

1 Mercury loading within the Selenga River Basin and Lake 2 Baikal, Siberia

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25

26 Abstract

27 Mercury (Hg) loading in Lake Baikal, a UNESCO world heritage site, is growing and poses a
28 serious health concern to the lake's ecosystem due to the ability of Hg to transform into a toxic

29 form, known as methylmercury (MeHg). Monitoring of Hg into Lake Baikal is spatially and
30 temporally sparse, highlighting the need for insights into historic Hg loading. This study reports
31 measurements of Hg concentrations from water collected in August 2013 and 2014 from across
32 Lake Baikal and its main inflow, the Selenga River basin (Russia, Mongolia). We also report
33 historic Hg contamination using sediment cores taken from the south and north basins of Lake
34 Baikal, and a shallow lake in the Selenga Delta. Field measurements from August 2013 and 2014
35 show high Hg concentrations in the Selenga Delta and river waters, in comparison to pelagic lake
36 waters. Sediment cores from Lake Baikal show that Hg enrichment commenced first in the south
37 basin in the late-19th century, and then in the north basin in the mid-20th century. Hg flux was also
38 20-fold greater in the south basin compared to the north basin sediments. Hg enrichment was
39 greatest in the Selenga Delta shallow lake (Enrichment Ratio (ER) = 2.3 in 1994 CE), with
40 enrichment occurring in the mid- to late-20th century. Local sources of Hg are predominantly from
41 gold mining along the Selenga River, which have been expanding over the last few decades. More
42 recently, another source is atmospheric deposition from industrial activity in Asia, due to rapid
43 economic growth across the region since the 1980s. As Hg can bioaccumulate and biomagnify
44 through trophic levels to Baikal's top consumer, the world's only truly freshwater seal (*Pusa*
45 *sibirica*), it is vital that Hg input at Lake Baikal and within its catchment is monitored and
46 controlled.

47 Keywords: Mercury, mining, atmospheric deposition, lake sediments

48

49 **1. Introduction**

50 Mercury (Hg) is a global pollutant of concern and has both natural and anthropogenic sources.

51 Once emitted, most inorganic Hg can remain in the atmosphere for up to 12 months (Corbitt et al.,

2011) and can be transported across the world. Gaseous and particulate Hg emitted into the atmosphere is transformed into Hg (II), which is then deposited onto the landscape via wet and dry deposition (Bergan and Rodhe, 2001). Atmospherically emitted Hg will cycle between short-term stores (<10 years) in the atmosphere, terrestrial environments, and surface ocean waters, before being sequestered long-term into terrestrial soils and sediments, ocean margins and the deep ocean (Amos et al., 2014). Within aquatic environments, methylating bacteria can transform Hg (II) into a toxic organic form, known as methylmercury (MeHg). This organic form makes Hg especially harmful within aquatic ecosystems as it can bioaccumulate and biomagnify in foodwebs. Due to the toxicity of MeHg, the Minamata Convention was set up in 2017 to reduce the impact that human activities have on Hg releases to the environment (UN Environment, 2017). Hg cycling in aquatic environments may be affected by dissolved organic carbon (DOC), pH, temperature, redox conditions, sulfate concentrations and microbial activity, which control methylation (transformation of Hg into MeHg) and demethylation (transformation of MeHg into Hg) processes (Hintelmann et al., 1995; Kelly et al., 2003; French et al., 2014). Environmental changes associated with warming (e.g. increased weathering, temperature, productivity and organic loadings) can also affect Hg cycling, by stimulating methylation and inhibiting photodecomposition, due to increasing primary productivity and DOC concentrations which reduce light penetration in the water column (Hammerschmidt et al., 2006).

Lake Baikal is a UNESCO World Heritage Site and is internationally important for its high levels of water purity and endemism (Fig. 1). Gold mining began in Lake Baikal's catchment with the discovery of the Ildikan deposit in the mid-1800s (Maruev, 2018). Small-scale gold mining operations use Hg to extract gold from ore in a process of amalgamation and distillation. The first gold extraction processes using Hg started along the Kharaa River, in the basin of the Amur River

75 in 1837 CE (common era), and in the basin of the Selenga River (Lake Baikal's primary inflow)
76 in 1841 CE (Misyurkeeva, 2009; Maruev, 2018). Between 1860-1890 CE 40% of all gold in Russia
77 was mined in the Baikal region, with Hg used in the extraction before being disposed in rivers and
78 dispersed into the atmosphere (Maruev, 2018). Since the 1950s, the use of Hg in gold extraction
79 has stopped in the Russian region of the Baikal catchment, but continues in the Mongolian Selenga
80 River basin (Misyurkeeva, 2009). Over the last few decades, gold extraction along the Selenga
81 River has increased, with over 700 mines currently in operation in the Baikal catchment within
82 Mongolia (Brunello et al., 2004; Pietron et al., 2017), and the largest gold mining operation, the
83 Zaamar Goldfield, situated within the Mongolian Selenga River basin (Tumenbayer et al., 2000;
84 Chalov et al., 2015; Pietron et al., 2017). Recent studies report the Lake Baikal catchment and
85 Selenga River basin to be heavily polluted from these gold extraction activities (Brunello et al.,
86 2012; Thorslund et al., 2012; 2016; Brumbaugh et al., 2013; Chalov et al., 2015; Jarsjö et al., 2017;
87 Hampton et al., 2018).

88 Within the past decade, MeHg bioaccumulation has been observed in Baikal's pelagic foodweb
89 (Perrot et al., 2010; 2012; Ciesielski et al., 2016). High Hg concentrations have been reported in
90 fish from the Selenga River basin, which are above the recommended thresholds for human
91 consumption (Kaus et al., 2017), and in the water reservoir north of Irkutsk in the Baikal region
92 (Koval et al., 1999). Analyses of the livers and muscle of the Baikal Seal (*Pusa sibirica*), have also
93 shown Hg contamination within the lake's top consumer in the 1960s and 1970s, before declining
94 to present (2013 CE) in response to reduced atmospheric Hg emissions from Europe and Russia
95 (Ozersky et al., 2017).

96 Recent and current levels of Hg contamination at Lake Baikal are largely unknown due to sparse
97 records of Hg measurements and the lack of historical Hg loading records for the region. Within

98 this study, we have undertaken the first Hg assessment for Lake Baikal in 20 years (Leermakers et
99 al., 1996), and aim to address the following research questions: 1) is the Selenga River basin a
100 major source of Hg into Lake Baikal, and (2) has there been Hg enrichment in the Selenga Delta
101 and Lake Baikal since the onset of gold mining and development in the region?

102 **2. Materials and Methods**

103 **2.1. Study sites and field collection**

104 Lake Baikal can be divided into three main basins (south, central and north) with the central basin
105 separated from the south basin by the Buguldeika Ridge and the more than 20 km wide Selenga
106 River Delta. The Selenga River, which is approximately 943 km long (Nadmitov et al., 2015), is
107 the main tributary into Lake Baikal and contributes over 60% of annual flow into the lake. It
108 originates in the Khangai Mountains, northern Mongolia, and accounts for over 80% (over 447,000
109 km²) of Baikal's catchment (Nadmitov et al., 2015). The majority of the Selenga River basin is
110 situated in Mongolia (282,349 km²) rather than Russia (148,060 km²), with the basin covering
111 almost 20% of the total land area in Mongolia (Nadmitov et al., 2015). The Selenga River branches
112 into the Selenga Delta, the world's largest freshwater inland delta (Logachev, 2003), and a Ramsar-
113 designated floodplain wetland, which is internationally important for high rates of biodiversity and
114 migratory bird habitat (Scholz and Hutchinson, 2000).

115 The region around Lake Baikal became one of the most highly Hg polluted regions in Siberia,
116 following industrialization of the catchment between the 1950s and 1990s (Koval et al., 1999).
117 The largest cities and main industrial districts in Mongolia (Ulaanbaatar, Erdenet and Darkhan)
118 are situated along the main tributaries of the Selenga River, namely the Tuul, Orkhon and Kharaa
119 rivers, respectively. In Russia, Ulan Ude and Selenginskii are situated along the Selenga River
120 (Kasimov et al., 2017). Other major polluting cities and towns within Lake Baikal's catchment and

121 airshed include Irkutsk, Gusinoozersk and Severobaykalsk. Notorious industrial Hg emitters in the
122 region include metallurgical plants which produce Hg directly, chemical and electrical plants,
123 where Hg is an element in the manufacturing process, and coal and oil fired thermal electric power
124 plants, where Hg is recovered (Vasiliev et al., 1998). Chemical industries are prominent within the
125 Irkutsk-Cheremkhovo industrial zone and are a major concern for Hg pollution (Koval et al.,
126 1999). Other major regional Hg pollution sources include the Gusinoozersk State Regional Power
127 Plant (a coal-fired power plant), and the Selenginsk Pulp and Cardboard Mill within the Selenga
128 River basin, which began operating in 1974 CE and continued as an open system until 1990 CE
129 (Pisarksy et al., 2005; Nikanorov et al., 2012; Nomokonova et al., 2013). Industrial activity around
130 the shores of Lake Baikal began in the 20th century, and the Baikal Pulp and Paper Mill (BPPM),
131 which was in operation between 1966 to 2013, was a suggested point source of Hg (Brunello et
132 al., 2004).

133 Five sites were selected within Lake Baikal for surface water sampling to represent the main
134 basins, including the south basin (BAIK13-8), the shallow waters off the Selenga Delta (BAIK13-
135 10), the central basin (BAIK13-12), within Maloe More Bay off the central basin (BAIK13-14),
136 and the Upper Angara River in the north basin (BAIK13-19) (Fig. 1; Table S2). Maloe More Bay
137 is a vulnerable region of Lake Baikal, currently affected more than deeper water sites by
138 anthropogenic influence (Timoshkin et al., 2016). Additionally, water samples at five sites from
139 the Selenga Delta branches (SDB01 to SDB05), fourteen sites from Selenga Delta shallow water
140 bodies (SLNG01, SLNG03-SLNG15), three sites from the Selenga River (B13-8-11, B13-8-20
141 and B13-8-26), and one shallow lake (Black Lake; BRYT) within the upstream section of the
142 Siberian Selenga River basin were analysed for Hg (Fig. 1; Table S2).

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144 Prior to water sample collection, bottles (120 mL PFA Savillex) were soaked in 5% Decon 90
145 solution for 24 hours, followed by multiple rinses of deionized water and then soaked in 1 M super
146 pure HCl for another 24 hours. This was then followed by extensive rinsing in deionized water and
147 double-bagging after drying. Unfiltered samples were acidified with 1.25 mL analytical grade HCl
148 (Romil Superpure 10M) and stored at 4°C prior to analyses. Short sediment cores (< 65 cm) were
149 collected using an *UWITEC* gravity corer (*UWITEC* Ltd., Austria) fitted with a 6.3 cm internal
150 diameter Perspex[®] acrylic tube (*UWITEC* Ltd.) in August 2013 from BAIK13-10 (core: BAIK13-
151 10A, water depth = 66 m), BAIK13-19 (core: BAIK13-19B, water depth = 460 m), and in March
152 2014 from SLNG04 (core: SLNG04-C, water depth = 1.3 m) (Fig. 1; Table S1; S2). The sediment
153 cores were extruded in the field at 0.2 cm (BAIK13-10A and 19B) or 0.5 cm (SLNG04-C) intervals
154 using a vertical extruder. Extruded sediment samples were stored in Whirlpak[®] bags, shipped to
155 University College London (UCL), London, UK and University of Nottingham, UK, and stored at
156 -20°C until processing. Radiometric chronologies for sediment core BAIK13-10A and BAIK13-
157 19B have been previously published in Roberts et al. (2018), and for SLNG04-C in Adams et al.
158 (2018) (Fig. S1). These ²¹⁰Pb chronologies were constructed using the constant rate of supply
159 (CRS) dating model (Appleby, 2001), and independently verified using ¹³⁷Cs.

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162 **2.2. Laboratory analysis of Hg concentrations in water samples**

163 Hg in water samples was analysed at the Environmental Mercury Analytical Facility at UCL, UK.
164 0.25 mL concentrated HCl (Rohm, pure grade) and 0.25 mL 0.1 N BrO³⁻/Br⁻ (purified) was added
165 to each 45 mL water sample, which was then sealed for 30 minutes, had 15 µg/L 12% NH₂OH-
166 HCl added, and diluted to 50 mL. Hg concentrations were analysed using gold trap cold vapour-
167 atomic fluorescence spectrometry (CV-AFS) following reduction with SnCl₂ (US EPA, 2002).
168 Detection limit is 0.4 ng/L; measurement errors for the Hg concentrations of less than 4 ng/L were
169 0.4 ng/g, and 10% for concentrations greater than 4 ng/L. Standard solutions and quality control
170 blanks were measured after every three samples to monitor measurement stability.

171 **2.3. Laboratory analysis of Hg concentrations in sediments**

172 Sediment samples were freeze-dried using a Thermo Modulyo D freeze-drier in which samples
173 were kept for several days at temperatures of -40°C until the pressure was < 100 mTorr. Freeze-
174 dried sediment samples were analysed at a temporal resolution of 5 – 20 years for BAIK13-10A
175 and BAIK13-19B. For SLNG04, samples were analysed through the core at a temporal resolution
176 of approximately 15 years. Hg analyses on sediment samples followed procedures in Yang et al.
177 (2010a). For each sample, approximately 0.2 g fine powdered freeze-dried sediment, was digested
178 with 8 mL of a 1:3 mixture of HNO₃ and HCl (aqua regia) at 100°C on a hotplate for 2 hours in
179 rigorously acid-leached 50 mL Teflon digestion tubes. Following digestion, samples were diluted
180 to 50 mL with deionized water, capped and mixed. Digested solutions were then analysed for Hg
181 using cold vapour-atomic fluorescence spectrometry (CV-AFS), following reduction with SnCl₂.
182 Standard reference material (GBW07305; certified Hg value of 100 ± 10.0 ng/g and measured
183 mean value is 104 ng/g, with RSD = 4.3 ng/g (n=3)), and sample blanks were digested with every
184 20 samples.

185 **2.4. Hg enrichment and total fluxes**

186 To examine trends in Hg loading over time, total Hg fluxes were calculated using the
187 radiometrically-derived sedimentation rates (Fig. S1). Standard enrichment factors could not be
188 calculated as lithogenic element data (for example Al, Li and Ti; Ribeiro et al., 2018) were not
189 available for the cores. Instead, Hg enrichment ratios (ER) were calculated by normalising Hg
190 concentrations in sediments deposited after 1850 CE, as determined from the age-depth model, by
191 the natural baseline (mean Hg concentrations prior to 1850 CE) (BAIK13-10A baseline mean =
192 30.4 ± 6.4 ng/g; BAIK13-19B = 35.5 ± 6.6 ng/g; SLNG04-C = 22.6 ± 1.2 ng/g). The calculated
193 ER therefore represent a comparative ratio of background vs post-1850 Hg concentrations (Yang
194 et al., 2010b). A baseline date of 1850 CE was chosen to take into account global atmospheric
195 contamination from industrialisation, despite the main regional development and expansion in the
196 Lake Baikal catchment region beginning in the 1900s (Brunello et al., 2004). An ER of > 1.4
197 demonstrates that post-1850 Hg concentrations are in exceedance of baseline by 2 SD, suggesting
198 post-1850 anthropogenic pollution. To examine trends in Hg loading, constrained cluster and
199 broken stick analyses were conducted on Hg concentration profiles from the three sediment cores,
200 to determine points of significant change, using the rioja package in R (version 3.5.2; R Core
201 Team, 2018) (Juggins, 2017).

202 **3. Results & Discussion**

203 **3.1. Spatial patterns and modern Hg sources**

204 Water Hg concentrations ranged between 5.3 and 10.1 ng/L in the Selenga Delta shallow water
205 bodies and between 0.3 and 5.5 ng/L in the Selenga Delta branches with a decreasing trend from
206 the Selenga River to the mouth of the delta (Fig. 1). Along the Selenga River, Hg concentrations
207 ranged from 6.0 to 8.1 ng/L with highest values at the furthest upstream locations near the town of

208 Ust-Kyakhta (B13-8-26) (Fig. 1). Black Lake (BRYT), within the Selenga River basin, had the
209 lowest Hg concentration of the shallow lakes, at 4.2 ng/L (Fig. 1). In the waters of Lake Baikal,
210 Hg concentrations reached 3.2 ng/L at the one site (BAIK13-19) in the north basin, near the Upper
211 Angara and ranged from below the limit of detection to 1.6 ng/L in the south and central basin lake
212 waters (Fig. 1), while near the Selenga Delta at BAIK13-10 the Hg concentration was 1.6 ng/L.

213 The spatial gradient from higher Hg concentrations in the upstream Selenga River to low
214 concentrations in Lake Baikal is expected due to the mining activity along the Selenga River, and
215 industrial activities in the cities of Ulan Ude and Selenginsk (Fig. 1). With the exception of
216 SLNG07, concentrations in the Selenga Delta shallow lakes are consistently higher than in the
217 Selenga Delta branches, and are higher than concentrations found in Lake Baikal. Mercury
218 concentrations are at their highest and most variable in lakes on the east side of the Delta but are
219 similar amongst lakes on the west side (Fig. 1). Single spot samples raise uncertainty regarding
220 their spatial and temporal representativity and should be interpreted with caution. Nevertheless,
221 the water Hg concentrations are likely indicating that the lakes of the Selenga Delta are acting as
222 retention ponds for Hg contamination within the Selenga River basin and preventing it from
223 entering Lake Baikal. River deltas are known hotspots for geochemical retention and
224 transformations, which may be controlled by seasonal and hydrological factors, including sediment
225 load and flow (Lychagin et al., 2015; Chalov et al., 2016). As most of the Hg in rivers is particle-
226 bound, much of it will tend to deposit in the smaller branches and shallow water bodies of the
227 Selenga Delta, as flow decreases (Amos et al., 2014). However, the fraction of the suspended
228 particle load in rivers that is buried is highly variable depending on freshwater discharge rates and
229 the physical characteristics of different deltas (Amos et al., 2014).

230 Lake Baikal surface water Hg concentrations in August 2013 (mean 1.52 ± 1.14 ng/L) were higher
231 than previously published values of 0.14 – 0.77 ng/L in June 1992 – 1993 (Meuleman et al., 1995;
232 Baeyens et al., 2002). The slightly elevated Hg concentration observed in the north basin at
233 BAIK13-19 (3.2 ng/L) are consistent with the suggestion that there is a nearby riverine source,
234 however, there is no supporting evidence that the Upper Angara River is impacting the water Hg
235 concentrations, through contamination from industry in the north basin catchment. The largest
236 town in this area is Severobaykalsk, and the largest village settlement previously reported is
237 Nizhneangarsk (Rose et al., 1998). The Baikal-Amur railroad also travels through this region. The
238 main Hg sources in Severobaykalsk are from fossil-fuel combustion facilities, waste incineration
239 processes and chemical or electrical industries. These sources have been demonstrated in past
240 studies to contribute to the higher than expected spheroidal carbonaceous particle (SCP)
241 concentrations in the north basin of Lake Baikal (BAIK28; Rose et al., 1998). Alongside
242 anthropogenic sources, another possible source of Hg into Lake Baikal is from the hydrothermal
243 vents at the bottom of the lake, which form as a result of the active tectonic rift boundary (Crane
244 et al., 1991; Kipfer et al., 1996). This geothermal activity mainly occurs in the north basin of Lake
245 Baikal and releases Hg into the sediments and water column via the hydrothermal waters which
246 are enriched in metals (Crane et al., 1991; Kipfer et al., 1996). Isotope ratios of Hg can be used to
247 distinguish between sources; however, it has been suggested that hydrothermal discharge along
248 fault lines at the bottom of Lake Baikal causes only a minor impact on the lake water chemistry
249 (Granina et al., 2007).

250 **3.2. Historic trends of sediment Hg contamination**

251 Hierarchical cluster analysis indicates that sedimentary Hg concentrations at BAIK13-10 increase
252 significantly at c. 1840 CE from 39 ng/g to 48 ng/g. At BAIK13-19, sedimentary Hg concentrations

253 increase towards the top of the core, with concentrations increasing significantly after 1920 CE
254 and remaining elevated to the surface (Fig. 2). While only two samples comprise the post-1940s
255 timeframe at BAIK13-19, they display similar concentrations of 53 and 51 ng/g. Hg concentrations
256 at SLNG04 showed a gradually increasing trend beginning c. 1950 CE, with a significant increase
257 in Hg concentration (c. 1960 CE) that continue to increase until a maximum concentration of 56
258 ng/g at c. 1990 CE. Sediment concentrations at SLNG04 then declined slightly after 1990 CE but
259 have remained relatively steady during the past two decades (Fig. 2). Sediment Hg concentrations
260 in Lake Baikal and the Selenga Delta are comparable with previous studies from Lake Baikal,
261 which reported values between c. 40 – 70 ng/g over a 16 cm sediment core depth, collected in 1990
262 CE (with no published sediment core chronology) (Leermakers et al., 1996).

263 Maximum and contemporary Hg concentrations show an approximate doubling of concentration
264 after 1945 CE across the sampled region, with recent concentrations close to 50 ng/g at all sites.
265 Sediments from BAIK13-10 show Hg enrichment, with Enrichment Ratios (ERs) ranging between
266 1.6 and 1.7 from 1910 CE to 2013 CE (Fig. 2). Similarly, the BAIK13-19 sediment core from
267 nearby the Upper Angara River in the north basin shows Hg enrichment in the upper sediments,
268 with ERs ranging between 1.2 and 1.5 from 1880 CE to 1960 CE (Fig. 2). Sediments from SLNG04
269 indicate little enrichment of Hg (ER c. 1.0) until the mid-20th century when Hg enrichment quickly
270 increased and was consistently > 1.4 between c. 1960 CE and 2013 CE (Fig. 2). Hg enrichment
271 peaks at c. 1990 CE at SLNG04 with an ER of 2.3, but declined to 1.9 by 2013 CE.

272 Total fluxes of Hg show higher values post-1850 CE, compared to pre-1850 CE, in both the south
273 basin (BAIK13-10) and north basin (BAIK13-19) sediment cores from Lake Baikal. However,
274 post-1850 CE Hg flux was 20-fold greater in the south basin compared to the north basin sediment
275 core (Fig. 2). In BAIK13-10, Hg fluxes ranged from 0.26 ng/cm²/yr in 1910 CE to 6.32 ng/cm²/yr

276 in 2013 CE (Fig. 2), whereas in the north basin (BAIK13-19) a smaller range in Hg flux is recorded
277 in the sediments over the post-1850 CE period (from 0.38 ng/cm²/yr in 1880 CE to 0.43 ng/cm²/yr
278 in 2013 CE (Fig. 2)). Due to limitations of radiometric dating, SLNG04 Hg flux can only be
279 calculated from the mid-20th century, but fluxes show a distinct increase between c. 1945 CE and
280 c. 1995 CE, from 2.3 to 11.0 ng/cm²/yr. Since c. 1995 CE, Hg flux at SLNG04 has declined slightly
281 to 8.1 ng/cm²/yr (Fig. 2).

282 Both modern water samples and sedimentary records from Lake Baikal show that lakes in the
283 Selenga Delta appear to retain Hg. In the sedimentary records this retention effect is apparent as
284 Hg enrichment levels in Selenga Delta sediment core (SLNG04: mean post-1850 = 6.47 ± 3.01
285 ng/cm²/yr) reach over 2-fold greater than baseline concentrations, which is a slightly higher range
286 than in the south basin sediments (BAIK13-10: mean post-1850 = 2.85 ± 2.27 ng/cm²/yr) in Lake
287 Baikal close to the Selenga Delta system (Fig. 1), and 18-fold higher compared to in the north
288 basin sediment core (BAIK13-19: mean post-1850 = 0.35 ± 0.09 ng/cm²/yr) (Fig. 2). The higher
289 sedimentary Hg fluxes in these Selenga Delta lakes, compared to Lake Baikal, is also expected
290 due to their closer proximity to the sources of Hg pollution within the Selenga River area. It is
291 important to note, however, that these enrichment levels are similar to those found in remote lakes
292 in Uganda, North America, Europe and Arctic Alaska, where Hg concentrations were up to 3-fold
293 higher than those in the pre-industrial period (Swain et al., 1992; Fitzgerald et al., 2005; Engstrom
294 et al., 2007; Yang et al, 2010a), which indicates that Hg loading at Lake Baikal is not greater than
295 the global background Hg enrichment levels. These enrichment levels in remote lakes (Swain et
296 al., 1992; Fitzgerald et al., 2005; Engstrom et al., 2007; Yang et al, 2010a) relate to atmospheric
297 deposition sources and not riverine drainage of industrial areas. Furthermore, Lake Baikal
298 sediment records covering the last 6 million years show naturally elevated Hg concentrations in

299 the sediments during warmer climatic conditions (average Hg concentrations of 46 ± 11 ng/g
300 during warm periods and 27 ± 12 ng/g during cold periods), and anomalously high peaks in Hg
301 concentrations (between 210 – 420 ng/g) during volcanic events in the Baikal area (Gelety et al.,
302 2007). By comparison, Hg concentrations from BAIK13-10 and BAIK13-19 are only slightly
303 higher than the average Hg concentration during warmer periods (Gelety et al., 2007).

304 Hg enrichment levels are lower in the north basin (average post-1850 ER for BAIK13-19 = $1.3 \pm$
305 0.16) than the south basin (average post-1850 ER for BAIK13-10 = 1.6 ± 0.05) and Selenga Delta
306 lake (average post-1850 ER for SLNG04 = 1.6 ± 0.42). Moreover, ER results suggest an
307 enrichment of north basin (BAIK13-19) sediments after 1940 CE, whereas the south basin
308 (BAIK13-10) site near the Selenga Delta experienced enrichment much earlier at around 1910 CE.
309 Such temporal differences in the onset and overall magnitude of Hg enrichment between north and
310 south basins, and the Selenga Delta, suggest local scale sources of Hg contamination. Hg
311 enrichment of the south basin sediments in the early 1900s suggests the contribution of
312 contamination from local sources as a result of industrialization in the Lake Baikal catchment and
313 the adjacent areas drained by the Angara and Lena rivers. The mid-20th century onset of Hg
314 enrichment in the north basin is perhaps attributed to the development of the major town on the
315 north basin shores, Severobaykalsk, which was only founded in the 1970s and with the completion
316 of the Baikal-Amur Mainline railway.

317 All three sediment cores indicate increases in Hg flux in Lake Baikal post-1850 CE, but the
318 subsurface peak in SLNG04 Hg flux indicates a possible mid-1990s peak in the delivery of Hg to
319 the Selenga River/Lake Baikal system from both local and long-range sources. Adams et al. (2018)
320 recorded similar timing in decline of polycyclic aromatic hydrocarbons (PAHs), polychlorinated
321 biphenyls (PCBs), and dichlorodiphenyltrichloroethane (DDT) fluxes to SLNG04, while Rose et

322 al. (1998) recorded evidence of SCP concentration declines in Lake Baikal sediments after 1990,
323 likely indicating a regional decline in industrial coal and oil combustion in southeast Siberia. The
324 timing of this observed decline in anthropogenic contamination in the Lake Baikal region ties in
325 with the economic recession in the early 1990s following the collapse of the former Soviet Union
326 (Khanin, 2003; Adams et al. 2018). However, the decline in Hg flux at SLNG04 is not large and
327 remains elevated relative to pre- c. 1950 CE levels. Differences in Hg flux between Lake Baikal
328 and the Selenga Delta are also likely due to the high affinity of Hg for organic matter; Hg binds to
329 DOC and the Selenga Delta lakes receive a higher input of catchment derived DOC than the pelagic
330 regions of Lake Baikal (Yoshioka et al., 2002). Thus, the higher input of DOC bound Hg into the
331 Selenga Delta lakes could be a contributing factor to the elevated levels of Hg enrichment seen in
332 these lakes in comparison to Lake Baikal. Alternatively, the Selenga Delta might be receiving
333 greater impacts from local sources than Lake Baikal, as a result of more sediments being deposited
334 in the SLNG04 location, and therefore SLNG04 is actually more highly contaminated by Hg
335 inputs. The large differences in water column depths between the coring sites may also effect Hg
336 fluxes, as within deeper water sites at BAIK13-10 (66 m) and BAIK13-19 (460 m) more particulate
337 matter decomposition will occur within the water column, than in the shallow Selenga Delta site,
338 SLNG04 (1.3 m). In deeper waters, more particulate-bound Hg will be released during particle-
339 scavenged remineralization down the water column, as well as photo-reductive and photo-induced
340 micro-biological processes, resulting in the evasion of Hg fluxes reaching deeper water sediments
341 (O'Driscoll et al., 2003).

342 In summary, sedimentary profiles in the south and north basin of Lake Baikal are likely to reflect
343 of both local sources and long-range atmospheric deposition of Hg, however the retention of Hg
344 in the Selenga Delta reduces inputs to Lake Baikal from the Selenga River. As Hg can remain

345 within the atmosphere for up to a year, an important anthropogenic source of Hg to Lake Baikal
346 and its catchment area is likely to be atmospherically transported Hg from industrial centres, from
347 other urban areas in Russia and across the globe (Gelety et al., 2007; UNEP Global mercury
348 assessment, 2013). Air pollution controls and mitigation efforts in North America and Europe have
349 helped to reduce their Hg emissions from industrial activity. However, in Asia (mainly China and
350 India), Hg emissions have been rising since the 1990s due to the marked economic expansion
351 (Pacyna et al., 2016; Sundseth et al., 2017). Declines in Hg ER and flux at SLNG04 since the late-
352 1900s indicates that long-range transport of Hg from elsewhere in Asia is likely to be an important
353 contributor to the enrichment at Lake Baikal; lake sediment cores from remote regions in China
354 show a marked increase in China's metal air pollution from 1990 CE (Wan et al., 2019) continuing
355 to present day (Yang et al., 2010b; UNEP Global mercury Assessment, 2013).

356 **3.3 Implications for Lake Baikal**

357 The 2013/2014 surveys of water Hg concentrations across Lake Baikal and the Selenga River basin
358 show elevated levels of Hg in the Selenga River waters, in comparison to Lake Baikal waters, most
359 likely linked to gold mining and location of industrial centres (Brunello et al., 2004; Thorslund et
360 al., 2012; 2016; Brumbaugh et al., 2013; Chalov et al., 2015; Jarsjö et al., 2017). However,
361 although the Selenga Delta reduces the extent of Hg pollution entering the south and central basins
362 of Lake Baikal, the current state of the environment in Lake Baikal's catchment gives cause for
363 concern with respect to future contamination by Hg.

364 For example, re-emission of legacy Hg stores has become another important source of Hg pollution
365 to the landscape, which can be released via soil erosion and permafrost thaw (Yang, 2015).
366 Modelling of current Hg reservoirs by Amos et al. (2013) indicated that up to 60% of present-day

367 atmospheric deposition of Hg is legacy-derived, re-emitted from surface reservoirs. Hydro-
368 climatic modelling studies for the Selenga River basin predict an increase in temperatures,
369 precipitation and run off between 2010 – 2099 under a high greenhouse gas emission scenario
370 (Törnqvist et al., 2014), which may lead to shifts in Hg loading as a result of altered hydrology
371 and basin-scale permafrost degradation (Zhoa et al., 2010; Törnqvist et al., 2014). Legacy Hg input
372 into Lake Baikal and the Selenga River basin is likely to increase with regional climate warming,
373 as permafrost underlays a large proportion of the catchment area (Hampton et al., 2008; Moore et
374 al., 2009) and catchment loading of Hg from the subsequent increased erosion of catchment soils
375 (Yang, 2015). In western Europe, changes to the climate system in recent years have also led to
376 increased storm events, causing further increased instability of catchment soils, increasing the
377 mobility of particulate-bound Hg across the terrestrial landscape (Yang and Smyntek, 2014). Thus,
378 Hg which has previously been deposited and stored within the lake catchment can also act as a
379 source of anthropogenic Hg to the lake system (Yang et al., 2002; Rose et al., 2012). Hg pollution
380 in Lake Baikal and the Selenga River basin area could therefore be a result of the continuing Hg
381 use in gold extraction processes in Mongolia, plus historical legacy of past Hg used in the region,
382 including in Russian gold mining prior to 1950 CE and industrial practices, as well as long-range
383 transport of atmospheric Hg from regional and international industrial centres, from metal
384 smelters, chemical and electrical industries, coal combustion facilities and waste incineration
385 plants.

386 Lake Baikal is increasingly facing pressures from shoreline anthropogenic nutrient pollution from
387 inadequate sewage treatment (Timoshkin et al., 2016), as well as pressures from recent
388 atmospheric warming since the 1950s which has been driving limnological and ecosystem changes
389 (Hampton et al., 2008; 2014; 2015; Moore et al., 2009; Izmet'eva et al., 2016; Silow et al., 2016;

390 Roberts et al., 2018). These pressures, combined with the continued inputs of Hg from a variety
391 of sources, put the Lake Baikal ecosystem at risk from Hg inputs into the future. Efforts need to
392 be focussed on minimising Hg pollution to Lake Baikal and its catchment area, primarily by
393 eliminating the current use of Hg in the extraction process of small-scale gold mining operations
394 in Mongolia. Furthermore, global efforts, in accordance with the Minamata Convention need to
395 continue, to reduce industrial release of Hg emissions into the atmosphere, which is likely a sizable
396 contribution of contemporary Hg to Lake Baikal. Additionally, Hg levels need to be monitored on
397 the freshwater ecosystems of the Selenga Delta itself, as it is an important Ramsar site for
398 continental Eurasia, and demonstrates higher levels of Hg within the Lake Baikal catchment.

399 **5. Conclusions**

400 Mercury measurements from 2013/2014 demonstrate that the Selenga River is a major source of
401 anthropogenic Hg contamination into the Selenga Delta region and Lake Baikal, as a result of the
402 variety of sources of Hg within the Selenga River basin, including chemical (mainly the
403 manufacturing of chlorine) and electrical plants where Hg is an element in the manufacturing
404 process, metallurgical plants which produce Hg directly, coal and oil fired electric power plants,
405 and current gold mining activity within the Mongolian Selenga River basin. The low Hg
406 concentrations within Lake Baikal waters could be attributed to retention within the Selenga Delta
407 system, which contains higher water Hg concentrations, and a result of dilution by the large volume
408 of Lake Baikal. The highest water concentrations within Lake Baikal are seen at a north basin site
409 near the Upper Angara River. Moreover, spatiotemporal differences in the timing of Hg
410 enrichment in Lake Baikal and Selenga Delta sediments likely highlight key influences of local
411 and regional sources of Hg to Lake Baikal during the 19th and 20th centuries. Recent moderate
412 declines in ERs and fluxes may reflect declining local sources of Hg within the catchment.

413 However, as concentrations currently remain elevated above background levels in all sediment
414 cores, long-range atmospheric sources likely continue to be a key contributor of Hg pollution in
415 Lake Baikal. Moreover, Hg concentrations measured in the sediments are similar to measurements
416 taken in the 1990s (Leermakers et al., 1996) and over warm climatic periods (Gelety et al., 2007).
417 Thus, with the projected hydro-climatic changes in the region from previous modelling studies,
418 there is a necessity to continue monitoring of Hg contamination for the protection of Lake Baikal
419 and the Selenga catchment, to reduce Hg pollution of this unique aquatic ecosystem and the
420 deterioration of a globally important freshwater resource.

421 With rising unregulated mining activity along the Selenga River, it is vital to monitor Hg pollution
422 across the Baikal catchment, especially as MeHg has already been found to bioaccumulate within
423 Lake Baikal's pelagic foodweb (Ciesielski et al., 2016). Furthermore, recent and future climate
424 warming is likely to increase the transfer of different forms of Hg, such as Hg bound DOC across
425 the terrestrial landscape, from thawing permafrost and soil erosion (Zhoa et al., 2010; Rose et al.,
426 2012; Törnqvist et al., 2014) and greater fluvial inflows into connected rivers. These climate driven
427 processes might increase the Hg loading within the Selenga River basin, and ultimately into pelagic
428 Lake Baikal and its foodweb.

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439 7. References

440
441 **Adams, J.K., Martins, C.C., Rose, N.L., Shchetnikov, A.A., Mackay, A.W. (2018).** ‘Lake
442 sediment records of persistent organic pollutants and polycyclic aromatic hydrocarbons in
443 Southern Siberia mirror the changing fortunes of the Russian economy over the past 70 years.’
444 *Environmental Pollution*, **242**, 528-538.

445
446 **Appleby, P.G. (2001).** Chronostratigraphic techniques in recent sediments. In: Last, W.M., Smol,
447 J.P. (Eds.), *Tracking Environmental Change Using Lake Sediments, Volume 1: Basin Analysis,*
448 *Coring, and Chronological Techniques.* Kluwer Academic Publishers, Dordrecht, pp. 171–203.

449
450 **Amos, H.M., Jacob, D.J., Streets, D.G., Sunderland, E.M. (2013).** ‘Legacy impacts of all-time
451 anthropogenic emissions on the global mercury cycle.’ *Global Biogeochemical Cycles*, *27*, 410-
452 421.

453
454 **Amos, H.M., Jacob, D.J., Kocman, D., Horowitz, H.M., Zhang, Y., Dutkiewicz, S., Horvat,**
455 **M., Corbitt, E.S., Krabbenhoft, D.P. (2014).** ‘Global biogeochemical implications of mercury
456 discharges from rivers and sediment burial.’ *Environment Science & Technology*, **48**, 9514-9522.

457
458 **Baeyens, W., Dehandschutter, B., Leermakers, M., Bobrov, V.A., Hus, R., Baeyens-Volant,**
459 **D. (2002).** ‘Natural mercury levels in geological enriched and geological active areas: case study
460 of Katun River and Lake Teletskoye, Altai (Siberia).’ *Water, Air and Soil Pollution*, **142**, 375 –
461 393.

462
463 **Bergan, T., Rodhe, H. (2001).** ‘Oxidation of elemental mercury in the atmosphere: constraints
464 imposed by global scale modelling.’ *Journal of Atmospheric Chemistry*, **40**, (2), 191-212.

- 465
466 **Brunello, A.J., Molotov, V.C., Dugherkhuu, B., Goldman, C., Khamaganova, E., Strijhova,**
467 **T., Sigman, R. (2004).** Lake Baikal: Lake Basin Management Initiative Experience and Lessons
468 Learned Brief. 1 – 26.
- 469
470 **Brumbaugh, W.G., Tillitt, D.E., May, T.W., Javzan, Ch., Komov, V.T. (2013).** ‘Environmental
471 survey in the Tuul and Orkhon River basins of north-central Mongolia, 2010: metals and other
472 elements in streambed sediment and floodplain soil.’ *Environmental Monitoring Assessment*, **185**,
473 8991 – 9008.
- 474
475 **Chalov, S.R., Jarsjö, J., Kasimov, N.S., Romanchenko, A.O., Pietron, J., Thorslund, J.,**
476 **Promakhova, E.V. (2015).** ‘Spatio-temporal variation of sediment transport in the Selenga River
477 Basin, Mongolia and Russia.’ *Environmental Earth Sciences*, **73**, (2), 663 – 680.
- 478
479 **Chalov, S., Thorslund, J., Kasimov, N., Aybullatov, D., Ilyicheva, E., Karthe, D., Kositsky,**
480 **A., Lychagin, M., Nittrouer, J., Pavlov, M., Pietron, J., Shinkareva, G., Tarasov, M.,**
481 **Garmaev, E., Akhtman, Y., Jarsjo, J. (2016).** ‘The Selenga River delta: a geochemical barrier
482 protecting Lake Baikal waters.’ *Regional Environmental Change*, **17**, (7), 2039 – 2053.
- 483
484 **Ciesielski, T.M., Pastukhov, M.V., Leeves, S.A., Farkas, J., Lierhagen, S., Poletaeva, V.I.,**
485 **Jenssen, B.M. (2016).** ‘Differential bioaccumulation of potentially toxic elements in benthic and
486 pelagic food chains in Lake Baikal.’ *Environmental Science and Pollution Research*, **23**, 15593 –
487 15604.
- 488
489 **Corbitt, E.S., Jacob, D.J., Holmes, C.D., Streets, D.G., Sunderland, E.M. (2011).** ‘Global
490 source-receptor relationships for mercury deposition under present day and 2050 emissions
491 scenarios.’ *Environmental Science & Technology*, **45**, (24), 10477-10484.
- 492
493 **Crane, K., Hecker, B., Golubev, V. (1991).** ‘Heat flow and hydrothermal vents in Lake Baikal,
494 U.S.S.R.’ *Eos*, **72**, (52), 588-589.

- 495
496 **Engstrom, D.R., Balogh, S.J., Swain, E.B. (2007).** ‘History of mercury inputs to Minnesota
497 lakes: Influences of watershed disturbance and localised atmospheric deposition.’ *Limnology &*
498 *Oceanography*, **52**, (6), 2467-2483.
499
- 500 **Fitzgerald, W.F., Engstrom, D.R., Lamborg, C.G., Tseng, C-M., Balcom, P.H.,**
501 **Hammerschmidt, C.R. (2005).** ‘Modern and historic atmospheric mercury fluxes in northern
502 Alaska: global sources and arctic depletion.’ *Environmental Science & Technology*, **39**, 557-568.
503
- 504 **French, T.D., Houben, A.J., Desforges, J-P.W., Kimpe, L.E., Kokelj, S.V., Poulain, A.J.,**
505 **Smol, J.P., Wang, X., Blais, J.M. (2014).** ‘Dissolved organic carbon thresholds affect mercury
506 bioaccumulation in arctic lakes.’ *Environmental Science & Technology*, **48**, 3162-3168.
507
- 508 **Granina, L.Z., Klerkx, J., Callender, E., Leermakers, M., Golobokova, L.P. (2007).** ‘Bottom
509 sediments and pore waters near a hydrothermal vent in Lake Baikal (Frolikha Bay).’ *Russian*
510 *Geology and Geophysics*, **48**, (3), 237-246.
511
- 512 **Gelety, V.F., Kalmykov, G.V., Parkhomenko, I.Y. (2007).** ‘Mercury in the sedimentary deposits
513 of Lake Baikal.’ *Geochemistry International*, **45**, (2), 170-177.
514
- 515 **Hammerschmidt, C.R., Fitzgerald, W., Lamborg, C.H., Balcom, P.H., Tseng, C-M. (2006).**
516 ‘Biogeochemical cycling of methylmercury in lakes and tundra watersheds of arctic Alaska.’
517 *Environmental Science & Technology*, **40**, 1204-1211.
518
- 519 **Hampton, S.E., Izmet’eva, L.R., Moore, M.V., Katz, S.L., Dennis, B., Silow, E.A. (2008).**
520 ‘Sixty years of environmental change in the world’s largest freshwater lake – Lake Baikal, Siberia.’
521 *Global Change Biology*, **14**, 1947 – 1958.
522
- 523 **Hampton, S.E., Gray, D.K., Izmet’eva, L.R., Moore, M.V., Ozersky, T. (2014).** ‘The Rise and
524 Fall of Plankton: Long-term changes in the vertical distribution of algae and grazers in Lake
525 Baikal, Siberia.’ *PLOS ONE*, **9**, (2), 1–10.

- 526
527 **Hampton, S.E., Moore, M.V., Ozersky, T., Stanley, E.H., Polashenski, C.M., Galloway,**
528 **A.W.E. (2015).** ‘Heating up a cold subject: prospects for under-ice plankton research in lakes.’
529 *Journal of plankton research*, **37**, (2), 277 - 284.
- 530
531 **Hampton, S.E., McGowan, S., Ozersky, T., Virdis, S.G.P., Vu, T.T, Spanbauer, T.L. et al**
532 **(2018).** ‘Recent ecological change in ancient lakes.’ *Limnology and Oceanography*, **63**, (5), 2277
533 – 2304.
- 534
535 **Hintelmann, H., Welbourn, P.M., Evans, R.D. (1995).** ‘Binding of methylmercury compounds
536 by humic and fluvic acids.’ *Water, Air and Soil Pollution*, **80**, 1031-1034.
- 537
538 **Izmet’eva, L.R., Moore, M.V., Hampton, S.E., Ferwerda, C.J., Gray, D.K., Woo, K.H., et**
539 **al. (2016).** ‘Lake-wide physical and biological trends associated with warming in Lake Baikal.’
540 *Journal of Great Lakes Research*, **42**, 6–17.
- 541 **Jarsjö, J., Chalov, S.R., Pietron, J., Alekseenko, A.V., Thorslund J. (2017).** ‘Patterns of soil
542 contamination, erosion, and river loading of metals in a gold mining region of northern
543 Mongolia.’ *Regional Environmental Change*, **17**, 1991-2005.
- 544 **Juggins, S. (2017).** rioja: Analysis of Quaternary Science Data, R package version (0.9-15.1).
545 (<http://cran.r-project.org/package=rioja>).
- 546 **Kasimov, N., Karthe, D., Chalov, S. (2017).** ‘Environmental change in the Selenga River – Lake
547 Baikal Basin.’ *Regional Environmental Change*, **17**, 1945 – 1949.
- 548
549 **Kaus, A., Schäffer, M., Karthe, D., Büttner, O., Tümpling, W., Borchardt, D. (2017).**
550 ‘Regional patterns of heavy metal exposure and contamination in the fish fauna of the Kharaa
551 River basin (Mongolia).’ *Regional Environmental Change*, **17**, 2023 – 2037.
- 552
553 **Kelly, C.A., Rudd, J.W.M., Holoka, M.H. (2003).** ‘Effect of pH on mercury uptake by an aquatic
554 bacterium: implications for Hg cycling.’ *Environmental Science Technology*, **37**, 2941-2946.

- 555
556 **Khanin, G.I. (2003).** ‘The 1950s – the triumph of the Soviet economy.’ *Europe-Asia Studies* **55**,
557 1187-573 1212.
- 558
559 **Kipfer, R., Aeschbach-Hertig, W., Hofer, M., Hohmann, R., Imboden, D.M., Baur, H.,**
560 **Golubev, V., Klerkx, J. (1996).** ‘Bottom water formation due to hydrothermal activity in Frolikha
561 Bay, Lake Baikal, eastern Siberia.’ *Geochimica et Cosmochimica Acta*, **60**, (6), 961-971.
- 562
563 **Koval, P.V., Kalmychkov, G.V., Gelety, V.F., Leonova, G.A., Medvedev, V.I., Andrulaitis,**
564 **L.D. (1999).** ‘Correlation of natural and technogenic mercury sources in the Baikal polygon,
565 Russia.’ *Journal of Geochemical Exploration*, **66**, 277 – 289.
- 566
567 **Leermakers, M., Meuleman, C., Baeyens, W. (1996).** ‘Mercury distribution and fluxes in Lake
568 Baikal.’ In: Global and Regional mercury cycles: sources, fluxes and mass balances. Eds Baeyens
569 et al. Kluwer Academic Publishers. 303 – 315.
- 570
571 **Logachev, N.A. (2003).** ‘History and geodynamics of the Baikal rift.’ *Russian Geology and*
572 *Geophysics*, **44**, (5), 391 – 406.
- 573
574 **Lychagin, M.Y., Tkachenko, A.N., Kasimov, N.S., Kroonenberg, S. (2015).** ‘Heavy metals in
575 the water, plants, and bottom sediments of the Volga River mouth area.’ *Journal of Coastal*
576 *Research* **31**, (4), 859–868.
- 577
578 **Nadmitov, B., Hong, S., Kang, S.I., Chu, J.M., Gomboev, B., Janchivdorj, L., Lee, C-H.,**
579 **Khim, J, S. (2015).** ‘Large-scale monitoring and assessment of metal contamination in surface
580 water of the Selenga River Basin (2007 – 2009).’ *Environmental Science and Pollution Research*,
581 **22**, 2856 – 2867.
- 582
583 **Nikanorov, A.M., Reznikov, S.A., Mateev, A.A., Arakelyan, V.S. (2012).** ‘Monitoring of
584 Polycyclic Aromatic Hydrocarbons in the Lake Baikal basin in the areas of intensive
585 anthropogenic impact.’ *Russian Meteorology and Hydrology*, **37**, (7), 477 – 484.

- 586
587 **Nomokonova, E., Lin, S-C., Chen, G. (2013).** ‘Investigation of safety compliance and safety
588 participation as well as cultural influences: Using Selenginsk Pulp and Cardboard Mill in Russia
589 as an example.’ Proceedings of the Institute of Industrial Engineers Asian Conference Eds. Yi-
590 Kuei Lin, Yu-Chung Tsao, Shi-Woei Lin. Pp. 1001 – 1007.
591
- 592 **Maruev, V.A. (2018).** History of the gold-mining industry in the barring in the XIX-beginning of
593 the XX century. Transbaikal State University, PhD thesis, 223 pp (in Russian).
594
- 595 **Meuleman, C., Leermakers, M., Baeyens, W. (1995).** ‘Mercury speciation in Lake Baikal.’
596 *Water, Air and Soil Pollution*, **80**, 539 – 551.
597
- 598 **Misyurkeeva, Yu. (2009).** The barbarous "gold rush" in Mongolia poisons Selenga. Novaya
599 Buryatiya 39, 12-14. (in Russian).
600
- 601 **Moore, M.V., Hampton, S.E., Izmet'eva, L.R., Silow, E.A., Peshkova, E.V., Pavlov, B.K.**
602 **(2009).** ‘Climate change and the world’s “sacred sea” – Lake Baikal, Siberia.’ *Bioscience*, **59**, (5),
603 405 – 417.
604
- 605 **O’Driscoll, N.J.O., Beauchamp, S., Siciliano, S.D., Rencz, A.N., Lean, D.R.S. (2003).**
606 ‘Continuous analysis of dissolved gaseous mercury (DGM) and mercury flux in two freshwater
607 lakes in Kejimikujik Park, Nova Scotia: Evaluating mercury flux models with quantitative data.’
608 *Environmental Science & Technology*, **37**, 2226-2235.
609
- 610 **Ozersky, T., Pastukhov, M.V., Poste, A.E., Deng, X.Y., Moore, M.V. (2017).** ‘Long-term and
611 ontogenetic patterns of heavy metal contamination in Lake Baikal seals (*Pusa sibirica*).’
612 *Environmental Science & Technology*, **51**, 10316 – 10325.
613
- 614 **Pacyna, J.M., Travnikov, O., De Simone, F., Hedgecock, I.M., Sundseth, K., Pacyna, E.G.,**
615 **Steenhuisen, F., Pirrone, N., Munthe, J., Kindbom, K. (2016).** ‘Current and future levels of

616 mercury atmospheric pollution on a global scale.’ *Atmospheric Chemistry and Physics*, **16**, 12495
617 – 12511.

618
619 **Perrot, V., Epov, V.N., Pastukhov, M., Grebenschikova, V., Zouiten, C., Sonke, J.E.,**
620 **Husted, S., Donard, O.F.X., Amouroux, D. (2010).** ‘Tracing sources and bioaccumulation of
621 mercury in fish of Lake Baikal – Angara River using Hg isotopic composition.’ *Environmental*
622 *Science & Technology*, **44**, (21), 8030 – 8037.

623
624 **Perrot, V., Pastukhov, M.V., Epov, V.N., Husted, S., Donard, O.F.X., Amouroux, D. (2012).**
625 ‘Higher mass-independent isotope fractionation of methylmercury in the pelagic food web of Lake
626 Baikal (Russia).’ *Environmental Science & Technology*, **46**, 5902-5911.

627
628 **Pietron, J., Chalov, S.R., Chalova, A.S., Alekseenko, A.V., Jarsjö, J. (2017).** ‘Extreme spatial
629 variability in riverine sediment load inputs due to soil loss in surface mining areas of the Lake
630 Baikal basin.’ *Catena*, **152**, 82 – 93.

631 **Pisarsky, B.I., Hardina, A.M., Naganawa, H. (2005).** ‘Ecosystem evolution of Lake Gusinoe
632 (Transbaikal region, Russia).’ *Limnology*, **6**, 173 – 182.

633
634 **Ribeiro, C., Couto, C., Ribeiro, A.R., Maia, A.S., Santos, M., Tiritan, M.E., Pinto, E.,**
635 **Almeida, A.A. (2018).** ‘Distribution and environmental assessment of trace elements
636 contamination of water, sediments and flora from Douro River estuary, Portugal.’ *Science of the*
637 *Total Environment*, **639**, 1382-1393.

638
639 **Roberts, S.L., Swann, G.E.A., McGowan, S., Panizzo, V.N., Vologina, E.G., Sturm, M.,**
640 **Mackay, A.W. (2018).** ‘Diatom evidence of 20th century ecosystem change in Lake Baikal,
641 Siberia.’ *PLOS ONE*, **13**, (12), 1-20.

642
643 **Rose, N.L., Appleby, P.G., Boyle, J.F., Mackay, A.W., Flower, R.J. (1998).** ‘The spatial and
644 temporal distribution of fossil-fuel derived pollutants in the sediment record of Lake Baikal,
645 eastern Siberia.’ *Journal of Paleolimnology*, **20**, 151 – 162.

646

- 647 **Rose, N.L., Yang, H., Turner, S.D., Simpson, G.L. (2012).** ‘An assessment of the mechanisms
648 for the transfer of lead and mercury from atmospherically contaminated organic soils to lake
649 sediments with particular reference to Scotland, UK.’ *Geochimica et Cosmochimica Acta*, **82**, 113-
650 135.
- 651
- 652 **Scholz, C.A., Hutchinson, D.R. (2000).** ‘Stratigraphic and structural evolution of the Selenga
653 Delta Accommodation Zone, Lake Baikal Rift, Siberia.’ *International Journal of Earth Sciences*,
654 **89**, 212 – 228.
- 655
- 656 **Silow, E.A., Krashchuk, L.S., Onuchin, K.A., Pislegina, H.V., Rusanovskaya, O.O.,
657 Shimaraeva, S.V. (2016).** ‘Some recent trends regarding Lake Baikal phytoplankton and
658 zooplankton.’ *Lakes and Reservoirs and Management*, **21**, 40-44.
- 659
- 660 **Sundseth, K., Pacyna, J.M., Pacyna, E.G., Pirrone, N., Thorne, R.J. (2017).** ‘Global sources
661 and pathways of mercury in the context of human health.’ *International Journal of Environmental
662 Research and Public Health*, **14**, 105, 1-14.
- 663 **Swain, E.B., Engstrom, D.R., Brigham, M.E., Henning, T.A., Brezonik, P.L. (1992).**
664 ‘Increasing rates of atmospheric mercury deposition in midcontinental north America.’ *Science*,
665 **257**, 784-787.
- 666 **Thorslund, J., Jarsjö, J., Chalov, S.R., Belozerova, E.V. (2012).** ‘Gold mining impact on
667 riverine heavy metal transport in a sparsely monitored region: the upper Lake Baikal Basin case.’
668 *Journal of Environmental Monitoring*, **14**, (10), 2780 – 2792.
- 669
- 670 **Thorslund, J., Jarsjö, J., Wällstedt, T., Mörth, C.M., Lychagin, M.Y., Chalov, S.R. (2016).**
671 ‘Speciation and hydrological transport of metals in non-acidic river systems of the Lake Baikal
672 basin: Field data and model predictions.’ *Regional Environmental Change*, **17**, (7), 2007 – 2021.
- 673
- 674 **Timoshkin, O.A., Samsonov, D.P., Yamamuro, M., Moore, M.V., Belykh, O.I., Malnik, V.V.
675 et al. (2016).** ‘Rapid ecological change in the coastal zone of Lake Baikal (East Siberia): Is the

676 site of the world's greatest freshwater biodiversity in danger?' *Journal of Great Lakes Research*,
677 **42**, (3), 487-497.

678
679 **Törnqvist, R., Jarsjö, J., Pietron, J., Bring, A., Rogberg, P., Asokan, S.M. (2014).** 'Evolution
680 of the hydro-climate system in the Lake Baikal basin.' *Journal of Hydrology*, **519**, 1953-1962.

681
682 **Tumenbayer, B., Batbaya, M., Grayson, R. (2000).** 'Environmental hazard in Lake Baikal
683 watershed posed by mercury placer in Mongolia.' *World Placer Journal*, **1**, 134 – 159.

684
685 **United Nations Environmental Programme (UNEP) Global Mercury Assessment. (2013).**
686 'Global Mercury Assessment 2013 Sources, Emissions, Releases and Environmental Transport.
687 UNEP Chemicals Branch, Geneva, Switzerland.

688
689 **United Nations (UN) Environment. (2017).** 'Minamata Convention on Mercury.'
690 www.mercuryconvention.org

691
692 **United States Environmental Protection Agency. (2002).** Method 1631, Revision E: Mercury
693 in water by oxidation, purge and trap, and cold vapor atomic fluorescence spectrometry.
694 Washington, D.C, United States of America.

695
696 **Vasiliev, O.F., Obolenskiy, A.A., Yagolnitsler, M.A. (1998).** 'Mercury as a pollutant in Siberia:
697 sources, fluxes and a regional budget.' *The Science of the Total Environment*, **213**, 73-84.

698
699 **Wan, D., Song, L., Mao, X., Yang, J., Jin, Z., Yang, H. (2019).** 'One-century sediment records
700 of heavy metal pollution on the southeast Mongolian Plateau: Implications for air pollution trend
701 in China.' *Chemosphere*, **220**, 539-545.

702
703 **Yang, H., Rose, N.L., Battarbee, R.W. (2002).** 'Mercury and lead budgets for Lochnagar, a
704 Scottish Mountain Lake and its catchment.' *Environment, Science & Technology*, **36**, 1383-1388.

705

706 **Yang, H., Engstrom, D.R., Rose, N.L. (2010a).** ‘Recent changes in atmospheric mercury
707 deposition recorded in the sediments of remote equatorial lakes in the Rwenzori mountains,
708 Uganda.’ *Environmental Science & Technology*, **44**, 6570 – 6575.

709
710 **Yang, H., Battarbee, R.W., Turner, S., Rose, N.L., Derwent, R.G., Wu, G., Yang, R. (2010b).**
711 ‘Historic reconstruction of mercury pollution across the Tibetan Plateau using lake sediments.’
712 *Environmental Science & Technology*, **44**, 2918-2924.

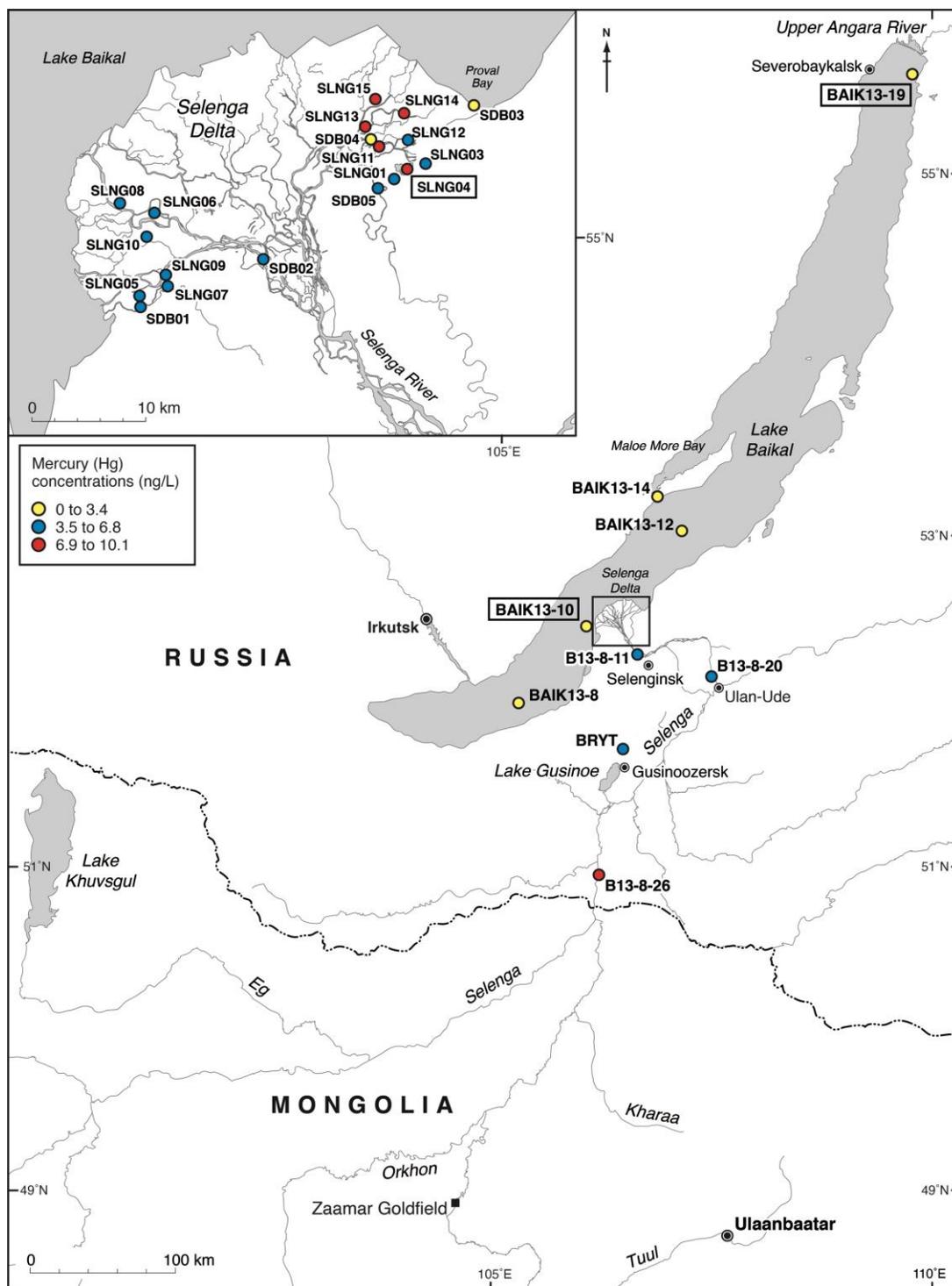
713• **Yang, H., Smyntek, P. (2014).** ‘The mercury record in Red Tarn sediments reveals air pollution
714 history and implications of catchment erosion.’ *Environment Science: Processes and Impacts*, **16**,
715 (11).

716
717 **Yang, H. (2015).** ‘Lake sediments may not faithfully record decline of atmospheric pollutant
718 deposition.’ *Environmental Science & Technology*, **49**, (21), 12607–12608.

719
720 **Yoshioka, T., Ueda, S., Khodzher, T., Bashenkhaeva, N., Korovyakova, I., Sorokovikova, L.,**
721 **Gorbunova, L. (2002).** ‘Distribution of dissolved organic carbon in Lake Baikal and its
722 watershed.’ *Limnology*, **3**, 159-168.

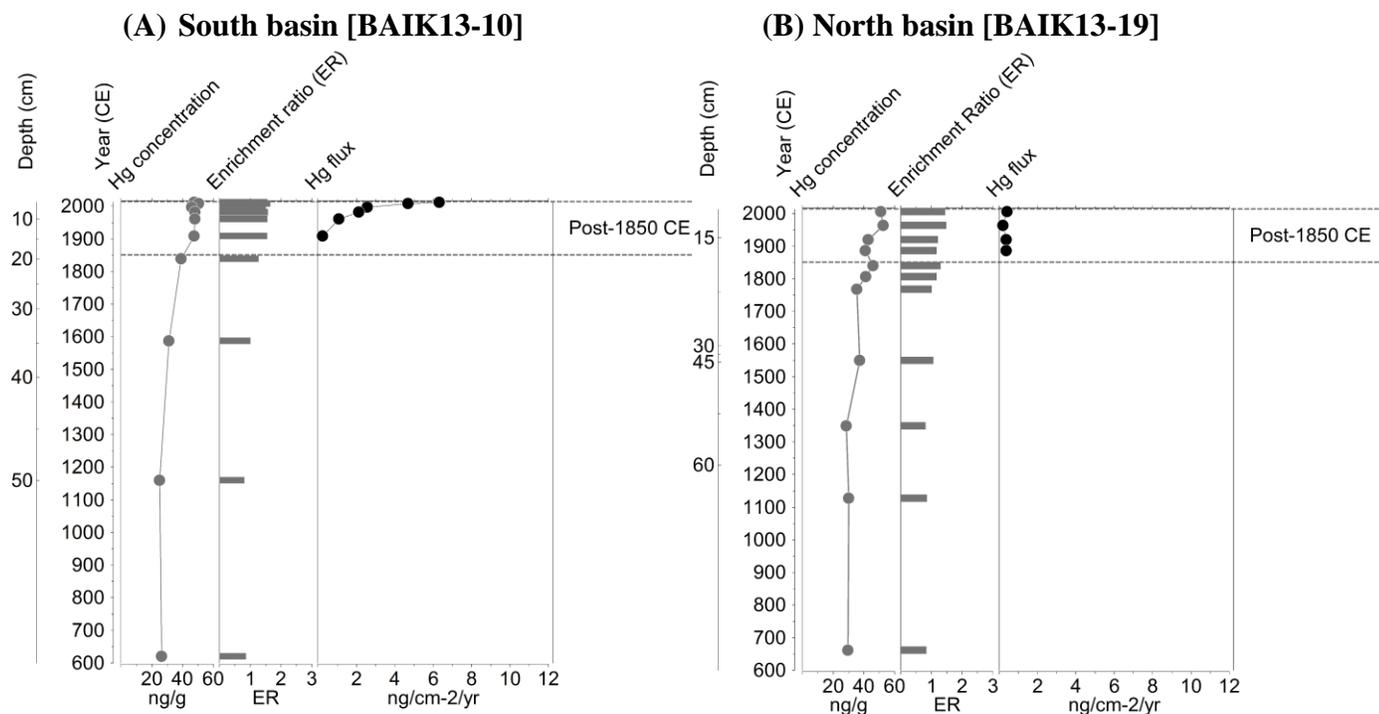
723
724 **Zhao, L., Qingbai, W., Marchenko, S.S., Sharkhuu, N. (2010).** ‘Thermal state of permafrost
725 and active layer in central Asia during the international polar year.’ *Permafrost and periglacial*
726 *processes*, **21**, 198–207.

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764 **Fig 1. Location of study sites and other key locations referred to in the text across Lake**
 765 **Baikal and the Selenga River catchment. Mercury (Hg) concentrations (ng/L) in surface**
 766 **water samples collected from the Selenga River, Selenga Delta and Lake Baikal.**

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(C) Selenga Delta [SLNG04]

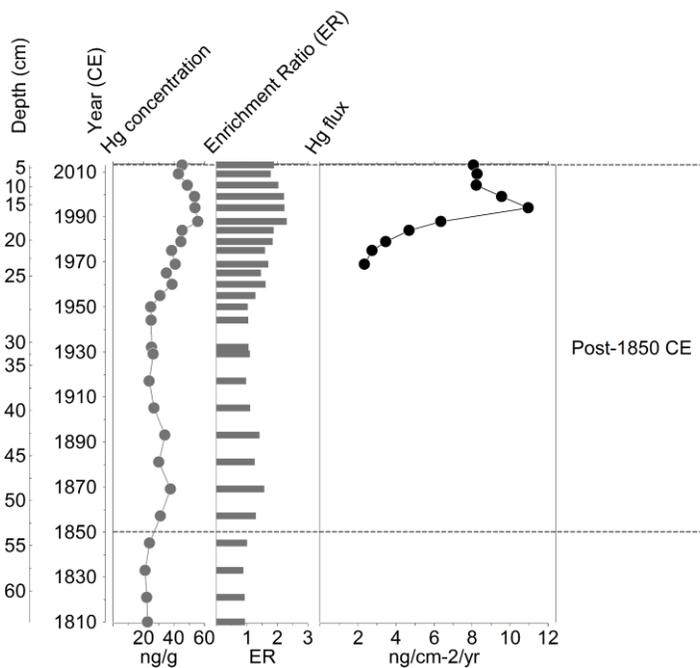


Fig 2. Sedimentary mercury (Hg) concentrations (ng/g), enrichment ratios (ER) and Hg fluxes (ng/cm²/yr) profiles from the (A) south basin [BAIK13-10], (B) north basin [BAIK13-19] in Lake Baikal and (C) Selenga Delta [SLNG04]. For SLNG04 all the dates beyond c. 1945 are extrapolations of constant background sedimentation rates pre-1980.