Peter Mafimisebi, Jemilah Abdulrahman

Email: mafimisebipeter.o@qmail.com, Email: jabdulrahman033@gmail.com

Department of Geology and Mineral Sciences, University of Ilorin. Nigeria.

Dear Editorial Manager,

I am enclosing herewith a manuscript entitled "GEOCHEMICAL SURVEY AND POTENTIAL MINERALIZATION INVESTIGATION OF ILERO AND ITS ENVIRONS OYO STATE, NIGERIA.

This work reflects research on deciphering possible mineralization in the Ilero and its environments, Oyo State to its role in mineral exploration and exploitation.

I believe this submission will be useful to your readers, as it addresses some critical methodologies for discovering economic minerals. The research stands to benefit students researching Earth Science Studies.

This manuscript has been submitted to sedimentary geology journal. The current version is a preprint which has not yet been peer reviewed. Subsequent versions of this manuscript may have slightly different content. If accepted, the final version of this manuscript will be available via the 'Peer-reviewed Publication DOI' link on the right hand side of this webpage"

# GEOCHEMICAL SURVEY AND POTENTIAL MINERALIZATION INVESTIGATION OF ILERO AND ITS ENVIRONS OYO STATE, NIGERIA

3 Mafimisebi O. Peter, Abdulrahman Jemilah,

4 Department of Geology and Mineral Science, University of Ilorin, Ilorin, Nigeria.

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#### Abstract

7 Ilero and its environs are part of the Kajola local Government area, Oyo State, and fall within the basement complex of southwestern Nigeria. The research work was focused on the geological mapping of the area and 8 9 the study of the different rock and minerals units to determine the mineral resources of the area using 10 petrographic, geochemical, and XRD analysis. The geochemical analysis showed the rocks and minerals to 11 be diversely composed of 54-97% SiO<sub>2</sub>, 0.1-20% Al<sub>2</sub>O<sub>3</sub>, 0.8-4% MgO, 0.5-6% Fe<sub>2</sub>O<sub>3</sub>, < 1% K<sub>2</sub>O, TiO<sub>2</sub> and minor amounts of trace elements including copper, manganese, and lead. The mineralogy of the rocks and ore 12 mineral comprised of quartz, biotite, hornblende, alkali feldspars, and tourmaline. There are also indications 13 14 of the presence of chlorite and some ferromagnesian minerals such as pyroxene and olivine. The rocks are 15 predominantly acid rocks along with a few basic rocks

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17 Key word: Petrographic, Provenance, Mineralization, Rocks, Ore Mineral.

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# 19 **1. Introduction**

The studied area is in the northwest of Okeho and Ilero, in Kajola local Government area, of Oyo-State. These areas fall within the Iseyin Schist belt, in the southwestern part of the Precambrian Complex geological province of Nigeria (Rahaman, 1976). As a result of less detailed work on the study area, not much information is available, and therefore the need for the delineation of various rock types of the different units of the basement complex in the area as well as the need for detailed petrographic and geochemical studies on the rocks for assessment of their petrogenesis. The project work is aimed at revealing the
 mineralization potential of the llero area.

# 27 1.1 Geology and Location

28 The studied area is located around Ilero within, Kajola local Government area, Oyo State. It lies within longitude 003<sup>0</sup> 15'00'' to 003<sup>0</sup> 26'00 and latitude 008''0500'' to 008<sup>0</sup> 01 '0<sup>0</sup>'N. The study area falls within 29 the Precambrian Basement Complex of Southwestern Nigeria and in the Iseyin Schist Belt of southwestern 30 Nigeria. It is characterized by occurrences of biotite granites, syenites, talcose, amphibolites, and bodies of 31 32 pegmatites and quartz-veins. The study area is accessible from Oyo-Iseyin, Okeho- Ilero, Igana, and Shaki -Ipapo -Otu Road (Rahaman, 1976). Numerous footpaths and un-tarred rural roads were also accessible. The 33 34 study area is more accessible during the dry season when the field aspect of this study was carried out. Accessibility would have been difficult during the rainy season when the drainage channels would have been 35 36 overflown and the forest will be thick thereby causing coverage of exposures and making navigation within 37 the area difficult.



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Fig. 2 Map of Regional Geology of the Study Area (Oyo State) within the basement complex in southwestern Nigeria (Rahaman, 1976).

#### 42 2. Materials and Methods

#### 43 2.1 The Petrographic Analysis

The petrographic analysis was used to study the mineralogical composition of rock samples by 44 studying the behavior of these minerals under the plane and crossed polarized light (Ekeleme, 2017). 45 46 This study involves cutting a thin section of rock samples and critically studying these thin sections 47 under a polarizing microscope to determine the optical properties of the samples. Basically, the thin section is made of a glass slide, a thin film of a rockslide (30 microns), and a cover slip. The minerals 48 used in the preparation of thin sections include Canada Balsam, Carborundum powder of 400, 600, 49 and 800 grits, and Araldite. The equipment used was the Logitech rock cutting machine, hot plate, 50 smooth glass, glass slide, diamond pen, etc. (Nwachukwu, 2011). The rock samples to be studied were 51 first cut into slices of thickness 3mm -4mm using the thin section (Logitech) machine, then the surface 52 53 of the rock was then smoothened by continuously grinding its surface on a glass plate using coarse grit 54 carborundum (Folorunso, 2013). The planned surface of the rock sample is then mounted on a glass slide with the use of Araldite and pressed together to remove air bubbles using forceps and placed on 55 the hot plate for a few minutes. The other end of the rock sample is further lowered by cutting a certain 56 57 thickness off using the cutting machine and is then ground on the glass plate using coarse grit carborundum powder followed by fine grit carborundum powder until the desired thickness of 30 58 microns (0.03mm) is achieved. At this thickness, light can pass through the rock sample thereby 59 revealing the mineral composition of such rock sample. A cover slip is used to cover the section with 60 61 the aid of the Canada balsam to prevent the composition from being altered. After the thin section of the rock, samples have been successfully prepared, then the mineralogical composition of these rock 62 samples is observed under a polarizing microscope. 63

#### 64 2.2 Geochemical Analysis

The first step in carrying out a geochemical analysis is pulverization. After the rock samples are selected for geochemical analysis, they are broken into small fragments with each fragment obtained from the cleanest, freshest portion of the entire rock sample to obtain distinct compositional results of the samples (Black, 1980). These small rock fragments are then pulverized (ground) such that there are no conspicuous grains observed in the sample. It is important to immediately transfer the pulverized sample into a tight sample holder to protect the composition of the bulk (Caby, 1981). The

geochemical analysis is basically used to analyze rock geochemistry, especially for the major, mir, and trace elements present in the rock and ore samples (Cox, 1979). The essence of geochemistry is to relate mineralogy with the chemistry of the rock. The major, minor, and trace element geochemical compositions were obtained by X-Ray fluorescence (XRF) while X-Rays diffraction (XRD) was used for mineralogical composition (Dada, 2006).

## 76 2.3 X-Ray Diffraction (XRD)

77 X-Ray diffraction is one of the important techniques for material characterization. The diffraction 78 experiments using x-rays help to study the structural properties of materials on an atomic scale 79 (Ahmed, 2015). The techniques are also used to measure crystallite size and calculate lattice strain, 80 and chemical composition, and determine phase diagrams as well (Arif, 2017). They are also one of the 81 most useful in the characterization of crystalline materials, such as metals, inter-metallic, ceramics, 82 minerals, polymers, plastics, or other inorganic or organic compounds., X-ray diffraction techniques 83 can be used to identify the phases present in samples from raw starting materials to finished product and to provide information on the physical state of the sample, such as grain size, texture, and the 84 crystal perfection. Most x-ray diffraction techniques are rapid and non-destructive, some instruments 85 are portable and can be transported to the sample (Ahmed, 2015). The type of information obtained 86 from x-ray diffraction studies ranges from sample composition to details of the crystal structure or the 87 state of orientation of the crystallites. Phase identification can be conducted on virtually all single 88 89 crystal or powder samples (Arif, 2017). Also, useful are measurements of the physical state of a sample 90 that detect differences from the ideal crystal (presence of defects, strain, etc). In powder diffraction, finely ground and homogenized samples are used. The sample may be held in thin-walled glass or 91 cellophane capillary tubes (Zandomeneghi, 2010). The x –rays (usually, K-radiation of wavelength 92 93 1.5418A emitted from copper) required for diffraction are mono-chromatized by filtering with the help of foils or crystal mono-chronometers. These x-rays are collimated and directed onto samples 94 (Rahman, 2016). In powder form, a significant number of crystallites can be expected to be oriented 95 to fulfill the Bragg condition for reflection from every possible interplanar spacing. 96

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#### 99 **3. Results and Discussion**

#### 100 *3.1 Geochemical*

101 The geochemical analysis showed the rocks and minerals to be diversely composed of 54-97% SiO<sub>2</sub>, 102 0.1-20% Al<sub>2</sub>O<sub>3</sub>, 0.8-4% MgO, 0.5-6% Fe<sub>2</sub>O<sub>3</sub>, < 1% K<sub>2</sub>O, TiO<sub>2</sub> and minor amounts of trace elements 103 including copper, manganese, and lead. These geochemical concentrations values obtained within the 104 study area are close to that reported by (Danbatta, 2007) for Precambrian rocks/ minerals in the Zuru 105 Schist Belt characterized by extremely low K<sub>2</sub>O; high Al<sub>2</sub>O<sub>3</sub>; and low TiO<sub>2</sub>, Cs, Ge, and Rb contents which 106 were in accordance with the orogenic nature of the Pan-African event.

107 The results of the geochemical analysis show that altered pegmatite (SO-1), syenite (SO-2), and 108 quartz vein (SO-4) are constituted with 54.8%, 63.61%, 98.86%, SiO<sub>2</sub>, low concentrations of

K<sub>2</sub>O, CaO, Fe<sub>2</sub>O<sub>3</sub>, MgO, TiO<sub>2</sub>, K<sub>2</sub>O, Na<sub>2</sub>Oand P<sub>2</sub>O<sub>5</sub> and minor trace elements (Table 1 and Fig. 3a-e). 109 110 These geochemical composition values suggest that the rocks underlying the area of llero and Okeho 111 possess a high abundance of silica that would support the granitic origin of the rocks (Dada, 2006). The amphibolite (SO-3) is however significantly rich in iron oxide, magnesia, and calcium oxide as to be 112 113 appreciated to be basic lithology. The Tourmaline ore mineral (SO-5) is a ubiquitous accessory mineral 114 in rocks of the Earth's crust. They can adjust their composition to suit a wide variety of environments, 115 and therefore display a remarkable range of instability in terms of pressure, temperature, fluid 116 composition, and host-rock composition.

The results of the geochemical analysis show that altered pegmatite (SO-1), syenite (SO-2), and quartz 117 118 vein (SO4) are constituted with 54.8%, 63.61%, 98.86%, SiO<sub>2</sub>, low concentrations of K<sub>2</sub>O, CaO, Fe<sub>2</sub>O<sub>3</sub>, MgO, TiO<sub>2</sub>, K<sub>2</sub>O, Na<sub>2</sub>Oand P<sub>2</sub>O<sub>5</sub> and minor trace elements (Table 4.1 and Fig. 4.1-4.5). These 119 120 geochemical composition values suggest that the rocks underlying the area of Ilero and Okeho possess a high abundance of silica that would support the granitic origin of the rocks (Dada, 2006). The 121 amphibolite (SO-3) is however significantly rich in iron oxide, magnesia, and calcium oxide as to be 122 appreciated to be basic lithology. The Tourmaline ore mineral (SO-5) is a ubiquitous accessory mineral 123 124 in rocks of the Earth's crust. They can adjust their composition to suit a wide variety of environments, 125 and therefore display a remarkable range of stability in terms of pressure, temperature, fluid

composition, and host-rock composition The correlation matrix for the data obtained (Table 2) shows 126 127 a wide variation in the correlation coefficient (r) between element pairs. For example, the r values 128 range from -0.05 between Rb and Al to 0.99 between Rb and Pb. A strong to very strong positive correlation occurs among Ti, Fe, S, and Cu positively correlated with each other. Considerable high 129 correlation was found between Rb and Pb (R = 0.99), S and Al (R = 0.69), S and Ti (R=0.81), S and Cu 130 (R=0.80), Cu and Al (R=0.80), Ti and Fe (R = 0.78). Other elements also showed a fair to strong 131 correlation as summarized in Table 2, (Okonkwo, 2013). According to Okonkwo and Folorunsho 132 (2013), the Fairly strong to very strong positive correlation between Ti, Fe, S, Cu, Al, Rb, and Pb in the 133 analyzed Rocks suggests the possibility of Ti-Fe-S-Cu-Al- Rb-Pb bearing minerals within the rocks of the 134 135 study area.

	SO-1	SO-2	SO-3	SO-4	SO-5
Rock/Mineral	Altered	Syenite	Amphibolite	Vein	Tourmaline
	Pegmatite			Quartz	(Ore mineral)
SiO <sub>2</sub>	54.80	63.61	39.23	96.86	61.82
TiO <sub>2</sub>	0.05	0.70	0.97	0.02	0.59
Al <sub>2</sub> O <sub>3</sub>	20.67	14.99	21.16	0.10	12.15
Fe <sub>2</sub> O <sub>3</sub>	0.57	5.19	30.98	0.54	3.72
MnO	0.02	0.32	1.32	0.03	0.37
MgO	4.20	3.82	3.69	0.79	5.85
CaO	0.45	0.44	4.38	0.04	3.66
Na <sub>2</sub> O	3.36	1.50	0.05	0.00	2.00
K <sub>2</sub> O	9.00	3.08	0.10	0.07	4.59
P <sub>2</sub> O <sub>5</sub>	0.12	0.15	3.61	0.31	0.50
LOI	22.12	3.00	2.60	0.60	2.12
SO <sub>2</sub>	0.04	0.05	0.08	0.00	0.08
CuO	0.03	0.01	0.03	0.00	0.03
PbO	0.01	0.02	0.00	0.07	0.01
Rb₂O	0.01	0.02	0.00	0.00	0.01
Cs <sub>2</sub> O	0.02	0.00	0.02	0.03	0.00
GeO <sub>2</sub>	3.08	1.00	1.03	0.40	1.00

136 Table 1. Result of Geochemical Analysis of Rocks and Minerals from Ilero Area.

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	Al	Fe	S	Ti	Cu	Pb	Rb	Ge
Al	1							
Fe	0.50	1						
S	0.69	0.58	1					
Ti	0.49	0.78	0.81	1				
Cu	0.80	0.39	0.79	0.36	1			
Pb	-0.16	-0.36	0.40	0.10	0.25	1		
Rb	-0.05	-0.26	0.51	0.18	0.37	0.99	1	
Ge	0.64	-0.19	0.07	-0.33	0.56	-0.16	-0.10	1

141 Table 2: Correlation between Ore Minerals of the Study Area





Fig 3a. Distribution of Major Oxides in Tourmaline (Ore Mineral).



Fig 3b. Distribution of Major Oxides in Altered Pegmatite.







# 162 3.2 X-Ray Diffraction Analysis (XRD)

163 The x-ray diffraction results are presented as charts in Figs. 4a-c. The mineral phases were 164 identified with respect to the International Committee on Powder Diffraction File (ICPDF) and their 165 abundances computer calculated (Zandomeneghi, 2010).





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Fig. 4a. XRD result of Syenite.







Fig. 4b XRD Chart of Amphibolite.





Fig. 4c. XRD result of Tourmaline (Ore mineral)

- 175 Table 3. Minerals Identified (Names of compounds and formulae) in Sample by X-Ray Diffraction

Sample	Mineral	Chemical Formula	Crystal System	Percentage
I.D	Identified			(%)
SO-2	Orthoclase	KAISi₃O <sub>8</sub>	Tectosilicate	56%
	Albite	NaAlSi₃O <sub>8</sub>	Tectosilicate	25%
	Quartz	SiO <sub>2</sub>	Hexagonal	
				19%
SO-3	Schorl	$NaFe^{3+}Al_6(BO_3)_3Si_6O_{18}(OH)_4$	Prismatic	50%
	Quartz	SiO <sub>2</sub>	Hexagonal	
				47%
SO-6	Actinolite	$Ca_2(Mg_5Fe^{2+}) Si_8O_{22}(OH)_2$	Monoclinic	69%

178 (Table 3) showed the XRD results of the mineral composition of the rocks within the llero area. 179 The XRD results showed that Albite (NaAlSi $_3O_8$ ) is found except for Tourmaline (Rahman, 2017). The 180 Syenite apart from being rich in Orthoclase had quartz (SiO<sub>2</sub>) and Albite. Syenite is a coarse-grained 181 intrusive igneous rock with a general composition like that of granite, but deficient in quartz, which, if present at all, occurs in relatively small concentrations (< 5%). Amphibolite had Actinolite and Albite. 182 Amphibolite is a dark green foliated rock predominantly composed of green hornblende, plagioclase, 183 184 and quartz, and a minor occurrence of black pyroxene, quartz, and epidote. Tourmaline (ore mineral) showed Schorl, and Quartz (Rahman et al., 2017; Arif, 2017). Tourmaline generally occurs as a minor, 185 186 accessory phase in rocks, but tourmalines, where tourmaline is a major mineral, are well known (Arif, 187 2017). Tourmaline forms in a wide variety of settings, including as a diagenetic mineral in buried 188 sedimentary basins, as a gangue mineral in ore deposits, associated with a contact, regional and 189 subduction-related metamorphism, in metasomatism, and as a significant mineral typically crystallizing 190 in fractionated igneous bodies (Arif, 2017). The peaks on the diffraction spectra (Fig. 4.6-4.8) signified 191 the abundance of each of these minerals in their different samples.

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## 193 3.3 Petrology and Heavy Mineral Study

194 Thin sections of rocks and ore minerals (Tourmaline (Ore mineral) from Ilero environs studied under the petrological microscope showed the rocks (Altered Pegmatite, Syenite, Amphibolite, and 195 196 Vein Quartz) to be coarse-grained textured with euhedral mineral crystals (Nwachukwu, 2011). In 197 general, the minerals present include quartz (20%), orthoclase (6%), hornblende (7%), tourmaline (2%), 198 and biotite (40%) (Black, 1980). (Fig 5a-j) shows the minerals observed both in plain polarized light (PPL) and under crossed Nicol (XPL). Quartz and biotite are the most abundant minerals observed. 199 200 Quartz constitutes about 20% of the mineral content of the Vein Quartz and its morphology rarely forms euhedral crystals except in some veins which are anhedral (Danbatta, 2007). Quartz is an 201 extremely common mineral in continental crustal rocks such as Granite, Pegmatite, etc. It is the most 202 203 common vein-forming mineral. Biotite is one of the most common rock-forming minerals (Folorunso, 204 2013). It is a major constituent of rocks of intermediate composition, such as granodiorites, diorites, monzonites, and some nepheline syenites. 205



208 Fig 5a: Photomicrograph of sample Vein Quartz. showing Q-Quartz, B- Biotite, OM-Opaque Mineral

- Magnification x4.

![](_page_13_Picture_5.jpeg)

Fig 5b: Photomicrograph of sample Altered Pegmatite. showing Q-Quartz, B-Biotite, OR-Orthoclase Feldspar

Magnificationx4.

![](_page_14_Picture_0.jpeg)

![](_page_14_Figure_1.jpeg)

![](_page_14_Picture_4.jpeg)

![](_page_14_Picture_5.jpeg)

![](_page_14_Figure_7.jpeg)

Fig 5d: Photomicrograph of Vein Quartz, showing B-Biotite. Magnification x4

![](_page_15_Picture_0.jpeg)

- 221 Fig 5e: Photomicrograph of Amphibolite showing B- Biotite, OR-Orthoclase Feldspar, H- Hornblende, OM-
- 222 Opaque Mineral. Magnification

![](_page_15_Picture_4.jpeg)

Fig 5f: Photomicrograph of sample Vein Quartz showing B-Biotite Magnification x4

![](_page_16_Picture_0.jpeg)

Fig5g: Photomicrograph of Vein Quartz showing B-Biotite, Q- Quartz Magnification x4

![](_page_16_Picture_3.jpeg)

Fig 5h: Photomicrograph of Tourmaline (Ore Mineral) showing B-Biotit, T-Tourmaline, Q-Quartz.
Magnification x4

![](_page_17_Picture_0.jpeg)

![](_page_17_Figure_1.jpeg)

![](_page_17_Picture_4.jpeg)

![](_page_17_Picture_5.jpeg)

Fig 5j: Photomicrograph of sample Altered Pegmatite showing B-Biotite, OR-Orthoclase Feldspar. Magnification
 x4

![](_page_18_Figure_0.jpeg)

Fig 5: Bar chart showing estimated modal mineralogical composition of granite under the microscope.

![](_page_18_Figure_3.jpeg)

Minerals	Optical Dramatics	Crossed Po	olarized	Plane Polarized	Percentage in Thin	
	Properties	(XPI)	:11	(ppi)	Section	
	Color	Bluish to milky				
Orrente	Relief:	Low relief		Calaardaaa	20%	
Quartz	Form/size	Anhedral		Colourless		
	Inclusion	Absent				
	Cleavage	Absent				
	Color	Brown		Brown	40%	
	Relief:	Low				
D:	Form/size	Euhedral				
Biotite	Inclusion	Absent				
	Cleavage	Two (2) cleavage				
		direction				
	Color	Green				
	Relief:	Low			25%	
Chlorite	Form/size	Euhedral		Green		
	Inclusion	Absent				
	Cleavage	Present				
	Color	Brown				
	Relief:	High		Brown	7%	
Hornblende	Form/size	Prismatic				
	Inclusion	Present				
	Cleavage	Present				
Tourmaline	Color	Dark Brown /	Yellow			
	Relief:	High				
	Form/size	Slender Prism			2%	
	Inclusion	Absent		Dark Brown		
	Cleavage	Present				
Orthoclase	Color	Grey				
Feldspar	Relief:	Low			6%	
	Form/size	Prismatic		Grey		
	Inclusion	Present				
	Cleavage	Present				
Total					100%	

251 Table 4.: Petrographical Analysis of Ilero Samples

#### 256 **3. Conclusion**

The correlation analysis of some selected elements from the result of the geochemical analysis shows 257 Ti, Fe, S, Cu, Al, Rb, and Pb among others to be strongly correlated. The conclusion drawn from the 258 259 integration of the geochemical analysis, and correlation analysis in this study is that the study area is 260 mineralized with gibbsite-cuprite-pyrite-galena-cassiterite-pollucite. The result of the petrographic study 261 reveals that the rocks and ore minerals in the study area are predominantly whitish-grey-greenish in color 262 with the textural compositions of porphyritic to fine-grained structures. Petrographically, the rocks and ore 263 minerals in the study area are composed of major rock-forming silicate minerals such as quartz, biotite, 264 hornblende, tourmaline, and feldspars. The XRD results showed that Albite, Orthoclase, guartz, Schorl, and 265 Actinolite are possible mineralized minerals in the samples of the study area. Based on the petrographical, XRD, and geochemical analysis obtained, the results suggest that the study area possesses highly 266 mineralized with economical minerals. 267

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