- 1 Climatic and cave setting influences on drip water fluorescent organic matter
- 2 over an annual cycle in Northwestern Spain and implications for fluorescent
- 3 laminations in stalagmites
- 4 Laura Sibylla Endres¹, Céline Jacquin², Saul González-Lemos³, Laura Rodriquez-Rodriquez³, Jakub
- 5 Sliwinski⁴, Nikita Kaushal¹, Oliver Kost¹ and Heather Marie Stoll¹
- 6
- ¹Geological Institute; ETH Zürich, Sonneggstrasse 5, 8092 Zürich, Switzerland
- 8 ² EAWAG; CH-8600 Dübendorf, Switzerland
- 9 ³ Departamento de Geología; Universidad de Oviedo, Spain
- ⁴ School of Earth and Environmental Sciences, University of St. Andrews; Queen's Terrace, St. Andrews

11 KY16 9TS

- 12
- 13 Corresponding author: Laura Sibylla Endres (<u>endres@erdw.ethz.ch</u>, Phone: +41 (0)44 632 86 13)
- 14

This is a non-peer reviewed preprint (v1), submitted to EarthArXiv. The manuscript is currently under review in Quaternary Research. Figures, Tables and Supplementary are placed at the end of the full text. We welcome Your feedback!

16 ABSTRACT

17 Speleothem fluorescence may elucidate past vegetation dynamics, while microscale fluorescent 18 laminations can provide annually resolved chronology. However, the origin of speleothem fluorescence 19 and the mechanism responsible for the formation of micrometer scale fluorescent lamination in 20 stalagmites, are not well constrained by monitoring studies. Here, we present results from a year-long 21 monthly drip water monitoring from seven locations in La Vallina cave (Northwestern Spain). 22 Fluorescence was quantified by excitation-emission matrices (EEM) spectrofluorometry. Five distinct 23 components were resolved by parallel factor analysis (PARAFAC) modeling, including previously 24 described humic-like, and protein-like components as well as an additional component suggesting a 25 contribution of indigenous fluorescence sourced from the bedrock. Variations in the overlying vegetation 26 and in the water reservoir age contribute to differences in the fluorescent components among different 27 drip sites. While some active stalagmites feature annual to sub-annual fluorescent laminae, the drip water 28 does not support higher abundance of humic-like fluorescence during the rainy season as a primary cause 29 for layer formation. The constancy in humic-like fluorescence, likely to arise due to abiotic interactions 30 with the bedrock, highlights possible other mechanisms on fluorescent layer formation, such as growth 31 rate control over the dilation of the fluorescence signal in the stalagmite.

32

33 **KEYWORDS**

34 banding, speleothem, monitoring, karst aquifer, climate reconstruction, proxy development,
 35 PARAFAC, spectrofluorometer, absorbance, EEM

37 INTRODUCTION

38 Fluorescence is a common feature in speleothems from many caves and is generated by the 39 dissolved organic matter (DOM) preserved in the speleothems. Speleothem fluorescence has been 40 recognized for many decades (O'Brien, 1956) and has been characterized in organics dissolved from the 41 stalagmite matrix (Pearson et al., 2020) as well as by surface scan in-situ analysis (Perrette et al., 2005). 42 The in-situ distribution of fluorescence in stalagmites has been studied in thin or thick sections using UV 43 excitation and detection on photographic film (Shopov et al., 1994) or detection of reflected fluorescence 44 using epifluorescence microscopes (Baker et al., 1993; Luetscher et al., 2021), fibre-optics extensions 45 connected to spectrofluorometer (McGarry and Baker, 2000), or on a polished section using a pulsed laser 46 of a fixed wave length (Quiers et al., 2015) and confocal laser scanning microscopy (CLSM) ((Sliwinski 47 and Stoll, 2021). From in-situ techniques, fine (5 to 1000 micron) scale alternations of high- and low 48 fluorescence intensity perpendicular to the growth axis, have been described from many stalagmites from 49 mid-latitude regions and this layering has been demonstrated to be annual in many cases (Baker et al., 50 2008; Proctor et al., 2000). The fluorescent layer intensity complexly interacts with speleothem trace 51 element chemistry (Huang et al., 2001; Sliwinski et al., 2022). In such stalagmites, several authors have 52 suggested the seasonal peak in fluorescence intensity is due to an increased in-flux of dissolved organic 53 matter (DOM) in drip waters during increased autumn leaf litter and rainfall (Borsato et al., 2007; 54 Kwiecien et al., 2022; McGarry and Baker, 2000), or alternately due to fine soil blown into the cave as an 55 aerosol (Fairchild and Baker, 2012; Luetscher et al., 2021). In addition to changes in DOM input, changes 56 in adsorption behavior at the crystal level, induced by variations in environmental and cave conditions 57 such as pH, redox state and cave ventilation, could also cause variations in the incorporation of the 58 fluorescent DOM ((Ramsever et al., 1997; Sliwinski et al., 2022). Where the processes driving fluorescent 59 layers can be demonstrated to be annual, then fluorescent banding can be exploited as a paleo-seasonality 60 indicator (Ramseyer et al., 1997; Blyth et al., 2008; Sliwinski and Stoll, 2021; Orland et al., 2012) or to improve age models through fluorescent layer counting (Kwiecien et al., 2022). However, due to the 61

limited number of monitoring studies and laboratory experiments available, direct evidence for 62 fluorescent banding formation is sparse and the physical process is currently not understood in detail. 63 64 Speleothem and drip water organics originate from surface vegetation (e.g. lignin 65 macromolecules), animal byproducts and soil processes; but also from biofilms and microbes living in the 66 karst environment (McGarry and Baker, 2000; Fairchild and Baker, 2012). In the carbonate aquifer, 67 organic matter can be degraded by microbial metabolism (Xia et al., 2022) or abiotically through 68 adsorption on the bedrock. In lab experiments by Jin and Zimmermann (2010), DOM adsorption on 69 carbonate host-rock was enhanced for high influx and long retention times. Adsorption in the same study 70 acts similar to a "buffer", because organics covering the active crystal surfaces have been shown to inhibit 71 bedrock leaching, and in contrast, low concentrations of DOM dissolved in water led to a release of 72 indigenous organic matter from the host rock. Further lab speleothem growth experiments indicate that 73 the presence of organic matter in the drip water can alter the fabric of the precipitated carbonates and 74 precipitation kinetics (Pearson et al., 2020). The chemical structure of the organic matter also influences 75 inorganic colloid coagulation (Wilkinson et al., 1997), and therefore the associated trace element transport 76 through the aquifer and incorporation into the calcite matrix (Blyth et al., 2008), especially for first-row 77 transition metals (Lindeman et al., 2021). 78 Only a minor part of the total DOM incorporated in speleothem fluoresces, often associated with 79 aromatic ring structures(Aiken, 2014). Fluorescence occurs due to a three-step process: 1) a molecule is 80 excited to a higher electronic state by absorption of a photon, 2) the molecule relaxes to lowest excited 81 singlet state (vibrational relaxation) and 3) the molecule emits light to return to ground state (Lakowicz, 82 2006). The possible total fluorescence yield depends on molecular structure and is higher for more "rigid" 83 structures, such as aromatic rings (Reynolds, 2014). Because fluorescence does not directly measure 84 organic molecules, but rather the interaction of functional groups with photons, fluorescence does not 85 provide a unique identification of the compounds present. Nonetheless, differing categories of 86 fluorescent DOM are preferentially excited by, and emit, at different wavelength ranges (Coble et al., 87 2014). In the liquid phase, such as drip waters or dissolved carbonate, three-dimensional excitation-

88 emission matrices (3D EEM) can be measured with modern spectrofluometers to provide characterization 89 of the range of DOM types present and has been used extensively to study fresh and marine waters over 90 the last decades. Two main categories of fluorescent DOM have been described: humic-like fluorescence, 91 which resembles spectral features of isolated humic and fulvic acids (Alberts and Takács, 2004), and 92 protein-like fluorescence, resembling fluorescence of amino acids such as trypophan and tyrosine (Coble 93 et al., 2014). Humic-like fluorescence is characterized by the presence of a double peak in the EEMs and 94 broader peaks, because it is thought to originate mostly from degradation of terrestrial biomaterial, which 95 typically consists of molecules of higher molecular weight and encompasses a number of different 96 compounds. Humic-like fluorescence also has been found to preferentially adsorb to the bedrock (Jin and 97 Zimmerman, 2010). Protein-like, in contrast, consists of more isolated compounds with lower molecular 98 weight and is thought to be more of microbial origin. Individual EEMs have been described by the main 99 peaks in excitation-emission wavelength pairs based on literature, e.g. Coble peak nomenclature (Coble et 100 al., 2014) or by ratios of the intensity at fixed excitation-emission wavelength pairs (e.g. the fluorescence 101 index, McKnight et al. (2001)). More recently, parallel factor analysis (PARAFAC, Stedmon and Bro 102 (2008)) has emerged to compile a model which describes the main fluorescent components and variability 103 over an entire dataset. The generation of a PARAFAC model enables further discussion about the 104 individual contributions to overall fluorescence and also allows to compare the dataset with other 105 published models through the OpenFluor database (Murphy et al., 2014a).

106 Speleothems beaker precipitation experiments incorporate peat water DOM (Pearson et al., 2020). 107 Prior studies of ground and cave drip waters have indicated dominant peaks in humic-like fluorescence, 108 with seasonally variable peak wavelengths (Baker and Genty, 1999). Compared to overlying soil water 109 fluorescence, spectra of drip water are likely to be blue-shifted, because of preferential adsorption on 110 bedrock (Philibert et al., 2022; Jin and Zimmerman, 2010) and continued microbial degradation. 111 Additionally, microbial fluorescence has been suggested to be important for cave waters (Birdwell and 112 Engel, 2010). Microbial ("protein-like") fluorescence could occur in the cave system, but also be related to livestock above the karst (Mudarra et al., 2011). In one study comparing EEM spectra of three 113

114 dissolved flowstone samples and the drip waters feeding them, the flowstones showed evidence of microbial peaks that are not seen in the drip waters feeding them (Pearson et al., 2020). PARAFAC 115 116 modeling is common in water research and has been used briefly by Rutlidge (2014, 2015) to investigate 117 DOM in drip waters after irrigation and also by Pearson et. al (2020). 118 In this present study, we have analyzed the evolution of fluorescent DOM in drip water at 119 multiple drip sites within La Vallina cave in NW Spain to contribute to a better understanding of the 120 spatial and temporal variations of organic matter in drip water and improve our understanding of 121 stalagmite fluorescence signals. We characterize the components of drip water fluorescence using a 122 PARAFAC model, and compare this model to cave properties, drip water chemistry and climatic 123 influences. The analysis is extended through measurements of fluorescence in the dissolved bedrock and 124 by CLSM images of actively growing stalagmites from the same cave. We evaluate the implications for 125 inferring past changes in landscape and climatic processes from stalagmite fluorescence, and for the 126 interpretation of annual to subannual fluorescence lamination in stalagmites.

127

128 **STUDY SITE**

129 Our study was conducted in La Vallina cave, situated near the town of Porrúa in Northwestern Spain (43°27'38''N, 5°04'04 "W, Fig.1). Under present-day conditions, the cave entrance is located 2.5 130 131 km from the coastline at an altitude of 70 m above the sea level. La Vallina formed within the 132 carboniferous dark sulfidic massive micritic limestone of the Barcaliente formation. The bedding is 133 subvertical, dipping 70° to the North. Barcaliente formation is rich in amorphous organic matter and classified as containing type II and III kerogen (Álvarez et al., 2019). Numerous block falls throughout 134 135 the cave are consistent with a seismically active shallow crust (Delavaud et al., 2012). Karstification has 136 led to several dolines, which can be recognized at the surface above the cave.

For this study, we selected seven locations within the cave system at which to monitor organics in
drip water for one year from May 2020 – May 2021. Except for Mushroom (*MUS*), all sampling locations

139 are positioned in the upper of two cave galleries. Vegetation and land area above the cave are 140 significantly influenced by anthropogenic activities. A former primary deciduous forest has been 141 largely exchanged with eucalyptus trees, which have been introduced in the region since the 19th 142 century for wood production. The stem diameter of the current eucalyptus trees indicates ages around 60 143 years. Compared to other trees, eucalyptus has deep roots and a high water demand, which may have an 144 effect on present-day cave hydrology. Furthermore, parts of the land have been exposed to fire activity 145 and are currently covered with ferns. Exceptions are a few remaining oak trees close to the sampling 146 location "Forest" (Fig.1). Other parts of the land above the cave are used as pasture for farm animals 147 today. The detailed conditions for each sampling location are presented in Table 1 and Supplementary 148 Table 1.

149 Present day regional climate, primarily controlled by the proximity to the North Atlantic ocean, is 150 mild and warm. It can be classified as "Cfb" [temperate, with a warm summer, but no dry season], 151 according to the Köppen-Geiger classification (Köppen, 1884; Peel et al., 2007). The closest weather 152 station (Llanes, 4.2km NE of the cave, at the coast) has recorded a precipitation of 1267 mm/yr and an 153 annual mean temperature of 15.3°C for the study period May 2020 - May 2021. The weather patterns over 154 Northern Iberia are predominantly influenced by the westerlies with most incoming precipitation 155 originating from Atlantic front systems and exhibit a long-term weighted annual mean $\delta^{18}0$ of -5.75 \pm 156 0.34 ‰ (Santander GNIP station, 2000-2015, IAEA/WMO (2022)).

157 METHODS

158 Experimental Methods

159 Field Campaign

Drip water samples were collected from May 2020 to May 2021 on a quasi-monthly basis. Biweekly sampling was carried out in autumn to evaluate the impact of a higher amount of rain and falling leaves on drip water chemistry. Four locations (*GRA*, *GLO*, *SKY*, *PLA*) were monitored over the entire year, *FOR* was monitored from September 2020-May 2021, *MUS* and *SNO* only for three dates in 2021.

164 From each location, samples for fluorescence measurements have been collected directly into amber glass 165 vials, which were previously combusted for 4 hours at 450°C. In a separate study (Kost et al., 2022), 166 splits of water samples on the same day and location were analyzed for stable isotopes, trace elements, 167 carbon isotopic ratio of dissolved inorganic carbon (DIC) and cave conditions (Temperature, RH, pCO2) 168 were recorded. The samples were shipped to ETH Zurich and stored in the dark at 4°C before 169 measurement. Measurements of bedrock thickness for all sampling locations (Table 1) were carried out 170 with a modified Leica Disto X310 (Heeb, 2009, 2010). The sampling frequency was chosen to identify 171 seasonal scale changes in drip water fluorescence. Given the length of the entire time series (only one 172 year), longterm trends and interannual variability cannot be evaluated. The bi-weekly or monthly 173 sampling is also not designed to capture the effect of fast-changing fissure flows; however, with one 174 exception (FOR), the sampled sites are classified at the boundary between seepage flow and seasonal drip 175 based on the variation in drip rate, suggesting a dominant matrix flow component (Kost et al 2022).

176

177 Drip Water Absorbance and Fluorescence Measurements

Drip water samples were stored at 4°C in a refrigerator unit without other biological samples or organic solvents. Samples were measured between one month and one year after collection and it has been double-checked that storage time does not impact the results by re-measuring selected older samples (Supplementary Figure 1). Previous studies showed a removal of especially tryptophan-like fluorescence after filtering the samples (Baker et al., 2007). Therefore, because no large colloids were expected in our drip waters and we were interested in total fluorescence reaching the speleothem, we did not filter the samples.

Drip water absorbance spectra and fluorescence excitation-emission matrices (EEM) were measured simultaneously with a Horiba AquaLog spectrofluorometer using 4 s integration time and a step-size of 5 or 3 nm, depending on measurement day. Quartz cuvettes with a pathlength of 1x1 cm were used and rinsed several times with nanoQ water between each measurement. During post-processing EEMs with 3 nm resolution were converted to the coarser 5 nm resolution by bicubic interpolation to

enable PARAFAC analysis and increase comparability between individual measurements days. EEMs
were exported from the R1/C1 AquaLog output with respective excitation and emission wavelengths of
240-600 nm and 240-800 nm. Further post-processing included NanoQ water blank subtraction, innerfilter effect correction, Raman/Rayleigh scatter removal and Raman normalization performed with eemR
(Massicotte, 2019). EEMs were visualized using the scientific colour map *batlow* (Crameri et al., 2020).
Intensity is given in normalized raman units (R. U.) and variability of data is expressed in the form of 1
standard deviation.

197

198 Confocal Laser Scanning Microscopy

199 Solid-phase fluorescence in three active stalagmites and one bedrock sample was investigated 200 with confocal laser scanning microscopy (CLSM). Measurements were performed at Scientific Center for 201 Optical and Electron Microscopy (ScopeM) at ETH Zurich using an Olympus Fluoview 3000 and the 202 methodology of Sliwinski and Stoll (2021). The instrumental settings result in an optical slice thickness 203 of less than 10 µm with an 10x objective. An excitation wavelength of 488 nm and detection window of 204 490-590 nm were used, and resulted in an optical slice thickness of less than 10 um with an 10x objective. 205 The instruments minimum excitation wavelength is 405 nm, so UV excitation in solid state was not 206 possible. Retrieved images were mirrored to represent the original orientation and a scale bar was added 207 using Fiji (Schindelin et al., 2012).

208

209 Liquid-Phase Fluorescence Measurement of Bedrock Sample

210 Contribution of the bedrock to drip water fluorescence was investigated by dissolving three times
211 2 mg aliquots of drilled powder samples in 4.5 ml of 0.025 M HCl, analogous to Pearson et al. (2020).

212 The aqueous solution was brought to a pH of 5-7 by titration with NaOH. Dissolved bedrock absorbance

213 spectra and fluorescence EEM were post-processed identical to drip water and measured on the same

instrument, but with an integration time of 10 s and a step size of 5 nm.

215 Numerical Analysis Methods

216 PARAFAC modeling

217 PARAllel FACtor (PARAFAC) analysis (Stedmon and Bro, 2008) was used to investigate DOM 218 fluorescence variability in the spatial and temporal domain. Over the last decade, PARAFAC modeling 219 has become a standard approach for EEM interpretation due to the possibility of recognizing site-specific 220 peaks instead of referring to general literature (Coble et al., 2014). A global five-component model was 221 computed from 65 corrected EEMs by staRdom (Pucher et al., 2019) with non-negative constraints, a model tolerance of 1e⁻⁸, 40 models and 10'000 maximum iterations. All models converged with these 222 223 settings and no additional outliers were found by visual. Validation with a regular split-half analysis was 224 not considered statistically robust due to the high variability between the sites and small sample size < 225 100 (Pucher et al., 2019). Instead, separate local PARAFAC models of the five main sampling sites were 226 computed with fewer components due to the small sample sizes of 9-13 samples. To compare local models to the global five-component model, shape-sensitive congruencies (SSC) (Wünsch et al., 2019) 227 228 were computed for each combination to quantify similarities between the individual components in the 229 two models (Supplementary Figure 2). 230 The final PARAFAC model was uploaded to OpenFluor database (Murphy et al., 2014b) to 231 compare peaks with published PARAFAC models. Model stability was assessed with the core 232 consistency diagnostics (CORCONDIA) by Bro and Kiers (2003) and was found to be 79.65%. 233 PARAFAC components were visualized using the scientific colour map *ajolla* (Crameri et al., 2020). 234 235 Calculation of Absorption Spectra and Slope Ratio

In addition to the fluorescent measurement, optical properties of DOM can be characterized by joint absorption measurements with the Aqualog spectrofluometer. The fraction of total DOM that absorbs UV and visible light, chromophoric dissolved organic matter, can overlap with the fraction of fluorescent DOM, but is not necessarily the same and analysing absorbance separately adds additional

240	information about the origin and composition of the samples. To this purpose, absorbance measured with
241	Aqualog was converted to the absorption coefficient α following Beer's law (Supplement Eq. 1).
242	We have used the absorption α as a quality test, whether the unfiltered samples are optically
243	dilute enough. Values for α were well below 0.1 for all samples and measured wavelengths except for
244	GRA on 7.8.2020, which showed evidence of mud contamination already during collection and was
245	subsequently treated as outlier. Two samples containing negative absorbance values have also been
246	disregarded for absorbance analysis.
247	From the entire absorption spectra, a slope ratio between a shorter (275-295 nm) over a longer
248	(350-400 nm) wavelength region was computed analogous to Helms et al. (2008). The two slopes, S ₂₇₅₋₂₉₅
249	and S ₃₅₀₋₄₀₀ , were computed by fitting a linear model through the natural logarithm of the absorption
250	spectra values in the respective wavelength regions. By convention, positive slope values are given for a
251	decrease with higher wavelengths in order to fit the expected exponential decay.
252	

253 **RESULTS AND DISCUSSION**

254 Dominant Modes of Drip Water Fluorescence and their Attribution

The principal peaks and modes of variability in fluorescence across all collected drip water samples are identified with PARAFAC analysis. We present a five-component model, which explains 88% of the total variance within the entire data set (Fig. 2a and Table 2) and serves as a base template for further interpretation of the individual locations. The components C1-C5 are normalized to unit intensity and are ordered by the fraction of variance explained.

The dominant feature of odd PARAFAC components (C1,C3,C5) is the occurrence of a double peak within the EEM (Fig. 2a). C1 shows two excitation peaks (Ex_{max}) at 240/320 nm and an emission maximum (Em_{max}) at 405 nm, which are traditionally assigned to Peaks A_c and C (Coble et al., 2014). These peaks have been widely recognized as humic-like components with high molecular weight and aromatic organic compounds derived from plant litter (Stedmon and Cory, 2014; Coble et al., 2014). In

265	this study, these are thought to originate mostly from vegetation above the cave and containing
266	compounds from plant litter such as lignin. The components C3 ($Em_{max} = 305 \text{ nm}, Ex_{max} = 240/280 \text{ nm}$),
267	and C5 (Em _{max} =455 nm, Ex _{max} =265/370 nm), appear to be blue and red shifted versions of C1,
268	respectively. C5 exhibits a shift towards longer wavelengths and similar patterns have been attributed to
269	soil fulvic acid (Yamashita and Jaffé, 2008) and terrestrial higher plants (Cory and McKnight, 2005).
270	Shifting of wavelengths towards a more blue spectrum (i.e., shorter wavelengths) is likely to appear due
271	to breaking of aromatic bonds in the original compounds by microbial activity (degradation). C3, shifted
272	towards shorter wavelengths, possibly indicates the presence of older and/or more processed material.
273	This kind of fluorescence is often present in marine planktonic environments, but has also been shown to
274	be produced within the lower soil or epikarst by planktonic, heterotrophic bacteria feeding from plant
275	litter (Xia et al., 2022) and is consistent with EEM of fossil (degraded oil) compounds (Zhou et al., 2013).
276	We have dissolved the organic-rich bedrock to investigate a possible contribution of indigenous
277	fluorescence due to bedrock leaching. A comparison of the bedrock-derived EEM (Fig. 2b) with
278	PARAFAC peak wavelengths shows a strong similarity to C3. Therefore, we propose that fluorescent
279	material within the organic-rich bedrock, which potentially is released during bedrock leaching, could be
280	a major source for C3 drip water fluorescence. To our knowledge, such a contribution of fossil organic
281	matter to drip water fluorescence has not been discussed in literature previously.
282	The two even components (C2, C4) both contain single peaks with low Ex-Em wavelengths and
283	narrow Ex/Em spectra, which has been attributed to organic compounds with a lower degree of
284	conjugation such as microbially produced amino acids (Coble et al., 2014), microbial metabolism (Elliott
285	et al., 2006), but also plant-derived simple phenols, indoles and tannins (Aiken 2014).
286	On comparison with the OpenFluor database (Murphy et al., 2014b), spectra similar (similarity
287	scores in Ex and $\text{Em} > 0.95$) to C2 have occurred in cold (arctic) water surroundings, described as
288	tyrosine-like (Chen et al., 2018), but have also been found in EEMs from melted ice cores and described
289	as more processed microbial signal and tannin-like (Dandrilli and McConnell, 2021). While a tannin input
290	signal from eucalyptus trees above the cave can't be excluded, we find microbial activity here the

291 favorable explanation, because the observed seasonal variability of C2 is more similar to the microbiallyaltered C3 and C4 components than to C1 and C5, attributed to higher plants. Moreover, C4 spectrum is 292 293 similar to the amino acid phenylalanine, which has been measured isolated by Wünsch et al. (2015). 294 Phenylalanine is a rather hydrophobic compound, e.g produced by algae/bacteria and consequently 295 expected to be a rather local signal originating from the lower soil/epikarst. 296 Characterisation of dissolved organic matter in cave drip water by EEM is limited, because EEM 297 spectral peaks do not have unique sources and cannot unambiguously be attributed to single compounds. 298 Additionally, PARAFAC modeling may create artifacts instead of biologically relevant signals, and 299 changing experiment conditions (e.g. pH or temperature of drip waters) can influence the peak intensities 300 and positions (Aiken, 2014). Nevertheless, as demonstrated by numerous applications of EEM and 301 PARAFAC in surface waters, the combination of EEM measurements and PARAFAC modeling can be 302 used successfully to describe properties of drip water fluorescence in more detail, with a comparably 303 rapid measurement method. The PARAFAC model reveals several, distinct fluorescent peaks of both, 304 humic-like and microbial origin. Specifically mentioned should as well be the absence of a further humic-305 like peak, often present in soils (UVC humic-like, Cuss et al. (2019)). Our C5 peak might contain some of 306 this signal, but it is much weaker than would be expected. Based on studies investigating active carbon 307 adsorption (Philibert et al., 2022), we suggest that this component might be preferentially adsorbed on the 308 bedrock and this is the reason why it is missing in the here presented PARAFAC model. Future sampling 309 including soil water would help clarify the relative contributions of soil and karst fluorescent organic 310 matter. In combination with the here presented fossil contribution, this points towards a rather complex 311 interaction between karst and drip water fluorescence, which will also be reflected in the speleothem. 312 More specific information on the compounds present and their origin could be provided by future studies 313 employing ESI-FTIR-MS (Lechleitner et al., 2017) or LC-OCD and characterizing the microbiology of 314 the drip waters.

316 Spatial Variation of Drip Water Optical Properties and Relationship to Cave Setting

Drip water optical properties, summarized in Table 1, vary substantially between sampling 317 318 locations. Overall mean fluorescence over the entire EEM region ranges from 10.88 +/- 0.23 R.U. (MUS) 319 to 47.40 +/- 13.34 R.U. (FOR). The absorption coefficient of UV and visible light shows an exponential 320 decrease with increasing wavelength across sampling locations (Fig. 3a). The observed decrease of the 321 absorption coefficient is fast compared to samples from the surface or swamps (e.g. Helms et al. (2008)), 322 likely due to the higher amount of filtering occurring in the carbonate aquifer. In PLA and GLO the 323 median absorption spectra further show secondary peaks around 295 nm and 350 nm, respectively. These 324 peaks could theoretically occur due to spectral interferences with larger colloids, but overall absorption coefficients of our samples are below $< 0.1 \text{ m}^{-1}$. Absorption coefficients in this bandwidth could also be 325 326 altered by compounds such as tryptophan or uric acid, which show local deviations in this wavelength 327 region, when measured in isolation (Spangenberg et al., 2021). PLA is located below pasture and a doline; 328 the presence of uric acid, originating from cattle, seems therefore plausible, but would need to be 329 confirmed by future testing of uric acid by liquid chromatography-mass spectromety. PLA also has a significantly different slope ratio (p < 0.05, Tukey-test), because the spectral slope S₂₇₅₋₂₉₅ is negative for 330 331 PLA due to the secondary peak (Fig 3b). Although a negative S₂₇₅₋₂₉₅ value is a deviation from the 332 expected exponential decay, similar features have been found in cave spring waters previously (Birdwell 333 and Engel, 2010).

334 Plotted by sampling location, the average and standard deviation fluorescence EEM plots show 335 substantial variations in peak distribution (Fig. 4). A key influence in our data set are the properties of 336 overlying land cover (Fig. 1), featuring a gradient from dominantly forest (FOR, SNO), to bushes (GLO, 337 GRA) and pasture (PLA, MUS, SKY) vegetation. Sampling locations FOR and SNO exhibit both similar 338 average patterns and variability, with the dominant contribution from the humic-like components C1 and 339 C5 (Fig. 4a,b). Above these locations, oak and eucalyptus trees dominate, therefore, the strong humic-like 340 fingerprint is consistent with a higher amount of plant litter and soil products expected to be transported 341 to the respective drip sites. In contrast, "protein-like" C2, C4 and also C3 ("fossil-like") are rather low.

342	GLO and GRA, located in shallow depths and covered by ferns and singular trees, exhibit comparable
343	total fluorescence, but much greater contribution of C2 and C4. PLA and MUS feature the lowest total
344	fluorescence and very limited variability. PLA is covered by grass/pasture and MUS by ferns and singular
345	trees. Despite similar landcover at PLA and SKY, SKY features higher overall fluorescence.
346	Fluorescence is generally expected to decrease with longer transport and residence time, since the
347	organic matter can be degraded by microbial activity (Fairchild and Baker, 2012) and especially humic-
348	like compounds are likely to adsorb on the bedrock (Jin and Zimmerman, 2010). Within our data set,
349	depth is not a dominant control on humic-like fluorescence (Fig. 5a). Highest values are found in FOR
350	and SNO, covered by forest, with a cover thickness of 14.35 m - 14.47 m. In contrast, humic-like
351	fluorescence is lower in the shallow locations GLO, GRA and no decrease can be found compared to the
352	deepest locations PLA and SKY. Yet, for sites of comparable landcover (GLO, GRA, MUS), the deepest
353	site (MUS) does feature lower total fluorescence which could be due to greater degradation in transit to
354	the deeper site. However, these three sites feature similar total C1+C5 fluorescence; therefore, it is also
355	possible that instead of a depth driven process, the additional livestock activity at GLO and GRA may
356	contribute to the additional C2, C3, and C4 intensity relative to MUS.
357	In this setting, the variance of drip water δ^{18} O may be useful indicator of drip water residence
358	time (Kost et al. 2022). This is because individual rain events feature considerable δ^{18} O variability (Stoll
359	et al., 2015; Moreno et al., 2020), but mixing within the karst reservoir leads to dampening of this signal
360	to varying degrees at different drip sites. The greatest dampening of event scale variation, ie. those with
361	lowest variance in drip water δ^{18} O, are inferred to be characterized by the longest residence times and
362	greatest mixing of infiltration events. Here, the δ^{18} O variance is calculated only for drip water sites
363	monitored for at least 9 samplings over at 9 month period, and is notably much greater for the FOR
364	location than for the other calculated sites (<i>PLA, SKY, GLO, GRA</i>). The site with much higher δ^{18} O
365	variance and shorter residence time, FOR, features higher total C1+C5 fluorescence, compared to other
366	sites with longer residence time Fig. 5b. We have also calculated the ratio C3/(C5+C1) to assess the

367 relative significance of reworked or fossil DOM relative to fresher humic- and fulvic-like inputs. This 368 ratio correlates strongly with the humification index, which has also previously been used to estimate the 369 ages of organic matter (Ohno, 2002). The C3/(C5+C1) ratio is lowest for the FOR site of shortest 370 reservoir time, consistent with least microbial consumption and/or more adsorption of high-molecular 371 weight components (Fig. 5c). These trends are consistent with observations elsewhere that longer 372 residence times lead to increased DOM processing through microbial transformation and adsorption 373 reactions (Blyth et al., 2016; Lechleitner et al., 2017). 374 Additionally, the water flux may influence the fraction of input fluorescence which arrives at the

375 cave drip site. Although at similar depth and reservoir mixing inferred from δ^{18} O variance, the much 376 lower driprate of *PLA* (1.22+/- 0.916 ml/min) than *SKY* (89.22 +/- 24.25 ml/min) correlates with a much 377 lower average total fluorescence intensity (Fig. 4b). The very high flux at *SKY* may saturate the 378 absorbance capacity of karst host rock. Conversely, a lower drip rate and flux at *PLA* may result in a 379 higher fraction of dissolved fluorescence components adsorbed by karst host rock and hence lower 380 concentrations emerging in cave drip water.

381 The fluorescent results therefore suggest the potential for multiple land surface origins of the 382 fluorescence components from humic and fulvic vegetation signals as well as potential livestock 383 contributions, further modified during transport through the karst system by microbial reworking and 384 adsorption on karst host rock (Fig. 6). Our spatial trends in total fluorescence and its composition 385 therefore suggest that at comparable bedrock depth, higher density of trees and shorter water residence 386 times favor higher relative contribution of fluorescence from humic-like and fulvic-like sources. 387 Fluorescence components more likely of microbial origin dominate at sites with longer water residence 388 times. For sites of comparable landcover and residence time, higher drip water flux (drip rate) correlates 389 with a higher total fluorescence.

391 Temporal Variation of Fluorescence and Relation to Climate

392 Temporal variation in fluorescence components at different locations

393 At a given drip site, the PARAFAC model components C1 and C5 feature more constant 394 concentrations over the year without strong seasonality (Fig. 7). In contrast, C2 and C4 intensity consists 395 of short-lived peaks and varies substantially on the seasonal timescale in most monitored locations, 396 especially in GLO and GRA. C3 appears more constant in some locations, but variable in other sites. Dips 397 in protein-like fluorescence (C2,C4), e.g. in October 2020, occur often concurrently across sampling sites. 398 In contrast, C3 peaks do not occur at the same time across drip sites, e.g. increased values in location 399 FOR in autumn are contrasted by especially low C3 values in SKY. FOR, which is the more shallow 400 sampling site and covered with denser vegetation, likely featuring rapid water transmission and shorter 401 reservoir times inferred from its high δ^{18} O variance, might be prone to receive an increased in-flux of 402 older soil material, e.g. from fissure flow. Despite the significant short term variation of some trace 403 elements, consistent with mixture of matrix and fissure derived water (Kost et al., 2022), components C1,C3 and C5 variability is smoothed and generally, humic-like fluorescence is present over the entire 404 405 year. This constancy might be related to longer water residence times and well-mixed reservoirs, but it 406 may also or additionally reflect a buffering of the concentration of humic components by sorption 407 processes in the karst reservoir. Abiotically, humic-like compounds are preferentially adsorbed by the 408 carbonate bedrock, while during low input of humic-like compounds, a large portion of the compounds 409 are likely to desorb again from the bedrock (Jin and Zimmerman, 2010). In the presented data set, this 410 process together with the release of indigenous organic matter from bedrock, might have weakened a 411 potential seasonal cycle, leading to basically constant values of humic-like fluorescence.

412

413 *Relationship of fluorescence to rainfall and temperature*

Relevant for the interpretation of the fluorescent signal within speleothems is further the
dependence on climatological parameters such as temperature and precipitation. We investigate this

416	hypothesis by comparing the fluorescence dataset to climatological data from the nearby weather station
417	Llanes (4.2km NE of the cave, at the coast). Two equally-sized groups of dry and wet conditions are
418	compared for each location based on total rainfall amount in the 10 days prior to sampling (Fig. 8, top
419	row). The humic-like and fulvic-like components C1 and C5 are higher during wet conditions only for the
420	location FOR, whereas all other locations show no change between the wet vs dry conditions, and PLA
421	shows lower C5 for wet conditions. For C3, FOR shows higher values in wet conditions, whereas other
422	sites (PLA, SKY) show increased values for dry conditions or no difference (GRA, GLO). This contrasting
423	behavior at FOR may reflect an aged soil source of C3 at this site and its fast transmission due to the low
424	reservoir time of FOR drip water, versus deeper sites where C3 may be more significantly sourced from
425	fossil bedrock OM and therefore decoupled from the precipitation regime.
426	In contrast, microbial components C2, C4 are enhanced for dry conditions across all drip sites.
427	Here, the difference in C4 between the two conditions is more pronounced for shallow locations FOR and
428	GLO, than for PLA and SKY. Because all monitored sites continued dripping throughout the 16 month
429	monitoring period, we infer that water is present continually within the karst. However, we hypothesize
430	that the presence of cavities in semi-dry aquifers enhance the conditions for microbial life (C2,C3,C4).
431	Shallow locations might react more readily because they dry out faster and carbon sources for microbes
432	might be more abundant. An alternative interpretation would be that the microbial population and its
433	contributions are diluted during wet periods. Similar results, an increased presence of autochthonous
434	microbial fluorescence for drier conditions, have been found experimentally for karstic subsurface-
435	environments (Xia et al., 2022). The here presented groups are too small for statistical tests, but a t-Test
436	on the difference between the means of humic components C1,C5 and microbial components C2,C4 over
437	data from all locations yields significantly higher values in dry conditions for microbial components (p-
438	value= 0.021, Supplementary Figure 3).
439	In contrast to precipitation, with the same methodology no effect of ambient temperature on the

440 fluorescence fingerprint could have been detected (Fig. 8, bottom row), which is justified by the constant

temperature conditions present within the karst (T_{mean} =13.96 °C, average seasonal variation with SD $\pm 0.32^{\circ}C$) (Fig. 8, bottom row).

443

444 Relationship of fluorescence to soil respiration, bedrock dissolution, and chelated metal transport

445 We compare the fluorescence signals with indicators of karst processes measured on the same 446 drip water samples. The carbon isotopic ratio of dissolved inorganic carbon (DIC) in drip water is 447 interpreted as an indicator of the soil and epikarst CO2 which in turn is driven by the aggregate autotrophic and heterotrophic respiration rate (Kost et al., 2022). More negative $\delta^{13}C_{DIC}$ indicates higher 448 449 respiration rates. At the shallow and rapid transmitting FOR location, the higher soil and epikarst 450 respiration rates coincide with the highest concentrations of C1 (and C5), suggesting either that that high 451 respiration rates contribute to high production of humic-like breakdown products, or that conditions 452 optimal for maximizing respiration are also optimal for maximizing the release of humic-like components 453 to waters (Fig. 9a). We suspect that at the other sites with long data series (SKY, PLA, GLO, GRA), 454 longer water reservoir times may attenuate the seasonal signal in dissolved fluorescence leaving the soil layer relative to the cycle of $\delta^{13}C_{DIC}$. This is because $\delta^{13}C_{DIC}$ is not sensitive to water residence time, when 455 456 it is set by continued CO_2 exchange with water throughout the infiltration pathway by deep-rooted trees (Breecker et al., 2012). No significant relationship is observed between the $\delta^{13}C_{DIC}$ and the fraction of 457 fresher versus the more degraded compounds (C3:(C1+C5) ratio) nor the concentration of the C2 and C4 458 459 compounds, at any site.

Within a given dripsite, the concentration of Sr has been proposed as an indicator of the intensity of bedrock dissolution, because all drip water Sr is bedrock sourced, and the concentration of Sr is not appreciably affected by precipitation of CaCO₃ prior to sampling of drip water (Kost et al., 2022). Across all of the sampled sites, there is a general weak positive correlation between the Sr concentration and the concentration of C1, driven primarily by the higher concentrations in the FOR site compared to other sites (Fig. 9b). This correlation may arise because greater soil and epikarst respiration and CO₂ are required to

attain greater degrees of bedrock dissolution, and these higher rates of respiration result in greater
production of humic-like components. The forested vegetation of the *FOR* site would especially favor
high respiration rates. A similar correlation is observed for Ca (Supplementary Figure 4). No significant
correlation is observed between the bedrock dissolution and the concentration of C3, or the ratios of other
components.

471 Highly insoluble transition metals and REE-like elements such as Cu or Y are inferred to be 472 transported in drip waters as complexes with organic ligands such as humic and fulvic acids (Hartland et 473 al., 2014; Hartland and Zitoun, 2018), and in stalagmites, seasonal layers of enhanced fluorescence often 474 coincide with elevated concentrations of these elements (Sliwinski and Stoll, 2021). However, a 475 comparison of C1 and C5 concentrations with the concentration of drip water Cu (Fig. 9c) (or Y; 476 Supplementary Figure 4) reveals no significant correlations within a given site or across all sites. This 477 lack of relationship suggests that the Cu and Y concentration in drip water may be controlled by more 478 unique components of the dissolved organic carbon pool than resolved by the C1 or C5 fluorescence, that 479 fluorescent DOM present is not necessarily attached to colloids, or that other controls of selective 480 scavenging of either Y or organic ligands decouple any initial relationship. The precise relationship 481 between Y (and transition metal transport) and dissolved organic components in drip waters may require 482 further analysis in typical karst settings, as the relationships to date have been explored most rigorously in 483 a hyperalkaline cave setting (Hartland et al., 2014).

484

485 **Relating Drip water fluorescence and speleothem fluorescence**

Annual scale fluorescent lamination has been reported in mid-latitude stalagmites (McGarry and Baker, 2000; Kwiecien et al., 2022; Luetscher et al., 2021; Sliwinski and Stoll, 2021) from this and other caves, and our (CLSM) images (Fig. 10 a-c) reveal some sectors in the actively growing stalagmite from beneath the *SKY*, and *SNO* drip locations with fine scale lamination which may be annual to subannual according to available age models from ¹⁴C (Sliwinski et al., 2022). These rapidly growing stalagmites feature high porosity and many crystal defects, which contribute to lateral discontinuity in fluorescent

492 lamination, as described in previous studies (Sliwinski and Stoll, 2021; Sliwinski et al., 2022). In our 493 study with CLSM, the selected excitation wavelength is 488 nm. Based on our analysis of dissolved 494 fluorescence, this range is suitable to stimulate fluorescence in the humic-like and fulvic-like region of C1 495 or C5, but not as significantly for C2, C3, or C4. Nonetheless, comparing solid state and dissolved 496 fluorescence is not straightforward. The polished bedrock sample, dominated by C3-like component in 497 dissolved form, also indicates pronounced fluorescence, concentrated in certain sectors, by this same 488 498 nm excitation wavelength in the solid state CLSM analysis of the stalagmites (Fig. 10d), suggesting that 499 this bedrock component, if present without further modification, would also be detected in our stalagmite 500 CLSM analyses.

501 While the prevailing hypothesis for fluorescent laminae in speleothems invokes an autumnal 502 pulse in fluorescent components (along with colloidally transported metals) with the onset of autumnal 503 rains and infiltration after a dry summer, our sampling has not detected any seasonal pulses of humic-like 504 or fulvic-like fluorescence in autumn or any other season in the drip water at any monitored sites (Fig. 7), 505 and we also do not detect a relationship between C1 or C5 with precipitation at SKY (Fig. 8). In previous 506 studies, fluorescent lamination has been attributed to humic- and fulvic-like components (e.g., McGarry 507 and Baker, 2000; Pearson et al., 2020), although we are aware of no full EEM characterization of the 508 speleothem fluorescence from annually laminated stalagmite samples. Because of the low concentrations 509 of organic matter, we have not been able to obtain EEM spectra of our actively growing stalagmites, so 510 the conventional inference of humic- and fulvic-like source components of the fluorescent laminae cannot 511 be tested in these active stalagmites. Given the absence of any seasonal variation in humic-like and fulvic-512 like drip water concentrations at our cave setting, several alternative processes may contribute to the 513 formation of annual to subannual fluorescent laminae. First, if the fluorescent lamina were dominated by 514 the C1 and C5 components, which have constant concentrations in drip water, it is possible that a 515 concentrated fluorescent layer reflects not changes in the delivery of fluorescent components, but rather 516 variable dilation of the fluorescence in the stalagmite as a function of the CaCO₃ deposition rate. An 517 analogous mechanism has been described for the generation of a seasonal Y peak in stalagmites, whereby

518 periods of slower stalagmite growth distribute an otherwise constant Y (or fluorescence) flux over a 519 shorter distance in the stalagmite, generating higher average concentrations in the slower growth portion 520 of the year (Sliwinski et al., 2022). Based on calculations from the monitored drip water Ca 521 concentrations and the cycle in cave air pCO₂, stalagmite growth at SKY is simulated to occur rapidly 522 during the winter season, and slowly during the summer season, with brief periods of negligible growth 523 simulated during the spring and fall (Kost et al., 2022). In such a stalagmite growth regime, the slow 524 growth during summer could lead to enhanced fluorescent intensity in the summer and weaker fluorescent 525 intensity in the winter layer. In the case that fluorescent laminae were dominated by C1 and C5 like 526 components, an alternative mechanism for the development of a fluorescent laminae could be an 527 increased absorption or trapping efficiency of the fluorescent components in the stalagmite during one 528 season, due to change in factor such as drip water pH or residence time on the stalagmite. Here, further 529 experiments testing the affinity of different organic ligands to Ca and trace elements would be beneficial 530 for interpretation.

531 In some of our monitored drip waters (GLO, SKY, to lesser extent FOR), the C3 component 532 exhibited greater temporal variation than C1 or C5. If this component were the main source of 533 fluorescence in the laminae, the variations in C3 concentration in drip water might contribute to modest 534 seasonal variation in speleothem fluorescence. However, the distinct drip water sites exhibited different 535 seasons of maximum C3 fluorescence, so if this were the dominant process behind laminations, it would 536 imply that the bright fluorescent laminae may not represent the same season in all stalagmites from a 537 given cave. C2 and C4 components exhibit consistent increase during dry intervals, so if these 538 components dominated the variability in the stalagmite laminae, it could contribute to systematic 539 oscillations on annual to subannual scale. However, a further caution is that the sites in our program 540 which exhibit large variations in C2, C3, and C4 components are also in sectors of the cave beneath at 541 least intermittent livestock activity. We cannot assess if such variations would characterize a natural system, or if they are enhanced by the variation in livestock presence. A final possible explanation could 542 543 be that the fluorescent layers are driven by a season of elevated in-situ microbial production on the



557

558 CONCLUSIONS AND IMPLICATIONS

559 Characterization of cave drip waters by full EEM matrices is a non-destructive, readily applied 560 technique for which PARAFAC analysis facilitates increasingly refined interpretations and contributes to 561 improved understanding of the natural dissolved organic matter dynamics in cave systems. We have 562 conducted a one-year monitoring of fluorescence in drip water at a monthly resolution in a cave system 563 with varying vegetation and land use at the surface above the cave. Our results indicate diverse 564 fluorescence contributions characterized by a five component PARAFAC model, which likely originate 565 from the overlying vegetation, sectors of livestock grazing above the cave, microbial processes and the 566 fossil bedrock. The contribution of fossil organic material in bedrock to drip water fluorescence is 567 suggested by EEMs of dissolved bedrock samples exhibiting a peak of Ex/Em 250/330 and 280/335, very 568 similar to one of the drip water components identified by PARAFAC models. To our knowledge, this is

the first study indicating a contribution of fossil fluorescent compounds to fluorescence in drip waters andpotentially the precipitating speleothems.

Across this cave system, the nature of fluorescence varies significantly among drip sites 571 572 characterized by different land cover and water reservoir times. Drip sites with dense tree cover and low 573 reservoir mixing times are characterized by the highest concentrations of humic-like and fulvic-like 574 fluorescence. Similar total fluorescence, but much lower humic-and fulvic-like components, 575 characterized sites with longer reservoir mixing times and mixed tree and fern cover. The site with 576 pasture above the drip site exhibited almost no humic-like fluorescence and the lowest total fluorescence. 577 For sites of comparable landcover and residence time, higher drip water flux (drip rate) correlates with a 578 higher total fluorescence concentration in drip water. Especially under long flow routes and long 579 reservoir times, drip sites fed by a low water flow, may lose a greater fraction of dissolved organic matter 580 to adsorption on karst host rock, reducing the final concentration arriving at the drip point in the cave. 581 The role of microbiological processes in the epikarst and within the caves, merits further exploration 582 because of the potential effect on interpretation of fluorescence properties of stalagmites. 583 In our data set, concentration of humic-like and fulvic-like fluorescence is constant and not 584 significantly enhanced seasonally in any drip sites. Higher concentrations of these components 585 accompany rainier periods only at a single drip site characterized by the shortest residence times. In 586 contrast, tyrosine-like and protein-like components of C2 and C4, hypthesized to originate from microbial 587 activity in the soil or epikarst, exhibit greater temporal variations in some drip sites, and increase in 588 concentration during drier periods across all drip sites. It is unclear if this trend reflects greater microbial 589 production during dry periods, or reduced dilution of a stable microbial population within the karst or 590 cave during dry periods.

591 These findings have several important implications for the interpretation of fluorescence and fine 592 scale annual to subannual fluorescent lamination in stalagmites. First, the surface vegetation signal of 593 humic like and fulvic like fluorescence may dominate the drip water fluorescence signal reaching 594 stalagmites, only in settings which feature low water reservoir times and limited microbial reworking. Our

data are consistent with a greater reduction in the concentration of fluorescent compounds in very slow flow drip sites compared to fast flow, likely due to the adsorption of fluorescent compounds to host karst rock. Interpretation of the fluorescent components in stalagmites therefore needs to consider the reprocessing reactions that operate between the surface environment and the cave. Additionally, a bedrock source of fossil fluorescent material, independent of the surface landscape or climate conditions at the time of stalagmite growth, may be significant in some cave systems.

601 Our finding of annual to sub-annual cyclic fluorescent layers in an actively growing stalagmite 602 beneath a drip characterized by invariant humic and fulvic-like components, challenges one conventional 603 interpretation for the origin of fluorescent laminae by a high autumnal influx of organic matter. Our 604 favored interpretation is that fluorescent laminae could be generated during seasons of slower stalagmite 605 growth in which an otherwise constant concentration of fluorescent components is concentrated over a 606 shorter distance in the stalagmite, generating higher average stalagmite fluorescence in the slower growth 607 portion of the year. In this case, the season of maximum fluorescence may not be identical among 608 multiple caves or even among multiple stalagmites in the same cave system, because season of stalagmite 609 growth depend on the timing and intensity of cave ventilation which can vary both among different caves 610 and within a given cave system. Thus, the fluorescent laminae should not be assumed to be a consistent 611 marker of a particular season nor used to reference the attribution of a particular season to observed cycles in δ^{13} C or trace elements. 612

613 Alternatively, fluorescent laminae could be dominated by variations in the non-humic and non-614 fulvic components of fluorescence. Certain of these components were systematically elevated during dry 615 intervals. If these components are most responsible for fluorescent lamination, then fluorescent laminae 616 may represent dry periods, exactly the opposite interpretation as the conventional hypothesis. In our sites, 617 variations in these components were not synchronous among multiple drip locations, indicating that this 618 process could also lead to seasonally diachronous fluorescent layers in different drip settings. 619 Nonetheless, additional monitoring studies in cave settings with less livestock influence would be 620 important to distinguish if large variations in drip water tyrosine-like, and protein-like components also

621 characterize systems in a more natural state. More generally, to more confidently reconcile the interpretation of fluorescent laminae in stalagmites, it will be important to identify which fluorescent 622 623 components are responsible for the annual to subannual scale lamination in stalagmites by EEM or other 624 analytical approaches. Some fluorescent components may be selectively incorporated in stalagmite calcite 625 and therefore play a larger role than predicted by their relative contribution to the drip water 626 fluorescence. Furthermore, it will be important to assess if the constancy of humic-like and fulvic-like 627 drip water concentrations seen in this cave is representative of other cave systems in which annual to 628 subannual fluorescent laminae are documented.

629 **COMPETING INTERESTS**

630 The authors declare that they have no known competing interests or personal relationships that631 could have appeared to influence the work reported in this paper.

632 ACKNOWLEDGEMENTS

633 This study was supported by ETH core funding and ETH Zürich [grant number ETH-1318-1].634

635 DATA AND CODE AVAILABILITY

Upon acceptance of this article, all data used in the study and codes for figure compilation will be uploaded to <u>https://github.com/lrndrs/Organics_Analysis</u>, last access: 30 December 2022) and given in the supplement provided with the article. The compiled data set of drip water organics and chemistry will also be uploaded to PANGEA (<u>https://www.pangaea.de/</u>, last access: 30 December 2022) and the PARAFAC model will be published in the OpenFluor database (<u>https://openfluor.lablicate.com</u>, last access: 30 December 2022).

643 **REFERENCES**

- 644 Aiken, G. R.: Fluorescence and dissolved organic matter: A chemist's perspective: Chapter 2, 35–74 pp.,
- 645 https://doi.org/10.1017/CBO9781139045452.005, 2014.
- 646 Alberts, J. J. and Takács, M.: Comparison of the natural fluorescence distribution among size fractions of
- 647 terrestrial fulvic and humic acids and aquatic natural organic matter, Org Geochem, 35, 1141–1149, 2004.
- 648 Álvarez, R., Ordóñez, A., Canteli, P., and de Miguel, E.: Unconventional gas resources in the Cantabrian
- 649 Zone (NW Spain): A comprehensive preliminary assessment, Geological Journal, 54, 2608–2620,
- 650 https://doi.org/10.1002/gj.3314, 2019.
- Baker, A. and Genty, D.: Fluorescence wavelength and intensity variations of cave waters, J Hydrol
- 652 (Amst), 217, 19–34, https://doi.org/10.1016/S0022-1694(99)00010-4, 1999.
- Baker, A., Smart, P. L., Edwards, R. L., and Richards, D. A.: Annual growth banding in a cave
- 654 stalagmite, Nature, 364, 518–520, 1993.
- Baker, A., Smith, C., Jex, C., Fairchild, I., Genty, D., and Fuller, L.: International Journal of Speleology
- Annually Laminated Speleothems: a Review, Int J Speleol, 37, 193–206, 2008.
- 657 Birdwell, J. E. and Engel, A. S.: Characterization of dissolved organic matter in cave and spring waters
- using UV-Vis absorbance and fluorescence spectroscopy, Org Geochem, 41, 270–280,
- 659 https://doi.org/10.1016/j.orggeochem.2009.11.002, 2010.
- Blyth, A. J., Baker, A., Collins, M. J., Penkman, K. E. H., Gilmour, M. A., Moss, J. S., Genty, D., and
- Drysdale, R. N.: Molecular organic matter in speleothems and its potential as an environmental proxy,
- 662 Quat Sci Rev, 27, 905–921, https://doi.org/10.1016/j.quascirev.2008.02.002, 2008.
- Blyth, A. J., Jex, C. N., Baker, A., Khan, S. J., and Schouten, S.: Contrasting distributions of glycerol
- dialkyl glycerol tetraethers (GDGTs) in speleothems and associated soils, Org Geochem, 69, 1–10, 2014.
- 665 Blyth, A. J., Hartland, A., and Baker, A.: Organic proxies in speleothems-New developments, advantages
- 666 and limitations, Quat Sci Rev, 149, 1–17, 2016.

- 667 Borsato, A., Frisia, S., Fairchild, I. J., Somogyi, A., and Susini, J.: Trace element distribution in annual
- 668 stalagmite laminae mapped by micrometer-resolution X-ray fluorescence: Implications for incorporation
- of environmentally significant species, Geochim Cosmochim Acta, 71, 1494–1512,
- 670 https://doi.org/10.1016/j.gca.2006.12.016, 2007.
- 671 Breecker, D. O., Payne, A. E., Quade, J., Banner, J. L., Ball, C. E., Meyer, K. W., and Cowan, B. D.: The
- 672 sources and sinks of CO2 in caves under mixed woodland and grassland vegetation, Geochim
- 673 Cosmochim Acta, 96, 230–246, 2012.
- Bro, R. and Kiers, H. A. L.: A new efficient method for determining the number of components in
- 675 PARAFAC models, J Chemom, 17, 274–286, https://doi.org/10.1002/cem.801, 2003.
- 676 Chen, M., Jung, J., Lee, Y. K., and Hur, J.: Surface accumulation of low molecular weight dissolved
- 677 organic matter in surface waters and horizontal off-shelf spreading of nutrients and humic-like
- 678 fluorescence in the Chukchi Sea of the Arctic Ocean, Science of the Total Environment, 639, 624–632,
- 679 https://doi.org/10.1016/j.scitotenv.2018.05.205, 2018.
- 680 Coble, P. G., Spencer, R. G. M., Baker, A., and Reynolds, D. M.: Aquatic Organic Matter Fluorescence,
- 681 75–122 pp., https://doi.org/10.1017/cbo9781139045452.006, 2014.
- 682 Cory, R. M. and McKnight, D. M.: Fluorescence spectroscopy reveals ubiquitous presence of oxidized
- and reduced quinones in dissolved organic matter, Environ Sci Technol, 39, 8142–8149,
- 684 https://doi.org/10.1021/es0506962, 2005.
- 685 Crameri, F., Shephard, G. E., and Heron, P. J.: The misuse of colour in science communication, Nat
- 686 Commun, 11, https://doi.org/10.1038/s41467-020-19160-7, 2020.
- 687 Cuss, C. W., Donner, M. W., Noernberg, T., Pelletier, R., and Shotyk, W.: EEM-PARAFAC-SOM for
- assessing variation in the quality of dissolved organic matter: simultaneous detection of differences by
- source and season, Environmental Chemistry, 16, 360–374, https://doi.org/10.1071/EN19016, 2019.
- 690 Dandrilli, J. and McConnell, J. R.: Polar ice core organic matter signatures reveal past atmospheric carbon
- 691 composition and spatial trends across ancient and modern timescales, Journal of Glaciology, 67, 1028-
- 692 1042, https://doi.org/10.1017/jog.2021.51, 2021.

- 693 Delavaud, E., Cotton, F., Akkar, S., Scherbaum, F., Danciu, L., Beauval, C., Drouet, S., Douglas, J.,
- Basili, R., Sandikkaya, M. A., and others: Toward a ground-motion logic tree for probabilistic seismic
- hazard assessment in Europe, J Seismol, 16, 451–473, 2012.
- Fairchild, I. J. and Baker, A.: Speleothem Science: From Process to Past Environments, John Wiley &Sons, 2012.
- 698 Hartland, A. and Zitoun, R.: Transition metal availability to speleothems controlled by organic binding
- ligands, Geochem Perspect Lett, 8, 22–25, https://doi.org/10.7185/geochemlet.1824, 2018.
- 700 Hartland, A., Fairchild, I. J., Müller, W., and Dominguez-Villar, D.: Preservation of NOM-metal
- 701 complexes in a modern hyperalkaline stalagmite: Implications for speleothem trace element
- 702 geochemistry, Geochim Cosmochim Acta, 128, 29–43, 2014.
- Heeb, B.: An all-in-one electronic cave surveying device, Cave Radio \& Electronics Group Journal, 72,
 8–10, 2009.
- Heeb, B.: A general calibration algorithm for 3-axis compass/clinometer devices, Cave Radio and
 Electronics Group Journal, 73, 2010.
- 707 Helms, J. R., Stubbins, A., Ritchie, J. D., Minor, E. C., Kieber, D. J., and Mopper, K.: Absorption spectral
- slopes and slope ratios as indicators of molecular weight, source, and photobleaching of chromophoric
- dissolved organic matter, Limnol Oceanogr, 53, 955–969, https://doi.org/10.4319/lo.2009.54.3.1023,
- 710 2008.
- Huang, Y., Fairchild, I. J., Borsato, A., Frisia, S., Cassidy, N. J., McDermott, F., and Hawkesworth, C. J.:
- Seasonal variations in Sr, Mg and P in modern speleothems (Grotta di Ernesto, Italy), Chem Geol, 175,
- 713 429–448, https://doi.org/https://doi.org/10.1016/S0009-2541(00)00337-5, 2001.
- 714 IAEA/WMO: Global Network of Isotopes in Precipitation. The GNIP Database., Accessible at:
- 715 http://www.iaea.org/water, 2022.
- 716 Jin, J. and Zimmerman, A. R.: Abiotic interactions of natural dissolved organic matter and carbonate
- aquifer rock, Applied Geochemistry, 25, 472–484, https://doi.org/10.1016/j.apgeochem.2009.12.012,
- 718 2010.

- 719 Köppen, W.: Die Wärmezonen der Erde, nach der Dauer der heissen, gemässigten und kalten Zeit und
- nach der Wirkung der Wärme auf die organische Welt betrachtet, Meteorologische Zeitschrift, 1, 5–226,
- 721 1884.
- 722 Kost, O., González-Lemos, S., Rodriguez-Rodriguez, L., Sliwinski, J., Endres, L., Haghipour, N., and
- 523 Stoll, H.: Relationship of seasonal variations in drip water δ 13 CDIC, δ 18 O and trace elements with
- surface and physical cave conditions of La Vallina Cave, NW Spain, Hydrology and Earth System
- 725 Sciences Disc., 1–42, https://doi.org/10.5194/hess-2022-386, 2022.
- 726 Kwiecien, O., Braun, T., Brunello, C. F., Faulkner, P., Hausmann, N., Helle, G., Hoggarth, J. A., Ionita,
- 727 M., Jazwa, C. S., Kelmelis, S., Marwan, N., Nava-Fernandez, C., Nehme, C., Opel, T., Oster, J. L.,
- 728 Perşoiu, A., Petrie, C., Prufer, K., Saarni, S. M., Wolf, A., and Breitenbach, S. F. M.: What we talk about
- vhen we talk about seasonality A transdisciplinary review, Earth Sci Rev, 225, 103843,
- 730 https://doi.org/10.1016/j.earscirev.2021.103843, 2022.
- 731 Lakowicz, J. R.: Principles of fluorescence spectroscopy, Springer, 2006.
- 732 Lechleitner, F. A., Dittmar, T., Baldini, J. U. L., Prufer, K. M., and Eglinton, T. I.: Molecular signatures
- of dissolved organic matter in a tropical karst system, Org Geochem, 113, 141–149,
- 734 https://doi.org/10.1016/j.orggeochem.2017.07.015, 2017.
- 735 Lindeman, I., Hansen, M., Scholz, D., Breitenbach, S. F. M., and Hartland, A.: Effects of organic matter
- complexation on partitioning of transition metals into calcite: Cave-analogue crystal growth experiments,
- 737 Geochim Cosmochim Acta, 317, 118–137, https://doi.org/10.1016/j.gca.2021.10.032, 2021.
- 738 Luetscher, M., Moseley, G. E., Festi, D., Hof, F., Edwards, R. L., and Spötl, C.: A Last Interglacial
- 739 speleothem record from the Sieben Hengste cave system (Switzerland): Implications for alpine
- paleovegetation, Quat Sci Rev, 262, https://doi.org/10.1016/j.quascirev.2021.106974, 2021.
- 741 Massicotte, P.: eemR: tools for pre-processing emission-excitation-matrix (EEM) fluorescence data, R
- package version, 1, 2019.

- 743 McGarry, S. F. and Baker, A.: Organic acid fluorescence: Applications to speleothem
- palaeoenvironmental reconstruction, Quat Sci Rev, 19, 1087–1101, https://doi.org/10.1016/S0277-
- 745 3791(99)00087-6, 2000.
- 746 McKnight, D. M., Boyer, E. W., Westerhoff, P. K., Doran, P. T., Kulbe, T., and Andersen, D. T.:
- 747 Spectrofluorometric characterization of dissolved organic matter for indication of precursor organic
- 748 material and aromaticity, Limnol Oceanogr, 46, 38–48, https://doi.org/10.4319/lo.2001.46.1.0038, 2001.
- 749 Moreno, A., Iglesias, M., Azorin-Molina, C., Pérez-Mejías, C., Bartolomé, M., Sancho, C., Stoll, H.,
- 750 Cacho, I., Frigola, J., Osácar, C., Muñoz, A., Delgado-Huertas, A., Blade, I., and Vimeux, F.: Spatial
- variability of northern Iberian rainfall stable isotope values: Investigating climatic controls on daily and
- monthly timescales, Atmospheric Chemistry and Physics Discussions, 1–34, https://doi.org/10.5194/acp-
- 753 2020-861, 2020.
- 754 Mudarra, M., Andreo, B., and Baker, A.: Characterisation of dissolved organic matter in karst spring
- 755 waters using intrinsic fluorescence: Relationship with infiltration processes, Science of the Total
- 756 Environment, 409, 3448–3462, https://doi.org/10.1016/j.scitotenv.2011.05.026, 2011.
- 757 Murphy, K. R., Bro, R., and Stedmon, C. A.: Chemometric Analysis of Organic Matter Fluorescence,
- 758 339–375 pp., https://doi.org/10.1017/cbo9781139045452.016, 2014a.
- 759 Murphy, K. R., Stedmon, C. A., Wenig, P., and Bro, R.: OpenFluor- An online spectral library of auto-
- fluorescence by organic compounds in the environment, Analytical Methods, 6, 658–661,
- 761 https://doi.org/10.1039/c3ay41935e, 2014b.
- 762 O'Brien, B. J.: 'After -glow' of cave calcite, National Speleological Society, 18, 50–51, 1956.
- 763 Ohno, T.: Fluorescence inner-filtering correction for determining the humification index of dissolved
- 764 organic matter, Environ Sci Technol, 36, 742–746, https://doi.org/10.1021/es0155276, 2002.
- 765 Orland, I. J., Bar-Matthews, M., Ayalon, A., Matthews, A., Kozdon, R., Ushikubo, T., and Valley, J. W.:
- 766 Seasonal resolution of Eastern Mediterranean climate change since 34ka from a Soreq Cave speleothem,
- 767 Geochim Cosmochim Acta, 89, 240–255, https://doi.org/10.1016/j.gca.2012.04.035, 2012.

- 768 Pearson, A. R., Hartland, A., Frisia, S., and Fox, B. R. S.: Formation of calcite in the presence of
- dissolved organic matter: Partitioning, fabrics and fluorescence, Chem Geol, 539, 119492,
- 770 https://doi.org/10.1016/j.chemgeo.2020.119492, 2020.
- 771 Peel, M. C., Finlayson, B. L., and McMahon, T. A.: Updated world map of the Köppen-Geiger climate
- 772 classification, Hydrol Earth Syst Sci, 11, 1633–1644, https://doi.org/10.5194/hess-11-1633-2007, 2007.
- Perrette, Y., Delannoy, J. J., Desmet, M., Lignier, V., and Destombes, J. L.: Speleothem organic matter
- content imaging. The use of a Fluorescence Index to characterise the maximum emission wavelength,
- 775 Chem Geol, 214, 193–208, https://doi.org/10.1016/j.chemgeo.2004.09.002, 2005.
- Philibert, M., Luo, S., Moussanas, L., Yuan, Q., Filloux, E., Zraick, F., and Murphy, K. R.: Drinking
- 777 water aromaticity and treatability is predicted by dissolved organic matter fluorescence, Water Res, 220,
- 118592, 2022.
- Proctor, C. J., Baker, A., Barnes, W. L., and Gilmour, M. A.: A thousand year speleothem proxy record of
- North Atlantic climate from Scotland, Clim Dyn, 16, 815–820, https://doi.org/10.1007/s003820000077,
 2000.
- 782 Pucher, M., Wünsch, U., Weigelhofer, G., Murphy, K., Hein, T., and Graeber, D.: StaRdom: Versatile
- real software for analyzing spectroscopic data of dissolved organic matter in R, Water (Switzerland), 11, 1–
- 784 19, https://doi.org/10.3390/w11112366, 2019.
- 785 Quiers, M., Perrette, Y., Chalmin, E., Fanget, B., and Poulenard, J.: Geochemical mapping of organic
- carbon in stalagmites using liquid-phase and solid-phase fluorescence, Chem Geol, 411, 240–247,
- 787 https://doi.org/10.1016/j.chemgeo.2015.07.012, 2015.
- 788 Ramseyer, K., Miano, T. M., D'Orazio, V., Wildberger, A., Wagner, T., and Geister, J.: Nature and origin
- of organic matter in carbonates from speleothems, marine cements and coral skeletons, Org Geochem, 26,
- 790 361–378, https://doi.org/10.1016/S0146-6380(97)00008-9, 1997.
- 791 Reynolds, D. M.: The Principles of Fluorescence, 3–34 pp.,
- 792 https://doi.org/10.1017/cbo9781139045452.004, 2014.

- 793 Rutlidge, H., Baker, A., Marjo, C. E., Andersen, M. S., Graham, P. W., Cuthbert, M. O., Rau, G. C.,
- 794 Roshan, H., Markowska, M., Mariethoz, G., and Jex, C. N.: Dripwater organic matter and trace element
- 795 geochemistry in a semi-arid karst environment: Implications for speleothem paleoclimatology, Geochim
- 796 Cosmochim Acta, 135, 217–230, https://doi.org/10.1016/j.gca.2014.03.036, 2014.
- 797 Rutlidge, H., Andersen, M. S., Baker, A., Chinu, K. J., Cuthbert, M. O., Jex, C. N., Marjo, C. E.,
- 798 Markowska, M., and Rau, G. C.: Organic characterisation of cave drip water by LC-OCD and
- fluorescence analysis, Geochim Cosmochim Acta, 166, 15–28, https://doi.org/10.1016/j.gca.2015.05.042,
 2015.
- 801 Schindelin, J., Arganda-Carreras, I., Frise, E., Kaynig, V., Longair, M., Pietzsch, T., Preibisch, S.,
- 802 Rueden, C., Saalfeld, S., Schmid, B., and others: Fiji: an open-source platform for biological-image
- analysis, Nat Methods, 9, 676–682, 2012.
- 804 Shopov, Y. Y., Ford, D. C., and Schwarcz, H. P.: Luminescent microbanding in speleothems: High-
- resolution chronology and paleoclimate, Geology, 1994.
- 806 Sliwinski, J. T. and Stoll, H. M.: Combined fluorescence imaging and LA-ICP-MS trace element
- 807 mapping of stalagmites: Microfabric identification and interpretation, Chem Geol, 581, 120397,
- 808 https://doi.org/10.1016/j.chemgeo.2021.120397, 2021.
- 809 Sliwinski, J. T., Kost, O., Endres, L., Iglesias, M., Haghipour, N., González-Lemos, S., and Stoll, H. M.:
- 810 Exploring soluble and colloidally transported trace elements in stalagmites: the strontium-yttrium
- 811 connection, Geochim Cosmochim Acta, https://doi.org/https://doi.org/10.1016/j.gca.2022.12.023, 2022.
- 812 Spangenberg, M., Bryant, J. I., Gibson, S. J., Mousley, P. J., Ramachers, Y., and Bell, G. R.: Ultraviolet
- absorption of contaminants in water, Sci Rep, 11, 1–8, 2021.
- 814 State Meteorological Agency of Spain: AEMET OpenData, Spain, 2021.
- 815 Stedmon, C. A. and Bro, R.: Characterizing dissolved organic matter fluorescence with parallel factor
- 816 analysis: a tutorial. (2008) 6:572-579, 2008.

- 817 Stedmon, C. A. and Cory, R. M.: Biological Origins and Fate of Fluorescent Dissolved Organic Matter in
- 818 Aquatic Environments, Aquatic Organic Matter Fluorescence, 278–300,
- 819 https://doi.org/10.1017/cbo9781139045452.013, 2014.
- 820 Stoll, H., Mendez-Vicente, A., Gonzalez-Lemos, S., Moreno, A., Cacho, I., Cheng, H., and Edwards, R.
- 821 L.: Interpretation of orbital scale variability in mid-latitude speleothem δ 180: Significance of growth rate
- 822 controlled kinetic fractionation effects, Quat Sci Rev, 127, 215–228,
- 823 https://doi.org/10.1016/j.quascirev.2015.08.025, 2015.
- 824 Wilkinson, K. J., N~gre, J.-C., and Buffie, J.: Contaminant Hydrology Coagulation of colloidal material
- 825 in surface waters: the role of natural organic matter, Journal of Contaminant Hydrology, 229–243 pp.,
- 826 1997.
- 827 Wünsch, U. J., Murphy, K. R., and Stedmon, C. A.: Fluorescence quantum yields of natural organic
- 828 matter and organic compounds: Implications for the fluorescence-based interpretation of organic matter
- 829 composition, Front Mar Sci, 2, 1–15, https://doi.org/10.3389/fmars.2015.00098, 2015.
- 830 Wünsch, U. J., Bro, R., Stedmon, C. A., Wenig, P., and Murphy, K. R.: Emerging patterns in the global
- distribution of dissolved organic matter fluorescence, Analytical Methods, 11, 888–893,
- 832 https://doi.org/10.1039/c8ay02422g, 2019.
- 833 Xia, F., Liu, Z., Zhao, M., Li, Q., Li, D., Cao, W., Zeng, C., Hu, Y., Chen, B., Bao, Q., Zhang, Y., He, Q.,
- Lai, C., He, X., Ma, Z., Han, Y., and He, H.: High stability of autochthonous dissolved organic matter in
- karst aquatic ecosystems: Evidence from fluorescence, Water Res, 220, 118723,
- 836 https://doi.org/10.1016/j.watres.2022.118723, 2022.
- 837 Yamashita, Y. and Jaffé, R.: Characterizing the interactions between trace metals and dissolved organic
- matter using excitation-emission matrix and parallel factor analysis, Environ Sci Technol, 42, 7374–7379,
- 839 https://doi.org/10.1021/es801357h, 2008.
- 840 Zhou, Z., Liu, Z., and Guo, L.: Chemical evolution of Macondo crude oil during laboratory degradation as
- characterized by fluorescence EEMs and hydrocarbon composition, Mar Pollut Bull, 66, 164–175,
- 842 https://doi.org/10.1016/j.marpolbul.2012.09.028, 2013.

843 LIST OF TABLES

844

Tab. 1

Component	Ex. Max	Em. Max	Description	Fmax c	orrela	ation		
(-)	(nm)	(nm)	(from literature and openfluor)	C1	C2	C3	C4	C5
C1	240,320	408	Humic-like ¹ , higher molecular weight vegetation	1	0.04	0.5	-0.12	0.62
C2	265	291	tyrosine-like ¹ , simple tannin ^{2,8} found in cold waters ⁷	0.04	1	0.5	0.5	0.08
C3	240, 280	333	planktonic Humic-like ¹ , microbially processed ³ , oil ⁴ indigenous OM from bedrock	0.5	0.5	1	-0.02	0.28
C4	255	277	amino acid-like ¹ , close spectral match to phenylalanine ⁹	-0.12	0.5	-0.02	1	-0.1
C5	256, 370	455	Soil fulvic acid ⁵ , terrestrial higher plants ⁶	0.62	0.08	0.28	-0.1	1

845

846 *Table 1:*

PARAFAC model component properties. Ex_{max} and Em_{max} give wavelength of peak excitationemission pairs (double peaks for humic-like components). Pearson correlation between individual
components across all samples is given in the column Fmax correlation. EEM spectra are described from
comparison to literature and the OpenFluor database. References from literature are ¹Coble et al., 2014 ;
²Aitken et al., 2014, ³Xia et al., 2022; ⁴Zhou et al., 2013 ; ⁵Yamashita and Jaffé, 2008 and ⁶Cory and
McKnight, 2005. Derived from OpenFluor are ⁷Chen et al., 2018; ⁸Dandrilli and McConnell, 2021 and
⁹Wünsch et al., 2015.

Tab. 2

Location	Cave Temp. (°C)	pCO2 (ppm)	Cover thickness (m)	Drip rate (ml/min)	δ ¹⁸ Ο	Overall fluorescence (R.U.)	FI	Spectral slope	C3:(C1+C5) ratio
(Abbr.)	monthly mean (1SD)	Max (Min)		Mean (1SD)	variance	Mean (1SD)	Mean (1SD)	Mean (1SD)	Mean (1SD)
Forest (FOR)	14.21 (0.56)	1496 (461)	14.47	0.35 (0.18)	0.477	47.4 (13.35)	1.66 (0.02)	1.14 (0.3)	0.25 (0.17)
Snowball (SNO)	14.21 (0.56)	1496 (461)	14.35	-	-	37.86 (12.15)	1.60 (0.03)	0.62 (0.21)	0.18 (0.14)
Gloria (GLO)	14.44 (0.27)	1582 (569)	8.72	0.22 (0.08)	0.09	22.56 (24.64)	1.65 (0.12)	0.99	0.76 (0.7)
Gravel (GRA)	13.9 (0.22)	2416 (509)	13.29	0.72 (0.38)	0.08	18.64 (19.32)	1.52 (0.16)	1.15	0.37 (0.25)
Skyscraper (SKY)	13.89 (0.17)	5390 (510)	27.62	89.22 (24.25)	0.05	26.76 (20.35)	1.72 (0.16)	0.83	0.47 (0.50)
Playground (PLA)	13.72 (0.16)	4199 (525)	29.45	1.22 (0.916)	0.04	20.34 (5.73)	1.73 (0.14)	-0.48	0.96 (0.57)
Mushroom (MUS)	13.6	-	23.21	-	-	10.87 (0.23)	1.55 (0.2)	0.91	0.22 (0.46)

855

856

857 *Table 2:*

858 Sampling locations discussed in this study, their overall cave properties and mean optical values.

859 Variability is given in the form of one standard deviation (1 SD). Overall fluoresecence over the entire

860 EEM is given in raman units (R.U.). FI is the fluorescence index (McKnight et al. 2001). Extended

861 description of the overlying vegetation and cave setting can be found in the Supplementary Table 1.

862

864 LIST OF FIGURES

865



866

867 *Figure 1.*

868 Location, sampling map of locations in cave, and landcover and climatic setting. (A) Reference 869 map with red circle denoting the cave location. (B) Plan map of the cave upper gallery (green outline) and 870 lower gallery (orange outline) indicating the location of drip water sampling (labeled drop symbols) 871 overlaid on aerial image from 2017 (Plan Nacional Ortofotografía Aérea - PNOA of Spanish IGN; 872 available at https://centrodedescargas.cnig.es, last access: 30 December 2022), and showing topographic 873 contours (white lines) with elevation above sea level (m). Pasture use and large trees are noted, as well as 874 the position of root systems penetrating the cave. (C) Cross section (X to X') of cave with projected 875 sampling locations. (D) Vegetation above sampling location FOR (May 2021), (E) Vegetation above

- 876 Sampling location PLA (May 2021). (F) Monthly mean air temperature and precipitation from Llanes
- 877 weather station over sampling period (State Meteorological Agency of Spain, 2021).



880 *Figure 2*.

Dripwater components and bedrock endmember EEM. A) Drip water 5-component PARAFAC model and its spectral loadings. Fluorescence intensity is normalized to unity. Spectral loadings maxima in emission and excitation wavelengths are indicated with solid and dashed lines, respectively. B) EEM of dissolved organic-rich bedrock. Fluorescence intensity is given in arbitrary unit (a.u.). Yellow triangles indicate peak positions of drip water PARAFAC component C3 for comparison.



887 *Figure 3*.

Absorption properties of drip water samples grouped by location. A) Loess smoothed median
absorption spectra for all locations. B) Spectral slope ratio values for individual absorbance
measurements.



893 *Figure 4.*

A) Mean and standard deviation (SD) of Excitation-Emission Matrices (EEM) for each sampling
location compiled over the monitoring period. Fluorescent intensities are given in Raman units (R.U.). B)
mean intensity of all PARAFAC components for each sampling location, stacked to illustrate the total
PARAFAC characterized fluorescence intensity at each location. Comp. 1 and Comp. 5 are described as
"humic-like" and "fulvic-like", respectively. Comp. 3 is described as "fossil like", Comp. 2 as "tyrosinelike", and Comp. 4 as "protein like".





902 Overview of the relationship between fluorescence and cave settings. A) comparison of the 903 summed C1 and C5 (humic-like and fulvic-like) components plotted vs ascending thickness of bedrock 904 cover over the drip sampling site. B) comparison of summed C1 and C5 components plotted vs the drip 905 water δ^{18} O variance, which increases for shorter drip water residence times. C) comparison of the ratio of 906 C3/(C1+C5) ratio of reworked or fossil components to humic-like and fulvic-like components, plotted vs 907 the drip water δ^{18} O variance, which increases for shorter drip water residence times.



908



Schematic of processes contributing to observed drip water fluorescence, from vegetation through
soil, epikarst, and into cave. Red labels denote potential end member sources of different fluorescent
organic component classes. Lavender boxes highlights proposed interactions between climate,
vegetation/plant litter and bedrock including dissolution or leaching of bedrock by respired CO2, and

adsorption of fluorescent matter to bedrock, as well as cave conditions potentially affecting incorporation

915 of fluorescent organic matter in speleothems.





918 Temporal evolution of the 5 components of the PARAFAC model throughout the monitored

919 period at each site. Each component y axis is given in normalized PARAFAC fluorescence intensity.





922 Relationship of climate conditions 10 days prior to drip water sampling date versus fluorescence 923 of each component for the five locations FOR, GLO, GRA, PLA and SKY. Upper row contrasts the 924 fluorescence during four dates of dry weather in the 10 days prior to sampling, with fluorescence during 925 four dates of wet weather in the 10 days prior to drip water sampling. The sample size of each group 926 (n=4) was limited by total length of time series FOR. Lower row shows an analogous comparisons of 927 groups during 4 dates of warm average surface air temperatures in the 10 days prior to drip water 928 sampling vs 4 dates of cold average surface air temperatures in the 10 days prior to drip water sampling. 929 The boxplot shows the median, first and third quartiles; the attached whisker extends from the hinges to 930 the extreme value no further than 1.5x inter-quartile range and data outside of whisker is plotted as single 931 points.







834 Relationship between fluorescence parameters and indicators of the dissolution and karst 935 conditions from drip water geochemistry. A) comparison of fluorescence components with the δ^{13} C of 936 DIC in drip water, indicator of soil and epikarst CO₂ and respiration rates. B) comparison of fluorescence 937 components with the drip water Sr concentration, an indicator of the extent of dissolution of karst 938 bedrock. C) comparison of fluorescence components with the total drip water Cu concentration. 939 Dripwater Cu is dominantly in the form of colloidal organic complexes.

Non-peer reviewed preprint, submitted to EarthArXiv





Solid-phase fluorescence features of actively growing speleothems and bedrock from La Vallina
Cave. A), stalagmite PG-2, sampled 5 m from drip water point PLA. B) stalagmite SKY, sampled
beneath the drip water sample point SKY. C) stalagmite SNO, sampled beneath the drip water sample
point SNO. D) polished bedrock. CLSM images were created with a Olympus Fluoview 3000 generated
by Excitation at 488 nm and detection at 490-590 nm; images were and post-processed with Fiji
(Schindelin et. al, 2012) and Adobe illustrator (Adobe Inc, 2019). Bomb spike age models for active
stalagmites are presented in Sliwinski et al. (2022)



Supplementary Figure 1. Comparison of dripwater sample FOR collected 13 November 2020. Left image is EEM acquisition made on 20 November 2020, and the right image is acquisition on 14 July 2021.



Supplementary Figure 2. Results from location-wise 3 component PARAFAC model reveal the presence of all five peaks as dominant features in single locations, but also across locations. Therefore, the 5-component model is able to represent the variability of the full dataset to a large degree. In detail: featuring high SSC values from 0.6-0.99, variability in location FOR is most similar to components C2,C5,C3; in GLO to C2,C1,C4; in GRA to C4,C5,(+Outlier); in PLA to C1,C3,C5 and in SKY to C3,C1,C4.



Supplementary Figure 3. Simplified to a humic-like and a protein-like component, the difference in intensity distribution in protein-like fluorescence between wet and dry conditions over the entire dataset is significant.



Supplementary Figure 4. Relationship between fluorescence parameters and indicators of the dissolution and karst conditions from dripwater geochemistry. Upper row, comparison of fluorescence components with dripwater Ca concentration, an indicator of the extent of dissolution of karst bedrock. Lower row, comparison of fluorescence components with the total dripwater Y concentration. Dripwater Y is dominantly in the form of colloidal organic complexes.

Supplementary Table 1. Description of vegetation and conditions above each drip location.

Location	Vegetation and conditions above/at sampling location				
(Abbr.)					
Forest (FOR)	Oak, chestnut and large eucalyptus trees; also Fern, brambles and gorse. Roots of eucalyptus tree reach cave ceiling.				
Snowball (SNO)	Oak, chestnut and large eucalyptus trees; also Fern, brambles and gorse. Roots of eucalyptus tree reach cave ceiling.				
Gloria (GLO)	Vegetation is dominated by Bushes and Fern, two isolated trees (eucalyptus and oak)				
	Doline is close by. Eucalyptus roots reach cave ceiling.				
Gravel (GRA)	Vegetation is dominated by Bushes and Fern, two isolated trees (eucalyptus and oak)				
	Doline is close by. Eucalyptus roots reach cave ceiling.				
	cave ceiling filled with helictites and eccentric soda straws				
Skyscraper (SKY)	Pasture, regularly used by cattle.				
	Several sink holes are nearby.				
Playground (PLA)	Pasture, regularly used by cattle.				
	Several sink holes are nearby.				
Mushroom (MUS)	Below hill slope, intermediate cover of some isolated trees and ferns.				

Supplementary Equation 1

Beer's law:

$$\alpha(\lambda) = 2.303 \, A(\lambda)/l$$

where $\alpha = absorption (m^{-1})$, A=Absorbance, l=cuvette pathlength (m) and $\lambda = wavelength (nm)$.