

1 Mercury loading within the Selenga River Basin and Lake 2 Baikal, Siberia

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25

26 Abstract

27 Mercury (Hg) loading in Lake Baikal, a UNESCO world heritage site, is growing and poses a
28 serious health concern to the lake's ecosystem due to the ability of Hg to transform into a toxic

29 form, known as methylmercury (MeHg). Monitoring of Hg into Lake Baikal is spatially and
30 temporally sparse, highlighting the need for insights into historic Hg loading. This study reports
31 measurements of Hg concentrations from water collected in August 2013 and 2014 from across
32 Lake Baikal and its main inflow, the Selenga River basin (Russia). We also report historic Hg
33 contamination using sediment cores taken from the south and north basins of Lake Baikal, and a
34 shallow lake in the Selenga Delta. Field measurements from August 2013 and 2014 show high Hg
35 concentrations in the Selenga Delta and river waters, in comparison to pelagic lake waters.
36 Sediment cores show temporal heterogeneity of Hg enrichment across Lake Baikal since the mid-
37 19th century, increasing first in the southern basin in the late-19th century, and increasing in the
38 north basin in the mid-20th century. Hg enrichment was greatest in the Selenga Delta shallow lake
39 (ER = 2.3 in 1994 CE), with enrichment occurring in the mid- to late-20th century. Local sources
40 of Hg are predominantly from gold (Au) mining along the Selenga River, which have been
41 expanding over the last few decades. More recently, another source is atmospheric deposition from
42 industrial activity in Asia, due to rapid economic growth across Asia since the 1980s. As Hg can
43 bioaccumulate and biomagnify through trophic levels to Baikal's top consumer, the world's only
44 truly freshwater seal (*Pusa sibirica*), it is vital that Hg input at Lake Baikal and within its catchment
45 is monitored and controlled.

46
47 Sediment cores show greatest Hg enrichment in the Selenga Delta water body in the mid to late
48 20th century in response to gold mining and industrial activities along the Selenga River.

49
50 **Keywords:** Pollution, mining, long-range atmospheric deposition, lake sediments, Russia

51

52 **1. Introduction**

53 Mercury (Hg) is a global pollutant of concern and has both natural and anthropogenic sources.
54 Once emitted, most inorganic Hg can remain in the atmosphere for up to 12 months (Corbitt et al.,
55 2011) and can be transported across the world. Gaseous and particulate Hg emitted into the
56 atmosphere is transformed into Hg (II), which is then deposited onto the landscape via wet and dry
57 deposition (Bergan and Rodhe, 2001). Atmospherically emitted Hg will cycle between short-term
58 stores (<10 years) in the atmosphere, terrestrial environments, and surface ocean waters, before
59 being sequestered long-term into terrestrial soils and sediments, ocean margins and the deep ocean
60 (Amos et al., 2014). Within aquatic environments, methylating bacteria can transform Hg (II) into
61 a toxic organic form, known as methylmercury (MeHg). This organic form makes Hg especially
62 harmful within aquatic ecosystems as it can bioaccumulate and biomagnify in foodwebs. Due to
63 the toxicity of MeHg, the Minamata Convention was set up in 2017 to reduce the impact that human
64 activities have on Hg releases to the environment (UN Environment, 2017). Hg cycling in aquatic
65 environments may be affected by dissolved organic carbon (DOC), pH, temperature, redox
66 conditions, sulfate concentrations and microbial activity, which control methylation
67 (transformation of Hg into MeHg) and demethylation (transformation of MeHg into Hg) processes
68 (Hintelmann et al., 1995; Kelly et al., 2003; French et al., 2014). Environmental changes associated
69 with warming (e.g. increased weathering, temperature, productivity and organic loadings) can also
70 affect Hg cycling, by stimulating methylation and inhibiting photodecomposition, due to
71 increasing primary productivity and DOC which reduce light penetration in the water column
72 (Hammerschmidt et al., 2006).

73 Lake Baikal is a UNESCO World Heritage Site and is internationally important for its high levels
74 of water purity and endemism. The lake can be divided into three main basins (south, central and

75 north) (Fig. 1) with the central basin separated from the south basin by the Buguldeika Ridge and
76 the more than 20 km wide Selenga River Delta. The Selenga River, which is approximately 943
77 km long (Nadmitov et al., 2015), is the main tributary into Lake Baikal and contributes over 60%
78 of annual flow into the lake. It originates in the Khangai Mountains, northern Mongolia, and
79 accounts for over 80% (over 447,000 km²) of Baikal's catchment (Nadmitov et al., 2015). The
80 majority of the Selenga River basin is situated in Mongolia (282,349 km²) rather than Russia (148,
81 060 km²), with the basin covering almost 20% of the total land area in Mongolia (Nadmitov et al.,
82 2015). The Selenga River branches into the Selenga Delta, the world's largest freshwater inland
83 delta (Logachev, 2003), and a Ramsar-designated floodplain wetland, which is internationally
84 important for high rates of biodiversity and migratory bird habitat (Scholz and Hutchinson, 2000).

85 **1.1. Sources of Hg in the Lake Baikal region**

86 Gold (Au) mining began in Lake Baikal's catchment with the discovery of the Ildikan deposit in
87 the mid-1800s (Maruev, 2018). Small-scale gold mining operations use Hg to extract Au from ore
88 in a process of amalgamation and distillation. The first Au extraction processes using Hg started
89 along the Kharaa River, in the basin of the Amur River in 1837 CE (common era), and in the basin
90 of the Selenga River in 1841 CE (Misyurkeeva, 2009; Maruev, 2018). Between 1860-1890 CE
91 40% of all Au in Russia was mined in the Baikal region, with Hg used in the extraction before
92 being disposed in rivers and dispersed into the atmosphere (Maruev, 2018). Since the 1950s, the
93 use of Hg in Au extraction has stopped in the Russian region of the Baikal catchment, but continues
94 in the Mongolian Selenga River basin (Misyurkeeva, 2009). Over the last few decades, Au
95 extraction along the Selenga River has increased, with over 700 mines currently in operation in
96 the Baikal catchment within Mongolia (Brunello et al., 2004; Pietron et al., 2017), and the largest
97 gold mining operation, the Zaamar Goldfield, situated within the Mongolian Selenga River basin

98 (Fig. 1) (Tumenbayer et al., 2000; Chalov et al., 2015; Pietron et al., 2017). Recent studies report
99 the Lake Baikal catchment and Selenga River basin to be heavily polluted from these Au extraction
100 activities (Brunello et al., 2004; Thorslund et al., 2012; 2016; Brumbaugh et al., 2013; Chalov et
101 al., 2015; Jarsjö et al., 2017; Hampton et al., 2018).

102 The Lake Baikal catchment and Selenga River basin were heavily industrialised between the 1950s
103 and the 1990s and became known as one of the most highly Hg polluted regions in Siberia (Koval
104 et al., 1999). The largest cities and main industrial districts in Mongolia (Ulaanbaatar, Erdenet and
105 Darkhan) are situated along the main tributaries of the Selenga River; Tuul, Orkhon and Kharaa
106 rivers, respectively, and in Russia, Ulan Ude and Selenginskii are situated along the Selenga River
107 (Kasimov et al., 2017), while the major cities and towns of Irkutsk, Gusinozersk and
108 Severobaykalsk are within Lake Baikal's catchment and airshed (Fig. 1). Notorious industrial Hg
109 emitters in the region include metallurgical plants which produce Hg directly, chemical and
110 electrical plants, where Hg is an element in the manufacturing process, and hydrocarbon/coal or
111 oil fired thermal electric power plants, where Hg is recovered (Vasiliev et al., 1998). Chemical
112 industries are prominent within the Irkutsk-Cheremkhovo industrial zone and are a major concern
113 for Hg pollution (Koval et al., 1999). Other major regional Hg pollution sources include the
114 Gusinozersk State Regional Power Plant (a coal-fired power plant), and the Selenginsk Pulp and
115 Cardboard Mill within the Selenga River basin, which began operating in 1974 CE and continued
116 as an open system until 1990 CE (Pisarksy et al., 2005; Nikanorov et al., 2012; Nomokonova et
117 al., 2013). Industrial activity around the shores of Lake Baikal began in the 20th century, and the
118 Baikal Pulp and Paper Mill (BPPM), which was in operation between 1966 to 2013, was a
119 suggested point source of Hg (Brunello et al., 2004).

120

121 **1.2. Hg toxicity in Lake Baikal**

122 Within the past decade, MeHg bioaccumulation has been observed in Baikal's pelagic foodweb
123 (Perrot et al., 2010; 2012; Ciesielski et al., 2016). High Hg concentrations have been reported in
124 fish from the Selenga River basin, which are above the recommended thresholds for human
125 consumption (Kaus et al., 2017), and within fish from the Bratsk water reservoir in the Baikal
126 region (Koval et al., 1999). Analyses on the livers and muscle of the Baikal Seal (*Pusa sibirica*),
127 has also shown Hg contamination within the lake's top consumer in the 1960's and 1970's, before
128 declining to present in response to reduced atmospheric Hg emissions from Europe and Russia
129 (Ozersky et al., 2017).

130 **1.3. Rationale and research questions**

131 Recent and current levels of Hg contamination at Lake Baikal are largely unknown due to sparse
132 records of Hg measurements and the lack of historical Hg loading records for the region. This
133 study undertakes the first Hg assessment for Lake Baikal in 20 years (Leermakers et al., 1996).
134 Herein, we report measurements of Hg concentrations from water samples collected in August
135 2013/2014 from across Lake Baikal and the Selenga River basin and establish historic Hg
136 contamination records over the past c. 200 years using lake sediment cores. Moreover, we assess
137 potential sources of contemporary and past Hg loading and transport within the catchment together
138 with the relative importance of local vs. long-range sources of contamination.

139 **2. Methodology**

140 **2.1. Study sites and field collection**

141 Five sites were selected within Lake Baikal for surface water sampling to represent the main
142 basins, including the south basin (BAIK13-8), the shallow waters off the Selenga Delta (BAIK13-
143 10), the central basin (BAIK13-12), within Maloe More Bay off the central basin (BAIK13-14),

144 and the Upper Angara River in the north basin (BAIK13-19) (Fig. 1; Table 1). Maloe More Bay is
145 a vulnerable region of Lake Baikal, currently affected more than deeper water sites by
146 anthropogenic influence (Timoshkin et al., 2016). Additionally, water samples at five sites from
147 the Selenga Delta branches (SDB01 to SDB05), fourteen sites from Selenga Delta shallow water
148 bodies (SLNG01, SLNG03-SLNG15), three sites from the Selenga River (B13-8-11, B13-8-20
149 and B13-8-26), and one shallow lake (Black Lake; BRYT) within the upstream section of the
150 Siberian Selenga River basin were analysed for Hg (Fig. 1; Table 1).

151
152 Prior to water sample collection, bottles (120 mL Savillex) were soaked in 5% Decon 90 solution
153 for 24 hours, followed by multiple rinses of deionized water and then soaked in 1 M super pure
154 HCl for another 24 hours. This was then followed by extensive rinsing in deionized water and
155 double-bagging after drying. Unfiltered samples were acidified with 1.25 mL analytical grade HCl
156 (Romil Superpure 10M) and stored at 4°C prior to analyses. Short sediment cores (< 65 cm) were
157 collected using an *UWITEC* gravity corer (UWITEC Ltd., Austria) fitted with a 6.3 cm internal
158 diameter Perspex[®] tube in August 2013 from BAIK13-10 (core: BAIK13-10A), BAIK13-19 (core:
159 BAIK13-19B), and in March 2014 from SLNG04 (core: SLNG04-C) (Fig. 1; Table 1; Table S1).
160 The sediment cores were extruded in the field at 0.2 cm (BAIK13-10A and 19B) or 0.5 cm
161 (SLNG04-C) intervals using a vertical extruder. Extruded sediment samples were stored in
162 Whirlpak[®] bags, shipped to University College London (UCL), London, UK and University of
163 Nottingham, UK, and stored at -20°C until processing. Radiometric chronologies for sediment
164 core BAIK13-10A and BAIK13-19B have been previously published in Roberts et al. (2018), and
165 for SLNG04-C in Adams et al. (2018) (Fig. S1).

166

167 Table 1. Site code, geographical location and mercury (Hg) concentrations (ng/L) of water samples measured in August 2013 and 2014.
 168 (SB – south basin, SD – Selenga Delta, CB – central basin, MM – Maloe More Bay, NB – north basin, UAR – Upper Angara River)
 169 (Fig. 1).

Code	Sample collection year	Location	North	East	Hg concentration (ng/L)
BAIK13-8	2013	Lake Baikal (SB)	51°44'37.9"	105°18'52.4"	0.00
BAIK13-10	2013	Lake Baikal (SB/SD)	52°11'07.0"	106°05'38.0"	1.40
BAIK13-12	2013	Lake Baikal (CB)	52°47'53.3"	107°09'28.8"	1.40
BAIK13-14	2013	Lake Baikal (MM)	53°02'29.4"	106°56'29.5"	1.60
BAIK13-19	2013	Lake Baikal (NB/UAR)	55°38'57.8"	106°46'57.7"	3.20
SDB01	2014	Selenga Delta Branch	52°18'51.1"	106°44'23.3"	0.34
SDB02	2014	Selenga Delta Branch	52°17'11.1"	106°38'22.3"	1.92
SDB03	2014	Selenga Delta Branch	52°15'16.5"	106°38'07.3"	4.50
SDB04	2014	Selenga Delta Branch	52°09'14.8"	106°19'28.1"	5.50
SDB05	2014	Selenga Delta Branch	52°11'19.6"	106°29'37.5"	5.53
SLNG01	2014	Selenga Delta water body	52°15'23.0"	106°39'42.9"	6.48
SLNG03	2014	Selenga Delta water body	52°16'09.4"	106°42'10.1"	6.44
SLNG04	2014	Selenga Delta water body	52°15'52.5"	106°40'35.6"	7.93
SLNG05	2014	Selenga Delta water body	52°09'46.7"	106°19'59.6"	5.92
SLNG06	2014	Selenga Delta water body	52°13'49.0"	106°21'11.0"	6.18
SLNG07	2014	Selenga Delta water body	52°10'16.7"	106°22'06.6"	5.27
SLNG08	2014	Selenga Delta water body	52°14'12.8"	106°18'27.6"	5.78
SLNG09	2014	Selenga Delta water body	52°10'51.6"	106°21'56.8"	5.81
SLNG10	2014	Selenga Delta water body	52°12'38.3"	106°20'33.3"	6.05
SLNG11	2014	Selenga Delta water body	52°16'52.8"	106°38'26.6"	8.17
SLNG12	2014	Selenga Delta water body	52°17'17.3"	106°40'45.8"	6.00
SLNG13	2014	Selenga Delta water body	52°17'56.1"	106°37'27.7"	8.17
SLNG14	2014	Selenga Delta water body	52°18'32.9"	106°40'27.0"	10.14
SLNG15	2014	Selenga Delta water body	52°19'12.6"	106°38'26.6"	8.17
B13-8-11	2013	Selenga River	52°03'02.6"	106°40'20.0"	6.00
B13-8-20	2013	Selenga River	51°08'15.9"	107°29'21.7"	6.10
B13-8-26	2013	Selenga River	50°31'49.1"	106°16'16.7"	8.10
BRYT	2014	Selenga River Basin Lake	51°24'14.2"	106°29'25.5"	4.16

170

171 **2.2. Laboratory analysis of Hg concentrations in water samples**

172 0.25 mL concentrated HCl (Romil, pure grade) and 0.25 mL 0.1 N $\text{BrO}_3^-/\text{Br}^-$ (purified) was added
173 to each 45 mL water sample, which was then sealed for 30 minutes, had 15 $\mu\text{g/L}$ 12% NH_2OH -
174 HCl added, and diluted to 50 mL. Hg concentrations were analysed using Au trap cold vapour-
175 atomic fluorescence spectrometry (CV-AFS) following reduction with SnCl_2 (US EPA, 2002).
176 Detection limit is 0.4 ng/L; measurement errors for the Hg concentrations of less than 4 ng/L were
177 0.4 ng/g, and 10% for concentrations greater than 4 ng/L. Standard solutions and quality control
178 blanks were measured after every three samples to monitor measurement stability.

179 **2.3. Laboratory analysis of Hg concentrations in sediments**

180 Freeze-dried sediment samples were analysed at a temporal resolution of 5 – 20 years for BAIK13-
181 10A and BAIK13-19B. For SLNG04, samples were analysed through the core at a temporal
182 resolution of approximately 15 years. Hg analyses on sediment samples followed procedures in
183 Yang et al. (2010a). For each sample, approximately 0.2 g fine powdered freeze-dried sediment,
184 was digested with 8 mL of a 1:3 mixture of HNO_3 and HCl (aqua regia) at 100°C on a hotplate for
185 2 hours in rigorously acid-leached 50 mL Teflon digestion tubes. Following digestion, samples
186 were diluted to 50 mL with deionized water, capped and mixed. Digested solutions were then
187 analysed for Hg using cold vapour-atomic fluorescence spectrometry (CV-AFS), following
188 reduction with SnCl_2 . Standard reference material (GBW07305; certified Hg value of 100 ± 10.0
189 ng/g and measured mean value is 104 ng/g, with $\text{RSD} = 4.3 \text{ ng/g}$ ($n=3$)), and sample blanks were
190 digested with every 20 samples.

191 **2.4. Hg enrichment and total fluxes**

192 To examine trends in Hg loading over time, total Hg fluxes were calculated using the
193 radiometrically-derived sedimentation rates (Fig. S1). Standard enrichment factors could not be

194 calculated as lithogenic element data (for example Al, Li and Ti; Ribeiro et al., 2018) were not
195 available for the cores. Instead, Hg enrichment ratios (ER) were calculated by normalising Hg
196 concentrations in sediments deposited after 1850 CE, as determined from the age-depth model, by
197 the natural baseline (mean Hg concentrations prior to 1850 CE) (BAIK13-10A baseline mean =
198 30.4 ± 6.4 ng/g; BAIK13-19B = 35.5 ± 6.6 ng/g; SLNG04-C = 22.6 ± 1.2 ng/g). The calculated
199 ER therefore represent a comparative ratio of background vs post-1850 Hg concentrations (Yang
200 et al., 2010b). A baseline date of 1850 CE was chosen to take into account global atmospheric
201 contamination from industrialisation, despite the main regional development and expansion in the
202 Lake Baikal catchment region beginning in the 1900s (Brunello et al., 2004). An ER of > 1.4
203 demonstrates that post-1850 Hg concentrations are in exceedance of baseline by 2 SD, suggesting
204 post-1850 anthropogenic pollution. To examine trends in Hg loading, constrained cluster and
205 broken stick analyses were conducted on Hg concentration profiles from the three sediment cores,
206 to determine points of significant change, using the rioja package in R (version 3.5.2; R Core
207 Team, 2018) (Juggins, 2017).

208 **3. Results**

209 **3.1. Spatial distribution of Hg concentrations**

210 Water Hg concentrations ranged between 5.3 and 10.1 ng/L in the Selenga Delta shallow water
211 bodies and between 0.3 and 5.5 ng/L in the Selenga Delta branches with a decreasing trend from
212 the Selenga River to the mouth of the delta (Fig. 2 and 3). Along the Selenga River, Hg
213 concentrations ranged from 6.0 to 8.1 ng/L with highest values at the furthest upstream locations
214 near the town of Ust-Kyakhta (B13-8-26) (Fig. 1). Black Lake (BRYT), within the Selenga River
215 basin, had the lowest Hg concentration of the shallow lakes, at 4.2 ng/L (Fig. 2). In the waters of
216 Lake Baikal, Hg concentrations reached 3.2 ng/L at the one site (BAIK13-19) in the north basin,

217 near the Upper Angara and ranged from below the limit of detection to 1.6 ng/L in the south and
218 central basin lake waters. (Fig. 2), while near the Selenga Delta at BAIK13-10 the Hg
219 concentration was 1.6 ng/L.

220 **3.2. Historic trends of Hg contamination**

221 Hierarchical cluster analysis indicates that Hg concentrations at BAIK13-10 increase significantly
222 at c. 1840 CE. At BAIK13-19, Hg concentrations increase towards the top of the core, with
223 concentrations increasing significantly post-1920 CE and remaining elevated to the surface (Fig.
224 4). While only two samples comprise the post-1940s timeframe at BAIK13-19, they display similar
225 concentrations of 53 and 51 ng/g. Hg concentrations at SLNG04 showed a gradually increasing
226 trend beginning c. 1950 CE, with a significant increase in Hg concentration (c. 1960 CE) that
227 continue to increase until a maximum concentration of 56 ng/g is reached at c. 1990 CE.
228 Concentrations at SLNG04 then declined slightly post-1990 CE but have remained relatively
229 steady during the past two decades (Fig. 4).

230 Maximum and contemporary Hg concentrations show an approximate doubling of concentration
231 after 1945 CE across the sampled region, with recent concentrations close to 50 ng/g at all sites.
232 Sediments from BAIK13-10 show Hg enrichment, with ERs ranging between 1.6 and 1.7 between
233 1910 CE and 2013 CE (Fig. 4). Similarly, the BAIK13-19 sediment core from nearby the Upper
234 Angara River in the north basin show Hg enrichment in the upper sediments, with ER's ranging
235 between 1.2 and 1.5 from 1880 CE to 1960 CE (Fig. 4). Sediments from SLNG04 indicate little
236 enrichment of Hg (ER c. 1.0) until the mid-20th century when Hg enrichment quickly increased
237 and was consistently > 1.4 between c. 1960 CE and 2013 CE (Fig. 4). Hg enrichment peaks at
238 SLNG04 in c. 1990 CE with an ER of 2.3 has declined to 1.9 by 2013 CE.

239 Total fluxes of Hg show higher values post-1850 CE, compared to pre-1850 CE, in both the south
240 basin (BAIK13-10) and north basin (BAIK13-19) sediment cores from Lake Baikal. However,
241 post-1850 CE Hg flux was 20-fold greater in the south basin compared to the north basin sediment
242 core (Fig. 4). In BAIK13-10, Hg fluxes ranged from $0.26 \text{ ng cm}^{-2} \text{ yr}^{-1}$ in 1910 CE to 6.32 ng cm^{-2}
243 yr^{-1} in 2013 CE (Fig. 4), whereas in the north basin (BAIK13-19) a smaller range in Hg flux is
244 recorded in the sediments over the post-1850 CE period (from $0.38 \text{ ng cm}^{-2} \text{ yr}^{-1}$ in 1880 CE to 0.43
245 $\text{ng cm}^{-2} \text{ yr}^{-1}$ in 2013 CE (Fig. 4). Due to limitations of radiometric, SLNG04 Hg flux can only be
246 calculated from the mid-20th century, but fluxes show a distinct increase between c. 1945 CE and
247 c. 1995 CE, from 2.3 to $11.0 \text{ ng cm}^{-2} \text{ yr}^{-1}$. Since c. 1995 CE, Hg flux at SLNG04 has declined
248 slightly to $8.1 \text{ ng cm}^{-2} \text{ yr}^{-1}$ (Fig. 4).

249 **4. Discussion**

250 **4.1 Spatial patterns and modern Hg sources**

251 Mercury concentrations in surface waters span a spatial gradient from higher concentrations in the
252 upstream Selenga River to low concentrations in Lake Baikal. This pattern is expected due to the
253 mining activity along the Selenga River, and industrial activities in the cities of Ulan Ude and
254 Selenginsk (Fig. 1). There is a caveat regarding the interpretation of these single spot samples in
255 2013/2014, due to the uncertainty in whether these single measurements are representative of the
256 whole year and lake/drainage basin. With the exception of SLNG07, concentrations in the Selenga
257 Delta shallow lakes are consistently higher than in the Selenga Delta branches, and are higher than
258 concentrations found in Lake Baikal. Mercury concentrations are at their highest and most variable
259 in lakes on the east side of the Delta but are similar amongst lakes on the west side (Fig. 2). River
260 deltas are known hotspots for geochemical retention and transformations, which may be controlled

261 by seasonal and hydrological factors, including sediment load and flow (Lychagin et al 2015;
262 Chalov et al 2016). As most of the Hg in rivers is particle-bound, much of it will tend to deposit
263 in the smaller branches and shallow water bodies of the Selenga Delta, as flow decreases (Amos
264 et al., 2014). However, the fraction of the suspended particle load in rivers that is buried is highly
265 variable depending on freshwater discharge rates and the physical characteristics of different deltas
266 (Amos et al., 2014). Therefore, it is likely that the lakes of the Selenga Delta are acting as retention
267 ponds for Hg contamination within the Selenga River basin and preventing it from entering Lake
268 Baikal. The retention effect of the Selenga Delta is also apparent from the sedimentary records, as
269 Hg fluxes are 2-fold higher in the Selenga Delta sediment core (mean post-1850 = 6.47 ± 3.01
270 $\text{ng/cm}^2/\text{yr}$) compared to in the south basin (mean post-1850 = 2.85 ± 2.27 $\text{ng/cm}^2/\text{yr}$) and 18-fold
271 higher compared to in the north basin sediment core (mean post-1850 = 0.35 ± 0.09 $\text{ng/cm}^2/\text{yr}$) (Fig.
272 4). The higher sedimentary Hg fluxes in these Selenga Delta lakes compared to Lake Baikal is also
273 expected due to their closer proximity to the sources of Hg pollution within the Selenga River area.

274 Lake Baikal surface water Hg concentrations in August 2013 (mean 1.52 ± 1.14 ng/L) were higher
275 than previously published values of $0.14 - 0.77$ ng/L in June 1992 – 1993 (Meuleman et al., 1995;
276 Baeyens et al., 2002). The slightly elevated Hg concentration observed in the north basin at
277 BAIK13-19 (3.2 ng/L) also highlights the importance of the Upper Angara River as a source, with
278 contamination from industry in the north basin catchment. The largest town in this area is
279 Severobaykalsk, and the largest village settlement is Nizhneangarsk (Rose et al., 1998). The
280 Baikal-Amur rail-road also travels through this region. The main Hg sources in Severobaykalsk
281 are from fossil-fuel combustion facilities, waste incineration processes and chemical or electrical
282 industries. Alongside anthropogenic sources, another possible source of Hg into Lake Baikal is
283 from the hydrothermal vents at the bottom of the lake, which form as a result of the active tectonic

284 rift boundary (Crane et al., 1991; Kipfer et al., 1996). This geothermal activity mainly occurs in
285 the north basin of Lake Baikal and releases Hg into the sediments and water column via the
286 hydrothermal waters which are enriched in metals (Crane et al., 1991; Kipfer et al., 1996). Isotope
287 ratios of Hg can be used to distinguish between sources; however, it has been suggested that
288 hydrothermal discharge along fault lines at the bottom of Lake Baikal causes only a minor impact
289 on the lake water chemistry (Granina et al., 2007).

290 **4.2 Decadal-scale trends**

291 Sedimentary Hg concentrations in Lake Baikal and the Selenga Delta are comparable with
292 previous studies from Lake Baikal, which reported values between c. 40 – 70 ng/g over a 16 cm
293 sediment core depth, collected in 1990 CE (with no published sediment core chronology)
294 (Leermakers et al., 1996). Both modern water samples and sedimentary records from Lake Baikal
295 show that lakes in the Selenga Delta appear to be acting as Hg filters. In the sedimentary records
296 this filter effect is apparent as Hg enrichment levels in SLNG04 sediments reach over 2-fold
297 greater than baseline concentrations, which is a slightly higher range than BAIK13-10A sediments
298 in Lake Baikal close to the Selenga Delta system (Fig. 2 and 3). It is important to note, however,
299 that these enrichment levels are not dissimilar to those found in remote lakes in Uganda, North
300 America, Europe and Arctic Alaska, where Hg concentrations were up to 3-fold higher than those
301 in the pre-industrial period (Swain et al., 1992; Fitzgerald et al., 2005; Engstrom et al., 2007; Yang
302 et al, 2010a), which indicates that Hg loading at Lake Baikal is not greater than the global
303 background Hg enrichment levels. These enrichment levels in the above remote lakes (Swain et
304 al., 1992; Fitzgerald et al., 2005; Engstrom et al., 2007; Yang et al, 2010a) relate to atmospheric
305 deposition sources and not riverine draining industrial areas. Furthermore, it is interesting to
306 consider that Lake Baikal sediment records covering the last 6 million years show naturally

307 elevated Hg concentrations in the sediments during warmer climatic conditions (average Hg
308 concentrations of 46 ± 11 ng/g during warm periods and 27 ± 12 ng/g during cold periods), and
309 anomalously high peaks in Hg concentrations (between 210 – 420 ng/g) during volcanic events in
310 the Baikal area (Gelety et al., 2007). By comparison, Hg concentrations from BAIK13-10 and
311 BAIK13-19 are only slightly higher than the average Hg concentration during warmer periods and
312 fall within the range over the last 6 million years (Gelety et al., 2007).

313 In the north basin sediments, the Hg enrichment levels are lower (average post-1850 ER for
314 BAIK13-19 = 1.3 ± 0.16) than the south basin (average post-1850 ER for BAIK13-10 = 1.6 ± 0.05)
315 and Selenga Delta lake (average post-1850 ER for SLNG04 = 1.6 ± 0.42). ER results also suggest
316 an enrichment of north basin (BAIK13-19) sediments post 1940 CE, whereas the south basin
317 (BAIK13-10) site nearby the Selenga Delta experienced enrichment much earlier at around 1910
318 CE. Mercury enrichment of these sediments in the south basin in the early 1900s suggests the
319 contribution of contamination from local sources as a result of industrialisation in the Lake Baikal
320 catchment and the adjacent areas drained by the Angara and Lena rivers. In contrast to the south
321 basin, north basin sediments show a later onset in the mid-20th century of Hg enrichment, perhaps
322 because the major town on the north basin shores, Severobaykalsk, was only founded in the 1970's
323 and with the completion of the Baikal-Amur Mainline (BAM).

324 All three Lake Baikal cores indicate increases in Hg flux in Lake Baikal post-1850 CE, but the
325 subsurface peak in SLNG04 Hg flux indicates a rise in the sedimentation rate and a possible mid-
326 1990s peak in the delivery of Hg to the Selenga River/Lake Baikal system from both local and
327 long-range sources. Adams et al. (2018) recorded similar timing in decline of PAH, PCB, and DDT
328 fluxes to SLNG04, while Rose et al. (1998) recorded evidence of SCP concentration declines in
329 Lake Baikal sediments post-1990, likely indicating a regional decline in industrial coal and oil

330 combustion in southeast Siberia. The timing of this observed decline in anthropogenic
331 contamination in the Lake Baikal region ties in with the economic recession in the early 1990's
332 following the collapse of the former Soviet Union (Khanin, 2003; Adams et al. 2018). However,
333 the decline in Hg flux at SLNG04 is not large and remains elevated relative to pre- c. 1950 CE
334 levels. Differences in Hg flux between Lake Baikal and SLNG04 are also likely due to the high
335 affinity of Hg for organic matter; Hg binds to DOC and the Selenga Delta lakes receive a higher
336 input of catchment derived DOC than the pelagic regions of Lake Baikal (Yoshioka et al., 2002).
337 Other factors which affect Hg cycling in aquatic environments are temperature, redox conditions,
338 pH and microbial activity, which influence the Hg species transformation via methylation and
339 demethylation processes and biological uptake of MeHg (Hintelmann et al., 1995; French et al.,
340 2014). Thus, the higher input of DOC bound Hg into the Selenga Delta lakes could be a
341 contributing factor to the elevated levels of Hg enrichment seen in these lakes in comparison to
342 Lake Baikal. Alternatively, the Selenga Delta might be receiving greater impacts from local
343 sources than Lake Baikal, as a result of more sediments being deposited in the SLNG04 location,
344 and therefore SLNG04 is actually more highly contaminated by Hg inputs.

345 In summary, sedimentary profiles in the south basin of Lake Baikal are likely to be largely a
346 reflection of both local sources and long-range atmospheric deposition of Hg, however the
347 filtration effect of the Selenga Delta reduces the input of Hg pollution entering Lake Baikal from
348 the Selenga River. As Hg can remain within the atmosphere for up to a year, an important
349 anthropogenic source of Hg to Lake Baikal and its catchment area is likely to be atmospherically
350 transported Hg from industrial centres, from other urban areas in Russia and across the globe
351 (Gelety et al., 2007; UNEP Global mercury assessment, 2013). Air pollution controls and
352 mitigation efforts in North America and Europe have helped to reduce their Hg emissions from

353 industrial activity. However, in Asia (mainly China and India), Hg emissions have been rising
354 since the 1990s due to the marked economic expansion (Pacyna et al., 2016; Sundseth et al., 2017).
355 Thus since the late-1900's, long-range transport of Hg from elsewhere in Asia is likely to be
356 contributing to the enrichment at Lake Baikal; lake sediment cores from remote regions in China
357 show a marked increase in China's metal air pollution from 1990 CE (Wan et al., 2019) continuing
358 to present day (Yang et al., 2010b; UNEP Global mercury Assessment, 2013).

359 **4.3 Implications for Lake Baikal**

360 The 2013/2014 surveys of Hg concentrations across Baikal and the Selenga River basin area show
361 elevated levels of Hg in the Selenga River waters, in comparison to Lake Baikal waters, most likely
362 linked to gold mining and location of industrial centres (Brunello et al., 2004; Thorslund et al.,
363 2012; 2016; Brumbaugh et al., 2013; Chalov et al., 2015; Jarsjö et al., 2017). However, although
364 the Selenga Delta reduces the extent of Hg pollution entering the south and central basins of Lake
365 Baikal, the current state of the environment around Lake Baikal and its catchment gives cause for
366 concern with respect to future contamination by Hg.

367 For example, alongside present atmospheric deposition of Hg, it is also important to take into
368 consideration the effect of legacy Hg on the landscape. Re-emission of legacy Hg stores has
369 become another important source of Hg pollution to the landscape, which can be released via soil
370 erosion and permafrost thaw (Yang, 2015). Legacy Hg input into Lake Baikal and the Selenga
371 River basin is likely to increase with regional climate warming, as permafrost underlays a large
372 proportion of the catchment area. Modelling of current Hg reservoirs by Amos et al. (2013)
373 indicated that up to 60% of present-day atmospheric deposition of Hg is legacy-derived, re-emitted
374 from surface reservoirs. Moreover, recent increases in Hg concentrations within the lake sediments

375 could be impacted by permafrost thaw in the Lake Baikal basin (Hampton et al., 2008; Moore et
376 al., 2009) and catchment loading of Hg from the subsequent increased erosion of catchment soils
377 (Yang, 2015). In western Europe, changes to the climate system in recent years have also led to
378 increased storm events, causing further increased instability of catchment soils, increasing the
379 mobility of particulate-bound Hg across the terrestrial landscape (Yang and Smyntek, 2014). Thus,
380 Hg which has previously been deposited and stored within the lake catchment can also act as a
381 source of anthropogenic Hg to the lake system (Yang et al., 2002; Rose et al., 2012). Hg pollution
382 in Lake Baikal and the Selenga River basin area could therefore be a result of the continuing Hg
383 use in gold extraction processes in Mongolia, plus historical legacy of past Hg used in Russian
384 gold mining prior to 1950 CE, as well as long-range transport of atmospheric Hg from regional
385 and international industrial centres, from metal smelters, chemical and electrical industries, coal
386 combustion facilities and waste incineration plants.

387 Lake Baikal is increasingly facing pressures from shoreline anthropogenic nutrient pollution from
388 inadequate sewage treatment (Timoshkin et al., 2016), as well as pressures from recent
389 atmospheric warming since the 1950s which has been driving limnological and ecosystem changes
390 (Hampton et al., 2008; 2014; 2015; Moore et al., 2009; Izmet'eva et al., 2016; Silow et al., 2016;
391 Roberts et al., 2018). All these pressures combined put the Lake Baikal ecosystem at continued
392 risk from Hg inputs into the future. Efforts need to be focussed on minimising Hg pollution to
393 Lake Baikal and its catchment area, by eliminating the current use of Hg in the extraction process
394 of small-scale gold mining operations in Mongolia. Furthermore, global efforts, in accordance with
395 the new international treaty for the Minamata Convention need to continue, to reduce industrial
396 release of Hg emissions into the atmosphere. Mercury levels need to be monitored on the

397 freshwater ecosystems of the Selenga Delta itself, as it is an important Ramsar site for continental
398 Eurasia.

399 **5. Conclusions**

400 Mercury measurements from 2013/2014 suggest that the Selenga River is a major source of
401 anthropogenic Hg contamination into the Selenga Delta region and Lake Baikal, due to the
402 chemical (mainly the manufacturing of chlorine) and electrical plants where mercury is an element
403 in the manufacturing process, metallurgical plants which produce mercury directly,
404 hydrocarbon/coal or oil fired electric power plants and gold mining activity within the Mongolian
405 Selenga River basin. The low Hg concentrations within the lake waters could be attributed to the
406 retention effect of the Selenga Delta system and a result of dilution by large volume of Lake Baikal
407 and retention within the Selenga Delta. However, it is interesting that the highest concentrations
408 in the pelagic lake water are seen at a north basin site near the Upper Angara River. Sediments by
409 the Upper Angara River show Hg enrichment post-1940, but sediments near the Selenga Delta
410 showed evidence of enrichment above background levels much earlier, post-1850 CE. Hg
411 concentrations measured in the sediments are similar to measurements taken in the 1990's
412 (Leermakers et al., 1996) and over warm climatic periods (Gelety et al., 2007). Although there is
413 evidence of contamination in the Selenga Delta sediments (ERs over 1.4 in SLNG04), the pollution
414 levels are also modest. Thus, the current levels of Hg contamination within the Selenga River basin
415 highlight the necessity for the protection of Lake Baikal and the Selenga catchment, to reduce Hg
416 pollution of this unique aquatic ecosystem and the deterioration of a globally important freshwater
417 resource.

418 With rising unregulated mining activity along the Selenga River, it is vital to monitor Hg pollution
419 across the Baikal catchment, especially as MeHg has already been found to bioaccumulate within

420 Lake Baikal's pelagic foodweb (Ciesielski et al., 2016). Furthermore, recent and future climate
421 warming is likely to increase the transfer of different forms of Hg, such as Hg bound DOC across
422 the terrestrial landscape, from thawing permafrost and soil erosion (Rose et al., 2012) and greater
423 fluvial inflows into connected rivers. These climate driven processes will increase the Hg loading
424 within the Selenga River basin, and ultimately into pelagic Lake Baikal and its foodweb.

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434 **7. References**

435
436 **Adams, J.K., Martins, C.C., Rose, N.L., Shchetnikov, A.A., Mackay, A.W. (2018).** 'Lake
437 sediment records of persistent organic pollutants and polycyclic aromatic hydrocarbons in
438 Southern Siberia mirror the changing fortunes of the Russian economy over the past 70 years.'
439 *Environmental Pollution*, **242**, 528-538.

440
441 **Amos, H.M., Jacob, D.J., Streets, D.G., Sunderland, E.M. (2013).** 'Legacy impacts of all-time
442 anthropogenic emissions on the global mercury cycle.' *Global Biogeochemical Cycles*, **27**, 410-
443 421.

444

- 445 **Amos, H.M., Jacob, D.J., Kocman, D., Horowitz, H.M., Zhang, Y., Dutkiewicz, S., Horvat,**
446 **M., Corbitt, E.S., Krabbenhoft, D.P. (2014).** ‘Global biogeochemical implications of mercury
447 discharges from rivers and sediment burial.’ *Environment Science & Technology*, **48**, 9514-9522.
448
- 449 **Baeyens, W., Dehandschutter, B., Leermakers, M., Bobrov, V.A., Hus, R., Baeyens-Volant,**
450 **D. (2002).** ‘Natural mercury levels in geological enriched and geological active areas: case study
451 of Katun River and Lake Teletskoye, Altai (Siberia).’ *Water, Air and Soil Pollution*, **142**, 375 –
452 393.
453
- 454 **Bergan, T., Rodhe, H. (2001).** ‘Oxidation of elemental mercury in the atmosphere: constraints
455 imposed by global scale modelling.’ *Journal of Atmospheric Chemistry*, **40**, (2), 191-212.
456
- 457 **Brunello, A.J., Molotov, V.C., Dugherkhuu, B., Goldman, C., Khamaganova, E., Strijhova,**
458 **T., Sigman, R. (2004).** Lake Baikal: Lake Basin Management Initiative Experience and Lessons
459 Learned Brief. 1 – 26.
460
- 461 **Brumbaugh, W.G., Tillitt, D.E., May, T.W., Javzan, Ch., Komov, V.T. (2013).** ‘Environmental
462 survey in the Tuul and Orkhon River basins of north-central Mongolia, 2010: metals and other
463 elements in streambed sediment and floodplain soil.’ *Environmental Monitoring Assessment*, **185**,
464 8991 – 9008.
465
- 466 **Chalov, S.R., Jarsjö, J., Kasimov, N.S., Romanchenko, A.O., Pietron, J., Thorslund, J.,**
467 **Promakhova, E.V. (2015).** ‘Spatio-temporal variation of sediment transport in the Selenga River
468 Basin, Mongolia and Russia.’ *Environmental Earth Sciences*, **73**, (2), 663 – 680.
469
- 470 **Chalov, S., Thorslund, J., Kasimov, N., Aybullaev, D., Ilyicheva, E., Karthe, D., Kositsky,**
471 **A., Lychagin, M., Nittrouer, J., Pavlov, M., Pietron, J., Shinkareva, G., Tarasov, M.,**
472 **Garmaev, E., Akhtman, Y., Jarsjo, J. (2016).** ‘The Selenga River delta: a geochemical barrier
473 protecting Lake Baikal waters.’ *Regional Environmental Change*, **17**, (7), 2039 – 2053.
474

- 475 **Ciesielski, T.M., Pastukhov, M.V., Leeves, S.A., Farkas, J., Lierhagen, S., Poletaeva, V.I.,**
476 **Jenssen, B.M. (2016).** ‘Differential bioaccumulation of potentially toxic elements in benthic and
477 pelagic food chains in Lake Baikal.’ *Environmental Science and Pollution Research*, **23**, 15593 –
478 15604.
- 479
- 480 **Corbitt, E.S., Jacob, D.J., Holmes, C.D., Streets, D.G., Sunderland, E.M. (2011).** ‘Global
481 source-receptor relationships for mercury deposition under present day and 2050 emissions
482 scenarios.’ *Environmental Science & Technology*, **45**, (24), 10477-10484.
- 483
- 484 **Crane, K., Hecker, B., Golubev, V. (1991).** ‘Heat flow and hydrothermal vents in Lake Baikal,
485 U.S.S.R.’ *Eos*, **72**, (52), 588-589.
- 486
- 487 **Engstrom, D.R., Balogh, S.J., Swain, E.B. (2007).** ‘History of mercury inputs to Minnesota
488 lakes: Influences of watershed disturbance and localised atmospheric deposition.’ *Limnology &*
489 *Oceanography*, **52**, (6), 2467-2483.
- 490
- 491 **Fitzgerald, W.F., Engstrom, D.R., Lamborg, C.G., Tseng, C-M., Balcom, P.H.,**
492 **Hammerschmidt, C.R. (2005).** ‘Modern and historic atmospheric mercury fluxes in northern
493 Alaska: global sources and arctic depletion.’ *Environmental Science & Technology*, **39**, 557-568.
- 494
- 495 **French, T.D., Houben, A.J., Desforges, J-P.W., Kimpe, L.E., Kokelj, S.V., Poulain, A.J.,**
496 **Smol, J.P., Wang, X., Blais, J.M. (2014).** ‘Dissolved organic carbon thresholds affect mercury
497 bioaccumulation in arctic lakes.’ *Environmental Science & Technology*, **48**, 3162-3168.
- 498
- 499 **Granina, L.Z., Klerkx, J., Callender, E., Leermakers, M., Golobokova, L.P. (2007).** ‘Bottom
500 sediments and pore waters near a hydrothermal vent in Lake Baikal (Frolikha Bay).’ *Russian*
501 *Geology and Geophysics*, **48**, (3), 237-246.
- 502
- 503 **Gelety, V.F., Kalmykov, G.V., Parkhomenko, I.Y. (2007).** ‘Mercury in the sedimentary deposits
504 of Lake Baikal.’ *Geochemistry International*, **45**, (2), 170-177.

- 505
506 **Hammerschmidt, C.R., Fitzgerald, W., Lamborg, C.H., Balcom, P.H., Tseng, C-M. (2006).**
507 ‘Biogeochemical cycling of methylmercury in lakes and tundra watersheds of arctic Alaska.’
508 *Environmental Science & Technology*, **40**, 1204-1211.
509
- 510 **Hampton, S.E., Izmet’eva, L.R., Moore, M.V., Katz, S.L., Dennis, B., Silow, E.A. (2008).**
511 ‘Sixty years of environmental change in the world’s largest freshwater lake – Lake Baikal, Siberia.’
512 *Global Change Biology*, **14**, 1947 – 1958.
513
- 514 **Hampton, S.E., Gray, D.K., Izmet’eva, L.R., Moore, M.V., Ozersky, T. (2014).** ‘The Rise and
515 Fall of Plankton: Long-term changes in the vertical distribution of algae and grazers in Lake
516 Baikal, Siberia.’ *PLOS ONE*, **9**, (2), 1–10.
517
- 518 **Hampton, S.E., Moore, M.V., Ozersky, T., Stanley, E.H., Polashenski, C.M., Galloway,**
519 **A.W.E. (2015).** ‘Heating up a cold subject: prospects for under-ice plankton research in lakes.’
520 *Journal of plankton research*, **37**, (2), 277 - 284.
521
- 522 **Hampton, S.E., McGowan, S., Ozersky, T., Viridis, S.G.P., Vu, T.T., Spanbauer, T.L. et al**
523 **(2018).** ‘Recent ecological change in ancient lakes.’ *Limnology and Oceanography*, **63**, (5), 2277
524 – 2304.
525
- 526 **Hintelmann, H., Welbourn, P.M., Evans, R.D. (1995).** ‘Binding of methylmercury compounds
527 by humic and fluvic acids.’ *Water, Air and Soil Pollution*, **80**, 1031-1034.
528
- 529 **Izmet’eva, L.R., Moore, M.V., Hampton, S.E., Ferwerda, C.J., Gray, D.K., Woo, K.H., et**
530 **al. (2016).** ‘Lake-wide physical and biological trends associated with warming in Lake Baikal.’
531 *Journal of Great Lakes Research*, **42**, 6–17.
- 532 **Jarsjö, J., Chalov, S.R., Pietron, J., Alekseenko, A.V., Thorslund J. (2017).** ‘Patterns of soil
533 contamination, erosion, and river loading of metals in a gold mining region of northern
534 Mongolia.’ *Regional Environmental Change*, **17**, 1991-2005.

- 535 **Juggins, S. (2017).** rioja: Analysis of Quaternary Science Data, R package version (0.9-15.1).
536 (<http://cran.r-project.org/package=rioja>).
- 537 **Kasimov, N., Karthe, D., Chalov, S. (2017).** ‘Environmental change in the Selenga River – Lake
538 Baikal Basin.’ *Regional Environmental Change*, **17**, 1945 – 1949.
- 539
- 540 **Kaus, A., Schäffer, M., Karthe, D., Büttner, O., Tümpling, W., Borchardt, D. (2017).**
541 ‘Regional patterns of heavy metal exposure and contamination in the fish fauna of the Kharaa
542 River basin (Mongolia).’ *Regional Environmental Change*, **17**, 2023 – 2037.
- 543
- 544 **Kelly, C.A., Rudd, J.W.M., Holoka, M.H. (2003).** ‘Effect of pH on mercury uptake by an aquatic
545 bacterium: implications for Hg cycling.’ *Environmental Science Technology*, **37**, 2941-2946.
- 546
- 547 **Khanin, G.I. (2003).** ‘The 1950s – the triumph of the Soviet economy.’ *Europe-Asia Studies* **55**,
548 1187-573 1212.
- 549
- 550 **Kipfer, R., Aeschbach-Hertig, W., Hofer, M., Hohmann, R., Imboden, D.M., Baur, H.,
551 Golubev, V., Klerkx, J. (1996).** ‘Bottom water formation due to hydrothermal activity in Frolikha
552 Bay, Lake Baikal, eastern Siberia.’ *Geochimica et Cosmochimica Acta*, **60**, (6), 961-971.
- 553
- 554 **Koval, P.V., Kalmychkov, G.V., Gelety, V.F., Leonova, G.A., Medvedev, V.I., Andrulaitis,
555 L.D. (1999).** ‘Correlation of natural and technogenic mercury sources in the Baikal polygon,
556 Russia.’ *Journal of Geochemical Exploration*, **66**, 277 – 289.
- 557
- 558 **Leermakers, M., Meuleman, C., Baeyens, W. (1996).** ‘Mercury distribution and fluxes in Lake
559 Baikal.’ In: Global and Regional mercury cycles: sources, fluxes and mass balances. Eds Baeyens
560 et al. Kluwer Academic Publishers. 303 – 315.
- 561
- 562 **Logachev, N.A. (2003).** ‘History and geodynamics of the Baikal rift.’ *Russian Geology and
563 Geophysics*, **44**, (5), 391 – 406.
- 564

- 565 **Lychagin, M.Y., Tkachenko, A.N., Kasimov, N.S., Kroonenberg, S. (2015).** ‘Heavy metals in
566 the water, plants, and bottom sediments of the Volga River mouth area.’ *Journal of Coastal*
567 *Research* **31**, (4), 859–868.
- 568
- 569 **Nadmitov, B., Hong, S., Kang, S.I., Chu, J.M., Gomboev, B., Janchivdorj, L., Lee, C-H.,**
570 **Khim, J, S. (2015).** ‘Large-scale monitoring and assessment of metal contamination in surface
571 water of the Selenga River Basin (2007 – 2009).’ *Environmental Science and Pollution Research*,
572 **22**, 2856 – 2867.
- 573
- 574 **Nikanorov, A.M., Reznikov, S.A., Mateev, A.A., Arakelyan, V.S. (2012).** ‘Monitoring of
575 Polycyclic Aromatic Hydrocarbons in the Lake Baikal basin in the areas of intensive
576 anthropogenic impact.’ *Russian Meteorology and Hydrology*, **37**, (7), 477 – 484.
- 577
- 578 **Nomokonova, E., Lin, S-C., Chen, G. (2013).** ‘Investigation of safety compliance and safety
579 participation as well as cultural influences: Using Selenginsk Pulp and Cardboard Mill in Russia
580 as an example.’ Proceedings of the Institute of Industrial Engineers Asian Conference Eds. Yi-
581 Kuei Lin, Yu-Chung Tsao, Shi-Woei Lin. Pp. 1001 – 1007.
- 582
- 583 **Maruev, V.A. (2018).** History of the gold-mining industry in the barring in the XIX-beginning of
584 the XX century. Transbaikal State University, PhD thesis, 223 pp (in Russian).
- 585
- 586 **Meuleman, C., Leermakers, M., Baeyens, W. (1995).** ‘Mercury speciation in Lake Baikal.’
587 *Water, Air and Soil Pollution*, **80**, 539 – 551.
- 588
- 589 **Misyurkeeva, Yu. (2009).** The barbarous "gold rush" in Mongolia poisons Selenga. *Novaya*
590 *Buryatiya* 39, 12-14. (in Russian).
- 591
- 592 **Moore, M.V., Hampton, S.E., Izmet’eva, L.R., Silow, E.A., Peshkova, E.V., Pavlov, B.K.**
593 **(2009).** ‘Climate change and the world’s “sacred sea” – Lake Baikal, Siberia.’ *Bioscience*, **59**, (5),
594 405 – 417.
- 595

- 596 **Ozersky, T., Pastukhov, M.V., Poste, A.E., Deng, X.Y., Moore, M.V. (2017).** ‘Long-term and
597 ontogenetic patterns of heavy metal contamination in Lake Baikal seals (*Pusa sibirica*).’
598 *Environmental Science & Technology*, **51**, 10316 – 10325.
599
- 600 **Pacyna, J.M., Travnikov, O., De Simone, F., Hedgecock, I.M., Sundseth, K., Pacyna, E.G.,**
601 **Steenhuisen, F., Pirrone, N., Munthe, J., Kindbom, K. (2016).** ‘Current and future levels of
602 mercury atmospheric pollution on a global scale.’ *Atmospheric Chemistry and Physics*, **16**, 12495
603 – 12511.
604
- 605 **Perrot, V., Epov, V.N., Pastukhov, M., Grebenshchikova, V., Zouiten, C., Sonke, J.E.,**
606 **Husted, S., Donard, O.F.X., Amouroux, D. (2010).** ‘Tracing sources and bioaccumulation of
607 mercury in fish of Lake Baikal – Angara River using Hg isotopic composition.’ *Environmental*
608 *Science & Technology*, **44**, (21), 8030 – 8037.
609
- 610 **Perrot, V., Pastukhov, M.V., Epov, V.N., Husted, S., Donard, O.F.X., Amouroux, D. (2012).**
611 ‘Higher mass-independent isotope fractionation of methylmercury in the pelagic food web of Lake
612 Baikal (Russia).’ *Environmental Science & Technology*, **46**, 5902-5911.
613
- 614 **Pietron, J., Chalov, S.R., Chalova, A.S., Alekseenko, A.V., Jarsjö, J. (2017).** ‘Extreme spatial
615 variability in riverine sediment load inputs due to soil loss in surface mining areas of the Lake
616 Baikal basin.’ *Catena*, **152**, 82 – 93.
- 617 **Pisarsky, B.I., Hardina, A.M., Naganawa, H. (2005).** ‘Ecosystem evolution of Lake Gusinoe
618 (Transbaikal region, Russia).’ *Limnology*, **6**, 173 – 182.
619
- 620 **Ribeiro, C., Couto, C., Ribeiro, A.R., Maia, A.S., Santos, M., Tiritan, M.E., Pinto, E.,**
621 **Almeida, A.A. (2018).** ‘Distribution and environmental assessment of trace elements
622 contamination of water, sediments and flora from Douro River estuary, Portugal.’ *Science of the*
623 *Total Environment*, **639**, 1382-1393.
624

- 625 **Roberts, S.L., Swann, G.E.A., McGowan, S., Panizzo, V.N., Vologina, E.G., Sturm, M.,**
626 **Mackay, A.W. (2018).** ‘Diatom evidence of 20th century ecosystem change in Lake Baikal,
627 Siberia.’ *PLOS ONE*, **13**, (12), 1-20.
- 628
- 629 **Rose, N.L., Appleby, P.G., Boyle, J.F., Mackay, A.W., Flower, R.J. (1998).** ‘The spatial and
630 temporal distribution of fossil-fuel derived pollutants in the sediment record of Lake Baikal,
631 eastern Siberia.’ *Journal of Paleolimnology*, **20**, 151 – 162.
- 632
- 633 **Rose, N.L., Yang, H., Turner, S.D., Simpson, G.L. (2012).** ‘An assessment of the mechanisms
634 for the transfer of lead and mercury from atmospherically contaminated organic soils to lake
635 sediments with particular reference to Scotland, UK.’ *Geochimica et Cosmochimica Acta*, **82**, 113-
636 135.
- 637
- 638 **Scholz, C.A., Hutchinson, D.R. (2000).** ‘Stratigraphic and structural evolution of the Selenga
639 Delta Accommodation Zone, Lake Baikal Rift, Siberia.’ *International Journal of Earth Sciences*,
640 **89**, 212 – 228.
- 641
- 642 **Silow, E.A., Krashchuk, L.S., Onuchin, K.A., Pislegina, H.V., Rusanovskaya, O.O.,**
643 **Shimaraeva, S.V. (2016).** ‘Some recent trends regarding Lake Baikal phytoplankton and
644 zooplankton.’ *Lakes and Reservoirs and Management*, **21**, 40-44.
- 645
- 646 **Sundseth, K., Pacyna, J.M., Pacyna, E.G., Pirrone, N., Thorne, R.J. (2017).** ‘Global sources
647 and pathways of mercury in the context of human health.’ *International Journal of Environmental*
648 *Research and Public Health*, **14**, 105, 1-14.
- 649 **Swain, E.B., Engstrom, D.R., Brigham, M.E., Henning, T.A., Brezonik, P.L. (1992).**
650 ‘Increasing rates of atmospheric mercury deposition in midcontinental north America.’ *Science*,
651 **257**, 784-787.
- 652 **Thorslund, J., Jarsjö, J., Chalov, S.R., Belozerova, E.V. (2012).** ‘Gold mining impact on
653 riverine heavy metal transport in a sparsely monitored region: the upper Lake Baikal Basin case.’
654 *Journal of Environmental Monitoring*, **14**, (10), 2780 – 2792.

- 655
- 656 **Thorslund, J., Jarsjö, J., Wällstedt, T., Mörth, C.M., Lychagin, M.Y., Chalov, S.R. (2016).**
- 657 ‘Speciation and hydrological transport of metals in non-acidic river systems of the Lake Baikal
- 658 basin: Field data and model predictions.’ *Regional Environmental Change*, **17**, (7), 2007 – 2021.
- 659
- 660 **Timoshkin, O.A., Samsonov, D.P., Yamamuro, M., Moore, M.V., Belykh, O.I., Malnik, V.V.**
- 661 **et al. (2016).** ‘Rapid ecological change in the coastal zone of Lake Baikal (East Siberia): Is the
- 662 site of the world’s greatest freshwater biodiversity in danger?’ *Journal of Great Lakes Research*,
- 663 **42**, (3), 487-497.
- 664
- 665 **Tumenbayer, B., Batbaya, M., Grayson, R. (2000).** ‘Environmental hazard in Lake Baikal
- 666 watershed posed by mercury placer in Mongolia.’ *World Placer Journal*, **1**, 134 – 159.
- 667
- 668 **United Nations Environmental Programme (UNEP) Global Mercury Assessment. (2013).**
- 669 ‘Global Mercury Assessment 2013 Sources, Emissions, Releases and Environmental Transport.
- 670 UNEP Chemicals Branch, Geneva, Switzerland.
- 671
- 672 **United Nations (UN) Environment. (2017).** ‘Minamata Convention on Mercury.’
- 673 www.mercuryconvention.org
- 674
- 675 **United States Environmental Protection Agency. (2002).** Method 1631, Revision E: Mercury
- 676 in water by oxidation, purge and trap, and cold vapor atomic fluorescence spectrometry.
- 677 Washington, D.C, United States of America.
- 678
- 679 **Vasiliev, O.F., Obolenskiy, A.A., Yagolnitsler, M.A. (1998).** ‘Mercury as a pollutant in Siberia:
- 680 sources, fluxes and a regional budget.’ *The Science of the Total Environment*, **213**, 73-84.
- 681
- 682 **Wan, D., Song, L., Mao, X., Yang, J., Jin, Z. (2019).** ‘One-century sediment records of heavy
- 683 metal pollution on the southeast Mongolian Plateau: Implications for air pollution trend in China.’
- 684 *Chemosphere*, **220**, 539-545.
- 685

- 686 **Yang, H., Rose, N.L., Battarbee, R.W. (2002).** ‘Mercury and lead budgets for Lochnagar, a
687 Scottish Mountain Lake and its catchment.’ *Environment, Science & Technology*, **36**, 1383-1388.
688
- 689 **Yang, H., Engstrom, D.R., Rose, N.L. (2010a).** ‘Recent changes in atmospheric mercury
690 deposition recorded in the sediments of remote equatorial lakes in the Rwenzori mountains,
691 Uganda.’ *Environmental Science & Technology*, **44**, 6570 – 6575.
692
- 693 **Yang, H., Battarbee, R.W., Turner, S., Rose, N.L., Derwent, R.G., Wu, G., Yang, R. (2010b).**
694 ‘Historic reconstruction of mercury pollution across the Tibetan Plateau using lake sediments.’
695 *Environmental Science & Technology*, **44**, 2918-2924.
- 696• **Yang, H., Smyntek, P. (2014).** ‘The mercury record in Red Tarn sediments reveals air pollution
697 history and implications of catchment erosion.’ *Environment Science: Processes and Impacts*, **16**,
698 (11).
699
- 700 **Yang, H. (2015).** ‘Lake sediments may not faithfully record decline of atmospheric pollutant
701 deposition.’ *Environmental Science & Technology*, **49**, (21), 12607 – 12608.
702
- 703 **Yoshioka, T., Ueda, S., Khodzher, T., Bashenkhaeva, N., Korovyakova, I., Sorokovikova, L.,
704 Gorbunova, L. (2002).** ‘Distribution of dissolved organic carbon in Lake Baikal and its
705 watershed.’ *Limnology*, **3**, 159-168.
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738 **Fig 1. Location of study sites and other key locations referred to in the text across Lake**
739 **Baikal and the Selenga River catchment.**

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770 **Fig 2. Mercury (Hg) concentrations (ng/L) in water samples collected from the Selenga River,**
771 **Selenga Delta and Lake Baikal.**

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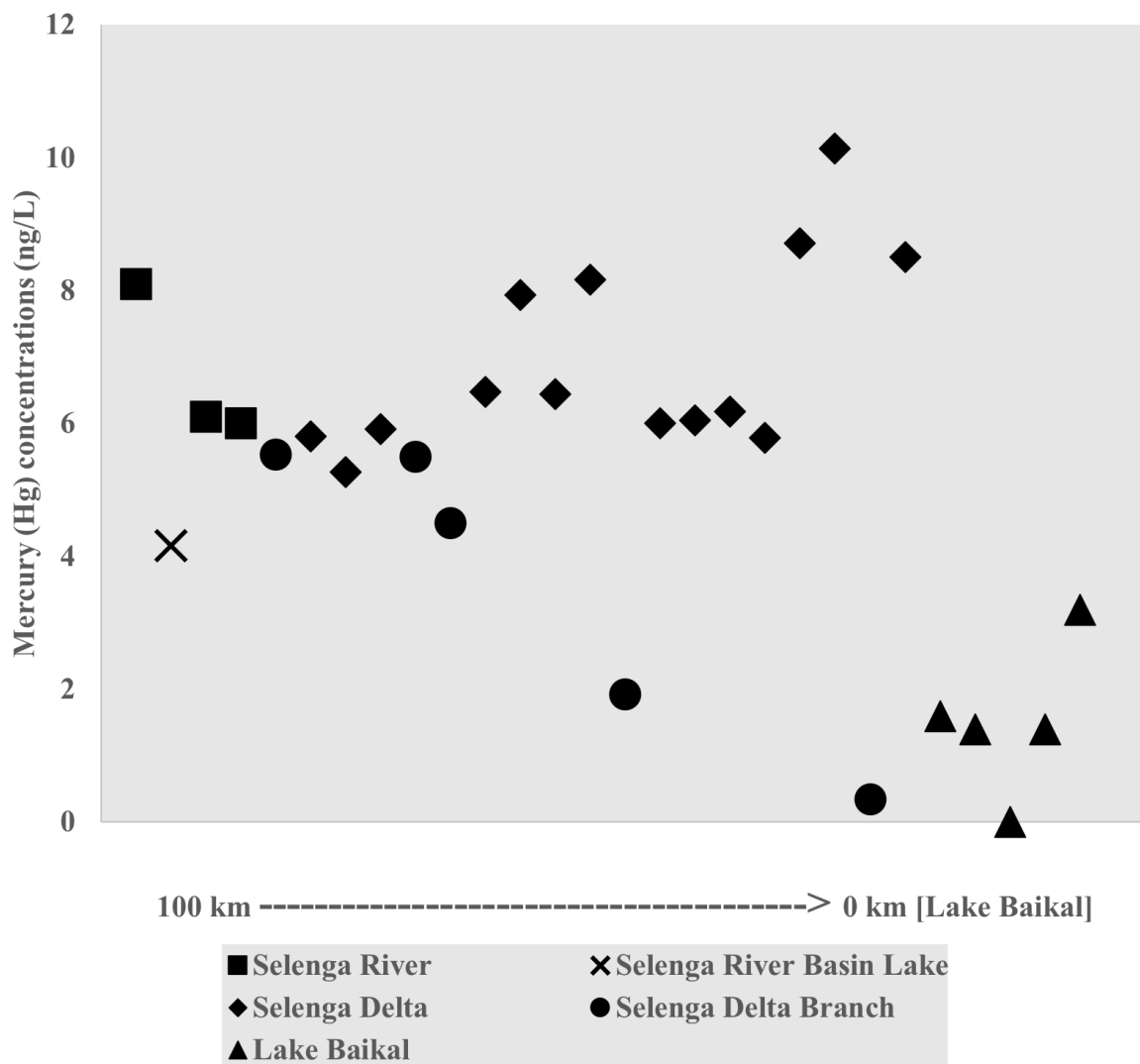
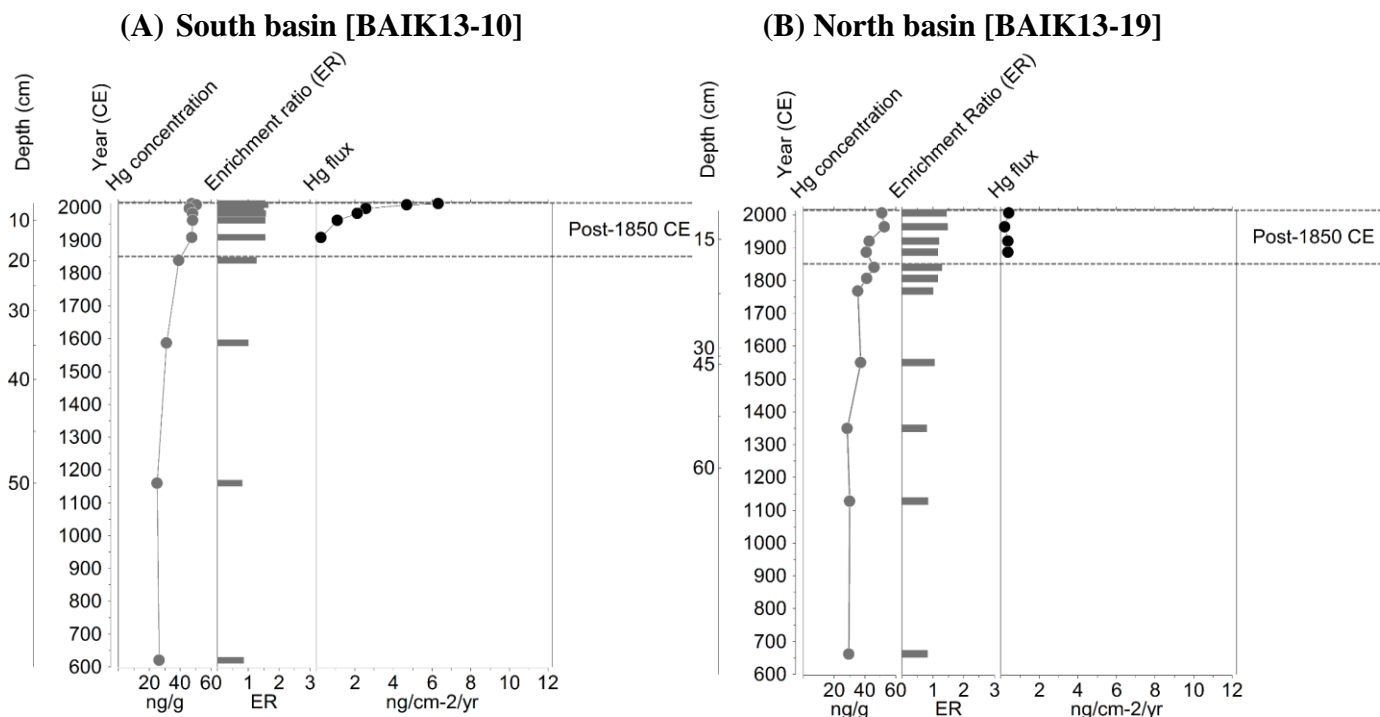


Fig 3. Mercury (Hg) concentrations (ng/L) in water samples collected at each site (grouped by Selenga River, Selenga River Basin Lake, Selenga Delta, Selenga Delta Branch and Lake Baikal) plotted along a distance gradient from the Selenga River to Lake Baikal.

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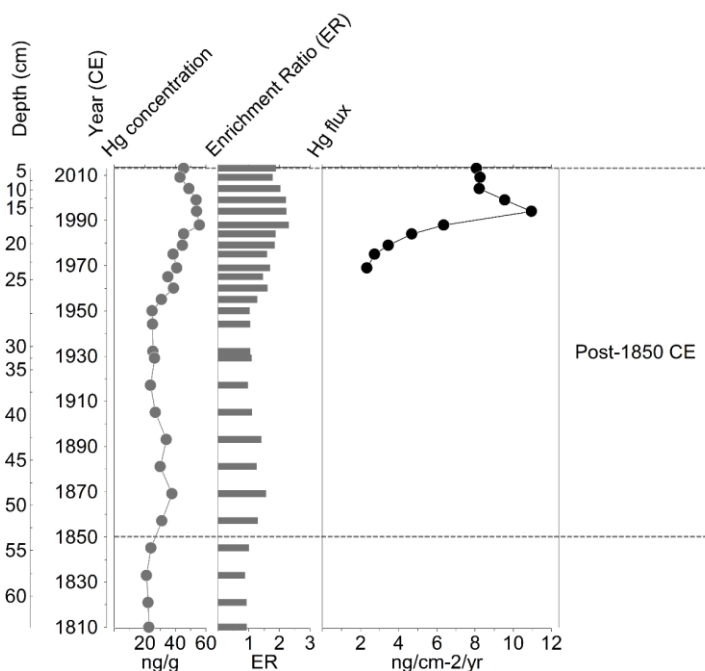
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(C) Selenga Delta [SLNG04]



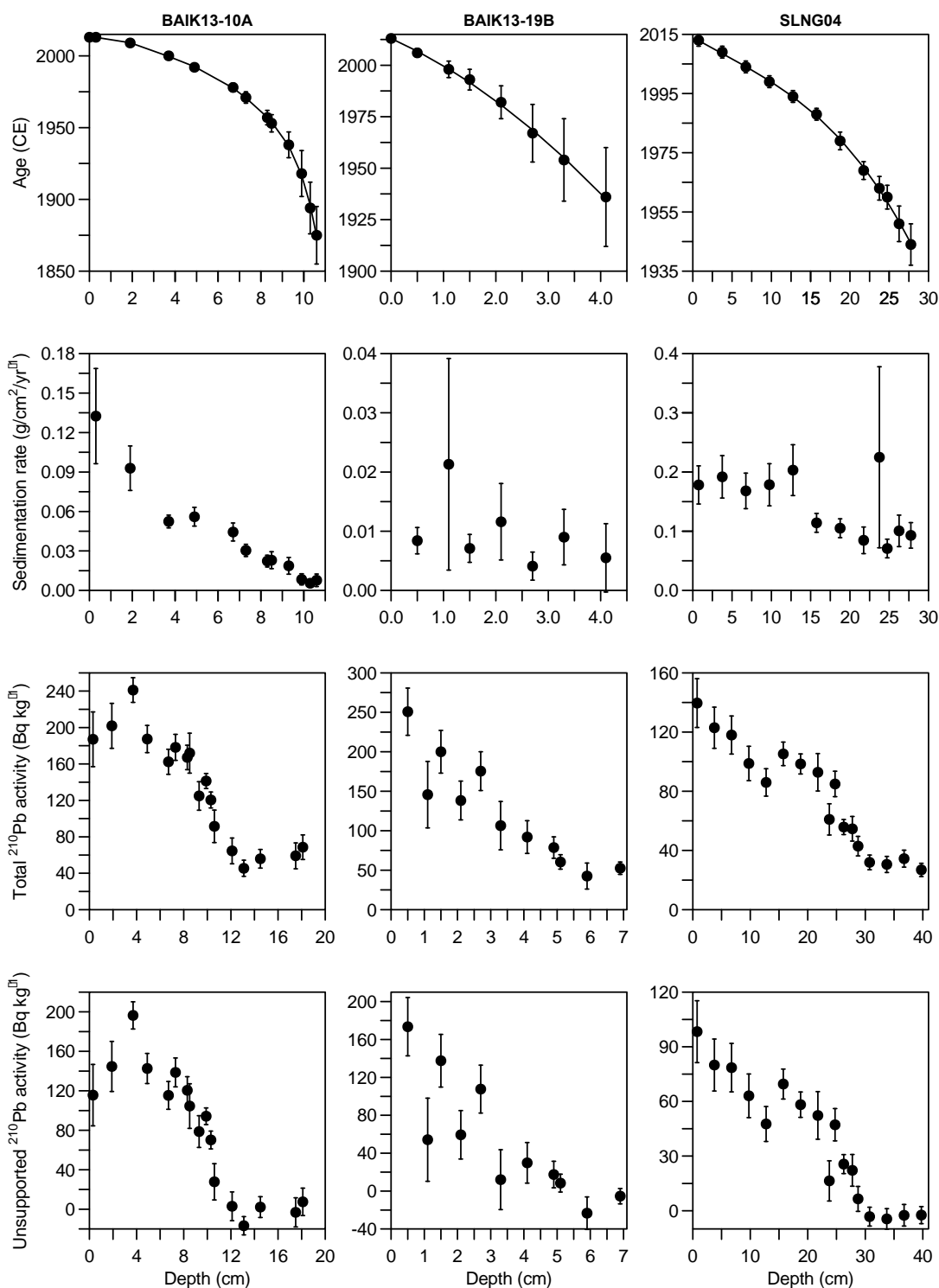
818 **Fig 4. Sedimentary mercury (Hg) concentrations (ng/g), enrichment ratios (ER) and Hg**

819 **fluxes (ng/cm²/yr) profiles from the (A) south basin [BAIK13-10], (B) north basin [BAIK13-**

820 **19] in Lake Baikal and (C) Selenga Delta [SLNG04]. For SLNG04 all the dates beyond c.**

821 **1945 are extrapolations of constant background sedimentation rates pre-1980.**

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850 **Fig S1. ²¹⁰Pb chronology and sedimentation rates for BAIK13-10A (south basin), BAIK13-**
851 **19B (north basin) (Roberts et al., 2018) and SLNG04 (Selenga Delta) (Adams et al., 2018)**
852 **sediment cores.**