

1 **Cohesional behaviours in volcanic material and the implications for deposit architecture.**

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18 **Abstract**

19 Pyroclastic Density Currents (PDCs) are hazardous, multiphase currents of heterogeneous volcanic material  
20 and gas. Their high mobility can be partially attributed to fluidisation mechanisms. Moisture (as liquid or gas)  
21 can enter a PDC through external (e.g., interaction with bodies of water) or internal (e.g., initial eruptive  
22 activity style) processes and presence of moisture can be recorded within distinct deposit layers. We use  
23 analogue experiments to explore the behaviour of volcanic material with increasing moisture percentages  
24 from 0.00 – 10.00%. Our results show that: 1) the cohesivity of ignimbrite material changes with the addition  
25 of small amounts of moisture; 2) Small increases in moisture content change the flow behaviour from a free-  
26 flowing material to a non-flowable material; 3) changes in moisture can affect the formation of gas escape  
27 structures, and fluidisation profiles, 4) gas flow through a deposit can lead to a moisture profile and resulting  
28 mechanical heterogeneity within the deposit and 5) where gas escape structure growth is hindered by  
29 cohesivity driven by moisture, pressure can increase and release in an explosive fashion. This work highlights  
30 how a suite of dynamic and varied gas escape morphologies can form within the deposit resulting from  
31 moisture content heterogeneity, explaining variation in gas escape structures as well as providing a potential  
32 mechanism for secondary eruptions.

33 Key Words: Cohesion, Gas escape, Fluidisation, Secondary eruptions, Pyroclastic Density Currents, Deposit

## 34 Introduction

35 Pyroclastic density currents (PDC) are hazardous, multiphase, rapidly moving, often high-temperature currents  
36 of heterogeneous volcanic material and gas. The high mobility of PDCs has been attributed to fluidisation  
37 (Sparks, 1976; 1978; Wilson, 1984; Branney and Kokelaar, 1992, 2002; Breard et al., 2023) where the upward  
38 movement of gas supersedes the force of gravity and supports the flow (Sparks, 1976; Branney and Kokelaar,  
39 2002; Cocco et al., 2014). The ability of a current to fluidise and flow can be described by its ‘flowability’, which  
40 depends upon interparticle forces (Van der Waals, electrostatic or capillary forces). These forces can be  
41 influenced by bulk composition and material physical properties such as particle size, density, shape, and  
42 moisture content (Rios., 2006; Leturia et al., 2014).

43 Fluidisation in PDCs can be initiated during formation and maintained throughout the course of the flow by  
44 continued fluidisation mechanisms such as substrate evaporation (i.e., steam generated from interaction with  
45 surfaces with moisture content or bodies of water), bulk self-fluidisation or ambient air entrainment (Sparks,  
46 1978; Branney and Kokelaar, 2002; Chedeville and Roche, 2015; Breard et al., 2023). Sedimentation fluidisation  
47 (or, hindered settling) and particle-self fluidisation is where compaction from particle settling and sorting  
48 causes interstitial fluid movement. On or after deposition, the material will defluidise which can lead to  
49 segregation of material through gas escape structures, which form concentrated areas of particle sorting (i.e.,  
50 fines depleted elutriation pipes) and segregation of particles (Wilson, 1980; Cas and Wright, 1991).

51 Previous analogue investigations into fluidisation behaviours of volcanic material and segregations structures  
52 have been completed on dry (0%; Wilson 1980; 1984) and saturated (80+-15%; Roche et al., 2001) natural  
53 ignimbrite (herein defined as the deposits of PDCs, typically consisting of poorly-sorted ash and pumice or  
54 scoria (Giordano & Cas 2021). Experiments completed by Wilson (1980; 1984) used non-cohesive, poorly-  
55 sorted ignimbrite mixtures, and added an influx of gas into the deposit. This resulted in poor fluidisation  
56 behaviours and the formation of gas escape structures determined by particle size and density. Roche et al  
57 (2001) explored contrasting aqueous fluidisation in a water-saturated deposit of volcanic material. These  
58 results determined that fluid-escape pipes form readily with low water flux and contain localised segregation  
59 of particle sizes and densities. From both experiments, we can determine that natural ignimbrite material will  
60 demonstrate aggregative (i.e., inhomogeneous; Branney and Kokelaar, 2002) fluidisation resulting from  
61 particle size and density range, irrespective of the fluidising medium.

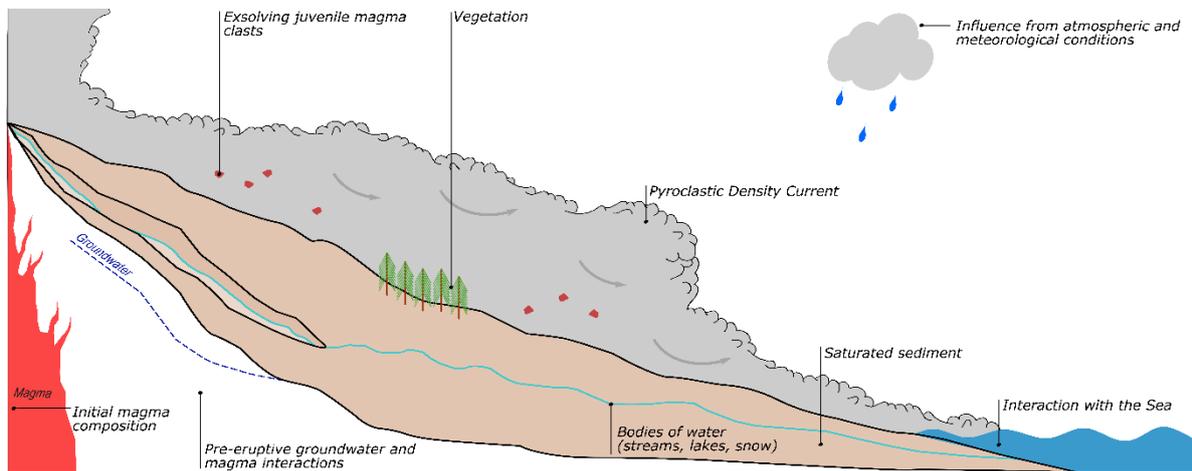
62 Understanding how moisture impacts powder material has important industrial applications. Experiments  
63 have explored fluidisation behaviours of industrial material with the addition of small volumes of moisture  
64 (i.e., through adsorption of ambient humidity). With the introduction of moisture into a material, Van der  
65 Waals forces are no longer dominant, and liquid bridges connect particles through capillary cohesion; resulting  
66 in poor fluidisation behaviours (Wormsbecker and Pugsley, 2008; Ludwig et al., 2020; Yehuda and Kalman,  
67 2020). A study by Wormsbecker and Pugsley (2008) looked at gas fluidisation behaviours on a semi-saturated  
68 (30, 20, 15 and 5 wt.% moisture) powder. Results showed a dominant change in fluidisation behaviour

69 associated with the addition of moisture, which were observed in conjunction with the drying states of the  
 70 material from 30 to 5 wt.%.

71 The experiments detailed herein experimentally assess the impact of the addition of small volumes of  
 72 moisture within natural volcanic material. We identify the resulting variations in terms of fluidisation and  
 73 particle segregation behaviours.

74 **Moisture in PDCs and their resulting deposits**

75 Moisture (i.e., water vapour or liquid water) can enter a PDC system during formation at source or as they  
 76 propagate (**Error! Reference source not found.**). Eruption columns can be water-rich due to  
 77 phreatomagmatism (Self and Sparks, 1978; Hurwitz, 2003; Houghton et al., 2015) or atmospheric conditions  
 78 (Vecino et al., 2022). During transport, internal clasts of juvenile magma will exsolve and release water vapour  
 79 and other volatiles. Experiments have highlighted how magmatic clasts may still hold residual water content of  
 80 0.6 – 0.8 wt. % during transport (Sparks et al., 1978).



81

82 **Fig.1** a PDC interacting with sources of moisture across a landscape which have the potential to enter the PDC  
 83 system and resulting deposits.

84 Externally, moisture may be introduced through a dynamic mix of atmospheric (e.g., humidity; Pepin et al.,  
 85 2017; Camuffo, 2019), topographic (e.g., height; Barclay et al., 2006; Duane et al., 2008; Hartman, 2016),  
 86 climatic (e.g., global location; Barclay et al., 2006) and meteorological conditions (e.g., precipitation).

87 Furthermore, periods of intense rainfall have been suspected and observed to affect the onset of volcanic  
 88 activity (Barclay et al., 2006; Sahoo et al., 2022 and references therein). Matthews et al., (2009) documented  
 89 that within 24 hours of heavy rainfall, the probability of lava dome collapse at Soufriere Hills Volcano,  
 90 Montserrat (during the period 1998-2003) increased, resulting in higher moisture availability to the resulting  
 91 PDCs.

92 Interaction with external bodies of water (i.e., streams, lakes, sea, snow; Darteville et al., 2002; Cole et al.,  
 93 1998; 2002), water saturated substrate (Moyer and Swanson, 1987; Brown and Branney, 2013; Gilbertson et

94 al., 2020) or by the incorporation of vegetation (as observed in Mount Pelé, 1902; Mount St Helens, 1980;  
95 Montserrat, 2002 and Fuego Volcano, 2018 ) can also contribute to moisture within the flow system.  
96 Therefore, we expect moisture content in PDCs to be highly variable in time and space.

97 The presence of moisture within PDCs can be demonstrated through varying degrees within an ignimbrite  
98 deposit. Moisture has been linked to the formation of wet ash aggregates (e.g., pellets) in ignimbrite deposits  
99 (Brown et al., 2010), by the presence of elutriation pipes that are derived from areas of evaporating moisture  
100 (i.e., vegetation or water-laden sediments) or by secondary hydroeruptions forming in deposits overlying  
101 moisture-rich areas (e.g., Mount St. Helens; Moyer and Swanson, 1987). The influence of these relatively small  
102 additions of moisture into a PDC system has been largely ignored in analogue and experimental studies, due to  
103 the difficulty associated using and controlling the characteristics of cohesive material. Therefore, prior to the  
104 experiments, material is often dried to remove any residual moisture (Druitt et al., 2004, 2007; Girolami et al.,  
105 2008; 2015).

### 106 **Capillary Cohesion**

107 The presence of moisture in a PDC, or in a subsequent deposit, will result in cohesive forces within the  
108 interparticulate space. A PDC can reach temperatures  $> 1000^{\circ}\text{C}$  and the resulting deposit can maintain high  
109 temperatures for extended periods of time (Dufek, 2016; Riehle et al., 1995), and it has been assumed that at  
110 these temperatures the dominant cohesive forces will be electrostatic and Van der Waals forces (Branney and  
111 Kokelaar, 2002). However, with increasing distance and entrainment, temperatures will lower (Benage et al.,  
112 2016) and the introduction of moisture will likely lead to the formation of capillary bridges ('capillary  
113 condensation'; Ma et al., 2019), resulting in a change of the dominant interparticulate forces . This is observed  
114 by Telling et al., (2013), where electrostatic attraction has been observed to be dominant only where humidity  
115 was lower than 71% and by Chigira and Yokoyama (2005), where capillary cohesion became the dominant  
116 cohesive force with the addition of moisture into a granular material.

117 Previous studies have shown that an increase in water content and moisture leads to a drastic change in the  
118 physical properties of a bulk material. For example, in sands, capillary forces were seen to affect the tensile  
119 strength of a material until reaching a water saturated state (Kim and Sture, 2008; Chen et al., 2021).

120 Therefore, at lower temperatures, it is highly likely that the introduction of moisture into the dynamic  
121 (pyroclastic current) and static (deposited sedimentary packages from a current) regions will introduce  
122 variations in material properties (e.g., as a sedimentary package). Changes in tensile strength may determine  
123 how resistant a material is to shear and erode and are important in understanding the flow properties of a  
124 material (Pierrat and Caram, 1997; LaMarche et al., 2016). Within a PDC deposit, such changes may also  
125 influence defluidisation through gas escape.

126 Here, we investigate how capillary cohesion, through the introduction of water, may affect volcanic material in  
127 a static state. We present the results of analogue experiments that test how changes in cohesion of volcanic

128 material affects its fluidisation. Our results provide new and novel insights into the variation of gas escape  
129 behaviours in a defluidising PDC deposit.

## 130 Methodology

### 131 Source material and Sample Preparation

132 A range of unconsolidated material collected in 2009 from the 2006 Tungurahua, Ecuador, eruptions (provided  
133 by U. Küppers, LMU Munich) has been subjected to a range of characterisation tests to elucidate flowability  
134 properties and variations with moisture content.

135 The ignimbrite material is dark grey/brown and andesitic in chemistry (Eychenne et al., 2012) and samples  
136 were sieved into varying size fractions, with specific particle size distributions created for the suite of analysis.  
137 All samples were dried in an 80°C oven for 24 hours to ensure the removal of residual and adsorbed moisture  
138 and agglomerations were broken-up by sieving prior to addition of water. For the series of characterisation  
139 tests, water was added to the samples based on weight percentage (0.00, 0.25, 0.50, 1.00, 2.50, 5.00, 7.50,  
140 10.00 %). Samples were stirred thoroughly to ensure a homogeneous moisture distribution.

### 141 Material characterisation and cohesive behaviour tests

#### 142 Particle Size Analysis

143 Particle analysis of the ignimbrite material was undertaken using a CAMSIZER X2. This uses particle imaging to  
144 build particle shape and size characteristics for dry samples. Particles were sieved prior to using the CAMSIZER  
145 with samples <1000 µm were used. Any results from the CAMSIZER erroneously returned as >1460 µm were  
146 removed.

#### 147 Geldart's Classification of Powders

148 Geldart (1973) classified powders into four distinctive groups (A-D) defined by fluidisation behaviours, which  
149 vary dependent on particle size and density; resultant flowability properties are described as 'very poor' to  
150 'excellent'. Group C, the finest material (< 20 µm), is dominated by interparticulate forces. Group D (>1 mm)  
151 requires an increased gas velocity to fluidise. Group C and D present passable to very poor fluidisation  
152 behaviours and would typically express as slugging, channelling, and spouting behaviours (Leturia et al., 2014).  
153 Group A (30 – 100 µm) and B (100 µm – 1mm) powders show the best fluidisation behaviours overall and are  
154 most likely to have good flowability, typically expanding under fluidisation.

155 Volcanic materials used in these experiments (Fig. 2) have particle size distributions from 2.5 to 1000 µm. and  
156 should readily exhibit fluidisation behaviours typically of Groups A and B in Geldart's classification.

#### 157 Bulk and Tapped Density

158 Bulk and tapped density measurements describe the mass and volume ratio of a material, without and with  
159 packing respectively (Amidon et al., 2017). Tapped density experiments remove interparticulate voids. The  
160 differences within the bulk and tapped density measurements correspond to the cohesive properties of the  
161 particles (Deb et al., 2018) and can be affected by shape and size of material (Amidon et al., 2017).

162 Bulk and tapped density were calculated for dry sample herein to characterize cohesive behaviour prior to the  
 163 addition of water (method adapted from USP: Bulk and Tapped Density of Powders, 2015).

164 Bulk density was obtained by pouring 100g of the volcanic material into a 250mL cylinder and levelling where  
 165 needed. The unsettled volume was measured, and bulk density calculated using Equation 1. This procedure  
 166 was completed three times per sample.

$$167 \quad \rho_b = \frac{m}{V_0} \quad [1]$$

168  $m = \text{mass (g)}$

169  $V_0 = \text{unsettled apparent volume (mL)}$

170 The cylinder was tapped at 150 taps/min, with volume measured every minute until leveled. Using the  
 171 unsettled apparent volume and final tapped volume, the tapped density (Eq. 2), Carr's Index (Eq. 3) and the  
 172 Hausner ratio (Eq. 4) could be calculated.

$$173 \quad \rho_t = \frac{m}{V_f} \quad [2]$$

174  $m = \text{mass (g)}$

175  $V_f = \text{final tapped volume (mL)}$  (Moondra *et al.*, 2018)

176 The Carr's Index and Hausner Ratio are indicative of flowability and interparticulate behaviours (Hausner,  
 177 1981) and are a useful tool in determining a materials ability to fluidise and flow (Table 1) .

178 The Carr's Index measures the strength and compressibility of a material (Equation 3; Moondra *et al.*, 2018).

$$179 \quad CI = 100 \left( \frac{\rho_t - \rho_b}{\rho_t} \right) \quad [3]$$

180 The Hausner Ratio determines how certain powders will behave i.e., flowability and fluidisation (Equation 4, Yu  
 181 and Hall, 1994; Abdullah and Geldart, 1999).

182

$$183 \quad HR = \frac{\rho_t}{\rho_b} \quad [4]$$

184

185 **Table 1** Relationship between Carr's Compressibility Index, Hausner Ratio, and flowability behaviours. From  
 186 (Gorle and Chopade, 2020).

| CI      | HR          | Flowability    |
|---------|-------------|----------------|
| ≤10     | 1.00 – 1.11 | Excellent      |
| 11 – 15 | 1.12 – 1.18 | Good           |
| 16 – 20 | 1.19 – 1.25 | Fair           |
| 21 – 25 | 1.26 – 1.34 | Passable       |
| 26 – 31 | 1.35 – 1.45 | Poor           |
| 32 – 37 | 1.46 – 1.59 | Very Poor      |
| > 38    | >1.60       | Very Very Poor |

187

188 **Angle of Repose**

189 The Angle of Repose (AoR) refers to the static friction coefficient and the angle of internal friction, and can be  
 190 investigated through static (funnel) and dynamic (rotating cylinder drum) methods (Al-Hashemi and Al-  
 191 Amoudi, 2018). AoR results are attributed to understanding the flowability of a material (Table 2).

192 **Table 2** Flowability based on angle of repose results (Al-Hashemi and Al-Amoudi., 2018).

| Flowability                        | Angle of Repose (°) |
|------------------------------------|---------------------|
| <b>Very free flowing</b>           | <30                 |
| <b>Free flowing</b>                | 30 - 38             |
| <b>Fair to passable flow</b>       | 38 - 45             |
| <b>Cohesive</b>                    | 45 - 55             |
| <b>Very Cohesive (non-flowing)</b> | >55                 |

193

194 To determine the static angle of repose (SAoR) for each experiment, 100 g of material was released from a  
 195 funnel held 3.5 cm over a circular platform (Av diameter = 12 cm). The height of the cone was measured, and  
 196 the angle of repose calculated using Equation 5 (Al-Hashemi and Al-Amoudi, 2018). Where material did not  
 197 release freely from the funnel, material was lightly agitated. If the height of the cone reached the base of the

198 funnel, then the funnel was incrementally moved vertically to accommodate the growing cone. This was  
199 repeated three times for each experiment.

200

$$201 \quad SAoR (^{\circ}) = \tan^{-1} \frac{2h}{D} \quad [6]$$

202  $h = \text{height (mm)}$

203  $D = \text{base diameter (mm)}$

204 Dynamic Angle of Repose (DAoR) was determined by rotating 100 g of material in a clear cylindrical drum at a  
205 constant rate (Smith, 2020). This was recorded on video and critical angle (the maximum angle prior to  
206 collapse) measurements analyzed using ImageJ (Schneider et al., 2012). This was completed three times.

### 207 Fluidisation Behaviour Tests

208 Experiments to determine the fluidisation behaviours of the volcanic material with increasing moisture  
209 contents were completed using a rectangular, near-2D fluidisation chamber with a porous base (following  
210 Gilbertson et al., 2020). Material (200.0 g) was measured, a weight % of water was added, and the  
211 homogeneous sample was then placed into the chamber and carefully levelled. A manometer probe recorded  
212 basal pore pressure changes during each experiment. Gas velocity (L/min) was increased incrementally until  
213 either a stable, channelised bubbling fluidisation state was achieved, or large amounts of winnowing or  
214 pressure build-up occurred. To limit the effects of drying from basal air flow, experiments were carried out  
215 with gradual increases in gas flow rate (0.8 – 3.00 L/min/min for dry sediments and 6.80 – 11.00 L/min/min for  
216 moisture added sediments) over a period of 01:11 – 23:51 minutes.

217 Results

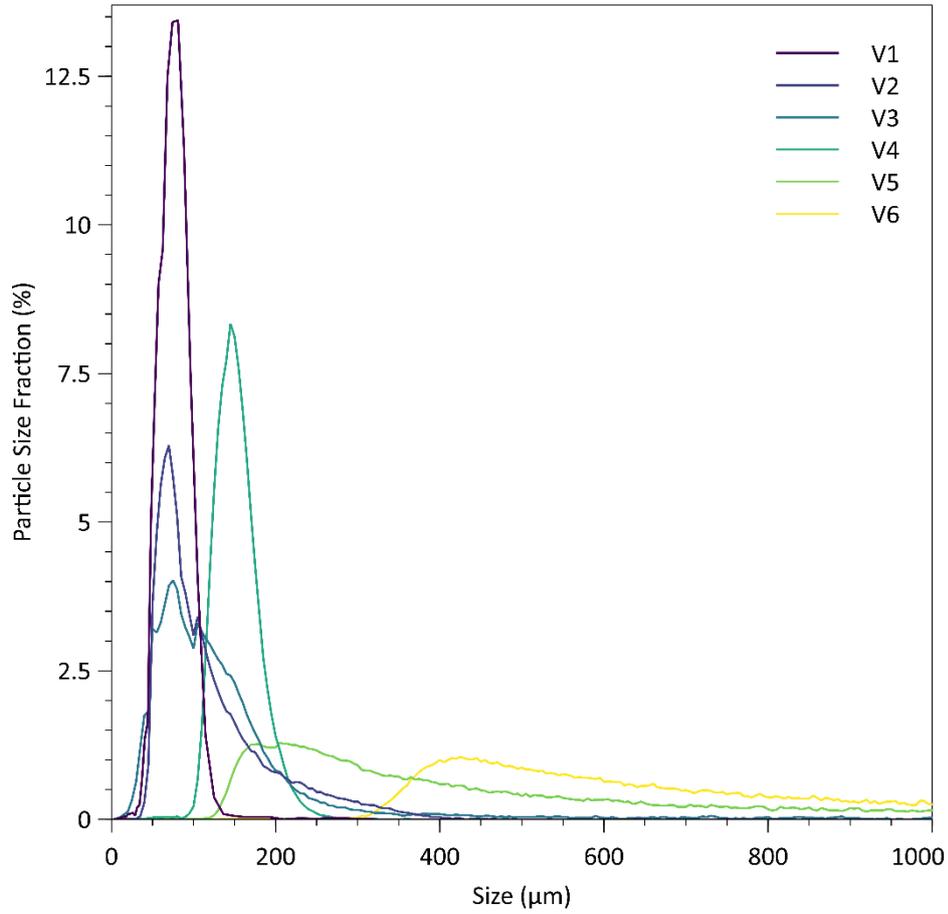
218 Particle Size Analysis

219 The experiments were completed using volcanic material (V1 – V6). All samples are moderately to very well  
 220 sorted (Fig. 2). Samples V1, V4, V5 and V6 were sieved into desired particle size distributions, whereas samples  
 221 V2 and V3 were kept as natural volcanic particle size distributions (ranging from >74 - 300  $\mu\text{m}$ ). Particle size  
 222 and sorting characterisation of each sample is presented in Table 3. As highlighted above, all samples fall into  
 223 groups A and B of Geldart’s classification, indicating that they should display excellent to fair flowability  
 224 (Geldart, 1973).

225 **Table 3** Table showing particle size mean (logarithmic), particle size median (log), particle range, fines content,  
 226 geometric mean, logarithmic ( $\Phi$ ) Method of Moments used for Mean, Sorting, Skewness, and Kurtosis. Sauter  
 227 mean diameter calculated from Breard et al (2019). Geldart Group (1973) based on size of particle.

228

| Material | Particle Size Mean ( $\bar{x}$ ) $\phi$ | Particle Size Median $\phi$ | Particle Size Range ( $\mu\text{m}$ ) | Fines Content (%) | Sorting Index ( $\sigma$ ) $\phi$ | Sorting ( $\sigma_G$ ) | Skewness ( $Sk$ ) $\phi$ | Kurtosis ( $K$ ) $\phi$ | Sauter Mean (mm) | Geometric Mean ( $\mu\text{m}$ ) | Geldart Group |
|----------|---|-----------------------------|---------------------------------------|-------------------|-----------------------------------|------------------------|--------------------------|-------------------------|------------------|----------------------------------|---------------|
| V1       | 3.776                                   | 3.734                       | 2.5 – 297.3                           | 35.76             | 0.428                             | Well                   | 1.891                    | 17.00                   | 0.07             | 72.93                            | A             |
| V2       | 3.215                                   | 3.320                       | 15 - 425                              | 21.41             | 0.710                             | Moderate               | -0.524                   | 2.576                   | 0.11             | 107.5                            | A, B          |
| V3       | 3.118                                   | 3.140                       | 5 - 1000                              | 19.57             | 0.868                             | Moderate               | -0.562                   | 4.140                   | 0.12             | 115.0                            | A, B          |
| V4       | 2.703                                   | 2.710                       | 20 – 650                              | 0.19              | 0.252                             | Very well              | 0.141                    | 8.536                   | 0.15             | 153.5                            | A, B          |
| V5       | 1.508                                   | 1.568                       | 10 – 1000                             | 0.11              | 0.758                             | Moderate               | - 0.103                  | 2.480                   | 0.42             | 347.3                            | A, B          |
| V6       | 0.833                                   | 0.812                       | 10 - 1000                             | 0.05              | 0.445                             | Well                   | 0.589                    | 8.277                   | 0.73             | 557.1                            | A, B          |



229

230 **Fig. 2** Particle size analysis of volcanic material.

231 Bulk/Tapped Density

232 **Table 4** Loose and tapped bulk density, the Hausner ratio, Carr Index and Flowability.

| Material | Loose Bulk Density (kg m <sup>-3</sup> ) | Tapped Bulk Density (kg m <sup>-3</sup> ) | Hausner Ratio | Carr Index | Flowability |
|----------|--|---|---------------|------------|-------------|
| V1       | 1310                                     | 1420                                      | 1.08          | 7.73       | Excellent   |
| V2       | 1320                                     | 1550                                      | 1.18          | 15.13      | Good        |
| V3       | 1380                                     | 1610                                      | 1.17          | 14.28      | Good        |
| V4       | 1320                                     | 1420                                      | 1.07          | 6.59       | Excellent   |
| V5       | 1280                                     | 1440                                      | 1.12          | 10.68      | Excellent   |

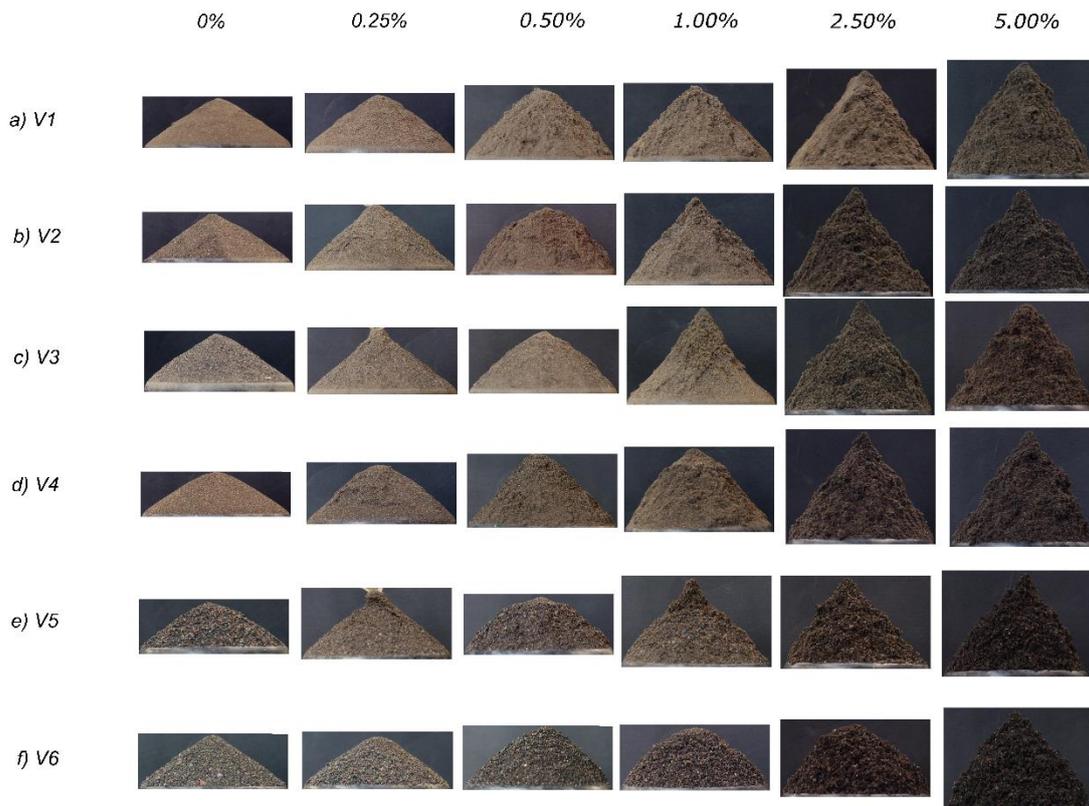
|    |      |      |      |       |      |
|----|------|------|------|-------|------|
| V6 | 1180 | 1370 | 1.15 | 13.37 | Good |
|----|------|------|------|-------|------|

233

234 The bulk and tapped densities were calculated for volcanic samples ranging in sizes from finest (V1 – 3.8  $\phi$ ) to  
 235 coarsest (V6 – 0.8  $\phi$ ). With increasing particle size, bulk and tapped densities generally decrease (Table 4).  
 236 Material flowability, as determined by the Hausner Ratio and Carr Index, is good (V2, V3, V6) and excellent (V1,  
 237 V4, V5) under the 0% moisture conditions. The change in flowability between V5 and V6 likely reflects the  
 238 large increase in geometric mean from 347 (V5) to 557 (V6) (Table 3). V1, V4 and V5 all show excellent  
 239 flowability, presumed to be related to the smaller particle range in V1 and V4 and having a low fines content in  
 240 V5 (0.11%) (Table 3).

241 **Angle of Repose**

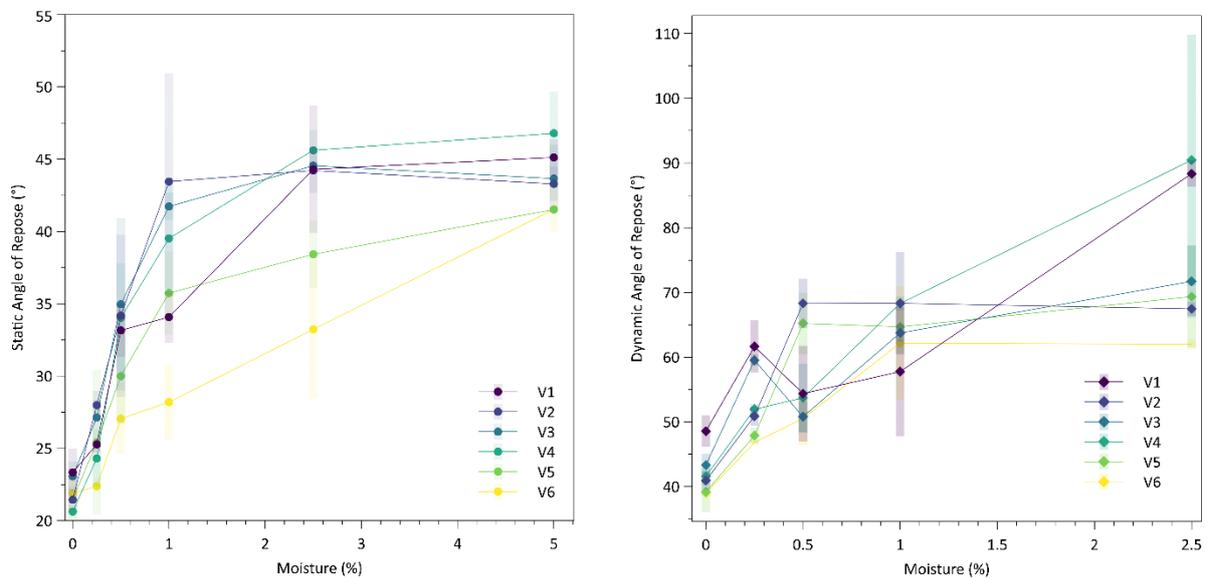
242 Static angle of repose (SAoR) increases with increasing across all volcanic samples (V1-6; Fig. 3). For the 0%  
 243 moisture condition the SAoR ranges from 21° (V2, V4, V5) to 23° (V1, V3). Interestingly, these results show that  
 244 under 0% moisture conditions, the SAoR is broadly similar (within 2°) regardless of particle size or sorting (Fig.  
 245 4a).



246

247 **Fig. 3** Representative static angle of repose (SAoR) cone formation of V1 – V6.

248 When increasing moisture contents to 5% the SAoR values increase to approximately double those achieved with  
 249 0% moisture, approaching 42° (V5, V6) to 47° (V4). However, this relationship is not linear with increasing  
 250 moisture content (Fig 4a). All materials show a rapid increase in SAoR with moisture to around 25°. However,  
 251 beyond a moisture content of 0.5% a division is evident between the finer and course mixtures; those with Sauter  
 252 mean diameters below 0.3 mm (V4, V3, V2) quickly increase to SAoR values of ~45° at moisture contents of 1-2  
 253 %, before plateauing and becoming invariant with additional moisture content. Mixtures with higher Sauter  
 254 diameters (V1, V5, V6) show a more gradual increase in SAoR with moisture content. V5, with a Sauter mean  
 255 diameter of 0.42 mm has somewhat intermediate behaviour, while V6 with a Sauter diameter of 0.73mm shows  
 256 a more *linear* relationship for SAoR with moisture between 0.5 – 5%. This indicates that SAoR shows distinct  
 257 sensitivity to increase in water in the materials and that relatively small weight percentages can produce very  
 258 different cohesivities within the mixtures It is notable that fines-rich mixtures are particularly sensitive to  
 259 moisture related cohesion, notably at <2 wt. %.



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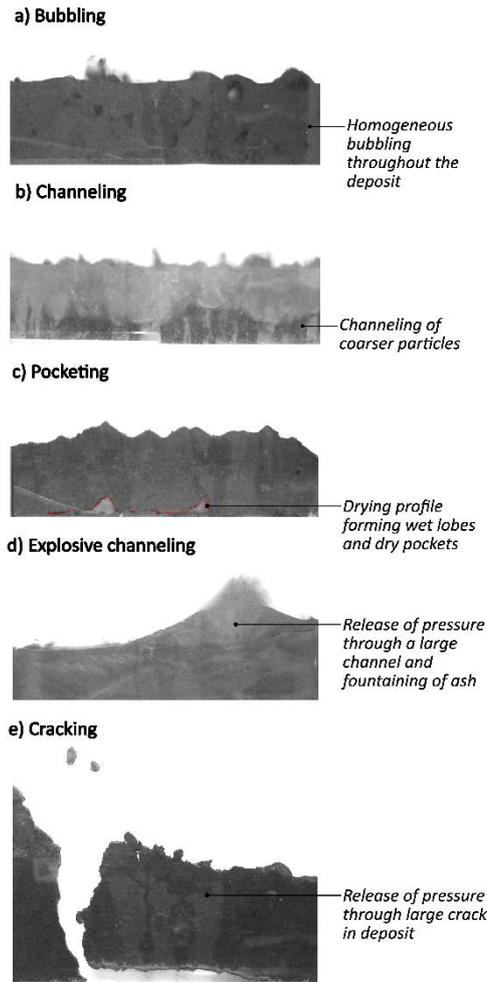
261 **Fig. 4** a) SAoR for volcanic material with varying moisture percentages (0.00, 0.25, 0.50, 1.00, 2.50 and 5.00%  
 262 with standard deviation error bars. b) DAoR critical angle of volcanic material with varying moisture  
 263 percentages (0.00, 0.25, 0.50, 1.00 and 2.50%).

264 Figure 4 also shows the relationship of Dynamic Angle of Repose experiments (Fig 4b). Generally, and similar to  
 265 the SAoR result, there is an increase in the DAoR with increasing moisture. However, in experiments with  
 266 increasing moisture levels (> 2.50%) the material was observed to clump, slide, and stick to the outer walls of  
 267 the drum, complicating the results. Nonetheless, it is important to observe that the Sauter mean relationships  
 268 detected within the SAoR experiments are not replicated in the DAoR setting.

269 **Fluidisation experiments**

270 Fluidisation behaviours were described via sidewall video analysis of the fluidisation chamber. The visual  
 271 observation of gas escape structures (i.e., bubbling, channeling, pocketing, explosive channeling, cracking (Fig.  
 272 5 a-e) and gas velocity measurements have been recorded at varying moisture levels (Fig. 6 a - f).

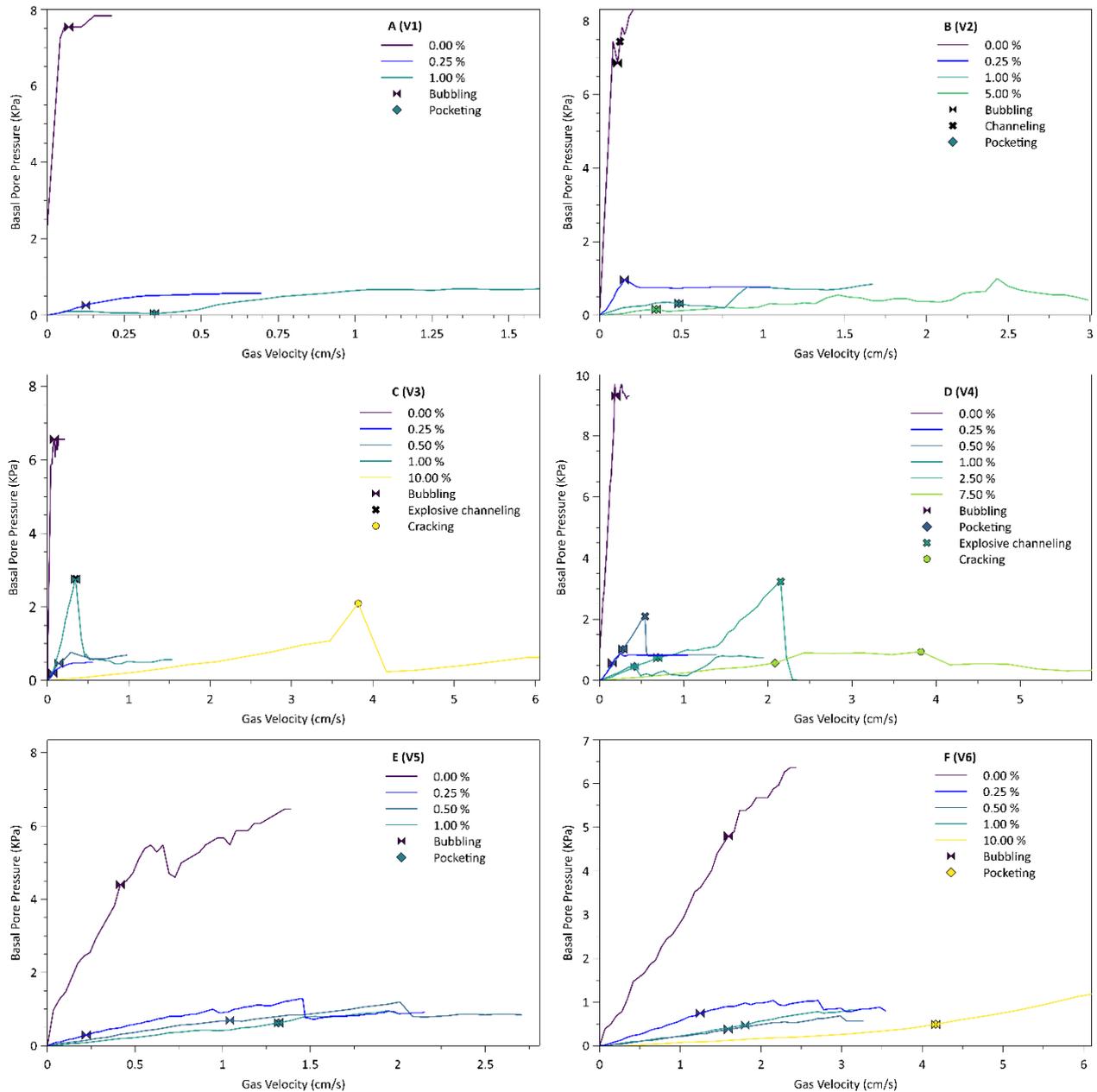
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286 **Fig. 5** a-e shows examples of the descriptive structures from across the experiments.

287 Bubbling gas escape (Fig. 5a) is seen initially in most experiments, where gas bubbles rise from the influx of gas  
288 within the deposit. With increasing gas flux, this can lead to channeling, where material is sorted through  
289 vertical channel or via pipe structures forming within the deposit (Fig. 5b). Drying profiles that migrate through  
290 the deposit are shown in Fig. 5c and as this migrates with non-uniformity in the vertical deposit, formation of  
291 areas of wet lobes and dry pockets can be observed (Fig. 5c). Where the material is dry and bubbling, this is  
292 referred to as pocketing. Explosive channeling can also be observed in some experiments (Fig. 5d), as the  
293 material dries, the upper wet deposit inhibits gas escape and causes a pressure increase and the subsequent  
294 release (Online Resource 1). Finally, under the highest moisture contents, material does not form any of the  
295 previous gas escape structures outlined above, instead, pressure builds until the deposit fractures into cracks  
296 where gas can easily permeate through (Fig. 5e; Online Resource 2).

297  
298



299 **Fig. 6** A-F. Fluidisation profiles of V1 – V6 with increasing moisture (0.00 – 10.00 %). Symbols show gas escape  
 300 structure formation.

301 **0.00% Moisture**

302 At 0.00% moisture for samples with moderate sorting (i.e., V2, V3, V5), fine material escapes through gas  
 303 escape channels (Fig. 5b) in the lower area of the deposit. The observation of minimum bubbling ( $U_{mb}$ ) is first  
 304 seen in the upper fine fraction of the deposit at  $U_{mb}$  0.11 (V2), 0.08 (V3), and 0.42 (V5) cm/s. There is often a  
 305 separation of fines bubbling in the upper layer, a mid-area of coarse channeling ( $U_{mc}$ ) at 0.13 (V2) and 0.10 (V3)  
 306 cm/s as fines are being elutriated, and a coarse material layer at the base of the deposit. Bubbling only affects  
 307 the finer material.

308 In volcanic mixtures that are well to very well sorted (i.e., V1, V4) bubbles rise uniformly throughout the whole  
 309 deposit with a  $U_{mb}$  of 0.07 (V1) and 0.19 (V4) cm/s. Within the more coarse, well sorted, material (V6) bubbles

310 in a sluggish motion from the base of the deposit, with mostly bubbling ( $U_{mb}$  1.60 cm/s) occurring in the upper  
311 half of the deposit and channeling in the lower. This reflects the slight particle size variation of material used,  
312 and therefore the  $U_{mb}$  of the coarser material (Fig. 5b).

### 313 0.25% Moisture

314 At 0.25% moisture contents, similar behaviours are observed for V3 ( $U_{mb}$  0.069 cm/s), V5 ( $U_{mb}$  0.22 cm/s) and  
315 V6 ( $U_{mb}$  1.25 cm/s) as described for 0.00% moisture. For V2 ( $U_{mb}$  0.15 cm/s), V1 ( $U_{mb}$  0.13 cm/s) and V4 ( $U_{mb}$   
316 0.15 cm/s), bubbling begins at the base of the deposit. However, as the surrounding wet deposit begins to dry,  
317 this dry material becomes incorporated into the bubbling deposit. In V2, we again see a separation of  
318 channeling and bubbling in the lower and upper deposit.

### 319 0.50% Moisture

320 At 0.50% moisture, a drying profile can be observed throughout most of the deposit (V4, V5, V6). In the V4  
321 sample, as drying at the base moves throughout the deposit, dry material begins to bubble ( $U_{mb}$  0.28 cm/s),  
322 and pressure slowly increases. This is released suddenly (explosive channeling) at 0.54 cm/s through a large  
323 channel which cuts through the moist, upper part of the deposit. As the surrounding wet material then begins  
324 to dry, it then becomes incorporated into the bubbling deposit. In the V5 sample, the drying profile forms  
325 lobes of wet material and pockets of dry material. The dry pockets slowly grow until reaching the upper  
326 deposit and begin to bubble ( $U_{mb}$  1.04 cm/s). With continued drying as the experiment progresses, similar  
327 behaviours to the 0.25% and 0.00% moisture levels experiments are observed. After the drying profile has  
328 moved through the deposit of V6, similar behaviours to 0.25% and 0.00% moisture levels are observed ( $U_{mb}$   
329 1.60 cm/s).

330 For the V3 material, channels of coarser material begin to slowly move towards the surface. Material begins to  
331 dry and is then incorporated into the bubbling deposit ( $U_{mb}$  0.14 (V3)).

### 332 1.00% Moisture

333 At 1.00% moisture, V1, V2, and V4 show material at the base of the deposit drying in pockets. The dry material  
334 begins to bubble ( $U_{mb}$  0.35 (V1), 0.49 (V2), 0.42 (V4) cm/s) and as the surrounding wet material begins to dry, it  
335 is incorporated into the bubbling deposit. In V5 and V6, a distinctive drying profile moves throughout the  
336 deposit. Again, this creates dry pockets of bubbling material ( $U_{mb}$  1.32 (V5), 1.81 (V6) cm/s) and wet lobes. In  
337 V3, pressure slowly builds as gas velocity is increased. Pressure is suddenly released through the formation of  
338 an explosive channel (0.35 cm/s). The dry deposit then begins to bubble ( $U_{mb}$  0.35 cm/s) and is slowly  
339 incorporated into the surrounding drying material.

### 340 2.50% Moisture

341 At 2.50%, behaviours of V4 show similar results to 1.00% moisture content: as the base dries, bubbling pockets  
342 are formed ( $U_{mb}$  0.70 cm/s) in-between lobes of wet material. In V4, pressure builds until it is suddenly  
343 released through an explosive channel (2.15 cm/s).

344 **5.00% Moisture**

345 At 5.00% V2 shows the deposit drying at the base which forms drying and bubbling ( $U_{mb}$  0.35 cm/s) in pockets,  
346 and wet lobes.

347 **7.50% Moisture**

348 At 7.50%, a clear drying profile forms through the V4 deposit, cracks begin to form and move through the  
349 deposit until reaching the top and collapsing into pieces (3.82 cm/s). As gas was moving through cracks, there  
350 was no dramatic rise and release in pressure.

351 **10.00% Moisture**

352 Finally, at 10.00%, V3 forms a clear drying profile within the deposit. Pressure builds before being released  
353 suddenly at 3.82 cm/s. This forms a large crack in-between wet material. V6 shows a clear drying profile, as  
354 pressure slowly rises as small pockets eventually form and dry material begins to bubble (4.17 cm/s).

355 **Key Observations**

356 These fluidisation experiments clearly demonstrate how small additions of water into pyroclastic material can  
357 greatly impact fluidisation behaviours and resulting gas escape structures of a defluidising volcanic deposit.  
358 Two key observations are apparent in the experiments: 1) the drying profile, and 2) pressure build up and  
359 release.

360 The dynamics of the drying profile, as the moisture content is impacted by the fluidising gas, exert a strong  
361 control on the distribution of gas escape features, with variations across the grain size of the materials.

362 As gas flux is increased, a drying profile can move from the base to the top of the deposit. The drying profile  
363 forms more easily within the coarser materials (V3 – V6). The profile initially rises uniformly across the bed,  
364 before becoming irregular as it reaches the top of the deposit. These profiles are noted as they highlight  
365 vertical and lateral moisture heterogeneity within the deposit and their irregular structure determines the  
366 formation of drying pockets and wet lobes (Fig 5c). At low moisture percentages (< 2.50%) the drying pockets  
367 bubble and the wet lobes begin to dry before being incorporated into the pockets. However, at high moisture  
368 contents (> 2.5%) moisture rich lobes remain throughout the experiment, even at high gas velocities. This  
369 shows that within a defluidising deposit, a drying profile will lead to lateral and vertical variations in moisture.

370 In experiments with moisture contents of 0.50 – 10.00%, explosive channelling (V3, V4) and cracking (V3, V4)  
371 can occur. Across the experiments with 0.50 – 5.00% moisture contents, a wet impermeable cap was observed  
372 to form above the drier underlying deposits, with progressive drying of the vertical profile. Pressure builds  
373 under the cohesive cap and continues to rise with increasing gas velocity. This eventually results in explosive  
374 channeling and a sudden basal pressure drop as the overburden pressure is exceeded. In higher moisture  
375 levels (5.00 - 10.00%) experiments, the deposit does not dry as a relatively uniform rising profile. Instead,  
376 pressure builds as the gas velocity is increased until cracks form in the deposit. These cracks act as effective  
377 gas escapes and release the pore pressure.

378 **Discussion**

379 The impact of moisture on ignimbrite material and PDC behaviour is relatively understudied, with previous  
380 detailed investigations having traditionally focused on dry (Wilson, 1980) and saturated (Roche et al., 2001)  
381 extremes. However, direct observations have shown that variable amounts of moisture can enter a PDC  
382 system (Cole et al., 1998, 2002; Lipman, 2019; Vecino et al., 2022) and accretionary lapilli and ash pellets are  
383 believed to provide evidence for the presence of moisture within PDCs (Branney and Kokelaar, 2002; Brown  
384 and Branney, 2004; Druitt, 2014). Our results show that with increasing moisture content: 1) the cohesivity of  
385 ignimbrite material alters drastically, even with very small weight-percent amounts (> 0.50%); 2) an increase in  
386 moisture can change flow property behaviours from a free flowing to a non-flowing material; 3) changes in  
387 moisture affects fluidisation profiles and gas escape structures; 4) a defluidising deposit can lead to a drying  
388 profile, and therefore lateral and vertical heterogeneity within the deposit; and 5) pressure can increase where  
389 gas escape is hindered by moisture, which can cause dramatic releases of pressure in an explosive fashion.  
390 Here we discuss the implications of these findings for the flow behaviour of a PDC and variations in deposit  
391 sedimentology and broader architecture.

392 **Material Behaviours**

393 These experiments show that in a dry (0%) sample, material with a grainsize of 5 – 1000  $\mu\text{m}$  has low cohesivity.  
394 The cohesive nature of dry material heavily depends on the fine ash content of the material. For example,  
395 angle of repose is highest in samples V1 and V3 which have some of the highest % of fines (Table 3.1).  
396 However, results indicate that sorting plays a key role; V5 has one of the lowest fines contents (0.11 %) but has  
397 the same SAoR value as V2 which has a high fines content of 21.41 %. V2 and V5 have a large size range, similar  
398 sorting indexes (V2 – 0.710, V5 – 0.758  $\phi$ ) and the lowest kurtosis values (V2 – 2.58, V5 – 2.48  $\phi$ ). When dry,  
399 all samples have good flowability and are said to be free flowing (i.e., when fluidised the deposit demonstrates  
400 bubbling and channeling throughout the deposit). Results from the Bulk/Tapped density experiments show  
401 that V1 and V4 exhibit some of the best flowing behaviors. V1 is well sorted and displays the largest volumes  
402 of fines (35.76%) whereas V4 is very well sorted and has one of the lowest volumes of fine material (0.11%).  
403 The excellent flowability seen in V4 may result from its sorting and resulting packing behaviour, which is  
404 known to affect flow behaviour (Breard et al., 2023). The DAoR results show contrasting behaviours versus the  
405 SAoR results when comparing against the Sauter mean diameter. This suggests that particle size controls are  
406 likely to be more important in understanding the remobilisation of static deposits than in the flowability of  
407 particles already in motion.

408 Such an observation has wide ranging implications. For example, Breard et al (2023) suggests that long run-out  
409 distances in block and ash flows (BAFs) were a result of large degrees of fragmentation, with the current  
410 becoming more fines rich, and subsequently the deposit displaying higher packing. These particle size changes  
411 result in a dynamically evolving flow where fines formation and increasing packing behaviour reflect elevated  
412 pore-pressure within the flow (Breard et al., 2023). Our experiments (V1, V4) show that both fines content and  
413 packing can contribute to good flowability behaviours, with implications for run-out distances.

414 The material with the larger volume of fines (V1, V2) is shown to exhibit more cohesive behaviours with  
 415 increasing moisture (i.e., higher SAoR angle). V4, which is more well sorted, has the highest increase in SAoR  
 416 values. Our work demonstrates how important the role of moisture can be, at small amounts, in changing  
 417 flowability behaviours. We can build a hypothesis that the addition of moisture into a PDC during propagation,  
 418 with increasing fragmentation and packing, can be a factor in controlling run-out distances – higher moisture  
 419 contents reduce flowability so may reduce maximum runout distances, particularly in flows with enhanced  
 420 fragmentation.

421 We can then assume that during deposition from a PDC the fine ash fraction of a deposit, and material with  
 422 increased packing, may be more influenced by water – and therefore hold a higher moisture content during  
 423 deposition. Within a material with >30% volume of fines, the stress forces begin to be dominated by the fine  
 424 fraction (Li et al., 2020; Breard et al., 2023). A large volume of fines, both with and without moisture, may  
 425 dramatically alter the deposition and the preservation potential of these layers. This has implications for  
 426 understanding deposit architecture behaviours (e.g., gas escape structures) or extents (e.g., erosion of ash-rich  
 427 layers).

#### 428 Gas Escape Structures

429 A variety of gas escape structures were observed in the fluidisation experiments, with many related to  
 430 moisture content. Here we define three main types of behaviour (Table 5). In Type 1 (0.00 - < 0.50%) we see  
 431 partial fluidisation and segregation of heterogenous material through bubbling and channeling. In material  
 432 with a smaller size range, small vertical bubbling occurs across the entirety of the deposit. During Type 2 (0.50  
 433 – 5.00%), an irregular drying profile develops and moves through the deposit from the base. As the drying  
 434 profile grows, dry pockets of bubbling material begin to form in between irregular lobes of wet material.  
 435 Explosive channeling also occurs, which releases pressure and facilitates quicker drying of the whole deposit.  
 436 Finally, during Type 3 (7.50 – 10.00%), similar lobe and pocket structures are formed to Type 2 but are  
 437 accompanied by cracking processes, where fractures in the wet material form to accommodate rapid gas  
 438 escape.

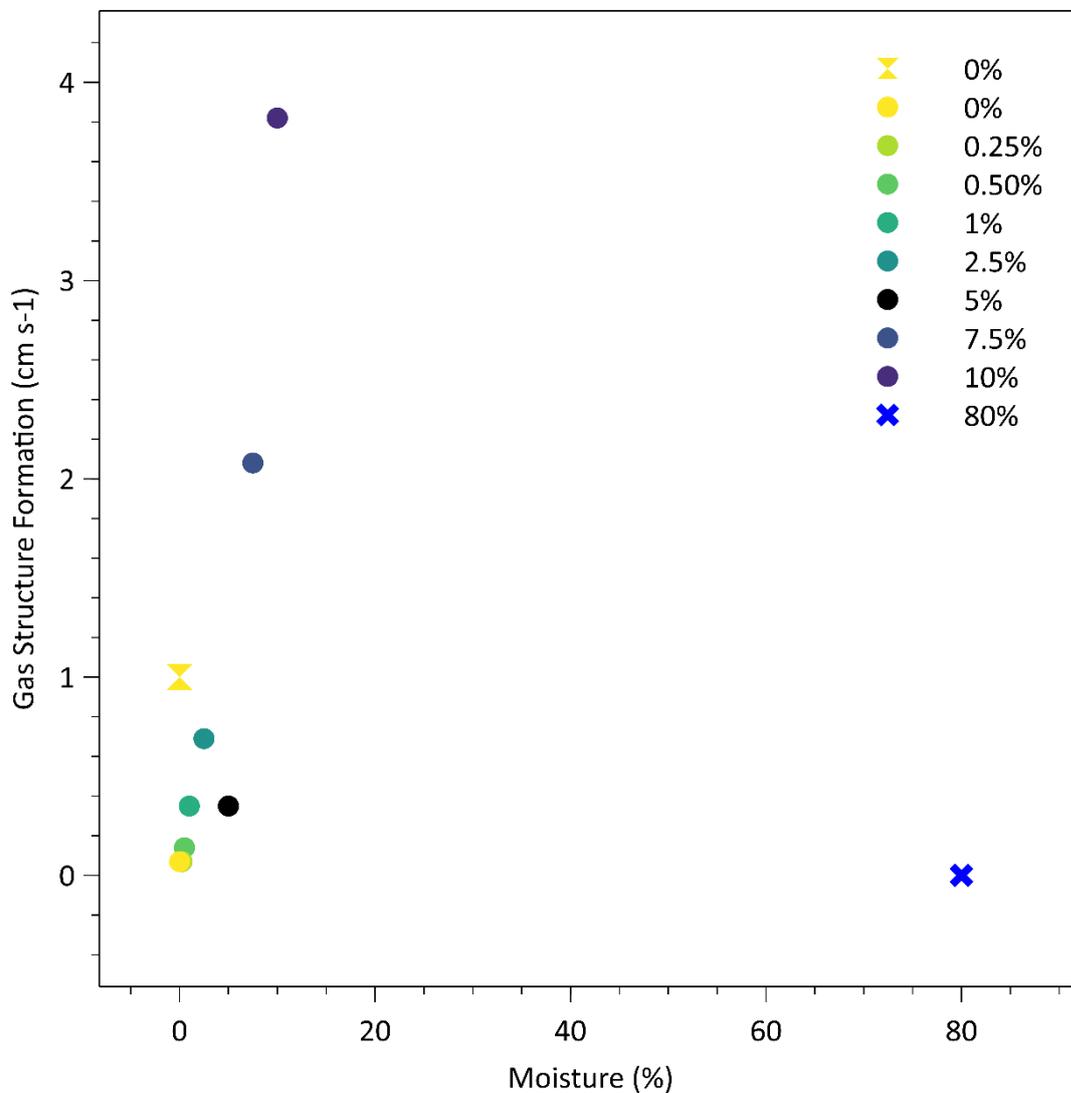
439 **Table 5** Varying types of behaviour of gas escape observed with increasing moisture in volcanic material.

|                | Type 1       | Type 2       | Type 3       |
|----------------|--------------|--------------|--------------|
| Moisture Range | 0.00 – 0.25% | 0.50 – 5.00% | 7.50, 10.00% |
| Bubbling       | Yes          | Yes          | Yes          |
| Channeling     | Yes          | Yes          | No           |
| Drying Profile | No           | Yes          | Yes          |

|                      |    |     |     |
|----------------------|----|-----|-----|
| Pocketing            | No | Yes | Yes |
| Explosive Channeling | No | Yes | No  |
| Cracking             | No | No  | Yes |

440

441 Roche et al., (2001) investigated the water fluidisation behaviour of ignimbrite material where material was  
 442 saturated and subjected to an increase in fluid velocity. Similarly, the findings of Wilson (1980; 1984) and the  
 443 experiments herein demonstrate gas escape structures forming from aggregative behaviour. The gas escape  
 444 structures were all fines depleted and rich in dense and coarse material. However, the aqueous gas escape  
 445 structures (pipes) were observed to form at lower fluid velocities than the aerated structures (Fig. 7).



446

447 **Fig.7** showing minimum gas velocity of gas escape structure formation for dry, moisture-influenced, and  
 448 saturated deposits. Wilson (1980) values (hourglass) based on first formation of pipes (0% wt.). The values of  
 449 our results (circles) are from first formation of gas escape structure seen in V1-V6 at varying moisture

450 percentages (i.e., bubbling at lower %'s, explosive cracking at the highest %'s). Roche et al., (2001) values  
451 (cross) are on initial pipe formation (80% wt.).

452 Figure 7 shows that by increasing moisture within a sample, higher gas velocities are required for aggregative  
453 fluidisation. However, between 10 and 80% moisture there is a change in the dominant fluidising medium,  
454 from gas to water. Instead of impeding early fluidisation structures, a large increase in moisture leads to more  
455 regular structures forming. This can be explained by changing particle-water states with increasing moisture.  
456 Aggregative fluidisation mechanisms will result in the segregation of particles through gas escape structures,  
457 where fines are winnowed. The nature of segregation will depend on the particle concentration and the size,  
458 shape, density, and relative proportions of clasts (Sparks, 1976; Wilson, 1980; 1984; Branney and Kokelaar,  
459 2002). We find that the moisture content of the deposit also controls this process; segregation structures can  
460 change dynamically with drying or become hindered with increasing moisture influence. This is due to our  
461 material being in a predominantly capillary state (Kim and Hwang., 2003; Kim and Sture, 2008). At higher levels  
462 of moisture, particles reach a more saturated state, are completely supported by capillary bonds and  
463 fluidisation behaviours are no longer inhibited (as seen in Roche et al., 2001; Kim and Hwang., 2003). We  
464 observe that even small influences of moisture (as low as 0.50% by weight) into volcanic material may control  
465 the formation and nature of gas escape structures.

#### 466 **Application to natural deposits**

467 Our results show introducing moisture into volcanic materials may cause changes in gas escape morphology.  
468 Gas escape structures have been recorded and described extensively within field volcanological literature (e.g.  
469 Fisher and Schmincke, 1984; Pacheco-Hoyos et al., 2020). They have been described as pods and pipes  
470 displaying single or branching patterns, or as lenticular, curvilinear, and crescentic shaped (Wilson, 1980;  
471 Branney and Kokelaar, 2002; Pacheco-Hoyos et al., 2020). They can be spatially arranged within individual  
472 layers or can move through multiple layers and are often fines depleted. Our results demonstrate varied  
473 morphologies, including vertical channels, sub-vertical cracks, and pods (created by moisture-rich lobes and  
474 dry pockets).

475 Changes in gas escape structures in ignimbrites are thought to be dominated by heterogeneity within  
476 ignimbrite material (e.g., size, density, shape etc.; Wilson, 1984; Pacheco-Hoyos et al., 2020). We propose that  
477 varying moisture levels will also influence changes in gas escape morphology and may explain circumstances  
478 where morphological changes are observed when other conditions appear unchanged. More detailed  
479 documentation of morphology of field examples may allow for improved interpretations of depositional  
480 environment.

481 Morphological changes in gas escape structures also have implications for drying profiles and resulting  
482 heterogeneities within a deposit. In our experiments, the formation of a drying profile demonstrates both  
483 vertical and lateral variation due to an undulating contact between wet and dry material resulting in vertical  
484 and lateral changes in the tensile strength of a deposit. This may have implications for erodibility,  
485 remobilisation (as well as preservation potential), and subsequent deposition of certain layers within a deposit.

486 **Mechanism for secondary eruptions**

487 Secondary eruptions in ignimbrite deposits form due to the interaction between water and hot material (Van  
488 Westen and Daag, 2005). When in contact with a hot ignimbrite, steam will expand and cause sudden  
489 explosive decompression. Secondary eruptions form large craters (20 – 80m depth), can remobilize large  
490 volumes of ignimbrite material and can occur for years after the initial eruption (the 1991 Mount Pinatubo  
491 generated secondary eruptions for up to a year; Riehle et al., 1995; Van Westen and Daag, 2005). Riehle et al.,  
492 (1995) modelled cooling, degassing and compaction behaviours within ignimbrites. High temperatures were  
493 most likely to remain elevated within deposits >50 m thick, with temperatures cooling mostly by groundwater  
494 and some influence from rainfall. Keating et al (2005) modelled that the addition of water on a hot deposit can  
495 result in exceeded pore pressure, in turn exceeding the overburden pressure. This can result in secondary  
496 eruptions.

497 Moyer and Swanson (1987) described three styles of secondary eruptions - passive degassing (least explosive),  
498 ash fountaining and explosive cratering (most explosive) - controlled by thermal energy and the permeability  
499 of the overlying material. Analogue experiments investigating the mechanisms of secondary hydroeruptions  
500 have been explored by Gilbertson et al (2020). They identified that vertical changes in size fractions, and  
501 therefore a vertical profile of minimum fluidisation velocities, resulted in secondary hydroeruptions. In these  
502 experiments, a deposit capped with coarser material formed an upward doming bed leading to an explosive  
503 release of material. This was due to a drag-induced system. The fine-particle layer below acted as a lower  
504 minimum fluidisation layer that was unable to fluidise the overlying, coarser layer, resulting in pressure  
505 increase and release.

506 Secondary eruptions are associated with subaerial PDC deposits, and also with sub-aqueous PDCs (Krakatau,  
507 1883; Mandeville et al., 1996) and sub-marine sediments (Hovland et al., 2002; Rogers, 2015; Cojean et al.,  
508 2021) and Martian impact craters (Boyce et al., 2012). Martian pits in impact craters are thought to be formed  
509 by water loss from exsolving vaporizing water, creating streams of gas and the funneling out of fine ejecta  
510 towards the surface (Boyce et al., 2012). Pockmarks are circular craters that form in the sub-aqueous  
511 environment: deep marine, lakes, lacustrine (Cojean et al., 2021). They can form from the upward movement  
512 of fluid (liquid or gas), the trapping of the fluid by an impermeable cap, or from gradual accumulation of the  
513 fluid until the gas is released. The fine mud particles within these environments can be extremely cohesive,  
514 and therefore display a very high tensile strength, which is what can form the impermeable overburden  
515 material in these environments (Rogers, 2015). Pockmarks have also been known to form in association with  
516 slides and slumps (Hovland et al., 2002).

517 Analogous to all the mechanisms for secondary hydroeruptions is an active moving pore pressure (i.e.,  
518 vaporization in meteorites) as well as a vertical variation in the permeability of a material (i.e., fine, or mud-  
519 rich material, minimum fluidisation velocities). Results from our experiments show that increasing moisture  
520 levels within the fluidised deposit can lead to impermeable layers forming through drying. By increasing  
521 moisture throughout our experiments, we exhibit passive degassing (0%), ash fountaining (> 0.50% wt.) and

522 explosive cratering (> 0.50% wt.) behaviours as described in Moyer and Swanson (1987). After lithostatic  
523 pressure of the impermeable wet cap is overcome, explosive channeling (> 0.50% wt.) and cracking (> 7.50%  
524 wt.) occurs (named *explosive cratering* by Moyer and Swanson., 1987). Similarly, to these works, our results  
525 herein demonstrate the impact of intermediate permeability on secondary eruptive styles. We argue that the  
526 change from passive degassing to explosive cratering is not only a reflection of thermal energy in the system  
527 (as our material cooled significantly throughout the length of experiment), but also by internal degassing of a  
528 partially fluidised deposit.

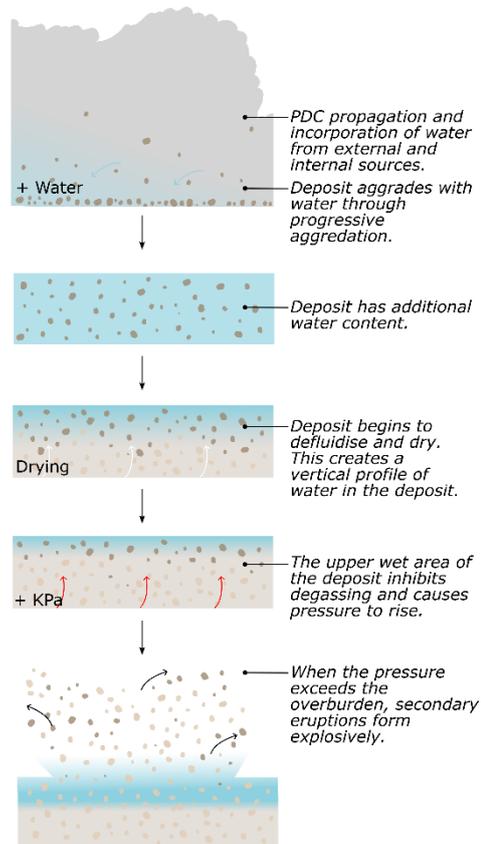
529 Critically, our results suggest a potential new mechanism for secondary eruptions that form in a moisture  
530 influenced material (Fig. 8a). In our experiments, the addition of water during deposition results in increased  
531 cohesion and resulting tensile strength. As the deposit dries from the base, we see a shift in gas escape as the  
532 material begins to dry and bubble. In our model (Fig. 8a), the upper moisture rich layer inhibits passive de-  
533 gassing and leads to increased pore-pressure. With increasing pressure in the deposit, the overburden strength  
534 of wet material is compromised. The result is a sudden pressure release by explosive channeling and cracking,  
535 which mimics similar behaviours seen in secondary eruptions in ignimbrite deposits. Additionally, in a dry  
536 deposit later moistened by water (i.e., precipitation), the upper moisture-rich layers of material will create an  
537 overall denser material (Fig. 8b).

### 538 **Application to field examples**

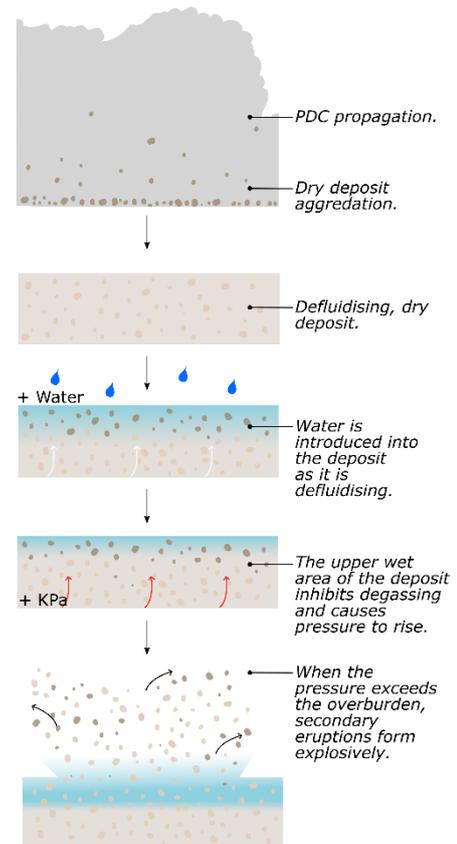
539 Secondary eruptions were observed following the Mt St. Helens 1980 and Mt Pinatubo 1991 events (Keating,  
540 2005) and were attributed to variations in the permeability of ignimbrite deposits caused by the presence of  
541 water (e.g., rainfall and lacustrine environments) (Moyer and Swanson, 1987; Manville et al., 2002). It is  
542 thought that high pressure towards the base of these ignimbrites, caused by vaporization of water, led to low-  
543 permeable layers preventing the balancing of pore pressures throughout the deposit, which resulted in  
544 explosive depressurization (Keating, 2005). Keating (2005) suggests that after emplacement, hydrological re-  
545 establishment may begin to occur and interaction with hot overlying ignimbrite material may result in the  
546 formation of secondary hydroeruptions.

547 Our moisture influenced model may provide an explanation for the observations of secondary eruptions in  
548 deposits that have aggraded with the presence of water (e.g., secondary eruptions followed the previous  
549 location of the Rogue River; Druitt and Bacon, 1986) (Fig. 8a) and that have interacted with rain (e.g., Mt  
550 Pinatubo, Daag and Westen, 1996) (Fig. 8b). Rainfall may create a moisture-rich cap to the deposit that is  
551 impermeable to degassing from the lower deposit. The increased moisture from the rain would result in an  
552 increased cohesivity, and therefore tensile strength, of the material. With gas escape inhibited, pressure may  
553 continue to build until the overburden pressure is reached, and degassing is then allowed to escape through a  
554 secondary eruption in the deposit.

a) Defluidising wet deposit.



b) Defluidising dry deposit with external water addition.



555

556 **Fig. 8** moisture-influenced model of secondary eruption formation by a) a defluidising wet deposit and b) a

557 defluidising dry deposit with external influences of water.

558 **Conclusion**

559 This work highlights the importance of moisture introduced into PDCs both in terms of PDC flow dynamics as  
560 well as the characteristics of the resultant deposits. Our results show that: 1) the cohesivity of ignimbrite  
561 material changes drastically, even though relatively small additions of moisture amounts (> 0.50%); 2) an  
562 increase in moisture can entirely alter flow property behaviour from a free flowing to a non-flowing material;  
563 3) changes in moisture impact fluidisation profiles and gas escape structures; 4) a defluidising deposit can lead  
564 to a moisture profile, and therefore lateral and vertical heterogeneity within the deposit, and 5) pressure can  
565 increase where gas escape is hindered by cohesive substrates driven by moisture content, resulting in  
566 secondary eruptions. This builds on previous models of secondary eruptions in deposits and supports the idea  
567 that they are formed because of the development of an impermeable capping layer, here created by the  
568 addition of moisture. This work further proposes that moisture within a defluidising deposit profile may hinder  
569 or change the formation of gas escape structures, which can then also lead to pressure increase and release,  
570 with significant implications for the interpretations of the sedimentary structures within the deposits. Overall,  
571 the work has shown the critical role of moisture within PDC dynamics and the implications for the erodibility of  
572 a material, preservation potential and the broader understanding of deposit architecture.

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