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Barometers behaving badly: Assessing the influence of analytical and experimental uncertainty on clinopyroxene thermobarometry calculations at crustal conditions

Key Words: Clinopyroxene, Thermobarometry, Analytical uncertainty, Experimental uncertainty, Monte Carlo

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21

23 **Abstract**

24 The composition of clinopyroxene and Clinopyroxene-Liquid (Cpx-Liq) pairs are
25 frequently used to calculate crystallization/equilibration pressures in igneous systems.
26 While canonical uncertainties are often assigned to calculated pressures based on fits to
27 calibration or test datasets, the sources of these uncertainties (and thus ways to reduce them)
28 have not been rigorously assessed. We show that considerable uncertainties in calculated
29 pressures arise from analytical error associated with Electron Probe Microanalyser (EPMA)
30 measurements of Cpx. Specifically, low X-ray counts during analysis of elements with
31 concentrations <1 wt% resulting from insufficient count times and/or low beam currents
32 yield highly imprecise measurements (1σ errors of 10–40% for Na_2O).

33 Low analytical precision propagates into the calculation of pressure-sensitive mineral
34 components such as jadeite. Using Monte Carlo approaches, we demonstrate that elemental
35 variation resulting from analytical precision alone generates pressures spanning ~4 kbar
36 (~15 km) for a single Cpx and ~6 kbar for a single Cpx-Liq pair using popular barometry
37 expressions. In addition, analytical uncertainties in mineral compositions produce highly
38 correlated arrays between pressure and temperature that have been previously attributed to
39 transcrustal magma storage. Before invoking such geological interpretations, the more
40 mundane origin from analytical imprecision must be ruled out. Most importantly, low
41 analytical precision does not just affect the application of barometers to natural systems; it
42 has also affected characterization of Cpx in experimental products used to calibrate and test
43 barometers. The impact of poor precision on each individual measurement is also often
44 magnified by the small number of measurements made within experimental charges,
45 meaning that low analytical precision and true variability in mineral compositions have not
46 been sufficiently mediated by averaging multiple EPMA analyses. We compile the number

47 of Cpx measurements performed in N=295 experiments used to calibrate existing
48 barometers, and N=459 new experiments, finding ~40% of experiment charges were
49 characterized by ≤ 5 individual Cpx analyses. Insufficient characterization of the true
50 composition of experimental phases likely accounts for the fact that all Cpx-based
51 barometers exhibit large errors (± 3 kbar) when tested using global experimental datasets.

52 We suggest specific changes to analytical and experimental protocols, such as increased
53 count times and/or higher beam currents when measuring low concentration elements in
54 relatively beam resistant Cpx in experiments and natural samples. We also advocate for
55 increasing the number of analyses per experimental charge, resolving interlaboratory
56 analytical offsets and improving data reporting. Implementing these changes is essential to
57 produce a more robust dataset to calibrate and test the next generation of more precise and
58 accurate Cpx-based barometers. In turn, this will enable more rigorous investigation of
59 magma storage geometries in a variety of tectonic settings (e.g., distinguishing true
60 transcrustal storage vs. storage in discrete reservoirs).

61 **1. INTRODUCTION**

62 Constraining the conditions under which magma is stored and transported within the crust is
63 of critical importance to understanding volcanic systems (e.g., McGuire et al., 2017). It has
64 long been recognised, based on thermodynamic principles and from chemical characterization
65 of experimental products, that the composition of igneous minerals and co-existing liquids can
66 be used to place constraints on the pressures, temperatures, and water concentrations at which
67 these phases grew and equilibrated (e.g., Bacon and Carmichael, 1973; Brown and Parsons,
68 1981; Lindsley and Andersen, 1983; Putirka, 1999, 1997, 2008). Many of the most commonly
69 used mineral-based igneous thermobarometers revolve around clinopyroxene (Cpx), perhaps
70 because this phase is relatively abundant in a wide variety of different volcanic systems and
71 tectonic settings (e.g., mid-oceanic ridges, oceanic islands, and volcanic arcs), and in a wide

72 range of lava compositions (basalts to rhyolites). Cpx-based equilibria are sensitive to pressure
73 because there is a large change in volume associated with the exchange of Na and Al (the
74 jadeite component, $\text{NaAlSi}_2\text{O}_6$) between liquid (Liq) and Cpx. There are also relatively large
75 volume changes between different Cpx components (e.g., jadeite and diopside-hedenbergite,
76 $\text{CaMgSi}_2\text{O}_6$ - $\text{CaFeSi}_2\text{O}_6$, Putirka, 2016; Putirka et al., 1996). The exchange of jadeite and
77 diopside-hedenbergite between clinopyroxene and liquid is also sensitive to temperature, so
78 the abundance of these components can be used as a thermometer (e.g., Putirka, 1999).

79 By parameterizing the relationship between phase composition and intensive parameters in
80 experiments, measurements of natural crystals and co-existing equilibrium liquids can be used
81 to calculate pressures and temperatures (e.g., Neave and Putirka, 2017; Petrelli et al., 2020;
82 Putirka, 1999, 2008; Wang et al., 2021). The vast majority of published calibrations have a
83 structure rooted in thermodynamics, both in terms of the form of the expression, and the
84 compositional components included (e.g., jadeite in Cpx).

85 For example, both the Putirka (1996) eqP1 and Putirka (2008) eq30 models for Cpx-Liq
86 barometry have the same general form informed by thermodynamics:

$$87 \quad P = a + \frac{bT}{10^4} + \frac{cT}{10^4} \ln(\text{Jd}^{\text{Cpx-Liq}}) + \dots \quad \text{[Equation 1]}$$

88 Additional empirical composition terms are added to improve regression statistics (Neave and
89 Putirka, 2017), although sometimes these terms can also be approximately tied to
90 thermodynamic reasoning. For example, the $X_{\text{Na}}^{\text{Liq}} X_{\text{Al}}^{\text{Liq}}$ term in Putirka (1996) eqP1 implies
91 identical activity coefficients for Na and Al in the liquid. In general, an increasing number of
92 empirical terms have been added to barometry equations to improve the fit between calculated
93 and experimental pressures in calibration datasets as more experimental data have become
94 available. For example, while both eqP1 (P1996) and eq30 (P2008) have additional empirically
95 derived terms for the sum of the cation fraction of Na and Al, eq30 has additional terms for the

96 liquid Mg#, the diopside-hedenbergite component in the Cpx, and the log of liquid Fe, Mg and
97 K cation fractions.

98 More recently, machine learning algorithms have been calibrated for equilibria involving Cpx
99 (Higgins et al., 2022; Jorgenson et al., 2022; Petrelli et al., 2020). Instead of creating equations
100 with specific terms informed by thermodynamics (as in Equation 1), these machine learning
101 techniques simply input selected oxide data from a training dataset into the machine learning
102 algorithm, along with the predictor variable (e.g., pressure or temperature; Petrelli et al., 2020).

103 When discussing errors and uncertainties associated with different thermobarometers in this
104 manuscript, we follow the National Physics Laboratory terminology guidelines (Bell, 2001).

105 Error is taken as the difference between the measured value and the true value; if a barometer
106 calculates a pressure of 3 kbar but the experiment was performed at 2 kbar, the error is +1 kbar.

107 Uncertainty quantifies the doubt about the measurement result, so an error with an unknown
108 magnitude is referred to as an uncertainty. A random uncertainty or error means that if the
109 measurement is repeated, a different value is obtained each time, but averaging of sufficient
110 measurements converges on the true value. Random uncertainties may follow a normal or
111 Poisson distribution (or other unskewed distribution). However, the central limit theorem states
112 that as the sample size grows, a distribution of sample means approximates a normal
113 distribution (even if the data aren't normally distributed). Systematic uncertainties or errors
114 cause the measurement to be offset from the true value, so additional measurements do not help
115 to converge on the correct answer, meaning that more measurements produce a more precise,
116 but inaccurate result.

117 The success of a given barometer or thermometer is normally assessed by comparing predicted
118 pressures and temperatures with experimentally determined pressures and temperatures. The
119 goodness of fit is typically assessed using R^2 values, estimates of the standard error estimate

120 (SEE) or the root-mean-square error (RMSE). In general, Cpx and Cpx-Liq thermometers have
121 high R^2 values (>0.8) and relatively low SEE (e.g., ± 20 – 100 K, e.g., Putirka, 2008, Petrelli et
122 al., 2020). These SEEs represents only a 2-10% error considering the temperatures of most
123 magmatic systems (~ 1000 – 1400 K). In contrast, barometers commonly have SEE of 1.4-5 kbar
124 ($\sim \pm 6$ – 19 km, using $\rho = 2700$ kg/m³ throughout this paper), which corresponds to very large
125 percentage errors given the depths of interest in many volcanic systems. For example, these
126 large SEEs span the entire crustal column in many tectonic settings (e.g., 5–8.5 km in MORB,
127 White et al., 1992, 14–24 km in Hawai'i, Leahy et al., 2010). This means that pressures
128 calculated on individual Cpx analyses do not have sufficient precision/resolution to reliably
129 identify upper, mid or lower crustal storage, or distinguish storage in distinct magma reservoirs
130 separated by ~ 1 – 2 km (unlike melt inclusion saturation pressures, which can achieve such
131 precision at relatively low pressures, Lerner et al., 2021; Wieser et al., 2022; Wieser et al.,
132 2021). Even in arcs with Moho pressures of 8 ± 3 kbar (based on $\sim 30 \pm 12$ km from Profeta et al.,
133 2016), these SEE estimates can only just distinguish between upper, mid and lower crustal
134 storage. As many of the uncertainties associated with thermobarometers are random
135 uncertainties, averaging multiple experiments (or natural Cpx analyses) can result in significant
136 improvements. For example, Putirka et al. (1996) showed that the SEE on individual
137 experiments using their eqP1 is 1.36 kbar, but can be reduced to 0.32 kbar if they average
138 experiments conducted at the same pressure. The importance of such averaging is discussed
139 in detail in Section 2.3.1.

140 In general, more recent calibrations of mineral-only and mineral-melt barometers state smaller
141 SEEs (e.g., ± 1.4 kbar for Cpx-Liq from Neave et al., 2017, ± 1.66 kbar for Cpx-only from Wang
142 et al., 2021, vs. ± 3.6 kbar for equation 30 and ± 5 kbar for eq32c of Putirka, 2008). However,
143 statistics quoted by different studies are not directly comparable. For example, the ± 1.4 kbar
144 SEE commonly stated by petrological studies using the Neave and Putirka (2017) barometer

145 describes the model fit to the calibration dataset of 113 experiments. When this
146 thermobarometer is applied to their global compilation of experimental data that span a wider
147 compositional range than was used in the calibration, the error is ± 3.6 – 3.8 kbar. When applied
148 to test data with compositions more similar to the calibration dataset (i.e., tholeiitic basalts),
149 the Neave and Putirka (2017) barometer return errors similar to the quoted error of 1.4 kbar.
150 The ± 1.66 kbar error from Wang et al. (2021) describes the overall fit to 100 random splits into
151 training and validation of the calibration dataset. However, when applied to global datasets
152 (including data not used at any point during model tuning), this barometer has a SEE of ± 3.68
153 kbar. Finally, the Cpx-only and Cpx-Liq machine-learning barometer of Petrelli et al. (2020)
154 have SEEs of ± 3.1 kbar and ± 3.2 kbar, respectively, when applied to a subset of $N=119$
155 experiments not used for calibration. The “global” statistics for newer barometers are more
156 similar to the larger quoted errors of Putirka (2008), who calculate statistics using a global
157 calibration dataset.

158 It is also important to recognise that the quoted SEE describes the overall fit across the range
159 of pressures or temperatures being considered. Close to the edges of the distribution (e.g., very
160 shallow or very deep pressures), the error can be larger than the SEE, and towards the centre
161 of the distribution, the error can be smaller. Additionally, the SEE estimated from a barometer
162 between 0–40 kbar (e.g., Putirka, 1996) is not directly comparable to one estimated between
163 0–20 kbar (e.g., Petrelli et al., 2020), which can also make comparison of statistics from
164 different barometers challenging.

165 Despite the fact that it is difficult to directly compare statistics, we can say in general that the
166 majority of Cpx-based barometers show large SEE (± 2 – 5 kbar, 7.6–19 km) when applied to
167 global datasets. Given that significantly greater precision in crustal barometry is needed to
168 address many of the key issues in igneous petrology (McGuire et al., 2017), we try to reconcile

169 the source of these large errors, to identify ways in which to improve Cpx-based barometric
170 methods in future.

171 **1.1. A new test dataset – ArcPL**

172 The majority of Cpx-based barometers were calibrated using experiments present in the LEPR
173 dataset compiled in 2008 (Library of Experimental Phase Relationships; Hirschmann et al.,
174 2008, e.g., N=850 experiments in Petrelli et al., 2020). For clarity, in this paper the word
175 experiment is used to represent a single experimental run/capsule/charge. In addition to using
176 experimental data within LEPR, we also compile Cpx-bearing experiments on variably hydrous
177 compositions ranging from basalt to rhyolite at crustal conditions (<15 kbar) that were not
178 included in LEPR. Most of the studies represented in this new dataset were published after
179 LEPR, and have not been added to it yet (Almeev et al., 2013; Alonso-Perez et al., 2009;
180 Andújar et al., 2015; Blatter et al., 2013, 2017; Bogaerts et al., 2006; Cadoux et al., 2014;
181 Costa, 2004; Erdmann and Koepke, 2016; Erdmann et al., 2016; Feig et al., 2010; Firth et al.,
182 2019; Hamada and Fujii, 2008; Husen et al., 2016; Krawczynski et al., 2012; Mandler et al.,
183 2014; Melekhova et al., 2015; Mercer and Johnston, 2008; Nandedkar et al., 2014; Neave et
184 al., 2019; Parat et al., 2014; Parman et al., 2011; Pichavant and Macdonald, 2007; Rader and
185 Larsen, 2013; Riker et al., 2015; Ulmer et al., 2018; Waters et al., 2021). We also include a
186 small number of pre-2008 experiments which were not included in LEPR (Berndt, 2004;
187 Rutherford et al., 1985; Sisson et al., 2005). We call our new dataset ArcPL (Arc post-LEPR,
188 N=469). One advantage of ArcPL is that many of the experiments were conducted relatively
189 recently, meaning that it was easier to obtain additional analytical information from the authors,
190 and more information on analytical conditions was generally presented in the text and
191 supporting information. We use these two datasets of experiments (LEPR and ArcPL) to place
192 quantitative constraints on sources of uncertainty when calibrating and testing Cpx-based
193 barometers.

1.2. Sources of uncertainty

Thermobarometer calibrations based on experimental products are subject to three main sources of uncertainty:

1) Uncertainties relating to the regression method used to calibrate thermobarometry equations and models (e.g., extrapolation, overfitting, incomplete data, Fig. 1a).

2) Analytical uncertainties associated with measurements of minerals and glasses, normally by electron probe microanalyser (EPMA, e.g., insufficient counts, heterogeneity in primary standards within a single chip and between chips, beam damage, instrument miscalibration, Fig. 1b).

3) Experimental errors (e.g., crystallization under disequilibrium conditions, uncertainty in the pressure in a piston-cylinder, Fig. 1c), as well as additional issues associated with analysing small experimental products (Fig. 1d).

Each of these are discussed in detail below.

1.2.1 Model formulation and calibration

One unavoidable reason why barometers have higher percentage errors than thermometers relates to thermodynamics; namely, major and minor element variations in available igneous phases are simply more sensitive to temperature than pressure, because relevant reactions have larger changes in entropy (relating to T) than volume (relating to P). This is in contrast to subsolidus metamorphic systems where a relatively large number of very P-sensitive phase transitions and chemical reactions are available (Powell and Holland, 2008). The paucity of P-sensitive reactions between solid mineral phases and silicate melts reflects the ability of liquid-rich igneous systems to absorb changes in volume (and to a lesser extent temperature and composition) without drastically changing the composition of the coprecipitating phases (Putirka, 2008). Thus, it is expected that regressions of experimental Cpx compositions against

219 pressure will have relatively poor predictive power (Fig. 1a), as many variables other than
220 pressure have more influence on the reaction (e.g., temperature, water content, melt
221 composition).

222 However, there are a number of sources of uncertainty that can be introduced during the
223 calibration of barometric models which decrease the accuracy and precision of barometers
224 beyond some theoretical limit determined by the thermodynamics of each reaction. Identifying
225 these is vital to produce the best possible barometers for igneous systems.

226 Firstly, the calibration dataset may span a restricted compositional, pressure or temperature
227 range, and empirical fits developed on this dataset may return poor results when extrapolated
228 outside this range (Fig. 1a). For example, Neave and Putirka (2017) cautioned users about the
229 limited applicability of their model in highly oxidised and alkaline systems, as it was
230 predominantly calibrated using tholeiites. Similarly, Wang et al. (2021) attributed the poorer
231 performance of their barometer at pressures >12 kbar to the lack of high-pressure experiments
232 in their calibration dataset. Models can also be overfitted to calibration data. This is a particular
233 problem for certain machine-learning algorithms such as extra tree regressors which can
234 perfectly fit the training data, but produce erratic results on test data (see section 2.2).

235 A second source of uncertainty relates to the fact that not all experimental studies report the
236 concentrations of all the major and minor elements in Cpx and Liq used when performing
237 thermobarometry calibrations. This is problematic for several reasons. Firstly, missing
238 elements from the quantification routine during EPMA analysis can affect the concentration of
239 elements which are measured. For example, EPMA analyses calculating water by difference
240 without a subsequent iterative matrix correction produce water estimates that are low by as
241 much as 1 to 2 wt% absolute (Roman et al., 2006). This is equally true for other unanalyzed
242 elements (Fournelle et al., 2020). Missing elements also cause issues during regression

243 analysis. It is generally considered that up to 5% missing data on each attribute (i.e., a single
244 oxide) are manageable during regression, 5–15% requires use of more sophisticated
245 approaches, and >15% can severely impact the prediction accuracy of the model on unseen
246 data (Pyle, 1999; Twala, 2009). Unfortunately, the percentage of missing data in experimental
247 compilations used to calibrate barometers is well within the severe category. For example, in
248 the LEPR dataset used by Petrelli et al. (2020) and Putirka (2008), 66% of experiments don't
249 report H₂O contents in the glass phase and 44% do not report Cr₂O₃ in the Cpx. Yet, both these
250 elements feature heavily in parameterizations of Cpx-only and Cpx-Liq barometry, and the
251 presence/absence of Cr actually affects phase stability (discussed in detail below, Voigt et al.,
252 2017).

253 There is abundant literature, but little consensus, about how to deal with missing values during
254 regression analysis (Emmanuel et al., 2021). The simplest way is list-wise deletion, where
255 experiments with any missing values for any oxides are simply removed from the calibration
256 dataset. However, in the experimental datasets used to calibrate thermobarometers, which have
257 large amounts of missing data, list-wise deletion can result in a dramatic reduction in the
258 number of observations used for calibration. For example, 54% of experiments in the
259 calibration dataset of Petrelli et al. (2020) do not report P₂O₅ in glass, 56% do not report Cr₂O₃
260 in glass, and 69% do not report K₂O in Cpx. List-wise deletion would leave only 31/850
261 experiments, which is far too small a calibration dataset. The software JMP used by Putirka
262 (2008, and previous papers) uses list-wise deletion by default; the shrinking dataset issue was
263 mitigated by excluding elements from the calibration which result in lots of rows being deleted,
264 but likely do not have a major influence on the equilibria of interest (e.g., MnO in liquid, K₂O
265 in Cpx).

266 List-wise deletion can be very problematic if missing values reflect an underlying bias in the
267 measurement procedure (termed “missing not at random” by Rubin, 1976). For example, it is

268 far more likely that H₂O-poor or nominally anhydrous experiments do not report H₂O in
269 experimental glasses compared with very water-rich experiments. Removing experiments with
270 no reported water could easily bias the remaining dataset towards compositions and phase
271 assemblages found in more hydrous magmas, meaning the resulting model would be poorly
272 calibrated for relatively anhydrous melts. For H₂O-poor experiments, it is far better to enter
273 H₂O=0 into the dataset than to remove the experiment. Missing H₂O data has been dealt with
274 in several ways. Petrelli et al. (2020) fill all missing values for H₂O (and any other elements)
275 with zeros.

276 Alternatively, missing values can be imputed (i.e., estimated). For example, Putirka (2008)
277 replaced missing H₂O data with 100 minus the sum of major element totals (i.e., volatiles by
278 difference). While this method is associated with large uncertainty unless calibrated using
279 hydrous standards (Hughes et al., 2019), it is likely better than replacing missing values with
280 the mean or median of the dataset (which can create issues when data are not missing at random,
281 as discussed above for H₂O; Twala, 2009).

282 The presence/absence of reported Cr₂O₃ (or Cr) data in Cpx demonstrates one challenge
283 associated with imputation for experimental datasets. In many cases, experiments on synthetic
284 starting materials do not report Cr concentrations in Cpx or Liq because no Cr was added (e.g,
285 Hamada and Fujii, 2008), whereas in other studies, Cr was likely present but not reported. For
286 example, Parat et al. (2014) use a natural starting material with ~187 ppm Cr, but report no
287 Cr₂O₃ data in Cpx or glasses. Using an approximate partition coefficient (~10, Hart and Dunn,
288 1993), their Cpx could contain ~0.1–0.2 wt% Cr₂O₃. Finally, there are experiments where it is
289 unclear whether Cr is present or not. For example, the natural starting materials of Cadoux et
290 al. (2014) are sufficiently evolved (67–71 wt% SiO₂) that it is unlikely there is very much Cr
291 left, but the actual Cr content of the starting material (and experimental Cpx) are not reported.

292 The paucity of Cr data (resulting from Cr-free starting materials and the lack of Cr
293 measurements in Cr-bearing starting materials) has been discussed in detail in the context of
294 MORB differentiation and liquid-barometry by Voigt et al. (2017). Voigt et al. (2017) showed
295 that Cr-containing experiments stabilize Cpx at higher temperatures than Cr-free experiments.
296 Pressures calculated using their melt geobarometer for Cr=0 versus true Cr contents in MORBs
297 differ by up to 1.5 kbar. Thus, replacing missing values with the mean of the dataset, or any
298 other prediction, would fail to capture the fact that there truly was no Cr, which has likely
299 influenced Cpx stability. We quantitatively evaluate the effect of missing Cr data on the Cpx-
300 only and Cpx-Liq thermobarometers in Section 2.

301 *1.2.2 Analytical Uncertainty*

302 In addition to thermodynamic considerations (larger changes in entropy compared with
303 volume), the larger percentage errors associated with barometers compared with thermometers
304 may result from the lower concentration of elements that are important in barometry compared
305 with thermometry, meaning analytical errors are larger (e.g., Na in Cpx for pressure vs. Mg-Fe
306 for temperature). A number of random and systematic uncertainties can arise during EPMA
307 analyses. Each wavelength dispersive spectrometer (WDS) is calibrated for a specific element
308 using the relationship between the peak-background ratio and the concentration of that element
309 in a primary standard material. In turn, this relationship is used to predict the element
310 concentration in an unknown material based on the measured peak-background ratio. A matrix
311 correction is applied to account for the influence of the specific material analysed on the
312 intensity of measured X-rays (e.g., atomic number effects, absorption, and fluorescence).

313 The generation of X-rays from a sample excited by an electron beam is a random process (like
314 radioactive decay), meaning that the emission of any given X-ray cannot be predicted.
315 However, if X-ray arrivals are collected over a long enough time interval, the average number
316 of X-ray arrivals per unit time is a function of the rate of X-ray production. X-ray production

317 is determined by the specific element and electron shell ionization efficiencies, element
 318 concentration, beam current, beam voltage and detector take off angle. The instrumental
 319 efficiency of X-ray detection is controlled by the WDS crystal size/geometric efficiency and
 320 detector efficiency (as well as the vacuum). X-ray intensities are then normalized to beam
 321 current, yielding counts per second per nA. Instrumental specific effects are accounted for
 322 when converting counts into concentration by performing the X-ray measurements using the
 323 same beam energy and spectrometer on both the primary standard and the unknown material
 324 to produce a k-ratio, which should be identical within measurement error for all instruments
 325 using the same element, X-ray emission line, beam energy and takeoff angle (assuming the
 326 high voltage calibration and effective take off angle of the spectrometers are sufficiently
 327 similar).

328 The precision of an EPMA measurement is determined by the total accumulated number of X-
 329 rays counted by the WDS detector at the peak and background positions. Errors directly
 330 resulting from the fundamentally random process of X-ray generation and detection are termed
 331 counting statistics. When averaged over sufficient time period, X-ray counts follow a Poisson
 332 distribution which approximates a normal distribution at sufficiently high counting rates (i.e.,
 333 the central limit theorem). Errors resulting from counting statistics can be estimated using two
 334 main methods.

335 Equation 2 can be used to predict the relative error (100*standard deviation (σ)/concentration)
 336 for a given element in an unknown sample using information from count rates in the primary
 337 standard (Weill et al., 2013), the relative concentration of the element in the primary standard
 338 and sample, and the time spent counting:

$$339 \quad \text{Error (\%)} = 100 \times \sqrt{\frac{\frac{X_{\text{sample}}(P_{\text{std}} - B_{\text{std}})}{X_{\text{std}}} + 2B_{\text{std}}}{\left(\frac{X_{\text{sample}}(P_{\text{std}} - B_{\text{std}})}{X_{\text{std}}}\right)^2 t}}}{\quad}} \quad \text{(Equation 2)}$$

340 where P_{std} and B_{std} are the counts per second (cps) of the peak and background for the element
 341 in the primary standard, X_{std} is the element concentration in the primary standard (wt%), X_{sample}
 342 is the concentration in the unknown sample (wt%), and t is the count time on the unknown
 343 sample (s).

344 Equation 2 assumes that the total count times on the peak and background are the same and
 345 does not account for the different matrices of the primary standard and the sample. It can be
 346 made more versatile by including a factor for the probe current (I), substituting counts per
 347 second on the primary standard with counts per second per nA ($P_{I, std}$ and $B_{I, std}$):

$$348 \quad \text{Error (\%)} = 100 \times \sqrt{\frac{\frac{X_{sample} (I P_{I, std} - I B_{I, std})}{X_{std}} + 2 I B_{I, std}}{\left(\frac{X_{sample} (I P_{I, std} - I B_{I, std})}{X_{std}}\right)^2 t}} \quad \text{(Equation 3)}$$

349
 350 When the measured total counts ($P_{Tot, sample}$) on the peak and background ($B_{Tot, sample}$) are known,
 351 A more precise way of calculating the relative error for a specific element is given by:

$$352 \quad \text{Error (\%)} = 100 \times \frac{\sqrt{\frac{P_{Tot, sample}}{t^2} + \frac{B_{Tot, sample}}{t^2}}}{\left(\frac{P_{Tot, sample}}{t} - \frac{B_{Tot, sample}}{t}\right)} \quad \text{(Equation 4)}$$

353 Equations 2–4 yield very similar results (often within ~1%), with slight discrepancies resulting
 354 from the fact that Equation 4 accounts for differences in the matrix of the sample and standard,
 355 which affects the intensity of generated X-rays, and slight differences in the background count
 356 rate on the sample and standard. The main advantage of Equations 2–3 is that they can be used
 357 predictively to assess the approximate precision on an unknown using count rates from primary
 358 standards on a given spectrometer and crystal. Equation 4 is best applied after an analysis has
 359 been conducted, and is what most commercial EPMA software uses to output 1 sigma errors
 360 for a specific analysis.

361 Uncertainties resulting from counting are almost always random (i.e., following a Poisson
362 distribution), whereas instrument calibration can introduce systematic errors. Historically,
363 because the accuracy of the physical models used for performing matrix corrections were
364 limited (though today that is generally no longer the case), EPMA analyses of geological
365 materials have tended to use primary standards with similar matrices to the target analytes (e.g.,
366 Na in Albite for analysis of Na in feldspar, Na in jadeite for analysis of Na in pyroxene). The
367 most common primary standards are from the Smithsonian National Museum of Natural
368 History (NMNH) collections (Jarosewich, 2002; Jarosewich et al., 1980). However, it is well
369 documented that different chips of the same natural NMNH standard may differ from published
370 compositions determined by wet chemistry, and that heterogeneity can also exist within a single
371 chip (Fournelle, 2012; Jarosewich et al., 1980; Rose et al., 2008). As each EPMA laboratory
372 has its own set of primary standards, for any given named standard (e.g. San Carlos olivine),
373 heterogeneity between the different chips in different laboratories can result in systematic
374 offsets between measurements made in these laboratories. Different laboratories also rely on
375 different primary standards to calibrate a specific element, increasing the variability. Some
376 laboratories also utilize “home grown” materials obtained from their mineral collections or
377 synthetically produced “in-house” standards, and some commercially available standard
378 materials may not be what are advertised. For example, some commercial “San Carlos olivine”
379 has not been split from the Smithsonian mineral collection, but rather was merely collected
380 from the same geographical vicinity.

381 The resulting systematic offsets in instrument calibration can be identified using round-robins,
382 where the same material is analysed as an unknown in different laboratories using their routine
383 calibration strategy. Hunt et al. (1998) distributed a Lipari Obsidian sample to five laboratories,
384 and Kuehn et al. (2011) distributed 3 different glass compositions (rhyolite, phonolite, basalt)
385 to 27 laboratories. The Kuehn round-robin identified a number of outlier laboratories with

386 large systematic errors, especially with regards to the correction of time dependent intensity
387 (TDI) effects, sometimes referred to as volatile element or beam damage effects. These changes
388 in emitted intensity are usually due to ion mobility within the interaction volume (Morgan and
389 London, 2005 and Hughes et al., 2019) and are observable on not only highly mobile alkali
390 elements such as Na and K (which can dramatically decrease in intensity during the
391 measurement and therefore must be extrapolated back to their zero time intensities), but also
392 for less mobile elements such as Si and Al, whose apparent intensities increase as the alkali
393 element intensities decrease due to migration towards subsurface charge accumulation. Such
394 TDI corrections can range from under a few percent (relative) to 30–100%. Thus, changes in
395 counts during beam exposure should be carefully inspected for each element in representative
396 samples when performing a study on hydrous/alkaline melt compositions and used to inform
397 decisions about which elements to perform TDI corrections for. As many Cpx-Liq
398 thermobarometers are parameterized in terms of Na and K (and Si and Al) in the liquid (e.g.,
399 Neave and Putirka, 2017, eq30–31 of Putirka, 2008), different TDI correction routines used
400 to measure different experiments could add significant scatter to an experimental dataset
401 compiled from many different laboratories.

402 At various points over the last few decades, authors have used smaller round-robins to identify
403 systematic offsets in glass measurements and have corrected data compiled from these different
404 laboratories to produce a more consistent calibration dataset. For example, Yang et al. (1996)
405 performed corrections to glass data collected at both Massachusetts Institute of Technology
406 (MIT) and the Smithsonian prior to performing modelling and thermobarometric calculations.
407 Most recently, Gale et al. (2013) performed a round-robin using VG2 to obtain correction
408 factors for MORB glasses relative to EPMA analyses from Lamont Doherty Earth Observatory
409 (hereafter, Lamont; Fig. 2). To our knowledge, no such reconciliation has been performed for

410 datasets used to calibrate Cpx-based barometers (or any other mineral barometers), so we
411 quantify the possible effects of interlaboratory offsets in Section 2.1.

412 Additional systematic and random analytical uncertainty during EPMA analyses can result
413 from problems relating to beam damage/charging of materials under the electron beam, poor
414 sample preparation (e.g., bad polishing, sample tilt) and drift of the instrument (e.g., stage Z
415 height, changes in the temperature and pressure in the room; Fig. 1b). It should also be
416 mentioned that other sources of systematic analytical inaccuracy can be introduced from
417 improperly calibrated instruments, for example assumed dead time constants and picoammeter
418 readings from poorly maintained instruments. However, these are almost impossible to
419 quantify retrospectively from published data, so are not discussed further.

420 *1.2.3 Experimental scatter*

421 Analysis of the experimental products used to calibrate thermobarometers are subject to the
422 analytical uncertainty discussed above, along with several other sources of uncertainty (Fig.
423 1c). Firstly, experimental phases may be heterogeneous because of zoning resulting from
424 changing P-T-fO₂-H₂O conditions during the run, remnant seed crystals, or kinetic effects
425 during growth and sample quench. Capsules may also experience Fe and H⁺ exchange with
426 samples and assemblies (Botcharnikov et al., 2005; Gaetani and Grove, 1998), resulting in poor
427 redox buffering. Relict crystals from the starting material may be present if the starting material
428 was not fully melted (Mutch et al., 2016). The relatively small size of experimental products
429 also enhances EPMA uncertainty relative to natural crystals, because the interaction volume of
430 the electron beam may directly enter other phases, and neighbouring phases may influence the
431 measurement through secondary fluorescence (Llovet and Galan, 2003, Fig. 1d).

432 There are also a number of sources of uncertainty that can arise from the experimental set up.
433 Experiments conducted at <~5–7 kbar were mostly performed using TMZ/MHZ vessels

434 (Sisson and Grove, 1993a, 1993b) and internally heated pressure vessels (IHPV, e.g., Cadoux
435 et al., 2014; Di Carlo, 2006; Hamada and Fujii, 2008; Parat et al., 2014). Pressure in IHPV
436 experiments is monitored with strain gauge manometers, Heise Bourdon tube gauges, or digital
437 pressure transducers, which have a precision of $\sim 0.01\text{--}0.04$ kbar (1–4 MPa, Koepke et al.,
438 2018; Pichavant, 2002). Pressure variations during experiments are mostly within ~ 0.05 kbar
439 (e.g., Parat et al., 2014). Because pressure is transmitted via a gas to the capsule, excluding
440 calibration offsets in gauges/manometers/transducers, an experimental pressure recorded in
441 one lab is likely to be similar to that in another lab.

442 In contrast, most experiments conducted at > 5 kbar were performed in piston cylinders (Baker
443 and Egger, 1987; Gaetani and Grove, 1998; Mercer and Johnston, 2008), where a solid-
444 medium transmits pressure to an enclosed capsule. Pressure in the piston cylinder apparatus is
445 commonly monitored using a Heise gauge, and maintained to within ± 0.5 kbar (Hamada and
446 Fujii, 2008). In solid-medium experiments the pressure experienced by the experimental
447 capsule can differ from the pressure applied to the hydraulic piston (measured with the Heise
448 gauge), with the correction between the two often referred to as a friction correction
449 (Condamine et al., 2022; Tamayama and Eyring, 1967). This friction/pressure correction can
450 be evaluated for each experimental set up by running experiments for reactions which are well
451 constrained in P-T space, and applying a shift to account for any offsets between the observed
452 and predicted transition.

453 However, different laboratories investigating Cpx equilibria correct for friction in different
454 ways, which could easily introduce bias to an experimental dataset compiled from many
455 different studies. For example, Mercer and Johnston (2008) and Draper and Johnston (1992)
456 do not include a friction correction. Likewise, Bartels et al. (1991) do not apply a correction,
457 because their offsets calculated from the melting point of Au are only 0–1 kbar, which is within
458 the uncertainty of the Au calibration (Akella and Kennedy, 1971). Grove et al. (2003) corrected

459 for the offset using the reaction between anorthite+gehlenite+corundum \rightarrow Ca-tshermakite
460 (now kushiroite, Hays, 1966), and although they do not give the magnitude of the offset, they
461 stated that after this correction pressures are accurate to ± 0.5 kbar. Gaetani and Grove (1998)
462 used this Ca-tshermakite/kushiroite reaction in addition to the melting point of Au, resulting in
463 a friction correction of 300 MPa. In contrast to this constant offset applied to 12, 16 and 20
464 kbar experiments, Hamada and Fuji (2008) apply a -10% correction for their 4 kbar
465 experiments, and a -15% correction for their 7 kbar experiments based on a calibration using
466 the water solubility of albite melt (Behrens et al., 2001). Ulmer et al. (2018) stated that they
467 follow the experimental approach of Villiger (2004), who applied a correction of -3%
468 calibrated against the quartz-coesite transition at 3.07 GPa and 1000°C (Bose and Ganguly,
469 1995) and fayalite+qtz \rightarrow orthoferrosilite reaction at 1.41 GPa and 1000°C (Bohlen et al.,
470 1980). Finally, Blatter et al. (2013, 2017) calibrated at each pressure by bracketing the CsCl
471 melting curve (Bohlen and Boettcher, 1982).

472 The wide variation in the reactions chosen for calibration (and the variation in the pressures
473 and temperatures of these calibration reactions) means that it is perfectly plausible that
474 systematic offsets still exist between different laboratories, up to the magnitude of the
475 corrections applied (e.g. ~ 10 – 15% , or a few kbar). For example, Johannes et al. (1971) showed
476 that measurements of the pressure of the albite \rightarrow jadeite+quartz reaction at 600°C using the
477 same starting material in six different laboratories varied from 15.7 kbar to 16.8 kbar
478 Additionally, Condamine et al. (2022) showed that friction can decrease during a run (from
479 29% in 6 hr runs to 21% in 24 hr runs), which means that the true pressure experienced by the
480 capsule may also be influenced by the length of the experimental run, which varies greatly
481 between studies. Johannes et al. (1971) also pointed out that the friction correction can also
482 vary between runs conducted with identical protocols, because of a softer than average batch
483 of the solid-medium material, or more efficient wall lubrication. This means that a piston

484 cylinder experiment reported at 10 kbar in one laboratory may not have experienced the same
485 pressures as an experiment performed at 10 kbar in another laboratory, and variations may exist
486 even within experiments run at 10 kbar in a single laboratory. Uncertainties in the true pressure
487 experienced by the charge explains why experimentalists tend to space out their piston-cylinder
488 pressures within a single study, to ensure that trends are at least coherent. However, when all
489 these data are compiled to calibrate a barometer, offsets on the order of 0.5–1 kbar will increase
490 scatter, and could create systematic uncertainties based on different laboratories investigating
491 different regions of PT space.

492 Temperature variations in piston cylinders are generally measured with thermocouples (e.g.,
493 Pt-Rh, Baker and Egger, 1987, W-Re, Mercer and Johnston, 2008), although the exact
494 geometry varies in different laboratories. Although thermocouples are reasonably precise and
495 accurate ($\pm 10^\circ\text{C}$), the larger source of error results from thermal gradients within the piston
496 cylinder, requiring a correction to be made to account for the higher temperatures in the hotspot
497 where the experimental capsule is placed versus the position of the thermocouple tip. These
498 corrections depend on the capsule geometry and assembly, but can easily reach 20–40°C
499 (Brugman et al., 2021; Grove et al., 2003). Variable thermal gradients between the capsule and
500 thermocouple could result in systematic differences between runs and laboratories, and thermal
501 gradients within capsules themselves (Harlow, 1997) can cause crystals grown in different parts
502 of the capsule to show different compositions (increasing the compositional scatter seen in
503 experiments). Putirka et al. (1996) stated that the most consistent pyroxene analyses are when
504 rim compositions are analysed along with the neighbouring liquid interface, perhaps helping to
505 mitigate the effect of thermal gradients (and thus compositional variations throughout the
506 capsule).

507

508 **2.1. Quantifying the effect of interlaboratory EPMA analysis offsets**

509 The dataset of experiments from LEPR used by Putirka (2008) and Petrelli et al. (2020) to
510 calibrate their Cpx \pm Liq thermobarometers lists 45 unique experimental laboratories, with a
511 relatively small number of laboratories performing a large proportion of experiments
512 (MIT=25%, Penn State=7%, Lamont=6.4%, Hannover and University of Oregon=4%). If all
513 laboratories investigated the same P-T-X conditions, and interlaboratory offsets were normally
514 distributed about the nominally correct value, inter-laboratory EPMA offsets would simply add
515 random uncertainty to calibration and test datasets. However, as different experimental
516 laboratories, which mostly use a specific EPMA instrument, tend to target different research
517 questions, the occurrence of these analytical offsets in the experimental dataset is not uniformly
518 distributed, and could thus lead to systematic uncertainty. For example, in the LEPR dataset,
519 at pressures >10 kbar, experiments performed at Australia National University (ANU) tend to
520 focus on higher SiO₂ liquids than experiments performed at MIT or University of Tasmania
521 (Supporting Fig. 1–2). Thus, interlaboratory offsets mean the calibration could be skewed as a
522 function of liquid composition.

523 Interlaboratory offsets can also affect the statistics determined from a test dataset if a specific
524 laboratory has an offset relative to the average offset of the LEPR dataset (or even the average
525 value in a specific P-T-X region of calibration dataset). It is noteworthy that the relative
526 contribution of different laboratories at crustal pressures (0–13 kbar) has changed dramatically
527 between our newly compiled ArcPL dataset and LEPR. For example, experiments conducted
528 at Hannover account for 40% of the ArcPL, but just 1% of the LEPR dataset, while MIT
529 comprises 10.4% of the LEPR dataset and 10.5% of the ArcPL dataset. If, say the EPMA lab
530 at Hannover (or any other laboratory which has greater output since LEPR was compiled) had
531 an offset from one of the dominant EPMA in the LEPR dataset used for calibration, this could

532 help to account for the fact that new experimental data are predicted poorly by existing
533 barometry calibrations.

534 In general, interlaboratory analytical differences from glass round-robins are <10% (Gale et
535 al., 2013; Hunt et al., 1998; Kuehn et al., 2011), although occasionally it is noted that one or
536 more laboratories produce discrepancies >10% (their identity is kept anonymous in Hunt et al.,
537 1998). Compared to other analytical techniques, relatively little attention has been paid to
538 interlaboratory EPMA offsets in petrology, with most studies being conducted on silicate
539 glasses by the tephrochronology community (Hunt et al., 1998; Kuehn et al., 2011). To our
540 knowledge, no round-robins have been conducted on pyroxene or other silicate minerals. As
541 many of the EPMA's used to measure the pre-2008 experiments in the LEPR dataset have been
542 decommissioned, we conclude that we will never be able to fully determine the exact value of
543 offsets for Cpx. In the absence of other data, we assume that the interlaboratory offsets for
544 VG2 glass reported by Gale et al. (2013; Fig. 2a-b) are a first-order estimate of the
545 interlaboratory differences that may have occurred during Cpx analysis. The round-robin
546 conducted by Gale et al. (2013) is particularly useful because it targeted a number of
547 laboratories which have analysed experimental charges used to calibrate and test various
548 thermobarometers.

549 To assess whether 0–10% offsets between different EPMA laboratories could adversely affect
550 thermobarometric calibrations and assessment using test datasets, we consider the average
551 reported Cpx and glass compositions from experiments from Krawczynski et al. (2012)
552 analysed on the MIT EPMA. We multiply these measured compositions by the correction
553 factors from Gale et al. (2013) to obtain the compositions that would have been measured on
554 the Lamont EPMA. Using the interlaboratory comparisons to Lamont from Gale et al. (2013),
555 we also estimate the composition that would have been measured at a number of other EPMA
556 labs. We then calculate pressures and temperatures for these corrected compositions by

557 iterating eq32d (T) and eq32b (P) for Cpx-only thermobarometry, and eq33 (T) and eq31 (P)
558 for Cpx-Liq thermobarometry (equations from Putirka 2008).

559 The range of calculated pressures resulting from these laboratory offsets is significant (Fig. 2c-
560 f), and varies greatly between different experiments and different thermobarometry expressions
561 (Fig. 2, Supporting Figs. 3–5). Cpx-Liq pressures that would have been calculated from mineral
562 compositions measured in different laboratories show offsets from one another of ~0.5 to 6
563 kbar (Fig. 2d, f, Supporting Figures 3–5), while Cpx-only pressures show offsets of ~3 to 5
564 kbar (Fig. 2c, e, Supporting Figures 3–5). Interlaboratory offsets also impact calculated
565 temperatures (~10 to 50 K), but to a lesser extent. These interlaboratory offsets likely contribute
566 to the overall scatter between calculated versus experimental pressures in calibration and test
567 datasets. In fact, it is noteworthy that our calculated offsets are similar in magnitude to quoted
568 SEEs on barometers. These offsets also affect natural samples which barometry equations are
569 applied to; it is highly concerning that calculated pressures may vary by 3–5 kbar based solely
570 on the EPMA laboratory used to perform analyses.

571 **2.2. Variability in treatment of Chromium**

572 As discussed in the introduction, many experiments in the calibration dataset do not report Cr
573 data, yet the Cpx-only thermobarometers of Wang et al. (2021, eq1, 2, 32dH), Putirka (2008,
574 eq32b, 32d) and Petrelli et al. (2020) include a term for the Cr content of the Cpx. For Cpx-Liq
575 thermobarometry, only the model of Petrelli et al. (2020) is parameterized directly in terms of
576 the Cr content of the Cpx and the liquid. However, Cr in the Cpx is used to calculate the CrCaTs
577 component of the Cpx, which is then used to calculate the diopside-hedenbergite (DiHd)
578 component:

$$579 \text{CrCaTs} = 0.5 * \text{Cr}_{\text{cat frac}}$$

$$580 \text{DiHd} = \text{Ca}_{\text{cat frac}} - \text{CaTs} - \text{CaTi} - \text{CrCaTs}.$$

581 The DiHd component is included in the Cpx-Liq barometers of Putirka (2008, eq30 and 31),
582 Neave and Putirka (2017), and the Cpx-only barometers of Putirka (2008, eq32a, 32b).
583 Additionally, even Cpx-Liq or Cpx-only barometers which have no compositional term
584 dependent on Cr in Cpx are normally parameterized in terms of Cpx cation fractions, either
585 directly or in terms of components like jadeite which are calculated from cation fractions. If Cr
586 is present in the sample, but not measured, the calculated cation fractions of other elements
587 such as Na (and therefore jadeite) are artificially high. Thus, a wide range of barometers are
588 directly or indirectly sensitive to the Cr content of the Cpx.

589 The proportion of missing Cr data is very similar in the calibration dataset and ArcPL. Missing
590 Cr data in the calibration dataset may have resulted in the relationship between components
591 dependent on Cr and P and T being incorrectly parameterized (or correctly parameterized, but
592 with the addition of significant noise). Missing Cr data in our test dataset will also cause noise
593 when comparing predicted versus experimental pressures, because if Cr was present but not
594 reported, calculated pressures and temperatures using true Cr contents may differ from those
595 calculated using the Cr=0 wt% (and would perhaps lie closer to the 1:1 line). To investigate
596 the possible offsets caused by not reporting Cr when it is actually present, we calculate
597 pressures and temperatures for all Cpx and Cpx-Liq pairs in the ArcPL dataset with reported
598 Cr data, and compare this to calculations using Cr=0 wt% (Fig. 3).

599 Different barometers show different sensitivity to Cr. For the Wang et al. (2021) Cpx-only
600 barometer and thermometer, there is a clear correlation between the change in pressure and the
601 Cr content of the Cpx, with the most Cr-rich Cpx showing pressures up to 2.8 kbar too high
602 when Cr is set to 0 wt% (Fig. 3a). The effect on temperature is proportionally smaller (~ 44 K,
603 Fig. 3b). Strong correlations with true Cr content are also seen for pressures calculated using
604 Putirka (2008) eq32d-32b, with max offsets of 2 kbar (and 16 K for temperature). In contrast,
605 the change in pressure using the Petrelli et al. (2020) Cpx-only barometer shows no apparent

606 relationship to the Cr content of the Cpx, and shows significantly larger variations than the
607 other barometers (max $\Delta P=4.9$ kbar, $\Delta T=125$ K). This unpredictable response reflects the use
608 of decision trees, where the Cr content affects the route taken through the tree.

609 For Cpx-Liq thermobarometers, iteration of Putirka (2008) eq33 (T) with eq30 (P), or eq33 (T)
610 with Neave and Putirka (2017, P, red triangles) shows very little sensitivity to Cr content (apart
611 from a single Cpx-Liq pair showing a large change for all equations, Fig. 3c). Iteration of
612 various thermometers with eq32c for pressure shows slightly more sensitivity, with offsets of
613 up to $\Delta P=1$ kbar and $\Delta T=10$ K. As for their Cpx-only thermobarometer, the extra-trees
614 regressions for Cpx-Liq of Petrelli et al. (2020) are significantly more sensitive to Cr, with
615 temperatures varying by up to ± 100 K simply by changing the Cr content, and no clear
616 correlation between the offset and the actual Cr content of the Cpx. Overall, these comparisons
617 show that the presence/absence of Cr data in both calibration and test datasets can clearly
618 introduce noticeable uncertainty in terms of calculated pressures (>1 kbar).

619

620 **2.3. Quantifying the effects of analytical versus experimental variability in test and** 621 **calibration datasets**

622 Variability in measured phase compositions within a single experiment can also affect
623 calibration and test datasets. Analytical uncertainty associated with the random process of X-
624 ray generation can produce variability in measured phase compositions, even if the
625 experimental phases themselves are entirely homogeneous. Because analytical uncertainty
626 resulting from counting statistics is normally distributed and random, if infinite numbers of
627 compositionally homogenous Cpx crystals are analysed from a single experiment, the mean
628 composition will be the same regardless of the precision of each analysis. However, if only a
629 very small number of Cpx are measured, analytical uncertainty can easily yield a reported

630 average composition that is significantly different from the true composition. In the more likely
631 scenario that experimental Cpx are also chemically heterogeneous (e.g., zoned or sector zoned),
632 large numbers of EPMA analyses are required to correctly characterize the average
633 composition of each experiment, even if each EPMA analysis is highly precise. For example,
634 Neave et al. (2019) noted that their Cpx show highly variable Al and Ca contents, which they
635 attributed to sector zoning. They suggested based on their observations of chemical
636 heterogeneity that at least 20–40 analyses are required to meaningfully characterize the
637 composition of Cpx.

638 *2.3.1. Are we sufficiently averaging experimental and analytical variability?*

639 To address the possible influence of analytical and experimental variability, we compile the
640 number of reported Cpx analyses in the 459 experimental charges from the ArcPL dataset to
641 investigate how many publications approach the Neave et al. (2019) recommendation of 20–
642 40 analyses. We supplement the ArcPL dataset with a compilation of Cpx measurements in
643 295 experimental charges in the LEPR dataset conducted at 0–13 kbar in arc-like compositions.
644 We note that we do not know for each individual experimental charge whether the number of
645 analyses reported in the associated paper represents measurements of N discrete Cpx crystals,
646 or N analyses, with some Cpx being characterized by more than 1 EPMA spot (e.g., perhaps
647 10 analyses from 8 Cpx crystals).

648 In 40 experimental charges (5% of the combined dataset) only a **single** Cpx analysis was
649 performed, 47 (6%) performed just 2 Cpx analyses, 67 (9%) performed 3 Cpx analyses, and 77
650 (10%) performed 4 Cpx analyses (Fig. 4a-b). Overall, 43% of the compiled experimental runs
651 were characterized by ≤ 5 Cpx analyses, 25% by ≥ 10 Cpx analyses, and only 5% by ≥ 20 Cpx
652 analyses (as recommended by Neave et al., 2019).

653 Concerningly, the experiments with the lowest number of analyses tend to be concentrated at
654 lower pressures, where Na₂O contents are generally lower and more susceptible to large
655 analytical uncertainties (Fig. 4a). It is notable that the discrepancy between the calculated
656 pressure using the reported Cpx composition and the true experimental pressure increases with
657 decreasing number of Cpx analyses (for the Wang et al. 2021 and Putirka, 2008 Cpx-only
658 barometers, Fig. 4c, d). This indicates that insufficient averaging of analytical variability or
659 true phase variability (e.g., zoning) is affecting the performance of barometers. As the median
660 number of analyses per experiment is 6 for both our new dataset and the subset of LEPR we
661 have compiled the number of analyses for, we assume the statistics given above are
662 representative of the entire LEPR database used for literature thermobarometry calibrations.

663 Assessing the relative influence of analytical uncertainty and experimental variability in
664 experimental compositions is vital to address how barometry calibrations can be improved. If
665 variability in measured Cpx compositions results from true variation in the composition of that
666 phase (e.g. zoning and other disequilibrium processes), it means we must direct more attention
667 to understanding and identifying disequilibrium in experimental products, measure very large
668 numbers of phases from each experiment (e.g., Neave et al., 2019), and possibly redo
669 experiments that have not sufficiently approached equilibrium (or remove these experiments
670 from the calibration dataset). In contrast, if analytical variability is the primary culprit, it means
671 we need to direct our attention to improving EPMA analyses of experimental products.
672 Hereafter, we refer to true variation in Cpx compositions in experimental charges as “phase
673 variability” and variability resulting from EPMA analyses as “analytical precision.”

674 2.3.2. *Distinguishing analytical and experimental variability*

675 Unfortunately, the information required to quantify the relative influence of phase variability
676 versus analytical precision is largely absent from the information provided in most legacy
677 experimental studies. In-text pdf tables are normally used to report the mean composition and

678 standard deviation of each phase in each experimental charge. It is very unusual for the
679 individual EPMA analyses from each experiment to be reported, except in the most recent
680 publications (e.g., Erdmann et al., 2016; Neave et al., 2019; Waters et al., 2021). This makes it
681 difficult to assess the true variability in Cpx compositions, because the quoted mean and
682 standard deviation for each phase in each experiment only provide a good description of the
683 underlying data if it is normally distributed, and lacks covariance between oxides. Yet, almost
684 all of the full experimental datasets we have been able to obtain (Blatter et al., 2013;
685 Krawczynski et al., 2012; Melekhova et al., 2015; Neave et al., 2019) do not show normally
686 distributed elemental variations, and exhibit significant correlations amongst different oxides
687 (e.g., Fig. 5a, Supporting Fig. 6–15).

688 The covariance structure of oxides in each phase gives important clues into the source of
689 variability in experimental products. For example, strong correlations between $\text{SiO}_2\text{-TiO}_2$,
690 $\text{SiO}_2\text{-Al}_2\text{O}_3$, $\text{SiO}_2\text{-MgO}$, $\text{Al}_2\text{O}_3\text{-MgO}$ in Cpx are indicative of sector zoning (e.g., Fig. 3b,
691 Neave et al., 2019; Ubide et al., 2019, Supporting Fig. 6–14), whereas elements with variability
692 arising from EPMA counting statistics will be uncorrelated (e.g., $\text{MnO-Na}_2\text{O}$, Fig. 5c).
693 Importantly, when the full data are not reported, simulations using the reported mean and
694 standard deviation provide a poor match to the real experimental data when a strong covariance
695 structure is present (e.g., Fig. 5b, simulated Cpx as yellow dots, measured Cpx as red dots from
696 Neave et al., 2019). The poor match of data simulated using just the mean and standard
697 deviation means that it is nearly impossible to simulate the true variability in the LEPR dataset,
698 where individual analyses could not be recovered.

699 Similarly, the information required to reconstruct the analytical precision for each element is
700 seldom (if ever) reported. Ideally, authors would report the software-calculated sigma values
701 for each analysis, which uses the number of counts on the peak and background of a specific
702 EPMA spot to estimate the uncertainty related to counting statistics (Equation 4). Alternatively,

703 more approximate estimates of error can be calculated from Equations 2–3, but this requires
704 knowledge of the counts on the primary calibration standards, as well as the beam current and
705 count times used for each element. We were unable to find any relevant experimental papers
706 reporting software-calculated errors or sufficient information to use Equations 2–4. However,
707 we were able to obtain software-calculated precisions upon request from the authors for a
708 subset of experimental Cpx analyses from Krawczynski et al. (2012) conducted on the old MIT
709 JEOL 733 and the newer MIT JEOL8200 (installed ~2007), and from Neave et al. (2019) on
710 the Hannover Cameca SX100 (installed ~2001, decommissioned ~2021). We supplement these
711 estimates with precision from natural pyroxene analyses on the University of Cambridge
712 Cameca SX100 (Gleeson et al., 2021) and secondary standard analyses for the Kakanui Augite
713 from the Oregon State University Cameca SX100. We also obtained peak and background
714 counts from primary standard analyses on the MIT JEOL 733 (installed in the 80s) and the
715 Bristol JEOL JXA8530F (installed ~2012) to model precision as a function of count time and
716 beam current.

717 For the following discussion, we focus on Na to highlight an element affected by analytical
718 error that is pertinent to the majority of Cpx barometry parameterizations, because of its
719 relationship to jadeite content. As expected, compiled analytical precisions from different
720 EPMA systems increase in magnitude as Na₂O content decreases below 1 wt% (Fig. 6a, c),
721 closely following the trajectory of modelled precision curves. The exact trajectory of these
722 precision versus concentration curves depends on spectrometer efficiency (ratio of peak:
723 background counts in a primary standard), as well as the energy of the beam (voltage, current)
724 and the count time. As the majority of Cpx analyses are performed at 15 kV, variations in beam
725 current and count time can be combined into a single variable, $I \cdot t$ (10s and 20 nA gives
726 identical counting statistics to 20 s and 10 nA).

727 Using count data from the Bristol JEOL JXA8530F TAP crystal for Na on an Albite primary
728 standard, we calculate that measurement of a Cpx with $\text{Na}_2\text{O}=0.13$ wt% if $I^*t=60$ nA s has a
729 precision of 25% (e.g., 10 s, 6 nA), while measurement if $I^*t=400$ nA s has a precision of 9.7%
730 (e.g., 20s, 20 nA, Fig. 6c). For the MIT JEOL 733 with a TAP crystal calibrated for Na on
731 Albite, the precision for $\text{Na}_2\text{O}=0.13$ wt% is 60% for $I^*t=60$ nA s, and 23% for $I^*t=400$ nA s
732 (Fig. 6a). The lower precision on the Bristol versus MIT JEOL for a given I^*t value reflects
733 the increase in detector efficiency on the newer Bristol instrument. This can also be seen by
734 comparing the software-reported precision for the old MIT JEOL 733 (orange dots, Fig. 6a) to
735 the newer MIT JEOL 8200 (red dots, Fig. 6a), where all other variables are kept constant (same
736 calibration routine, same current and count times).

737 When information is available on both the software-reported analytical precision for each
738 individual EPMA spot and the true phase variability observed in measured Cpx compositions
739 within a single experimental charge, these can be compared. If variability for a specific element
740 is dominated by analytical precision, the percentage variation calculated from multiple
741 measurements of Cpx in that experimental charge ($100 \cdot \sigma / \text{Mean concentration}$) will be the
742 same as the % precision value predicted from counting statistics. In contrast, if true phase
743 variability is dominant (e.g. zoning, disequilibrium), the observed variability in measured Cpx
744 compositions will greatly exceed the analytical precision. We were only able to obtain
745 estimates of analytical precision for the experiments of Krawczynski et al. (2012) and Neave
746 et al. (2019). In both sets of experiments, the analytical precision for Na_2O is very similar to
747 the observed variability for measurements of Na_2O in Cpx (Fig. 6a, c). In contrast, the
748 variability of Al_2O_3 in experimental Cpx greatly exceeds the analytical precision, indicating
749 true phase variability (Fig. 6b, d). In general, variability in elements with >1 wt% concentration
750 in the experiments of Neave et al. (2019) and Krawczynski et al. (2012) significantly exceed
751 the analytical precision, while lower abundance elements (<1 wt%) show variability similar to

752 that expected from counting statistics (Fig. 6, Supporting Figs. 16–17). While the strong
753 correlations between Al₂O₃ versus TiO₂ and Al₂O₃ versus SiO₂ in the experiments of Neave et
754 al. (2019) are strongly indicative of sector zoning (compare Supporting Figs. 6, 11), the weaker
755 correlations for the Krawczynski et al. (2012) data make it hard to know the exact origin of
756 true phase heterogeneity without detailed elemental mapping of crystals (Supporting Fig. 8).

757 We wish to determine whether elemental variability for the other experiments in the LEPR
758 database are similarly controlled by analytical precision for Na₂O, and true phase variability
759 for higher concentration elements. While we cannot reconstruct the exact analytical precisions
760 for these studies, we can use the curves calculated for the new Bristol JXA8530F and old MIT
761 JEOL733 probes as end-member examples of detector efficiency to estimate precision for
762 different I*t values. This approach is supported by the fact that the software-calculated Cameca
763 SX100 errors from Hannover lie within the modelled lines for these two probes (Fig. 6c). To
764 get an idea of common I*t values, we compile count times and beam currents from 39 randomly
765 selected experiments in LEPR used to calibrate the Putirka (2008) and Petrelli et al. (2020)
766 Cpx-only and Cpx-Liq expressions, as well as a number of Cpx containing experiments
767 conducted since 2008 (see Supporting Table 1). This compilation is also hampered by
768 insufficient reporting of analytical data. Of the 39 LEPR experimental studies, 31 stated the
769 beam current while only 16 (~40%) gave the count time for Na₂O. The 22 experiments
770 published since 2008 are somewhat better, with 21 providing beam current, and 14 (~60%)
771 providing count times for Na₂O. Although incomplete, our compilation shows that I*t mostly
772 varies between 60 and 400 nA s (Supporting Table 1). Interestingly, many papers include
773 statements to the effect that well known or established analytical conditions were used for
774 analysis without actually reporting what these were. The range of compiled I*t values shows
775 that this is clearly insufficient, as there is no such thing as “normal” analytical conditions
776 (Supporting Table 1).

777 When the worst-case scenario (60 nA s on the MIT JEOL733, orange line) and best-case
778 scenario (400 nA s on the Bristol JX8530F, blue dashed line) are overlain on the reported 1
779 sigma values for element variability within a single experimental charge from LEPR (Fig. 6e-
780 f), it is apparent that the majority of Cpx Na₂O variability reported within single experiments
781 results from analytical precision. The various measures of software-reported precision we
782 compiled for different EPMA facilities (e.g., Cambridge, OSU) lie in the middle of the cloud
783 defined by LEPR experiments, further supporting this assertion. The small number of points
784 that lie above the MIT JEOL733 line could be influenced by experimental scatter, or even lower
785 spectrometer efficiency than the MIT JEOL733 (or I*t<60 nA s). In contrast to Na₂O, Al₂O₃
786 in LEPR experiments show significantly more variability than software-estimated precision,
787 indicating that features such as sector zoning may be nearly ubiquitous in experimental Cpx
788 (Fig. 6f).

789 Our modelling and compilations of analytical precision highlight a major issue with the way
790 EPMA precision is being assessed and reported within the petrology community. If mentioned
791 at all, precision is generally discussed with reference to repeated measurements of secondary
792 standard materials (e.g., Neave et al., 2019; Waters et al., 2021). Although secondary standards
793 are very helpful to assess accuracy and drift during an analytical session, they only provide
794 useful insights into precision if secondary standards and samples have similar concentrations
795 for a given element. This is a particular problem for Cpx, because the commonly used Kakanui
796 augite secondary standard has 1.2–1.3 wt% Na₂O (Fig. 6b), whereas experimental Cpx
797 produced at <10 kbar have median Na₂O contents of only 0.55 wt% (and most crustal Cpx have
798 similarly low concentrations). The theoretical lines calculated for the MIT JEOL 733 show that
799 analysis of Kakanui augite would indicate 5–10% error, but the lower Na₂O contents of crustal
800 Cpx yield errors of up to ~40% (Fig. 6c). Thus, secondary standards must only be used to

801 estimate precision for elements where the standard and sample have very similar
802 concentrations.

803 *2.3.3. Effect of major element variability on calculated pressures*

804 For experiments where individual Cpx measurements within a single experimental charge
805 could be obtained, we calculate Cpx-only pressures for each reported measurement (Erdmann
806 et al., 2016; Neave et al., 2019, Blatter et al., 2013; Krawczynski et al., 2012; Melekhova et
807 al., 2015). Despite the fact that Cpx within a single experimental charge experienced a narrow
808 range of pressures within their capsule, calculated pressures span up to ~ 4 kbar (Fig. 7a-e).
809 The Cpx-only barometer of Putirka (2008) shows the widest range in most cases (red
810 diamonds), with the barometers of Wang et al. (2021, black dots) and Petrelli et al. (2020, cyan
811 squares) showing a slightly smaller range for a given experiment. The range of calculated
812 pressures in a single experiment is in many cases comparable to the reported 1 sigma error on
813 the barometer (coloured error bar). Visually, it is easy to see how this variation can lead to
814 large reported errors on barometers if only a small number of Cpx are analysed. For example,
815 if Neave et al. (2019) had measured just 1 Cpx, the pressure calculated using the Cpx-only
816 barometer and thermometer of Putirka (2008, eq32a-32d) could range from ~4 to 10 kbar. In
817 turn, when plotted against experimental pressure (3 kbar), this pressure calculation could lie 1
818 to 7 kbar off the 1:1 line. When all experiments in a test dataset are considered, these offsets
819 cause very large RMSE/SEE estimates on the barometer.

820 However, it is still unclear from these discussions whether major elements (predominantly
821 controlled by true phase variability) or minor elements (controlled by analytical precision) are
822 producing this spread of calculated pressures in a single experimental charge. Qualitatively,
823 when we plot the calculated pressures from five experiments against Na₂O for two barometers,
824 the strong correlation between calculated pressures and Na indicates that analytical precision
825 is a major contributor to the spread in calculated pressures (Fig. 7f).

826 To investigate the relative effect of analytical precision versus true phase variability more
827 quantitatively, we use Monte Carlo simulations implemented in the Python3 thermobarometry
828 tool Thermobar (Wieser et al., 2022b). For specific experimental charges from Krawczynski et
829 al. (2012) and Neave et al. (2019), we create 5000 synthetic Cpx compositions with the
830 concentration of each element distributed normally about the mean composition of the Cpx
831 from each experiment, and the standard deviation equal to the average reported 1 sigma error
832 from EPMA software for analyses from that experimental charge. Each major element in each
833 of the 5000 Cpx is sampled randomly from a normal distribution. The uncoupled variations of
834 these different oxides means that cation sums in these simulated Cpx do not lie substantially
835 outside the distribution of cation sums in the measured Cpx (Supporting Fig. 18-19). We then
836 calculate pressures and temperatures for these synthetic Cpx using a variety of Cpx-only
837 barometers (Fig. 8–9). To aid visualization of the spread of simulated pressures and
838 temperatures, we contour the results of the Monte Carlo simulations and overlay contours
839 incorporating 67% and 95% of simulated pressures and temperatures using the Python3 tool
840 Pyrolite (Williams et al., 2020). If analytical variability is the dominant cause of pressure-
841 temperature variability, the Monte Carlo simulations will encompass the variability of
842 calculated P and T for each measured Cpx composition. In contrast, if true phase variability
843 dominates, calculated P and T for measured Cpx compositions (yellow and red dots) will plot
844 well outside the PT region defined by the simulations (Fig. 8-9).

845 For experiment Y0272–1 from Neave et al. (2019), the 95% contour around the Monte Carlo
846 simulation almost completely incorporates the spread of calculated pressures and temperatures
847 in individual Cpx measured in that experiment for all three Cpx-only thermobarometers
848 (yellow dots, Fig. 8a-c, see also Supporting Fig. 20). Thus, while experimental products show
849 major element variability resulting from sector zoning (Neave et al., 2019), the dominant
850 control on calculated pressures and temperatures results from analytical precision. For Cpx-

851 Liq calculations, no EPMA-estimates of glass analytical uncertainties could be obtained for
852 Neave et al. (2019), so calculations were performed using the average glass composition paired
853 with simulated Cpx compositions (so this calculation underestimated the true variability caused
854 by analytical uncertainty). Despite neglecting analytical variability in the glass phase, the span
855 of P-T in measured Cpx-Liq pairs is very similar to our Monte Carlo simulations for Putirka
856 (2008) equation 33 for temperature, iterated with either equation 30 (Fig. 8e) or Neave and
857 Putirka (2017) for pressure (Fig. 8e and Fig. 8d, respectively). It is noticeable on these figures
858 that the machine-learning-based Cpx-only and Cpx-Liq thermobarometers from Petrelli et al.
859 (2020) show significantly less spread in calculated pressures and temperatures both for
860 measured experimental products, and our Monte Carlo simulations (Fig. 8c, 8f). However,
861 using a different experiment, this barometer shows a much larger spread (Supporting Fig. 20),
862 illustrating the unintuitive behaviour of regression tree machine-learning algorithms versus
863 empirical expressions.

864 In the experiments of Krawczynski et al. (2012), Cpx in a single experiment have compositions
865 such that sometimes jadeite is calculated using Na and sometimes using Al. This complicates
866 the relationship between the spread of pressures and the analytical precision because Al is
867 mostly affected by true phase variability (e.g., sector zoning) while Na is mostly affected by
868 analytical precision. Our simulations do not show the same P-T extent as measured values.
869 However, if one imagines shifting the simulation away from the mean, the spread shown by
870 Cpx where jadeite is calculated from Na (yellow dots) is reasonably similar to the simulation,
871 while Cpx with jadeite calculated from Al (red dots) lie well outside the Monte Carlo
872 simulation (Fig. 9).

873 Overall, these comparisons indicate that apparent phase variability resulting from analytical
874 precision alone can yield a wide range of pressures. Although analytical precision seems to
875 dominate the spread of calculated pressures in experiments where jadeite is calculated using

876 Na, measuring sufficient Cpx to sufficiently average true phase variability is vital where jadeite
877 is calculated from Al. When either source of elemental variability is insufficiently averaged by
878 measuring a large number of Cpx, this can lead to large calculated errors for barometers. For
879 x-y plots of calculated versus experimental pressure, the effect of measuring only a single Cpx
880 on the y axis value is easy to visualize from Fig. 8–9.

881 We investigate whether insufficient averaging of analytical and/or true phase variation can
882 explain the notable decline in the discrepancy between predicted and experimental pressure
883 with increasing number of Cpx analyses per experiment (Fig. 4c-d). We use software-reported
884 estimates of analytical precision from an experimental charge of Neave et al. (2019) to produce
885 5000 synthetic Cpx compositions. This simulates what would happen if 5000 discrete analyses
886 were made of entirely homogenous experimental products, with variability in measured oxides
887 resulting from counting statistics. We then calculate the composition obtained from sampling
888 discrete numbers of Cpx (N=1 to N=100), and feed these averaged compositions into various
889 Cpx-only barometers. We calculate the discrepancy between the calculated pressure for this
890 averaged composition, and the pressure calculated from the mean composition of all 5000 Cpx.
891 For each discrete number of averaged Cpx analyses, we overlay the mean discrepancy (black
892 diamond), and the 95% quantile (yellow star) on the data (Fig. 10a-b). We repeat this process
893 700 times using a for loop in Python, meaning we have simulated randomly sampling one Cpx
894 700 different times, two Cpx 700 times, etc. A schematic showing the for loop and subsampling
895 routine is provided in Supporting Fig. 21.

896 Fig. 10a-b demonstrates that when only one EPMA analysis is taken from a homogenous Cpx
897 population, the calculated pressure can differ from the pressure calculated from the true
898 composition by up to ± 3 –4 kbar for Putirka (2008, Fig. 10a) and ± 2.5 –3.5 kbar for Wang et al.
899 (2021, Fig. 10b). The mean absolute discrepancy for N=1 Cpx is ± 0.89 kbar for Putirka (2008)
900 and ± 0.55 kbar for Wang et al. (2021, blue stars), while the 95% quantile is ± 2.09 kbar for

901 Putirka (2008) and ± 1.39 kbar for Wang et al. (2021, yellow stars, Fig. 10a-b). The absolute
902 discrepancy from each experiment, the mean, and 95% quantiles decline very rapidly between
903 $N=1$ and $N=7$ Cpx. When >7 Cpx are averaged, there are very few experiments >1.5 kbar from
904 the mean value calculated from all 5000 Cpx, and very few individual experiments >1 kbar
905 ($\sim 5\%$). However, it must be remembered that we are only simulating analytical uncertainty;
906 analyses will also be affected by true heterogeneity in experimental products.

907 To investigate the combined effect of analytical and experimental variability (i.e., the measured
908 variability in a given experiment), we repeatedly resample the 18 reported Cpx from
909 experiment B1038 of Krawczynski et al. (2012), averaging different numbers of measured Cpx
910 compositions to create a theoretical “average” composition (see supporting Fig. 22 for a
911 schematic of this loop). For $N=1$ Cpx, the mean offsets reach ~ 2 kbar, with 95% quantiles of
912 $\sim 4\text{--}5$ kbar (Fig. 10c-d).

913 We also consider the effect of sampling on Cpx-Liq barometry, using individual measurements
914 of experimental Cpx and glass from experiment B1038 from Krawczynski et al. (2012). First,
915 we match up all possible pairs of measured experimental Cpx and Liq ($N=360$ pairs). Each
916 pass through the loop, we randomly select 10 Liq which have been paired with a given Cpx,
917 and calculate an average Liq composition. We then consider averaging different numbers of
918 Cpx paired with these average Liq compositions (see Supporting Fig. 23 for schematic of this
919 loop). This approach of using average liquid compositions means we are only investigating the
920 effect of how many Cpx analyses are averaged, because there is no clear correlation in between
921 number of Cpx and Liq analyses per experiment in our compiled dataset. Depending on the
922 choice of barometer, mean offsets decline rapidly with increasing N from $\sim \pm 2$ kbar for $N=1$
923 Cpx, with a 95% quantile of $\sim \pm 4\text{--}5$ kbar (Fig. 10e-f).

924 Importantly, the large offsets for N=2–4 Cpx highlight a particular problem that can arise when
925 measuring only a small number of Cpx. Specifically, when a Cpx has Na-Al contents such that
926 the jadeite content is zero, averaging the composition of that Cpx with a second Cpx with Jd>0
927 can result in a very low (but non-zero) jadeite content. While Jd=0 will return a NaN for
928 pressure (not a number), because most Cpx-Liq thermobarometers contain a log term involving
929 the jadeite content (and $\log(0) = \text{NaN}$), the log of a very low but non-zero jadeite content
930 produces a very large negative number, yielding an anomalous pressure (see Supporting
931 Information Fig. 24).

932 Our simulations (Fig. 10) demonstrate that insufficient EPMA analyses to characterize the true
933 composition of experimental phases can account for the observation that experiments
934 characterized by a small number of analyses can show large discrepancies between calculated
935 and experimental pressures (Fig 4c-d). Most simply, imagine that barometers perfectly predict
936 the relationship between composition and pressure. Even in this perfect scenario, measuring a
937 small number of Cpx (\pm Liq) in each experiment can result in the measured composition not
938 being representative of the true composition, resulting in offsets between calculated and
939 experimental pressures. Of course, in some cases, a single measurement will obtain the correct
940 value (i.e. sampling the mean of the normal distribution by fluke), accounting for the fact that
941 some experiments analysing a small number of Cpx show small offsets on Fig 4 and 10.
942 However, the smaller the number of Cpx analyses performed, the more likely it is that the
943 composition of the Cpx obtained differs from the true composition. In reality, barometers are
944 not perfectly accurate, which explains why there are still offsets between calculated and
945 experimental pressure even when larger numbers of Cpx are analysed.

946 The y axis in these subsampling simulations represents the distance a single experiment may
947 lie from the 1:1 line in a plot of experimental pressure versus calculated pressure, simply
948 because of insufficient averaging of analytical and experimental variability. Given that ~43%

949 of experiments we have compiled performed ≤ 5 Cpx analyses per experiment, even if
950 barometers perfectly recreate the relationship between pressure and temperature for the true
951 composition of Cpx and Cpx-Liq pairs from that experiment, we would expect scatter about a
952 1:1 line of $\sim \pm 2\text{--}3$ kbar based on insufficient averaging. In reality, the scatter off the 1:1 line
953 will be larger than this, because insufficient averaging of experimental products affects the
954 calibration of barometers, as well as the assessment of these barometers using test datasets.
955 Thus, we suggest that insufficient averaging of analytical precision (with a contribution from
956 true phase variability) is the ultimate reason barometers have SEE of $\sim 2\text{--}4$ kbar when applied
957 to global test datasets, regardless of the exact calibration strategy.

958 **2.4. Implications for thermobarometry on natural systems**

959 The Monte Carlo simulations of calculated pressures and temperatures resulting from analytical
960 uncertainty alone shown in Figures 7–8 for Cpx and Cpx-Liq are equally applicable to natural
961 systems. Namely, even if erupted Cpx are entirely homogenous and come from a single magma
962 storage region at a single pressure and temperature, calculated pressures may span ~ 4 kbar
963 using Cpx-only thermobarometry, and ~ 6 kbar using Cpx-Liq thermobarometry simply
964 resulting from analytical precision (assuming analytical conditions similar to those used by
965 Neave et al., 2019 and Krawczynski et al., 2012, i.e., 100–150 nA s). When plotted in pressure-
966 temperature space, completely random, normally distributed analytical error produces a
967 strongly correlated pressure-temperature array (Fig. 8–9). It is crucial to recognise that a wide
968 spread of pressures calculated for a natural system using popular Cpx-only and Cpx-Liq
969 barometers should not automatically be interpreted as representing magma storage across a
970 broad region of the crust (i.e., transcrustal magma storage). Rather it should be assumed that
971 this an artifact of analytical uncertainty until proven otherwise. Specifically, Monte Carlo
972 simulations with errors determined from software-calculated analytical uncertainty (rather than
973 secondary standards) should be used to calculate the spread in pressures and temperatures that

974 result from analytical precision. If the spread of pressures and temperatures from natural
975 samples exceeds the simulated spread (and even if it does not), the role of P-T spread resulting
976 from sector zoning and other disequilibrium features should also be investigated (e.g., Hammer
977 et al., 2016). True transcrustal storage should only be invoked after ruling out these null
978 hypotheses.

979 Although the overall structure of pressure and temperature estimates should certainly be
980 considered, averages of many individual measurements help to eliminate analytical and true
981 phase variability. For example, Putirka et al. (1996) showed that the fit between calculated and
982 experimental pressures are substantially improved when all experiments conducted at a given
983 pressure are averaged, rather than considered individually (compare their Fig. 3a and Fig 4a).
984 To avoid averaging out true variations in magma storage conditions, it may be best to perform
985 and average multiple analyses within the core and rim of any given Cpx (e.g., Klügel et al.,
986 2005).

987 **3. SUMMARY AND FUTURE DIRECTIONS**

988 There is broad consensus that improved methods for estimating the pressures of igneous
989 processes and magma storage is vital to advance the field of igneous petrology (Hilley et al.,
990 2022; McGuire et al., 2017). In the preceding sections we have highlighted a number of sources
991 of uncertainty affecting Cpx-only and Cpx-Liq barometry, which are two of the most popular
992 barometry tools. We show that insufficient averaging of measurements made with low
993 analytical precision, combined with heterogeneity in experimental products, and
994 interlaboratory offsets fundamentally limit the precision of Cpx-based barometric estimates to
995 $\pm 2\text{--}4$ kbar for crustal Cpx. Below, we highlight a number of ways in which we can improve
996 how experimental products are analysed and reported to improve the future calibration and
997 application of Cpx-based barometers. These recommendations can be summarized as: 1)

998 collect more counts, 2) measure more phases in each experimental charge, 3) address
999 interlaboratory biases and 4) better data reporting.

1000 **3.1. Can we simply collect more counts for Na in Cpx?**

1001 Our Monte Carlo simulations demonstrate that analytical precision associated with the
1002 measurement of Na₂O resulting from insufficient count times and/or beam currents is a major
1003 source of uncertainty affecting Cpx-only and Cpx-Liq barometry. Poor precision affects the
1004 experimental data used for barometer calibration and testing, as well as calculations of pressure
1005 in natural sample suites. Many papers suggest that they are using such short analysis times and
1006 low beam currents because of fears of Na migration (e.g., Neave et al., 2019). Although beam
1007 migration is a clear concern in hydrous glasses and Na-rich feldspars, to our knowledge there
1008 are no reports of Na migration within crustal pyroxenes.

1009

1010 To investigate the beam sensitivity of pyroxene, we perform tests of Na mobility during
1011 analysis at 15 kV, and 20, 40 and 100 nA using a 1 μm spot on the Oregon State Cameca
1012 SX100. We track changes in peak intensity, and calculated Na concentrations using typical
1013 analytical routines for a Na-rich end-member jadeite (~7.1 wt% Na₂O) and a Kakanui Augite
1014 with lower Na contents (~1.2 wt% Na₂O), as well as a hydrous rhyolite glass for comparison.

1015

1016 Peak intensities in the hydrous rhyolite glass show a rapid decrease with increasing exposure,
1017 even at 20 nA, as is widely documented (e.g., Morgan and London, 2005, Fig. 11a, black
1018 triangles). The jadeite is significantly less beam-sensitive, with peak counts only decreasing by
1019 15-20% after ~ 5 minutes (and only at 40 and 100 nA, Fig. 11a-b). In contrast, we see no
1020 evidence for a decrease in Na peak intensities collected on the Kakanui augite even at 100 nA
1021 over ~6 minutes (Fig. 11b).

1022

1023 We also perform 6 repeated analyses at 100 nA at the same stage position on the jadeite and
1024 Kakanui augite, with a total beam-on time of ~60 s per analysis. Peak count times were Na (3.3
1025 s), Ca (10 s), Al (10 s), Si (10 s) and Ti (10 s). We plot the concentration of each element
1026 quantified from the peak and background against the average beam-on time for each analysis.
1027 For the jadeite, the concentration of Na₂O declines reasonably coherently by ~10% after ~6
1028 minutes of beam exposure (Fig. 11c). For the Kakanui Augite there is no coherent decrease in
1029 Na₂O, with concentrations varying well within the expected threshold given the low count time
1030 (Fig. 11d). Further details of additional tests are provided in the supporting information
1031 (Supporting Figs. S25-30). Overall, we conclude that in pyroxenes with relatively low Na₂O
1032 contents, there is no need to restrict count times or beam currents, as Na migration seems
1033 negligible.

1034

1035 Given the lack of measurable Na migration in natural Cpx, we suggest that the simplest way to
1036 increase precision on reported Na in future experiments and natural samples is a combination
1037 of longer count times and/or a higher beam current during Cpx analyses. Increasing the count
1038 time at a specific beam current is the best strategy if the interaction volume needs to be
1039 minimised (e.g., for analyses of small experimental products), while increasing the beam time
1040 and current together is most efficient if a tiny interaction volume is not critical.

1041

1042 In many analytical routines, increasing the count time on Na will not increase the total
1043 analytical time if other spectrometers are already counting minor elements for longer (e.g., Cr
1044 and Mn on the LIF). Additionally, in a number of studies, Na and Al are both being measured
1045 on the TAP or LTAP (e.g., Hammer et al., 2016; Krawczynski et al., 2012; McCane, 2022;
1046 Wieser et al., 2022a), with Al being counted significantly longer than Na, despite its higher

1047 concentration in Cpx. If total analytical time is an issue, count times can be adjusted on the
1048 TAP/LTAP to count for longer on Na instead.

1049

1050 Although they have great utility for assessing accuracy and drift, secondary standards should
1051 only be used to assess precision if the standard has a very similar elemental concentration to
1052 the analyte. If elements are present in lower concentrations in natural samples, software-
1053 calculated precisions should be used to assess uncertainties resulting from counting statistics.
1054 Overall, we suggest that users optimize their EPMA acquisition parameters to achieve <5%
1055 precision for the range of Na₂O contents found in their samples.

1056 **3.2. Perform more measurements in each experimental charge.**

1057 Analytical noise is particularly problematic for calibration datasets if only a small number of
1058 Cpx in each experiment are measured. The large number of experiments performing <5
1059 individual Cpx measurements, combined with low precision on Na₂O measurements, explains
1060 why no Cpx-based barometers can predict pressure in a global dataset of experiments with
1061 errors smaller than ± 2 –5 kbar (despite many different calibration strategies). Even if analytical
1062 precision is improved by increasing the number of counts, performing multiple Cpx analyses
1063 remains vital to minimize the effects of experimental heterogeneity (e.g., sector zoning, Neave
1064 et al. 2019). This is particularly true if jadeite is calculated from Al, or when using Cpx-Liq
1065 barometers relying on Al content (e.g., eq32c, Putirka, 2008) as Al and other major elements
1066 in Cpx and glass are influenced more by true experimental phase variability than analytical
1067 uncertainty.

1068 **3.3. Quantify and resolve interlaboratory offsets for glasses and silicate minerals**

1069 Even if analytical conditions are optimized to make measurements more precise, and a larger
1070 number of products are measured in each experiment, interlaboratory offsets may still introduce
1071 random, or even systematic offsets into barometry calibrations. To properly resolve the extent

1072 of interlaboratory biases, we suggest that mineral round-robins are conducted on the current
1073 generation of EPMA at institutions which are contributing a significant amount of
1074 experimental data (and ideally between all laboratories). Such round-robins will be vital to
1075 determine whether our assumption that mineral offsets may be as large as offsets measured for
1076 glasses is correct. It may be that by applying such corrections, more precise barometry
1077 calibrations can be obtained.

1078 Longer-term, it is not practical to rely on large round-robins among all possible EPMA labs
1079 that may wish to perform mineral-melt barometry on natural samples, or conduct experiments
1080 used to calibrate these expressions. These interlaboratory biases highlight the need for the
1081 development of a new generation of homogenous reference materials to replace the
1082 heterogeneous NMNH standards. One recent suggestion, put forward in an open letter to the
1083 microanalysis community with over 90 co-signers
1084 (<https://probesoftware.com/smf/index.php?topic=1415.0>), is to create a global reference set of
1085 high purity, stoichiometric end-member synthetic compositional standards in approximately
1086 500 to 1000 gram quantities. This would ensure that every e-beam microanalysis laboratory in
1087 the world could readily obtain sufficient material to last decades of polishing, use, re-polishing
1088 and re-use. Even better, such synthetic standard materials can always be reproduced in the
1089 future as necessary, since they would be selected such that their synthesis would be well-
1090 constrained by both purity and thermodynamics. These efforts are being formalized within the
1091 focused interest group microanalytical standards (FIGMAS) committee of the Microbeam
1092 Analysis Society (<https://the-mas.org/about-us/focused-interest-group-figmas/>).

1093 **3.4. Better reporting of compositional, analytical and metadata**

1094 It may be possible to improve barometry calibrations by excluding experiments with numbers
1095 of analyses below a certain threshold, or experiments where phases were only briefly
1096 characterized to identify phase occurrence, rather than to provide reliable phase compositions.

1097 At a minimum, filtering based on numbers of analyses will require this information to be
1098 copied over from pdf tables in papers and entered into the LEPR dataset. However, to evaluate
1099 analytical noise within a single experiment rigorously, we also need an estimate of the
1100 analytical precision. For example, it may be more rigorous to include an experiment where
1101 only two Cpx were measured but at high precision (e.g., Na = 10%) versus an experiment with
1102 five Cpx with low analytical precision (e.g., Na = 50%, Fig. 6b). However, this information
1103 cannot be obtained for the majority of the LEPR database, particularly as many of the EPMA
1104 instruments used for these measurements have been decommissioned.

1105 Thus, for future experimental work, we suggest that authors report the following information:

- 1106 1. The beam current, voltage, crystal, primary calibration material, and peak and
1107 background count times for every element.
- 1108 2. The software-calculated 1 sigma value for each analysis.
- 1109 3. The elemental data for every spot analysis of every phase in each experimental charge
1110 and every natural mineral analysis, rather than providing a mean composition and
1111 standard deviation.
- 1112 4. Detailed information on how thermal gradients were assessed, and any friction
1113 corrections.

1114 Providing this information will allow future attempts at calibrating barometers to better filter
1115 the underlying data. We also encourage authors to think carefully about the influence of minor
1116 elements such as Cr on Cpx phase stability (Voigt et al., 2017) and to carefully report whether
1117 an attempt was made to measure an element but it was below the detection limit, or whether
1118 no measurement attempt was made. If an element was below detection limit, and an estimate
1119 of the detection limit is provided, the number can be more reliably imputed than when no
1120 information is given.

1121 **3.5. Remeasure existing experimental products**

1122 We recognise that experimental studies are seldom performed with the sole aim of calibrating
1123 thermobarometers. Instead, the authors may have simply wanted to constrain a phase diagram,
1124 so only a small number of Cpx were analysed in each experiment to confirm phase occurrence,
1125 and there was no reason for analytical conditions to be optimized for low concentration
1126 components like Na₂O, or to measure all elements (e.g., Cr₂O₃ in Cpx and glass). Although it
1127 will require a significant community effort to find and share the experimental charges, this is
1128 likely much less effort than redoing experiments from scratch, and it would be worthwhile to
1129 reanalyse a large proportion of the experimental charges compiled in LEPR. This would take
1130 advantage of the higher precision of modern EPMA instruments (compare the precision of the
1131 old and new MIT JEOL on Fig. 6c) and help to fill in the large amount of missing data in LEPR
1132 (Cr₂O₃, H₂O, P₂O₅ etc). Ideally, reanalysis would take place on a single EPMA instrument, or
1133 on a set of instruments where secondary standards are exchanged to correct for interlaboratory
1134 biases.

1135 **5. CONCLUSIONS**

1136 In their current state, Cpx-based barometers struggle to precisely and accurately constrain the
1137 location of magma storage in the crust (with errors of ± 2 – 5 kbar, $\sim \pm 8$ – 11 km). In addition to
1138 fundamental thermodynamic limitations, we suggest that the poor performance of barometers
1139 results from the fact that the true composition of Cpx in experiments has not been precisely
1140 determined, because insufficient analyses were conducted to average out low analytical
1141 precision and true phase variability. Calibrating and testing Cpx-based barometers has been
1142 further hindered by interlaboratory offsets during EPMA analyses, and large amounts of
1143 missing elemental data reported by experimental studies. We suggest that pressures calculated
1144 from individual Cpx analyses in both experiments and natural samples must be evaluated in
1145 the context of the expected range of pressures obtained from the propagation of analytical

1146 errors for that specific study. To invoke true transcrustal storage, it must be demonstrated that
1147 the range of calculated pressures greatly exceeds that expected from analytical precision alone.
1148 We believe a new generation of more precise barometers could be calibrated through a
1149 community effort to obtain an experimental dataset which properly averages analytical
1150 imprecision (for low concentration elements such as Na₂O) and true phase variability (e.g.,
1151 Al₂O₃ concentrations affected by sector zoning). There is growing recognition in the Machine
1152 Learning community that big data are not as important as good data. In a recent interview
1153 (2022), Machine Learning expert Andrew Ng states “In many industries where giant datasets
1154 simply do not exist, I think the focus has to shift from big data to good data.” We suggest that
1155 the same reasoning should be applied to petrological thermobarometers. Ideally, an
1156 independent high quality test dataset would be isolated during model calibration and tuning to
1157 allow robust estimates of the precision that can be expected when these methods are applied to
1158 “unknown” samples. Improving Cpx-based barometers is vital for reliable interpretation of a
1159 volcanic plumbing system geometry (e.g., distinguishing between a single reservoir, discrete
1160 reservoirs, and true transcrustal magmatic systems).

1161 **6. DATA AND CODE AVAILABILITY**

1162 Jupyter Notebooks used to produce figures 2 to 11 in this paper are available on Penny Wieser’s
1163 Github. https://github.com/PennyWieser/BarometersBehavingBadly_Wieser2022. Upon final
1164 acceptance version of this will be archived on Zenodo.

- 1165 • Supporting_Data_1.xlsx –Contains Interlaboratory offsets from Gale et al. (2013), the
1166 Cpx calibration dataset used by Keith Putirka which we use to examine the prevalence of
1167 missing data, and the compilation of Cpx-Liq experiments used in this study (i.e. ArcPL).
- 1168 • Supporting_Data_2.xlsx – Analysis of individual phases from the experiments of
1169 Krawczynski et al. (2012).

- 1170 • Supporting_Data_3.xlsx – Estimates of analytical precision from a subset of
1171 experimental analyses by Krawczynski et al. (2012),
- 1172 • Supporting_Data_4.xlsx –All Cpx-bearing experiments from the LEPR (downloaded
1173 in 2021).
- 1174 • Supporting_Data_5.xlsx – Analysis of individual phases from the experiments of
1175 Neave et al. (2019).
- 1176 • Supporting_Data_6.xlsx – Analysis of individual phases from the experiments of
1177 Erdmann et al. (2016), from their supporting information
- 1178 • Supporting_Data_7.xlsx – Analysis of individual phases from the experiments of
1179 Melekhova et al. (2015).
- 1180 • Supporting_Data_8.xlsx – Analysis of individual phases from the experiments of
1181 Waters et al. (2021).
- 1182 • Supporting_Data_9.xlsx – Analysis of individual phases from the experiments of
1183 Blatter et al. (2013)
- 1184 • Supporting_Data_9.xlsx – Analysis of individual phases from the experiments of
1185 Blatter et al. (2013)
- 1186 • Supporting_Data_10.xlsx – Investigation of Na migration on the OSU SX100

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1202 REFERENCES

- 1203 Akella, J., Kennedy, G.C., 1971. Melting of gold, silver, and copper-proposal for a new high-pressure
1204 calibration scale. *J. Geophys. Res.* 76, 4969–4977. <https://doi.org/10.1029/JB076i020p04969>
- 1205 Almeev, R.R., Holtz, F., Ariskin, A.A., Kimura, J.-I., 2013. Storage conditions of Bezymianny Volcano
1206 parental magmas: results of phase equilibria experiments at 100 and 700 MPa. *Contrib
1207 Mineral Petrol* 166, 1389–1414. <https://doi.org/10.1007/s00410-013-0934-x>
- 1208 Alonso-Perez, R., Müntener, O., Ulmer, P., 2009. Igneous garnet and amphibole fractionation in the
1209 roots of island arcs: experimental constraints on andesitic liquids. *Contrib Mineral Petrol*
1210 157, 541–558. <https://doi.org/10.1007/s00410-008-0351-8>
- 1211 Andújar, J., Scaillet, B., Pichavant, M., Druitt, T.H., 2015. Differentiation Conditions of a Basaltic
1212 Magma from Santorini, and its Bearing on the Production of Andesite in Arc Settings. *Journal
1213 of Petrology* 56, 765–794. <https://doi.org/10.1093/petrology/egv016>
- 1214 Bacon, C.R., Carmichael, I.S.E., 1973. Stages in the P-T path of ascending basalt magma: An example
1215 from San Quintin, Baja California. *Contr. Mineral. and Petrol.* 41, 1–22.
1216 <https://doi.org/10.1007/BF00377648>
- 1217 Baker, D.R., Eggler, D.H., 1987. Compositions of anhydrous and hydrous melts coexisting with
1218 plagioclase, augite, and olivine or low-Ca pyroxene from 1 atm to 8 kbar: Application to the
1219 Aleutian volcanic center of Atka. *American Mineralogist* 72.
- 1220 Bartels, K.S., Kinzler, R.J., Grove, T.L., 1991. High pressure phase relations of primitive high-alumina
1221 basalts from Medicine Lake volcano, northern California. *Contr. Mineral. and Petrol.* 108,
1222 253–270. <https://doi.org/10.1007/BF00285935>
- 1223 Behrens, H., Meyer, M., Holtz, F., Benne, D., Nowak, M., 2001. The effect of alkali ionic radius,
1224 temperature, and pressure on the solubility of water in MAISI3O8 melts (M=Li, Na, K, Rb).
1225 *Chemical Geology* 174, 275–289. [https://doi.org/10.1016/S0009-2541\(00\)00320-X](https://doi.org/10.1016/S0009-2541(00)00320-X)
- 1226 Bell, S., 2001. Good Practice Guide No. 11, The Beginners guide to the uncertainty of measurement.
1227 National Physics Laboratory.
- 1228 Berndt, J., 2004. An Experimental Investigation of the Influence of Water and Oxygen Fugacity on
1229 Differentiation of MORB at 200 MPa. *Journal of Petrology* 46, 135–167.
1230 <https://doi.org/10.1093/petrology/egh066>
- 1231 Blatter, D.L., Sisson, T.W., Hanks, W.B., 2017. Voluminous arc dacites as amphibole reaction-
1232 boundary liquids. *Contrib Mineral Petrol* 172, 27. [https://doi.org/10.1007/s00410-017-1340-](https://doi.org/10.1007/s00410-017-1340-6)
1233 6
- 1234 Blatter, D.L., Sisson, T.W., Hanks, W.B., 2013. Crystallization of oxidized, moderately hydrous arc
1235 basalt at mid- to lower-crustal pressures: implications for andesite genesis. *Contrib Mineral
1236 Petrol* 166, 861–886. <https://doi.org/10.1007/s00410-013-0920-3>

1237 Bogaerts, M., Scaillet, B., Auwera, J.V., 2006. Phase Equilibria of the Lyngdal Granodiorite (Norway):
1238 Implications for the Origin of Metaluminous Ferroan Granitoids. *Journal of Petrology* 47,
1239 2405–2431. <https://doi.org/10.1093/petrology/egl049>
1240 Bohlen, S.R., Boettcher, A.L., 1982. The quartz \rightleftharpoons coesite transformation: A precise determination
1241 and the effects of other components. *J. Geophys. Res.* 87, 7073–7078.
1242 <https://doi.org/10.1029/JB087iB08p07073>
1243 Bohlen, S.R., Essene, E.J., Boettcher, A.L., 1980. Reinvestigation and application of olivine-quartz-
1244 orthopyroxene barometry. *Earth and Planetary Science Letters* 47, 1–10.
1245 [https://doi.org/10.1016/0012-821X\(80\)90098-9](https://doi.org/10.1016/0012-821X(80)90098-9)
1246 Botcharnikov, R.E., Koepke, J., Holtz, F., McCammon, C., Wilke, M., 2005. The effect of water activity
1247 on the oxidation and structural state of Fe in a ferro-basaltic melt. *Geochimica et*
1248 *Cosmochimica Acta* 69, 5071–5085. <https://doi.org/10.1016/j.gca.2005.04.023>
1249 Brown, W.L., Parsons, I., 1981. Towards a more practical two-feldspar geothermometer. *Contr.*
1250 *Mineral. and Petrol.* 76, 369–377. <https://doi.org/10.1007/BF00371478>
1251 Brugman, K., Phillips, M.G., Till, C.B., 2021. Experimental Determination of Mantle Solidi and Melt
1252 Compositions for Two Likely Rocky Exoplanet Compositions. *JGR Planets* 126.
1253 <https://doi.org/10.1029/2020JE006731>
1254 Cadoux, A., Scaillet, B., Druitt, T.H., Deloule, E., 2014. Magma Storage Conditions of Large Plinian
1255 Eruptions of Santorini Volcano (Greece). *Journal of Petrology* 55, 1129–1171.
1256 <https://doi.org/10.1093/petrology/egu021>
1257 Condamine, P., Tournier, S., Charlier, B., Médard, E., Triantafyllou, A., Dalou, C., Tissandier, L.,
1258 Lequin, D., Cartier, C., Füre, E., Burnard, P.G., Demouchy, S., Marrocchi, Y., 2022. Influence of
1259 intensive parameters and assemblies on friction evolution during piston-cylinder
1260 experiments. *American Mineralogist* 107, 1575–1581. <https://doi.org/10.2138/am-2022-7958>
1261
1262 Costa, F., 2004. Petrological and Experimental Constraints on the Pre-eruption Conditions of
1263 Holocene Dacite from Volcan San Pedro (36 S, Chilean Andes) and the Importance of Sulphur
1264 in Silicic Subduction-related Magmas. *Journal of Petrology* 45, 855–881.
1265 <https://doi.org/10.1093/petrology/egg114>
1266 Di Carlo, I., 2006. Experimental Crystallization of a High-K Arc Basalt: the Golden Pumice, Stromboli
1267 Volcano (Italy). *Journal of Petrology* 47, 1317–1343.
1268 <https://doi.org/10.1093/petrology/egl011>
1269 Draper, D.S., Johnston, A.D., 1992. Anhydrous PT phase relations of an Aleutian high-MgO basalt: an
1270 investigation of the role of olivine-liquid reaction in the generation of arc high-alumina
1271 basalts. *Contr. Mineral. and Petrol.* 112, 501–519. <https://doi.org/10.1007/BF00310781>
1272 Emmanuel, T., Maupong, T., Mpoeleng, D., Semong, T., Mphago, B., Tabona, O., 2021. A survey on
1273 missing data in machine learning. *J Big Data* 8, 140. <https://doi.org/10.1186/s40537-021-00516-9>
1274
1275 Erdmann, M., Koepke, J., 2016. Silica-rich lavas in the oceanic crust: experimental evidence for
1276 fractional crystallization under low water activity. *Contrib Mineral Petrol* 171, 83.
1277 <https://doi.org/10.1007/s00410-016-1294-0>
1278 Erdmann, S., Martel, C., Pichavant, M., Bourdier, J.-L., Champallier, R., Komorowski, J.-C., Cholik, N.,
1279 2016. Constraints from Phase Equilibrium Experiments on Pre-eruptive Storage Conditions in
1280 Mixed Magma Systems: a Case Study on Crystal-rich Basaltic Andesites from Mount Merapi,
1281 Indonesia. *J. Petrology* 57, 535–560. <https://doi.org/10.1093/petrology/egw019>
1282 Feig, S.T., Koepke, J., Snow, J.E., 2010. Effect of oxygen fugacity and water on phase equilibria of a
1283 hydrous tholeiitic basalt. *Contrib Mineral Petrol* 160, 551–568.
1284 <https://doi.org/10.1007/s00410-010-0493-3>
1285 Firth, C., Adam, J., Turner, S., Rushmer, T., Brens, R., Green, T.H., Erdmann, S., O'Neill, H., 2019.
1286 Experimental constraints on the differentiation of low-alkali magmas beneath the Tonga arc:

1287 Implications for the origin of arc tholeiites. *Lithos* 344–345, 440–451.
1288 <https://doi.org/10.1016/j.lithos.2019.07.008>

1289 Fournelle, J., 2012. Complications with Using Natural Minerals as Microbeam Standards: Pyroxenes.
1290 Presented at the American Geophysical Union, Fall Meeting 2012, abstract id. V23C-2827.

1291 Gaetani, G.A., Grove, T.L., 1998. The influence of water on melting of mantle peridotite.
1292 *Contributions to Mineralogy and Petrology* 131, 323–346.
1293 <https://doi.org/10.1007/s004100050396>

1294 Gale, A., Dalton, C.A., Langmuir, C.H., Su, Y., Schilling, J.-G., 2013. The mean composition of ocean
1295 ridge basalts: MEAN MORB. *Geochem. Geophys. Geosyst.* 14, 489–518.
1296 <https://doi.org/10.1029/2012GC004334>

1297 Gleeson, M.L.M., Gibson, S.A., Stock, M.J., 2021. Upper Mantle Mush Zones beneath Low Melt Flux
1298 Ocean Island Volcanoes: Insights from Isla Floreana, Galápagos. *Journal of Petrology* 61,
1299 egaa094. <https://doi.org/10.1093/petrology/egaa094>

1300 Grove, T.L., Elkins-Tanton, L.T., Parman, S.W., Chatterjee, N., Montener, O., Gaetani, G.A., 2003.
1301 Fractional crystallization and mantle-melting controls on calc-alkaline differentiation trends.
1302 *Contributions to Mineralogy and Petrology* 145, 515–533. [https://doi.org/10.1007/s00410-](https://doi.org/10.1007/s00410-003-0448-z)
1303 [003-0448-z](https://doi.org/10.1007/s00410-003-0448-z)

1304 Hamada, M., Fujii, T., 2008. Experimental constraints on the effects of pressure and H₂O on the
1305 fractional crystallization of high-Mg island arc basalt. *Contrib Mineral Petrol* 155, 767–790.
1306 <https://doi.org/10.1007/s00410-007-0269-6>

1307 Hammer, J., Jacob, S., Welsch, B., Hellebrand, E., Sinton, J., 2016. Clinopyroxene in postshield
1308 Haleakala ankaramite: 1. Efficacy of thermobarometry. *Contrib Mineral Petrol* 171, 7.
1309 <https://doi.org/10.1007/s00410-015-1212-x>

1310 Harlow, G.E., 1997. K in clinopyroxene at high pressure and temperature; an experimental study.
1311 *American Mineralogist* 82, 259–269. <https://doi.org/10.2138/am-1997-3-403>

1312 Hart, S.R., Dunn, T., 1993. Experimental cpx/melt partitioning of 24 trace elements. *Contr. Mineral.*
1313 *and Petrol.* 113, 1–8. <https://doi.org/10.1007/BF00320827>

1314 Hays, J., 1966. Lime-alumina-silica. *Carnegie Inst Washington Yearbook* 65.

1315 Higgins, O., Sheldrake, T., Caricchi, L., 2022. Machine learning thermobarometry and chemometry
1316 using amphibole and clinopyroxene: a window into the roots of an arc volcano (Mount
1317 Liamuiga, Saint Kitts). *Contrib Mineral Petrol* 177, 10. [https://doi.org/10.1007/s00410-021-](https://doi.org/10.1007/s00410-021-01874-6)
1318 [01874-6](https://doi.org/10.1007/s00410-021-01874-6)

1319 Hilley, G., 2022. SZ4D Implementation Plan. Stanford Digital Repository.
1320 <https://doi.org/10.25740/HY589FC7561>

1321 Hughes, E.C., Buse, B., Kearns, S.L., Blundy, J.D., Kilgour, G., Mader, H.M., 2019. Low analytical totals
1322 in EPMA of hydrous silicate glass due to sub-surface charging: Obtaining accurate volatiles by
1323 difference. *Chemical Geology* 505, 48–56. <https://doi.org/10.1016/j.chemgeo.2018.11.015>

1324 Hunt, J., Clift, P., Lacasse, C., Vallier, T., Werner, R., 1998. INTERLABORATORY COMPARISON OF
1325 ELECTRON PROBE MICROANALYSIS OF GLASS GEOCHEMISTRY. *Proceedings of the Ocean*
1326 *Drilling Program, Scientific Results, Vol. 152.*

1327 Husen, A., Almeev, R.R., Holtz, F., 2016. The Effect of H₂O and Pressure on Multiple Saturation and
1328 Liquid Lines of Descent in Basalt from the Shatsky Rise. *Journal of Petrology* 57, 309–344.
1329 <https://doi.org/10.1093/petrology/egw008>

1330 Jarosewich, E., 2002. Smithsonian Microbeam Standards. *J. Res. Natl. Inst. Stand. Technol.* 107, 681.
1331 <https://doi.org/10.6028/jres.107.054>

1332 Jarosewich, E., Nelen, J.A., Norberg, J.A., 1980. Reference Samples for Electron Microprobe
1333 Analysis*. *Geostandards and Geoanalytical Research* 4, 43–47.
1334 <https://doi.org/10.1111/j.1751-908X.1980.tb00273.x>

1335 Johannes, W., Bell, P.M., Mao, H.K., Boettcher, A.L., Chipman, D.W., Hays, J.F., Newton, R.C., Seifert,
1336 F., 1971. An interlaboratory comparison of piston-cylinder pressure calibration using the

1337 albite-breakdown reaction. *Contr. Mineral. and Petrol.* 32, 24–38.
1338 <https://doi.org/10.1007/BF00372231>

1339 Jorgenson, C., Higgins, O., Petrelli, M., Bégué, F., Caricchi, L., 2022. A Machine Learning-Based
1340 Approach to Clinopyroxene Thermobarometry: Model Optimization and Distribution for Use
1341 in Earth Sciences. *JGR Solid Earth* 127. <https://doi.org/10.1029/2021JB022904>

1342 Klügel, A., Hansteen, T.H., Galipp, K., 2005. Magma storage and underplating beneath Cumbre Vieja
1343 volcano, La Palma (Canary Islands). *Earth and Planetary Science Letters* 236, 211–226.
1344 <https://doi.org/10.1016/j.epsl.2005.04.006>

1345 Koepke, J., Botcharnikov, R.E., Natland, J.H., 2018. Crystallization of late-stage MORB under varying
1346 water activities and redox conditions: Implications for the formation of highly evolved lavas
1347 and oxide gabbro in the ocean crust. *Lithos* 323, 58–77.
1348 <https://doi.org/10.1016/j.lithos.2018.10.001>

1349 Krawczynski, M.J., Grove, T.L., Behrens, H., 2012. Amphibole stability in primitive arc magmas:
1350 effects of temperature, H₂O content, and oxygen fugacity. *Contrib Mineral Petrol* 164, 317–
1351 339. <https://doi.org/10.1007/s00410-012-0740-x>

1352 Kuehn, S.C., Froese, D.G., Shane, P.A.R., 2011. The INTAV intercomparison of electron-beam
1353 microanalysis of glass by tephrochronology laboratories: Results and recommendations.
1354 *Quaternary International* 246, 19–47. <https://doi.org/10.1016/j.quaint.2011.08.022>

1355 Leahy, G.M., Collins, J.A., Wolfe, C.J., Laske, G., Solomon, S.C., 2010. Underplating of the Hawaiian
1356 Swell: evidence from teleseismic receiver functions: Underplating of the Hawaiian Swell.
1357 *Geophysical Journal International* 183, 313–329. <https://doi.org/10.1111/j.1365-246X.2010.04720.x>

1359 Lerner, A.H., Wallace, P., Shea, T., 2021. The petrologic and degassing behavior of sulfur and other
1360 magmatic volatiles from the 2018 eruption of Kilauea, Hawai'i: melt concentrations, magma
1361 storage depths, and magma recycling. *Bulletin Volcanology* 83:43, 1–32.

1362 Lindsley, D.H., Andersen, D.J., 1983. A two-pyroxene thermometer. *J. Geophys. Res.* 88, A887.
1363 <https://doi.org/10.1029/JB088iS02p0A887>

1364 Llovet, X., Galan, G., 2003. Correction of secondary X-ray fluorescence near grain boundaries in
1365 electron microprobe analysis: Application to thermobarometry of spinel lherzolites.
1366 *American Mineralogist* 88, 121–130. <https://doi.org/10.2138/am-2003-0115>

1367 Mandler, B.E., Donnelly-Nolan, J.M., Grove, T.L., 2014. Straddling the tholeiitic/calc-alkaline
1368 transition: the effects of modest amounts of water on magmatic differentiation at Newberry
1369 Volcano, Oregon. *Contrib Mineral Petrol* 168, 1066. <https://doi.org/10.1007/s00410-014-1066-7>

1370

1371 McCane, J., 2022. VARIATION IN CLINOPYROXENE TEXTURE, COMPOSITION, AND CRYSTALLIZATION
1372 DEPTH OF LATE CRETACEOUS TO EARLY EOCENE LAMPROPHYRIC ROCKS FROM ALKALINE
1373 CALC-ALKALINE MAGMATIC COMPLEXES OF MONTANA, USA. Masters of Science,
1374 Department of Geosciences, Colorado State University.

1375 McGuire, J., Plank, T., Barrientos, S., et al, 2017. The SZ4D initiative: Understanding the processes
1376 that underlie Subduction zone hazards in 4D. A vision document submitted to the National
1377 Science Foundation.

1378 Melekhova, E., Blundy, J., Robertson, R., Humphreys, M.C.S., 2015. Experimental Evidence for
1379 Polybaric Differentiation of Primitive Arc Basalt beneath St. Vincent, Lesser Antilles. *Journal*
1380 *of Petrology* 56, 161–192. <https://doi.org/10.1093/petrology/egu074>

1381 Mercer, C.N., Johnston, A.D., 2008. Experimental studies of the P–T–H₂O near-liquidus phase
1382 relations of basaltic andesite from North Sister Volcano, High Oregon Cascades: constraints
1383 on lower-crustal mineral assemblages. *Contrib Mineral Petrol* 155, 571–592.
1384 <https://doi.org/10.1007/s00410-007-0259-8>

1385 Morgan, G.B., London, D., 2005. Effect of current density on the electron microprobe analysis of
1386 alkali aluminosilicate glasses. *American Mineralogist* 90, 1131–1138.
1387 <https://doi.org/10.2138/am.2005.1769>

1388 Mutch, E.J.F., Blundy, J.D., Tattitch, B.C., Cooper, F.J., Brooker, R.A., 2016. An experimental study of
1389 amphibole stability in low-pressure granitic magmas and a revised Al-in-hornblende
1390 geobarometer. *Contrib Mineral Petrol* 171, 85. <https://doi.org/10.1007/s00410-016-1298-9>
1391 Nandedkar, R.H., Ulmer, P., Müntener, O., 2014. Fractional crystallization of primitive, hydrous arc
1392 magmas: an experimental study at 0.7 GPa. *Contrib Mineral Petrol* 167, 1015.
1393 <https://doi.org/10.1007/s00410-014-1015-5>
1394 Neave, D.A., Bali, E., Guðfinnsson, G.H., Halldórsson, S.A., Kahl, M., Schmidt, A.-S., Holtz, F., 2019.
1395 Clinopyroxene–Liquid Equilibria and Geothermobarometry in Natural and Experimental
1396 Tholeiites: the 2014–2015 Holuhraun Eruption, Iceland. *Journal of Petrology* 60, 1653–1680.
1397 <https://doi.org/10.1093/petrology/egz042>
1398 Neave, D.A., Putirka, K.D., 2017. A new clinopyroxene-liquid barometer, and implications for magma
1399 storage pressures under Icelandic rift zones. *American Mineralogist* 102, 777–794.
1400 <https://doi.org/10.2138/am-2017-5968>
1401 Parat, F., Streck, M., Holtz, F., Almeev, R.R., 2014. Experimental study into the petrogenesis of
1402 crystal-rich basaltic to andesitic magmas at Arenal volcano. *Contributions to Mineralogy and*
1403 *Petrology*.
1404 Parman, S.W., Grove, T.L., Kelley, K.A., Plank, T., 2011. Along-Arc Variations in the Pre-Eruptive H₂O
1405 Contents of Mariana Arc Magmas Inferred from Fractionation Paths. *Journal of Petrology* 52,
1406 257–278. <https://doi.org/10.1093/petrology/egq079>
1407 Petrelli, M., Caricchi, L., Perugini, D., 2020. Machine Learning Thermo-Barometry: Application to
1408 Clinopyroxene-Bearing Magmas. *J. Geophys. Res. Solid Earth* 125.
1409 <https://doi.org/10.1029/2020JB020130>
1410 Pichavant, M., 2002. Physical conditions, structure, and dynamics of a zoned magma chamber:
1411 Mount Pelée (Martinique, Lesser Antilles Arc). *J. Geophys. Res.* 107, 2093.
1412 <https://doi.org/10.1029/2001JB000315>
1413 Pichavant, M., Macdonald, R., 2007. Crystallization of primitive basaltic magmas at crustal pressures
1414 and genesis of the calc-alkaline igneous suite: experimental evidence from St Vincent, Lesser
1415 Antilles arc. *Contrib Mineral Petrol* 154, 535–558. [https://doi.org/10.1007/s00410-007-](https://doi.org/10.1007/s00410-007-0208-6)
1416 [0208-6](https://doi.org/10.1007/s00410-007-0208-6)
1417 Powell, R., Holland, T.J.B., 2008. On thermobarometry. *J Metamorph Geol* 26, 155–179.
1418 <https://doi.org/10.1111/j.1525-1314.2007.00756.x>
1419 Profeta, L., Ducea, M.N., Chapman, J.B., Paterson, S.R., Gonzales, S.M.H., Kirsch, M., Petrescu, L.,
1420 DeCelles, P.G., 2016. Quantifying crustal thickness over time in magmatic arcs. *Sci Rep* 5,
1421 17786. <https://doi.org/10.1038/srep17786>
1422 Putirka, K., 1999. Clinopyroxene + liquid equilibria to 100 kbar and 2450 K. *Contributions to*
1423 *Mineralogy and Petrology* 135, 151–163. <https://doi.org/10.1007/s004100050503>
1424 Putirka, K., 1997. Magma transport at Hawaii: Inferences based on igneous thermobarometry.
1425 *Geology* 25, 69. [https://doi.org/10.1130/0091-7613\(1997\)025<0069:MTAHIB>2.3.CO;2](https://doi.org/10.1130/0091-7613(1997)025<0069:MTAHIB>2.3.CO;2)
1426 Putirka, K., Johnson, M., Kinzler, R., Longhi, J., Walker, D., 1996. Thermobarometry of mafic igneous
1427 rocks based on clinopyroxene-liquid equilibria, 0-30 kbar. *Contributions to Mineralogy and*
1428 *Petrology* 123, 92–108. <https://doi.org/10.1007/s004100050145>
1429 Putirka, K.D., 2008. Thermometers and Barometers for Volcanic Systems. *Reviews in Mineralogy and*
1430 *Geochemistry* 69, 61–120. <https://doi.org/10.2138/rmg.2008.69.3>
1431 Pyle, D., 1999. *Data Preparation for Data Mining*. Morgan Kaufman, San Francisco.
1432 Rader, E.L., Larsen, J.F., 2013. Experimental phase relations of a low MgO Aleutian basaltic andesite
1433 at XH₂O = 0.7–1. *Contrib Mineral Petrol* 166, 1593–1611. [https://doi.org/10.1007/s00410-](https://doi.org/10.1007/s00410-013-0944-8)
1434 [013-0944-8](https://doi.org/10.1007/s00410-013-0944-8)
1435 Riker, J.M., Blundy, J.D., Rust, A.C., Botcharnikov, R.E., Humphreys, M.C.S., 2015. Experimental phase
1436 equilibria of a Mount St. Helens rhyodacite: a framework for interpreting crystallization
1437 paths in degassing silicic magmas. *Contrib Mineral Petrol* 170, 6.
1438 <https://doi.org/10.1007/s00410-015-1160-5>

1439 Rose, T.R., Sorenson, S., Post, J., 2008. The impurities in the Rockport fayalite microbeam standard:
1440 How bad are they? Presented at the American Geophysical Union, Fall Meeting 2009,
1441 abstract id. V31E-2008.

1442 Rubin, D.B., 1976. Inference and missing data. *Biometrika* 63, 581–592.
1443 <https://doi.org/10.1093/biomet/63.3.581>

1444 Rutherford, M.J., Sigurdsson, H., Carey, S., Davis, A., 1985. The May 18, 1980, eruption of Mount St.
1445 Helens: 1. Melt composition and experimental phase equilibria. *J. Geophys. Res.* 90, 2929.
1446 <https://doi.org/10.1029/JB090iB04p02929>

1447 Sisson, T.W., Grove, T.L., 1993. Temperatures and H₂O contents of low-MgO high-alumina basalts.
1448 *Contr. Mineral. and Petrol.* 113, 167–184. <https://doi.org/10.1007/BF00283226>

1449 Sisson, T.W., Ratajeski, K., Hankins, W.B., Glazner, A.F., 2005. Voluminous granitic magmas from
1450 common basaltic sources. *Contrib Mineral Petrol* 148, 635–661.
1451 <https://doi.org/10.1007/s00410-004-0632-9>

1452 Tamayama, M., Eyring, H., 1967. Study of Pressure Calibration and Pressure Distribution in a Piston-
1453 Cylinder High Pressure Press. *Review of Scientific Instruments* 38, 1009–1018.
1454 <https://doi.org/10.1063/1.1720958>

1455 Twala, B., 2009. AN EMPIRICAL COMPARISON OF TECHNIQUES FOR HANDLING INCOMPLETE DATA
1456 USING DECISION TREES. *Applied Artificial Intelligence* 23, 373–405.
1457 <https://doi.org/10.1080/08839510902872223>

1458 Ubide, T., Mollo, S., Zhao, J., Nazzari, M., Scarlato, P., 2019. Sector-zoned clinopyroxene as a
1459 recorder of magma history, eruption triggers, and ascent rates. *Geochimica et Cosmochimica*
1460 *Acta* 251, 265–283. <https://doi.org/10.1016/j.gca.2019.02.021>

1461 Ulmer, P., Kaegi, R., Müntener, O., 2018. Experimentally Derived Intermediate to Silica-rich Arc
1462 Magmas by Fractional and Equilibrium Crystallization at 1.0 GPa: an Evaluation of Phase
1463 Relationships, Compositions, Liquid Lines of Descent and Oxygen Fugacity. *Journal of*
1464 *Petrology* 59, 11–58. <https://doi.org/10.1093/petrology/egy017>

1465 Villiger, S., 2004. The Liquid Line of Descent of Anhydrous, Mantle-Derived, Tholeiitic Liquids by
1466 Fractional and Equilibrium Crystallization--an Experimental Study at 1{middle dot}0 GPa.
1467 *Journal of Petrology* 45, 2369–2388. <https://doi.org/10.1093/petrology/egh042>

1468 Voigt, M., Coogan, L.A., von der Handt, A., 2017. Experimental investigation of the stability of
1469 clinopyroxene in mid-ocean ridge basalts: The role of Cr and Ca/Al. *Lithos* 274–275, 240–253.
1470 <https://doi.org/10.1016/j.lithos.2017.01.003>

1471 Wang, X., Hou, T., Wang, M., Zhang, C., Zhang, Z., Pan, R., Marxer, F., Zhang, H., 2021. A new
1472 clinopyroxene thermobarometer for mafic to intermediate magmatic systems. *Eur. J.*
1473 *Mineral.* 33, 621–637. <https://doi.org/10.5194/ejm-33-621-2021>

1474 Waters, L.E., Cottrell, E., Coombs, M.L., Kelley, K.A., 2021. Generation of Calc-Alkaline Magmas
1475 during Crystallization at High Oxygen Fugacity: An Experimental and Petrologic Study of
1476 Tephra from Buldir Volcano, Western Aleutian Arc, Alaska, USA. *Journal of Petrology* 62,
1477 ega104. <https://doi.org/10.1093/petrology/egaa104>

1478 Weill, D., Rice, J., Shaffer, M., Donovan, J.J., 2013. Electron Beam MicroAnalysis Theory and
1479 Application.

1480 White, R.S., McKenzie, D., O’Nions, R.K., 1992. Oceanic crustal thickness from seismic measurements
1481 and rare earth element inversions. *J. Geophys. Res.* 97, 19683.
1482 <https://doi.org/10.1029/92JB01749>

1483 Wieser, P., Petrelli, M., Lubbers, J., Wieser, E., Ozaydin, S., Kent, A., Till, C., 2022. Thermobar: An
1484 open-source Python3 tool for thermobarometry and hygrometry. *Volcanica* 5, 349–384.
1485 <https://doi.org/10.30909/vol.05.02.349384>

1486 Wieser, Penny E., Edmonds, M., Gansecki, C., Maclennan, J., Jenner, F.E., Kunz, B., Antoshechkina, P.,
1487 Trusdell, F., Lee, R.L., Eimf, 2022. Explosive Activity on Kilauea’s Lower East Rift Zone Fueled
1488 by a Volatile-Rich, Dacitic Melt. *Geochem Geophys Geosyst* 23.
1489 <https://doi.org/10.1029/2021GC010046>

1490 Wieser, P. E., Iacovino, K., Matthews, S., Moore, G., Allison, C.M., 2022. VESlcal: 2. A Critical
1491 Approach to Volatile Solubility Modeling Using an Open-Source Python3 Engine. Earth and
1492 Space Science 9. <https://doi.org/10.1029/2021EA001932>
1493 Wieser, P.E., Lamadrid, H., Maclennan, J., Edmonds, M., Matthews, S., Iacovino, K., Jenner, F.E.,
1494 Gansecki, C., Trusdell, F., Lee, R.L., Ilyinskaya, E., 2021. Reconstructing Magma Storage
1495 Depths for the 2018 Kīlauean Eruption From Melt Inclusion CO₂ Contents: The Importance
1496 of Vapor Bubbles. Geochem Geophys Geosyst 22. <https://doi.org/10.1029/2020GC009364>
1497 Williams, M., Schoneveld, L., Mao, Y., Klump, J., Gosses, J., Dalton, H., Bath, A., Barnes, S., 2020.
1498 pyrolite: Python for geochemistry. JOSS 5, 2314. <https://doi.org/10.21105/joss.02314>
1499 Yang, H.-J., Kinzler, R.J., Grove, T.L., 1996. Experiments and models of anhydrous, basaltic olivine-
1500 plagioclase-augite saturated melts from 0.001 to 10 kbar. Contributions to Mineralogy and
1501 Petrology 124, 1–18. <https://doi.org/10.1007/s004100050169>
1502

1503 **Figure Captions**

1504 Figure 1 – Schematic diagram showing the three main categories of uncertainty associated with
1505 thermobarometry.

1506 Figure 2 – Assessing the influence of interlaboratory biases on Cpx-only and Cpx-Liq
1507 thermobarometry. a-b) Interlaboratory correction factors from Gale et al. (2013) relative to
1508 Lamont (plotting at 1, 1). c-d) Calculated Cpx-only and Cpx-Liq pressures and temperatures
1509 for the average reported composition from Experiment 41c-106 of Krawczynski et al. (2012,
1510 4.9 kbar, 1248.15 K), corrected as if Cpx and Liq compositions were measured in different
1511 laboratories. e-f) as for c-d, using Experiment B1038 (8 kbar, 1323.15 K). Additional
1512 experiments are shown in Supplementary Figs. 2–4. Commonly stated 1 σ errors for each
1513 thermobarometer are shown for comparison. The SEE for eq32b is the fit to N=1173 data,
1514 eq32d is the fit to N=910 anhydrous experiments, and eq33 is the fit to N=1174 data (all from
1515 figures in Putirka, 2008). The SEE for Neave and Putirka (2017) is that given in their abstract.

1516 Fig. 3 – For each of the N=852 experiments in ArcPL, we calculate pressure (a-b) and
1517 temperature (c-d) using the measured Cr content (shown on the x axis), and using Cr=0 wt%.
1518 The difference between the calculation using measured Cr and Cr=0 wt% is the y coordinate.
1519 Different colors and symbols represent different thermobarometry combinations used to

1520 calculated these pressures and temperatures. Commonly stated 1σ errors for these
1521 thermobarometer are shown for comparison. In addition to those described in the caption of
1522 Fig. 2 from Putirka (2008), eq32c is the fit to a global dataset, the SEE from Petrelli et al.
1523 (2020) are the fit to a test dataset, and the error from Putirka (1996) eqT1 is that stated in the
1524 conclusion of that paper.

1525 Fig. 4. Number of discrete Cpx analyses performed in each experimental charges plotted
1526 against experimental pressure (a), and as a histogram with a bin width of 1 (b). (c-d) For each
1527 experimental run, we calculate the absolute difference between the pressure calculated using
1528 the Cpx-only barometers of Wang et al. (2021) and Putirka (2008) eq32d-32b and the
1529 experimental pressure. In general, experimental runs with a smaller number of Cpx analyses
1530 show the largest absolute discrepancies. Symbols are semi-transparent, so darker colors
1531 represent a tighter clustering of data. The blue stars show the mean value for each discrete
1532 number of Cpx in c and d, and for each pressure bin (0 ± 0.5 kbar, 1 ± 0.5 kbar etc.). The yellow
1533 stars show the 95% quantile.

1534

1535 Figure 5 – a) Correlation matrix showing the correlation coefficient (R^2) between different
1536 oxides in the 42 clinopyroxenes measured by Neave et al. (2019) in experiment Y0201–2, with
1537 cells colored based on this v R^2 value. b-c) 500 synthetic Cpx analyses with major elements
1538 distributed normally using the reported mean and standard deviation of the 42 clinopyroxenes
1539 measured in the experiment (yellow dots) versus measured compositions in red dots.

1540

1541 Figure 6 – Analytical precision as a function of Na_2O content and analytical conditions. a)
1542 Theoretical precision versus concentration curves calculated for 4 different beam conditions
1543 on the MIT JEOL733 are shown in orange. EPMA software-calculated precisions for individual

1544 Cpx analyses are shown in red circles from Krawczynski et al. (2012) on the MIT JEOL8200
1545 and orange circles from the older JEOL733. Phase variation/precision calculated from each
1546 experiment of Krawczynski et al. (2012, combining analyses from both probes) are shown as
1547 yellow squares. Analyses were performed with $I^*t = 150$ nA s. b) same for Al_2O_3 . c) Software-
1548 calculated precision for Cpx measurements from Neave et al. (2019) on the University of
1549 Hannover SX100 (cyan dots) with phase variation overlain (blue squares). The necessary count
1550 rates to use equation 2 could not be obtained from this instrument, so we overlay curves
1551 calculated using various analytical conditions for the old MIT JEOL733 and newer Bristol
1552 JEOL JXA8530F. c) Same for Al_2O_3 . Theoretical lines are not shown, as there is substantially
1553 more variation in instrument efficiency and analytical conditions between different labs for Al
1554 compared to Na (See Supporting Table 1). d-e) Experimental precision for Cpx in different
1555 experimental charges reported in the LEPR dataset are shown as grey dots. We also overlay
1556 software-calculated precisions from MIT and Hannover, as well as measurements on the
1557 University of Cambridge Cameca SX100 from Gleeson et al. (2021), and measurements of
1558 Kakanui augite on the Oregon State Cameca SX100. We overlay theoretical precision lines
1559 representing best-case (newer Bristol JEOL JXA8530F, $I^*t = 400$ nA s) and worst-case
1560 scenarios (older MIT JEOL733, $I^*t = 60$ nA s).

1561

1562 Fig. 7. a-e) Pressures and temperatures calculated from individual Cpx compositions from
1563 specific experiments using Cpx-only thermobarometers from Petrelli et al. (2020), Putirka
1564 (2008), and Wang et al. (2021). Error bars showing published/commonly quoted 1 sigma error
1565 each thermobarometer (see caption for Fig. 2-3). a) shows the pressures and temperatures
1566 calculated from the 42 Cpx from experiment Y0200–3 of Neave et al. (2019). The yellow star
1567 shows the P-T conditions at which the experiment was conducted. b) shows experiment BM49
1568 from Melekhova et al. (2015), c) shows Exp CE10 from Erdmann et al. (2016), d) shows Exp

1569 41c-110 from Krawczynski et al. (2012), and e) shows Exp2374 from Blatter et al. 2013. In f),
1570 for 3 different experiments, we show the strong correlation between Na₂O in the Cpx and
1571 calculated pressure using the Putirka (2008) barometer (diamonds) and the Wang et al. (2021)
1572 barometer (circles). Experiments showing a larger spread of Na₂O values (Neave et al. 2019,
1573 Melekhova et al. 2015) show a much wider spread of pressures than the Blatter et al. (2013)
1574 experiment.

1575 Figure 8– Monte Carlo simulations showing the spread of Cpx-only and Cpx-Liq pressures
1576 resulting from electron microprobe-calculated analytical precision of experimental Cpx from
1577 Exp Y0272–1 of Neave et al. (2019). Measured Cpx compositions are overlain in a-c, and all
1578 possible combinations of measured Cpx and Liquids in d-e). The error bar on each plot shows
1579 the reported error on each thermobarometer, and the yellow star shows the analytical
1580 conditions. In a-b, and d-e) including experimental conditions compresses the plot, so the
1581 direction is marked with a yellow arrow. In d-f), Monte Carlo simulations show errors for Cpx
1582 and the average glass composition. Monte Carlo simulations and thermobarometry calculations
1583 performed in Thermobar, contours showing 67% and 95% of simulations calculated using
1584 Pyrolite. Supporting Fig. 20 shows the same plots for experiment B1084–10.

1585

1586 Figure 9 -As in Fig. 8 but for Exp B1038 from Krawczynski et al. (2012). In d-f), Monte Carlo
1587 simulations show errors for both Cpx and Glass. The 95% contour using only analytical noise
1588 for Cpx (and not glass) is shown in grey for comparison with Fig. 8 (where the analytical error
1589 on glass analyses from Neave et al. 2019 was not available).

1590

1591 Fig. 10 – Demonstration of the absolute pressure offset that can be introduced by measuring
1592 insufficient Cpx from any given experimental study. a-b) Groups of 1 to 100 Cpx were

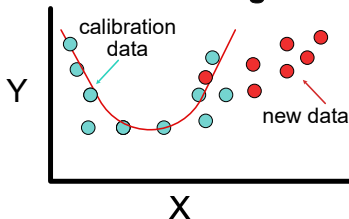
1593 randomly sampled from an underlying distribution of 5000 Cpx simulated using reported
1594 EPMA errors for the experiment of Neave et al. (2019) Y0272–1. The difference in pressure
1595 calculated for this group is compared to the average pressure calculated for all 5000 Cpx, and
1596 plotted as a single dot (see Supporting Fig. 21). c-d) To simulate the combined influence of
1597 true phase variability and analytical errors, we repeatedly resample and average the 18
1598 measured Cpx from Krawczynski et al. (2012, see Supporting Fig. 22). e-f) Random sampling
1599 of measured Cpx-Liq pairs from Krawczynski et al. (2012, see Supporting Fig. 23). On all
1600 plots, the y axis shows the absolute discrepancy between the pressure calculated during each
1601 run through the for loop and the average pressure. In a-d), this is the average PT for the entire
1602 simulated cloud of Cpx, in e-f) it's the average for the reported mean Cpx and glass
1603 composition from the experiment). For each discrete number of Cpx, we calculate the mean
1604 (blue star) and 95% quantile (yellow star).

1605 Fig. 11 – Investigation of Na migration. a-b) tracking the number of counts on the Na peak
1606 with increasing beam exposure on a hydrous rhyolite, a jadeite and a Kakanui augite.
1607 Polynomial fits are included through the data to help with visualization. Note the compressed
1608 y axis scale on b. c-d) Change in elemental concentration for six repeated quantitative analyses
1609 on the same location in jadeite versus Kakanui augite. The approximate beam exposure time is
1610 shown as a secondary axis (although different elements were acquired at slightly different
1611 points during each acquisition).

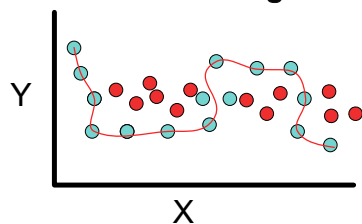
1612

(a) Model Errors

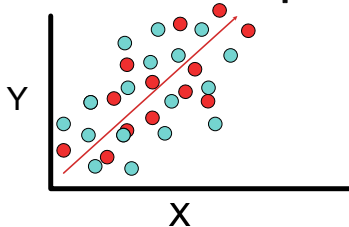
Calibration range



Model overfitting



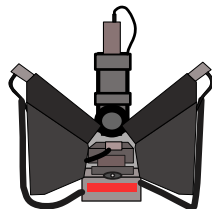
Weak relationship



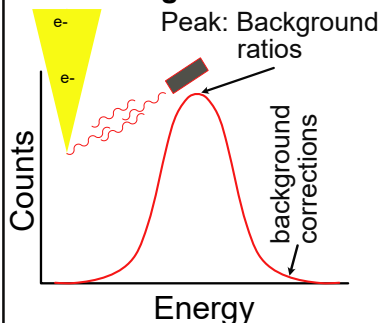
Incomplete data

SiO ₂	TiO ₂	H ₂ O	...	Cr ₂ O ₃
51	2	1	...	0.4
53	1.9	6		×
47	1.8	2.6		0.03
51	2.1	×		×
56	2.3	×		0.01

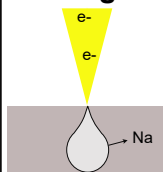
(b) Analytical Errors



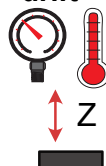
Counting Statistics



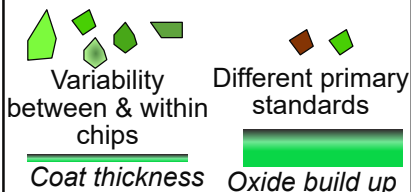
Beam Damage



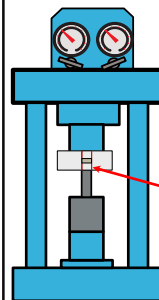
Instrument drift



Primary standards



(c) Experimental Errors



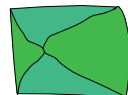
Thermal gradients

H⁺, Fe

Fe/H⁺ loss, poor/ no fO₂ buffering

Relict crystals

Phase Inhomogeneity

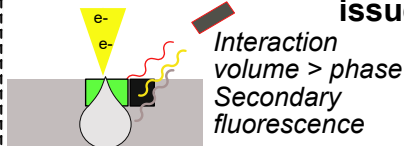


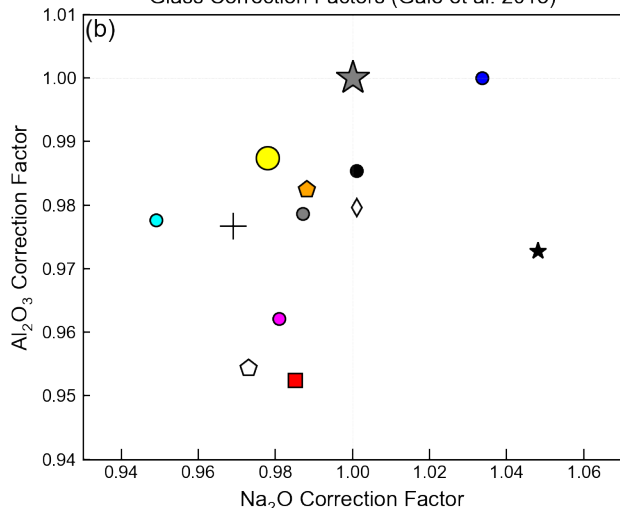
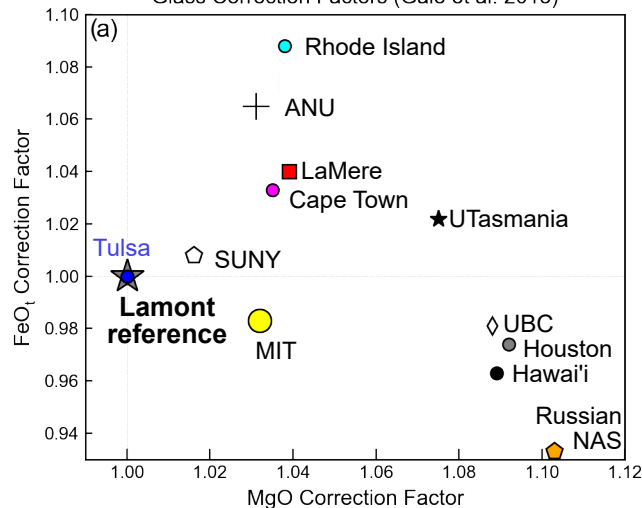
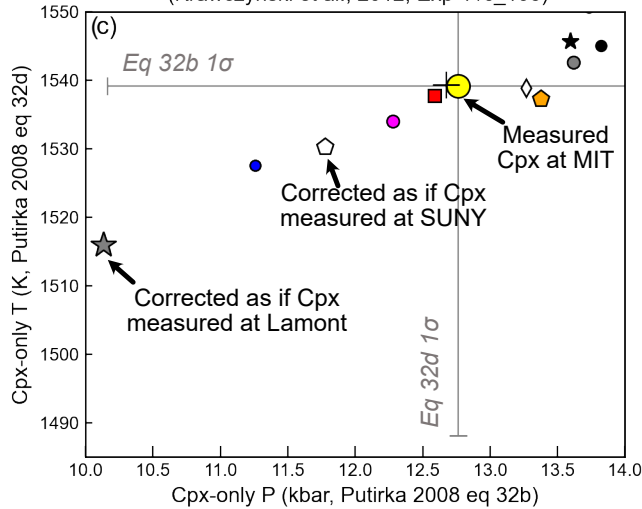
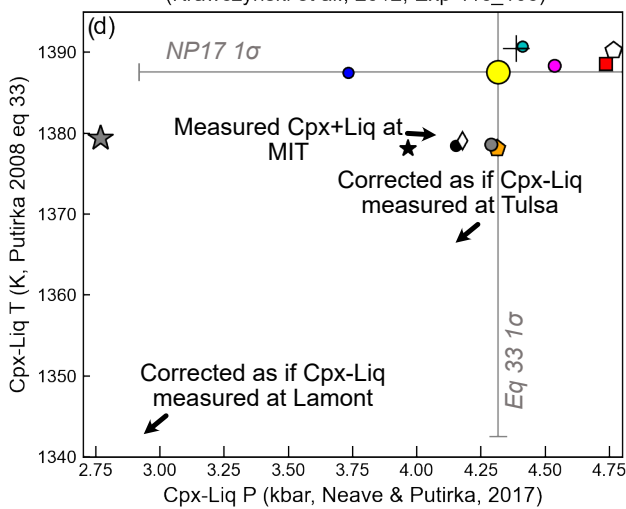
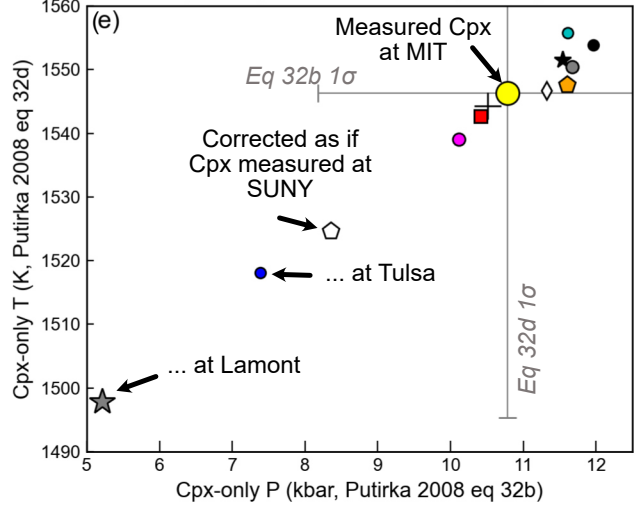
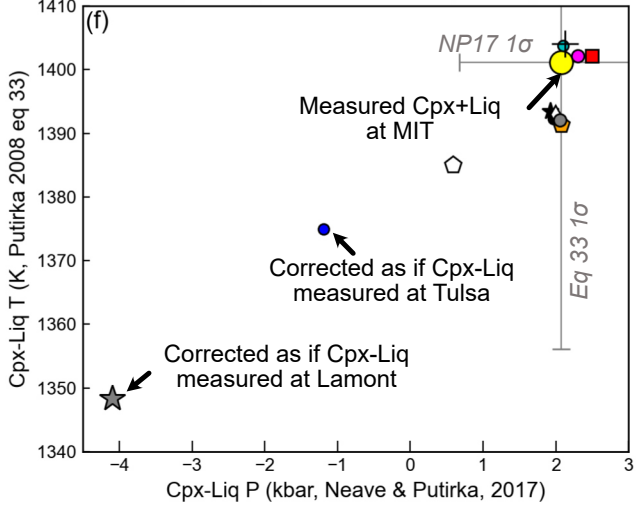
Insufficient time to reach equilibrium

Missing elements vs. real systems

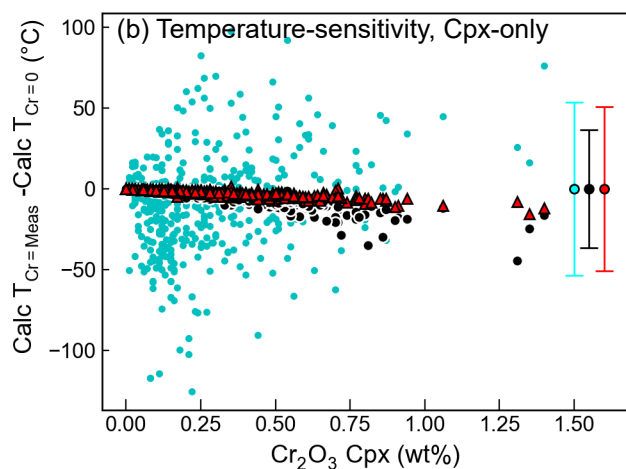
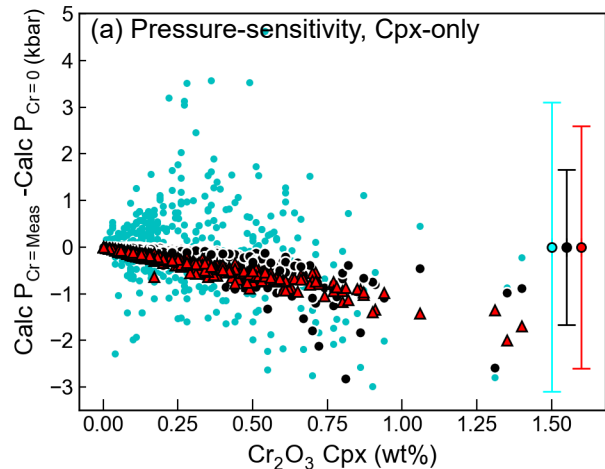


(d) Combined analytical + exp issues

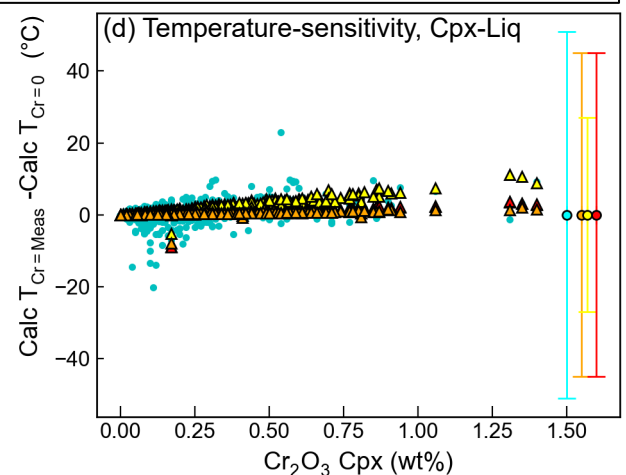
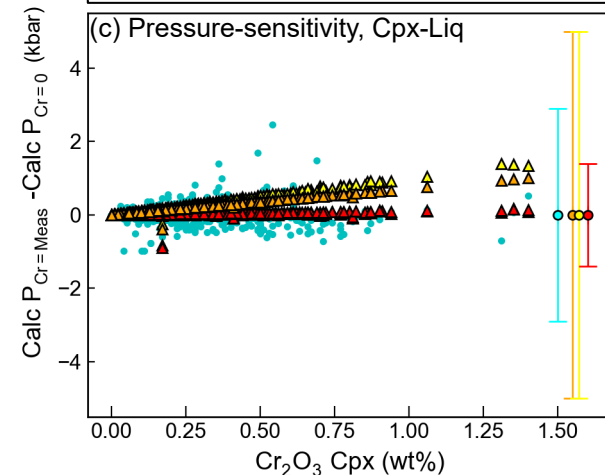


Average Cpx-Only PT
(Krawczynski et al., 2012, Exp 41c_106)Average Cpx-Liq PT
(Krawczynski et al., 2012, Exp 41c_106)Average Cpx-Only PT
(Krawczynski et al., 2012, Exp B1038)Average Cpx-Liq PT
(Krawczynski et al., 2012, Exp B1038)

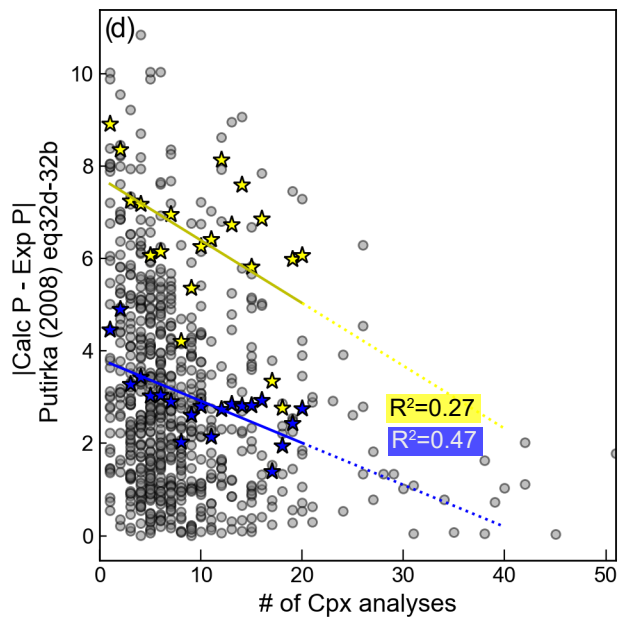
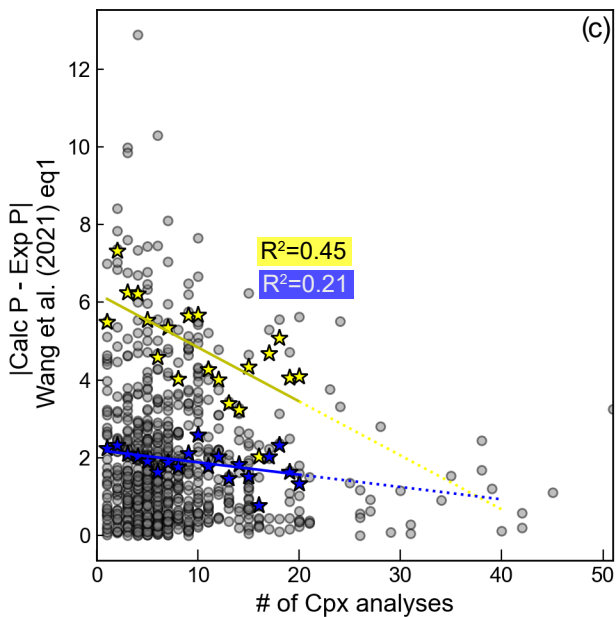
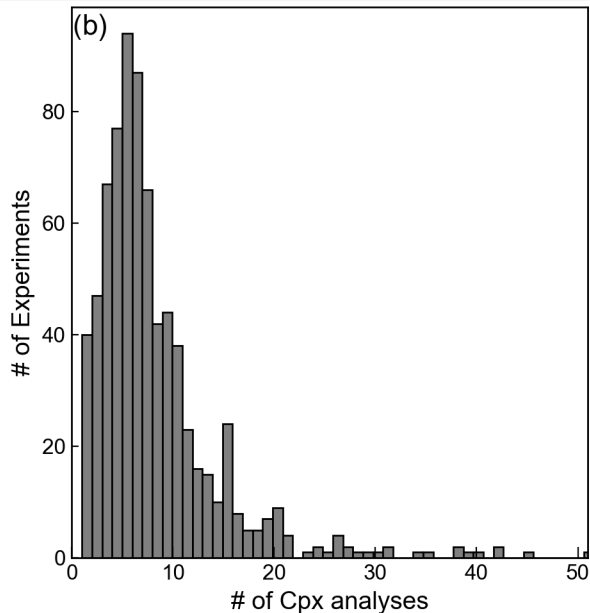
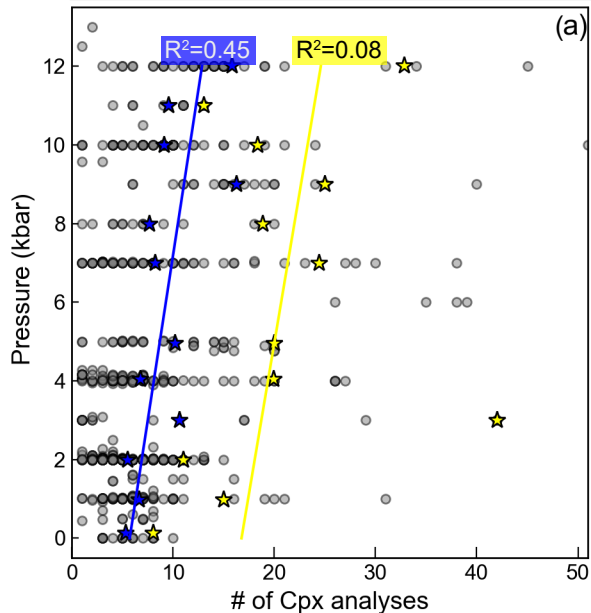
● Petrelli et al. (2020) ▲ P2008 eq32d-32b ● Wang et al. (2021)



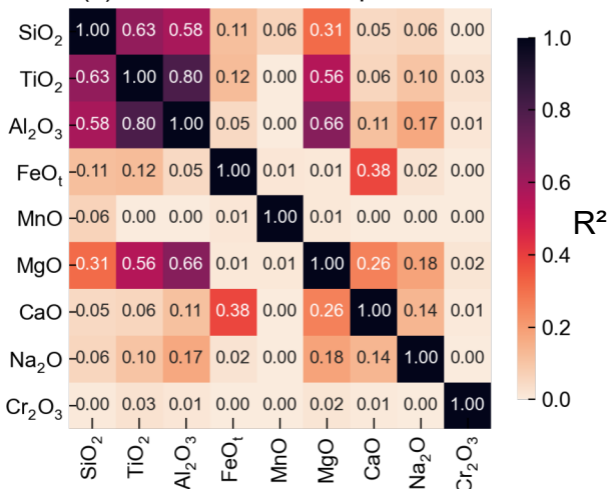
● Petrelli et al. (2020) ▲ P2008 eq33-Neave and Putirka (2017) ▲ P2008 eq33-32c ▲ P1996 T1 - P2008 32c



○ 1 Experimental charge ☆ 95% Quantile_{>4 Cpx} ★ Mean_{>4 Cpx} — Linear Regression

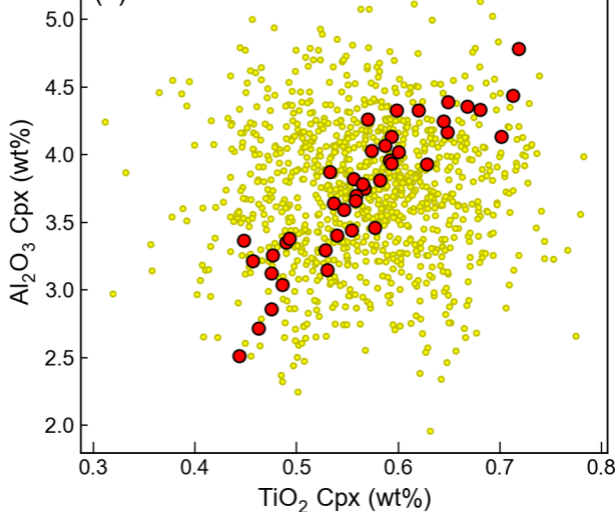


(a) Correlation matrix Cpx

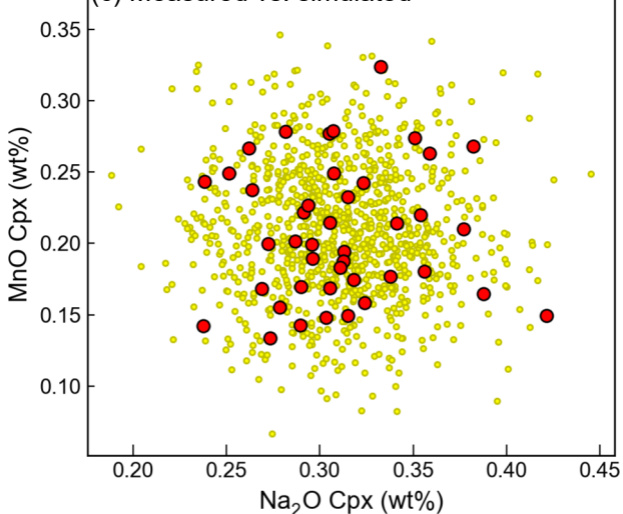


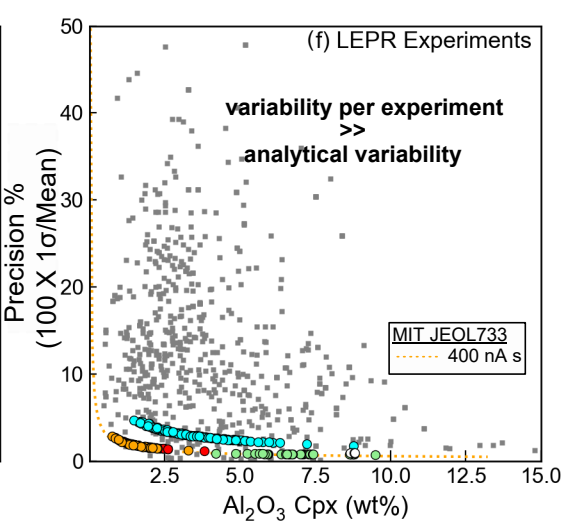
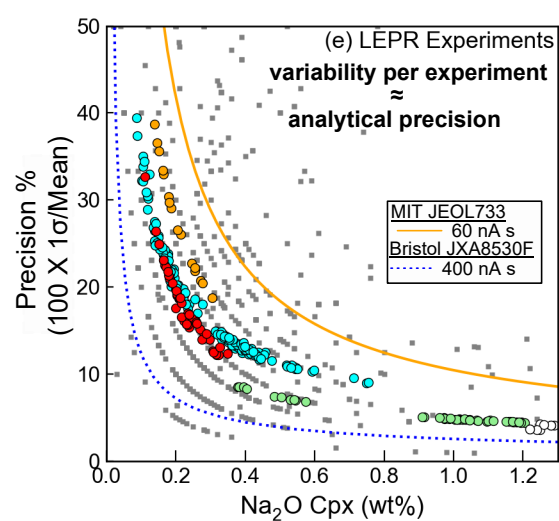
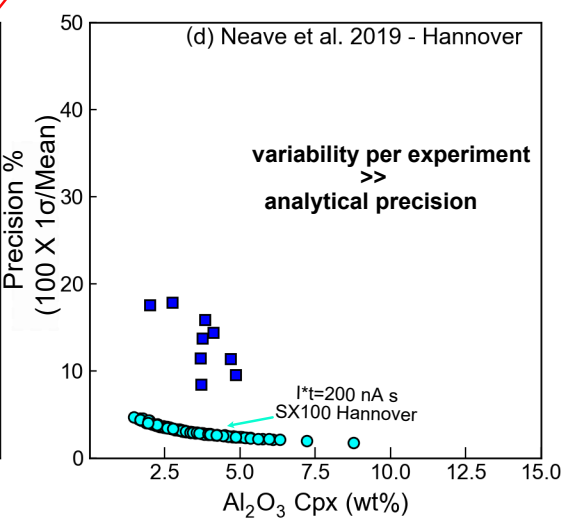
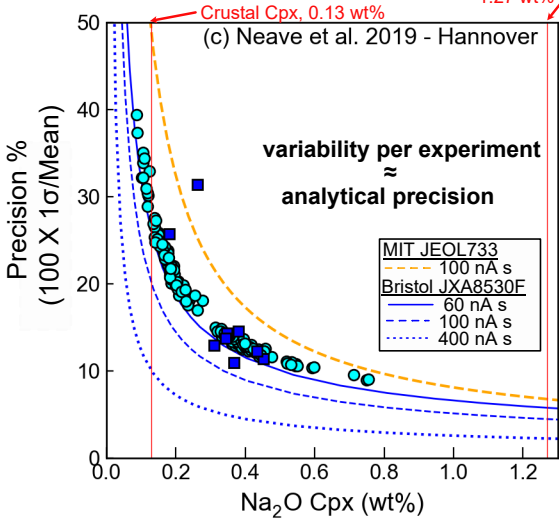
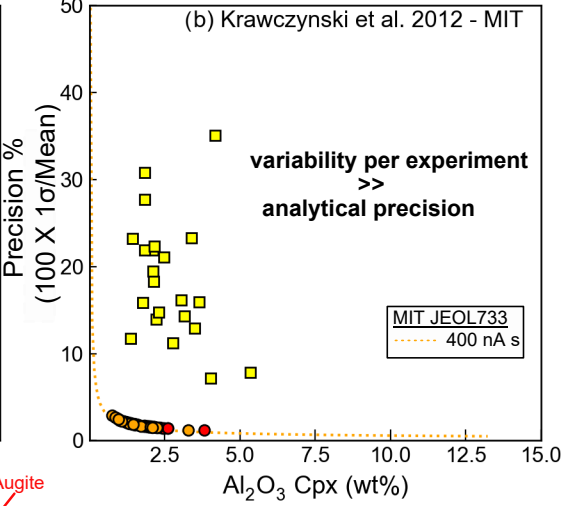
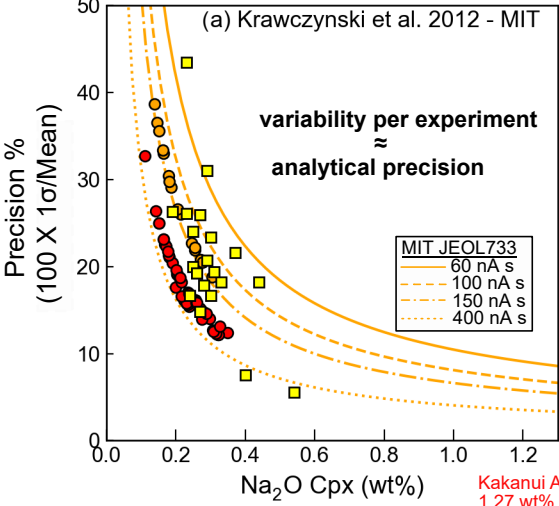
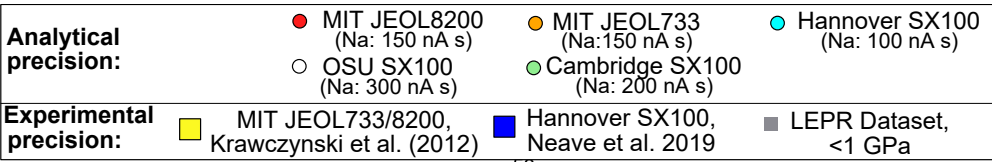
- Measured Cpx
- Simulated Cpx (using mean and σ)

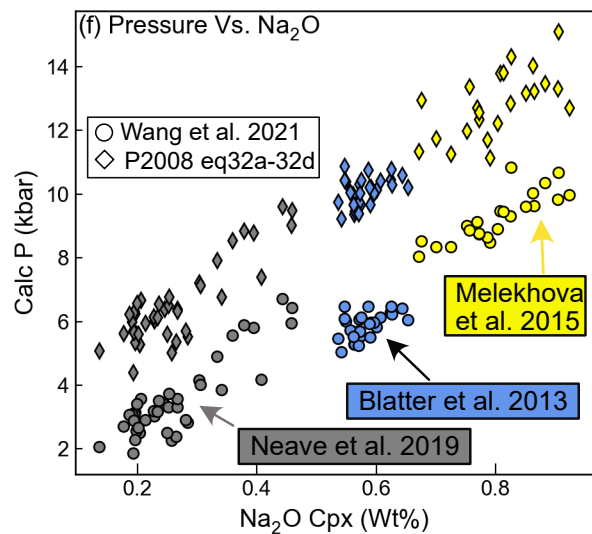
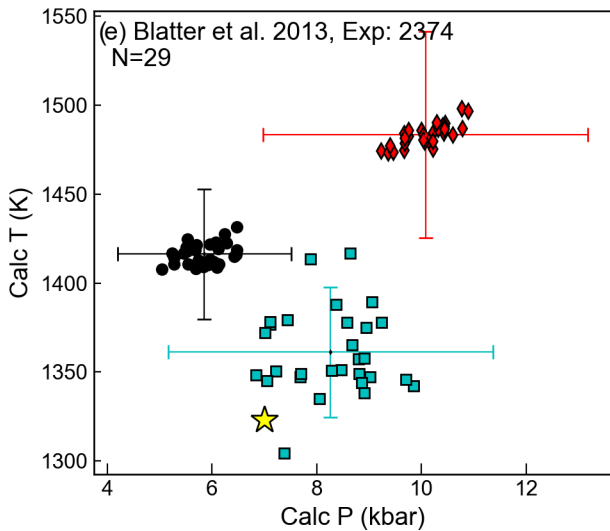
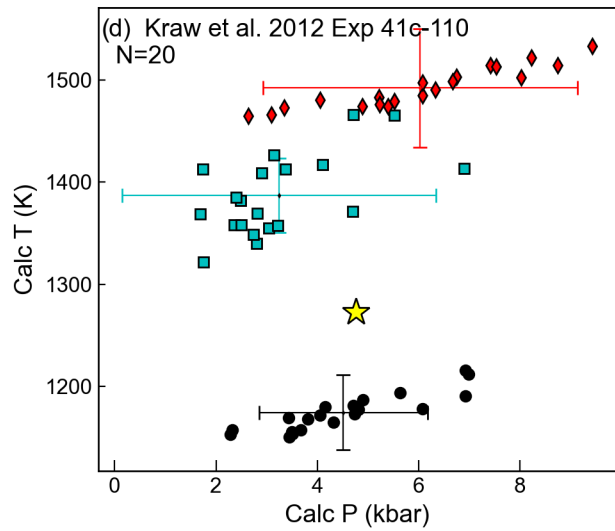
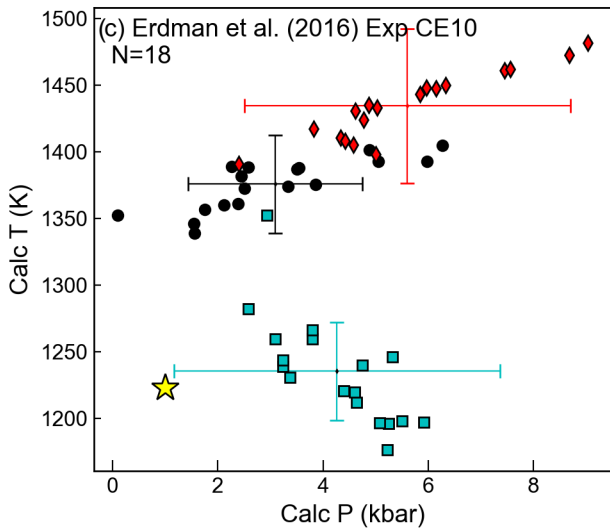
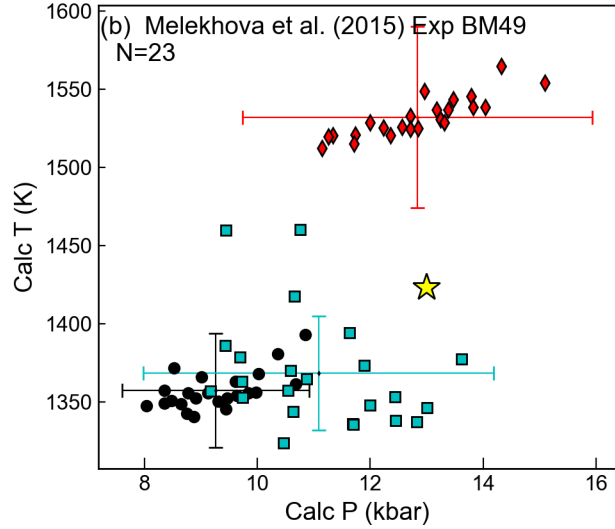
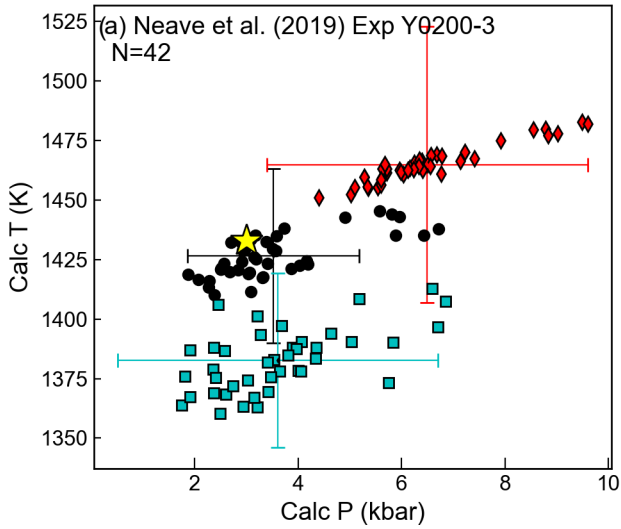
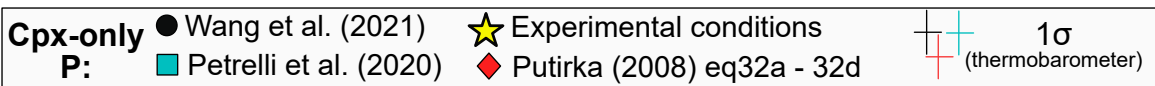
(b) Measured vs. simulated



(c) Measured vs. simulated





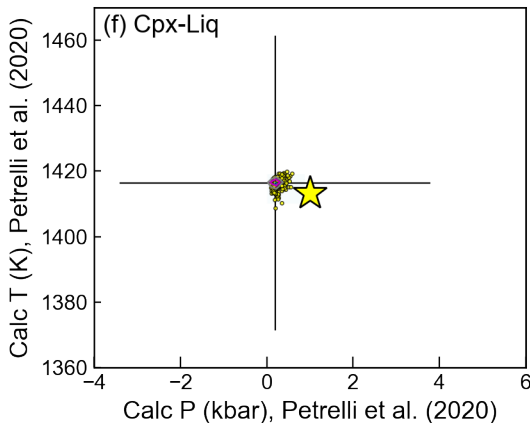
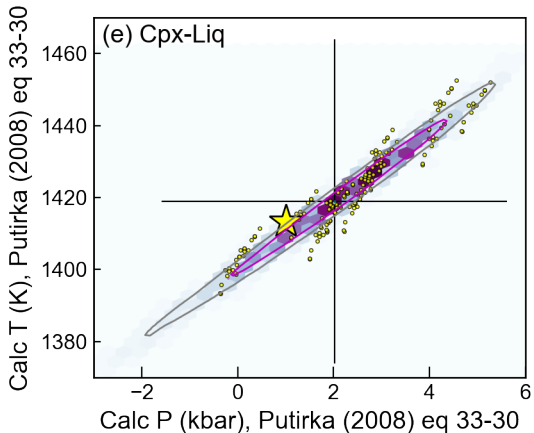
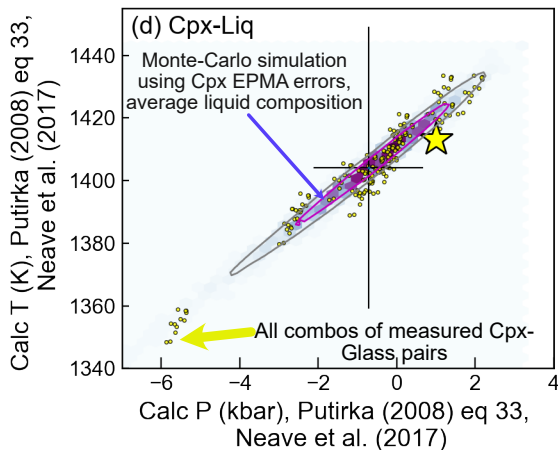
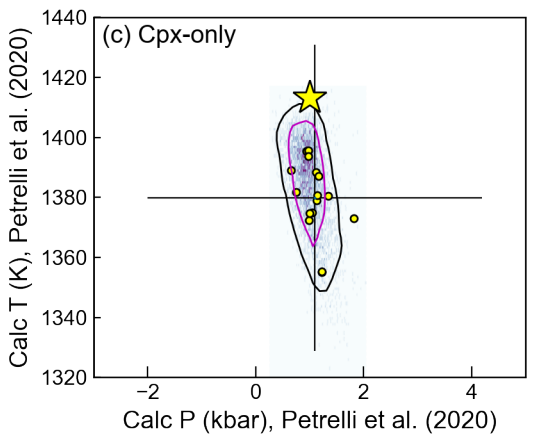
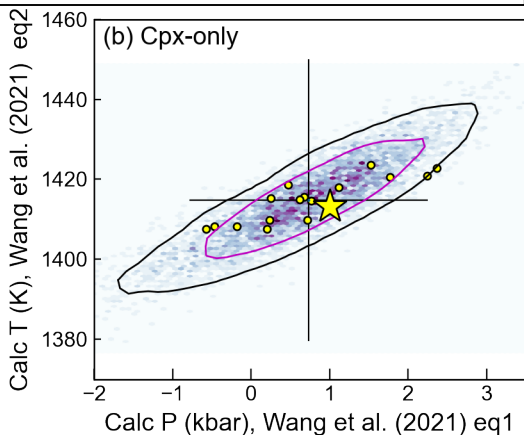
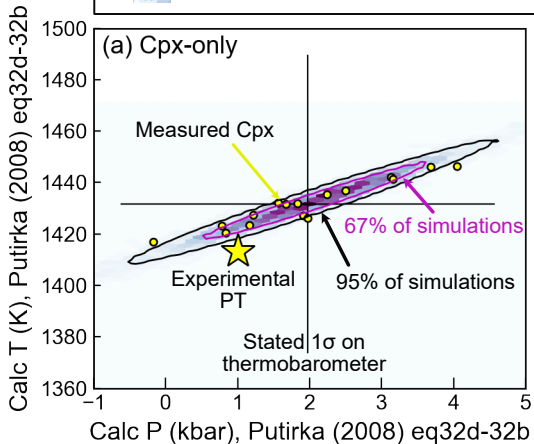



Y0272-1 - Neave et al. (2019) 1 kbar, 1140°C, 96 hours, C-Pt capsule

Monte Carlo simulation (EPMA errors)


Reported 1σ on thermobarometer

Experimental PT

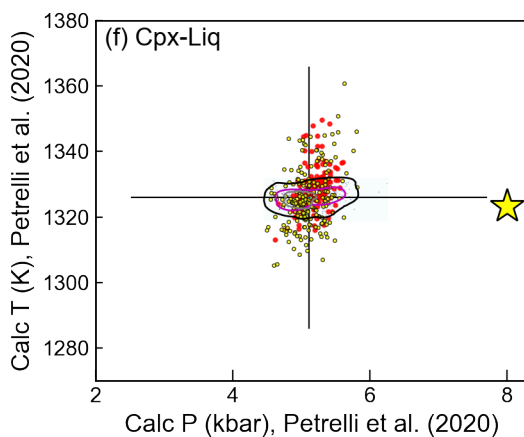
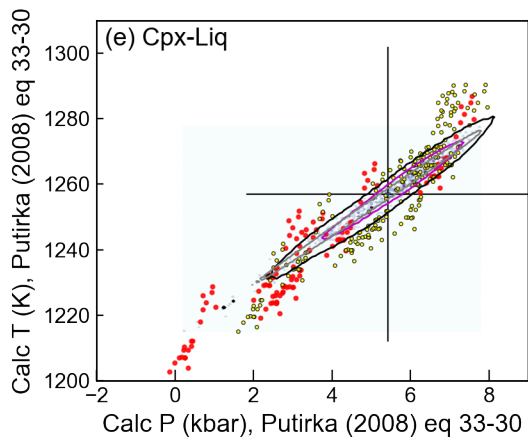
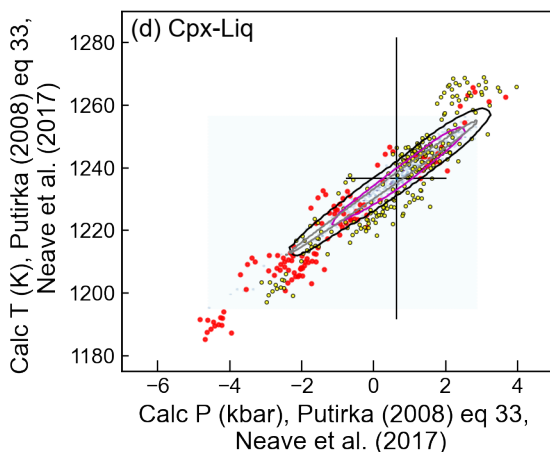
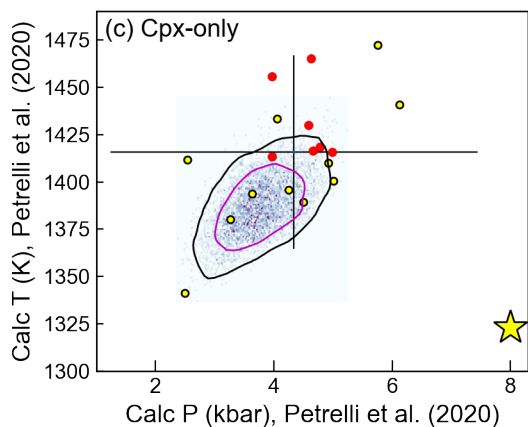
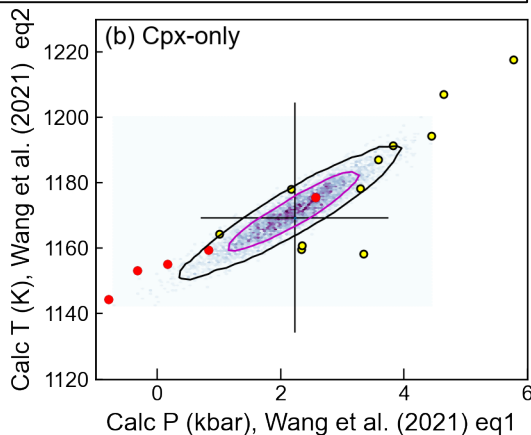
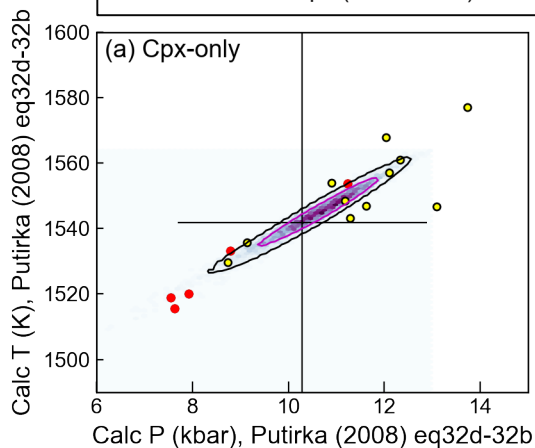


B1038 - Krawczynski et al. (2012), 8 kbar, 1323 K, 7.4 hours, Au caps
 Monte Carlo simulation (EPMA errors)

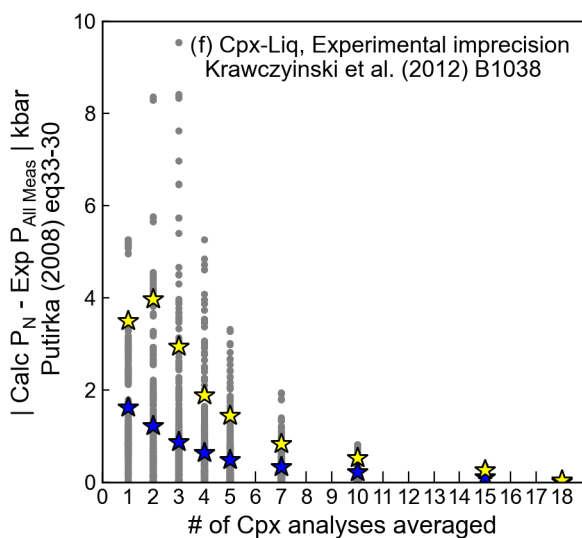
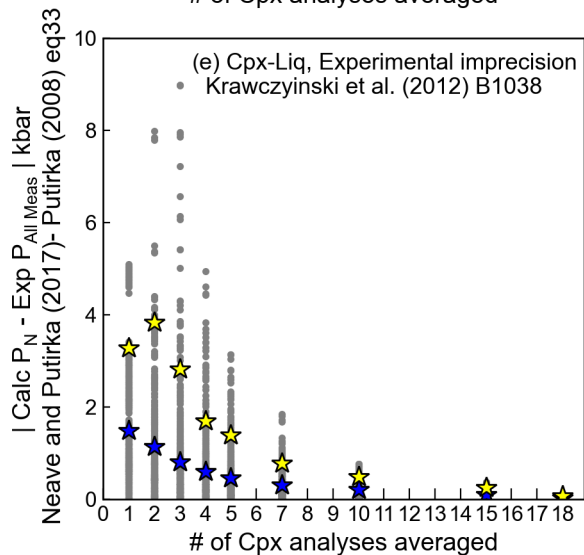
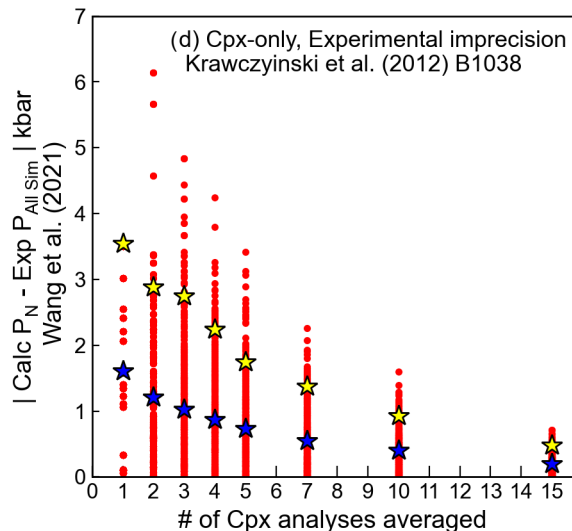
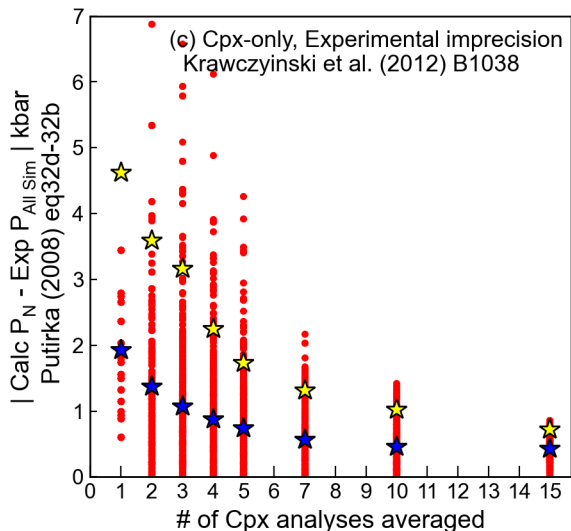
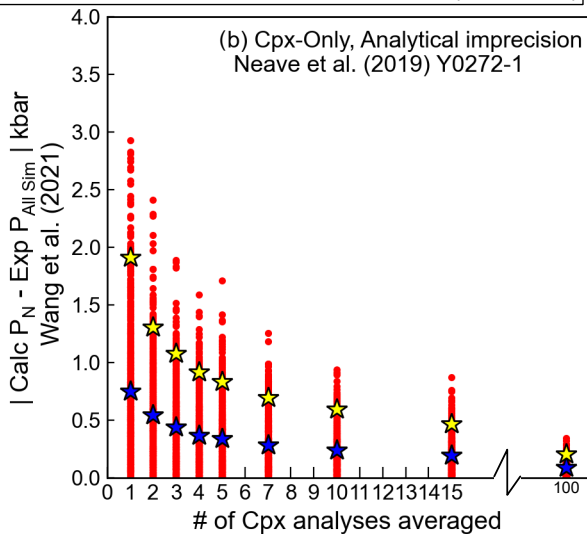
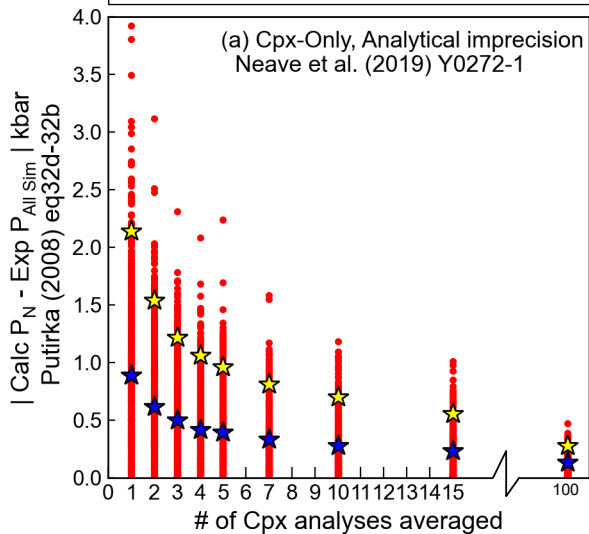
 Reported 1σ on thermobarometer

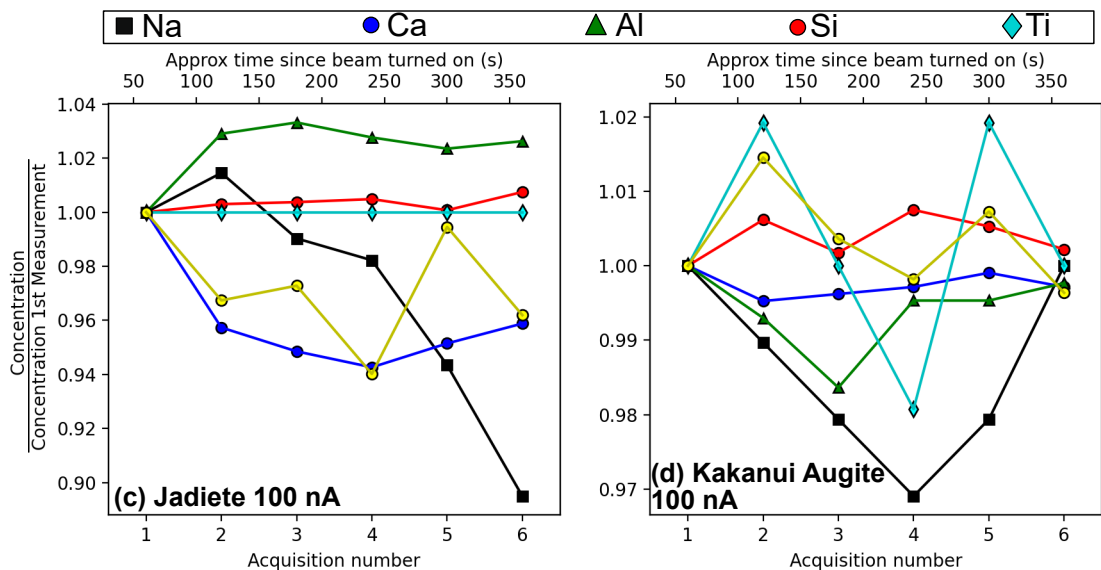
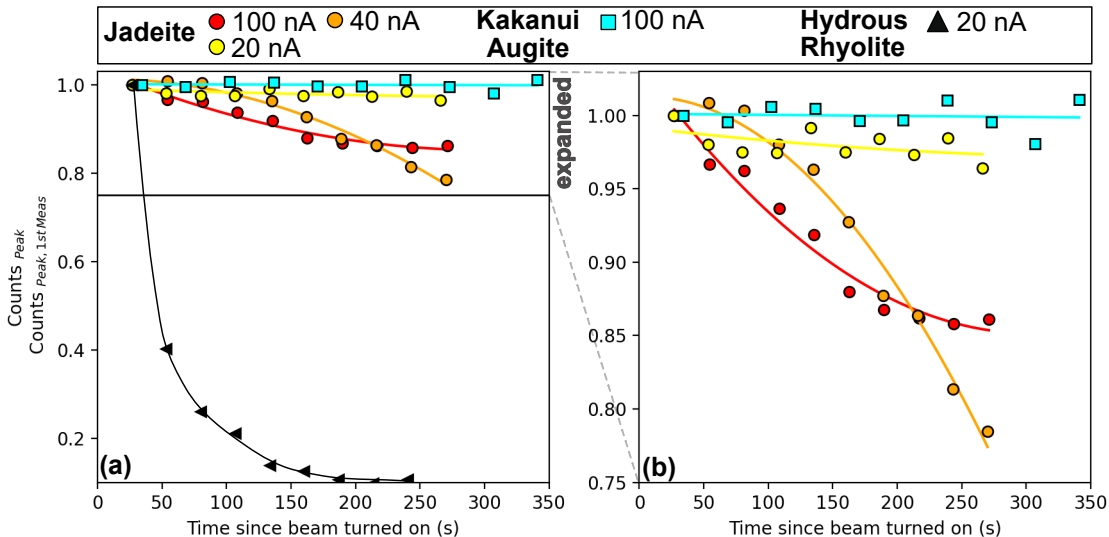
 Experimental PT

 Measured Cpx (Jd from Na)

 Measured Cpx (Jd from Al)


• Av PT per run through for loop ★ Mean discrepancy (for each N) ☆ 95% Quantile (for each N)





Supplementary Information for Barometers Behaving Badly

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7. Department of Earth and Planetary Sciences, Washington University in St. Louis, 1 Brookings Drive, St. Louis, MO 63130, USA

1. Compilation of beam currents and count times used for Na

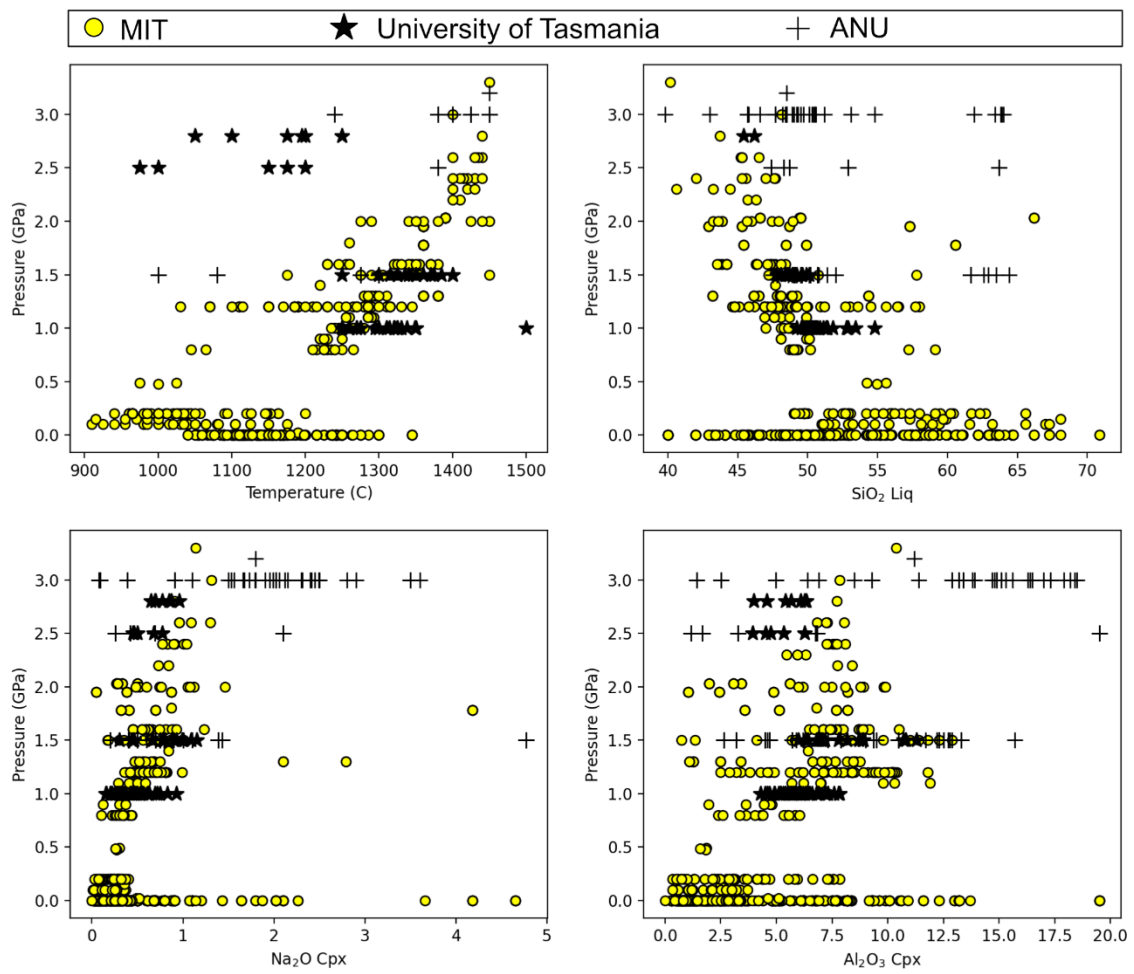
Supporting Table 1- *Compilation of beam current and count times for Na analyses in a subset of the LEPR dataset conducted at <1 GPa used to calibrate most existing thermobarometers. Missing or incomplete/ambiguous data are colored orange.*

Reference	Beam Current (I)	Count time (t)	I*t	Instrument
Akella (1976)				
Baker and Egger (1987)	12 nA			Penn State + Smithsonian
Gee and Sack (1988)	30 nA	2-10 s	60-300	JEOL Superprobe Northwestern, ARL UC Berkeley
Carroll and Wyllie (1989)	5 nA			JEOL at Caltech, Cameca at Brown
Kennedy et al. (1990)	10 nA			MIT MAC-5 and JEOL 733

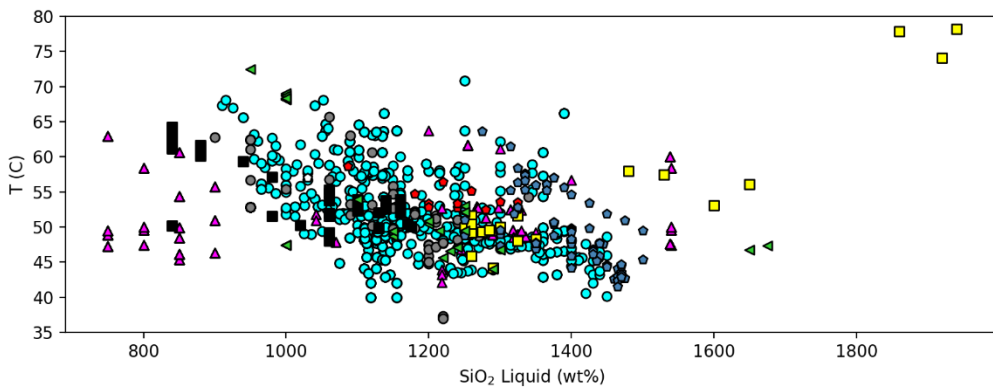
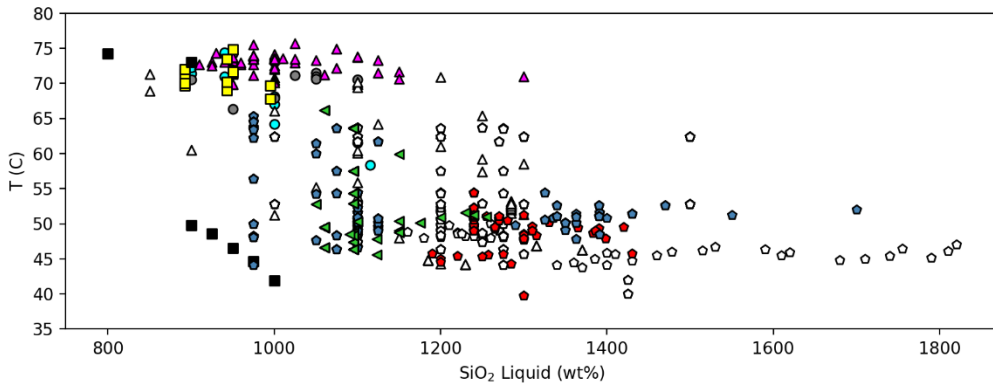
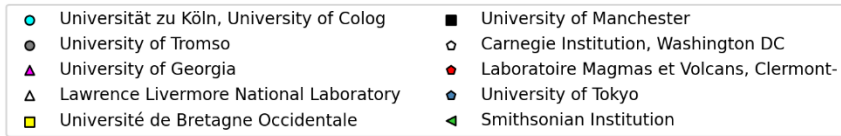
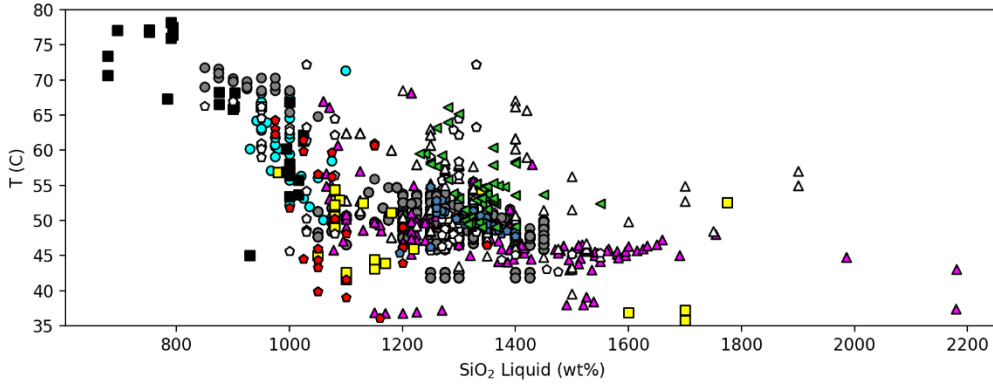
Nielsen et al. (1992)	20 nA	10 s	200	SX50, OSU
Fram and Longhi (1992)	5 nA			
Rushmer (1993)				SX50 (no loc, authors at ETH)
Baker et al. (1994)	10 nA			JEOL 733, Caltech
Baker et al. (1994)				MIT MAC-5 and JEOL 733
Draper and Johnston (1992)	20 nA	10 s	200	Cameca SX50, UOregon
Kawamoto (1996)				JEOL JSM-840 e, University of Tokyo
Springer and Seck (1997)	10 nA	5 s	50	camebax
Tsuruta and Takahashi (1998)	12 nA	10-40 s (don't specify for each element)	120-480	JEOL-JXA8800, Tokyo institute of technology
Métrich and Rutherford (1998)	15 nA			Camebax, Brown
Blundy et al. (1998)	15 nA			JEOL 833, University of Bristol
Draper and Green (1999)	20 nA	15 s	300 s	SX50, Macquarie
McCoy and Lofgren (1999)				JEOL JXA-8900R, Smithsonian
Wang and Takahashi (1999)	No information whatsoever on how phase compositions were measured (not even analytical technique)			
Minitti and Rutherford (2000)	15 nA			
Tielpo et al. (2000)				JEOL JXA-840A, Pavia
Blatter and Carmichael (2001)	20 nA			SX50, UC Berkeley
Wood and Trigila (2001)	15 nA	10-60 (don't say which element is which)	150 - 900	JEOL 8600, Bristol
Berndt et al. (2001)	18 nA	5 s	90	Camebax,
Toplis and Corgne (2002)	10 nA	Says 10s major 30s minor elements (no list)	300	Cameca SX50, Nancy
Scaillet (2003)	6 nA	10 s	60	(authors at Orleans)
Pertermann (2003)	7.5-15 nA	30 s	225-450	JXA8900R, Minnesota
Prouteau (2003)	6 nA	10 s	60	SX50, Orleans
Wasylenki (2003)	30 nA			Caltech JEOL733
Laporte et al. (2004)	15 nA	10s	150	SX-100, authors at Clermont-Ferrand
Maaløe (2004)	Uses EDS – no further information about analytical conditions. E.g. unclear if any standards used.			JEOL-2400 SEM
Barclay (2004)	20 nA	10 s	200	SX50, UC Berkeley
Feig et al. (2006)	15 nA	5 s	75	SX100,
Di Carlo (2006)	6 nA	10 s	60	SX-50, Orleans

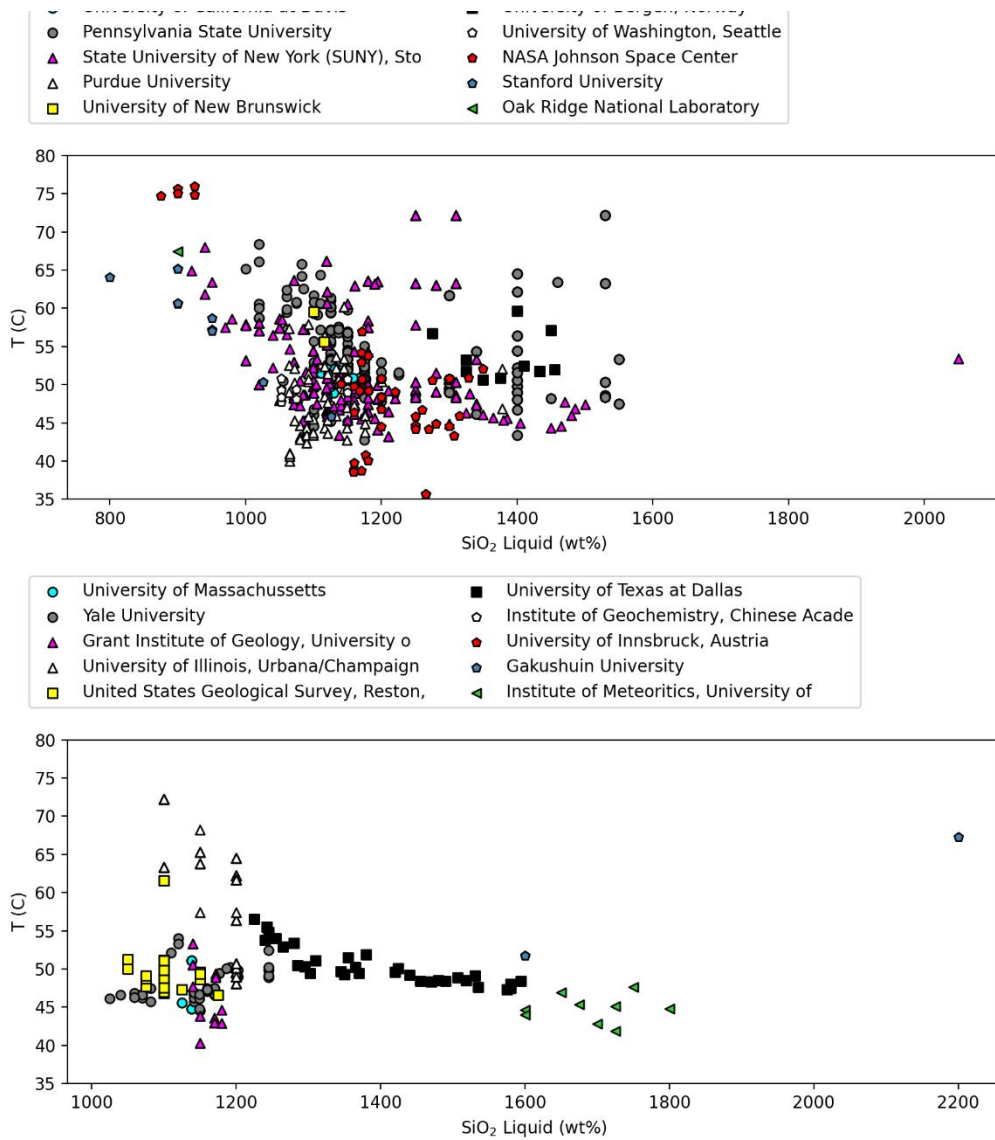
Scoates et al. (2006)	10 nA	10 s	100	Cameca SX-50, Universite´ de Pierre
Alonso-Perez et al. (2009)	20 nA			SX50/JEOL 8600
Pertermann and Lundstrom (2006)	Use EDS on JEOL JSM 840 SEM with natural and synthetic standards. Some repeat analyses on JXA 8900, no mention of analytical conditions.			
new compilation				
Costa (2004)	12 nA	10 s	120	SX50 Orleans
Berndt (2004)	15 nA	5 s	75	SX100 Hannover
Pichavant and Macdonald (2007)	6 nA	6- 10 s	36-60	2 different Cameca, Orleans
Hamada and Fujii (2008)	12 nA			JEOL JXA8800R,
Feig et al. (2010)	15 nA	5 s	75	SX100, nd
Krawczynski et al. (2012)	10 nA	15 s	150	JEOL
Mandler et al. (2014)				JEOL JXA8200 Superprobe
Rader and Larsen (2013)	10 nA			Cameca SX-50, University of Alaska
Blatter et al. (2013)	15 nA	20 s	300	JEOL JXA-8900 Menlo Park
Almeev et al. (2013)	15 nA	8 s	120	SX100 Hannover
Cadoux et al. (2014)	6 nA	10 s	60	SX50 Orleans
Parat et al. (2014)	10 nA	10 s	100	SX100 montpellier
Melekhova et al. (2015)	No info for mineral analyses			SX100 Bristol
Andújar et al. (2015)	6 nA	10 s	60	SX50 Orleans
Nandedkar et al. (2014)	20 nA	20 s	400	JEOL JXA8200, nd, ETH?
Erdmann et al. (2016)	10 nA	10-20 s	100-200	
Husen et al. (2016)	10 nA	10 s	100	SX100 Hannover
Koepke et al. (2018)	15 nA	5 s	75	SX100,
Ulmer et al. (2018)	20 nA	20-30 s	400-600	ARL SEMQ/ SX50/JEOL JXA8200.
Neave et al. (2019)	10 nA	10 s	100	SX100 Hannover
Firth et al. (2019)	10 nA			SX100 Anu, Canberra, Macquarie
Waters et al. (2021)	10 nA	20 s	200	JEOL 8900 Superprobe, NHM

2. Further investigation of interlaboratory offsets

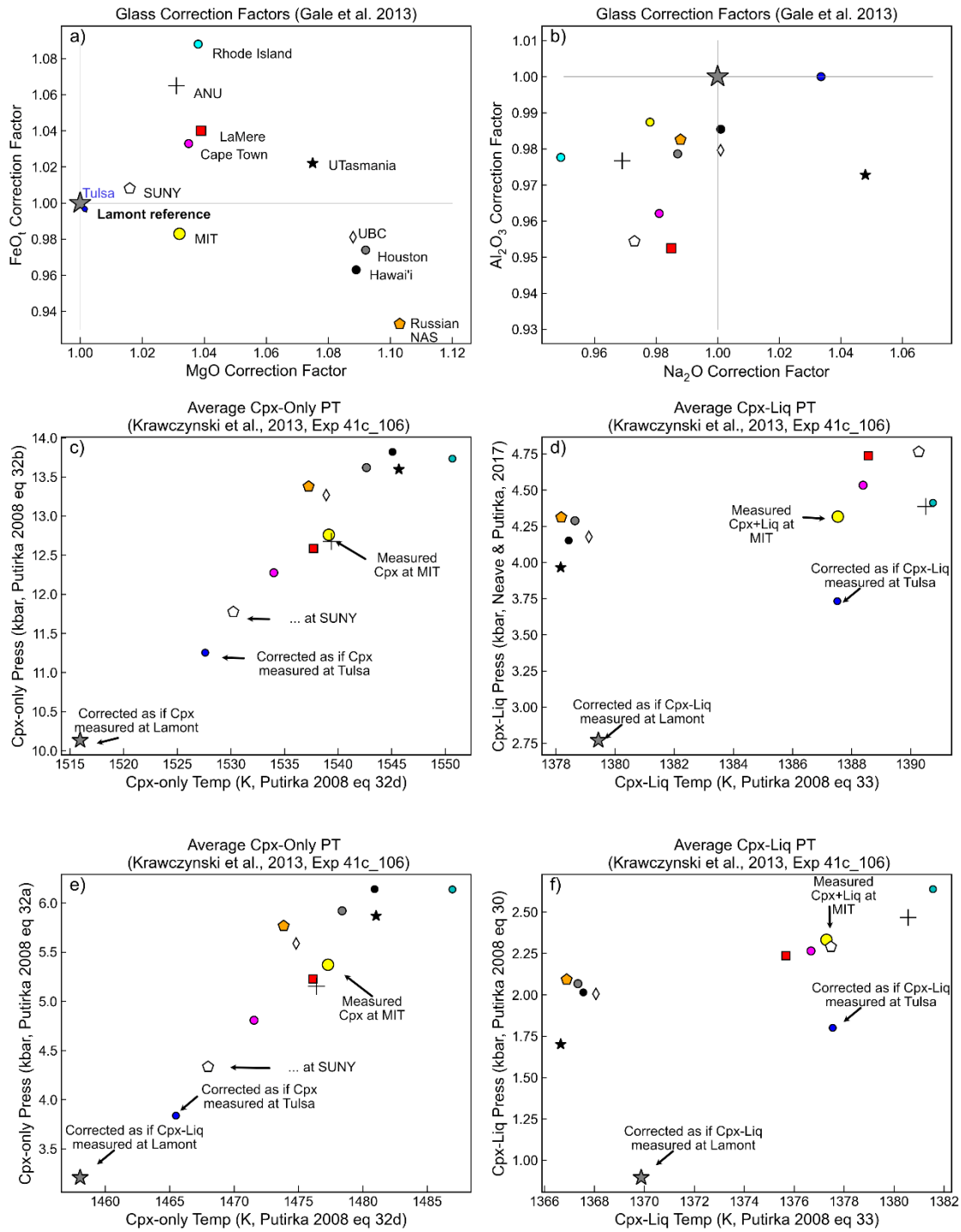


Supplementary Fig. 1 – Non-uniform distribution of P-T-X space covered by different laboratories.

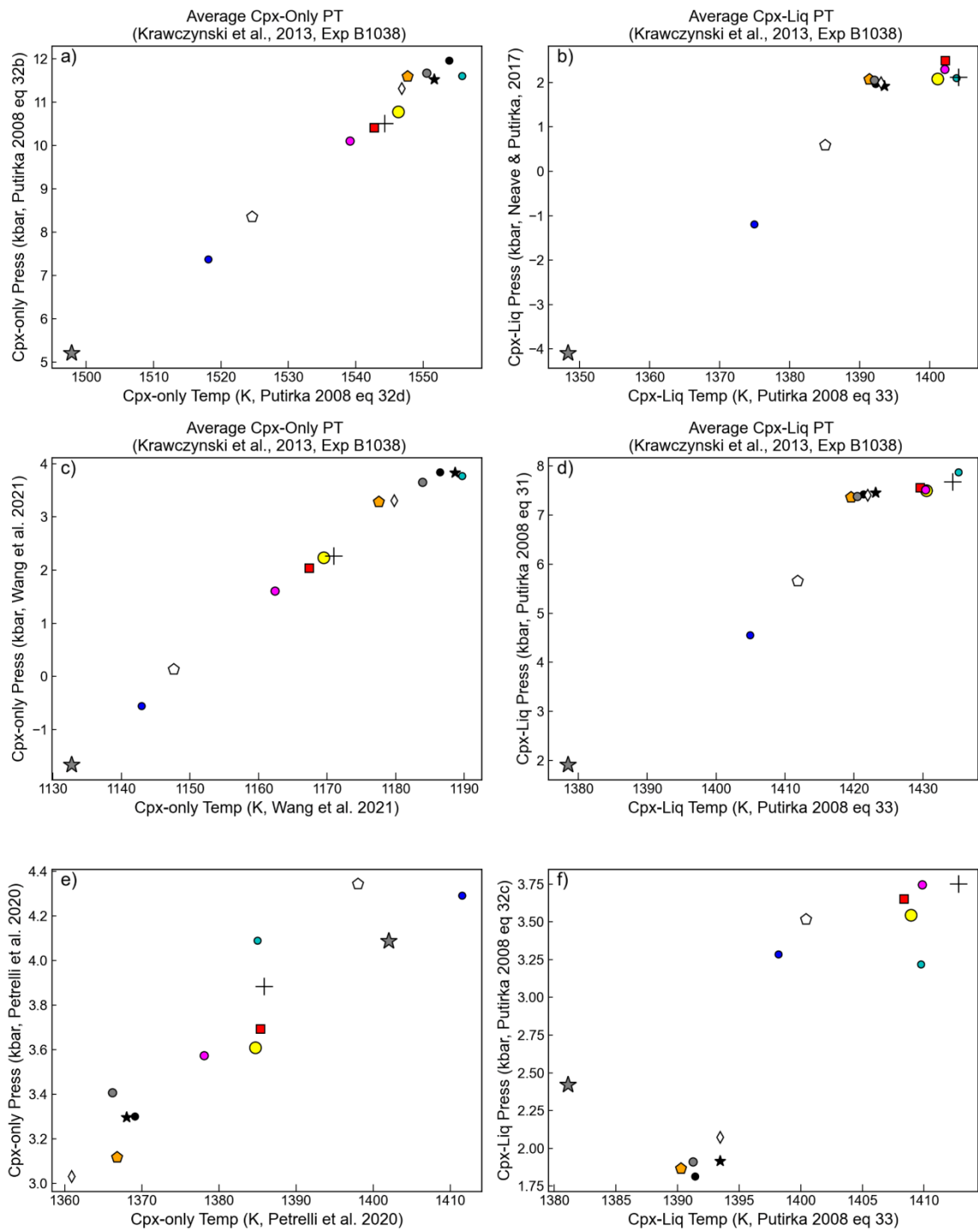




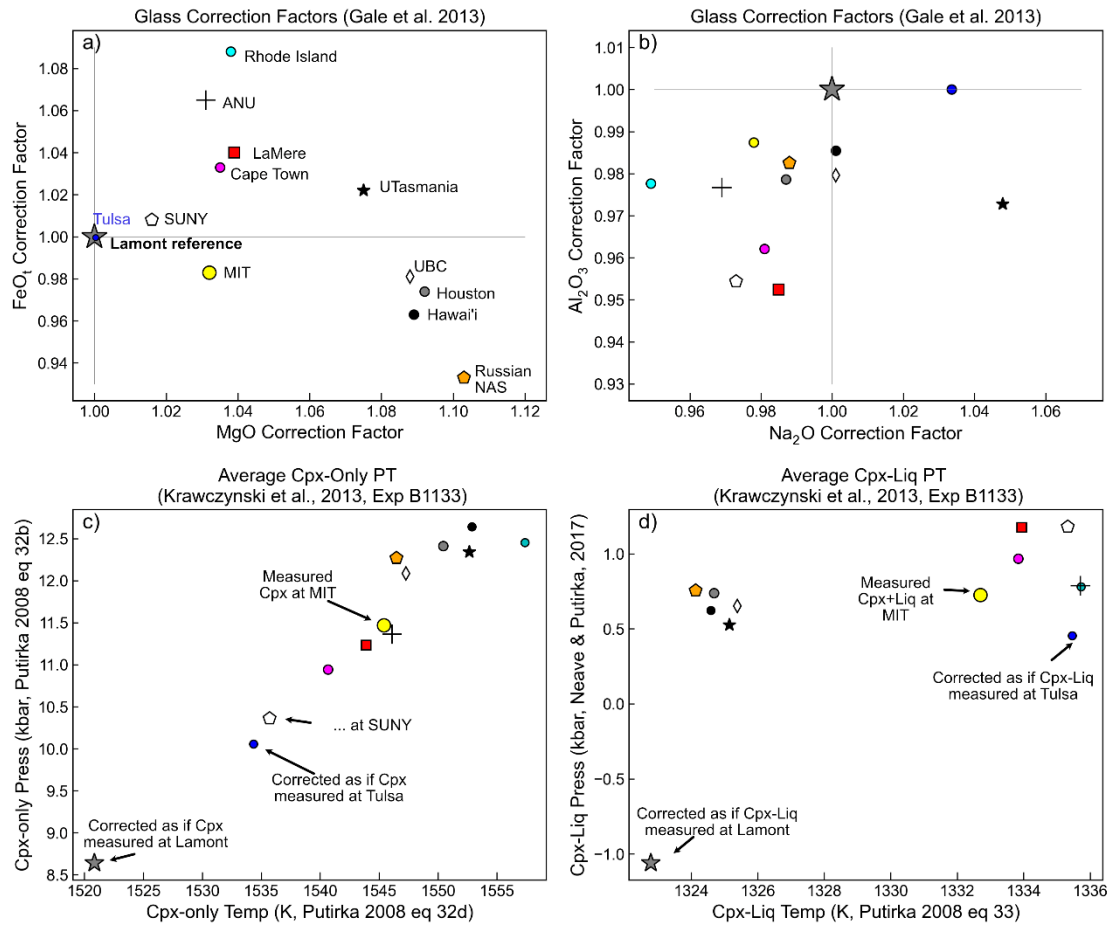
Supplementary Fig. 2 – Range of SiO₂-T space covered by 47 different laboratories in the LEPR dataset.



Supporting Fig. 3 – First 4 panels as in main text, panel e-d showing the differences using eq32a for Cpx-only pressures (e) and equation 30 for Cpx-Liq pressures (d).

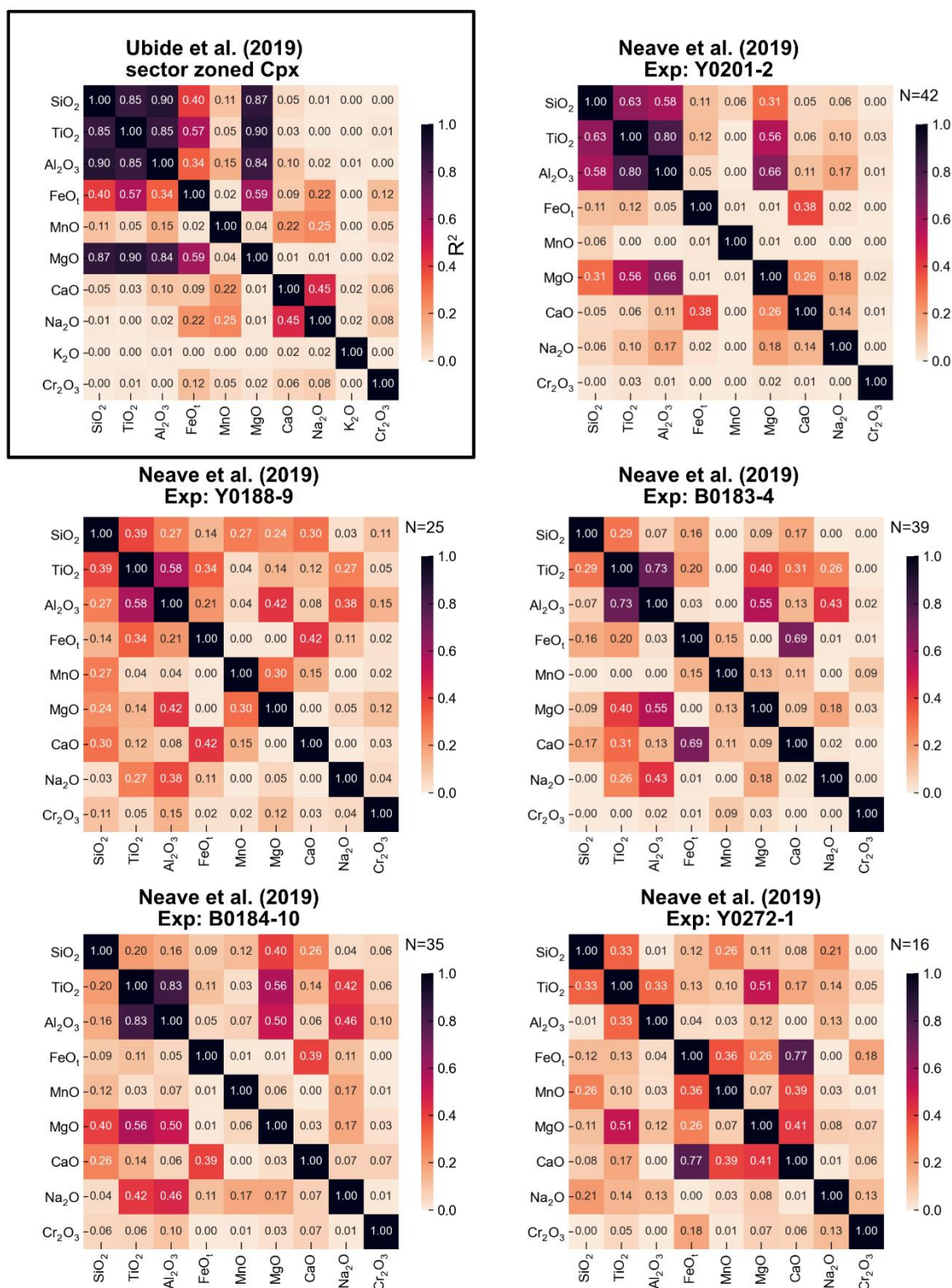


Supporting Fig. 4 – Using the same experiment as the bottom two panels in Fig. 2 of the main text for B1038 but showing discrepancies using different sets of barometers.

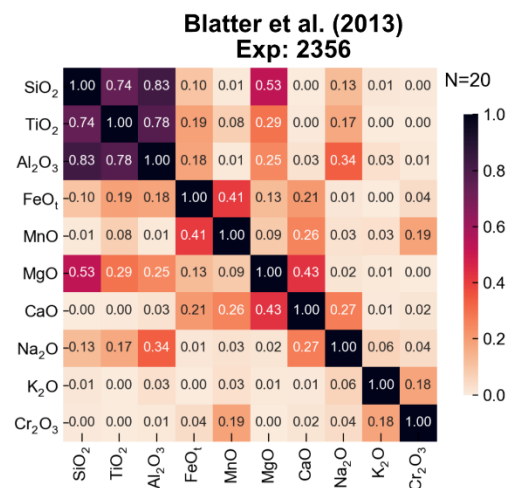
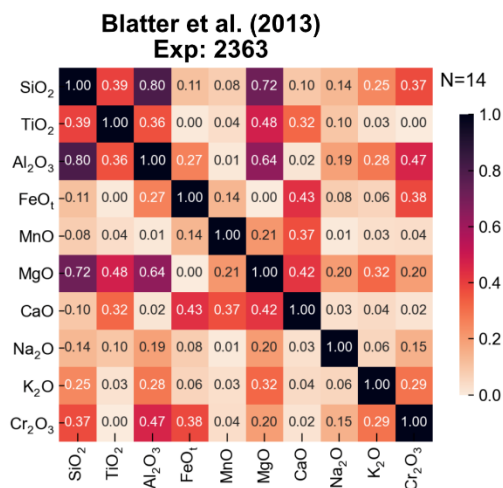
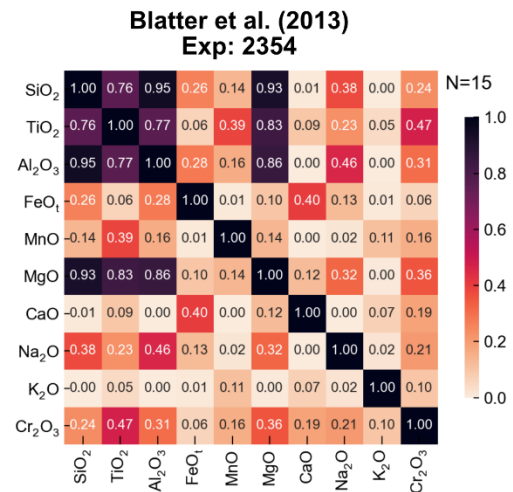
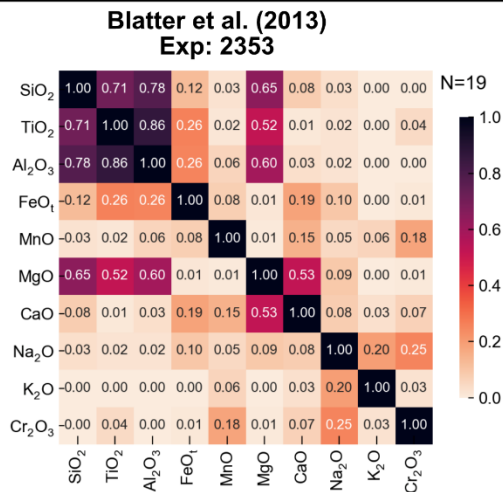
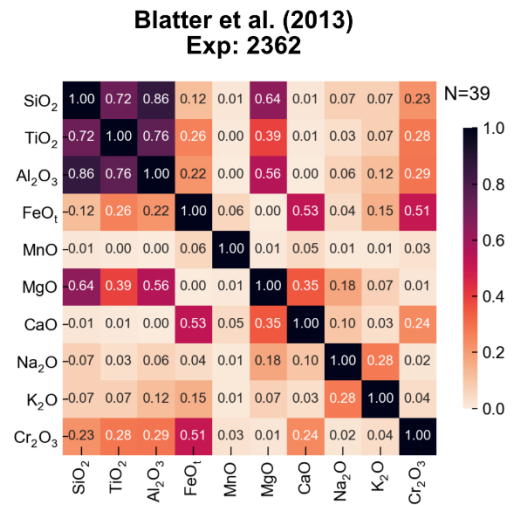
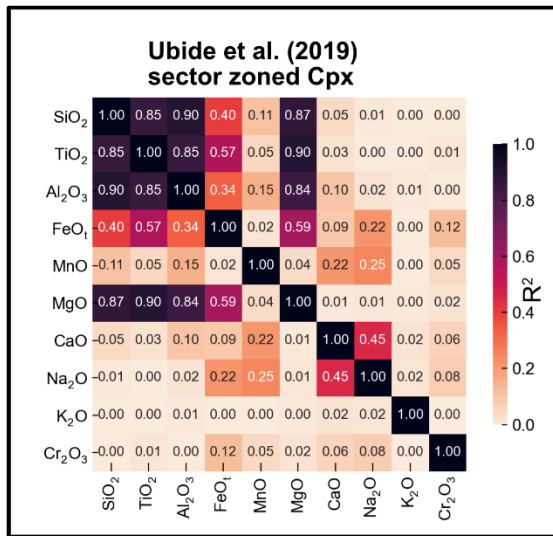


Supporting Fig. 5 – As for Fig. 2 of the main text but showing offsets for Experiment B1133.

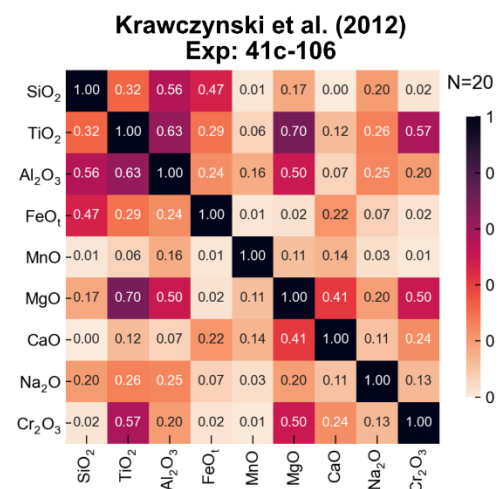
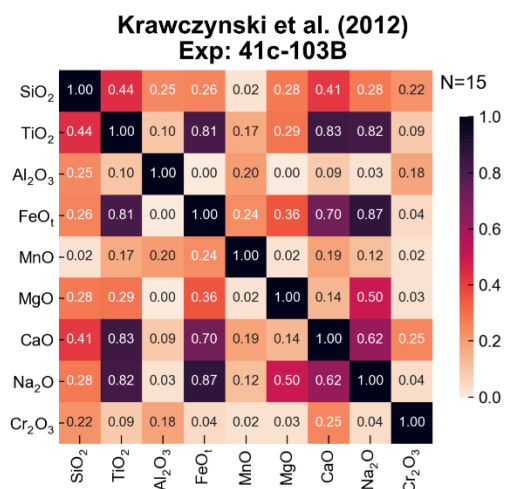
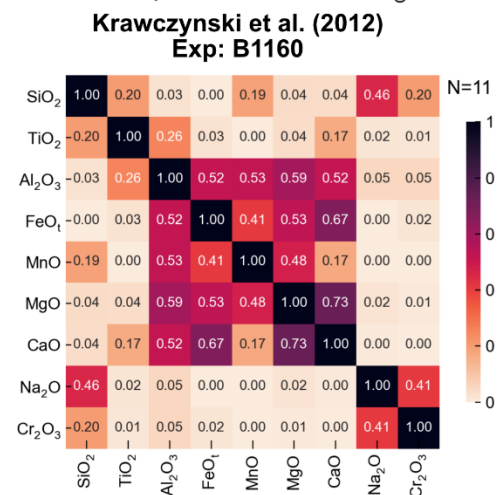
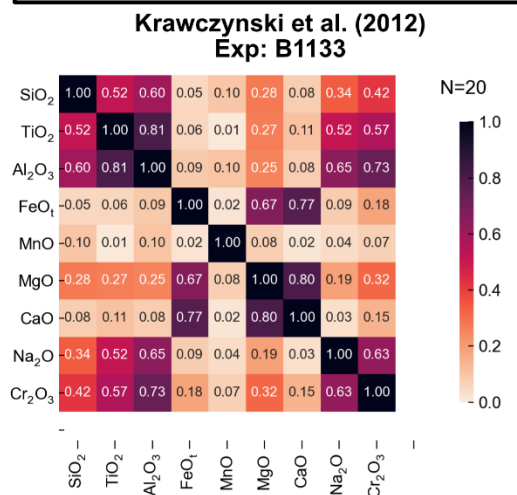
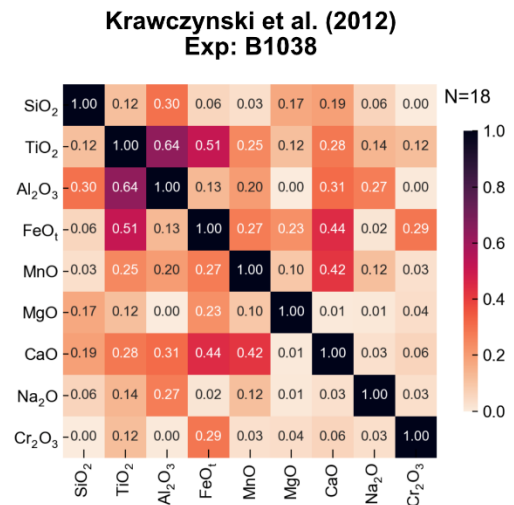
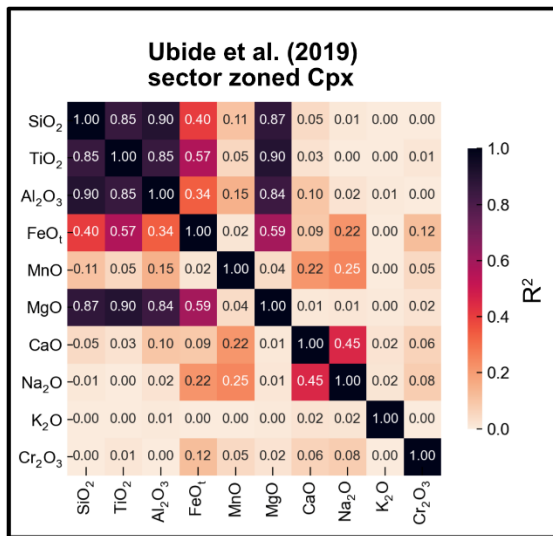
3. Covariance in Experimental Cpx Compositions



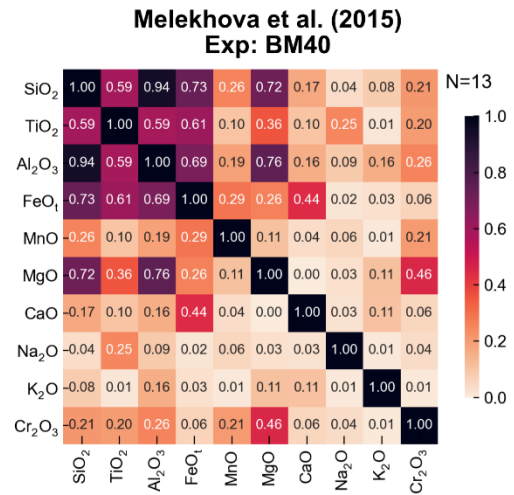
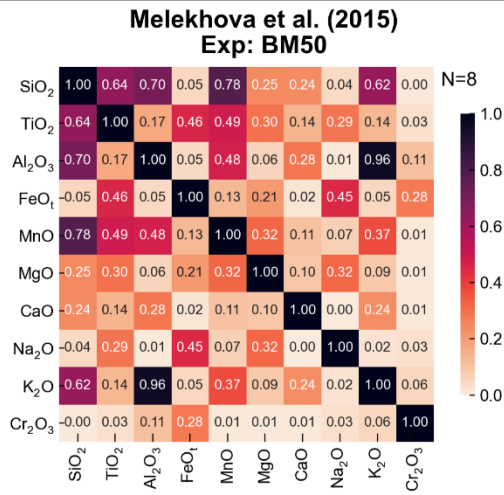
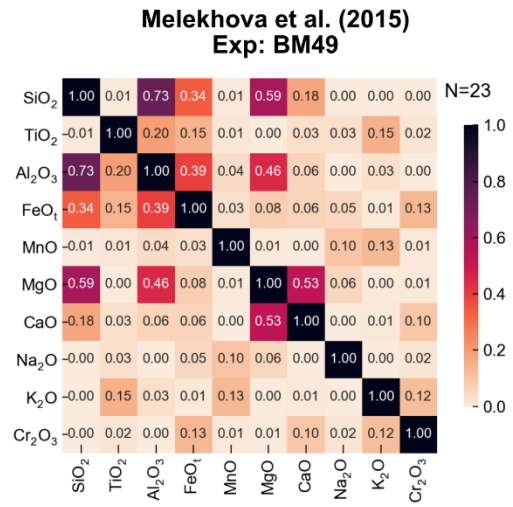
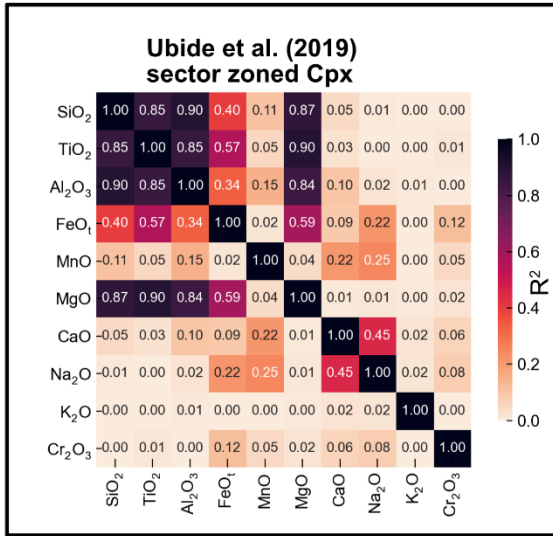
Supporting Fig. 6. Correlation matrix of pyroxenes from different experiments of Neave et al. (2019) with the color bar showing the R² value. The correlation matrix for the sector-zoned Cpx of Ubide et al. (2019) are shown for reference.



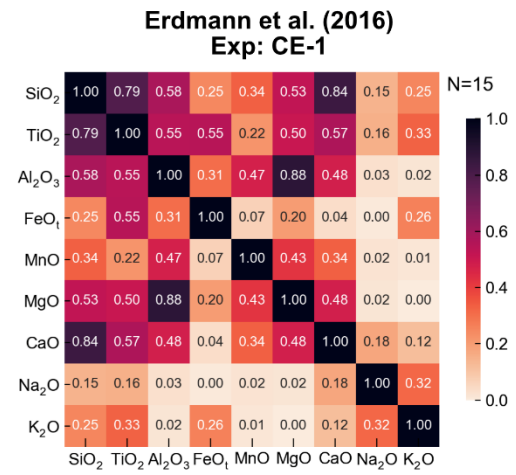
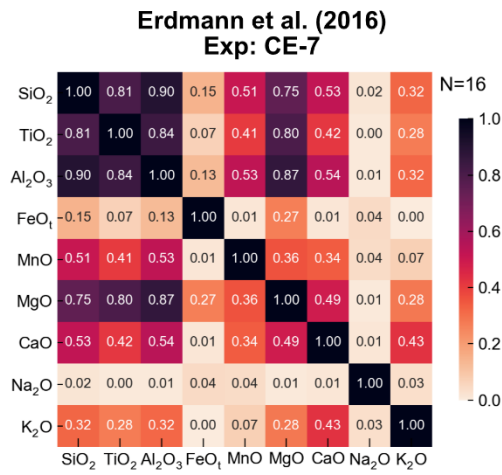
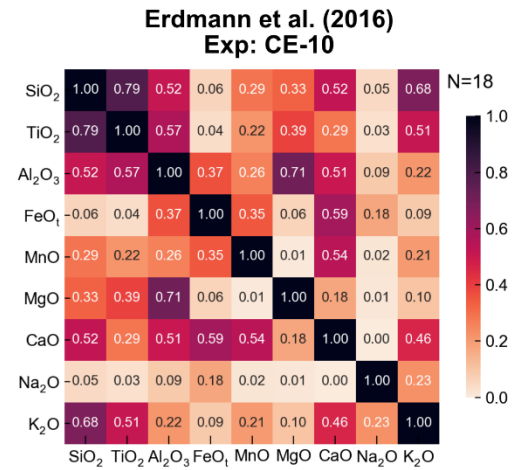
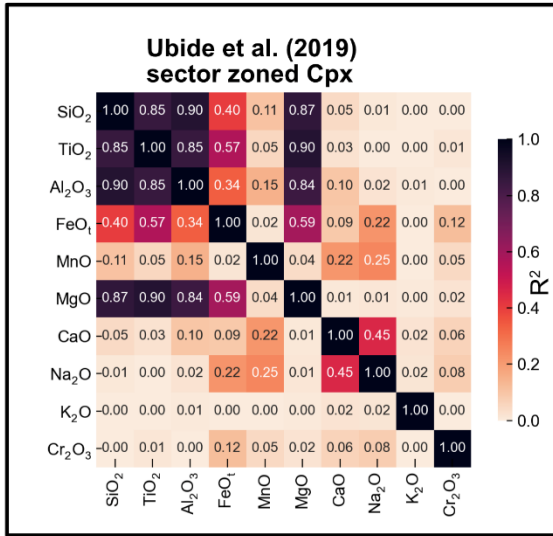
Supporting Fig 7 - Correlation matrix of pyroxenes from different experiments of Blatter et al. (2013) with the color bar showing the R² value. The correlation matrix for the sector-zoned Cpx of Ubide et al. (2019) are shown for reference.



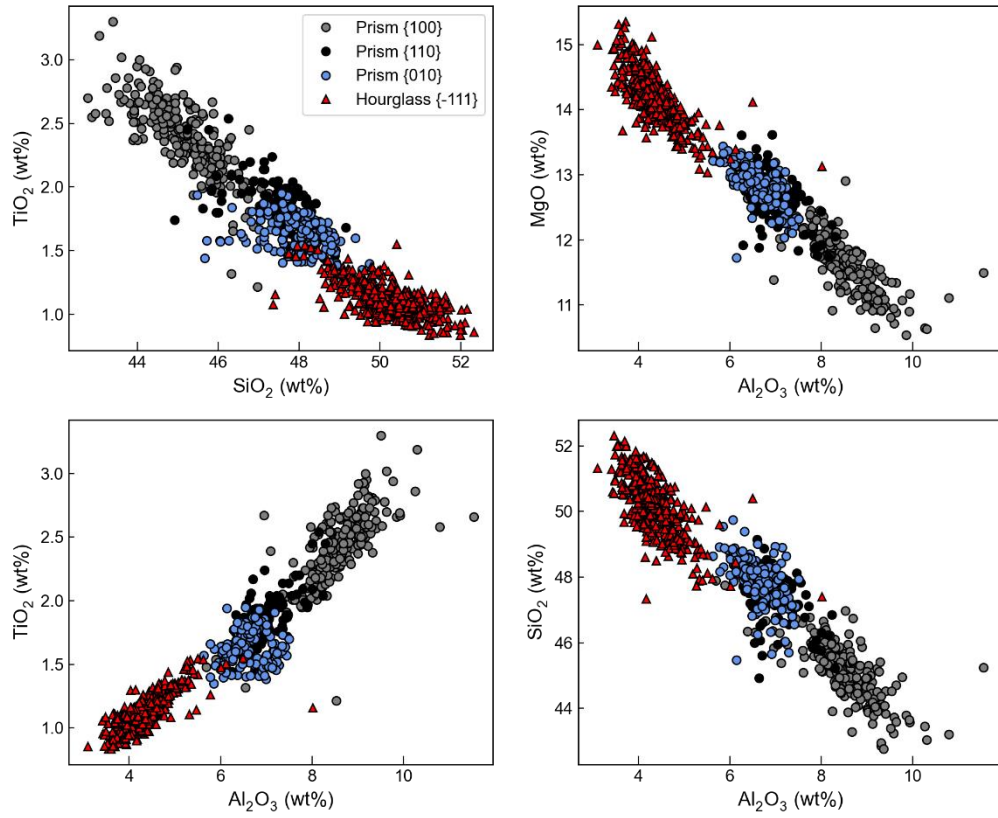
Supporting Fig 8 - Correlation matrix of pyroxenes from different experiments of Krawczynski et al. (2012) with the color bar showing the R² value. The correlation matrix for the sector-zoned Cpx of Ubide et al. (2019) are shown for reference.



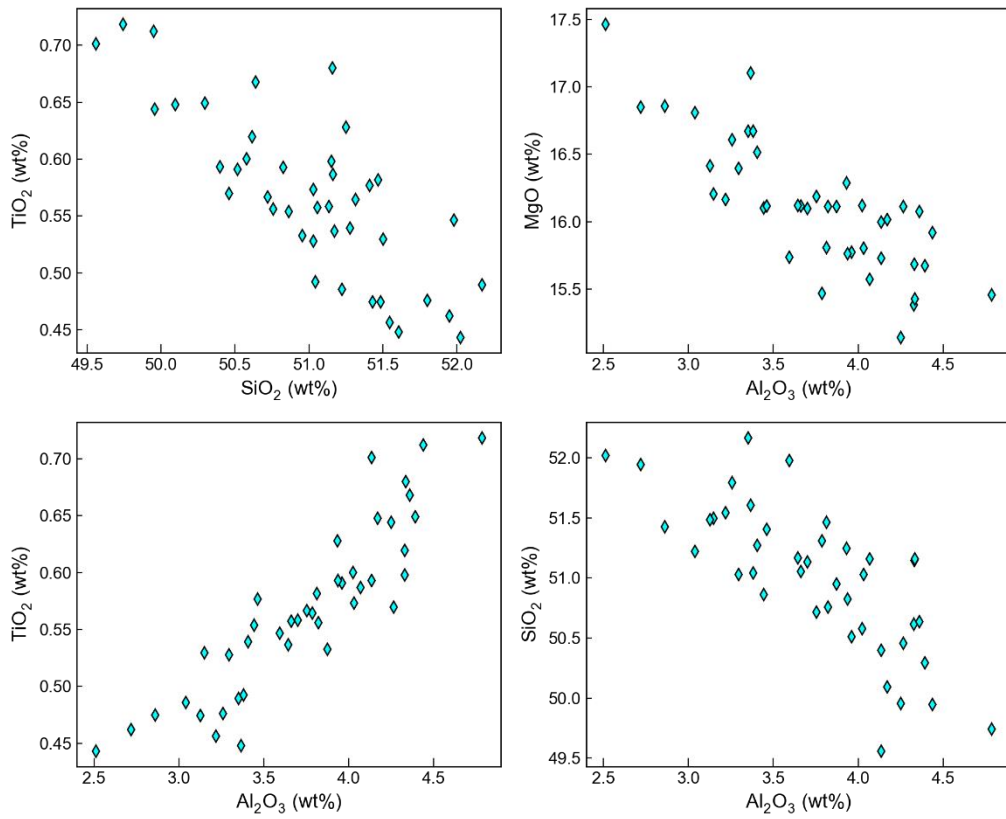
Supporting Fig 9 - Correlation matrix of pyroxenes from different experiments of Melekhova et al. (2015) with the color bar showing the R² value. The correlation matrix for the sector-zoned Cpx of Ubide et al. (2019) are shown for reference.



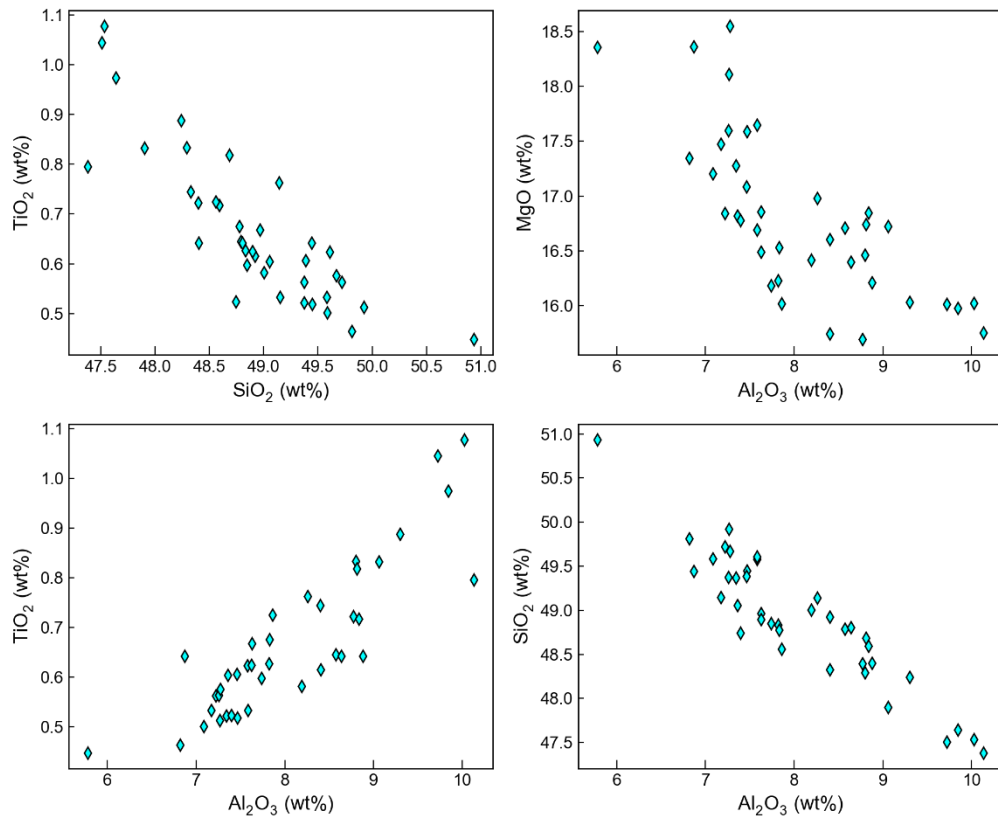
Supporting Fig 10 - Correlation matrix of pyroxenes from different experiments of Erdmann et al. (2016) with the color bar showing the R^2 value. The correlation matrix for the sector-zoned Cpx of Ubide et al. (2019) are shown for reference.



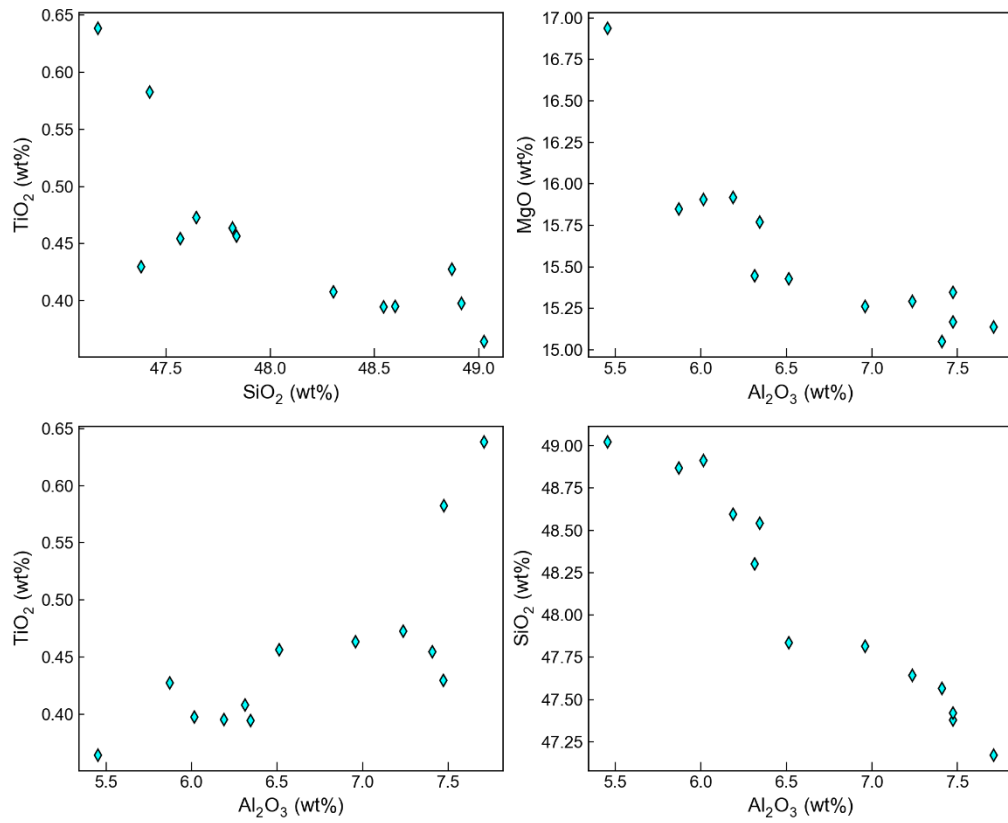
Supporting Fig. 11 : Major element correlations in sector zoned pyroxenes from Ubide et al. (2019), colored by sector.



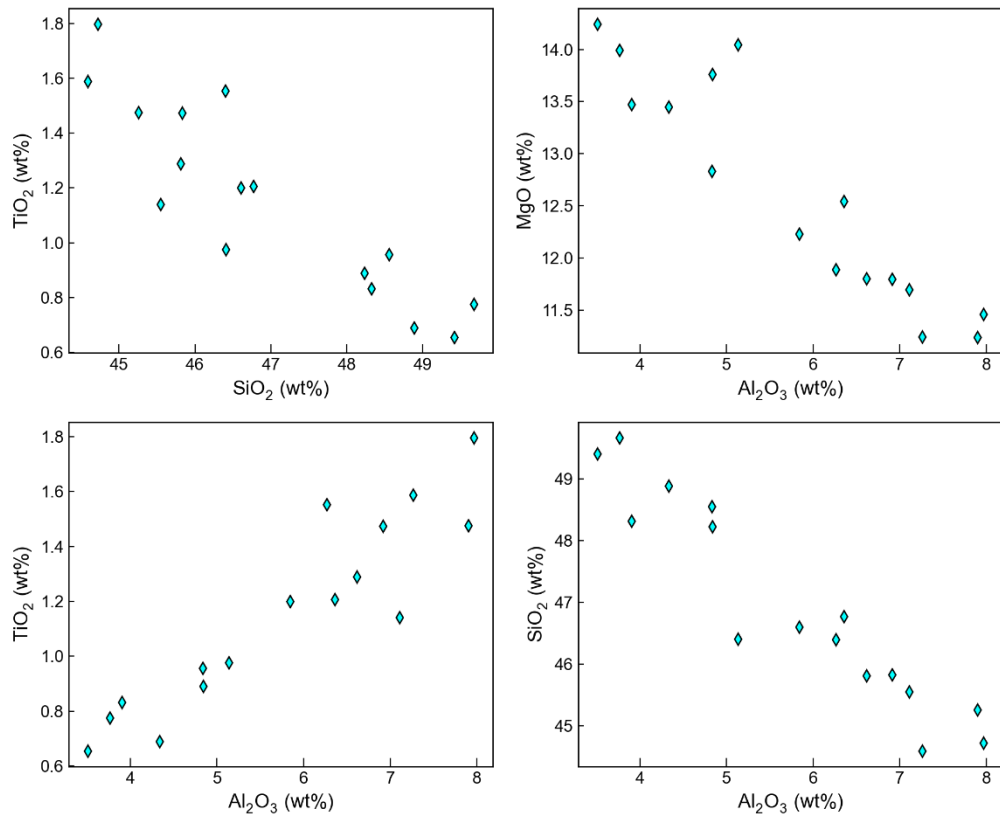
Supporting Figure 12 – Correlations between elements in Exp. Y0201-2 from Neave et al. (2017).



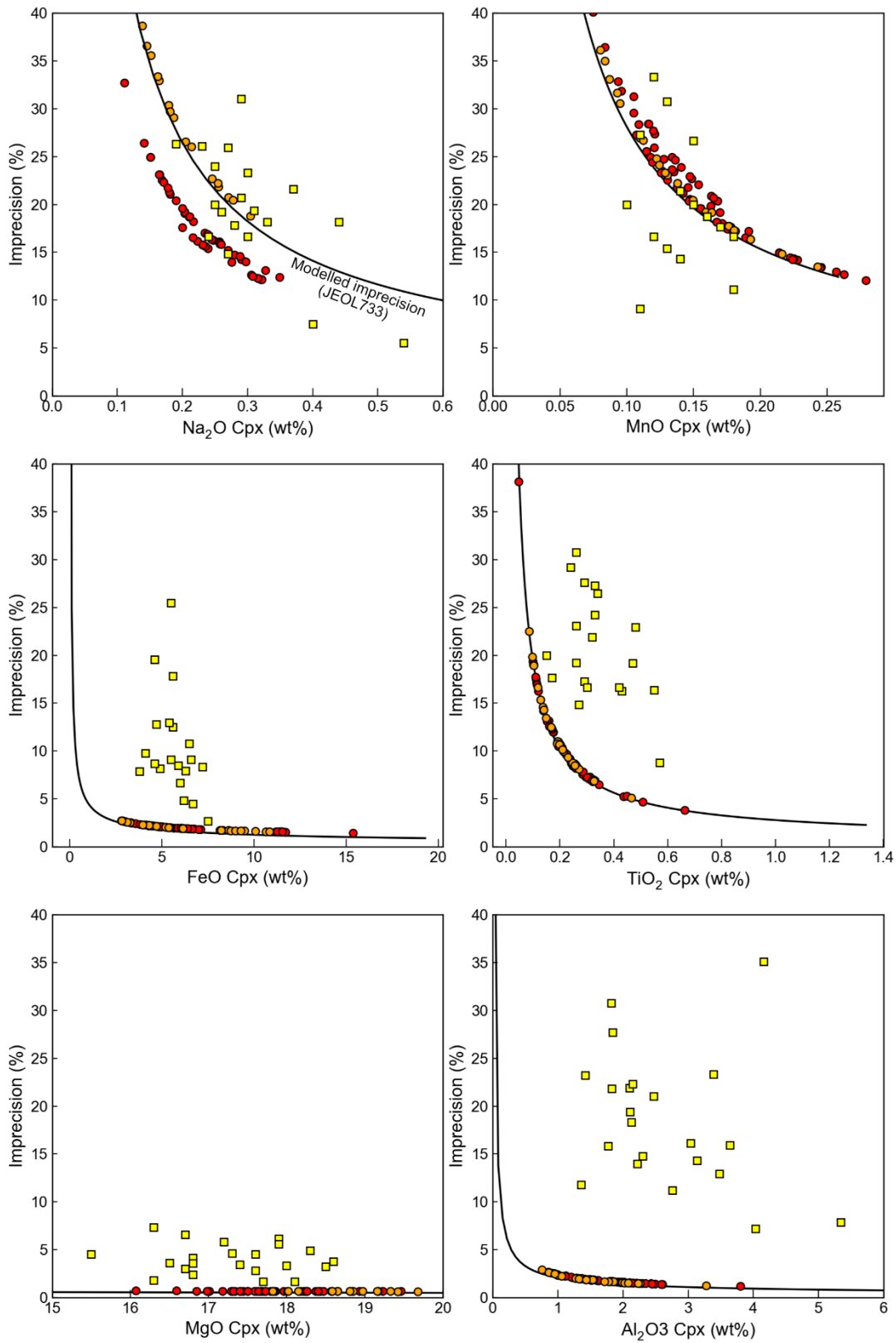
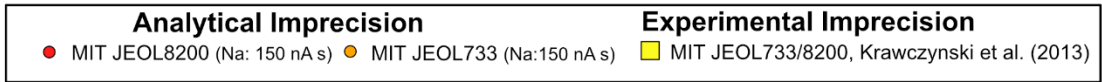
Supporting Figure 13 – Correlations between elements in Exp. 2362 from Blatter et al. (2013)



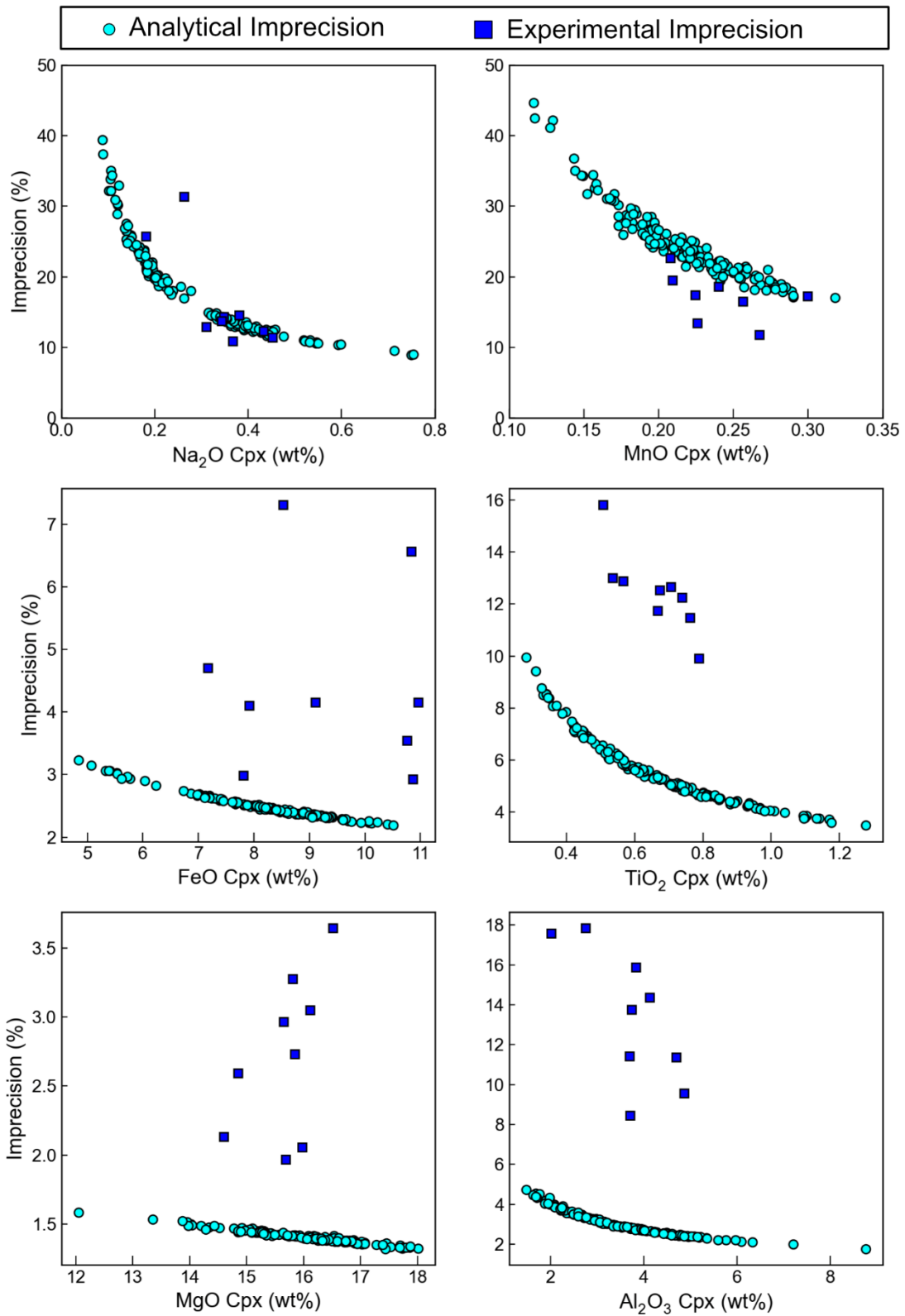
Supporting Figure 14 – Correlations between elements in Exp. BM40 from Melekhova et al. (2015)



Supporting Figure 15 – Correlations between elements in Exp. CE-7 from Erdmann et al. (2016).

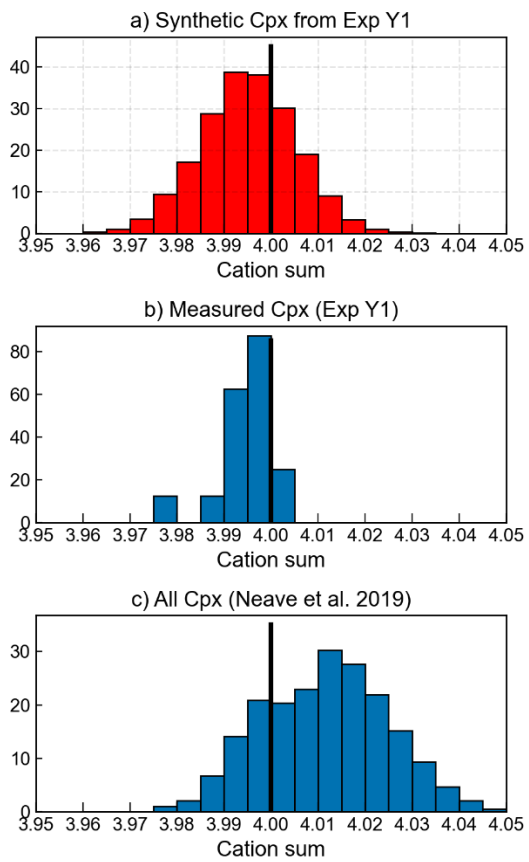


Supporting Fig. 16 – Figures comparing analytical imprecision (orange and red dots) to the variability observed in each experiment of Krawczynski et al. (2012- yellow squares).

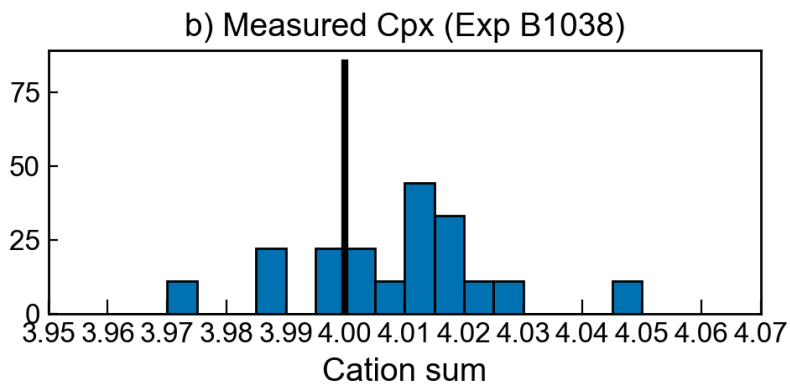
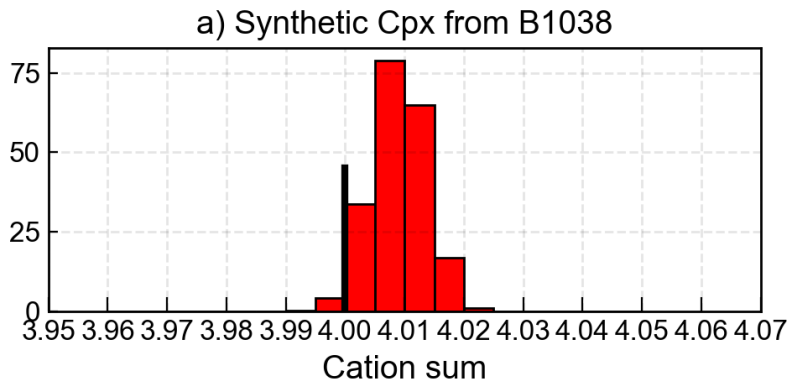


Supporting Fig. 17 - Figures comparing analytical imprecision (cyan dots) to the variability observed in each experiment of Neave et al. (2019; blue squares).

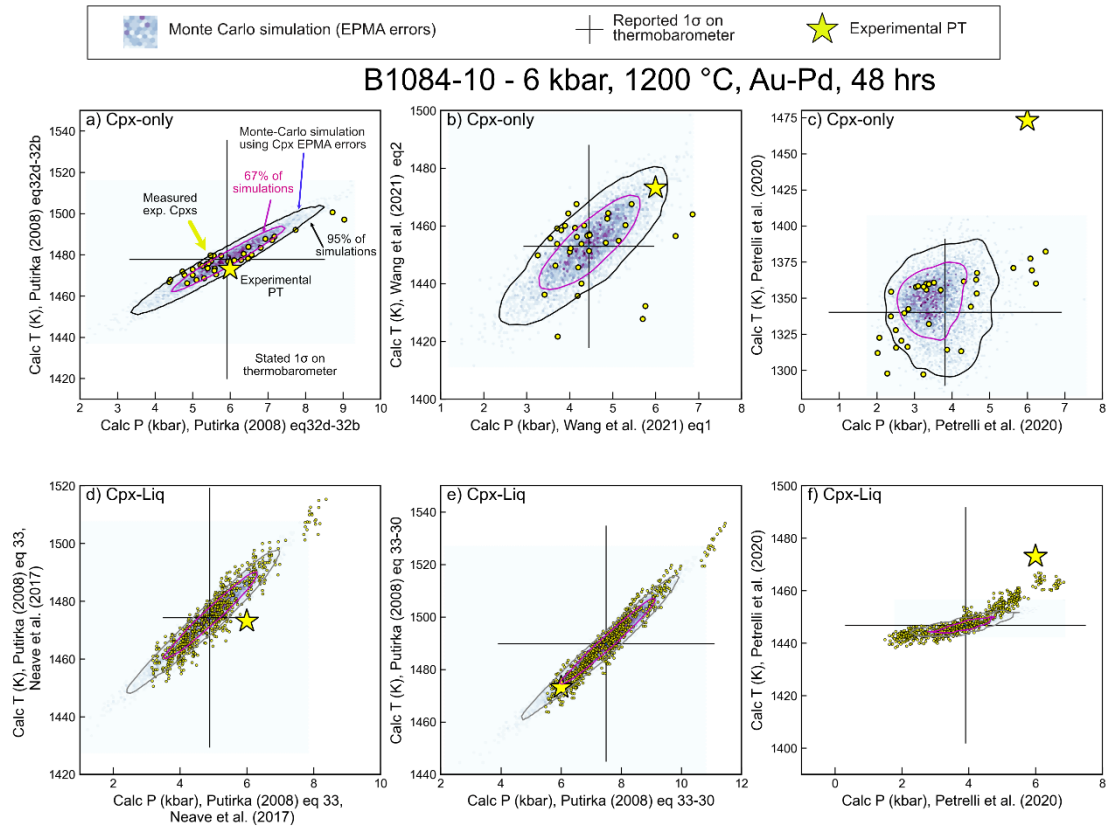
4. Additional Information regarding Monte Carlo Simulations



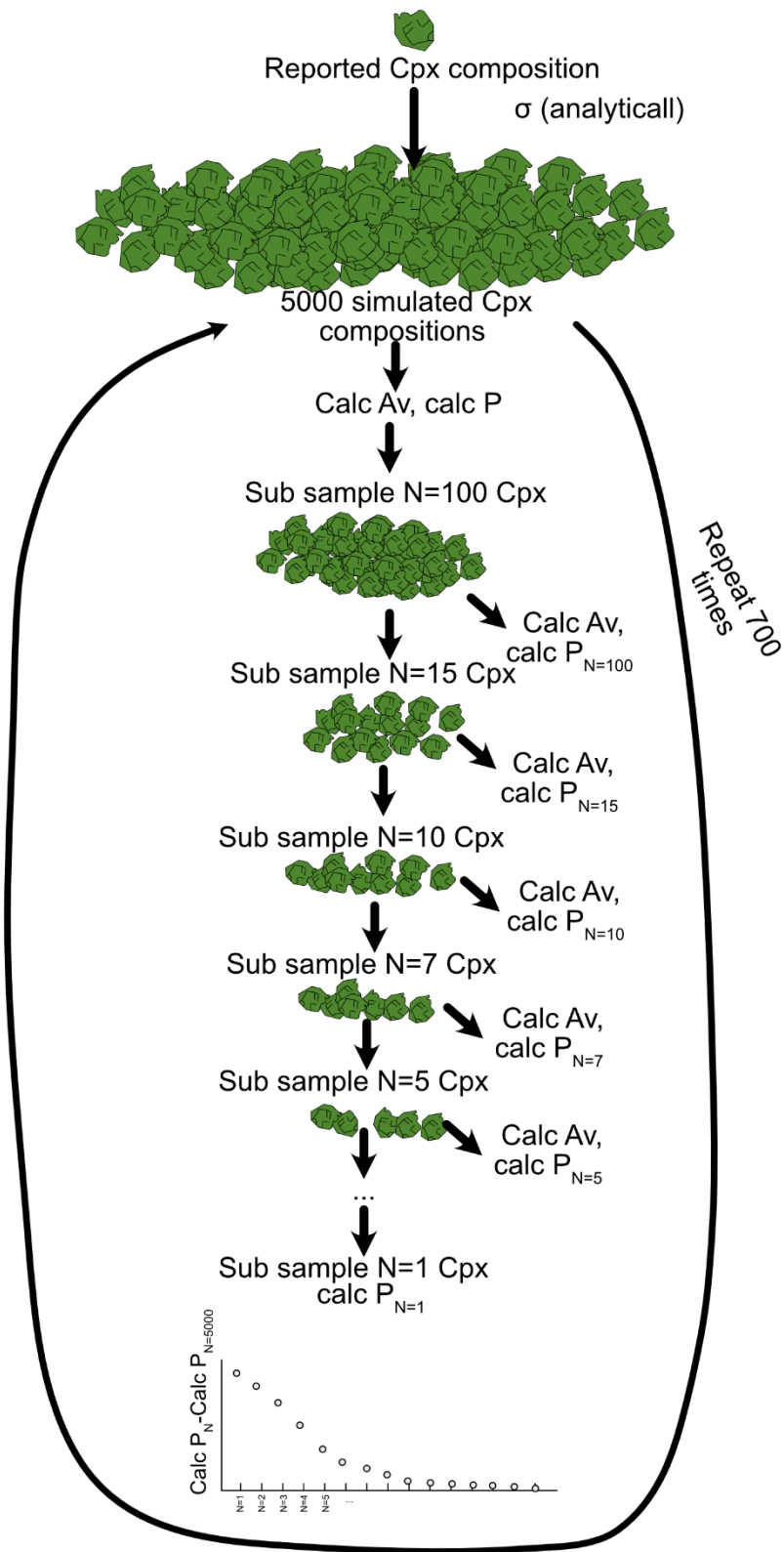
Supporting Fig. 18 – Distribution of cation sums in synthetic Cpx (part a) vs measured Cpx in that experiment, and measured Cpx in all experiments of Neave et al. (2019). Y axis shows probability density.



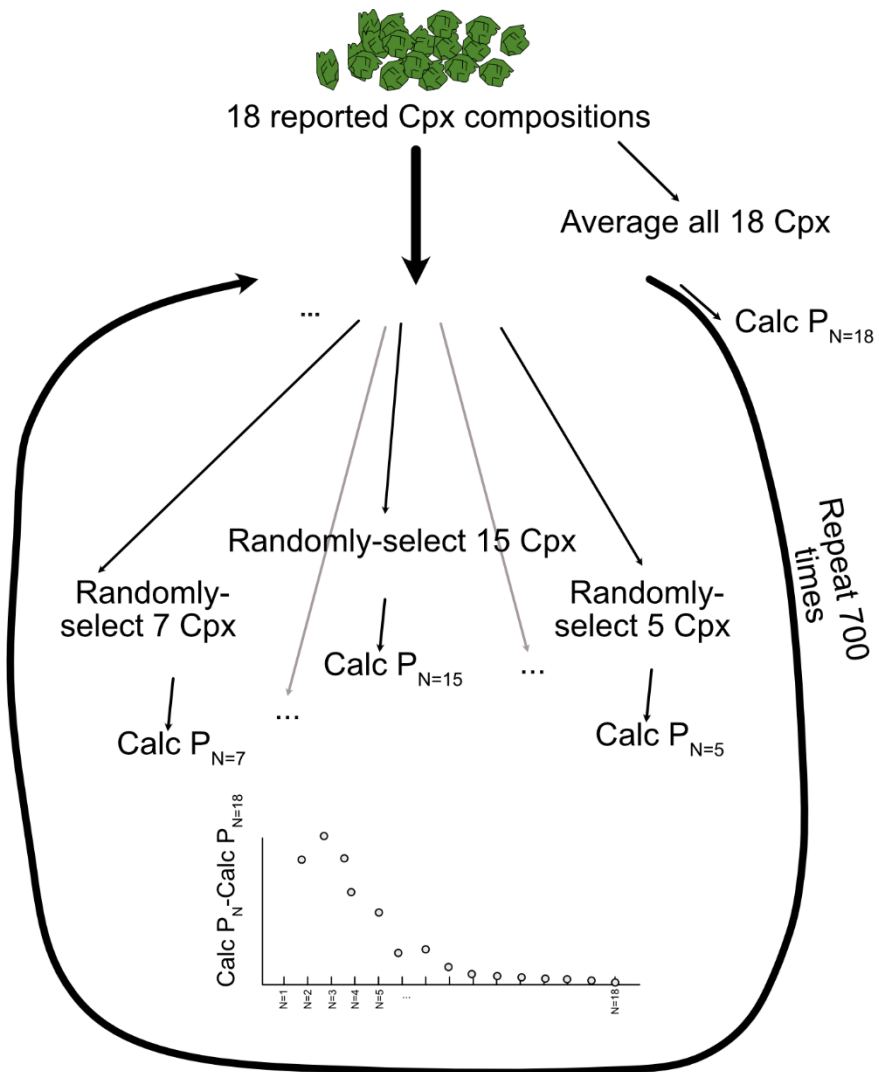
Supporting Fig. 19 – Cation sums for simulated Cpx from Krawczynski et al. (2012) compared to those in the experiment we are trying to simulate.



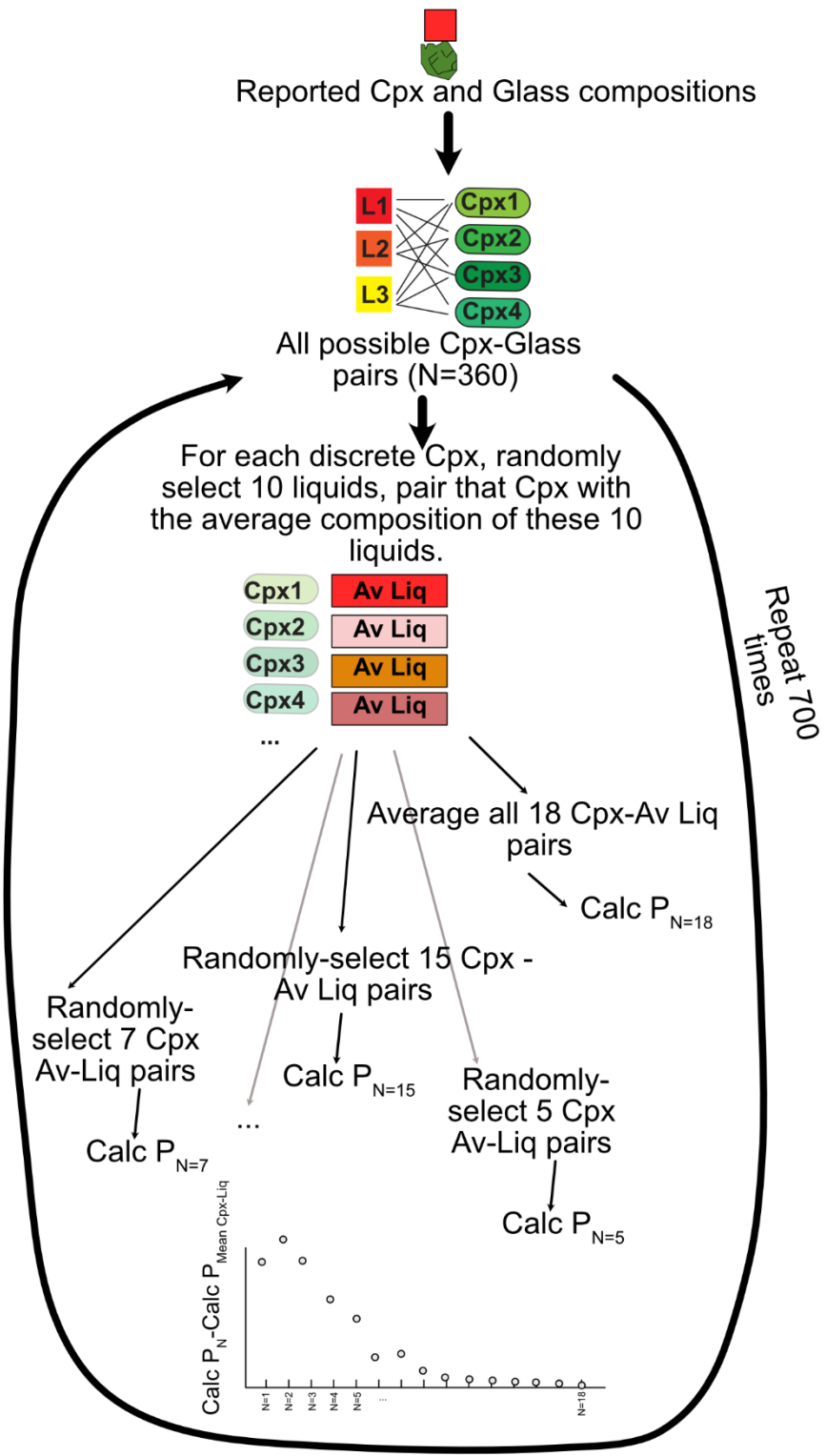
Supporting Figure 20 – As for Fig. 8 in the main text, but showing experiment B1084-10 from Neave et al. (2019).



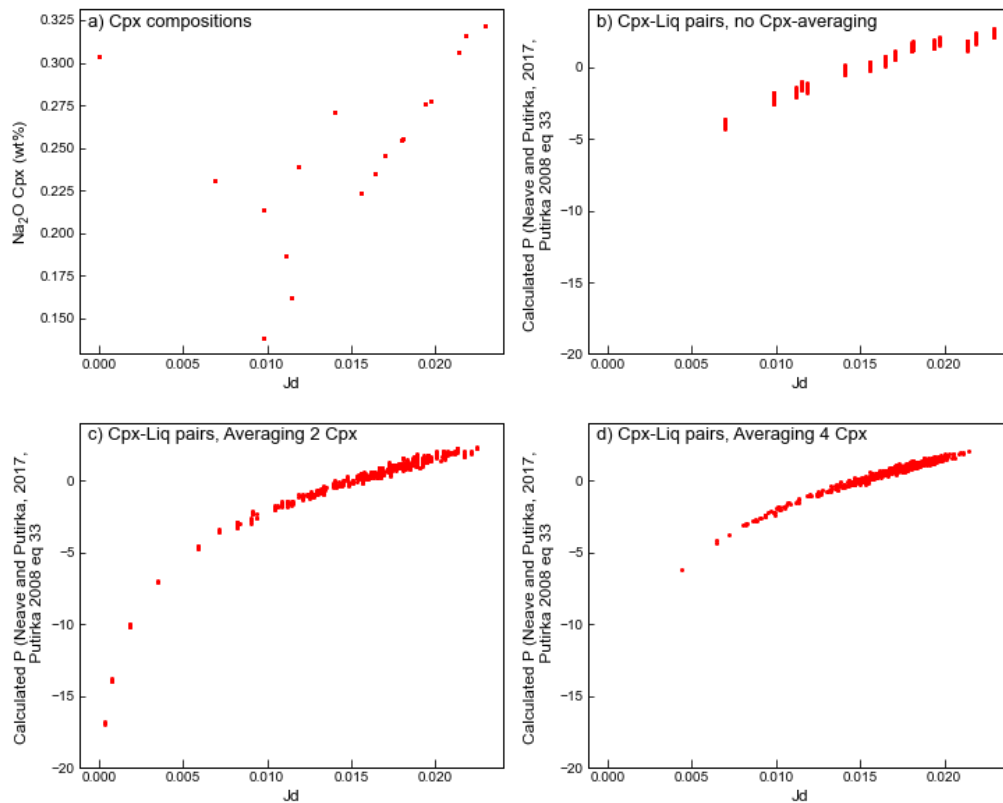
Supporting Fig. 21 – Schematic of the for-loop used to calculate the discrepancy in pressure as a function of the number of Cpx averaged for the Neave et al. (2017) experiment shown in Fig. 10a-b.



Supporting Fig 22– Schematic of the for loop used for subsampling the Cpx of Krawczynski et al. (2012) in Fig 10 c-d of the main text.



Supporting Fig 23– Schematic of the for loop used for subsampling the Cpx-Liq experiments of Krawczynski et al. (2012) in Fig 10 e-f of the main text.

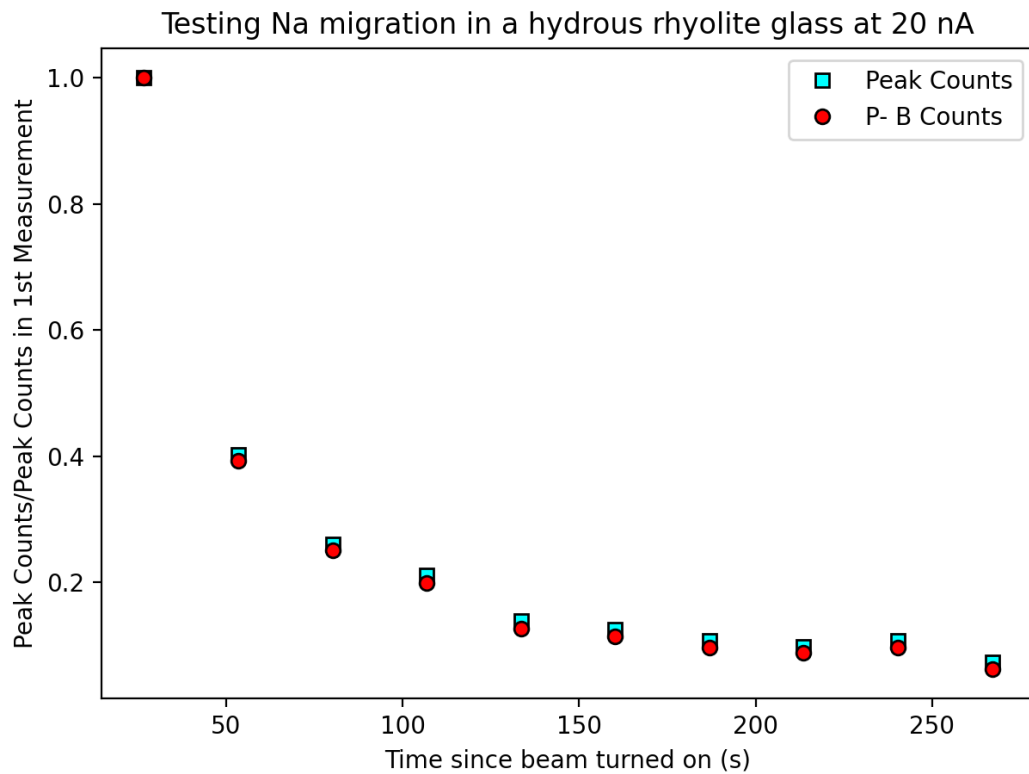


Supporting Fig. 24 –a) Cpx compositions from Experiment B1038 of Krawczynski et al. (2012). b) calculated Cpx-Liq pressures vs. Jd for measured Cpx matched with various liquid compositions. C- d) Cpx-Liq pressures averaging 2 and 4 Cpx. In c), a number of Cpx have very low, but non-zero Jd contents, resulting in strongly negative calculated pressures.

5. Investigating Na Migration in Pyroxene during EPMA analyses.

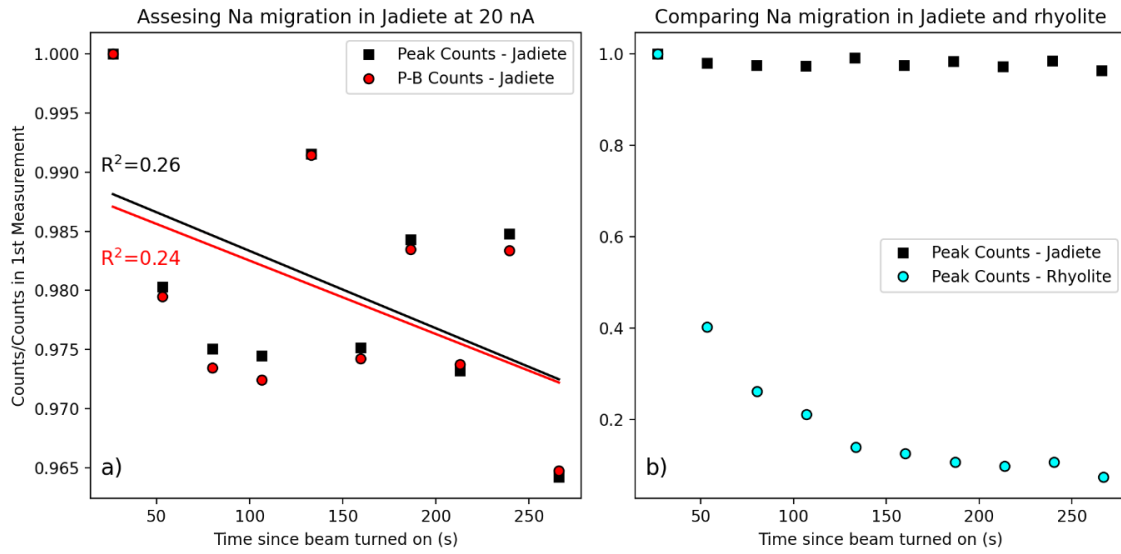
Analyses were performed on the SX100 at Oregon State at 15 kV with a beam size of 1 μm , using beam currents of 20, 40 and 100 nA. Na was analysed on the LTAP crystal, using Labradorite as a primary calibration standard (calibration performed at 20 nA).

We use three separate methods to track whether Na is migrating under the electron beam. First, we use the P-B-P-B “Subcounting” routine in the PeakSight software. The software splits the acquisition time of any given elements into 10 windows. Then, it measures the Peak counts, Background1 counts, Background 2 counts for each the first measurement. It then cycles through this for as many N as are specified. To validate this method, we first track the change in Na counts within a hydrous rhyolitic glass (3 wt% H₂O), which should show extensive migration. We plot the change in peak counts relative to the first measurement. Peak and P-B counts decline rapidly, to values only 0.4X the original after 50s of beam exposure, with a tail off to very low Na counts after 100s (Supporting Fig. 22).

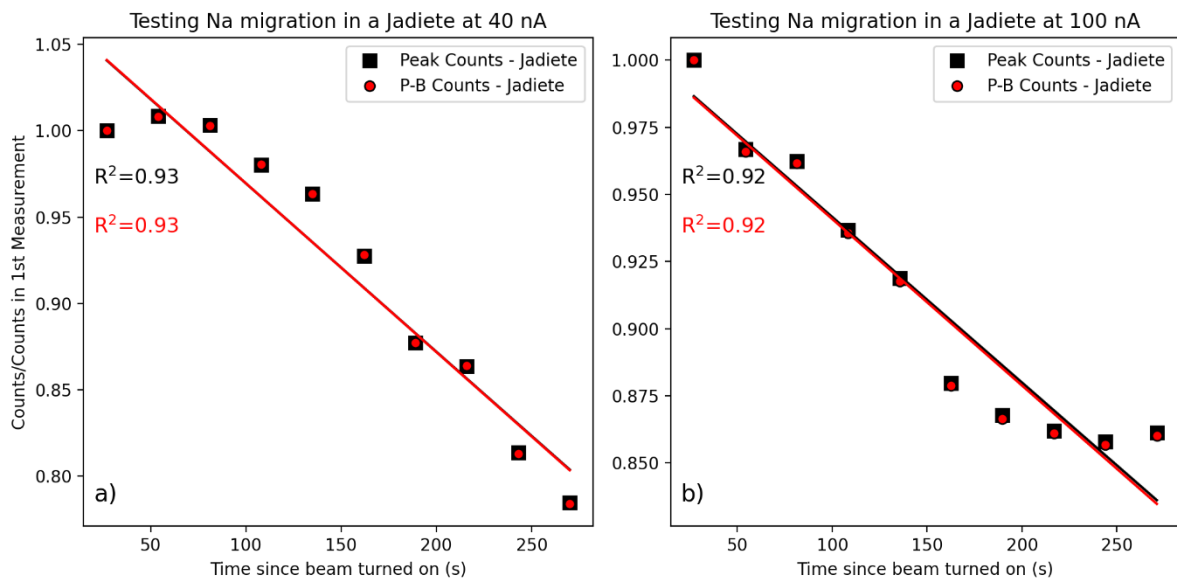


Supporting Figure 25 – Tracking Na counts in a hydrous rhyolite glass using the sub-counting P-B-P routine at 20 nA over >250 s.

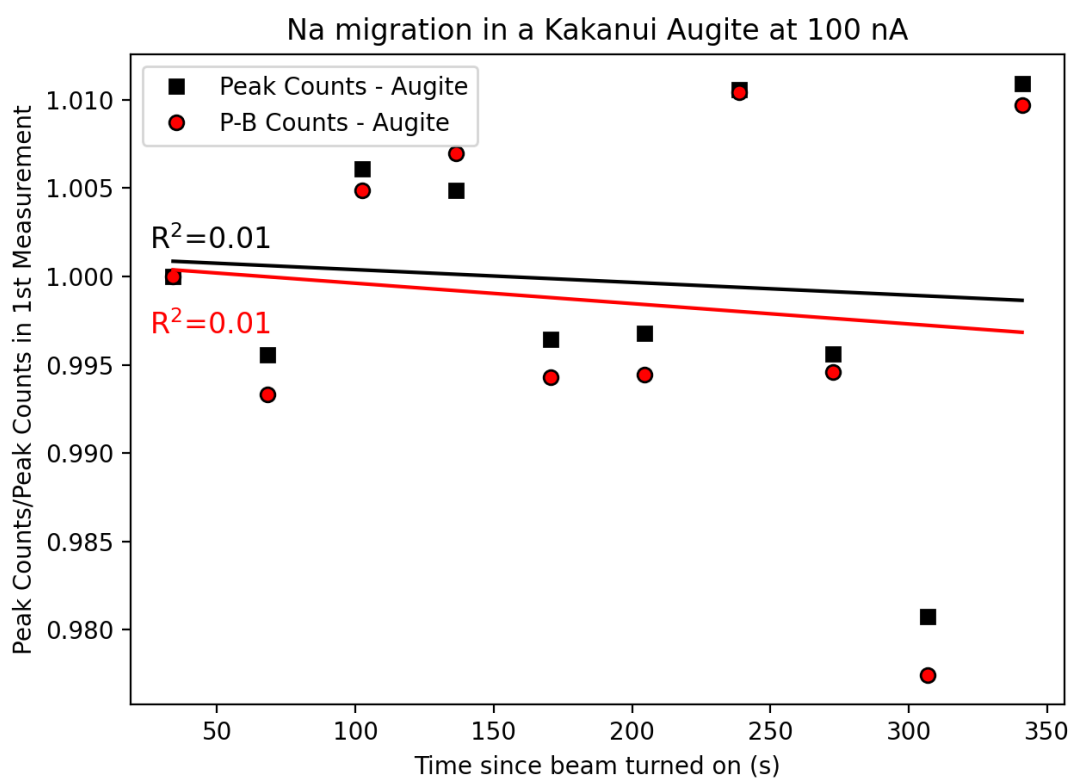
In contrast, when we perform the same P-B-P routine on a Jadeite standard, counts only decline by ~4% over 250 s at 20 nA (Supporting Figure 23a), with a small r^2 value (compare to rhyolite in Supporting Figure 23b). At higher currents (40 and 100nA) counts decline by 15-20% over 250 s (Supporting Figure 24). In contrast, even at 100 nA, Na Peak counts, and Peak-Background counts show no noticeable change with time in natural augite (Kakanui Augite, 1.1 wt% Na, Supporting Figure 25). These subcounting routines demonstrate that while glasses are highly beam sensitive and undergone substantial beam loss, Jadeite is only very slightly beam sensitive at 20nA over prolonged count periods, and only shows strong correlations between peak counts and time at 40 nA and 100 nA. In contrast, even for prolonged count times (>250s) and very high probe currents (100 nA), natural pyroxenes undergo no noticeable migration of Na.



Supporting Figure 26 – a) Tracking peak and peak – background (P-B) counts in Jadeite vs. time at 20 nA. b) The changes are entirely overwhelmed by those seen in rhyolites.



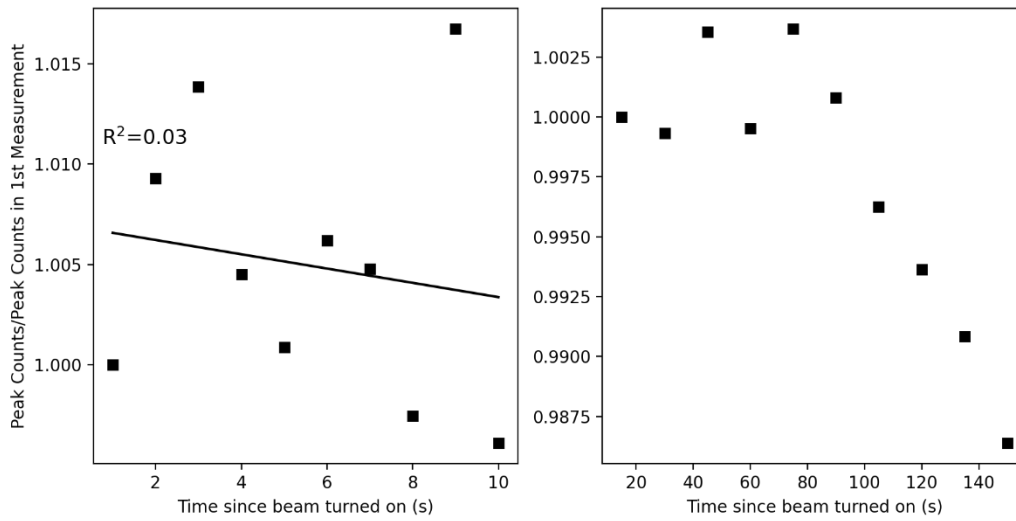
Supporting Figure 27 – Tracking peak and peak – background (P-B) counts in Jadeite vs. time at 40 nA (a) and 100 nA (b).



Supporting Figure 28 – Tracking changes in Na counts with time in Kakanui Augite at 100 nA.

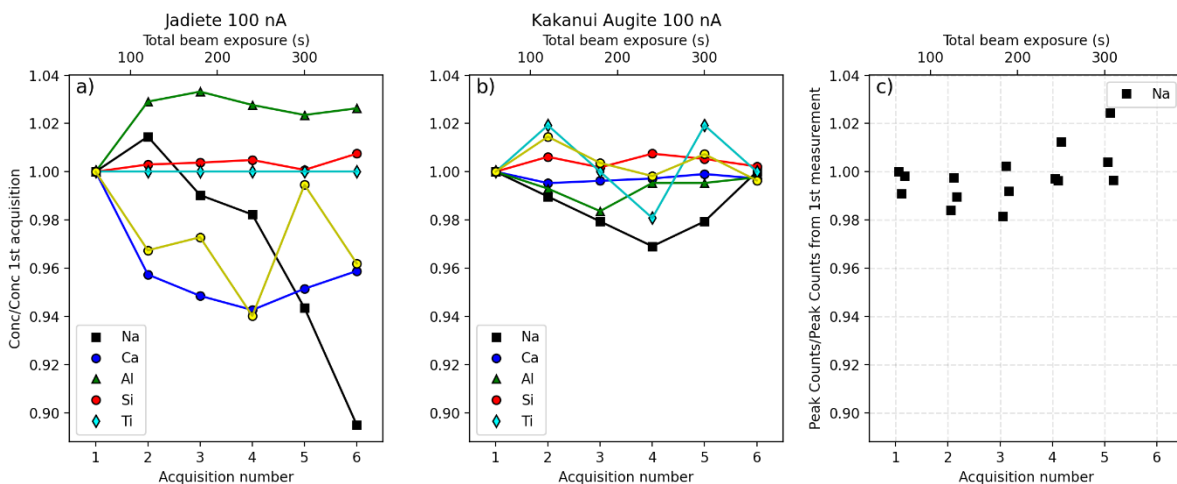
Beam sensitivity of Jadeite is important in that it is often used as a primary calibration standard for Na in pyroxene. Given that P-B-P routines are missing gaps in time as the spectrometer position is constantly being adjusted, we also perform a routine where the Na peak is measured in short increments, without measuring backgrounds (as Supporting Figs. 22-25 show that Peak-counts and P-B counts show very similar trends). First, we count on the peak ten times for 1-second intervals and then count on the peak ten times over 15 s intervals (Supporting Figure 26). These graphs demonstrates that, even at 100 nA, Jadeite only begins to lose Na after 80s of beam exposure. This indicates that at common calibration conditions (e.g., 20-40 nA, 10-30s), Na migration in Jadeite is not a major problem.

Peak-only, Na migration in Jadeite at 100 nA



Supporting Figure 29 – Tracking Peak counts on Na over 10s and 150s (in 10 discrete measurement intervals).

We perform one additional set of tests at 100 nA to investigate changes in elemental concentrations where we perform a relatively short (1 minute total beam exposure) analytical routine for Na-Ca-Al-Si-Ti on a single point on Jadeite, and obtain a quantitative analysis for this point (in terms of wt%, Supporting Figure 27a). We then perform a repeat analysis without moving the stage. We compare element concentrations from 6 repeated measurements to those measured in the first spot. For Jadeite, the only element showing noticeable unidirectional changes is Na, declining by ~10% relative after 6 minutes of beam exposure. For Kakanui augite, we use the same routine, but also split each measurement of the Na peak into 3 P-P-P subcount routines (3 seconds each), and apply the zero-time-intersection correction in the software. None of the 6 acquisitions show any coherent trends between Na counts and time (Supporting Figure 27c) so very similar results would have been achieved without using a zero time intercept.



Supporting Figure 30 – a) Tracking changes in elemental concentrations during 6 repeated analyses (60s of beam exposure each) in the same stage position on Jadeite, with no zero-time-intercept correction. b) Same for Kakanui augite, but using a zero-time-intercept correction for Na

based on 3 subcounts on the Na Peak during each acquisition. In c), the counts for these are shown. There are no coherent changes with time above the noise of the measurement.

Additional References

- Akella, J., 1976. Garnet pyroxene equilibria in the system CaSiO₃-MgSiO₃-Al₂O₃ and in a natural mineral mixture. *American Mineralogist*, V61, pp 589-598
- Almeev, R.R., Holtz, F., Ariskin, A.A., Kimura, J.-I., 2013. Storage conditions of Bezymianny Volcano parental magmas: results of phase equilibria experiments at 100 and 700 MPa. *Contrib Mineral Petrol* 166, 1389–1414. <https://doi.org/10.1007/s00410-013-0934-x>
- Alonso-Perez, R., Müntener, O., Ulmer, P., 2009. Igneous garnet and amphibole fractionation in the roots of island arcs: experimental constraints on andesitic liquids. *Contrib Mineral Petrol* 157, 541–558. <https://doi.org/10.1007/s00410-008-0351-8>
- Andújar, J., Scaillet, B., Pichavant, M., Druitt, T.H., 2015. Differentiation Conditions of a Basaltic Magma from Santorini, and its Bearing on the Production of Andesite in Arc Settings. *Journal of Petrology* 56, 765–794. <https://doi.org/10.1093/petrology/egv016>
- Baker, D.R., Egger, D.H., 1987. Compositions of anhydrous and hydrous melts coexisting with plagioclase, augite, and olivine or low-Ca pyroxene from 1 atm to 8 kbar: Application to the Aleutian volcanic center of Atka. *American Mineralogist* 72.
- Baker, M.B., Grove, T.L., Price, R., 1994. Primitive basalts and andesites from the Mt. Shasta region, N. California: products of varying melt fraction and water content. *Contrib. Mineral. and Petrol.* 118, 111–129. <https://doi.org/10.1007/BF01052863>
- Barclay, J., 2004. A Hornblende Basalt from Western Mexico: Water-saturated Phase Relations Constrain a Pressure-Temperature Window of Eruptibility. *Journal of Petrology* 45, 485–506. <https://doi.org/10.1093/petrology/egg091>
- Berndt, J., 2004. An Experimental Investigation of the Influence of Water and Oxygen Fugacity on Differentiation of MORB at 200 MPa. *Journal of Petrology* 46, 135–167. <https://doi.org/10.1093/petrology/egh066>
- Berndt, J., Holtz, F., Koepke, J., 2001. Experimental constraints on storage conditions in the chemically zoned phonolitic magma chamber of the Laacher See volcano. *Contrib Mineral Petrol* 140, 469–486. <https://doi.org/10.1007/PL00007674>
- Blatter, D.L., Carmichael, I.S.E., 2001. Hydrous phase equilibria of a Mexican high-silica andesite: A candidate for a mantle origin? *Geochimica et Cosmochimica Acta* 65, 4043–4065. [https://doi.org/10.1016/S0016-7037\(01\)00708-6](https://doi.org/10.1016/S0016-7037(01)00708-6)

- Blatter, D.L., Sisson, T.W., Hanks, W.B., 2013. Crystallization of oxidized, moderately hydrous arc basalt at mid- to lower-crustal pressures: implications for andesite genesis. *Contrib Mineral Petrol* 166, 861–886. <https://doi.org/10.1007/s00410-013-0920-3>
- Blundy, J.D., Robinson, J.A.C., Wood, B.J., 1998. Heavy REE are compatible in clinopyroxene on the spinel lherzolite solidus. *Earth and Planetary Science Letters* 160, 493–504. [https://doi.org/10.1016/S0012-821X\(98\)00106-X](https://doi.org/10.1016/S0012-821X(98)00106-X)
- Cadoux, A., Scaillet, B., Druitt, T.H., Deloule, E., 2014. Magma Storage Conditions of Large Plinian Eruptions of Santorini Volcano (Greece). *Journal of Petrology* 55, 1129–1171. <https://doi.org/10.1093/petrology/egu021>
- Carroll, M.R., Wyllie, P.J., 1989. Experimental Phase Relations in the System Tonalite-Peridotite-H₂O at 15 kb; Implications for Assimilation and Differentiation Processes near the Crust-Mantle Boundary. *Journal of Petrology* 30, 1351–1382. <https://doi.org/10.1093/petrology/30.6.1351>
- Costa, F., 2004. Petrological and Experimental Constraints on the Pre-eruption Conditions of Holocene Dacite from Volcan San Pedro (36 S, Chilean Andes) and the Importance of Sulphur in Silicic Subduction-related Magmas. *Journal of Petrology* 45, 855–881. <https://doi.org/10.1093/petrology/egg114>
- Di Carlo, I., 2006. Experimental Crystallization of a High-K Arc Basalt: the Golden Pumice, Stromboli Volcano (Italy). *Journal of Petrology* 47, 1317–1343. <https://doi.org/10.1093/petrology/egl011>
- Draper, D.S., Green, T.H., 1999. P–T phase relations of silicic, alkaline, aluminous liquids: new results and applications to mantle melting and metasomatism. *Earth and Planetary Science Letters* 170, 255–268. [https://doi.org/10.1016/S0012-821X\(99\)00111-9](https://doi.org/10.1016/S0012-821X(99)00111-9)
- Draper, D.S., Johnston, A.D., 1992. Anhydrous PT phase relations of an Aleutian high-MgO basalt: an investigation of the role of olivine-liquid reaction in the generation of arc high-alumina basalts. *Contr. Mineral. and Petrol.* 112, 501–519. <https://doi.org/10.1007/BF00310781>
- Erdmann, S., Martel, C., Pichavant, M., Bourdier, J.-L., Champallier, R., Komorowski, J.-C., Cholik, N., 2016. Constraints from Phase Equilibrium Experiments on Pre-eruptive Storage Conditions in Mixed Magma Systems: a Case Study on Crystal-rich Basaltic Andesites from Mount Merapi, Indonesia. *J. Petrology* 57, 535–560. <https://doi.org/10.1093/petrology/egw019>
- Feig, S.T., Koepke, J., Snow, J.E., 2010. Effect of oxygen fugacity and water on phase equilibria of a hydrous tholeiitic basalt. *Contrib Mineral Petrol* 160, 551–568. <https://doi.org/10.1007/s00410-010-0493-3>
- Feig, S.T., Koepke, J., Snow, J.E., 2006. Effect of water on tholeiitic basalt phase equilibria: an experimental study under oxidizing conditions. *Contrib Mineral Petrol* 152, 611–638. <https://doi.org/10.1007/s00410-006-0123-2>

- Firth, C., Adam, J., Turner, S., Rushmer, T., Brens, R., Green, T.H., Erdmann, S., O'Neill, H., 2019. Experimental constraints on the differentiation of low-alkali magmas beneath the Tonga arc: Implications for the origin of arc tholeiites. *Lithos* 344–345, 440–451. <https://doi.org/10.1016/j.lithos.2019.07.008>
- Fram, M., Longhi, John, 1992. Phase equilibria of dikes associated with Proterozoic anorthosite complexes. *American Mineralogist*.v77, 605–616
- Gee, L.L., Sack, R.O., 1988. Experimental Petrology of Melilite Nephelinites. *Journal of Petrology* 29, 1233–1255. <https://doi.org/10.1093/petrology/29.6.1233>
- Hamada, M., Fujii, T., 2008. Experimental constraints on the effects of pressure and H₂O on the fractional crystallization of high-Mg island arc basalt. *Contrib Mineral Petrol* 155, 767–790. <https://doi.org/10.1007/s00410-007-0269-6>
- Husen, A., Almeev, R.R., Holtz, F., 2016. The Effect of H₂O and Pressure on Multiple Saturation and Liquid Lines of Descent in Basalt from the Shatsky Rise. *Journal of Petrology* 57, 309–344. <https://doi.org/10.1093/petrology/egw008>
- Kawamoto, T., 1996. Experimental constraints on differentiation and H₂O abundance of calc-alkaline magmas. *Earth and Planetary Science Letters* 144, 577–589. [https://doi.org/10.1016/S0012-821X\(96\)00182-3](https://doi.org/10.1016/S0012-821X(96)00182-3)
- Kennedy, A.K., Grove, T.L., Johnson, R.W., 1990. Experimental and major element constraints on the evolution of lavas from Lihir Island, Papua New Guinea. *Contr. Mineral. and Petrol.* 104, 722–734. <https://doi.org/10.1007/BF01167289>
- Koepke, J., Botcharnikov, R.E., Natland, J.H., 2018. Crystallization of late-stage MORB under varying water activities and redox conditions: Implications for the formation of highly evolved lavas and oxide gabbro in the ocean crust. *Lithos* 323, 58–77. <https://doi.org/10.1016/j.lithos.2018.10.001>
- Krawczynski, M.J., Grove, T.L., Behrens, H., 2012. Amphibole stability in primitive arc magmas: effects of temperature, H₂O content, and oxygen fugacity. *Contrib Mineral Petrol* 164, 317–339. <https://doi.org/10.1007/s00410-012-0740-x>
- Laporte, D., Toplis, M.J., Seyler, M., Devidal, J.-L., 2004. A new experimental technique for extracting liquids from peridotite at very low degrees of melting: application to partial melting of depleted peridotite. *Contributions to Mineralogy and Petrology* 146, 463–484. <https://doi.org/10.1007/s00410-003-0509-3>
- Maaloe, S., 2004. The PT-phase relations of an MgO-rich Hawaiian tholeiite: the compositions of primary Hawaiian tholeiites. *Contrib Mineral Petrol* 148, 236–246. <https://doi.org/10.1007/s00410-004-0601-3>

- Mandler, B.E., Donnelly-Nolan, J.M., Grove, T.L., 2014. Straddling the tholeiitic/calc-alkaline transition: the effects of modest amounts of water on magmatic differentiation at Newberry Volcano, Oregon. *Contrib Mineral Petrol* 168, 1066. <https://doi.org/10.1007/s00410-014-1066-7>
- McCoy, T.J., Lofgren, G.E., 1999. Crystallization of the Zagami shergottite: an experimental study. *Earth and Planetary Science Letters* 173, 397–411. [https://doi.org/10.1016/S0012-821X\(99\)00241-1](https://doi.org/10.1016/S0012-821X(99)00241-1)
- Melekhova, E., Blundy, J., Robertson, R., Humphreys, M.C.S., 2015. Experimental Evidence for Polybaric Differentiation of Primitive Arc Basalt beneath St. Vincent, Lesser Antilles. *Journal of Petrology* 56, 161–192. <https://doi.org/10.1093/petrology/egu074>
- Métrich, N., Rutherford, M.J., 1998. Low Pressure Crystallization Paths of H₂O-Saturated Basaltic-Hawaiitic Melts from Mt Etna: Implications for Open-System Degassing of Basaltic Volcanoes. *Geochimica et Cosmochimica Acta* 62, 1195–1205. [https://doi.org/10.1016/S0016-7037\(98\)00048-9](https://doi.org/10.1016/S0016-7037(98)00048-9)
- Minitti, M.E., Rutherford, M.J., 2000. Genesis of the Mars Pathfinder ?sulfur-free? rock from SNC parental liquids. *Geochimica et Cosmochimica Acta* 64, 2535–2547. [https://doi.org/10.1016/S0016-7037\(00\)00366-5](https://doi.org/10.1016/S0016-7037(00)00366-5)
- Nandedkar, R.H., Ulmer, P., Müntener, O., 2014. Fractional crystallization of primitive, hydrous arc magmas: an experimental study at 0.7 GPa. *Contrib Mineral Petrol* 167, 1015. <https://doi.org/10.1007/s00410-014-1015-5>
- Neave, D.A., Bali, E., Guðfinnsson, G.H., Halldórsson, S.A., Kahl, M., Schmidt, A.-S., Holtz, F., 2019. Clinopyroxene–Liquid Equilibria and Geothermobarometry in Natural and Experimental Tholeiites: the 2014–2015 Holuhraun Eruption, Iceland. *Journal of Petrology* 60, 1653–1680. <https://doi.org/10.1093/petrology/egz042>
- Nielsen, R.L., Gallahan, W.E., Newberger, F., 1992. Experimentally determined mineral-melt partition coefficients for Sc, Y and REE for olivine, orthopyroxene, pigeonite, magnetite and ilmenite. *Contrib Mineral Petrol* 110, 488–499. <https://doi.org/10.1007/BF00344083>
- Parat, F., Streck, M., Holtz, F., Almeev, R.R., 2014. Experimental study into the petrogenesis of crystal-rich basaltic to andesitic magmas at Arenal volcano. *Contributions to Mineralogy and Petrology*, v168. <https://doi.org/10.1007/s00410-014-1040-4>
- Pertermann, M., 2003. Anhydrous Partial Melting Experiments on MORB-like Eclogite: Phase Relations, Phase Compositions and Mineral-Melt Partitioning of Major Elements at 2-3 GPa. *Journal of Petrology* 44, 2173–2201. <https://doi.org/10.1093/petrology/egg074>

- Pertermann, M., Lundstrom, C.C., 2006. Phase equilibrium experiments at 0.5 GPa and 1100–1300 °C on a basaltic andesite from Arenal volcano, Costa Rica. *Journal of Volcanology and Geothermal Research* 157, 222–235. <https://doi.org/10.1016/j.jvolgeores.2006.03.043>
- Pichavant, M., Macdonald, R., 2007. Crystallization of primitive basaltic magmas at crustal pressures and genesis of the calc-alkaline igneous suite: experimental evidence from St Vincent, Lesser Antilles arc. *Contrib Mineral Petrol* 154, 535–558. <https://doi.org/10.1007/s00410-007-0208-6>
- Prouteau, G., 2003. Experimental Constraints on the Origin of the 1991 Pinatubo Dacite. *Journal of Petrology* 44, 2203–2241. <https://doi.org/10.1093/petrology/egg075>
- Rader, E.L., Larsen, J.F., 2013. Experimental phase relations of a low MgO Aleutian basaltic andesite at $X_{H_2O} = 0.7–1$. *Contrib Mineral Petrol* 166, 1593–1611. <https://doi.org/10.1007/s00410-013-0944-8>
- Rushmer, T., 1993. Experimental high-pressure granulites: Some applications to natural mafic xenolith suites and Archean granulite terranes. *Geol* 21, 411. [https://doi.org/10.1130/0091-7613\(1993\)021<0411:EHPGSA>2.3.CO;2](https://doi.org/10.1130/0091-7613(1993)021<0411:EHPGSA>2.3.CO;2)
- Scaillet, B., 2003. Experimental Constraints on the Relationships between Peralkaline Rhyolites of the Kenya Rift Valley. *Journal of Petrology* 44, 1867–1894. <https://doi.org/10.1093/petrology/egg062>
- Scoates, J.S., Lo Cascio, M., Weis, D., Lindsley, D.H., 2006. Experimental constraints on the origin and evolution of mildly alkalic basalts from the Kerguelen Archipelago, Southeast Indian Ocean. *Contrib Mineral Petrol* 151, 582–599. <https://doi.org/10.1007/s00410-006-0070-y>
- Springer, W., Seck, H.A., 1997. Partial fusion of basic granulites at 5 to 15 kbar: implications for the origin of TTG magmas. *Contributions to Mineralogy and Petrology* 127, 30–45. <https://doi.org/10.1007/s004100050263>
- Toplis, M.J., Corgne, A., 2002. An experimental study of element partitioning between magnetite, clinopyroxene and iron-bearing silicate liquids with particular emphasis on vanadium. *Contrib Mineral Petrol* 144, 22–37. <https://doi.org/10.1007/s00410-002-0382-5>
- Tsuruta, K., Takahashi, E., 1998. Melting study of an alkali basalt JB-1 up to 12.5 GPa: behavior of potassium in the deep mantle. *Physics of the Earth and Planetary Interiors* 107, 119–130. [https://doi.org/10.1016/S0031-9201\(97\)00130-1](https://doi.org/10.1016/S0031-9201(97)00130-1)
- Ubide, T., Mollo, S., Zhao, J., Nazzari, M., Scarlato, P., 2019. Sector-zoned clinopyroxene as a recorder of magma history, eruption triggers, and ascent rates. *Geochimica et Cosmochimica Acta* 251, 265–283. <https://doi.org/10.1016/j.gca.2019.02.021>

- Ulmer, P., Kaegi, R., Müntener, O., 2018. Experimentally Derived Intermediate to Silica-rich Arc Magmas by Fractional and Equilibrium Crystallization at 1.0 GPa: an Evaluation of Phase Relationships, Compositions, Liquid Lines of Descent and Oxygen Fugacity. *Journal of Petrology* 59, 11–58. <https://doi.org/10.1093/petrology/egy017>
- Wang, W., Takahashi, E., 1999. Subsolidus and melting experiments of a K-rich basaltic composition to 27 GPa; implication for the behavior of potassium in the mantle. *American Mineralogist* 84, 357–361. <https://doi.org/10.2138/am-1999-0319>
- Wasylenki, L.E., 2003. Near-solidus Melting of the Shallow Upper Mantle: Partial Melting Experiments on Depleted Peridotite. *Journal of Petrology* 44, 1163–1191. <https://doi.org/10.1093/petrology/44.7.1163>
- Waters, L.E., Cottrell, E., Coombs, M.L., Kelley, K.A., 2021. Generation of Calc-Alkaline Magmas during Crystallization at High Oxygen Fugacity: An Experimental and Petrologic Study of Tephros from Buldir Volcano, Western Aleutian Arc, Alaska, USA. *Journal of Petrology* 62, ega104. <https://doi.org/10.1093/petrology/egaa104>
- Wood, B.J., Trigila, R., 2001. Experimental determination of aluminous clinopyroxene–melt partition coefficients for potassic liquids, with application to the evolution of the Roman province potassic magmas. *Chemical Geology* 172, 213–223. [https://doi.org/10.1016/S0009-2541\(00\)00259-X](https://doi.org/10.1016/S0009-2541(00)00259-X)