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9	Figures are at the end of this document.
10 11	A new calibration of the OPAM thermobarometer for anhydrous and hydrous mafic
12	systems
13	
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## 18 ABSTRACT

Melt-based thermobarometers are essential tools to recover pre-eruptive magma storage 19 conditions through their application to bulk rock and liquid chemistry. In active volcanic 20 21 systems, thermobarometric results can be combined with independent geophysical data during 22 or after an eruption to validate conceptual models. In this contribution, we revisit the thermobarometer for melts equilibrated with the mineral assemblage of olivine + plagioclase 23 24 + augitic clinopyroxene (OPAM). We first demonstrate that the most widely applied OPAM thermobarometer suffers from both random and systematic uncertainty even for anhydrous 25 26 melts, and that the uncertainty increases proportionally with melt H<sub>2</sub>O. To address this issue, 27 we use a modern compilation of anhydrous and hydrous OPAM-saturated experiments to regress a new melt-based OPAM thermometer and barometer. Our new equations recover a 28 29 validation dataset with a standard error estimate (SEE) of  $\pm 1.1$  kbar and  $\pm 35$  °C for pressure 30 and temperature respectively, as well as a low systematic uncertainty that does not depend on 31 melt H<sub>2</sub>O. Additionally, we present a novel statistical approach to determine the probability 32 that a given melt is OPAM-saturated, which can be used alongside petrographic observations. 33 Our thermobarometer and saturation test are presented as a straightforward R script which reads from an input csv file to be populated with natural data. We benchmark the new 34 calibrations on the products of the 2015 eruption of Wolf Volcano (Isabela Island, Galápagos) 35 36 and the 2014-2015 Holuhraun eruption (Iceland), both of which have independent geophysical 37 estimates of magma storage that agree well with our thermobarometric results.

38

#### **39 KEYWORDS**

40 Multiple saturation; OPAM; thermobarometry; basalt; Galápagos; Iceland

41

42 INTRODUCTION

43 Thermobarometry, the retrieval of pressure (P) and temperature (T) information from magmas, provides a quantitative window into the physicochemical structure of magmatic 44 systems (Blundy and Cashman, 2008; Grove et al., 1992; Neave and Putirka, 2017; Putirka, 45 46 2008). Thermobarometric equations relate T (°C) and P (kbar) to the measured composition (X) of a melt and/or mineral(s) using equilibrium crystallisation experiments (Putirka, 2008). 47 Equations may have a general form grounded firmly in thermodynamics with additional 48 49 empirical terms to improve the goodness of fit (e.g., Masotta et al., 2013; Neave and Putirka, 2017; Putirka, 2016), or be purely empirical and consist only of compositional variables with 50 51 or without P and T terms as independent variables (e.g., Jorgenson et al., 2022; Ridolfi and Renzulli, 2012). Irrespective of the calibration strategy, thermobarometers have significant 52 scope as petrological volcano monitoring tools. For example, P data can independently validate 53 54 geophysical estimates of pre-eruptive magma storage depth during or after eruptions (Hartley 55 et al., 2018; Stock et al., 2018; Ubide et al., 2023) or catalogue past changes in magma P and T for eruptions of volcanoes that lack a protracted historical monitoring record (Bouvet de 56 57 Maisonneuve et al., 2021; Higgins et al., 2021a; Jorgenson et al., 2024; Weber et al., 2020). Therefore, access to accurate, precise, and readily applicable thermobarometers is an essential 58 component of volcanic hazard management and eruption forecasting. 59

Melt-based thermobarometers (e.g., Blundy, 2022; Grove et al., 1992; Helz and 60 61 Thornber, 1987; Wilke et al., 2017; Yang et al., 1996) are a particularly useful implementation 62 of the method as *P*-*T* estimates can be derived from the chemistry of a single phase which is 63 often easily recovered from matrix glass, melt inclusions, or whole-rock data. However, to be reliably applied, the input melt composition must represent a true equilibrium liquid, free from 64 65 the effects of magma mixing or crystal accumulation which are rife in magmatic systems (Cashman and Blundy, 2013; Kent et al., 2010; Passmore et al., 2012). Melt-based 66 67 thermobarometry equations regress melt composition against P or T by assuming that the melt is buffered by a discrete, low-variance phase assemblage; the mineral proportions change in response to P, T, and bulk composition, imparting a predictable chemical control on the coexisting melt. Such "multiple saturation thermobarometers" are broadly applicable in both high-melt-fraction and low-melt-fraction magmatic systems, recording P-T conditions at the point where melts decouple (via eruption or melt migration) at a speed that is too rapid for further chemical equilibration with the buffering crystal assemblage (Solano et al., 2014; Jackson et al., 2018; Lissenberg et al., 2019; Blundy, 2022).

The OPAM thermobarometer is a melt-based multiple-saturation thermobarometer 75 76 relying on equilibrium between the mineral assemblage olivine + plagioclase + augitic clinopyroxene (the olivine gabbro cotectic; Herzberg, 2004). Early work showed that the 77 location of the OPAM cotectic was strongly P dependent, and could be parameterised as a 78 79 function of melt major element composition (Ca+Mg+Al+Si+Fe+Na; Shi, 1993, 1992). Later, 80 the effect of minor elements (K, Ti, Cr) were deemed significant to capture the melt composition effect within the form of the cotectic equation (Grove et al., 1992; Voigt et al., 81 82 2017; Yang et al., 1996). The Yang et al (1996) equations remain the most widely applied calibration of the OPAM cotectic (e.g., Baxter et al., 2023; Bell et al., 2021; Hartley et al., 83 2018; Stock et al., 2018); this model comprises three empirical equations which regress the 84 molar proportion of three melt-compositional cations (Al, Ca, Mg) as separate functions of Na, 85 86 K, Fe, Ti, Si, and P (kbar). Originally, these P-dependent equations were developed as a 87 method to model basaltic fractional crystallisation and were not strictly proffered as a 88 barometer (Yang et al., 1996). The calibration experiments were predominantly performed at 1 atmosphere (1 atm; n=164) and extrapolated across a range of crustal magma storage 89 90 conditions (>1 atm to 10 kbar) using only a small number of high-P experiments (n=26). The hydrous, high-P experiments used to calibrate the Yang et al (1996) equations span up to  $\sim 2$ 91 92 wt.% H<sub>2</sub>O (Baker and Eggler, 1987), theoretically limiting their applicability to low-H<sub>2</sub>O ocean

93 island and mid-ocean ridge settings. However, the explicit effect of H<sub>2</sub>O was not tested, even though OPAM-saturated (OPAM<sub>SAT</sub>) melts are stable across a wide range of melt H<sub>2</sub>O content 94 (0-7.7 wt% H<sub>2</sub>O; Fig. 1a), with increasing H<sub>2</sub>O supressing the T and MgO (wt%) of OPAM 95 saturation (Fig. 1b). In general, constraining the effect of H<sub>2</sub>O on thermobarometric models is 96 vital given that the addition of even moderate concentrations (much less than 2 wt%) to 97 anhydrous compositions can have a drastic effect on phase equilibria in basaltic systems 98 99 (Almeev et al., 2012, 2007; Husen et al., 2016). Yang et al., (1996) also presented a separate thermometric equation that regressed T (°C) as a function of the same melt composition 100 101 variables (Na, K, Fe, Ti, Si) plus P as an independent variable. The quoted uncertainty of the Yang et al (1996) thermometer is  $\pm 20$  °C, based on the observation that 170 of the 190 calibrant 102 experiments are recovered with a T residual of  $\pm 20$  °C or less. However, this assessment of 103 104 uncertainty was self-validated by Yang et al (1996), calculated with the calibration experimental dataset rather than an independent set of experiments. 105

Hartley et al., (2018) first implemented the three Yang et al (1996) *P*-dependent compositional equations as a cohesive barometer, returning the *P* that minimised the  $\chi^2$  misfit between measured Al, Ca, and Mg content and those predicted by the three equations. The Yang et al (1996) equations were solved at 0.01 kbar intervals between -5 and 15 kbar until the misfit  $\chi^2$  was minimised, indicating the unique *P* of convergence. The misfit  $\chi^2$  is defined by the general equation:

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113 
$$\chi^2 = \sum_{i=1}^{3} \left( \frac{X_i - Y_i}{\sigma_i} \right)^2$$
(1)

114

where  $X_i$  are the measured cation fractions of Al, Ca, and Mg,  $Y_i$  are the cation fractions predicted from the three Yang et al (1996) equations, and  $\sigma_i$  is a term for analytical uncertainty 117 that allows some leeway in the convergence. As well as providing the P that minimised the misfit  $\chi^2$ , Equation 1 also returns a measure of the convergence quality or probability of fit (P<sub>F</sub>). 118 Hartley et al (2018) repurposed this quantitative information as a probability that the melt was 119 120  $OPAM_{SAT}$ , using it to screen natural melts that would not yield a reliable P of three-phase saturation. A low P<sub>F</sub> indicates that the three Yang et al (1996) equations do not intersect closely 121 in multidimensional space outside of analytical uncertainty and so there is no unique P solution 122 for the input melt. Hartley et al (2018) suggested setting a cut off for  $P_F$  at <0.8, which was 123 determined by observing the P<sub>F</sub> above which the P of an experimental dataset was reproduced 124 125 with low random and systematic uncertainty by the Yang et al (1996) equations, applied 126 through Equation 1. Their test dataset contained 157 OPAM<sub>SAT</sub> experiments including some of those used to calibrate the Yang et al (1996) equations as well as several more recent studies. 127 128 Hartley et al (2018) found that the systematic (+1 kbar) and random (±2.5 kbar) uncertainties when using the whole dataset dropped after removing around 40% of the OPAM<sub>SAT</sub> 129 experiments, which all had  $P_F < 0.8$ , yielding new lower systematic (+0.34 kbar) and random 130 131 (±1.32 kbar) uncertainties for the P<sub>F</sub>-filtered dataset. However, by using experiments from the Yang et al (1996) equations in their test dataset, the model performance was, again, partially 132 self-validated by Hartley et al (2018). Further, the removal of around 40% OPAM<sub>SAT</sub> 133 experiments raises questions about the reliability of their quoted  $\pm 1.32$  kbar barometer 134 135 uncertainty if no methodological or analytical reason for their omission can be found.

Baxter et al., (2023) released a Python-based implementation of Equation 1, mimicking the approach of Hartley et al (2018). Their code provides an open-access, reproducible method to recover *P* estimates and P<sub>F</sub> for OPAM<sub>SAT</sub> natural melts via the three Yang et al (1996) cotectic equations. The Yang et al (1996) *P*-dependent thermometer was omitted from their code. Baxter et al., (2023) tested the Hartley et al (2018) method using a near identical workflow, collecting a dataset which included Yang et al (1996) calibrant experiments plus a suite of newer studies (n=250). Using a filter of  $P_F \ge 0.8$ , they found that only 92 experiments remained (37%) which had been reproduced with a mean absolute error of ±1.13 kbar. The Baxter et al., (2023) test of the Hartley et al (2018) method contains the same issues of selfvalidation and unexplained filtering of OPAM<sub>SAT</sub> experiments which remain to be justified.

The Yang et al (1996) cotectic equations have also featured in other implementations 146 of the melt-based OPAM thermobarometer. Michael and Cornell (1998) rearranged the Ca 147 148 Yang et al (1996) equation for P and used it as a stand-alone barometer; rearranging multiple linear regressions in this way is mathematically incorrect (Hartley et al., 2018). Voigt et al 149 150 (2017) presented an updated parameterization of the three *P*-dependent Yang et al (1996) equations by considering Cr,  $Fe^{2+}$  and  $Fe^{3+}$  as additional variables, although there is no 151 reproducible method with which to apply and test their equations (e.g., Baxter et al., 2023). 152 153 Kelley and Barton (2008) calculated the liquid compositions at 0.1 kbar increments with the three Yang et al (1996) cotectic equations, converted them to normative mineral compositions, 154 regressed P against the calculated and original compositions, and finally computed the mean 155 156 P. This unduly complex method performs significantly worse than the original Yang et al (1996) barometer applied via the approach of Hartley et al (2018) when tested using OPAM<sub>SAT</sub> 157 experiments (Baxter et al., 2023). 158

In this contribution, we re-visit the OPAM thermobarometer using the wealth of new 159 160 experimental data available since the implementation of Yang et al (1996). An experimental 161 dataset of OPAM<sub>SAT</sub> experiments with no obvious quality issues are compiled. These 162 experiments span 0-7.7 wt% H<sub>2</sub>O and 0.5-10 kbar, encompassing the full crustal spectrum of OPAM<sub>SAT</sub> melts. By first omitting studies which were used to calibrate the Yang et al (1996) 163 164 equations, we test the robustness of the filtering method of Hartley et al (2018) and the quoted Yang et al (1996) P uncertainty (±1.13 kbar) of Baxter et al., (2023) using a truly independent 165 166 test dataset. Based on these findings, we highlight the large true uncertainty associated with 167 the Yang et al (1996) thermometer and barometer, exacerbated by increasing melt H<sub>2</sub>O. In light of this, we present a new empirical thermometer and barometer for OPAM<sub>SAT</sub> melts at 168 anhydrous-hydrous conditions and crustal pressures (0.5-10 kbar) as a function of melt 169 170 composition. We also introduce a new statistical method for determining the probability that a melt is OPAM saturated which allows for application of our geothermobarometer to natural 171 melts where petrological constraints on multiple saturation are unavailable. Our equations are 172 173 applied to basaltic erupted products from the 2015 eruption of Wolf Volcano (Isabela Island, Galápagos archipelago) and the 2014-2015 Holuhraun eruption (Iceland) which both have 174 175 extensive petrographic, geochemical, and geophysical records (Halldórsson et al., 2018; 176 Hartley et al., 2018; Stock et al., 2020, 2018).

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178 METHODS

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## 180 Database Assembly and Filtering

181 OPAM<sub>SAT</sub> experiments were collected from the Library of Experimental Phase Relations (LEPR; Hirschmann et al., 2008), corrected for typos and copy errors, and 182 supplemented with experiments published after circa 2008. Experiments conducted at 1 atm 183 were excluded to avoid issues of rapid disequilibrium crystal growth in 1 atm experiments 184 185 (Mollo et al., 2010; Ziberna et al., 2017) and to prevent an overweighted regression at pressures 186 (depths) irrelevant to pre-eruptive magma storage (Neave and Putirka, 2017). Any studies not 187 reporting the abundances of mineral run products were not considered further. The upper P was limited to 10 kbar where OPAM<sub>SAT</sub> data become sparse, resulting in extrapolation over 188 189 large P intervals. Ten kbar (~32 km in oceanic crust; Putirka, 1997) encapsulates the thickness 190 of most oceanic crust and is the *P* limit of the thermobarometer calibration of Yang et al (1996). 191 No filter was used for melt H<sub>2</sub>O as the pre-existing H<sub>2</sub>O content of a magma is rarely known 192 which would limit the confident application of our updated equations. Experiments crystallising major silicate phases outside of OPAM (biotite, amphibole, orthopyroxene, 193 194 garnet) were removed. However, we retained experiments which crystallised Fe-oxide phases, 195 despite Putirka (2008) finding that strictly OPAM<sub>SAT</sub> experiments (i.e., without Fe-oxides) led to an improved fit, as removing these experiments may also limit the model applicability in 196 natural systems. Removing experiments crystallising major silicate phases outside of OPAM 197 198 is important as the critical assumption of an OPAM thermobarometer is that the melt is buffered by a common assemblage of olivine + plagioclase + augitic clinopyroxene whose compositions 199 200 and proportions change as a function of P and T. A total alkali filter was not initially applied, although we discuss the role of alkalis later; while Hartley et al (2018) state that the OPAM 201 thermobarometer should be avoided for transitional or alkali basalts due to poor performance, 202 203 this is at odds with the Yang et al (1996) equations which are calibrated using some alkalic melt compositions. Indeed, our compiled OPAMSAT experiments span to alkalic melt 204 compositions (Fig. 1c). 205

206 The liquid cation fractions of experimental melts were calculated using the method outlined in Table 1 of Putirka (2008), and are denoted by X followed by the oxide subscript 207 (e.g.,  $X_{AlO_{15}}$ ). Additional experiments were removed which did not report  $X_{KO_{05}}$  (i.e., where it 208 is not clear whether potassium is legitimately absent or not reported; 29 experiments) and/or 209 210 which exhibited iron loss or gain greater than 20% (7 experiments). Outliers, defined as an 211 abnormal value at odds with the regular distribution of the dataset, were subsequently appraised 212 in the remaining data by inspecting x-y plots of liquid cation fractions. The potential for outliers 213 is high due to the difficulty of performing and analysing experimental run products. However, 214 the dataset appeared largely coherent, and few outliers were identified. The exception was scatter at high and low  $X_{CaO}$ , which was removed using a filter of 0.04 <  $X_{CaO}$  < 0.14 (n=12), 215 216 and a single experiment (Y0206-1) from Neave et al (2019) which showed anomalously

217 heterogeneous glass compositions. Our final OPAM<sub>SAT</sub> compilation (n = 249) is reported in Supplementary Table 1a, including the mean and range of key variables (Supplementary Table 218 1b). A full reference list for all experiments can be found in Supplementary Table 1c. To our 219 220 knowledge, this dataset consists of good-quality OPAM<sub>SAT</sub> experiments for calibrating or testing OPAM thermobarometers. However, to address our earlier comment on the limitation 221 of the Baxter et al., (2023) and Hartley et al (2018) studies, we subset our OPAM<sub>SAT</sub> 222 223 compilation (n=249) to remove all calibrant experimental studies named in Table 4 of Yang et al (1996) when testing the performance of the Yang et al (1996) barometer and thermometer 224 225 (n=199; marked in Supplementary Table 1a).

226

# 227 Asserting OPAM saturation

228 Determining OPAM saturation is critical for reliable application of the 229 thermobarometric equations which rely on this equilibrium assemblage. We stress that the most 230 valid method for determining multiple saturation is to use petrographic indicators such as 231 texturally equilibrated OPAM mineral phases (i.e., euhedral, not resorbed). For bulk rock analyses of magmas considered as putative melts, "petrological cannibalism" (Cashman and 232 Blundy, 2013) may make quantitative textural analysis (e.g., crystal size distributions) and/or 233 chemical mapping essential to elucidate exogenous OPAM crystal populations (Higgins et al., 234 235 2021b; Neave et al., 2017, 2014; Sheldrake and Higgins, 2021). For example, a hallmark of 236 less-evolved basaltic magmas is high-An# phenocrysts, or phenocryst cores, at disequilibrium 237 with their matrix glass (Neave and Namur, 2022). Significant crystal accumulation may be 238 identified by strong correlations between modal mineral abundances and mineral-compatible 239 trace elements, although it can also be necessary to identify and correct for an accumulated mush liquid in addition (Passmore et al., 2012). Magmas which have accumulated crystals 240 241 should have their bulk chemistry corrected by mass balance, akin to a post-entrapment

crystallisation correction for melt inclusions, before using any melt-based thermobarometer. 242 Finally, we note that the petrographic absence of a phase in hand sample does not preclude its 243 presence in the crystallising assemblage. This is often termed "cryptic fractionation" and 244 245 manifests itself in mid-ocean ridge basalts as the "pyroxene paradox" whereby the chemical mass balance between minerals, glass, and whole rock requires clinopyroxene that is 246 conspicuously lacking as a phenocryst (Dungan and Rhodes, 1978; Grove et al., 1992; 247 248 Herzberg, 2004; Kelley and Barton, 2008; Lissenberg et al., 2019). Based on the range of processes which may prevent a magma from representing an OPAM<sub>SAT</sub> melt, it is preferable to 249 250 use clean, unaltered, microlite-free glass compositions in petrographically characterised 251 magmas as the input for OPAM thermobarometry.

An alternative, or a complement, to petrographic analysis is a melt-based binary 252 253 classifier to distinguish OPAM<sub>SAT</sub> melts from OPAM undersaturated (OPAM<sub>UNSAT</sub>) melts. A 254 classifier is particularly useful for processing large regional or global datasets where original samples cannot be obtained, or whole rock data where multiple saturation cannot be 255 256 petrographically determined. Previous methods include graphical (Maclennan et al., 2001; 257 Yang et al., 1996), empirical (Kelley and Barton, 2008), and statistical or probabilistic (Baxter et al., 2023; Hartley et al., 2018) approaches. The latter remains the most widely applied but 258 suffers from false negatives whereby OPAM<sub>SAT</sub> experimental melts with no explainable defect 259 260 are classified as OPAM<sub>UNSAT</sub> melts (Baxter et al., 2023; Hartley et al., 2018). A saturation 261 classifier must work to limit, as much as feasibly possible, the presence of false positives (false report of OPAM<sub>SAT</sub> melts) or false negatives. The aim of an OPAM saturation binary classifier, 262 or any binary classifier for that matter, is to identify or engineer differences between the two 263 264 categories of data so they can be mathematically discriminated from one another. We have developed a simplified statistical approach, presented as an R script (R Core Team, 2023), 265

which returns a probability of OPAM saturation ( $P_{SAT}$ ) using our up-to-date experimental dataset.

We first consider a set of critical oxides which must be reported and non-zero (i.e., 268 269 chemically measured in either an experimental or natural melt) to use our saturation test. The critical oxides are chosen by inspecting our OPAM<sub>SAT</sub> compilation and are expressed in wt% 270 in their raw form: SiO<sub>2</sub>, Al<sub>2</sub>O<sub>3</sub>, MgO, FeO, CaO, Na<sub>2</sub>O, K<sub>2</sub>O, TiO<sub>2</sub>. For the entire saturation 271 272 test process, all experimental and natural melts are normalised to 100 wt% anhydrous considering just these eight critical oxides. The eight critical oxides also perform an essential, 273 274 non-negotiable, first-pass test for identifying OPAM<sub>SAT</sub> melts, by using them to define a series of OPAM convex hulls. A convex hull is a polygon that connects the outer ring of points in an 275 X-Y plot, in this case a Harker plot of two critical oxides (e.g., SiO<sub>2</sub>[x] versus Al<sub>2</sub>O<sub>3</sub>[y]; Fig. 276 277 2a). The OPAM convex hull is expanded to account for an average relative analytical uncertainty of  $\pm 5\%$ . We define the OPAM convex hull for the 28 pairwise combination of the 278 8 critical oxides and use it as an initial test to verify if a natural or experimental melt of 279 280 unknown saturation state could be OPAM<sub>SAT</sub>. In other words, if a melt subjected to the saturation test does not fall inside all 28 OPAM convex hulls, the melt does not represent a 281 reasonable approximation of an OPAM<sub>SAT</sub> melt in multi-dimensional space based on our 282 current experimental knowledge. The OPAM convex hull test provides a rapid method to 283 284 preliminarily screen melts that are being applied beyond the range of current OPAM<sub>SAT</sub> 285 experiments.

Our OPAM<sub>SAT</sub> experimental dataset is combined with OPAM<sub>UNSAT</sub> melts from the large experimental compilation of Weber and Blundy (2023). The OPAM<sub>UNSAT</sub> melts are those experiments that contain up to two of the three major OPAM mineral phases (olivine  $\pm$ plagioclase  $\pm$  augitic clinopyroxene) plus or minus Fe-oxides, and are performed under pressurised conditions at  $\leq 10$  kbar. Further, we filter these OPAM<sub>UNSAT</sub> melts to ensure they too are within the 28 OPAM convex hulls as described above. The result is an OPAM<sub>UNSAT</sub> dataset that is coincidentally identical in size to our OPAM<sub>SAT</sub> dataset (n=249), together forming our "saturation dataset" (n = 498; Fig. 2; Supplementary Table 2a; full references in Supplementary Table 2b). These 498 experiments are used to define the probability that a melt of unknown saturation state (e.g., a natural melt composition) is OPAM<sub>SAT</sub> given that it first satisfies the OPAM convex hull test.

297 The experimental data of the saturation dataset are combined with the natural (unknown) dataset to produce a single, coherent "input dataset" for comparison. Next, all melts 298 299 within the input dataset have their oxide melt chemistry transformed. Raw compositional data carry only relative information and are subject to nonnegative and constant-sum (100 wt%) 300 constraints (Boschetty et al., 2022). Instead, ratios between compositional data better measure 301 302 variability which can be achieved using log-ratio transformations that map data into an 303 unconstrained Euclidian space (Chayes, 1971). We use the isometric log ratio (ilr) which yields seven isometric log ratios (one less than the number of critical oxides) that preserve the 304 305 geometric properties of the raw data and have a non-singular covariance matrix (Boschetty et al., 2022; Egozcue et al., 2003; Equation 2). The ilr is calculated using the "ilr" function in the 306 "compositions" package (Boogaart et al., 2023) of R (R Core Team, 2023) as follows: 307

308

309 
$$ilr(x) = \sqrt{\frac{i}{i+1} ln\left[\frac{g(x_1, \dots, x_i)}{x_{i+1}}\right]}, i = 1, 2, \dots, D-1 \quad (2)$$

310

where, x is a raw melt compositional analysis, i is a specific part, D is the total number of parts (8, the number of critical oxides), and  $g(x_i)$  is the geometric mean of the parts of x given by:

314 
$$g(x_i) = \left(\prod_{i=1}^n x_i\right)^{\frac{1}{n}}$$
(3)

315

Next, using the newly ilr-transformed data, a Euclidian distance matrix is calculated for theinput dataset:

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319 
$$d_E = \sqrt{\sum_{i=1}^{n} (x_i - y_i)^2}$$
(4)

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where d<sub>E</sub> is the Euclidean distance matrix, i is the number of parts (7, the number of isometric
log ratios from Equation 2), and x and y are the two compositional analyses being considered.
The entire Euclidian distance matrix is normalised between 0 (closest) and 1 (furthest)
according to:

324 
$$d_{E(norm)} = \frac{d_E^i - \min(d_E)}{\max(d_E) - \min(d_E)}$$
(5)

325 where  $d_E^i$  is a single entry in the Euclidian distance matrix  $(d_E)$ .

To predict whether an unknown (natural) entry in the input dataset is OPAM<sub>SAT</sub> or OPAM<sub>UNSAT</sub> we use the n closest experimental points according to the scaled distance matrix  $(d_{E(norm)})$ . The sensitivity to n is assessed in the results section. The n closest scaled distances are used in a weighted equation which returns the probability of OPAM saturation ( $P_{SAT}$ ): 330

331 
$$P_{SAT} = \sum_{i=1}^{n} \left[ \left( 1 - d_{E(norm)}^{i} \right) * \frac{V_{SAT}}{n} \right]$$
(6)

332

333 where  $V_{SAT}$  is the numerical saturation value of each of the n points ( $V_{SAT} = 0$  for an 334 OPAM<sub>UNSAT</sub> melt;  $V_{SAT} = 1$  for an OPAM<sub>SAT</sub> melt). If Equation 6 returns  $\geq 0.5$  then the

unknown value is considered more likely to be OPAM<sub>SAT</sub> and if it returns <0.5 then it is 335 considered more likely to be OPAM<sub>UNSAT</sub>. The above mathematics reduces to a quantitative 336 question, which is posed for each compositional datapoint of unknown saturation state in the 337 338 input dataset in turn: "of the n closest experimental points of known saturation state to our datapoint of unknown saturation state, how many of those n experiments are OPAM<sub>UNSAT</sub> and 339 how many are OPAM<sub>SAT</sub> in scaled multidimensional chemical space"? Equations 2 and 3 340 341 transform the experimental and unknown data, Equations 4 and 5 define distance using the newly transformed data, and Equation 6 weights the n experiments by their distances to the 342 343 unknown melt. This approach has two advantages: (i) unevenly distributed experiments in P-T-X space are accounted for by Equation 6, giving a smaller weight to chemically distal 344 experiments; (ii) Equation 6 has a probabilistic meaning such that an unknown melt sitting in 345 346 a part of multidimensional chemical space which is occupied by many OPAM<sub>SAT</sub> melts will return a high saturation probability, whereas a diffuse mixture of OPAM<sub>SAT</sub> and OPAM<sub>UNSAT</sub> 347 experiments will return a lower saturation probability. 348

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## 350 Initial variable selection, regression methodology, and model validation

To derive an improved OPAM thermobarometer we regress new T and P equations 351 using our OPAM<sub>SAT</sub> experimental compilation (n = 249; Supplementary Table 1a). Regression 352 353 identifies the "best fit" equation that describes a dataset by minimising a cost function, in this 354 case (and most typically) the sum of the squared residuals. We follow the reverse approach to 355 multilinear regression which initially includes all possible variables in the fitting process, fits a function, identifies the variable of lowest significance, and removes it from the fit. This 356 357 process is performed iteratively using the "lm" function in the "stats" package of R (R Core Team, 2023) until all variables and coefficients in the fit are significant. The t-value and p-358 359 value are the common measures of significance, where the p-value is the probability that any observed relationship is due to chance and the t-value is the ratio between a fit coefficient and its standard error. Fit coefficients with higher t-values (e.g., >3) and low p-values (<0.01) indicate a good fit to the data. We use a stopping rule such that all variables must have t-values >3 to be considered a converged fit. A small number of additional outliers (P = 6; T = 3) were identified during the fitting process, based on residuals 1.5 times higher than the interquartile range of all residuals, and these were omitted from the regression (e.g., Blundy, 2022).

366 The starting variables to initialise the iterative regression process should logically describe the OPAM cotectic position as a function of P-T-X. The OPAM cotectic is P-sensitive, 367 368 largely due to expansion of the clinopyroxene stability field at higher P (Bender et al., 1978; Grove et al., 1992; Lissenberg et al., 2019), with increasing melt H<sub>2</sub>O content also suppressing 369 plagioclase saturation to lower T at fixed P (Sisson and Grove, 1993; Yoder, 1965). CaO-MgO-370 Al<sub>2</sub>O<sub>3</sub>-SiO<sub>2</sub> (CMAS; Libourel et al., 1989; Presnall et al., 1978; Shi, 1992; Yang et al., 1996) 371 in addition to FeO (Shi, 1993), Na<sub>2</sub>O, K<sub>2</sub>O, and TiO<sub>2</sub> (Yang et al., 1996) are the major elements 372 which have previously been used to characterise the OPAM cotectic (i.e., the eight critical 373 oxides used for our OPAM saturation test). Increasing Cr<sub>2</sub>O<sub>3</sub> promotes clinopyroxene 374 saturation, which prompted the reformulation of the three Yang et al (1996) cotectic equations 375 376 by Voigt et al (2017). However, we omit Cr<sub>2</sub>O<sub>3</sub> from our fit routine due to a lack of consistent reporting in both experimental and natural data which would prevent its wide application and 377 robust testing. Oxides are normalised to 100 wt% (anhydrous) and converted to liquid cation 378 fractions (e.g.,  $X_{AlO_{1,5}}$ ) for fitting (see Table 1 of Putirka, 2008 for an example conversion). 379 Composite variables implicated by other studies of basaltic melt equilibria such as  $X_{SiO_2} * X_{TiO_2}$ , 380  $(X_{SiO_2})^2$  (Yang et al., 1996), Ca#  $(X_{CaO} / [X_{CaO} + X_{NaO_{0.5}}])$ , Al#  $(X_{AlO_{1.5}} / [X_{AlO_{1.5}} + X_{SiO_2}])$ 381 (Neave and Namur, 2022), and  $X_{CaO}/X_{AlO_{1.5}}$  (Voigt et al., 2017) were also included. The latter 382 383 positively correlates with clinopyroxene saturation T as well as the Mg# of the liquidus

clinopyroxene (Voigt et al., 2017), whereas alkalis modulate olivine stability (Grove et al.,
1992; Meen, 1990).

Around 10 points or more are required per fit variable (Harrell, 2015) meaning our 13 386 387 initial variables require around 130 experiments. Therefore, we are required to split our experimental dataset into two roughly equal parts: a "training dataset" (n = 126) to regress the 388 model and a "validation dataset" (n = 123) to recover independent performance statistics. We 389 390 partition entire experimental studies into each dataset, rather than splitting all data randomly, to assess the role of systematic error between experimental studies (see Discussion). We 391 392 achieve this using a splitting algorithm which repeatedly partitions studies randomly into the 393 training and validation dataset until the number of entries in each dataset are 125±5. There is 394 sufficient chemical overlap between the two datasets (Supplementary Fig. 1). Once the 395 thermometer and barometer are regressed using the training dataset, the validation dataset is 396 used to recover the following metrics of average model performance: the standard error estimate of the fit (SEE; Equation 7); the slope, intercept, and  $R^2$  of a plot of predicted (X) 397 398 versus experimental P or T (Y; Piñeiro et al., 2008); and the mean absolute error (MAE; 399 Equation 8).

400

401 
$$SEE = \sqrt{\frac{1}{N} \sum_{i=1}^{N} (predicted_i - experimental_i)^2}$$
(Equation 7)

402

403 
$$MAE = \frac{1}{N} \sum_{i=1}^{N} |predicted_i - experimental_i| \qquad (Equation 8)$$

404 Our thermobarometer and saturation test are presented as a straightforward R script 405 which reads from an input csv file to be populated with melt data. The R script, input csv file 406 template, and simple instructions for use (i.e., downloading R, opening and running the script) are provided in the Supplementary Material and on the code repository GitHib
(https://github.com/OJHiggins/HS2024\_OPAM). The R script output retains the original data
structure of the input csv file and manages unlimited additional descriptor columns (field
location, sample names, etc).

411

- 412 **RESULTS**
- 413

### 414 Independent test of the Yang et al (1996) OPAM thermobarometer

415 To independently test the performance of the Yang et al (1996) thermobarometer we use our OPAM<sub>SAT</sub> compilation minus the experimental studies named in Table 4 of Yang et al 416 417 (1996; n=199). The barometer calculations were made using the Yang et al (1996) 418 implementation of Baxter et al (2023) which applies the method of Hartley et al (2018) and so 419 also returns their probability of OPAM<sub>SAT</sub> (P<sub>F</sub>). The Yang et al (1996) thermometer is not applied in Baxter et al (2023) so we instead use the liquid cation output from Baxter et al (2023) 420 421 and calculate T ourselves using the Yang et al (1996) thermometer equation. The limit of  $H_2O$ 422 for the Yang et al (1996) thermobarometer is not officially stated but is probably capped at  $\sim 2$ wt% H<sub>2</sub>O based on the upper concentration in their calibration experiments. To inspect the 423 uncertainty of the Yang et al (1996) thermobarometer as a function of melt H<sub>2</sub>O, we calculate 424 the SEE (Equation 7) and MAE (Equation 8) for experiments with an  $H_2O \le H_2O_{max}$ . We vary 425 426 H<sub>2</sub>O<sub>max</sub> between 0 and 7.2 wt% H<sub>2</sub>O to yield an uncertainty metric of the Yang et al (1996) equations at every melt H<sub>2</sub>O content in the test dataset. Whilst melt H<sub>2</sub>O exceeding  $\sim 2$  wt% 427 would likely result in a significant drop in the Yang et al., (1996) thermobarometer 428 429 performance given the significant melt-compositional control of melt H<sub>2</sub>O (Fig. 1), we would expect their thermobarometer to reproduce experimental values close to the quoted 430 431 uncertainties at anhydrous and low-H<sub>2</sub>O conditions.

Of the 199 input OPAM<sub>SAT</sub> experiments in our Yang et al., (1996) test dataset, only 7 432 pass the Hartley et al (2018) saturation filter of  $P_F \ge 0.8$  which is 4% of all experiments or 13% 433 of anhydrous (0 wt% H<sub>2</sub>O) experiments. As outlined above, our OPAM<sub>SAT</sub> compilation 434 435 contains no obvious experimental outliers and so should be a robust test of the number of false negatives produced by an OPAM saturation classifier. Our observation is consistent with that 436 of Baxter et al (2023) where only 92 of their 250 OPAM experiments have a  $P_F \ge 0.8$  (37%) 437 and only 23 of these were conducted at >1 atm (9%). Hartley et al (2018) deliberately tuned 438 their P<sub>F</sub> to limit false positives which, in a binary classifier, will be counterbalanced by a higher 439 440 number of false negatives; whilst we acknowledge the important step forward by Hartley et al (2018) in the pre-screening of OPAM<sub>SAT</sub> melts before applying the Yang et al., (1996) 441 442 thermobarometer, the false negative rate is very high and likely underrepresented by Baxter et 443 al (2023).

444 The Yang et al., (1996) predicted T and experimental T show poor correspondence, with an increasing overestimation of T with increasing melt  $H_2O$  (Fig. 3a). The few OPAM<sub>SAT</sub> 445 446 experiments with  $P_F \ge 0.8$  (large symbols in Fig. 3a) generally lie close to the 1:1 line. Despite a strong relationship between the quality of Yang et al., (1996) T prediction and melt H<sub>2</sub>O, 447 several anhydrous experiments still show considerable and systematic T overestimation. The 448 relationship between uncertainty and melt H<sub>2</sub>O can be more clearly observed in Fig. 3b, where 449 450 both the SEE and MAE increase quasi-linearly between 0 wt% H<sub>2</sub>O (n = 53) and  $\leq$ 7.2 wt% 451 H<sub>2</sub>O (n = 199). At anhydrous conditions, the SEE is  $\pm 35$  °C and the MAE is  $\pm 29$  °C which is 452 more than the  $\pm 20$  °C quoted by Yang et al., (1996) but comparable to many widely applied melt-based and mineral-based thermometers (e.g., Putirka, 2008). However, the Yang et al., 453 454 (1996) calibration is theoretically applicable to melts with up to  $\sim 2 \text{wt}\%$  H<sub>2</sub>O based on the calibration dataset and is often applied in low-H2O, but not anhydrous, natural systems (e.g., 455 456 Iceland, the Galápagos; Hartley et al., 2018; Stock et al., 2018). As an example, the H<sub>2</sub>O of 457 matrix melts and nominally anhydrous minerals in basalts from the Galápagos span 0.06-0.77 458 wt% and 0.09-2.54 wt% respectively (Gleeson et al., 2021; Fig. 3b). At the upper limit of these 459 natural melt H<sub>2</sub>O contents, the SEE is  $\pm 69$  °C and the MAE is  $\pm 55$  °C in addition to the 460 systematic uncertainty highlighted in Figure 3a.

The Yang et al., (1996) predicted P and experimental P show the same relationship as 461 for T (Fig. 3c), with a systematic overestimation of predicted P with increasing melt  $H_2O$ . 462 463 Again, the few experiments with  $P_F \ge 0.8$  (large symbols in Fig. 3c) generally lie close to the 1:1 line. Contrary to the suggestion of Baxter et al (2023), we observe no strong relationship 464 465 between P<sub>F</sub>, offset between predicted and experimental P, and melt alkali content (Na<sub>2</sub>O+K<sub>2</sub>O [wt%]; square symbols in Fig. 3c). We question the utility of their melt alkali filter given that 466 changing the maximum permissible Na<sub>2</sub>O+K<sub>2</sub>O of an input melt from 6.16 wt% to 5 wt% 467 468 resulted in a reduction of only  $\pm 0.03$  kbar in the MAE according to Table 1 of Baxter et al (2023). The Yang et al., (1996) MAE of  $\pm 1.13$  kbar quoted by Baxter et al (2023) is a 469 substantial underestimate; for anhydrous melts using our independent dataset the SEE is  $\pm 3.29$ 470 471 kbar and the MAE is  $\pm 2.54$  kbar, more than double their quoted value. These results underscore the significant effect that filtering large quantities of OPAM<sub>SAT</sub> experiments can have on the 472 quoted uncertainty. For H<sub>2</sub>O contents in the range of melts from the Galápagos, the SEE and 473 MAE rise to  $\pm 4.96$  and  $\pm 3.83$  kbar respectively (Gleeson et al., 2021; Fig. 3d). In short, the 474 475 Yang et al (1996) barometer suffers from significant systematic (Fig. 3c) and random (Fig. 3d) 476 uncertainty when tested by an independent OPAM<sub>SAT</sub> dataset, even at anhydrous conditions.

477

### 478 **OPAM**<sub>SAT</sub> prediction

To assess the performance of our binary classifier for  $OPAM_{SAT}$  melts as a function of n, we treat each experiment from the saturation dataset (n = 498) in turn as an unknown and predict the saturation state ( $OPAM_{UNSAT}$  or  $OPAM_{SAT}$ ). We score our approach using the number of true positives (TP), false positives (FP), true negatives (TN), and false negatives
(FN). The combination of these metrics defines the accuracy (A) and precision (P) of the
algorithm, as well as the true positive rate (TPR), false positive rate (FPR), true negative rate
(TNR), and false negative rate (FNR) (Equation 9a–f; Fig. 4):

486

487 
$$TPR = \frac{TP}{TP + FN}$$
 (Equation 9a)

$$FPR = \frac{FP}{FP + TN}$$
 (Equation 9b)

$$TNR = 1 - FPR \qquad (Equation 9c)$$

$$FNR = 1 - TPR \qquad (Equation 9d)$$

491 
$$A = \frac{TP + TN}{TP + TN + FP + FN}$$
 (Equation 9e)

492 
$$P = \frac{TP}{TP + FP}$$
 (Equation 9f)

493

The predictive performance of our OPAM binary classifier varies as a function of n, where n is the number of points considered to define the local state of OPAM saturation in multidimensional space, and can be divided into three broad behaviours (grey shading in Fig. 497 4; i-iii):

Predictive performance is good (accuracy=0.81-0.74; precision=0.91-0.78) but erratic 498 (i) or "spiky" ( $2 \le n < 10$ ). The FPR increases with n but remains broadly low (0.06-0.20). 499 Classifier performance begins to drop off as n increases, with accuracy (0.75-0.63) and 500 (ii) precision (0.81–0.70) decreasing over a wide range of n (11 $\le$ n<40). The FNR increases 501 and the TPR decreases in tandem but the FPR and TNR both remain stable. 502 All performance metrics largely plateau at  $n \ge 40$ , except for an increasing FNR (0.5-503 (iii) 0.68) and decreasing TPR (0.47-0.32). 504

505 Using the calculated performance metrics, our preferred n value is 11, coinciding with 506 the end of the spiky behaviour whilst maximising predictive performance (ii; Fig. 4; 507 Supplementary Table 3). At n=11, our approach has an accuracy of 0.75, a precision of 0.78, a 508 TPR of 0.7, and a FPR of 0.2. Therefore, when assessing the saturation state of a natural melt 509 of unknown saturation state, the saturation states of the 11 closest experimental melts are considered. Our approach is not a "model" sensu stricto, but rather a probability map of local 510 compositional space detailing the likelihood that a given melt is OPAM<sub>SAT</sub>. Theoretically, our 511 method can continuously evolve as experiments are added, particularly at the margins of the 512 513 OPAM convex hulls. However, whilst the performance metrics of our model offer improvements over the original saturation test of Hartley et al (2018), our model does not 514 unequivocally identify OPAM<sub>SAT</sub> melts (Fig. 4). It is highly likely that the diffuse overlap in 515 516 chemical space between OPAM<sub>SAT</sub> and OPAM<sub>UNSAT</sub> melt compositions (e.g., Fig. 2) makes a 517 perfect mathematical solution impossible. As such, our saturation test should be used in 518 combination with petrographic and geochemical indicators where possible and should always 519 be usurped by these methods. The key advantage of our approach is that it acknowledges the 520 local diffuseness of the multidimensional space occupied by OPAM<sub>SAT</sub> and OPAM<sub>UNSAT</sub> melts 521 when calculating  $P_{SAT}$ .

522

## 523 New OPAM thermobarometer performance

524 The successful fits for P (Equation 10a) and T (Equation 10b) of OPAM<sub>SAT</sub> melts as a 525 function of the melt composition are as follows:

526

527 
$$P(kbar) = 90.74 - (257.77 * X_{SiO_2}) + (33.99 \cdot X_{AlO_{1.5}}) + (106.83 * X_{MgO})$$

528 
$$-(194.34.X_{Ca0}) - (50.55.X_{Na0_{0.5}}) + (189.62.(X_{Si0_2})^2) (Equation 10a)$$

530 
$$T(^{\circ}C) = -9201 + (10331 * X_{SiO_2}) + (7128 * X_{FeO}) + (10004 * X_{MgO}) + (7691 * X_{FeO})$$

531 
$$X_{NaO_{0.5}}$$
 + (9658 \*  $X_{KO_{0.5}}$ ) + (9574 \*  $Al\#$ ) + (19921 \* [ $X_{SiO_2} * X_{TiO_2}$ ]) + (1303 \*

- 532  $[X_{Ca0}/X_{AlO_{1,5}}]$  (Equation 10b)
- 533

where *X* followed by the oxide subscript denotes liquid cation fractions calculated using the method outlined in Table 1 of Putirka (2008) on an anhydrous basis, all Fe is considered as FeO, and Al#  $(X_{AlO_{1.5}}/[X_{AlO_{1.5}} + X_{SiO_2}])$  is on a molar basis. Coefficient standard errors, pvalues, and t-values are listed in Supplementary Table 4, with scatter assumed to include normally distributed error (e.g., analytical and experimental uncertainty).

539 Performance metrics for our new OPAM thermometer and barometer are calculated using the same validation dataset (n = 123). The new thermometer has a SEE of  $\pm 35$  °C and an 540 MAE of  $\pm 30$  °C, with low systematic offset relative to the 1:1 line (grey line in Fig. 5a) shown 541 by slope (1.00) and intercept (7 °C) values close to 1 and 0, respectively (Fig. 5a). For the 542 543 closest comparison to the quoted uncertainty of  $\pm 20$  °C for the Yang et al (1996) thermometer, the MAE of our own training dataset is  $\pm 21$  °C although our MAE is applicable for a much 544 wider range of melt H<sub>2</sub>O. The new barometer has a SEE of  $\pm 1.13$  kbar and an MAE of  $\pm 0.9$ 545 546 kbar (Fig. 5b). As with our thermometer, the systematic offset relative to the 1:1 line of our new barometer is modest (slope = 0.98, intercept = 0.17 kbar). It is difficult to compare 547 uncertainty with the Yang et al (1996) barometer as we have asserted that truly independent 548 test statistics were not produced by Hartley et al (2018) or Baxter et al (2023; Fig. 3). However, 549 the quoted uncertainty of Baxter et al (2023) on their partially self-validated and heavily filtered 550 551 test dataset is an MAE of  $\pm 1.13$  kbar. Our performance statistics are calculated using an independent validation dataset and are applicable to a range of melt H<sub>2</sub>O contents spanning 0 552 to 7.7 wt%. Furthermore, the MAE of our new P calibration is similar to the quoted MAE value 553 554 of the gabbronorite (CHOMPI) melt-based barometer of Blundy (2022; MAE=±1.3 kbar) and the SEE of the clinopyroxene-liquid barometer of Neave and Putirka (2017; SEE=±1.4 kbar).
Therefore, our calibration can complement other independent models and, indeed, produces
comparable results (see Discussion).

558 Given the strong control of melt H<sub>2</sub>O on the uncertainty of the Yang et al., (1996) thermobarometer performance (Fig. 3), we also assess any systematic uncertainty in P and T559 residuals (experimental P or T minus predicted P or T) for our new OPAM thermobarometric 560 equations (Equation 10a, Equation 10b) as a function of melt H<sub>2</sub>O. As the melt H<sub>2</sub>O 561 concentration is universally higher in volcanic arcs (Wallace, 2005) compared to mid-ocean 562 ridge basalts (MORB) and ocean island basalts (OIB; Michael, 1995; Sobolev and Chaussidon, 563 1996), we also categorise experimental studies as either arc or MORB/OIB where these labels 564 are unambiguous (Fig. 5c, Fig. 5d). When using these categories there is a tendency for the T565 566 of OIB and MORB experiments to be underestimated, with an average residual of +24 °C compared to a mean of +7 °C for arc experiments (Fig. 5c). For P, there is no clear relationship 567 between the calculated residuals and geodynamic setting, with relatively sharp distributions 568 569 centred on zero (Fig. 5d). Overall, compared to the very large effect observed for the Yang et al., (1996) equations (Fig. 3), melt H<sub>2</sub>O only weakly dictates systematic uncertainty in T (Fig. 570 5c) and has little effect on systematic uncertainty in P (Fig. 5d) for our new OPAM 571 thermobarometer. 572

573

#### 574 **DISCUSSION**

575

### 576 Perspectives on thermobarometer performance and usability

577 The calibration and testing of our improved OPAM barometer (Equation 10a) and 578 thermometer (Equation 10b) as a function of OPAM<sub>SAT</sub> melt composition has highlighted three 579 considerations for future thermobarometers.

In a correlation matrix between SiO<sub>2</sub>, TiO<sub>2</sub>, Al<sub>2</sub>O<sub>3</sub>, FeO, MgO, CaO, Na<sub>2</sub>O, K<sub>2</sub>O, T, 580 and P using our newly compiled OPAM<sub>SAT</sub> dataset (n=249; Supplementary Fig. 2), P is the 581 third most correlated variable with T and T is the most correlated variable with P (correlation 582 583 coefficient = 0.51). The original Yang et al., (1996) thermometer regresses T as a function of OPAM<sub>SAT</sub> melt composition and P, in contrast to Equation 10b which is a P-independent 584 thermometer. Whilst the addition of a *P* term undoubtedly leads to an improved fit for many 585 586 thermometers (e.g., Neave and Putirka, 2017; Putirka, 2016, 2008), the incorporation of a P term makes it difficult to deconvolve whether any relationship between P and T is 587 588 methodological or thermodynamic. In other words, can P and T be considered two truly independent variables? The strong correlation between P and T is a thermodynamic 589 590 consequence for high-melt-fraction, hydrous experiments close to the liquidus due to the strong 591 dependency of volatile solubility and P on liquidus T (Bohlen et al., 1982). However, in lower-592 melt-fraction experiments further from the liquidus, some of the correlation between P and Twill also be partly due to the methodological tendency to perform high-T experiments at lower 593 594 P and vice versa. This tendency relates to the need for experimentalists to have large melt pools to analyse in their experimental charges, requiring a methodological consideration of liquidus 595 596 T which is a function of P. By applying P-dependent thermometers and T-dependent barometers in natural systems, this experimental bias may consequently be introduced into 597 598 geological interpretations of natural systems, leading to spurious correlations between the two 599 variables. Therefore, we advise that P and T remain purely independent when calibrating 600 thermobarometers, or that both dependent and independent equations are provided where alternative constraints on P or T are available (e.g., Putirka, 2016). 601

602 Our binary classifier to determine the probability that a melt is  $OPAM_{SAT}$  highlights 603 the need for new standards in how binary classifiers are benchmarked and tested between 604 studies. In the case of testing thermobarometric equations (e.g., Equation 10a, Equation 10b), 605 the SEE (Equation 2) and MAE (Equation 3) are standardised metrics which can robustly 606 compare different thermobarometer performance between studies (e.g., Fig. 3, Fig. 5). In contrast, testing classifiers requires a clear description of prior assumptions, akin to the prior 607 608 assumption term described in the equation of Bayes Theorem. For example, the performance metrics for our new binary classifier outlined in Figure 4, such as the accuracy and precision, 609 are valid for the prior assumptions that: (i) the experimental melt is in equilibrium with between 610 611 one and three of olivine + plagioclase + clinopyroxene, with or without Fe-oxides; (ii) the melt of unknown saturation state, be it a newly performed experimental melt or a natural melt 612 613 composition, passes the OPAM convex hull test. The latter assumption is determined mathematically by our model and marked in the output csv files generated by our R script, yet 614 the former requires at least some petrographic or geochemical knowledge from the user prior 615 616 to model application. Consistently stating these prior assumptions is vital if the performance 617 of future new and improved OPAM classifiers are to be fairly and consistently compared with existing models. 618

619 The residuals for experimental P or T minus predicted P or T of our new models are generally centred at or close to zero and there is little systematic uncertainty identified by the 620 621 fit between predicted and measured values (Fig. 5). Despite this, there is clear clustering of individual studies on either side of the 1:1 line between predicted and experimental P or T (Fig. 622 623 5), demonstrating that inter-study effects partly control the thermobarometer uncertainty 624 according to the validation dataset statistics. This intra-study clustering effect is often apparent 625 in machine learning thermobarometers where multiple train-validation dataset splits show the effect that partitioning of studies between these two datasets can have on model performance 626 627 (Higgins et al., 2021a; Jorgenson et al., 2022; Petrelli et al., 2020). Understanding, constraining, and correcting for these inter-study or inter-laboratory effects is central to 628

developing future barometers which can reliably predict global datasets with uncertaintiescomparable to those quoted for the original regression.

631

## 632 Application to two geophysically monitored volcanic eruptions

The *P*-*T*-*X* of magma during storage and ascent plays a leading role in modulating 633 eruptive dynamics (Higgins and Caricchi, 2023; Popa et al., 2021). Therefore, quantitative 634 635 tools to describe pre-eruptive magmatic *P*-*T*-*X*, such as the new thermobarometric equations presented here, can be useful to: (i) petrologically monitor active volcanoes during protracted 636 637 eruptions, particularly to compare evolving geochemical and geophysical signals (Hartley et 638 al., 2018; Longpré et al., 2014; Magee et al., 2018, 2020; Saunders et al., 2012; Stock et al., 2018); (ii) understand medium-term and long-term evolution in magma storage conditions by 639 640 studying multiple temporally constrained volcanic deposits (Bouvet de Maisonneuve et al., 641 2021; Forni et al., 2018; Giordano and Caricchi, 2022; Higgins et al., 2021a; Smithies et al., 2023; Weber et al., 2020). To assess the utility of our new thermobarometer for studying 642 643 volcanic systems, we apply our calibration to the OPAM<sub>SAT</sub> melts of two case study eruptions (the 2015 eruption of Wolf Volcano, Galápagos; the 2014-2015 Holuhraun eruption, Iceland). 644 Both eruptions have been geophysically, geochemically, and petrographically studied in detail, 645 providing a means of independent validation and comparison with our updated 646 647 thermobarometric predictions. For the two case study eruptions we present P and T predictions 648 with our new thermobarometer for both unfiltered (all samples which pass the OPAM convex hull test; n = 76 of 76 for Wolf and n = 102 of 114 for Holuhraun) and filtered (P<sub>SAT</sub>>0.5; n =649 70 of 76 for Wolf and n = 45 of 102 for Holuhraun) data. In doing so we keep track of any 650 651 compositional bias, if present, introduced to our P and T predictions by the probabilistic filtering. The full dataset of melt compositions for both volcanic systems are presented in 652 653 Supplementary Table 5.

654

655 The 2015 eruption of Wolf Volcano, Galápagos

The Galápagos Archipelago is one of the most volcanically active regions on Earth, 656 657 with eruptions occurring at a repose period of ~2 years (Bernard et al., 2019). The archipelago is the surface manifestation of the Galápagos plume which impinges on the base of ~10 Ma-658 old, ~11 km-thick oceanic crust (Feighner and Richards, 1994). Wolf Volcano, on the north of 659 660 Isabela Island, is an active volcano in the western Galápagos (Geist et al., 2005). The 2015 eruption produced ~0.01 km<sup>3</sup> of monotonous basaltic lava and tephra from circumferential and 661 662 radial vents (Bernard et al., 2019; Stock et al., 2020, 2018). We apply our new OPAM thermobarometer to the 2015 whole rock and matrix glass measurements from Wolf Volcano 663 (Stock et al., 2020, 2018). All petrological P (kbar) estimates are converted to depth (km) 664 665 relative to the surface using a polynomial fit to the Hawaiian ocean crust *P*-depth relationship of Putirka (1997; Equation 11;  $R^2 = 0.96$ ) as per Stock et al. (2018): 666

667

668 
$$Depth(km) = -0.02086^2 P(kbar) + 3.36352 P(kbar) + 0.46170$$
 (Equation 11)  
669

670 All whole rock and matrix glass data bar six measurements return a strong indicator for OPAM saturation ( $P_{SAT}$ ) using our new saturation test (0.81±0.16; mean±1 $\sigma$ ), agreeing with 671 petrographic and thermodynamic modelling results which attest to three-phase saturation 672 673 (Stock et al., 2020). Due to the strong indication for OPAM saturation in all samples, unfiltered and filtered ( $P_{SAT} \ge 0.5$ ) melts record an identical *T* of 1151±15 °C (mean±1 $\sigma$ ) using Equation 674 675 10b (Fig. 6a). The restricted whole rock compositions result in a tighter range of T (1172 $\pm$ 2 °C) compared to matrix glass (1144±11 °C). The overall OPAM<sub>SAT</sub> T range is in strong 676 agreement with clinopyroxene thermometry for glomerocrystic (1164±11 °C), phenocrystic 677 (1151±9 °C), and tephra-hosted (1164±15 °C) clinopyroxene from Stock et al. (2018; Fig. 6a). 678

New OPAM P estimates for unfiltered Wolf samples (n = 76), calculated with Equation 679 10a and converted to depth (km) using Equation 11, are 8.6±2.2 km for matrix glass and 680 9.6±0.8 km for whole rock. The OPAM<sub>SAT</sub> depths compare well with estimates derived using 681 682 the Yang et al., (1996) barometer and  $P_F \ge 0.8$  filter of Hartley et al., (2018; 9.7±2.8 km; A and B vs L in Fig. 6b). Filtering our new OPAM<sub>SAT</sub> depths (P<sub>SAT</sub>≥0.5; n=70) yields near identical 683 values for matrix glass (8.6±2.3 km; C in Fig. 6b) and identical values for whole rock (9.6±0.8 684 685 km; D in Fig. 6b) compared to the unfiltered results. The three different textural sub-classes of clinopyroxene identified by Stock et al. (2018) record appreciably wider and systematically 686 687 deeper storage depths (glomerocrystic =  $9.7\pm2.5$  km, phenocrystic =  $11\pm2.8$  km, tephra-hosted =  $10\pm3.1$  km; mean $\pm1\sigma$ ; I, J, and K in Fig. 6b), where P is calculated using clinopyroxene-melt 688 barometry (Neave and Putirka, 2017). Inversion of interferometric synthetic aperture radar 689 690 (InSAR) measurements indicates a shallow source at ~1 km depth (E in Fig. 6b) and deep 691 sources at ~6.1 to 8.8 km (F, G, and H in Fig. 6b). Encouragingly, our new OPAM barometer 692 produces depth results which are consistent with the depth of the deep magma source identified 693 by InSAR inversion (F, G and H in Fig. 6a). Therefore, we corroborate the findings of Stock et al (2018) that magmas feeding the 2015 Wolf eruption were sourced predominantly from 694 695 the lower crust and not the shallow-crustal InSAR source at ~1 km (E in Fig. 6b). In detail, the mean values of all petrographic indicators of magma storage depth (pressure) are 696 697 systematically higher than those from the InSAR inversions. As alluded to by Stock et al 698 (2018), we interpret this offset as InSAR signals returning deformation specifically at the top 699 of a vertically protracted magma storage region whereas petrological estimates are recording 700 crystallisation and/or equilibration deeper within that same storage region. We note that the 701 mean depths recovered using our new OPAM barometer are most consistent with the InSAR 702 results (Fig. 6b).

#### 704 *The 2014-2015 Holuhraun eruption, Iceland*

The island of Iceland is the surface expression of a plume-ridge interaction between the 705 Mid-Atlantic Ridge and a mantle plume, resulting in anomalously thickened oceanic crust (15-706 707 44 km; Darbyshire et al., 2000; Jenkins et al., 2018) due to an increased mantle potential temperature that enhances melt supply. A variable mantle melt flux has been posited as a 708 709 control on magma storage pressures on Iceland (Baxter et al., 2023), producing highly variable 710 depths for the erupted melts from various volcanic zones. The 2014-2015 Holuhraun eruption was a basaltic fissure event that originated from within the Bárðarbunga volcanic system in 711 central Iceland, producing >1.4 km<sup>3</sup> of lava and tephra (Pedersen et al., 2017). The eruption 712 713 represents a remarkable case study for testing models of magma storage conditions due to 714 extensive pre- and syn-eruptive geochemical, seismic, and ground deformation datasets 715 (Ágústsdóttir et al., 2016; Gauthier et al., 2016; Gudmundsson et al., 2016; Halldórsson et al., 716 2018; Hartley et al., 2018; Sigmundsson et al., 2015). Further, it allows us to directly verify our new OPAM thermobarometer against the results of Hartley et al. (2018), who developed 717 718 their implementation of the Yang et al., (1996) barometer to assess the evolution of magma storage depths during the 6-month Holuhraun eruption. We apply our new OPAM 719 720 thermobarometer to melt inclusion and matrix glass measurements from the Holuhraun eruption and, for direct comparison with Hartley et al. (2018), our P (kbar) estimates are 721 722 converted to depth (km) relative to the surface using a fixed crustal density of 2.86x10<sup>6</sup> kg/m<sup>3</sup>. 723 Unlike samples from Wolf Volcano (Fig. 6a), Holuhraun melts show a lower and more diverse range of  $P_{SAT}$  using our new saturation test (0.46±0.19; mean±1 $\sigma$ ; Fig. 7a). Considering 724 all compositions unfiltered for  $P_{SAT}$  (n = 102), T ranges from 1135 to 1254 °C (1193±31 °C) 725 726 except for a single outlier at 1098 °C (Fig. 7a). Three P values, not plotted but included in the P statistics, return negative P estimates. When our new OPAM P results are converted to depth, 727 728 unfiltered matrix glass (7.1±1.7 km; A in Fig. 7b) and melt inclusions (8.8±4.4 km; B in Fig.

729 7b) overlap. Using a filter of  $P_{SAT} \ge 0.5$  (n = 45), only 2 of 9 matrix glass values remain (5.2 730 km and 8.7 km; C in Fig. 7b), whereas melt inclusions cover a similar range to the unfiltered data (9.3±4.7 km; D in Fig. 7b). The wide spread of the melt inclusion depths is consistent with 731 732 clinopyroxene-liquid barometry (Neave and Putirka, 2017) applied to clinopyroxene crystals that were screened for textural equilibrium (Halldórsson et al., 2018; 8.2±5 km; G in Fig. 7b). 733 The unfiltered depth results of the Yang et al (1996) equations implemented by Hartley 734 735 et al (2018) are similar to our unfiltered OPAM depths for both matrix glass (6.6±2.4 km; H in Fig. 7b) and melt inclusions (10.2±4.2 km; I in Fig. 7b). However, the Hartley et al (2018) 736 737 OPAM depths filtered for  $P_f \ge 0.8$  are notably deeper and less dispersed for melt inclusions (11.9±3.5 km; K in Fig. 7b) compared to their unfiltered values, whereas matrix glass (6.8±2.4 738 km; J in Fig. 7b) are near identical to their unfiltered values. This observation shows that the 739 740 saturation filter of Hartley et al (2018) has a much stronger effect on skewing P to higher values 741 for certain melt compositions than our new saturation test (Fig. 7b). Given the very-high false negative rate of the saturation test of Hartley et al (2018) there may be a compositional bias 742 743 which is artificially deepening values as a result of applying their probabilistic filter. As with observations from Wolf Volcano, our OPAM barometry results are consistent with independent 744 geophysical estimates of magma storage via the range of pre-eruptive seismicity (3-9 km; 745 Sigmundsson et al., 2015) and the most common earthquake hypocentral depth (6±1 km; 746 747 Ágústsdóttir et al., 2016), particularly for matrix glass. Therefore, we concur with Hartley et al 748 (2018) that the 2014-2015 Holuhraun eruption represents a case study in which petrological 749 and geophysical data can be reconciled.

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## 751 CONCLUSIONS

We have calibrated a new barometer (Equation 10a) and thermometer (Equation 10b) for melts equilibrated with the mineral assemblage of olivine + plagioclase + augitic 754 clinopyroxene  $\pm$  Fe-oxides, using an expanded dataset of anhydrous and hydrous experiments 755 (Fig. 1, Fig. 2). The necessity to calibrate an updated OPAM thermobarometer arises from the most widely used calibration (Yang et al., 1996) being unable to reliably reproduce an 756 757 independent dataset of experimental pressures and temperatures (Fig. 3). Previous attempts to calculate the true predictive performance of the Yang et al (1996) barometer by Hartley et al 758 (2018) and Baxter et al., (2023) were hindered by: (i) testing datasets which were not 759 760 independent of the Yang et al (1996) calibration dataset; and (ii) a statistical filtering process for OPAM saturation which removed large swathes of experiments that were demonstrably 761 762 suitable for OPAM barometry. Therefore, the mean absolute error (MAE) of ±1.13 kbar quoted by Baxter et al., (2023) for the Yang et al (1996) barometer is a substantial underestimate. 763 764 Rather, our independent test dataset yielded a MAE of  $\pm 2.54$  kbar and a SEE of  $\pm 3.29$  for the 765 Yang et al (1996) barometer which increases significantly with melt H<sub>2</sub>O (Fig. 3d). The Yang 766 et al., (1996) thermometer performs passably at anhydrous conditions (SEE =  $\pm 35$  °C; MAE =  $\pm 29$  °C) but the uncertainty increases dramatically with increasing melt H<sub>2</sub>O even at 767 768 concentrations well within the range of their original calibration dataset (Fig. 3b).

769 In contrast with the Yang et al (1996) P and T equations, our new OPAM barometer and thermometer accurately reproduce an independent validation dataset of P and T, with a 770 SEE of  $\pm 1.1$  kbar and  $\pm 35$  °C, respectively, and low systematic offset between measured and 771 772 predicted values (Fig. 5a, Fig. 5b). Importantly we also see no significant systematic P offset 773 as a function of H<sub>2</sub>O (Fig. 5d) up to 7.7 wt% in OPAM<sub>SAT</sub> melts (Fig. 1), and only minor T 774 offset which is within the average thermometer uncertainty (Fig. 5c). We also provide a binary classifier to predict the probability that a melt of unknown saturation state is OPAM saturated 775 776 based on a weighted distance matrix in multidimensional chemical space. Our new method improves on existing approaches, particularly with respect to false negatives (falsely predicting 777 that a melt is not OPAM<sub>SAT</sub>), yielding an accuracy of 0.75 and a precision of 0.78 (Fig. 4). The 778

new binary classifier of OPAM saturation should be used in conjunction with detailed
petrographic and geochemical analysis where possible, owing to the overlap of OPAM<sub>SAT</sub> and
OPAM<sub>UNSAT</sub> melts in multidimensional chemical space which precludes the development of a
perfectly efficient binary chemical classifier (Fig. 2). Our thermobarometer and saturation test
are presented as a simple R script (R Core Team, 2023) which takes user input via a csv file of
melt compositional data.

785 To test our implementation, we used two recent volcanic eruptions which have been petrographically, geochemically, and geophysically well studied (the 2015 eruption of Wolf 786 787 Volcano, Isabela Island, Galápagos; the 2014-2015 Holuhraun eruption, Iceland). In both cases, our updated OPAM barometer is consistent with existing, independent petrological and 788 geophysical magma depth estimates but provides constraints with the lowest uncertainties to 789 790 date (Fig. 6, Fig. 7). Hence, given the relative ease and rapidity of whole-rock and glass analysis 791 afforded by modern sample preparation and analytical equipment, our new thermobarometer 792 can complement syn-eruptive geophysical measurements to aid hazard management during on-793 going volcanic crises (Re et al., 2021). Additionally, our calibration may have sufficient P and T resolution (SEE of  $\pm 1.1$  kbar and  $\pm 35$  °C, respectively) to identify discrete changes in magma 794 795 storage conditions when studying the time-integrated stratigraphic record of past eruptions.

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## 797 FIGURE CAPTIONS

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Fig.1: (a) Temperature (°C) versus pressure (kbar); (b) melt MgO (wt%) versus melt CaO/Al<sub>2</sub>O<sub>3</sub>; and (c) melt SiO<sub>2</sub> (wt%) versus melt Na<sub>2</sub>O+K<sub>2</sub>O (wt%) for OPAM<sub>SAT</sub> equilibrium experiments, colour contoured for melt H<sub>2</sub>O (wt%). Grey symbols are hydrous experiments for which a H<sub>2</sub>O content is not reported. Diamonds are experiments from studies used to calibrate the Yang et al (1996) cotectic equations. The black curvilinear boundary in (c) is the alkalinesubalkaline transition boundary (Irvine and Baragar, 1971). Melt compositions are normalised
to 100 wt% anhydrous with all Fe as FeO.

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807 Fig. 2: (a) Melt SiO<sub>2</sub> (wt%) versus melt  $Al_2O_3$  (wt%); and (b) melt SiO<sub>2</sub> (wt%) versus melt MgO (wt%) for OPAM<sub>SAT</sub> and OPAM<sub>UNSAT</sub>. OPAM<sub>SAT</sub> experiments were compiled for this 808 study and are presented in Supplementary Table 1a. OPAM<sub>UNSAT</sub> experiments (olivine ± 809 810 plagioclase  $\pm$  augitic clinopyroxene  $\pm$ Fe-oxides) are from the compilation of Weber and Blundy (2023; Supplementary Table 2a). The OPAM convex hull is a polygon connecting the 811 812 outermost OPAM-saturated points in X-Y space with  $\pm 5\%$  relative added to account for analytical uncertainty (black field in both figures). Melts from the 2015 eruption of Wolf 813 814 Volcano (Galápagos; Stock et al., 2018, 2020) and the 2014-2015 Holuhraun eruption (Iceland; 815 Hartley et al., 2018) are also plotted (see Discussion; Supplementary Table 5). All experimental 816 and natural melt compositions are plotted normalised to the 8 critical oxides used for the OPAM saturation test. 817

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819 Fig. 3: Performance of the Yang et al (1996) thermometer and barometer for an independent 820 test of 199 OPAM<sub>SAT</sub> experiments from studies not used in the original Yang et al (1996) calibration. (a) The predicted temperature (°C) versus experimental temperature (°C). (b) The 821 822 SEE and MAE for temperature (°C) as a function of melt H<sub>2</sub>O (wt%). The line for n shows the 823 number of experiments with the corresponding melt H<sub>2</sub>O content on the x axis or less used to 824 calculate the SEE or MAE. (c) The predicted pressure (kbar) versus experimental pressure (kbar). (d) The SEE and MAE for pressure (kbar) as a function of melt H<sub>2</sub>O (wt%). In (a) and 825 826 (c) points are colour coded for melt H<sub>2</sub>O (wt%) and large symbols indicate experiments which pass the statistical filter of Hartley et al (2018) with a  $P_F \ge 0.8$  (see text for details). Solid line is 827 828 1:1 and dashed lines are the bounds of the quoted uncertainty for the Yang et al (1996) thermometer ( $\pm 20$  °C) and barometer ( $\pm 1.13$  kbar) respectively. In (b) and (d) the ranges of H<sub>2</sub>O (wt%) measured for Galápagos melts and calculated from nominally anhydrous minerals are shown (Gleeson et al., 2021).

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Fig. 4: Performance (Equation 9a-9f) of our statistical test for OPAM<sub>SAT</sub> melts as a function of n (the number of adjacent experiments considered in multidimensional space when testing if a natural melt is OPAM<sub>SAT</sub>). Grey boxes (i, ii, iii) highlight changes in test performance as a function of n and are discussed in text. The chosen n used for our statistical test is 11 (thick dashed line) at the start of region (ii).

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Fig. 5: The performance of our new OPAM thermobarometer using the compiled training and 839 840 validation data (Supplementary Table 1a). (a) Experimental temperature (°C) versus predicted 841 temperature (°C). (b) Experimental pressure (kbar) versus predicted pressure (kbar). (c) Experimental temperature (°C) minus predicted temperature (°C) and (d) experimental pressure 842 843 (kbar) minus predicted pressure (kbar) as a function of classified geodynamic setting (lines) and experimental melt H<sub>2</sub>O (wt%; points). In (a) and (b) summary statistics for training (grey 844 points) and validation (coloured points) datasets are provided in the plot area. Grey solid line 845 is the 1:1 line, black dashed lines are the bounds of the quoted validation SEE, and the black 846 solid line is a linear fit between predicted (x) and experimental (y) values. In (c) and (d) 847 848 geodynamic settings for experimental studies are either arc or ocean island basalt (OIB) plus 849 mid-ocean ridge basalt (MORB).

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Fig. 6: (a) Pressure (kbar) versus temperature (°C) using our new OPAM thermobarometer for
all matrix glass and whole rock data from the 2015 eruption of Wolf Volcano (Galápagos;
Stock et al., 2018). Data points are colour contoured for P<sub>SAT</sub> using our binary classifier for

 $OPAM_{SAT}$  melts. Clinopyroxene T for the three textural subclasses of clinopyroxene from 854 Stock et al., (2018) are shown as mean $\pm 1\sigma$  ranges for comparison. (b) P converted to depth 855 using Equation 11 for petrological and geophysical methods discussed in text, with letters 856 857 corresponding as follows: A = mean $\pm 1\sigma$  of all matrix glass values using Equation 10a; B = mean $\pm 1\sigma$  of all whole rock values using Equation 10a; C = mean $\pm 1\sigma$  of matrix glass values 858 with  $P_{SAT} \ge 0.5$  using Equation 10a;  $D = \text{mean} \pm 1\sigma$  of whole rock values with  $P_{SAT} \ge 0.5$  using 859 Equation 10a; E = pre-eruptive shallow source from Stock et al., (2018) using InSAR; F, G, H 860 = syn-eruptive deep source of the circumferential fissure phase, caldera fill phase, and the entire 861 eruption, respectively, from Stock et al., (2018) using InSAR; I, J, K = clinopyroxene-liquid 862 barometry (Neave and Putirka, 2017) for glomerocrystic, phenocrystic, and tephra-hosted 863 clinopyroxene, respectively, from Stock et al., (2018). The Moho depth is shown by the dashed 864 865 line (Feighner and Richards, 1994).

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Fig. 7: (a) Pressure (kbar) versus temperature (°C) using our new OPAM thermobarometer for 867 868 all matrix glass and melt inclusion data from the 2014-2015 Holuhraun eruption (Iceland; Hartley et al., 2018). Data points are colour contoured for P<sub>SAT</sub> using our binary classifier for 869 OPAM<sub>SAT</sub> melts. (b) P converted to depth using a fixed density of  $2.86*10^6$  kg/m<sup>3</sup> for 870 petrological and geophysical methods discussed in text, with letters corresponding as follows: 871 872 A = mean $\pm 1\sigma$  of all matrix glass values using Equation 10a; B = mean $\pm 1\sigma$  of all melt inclusion 873 values using Equation 10a;  $C = \text{mean}\pm 1\sigma$  of matrix glass values with  $P_{SAT} \ge 0.5$  using Equation 10a; D = mean $\pm 1\sigma$  of melt inclusion values with P<sub>SAT</sub> $\geq 0.5$  using Equation 10a; E = pre-eruptive 874 seismicity (Sigmundsson et al., 2015); F = most common earthquake hypocentral depth 875 876 (Ágústsdóttir et al., 2016); G = clinopyroxene-liquid barometry (Neave and Putirka, 2017) for clinopyroxene crystals screened for textural equilibrium (Halldórsson et al., (2018); H, I = 877 878 OPAM barometry of glass and melt inclusions respectively using the Yang et al (1996) equations implemented by the approach of Hartley et al (2018); J, K = as for H and I but with data filtered for  $P_{f} \ge 0.8$  according to the saturation test of Hartley et al (2018).

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## 882 DATA AVAILABILITY

The data tables used in this article and additional supplementary figures described in text are 883 available in its online supplementary material. The OPAM thermobarometer code and 884 885 saturation test are available on GitHub combined R script as а (https://github.com/OJHiggins/HS2024 OPAM). 886

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Fig.1: (a) Temperature (°C) versus pressure (kbar); (b) melt MgO (wt%) versus melt CaO/Al<sub>2</sub>O<sub>3</sub>; and (c) melt SiO<sub>2</sub> (wt%) versus melt Na<sub>2</sub>O+K<sub>2</sub>O (wt%) for OPAM<sub>SAT</sub> equilibrium experiments, colour contoured for melt H<sub>2</sub>O (wt%). Grey symbols are hydrous experiments for which a H<sub>2</sub>O content is not reported. Diamonds are experiments from studies used to calibrate the Yang et al (1996) cotectic equations. The black curvilinear boundary in (c) is the alkalinesubalkaline transition boundary (Irvine and Baragar, 1971). Melt compositions are normalised to 100 wt% anhydrous with all Fe as FeO.

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Fig. 2: (a) Melt SiO<sub>2</sub> (wt%) versus melt Al<sub>2</sub>O<sub>3</sub> (wt%); and (b) melt SiO<sub>2</sub> (wt%) versus melt MgO (wt%) for OPAM<sub>SAT</sub> and OPAM<sub>UNSAT</sub>. OPAM<sub>SAT</sub> experiments were compiled for this study and are presented in Supplementary Table 1a. OPAM<sub>UNSAT</sub> experiments (olivine  $\pm$ 

1229 plagioclase  $\pm$  augitic clinopyroxene  $\pm$ Fe-oxides) are from the compilation of Weber and Blundy (2023; Supplementary Table 2a). The OPAM convex hull is a polygon connecting the 1230 outermost OPAM-saturated points in X-Y space with  $\pm 5\%$  relative added to account for 1231 1232 analytical uncertainty (black field in both figures). Melts from the 2015 eruption of Wolf Volcano (Galápagos; Stock et al., 2018, 2020) and the 2014-2015 Holuhraun eruption (Iceland; 1233 Hartley et al., 2018) are also plotted (see Discussion; Supplementary Table 5). All experimental 1234 1235 and natural melt compositions are plotted normalised to the 8 critical oxides used for the OPAM 1236 saturation test.





Fig. 3: Performance of the Yang et al (1996) thermometer and barometer for an independent
test of 199 OPAM<sub>SAT</sub> experiments from studies not used in the original Yang et al (1996)

1240 calibration. (a) The predicted temperature (°C) versus experimental temperature (°C). (b) The SEE and MAE for temperature (°C) as a function of melt H<sub>2</sub>O (wt%). The line for n shows the 1241 number of experiments with the corresponding melt H<sub>2</sub>O content on the x axis or less used to 1242 1243 calculate the SEE or MAE. (c) The predicted pressure (kbar) versus experimental pressure (kbar). (d) The SEE and MAE for pressure (kbar) as a function of melt H<sub>2</sub>O (wt%). In (a) and 1244 1245 (c) points are colour coded for melt H<sub>2</sub>O (wt%) and large symbols indicate experiments which pass the statistical filter of Hartley et al (2018) with a  $P_F \ge 0.8$  (see text for details). Solid line is 1246 1:1 and dashed lines are the bounds of the quoted uncertainty for the Yang et al (1996) 1247 thermometer (±20 °C) and barometer (±1.13 kbar) respectively. In (b) and (d) the ranges of 1248 H<sub>2</sub>O (wt%) measured for Galápagos melts and calculated from nominally anhydrous minerals 1249 1250 are shown (Gleeson et al., 2021).



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Fig. 4: Performance (Equation 9a-9f) of our statistical test for  $OPAM_{SAT}$  melts as a function of n (the number of adjacent experiments considered in multidimensional space when testing if a natural melt is  $OPAM_{SAT}$ ). Grey boxes (i, ii, iii) highlight changes in test performance as a function of n and are discussed in text. The chosen n used for our statistical test is 11 (thick dashed line) at the start of region (ii).



1259 Fig. 5: The performance of our new OPAM thermobarometer using the compiled training and validation data (Supplementary Table 1a). (a) Experimental temperature (°C) versus predicted 1260 temperature (°C). (b) Experimental pressure (kbar) versus predicted pressure (kbar). (c) 1261 Experimental temperature (°C) minus predicted temperature (°C) and (d) experimental pressure 1262 1263 (kbar) minus predicted pressure (kbar) as a function of classified geodynamic setting (lines) 1264 and experimental melt H<sub>2</sub>O (wt%; points). In (a) and (b) summary statistics for training (grey 1265 points) and validation (coloured points) datasets are provided in the plot area. Grey solid line 1266 is the 1:1 line, black dashed lines are the bounds of the quoted validation SEE, and the black

solid line is a linear fit between predicted (x) and experimental (y) values. In (c) and (d)
geodynamic settings for experimental studies are either arc or ocean island basalt (OIB) plus
mid-ocean ridge basalt (MORB).



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Fig. 6: (a) Pressure (kbar) versus temperature (°C) using our new OPAM thermobarometer for 1272 all matrix glass and whole rock data from the 2015 eruption of Wolf Volcano (Galápagos; 1273 Stock et al., 2018). Data points are colour contoured for P<sub>SAT</sub> using our binary classifier for 1274 1275  $OPAM_{SAT}$  melts. Clinopyroxene T for the three textural subclasses of clinopyroxene from 1276 Stock et al., (2018) are shown as mean $\pm 1\sigma$  ranges for comparison. (b) P converted to depth using Equation 11 for petrological and geophysical methods discussed in text, with letters 1277 1278 corresponding as follows: A = mean $\pm 1\sigma$  of all matrix glass values using Equation 10a; B = 1279 mean $\pm 1\sigma$  of all whole rock values using Equation 10a; C = mean $\pm 1\sigma$  of matrix glass values with  $P_{SAT} \ge 0.5$  using Equation 10a; D = mean  $\pm 1\sigma$  of whole rock values with  $P_{SAT} \ge 0.5$  using 1280 Equation 10a; E = pre-eruptive shallow source from Stock et al., (2018) using InSAR; F, G, H 1281 = syn-eruptive deep source of the circumferential fissure phase, caldera fill phase, and the entire 1282 eruption, respectively, from Stock et al., (2018) using InSAR; I, J, K = clinopyroxene-liquid 1283 barometry (Neave and Putirka, 2017) for glomerocrystic, phenocrystic, and tephra-hosted 1284

clinopyroxene, respectively, from Stock et al., (2018). The Moho depth is shown by the dashedline (Feighner and Richards, 1994).





Fig. 7: (a) Pressure (kbar) versus temperature (°C) using our new OPAM thermobarometer for 1289 1290 all matrix glass and melt inclusion data from the 2014-2015 Holuhraun eruption (Iceland; Hartley et al., 2018). Data points are colour contoured for P<sub>SAT</sub> using our binary classifier for 1291 OPAM<sub>SAT</sub> melts. (b) P converted to depth using a fixed density of  $2.86*10^6$  kg/m<sup>3</sup> for 1292 1293 petrological and geophysical methods discussed in text, with letters corresponding as follows: 1294 A = mean $\pm 1\sigma$  of all matrix glass values using Equation 10a; B = mean $\pm 1\sigma$  of all melt inclusion values using Equation 10a;  $C = \text{mean}\pm 1\sigma$  of matrix glass values with  $P_{SAT} \ge 0.5$  using Equation 1295 1296 10a; D = mean $\pm 1\sigma$  of melt inclusion values with P<sub>SAT</sub> $\geq 0.5$  using Equation 10a; E = pre-eruptive 1297 seismicity (Sigmundsson et al., 2015); F = most common earthquake hypocentral depth (Ágústsdóttir et al., 2016); G = clinopyroxene-liquid barometry (Neave and Putirka, 2017) for 1298 clinopyroxene crystals screened for textural equilibrium (Halldórsson et al., (2018); H, I = 1299 OPAM barometry of glass and melt inclusions respectively using the Yang et al (1996) 1300 1301 equations implemented by the approach of Hartley et al (2018); J, K = as for H and I but with data filtered for  $P_{f} \ge 0.8$  according to the saturation test of Hartley et al (2018). 1302