1	Nitrogen Availability and Denitrification in Urban Agriculture and Regreened Vacant Lots
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Abstract

35	Many cities demolish abandoned homes and create regreened vacant lots (RVLs), and an
36	increasingly popular, high-intensity use of RVLs is as urban agriculture (UA) sites. UA may
37	potentially result in higher nitrogen (N) runoff to aquatic ecosystems, but this potential has not
38	been quantified. We examined the role that varying land reuse intensity plays in determining
39	potential for N export via runoff or leaching, focusing on soil N availability and N removal
40	capacity via denitrification. We contrasted three levels of land use intensity for vacant parcels:
41	intact vacant properties, turfgrass RVLs, and regreened UA lots in Buffalo, NY. We examined
42	soil N and C availability, denitrification potential, and isotopic evidence of denitrification.
43	Land use intensity only affected soil properties in surficial soil horizons. Total N was
44	2.5x higher in UA soils (mean = 0.51%) than non-UA (mean = 0.21%). Soil nitrate was 2.6x
45	higher in winter (mean = 12.4 μ g NO ₃ ⁻ -N g ⁻¹) than summer (mean = 4.7 μ g NO ₃ ⁻ -N g ⁻¹) and was
46	generally higher in UA soils. Despite higher soil N availability at UA sites, there were no
47	differences in denitrification potential between UA and non-UA sites (mean = $620 \text{ ng } N_2\text{O-N } \text{g}^{-1}$
48	h ⁻¹). Isotopic evidence further confirms that denitrification was not a major sink of N. Although
49	UA had high N availability compared to non-UA sites and low rates of denitrification, UA only
50	has moderate potential for runoff-driven N export, as nitrate concentrations were substantially
51	lower than values typical for conventional agricultural soils.

52

53 Keywords: urban agriculture, vacant lots, green infrastructure, nitrogen, nitrate, denitrification

Introduction

55 Urban environments are hotspots of nitrogen (N) and phosphorus export to aquatic 56 ecosystems, where pollution by these important macronutrients can drive eutrophication and 57 harmful algal blooms (Grogan et al., 2023). Nutrient export to aquatic systems from the urban 58 environment is primarily driven by runoff during precipitation events that results from a large 59 percentage of impervious surfaces (Brezonik & Stadelmann, 2002). In cities with combined 60 sewer systems, stormwater combines with sewage during high flows and overwhelms the 61 capacity of wastewater treatment plants, with excess untreated water containing high nutrient 62 loads routed to waterways (Field & Struzeski, 1972). While sewage from these combined sewer 63 overflow (CSO) events can be an important component of overall nutrient loads to receiving 64 aquatic ecosystems, nutrients mobilized by runoff from green spaces can also contribute 65 substantial amounts to these nutrient loads. For example, Hobbie et al. (2017) found that, in St. 66 Paul, MN, residential fertilizer comprised 37-59% of N supplied to a watershed, and that 67 stormwater runoff, exclusive of sewage inputs, comprised 37-79% of total N export from the 68 watershed. Similarly, in a study of a high-to-medium housing density area in Florida, fertilizer 69 runoff contributed 6-25% of N export from the watershed (Yang & Toor, 2017). Runoff N from 70 lawns is likely to be highest when the flow path to the nearest impervious surface is short, such 71 as in dense urban areas (Suchy et al., 2021).

To reduce stormwater pollution from urban ecosystems, cities have implemented management plans that increasingly focus on engineered green infrastructure (such as rain gardens) as a pathway to mitigation of runoff volumes and resulting nutrient pollution (Hopkins et al., 2018). Many Rust Belt cities such as Detroit, Michigan and Buffalo, New York have experienced population decline and widespread abandonment and subsequent decay of properties

77	(Hackworth, 2016; Silverman et al., 2013). To combat this, many of these cities have engaged in
78	programs of active demolition and regreening of abandoned properties. The resulting regreened
79	vacant lots (RVLs), have been recognized over the past decade for their usefulness as a low
80	intensity but extensive form of green infrastructure (Kelleher et al., 2020; Newman et al., 2019;
81	Shuster et al., 2014, 2015). Demolition of abandoned structures decreases impervious surface
82	cover of RVLs, and planting of vegetation potentially increases evapotranspiration and improves
83	soil hydraulic properties (Qiu et al., 2013, 2017), in turn reducing storm flows in sewer systems
84	and mitigating CSOs (Kelleher et al., 2020). For example, in Buffalo, NY, infiltration capacity of
85	RVLs was increased by greater than 50% compared to pre-demolition state (Kelleher et al.,
86	2020). Buffalo includes RVLs, produced as a result of its urban blight management program, in
87	its EPA-mandated CSO control plan (Buffalo Sewer Authority, 2014). Despite their recognized
88	hydrologic benefits, little attention has been paid to whether RVLs act as sinks of nutrients, as in
89	the case of other, engineered green infrastructure, or whether they can produce nutrient pollution
90	in surface runoff, partially counteracting their hydrologic benefits.
91	RVLs have documented costs and benefits, the balance of which largely depends on how
92	they are managed. RVLs targeted for high-intensity reuse, such as debris removal and replanting,
93	are associated with many community health outcomes including decreased levels of violence,
94	decreased mortality, and increased feelings of safety (Garvin et al., 2013; Kondo et al., 2018)
95	while unmanaged RVLs, in addition to being hotspots for crime, can increase the incidence of
96	zoonotic diseases (Gulachenski et al., 2016). A common, high-intensity use of RVLs is urban
97	agriculture (UA). UA has been the focus of increasing interest in recent decades (Heckler, 2012;
98	Kleszcz, 2018) and can provide nutrition benefits to low income individuals in inner-city
99	communities (Gudzune et al., 2015) as well as fostering community engagement (Ohmer et al.,

100	2009). As UA is often associated with intensive organic agriculture, UA also has the benefit of
101	not contributing new inorganic N to the global N cycle, sourcing N instead from manure,
102	compost or from N already present in the urban environment (Erisman et al., 2008). RVLs used
103	for UA are characterized by a high degree of soil amendments, with nutrient rich materials such
104	as compost or manure imported to increase levels of N and P (Taylor & Lovell, 2015;
105	Wielemaker et al., 2019). These high levels of nutrient inputs to UA suggest high potential for
106	export of these nutrients in runoff, and this potential needs to be quantified to properly weigh the
107	costs and benefits of UA as a potential use for RVLs (Meharg, 2016; Sieczko et al., 2023).
108	Denitrification can remove N from ecosystems by converting nitrate (NO3 ⁻) to inert N2
109	gas, a highly desirable outcome from a nutrient runoff perspective. Denitrification is an
110	anaerobic microbial process that requires anoxic environments and that uses labile organic
111	carbon as a substrate to reduce NO3 ⁻ and obtain energy. Denitrification rates are negatively
112	correlated with nitrate runoff in urban lawns (Suchy et al., 2023). Although denitrification
113	requires anoxic environments, substantial denitrification occurs in oxic soils in microsites where
114	oxygen has been depleted (e.g. soil aggregate interiors), especially in fine-textured soils (Tiedje
115	et al., 1984). Researchers frequently measure potential denitrification in upland agricultural
116	environments where typical rates range from 27 ng N ₂ O-N $g^{-1} h^{-1}$ to 932 ng N ₂ O-N $g^{-1} h^{-1}$
117	(D'Haene et al., 2003; Wang et al., 2022). Potential denitrification rates have been measured in
118	urban lawn environments as well, and in a meta-analysis rates ranged widely, from <1 ng N2O-N
119	$g^{-1} h^{-1}$ to >62000 ng N ₂ O-N $g^{-1} h^{-1}$ (Reisinger et al., 2016), though other studies show a much
120	narrower range of rates, comparable to agricultural soils (Suchy et al., 2021).
121	The contribution of RVLs to N export from urban watersheds is a function of the balance

122 of N inputs to these systems and the capacity for these systems to retain and ultimately remove

123 these nutrients. The quantities of inputs are in turn likely to be driven by reuse intensity of RVLs 124 (e.g. unmaintained lawn vs. urban agriculture) and the associated management practices that are 125 employed. RVL reuse intensity is also likely to change the rates and occurrence of denitrification 126 by modifying substrate supply and the favorability of environmental conditions. For example, the 127 soils of RVLs have been altered as a byproduct of demolition and regrading, resulting in 128 compaction that can result in more anaerobic conditions favorable to denitrifiers (Li et al., 2014). 129 Variable demolition procedures also contribute greatly to the high variability of soil hydraulic 130 properties that have been observed in RVLs in Buffalo (Kelleher et al., 2020). Finally RVLs that 131 have UA are frequently tilled and amended with compost and manure, altering soil structure and 132 greatly increasing organic carbon and nitrogen available to microbes for denitrification 133 (Wielemaker et al., 2019).

134 This study aims to address how vacant lot regreening may impact the N cycle in these 135 urban ecosystems, with the objective of understanding the potential for these ecosystems to be a 136 source or sink of N that can result in pollution of urban aquatic ecosystems. Specifically, we ask: 137 How do various land reuses of vacant lots determine a) N content available for export and b) N 138 removal via denitrification? We hypothesized that, compared to undemolished vacant properties, 139 soil N content and denitrification activity would be higher in urban agriculture sites and lower in 140 regreened turfgrass. We reasoned that urban agriculture receives soil amendments from farmers 141 and so can be expected to have higher N content including NO₃⁻ as well as labile organic carbon. 142 By contrast, turfgrass sites have experienced soil disturbances such as burial and mixing which 143 could be expected to decrease nutrient content, and also generally receive no applied fertilizer. 144 We examined sites in Buffalo, NY, a city with a high amount of property vacancy and RVLs, to 145 test the hypothesis.

Methods

147 Site Description

148 Buffalo, New York (42°53.187' N, 78°52.7022' W) is located in the far west of the state, 149 on the eastern shore of Lake Erie. The climate is humid continental (Köppen classification: 150 Dfb/Dfa) with mean annual precipitation of 640 mm and mean annual temperature of 9 °C. The 151 natural vegetation of the area consists of temperate hardwood forest, although most land within 152 the region is in agriculture or developed land uses (Multi-Resolution Land Characteristics 153 Consortium (U.S.), 2019). Soils consist predominantly of urban anthrosols developed from 154 glacial till and glaciolacustrine sediments (Cline & Marshall, 1977). As a result of city-led 155 demolitions from 2001-2013, there have been 404 acres (163.5 ha) of impervious surfaces 156 converted to RVLs across the city (Buffalo Sewer Authority, 2014). Demolished lots have 157 typically been converted to lawns, although more intensive redevelopment has occurred in some 158 areas.

159 Work for this study was conducted on lots within the urban core of Buffalo (Figure 1), 160 focusing on the west and southeast regions of the city, regions that have experienced substantial 161 depopulation and urban decay. Sites were selected for proximity to one another and availability 162 for research purposes. All sites were privately owned. Work was conducted at four sites with 163 intact vacant homes (hereafter intact sites), four demolished home sites that were replanted in 164 turfgrass (turfgrass RVL sites), and four demolished home sites that have been converted to UA 165 (agriculture sites). The intact and turfgrass sites are mowed periodically. The UA sites are used 166 for intensive vegetable production, receiving regular tillage and soil amendments of manure 167 and/or compost.



Figure 1 Location of study sites within the city of Buffalo, NY.

169 Soil Sampling

170 Two soil sampling campaigns were conducted between October and December 2021

171 (non-growing season) and between May and July 2022 (growing season), as plant demand for N

- 172 varies substantially between these seasons. At the agriculture and turfgrass sites, two soil profiles
- 173 were taken at each site, one within the former foundation (fill soil) and one within the yard



Figure 2 Sampling regime: sampling in fill and yard soils at agricultural and turfgrass sites and in yard soils at intact sites. Four sites of each type were sampled. Abbreviations for each of these location types are; AF: agricultural fill, AY: agricultural yard, IY: intact yard, TF: turfgrass fill, TY: turfgrass yard

- 174 (Figure 2), as pervious work has shown that soil properties can differ substantially between
- 175 former foundation and yard locations on regreened vacant lots (Kelleher et al., 2020; Shuster et
- al., 2014, 2015). For the intact sites, only one soil profile was taken from the yard. Sampling
- 177 locations within each lot were randomly determined using ArcGIS. Soil samples were collected
- using an 8.26 cm bucket auger at the following intervals: 0-10 cm, 15-25 cm, 50-60 cm, and 95-
- 179 105 cm. Intact soil cores (77 mm x 51 mm) were taken at the same intervals for bulk density
- 180 analysis and were collected using a slide hammer corer. To maintain somewhat uniform soil
- 181 moisture conditions among samples, no soil samples were taken within 24 h of a precipitation
- 182 event greater than 7 mm. Soils were stored at field moist conditions at 4 °C between sampling
- and analysis (approximately 1 week for all chemical analyses except for pH and elemental
- 184 analysis). All soils were sieved through a < 2 mm sieve prior to analysis.

185 Soil Physical and Chemical Analyses

Soil texture was determined via the hydrometer method, following the protocol of Gee and Bauder (1986). Gravimetric water content was measured on the sieved < 2 mm soil fraction by heating the soil in an oven at 115 °C to a constant weight. Soil cores were oven dried and weighed to determine bulk density. Soil pH was measured in a 2:1 water:soil slurry (Robertson et al., 1999) using a Fisherbrand pH meter with Orion probe (Fisher Scientific, Hampton, NH).

191 Soil nitrate (NO₃⁻) and nitrite (NO₂⁻) were extracted using 2M KCl at a 5:1 solution:soil 192 ratio. Soil subsamples (9 g) were shaken for two hours on an end-over shaker and filtered 193 through a 0.2 µm nylon membrane filter. Concentrations were determined by colorimetric 194 analysis on a San ++ segmented flow analyzer (Skalar Analytical B.V., Breda, Netherlands), 195 using the EPA cadmium reduction method (Eaton & Franson, 2005; U.S. EPA, 1993). Total soil 196 carbon and nitrogen were measured by combustion analysis. Samples were dried at 40 °C to a 197 constant weight and then ground on a roller mill so that 88% or more of the sample passed 198 through a 100 mesh sieve. Ground samples were sent to the Cornell Nutrient Analysis 199 Laboratory for elemental analysis on a Primacs SNC-100 analyzer (Skalar Analytical B.V., 200 Breda, Netherlands).

Water-extractable organic carbon (WEOC) was measured on all samples as an index of
microbially-available carbon substrates, following the protocol of Elliot et al. (1999). Briefly,
13g of soil (field moist weight) was mixed with Type 1 water at a ratio of 3:1 water:soil. Soils
were shaken overnight on an end-over shaker, filtered through a 0.2 μm nylon membrane filter,
and analyzed for dissolved organic carbon using a TOC-V_{CSH} combustion analyzer (Shimadzu
Corporation, Kyoto, Japan). WEOC was then calculated on a per gram of dry soil basis using the
gravimetric water content of each soil.

Soil Denitrification Activity

209 Soil denitrification potential was measured using denitrification enzyme assays following 210 the methods outlined in Groffman et al. (1999). To summarize, 25 g of sample was weighed into 211 125 ml Erlenmeyer flasks. One quarter of all samples were run in triplicate for quality control. 25 ml of DEA medium containing 100 mg l⁻¹ NO₃⁻-N, 40 mg l⁻¹ dextrose, and 10 mg l⁻¹ 212 213 chloramphenicol was added to each flask, and the flasks were stoppered with a butyl rubber 214 stopper. The flasks were then purged of oxygen by four cycles of flushing with N₂ gas followed 215 by evacuation for one minute. Acetylene was then added to each flask, which inhibits the last 216 step of denitrification, the reduction of nitrous oxide (N₂O) to N₂, allowing the quantification of 217 denitrification via measurement of N₂O production. Acetylene was purified by running industrial 218 grade acetylene gas through two traps of concentrated sulfuric acid followed by a water trap, and 219 purified acetylene gas was added to the headspace of each flask at the equivalent of 10% of 220 headspace volume. Flasks were incubated at 22 °C on a rotary shaker table and headspace gas 221 samples were taken at 30 and 90 minutes, with an additional quality control sample taken at 60 222 minutes in the case of replicates. Headspace gas samples were placed in evacuated 11 ml 223 Exetainer vials. The headspace gas samples were analyzed for N₂O concentration using a GC-224 2014 GHG-3 gas chromatograph equipped with an electron capture detector (Shimadzu 225 Corporation, Kyoto, Japan).

To complement these laboratory assays, we measured the isotopic composition of soil NO₃⁻. This provides an integrated, semi-quantitative measure of the total amount of denitrification a pool of NO₃⁻ has undergone, as denitrification strongly fractionates against heavy isotopes. Remaining NO₃⁻ in the soil thus becomes progressively enriched in the heavy isotope as the extent of denitrification progresses in soils (Burns et al., 2009). The isotopic

231 composition of extracted NO₃⁻ was analyzed using the denitrifier method (Casciotti et al., 2002). 232 Briefly, KCl solutions of NO₃⁻ were added to colonies of denitrifying bacteria that have been 233 genetically modified to only produce nitrous oxide (N₂O) rather than N₂. The isotopic 234 composition of the precursor NO3⁻ is then calculated from the isotopic values of nitrogen and 235 oxygen measured on the N_2O gas using an isotope ratio mass spectrometer (IRMS). Samples 236 were pre-treated for removal of (NO₂⁻) following the procedure outlined in Granger & Sigman 237 (2009) if they contained NO₂⁻ concentrations greater than 5% of NO₃⁻. KCl extracts from the 238 non-growing season were analyzed at UC Davis Stable Isotope Facility using a ThermoFinnigan 239 GasBench + PreCon trace gas concentration system interfaced to a Delta V Plus IRMS (Thermo Scientific, Bremen, Germany). Growing season KCl extracts were analyzed at the UC Riverside 240 241 Facility for Isotope Ratio Mass Spectrometry on a Delta-V Advantage IRMS interfaced to a 242 GasbenchL (Thermo Scientific, Bremen, Germany).

243 Statistical Analysis

Statistical analysis was performed in R (R Core Team, 2022). For analysis, we grouped data by location type, comprised of the unique combinations of land use and soil type, for a total of five location types: UA on foundation fill (AF), UA on former yards (AY), turfgrass on foundation fill (TF), turfgrass on former yards (TY), and intact yards (IY). We analyzed data from both seasons (growing and non-growing season) jointly, and we constructed separate models for each sampling depth interval.

For each soil property of interest at each depth, we tested for differences between location types by performing a Location Type x Season two-way ANOVA. In no model was the Location Type x Season interactive term significant, so we removed the interactive term from all models via a stepwise selection procedure. The results presented here are thus from models that

only include terms for the main effects of Location Type and Season. For ANOVAs where the
Location Type term was determined to be significant, we used Tukey's test (Tukey, 1949) for
post-hoc comparisons among location types. In these models, all parameters except pH were
log₁₀-transformed for variance stabilization. When reporting group means and standard errors in
the results, however, we report the arithmetic mean of untransformed data.

259 To analyze the isotopic NO_3^- data for semi-quantitative evidence of denitrification, we jointly examined δ^{15} N and δ^{18} O values, comparing these to literature values of the isotopic 260 261 composition of known NO₃⁻ sources (Kendall, 1998). We then visually looked for evidence of 262 increases in isotopic enrichment with each increase in depth along the "denitrification line," the slope of which (a 0.5 % increase in δ^{18} O with a 1 % increase in δ^{15} N) describes how 263 264 denitrification typically enriches ¹⁵N and ¹⁸O in residual NO₃⁻ pools. This approach assumes that 265 most NO₃⁻ is sourced from surficial soil layers and that progressive enrichment at depth is a 266 signal of denitrification as nitrate migrates vertically in the soil profile.

267

Results

268 Although our analysis examined soil properties throughout a 1 m soil profile, significant 269 differences among land uses for these soil properties were almost exclusively confined to the 270 uppermost soil sampling interval (0-10 cm). We therefore focus the following results and 271 discussion on the results from this uppermost sampling interval, unless explicitly noted otherwise 272 in the text. Full results for all sampled depths are presented in the Supporting Information 273 (Error! Reference source not found., Error! Reference source not found.), and additional 274 figures that display data for all soil depths are also provided in the Supporting Information 275 (Error! Reference source not found., Error! Reference source not found.). Full tables of

ANOVA results and post hoc comparisons are presented in the Supporting Information (Error!
Reference source not found.).

278 Soil Biogeochemical Properties

279 Soil NO₃⁻ concentrations differed greatly between seasons, but there were few differences 280 in NO₃⁻ concentrations among location types. Across all location types, NO₃⁻ concentrations were 2.6x higher in the winter (mean = 12.4 μ g NO₃⁻-N g⁻¹) than in the summer (mean = 4.6 ug 281 $NO_3^{-}Ng^{-1})(p < 0.001)$ (Error! Reference source not found.). Across both the growing and 282 283 non-growing season, NO_3^{-} concentrations tended to be higher at agricultural locations compared 284 to other location types, but the only significant differences were between the agricultural location types (AY: mean = 12.6 μ g NO₃⁻N g⁻¹; AF: mean = 12.7 μ g NO₃⁻N g⁻¹) and the TY location 285 type (TY: mean = $4.5 \ \mu g \ NO_3^{-}N \ g^{-1}$) (p = .011 and .0097 for comparisons of TY with AY and 286 287 AF, respectively) (Figure 3A). 288 Total N tended to be higher at both UA locations (AF and AY, pooled mean = 0.52%) 289 than non-UA locations (pooled mean = 0.21%) (p<0.001), though AF sites were not significantly 290 higher than IY sites (Figure 3C). There was no effect of season on total N. Differences between 291 location types ranged from a 2.9x difference at AY (mean = 0.58%) compared to TY (mean = 292 0.20%) to 2.5x greater at AY compared to IY (mean = 0.23%) (Error! Reference source not 293 found.).

Water-extractable organic carbon was significantly higher at UA sites compared to non-UA sites, but there were no seasonal effects (Figure 3B). AF and AY sites were 2.8x higher in WEOC (pooled mean = 0.11 mg C g^{-1}) than non-UA location types (pooled mean = 0.04 mg C d^{-1}) (p<0.001) (Error! Reference source not found.Error! Reference source not found.).

298	Differences ranged from a 4.4x greater concentration at AY	(mean = 0.11)) than TY ((mean =
	L/ L/	`	/	`

0.03) to a 2.1x greater concentration at AF (mean = 0.10) than TF (mean = 0.05).

Total carbon tended to be higher at UA locations than non-UA locations and showed no obvious seasonal trends across location types. UA location types (pooled mean = 8.74%) were 1.9x higher than non-UA location types (pooled mean = 4.61%) (Error! Reference source not found., Figure 3A). Differences ranged from a 2.1x greater proportion at AY (mean = 9.51%) than TY (mean = 4.48%) (p<0.001) to a 1.7x greater proportion at AF (mean = 7.97%) than IY (mean = 4.78%) (p<0.05) (Error! Reference source not found.).

306Turfgrass sites generally had a higher pH than intact and agricultural sites and showed no307obvious seasonal trends across location types. Although turfgrass location types showed higher

308 pH (pooled mean = 7.94) than agricultural and intact (pooled mean = 7.57) (Error! Reference

309 source not found. AFigure 3) the only significant difference between location types was that pH

310 was 0.43 units greater at TF (mean = 7.96) sites than AY (mean = 7.53) (p<0.05) (Error!

- 311 **Reference source not found.**).
- 312 Soil Physical Properties

All soils had high clay content, with the majority of soils classified as clay soils (USDA-NRCS, 2018). The mean clay content of surface soils was 44% and there were no significant differences in clay content between location types. The proportion of the coarse fraction (> 2 mm) in samples likewise varied widely (229% to <1% of the weight of the > 2mm fraction), typically composed of patches of building rubble that likely contributed to preferential flow. Bulk density differed between location types, with UA soils having generally lower bulk densities than non-UA soils (Figure 3E). AY sites had lower bulk density (mean = 0.69 g cm⁻³)

- 320 than all other location types, containing high amounts of organic debris in the sites we sampled.
- 321 AF (mean = 1.11 g cm^{-3}) was not significantly lower only than any non UA soil types.



322

323 Figure 3 Physical and chemical soil properties by location type and season at the 0.05 m depth. (Location type abbreviations; AF:

324 agricultural fill, AY: agricultural yard, IY: intact yard, TF: turfgrass fill, TY: turfgrass yard). Note that the y axis for each plot is log

325 scale. Shared letters between two or more location types indicate no significant difference between respective location types. Bulk

326 density was only measured in the non-growing season.



328 Figure 4 Chemical soil properties by location type and depth. Dots are mean values and error bars represent the standard error of the

mean. Points are vertically staggered at each depth interval for ease of comparison. Plots A-D are summer data and E-H are winter.

330 Note that the scales are different between seasons.

331 Denitrification Activity

332 Denitrification potential was not significantly different among any of the site types at any depth. The mean denitrification rate in the surficial sampling interval was 200 ng N₂O-N g⁻¹ h⁻¹ 333 334 (Figure 3F). Across all sites, denitrification potential decreased substantially with depth (Figure 335 4D,H), consistent with lesser substrate availability in lower horizons (Figures 4A,E,C,G). 336 Although there were no differences between denitrification potential at location types, the 337 availability of substrates for denitrification were significantly, positively correlated with 338 denitrification potential across all land uses and depths. Denitrification potential was significantly positively correlated with WEOC (R^2 =.3837, p<.001, Figure 5A), NO₃⁻ (R^2 =.3257, 339 p < 0.001, Figure 5B), total N (R²=.4342, p< 0.001, Figure 5C), and total C (R²=.2483, p < 340 341 0.001, Error! Reference source not found.B), and negatively correlated with soil pH (R²=.2326, p <.001, Error! Reference source not found.A). There was no significant effect of 342 343 season on the slopes of these linear regression relationships, although for the relationship 344 between denitrification potential and NO_3^{-} , the intercept of the relationship was higher in 345 summer than winter.

Very few sites exhibited progressive enrichment of NO₃⁻ δ^{15} N and δ^{18} O along the 346 expected slope of denitrification with increasing soil depth, indicating that little denitrification 347 348 occurred in these soils as soil water infiltrates deeper soil horizons, or that the signal of any denitrification activity was overwhelmed by endogenous NO₃⁻ production in these horizons. δ^{15} N 349 and δ^{18} O values of extracted NO₃⁻ fell almost entirely within the envelope of expected values for 350 351 fertilizer and manure inputs (Figure 6). Four of the 34 0.95-1.05m samples (all from non-UA location types) with low NO₃⁻ concentrations did show an increase in both δ^{15} N and δ^{18} O 352 353 between 0.50-0.60m and 0.95-1.05m depths, though these did not follow the consistent slope 354 expected in the case denitrification.



Figure 5 Multiple linear regressions of denitrification potential by NO3⁻, WEOC, and total N for all depths. Shaded areas represent the 95% confidence interval of the regression lines. Note that both axes are log scale.





Figure 6 NO₃⁻ isotopic ratios of samples at all depths, with lines connecting samples sharing the same depth profile. Boxes indicate ranges of expected isotopic ratios of δ^{15} N and δ^{18} O based on sources of NO₃⁻ and the black arrow is the slope that isotopic enrichment with denitrification is expected to follow (Kendall, 1998).

364 **4. Discussion**

365 4.1. Potential for N Runoff from RVLs

366 Urban agriculture has a moderate potential to be an increased source of N to local water

- 367 bodies via surface runoff compared to other uses of RVLs, particularly in winter, when NO₃⁻
- 368 concentrations are elevated. Soil N concentrations, including NO₃⁻, were moderately higher at
- 369 agricultural locations than non-agricultural locations, with NO₃⁻ in particular being higher in

surface soil horizons during winter. Despite higher N concentrations and higher availability of
carbon substrate, soils at agricultural locations did not exhibit significantly higher denitrification
than non-agriculture, as measured either through denitrification potential or increasing isotopic
enrichment of NO³⁻ with depth. Increased N supply is thus not coupled to increased removal
capacity, and this decoupling results in conditions that are favorable to surface N runoff (Suchy
et al. 2023).

376 Although the decoupling of N supply from denitrification, documented here, will likely 377 favor increased N runoff, this is tempered by the fact that our observed soil NO₃⁻ concentrations 378 are not anomalously high, compared to other measurements found in the literature. Nitrate has 379 long been understood to be the most leachable form of N, and NO₃⁻ tends to dominate N export 380 in anthropogenic landscapes (Basu et al., 2011). The farms that participated in this study used 381 only organic fertilizers, and the lack of inorganic NO_3^- fertilizers applied in this setting is likely 382 to be a major factor responsible for the low soil NO3⁻ concentrations compared to conventional 383 agriculture. Growing-season NO₃⁻ concentrations for UA soils in this study are comparable to other UA soils found in temperate climate zones (mean = $6.4 \mu g N g^{-1}$), as reported by Alvarez-384 385 Campos & Evanylo (2019). Our observed concentrations were at the low end compared to values reported for organic farms, where summer concentrations ranged from 9.6-94 μ g N g⁻¹ (Ouyang 386 387 et al., 2018; Xie et al., 2021; Zhang et al., 2019). These concentrations are also considerably lower than the 20 µg N g⁻¹ end-of-season values recommended for maximum crop yields in 388 389 conventional agriculture in the US (Sullivan et al., 2003). The NO₃⁻ concentrations of this 390 study's non-UA sites were also low compared to those in soils of similar plant communities. 391 These concentrations were between the values observed by others for suburban lawns (mean = 9.4 μ g NO₃⁻N g⁻¹) (Martinez et al., 2014) and those observed in unfertilized pastures (mean = 392

393 3.1 µg NO₃⁻-N g⁻¹) (Jarvis & Hatch, 1994). Leaching of N to groundwater is also unlikely to be 394 substantial because of the soils' high clay content that limits hydraulic conductivity and low 395 NO₃⁻ concentrations at depth at all locations. Although we thus assess the risk of N leaching to 396 groundwater to be low due to the heavy clay soils at our sites, localities with particularly course 397 textured soils should conduct similar studies to confirm that this still holds in such conditions. 398 Although we conclude that the potential for nitrate runoff from RVLs is only moderate, 399 the potential for export of particulate organic N and dissolved organic N may exist at our UA 400 sites. Although nitrate is often the dominant N source of concern in runoff, organic N forms can 401 contribute substantially to total N loads (Hobbie et al., 2017; Jani et al., 2020). Total N at UA 402 locations were similar to 0.7 %N measured in raised garden beds (Salomon et al., 2020) and 0.38 403 %N in urban agriculture allotments (Edmondson et al., 2014) and were much higher than typical 404 values measured in farms across the Midwest that use chemical fertilizers (mean = 0.21 %N) 405 (Laboski et al., 2008). Total N percentages at TY, TF, and IY sites, in counter to this, were 406 toward the lower end of the range of values between 0.05 %N and 0.6 %N in a nationwide study 407 of lawns (Selhorst & Lal, 2012). The higher values for total N in UA soils suggest that 408 particulate N and dissolved organic N thus may drive increased N export at UA compared to 409 lower intensity uses (Mulvaney et al., 2009). Erosion control, careful timing of fertilizer 410 application, and appropriate tillage practices are key controls on organic N export that should be 411 implemented in the context of UA and other uses of RVLs (Sieczko et al., 2023; Whittinghill & 412 Sarr, 2021).

Despite the increased N supply, the potential for lower runoff may exist in UA compared to other landscape use types, driven by the physical properties of the soil. We observed generally lower bulk density in UA than other land uses. Organically amended soils are often lower in bulk

density and greater porosity associated with the formation of soil aggregates (Aggelides &
Londra, 2000). This has been shown in several cases to increase infiltration, resulting in lower
runoff amounts with rainfall (Spargo et al., 2006). Evanylo et al, (2008) for example found lower
bulk density and greater porosity, time to runoff generation, and infiltration rates in a compost
fertilized soil with total N similar to this study's sites than an adjacent unfertilized soil, resulting
in lower NO3⁻ and total N loads in addition to lower runoff volumes.

422 The observation of higher NO₃⁻ in winter is typical of non-growing season conditions, 423 where assimilatory N demand by plants is negligible (Laine et al., 1994, Sieczko et al. 2023). 424 Elevated winter NO₃⁻ concentrations in UA is of particular concern for runoff events given the 425 common occurrence of springtime eutrophication events, even in locations relatively unimpacted 426 by anthropogenic nutrient loading (Venugopalan et al., 1998). The occurrence of winter runoff 427 events is potentially increased in climates that experience substantial snowfall. While early 428 snowmelt infiltrates completely, late snowmelt can contribute to overland flow and runoff at 429 rates similar to impervious surfaces when soils infiltration capacity is reduced and exceeded by 430 snowmelt (Bengtsson & Westerström, 1992). Rain on snow events further contribute to runoff in 431 a similar manner. Adding to this concern is that rapid snowmelt, rain on snow, and other extreme 432 variability in winter weather are expected to increase with climate change (Casson et al., 433 2019). We thus assess that, if UA does contribute substantially to nutrient export to aquatic 434 ecosystems, it is most likely to occur during the non-growing season. Though the setup of runoff 435 collectors on private property accessible by the public was not feasible for this study, runoff 436 generation and concentrations should be investigated in future studies, as should the 437 concentration and loads of possible runoff events.

438 **4.2. Drivers of Denitrification**

439 We found that total N was a better predictor of denitrification potential than NO_3^- , even 440 though NO₃ is the immediate substrate requirement for denitrification. NO₃ is highly mobile in 441 the soil and is readily taken up by plants, and NO_3^- concentrations are known to respond rapidly 442 to precipitation events (Nangia et al., 2010; Yahdjian & Sala, 2010). We also found that soil 443 NO_3 concentrations were more highly variable within location types (average CV=0.75) than 444 total N, (average CV=0.52). These results suggest that NO_3^- levels fluctuate more so than total N 445 and that individual point measurements of NO_3^- may not be highly representative of total nitrate 446 availability over the weekly time scales that denitrification enzymes persist in soils. 447 In addition to measures of N and C, bulk density was a moderately strong predictor of 448 denitrification, but with a counterintuitively inverse relationship. It may be expected that higher 449 bulk density and lower pore volumes would favor anaerobic conditions for denitrification, but 450 denitrification was higher in low-bulk density soils. This is likely driven by high levels of soil 451 amendments at UA sites that increase fertility and decrease bulk density. We thus conclude that 452 bulk density is likely not directly driving denitrification rates but is instead acting as a proxy for 453 soil amendment levels (Aggelides & Londra, 2000). Similarly, while denitrification is known to 454 be suppressed at low soil pH (<5) (ŠImek & Cooper, 2002), we found an inverse relationship 455 between soil pH and denitrification potential, across the relatively limited pH gradient (7.03-456 8.90) examined here. We speculate that the inverse relationship between soil pH and 457 denitrification is again due to organic amendments in the surface horizon of UA locations that 458 contribute organic acidity to the soil and lower pH, and pH is thus a proxy for substrate 459 availability and does not directly influence denitrification at the circumneutral pH levels of the 460 soils examined.

461	Denitrification potential in UA sites was fairly typical for that of organic agriculture
462	(Ouyang et al., 2018; Yin et al., 2015). Although not significant, denitrification potential at UA
463	sites had higher mean values (UA pooled mean = 811 ng N ₂ O-N g ⁻¹ h^{-1}) than at almost all non-
464	UA location types (non-UA pooled mean = 498 ng N ₂ O-N g ⁻¹ h ⁻¹). Compared to agriculture using
465	inorganic fertilizers in temperate regions (D'Haene et al., 2003; Malique et al., 2019; Yin et al.,
466	2015), denitrification potential rates at the UA sites were fairly high. Denitrification potential
467	rates were fairly low among non-UA soils compared to the values found by other researchers,
468	though denitrification rates in lawns tend to be quite variable (Reisinger et al., 2016). Despite
469	this, other researchers have found lawns with similar denitrification potential (Hall et al., 2016;
470	Suchy et al., 2021). A possible explanation for lower rates in this study is that lawns measured in
471	most studies are occupied properties, and are generally fertilized and watered regularly, in
472	contrast to the non-UA sites in this study. While denitrification potential was lower at these sites,
473	N inputs are also much lower on these unfertilized turfgrass sites, compared to residential lawns
474	and UA sites.

475 **Conclusion:**

476 Urban agriculture has the potential to modestly increase N export through runoff. 477 Nitrogen availability is higher in UA than other land uses, but this is not counteracted by higher 478 denitrification rates that result in N removal. In particular NO₃⁻, the most leachable form of N, is 479 only slightly higher at UA sites, compared to other land use types. Runoff may be a larger 480 concern in the winter largely due to elevated NO3⁻ concentrations in the non-growing season 481 combined with the increased risk of runoff during snowmelt. Overall, we assess that the small 482 increase in N availability from UA use of RVLs does not pose a sizeable risk of increased N 483 export, particularly when its land cover percentage is relatively low, such as in Buffalo. Given

484	the risk of small increased N export and the positive impacts that UA can provide such as social
485	cohesion, nutritional benefits, and reuse of N from imported nutrients, we conclude that UA is
486	most likely a net socio-ecological and economic benefit for the communities it serves.
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- 744 Both authors contributed to the study conception and design. Data collection, analysis, and the
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