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- 7
- 8 Time-resolved infrared stimulated luminescence of the blue and yellow-green emissions –
- 9 insights into charge recombination in chemically and structurally different alkali feldspars
- 10

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17

18 Abstract

19 Time-resolved luminescence measurements can be used to explore luminescence processes in 20 minerals and the defects involved. It has also been applied to feldspars and knowledge has been gained 21 regarding potential crystal defects associated with luminescence productions in these minerals, but 22 also regarding processes governing electron-hole recombination leading to luminescence emission.

Here we present time-resolved infrared stimulated luminescence (IRSL) signals measured for a range of mineralogically well characterised single crystal alkali feldspars. We explore time-resolved luminescence for the blue (~410 nm) and the yellow-green emission (~550 nm) in response to different irradiation doses and by comparing different IRSL signals. Firstly, we explore whether the lifetimes measured represent excited state or recombination lifetimes. Secondly, we investigate sampledependent changes in blue and yellow-green time-resolved signals and link those to physical properties of the samples.

Our results show that the timescales on which the blue and the yellow-green emission occur differ
 significantly, with the blue signal on the μs-scale, and the yellow-green emission on the ms-scale. We

32 do not observe any dependence of the time-resolved signal on signal integration, dose given or IRSL 33 signal measured. However, inter-sample variability is shown for both emissions. In the blue we only 34 observe small differences in decay time scale between single-phase feldspars and perthites, however 35 larger differences are measured between samples that were artificially disordered compared to 36 ordered feldspars. Longer lifetimes observed for disordered feldspars are suggested to be linked to 37 either changes in the recombination centre or to increased band-tail states transport due to an increase in the width or density of the sub-conduction band-tail states. The data indicates the potential 38 39 of using time-resolved IRSL of the blue emission to get an indication of the state of order of a feldspar. 40 For the yellow-green emission slow signal decays are observed for single-phase feldspars, likely 41 indicating a spin-forbidden transition. Interestingly, similar lifetimes were observed for K- and Na-42 feldspar end members.

43 Keywords

44 Time-resolved luminescence, lifetime, feldspar, recombination, band-tail states

45 1 Introduction

46 Conventional optically stimulated luminescence measurements use continuous wave (CW) stimulation 47 to excite trapped electrons and enable their migration through the crystal lattice, resulting in 48 recombination at a recombination centre. If the recombination process is radiant it leads to the 49 emission of photons. In time-resolved luminescence measurements samples are stimulated using pulsed light, meaning that a time-resolved measurement consists of multiple stimulation light pulses, 50 51 separated by so-called off-times, during which no optical stimulation occurs (Sanderson and Clark, 52 1994; Lapp et al., 2009). This method not only enables an improved discrimination of excitation and 53 emission wavelengths, but also gives information on recombination processes in minerals (Sanderson 54 and Clark, 1994). During on-times of pulsed stimulation, electrons are excited from the ground state 55 of the electron trapping centres and migrate to recombination centres, while the off-time allows 56 investigations into the decay of different luminescence emissions after the optical stimulation has 57 been turned off. Thus, time-resolved luminescence measurements make it possible to measure 58 recombination lifetimes and excited state lifetimes of the luminescence centres.

59 Feldspars are the most abundant mineral in the Earth's crust. Feldspars are complex framework 60 silicates, which occur in a variety of forms along two solid solution series (KAlSi₃O₈ – NaAlSi₃O₈, termed 61 alkali feldspars, and NaAlSi₃O₈ – CaAl₂Si₂O₈, termed plagioclase). The chemical composition and water 62 content of the melt, pressure conditions and temperature gradients during cooling influence the type 63 of feldspar which crystallises from the melt (e.g. Deer et al., 2013). The temperature gradient at which 64 feldspars cool dictate their structural state: the site occupancy of Al³⁺ ions on the (Si, Al)-framework. 65 Feldspars with a high structural state have a disordered framework, whereas feldspars with a low 66 structural state are ordered (cf. high sanidine compared to low microcline, high albite compared to 67 low albite). To accommodate K⁺ and Na⁺ ions, which have very different ionic radii, in the alkali feldspar 68 crystal during cooling, exsolution occurs and perthites form (e.g. Deer et al., 2013). Especially alkali 69 feldspars have been studied extensively for geochronological applications (e.g. Godfrey-Smith et al., 70 1988; Hütt et al., 1988; Duller, 1997; Buylaert et al., 2012). For feldspars, time-resolved luminescence 71 measurements have previously been performed to gain insights into electron-hole recombination 72 processes (e.g. Sanderson and Clark, 1994; Jain and Ankjærgaard, 2011) and to understand defects 73 involved in luminescence production and their locations within the crystal lattice (e.g. Clark and Bailiff, 74 1998). Despite this previous research, it is still unknown, how different chemistry and structure of 75 feldspars impact luminescence production and how these factors influence excited state and 76 recombination lifetimes of feldspar infrared stimulated luminescence (IRSL). In this paper we use time-77 resolved IRSL with detection in the blue (~410 nm) and yellow-green (~550 nm) wavelength regions to 78 explore recombination processes and potentially involved defects in chemically and structurally 79 different alkali feldspars.

80 Time-resolved luminescence of feldspars with emissions in the UV, blue, yellow-green or red to near-81 infrared, have been investigated, mainly to understand the defects and transitions involved in the 82 recombination process (cf. Clark et al., 1997; Clark and Bailiff, 1998; Tsukamoto et al., 2006; Prasad et 83 al., 2016). The blue luminescence emission has been associated with a hole centre located on Al-O-Al 84 bridges (Finch and Klein, 1999; Riedesel et al., 2021a) and it has been shown that its intensity and 85 stability is dependent on the degree of order on the Si, Al-framework and on the presence and type of 86 interfaces in perthites (Riedesel et al., 2021a). However, it is unknown whether lifetimes of the blue 87 emission are also affected by changes to the crystal structure. Previous research investigating lifetimes 88 of the blue emission indicated very fast decaying signals with lifetimes on the sub-us to few us-scale (e.g. Clark et al., 1997; Clark and Bailiff, 1998). Besides those very fast lifetimes, Ankjærgaard and Jain 89 90 (2010) observed very long lifetimes for the blue emission on the ms- to s-scale. These authors suggest 91 band-tail state transport as the source of these long lifetimes.

For the yellow-green emission two likely defect sites have been proposed: (i) The emission at ~560 nm has been assigned to Mn²⁺ substituting for Ca²⁺ in plagioclases resulting in a spin-forbidden transition with long lifetimes on the ms-scale (e.g. Geake et al., 1971, 1973, 1977; Telfer and Walker, 1978). These long lifetimes are due to the transition within the 3d shell, linked to the electrons being influenced by the crystal field strength, site symmetry and them being strongly coupled to lattice vibrations (Shi et al., 2010) ; (ii) Alternatively, the emission was suggested to result from Mn²⁺ substituting for Al³⁺ on tetrahedral sites (T sites) in alkali feldspars, creating a lattice stabilising hole centre on Si⁴⁺-O⁻-Mn²⁺

99 (Telfer and Walker, 1978; Kirsh et al., 1987). This second possibility would result in shorter lifetimes, 100 due to the defect being located at a lattice site. Clark and Bailiff (1998) have indeed observed very fast 101 lifetimes (on the ns-scale) for the emission at ~560 nm, supporting a lattice site defect for the yellow-102 green emission. From phosphorescence data Prasad et al. (2016) observed dominant lifetimes of the 103 yellow-green emission of the order of hundreds of microseconds to less than a fraction of a millisecond 104 – also an observation against a spin forbidden transition of a defect located on cation sites (M sites). 105 Riedesel et al. (2021a) observed a strong yellow-green emission centred around 560 nm in alkali 106 feldspar end members albite and microcline, with all three samples having zero to negligible 107 concentrations of MnO as determined by X-ray fluorescence analysis. Thus, the source of the yellow-108 green emission in chemically different alkali feldspars needs further investigation.

109 Besides using time-resolved luminescence to link certain defect types to luminescence emissions, 110 feldspar time-resolved luminescence has been used to improve the understanding of electron-hole 111 recombination processes and to infer possibilities to extract a more stable luminescence signal for 112 dating. As part of this work, it has been debated whether the time-resolved luminescence signal of 113 feldspars represents the excited state lifetime of the luminescence centre (the time needed for the 114 electron to transition from the excited state to the ground state of the luminescence centre, cf. e.g. 115 Clark et al., 1997) or the recombination lifetime (the time needed for the excited electron to move 116 from the electron trapping centre to a recombination centre) (e.g. Ankjærgaard et al., 2009; 117 Tsukamoto et al., 2010), and which role band-tail states play in this process (e.g. Jain and Ankjærgaard, 2011; Pagonis et al., 2012). 118

119 The current understanding of feldspar luminescence is primarily based on a proximity model, where 120 the electron-hole pairs are separated by a distribution of distances and where it is understood that 121 recombination will occur at the nearest recombination centre (e.g. Huntley, 2006; Jain et al., 2012). 122 Feldspars containing a higher density of recombination centres will thus have less stable luminescence 123 signals, compared to feldspars with a lower density, because it will be more likely for electrons to 124 recombine at a proximal luminescence centre, even more likely through tunnelling processes (e.g. 125 Huntley, 2006; Jain et al., 2012). Subsequently, one would expect faster recombination lifetimes in a 126 system with a higher density of recombination centres, compared to systems with a lower 127 recombination centre density. Ankjærgaard et al. (2009) explored whether the time-resolved luminescence signal of feldspars (green (532 nm) laser or blue (470 nm) LED stimulation and UV 128 129 detection) reflected the excited state or the recombination lifetime. They found arguments for either 130 of the two processes controlling the lifetimes measured. One argument supporting the recombination lifetime hypothesis was shown by time-resolved optically stimulated luminescence (OSL) measured at 131 132 different stimulation temperatures resulting in the same decay shape, i.e. not giving rise to thermal

133 quenching. However, Ankjærgaard et al. (2009) also found counter arguments, thus supporting the 134 excited state lifetime hypothesis: A systematic decrease in lifetime with increasing preheat 135 temperature was observed for feldspar time-resolved OSL. The authors argued that longer lifetimes 136 were expected for higher preheat temperatures, as a high preheat would result in a depletion of 137 recombination centres, making recombination more difficult and thus increasing the recombination 138 lifetime. Ankjærgaard et al. (2009) further explored whether different stimulation times could affect 139 the lifetimes and decay shape of the signals, expecting an increase in lifetimes with increasing 140 stimulation time. If the lifetimes measured would correspond to recombination lifetimes, then one 141 would expect longer lifetimes with decreasing hole concentrations due to longer stimulation lifetimes. 142 However, no clear trend was observed. Tsukamoto et al. (2010) compared time-resolved OSL and time-143 resolved OSE (optically stimulated exo-electron) measurements of quartz, feldspar and NaCl and found 144 that the decay of time-resolved OSE signals are faster than time-resolved OSL in the case of NaCl and 145 quartz, with time-resolved OSE signals decaying on timescales <1 µs, while NaCl and quartz time-146 resolved OSL signals decay with a lifetime of ~40 µs. time-resolved OSE and time-resolved OSL of 147 feldspars decay on similar timescales. The very fast decaying time-resolved OSE signals are understood 148 as rapid emptying of the electron population in the conduction band immediately after turning off the 149 stimulation light source. In contrast, the slower decay of the time-resolved OSL signals is interpreted 150 as arising from relaxation processes within the recombination centre, thus being governed by the 151 excited state lifetime of these defects and the internal transitions.

152 Since the physical source of various time-resolved luminescence signals is still under debate, and it is 153 still unclear whether the off-time of time-resolved signals represents the excited or the recombination 154 lifetime, different approaches have been used to describe time-resolved luminescence signals, particularly focussed on the off-time signal. Most commonly a sum of multiple first order exponential 155 156 functions have been used to describe the time-resolved off-time signal (e.g. Clark et al., 1997; Clark 157 and Bailiff, 1998; Tsukamoto et al., 2006; Ankjærgaard et al., 2009). However, feldspar luminescence 158 is not a first order process (e.g. Huntley, 2006) and thus it has been proposed that fitting a sum of 159 multiple first order exponential functions might be inappropriate (e.g. Ankjærgaard, 2009; Pagonis et 160 al., 2012, 2016). Jain and Ankjærgaard (2011) thus visually separated the off-time signal into a "fast" 161 and a "slow" time-resolved signal, with the "fast" signal describing the initial decay, and the "slow" 162 signal governing the time-resolved signal after 70 µs in the off-time. Pagonis et al. (2012) applied a 163 linear combination of exponential and stretched exponential functions. These authors interpreted the 164 part of the time-resolved signal fitted by a stretched exponential as being caused by transport via the 165 band-tail states. Later Pagonis et al. (2016) developed analytical equations describing feldspar time-166 resolved luminescence, which are based on the nearest-neighbour recombination model by Jain et al. 167 (2012). However, there are still many unknowns in the physical processes involved in luminescence

production in feldspars, so no definite answer can yet be given regarding the optimal signal analysisprocedure.

170 The research presented in this paper is focussed on creating a better understanding of defects and processes involved in blue and yellow-green luminescence production. Therefore, we measure time-171 172 resolved luminescence resulting from pulsed IR stimulation with emissions in the blue and yellow-173 green wavelength region of a suite of representative single crystal alkali feldspars including single-174 phase feldspars, perthites and artificially disordered feldspars. These are explored to investigate 175 whether the measured time-resolved luminescence signal and its lifetimes and decay shape are 176 governed by the recombination or excited state lifetime and which role band-tail states might 177 potentially play. We compare time-resolved luminescence resulting from different IRSL signals, 178 different irradiation doses and of chemically and structurally different feldspars.

179 2 Materials and methods

180 2.1 Samples

181 Samples investigated here are single crystal alkali feldspar specimens. Their chemical composition, 182 structural state and mineral phases present have been characterised previously and details can be 183 found in Riedesel et al. (2021a). The sample suite represents the chemical and structural range of the 184 alkali feldspar solid solution series and its end members. The chemical composition, mineral phases 185 present and the structural state of the samples was determined using X-ray fluorescence and X-ray diffraction and a summary of the results is given in Table 1. FSM-13 is a single-phase microcline (98.5 186 187 % K-feldspar) and CLBR a single-phase albite (0.5 % K-feldspar). FSM-3 (82.5 % K-FS), FSM-6 (74.4 % K-FS) and FSM-5 (74.8 % K-FS) are perthites. FSM-3 and FSM-6 are cryptoperthites consisting of albite 188 189 and microcline (FSM-3) or orthoclase (FSM-6). FSM-5 is a macroperthite consisting of albite and 190 microcline. FSM-13LH and FSM-6LH are artificially disordered samples. To obtain these samples, 191 powdered sample material of FSM-13 and FSM-6 was heated in Pt crucibles in a furnace to 1050 °C 192 and then rapidly cooled to room temperature to retain the disordered high-temperature structure, 193 with Al³⁺ ions distributed randomly across tetrahedral sites on the (Si,Al)-framework. Details of these 194 experiments and a comparison of the X-ray diffraction pattern of the ordered/disordered sample pairs 195 can be found in Riedesel et al. (2021a). For each sample three aliquots were measured.

Table 1. Details regarding the chemical composition and mineral phases present for samples investigated. The
 chemical composition in % feldspars, was done using stoichiometric conversion of the semi-quantitative XRF data.
 Present phases estimated based on semi-quantitative XRD analyses.

Samp		Origin		Che	mical cor	nposit	tion (F	S %)		Phases presen	t		
Samp	ie iD	Origin	к	-FS	ſ	Na-FS		Ca-FS	Microcline	Orthoclase		Sanidine	 Albite
FSM-1	13	Brazil	9	8.5	1	1.5		0.0	100.0	-		-	
FSM-1	L3LH	Brazil	9	8.5	1	1.5		0.0	100.0	-		-	

FSM-3	Granite pegmatite, Toe Head, South Harris, Scotland, UK (Cunningham, 1981)	82.5	17.2	0.3	78.0	-	-	22.0
FSM-5	Unknown	74.8	25.20	0.0	57.0	-	-	43.0
FSM-6	Granite pegmatite, Trezaise Quarry, Cornwall, UK (see Ussher et al., 1909)	74.4	25.3	0.3	-	38.0	-	62.0
FSM-6LH	Trezaise Quarry, Cornwall, UK	74.4	25.3	0.3	-	-	100.0	-
CLBR	Pegmatite, Golonca District, Minas Gerais, Brazil (Cassadanne and Roditi, 1996)	0.5	99.3	0.2	-	-	-	100

199

200 **2.2 Instrumentation and measurement protocol**

201 Time-resolved luminescence measurements were made on a Risø TL/OSL DA20 reader equipped with a ⁹⁰Sr/⁹⁰Y beta source delivering ~0.1 Gy s⁻¹ at the sample position and a Detection And Stimulation 202 203 Head (DASH) including an automated filter changer (Lapp et al. 2015). For the measurements 204 presented in this paper infrared stimulated luminescence (IRSL) was detected using a PDM 9107Q-AP-205 TTL-03 (sensitive wavelength region: 160-630 nm) photomultiplier tube. For detection of the blue 206 emission (~410 nm) we used a combination of Schott BG39 (2 mm) and BG3 (3 mm) filters and for the 207 yellow-green emission (~550 nm) a combination of Schott BG39 (2 mm) and OG550 (2 mm) filters. 208 Dependent on the brightness of the sample ND 1.0 or ND 2.0 filters were added to the chosen filter 209 combination. The emission windows of the individual filters and the chosen filter combinations are 210 displayed in Figs. 1D and 1E and a comparison of the transmission on linear and logarithmic scales can 211 be found in Fig. S1 in the supplementary material. The stimulation was achieved using 850 nm (300 212 mW cm⁻²) IR LEDs, operating at 90 % power.

213 Time-resolved luminescence measurements were possible through a photon timer attachment 214 (TimeHarp 260) in combination with the pulsed optically stimulated luminescence (OSL) unit (Lapp et 215 al., 2009). The TimeHarp 260 photon timer enables the detection of individual photons with a time 216 resolution of up to 1 ns. The pulsed IRSL plug-in board allows on- and off-times to be adjusted between 217 0.6 µs and 10 s. Additionally, different gating intervals can be selected. For time-resolved IRSL 218 measurements with emission detection in the blue wavelength region we used on-times of 50 μ s, 219 followed by an off-time of 200 µs. Due to the much slower decay of the yellow-green time-resolved 220 IRSL signal, on-times were 5 ms, followed by an off-time of 45 ms. By defining these on-times for the 221 blue and yellow-green emissions, respectively, we ensured that the time-resolved signal in some of 222 the samples investigated reached a stable on-time signal, prior to switching off the LEDs. The time-223 resolved IRSL signals for both emissions were recorded over a total stimulation time of 200 s.

224 Time-resolved IRSL measurements were performed using a post-IR₅₀IRSL₂₂₅ protocol with a 60 s 225 preheat at 250 °C. The IR stimulation time was 200 s for the IRSL₅₀ and post-IR IRSL₂₂₅ signals. Prior to 226 each measurement using the post-IR IRSL protocol the samples were annealed to 450 °C to ensure 227 complete signal resetting (Table 2). All samples were irradiated after the 450 °C clean-out. Blue time-228 resolved IRSL measurements were performed after irradiation doses of 50, 200 and 800 Gy. Yellow-229 green time-resolved IRSL measurements were conducted after doses of 200 or 800 Gy. In case of 230 samples FSM-5 and FSM-6 the yellow-green emission was not intense enough after a dose of 200 Gy, 231 thus these samples were irradiated with 400 Gy instead. Results of these dose-dependent 232 measurements are presented in section 3.1.3 for the blue emission. Results of these experiments for 233 the yellow-green emission are not shown, but results obtained from fitting are given in Tables S3a and S3b in the supplementary material. 234

235 Table 2: Measurement protocol used.

Step	Treatment	Purpose
1	TL to 450 °C, 1 °C s ⁻¹	Signal clean out
2	Beta dose 50, 200, 400 or 800 Gy	Irradiation, dose dependent on experiment
3	Preheat to 250 °C, 2 °C s ⁻¹ , 60 s	
4	IRSL ₅₀ , 2 °C s ⁻¹ , 200s	Obtain time-resolved IRSL ₅₀ signal
	Blue emission: 50 μ s on-time, 200 μ s off-time	
	Yellow-green emission: 5 ms on-time, 45 ms off-	
	time	
5	post-IR IRSL ₂₂₅ , 2 °C s ⁻¹ , 200s	Obtain time-resolved post-IR IRSL225 signal
	Blue emission: 50 μ s on-time, 200 μ s off-time	
	Yellow-green emission: 5 ms on-time, 45 ms off-	
	time	
-		

236 Fitting of time-resolved luminescence signals was done in R using the nls() function (Bates and DeRoy, 237 2018) using equation 1 for the on-time signal and equation 2 for the off-time signal, where I is the 238 intensity at time t, A_i is the saturation intensity of the i'th component and a_i the intensity at time t_1 , 239 where t_1 is the on-time duration, and k is a constant, representing a stable linear background 240 (Chithambo, 2003; Tsukamoto et al., 2006). The data were normalised to the intensity of the last data 241 point of the on-time. The sole purpose of fitting the data was to enable a numerical comparison 242 between the samples used in this paper and to also be able to compare the data obtained here to 243 previously published time-resolved IRSL data (e.g. Clark and Bailiff, 1998; Tsukamoto et al., 2006, 244 2010). It is not our intention to make any suggestions regarding the physical processes by using the 245 numerical results obtained through fitting the data using a sum of multiple first order exponentials. 246 Later sections discussing potential physical properties of the samples and their influence on the timeresolved luminescence signals are based on qualitative descriptions and sample-to-samplecomparisons.

$$I(t) = \sum A_i \left[1 - \exp\left(\frac{-t}{\tau_1}\right) \right]$$
[1]

$$I(t) = \sum a_i \exp\left[-\left(\frac{t-t_1}{\tau_1}\right)\right] + k$$
[2]

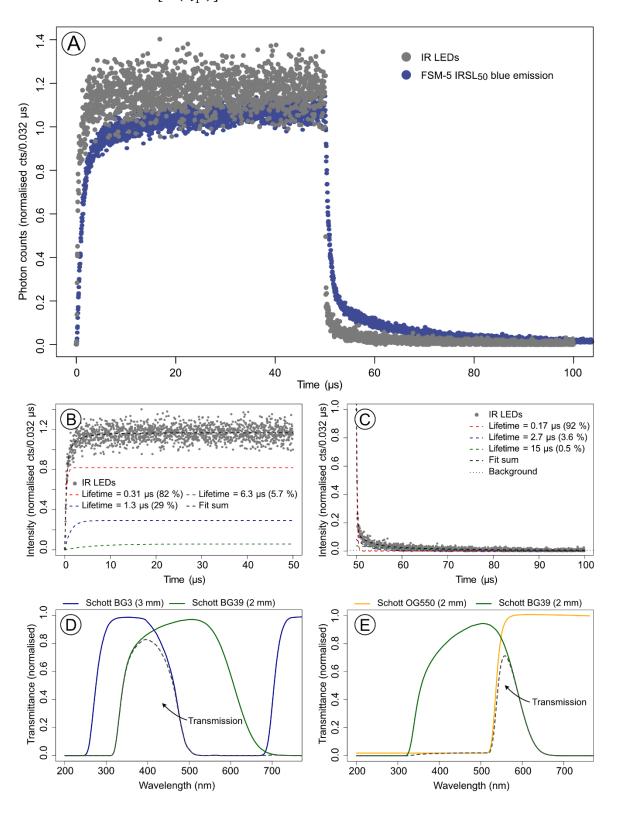


Fig. 1. A) The response of the IR LEDs measured through a ND filter for deadtime correction, and the signal was recorded using a red-sensitive photomultiplier tube (PMT). The signal of FSM-5, the fastest decaying signal from all samples measured, was recorded through the BG39 + BG3 combination visualised in D and was detected using a UV-sensitive PMT. This sample was chosen in comparison as it has the fastest decay in the off-time. B) Fitted LED on-time and C) off-time signal. The signal was normalised to the last point of the on-time, and the scatter of the data means that some measurement points lie above 1. D) Transmission of the BG3 and BG39 filter combination.

251

252 3 Blue time-resolved IRSL emission – Results and Discussion

253 The blue luminescence emission (~410 nm) is the emission commonly detected when using feldspars 254 for luminescence dating, thus this emission and potential variations of this emission between different 255 feldspars is of particular interest. In the following we investigate variations in the lifetimes of on- and off-times of the blue time-resolved IRSL signals as a function of integration interval, irradiation dose, 256 257 and IRSL signal investigated. To enable a direct comparison all on- and off-time signals were fitted using 258 a sum of multiple first order exponential functions. We have also attempted to use a linear sum of 259 single exponential function and stretched exponential function (Pagonis et al., 2012) to describe our 260 data. However, fitting our data using the equation given by Pagonis et al. (2012) in R and SigmaPlot 261 yielded parameter values of infinity. The only samples we were able to fit using a linear sum of single 262 exponential function and stretched exponential function were the disordered samples FSM-13LH and 263 FSM-6LH. Thus, we only present lifetimes based on fitting with a linear sum of multiple first order 264 exponential functions.

265 **3.1 Lifetimes**

In the following we first investigate the effects of changing the integration interval, the IRSL signal (IRSL₅₀ compared to post-IR IRSL₂₂₅) or the given dose on the on- and off-time lifetimes. To enable a numerical comparison, we here use the results obtained from fitting using a sum of multiple exponentials.

270 All results presented in the following sections are based on time-resolved luminescence signal fitting 271 using between one and four exponential functions (eq. 1 and 2). The number of components used for fitting is based on the lowest residual signal obtained from fitting with different numbers of 272 273 components, thereby avoiding redundancy in the fitting. Using this approach, we obtained lifetimes 274 ranging from 0.5 µs to over 100 µs. Faster lifetimes could not be detected due to the decay time of the 275 IR LEDs (~0.3 µs, cf. Fig. 1A, B, C). Generally, lifetimes could be arranged into three lifetime groups for the on-time signal: $<5 \ \mu$ s, 8-20 μ s and $> 20 \ \mu$ s, and into four lifetime groups for the off-time: $< 1 \ \mu$ s, 3-276 277 9 μ s, 12-16 μ s and > 20 μ s. This grouping was done by identifying similar lifetimes between different 278 samples and between the two signals investigated. The grouping is only used for visualisation purposes 279 in figures and tables. Differences in on- and off-time lifetimes might be related to the different time

- 280 intervals used for on- and off-times: 50 μs on-time compared to 200 μs off-time, leading to an
- 281 improved description of the longer off-time duration compared to the shorter on-time. All numerical
- 282 fitting results can be obtained from the supplementary material.

284 Integration intervals

285 The time-resolved IRSL signals were recorded over a stimulation period of 200 s, for the IRSL₅₀ and 286 post-IR IRSL225 signal, respectively. An example of IRSL decay curves measured during the on- and off-287 time is given for sample FSM-13 in Fig. S2. Using the software PTanalyse, we extracted photon arrival 288 time distributions of the following time intervals for a comparison: 0-2 s, 0-5 s, 0-10 s, 0-50 s and 0-289 200 s, 10-50 s, 50-100 s and 100-200 s. To be able to make an absolute comparison, we selected two 290 samples, FSM-3 and CLBR, and fitted the extracted photon arrival time distributions of the on- and off-291 time of the IRSL₅₀ and post-IR IRSL₂₂₅ signals using equation 1 and 2. Figure S3 in the supplementary 292 material shows the results of FSM-3, results for CLBR are not shown, but are similar to those of FSM-293 3. We did not observe any significant and consistent changes in lifetime with integration interval 294 selected for either of the IRSL signals or any samples. However, increasing the integrated time interval 295 resulted in a better goodness of fit, expressed as the square sum of residuals of the fit, with the best 296 fit obtained when integrating the photon arrival time distribution of the entire IRSL signal (total 200 s 297 stimulation). Subsequently, all data presented in this paper are based on whole IRSL signal integration.

298 Signals investigated – IRSL₅₀ compared to post-IR IRSL₂₂₅

Post-IR IRSL signals are observed to be more stable compared to the IRSL₅₀ signal measured as part of the post-IR IRSL protocol (e.g. Thomsen et al. 2008), likely due to the post-IR IRSL signal probing more distant electron-hole-pairs (Jain and Ankjærgaard, 2011). Time-resolved IRSL signal differences have also been observed: Jain and Ankjærgaard (2011) defined a "fast" and a "slow" time-resolved signal and found different behaviour of these two for the IRSL and post-IR IRSL signals measured in the UV at elevated temperatures, and that the relative signal intensity between the fast and slow signal contribution changes.

306 Here we compare lifetimes obtained for the two IRSL signals measured in the blue using a post-IR₅₀ IRSL₂₂₅ protocol. Figure 2 shows lifetimes of all samples obtained by fitting on- and off-time signals of 307 the IRSL₅₀ and post-IR IRSL₂₂₅. Figs. 2A and B show a direct comparison of all measured lifetimes and 308 309 their distribution around a 1:1 line, for the on- and off-times respectively. Especially at faster lifetimes 310 and in the off-time signal, IRSL₅₀ and post-IR IRSL₂₂₅ yield similar results and even the slower signals 311 show results scattered around the 1:1 line (Fig. 2 A, B). Sample-dependent lifetimes presented in Figs. 312 2C and D indicate variations in the presence of different lifetimes in the samples investigated. Further 313 details regarding sample-to-sample variations are discussed qualitatively in section 3.2. However, what 314 is also visible from Figs. 2C and D is that in some samples, some lifetimes only occur in either the IRSL₅₀ 315 or the post-IR IRSL225 signal (e.g. Lifetime 1 in the on-time of sample FSM-6). Such points are not visible 316 from the diagrams in Figs. 2A and B. Despite these minor deviations and some observed variability, it

317 can be concluded that overall similar lifetimes can be obtained for all samples regardless of the IRSL

318 signal chosen.

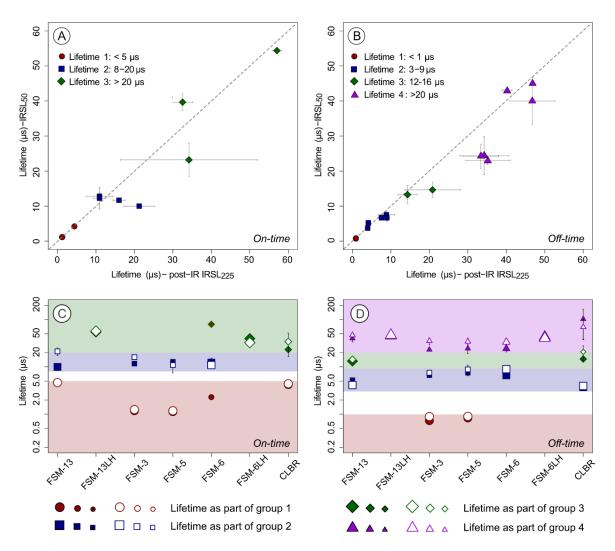


Fig. 2. Comparison of lifetimes obtained for the IRSL₅₀ and post-IR IRSL₂₂₅ signals for the on-time (A) and offtime (B) of the blue emission (50 Gy given dose). The dashed line in the graphs represent the one-to-one line. The data shown here is for all samples measured. The same data is shown in (C) and (D) where the fitted lifetime components are resolved for the individual samples. The data points in all figures represent the mean of three aliquots and their standard deviation. The diamond with red border (C) shows a lifetime, which only observed in one aliquot. The different sizes of the symbols in subfigures C and D indicate how prominent the component is within the given sample; the larger the symbol the more prominent is the component.

319

320 Dose dependency

- 321 Ankjærgaard et al. (2009) compared time-resolved OSL (470 nm LED and 532 nm laser stimulated)
- 322 lifetimes measured in the UV for different feldspars following irradiation doses of 5 Gy and 1000 Gy,
- 323 dependent on sample brightness, and showed that lifetime determination is independent of the dose
- 324 administered. However, Ankjærgaard et al. (2009) did not explore IRSL or post-IR IRSL signals. Here we
- 325 show the results of a systematic investigation of potential lifetime-dependencies for the blue emission
- 326 (IRSL₅₀ and post-IR IRSL₂₂₅) for irradiation doses of 50 Gy, 200 Gy and 800 Gy (Fig. 3). The fitting results

327 presented in direct comparison for the different doses in Fig. 3 scatter around the one-to-one line,

indicating independence of the lifetimes of the doses given and thus supporting earlier work byAnkjærgaard et al. (2009) for OSL of feldspars.

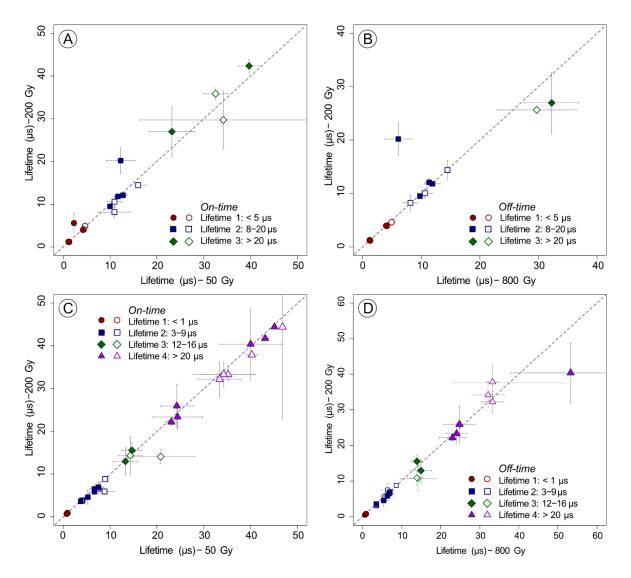


Fig. 3. Testing potential dose dependencies on the lifetimes of the IRSL₅₀ (filled symbols) and the post-IR IRSL₂₂₅ signal (open symbols) signal of the blue IRSL emission (on-time: A and B, off-time: C and D). Four doses were tested for this experiment, where the results of lifetimes obtained in response to 50 Gy are compared to 200 Gy (A and C) and 200 Gy are compared to 800 Gy (B and D).

330

331 Lifetimes compared to literature values

Measurements performed in this paper resulted in lifetimes spanning the following time scales: Ontime: (1) < 5 μ s, (2) 8-20 μ s, (3) > 20 μ s. Off-time: (1) < 1 μ s, (2) 3-9 μ s, (3) 12-16 μ s, (4) > 20 μ s. These lifetimes, obtained for the IRSL₅₀ and post-IR IRSL₂₂₅ signal, are in good agreement with previous work: Clark and Bailiff (1998) presented very fast lifetimes of different emissions measured for various chemically different feldspar samples using pulsed 850 nm laser stimulation. Their lifetimes range from a few ns to the μ s-scale. Due to instrumental limitations, we are unable to detect lifetimes faster than 338 0.3 μ s, but lifetimes measured on the μ s-scale in this paper are similar to those obtained by Clark and 339 Bailiff (1998). Using the sum of three exponential functions to describe the off-time decay, Tsukamoto 340 et al. (2006) measured blue IRSL lifetimes of 0.61, 2.4 and 19 µs for a Na-feldspar sample (grain 341 mixture) and 1.1, 4.3 and 19 µs for a K-feldspar sample (grain mixture). These values and the number 342 of exponentials used to describe the decay are similar to the results obtained in the present study (cf. 343 Fig. 2). Ankjærgaard et al. (2009) compared lifetimes of chemically different single crystal feldspars and 344 feldspar grain mixtures. These authors observed the UV emission of 532 nm laser and 470 nm diode 345 stimulated feldspars and enabled a comparison by using the sum of multiple first order exponentials 346 for fitting their off-time signals. The feldspars investigated showed lifetimes for the UV emission on 347 the μ s-scale, with the fastest lifetimes (<0.1 μ s) obtained through pulsed 532 nm laser stimulation. 348 Longest lifetimes measured ranged from ~10 µs to <100 µs, and are thus within the range of lifetimes 349 obtained in the present study, but it should be noted that stimulation and emission wavelengths differ 350 from our study.

351 Excited state vs. recombination lifetime?

352 Here we compared lifetimes of different feldspars, measured using two different IRSL signals, obtained from different IRSL integration intervals, and after different irradiation doses. Figs. 2 and 3, and Fig. S2 353 354 showed that the lifetimes seem to be largely independent of the given dose, the signal integration 355 interval and IRSL signal measured. This observation supports the excited state lifetime hypothesis if 356 one assumes a random distribution of donor and acceptor pairs in a proximity model. However, the 357 recombination lifetime might still be considered as a valid explanation, if the defect density is so high 358 that changes to e.g. the number of populated defects through different doses might be so small that 359 changes to the lifetime cannot be observed.

360 **3.2 Sample-to-sample variations in the blue time-resolved IRSL**₅₀ signal

We showed in section 3.1 that neither the on- nor the off-time of the blue IRSL signal depends on the 361 362 IRSL signal integration interval, or the dose given. We also showed that the IRSL₅₀ signal and the post-363 IR IRSL225 signal result in similar lifetimes. Thus, these findings support the excited state lifetime 364 hypothesis, unless the defect density is too high to observe any potential changes of the populated 365 defects and their effect on the lifetimes. In the following we investigate time-resolved IRSL (blue 366 emission) of a suite of chemically and structurally well-constrained alkali feldspars for which Riedesel et al. (2021a, b) have demonstrated how their different compositions and structure impact the blue 367 and infrared luminescence emissions and fading rates. This will enable us to study variations in time-368 369 resolved luminescence signals and their potential physical causes. Since no dependence on dose and 370 IRSL signal investigated were found in the previous experiments all results shown in the following are 371 based on the IRSL₅₀ signal which was recorded after a dose of 200 Gy. The discussion is based on qualitative comparisons between samples, as we will make some statements regarding potential physical processes in feldspars. Except in cases where time-resolved signals can be fitted using a single exponential function, no quantitative assessments are made, to avoid confusion about the kinetic order of feldspar luminescence.

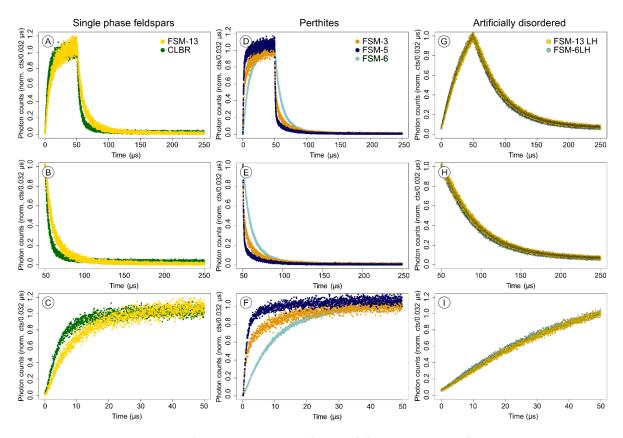


Fig. 4. Time-resolved IRSL₅₀ of the blue emission (200 Gy) for single phase feldspars, perthites and artificially disordered samples. A, D and G show the whole signals, B, E and H the off-time signal and C, F and I the on-time.

376

377 Figure 4 shows a comparison of blue time-resolved $IRSL_{50}$ signals obtained for the different samples. 378 For the seven feldspars investigated we observed a range of different time-resolved luminescence 379 signals. Whilst all samples show an on-time signal increase and an off-time decay on the µs-scale, there 380 are subtle differences between samples. The fastest rise and decay can be found in macroperthite 381 FSM-5 (Fig. 4D-F), followed by K-rich cryptoperthite FSM-3, where the short lifetime components 382 appear to be dominating the shape. Perthitic sample FSM-6 and the two single phase feldspars 383 microcline FSM-13 and albite CLBR show similar time-resolved signals, although CLBR shows faster 384 signal increase and decay compared to FSM-13 (Fig. 4A-C). The slowest rise and decay of the time-385 resolved IRSL signal can be found in the two artificially disordered feldspars FSM-13LH and FSM-6LH 386 (Fig. 4G-I), where the longer lifetime components appear to be dominating the shape. A direct 387 comparison between the artificially disordered samples FSM-13LH and FSM-6LH and their ordered 388 counterparts is shown in Figure 5. When comparing the artificially disordered samples FSM-13LH and FSM-6LH it is evident that the two samples show very similar on- and off-time signal behaviour, with
 samples FSM-6LH showing only slightly faster signals. Fitting of the time-resolved luminescence signals
 of these two artificially disordered samples was possible by using a single exponential function plus a
 constant background signal (i.e. k in eq. 2) and lifetimes obtained are around ~45 µs.

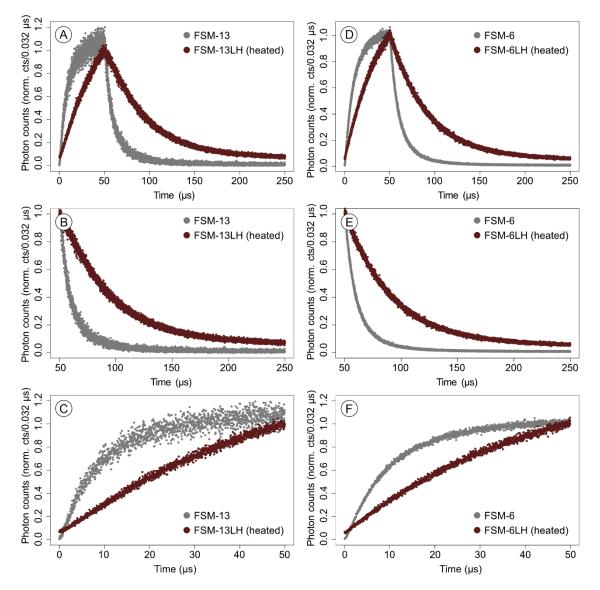


Fig. 5. Time resolved signals of the blue IRSL₅₀ emissions (200 Gy) of unheated and heated sample pairs FSM-13 and FSM-13LH (A, B, C) and FSM-6 and FSM-6LH (D, E, F).

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This slow rise and decay and the lifetimes obtained from fitting using a single exponential function are similar to results obtained for quartz elsewhere (e.g. Bailiff, 2000; Chithambo 2003; 2007). One other feature of quartz time-resolved luminescence signals is a significant decrease in lifetime with increasing stimulation temperature, which has been associated with thermal quenching (Bailiff, 2000; Chithambo, 2003). Bailiff (2000) observed a decrease in the decay lifetime of synthetic quartz from~40 µs to 8 µs, when increasing the stimulation temperature from room temperature to 210 °C. These observations are supported by similar experiments made by Chithambo (2003) on natural sand-sized 401 quartz. To test potential influence from thermal quenching on lifetimes and time-resolved 402 luminescence signals of artificially disordered feldspars, we measured time-resolved IRSL signals at 403 different stimulation temperatures. The signals were recorded using the protocol described in Table 2. 404 We varied the stimulation temperature of the post-IR IRSL signal (step 5 in Table 2) from 50 °C to 225 405 °C in 25 °C steps. In contrast to observations made on quartz, our artificially disordered feldspars do 406 not show thermal quenching (cf. Fig. 6) and lifetimes measured remain at a constant level around ~45 407 µs. Thus, we conclude that artificial disordering influences the lifetime of the blue emission, however, 408 the resulting emission and related processes seem to differ from what has been observed for quartz, 409 and thus do not show any thermal quenching effects.

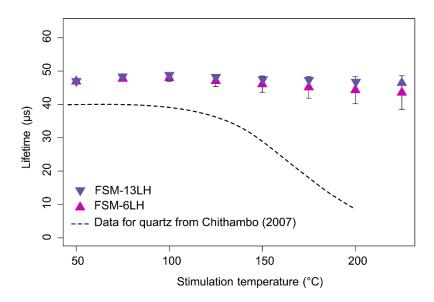


Fig. 6. Effect of varying the stimulation temperature of the post-IR IRSL signal on the lifetimes of the second IRSL signal in the post-IR₅₀ IRSL₅₀₋₂₂₅ °C protocol on disordered (heated) samples FSM-13LH and FSM-6LH. The lifetimes are compared to data obtained from quartz (Chithambo, 2007). Data points from this study represent the average and standard deviation of measurements of three aliquots.

410

The sample-to-sample variations in rise and decay of the time-resolved IRSL signals are striking, particularly showcasing the changes observed for the ordered and disordered samples pairs. Whilst we observe these variations in time-resolved signals, we so far cannot explain what causes these. However, these samples have been investigated previously and thus some information regarding their luminescence characteristics are known, enabling potential links between existing information and the newly gathered time-resolved results.

When comparing TL emission spectra and fading rates of the same single-phase feldspars and perthites as explored in the present study, Riedesel et al. (2021a) observed differences in emission intensities and fading rates of the samples investigated. Whilst single-phase feldspars (microcline and albite) only showed weak blue luminescence and no fading, cryptoperthites were characterised to have intense and broad blue emissions and fading rates ranging from ~2 to ~7 %/decade. Macroperthites exhibited 422 the highest fading rates with up to 12 %/decade. Riedesel et al. (2021a) found that these observations 423 support a location for the defect responsible for the blue emission on Al-O-Al bridges (cf. Finch and 424 Klein, 1999). The data presented by Riedesel et al. (2021a) suggests that potentially higher densities of 425 these defects can be found along the interfaces in perthites. When comparing time-resolved IRSL 426 measurements of single-phase feldspars, albite (CLBR) and microcline (FSM-13), and perthites (FSM-3, 427 FSM-5 and FSM-6), slightly slower time-resolved IRSL on-time rise and off-time signal decays can be 428 found for single-phase feldspars compared to perthites consisting of albite and microcline (cf. FSM-3 429 and FSM-5, Fig. 4A-C compared with Fig. 4D-F). Contrastingly, perthite FSM-6 shows time-resolved IRSL 430 signals similar to the two single-phase feldspars investigated. Interestingly FSM-6, a perthite, consists 431 of orthoclase and albite, indicating a slightly more disordered structure in FSM-6 compared to the 432 other two perthites.

433 Riedesel et al. (2021a) also compared TL emission spectra and fading rates of the same ordered and 434 disordered sample pairs as investigated here. They found that the blue luminescence emission intensity and the IRSL fading rate increased with framework disorder. This was interpreted as the result 435 436 of an increase in recombination centre density due to framework disorder. A denser defect population 437 would, if the recombination lifetime hypothesis holds true, result in a faster time-resolved 438 luminescence signal. However, the contrary is observed here (cf. Fig. 5), with disordered feldspars 439 (FSM-13LH, FSM-6LH) showing the slowest time-resolved signals. This potentially indicates a change in 440 the blue luminescence centre properties and thus a change to the excited state lifetime of this 441 luminescence centre.

Whilst large differences in fading rate and blue luminescence emission intensity were found across different feldspars, electron trapping centres, understood to be involved in feldspar IRSL (e.g. Prasad et al., 2017; Kumar et al. 2018), seem to be largely independent of changes to the sample chemistry or structure (Riedesel et al., 2019, 2021b). Riedesel et al. (2019, 2021b) showed that the electron trapping centre depth as well as emissions related to trapping and retrapping at electron trapping centres are similar across the alkali feldspar group.

448 If the electron trapping centres are independent of the sample's chemistry or structural state, then 449 changes in lifetime of the blue luminescence emission in ordered and disordered sample pairs and 450 across the chemical range of alkali feldspars, are either related to changes in the blue luminescence 451 centre itself or to parts of the crystal unrelated to electron trapping centres. Ankjærgaard and Jain 452 (2010) and Jain and Ankjærgaard (2011) suggested that the slower part of the TR decay could result 453 from recombination via the band-tail states. Pagonis et al. (2012) further investigated this hypothesis 454 by exploring time-resolved IRSL signals of four feldspar museum specimens. Pagonis et al. (2012) 455 applied a fitting approach which combined a single exponential and a stretched exponential function,

where the authors related the results of the stretched exponential to the band-tail states. The observed increase in time-resolved IRSL lifetime with increasing disorder of the framework in this paper could be related to a higher density of band-tail states or a wider band-tail due to disorder of the crystal. A wider or denser sub-conduction band tail could potentially result in an increased transport of charge via these states – resulting in slower recombination processes than would be expected for excited state tunnelling (cf. Jain and Ankjærgaard, 2011).

462 Additionally, Tsukamoto et al. (2006) found that longer lifetime components (> 20 μ s) were more 463 stable in laboratory storage experiments. When comparing fading results of the samples investigated 464 here (cf. Riedesel et al., 2021a) then a positive relationship between increasing fading rate and 465 increasing longer lifetime components can be found for the ordered and disordered sample pairs (FSM-466 13/FSM-13LH and FSM-6/FSM-6LH). However, macroperthite FSM-5 was amongst the samples 467 showing the highest fading, with an IRSL₅₀ fading rate of ~9 %/decade, but it also shows the fastest on-468 and off-time signals (cf. Fig. 4D, E, F) amongst the samples presented here. Thus, based on the present 469 data set, no clear relationship between fading rate and time-resolved signal rise and decay can be 470 established.

471 4 Yellow-green time-resolved IRSL emission – Results and Discussion

472 The yellow-green emission is rarely used in luminescence dating applications, however, it has been 473 identified as a common emission in many plagioclase feldspars (e.g. Geake et al., 1971, 1973, 1977; 474 Telfer and Walker, 1978), and was also recorded in microcline samples (c.f. Rendell and Clarke, 1997; 475 Riedesel et al., 2021a). Whilst being present as a distinct emission in some samples, the broad emission 476 recorded for perthites often spans from the blue into the yellow wavelength region (cf. Riedesel et al., 477 2021a) and might thus include this emission. Two potential defects have been proposed for causing this emission: (i) Mn²⁺ substituting for Ca²⁺ on M sites (e.g. Geake et al., 1971, 1973, 1977; Telfer and 478 Walker, 1978) or (ii) Mn²⁺ substituting for Al³⁺ on T sites (Telfer and Walker, 1978; Kirsh et al., 1987; 479 Clark and Bailiff, 1998). In the following we show the recorded time-resolved IRSL signals for the 480 481 yellow-green emission and the sample-to-sample variations we observed for this emission. The results 482 shown in this section are based on time-resolved measurements of the IRSL₅₀ signal, measured after 483 an irradiation dose of 200 Gy (or 400 Gy, in case of FSM-5 and FSM-6).

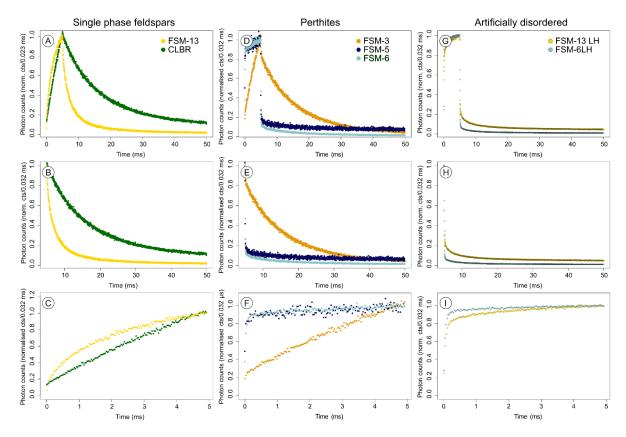


Fig. 7. Time-resolved IRSL₅₀ of the yellow-green emission for single phase feldspars, perthites and artificially disordered samples. The signals were recorded after an irradiation dose of 200 Gy (400 Gy in case of FSM-5 and FSM-6). A, D and G show the whole signals, B, E and H the off-time signal and C, F and I the on-time.

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For all samples we observed time-resolved IRSL signals on the ms-scale (Fig. 7), which is much slower compared to most of the observations made for the blue emission (Fig. 4). However, the different samples contain varying proportions of the slower part of the TR signal. Especially in perthitic and disordered samples the yellow-green emission is dominated by a very fast component, visible by the sharp increase at the beginning of the on-time and the sharp decrease at the start of the off-time (cf. Figs. 7 D-I). TR on-and off-time signals of single-phase feldspars FSM-13 and CLBR and perthite FSM-3 are significantly slower compared to the other samples (cf. Fig. 7A-C, D-F).

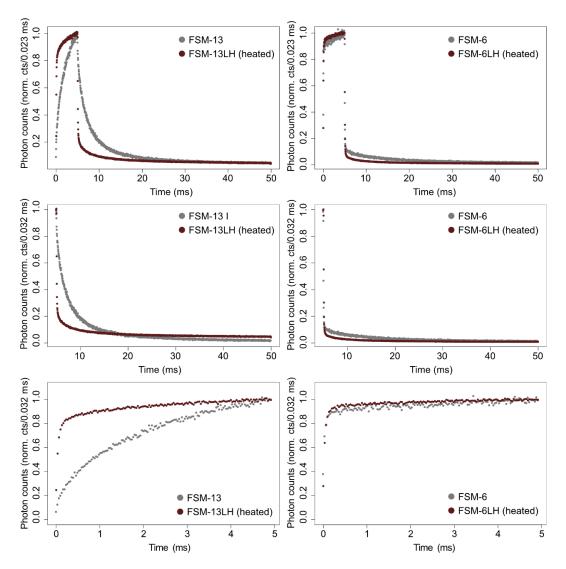


Fig. 8. Time resolved signals of the yellow-green IRSL₅₀ emissions of unheated and heated sample pairs FSM-13 andFSM-13LH (heated) (A, B, C) and FSM-6 andFSM-6LH (heated) (D, E, F).

492

493 A comparison of the ordered and disordered sample pairs reveals that the yellow-green emission 494 becomes faster during on- and off-times when the sample is disordered (Fig. 8). The effect is more 495 visible in the case of FSM-13 and FSM-13LH, compared to FSM-6 and FSM-6LH. However, from TL 496 emission spectra it becomes apparent that disordering results in a significant increase in blue TL 497 emission intensity (Riedesel et al., 2021a). This could potentially result in breakthrough of this strong 498 blue emission into the emission window we used to record the yellow-green emission (cf. transmission 499 of the filter combination on a log-scale in Fig. S1B). This might also explain the fast initial rise and decay 500 in samples FSM-3 and FSM-5. Alternatively, the fast signals recorded for the yellow-green emission 501 could originate from the defect being located on a tetrahedral site, as previously suggested by Clark 502 and Bailiff (1998). These authors observed very fast lifetimes for the yellow-green emission, even on 503 the ns-scale, when fitting their time-resolved signals.

504 Whilst we proposed a potential influence of band-tail states as the explanation for longer lifetimes of 505 the blue emission in disordered samples, this hypothesis would not hold in case of the yellow-green 506 emission, as the slowest signal rise and decay is found in ordered end member samples.

507 Observing significant differences in the time scale of signal rise and decay of different samples 508 investigated here could indicate that the emission indeed results from two different defect types. Long 509 lifetimes obtained here for single-phase microcline FSM-13 and albite CLBR suggest a metal ion site 510 defect as source of the yellow-green emission, even in alkali feldspars, supporting the Mn²⁺ 511 substituting for Ca²⁺ on M site hypothesis by e.g. Telfer and Walker (1978). In this case, a spin-forbidden 512 transition would explain the lifetimes on the ms-scale. However, short lifetimes (µs-scale) would be expected for a defect resulting from Mn^{2+} substituting for Al^{3+} on T sites (cf. Clark and Bailiff, 1998). In 513 514 this case the defect responsible for the yellow-green emission would be a lattice stabilising hole centre 515 on $Si^{4+}O^{-}Mn^{2+}$, as the hole centre on the O ion would help in charge neutralisation, due to the introduced imbalance by Mn^{2+} substituting for Al^{3+} (Telfer and Walker, 1978; Kirsh et al., 1987). If this 516 517 type of defect is indeed present in our samples, then one would expect that framework disorder 518 influences the lifetime of this signal, as is also shown in Fig. 8. Unfortunately, we cannot completely 519 rule out any influence of breakthrough of the blue emission, so no definite answer can be given here. 520 However, it is striking that the strong yellow-green emission can be found in the single-phase 521 microcline, a K-rich alkali feldspar end member, which also shows on-time increase and off-time 522 decrease on similar time-scales as seen for the albite specimen. This could indicate that whatever the type of defect, it is not restricted to plagioclase feldspars and might thus not be Mn²⁺ substituting for 523 Ca²⁺ on M sites. 524

525 5 Conclusions

In this paper we presented time-resolved infrared stimulated luminescence signals measured for a range of single crystal alkali feldspars, including K- and Na-feldspar end members (microcline and albite, both single phase), for the blue emission (~410 nm) and the yellow-green emission (~550 nm) using a post-IR₅₀ IRSL₂₂₅ protocol. The timescale on which the blue and the yellow-green emissions occur differ significantly; the blue signal on the µs-scale, and the yellow-green emission on the msscale. For each emission the lifetimes are independent of the signal integration interval, the IRSL signal investigated, and the dose given.

For the blue emission there is a slight difference in decay time scale between single-phase feldspars and perthites, however more significant differences are observed when the feldspars are artificially disordered. Our new observations show that artificially disordering the framework increases the rise and decay time of the on- and off-time, and that the signals can now be described using a single exponential fit. The lifetimes obtained from the disordered feldspars are similar to those measured for quartz elsewhere (c.f. Bailiff 2000, Chithambo 2003), but unlike for quartz the time-resolved signals here do not show any thermal quenching. The longer lifetimes observed for disordered feldspars may have two causes: 1. Disordering of the Si,Al-framework could result in a change of the blue recombination centre, resulting in longer excited state lifetimes of the defect itself. 2. Increased disorder of the lattice could cause changes to the band-tail state width or density. This in turn might increase the likelihood of recombination occurring via the band-tail states, resulting in longer recombination lifetimes (c.f. Jain and Ankjærgaard, 2011; Pagonis et al., 2012).

Time-resolved signals of the yellow-green emission vary between samples, with single-phase 545 546 microcline and albite specimens showing the slowest signals. These slow time scales could indicate a 547 spin-forbidden transition, as has been proposed for a defect resulting from Mn²⁺ substituting for Ca²⁺ 548 on M sites. However, the Ca content of the microcline specimen is negligible and the Mn content so 549 low that it could not be detected. In contrast to slow signals in the single-phase samples, perthitic and 550 disordered samples show a large proportion of a fast signal decay in the off-time, before a slow 551 component describes the remaining part of the time-resolved signal. The fast signal could either 552 indicate a different type of defect for this emission or, alternatively, it could be an artefact resulting 553 from breakthrough of the blue emission. This needs further investigation in future measurements.

The data in this paper shows the potential for using time-resolved IRSL with emission detected in the blue to gain information on the structural state of the feldspar investigated and thus helping in understanding variations in luminescence properties across a range of chemically and structurally different feldspars.

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Supplementary Material



663 insights into charge recombination in chemically and structurally different alkali feldspars

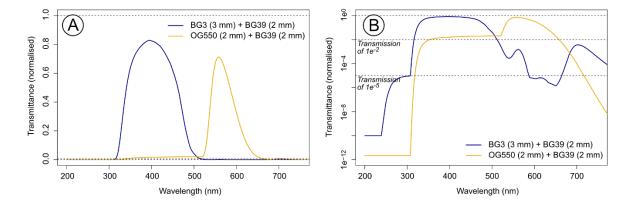


Fig. S1: Transmissions of filter combination used in this paper, with transmissions visualised on a linear (A) and on a logarithmic (B) scale.

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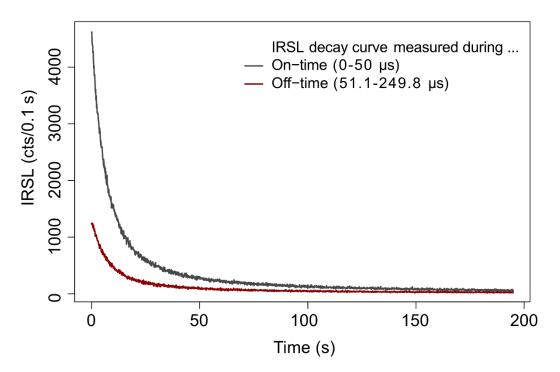


Fig. S2: IRSL₅₀ decay curves of sample FSM-13 measured during the on- (0-50 μ s, grey curve) and off-times (51.1-249.8 μ s, red curve) of the time-resolved luminescence measurements recorded in the blue emission window.

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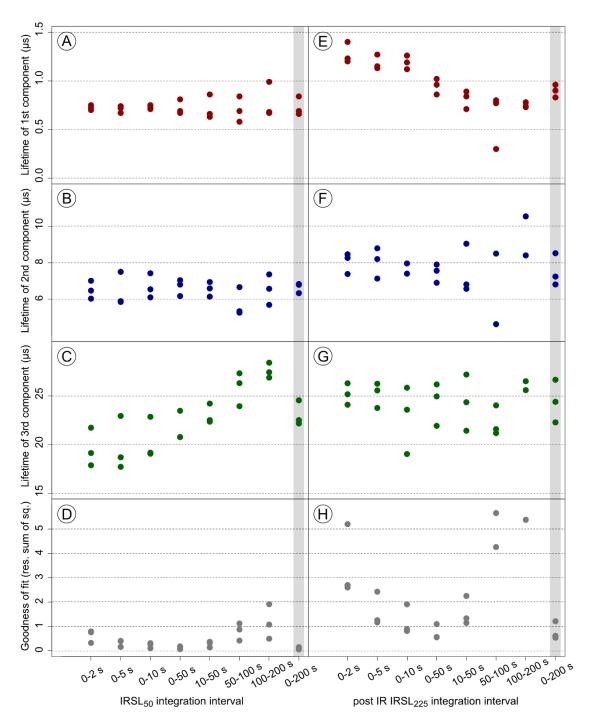


Fig. S3. Lifetimes and goodness of fit defined by the square sum of residuals for different integration time intervals for sample FSM-3. The data shown here is for the off-time signal of the blue emission and the two signals measured using the post-IR₅₀ IRSL₂₂₅ protocol, with results for the IRSL₅₀ signal in the left panels (A-D) and for the post-IR IRSL₂₂₅ on the right (E-H). The individual points show results for three aliquots measured.

Sample ID	Dose (Gy)	I ₁	τ1	l ₂	τ2	I ₃	τ3	I 4	τ4	k	residual sum of squares
FSM-13	50			1.00 ± 0.05	9.99 ± 0.27					0.01 ± 0.01	2.43 ± 0.72
FSM-13LH	50					1.56 ± 0.04	54.38 ± 0.76			0.04 ± 0.00	0.29 ± 0.07
FSM-3	50	0.61 ± 0.09	1.18 ± 0.18	0.42 ± 0.08	11.67 ± 0.22					-0.03 ± 0.01	0.50 ± 0.21
FSM-5	50	0.96 ± 0.05	1.13 ± 0.01	0.18 ± 0.03	12.79 ± 0.94					-0.10 ± 0.00	1.38 ± 0.47
FSM-6	50	0.29 ± 0.00	2.29 ± 0.19	0.76 ± 0.09	12.26 ± 3.09	0.25ª	79.39ª			0.00 ± 0.00	1.42 ± 0.90
FSM-6LH	50					1.33 ± 0.03	39.66 ± 2.33			0.05 ± 0.00	0.06 ± 0.02
CLBR	50	0.88 ± 0.05	4.21 ± 0.54			0.18 ± 0.09	23.19 ± 4.80			0.01 ± 0.06	0.42 ± 0.18
FSM-13	200			0.92 ± 0.05	9.49 ± 0.24					0.02 ± 0.00	3.75 ± 0.99
FSM-13LH	200					1.60 ± 0.05	53.26 ± 1.00			0.04 ± 0.00	0.55 ± 0.18
FSM-3	200	0.60 ± 0.09	1.18 ± 0.18	0.43 ± 0.07	11.81 ± 0.10					-0.02 ± 0.01	1.21 ± 0.36
FSM-5	200	0.76 ± 0.06	1.22 ± 0.04	0.26 ± 0.02	12.09 ± 0.77					-0.08 ± 0.01	3.00 ± 0.98
FSM-6	200	0.57 ± 0.19	5.58 ± 2.15	0.53 ± 0.19	20.19 ± 3.00					0.01 ± 0.02	3.72 ± 2.81
FSM-6LH	200					1.39 ± 0.03	42.34 ± 1.46			0.05 ± 0.00	0.44 ± 0.37
CLBR	200	0.91 ± 0.08	3.97 ± 0.19			0.17 ± 0.07	26.96 ± 5.78			-0.03 ± 0.01	1.15 ± 0.56
FSM-13	800			1.01 ± 0.05	9.77 ± 0.27					0.01 ± 0.01	6.46 ± 2.67
FSM-13LH	800			Not measured di	ue to very high si	gnal intensities s	saturating the PMT	despite adding	g an ND 2.0 filt	er.	
FSM-3	800	0.54 ± 0.04	1.28 ± 0.46	0.50 ± 0.06	11.93 ± 1.29	-	-		-	0.00 ± 0.01	3.40 ± 1.37
FSM-5	800	0.70 ± 0.04	1.27 ± 0.06	0.36 ± 0.02	11.33 ± 0.75					-0.06 ± 0.01	1.73 ± 0.49
FSM-6	800			0.57 ± 0.29	6.08 ± 2.34	0.48 ± 0.24	27.76 ± 13.17			0.00 ± 0.03	1.17 ± 0.84
FSM-6LH	800			Not measured d	ue to very high si	gnal intensities s	saturating the PMT	despite addin <u>a</u>	g an ND 2.0 filte	er.	
CLBR	800	0.87 ± 0.12	3.99 ± 0.19			0.17 ± 0.07	32.27 ± 4.65	-	-	-0.03 ± 0.01	3.34 ± 1.54

Table S1a: Lifetimes for the **blue IRSL**₅₀ emission recorded in the **on-time** of the pulsed IRSL signal. I_i is the normalised intensity of the fitted lifetime τ_i (μ s), k is a constant linear background. All values displayed are the average of three aliquots and the standard deviation. The data was normalised to the last data point of the on-time.

Sample ID	Dose (Gy)	I ₁	τ1	I ₂	τ2	I ₃	τ3	I 4	τ4	k	residual sum of squares
FSM-13	50			0.43 ± 0.13	5.25 ± 0.59	0.47 ± 0.14	13.27 ± 2.58	0.12 ± 0.02	39.99 ± 6.65	0.01 ± 0.00	0.88 ± 0.28
FSM-13LH	50							0.96 ± 0.02	45.08 ± 0.27	0.06 ± 0.00	0.63 ± 0.13
FSM-3	50	0.53 ± 0.08	0.73 ± 0.09	0.31 ± 0.05	6.65 ± 0.28			0.18 ± 0.04	23.08 ± 1.29	0.01 ± 0.00	0.11 ± 0.05
FSM-5	50	0.89 ± 0.04	0.83 ± 0.03	0.14 ± 0.02	7.52 ± 1.03			0.08 ± 0.01	24.39 ± 5.40	0.00 ± 0.00	0.16 ± 0.04
FSM-6	50			0.57 ± 0.10	6.64 ± 0.99			0.33 ± 0.05	24.24 ± 3.49	0.01 ± 0.00	0.58 ± 0.49
FSM-6LH	50							0.96 ± 0.01	43.08 ± 0.76	0.05 ± 0.00	0.21 ± 0.03
CLBR	50			0.78 ± 0.05	3.74 ± 0.26	0.16 ± 0.03	14.65 ± 2.13	0.03 ± 0.03	102.18 ± 64.17	0.06 ± 0.00	0.17 ± 0.08
FSM-13	200			0.40 ± 0.18	4.58 ± 1.63	0.45 ± 0.13	12.94 ± 3.26	0.11 ± 0.06	40.38 ± 8.46	0.001 ± 0.00	1.33 ± 0.45
FSM-13LH	200							0.99 ± 0.03	44.40 ±0.38	0.06 ± 0.00	1.12 ± 0.39
FSM-3	200	0.51 ± 0.10	0.67 ± 0.13	0.32 ± 0.05	5.84 ± 0.61			0.20 ± 0.03	22.16 ± 0.86	0.01 ± 0.00	0.27 ± 0.09
FSM-5	200	0.71 ± 0.05	0.81 ± 0.08	0.21 ± 0.02	6.92 ± 1.05			0.10 ± 0.02	23.33 ± 2.93	0.00 ± 0.00	0.38 ± 0.12
FSM-6	200			0.63 ± 0.11	6.50 ± 1.07			0.38 ± 0.08	25.90 ± 5.09	0.02 ± 0.01	1.66 ± 1.48
FSM-6LH	200							0.96 ± 0.00	41.71 ± 0.18	0.05 ± 0.00	0.64 ± 0.65
CLBR	200			0.82 ± 0.04	3.58 ± 0.06	0.18 ± 0.01	15.55 ± 1.79	0.03 ± 0.02	87.11 ± 31.14	0.03 ± 0.00	0.40 ± 0.24
FSM-13	800			0.51 ± 0.18	5.38 ± 1.20	0.47 ± 0.11	14.92 ± 2.69	0.06 ± 0.04	53.26 ± 15.37	0.01 ± 0.00	2.46 ± 1.32
FSM-13LH	800			Not measured d	ue to very high si	gnal intensities s	aturating the PM	IT despite addin	g an ND 2.0 filter	:	
FSM-3	800	0.45 ± 0.04	0.74 ± 0.26	0.36 ± 0.09	6.31 ± 1.64			0.21 ± 0.03	23.01 ± 3.36	0.01 ± 0.00	0.76 ± 0.36
FSM-5	800	0.64 ± 0.06	0.86 ± 0.03	0.29 ± 0.02	6.96 ± 0.34			0.12 ± 0.01	24.06 ± 2.65	0.01 ± 0.00	0.27 ± 0.06
FSM-6	800			0.59 ± 0.15	6.78 ± 1.32			0.37 ± 0.10	24.78 ± 4.03	0.01 ± 0.01	0.55 ± 0.47
FSM-6LH	800			Not measured d	ie to very high si	gnal intensities s	aturating the PM	IT despite addin	g an ND 2.0 filter	:	
CLBR	800			0.76 ± 0.09	3.49 ± 0.18	0.18 ± 0.02	13.91 ± 1.38	0.04 ± 0.04	68.07 ± 37.48	0.03 ± 0.00	1.15 ± 0.73

Table S1b: Lifetimes for the **blue IRSL**₅₀ emission recorded in the **off-time** of the pulsed IRSL signal. I_i is the normalised intensity of the fitted lifetime τ_i (μ s), k is a constant linear background. All values displayed are the average of three aliquots and the standard deviation. The data was normalised to the last data point of the on-time.

Sample ID	Dose (Gy)	I ₁	τ1	l ₂	τ2	I ₃	τ3	I 4	τ4	k	residual sum of squares
FSM-13	50	0.53 ± 0.04	4.66 ± 0.07	0.48 ± 0.07	21.25 ± 3.95					0.03 ± 0.02	2.08 ± 0.87
FSM-13LH	50					1.28 ± 0.02	57.05 ± 1.16			0.26 ± 0.01	0.10 ± 0.02
FSM-3	50	0.71 ± 0.07	1.25 ± 0.10	0.19 ± 0.01	16.00 ± 1.58					0.13 ± 0.07	0.96 ± 0.42
FSM-5	50	0.80 ± 0.06	1.18 ± 0.02	0.29 ± 0.04	10.92 ± 3.50					-0.09 ± 0.01	1.01 ± 0.56
FSM-6	50			0.95 ± 0.04	10.94 ± 1.56					0.05 ± 0.03	0.70 ± 0.37
FSM-6LH	50					1.16 ± 0.04	32.53 ± 2.63			0.09 ± 0.01	0.04 ± 0.02
CLBR	50	0.81 ± 0.20	4.40 ± 0.37			0.31 ± 0.26	34.18 ± 17.84			0.04 ± 0.06	0.44 ± 0.23
FSM-13	200	0.60 ± 0.03	4.93 ± 0.67			0.39 ± 0.07	24.92 ± 5.48			0.01 ± 0.01	4.34 ± 0.14
FSM-13LH	200					1.24 ± 0.05	52.93 ± 1.67			0.25 ± 0.00	0.17 ± 0.04
FSM-3	200	0.63 ± 0.05	1.20 ± 0.08	0.17 ± 0.01	14.48 ± 0.06					0.17 ± 0.08	2.25 ± 0.82
FSM-5	200	0.80 ± 0.04	1.21 ± 0.08	0.25 ± 0.01	8.16 ± 1.29					-0.07 ± 0.01	2.34 ± 0.73
FSM-6	200			0.93 ± 0.07	10.61 ± 1.28					0.04 ± 0.03	1.23 ± 0.70
FSM-6LH	200					1.21 ± 0.03	35.87 ± 1.39			0.09 ± 0.00	0.28 ± 0.22
CLBR	200	0.80 ± 0.16	4.14 ± 0.24			0.24 ± 0.25 ^b	29.73 ± 6.83 ^b			0.06 ± 0.01	1.64 ± 0.72
FSM-13	800	0.57 ± 0.06	4.63 ± 0.41	0.45 ± 0.03	16.77 ± 2.37					0.00 ± 0.01	7.14 ± 3.32
FSM-13LH	800			Not measured d	ue to very high sig	gnal intensities s	aturating the PMT	despite adding	an ND 2.0 filt	er.	
FSM-3	800	0.56 ± 0.14	1.26 ± 0.22	0.16 ± 0.01	14.39 ± 1.78					0.26 ± 0.12	6.52 ± 2.69
FSM-5	800	0.77 ± 0.02	1.23 ± 0.04	0.32 ± 0.02	8.25 ± 1.38					-0.06 ± 0.01	1.59 ± 0.48
FSM-6	800			0.97 ± 0.02	10.06 ± 0.88					0.03 ± 0.02	0.44 ± 0.29
FSM-6LH	800			Not measured d	ue to very high sig	gnal intensities s	aturating the PMT	despite adding	an ND 2.0 filte	er.	
CLBR	800	0.79 ± 0.21	3.95 ± 0.38			0.45ª	25.64ª			0.06 ± 0.00	5.26 ± 2.42

Table S2a: Lifetimes for the **blue post-IR IRSLI**₂₂₅ emission recorded in the **on-time** of the pulsed IRSL signal. I_i is the normalised intensity of the fitted lifetime τ_i (μ s), k is a constant linear background. All values displayed are the average of three aliquots and the standard deviation. The data was normalised to the last data point of the on-time.

Table S2b: Lifetimes for the **blue post-IR IRSL**₂₂₅ emission recorded in the **off-time** of the pulsed IRSL signal. I_i is the normalised intensity of the fitted lifetime τ_i (μ s), k is a constant linear background. All values displayed are the average of three aliquots and the standard deviation. The data was normalised to the last data point of the on-time.

Sample ID	Dose (Gy)	I ₁	τ1	l ₂	τ2	I ₃	τ3	I 4	τ4	k	residual sum of squares
FSM-13	50			0.45 ± 0.07	4.13 ± 0.27	0.35 ± 0.05	14.31 ± 2.46	0.12 ± 0.02	39.99 ± 6.65	0.01 ± 0.00	0.88 ± 0.28
FSM-13LH	50							0.96 ± 0.02	45.08 ± 0.27	0.06 ± 0.00	0.63 ± 0.13
FSM-3	50	0.65 ± 0.07	0.89 ± 0.07	0.14 ± 0.02	7.52 ± 0.89			0.18 ± 0.04	23.08 ± 1.29	0.01 ± 0.00	0.11 ± 0.05
FSM-5	50	0.75 ± 0.05	0.90 ± 0.03	0.27 ± 0.03	8.82 ± 2.09			0.08 ± 0.01	24.39 ± 5.40	0.00 ± 0.00	0.16 ± 0.04
FSM-6	50			0.77 ± 0.12	8.95 ± 0.27			0.33 ± 0.05	24.24 ± 3.49	0.01 ± 0.00	0.58 ± 0.49
FSM-6LH	50							0.96 ± 0.01	43.08 ± 0.76	0.05 ± 0.00	0.12 ± 0.03
CLBR	50			0.73 ± 0.20	3.96 ± 0.32	0.12 ± 0.06	20.81 ± 7.25	0.03 ± 0.03	102.18 ±	0.06 ± 0.05	0.31 ± 0.29
									64.17		
FSM-13	200			0.45 ± 0.13	3.81 ± 0.80	0.37 ± 0.09	14.33 ± 4.51	0.11 ± 0.07	62.84 ± 39.90	0.05 ± 0.01	1.96 ± 0.73
FSM-13LH	200							0.77 ± 0.02	44.37 ± 0.04	0.27 ± 0.00	0.46 ± 0.12
FSM-3	200	0.57 ± 0.06	0.84 ± 0.11	0.13 ± 0.01	6.62 ± 1.48			0.08 ± 0.02	33.25 ± 1.37	0.22 ± 0.07	2.45 ± 1.24
FSM-5	200	0.73 ± 0.02	0.84 ± 0.03	0.28 ± 0.02	5.90 ± 0.76			0.03 ± 0.01	33.28 ± 2.88	0.01 ± 0.00	0.31 ± 0.09
FSM-6	200			0.77 ± 0.13	8.81 ± 0.19			0.16 ± 0.07	32.15 ± 4.08	0.07 ± 0.01	0.89 ± 0.67
FSM-6LH	200							0.90 ± 0.00	37.86 ± 0.31	0.10 ± 0.00	0.47 ± 0.37
CLBR	200			0.72 ± 0.15	3.71 ± 0.11	0.11 ± 0.04	14.05 ± 1.67	0.08 ± 0.12	94.63 ± 42.70	0.12 ± 0.02	1.25 ± 0.70
FSM-13	800			0.38 ± 0.10	3.41 ± 0.42	0.48 ± 0.06	10.96 ± 3.37	0.13 ± 0.08	37.63 ± 14.26	0.05 ± 0.00	3.51 ± 1.50
FSM-13LH	800			Not measured a	lue to very high s	signal intensities	saturating the PN	ЛТ despite addin	g an ND 2.0 filter		
FSM-3	800	0.52 ± 0.11	0.84 ± 0.20	0.15 ± 0.04	7.80 ± 1.80			0.06 ± 0.02	37.87 ± 4.60	0.29 ± 0.11	7.88 ± 3.03
FSM-5	800	0.69 ± 0.01	0.87 ± 0.04	0.32 ± 0.02	6.01 ± 0.58			0.05 ± 0.01	32.31 ± 3.06	0.02 ± 0.00	0.29 ± 0.07
FSM-6	800			0.85 ± 0.06	9.10 ± 0.17			0.11 ± 0.05	34.26 ± 3.42	0.07 ± 0.01	0.29 ± 0.24
FSM-6LH	800			Not measured a	lue to very high s	signal intensities	saturating the PN	ЛТ despite addin	g an ND 2.0 filter		
CLBR	800			0.70 ± 0.15	3.52 ± 0.32	0.10 ± 0.04	11.18 ± 1.97	0.10 ± 0.14	133.61 ± 111.59	0.12 ± 0.00	4.33 ± 3.14

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Sample ID	Dose (Gy)	I ₁	τ1	I ₂	τ2	I ₃	τ3	I 4	τ4	k	residual sum of squares
FSM-13	200	0.20 ± 0.01	0.22 ± 0.08	0.83 ± 0.02	2.50 ± 0.34					0.09 ± 0.01	0.06 ± 0.03
FSM-13LH	200	0.64 ± 0.06	0.05 ± 0.00	0.12 ± 0.05	2.52 ± 0.43					0.26 ± 0.00	0.01 ± 0.01
FSM-3	200	0.09 ± 0.03	0.02 ± 0.01			1.87 ± 0.18	8.52 ± 0.61			0.18 ± 0.02	0.04 ± 0.01
FSM-5	400	0.39 ± 0.01	0.02 ± 0.00	0.14 ± 0.00	3.93 ± 0.72					0.51 ± 0.01	0.27 ± 0.14
FSM-6	400	0.43 ± 0.14	0.04 ± 0.00	0.38 ± 0.49	4.04 ± 1.46					0.36 ± 0.11	0.05 ± 0.04
FSM-6LH	200	0.66 ± 0.01	0.04 ± 0.00	0.07 ± 0.00	2.13 ± 0.28					0.28 ± 0.00	0.01 ± 0.00
CLBR	200	0.10 ^a	0.03ª			2.43 ± 0.21	11.12 ± 1.55			0.13 ± 0.01	0.05 ± 0.05

1.57 ± 0.20

 2.22 ± 0.10

 8.10 ± 0.67

9.66 ± 0.17

 0.09 ± 0.01

 0.27 ± 0.00

 0.20 ± 0.03

 0.53 ± 0.01

 0.31 ± 0.09

 0.27 ± 0.00

 0.12 ± 0.01

 0.01 ± 0.01

 0.02 ± 0.01

 0.02 ± 0.01

 0.19 ± 0.06

 0.09 ± 0.06

 0.02 ± 0.01

 0.01 ± 0.01

2.37 ± 0.19

 1.81 ± 0.08

 4.10 ± 2.38

6.16 ± 1.76

 1.82 ± 0.41

FSM-13

FSM-13LH

FSM-3

FSM-5

FSM-6

FSM-6LH

CLBR

800

800

800

800

800

800

800

 0.20 ± 0.00

 0.66 ± 0.02

 0.11 ± 0.02

 0.38 ± 0.04

 0.37 ± 0.12

 0.64 ± 0.01

0.06^a

 0.20 ± 0.04

 0.05 ± 0.00

 0.03 ± 0.01

 0.04 ± 0.00

 0.04 ± 0.00

 0.05 ± 0.00

0.04^a

 0.82 ± 0.03

 0.08 ± 0.00

 0.15 ± 0.06

0.65 ± 0.47

 0.08 ± 0.00

Table S3a: Lifetimes for the **yellow-green IRSL**₅₀ emission recorded in the **on-time** of the pulsed IRSL signal. I_i is the normalised intensity of the fitted lifetime τ_i (ms), k is a constant linear background. All values displayed are the average of three aliquots and the standard deviation. The data was normalised to the last data point of the on-time.

Sample ID	Dose (Gy)	I ₁	τ1	I ₂	τ2	I ₃	τ3	I 4	τ4	k	residual sum of squares
FSM-13	200	0.19 ± 0.03	0.24 ± 0.06	0.57 ± 0.02	2.19 ± 0.17	0.14 ± 0.04	9.58 ± 1.12			0.01 ± 0.00	0.05 ± 0.02
FSM-13LH	200	0.52 ± 0.02	0.12 ± 0.01	0.07 ± 0.04	6.09 ± 0.39					0.03 ± 0.02	0.15 ± 0.04
FSM-3	200	0.13 ± 0.01	0.11 ± 0.01			0.75 ± 0.04	11.04 ± 0.06			0.03 ± 0.01	0.14 ± 0.08
FSM-5	400	0.40 ± 0.01	0.09 ± 0.00			0.08 ± 0.00	10.18 ± 0.70			0.06 ± 0.00	0.39 ± 0.04
FSM-6	400	0.37 ± 0.13	0.10 ± 0.00	0.08 ± 0.06	5.35 ± 2.24	0.15 ± 0.23	32.98 ± 21.06			0.03 ± 0.02	0.19 ± 0.00
FSM-6LH	200	0.52 ± 0.00	0.11 ± 0.00	0.04 ± 0.00	6.55 ± 0.06					0.01 ± 0.00	0.17 ± 0.18
CLBR	200			0.17 ± 0.05	3.90 ± 1.34	0.75 ± 0.03	13.53 ± 0.90			0.08 ± 0.01	0.21 ± XX
FSM-13	800	0.20 ± 0.01	0.24 ± 0.03	0.58 ± 0.01	2.13 ± 0.09	0.13 ± 0.02	8.56 ± 0.96			0.01 ± 0.00	0.02 ± 0.01
FSM-13LH	800	0.53 ± 0.01	0.11 ± 0.00	0.04 ± 0.00	6.52 ± 0.06					0.02 ± 0.00	0.18 ± 0.02
FSM-3	800	0.24 ± 0.16	0.10 ± 0.01			0.51 ± 0.37	10.55 ± 0.51			0.04 ± 0.00	0.15 ± 0.12
FSM-5	800	0.31 ± 0.16	0.10 ± 0.02			0.23 ± 0.25	9.59 ± 0.12			0.11 ± 0.11	0.23 ± 0.05
FSM-6	800	0.32 ± 0.12	0.10 ± 0.01	0.11 ± 0.04	4.23 ± 2.34	0.23 ± 0.18	19.80 ± 11.31			0.02 ± 0.02	0.21 ± 0.04
FSM-6LH	800	0.52 ± 0.01	0.11 ± 0.00	0.05 ± 0.00	6.33 ± 0.15					0.01 ± 0.00	0.17 ± 0.01
CLBR	800	0.04 ^a	0.14ª	0.24 ± 0.02	5.34 ± 0.29	0.69 ± 0.04	14.90 ± 0.31			0.07 ± 0.02	0.06 ± 0.05

Table S3b: Lifetimes for the **yellow-green IRSL**₅₀ emission recorded in the **off-time** of the pulsed IRSL signal. I_i is the normalised intensity of the fitted lifetime τ_i (ms), k is a constant linear background. All values displayed are the average of three aliquots and the standard deviation. The data was normalised to the last data point of the on-time.

Table S4a: Lifetimes for the **yellow-green post-IR IRSL**₂₂₅ emission recorded in the **on-time** of the pulsed IRSL signal. I_i is the normalised intensity of the fitted lifetime τ_i (ms), k is a constant linear background. All values displayed are the average of three aliquots and the standard deviation. The data was normalised to the last data point of the on-time. ^aOnly one aliquot of the three aliquots measured revealed this lifetime.

Sample ID	Dose (Gy)	I ₁	τ1	I ₂	τ2	I ₃	τ3	I ₄	τ ₄	k	residual sum of squares
FSM-13	200	0.32 ± 0.03	0.12 ± 0.04	0.52 ± 0.04	1.92 ± 0.41					0.21 ± 0.02	0.07 ± 0.04
FSM-13LH	200	0.57 ± 0.00	0.05 ± 0.00	0.19 ± 0.00	2.06 ± 0.03					0.26 ± 0.00	0.01 ± 0.00
FSM-3	200	0.08 ± 0.01	0.03 ± 0.01			1.18 ± 0.20	7.14 ± 0.81			0.33 ± 0.03	0.11 ± 0.03
FSM-5	400	0.37ª	0.02ª	0.11ª	2.80ª					0.54	0.22
FSM-6	400	0.38 ± 0.02	0.04 ± 0.01	0.16 ± 0.01	2.35 ± 0.52					0.49 ± 0.00	0.06 ± 0.03
FSM-6LH	200	0.59 ± 0.00	0.04 ± 0.00	0.09 ± 0.01	1.59 ± 0.15					0.32 ± 0.00	0.01 ± 0.00
CLBR	200	0.24ª	0.04ª			2.39 ± 0.72	15.48 ± 4.07			0.31 ± 0.08	0.06 ± 0.06
FSM-13	800	0.32 ± 0.02	0.11 ± 0.01	0.52 ± 0.03	1.73 ± 0.18					0.19 ± 0.01	0.02 ± 0.01
FSM-13LH	800	0.56 ± 0.00	0.05 ± 0.00	0.19 ± 0.01	2.07 ± 0.12					0.27 ± 0.00	0.01 ± 0.00
FSM-3	800	0.08 ± 0.02	0.07 ± 0.07			1.10 ± 0.08	6.96 ± 0.56			0.35 ± 0.04	0.06 ± 0.02
FSM-5	800	0.33 ± 0.10	0.05 ± 0.03	0.09 ± 0.04	1.99 ± 0.63					0.58 ± 0.07	0.28 ± 0.11
FSM-6	800	0.44 ± 0.05	0.03 ± 0.00	0.13 ± 0.06	1.91 ± 0.30					0.46 ± 0.03	0.05 ± 0.04
FSM-6LH	800	0.59 ± 0.02	0.04 ± 0.00	0.11 ± 0.01	1.41 ± 0.36					0.31 ± 0.00	0.02 ± 0.01
CLBR	800	0.19ª	0.05ª			2.07 ± 0.45	12.35 ± 2.25			0.27 ± 0.06	0.02 ± 0.02

Table S4b: Lifetimes for the **yellow-green post-IR IRSL**₂₂₅ emission recorded in the **off-time** of the pulsed IRSL signal. I_i is the normalised intensity of the fitted lifetime τ_i (ms), k is a constant linear background. All values displayed are the average of three aliquots and the standard deviation. The data was normalised to the last data point of the on-time. ^aOnly one aliquot of the three aliquots measured revealed this lifetime.

Sample ID	Dose (Gy)	I ₁	τ ₁	l ₂	τ2	I ₃	τ3	I ₄	τ ₄	k	residual sum of squares
FSM-13	200	0.33 ± 0.05	0.18 ± 0.04	0.33 ± 0.04	1.58 ± 0.02	0.12 ± 0.05	10.13 ± 0.82			0.07 ± 0.01	0.09 ± 0.03
FSM-13LH	200	0.47 ± 0.00	0.12 ± 0.00	0.10 ± 0.00	2.81 ± 0.09	0.05 ± 0.00	16.35 ± 0.79			0.05 ± 0.00	0.11 ± 0.00
FSM-3	200	0.13 ± 0.01	0.12 ± 0.02			0.56 ± 0.04	9.54 ± 0.22			0.18 ± 0.03	0.44 ± 0.12
FSM-5	400	0.42 ± 0.02	0.09 ± 0.00			0.05 ± 0.00	9.64 ± 0.70			0.02 ± 0.00	0.31 ± 0.07
FSM-6	400	0.40 ± 0.02	0.09 ± 0.00	0.06 ± 0.01	2.46 ± 0.31	0.06 ± 0.01	12.67 ± 0.03			0.06 ± 0.01	0.16 ± 0.02
FSM-6LH	200	0.50 ± 0.00	0.11 ± 0.00	0.04 ± 0.00	3.57 ± 0.07	0.01 ± 0.00	28.54 ± 2.43			0.02 ± 0.00	0.16 ± 0.01
CLBR	200	0.22ª	0.14ª			0.57 ± 0.03	11.36 ± 0.18	0.21 ± 0.00^{b}	56.08 ± 17.86 ^b	0.19 ± 0.08	0.23 ± 0.11
FSM-13	800	0.32 ± 0.02	0.18 ± 0.01	0.36 ± 0.03	1.58 ± 0.13	0.10 ± 0.03	9.87 ± 1.10			0.07 ± 0.01	0.04 ± 0.01
FSM-13LH	800	0.46 ± 0.00	0.12 ± 0.00	0.10 ± 0.00	2.68 ± 0.04	0.05 ± 0.00	16.15 ± 0.32			0.05 ± 0.00	0.11 ± 0.00
FSM-3	800	0.14 ± 0.02	0.12 ± 0.01			0.60 ± 0.09	9.73 ± 0.99			0.14 ± 0.10	0.22 ± 0.13
FSM-5	800	0.40 ± 0.02	0.09 ± 0.00			0.07 ± 0.01	7.23 ± 0.30			0.02 ± 0.00	0.25 ± 0.03
FSM-6	800	0.41 ± 0.03	0.09 ± 0.00	0.05 ± 0.02	4.13 ± 3.30	0.05 ± 0.01	13.96 ± 1.24			0.06 ± 0.01	0.18 ± 0.02
FSM-6LH	800	0.50 ± 0.00	0.11 ± 0.00	0.05 ± 0.00	3.30 ± 0.16	0.02 ± 0.00	21.06 ± 2.90			0.03 ± 0.00	0.16 ± 0.00
CLBR	800	0.17ª	0.12ª			0.36 ± 0.20	8.30 ± 3.36	0.37 ± 0.11	22.71 ± 7.49	0.19 ± 0.08	0.10 ± 0.07