₂ saturation and connectivity with
 367 increasing injection ratio (Fig. 6).

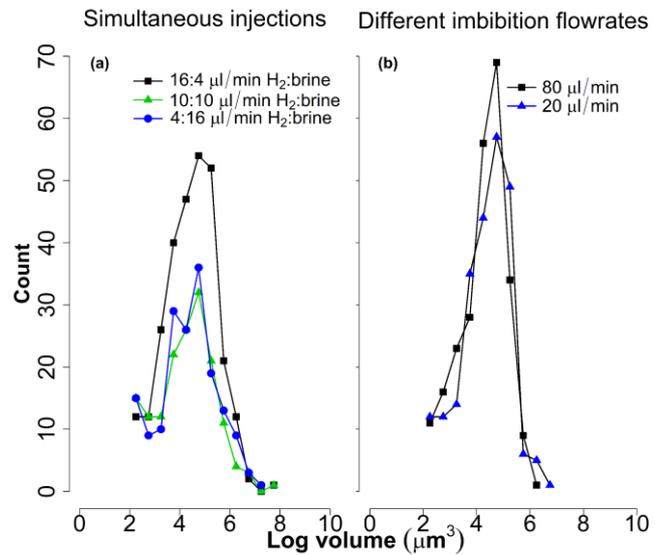


Fig. 7. **(a)** H₂ cluster size distributions during simultaneous injections of H₂ and brine at flowrate ratios of 16:4, 10:10 and 4:16 $\mu\text{l min}^{-1}$ H₂:brine and 5 MPa injection pressure. As the flowrate ratio of H₂ to brine increased, the number of intermediate size H₂ clusters increased and the radius of the biggest cluster increased. **(b)** Effect of flowrate during brine imbibition. All plots used a bin size of 10.

368 3.4 Effect of flowrate on hydrogen saturation and recovery

369 At constant pore fluid pressure of 5 MPa, increases in the flowrate during drainage from 20 μl
 370 min^{-1} to 80 $\mu\text{l min}^{-1}$, corresponding to bulk N_C of 1.7×10^{-8} to 6.8×10^{-8} , respectively, decreased
 371 the H₂ saturation from to 51.7% to 47.7% (Figure 4b and d). Correspondingly, increases in the
 372 brine flowrate during imbibition from 20 $\mu\text{l min}^{-1}$ to 80 $\mu\text{l min}^{-1}$, corresponding to bulk N_C of
 373 2.4×10^{-6} and 9.4×10^{-6} , respectively, reduced the S_{nw} from 11.5% to 7.2% (Fig. 4b and d).

374 In line with this, the H₂ cluster size distributions at the two flowrates showed that larger clusters
 375 were mobilized at 80 $\mu\text{l min}^{-1}$ (maximum cluster sizes of log 6.25 μm^3 at 80 $\mu\text{l min}^{-1}$ vs. log
 376 6.75 μm^3 at 20 $\mu\text{l min}^{-1}$; Fig. 7b). At both flowrates the largest H₂ clusters were still bigger than

377 the largest pore of $<\log 6 \mu\text{m}^3$ (Fig. 7b vs. Fig 6a), indicating that H_2 was trapped also as larger
378 ganglia.

379 *3.5 Secondary drainage and imbibition*

380 Secondary drainage and imbibition did not significantly change the H_2 saturation (47.9% and
381 7.0%) compared to primary drainage and imbibition (47.6% and 7.3%), based on results at 5
382 MPa and $80 \mu\text{l min}^{-1}$ flowrate (SI Fig. S4, Fig. 4d-e).

383 *3.6 Dry rock experiment*

384 In the dry-rock experiment, the brine entered the H_2 -filled dry rock via piston-like
385 displacement, (Fig. 8b), eventually recovering the H_2 entirely (Fig. 8d). Before all H_2 was
386 recovered, there was an intermediate stage where previously brine-filled pores (Fig. 8b)
387 showed several very small H_2 bubbles (Fig. 8d).

388 *3.7 Comparison to nitrogen*

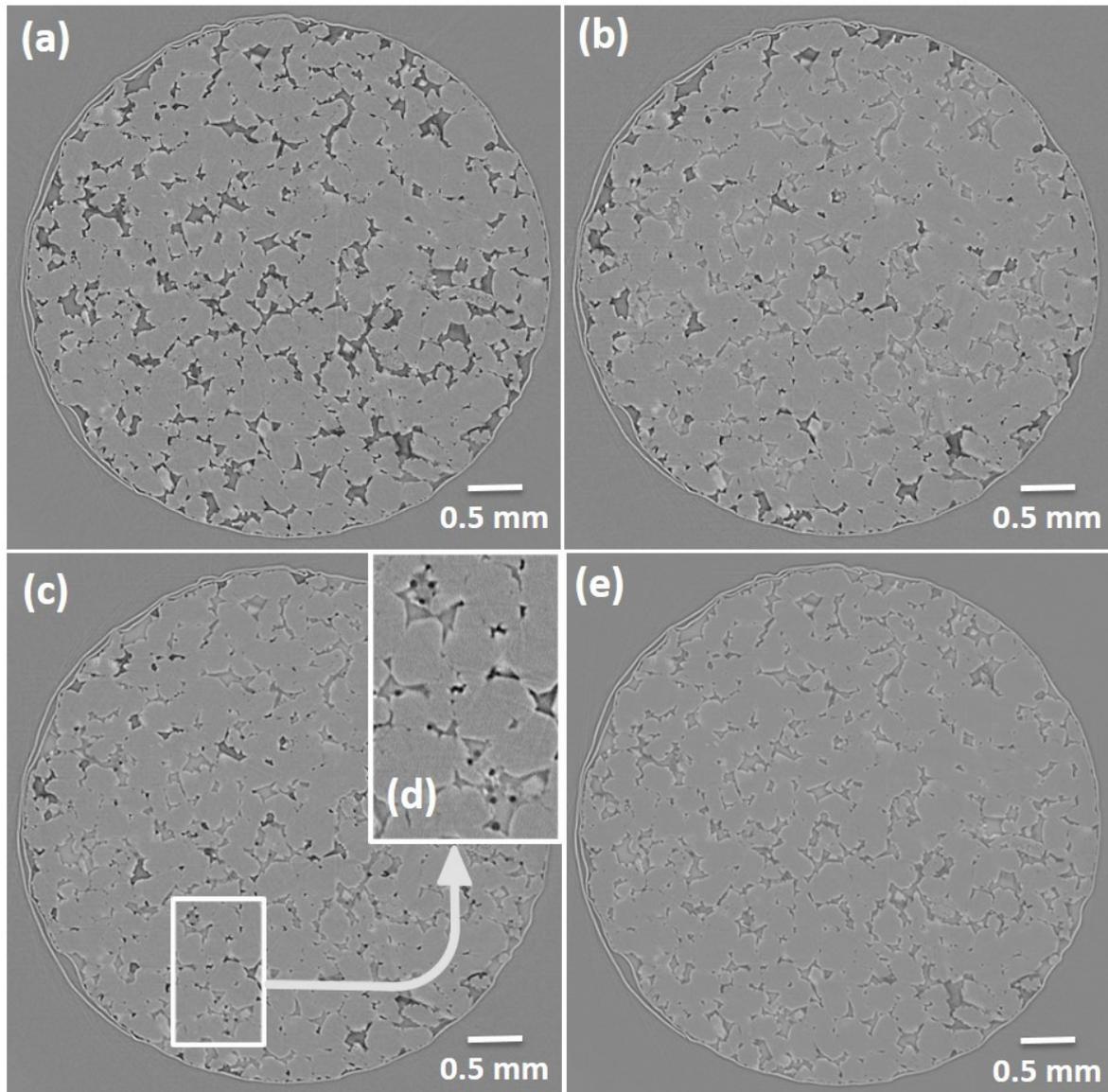
389 The N_2 saturation was similar to the H_2 saturation during drainage (48.7% vs. $51.7 \pm 0.66\%$,
390 respectively) but N_2 saturation after brine imbibition was much higher (33.9% vs. $11.5 \pm$
391 0.64% , respectively; Fig. 4e, SI Fig. S6). The N_c of H_2 and N_2 during drainage were 1.7×10^{-8}
392 and 3.5×10^{-8} , respectively.

393

394 **4. Discussion**

395 *4.1 Pore space*

396 The μCT -evaluated porosity of the Clashach sandstone from the segmented volume of the
397 water-saturated rock of 12.5-13.6% was within the range of the published porosities of 11.1-
398 14.4% for Clashach sandstone⁴²⁻⁴⁴. The distributions of pore size and pore throat size as
399 evaluated by μCT and MICP, respectively, suggested that mostly small pores of $<50 \mu\text{m}$ radii
400 were joined by very small throats of $<5 \mu\text{m}$ radii with a few small to intermediate size throats



401

402 Fig. 8. The dry-rock experiment summarized. (a) H₂-saturated, dry rock, (b) scan during brine imbibition shortly
 403 after appearance of the first brine in the rock, (c) scan during brine imbibition after 115 minutes, with several
 404 small, isolated H₂ bubbles inside pore bodies (d), and (e) brine-saturated rock after two hours of imbibition
 405 showing 100% recoverability of H₂.

406

407 of >5 to 90 μm radii in between (Fig. 2). The largest pore throat of 195 μm radius was probably
 408 measured at the surface of the rock core where the drilling process affected the pore space. The
 409 pore throat distribution for our Clashach sandstone sample was very similar to the pore throat
 410 distribution for Berea sandstone⁵⁴. Compared to the pore throat distributions for Bentheimer

411 sandstone and Doddington sandstone⁵⁴, our Clashach sandstone sample showed smaller pore
 412 throat sizes. Limestones generally show a wider pore throat size distributions than sandstones⁵⁴.

413 4.2 H₂ flow behaviour and trapping mechanisms

414 Hydrogen behaved as a non-wetting phase, filling the centre of the pores, with residual brine
 415 in the pore corners and throats (Fig. 3b and c), indicating a water wetting system. The largest
 416 H₂ cluster was much larger than discrete pores at any pore fluid pressure during drainage (Fig.
 417 5), indicating a good connectivity of the H₂⁵⁵. Hydrogen trapping occurred via snap-off of H₂
 418 ganglia (Fig. 9). Snap-off competes with piston-like, i.e. pore-filling, displacement during the
 419 displacement of a non-wetting fluid by a wetting fluid in porous media²¹, and is known as the
 420 swelling of water in the corner layers of a pore throat during water invasion in water-wet porous
 421 rocks until the threshold capillary pressure is exceeded, resulting in spontaneous filling of the
 422 throat with water and disconnection of the non-wetting phase which can lead to trapping²¹.
 423 Brine films around grains were not directly visible in the tomographic images (Fig. 3b and c)
 424

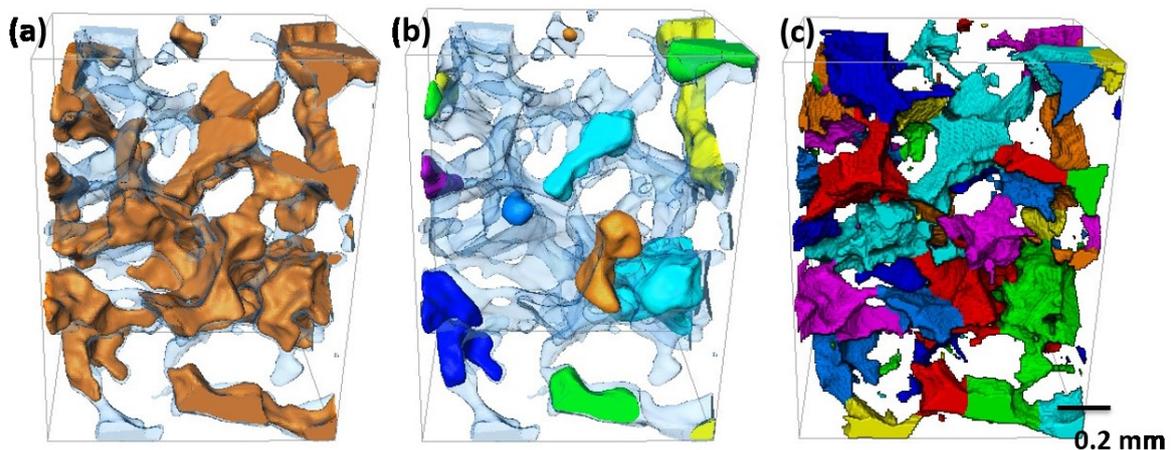


Fig. 9: Example of a snap-off event. **(a)** Labelled H₂-filled volume after drainage (orange), spanning over several pores, and total pore space (transparent blue) showing one large interconnected H₂-filled pore volume. **(b)** Labelled H₂-filled volume after brine imbibition (different coloured shades) and the total pore space (transparent blue) showing several, not connected H₂ ganglia and the snapped-off H₂ droplet (cobalt blue) in the centre, left hand side. **(c)** Pore body visualisation of the same volume.

425 but were revealed by subtraction of the water-saturated scan from the brine-saturated scan after
426 H₂ injection, following the registration of the brine-saturated scan after H₂ injection to the
427 water-wet scan (Fig. 3d and e). Figures 3d and 3e suggest that brine films were discontinuous
428 and very thin. When H₂ was injected into a dry rock, 100% of the injected H₂ could be recovered
429 (Fig. 8e) which substantiated the theory that sub-resolution brine films around grains and snap-
430 off of H₂ ganglia caused decreased H₂ recovery in experiments using an initially brine saturated
431 rock (UoE experiments). The occurrence of several very small H₂ bubbles in the dry-rock
432 experiment (Fig. 8d) indicated Roof snap-off⁵⁶ of H₂ ganglia.

433 *4.3 Effect of pore fluid pressure and hydrogen/brine injection ratio*

434 We observed no dependence of the H₂ saturation during drainage on pore fluid pressure,
435 considering that 2 out of 3 experiments at 7 MPa showed the same the same H₂ saturation of
436 ~50 % as at 2 and 5 MPa (Fig. 4a-c,f, and SI Fig. S3). The one experiment at 7 MPa which had
437 only 39.7% H₂ saturation (Fig. 4c, SI Fig. S3 and blue triangles in Fig. 5f) did also not have
438 the same H₂ cluster size distribution as the remaining experiments (Fig. 5f), despite using the
439 same experimental settings as for all other experiments at 7 MPa, and the log archives of the
440 pore fluid pressures and injected volumes revealed no abnormalities. A shift in the distribution
441 of cluster sizes can indicate a change in wettability, regardless of the measured H₂ saturation.
442 Yet, as two of the results at 7 MPa showed a similar distribution as at the other pressures, it
443 seems likely that this experiment is an outlier. The experiment was acquired after a filament
444 change on the μ CT apparatus, which implied that a slightly different part of the same rock core
445 was imaged (13.6% vs. 12.5% porosity). Yet, in principle this should not have affected the
446 results significantly, and subsequent experiments did return to show ~50% H₂ saturation, e.g.
447 the H₂ stability experiment (SI Fig. S2).

448 The observed constant drainage H₂ saturations with increasing pore fluid pressures from 2 to 7
449 MPa are in line with a lack of a dependence of the H₂ wettability on pressure increases from 2-

450 10 MPa in Berea and Bentheimer sandstone¹⁸, with only very small increases of $\sim 3\text{-}6^\circ$ in the
451 H_2 contact angles at pressure increases from 2 to 7 MPa in Basalt²¹ and clay²⁴ and quartz¹⁹, and
452 with a previous findings of no change in the characteristic trapping curves for CO_2 and N_2 at a
453 wide range of pressure and temperature conditions³⁵. The general anticipation of an increase in
454 gas saturation with injection pressure^{57, 58} may still be valid over pressure ranges larger than
455 the one investigated here. At unchanged wettability, the S_{nwi} is controlled by the capillary
456 pressure which in turn is controlled either by the fractional flow (during simultaneous injection)
457 or by the viscous force pressure drop (during single fluid phase injection). Thus, the
458 independence of the S_{nwi} from the pore fluid pressure indicates that the viscous pressure drop
459 is not being significantly altered by the changing pressure, e.g., the H_2 viscosity change is not
460 having a major impact on the force required to drive flow. Significant loss of H_2 from the gas
461 phase at higher pressures by dissolution into the brine is precluded by the low solubility of H_2
462 of $\sim 0.02 \text{ mol kg}^{-1}$ at 2.5 MPa⁵⁹. The H_2 stability experiments showed that H_2 saturation at 5
463 MPa did not change over a time period of ten hours (49.55% at time zero vs. 49.53% ten hours
464 after; SI Fig. S2), evidencing a stable result and no H_2 loss by dissolution.

465 Looking at the S_{nwr} data only (Fig. 4a-c,f), there was no clear dependency on pore fluid pressure
466 during imbibition. Any change may, however, have been masked by the high variation between
467 the two results at 7 MPa (4-21% Fig. 4c and f). Considering that the experiment at 7 MPa with
468 the low S_{nwi} of 40% could be identified as an outlier (see the discussion in the beginning of
469 section 4.3, and Fig. 5f, blue triangles) and that the S_{nwr} is a function of the S_{nwi} where a lower
470 S_{nwi} will tend to overestimate recovery^{54, 60}, we may disregard the S_{nwr} of 4%. The large increase
471 in trapped H_2 in the second full primary drainage and secondary imbibition experiment at 7
472 MPa (Fig. 4f) was probably due to the poorer initial H_2 connectivity⁶⁰. The H_2 cluster size
473 distribution for this experiment showed an increase in the number of intermediate size clusters
474 compared to the other experiments (Fig. 5g, darkyellow squares), without however shifting the

475 cluster distribution, suggesting no wettability change. This indicates that variation in one or
 476 more of the other thermophysical properties –density, viscosity, or IFT – have resulted in the
 477 impact on the pore scale fluid configuration (Figure 10). Meanwhile, a poorer initial H₂
 478 connectivity was not confirmed by the third repetition of primary drainage at 7 MPa (SI Fig.
 479 S3g). More experiments at 7 MPa are needed to confirm the result of increased trapping at
 480 higher pore fluid pressures.

481

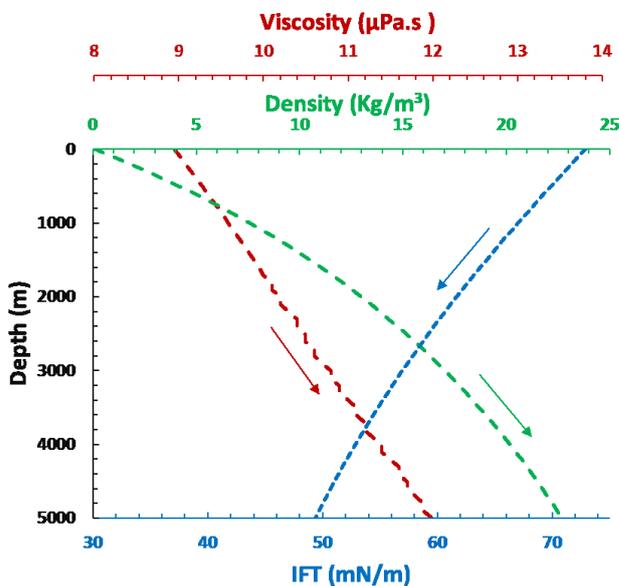


Fig. 10: Density (green line), viscosity (red line), and interfacial tension (blue line) of hydrogen versus depth (Hassanpouryouzband et al. 2021)⁶¹.

482

483 Given a hydrostatic gradient of ~ 10 MPa/km an increase of the S_{nwr} with increasing pore fluid
 484 pressure, as suggested by the second experiment at 7 MPa (Fig. 4f), would indicate that deeper
 485 aquifers are less favourable for H₂ storage operations. However, unlike our non-steady state
 486 experiments which showed barely any pressure difference between inlet and outlet, in a real
 487 H₂ storage operation, the well pressure is significantly higher than the reservoir pressure and
 488 the brine is not injected, but naturally flowing into previously H₂-saturated rock when the H₂

489 is recovered- again due to a pressure difference. Our displacement study results are hence
490 applicable to the fringe of the H₂-saturated zone, only, where pressure differences are very
491 small. A decreased H₂ recovery with depth would not align well with other criteria for an
492 economical and safe H₂ storage operation, such as a lower cushion gas requirement with
493 elevated depth⁵ and reduced risks for H₂-linked microbial activity at higher depths due to higher
494 temperatures⁶.

495 The pressure/depth effect on our results can be further related to the variation of key pore-scale
496 displacement parameters for H₂ with depth (Fig. 10): The IFT of H₂ reduces with depth whereas
497 the H₂ viscosity increases, both of which in theory should augment the displacement of H₂ with
498 water and reduce the S_{nwr} ^{62, 63}. However, increased S_{nwr} with decreasing IFT and increasing
499 viscosity has also been demonstrated⁶⁴ which may be due to a simultaneously increased
500 likelihood of unstable displacement/ non-uniform fronts^{10, 65} during both drainage and
501 imbibition processes at decreasing IFT and increasing viscosity, based on the augmenting effect
502 of both parameters on N_C (Eq. 1). Such unstable displacement processes at higher pore fluid
503 pressures, addition to a variation in the thickness of the brine thin films with pressure, may
504 explain the decrease in interconnected H₂-filled pore volume from one large cluster at 2-5 MPa
505 to three clusters at 7 MPa and have lead to increased snap-off and trapping during imbibition
506 (Fig. 4f).

507 The S_{nwr} of 10-21% was significantly less than the previously reported H₂ S_{nwr} of 41% for a
508 sandstone under ambient conditions⁴⁰. As mentioned previously, the short length rock sample
509 in Jha et al. (2021)⁴⁰ suggests that their results were affected by capillary end effects⁴⁶.
510 However, the applied the relatively high bulk brine N_C of 2.4×10^{-6} may have mobilised more
511 residual H₂ than under strict capillary regime conditions.

512 During simultaneous injection of H₂ and brine, which may be representative of H₂ injection
513 into hydrodynamic aquifers or simulate the far field conditions, H₂ saturation and H₂

514 interconnected pore volume increased with increasing H₂/brine injection ratio (Fig. 6). This
 515 indicated that a lower brine flow is favoured over high flow environments in terms of
 516 optimising the H₂ storage operation. The structure was apparently not percolating in any of the
 517 simultaneous injection experiments as opposed to during 100% H₂ injection (Fig. 6 vs. Fig.
 518 4b), yet considering significant pressure differences of up to 0.05 MPa between inlet and outlet
 519 in simultaneous injection experiments which were not observed in experiments injecting 100%
 520 H₂, the connections between the H₂ clusters may have been broken when the injection (and
 521 thereby the pressure gradient) was stopped for the scan.

522 *4.4 Effect of capillary number on initial and residual saturation*

523 Classical pore-scale displacement theory predicts little change in residual phase saturation in
 524 response to increases in flow rate until N_C exceeds 10^{-6} - 10^{-5} . However, for most subsurface
 525 reservoirs there will be rapid decreases at N_C of 10^{-4} or more, when viscous forces become
 526 dominant⁵⁴. The bulk brine N_C applied in this study (2.4 - 9.4×10^{-6}) was within the range of little
 527 saturation change but exceeded the threshold of $N_C < 10^{-6}$ for which the flow generally is said
 528 to be capillary dominated⁵⁴. This may indicate that viscous forces caused a significant effect of
 529 N_C on the S_{nwr} in our experiments (Fig. 4b and d), and these forces are likely to be even greater
 530 at local scale than at bulk⁵⁴. The H₂ cluster size distribution after imbibition at $N_C = 9.4 \times 10^{-6}$
 531 was shifted with respect to the distribution at $N_C = 2.4 \times 10^{-6}$ (Fig. 7b), indicating a change in the
 532 wetting behaviour and supporting previous findings of preferential desaturation of larger
 533 clusters at higher N_C ⁶⁶.

534 We observed a 4% decrease in S_{nwi} in our experiments when bulk N_C was increased from
 535 1.7×10^{-8} to 6.8×10^{-8} (Fig. 4b and d). Critical nonwetting phase N_C of 2×10^{-8} and 10^{-5} during
 536 imbibition have been reported for water–gas systems and water-oil systems, respectively⁶⁷,
 537 indicating that the threshold of $N_C < 10^{-6}$ for capillary dominated flow⁵⁴ is not rigid. However,
 538 considering reported increases in the H₂ saturation after drainage with increasing N_C from

539 7.7×10^{-7} to 7.7×10^{-5} ²⁰, and acknowledging the small observed difference in S_{nwi} , we cannot
540 exclude that the effect of flow rate was down to experimental variability.

541 *4.5 Comparison to nitrogen*

542 The N₂ saturation was comparable to the H₂ saturation during drainage at similar N_C of $1-3 \times 10^{-8}$
543 ⁸ but the S_{nwr} after imbibition was ~20 % higher for N₂ than for H₂ (Fig. 4b and e). Using N₂ as
544 a proxy for H₂ in experimental drainage and imbibition studies is hence not advisable.
545 Considering the high degree of N₂ trapping, the use of N₂ as a cushion gas for H₂ storage
546 operations which could reduce operational costs⁸ seems favourable. Our results are lower than
547 a reported 64% initial N₂ saturation during drainage and 43% residual N₂ saturation during brine
548 (200 g/L NaI +15 g/L CaCl₂) imbibition in a Berea sandstone (20-22% porosity) at 5.5 MPa
549 and 20°C³⁷, and higher than 43% N₂ saturation during drainage in a Bashijiqike tight sandstone
550 (5.6% porosity) at 8 MPa and ambient temperature³⁶. The trend in the differences of the N₂
551 saturation in the above studies follows the same trend as the differences in the porosities of the
552 studied sandstones, with the Clashach sandstone (14% porosity) being intermediate between
553 the two other rocks. This indicates that porosity differences between the different rock types
554 applied in the above experiments defined the observed N₂ saturations, yet differences in the
555 pore throats dimensions may equally have contributed or caused this. Hence S_{nwi} and S_{nwr}
556 depend strongly on (the local conditions within) each rock, and these rock type/local effects
557 may mask any effect of injection conditions. However, trends in the rock-specific behaviour
558 will be controlled by pressure and flow conditions. It follows from this that any rock selected
559 for H₂ storage operations has to be very well characterized, in particular with respect to
560 porosity, tortuosity and capillary pressure curves.

561 *4.6 Suitability of the Clashach sandstone for hydrogen storage*

562 It has been postulated that the low viscosity of H₂ will cause the gas to travel swiftly, making
563 it little suitable for displacing brine¹⁰ and causing low H₂ injectivity. In this work, we showed
564 that from an injectivity and recovery perspective, untreated Clashach sandstone is suitable for
565 underground H₂ storage. However, sandstones aged by exposure to humic acids may be more
566 suitable analogue rocks for experimental investigations of H₂ storage in porous media¹⁹.
567 Considering that aging has previously been shown to alter the wettability of H₂ brine-quartz
568 systems from highly water-wet toward intermediate-wet¹⁹ the stated H₂ saturations for our
569 untreated outcrop Clashach sandstone are expected to increase during drainage and decrease
570 during imbibition, further the increasing suitability for H₂ storage.

571

572 **5. Conclusion**

573 In this work, a prima facie examination of H₂ flow and displacement processes in porous rock
574 was carried out as a function of capillary numbers of $1.2-6.8 \times 10^{-8}$ for H₂ and $2.4-9.5 \times 10^{-6}$ for
575 brine, and of pore fluid pressures between 2-7 MPa. Results showed no clear relation between
576 pore fluid pressure and H₂ saturation with ~50% of the pore space saturated with H₂ during
577 drainage at all pressures, and 20%, 24% and 43% of the initial H₂ trapped at 2, 5 and 7 MPa,
578 respectively, during imbibition at a capillary number of 2.4×10^{-6} , indicating that higher
579 pressure, i.e. deeper reservoirs are less favourable for H₂ storage. Injection of brine at higher
580 capillary numbers reduced capillary trapping and increased H₂ recovery. Hydrogen recovery
581 was distinct from N₂ recovery, suggesting that N₂ is a poor proxy for H₂. Based on these results,
582 we recommend more shallow, lower pressure sites for future H₂ storage operations in porous
583 media. Future work should aim to measure the influence of pressures above 7 MPa, elevated
584 temperatures and rock aging on H₂ and brine displacement processes at a range of different
585 capillary numbers. Dynamic experiments using synchrotron light sources are recommended to
586 examine displacement mechanisms closely.

587

588 Author Contributions

589 The experimental kit was built by Butler and Thaysen. Experiments were carried out by Thaysen,
590 Butler, Hassanpouryouzband, Freitas. Alvarez-Borges, Atwood and Humphreys were the
591 supporting beamline scientists during the experiments at Diamond Light Source. Image
592 reconstruction and tomographic analysis was carried out by Thaysen. Interpretation of the results
593 was by Thaysen, Butler, Hassanpouryouzband, Krevor and Heinemann. The manuscript was
594 written by Thaysen with review by Butler, Krevor Heinemann, Hassanpouryouzband, Freitas and
595 Edlmann. All authors have given approval to the final version of the manuscript.

596

597 Declaration of interest

598 The authors declare no competing interest.

599

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608

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Supplementary Information

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Pore-scale imaging of hydrogen displacement and trapping in porous media

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814 Eike M. Thaysen^{2*}, Ian Butler¹, Damien Freitas¹, Aliakbar Hassanpouryouzband¹, Fernando815 Alvarez-Borges², Samuel C. Krevor³, Niklas Heinemann¹, Robert Atwood², Katriona

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Edlmann¹817 ¹School of Geoscience, Grant Institute, The King's Buildings, The University of Edinburgh, James Hutton Road,

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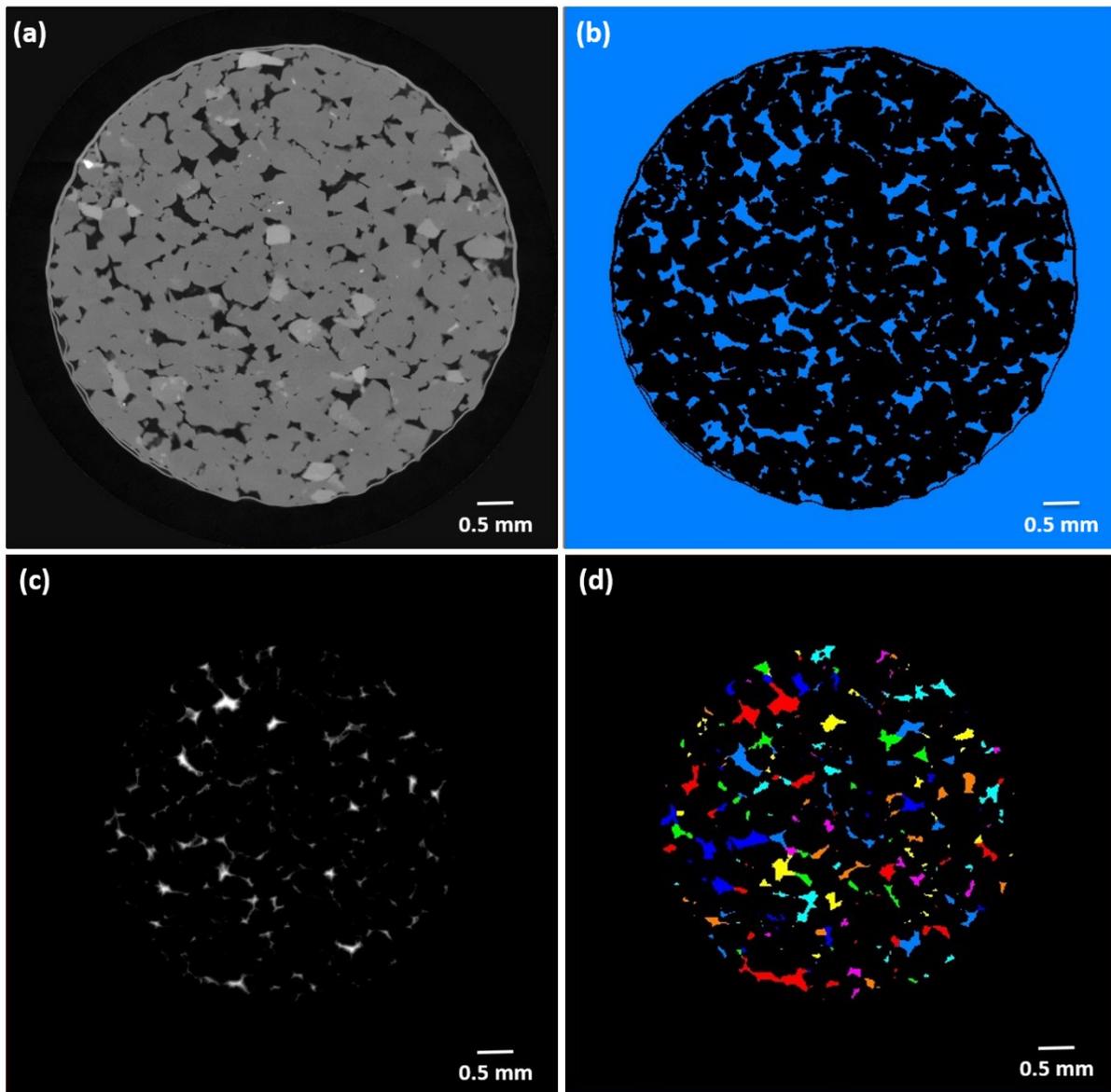
Edinburgh, EH9 3FE, United Kingdom

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² Diamond Light Source Ltd, Harwell Campus, Didcot OX11 0DE, UK820 ³ Department of Earth Science and Engineering, Imperial College London, London SW7 2AZ, United Kingdom

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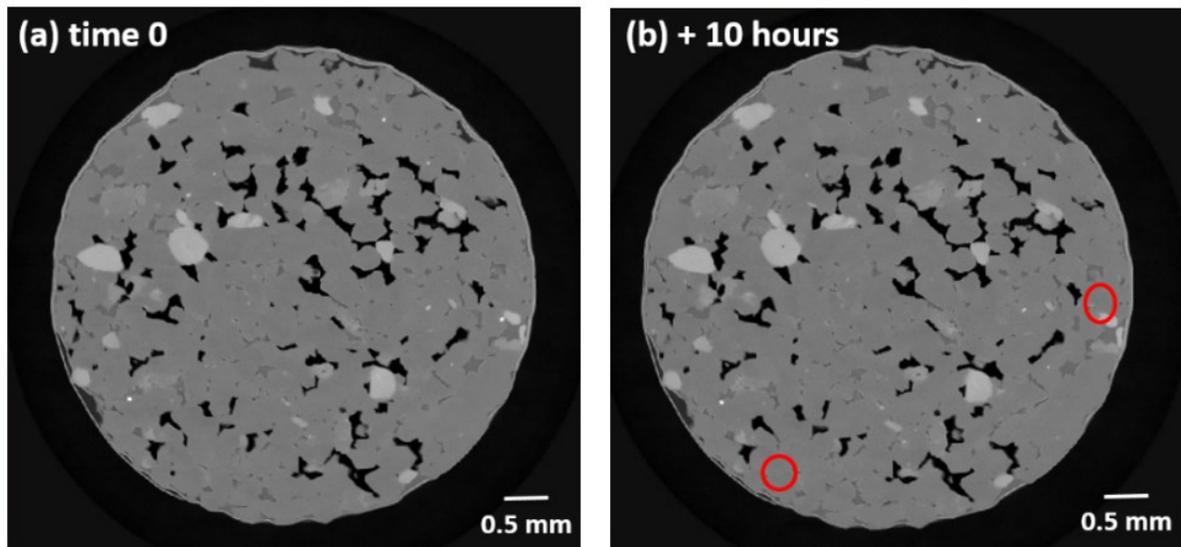
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SI Figure S1: Image processing workflow for UoE experiments. **(a)** The image after reconstruction and filtering. **(b)** The image after segmenting using a global threshold. **(c)** The outer part of the rock core touching the Al foil was removed through application of a mask, and a chamfer distance map of the pore space was calculated by assigning each voxel the distance from it to the nearest pore wall. **(d)** This was then used to separate the pore space into individual pores by calculating the water-shed basins of the distance map.



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831 SI Figure S2: H₂ stability at 7 MPa injection pressure and a flow rate of 20 $\mu\text{l min}^{-1}$ (capillary number $2 \cdot 10^{-8}$) at
832 time zero **(a)** and after ten hours **(b)**. Red circles indicate subtle differences in the hydrogen filled pore volume
833 after ten hours. The difference in the hydrogen saturation for the total scanned rock volume was within error at
834 49.55% and 49.53% for time 0 and after ten hours, respectively.

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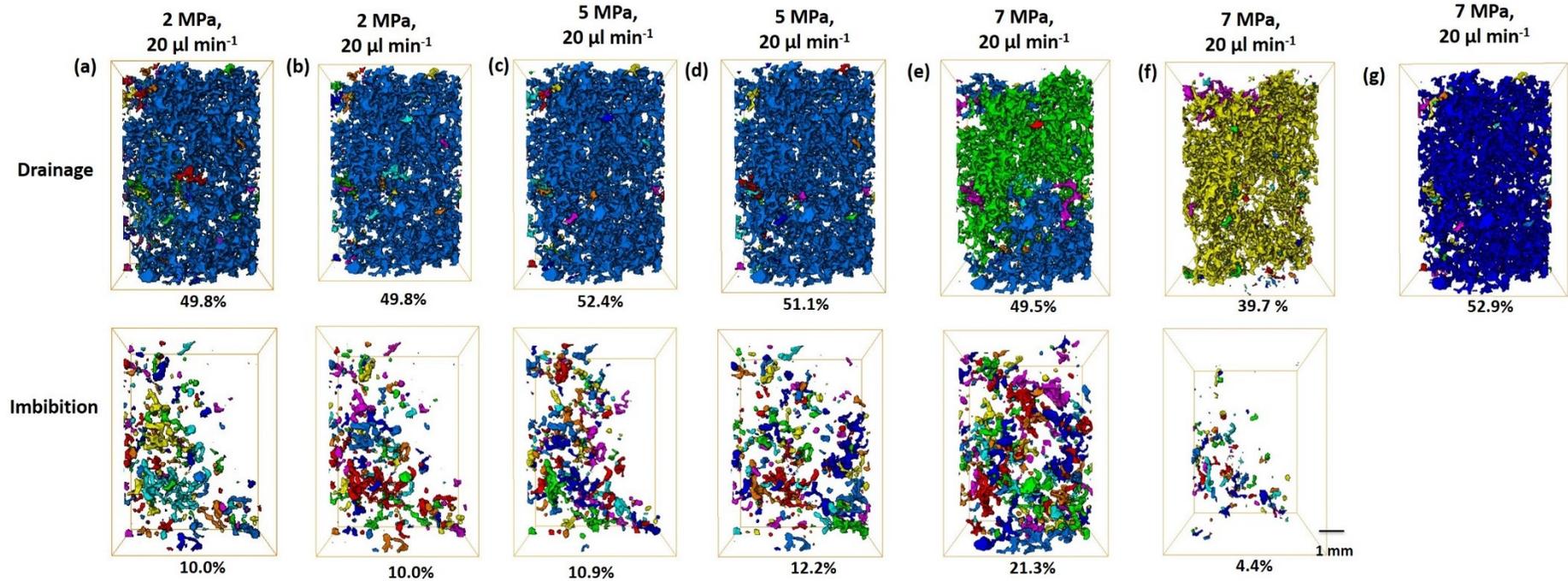
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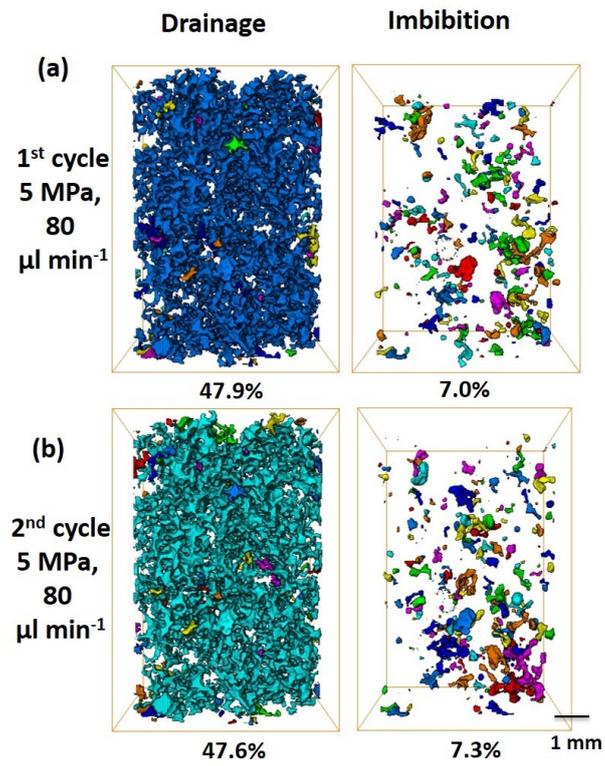
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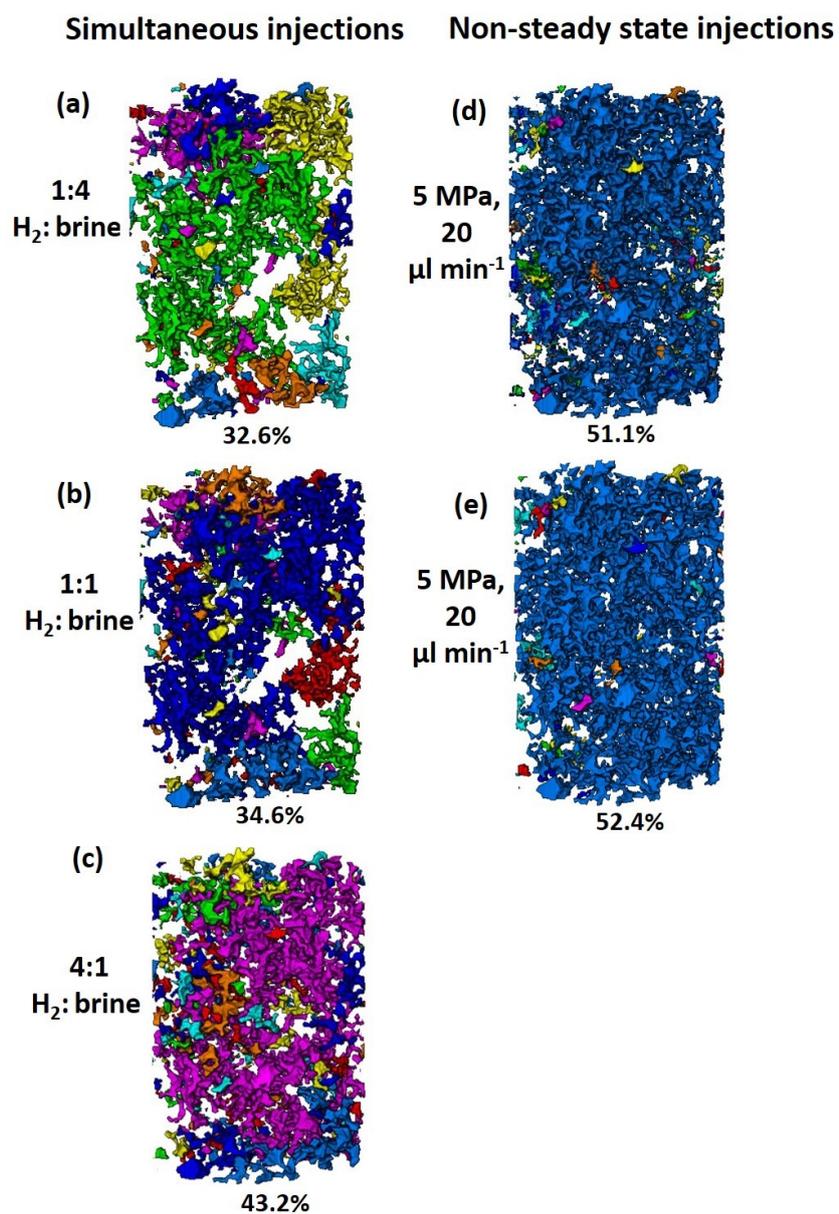
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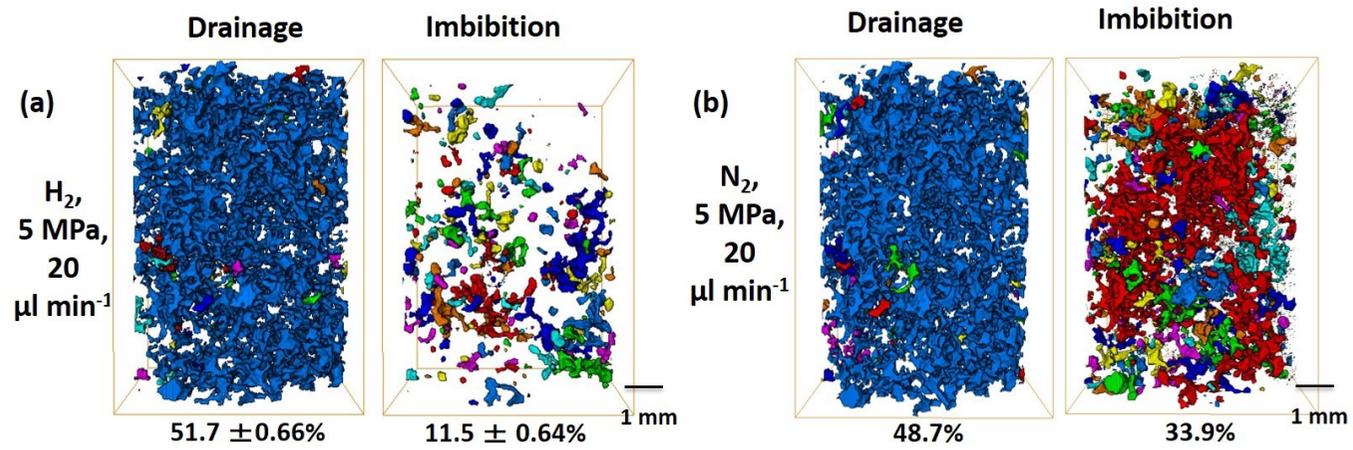
SI Figure S3: Enlarged view of 3D renderings of H_2 with saturation percentages in UoE experiments. Discrete clusters were rendered in colours, where mainly one colour marks one large, connected cluster and different colours indicate several, not connected clusters. **(a-d)** Effect of pore fluid pressure on H_2 clusters and saturation after drainage and after primary imbibition. **(a)** 2 MPa, **(b)** 5 MPa and **(c-e)** 7 MPa, all at a constant flow rate of $20 \mu\text{l min}^{-1}$ corresponding to capillary numbers of 1.7×10^{-8} and 2.4×10^{-6} , respectively. Experiments were repeated once. For experiments at 2 MPa and 5 MPa averages and standard errors for the H_2 saturation are reported. For experiments at 7 MPa, due to the discrepancy in the results, both results are visualized **(c-d)**.



SI Figure S4: Cyclic H_2 injection into Clashach sandstone at 5 MPa and $80 \mu\text{l min}^{-1}$ corresponding to a capillary number of 9.4×10^{-6} , showing no significant differences in H_2 saturation and H_2 connectivity after primary drainage and imbibition as compared to after secondary drainage and imbibition.



SI Figure S5: Comparison of the distribution of the H₂ clusters in the pore space between simultaneous H₂ and brine injection experiments (a-c) and the two repetitions of the non-steady state H₂ injection (d) and (e), all at 5 MPa and a total flow rate of 20 μl min⁻¹.



SI Figure S6: Nitrogen clusters and saturations during drainage and imbibition at 5 MPa pore fluid pressure and a flowrate of 20 $\mu\text{l min}^{-1}$