1 2 3	Fluoride concentration in groundwater and relationship with the sodium (Na+), potassium (K+), (Ca <sub>2</sub> +) and magnesium (Mg <sub>2</sub> +)
4	
5	Fluoride concentration in groundwater
6	
7	Andrea Machado <sup>1,2</sup> ; Antonio Samudio-Oggero <sup>1,2,5</sup> *; Héctor D.
8	Nakayama <sup>1,5*</sup> ; Gustavo Brozón B. <sup>1</sup> ; Víctor González <sup>3</sup> ; Claudia Avalos <sup>1</sup> ;
9	Carlos Méndez <sup>1</sup> ; Andreas Ries <sup>4</sup> ; Heriberto Núñez <sup>1,5</sup> ; Carlos Enciso <sup>2</sup>
10 11	<sup>1</sup> Universidad Nacional de Asunción, Centro Multidisciplinario de Investigaciones Tecnológicas, Campus Universitario San Lorenzo, Central, Paraguay.
12 13	<sup>2</sup> Universidad Nacional de Asunción, Facultad de Ciencias Agrarias, Carrera de Ingeniería Ambiental, Campus Universitario San Lorenzo, Central, Paraguay.
14 15	<sup>3</sup> Universidad Católica Nuestra Señora de la Asunción, Campus Carapeguá, Paraguarí, Paraguay.
16 17	<sup>4</sup> Universidad Nacional de Asunción, Facultad de Politécnica, Grupo de Investigación Bio y Materiales, Campus Universitario San Lorenzo, Central, Paraguay.
18 19	⁵Red Paraguaya de Estudios de la Exposición a Fluoruros y el Tratamiento de Fluorosis de la Asociación de Universidades Públicas del Paraguay.
20	* <u>asamudio@rec.una.py</u> ; <u>hnakayama@rec.una.py</u>
21	Abstract
22	Groundwater is the largest freshwater reserve on the planet, and its quality plays a
23	fundamental role in human well-being and economic development. However, it
24	sometimes contains potentially harmful compounds, such as fluoride in high
25	concentrations, which has led to the implementation of quality standards to ensure
	1

26	water potability. This study evaluates the concentrations of fluoride in groundwater
27	from the state of Paraguarí, Paraguay, and studies the joint occurrence of fluoride
28	with elements such as calcium, magnesium, sodium, and potassium. Forty-one
29	water samples were collected from boreholes at different locations in the state of
30	Paraguarí. The results showed that 7% of the samples exceeded the maximum
31	concentration limit allowed by the Paraguayan Standard and the World Health
32	Organization, this means concentrations were above 2 mg/L; this was observed
33	mainly in the district of Caapucú. These findings indicate that 93% of the wells meet
34	potability standards. Additionally, fluoride was found to correlate with the sodium
35	content in groundwater. The results suggest the need for continuous monitoring
36	and implementation of effective fluoride reduction technologies, especially in areas
37	with elevated concentrations.

38 Keywords: Groundwater, Fluoride, Spatial distribution,

# 39 **1.Introduction**

Groundwater is an essential resource for socioeconomic progress, providing
approximately 50% of the world's drinking water and accounting for 98% of the
unfrozen freshwater [1]. Despite its importance, its quality can be compromised by
the presence of naturally occurring inorganic compounds, which, depending on the
geomorphology of the region, can be potentially harmful for human consumption
[2-3-4].

Fluoride is one of the most common natural contaminants in drinking water. In lowconcentrations, this ion is frequently added to drinking water to prevent dental

48 caries, but in high concentrations, it can cause dental and skeletal fluorosis,

49	affecting	public	health	[5-6-7].
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50	In 1981, it was discovered that fluoride forms very strong hydrogen bonds with
51	amide functional groups in proteins [8]. Proteins are long-chain peptide polymers
52	and their macromolecular structure is governed by hydrogen bonds formed
53	between the N-H group of one amide fragment and the carbonyl group of another
54	fragment on a parallel chain. So these amide hydrogen bonds are responsible for
55	protein folding; when fluoride interferes with these hydrogen bonds, i.e. breaks
56	them down, the characteristic protein configuration is altered and some specific
57	protein properties might be lost. For instance, enzymes with a distorted spatial
58	structure are sometimes considered as deactivated enzymes.
59	The double-stranded structure of the DNA is also stabilized hydrogen bonds
60	involving amide groups. Any disruption of the hydrogen bonding of DNA by fluoride
61	would certainly cause damage to the cell.
62	The effects of fluoride on the nervous system and endocrine glands of mammals,
63	including the pineal gland have been extensively studied [9-10]. Concerns have risen
64	that fluoride exposure leads to neuronal damage, such as Parkinson's disease,
65	Alzheimer's disease, and a reduced intelligence quotient. Several studies have
66	shown that fluoride exposure leads to oxidative stress and lipid peroxidation in the
67	brain. Moreover, the production of enzymes responsible for the production of
68	energy in mitochondria can be inhibited. As a consequence, memory, and learning
69	ability in animals and humans have been observed to be negatively affected [9-10-
70	11-12-13-14-15-16-17].

71	Fluoride occurs naturally in air, soil, water, and plants. The fluoride concentration in
72	groundwater depends on geological and climatic factors, such as the presence of
73	minerals like fluorspar, cryolite, micas, and clay minerals [3-18-19-20].
74	In addition, its relationship with other elements such as calcium, magnesium,
75	sodium, and potassium is crucial to understanding its dynamics. It has been shown
76	that high fluoride concentrations often correlate with high sodium and low calcium
77	levels, while the presence of calcium and magnesium can mitigate the toxic effects
78	of fluoride [21-22].
79	In the state of Paraguarí, Paraguay, access to groundwater is vital for rural
80	communities. However, until now there are no systematic studies that evaluate
81	fluoride concentrations in this region, which raises health concerns, given that some
82	areas of the country are known for elevated concentrations of this ion in
83	groundwater [23-24-25]. This knowledge gap needs to be targeted through research
84	that identifies fluoride levels and their potential risks to the population.
85	The World Health Organization (WHO) establishes a recommended maximum limit
86	of 1.50 mg/L for fluoride in drinking water, a value also adopted by the Paraguayan
87	Water Quality Standard NP 24 001 80. Several studies have shown that the
88	presence of fluoride in concentrations above this limit is associated with the
89	occurrence of fluorosis and other health problems, especially in mountainous and
90	arid regions, where water mineralization is more significant [4-26-27-28].
91	In this scenario, the present study focuses on the state of Paraguarí, where,
92	although symptoms of fluorosis have been observed in some areas, standardized
93	studies on groundwater quality have not yet been carried out. Consequently, this

- 94 research aims to evaluate the fluoride concentrations in different groundwater
- 95 samples of this region. In addition, we try to correlate fluoride concentrations with
- 96 the presence of other ions such as calcium, magnesium, sodium, and potassium.

# 97 2. Methodology

- 98 2.1 Geographic location of the state of Paraguarí
- The region studied is located in the southwest of the eastern region of 99 Paraguay (Fig 1), limited to the state of Paraguarí. This state is geographically 100 surrounded in the north by the states of Cordillera and Caaguazú, in the south by 101 102 the state of Misiones, in the east by the states of Guairá and Caazapá, and in the 103 west by the states of Central and Ñeembucú. Paraguarí has a total area of 8,705 km<sup>2</sup> and a population of 8,719 inhabitants. The region's climate is moderate-humid, 104 105 with all-year-round rainfall and hot summers. The average annual temperature and rainfall are 22.2 °C and 1,384 mm, respectively. The rainiest periods are March-April 106 107 and October-November.



### 115 Fig 1. Location of the sampling points in the state of Paraguarí.

# 116 **2.2 Sampling and parameters**

117	The study included 41 wells officially enabled and in use for consumption,
118	according to SENASA records, distributed in the Department of Paraguarí, in addition
119	to the criteria, according to recommendations from experts from the Paraguayan
120	Network of Studies of Fluoride Exposure and the Treatment of Fluorosis. of CEMIT,
121	include wells with a minimum depth of 100 m [29] and wells that are in areas with
122	people diagnosed with dental or bone fluorosis. The 41 sampled points were as
123	detailed in the following table (Table 1).

District	Number of sampling points	Codification
Acahay	4	AC
Саариси	3	Cu
Carapegua	6	CA
Escobar	1	ES
La Colmena	2	СО
Mbuyapey	2	MB
Paraguarí	2	PA
Pirayu	2	PI
Quiindy	2	QU
Quyquyho	2	QY
San R. González	5	RG
Sapucai	2	SA
Yaguarón	4	YA
Үbycui	4	YB
Total	41	

#### 124 Table 1. Sampling points for each district in the state of Paraguarí.

125

All wells were deeper than 100m. Samples for fluoride determination were stored in 100 mL sterile polyethylene bottles, while those used for the analysis of the other ions were stored in 1L bottles. All bottles were previously washed with a 10% HNO3 (nitric acid) solution, following the laboratory's quality and biosafety protocols. The samples were sealed and properly stored before transport. All analyses were
carried out at the Water Quality Laboratory of the Multidisciplinary Center for
Technological Research.

The fluoride concentration in the groundwater samples was determined 133 potentiometrically utilizing an ion-selective electrode, as indicated in the Standard 134 135 Methods for the Examination of Water and Wastewater (SM-4500-F-C). An OAKTON 136 ION 6+ potentiometer, previously calibrated, was employed. For samples with 137 significant fluoride content, the elements calcium (Ca2+) and magnesium (Mg2+) were determined by EDTA titration (SM-3500-Ca B y SM-3500-Mg B, respectively), 138 while sodium (Na+) and potassium (K+) concentrations were quantified by atomic 139 absorption spectroscopy using an air-acetylene flame (SM-3111 B). A Schimadzu, AA-140 141 7000 (APHA-AWWA-WEF, 2017) unit was used.

## 142 **2.3 Statistical analysis**

143 The obtained fluoride concentrations were compared with the maximum permitted

144 levels established by the Paraguayan Drinking Water Standard (NP 24 001 80). To

- 145 evaluate a possible correlation between fluoride concentration and the
- 146 concentrations of calcium, magnesium, sodium, and potassium, the Pearson
- 147 correlation coefficient (r) was calculated using IBM SPSS Statistics software. Then,
- 148 the spatial distribution of fluoride was determined using the Kriging geostatistical
- interpolation technique, implemented in ArcGIS 10.1. software.

# 150 **3. Results and discussion**

## **3.1 Evaluation of fluoride concentrations at different points**

	152	Fluoride concentrations	recorded in the groundwater	samples ranged from 0.01 to
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- 153 3.30 mg/L (Table 2). Most of the samples presented low fluoride concentrations,
- 154 except Yere and Montiel Potrero, belonging to the Caapucú district. The pH values
- ranged between 5.26 and 7.29. García et al. (2023) [4] mention that an alkaline
- environment (7.6-8.6) with high bicarbonate concentration favors fluoride
- 157 dissolution in groundwater. In contrast, Gutiérrez and Alarcón-Herrera (2021) [30]
- 158 noted that at lower pH, fluoride speciation tends to show a higher proportion of F
- 159 complex ions.
- 160 The World Health Organization states that dental fluorosis occurs more likely when
- 161 fluoride concentrations in drinking water exceed 1.5 mg/L, while skeletal fluorosis
- tends to manifest at concentrations higher than 3 mg/L. Both conditions result from
- 163 chronic exposure to elevated levels of fluoride.

N⁰	District	Code Well	Prof. (m)	F mg/L-1	рН	Cond ms/cm
1	Acahay	AC114	160	0.01	6.14	42.3
2	Acahay	AC166	120	0.00	5.26	51.1
3	Acahay	AC203	100	0.01	5.8	9.92
4	Acahay	AC222	150	0.01	6.39	78.4
5	Caapucu	CU121	102	3.20	6.74	331
6	Caapucu	CU143	126	3.30	7.29	594
7	Caapucu	CU144	121	2.50	6.03	43.7
8	Carapegua	CA085	132	0.00	6.18	30.3
9	Carapegua	CA110	126	0.02	5.67	41.7
10	Carapegua	CA133	150	0.01	5.97	34.7
11	Carapegua	CA187	122	0.01	5.79	47.7
12	Carapegua	CA191	150	0.01	7.28	436
13	Carapegua	CA221	120	0.14	6.9	183.9
14	Escobar	ES031	152	0.00	5.64	48.2
15	La Colmena	CO167	150	0.01	6.18	131.6
16	La Colmena	CO286	104	0.04	6.2	52.6
17	Mbuyapey	MB036	124	0.00	5.82	46.7
18	Mbuyapey	MB092	150	0.01	5.85	49.5
19	Paraguari	PA054	150	0.01	6	37.8

164 **Table 2. Fluoride concentrations and recorded parameters.** 

Paraguari	PA125	108	0.02	6.84	198.3
Pirayu	PI195	104	0.00	5.79	9.88
Pirayu	PI253	120	0.00	6.09	7.69
Quiindy	QU211	150	0.00	5.83	51.1
Quiindy	QU235	117	0.00	5.78	81.5
Quyquyho	QY018	153	0.18	5.7	10.17
Quyquyho	QY145	112	0.52	5.85	9.42
San R. Gonzalez	RG039	105	0.00	6.19	111.6
San R. Gonzalez	RG098	110	0.04	5.87	29.6
San R. Gonzalez	RG142	105	0.00	6.14	36.7
San R. Gonzalez	RG154	105	0.00	6.81	62.9
San R. Gonzalez	RG232	150	0.68	6.4	35.3
Sapucai	SA194	132	0.00	6.83	154.3
Sapucai	SA283	169.5	0.01	5.84	43.6
Yaguaron	YA044	105	0.00	5.78	134.7
Yaguaron	YA198	138	0.01	5.89	64.4
Yaguaron	YA247	120	0.00	6.52	163.3
Yaguaron	YA252	146	0.00	5.68	53.4
Ybycui	YB010	105	0.00	6.02	117.1
Ybycui	YB056	100	0.01	6.01	31.2
Ybycui	YB171	125	0.00	5.89	43
Ybycui	YB257	121	0.00	5.48	53.6
	Paraguari Pirayu Pirayu Quiindy Quiindy Quyquyho Quyquyho San R. Gonzalez San R. Gonzalez Yaguaron Yaguaron Yaguaron Yaguaron Yaguaron Yaguaron Yaguaron Yaguaron Yaguaron Yaguaron Yaguaron	ParaguariPA125PirayuPI195PirayuPI253QuiindyQU211QuiindyQU235QuyquyhoQY018QuyquyhoQY145San R. GonzalezRG039San R. GonzalezRG142San R. GonzalezRG142San R. GonzalezRG142San R. GonzalezRG232San R. GonzalezRG232SapucaiSA194SapucaiSA283YaguaronYA044YaguaronYA247YaguaronYA247YaguaronYA252YbycuiYB010YbycuiYB010YbycuiYB171YbycuiYB257	Paraguari         PA125         108           Pirayu         PI195         104           Pirayu         PI253         120           Quiindy         QU211         150           Quiindy         QU235         117           Quyquyho         QY018         153           Quyquyho         QY145         112           San R. Gonzalez         RG039         105           San R. Gonzalez         RG098         110           San R. Gonzalez         RG142         105           San R. Gonzalez         RG142         105           San R. Gonzalez         RG154         105           San R. Gonzalez         RG232         150           Sapucai         SA194         132           Sapucai         SA283         169.5           Yaguaron         YA044         105           Yaguaron         YA247         120           Yaguaron         YA247         120           Yaguaron         YA247         120           Yaguaron         YA247         120           Yaguaron         YA252         146           Ybycui         YB010         105           Ybycui         YB171<	ParaguariPA1251080.02PirayuPl1951040.00PirayuPl2531200.00QuiindyQU2111500.00QuiindyQU2351170.00QuyquyhoQY0181530.18QuyquyhoQY1451120.52San R. GonzalezRG0391050.00San R. GonzalezRG1421050.00San R. GonzalezRG1421050.00San R. GonzalezRG1541050.00San R. GonzalezRG2321500.68SapucaiSA1941320.00SapucaiSA283169.50.01YaguaronYA2471200.00YaguaronYA2471200.00YbycuiYB0101050.00YbycuiYB1711250.00YbycuiYB2571210.00	ParaguariPA1251080.026.84PirayuPl1951040.005.79PirayuPl2531200.006.09QuiindyQU2111500.005.83QuiindyQU2351170.005.78QuyquyhoQY0181530.185.7QuyquyhoQY1451120.525.85San R. GonzalezRG0391050.006.19San R. GonzalezRG0981100.045.87San R. GonzalezRG1421050.006.14San R. GonzalezRG1541050.006.81San R. GonzalezRG2321500.686.4SapucaiSA1941320.005.88YaguaronYA0441050.005.78YaguaronYA2471200.006.52YaguaronYA2521460.005.68YbycuiYB0101050.006.02YbycuiYB1711250.005.89YbycuiYB2571210.005.48

165 166

Considering the maximum allowable concentration of fluoride (1.5 mg/L) according 167 to the Paraguayan Drinking Water Standard (NP 24 001 80), 93% of the sampled 168 169 wells (38 samples) meet national and international standards. However, 7% of the samples (3 wells) exceed this limit, with concentrations higher than 2 mg/L. Figure 2 170 171 shows the fluoride concentrations obtained, where wells CU121, CU143 and CU144 172 stand out, which greatly exceeded the permitted limit. Well CU143 recorded the highest concentration, with 3.30 mg/L. 173 174 These results are consistent with those obtained by the Ministry of Environment and Sustainable Development [6] (MADES, 2012) in the region 175 "Cuenca Hídrica del arroyo San Lorenzo", where 95.38% of the samples presented 176 177 concentrations between 0 and 0.5 mg/L. Similarly, Diez Pérez et al. (2019) [26]

178	reported that 8% of the samples in the district of Loreto, Department of
179	Concepción, exceeded the allowable limit. However, the results differ from those
180	reported by DIGESA in Borja [25], located in the state of Guairá, where fluoride
181	concentrations ranged from 0 to 21.8 mg/L, significantly exceeding the
182	concentrations observed in this study. It is important to note that these results
183	coincide with the observations of Núñez (2018) [25], who noted the presence of
184	natural fluoride concentrations equal to or higher than 1.5 mg/L in the departments
185	of Alto Paraná, Concepción, San Pedro, and Paraguarí.
186	Rehman et al. (2022) [7] suggest that elevated fluoride concentrations may
187	be associated with the presence of minerals such as clay, mica, dolomite, and
188	limestone, hydrogeological features also observed in the region of this study.
189	
190	
191	
192	
193	



Fluoride concentration (mg/L)

# Fig 2. Fluoride concentrations registered according to the Paraguayan Standard NP24 001 80.

197

## **3.2 Description of the spatial fluoride distribution**

- 199 Figure 3 shows the spatial distribution of fluoride in the state of Paraguarí estimated
- 200 by the Kriging interpolation method. The areas with the lowest concentration are
- 201 located in the northeast, while the highest concentrations are found in the
- southwest, suggesting a relatively uniform distribution over much of the area. The
- 203 areas represented in light and dark blue tones correspond to areas with low fluoride
- 204 concentrations, within the permitted limits. In contrast, the dark blue areas indicate
- 205 concentrations that exceed the maximum allowable limit. These results are
- 206 consistent with those reported by the Ministry of Environment and Sustainable
- 207 Development (MADES, 2012) [6], which observed a uniform distribution of fluoride
- 208 in the "Cuenca Hídrica del Arroyo San Lorenzo", without the presence of elevated
- 209 concentrations. Similarly, Durán et al. (2017) [31] documented a heterogeneous
- 210 geographic distribution of fluorides in the province of Tucumán, Argentina,
- 211 differentiating between areas with deficit, excess, and concentrations within the
- 212 recommended values.





# 215 3.3 Correlation of fluoride with calcium, magnesium, sodium, and

potassium.

217 Figures 4 and 5 show graphically possible correlations between fluoride and other

218 compounds (Ca, Mg, K, Na) in the groundwater. Only samples with significant

219 fluoride concentration were considered (5 wells). Numerical values of the measured

- 220 concentrations are given in Table 3.
- 221
- 222
- 223





226

Fig 4. Graphical detection of correlations between fluoride and sodium or fluoride and calcium. The red line is a linear fit of the sodium concentration versus fluoride concentration.



230

231 Fig 5. Graphical detection of correlations between fluoride and magnesium or

232 fluoride and potassium.

233	As can be seen from these figures, only sodium exists in a linear concentration
234	dependence on the fluoride concentration. The red line in Figure 4 is a linear fit of
235	these concentration data with the slope of 15,23 mg sodium / 1 mg fluoride. This
236	can be expressed as millimol Na per millimol F-, dividing the mass values by the
237	corresponding atomic masses, i.e., 15.24 mg / 22.99 g/mol = 0.66 mmol, and
238	1 mg /9.499 g/mol = 0.11 mmol. This means that for every fluoride anion in
239	solution, there are 6 sodium cations present (0.66/0.11 = 6).
240	These results are comparable with the findings of Rehman et al. (2022) [7], who
241	reported a moderate correlation between fluoride and $Mg^{2+}$ , $Ca^{2+}$ , and $K^+$ .
242	Podgorski and Berg (2022) [21] suggest that increased fluoride along with high Ca <sup>2+</sup>
243	and Mg <sup>2+</sup> concentrations are likely related to evaporative conditions and increased
244	salinity, reflected in parameters such as electrical conductivity (EC), total dissolved
245	solids (TDS), and elevated Na $^{+}$ and Cl- levels. Huepallara et al. (2021) [29] also
246	observed a strong positive correlation between sodium and fluoride (r = 0.985).
247	Kashyap et al. (2021) [20] mention that several studies have reported a strong
248	positive correlation between F- and Na $^{+}$ and K $^{+}$ , whereas a negative correlation
249	between F- and Ca <sup>2+</sup> and Mg <sup>2+</sup> ions has been documented.
250	However, the dissolution of fluoride-containing minerals should generate a
251	positive correlation with Ca <sup>2+</sup> and Mg <sup>2+</sup> . This could be explained by the reverse ion
252	exchange process, where Na <sup>+</sup> present in an aquifer mineral is exchanged for Ca <sup>2+</sup>
253	and Mg <sup>2+</sup> cations in groundwater. Ali et al. (2016) [22] studied global al water
254	contamination by fluoride, multiple investigations are cited that evidence the
255	correlation of fluoride with various ions. Jabal et al. (2014) and Rao (2009) [32-33]

- reported a positive correlation between F- and Na<sup>+</sup>, Ca<sup>2+</sup>, and Mg<sup>2+</sup>, while studies
- by Liu et al. (2015) [34] documented negative correlations with Ca<sup>2+</sup>, Mg<sup>2+</sup>, and K<sup>+</sup>.

Code Well	F mg/L-1	Ca+2	Mg+2	Na+	K+				
CU143	3.30	67.86	14.362	45.42	2.67				
CU121	3.20	24.53	7.070	39.95	9.33				
RG232	0.68	2.45	0.976	3.44	3.06				
QY145	0.52	2.45	0.731	1.33	3.58				
MB036	0.00	2.45	0.976	1.79	2.76				
PI253	0.00	1.64	0.732	0.74	1.25				

### Table 3. Cation values in wells with high, medium and low fluoride concentrations.

259

### 260 Table 4. Pearson correlation.

	F-	Ca+2	Mg+2	Na+2	K+
Coefficiet e correlation	1	.854*	.905*	.987**	.609
Sig. (bilateral)		0.030	0.013	0.000	0.199
Ν	6	6	6	6	6

261 \*. The correlation is significant at the 0.05 level (two-sided)

262 \*\*. The correlation is significant at the 0.01 level (two-sided)

263

## **4. Conclusions**

265 This work emphasizes the importance of future studies to evaluate the dynamics of

266 other related elements. This work emphasizes the importance of future studies to

267 evaluate the dynamics of other related elements, such as calcium, magnesium, sodium, and

268 potassium, which could be influencing the distribution and concentration of fluoride in the

- 269 region.
- 270 This study on fluoride concentration in groundwater in the state of Paraguarí
- 271 reveals that most of the analyzed samples present fluoride levels within the limits

- allowed by Paraguayan and WHO regulations. However, a small percentage of wells,
- 273 mainly in the district of Caapucú, exceed these limits, posing a potential risk to
- 274 public health, particularly concerning dental and skeletal fluorosis. Therefore, it is
- 275 suggested that specific monitoring and treatment strategies, such as the application
- of efficient technologies for fluoride reduction in affected water sources, should be
- 277 implemented to ensure the sustainability of drinking water sources and protect the
- 278 health of local communities

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