

1 **Mitigating Microbial Artifacts in Laboratory Research of H<sub>2</sub> Energy Geo-storage**

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14 **ABSTRACT**

15 Today our energy sectors are focusing on the marathon of CO<sub>2</sub> cut and coherently require  
16 progress in energy transition schemes to meet the UN climate change challenge and achieve a  
17 zero-emission target. Among these schemes, radioactive disposal, CO<sub>2</sub>, NH<sub>3</sub>, and H<sub>2</sub> geological  
18 storage are promising options for fixing anthropogenic waste, greenhouse gases and storing  
19 green energy in the depleted oil/gas reservoirs, deep saline aquifers, and salt caverns.  
20 Consequently, this could be achieved through rigorous research and development (R&D) projects  
21 involving laboratory-scale experiments. Despite the ubiquity of microorganisms in various  
22 environments, their potential impact on laboratory studies in fields outside of the biological  
23 sciences is not well established. In particular, their presence in research related to new energy  
24 technologies, such as hydrogen storage, poses a significant risk to experimental integrity.  
25 Microorganisms can consume hydrogen and other substances, leading to potentially misleading  
26 results. This oversight can have profound implications, especially when studying geological  
27 formations where microbial contamination might alter the properties and behaviours of reservoir  
28 rocks. Thus, it is crucial to incorporate sterile controls in experiments to accurately assess the  
29 influence of independent variables and to discern the specific effects of microbial presence. The  
30 effect of ultraviolet (UV), autoclave, oven heating, ethanol 75%, ethanol 95%, and gamma  
31 irradiation for cleaning microorganisms in the sand were investigated Interestingly, our  
32 experimental results revealed that gamma irradiation and autoclave heating are the most vibrant  
33 options for extinguishing microorganisms from the surface of the rock and saying no to the risk of  
34 experimental error in future work reflecting geological storage applications.

35 **Keywords:-** Geological storage, energy transition microorganisms, rock cleaning techniques,  
36 porous media, salt cavern

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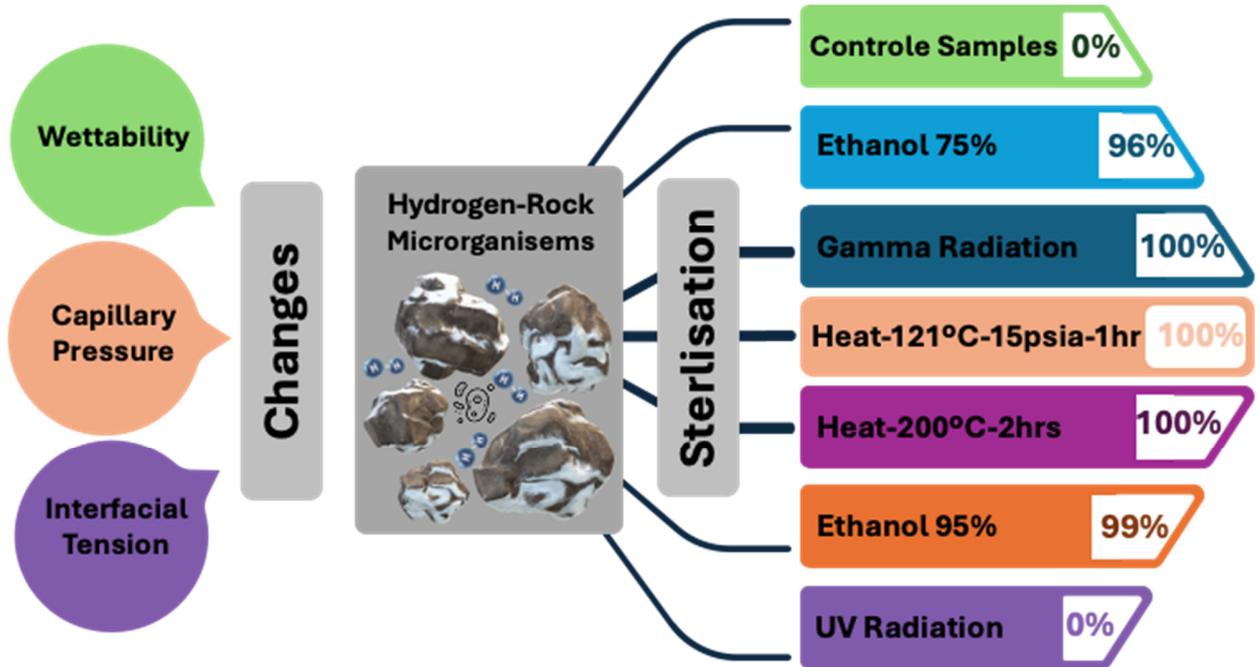
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47 **INTRODUCTION**

48 Geo-solutions are crucial in the phase of energy transition to achieve a carbon neutrality target  
 49 via compressed air, anthropogenic CO<sub>2</sub>, helium and hydrogen geological storage. Given that  
 50 microbes are widely distributed either in the subsurface or at surface and its presence may trigger  
 51 uncertainties and risks associated with long-term implementations of the anthropogenic waste  
 52 and energy fluids geological storage projects. Recent interest in H<sub>2</sub> and CO<sub>2</sub> storage in porous  
 53 media and salt caverns has developed a need for experimental models to evaluate the rate of  
 54 microorganism contamination for the application of geological hydrogen storage (GHS). The  
 55 variety of reservoir engineering experiments, for example measurement of porosity, permeability,  
 56 capillary pressure, interfacial tension, H<sub>2</sub>-brine core flooding, wettability, in-situ loss of H<sub>2</sub> and H<sub>2</sub>S  
 57 generation are conducted to quantify the potential of the reservoir rock performance for GHS <sup>1</sup>.  
 58 However, some bacteria either lying on the laboratory desk or present in the atmosphere could  
 59 contaminate surface of the rock resulting in artifacts during the assessment of these reservoir  
 60 properties.

61 Sterilization is needed without affecting the nature of the sample. There is no information what  
 62 microorganisms might be lurking inside the rock samples in the laboratory. The rock samples

63 need to be sterilized before release and kept in an isolation room or biological-free zone for further  
64 testing. A team of researchers conducted sterilization of perchlorates which have been observed  
65 on the surface of Mars. Results revealed that Martian UV flux made perchlorates to bactericidal.  
66 The surface of the planet is exposed to both UVC radiation of <280 nm and UVB of 280 to 315  
67 nm when compared to surface of the Earth <sup>2</sup>. Additionally, two components of the red planet  
68 including iron oxides and hydrogen peroxide induce a synergetic effect with irradiated  
69 perchlorates causing a 10.8-fold increase in cell death of *Bascillus subtilis* compared to cells  
70 exposed by UV rays for 60 seconds <sup>3</sup>. However, the absolute killing of bacteria from the rock  
71 remains a challenge through UV sterilization. The unpenetrated UV radiation could not produce  
72 the expected results due to the irregular shape of the rock and rough surface. <sup>4</sup>. This procedure  
73 is time-consuming with risk of contamination and did not confirm the irradiation of whole  
74 microcavities. Thus, a liquid chemical may be required which could penetrate tiny size capillaries  
75 of rock to influence the absolute sterilization effect. The sterilization of limestone rock using 96%  
76 and 70% of ethanol could not develop absolute aseptic conditions in deep natural cracks <sup>5</sup>. The  
77 environmental microbes interacted with the rock causing calcite dissolution and precipitation. In  
78 an overlooked phenomenon, it was illustrated that the entombment of microorganisms in Si-rich  
79 precipitate in the nutrient-depleted environment suggests Si mobilization at ambient conditions <sup>5</sup>.  
80 Hence, the phenomena might have jeopardized the mineralogy of the geological material. **Table**  
81 **S1** provides a summary of literature related with effect of different sterilization on the mineralogy  
82 of the different rocks and minerals for the applications of astrobiology and geoscience.

83 In previous work, autoclave technique (~ 121 °C, 15 psi and 30 min) was used to sterilize the  
84 rocks <sup>6</sup>. Additionally, dry heating (>200 °C and ~3 h) is also an efficient mean of sterilization.  
85 Nevertheless, it is reported that microorganisms could penetrate lengths of consolidated Berea  
86 sandstone rock quicker when the rock was sterilized by autoclaving compared to the dry heating.  
87 Additionally, autoclave resulted high chloride than dry heating resulting aggregated and uneven  
88 shaped of clays and decreased bacterial penetration rates. Hence the findings show that dry-heat  
89 (dry oven heating) is more appropriate technique when compared to autoclaving when  
90 investigation biotic and abiotic the Berea sandstone rock <sup>7</sup>. Moreover, Gamma-ray has been  
91 proven to be a strong sterilization technique. This is a strong ray and reveals quite effective results  
92 to completely sterilize the rock from both inside and outside. The Mars analogue rocks and  
93 minerals were sterilized with gamma. The high doses of the ray displayed no effect on the rock.  
94 However, the darkening of some minerals was observed due to gamma radiation still the  
95 technique was considered a feasible choice for sterilizing the Mars returned rock samples <sup>8</sup>.

96 Therefore, there is a pressing need to conduct comprehensive research to reveal the effects of  
 97 different sterilization techniques, for instance i. UV, ii. Ethanol concentrations (both 75% and  
 98 90%), iii. Oven heating (dry heating), iv. Autoclave heating (wet heating) and v) Gamma ray  
 99 irradiation. Thus, oven heating, autoclave heating and gamma irradiation on microbial-related  
 100 laboratory research environment, ruling out the potential microbial artifacts in laboratory condition  
 101 on H<sub>2</sub> and CO<sub>2</sub> geological.

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103     ▪ **EXPERIMENTAL RESEARCH**

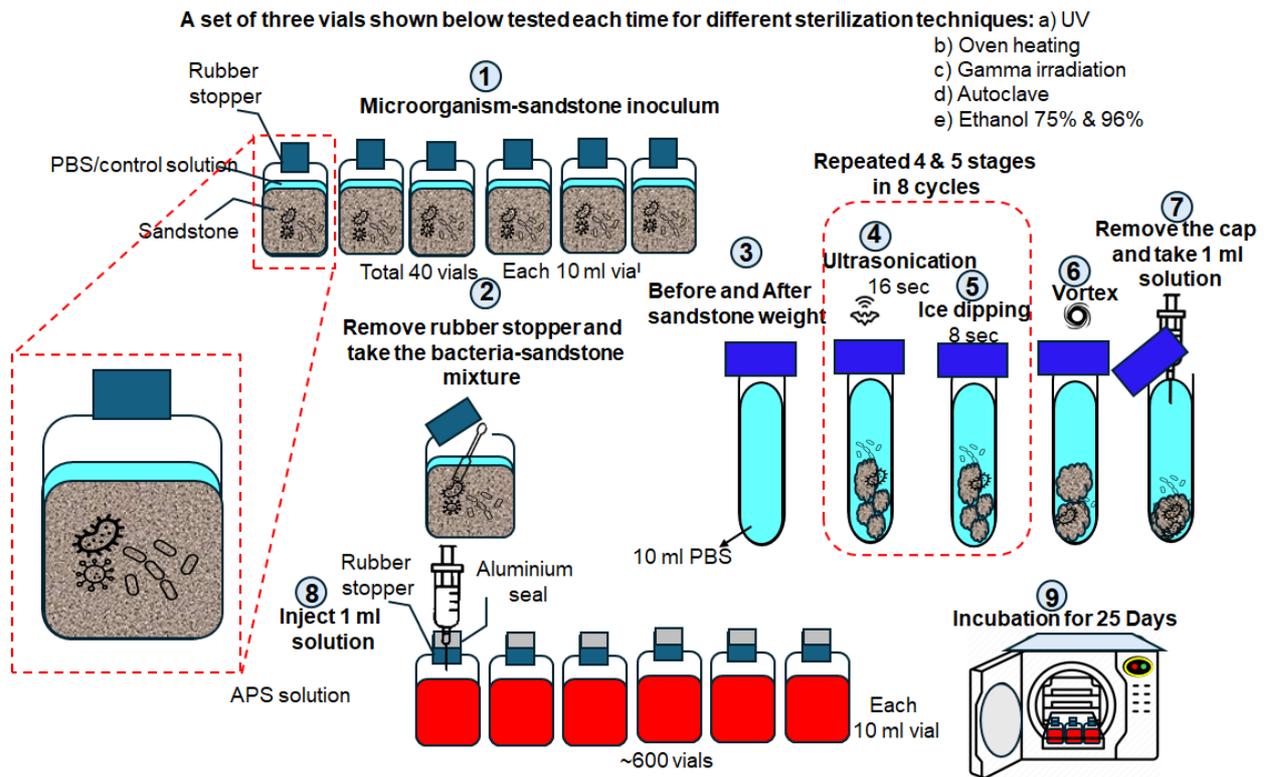
104 Generally, **Figure 1** illustrates research methodology used in this study to investigate the effect  
 105 of different sterilization techniques on sand-phosphate buffer saline (PBS)-microorganism  
 106 inoculum. **Table S2** enlists the number of bacteria-sand inoculum used to investigate the effect  
 107 of different techniques on killing the efficiency of bacteria in the rock. **Table 1** enlists the  
 108 techniques and factors used for the sterilization of sand.

109

110 Table 1 presents list of techniques and its parameters used for the sterilization of sand

Techniques	Temperature	Pressure	Intensity	Exposure time	Concentration
UV	Ambient	Ambient	280 to 100 nm	30 min both sides	NA
Autoclave	121°C	15psi	NA	1h	NA
Oven	200°C	Ambient	NA	2h	NA
Ethanol	Ambient	Ambient	NA	Washed 3 times and soaked 15 min	75wt% 95 wt%
Gamma ray	NA	NA	NA	32h	NA

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113 **Figure 1** shows the research methodology and various stages, in particular **1**. Preparation  
 114 of sand-PBS-microorganism inoculum and exposed them under different sterilization  
 115 methods, **2**. Taking of sand sample, **3**. Weighing of falcon tube containing 10 mL PBS  
 116 solution and around 2 grams of sand-PBS-microorganism inoculum, **4 - 5**. Ultrasonication  
 117 and ice dipping of the falcon tube with the inoculum, **6**. Rapid mixing of samples in falcon  
 118 tube using the vortex machine, and finally **7**. Take 1 ml of the solution from the Falcon  
 119 Tube, **8**. Injecting 1 ml of solution in 10 ml acid producing bacteria (APB) media at  $10^{-1}$   
 120 concentration and conducting serum dilution was conducted up to seven concentrations  
 121 and finally **9**. Incubation of the vials

122 **Table S3** enlists the chemicals and sand used in this work. APB solution media was  
 123 prepared as shown in **Table S4**. Over 600 vials were prepared with each 10 ml and filled  
 124 them with APB solution. **Table S4** and **Table S5** provide the list of chemicals used in this  
 125 work for the preparation of the growth media solution and phosphate buffer solution  
 126 (PBS). A consortium composed of *Bacillus sp.*, *Enterobacter sp.*, and *Cronabacter sp.*  
 127 bacteria were used in this study. 1 ml of each microorganism was taken from 10 ml stock

128 solution. Later, 1 ml of each microorganism solution was added in 50 ml PBS solution to  
129 achieve  $10^7$  dilution. Bacteria-sand solution was prepared using the procedure for  
130 instance, firstly a 50ml falcon tube was filled with PBS solution using a sterile pipette.  
131 Secondly, the spatulas were wrapped in aluminium foil and autoclaved under the liquid  
132 condition mode. Using the sterile spatulas, the sand was taken and added into 10ml glass  
133 vial weighing around  $\sim 15.7$ g of sand. Every time a fresh spatula was used to avoid  
134 contamination. **Table S6** illustrates the weight of the falcon tube with and without sand  
135 was measured. **Figure S1** shows 10 ml vials containing approximately 15.5 g sand. Later,  
136 3ml of cell suspension was added in each vial to use these vials for different tests. A total  
137 of 567 vials vials with 9 mL of anaerobic APB culture media were prepared **Figure S2 (A-C)**  
138 illustrates the methodology used for the preparation of 10 ml vials containing APB solution for the  
139 experiment. . We examined the effect of different cleaning techniques including UV, Ethanol 75%,  
140 Ethanol 96%, autoclave heating, oven heating and gamma irradiation using of incubation of the  
141 vials. **Supplementary information** provides list of all tests, equipment used and their procedure.

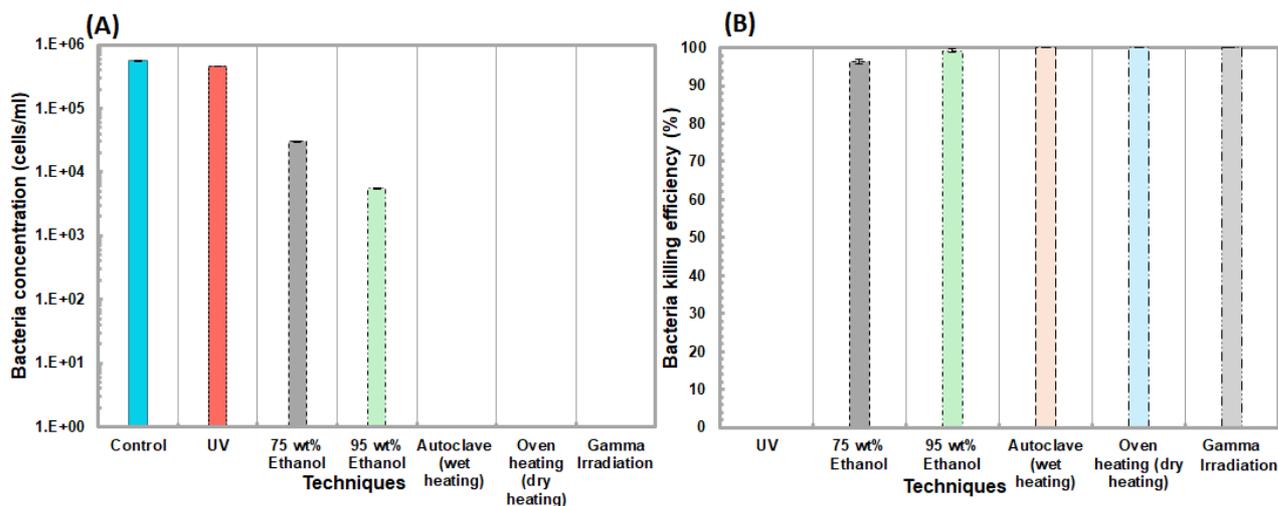
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## 143 ■ RESULTS AND DISCUSSIONS

144 **Figure 2A** illustrates the concentration of surviving cells in the sand after sterilization process.  
145 The control (no subjected to sterilization) indicated that the concentration of cells in the test bottles  
146 was in the order of  $10^6$  cell/g before sterilization treatment. MPN results indicate that autoclave,  
147 oven heating, and gamma irradiation were able to eliminate all cells in the sand. However, oven  
148 heating, autoclave and gamma rays could induce mineralogical changes. For instance, oven  
149 heating is reported for micro cracking in quartz minerology of sand <sup>9</sup>. Additionally, autoclave steam  
150 can carry microorganism and penetrate into the rock <sup>10</sup>. And gamma ray caused discoloration of  
151 quartz mineral in sand which is in line with previous finding <sup>8</sup>. **Figure S3** illustrates the glass of  
152 serum vials and sand changed to a blackish colour after treated using gamma irradiation. These  
153 changes could affect the in-situ reservoir properties in particular porosity, permeability, interfacial  
154 tension, capillary pressure and wettability at micron scale resulting in adverse effects on the  
155 laboratory research for geological H<sub>2</sub> storage.

156 Ethanol and UV radiation were not able to inactivate all the cells and a concentration between  $10^4$   
157 to  $10^6$  cells/g surviving bacteria were still detected. Our results provide a clear comparison of  
158 performance in cell sterilization among the different techniques. UV sterilization shows no effect  
159 on the sterilization of sand. We found that ethanol concentrations including 75 wt% and 95 wt%

160 illustrate the lowest killing efficiency more alive cells were detected compared to the other sterilisation methods  
 161 evaluated. These techniques seem to cause irreversible damage to cellular component which are  
 162 essential for their survival and illustrates 100% killing efficiency of bacteria in the sand (**Figure**  
 163 **2B**). **Figure 2B** shows the killing efficiency of each sterilization method. UV irradiation was the  
 164 inefficient technique with killing efficiency of 0%. This finding may be attributed to less penetration  
 165 of UV from glass of the vial into the rock. Ethanol 75% achieved a killing of 96.3%, whereas  
 166 ethanol 95% reached a killing of 99.2%. Although killing efficiency values are high, it is important  
 167 to consider that surviving population of cells was also high, in the order of  $10^5$  and  $10^4$ ,  
 168 respectively. This survival percentage could restore microbial activity in long-time core flooding  
 169 and salt cavern bioreactors experimental setups under the influence of anaerobic conditions.  
 170 Additionally, the effects of precipitation, change in mineralogy, and formation brittle were reported  
 171 in calcite, clay, and sand respectively after the use of Ethanol as a sterilizing substance. **Figure**  
 172 **S4** illustrates the total number of vials incubated after inoculated with cell suspension.



173  
 174 **Figure 2** Effect of different techniques on the sterilization of the sand (A) bacteria concentration  
 175 in cell/ml, and (B) bacteria killing efficiency in percent.

## 176 ■ CONCLUSIONS

177 This research concludes that autoclave heating, oven heating, and gamma irradiation methods  
 178 are effective in absolutely eliminating bio-life inside rock. However, gamma irradiation caused  
 179 discoloration of sand, and oven heating may induce micro-cracks, potentially compromising the  
 180 rock's integrity. We propose that autoclaving is the most suitable technique due to its low  
 181 operating temperature and pressure, making it ideal for sand sterilization. Additionally, rock  
 182 treated with 75% ethanol still exhibited a significant microbial survival rate, casting doubt on its

183 efficacy for long-term experiments, as bacteria could potentially regrow over time. Further  
184 research is recommended to investigate the effects of these methods on the rock's mineralogy,  
185 petrophysical properties, and surface behaviour for large-scale geological hydrogen storage  
186 experiments.

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1 **Supplementary Information**

2 **Mitigating Microbial Artifacts in Laboratory Research of H<sub>2</sub> Energy Geo-storage**

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34 **Table S1** provides a literature review table on effect of different sterilization techniques on  
 35 mineralogy of rocks related with geoscience and astrobiology applications

Sterilization technique	Overall Effect on Mineralogy	Geoscience Application	Astrobiology Application	Study
UV	<ul style="list-style-type: none"> <li>• Unpenetrated to core of the rock</li> </ul>	<ul style="list-style-type: none"> <li>• The unpenetrated UV radiation could not produce the expected results due to the irregular shape of the rock and rough surface.</li> <li>• It was examined that only one side of rock could be irradiated at a time and required a change of position of the sample during the exposure to UV</li> <li>• A photochemical lap consisting of four circular UV lamps allows adequate sterilization of rough surfaces of geological material while killing the microorganism possibly living in the core of the rock geological</li> </ul>	<ul style="list-style-type: none"> <li>• Martian UV flux made surface of perchlorates to bactericidal</li> <li>• The surface of the planet is exposed to both UVC radiation of &lt;280 nm and UVB of 280 to 315 nm when compared to surface of the Earth</li> <li>• Iron oxides and hydrogen peroxide induce a synergetic effect with irradiated perchlorates causing a 10.8-fold increase in cell death of <i>Bascillus substilis</i> compared to cells exposed by UV rays for 60 seconds</li> <li>• However, the absolute killing of bacteria from the rock remains a challenge through UV sterilization.</li> <li>•</li> </ul>	1-3
Ethanol 75% and Ethanol 95%	<ul style="list-style-type: none"> <li>• Calcite can precipitate.</li> <li>• Clay minerology can change.</li> <li>• Brittle film formed on sandstone</li> </ul>	<ul style="list-style-type: none"> <li>• High-Mg calcite can precipitate at ambient temperature via partially replacing water with 75% ethanol additionally it bypasses the hydration barrier.</li> <li>• Ethanol concentration could change the polymorph of calcite, vaterite or aragonite in Mg-free environment.</li> <li>• The specific polarizability decreases as ethanol concentration increased from 0, 10, and 20% v/v. Ethanol concentration on clay driven polarization relative to changes in clay minerology.</li> </ul>	Nil	4-9

		<ul style="list-style-type: none"> <li>Ethanol does not develop alkoxy silane-sandstone compatibility. Ethanol formed brittle film.</li> <li>Ethanol was used instead of water to prevent interaction of clay-water cation exchange reactions prior to start experiment.</li> </ul>		
Solarization	Soil	<ul style="list-style-type: none"> <li>Solarization was conducted through covering the soil with sterile transparent plastic film to trap solar radiation for heating which could rise temperature above 70 °C and considered as an adequate range to kill variety of plant pathogens.</li> </ul>	Nil	10
Oven heating	<ul style="list-style-type: none"> <li>Micro cracking in Quartz sandstone at very high temperature</li> </ul>	<ul style="list-style-type: none"> <li>Micro-cracking in quartz sandstone at the grain boundaries revealed increase in the porosity at 600 °C and within the grains and mineralogical changes at 750 °C.</li> <li>The structure of clay mineral collapse at 600 °C, and chlorite above 600 °C.</li> </ul>	Nil	11
Autoclave	<ul style="list-style-type: none"> <li>Could increase bacteria penetration in porous rock</li> </ul>	<ul style="list-style-type: none"> <li>Steam treatment provides a better solution to clean pests in soil.</li> </ul>	Nil	12
Gamma Irradiation	<ul style="list-style-type: none"> <li>Very low, discolor of the rock</li> </ul>	<p>Physico-chemical properties of natural sediments:</p> <ul style="list-style-type: none"> <li>Major part of Clay mineralogy unchanged</li> <li>pH slightly changes after irradiation</li> <li>Irradiation samples reduced the cation exchange capacity</li> </ul>	<p>Mars returned rock samples sterilization:</p> <ul style="list-style-type: none"> <li>Gamma photons from <sup>60</sup>Co (1.17 and 1.33 MeV)</li> <li>Doses as high as 3×10<sup>7</sup> rads</li> <li>No effect detected on basalt rock and quartz mineral</li> </ul>	13, 14

		<ul style="list-style-type: none"> <li>• Irradiated samples reduced iron oxide</li> <li>• Effects on organic and inorganic fraction were observed</li> </ul>	<ul style="list-style-type: none"> <li>• No change in concentration of elements before and after irradiation</li> <li>• Crystal structure of mineral not affected</li> <li>• No change in grain density except a small effect on halite</li> <li>• Specific surface area of rock was not effected.</li> <li>• Dose induces no radioactivity in the rock</li> <li>• No change in their isotopic composition</li> <li>• No change in chemical composition</li> <li>• No change in crystallographic structure</li> <li>• Effects in the visible and near-infrared spectral region</li> <li>• Discoloration of quartz</li> <li>• Darkening of quartz and halite</li> <li>• Increases thermoluminescence of quartz and plagioclase</li> </ul>	
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42 **Table S2** Preparation of bacteria-sandstone PBS solution and inoculated overnight

S.no	Sample	Sandstone weight	PBS solution	Number of 10 ml vials
1	Control A, B, C	15.7 g	3ml	3
2	Control A, B, C			
3	Control A, B, C			
4	Ethanol 95% A, B, C			
5	Ethanol 75% A, B, C			
6	Oven heating (200 °C) A, B, C			
7	Ultraviolet A, B, C			
8	Autoclave A, B, C			
9	Gamma irradiation A, B, C			

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45 **Figure S1** Approximately 40 sandstone vials were prepared. Each vial has a 10 ml capacity and  
 46 contains around 15.7 g of sandstone

47 **Table S3** list of chemical, rock and deionized water used in this work

Chemicals	Formula	Purity	Manufacturer	Quantity
Sandstone	NA	Silica white sand; super fine	Cook Industrial Minerals Pty. Ltd.	650 g
Beef extract (Lab Lemco)	NA	Total Nitrogen=12.4 %w/w Amino Nitrogen= 2.5 %w/w	Oxoid	1g
Tryptone	NA	Total Nitrogen=12.7 %w/w Amino Nitrogen=3.7 %w/w NaCl=0.4 % w/w		10g
D (+) Glucose	NA	≥99.5%	Sigma-Aldrich	5g
Sodium chloride	NaCl	≥99.0%	Sigma-Aldrich	20g
Phenol red	NA	354.38	Sigma-Aldrich	0.018g
Potassium phosphate monobasic	KH <sub>2</sub> PO <sub>4</sub>	≥99.0%	Sigma-Aldrich	0.4g
Potassium phosphate dibasic	K <sub>2</sub> HPO <sub>4</sub>	≥99.0%	Sigma-Aldrich	1.23g
DI-water	H <sub>2</sub> O	NA	In lab facility	1000ml

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49 **Table S4** Chemical composition of culture media APB 1000 ml

Chemicals	Formula	Purity	Manufacturer	Quantity
Beef extract (Lab Lemco)	NA	Total Nitrogen=12.4 %w/w Amino Nitrogen= 2.5 %w/w	Oxoid	1g
Tryptone	NA	Total Nitrogen=12.7 %w/w Amino Nitrogen=3.7 %w/w NaCl=0.4 % w/w		10g
D (+) Glucose	NA	≥99.5%	Sigma-Aldrich	5g
Sodium chloride	NaCl	≥99.0%	Sigma-Aldrich	20g
Phenol red	NA	354.38	Sigma-Aldrich	0.018g
DI-water	H <sub>2</sub> O	NA	In lab facility	1000ml

50 **Note:** the pH of the solution was maintained ~7.48 using caustic soda as a buffer solution

51

52 **Table S5** Chemical composition of PBS 1000 ml

Chemicals	Formula	Purity/Molecular Weight	Manufacturer	Quantity
Sodium chloride	NaCl	58.49	Chem-Supply	8.1g
Potassium phosphate monobasic	KH <sub>2</sub> PO <sub>4</sub>	≥99.0%	Sigma-Aldrich	0.4g
Potassium phosphate dibasic	K <sub>2</sub> HPO <sub>4</sub>	≥99.0%	Sigma-Aldrich	1.23g
DI-water	H <sub>2</sub> O	NA	In lab facility	1000ml

53 **Note:** the pH of the PBS solution was maintained at ~7 using caustic soda as a buffer solution

54

55 **Table S6** Weight of 10 ml falcon tube containing 10 ml PBS solution with and without sand

S.no	Date	Technique	Samples	PBS 10 ml without sandstone	PBS 10 ml with sandstone
1	12/12/2023	Control	A	16.59g	17.73g
2	12/12/2023	Control	B	16.60g	17.57g
3	12/12/2023	Control	C	16.62g	17.98g
4	12/12/2023	Ethanol 95%	A	16.58g	17.52g
5	12/12/2023	Ethanol 95%	B	16.58g	17.67g
6	12/12/2023	Ethanol 95%	C	16.68g	17.70g
7	13/12/2023	Control	A	16.34g	17.03g
8	13/12/2023	Control	B	16.36g	17.55g
9	13/12/2023	Control	C	16.35g	17.25g
10	13/12/2023	Ethanol 75%	A	16.65g	17.55g
11	13/12/2023	Ethanol 75%	B	16.63g	17.50g

<b>12</b>	13/12/2023	Ethanol 75%	C	16.70g	17.51g
<b>13</b>	13/12/2023	Oven heating 200°C	A	16.69g	17.41g
<b>14</b>	13/12/2023	Oven heating 200°C	B	16.64g	17.30g
<b>15</b>	13/12/2023	Oven heating 200°C	C	16.68g	17.48g
<b>16</b>	13/12/2023	Ultraviolet	A	16.69g	17.66g
<b>17</b>	13/12/2023	Ultraviolet	B	16.64g	17.81g
<b>18</b>	13/12/2023	Ultraviolet	C	16.63g	18.31g
<b>19</b>	13/12/2023	Autoclave	A	16.61g	17.40g
<b>20</b>	13/12/2023	Autoclave	B	16.66g	17.72g
<b>21</b>	13/12/2023	Autoclave	C	16.70g	17.65g
<b>22</b>	18/12/2023	Control	A	16.48g	17.59g
<b>23</b>	18/12/2023	Control	B	16.46g	17.49g
<b>24</b>	18/12/2023	Control	C	16.43g	17.68
<b>25</b>	18/12/2023	Gamma irradiation	A	16.61g	18.02g
<b>26</b>	18/12/2023	Gamma irradiation	B	16.66g	18.04g
<b>27</b>	18/12/2023	Gamma irradiation	C	16.62g	17.74g

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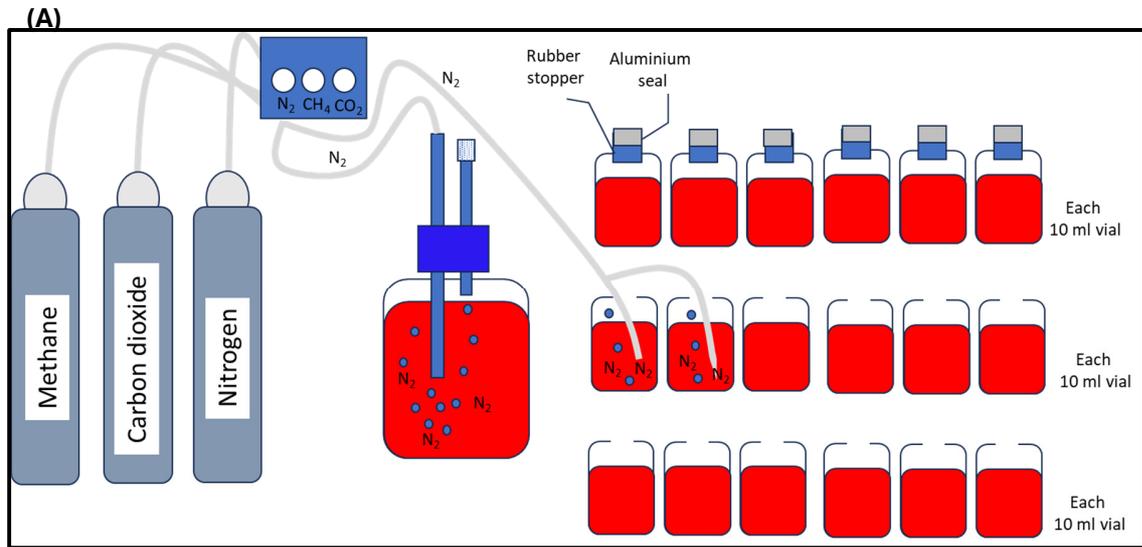
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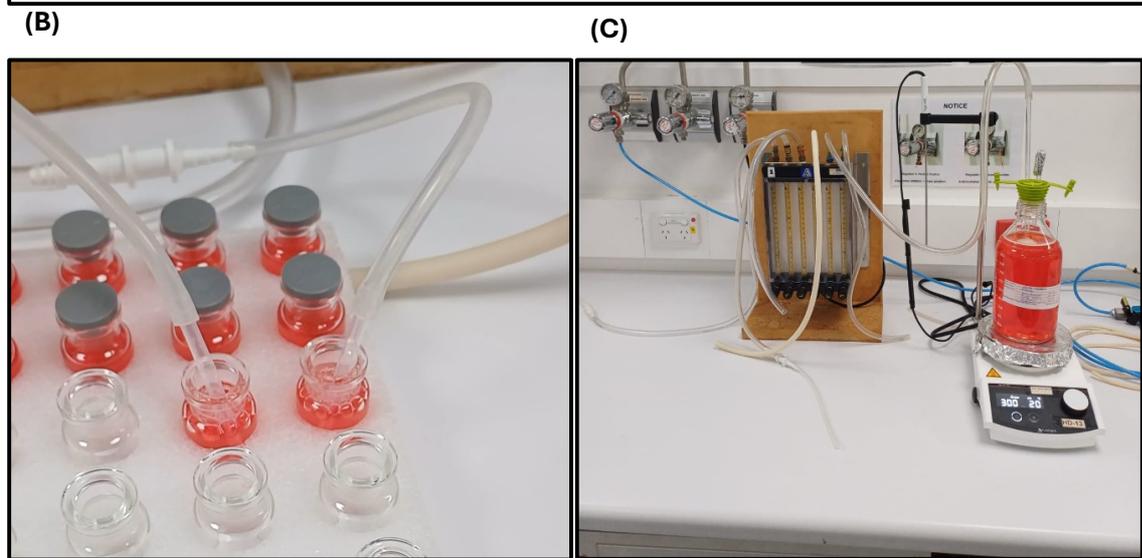
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72 **Figure S2** (A) Schematic of the mechanism used for preparation of 10 ml vials containing and  
73 deoxygenating them using  $N_2$  gas. (B-C) Figures show the deoxygenating of 10 ml vials and  
74 APB stock solution using  $N_2$  gas. All vials were autoclaved for 1h before inoculation of bacteria

75 • **UV**

76 The fume hood equipped with an exhaust fan was powered on. Later, 3 vials of sandstone-  
77 bacteria-PBS inoculum were put in the fume hood and UV tube was switched on with boarded  
78 up windows. UV experiment was conducted for 30 minutes. Each vial was rotated clockwise at  
79  $\sim 180^\circ$  without rubber seal with the interval of 15 mins. We took the sample using the spatula

80 from core of the vial for the inoculation of APB media to measure the efficiency of UV  
81 sterilization.

## 82 **Ethanol 95 wt% and 75 wt%**

83 Ethanol was used at two different concentrations to sterilize the rock sample. 95% and 75%  
84 ethanol solutions were prepared using Ethanol PURE 99.9%. 95 ml of ethane and 5 ml of water  
85 were added in 200ml reagent bottle to make a solution of Ethanol 95 wt%. Similarly, 75 wt% ethanol  
86 was prepared using 75 ml of the pure ethanol and 35 ml of water. There were 6 sandstone-  
87 bacteria-PBS vials each containing around 15.7 g of sandstone and 3ml PBS solution. 3 vials  
88 were washed 3 times with ethanol 95% and the other 3 vials were similarly washed with 75%  
89 ethanol. The ethanol was not completely removed from the vials so that the sandstone may  
90 remain soaked for at least 15 min. Finally, approximately 2 g of sandstone is taken from each  
91 sandstone-bacteria-PBS vial washed with the two different concentrations and added in falcon  
92 tube separately as enlisted in **Table S6**.

### 93 • **Oven**

94 Venticell 111-Eco line oven was used to sterilize the sandstone-bacteria-PBS solution. The oven  
95 was switched on and preheated until set temperature was reached which was 200 °C. We put  
96 three vials in the oven for approximately 2 hours at a constant 200 °C.

### 97 • **Autoclave**

98 Benchtop autoclave model 3870EL-D was used to sterilize the three vials. We have used liquid  
99 cycle mode which means 121 °C temperature and 15 psia pressure conditions. The vials were  
100 autoclaved for approximately 1 hour. Further, we have used a liquid cycle to autoclave APB and  
101 PBS solution contained in the 500 ml Reagent bottles.

### 102 • **Gamma ray irradiation**

103 Three vials of sandstone-bacteria-PBS inoculum samples to ChemCenter Government of  
104 Western Australia at Curtin University Campus for the irradiation of the gamma-ray. The  
105 samples were irradiated for at least 32 h. The center is equipped with a gamma irradiation unit  
106 named Gammacell 220.

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109      • **Ultrasonication**

110      The ultrasonication machine named Power Sonic 510, micro process controlled Benchtop  
 111      ultrasonic cleaner was used to sonicate each 10ml falcon tube for 15sec. Approximately 2g of  
 112      sand were added to falcon tubes containing 10 mL of PBS. Then, falcon tubes were sonicated  
 113      in cycle of ice dipping and sonication. Later, the falcon tube was dipped in the ice for 8sec and  
 114      both sonication and ice dipping for carried 8 times for each falcon tube.

115      • **Vortex mixer**

116      The vortex-Genie® 2, Mo BIO laboratories, Inc used to mix approximately 2 gm of sandstone  
 117      bacteria in 10 ml of falcon tubing containing 10 ml of PBS solution. A total of 27 falcon tubes  
 118      were passed through the vortex to achieve a homogenized mixture of the bacteria and  
 119      sandstone grain particles in the PBS solution to minimize the risk of error.

120      Serum dilution using 1 ml of PBS solution contaminated with sandstone-bacteria-PBBS mixture  
 121      exposed to different sterilization techniques. Serum dilution was conducted for each case  
 122      enlisted in **Table S7**. Serum dilution was conducted until 7 seven concentrations. We have a  
 123      total of 27 different cases and in each case serum dilution was repeated for 3 times totalling 21  
 124      concentrations for each case and 567 for all cases.

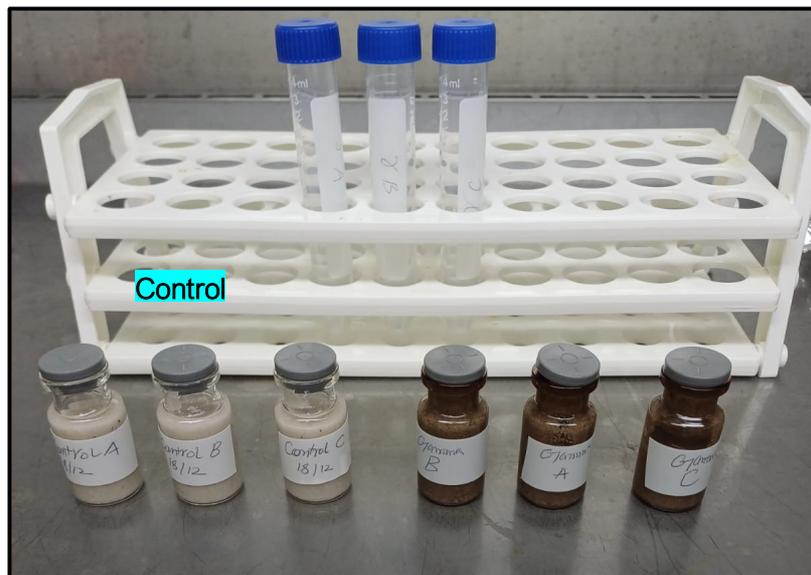
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126      **Table S7** Total number of vials prepared for the controls and different sterilization tests.

S.no	Technique	Samples	Number of 10 APS vials containing 1 ml of bacteria-sandstone-PBS solution from the falcon tube
1	Control	A	21
2	Control	B	21
3	Control	C	21
4	Ethanol 95%	A	21
5	Ethanol 95%	B	21
6	Ethanol 95%	C	21
7	Control	A	21
8	Control	B	21
9	Control	C	21
10	Ethanol 75%	A	21
11	Ethanol 75%	B	21
12	Ethanol 75%	C	21

13	Oven heating 200°C	A	21
14	Oven heating 200°C	B	21
15	Oven heating 200°C	C	21
16	Ultraviolet	A	21
17	Ultraviolet	B	21
18	Ultraviolet	C	21
19	Autoclave	A	21
20	Autoclave	B	21
21	Autoclave	C	21
22	Control	A	21
23	Control	B	21
24	Control	C	21
25	Gamma irradiation	A	21
26	Gamma irradiation	B	21
27	Gamma irradiation </td <td>C</td> <td>21</td>	C	21
Total			567

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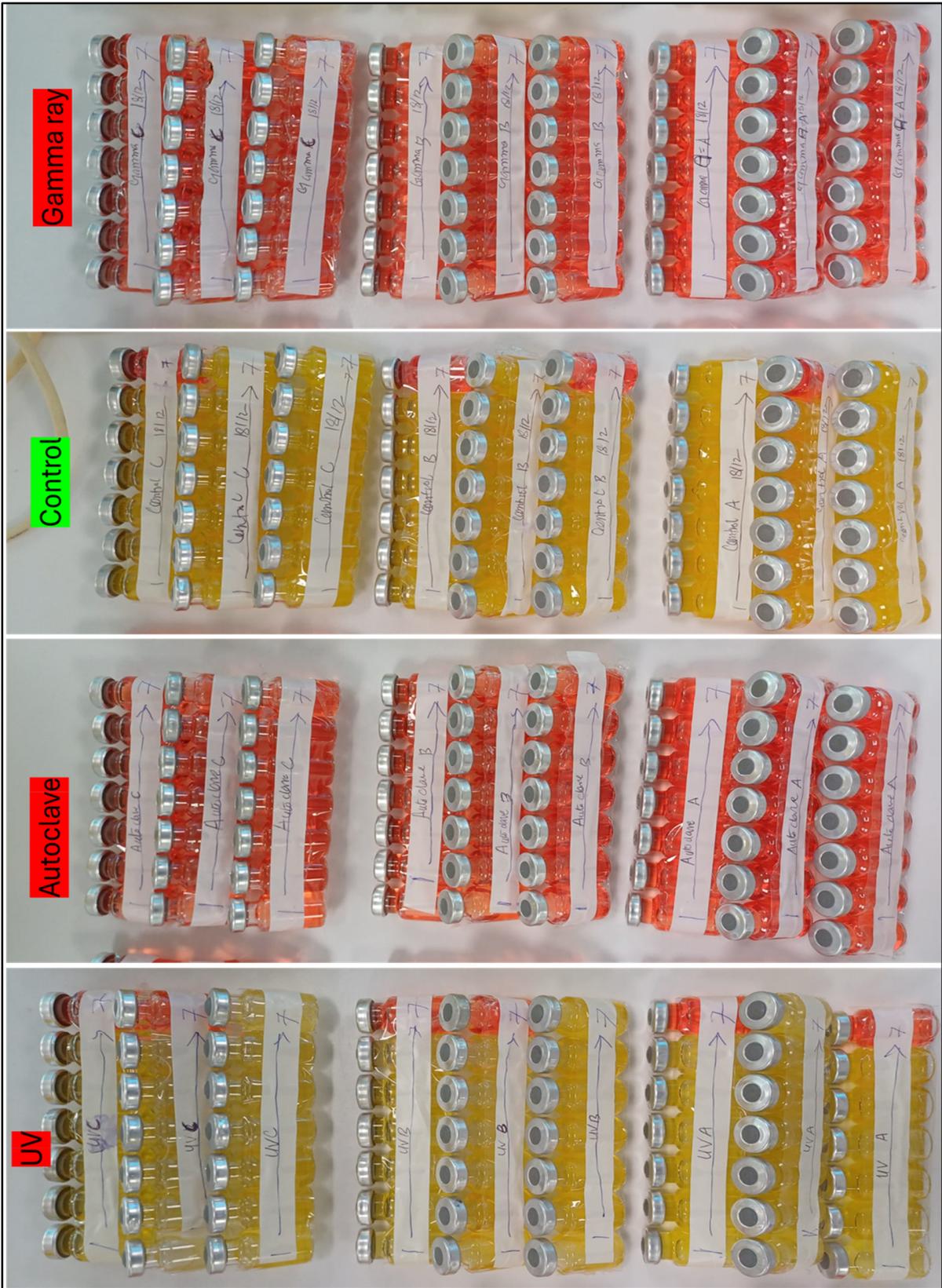


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129 **Figure S3** Illustrated difference between color of vials Control A, B, C (left), and Gamma ray A,  
 130 B, and C (right) samples

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134 **Figure S4** (A) Control from left ran parallel with Ethanol 95 wt% sterilization. (A-B) Another  
135 Control ran for Ethanol 75 wt%, Oven heating experiments, UV, and Autoclave sterilization  
136 techniques. (B) The second last Control was run in parallel with Gamma ray. There are 567  
137 vials each 10 ml illustrated in the Figure. We conducted each test at 7 serum concentrations, for  
138 example,  $10^{-1}$  to  $10^{-7}$ , and repeated it three times to determine if any deviation occurred in the  
139 values.

140 **Note:**-Red colour vials show no positive growth and 100% sterilization. Orange colour vials  
141 illustrate positive growth of microorganisms in the APB media solution.

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