

# 1 Persistent high-pressure magma storage beneath a near- 2 ridge ocean island volcano (Isla Floreana, Galápagos)

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## 11 ABSTRACT

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12 Volcanic evolution in ocean island settings is often controlled by variations in the chemistry and  
13 volumetric flux of magma from an underlying mantle plume. In locations such as Hawai'i or Réunion,  
14 this results in predictable variations in magma chemistry, the rate of volcanic activity, and the depth  
15 of magma storage with volcanic age and/or distance from the center of plume upwelling. These  
16 systems, however, represent outliers in global plume volcanism due to their high buoyancy flux,  
17 frequent eruptions, and large distance from any plate boundary. Most mantle plumes display clear  
18 interaction with nearby plate boundaries, influencing the dynamics of solid plume material in the  
19 upper mantle and the distribution of melt across regions of active volcanism. Yet, the influence of  
20 plume-ridge interaction and plume-ridge distance on the structure, characteristics, and evolution of  
21 magma storage beneath ocean island volcanoes remains under constrained. In this study, we  
22 consider the evolution of magmatic systems in the Galápagos Archipelago, a region of mantle plume

23 volcanism located 150-250 km south of the Galápagos Spreading Centre (GSC), focusing on the  
24 depth of magma storage during the eastward transport of volcanic systems away from the centre of  
25 plume upwelling. Geochemical analysis of gabbro xenoliths from Isla Floreana in the south-eastern  
26 Galápagos suggest that they formed at ~2-2.5 Ma, when the island was located close to the centre of  
27 plume upwelling. These nodules, therefore, provide rare insights into the evolution of volcanic  
28 systems in the Galápagos Archipelago, tracking variations in the magma system architecture as the  
29 Nazca plate carried Isla Floreana eastwards, away from the plume centre.

30 Mineral thermobarometry, thermodynamic modelling, and CO<sub>2</sub> fluid inclusion barometry reveal that  
31 Isla Floreana's plume-proximal stage of volcanic activity – recorded in the gabbro xenoliths – was  
32 characterized by the presence of high-pressure magma storage (> 25 km), below the base of the  
33 crust. In fact, we find no petrological evidence that sustained, crustal-level magma storage ever  
34 occurred beneath Isla Floreana. Our results contrast with the characteristics of volcanic systems in  
35 the western Galápagos above the current centre of plume upwelling, where mid-crust magma  
36 storage has been identified. We propose that this change in magmatic architecture of plume-  
37 proximal volcanic centres in the Galápagos – from high-pressure mantle storage at 2.5 Ma to mid-  
38 crustal storage at the present day – is controlled by the variations in plume-ridge distance. Owing to  
39 the northward migration of the GSC, the distance separating the plume stem and GSC is not  
40 constant, and was likely <100 km at 2.5 Ma, significantly less than the current plume-ridge distance  
41 of 150 – 250 km. We propose that smaller plume-ridge distances result in greater diversion of  
42 plume-material to the GSC, 'starving' the eastern Galápagos islands of magma during their initial  
43 formation and restricting the ability for these systems to develop long-lived crustal magma  
44 reservoirs.

## 45 1 INTRODUCTION

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46 Ocean island volcanoes, primarily formed through melting of anomalously hot, hydrous, or  
47 lithologically distinct mantle plumes, represent some of the most active volcanic systems in the  
48 world (Harpp and Weis, 2020; Herzberg et al., 2014; Matthews et al., 2021; Métrich et al., 2014; Neal  
49 et al., 2019; Sobolev et al., 2007; Weis et al., 2023). Owing to the deep-mantle origin of many mantle  
50 plumes, these volcanoes provide a snapshot into the dynamic nature of the Earth's mantle and the  
51 origin of large-scale mantle features identified through geophysical techniques (Cabral et al., 2013;  
52 Cottaar and Lekic, 2016; French and Romanowicz, 2015; Gleeson et al., 2021; Harpp and Weis, 2020;  
53 Jackson et al., 2008; Koppers et al., 2021). In addition, ocean island volcanoes can have significant  
54 impacts on local populations, infrastructure, and the global climate (Whitty et al., 2020; Wilson et  
55 al., 2014). Despite their scientific and societal importance, uncertainties remain regarding the  
56 growth and evolution of ocean island volcanoes, and the interplay between magma  
57 chemistry/storage and geodynamic setting (e.g., near-ridge vs intraplate; Cleary et al., 2020a; Geist  
58 et al., 2014a; Harpp and Geist, 2018a).

59 The magmatic architecture and volcanic evolution of ocean island volcanoes has traditionally been  
60 studied through petrological analysis of erupted volcanic products (e.g., lava and tephra) and  
61 analysis of the geodetic and seismic signatures of volcanic unrest (Bagnardi et al., 2013; Bell et al.,  
62 2021; Boudoire et al., 2019; Dayton et al., 2023; DeVitre et al., 2023b; Harpp and Geist, 2018; Neal et  
63 al., 2019; Stock et al., 2018). These approaches have demonstrated diversity in the characteristics of  
64 ocean island volcanoes, with variations in the composition and temperature of the mantle source,  
65 and the flux of magma into the lithosphere translating into dramatic differences in the chemistry and  
66 magmatic architecture of the volcanic plumbing systems (Baxter et al., 2023; Baxter and Maclennan,  
67 2024; Geist et al., 2014; Gleeson et al., 2020a; Maclennan, 2019). However, important questions  
68 remain, including the influence of geodynamic setting on the architecture of magmatic systems and  
69 the evolution of magmatic activity in near-ridge settings (Cleary et al., 2020). In addition, with the

70 growing consensus that magma reservoirs are dominated by crystal-rich, mushy domains, there is  
71 increasing recognition that petrological analysis of magmatic cumulates in addition to  
72 conventionally-examined erupted material can aid investigation of the chemical, thermal and  
73 structural history of ocean island magmatic plumbing systems (Cashman et al., 2017; Gao et al.,  
74 2022, 2016; Gleeson et al., 2020a; Horn et al., 2022; Sparks et al., 2019).

75 At ocean islands, direct access to the cumulate section of the underlying magmatic system is not  
76 often available. Instead, snapshots of the underlying mush are provided by cumulate nodules  
77 brought to the surface during volcanic eruptions (Clague and Bohrson, 1991; Holness et al., 2019;  
78 Lyons et al., 2007). These nodules range from melt-bearing mush fragments to fully solidified  
79 xenoliths and typically cover a range of mineralogical assemblages and bulk compositions (Gao et al.,  
80 2016; Horn et al., 2022; Stock et al., 2012). In many cases, cumulate xenoliths brought to the surface  
81 at post-shield volcanic systems (i.e. downstream of the main region of plume upwelling) record  
82 several hundred thousand to millions of years of activity at a single volcanic centre (Clague and  
83 Bohrson, 1991; Gao et al., 2016; Lyons et al., 2007). As a result, these cumulate xenoliths not only  
84 provide insights into the architecture of mush-dominated magmatic systems beneath ocean island  
85 volcanoes, but also the temporal evolution of the volcanoes from which they are collected.

86 In this study, we utilize recent developments in the analytical capabilities of Energy Dispersive  
87 Spectrometry (EDS) to extract quantitative, multi-scale petrological data from cumulate xenoliths in  
88 the Galápagos Archipelago. We couple this with depth constraints from thermodynamic models and  
89 fluid inclusion barometry to examine the formation and storage history of these nodules.

90 Specifically, we target Isla Floreana, a low melt-flux volcanic system located in the south-eastern  
91 Galápagos where ~2.5 Myr of magmatic activity is recorded in the composition of cumulate nodules  
92 found in lava and scoria deposits (Harpp et al., 2014; Lyons et al., 2007). We focus on a suite of  
93 gabbroic xenoliths that are hypothesized to have formed when the island was located close to the  
94 center of plume upwelling (2 - 2.5 Ma; Gleeson et al., 2020a; Lyons et al., 2007a). Our new data,

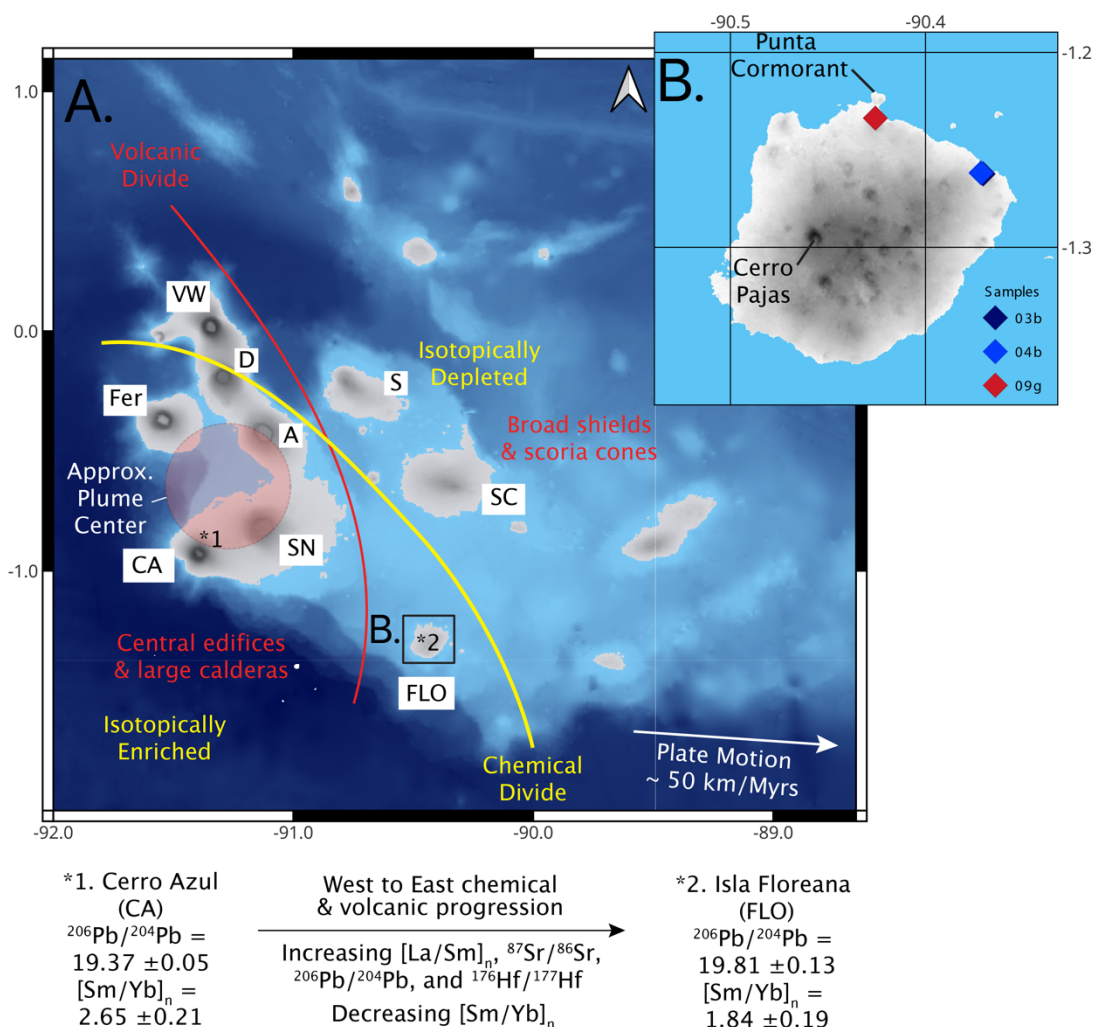
95 alongside recent analysis of the present-day Isla Floreana magmatic system (Gleeson et al., 2020a,  
96 2022), is used to evaluate the evolution of an ocean island magmatic system in a near-ridge setting,  
97 focusing on the depth of magma storage. Furthermore, comparison of our results to active volcanic  
98 centres in the present-day western Galápagos provides insights into the role of geodynamic settings  
99 in the construction of magmatic plumbing systems, owing to the migration of the Galápagos mantle  
100 plume away from the Galápagos Spreading Centre over the last 5 Myr (Mittelstaedt et al., 2012).

## 101 2 GEOLOGICAL BACKGROUND

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### 102 2.1 VOLCANIC HISTORY AND TECTONIC SETTING

103 The Galápagos Archipelago, located ~1000 km off the western coast of Ecuador, represents one of  
104 the most volcanically active regions in the world. Magmatic activity in the Galápagos is a  
105 consequence of adiabatic decompression melting in a chemically and lithologically heterogeneous  
106 mantle plume that is likely ~50-150°C hotter than the surrounding ambient mantle (Gibson et al.,  
107 2015; Hooft et al., 2003). Seismic tomography indicates that the current centre of plume upwelling is  
108 located beneath the islands of Isabela and Fernandina in the western Galápagos (Fig. 1), which are  
109 the most volcanically active islands in the archipelago (eruptions occur every 2-3 years on average;  
110 Villagómez et al., 2014). However, historical eruptions have also been documented in the eastern  
111 and northern Galápagos, on the islands of Santiago (1759, 1904 - 1906), Pinta (1928) and Marchena  
112 (1991), and volcanic activity elsewhere in the eastern Galápagos has been shown to be long-lived  
113 (Global Volcanism Program, 2013). For example, volcanic activity on the island of San Cristobal in the  
114 eastern Galápagos covers a period of more than 2 Myrs, with the most recent eruption dated at ~9  
115 ka (Mahr et al., 2016). Holocene to late Pleistocene eruptions have also been identified at several of  
116 the other eastern Galápagos islands, including Isla Floreana on the southern margin of the  
117 archipelago (~26 ka; Harpp et al., 2014a).



**Figure 1** – Map of the Galápagos Archipelago with the boundaries between different morphological and chemical domains indicated by the red and yellow lines, respectively. **A.** Topographic and bathymetric map of the Galápagos, data from GEBCO (General Bathymetric Chart of the Oceans). Plate motion estimate taken from Harpp et al. (2014). **B.** Topographic map of Isla Floreana using high resolution topographic data from the Terra Advanced Spaceborne Thermal Emission and Reflection Radiometer (ASTER) Global Digital Elevation Model (GDEM) Version 3 (ASTGTM) (~30m resolution). Both **A.**, and **B.** are created using the World Geodetic Survey (1984) Geographic Coordinate System with an equirectangular projection system. The mean and standard deviation of the  $^{206}\text{Pb}/^{204}\text{Pb}$  and  $[\text{Sm}/\text{Yb}]_n$  ratios, key chemical parameters that can be used to differentiate between the volcanic systems of Cerro Azul and Isla Floreana, are displayed below alongside the key characteristics defining the chemical progression from west to east along the southern margin of the archipelago. VW – Wolf; CA – Cerro Azul, SN – Sierra Negra, FLO – Isla Floreana, Fer – Isla Fernandina, D – Darwin, A – Alcedo, S – Santiago, SC – Santa Cruz.

118

119 The wide geographic distribution of recent volcanic activity in the Galápagos is likely related to  
 120 variations in lithospheric thickness across the archipelago (Gibson and Geist, 2010), extensional  
 121 faulting on the eastern islands (Schwartz et al., 2014), and the near-ridge tectonic setting. The

122 Galápagos mantle plume is centered roughly 150 – 250 km south of the east-west trending  
123 Galápagos Spreading Centre (GSC), a mid-ocean ridge spreading centre separating the Nazca and  
124 Cocos tectonic plates. Transfer of material from the Galápagos plume stem to the GSC (whether in  
125 the solid or melt phase) has a clear influence on the composition of erupted basalts along the ridge  
126 and may contribute to the volcanism observed in the Northern Galápagos Volcanic Province  
127 (Gleeson and Gibson, 2021; Mittal and Richards, 2017).

128 The distance between the Galápagos mantle plume and the GSC has not been constant over the last  
129 5 Myrs. Owing to the eastward motion of the Nazca tectonic plate and the north-east motion of the  
130 Cocos plate, the GSC is migrating north, away from the centre of Galápagos plume upwelling  
131 (Mittelstaedt et al., 2012; Wilson and Hey, 1995). In fact, ~5 – 12 Myr ago, the Galápagos mantle  
132 plume was likely located directly beneath the GSC, in a tectonic situation analogous to modern  
133 Iceland (Wilson and Hey, 1995). Since then, the separation distance between the mantle plume and  
134 spreading center has increased, with its northward migration punctuated by a series of southward  
135 ridge-jumps on the eastern GSC (the most recent of which occurred at ~1 Ma) that resulted in the  
136 formation of the Galápagos Transform Fault at ~91°W (Mittelstaedt et al., 2012). As a result, when  
137 the eastern Galápagos volcanic islands of Floreana and Santa Cruz were first constructed (at around  
138 2 – 3 Ma) the Galápagos mantle plume was located up to ~100 km closer to the GSC than it is at the  
139 present day. The 2-3 Myr old eastern Galápagos and the modern western Galápagos, therefore,  
140 formed in different geodynamic settings, which might have had a substantial influence on the  
141 characteristics of magma storage (Cleary et al., 2020; Harpp and Geist, 2018).

## 142 2.2 VARIATIONS IN VOLCANIC STYLE AND MORPHOLOGY

143 Across the Galápagos Archipelago, there are variations in the style of magmatic activity, which result  
144 in key chemical and morphological differences between the various volcanic centres. Broadly, these  
145 differences can be used to define two volcanic regions within the Galápagos Archipelago; a western  
146 volcanic domain with regular activity focused at 6 central shield volcanoes and an eastern domain

147 with less frequent eruptions and more spatially-distributed eruptive activity (Harpp and Geist,  
148 2018a; Fig. 1).

149 The western volcanic region, which encompasses the islands of Isabela and Fernandina, is  
150 characterized by the presence of 6 large shield volcanoes and the dissected shield of Volcán Ecuador  
151 on the northern margin of Isla Isabela (Geist et al., 2014, 2002). These volcanoes range from ~1100  
152 m to ~1700 m in height and are characterized by the presence of a large central caldera and an  
153 inverted soup-bowl morphology (Bernard et al., 2019; Geist et al., 2005; Naumann and Geist, 2000;  
154 Fig. 1c). The major element, trace element, and isotopic variations observed in erupted material at  
155 each of the western volcanic centers is typically relatively narrow (compared to the eastern  
156 Galápagos islands), indicating that mantle-derived magmas are efficiently mixed in their magmatic  
157 systems and buffered over a narrow range of temperatures by the regular input of new, mantle-  
158 derived magma into the system (Geist et al., 2014, 1995). Recent barometric work, using both  
159 petrological and geophysical monitoring data, indicates that the bulk of magma storage occurs in the  
160 mid-to-lower crust (7-15 km; Higgins and Stock, 2024; Stock et al., 2018), with a secondary magma  
161 storage region present at shallow depths beneath most of the western Galápagos shields (Bagnardi  
162 et al., 2013).

163 The eastern volcanic region encompasses the islands of Santa Cruz, Santa Fe, Floreana, Española, San  
164 Cristóbal and Santiago (Harpp and Geist, 2018). These islands are characterized by shallow slope  
165 gradients, distributed volcanism, monogenetic scoria cones, and lack of large central edifices (Fig.  
166 1d). There is little to no evidence to indicate the presence of buried calderas beneath the surface of  
167 any of the eastern Galápagos islands, which has been presented as evidence that these systems are  
168 unlikely to represent the late-stage evolution of a system analogous to the current western  
169 Galápagos shield volcanoes (Cleary et al., 2020). Unlike the western Galápagos shield volcanoes,  
170 there is a substantial major, trace element and isotopic heterogeneity in the erupted products of  
171 each eastern volcanic island (Bow and Geist, 1992; Harpp et al., 2014; Harpp and Geist, 2018). This



172 heterogeneity has been hypothesized to occur due to the lower flux of mantle-derived melts into  
173 these islands (relative to the western shields), which is insufficient to form a centralized, thermally  
174 buffered mush system capable of homogenizing chemical variations from incoming mantle melts  
175 (Gleeson et al., 2020a; Harpp and Geist, 2018). Instead, the magmatic systems are likely ephemeral,  
176 poorly connected, and are characterized by a range of melt temperatures and compositions. Melt  
177 inclusion and clinopyroxene-based barometry from Isla Floreana, which shows the greatest degree  
178 of compositional heterogeneity anywhere in the Galápagos, indicates that magma storage beneath  
179 the eastern Galápagos islands is substantially deeper than beneath the western Galápagos shield  
180 volcanoes and likely exceeds the depth of the Moho (~16 km; Gleeson et al., 2020a, 2022).

### 181 2.3 GALÁPAGOS PLUME HETEROGENEITY

182 The wide distribution of Holocene and late Pleistocene volcanism in the Galápagos has been used to  
183 unravel the compositional heterogeneity in the upwelling mantle plume (Geist et al., 1988; White et  
184 al., 1993). Several different mantle components have been identified or proposed, including  
185 enriched mantle reservoirs with highly radiogenic Pb and Sr isotope ratios (most prominently seen  
186 on Isla Floreana; Harpp et al., 2014a; White et al., 1993), primordial mantle domains retaining  
187 unradiogenic He isotope signatures ( $^3\text{He}/^4\text{He}$  up to ~29 R/R<sub>A</sub>; Isla Fernandina; Kurz et al., 2009; Kurz  
188 and Geist, 1999), and a depleted component whose origin (intrinsic plume material vs entrained  
189 upper mantle) remains debated (Gibson et al., 2012; Harpp and White, 2001; Hoernle et al., 2000).

190 Overall, at least 4 chemically and/or lithologically distinct components have been proposed (Geist et  
191 al., 1988; Gleeson et al., 2021, 2020b; Gleeson and Gibson, 2019; Harpp and White, 2001; Vidito et  
192 al., 2013). These components have a complex spatial arrangement that is long-lived on timescales of  
193 several 10s of Myrs (Hoernle et al., 2000). At the most basic level, the Galápagos can be split into an  
194 isotopically depleted north-eastern domain – where basalts have isotopic signatures that are similar  
195 to the global MORB field – and an isotopically enriched south-western domain (Gleeson et al., 2021;  
196 Harpp and Weis, 2020). The boundary separating these two domains is sub-parallel to the boundary

197 separating the western and eastern volcanic regions (yellow line, Fig. 1a). As a result, Volcán Wolf on  
198 northern Isabela falls within the western volcanic region (Harpp and Geist, 2018), but is chemically  
199 more closely related to volcanic systems in the north-eastern Galápagos (Harpp and Weis, 2020).  
200 Conversely, Isla Floreana in the southern Galápagos forms part of the eastern volcanic region (gentle  
201 slopes and an abundance of monogenetic scoria cones) yet displays the most radiogenic Sr and Pb  
202 isotope signatures of any volcanic island in the Galápagos and is thus associated with the chemically  
203 enriched south-western domain (Gleeson et al., 2020a; Harpp et al., 2014; Harpp and Weis, 2020).  
204 The chemical asymmetry of the Galápagos Archipelago is similar to the presence of the Loa and Kea  
205 geochemical trends in Hawai'i (Harpp and Weis, 2020). The main difference between the two  
206 systems is that the chemical asymmetry in the Galápagos is not orientated parallel to the direction  
207 of plate motion. As this chemical asymmetry in the Galápagos is long-lived on timescales >15 Myrs  
208 (Hoernle et al., 2000), the isotopic and trace element composition of erupted material at the surface  
209 of each individual volcanic system will change through time as the Nazca Plate moves east-southeast  
210 above the mantle plume. The anticipated chemical evolution of any particular volcanic system is  
211 controlled by the position of the volcano with respect to latitude; for example, in a central strip  
212 spanning west to east across the archipelago (e.g., Isla Fernandina, Volcán Alcedo, Isla Pinzón, and  
213 Isla Santa Cruz) the mantle source region will move across the boundary separating the enriched  
214 south-western domain and the depleted north-eastern domain leading to a temporal progression  
215 from enriched to depleted isotopic signatures (Wilson et al., 2022). In contrast, along the southern  
216 margin of the Galápagos platform (Volcán Cerro Azul, Volcán Sierra Negra, and Isla Floreana) the  
217 west-to-east progression is characterized by variations in the degree of enrichment and differences  
218 in key trace element and isotopic parameters (Fig. 1). Specifically, Isla Floreana basalts contain more  
219 radiogenic Sr and Pb isotope signatures relative to Volcán Cerro Azul and Volcán Sierra Negra, as well  
220 as offsets in many trace element ratios (e.g.,  $[Sm/Yb]_n$ , where n indicates normalization to the  
221 primitive mantle estimate of Sun and McDonough, 1989; Fig. 1; Harpp et al., 2014a). Consequently,  
222 volcanic evolution over millions of years at the southern margin of the Galápagos platform – the

223 region targeted in this study – should be associated with a systematic shift to more enriched isotopic  
224 signatures and concave Rare Earth Element patterns (high  $[La/Sm]_n$  but low  $[Sm/Yb]_n$  ratios).

## 225 3 METHODS

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226 The gabbroic xenoliths analysed in this study represent 2 – 5 cm diameter nodules collected from  
227 scoria cones in the north-east corner of Isla Floreana (17MMSG03b, 17MMSG04b; hereafter 03b and  
228 4b) and small fragments (exposed areas of  $\sim 0.5\text{-}3\text{ cm}^2$ ) found in the core of a blocky lava flow near  
229 Punta Cormorant on Isla Floreana's northern coast (17MMSG09; Fig. 1). Prior to analysis, all samples  
230 were cut into small blocks  $<1\text{ cm}$  thick with exposed areas ranging from  $<0.5\text{ - }5\text{ cm}^2$ . Lava samples  
231 containing small gabbroic fragments (17MMSG09; hereafter 09g1, 09g2, and 09g3) were cut to  
232 maximize the surface exposure of the gabbro. The cut samples were ground down using various  
233 grades of silicon carbide grit and then prepared for analysis by polishing the exposed surface areas  
234 with a  $1\text{ }\mu\text{m}$  diamond suspension.

### 235 3.1 FEG-SEM ANALYSIS

236 The polished gabbroic blocks were carbon coated to a thickness of  $\sim 10\text{ - }13\text{ nm}$  and analyzed using  
237 both the Zeiss Sigma HD Field Emission Gun Scanning Electron Microscope (FEG-SEM) at Cardiff  
238 University School of Earth and Environmental Science and the JEOL JSM IT800HL FEG-SEM at the  
239 University of California, Berkeley. Analyses were performed over a range of length-scales, from  
240 whole-sample compositional maps to high-precision analysis of major and minor elements in  
241 linescans and individual point analyses.

#### 242 3.1.1 Quantitative compositional mapping

243 Quantitative compositional maps of the exposed gabbroic regions were produced for all samples  
244 using the Cardiff Zeiss Sigma HD FEG-SEM and two Oxford Instruments X-Max 150 Energy Dispersive  
245 Spectrometers. Maps were created with a pixel size of  $25\text{ X }25\text{ }\mu\text{m}$  using a beam current of  $\sim 1.1\text{ nA}$

246 (determined through the choice of aperture size – 60  $\mu\text{m}$ ) and an accelerating voltage of 20 kV. To  
247 reduce background noise and increase spectral resolution, the signal process time was set to the  
248 AZtec preset 3 (~2 microseconds per X-ray). The dwell time was 0.05 seconds for each pixel, and a  
249 quantitative chemical composition was generated using a predefined calibration file (determined  
250 through analysis of mineral and metal standards; see Supplementary Material), and a XPP matrix  
251 correction routine (implemented through the associated AZtec software).

252 To assess the accuracy and precision of our map data, 15 mineral standards were mapped within the  
253 same analytical session as our gabbroic xenoliths. The estimated concentrations of all elements >2  
254 wt% are typically within 2% of the published standard value. Relative offsets between predicted and  
255 published values for oxides with lower concentrations are larger, in part owing to uncertainties in  
256 the published standard compositions and the presence of minor element heterogeneity within these  
257 samples. The analytical precision for each pixel was low (relative  $1\sigma$  precision of >39% for Na in  
258 Diopside – Astimex standard), but is substantially improved by binning multiple pixels. For example,  
259 generating 1000 compositions of each mapped standard by averaging the composition of 25 pixels in  
260 a randomly selected 5-by-5 grid indicates that the analytical precision of Na in Diopside  
261 (concentration of 0.3 wt%) in this binned region is ~7.5% (calculated as the relative  $1\sigma$  value). For  
262 elements present in concentrations > 1 wt% (e.g., Ca in Diopside), the analytical precision for each  
263 pixel is between 10% and 2%, improving to 3% to <1% for a 5-by-5 pixel region.

### 264 3.1.2 Map Data processing

265 Quantitative EDS mapping provides a wealth of information for each of the gabbroic xenoliths. Using  
266 this data, we can identify the major phases within a sample, constrain the compositional distribution  
267 of each phase, and determine the bulk sample composition. We used the Python3 package  
268 mineralML to identify the major phases present in each sample (Shi et al., 2023). mineralML uses  
269 Bayesian neural networks, trained on over 86,000 analyses of 17 minerals (data compiled from  
270 literature) and validated upon ~1 million analyses (data from EarthChem, GEOROC, and LEPR), to

271 predict the mineral phase for each chemical composition provided (i.e., every map pixel). mineralML  
272 also supplies the probability of classification for each analysis; we filter out pixels where the  
273 predicted probability is less than 0.6. The primary effect of this filtering step is the removal of pixels  
274 whose analytical totals are < 90 wt%, typically representing cracks and/or holes where high-quality  
275 analyses were not possible.

276 Once the mineral abundances and mean compositions have been determined (Fig. 2; Supplementary  
277 Information) a ‘bulk-rock’ composition can also be calculated by determining the weighted average  
278 of all pixels assigned a phase by mineralML:

$$279 \quad C = \frac{\sum_{i=0}^n C_i \times \rho_i}{\sum_{i=0}^n \rho_i}$$

280 Where  $C$  represents the bulk concentration of the chosen oxide;  $C_i$  indicates the measured  
281 concentration in pixel  $i$ ; and  $\rho_i$  represents the estimated density of the mineral located at each pixel  
282 (based on the identified mineral phase; see Supplementary Material). To estimate the uncertainty  
283 associated with these bulk concentration estimates, we generate 200 possible compositions for  
284 samples 03b and 04b where an area accounting for >33% of the total sample were randomly  
285 selected for each calculation. In general, the uncertainties estimated by this method are very low,  
286 typically less than 2% (relative  $1\sigma$  variation) for most oxides.

287 We consider the compositions estimated from samples 03b and 04b to be representative of their  
288 bulk-rock compositions. This is supported by the similarity between the SEM-derived bulk-rock  
289 estimates presented here and the bulk-composition of other Isla Floreana gabbros previously  
290 determined by XRF analysis (Lyons et al., 2007; see Supplementary Material). Furthermore, our error  
291 analysis indicates that samples 03b and 04b are homogeneous on scales smaller than the analyzed  
292 areas (see Supplementary Material). Nevertheless, we caution that the composition of samples  
293 09g1, 09g2, and 09g3 must be treated with care, as these samples are characterized by relatively

294 small areas ( $<0.5 \text{ cm}^2$ ), and the presence of heterogeneity on scales smaller than the pixel resolution  
295 of our large-scale maps (see below).

### 296 3.1.3 Qualitative, high-resolution chemical mapping

297 To supplement the large-scale maps and provide more information about any small-scale  
298 heterogeneity within each sample, qualitative chemical maps of orthopyroxene-clinopyroxene grain  
299 boundaries were also collected on the Cardiff Zeiss Sigma HD FEG-SEM. No standards were mapped  
300 alongside these high-resolution maps and so quantitative chemical compositions were not  
301 calculated. Each map was collected under identical analytical conditions to the large-scale maps, but  
302 with a pixel size of  $1.5 \times 1.5 \mu\text{m}$  (Fig. 3).

### 303 3.1.4 Linescans

304 In addition to the qualitative maps of clinopyroxene-orthopyroxene grain boundaries, EDS linescans  
305 were collected, using the Zeiss Sigma HD FEG-SEM and 2 X-Max EDS detectors, to provide  
306 quantitative chemical information of these regions. Five passes, with a dwell time of 3 seconds per  
307 spot on each pass, were carried out along each line (ranging from 140 to 220  $\mu\text{m}$  in length with a  
308 spatial resolution of  $1.5 \mu\text{m}$ ). Data collection was performed using a 1.1 nA beam current, a 20 kV  
309 accelerating voltage, and a process time of 3 (AZtec preset value). Quantification was performed  
310 using the same calibration file that was used for individual spot analyses (see Supplementary File),  
311 which was constructed using point analyses of a range of mineral and metal standards. To check the  
312 accuracy and precision of analyses, linescans were also performed on two clinopyroxene standards.  
313 Data from these standards indicate that the precision of analyses ranges from  $\sim 5\%$  ( $1\sigma$  relative  
314 standard deviation) for minor components such as  $\text{Al}_2\text{O}_3$  and  $\text{Na}_2\text{O}$  (0.6 wt% and 0.41 wt%  
315 respectively) to  $<0.5\%$  ( $1\sigma$  relative standard deviation) for  $\text{SiO}_2$  and  $\text{CaO}$ . Measured compositions of  
316 the secondary standard materials are typically within 1% of the preferred standard values.

317 3.1.5 Point Analyses

318 In addition to compositional maps and linescans, we perform high-precision point analyses of  
319 clinopyroxene, orthopyroxene and plagioclase crystals in all the Isla Floreana gabbros. The major and  
320 minor element composition of pyroxene crystals was determined through EDS analysis (using two X-  
321 Max 150 detectors) on the Cardiff Zeiss Sigma HD FEG-SEM. For these analyses, the beam was  
322 rastered over an area of 1  $\mu\text{m}$  by 1.5  $\mu\text{m}$  and analyses were performed using a 1.1 nA beam current  
323 and a 20 kV accelerating voltage. As with the compositional maps and EDS linescans, the process  
324 time was set as the AZtec preset 3, and a 15 second live dwell time was used for all EDS point  
325 analyses. The EDS calibration file was constructed via analyses of appropriate mineral and metal  
326 standards (see supplementary file).

327 The composition of plagioclase crystals hosted in the Isla Floreana gabbros was determined through  
328 EDS-WDS (Wavelength Dispersive Spectrometry) analysis on the JEOL JSM IT800HL FEG-SEM at UC  
329 Berkeley. The JEOL FEG-SEM provides a greater range in possible beam conditions than the Zeiss  
330 SigmaHD FEG-SEM in Cardiff, providing access to higher beam currents ( $\sim 100$  nA) suitable for minor  
331 element analysis by WDS. The concentration of most major and minor elements in the plagioclase  
332 crystals (i.e.,  $\text{SiO}_2$ ,  $\text{Al}_2\text{O}_3$ ,  $\text{FeO}_t$ ,  $\text{CaO}$ ,  $\text{Na}_2\text{O}$ , and  $\text{K}_2\text{O}$ ) was determined through EDS analysis using a  
333 single Ultim-Max 100  $\text{mm}^2$  detector. During analysis the beam was rastered over a 1-by-1.5  $\mu\text{m}$  area  
334 using a beam current of 2 nA and an accelerating voltage of 20kV. The process time for these  
335 analyses was set at the Aztec process time 5 ( $\sim 10$  microseconds). Following EDS analyses the beam  
336 current was increased to 100 nA and the concentration of MgO (and  $\text{K}_2\text{O}$  in some crystals), at the  
337 location of each EDS analysis, was determined via WDS using an Oxford Instruments Wave detector.  
338 Counts of Mg were collected for 40 seconds on peak using a TAP crystal, and background counts  
339 were collected for 20 seconds on either side of the peak. When measured, counts of K were also  
340 collected with a 40 second peak count time and 2\*20 second background count time using a PET  
341 crystal. Analysis of MgO and  $\text{K}_2\text{O}$  via WDS were calibrated through analysis (at 10nA) of a synthetic  
342 Forsterite standard and an Orthoclase standard, respectively.

343 Analyses of secondary standard materials, including clinopyroxene, plagioclase and garnet, were  
344 performed at regular intervals (~2 – 3 hours) to determine the accuracy and precision of analyses  
345 (see Supplement for full list of standard sources). These data demonstrate that our EDS analysis is  
346 typically accurate to within 1% for high concentration elements (i.e., > 2 wt%) within the secondary  
347 standard materials. Outputted error estimates (determined by the Aztec software), and the  
348 variability in the estimated composition of secondary standard materials, indicates that the  $1\sigma$   
349 relative precision of all EDS analyses is <0.5 - 1 % for elements present in high concentrations (> 2  
350 wt%) and increases to ~10-15% for analyses of minor elements (0.2 – 1 wt%). The relative  $1\sigma$   
351 precision of WDS analyses is estimated to be <3% for MgO in plagioclase (at ~0.12 wt%) and <1.5%  
352 for K<sub>2</sub>O (at 0.12 wt%). For all analyses we consider the detection limit to equivalent to  $3\sigma$ .

### 353 3.2 LASER ABLATION INDUCTIVELY COUPLED PLASMA MASS SPECTROMETRY

354 Clinopyroxene trace element abundances were determined using an ASI RESOLUTION 193 nm Laser  
355 Ablation system coupled to an Agilent 8900 triple quadrupole ICP-MS hosted in the CELTIC  
356 laboratory at Cardiff University. Data was collected for 2 minutes per analysis, including a 20 second  
357 delay prior to laser initiation, 60 seconds of data collection with the laser active, and a 40 second  
358 wash-out period. A 4.5 J/cm<sup>2</sup> fluence, a 10 Hz repetition rate, and an 80  $\mu$ m spot size was used for all  
359 analyses.

360 Data processing was carried out using the LaserTram software (Lubbers et al., 2021) with BCR-2g as  
361 the calibration standard. <sup>43</sup>Ca was used as the internal standard for all analyses and the accuracy of  
362 analysis was tracked through regular measurements of multiple secondary standards (Jochum et al.,  
363 2005). These analyses indicate that all elements reported here return values within ~5% of the  
364 published standard values. The relative  $1\sigma$  precision of analysis is typically <2 – 3% for most  
365 elements, although it is important to note that the concentration of most trace elements in the  
366 clinopyroxene unknowns are substantially lower than those in the glass secondary standards. Repeat  
367 analysis of clinopyroxene cores from our gabbroic samples indicates that the relative  $1\sigma$  analytical



368 precision is <5% for all light and middle Rare Earth Elements (REEs), increasing to ~8 – 10% for the  
369 heavy REEs.

### 370 3.3 FLUID INCLUSION ANALYSIS

371 The density of CO<sub>2</sub> in fluid inclusions (FIs) can provide a powerful barometer in CO<sub>2</sub>-dominated  
372 magmatic systems as CO<sub>2</sub> density is strongly related to the pressure and relatively insensitive to  
373 temperature (Hansteen and Klügel, 2008; Wieser et al., 2025). We quantify CO<sub>2</sub> densities in  
374 clinopyroxene and plagioclase-hosted FIs using microthermometry and calibrated Raman  
375 Spectroscopy following the methods of DeVitre and Wieser (2024). To isolate crystals containing FIs  
376 we carefully cut a small section (~5 mm x 5 mm x 1mm) of samples 03b and 04b following  
377 completion of EDS chemical analysis. These sections were then crushed and individual clinopyroxene  
378 and plagioclase crystals were picked from the crushed material and examined for the presence of  
379 FIs. These crystals were mounted in crystal bond and ground down using 1200 and 2500 grade  
380 silicon carbide paper. Crystals with visible fluid inclusions were then polished using 10000 grade  
381 paper and 1µm 3M polishing paper before they were flipped and polished on the other side. Once a  
382 thin wafer (< 300 µm) had been prepared with FIs clearly visible within ~100 µm of the surface,  
383 crystals were extracted from the crystal bond using a hot plate and Acetone and placed into  
384 separate vials. We primarily target fluid inclusions with no visible evidence of decrepitation. Most  
385 fluid inclusions in sample 04b appear as isolated inclusions with no clear connection to trails or  
386 fracture planes and are thus labelled as ‘apparent primary fluid inclusions’. Many of the inclusions  
387 measured in 03b are interpreted as secondary as they are located in trails (see Supplementary  
388 Material for full textural characterization).

389 Heating and cooling experiments were performed on a Linkam THMSG600 heating and freezing  
390 stage at UC Berkeley. Temperature was calibrated using the melting temperature of a pure H<sub>2</sub>O fluid  
391 inclusion (0°C) and the melting temperature of CO<sub>2</sub> (-56.6 °C) in a H<sub>2</sub>O-CO<sub>2</sub> fluid inclusion. We  
392 recorded the melting point/interval and homogenization temperature of each fluid inclusion with a

393 precision of  $\sim\pm 0.1^\circ\text{C}$  using the cycling technique (Hansteen and Klügel, 2008). Homogenization  
394 temperatures were converted into  $\text{CO}_2$  densities using equations 3.14 and 3.15 of Span and Wagner  
395 (1996), implemented in DiadFit (Wieser and DeVitre, 2024).

396 We also analyzed all fluid inclusions from sample 04b by calibrated Raman spectroscopy, following  
397 the methods of DeVitre and Wieser (2024). In summary, we utilize a WiTec Alpha 300R Raman  
398 spectrometer at the Department of Earth and Planetary Sciences at the UC Berkeley, with a green  
399 532.046 nm laser and an 1800 grating. Accounting for the small and variable size of fluid inclusions,  
400 we use 50-100X objectives for analysis. Laser powers of 6 mW were used throughout to minimize  
401 the effects of laser heating (DeVitre et al., 2023a; Hagiwara et al., 2019). Instrument drift was  
402 corrected using the  $\sim 1117$  and  $1447\text{ cm}^{-1}$  Ne emission lines (see DeVitre and Wieser, 2024; Lamadrid  
403 et al., 2017). All peak fitting was performed in the Python3 tool DiadFit (Wieser and DeVitre, 2024).  
404 The  $\text{CO}_2$  densities determined through Raman Spectroscopy and Microthermometry are typically  
405 equivalent within analytical uncertainties, consistent with prior work (DeVitre and Wieser, 2024). To  
406 check for the presence of other gaseous species that may have contributed to the derivation in  
407 observed melting temperatures from the theoretical melting point of  $\text{CO}_2$  (e.g.,  $\text{CO}$ ,  $\text{CH}_4$ ,  $\text{N}_2$ ), we  
408 analyzed fluid inclusion covering the complete range of observed melting temperatures at higher  
409 laser power (49 mW) and a 300 grating, resulting in higher signal strength and a wider range of  
410 wavenumbers recorded in a single acquisition.

411 To calculate pressure from the density of  $\text{CO}_2$  in a fluid inclusion using the  $\text{CO}_2$  equation of state, an  
412 estimate of the entrapment temperature is required. Fluid inclusion pressures will likely relax via  
413 diffusion creep during prolonged storage and cooling of the system (Hansteen and Klügel, 2008;  
414 Wanamaker and Evans, 1989) to equilibrate internal and external pressures. Consequently, under  
415 the assumption of full re-equilibration, we use the temperatures derived from clinopyroxene-  
416 orthopyroxene thermobarometry and Mg-in-plagioclase thermometry to invert the measured  
417 homogenization temperatures and calculated  $\text{CO}_2$  densities into pressure. These calculations were

418 performed using the CO<sub>2</sub> EOS of Span and Wagner (1996), implemented in CoolProp (Bell et al.,  
419 2014), and accessed through DiadFit (Wieser and DeVitre, 2024). Further calculations were  
420 performed to investigate the potential influence of H<sub>2</sub>O, He, and Ar on the calculated pressures (see  
421 Supplementary Material).

## 422 4 RESULTS & DISCUSSION

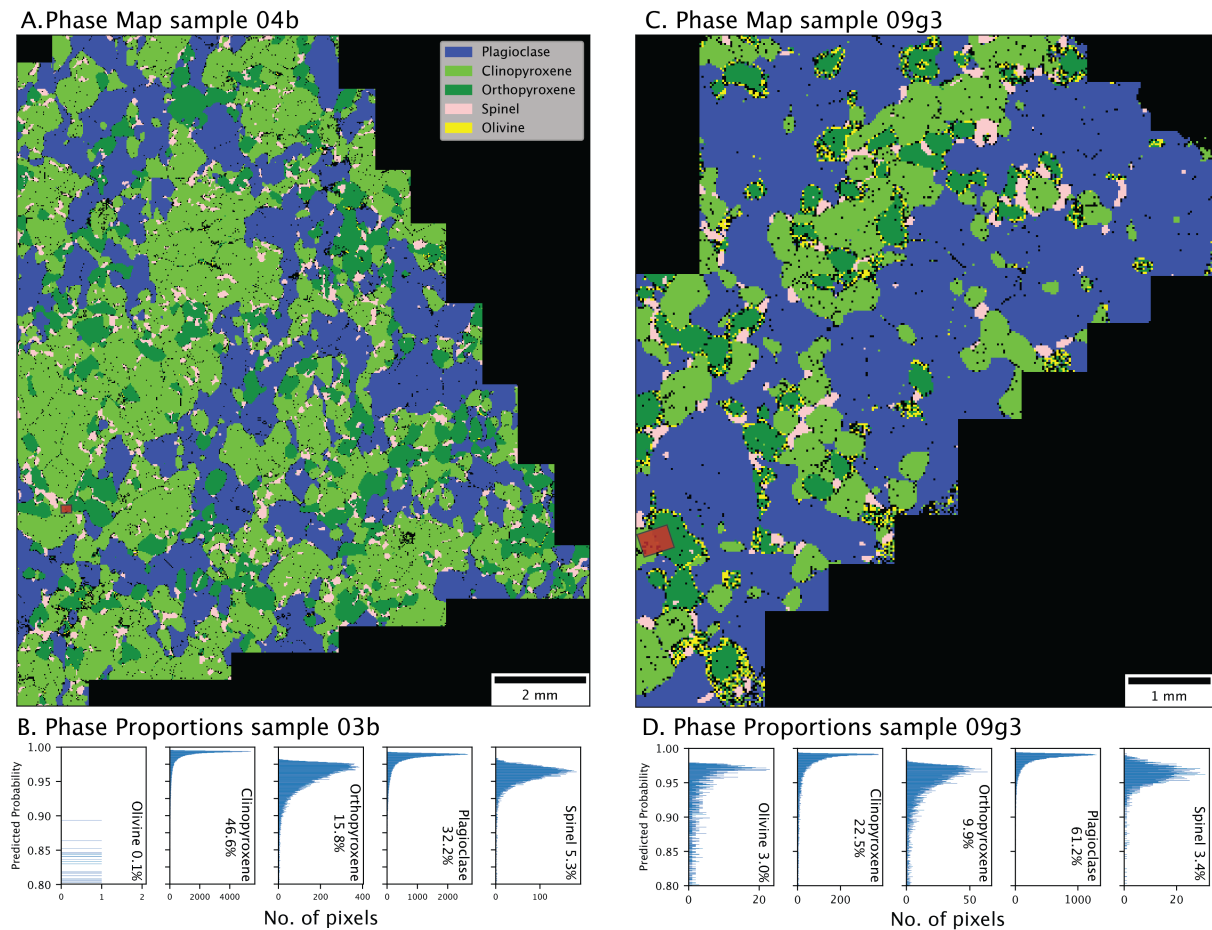
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### 423 4.1 MINERALOGICAL & CHEMICAL SIGNATURES IN THE CONTEXT OF GALÁPAGOS VOLCANISM

424 All gabbro samples considered here display a high degree of textural equilibrium, with 120° grain  
425 boundary angles at plagioclase triple junctions and a narrow range in grain sizes (mean size ~500  
426 μm; Gleeson et al., 2020a). While these textures restrict petrographic characterization of the  
427 xenoliths based on their original igneous texture (i.e., forming through mechanical settling vs. in situ  
428 crystallization) estimated bulk-rock compositions derived from the EDS maps are consistent with  
429 that expected from crystal fractionation (i.e., the residual solid assemblage), with density sorted  
430 settling of plagioclase and mafic mineral phases in a liquid-rich reservoir resulting in the observed  
431 compositional variations (see Supplementary Information).

432 EDS maps and associated phase predictions from mineralML also reveal a consistent mineralogical  
433 signature across our Isla Floreana gabbros (2 pyroxene + plagioclase + spinel) with no residual melt.  
434 All samples are dominated by plagioclase (32 – 69%) and clinopyroxene (22 – 49%), with lower  
435 proportions of orthopyroxene (1 – 16%) and an Al-rich spinel (typically < 5%; Fig. 2). No olivine is  
436 identified in samples 03b and 04b, with only a small number of olivine pixels (associated with  
437 orthopyroxene rims) in samples 09g1 – 3 (Fig. 2, 3). High-resolution chemical maps of the  
438 orthopyroxene grain boundaries confirm the presence of these inferred olivine pixels, revealing  
439 small crystals of olivine forming in orthopyroxene breakdown rims (Fig. 3). Consequently, the  
440 presence of minor olivine in samples 09g1-3 is unlikely to represent the equilibrium mineral

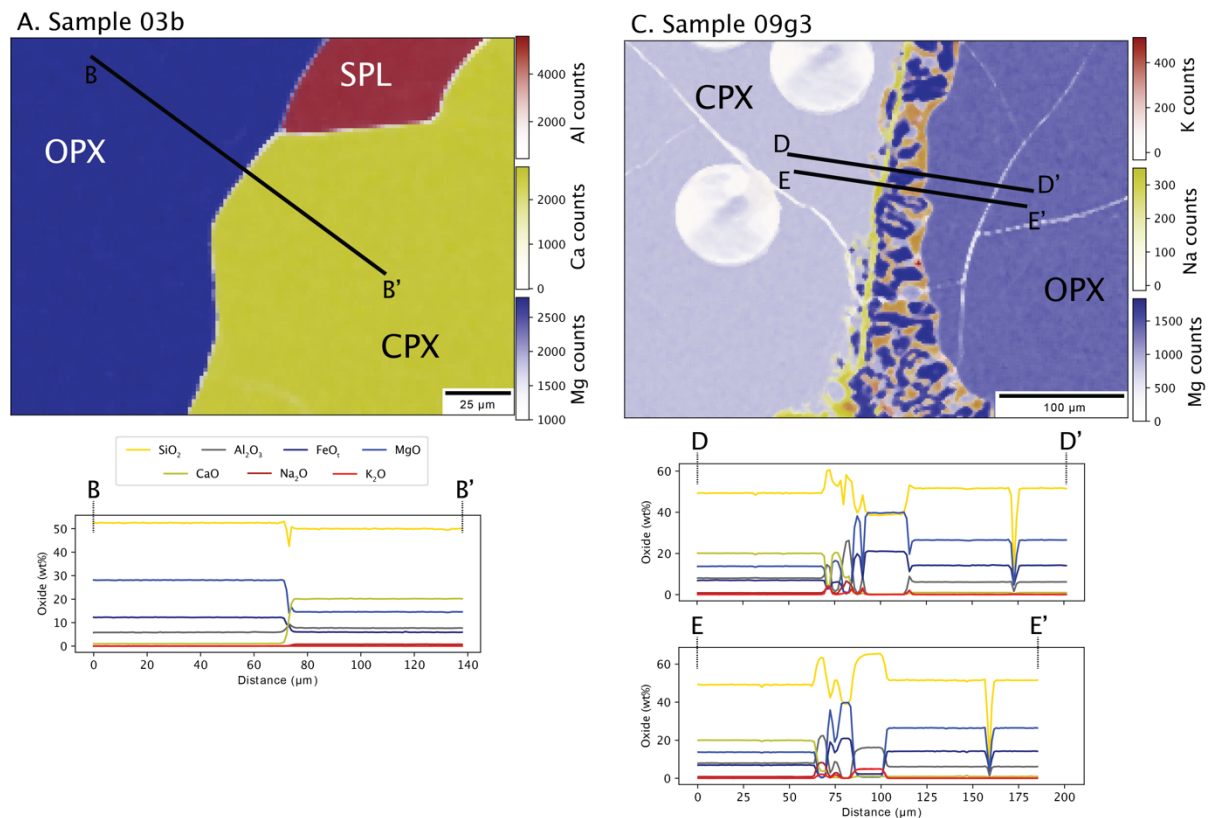
441 assemblage, indicating that all 5 samples considered here display essentially the same mineralogy.  
442 This is consistent with the mineralogy reported for gabbroic xenoliths collected from Isla Floreana by  
443 Lyons et al. (2007a), with substantial olivine only reported in samples with high MgO contents (> 20  
444 wt%).



**Figure 2** – Phase maps and associated modal abundances in characteristic gabbro samples. **A.** Phase map of sample 03b. This sample shows a clear 4 phase mineralogy, with no olivine present. **B.** Phase proportions, determined by mineralML (Shi et al. 2023), with the associated predicted probability for each analysis shown in the histograms. These confirm the absence of olivine in sample 03b. **C.** Phase map of sample 09g3. Pixels identified as olivine by MineralML can be seen in the rim regions of orthopyroxene grains. **D.** Phase proportions confirm the presence of olivine in this sample. Red boxes in panels **A.** and **C.** show the areas displayed in Fig. 3.

445  
446 The abundant plagioclase in our samples is, however, at odds with many petrographic and  
447 geochemical characteristics of Isla Floreana lavas. Plagioclase is absent or only present as a minor  
448 phase in most lavas exposed at the surface of Isla Floreana (< 1.5 Ma), and may represent  
449 disaggregated xenolith material where it does occur (Gleeson et al., 2020a; Harpp et al., 2014).

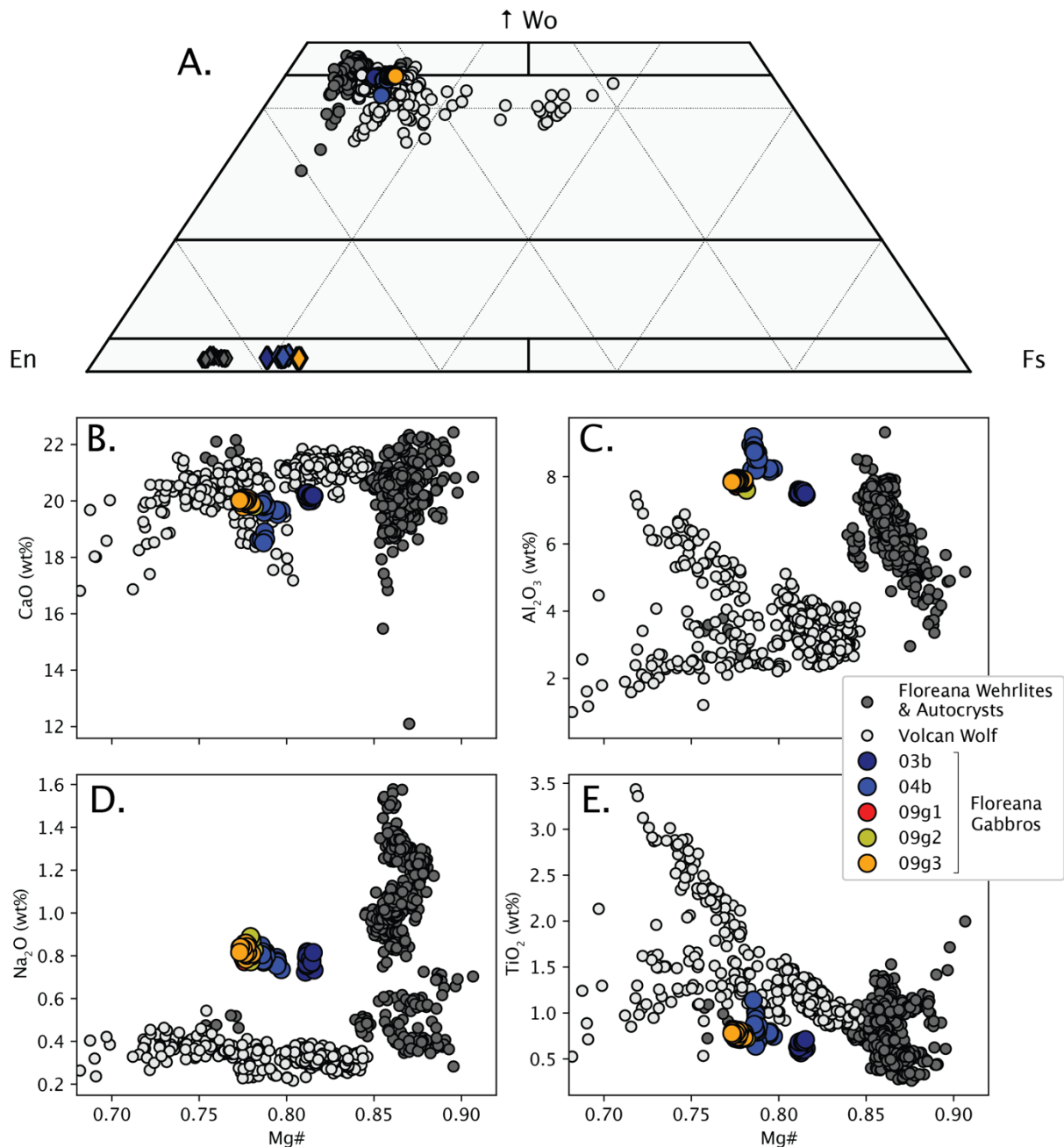
450 Similarly, most lavas on Isla Floreana are olivine-phyric, a phase conspicuously absent from our  
451 xenolith samples. Furthermore, erupted lava whole-rock major element trends – such as negatively  
452 correlated  $\text{Al}_2\text{O}_3$  vs MgO contents – indicate that erupted lava compositions are controlled by  
453 crystallization or addition of olivine and minor clinopyroxene, with plagioclase playing little to no  
454 role in the evolution of the Isla Floreana magmas (Harpp et al., 2014).



**Figure 3** – High-resolution EDS maps of clinopyroxene-orthopyroxene (cpx-opx) grain boundaries. **A.** EDS map of sample 03b demonstrates a sharp cpx-opx boundary without any breakdown features. **B.** An EDS linescan across this boundary reveals an absence of chemical zoning in the region near the cpx-opx boundary. **C.** EDS maps of sample 09g3 demonstrates the presence of an opx-cpx boundary characterised by the presence of an Mg-rich phase and an Si, K, and Na-rich phase. **D.&E.** Linescans across this boundary reveal the presence of olivine (Mg-rich) and silicate melt (Si-rich) in the rim zone separating the clinopyroxene and orthopyroxene crystals.

455 The differences between the Isla Floreana gabbros and the recent Isla Floreana magmatic system  
456 (detailed by Gleeson et al. (2022, 2020a)) extend beyond the mineralogy of these xenoliths. The  
457 composition of the clinopyroxene crystals found in the Isla Floreana gabbros contrasts with the  
458 composition of crystals collected from scoria cones and lava flows that are not associated with any  
459 xenolith group (termed ‘autocrysts’ by Gleeson et al., 2020a). The gabbroic clinopyroxenes are

460 characterized by relatively low Mg# (where Mg# = Mg/(Mg+Fe) molar), with crystals in samples  
 461 09g1-3 averaging Mg# ~0.77, slightly lower than those measured in samples 04b (~0.79) and 03b  
 462 (~0.81; Fig. 4). By contrast, autocrystic clinopyroxenes measured in the Isla Floreana lavas typically  
 463 have Mg# between 0.84 and 0.90 (Fig. 4).

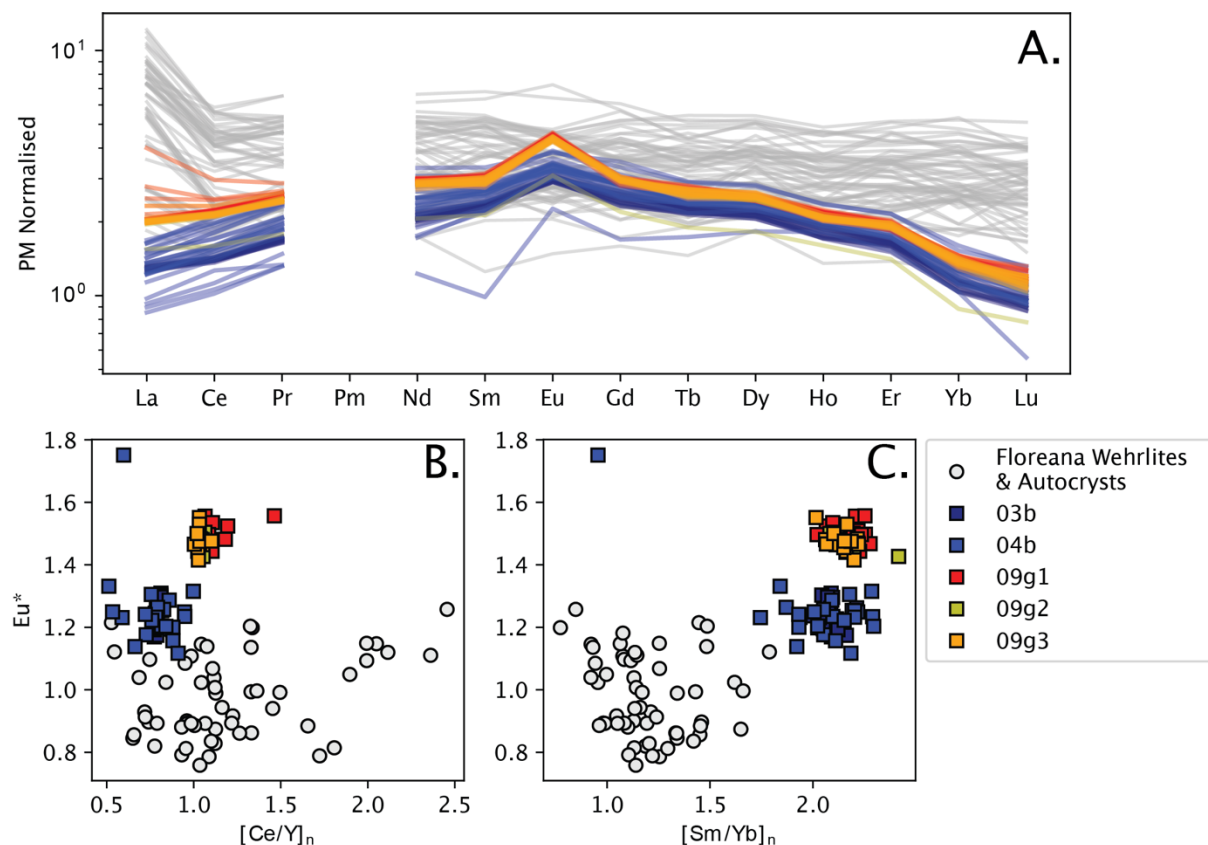


**Figure 4** – Clinopyroxene (and orthopyroxene) compositions from the Isla Floreana Gabbros compared against clinopyroxene data from Volcán Wolf (Stock et al. 2018), chosen as a ‘characteristic’ western Galápagos volcanic system, and pyroxene analyses from wehrlite xenoliths and clinopyroxene autocrysts from Isla Floreana based on the classification of Gleeson et al. (2020a). **A.** Pyroxene quadrilateral displaying the narrow range of compositions measured in the Isla Floreana gabbros relative to other datasets from the Galápagos Archipelago (made in Thermobar; Wieser et al. 2022). **B. – E.** Mg# vs CaO, Al<sub>2</sub>O<sub>3</sub>, Na<sub>2</sub>O, and TiO<sub>2</sub> respectively.

464 In addition, the trace element signature of the gabbroic clinopyroxene crystals are distinct from  
465 those measured in other clinopyroxene groups from Isla Floreana (autocrysts, wehrlite and dunite  
466 xenoliths; Gleeson et al., 2020a). The primitive mantle normalized REE patterns of the gabbro  
467 clinopyroxenes display a positive light-to-middle REE slope, and a negative slope from the middle to  
468 heavy REEs (Fig. 5). This pattern is expressed by the high  $[Sm/Yb]_n$  ratios measured in the gabbroic  
469 clinopyroxenes ( $>1.7$ ; where 'n' symbolizes normalization to the primitive mantle values of Sun and  
470 McDonough, 1989), which are significantly higher than those measured in the other clinopyroxene  
471 groups ( $1 - 1.5$ ; Fig. 5b,c). When converted into equilibrium melt compositions, using the  
472 parameterization of Sun and Liang (2012), our new clinopyroxene data exceed the measured  
473  $[Sm/Yb]_n$  ratio of any basalt erupted on the surface of Isla Floreana (Fig. 6). Taken together, these  
474 data indicate that the gabbroic xenoliths found in scoria and lava deposits on Isla Floreana are

475 unlikely to have formed from crystallization of their carrier liquid, or any magma composition that is  
476 related to basaltic compositions currently exposed on the surface of the island.

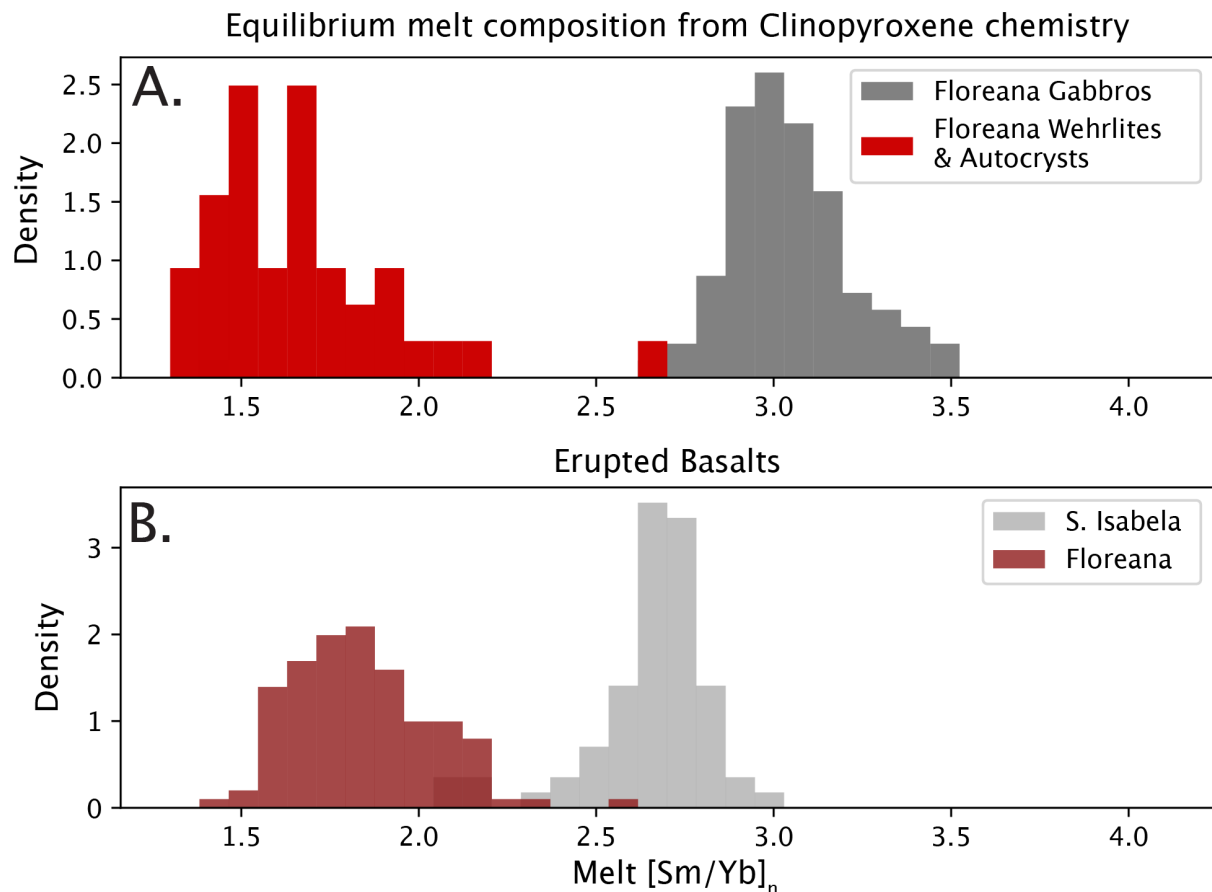
477 Prior work has revealed that the present style of volcanism on Isla Floreana (i.e., infrequent eruption  
478 of olivine-phyric basalts) has lasted for around 1 – 1.5 Myrs (Harpp et al., 2014). As the gabbro  
479 xenoliths show no clear geochemical relationship to this ‘recent’ period of volcanism on Isla  
480 Floreana, it is necessary to consider other possible origins. For example, at other ocean island  
481 volcanoes worldwide (e.g., Hualalai, Hawai’i; Gao et al., 2022), trace element and isotopic signatures  
482 have been used to identify the presence of gabbro xenoliths originating from the underlying oceanic  
483 crust. However, xenoliths of gabbroic cumulates associated with the Nazca oceanic crust (that are  
484 unrelated to Galapagos plume volcanism) would be expected to have very different chemical and



**Figure 5** – Trace element composition of Isla Floreana clinopyroxenes. **A.** REE diagram of clinopyroxenes from the Isla Floreana gabbros (colors) and published data from wehrlite xenoliths and clinopyroxene autocrysts (Gleeson et al. 2020a), normalized using the primitive mantle (PM) estimate of Sun and McDonough (1989). **B.** Eu anomaly ( $Eu^*$ ) vs  $[Ce/Y]_n$ , which is used to demonstrate the degree of incompatible trace element enrichment in each sample. **C.** The  $[Sm/Yb]_n$  content of the gabbroic clinopyroxenes is distinct from crystals measured in the wehrlite xenoliths.



485 mineralogical characteristics to those observed in the Isla Floreana gabbros. Most importantly, the  
486 clinopyroxene trace element (e.g., high  $[La/Yb]_n$  and  $[Sm/Yb]_n$  ratios) and radiogenic isotope  
487 systematics (from Lyons et al. 2007) confirm that the Isla Floreana gabbros must be formed by  
488 magmas generated at high pressure in the Galapagos mantle plume and cannot be associated with  
489 periods of ridge volcanism.



**Figure 6** – Equilibrium melt  $[Sm/Yb]_n$  compositions calculated from the Isla Floreana clinopyroxenes (A.) and published whole-rock data (B.). Clinopyroxene trace element data collected in this study (gabbros) and in Gleeson et al. (2020; wehrlites and antecrysts) are converted to equilibrium melt trace element compositions using the REE partitioning model of Sun and Liang (2012) at a temperature of 1150°C and a pressure of 0.5 GPa.

490 Consequently, we consider possible formation of these gabbros during an earlier period of Isla  
491 Floreana's magmatic history, prior to the period of current volcanism starting at ~1.5 Ma. At ~2 – 2.5  
492 Ma Isla Floreana was located close to the present-day position of Volcán Cerro Azul, near the centre  
493 of plume upwelling owing to the west-to-east motion of the Nazca tectonic plate (Argus et al., 2011;  
494 Harpp et al., 2014). Lavas erupted at western Galápagos shield volcanoes, including Volcán Cerro

495 Azul, are typically plagioclase-phyric (Geist et al., 2002; Naumann and Geist, 2000), possess  
496 overlapping clinopyroxene Mg# and plagioclase An compositions with the Isla Floreana gabbros  
497 (Stock et al., 2020, 2018). They also possess high melt  $[Sm/Yb]_n$  contents (Gibson and Geist, 2010) –  
498 indicating high pressure melting in the plume centre – that overlap with the equilibrium melt values  
499 predicted from the Isla Floreana gabbros (Fig. 6). In addition, radiogenic isotope data from a  
500 separate suite of Isla Floreana gabbros reveal Sr, Nd, and Hf isotope compositions that are distinct  
501 from the erupted Isla Floreana basalts, but overlap with the compositions measured from basalts in  
502 the western Galapagos (Volcán Cerro Azul and Volcán Sierra Negra). As a result, we propose that the  
503 Isla Floreana gabbros likely originate from a ‘plume proximal’ stage of Isla Floreana’s magmatic  
504 history when the island was located close the centre of plume upwelling – near the present-day  
505 location of Volcán Cerro Azul – at around 2-2.5 Ma.

506 The proposed age of these xenoliths is also supported by the EDS maps and linescans, which can  
507 identify the degree of chemical heterogeneity and zoning within individual crystal phases. Chemical  
508 zoning – be it concentric, patchy, or oscillatory – is a common feature of plagioclase and  
509 clinopyroxene crystals in volcanic and plutonic environments worldwide (Bennett et al., 2019;  
510 Lissenberg and MacLeod, 2016; Sanfilippo et al., 2020). Our data, however, indicates almost  
511 complete homogeneity across each crystal phase with only smooth core-rim zoning of An contents  
512 identified in some plagioclase crystals (see Supplementary Material). This indicates that there has  
513 been sufficient time for chemical diffusion to smooth (or remove) any chemical zoning, even for the  
514 slowest diffusing elements. For example, EDS linescans reveal an absence of zoning in minor  
515 components of clinopyroxene (e.g., Ti) over length scales of  $\sim 100 \mu\text{m}$ , an observation in consistent  
516 with late-stage crystallization of magma within a cumulate mush (Lissenberg et al. 2016 and others).  
517 Assuming a temperature of  $1000^\circ\text{C}$  and diffusivities from Cherniak and Liang, (2012) simple  
518 calculations of diffusive length scales (i.e.,  $x = \sqrt{4Dt}$ ) indicate that homogenization over these  
519 distances likely require timescales more than 350,000 kyrs. Furthermore, Ca-Na interdiffusion in  
520 plagioclase is often considered to be negligible in volcanic systems due to the extremely low

521 diffusivities ( $D_{Ca-Na}^{plag} \sim 7 * 10^{-25} m^2/s$  at 1000°C and  $\sim 2 * 10^{-21} m^2/s$  at 1250°C; Grove et al.,  
522 1984). Nevertheless, given the  $\sim 2$  Myr timescale proposed here, these diffusivities equate to Ca-Na  
523 diffusion length scales in plagioclase between 15 and 750  $\mu m$ . Even at the lower end of this range  
524 these distances are consistent with diffusive ‘smoothing’ of any original An zoning in plagioclase  
525 crystals during cooling of these xenoliths, consistent to the gentle core-rim profiles observed in a  
526 subset of crystals. Combined with the high degree of textural equilibrium observed in the Isla  
527 Floreana gabbros (Gleeson et al., 2020a) this chemical homogeneity supports the proposed long-  
528 term storage of these samples and their origin within an ancient ( $\sim 2 - 2.5$  Myr) magmatic system  
529 beneath Isla Floreana.

#### 530 4.2 THERMODYNAMIC CONSTRAINTS ON CUMULATE STORAGE AND SUBSOLIDUS HISTORY

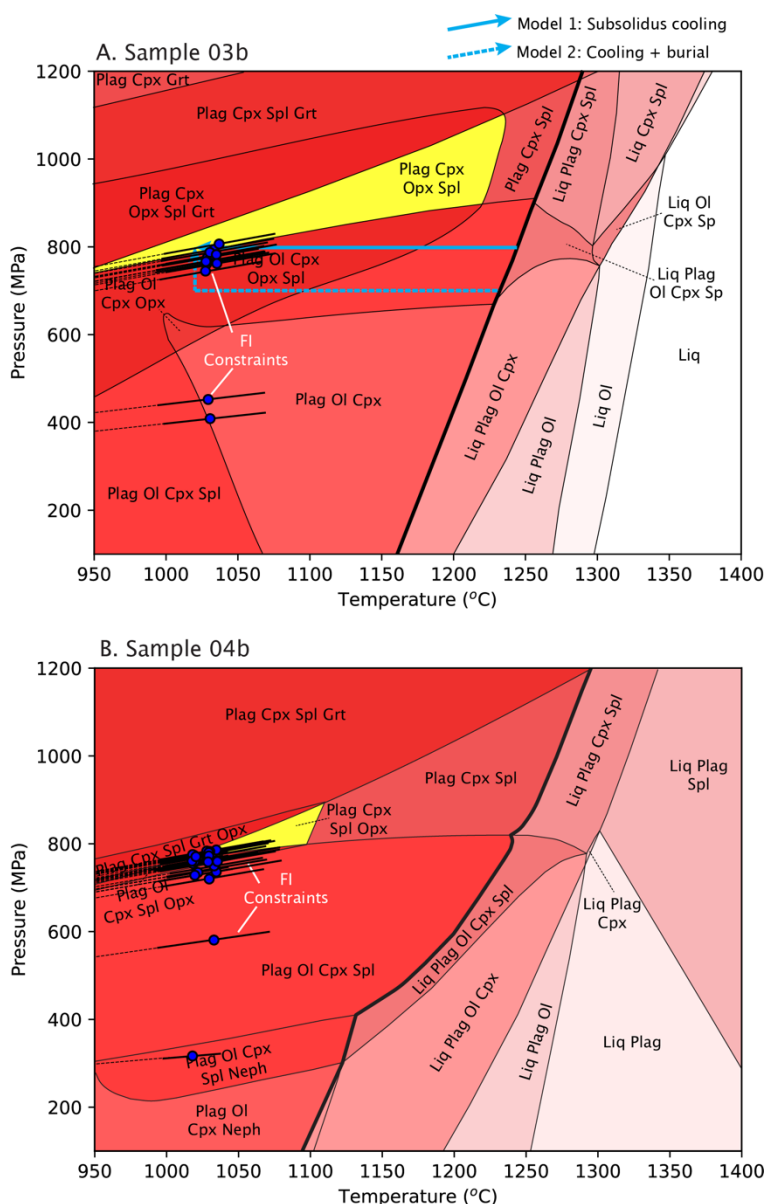
531 The mineralogy, trace element composition, and isotopic signature of the Isla Floreana gabbros links  
532 these samples to the plume-proximal stage of the island’s magmatic history at 2-2.5 Ma. However,  
533 there are several features within the Isla Floreana gabbros that contrast with the petrographic and  
534 chemical characteristics of the modern-day plume-proximal volcanic systems (e.g., Cerro Azul;  
535 Gleeson et al., 2020a; Harpp et al., 2014; Lyons et al., 2007; Naumann and Geist, 2000) and may  
536 provide insights into the storage conditions, and subsolidus history of these cumulate xenoliths.

537 The first difference lies in the absence of a stable olivine phase in the gabbros and the presence of  
538 orthopyroxene (Fig. 2). All shield volcanoes in the current western Galápagos erupt lavas saturated  
539 in some combination of olivine, clinopyroxene, and plagioclase (plus minor oxide phases), with  
540 orthopyroxene rare or absent (Geist et al., 2002; Naumann and Geist, 1999; Reynolds and Geist,  
541 1995). In contrast, all gabbros considered here contain abundant orthopyroxene macrocrysts (>5%),  
542 but no stable olivine. Furthermore, the clinopyroxene crystals in Isla Floreana gabbros contain  
543 anomalously high  $Na_2O$  contents (0.72 – 0.89 wt%) relative to data from volcanic systems in the  
544 western Galápagos (Fig. 4; Stock et al., 2020, 2018). Offsets are also seen in the clinopyroxene trace  
545 element contents. For example, our gabbroic clinopyroxenes contain clear Eu anomalies (defined as

546  $Eu^* = \frac{Eu_n}{\sqrt{Sm_n + Gd_n}}$ , ranging from ~1.15 – 1.35 in samples 03b and 04b, to 1.4 – 1.6 in samples 09g1,  
547 09g2, and 09g3 (Fig. 5). Similar  $Eu^*$  values are not seen in any other clinopyroxene data from Isla  
548 Floreana (Fig. 5; Gleeson et al., 2020a) and while clinopyroxene trace element data is not available  
549 for any western Galápagos shield volcano, the  $Eu^*$  values of erupted lavas in the western Galápagos  
550 does not exceed 1.1 (using compiled data from Gleeson et al., 2021 and Harpp and Weis, 2020).

551 As the Isla Floreana gabbros contain little chemical heterogeneity (i.e., an absence of chemical  
552 zoning) and high extents of textural equilibrium have previously been reported (Gleeson et al.,  
553 2020a), we hypothesize that the unusual chemical and mineralogical signatures of the Isla Floreana  
554 gabbros might result from chemical reactions that take place as the systems transition into  
555 equilibrium subsolidus assemblages. To investigate this, we use estimated whole-rock compositions  
556 from the EDS maps as inputs to thermodynamic models calibrated for sub-solidus igneous systems  
557 to examine the chemical and mineralogical changes that would occur during subsolidus cooling of  
558 cumulates. We use the Python3 package PetThermoTools (Gleeson & Wieser, 2024) to create phase  
559 diagrams for samples 03b and 04b using the pMELTS (Ghiorso et al., 2002), and Holland et al. (2018)  
560 thermodynamic models, implemented through alphaMELTS and MAGEMin (Riel et al., 2022; Smith  
561 and Asimow, 2005).

562 Results using the pMELTS thermodynamic model are shown in Fig. 7. The pMELTS simulations  
563 indicate that olivine is stable in these samples at pressure below ~750 MPa at 950°C, increasing to  
564 800 – 850 MPa at the solidus (1200 – 1250°C). Garnet – also absent from our samples – is predicted  
565 to be stable at pressures >750 MPa at 950°C, increasing to >1100 MPa at the solidus. Consequently,  
566 it is only between these phase boundaries (olivine-out and garnet-in reactions), at temperatures  
567 above 950°C, where our observed mineral assemblage is predicted to be stable (Fig. 7). This  
568 indicates a final pressure of equilibration for these gabbros >750 MPa, far greater than the pressures  
569 expected at the base of the crust beneath Isla Floreana (~380-450 MPa; Feighner and Richards,  
570 1994, using their assumed density of 2800 kg/m<sup>3</sup> for the crust). Calculations performed using the



**Figure 7** – Phase diagrams of samples 03b (A.) and 04b (B.) constructed in the NCFMASTOCr (Na, Ca, Fe, Mg, Al, Si, Ti, O, and Cr) system. All calculations are performed in PetThermoTools using the pMELTS thermodynamic model (Ghiorso et al. 2002, Gleeson & Wieser, 2024). P-T region where the observed mineralogy is recreated is highlighted in yellow. Also shown are the results of Fluid Inclusion (FI) barometry (solid lines indicate pressures calculated using clinopyroxene-orthopyroxene determined temperatures, dashed lines indicate pressures calculated using Mg-in-plagioclase determined temperatures), and the different subsolidus cooling and burial models used to investigate the phase compositions. For all calculations the  $Fe^{3+}/Fe_{tot}$  ratio was set to 0.10. Thicker line indicates the position of the solidus. FI pressures are determined by randomly sampling 1000 temperatures from the clinopyroxene-orthopyroxene thermometry results with the median (circles) and mean  $\pm 2$ \*standard deviation (solid lines) pressure and temperature results shown. Similar calculations are performed using the Mg-in-plagioclase determined temperatures, with the results displayed using the dashed black lines.

571 Holland et al. (2018) thermodynamic models, however, predict greater pressures for the garnet-in  
 572 and olivine-out reaction boundaries. For example, the olivine-out phase reaction occurs between

573 ~800 and 900 MPa for the Holland et al. (2018) thermodynamic model (see Supplementary  
574 Information).

575 In addition to estimating the final storage pressure of these cumulates, the pMELTS phase diagrams  
576 also provide insights into the phase reactions that occur during subsolidus cooling. For example, our  
577 phase calculations indicate that orthopyroxene is not stable at or above the solidus. In fact,  
578 assuming these gabbro nodules originated as a cumulate mush (melt + crystals), the initial cumulus  
579 phases were likely clinopyroxene + plagioclase + olivine +/- spinel, consistent with the cumulus  
580 assemblage predicted for active volcanic systems in the western Galápagos where orthopyroxene is  
581 rarely observed (Geist et al., 2002; Naumann and Geist, 1999; Stock et al., 2020). The presence of  
582 orthopyroxene therefore supports our hypothesis that significant chemical and mineralogical  
583 changes occurred during subsolidus cooling, which we examine in more detail using two possible  
584 subsolidus cooling models.

585 In the first model we assume that the system starts at the solidus at a pressure of 800 MPa and  
586 undergoes isobaric cooling to a temperature of 1020°C (Fig. 8) based on clinopyroxene-  
587 orthopyroxene thermobarometry. In our second model, we assume the solidified gabbroic mush  
588 zone undergoes an initial period of isobaric cooling at lower pressure, followed by a period of burial  
589 (increasing pressure) due to emplacement of lava/intrusions at shallower levels (Fig. 8). This second  
590 model is similar to that proposed by Lyons et al. (2007) and assumes emplacement of ~5 km of lava  
591 or shallow level intrusions following initial formation of the gabbro cumulates (approximately half of  
592 the total thickness of the Galápagos Platform). Ultimately, this leads to an increase in the system  
593 pressure from ~670 MPa to 800 MPa.

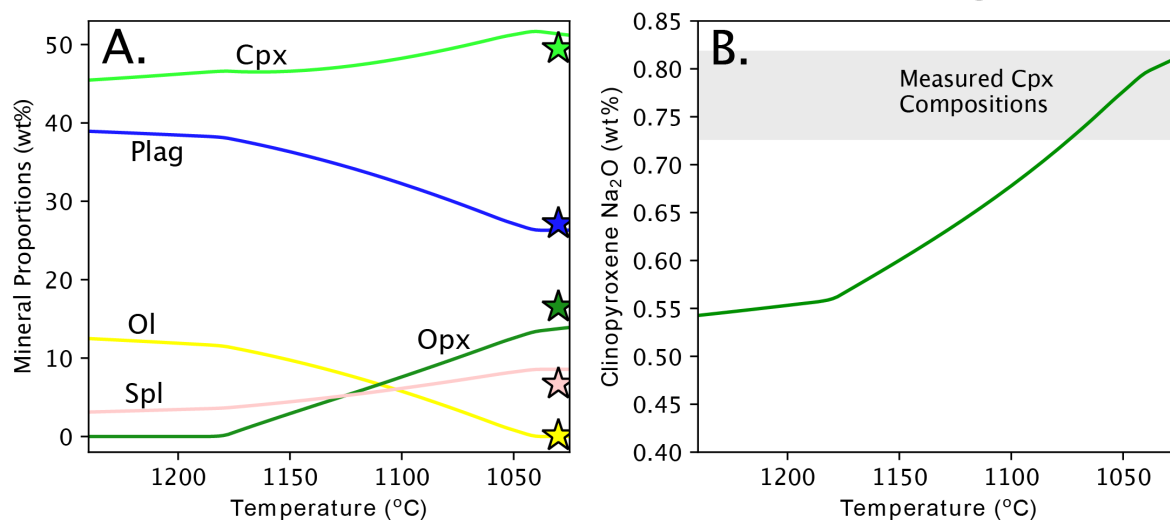
594 For both samples considered here the two models demonstrate a sequence of phase changes during  
595 cooling (and burial) that can be simplified as:

596 
$$\textit{plagioclase} + \textit{olivine} \rightarrow \textit{spinel} + \textit{orthopyroxene} (+\textit{clinopyroxene})$$

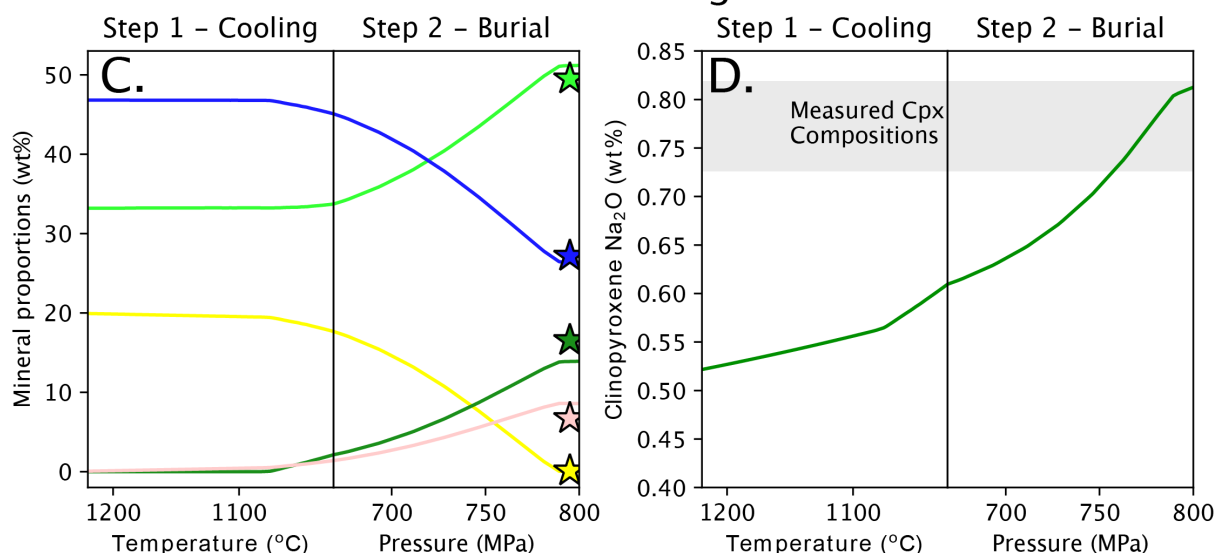
597 The decrease in the modal proportion of olivine and plagioclase in these samples during cooling (and  
598 pressurization, Fig. 8) requires a re-distribution of Na, Sr, and Eu – elements that are dominantly  
599 found in plagioclase – into other mineral phases. Our models reveal an increase in the Na<sub>2</sub>O content  
600 of clinopyroxene during cooling (by a factor of ~1.6-1.8X; Fig. 8). While MELTS cannot model Sr-Eu  
601 partitioning directly, we propose that these sub-solidus reactions may also be responsible for the  
602 strong positive Eu anomalies found in the gabbroic clinopyroxenes. Interestingly, in the subsolidus  
603 cooling models there is a decrease in the clinopyroxene TiO<sub>2</sub> content, consistent with the slightly  
604 lower concentration of TiO<sub>2</sub> in the gabbro clinopyroxenes compared to data from Volcán Wolf (Fig.  
605 4).

606 The thermodynamic models also provide a potential explanation for the presence of olivine crystals  
607 in orthopyroxene breakdown rims of samples 09g1-3, identified through high-resolution EDS  
608 mapping paired with automated mineral classification by mineral ML (Fig. 3). The stability of  
609 orthopyroxene is highly sensitive to pressure (Fig. 7), with replacement of orthopyroxene by olivine  
610 predicted to occur during depressurization. We suggest that the breakdown rims formed during  
611 ascent and decompression of these nodules either shortly prior to, or during eruption. What remains  
612 uncertain is why breakdown rims are observed in the 09g samples, but not in 03b and 04b. One  
613 important difference between the sample groups is that 03b and 04b were collected from  
614 explosively emplaced scoria cones, whereas 09g were found within the centre of a blocky lava flow.  
615 This could suggest that variations in ascent rate and/or surface cooling timescales could influence  
616 the formation of these orthopyroxene break-down rims.

### Model 1: Isobaric subsolidus cooling



### Model 2: Cooling & Burial



**Figure 8**—Phase proportions and mineral chemistry predicted during cooling and burial from the solidus. **A.** Mineral proportions predicted during isobaric cooling at 800 MPa. Color scheme as in Fig. 2. Stars indicate the measured mineral proportions derived from the EDS maps. **B.** Change in clinopyroxene Na<sub>2</sub>O contents during isobaric cooling. The grey region represented the range of measured clinopyroxene Na<sub>2</sub>O contents in sample 03b. **C. & D.** Mineral proportions and clinopyroxene chemistry, respectively, predicted during (i) isobaric cooling at 670 MPa, and (ii) burial to 800 MPa.

#### 617 4.3 FORMATION AND STORAGE DEPTHS

618 One of the key predictions from the thermodynamic modelling described above is that the absence  
 619 of olivine in these gabbros requires pressures greater than that found at the Moho, and thus  
 620 formation and/or storage below the base of the oceanic crust (Fig. 7). The thermodynamic modelling  
 621 does, however, contain significant uncertainties that are difficult to fully quantify or demonstrate



622 (including the offset between different models, the uncertainty in bulk-rock estimates, and the  
623 oxygen fugacity/water content of the subsolidus system). To more rigorously assess the pressure of  
624 gabbro formation and storage beneath Isla Floreana, we consider the available independent  
625 barometric evidence from both mineral chemistry and fluid inclusion CO<sub>2</sub> densities.

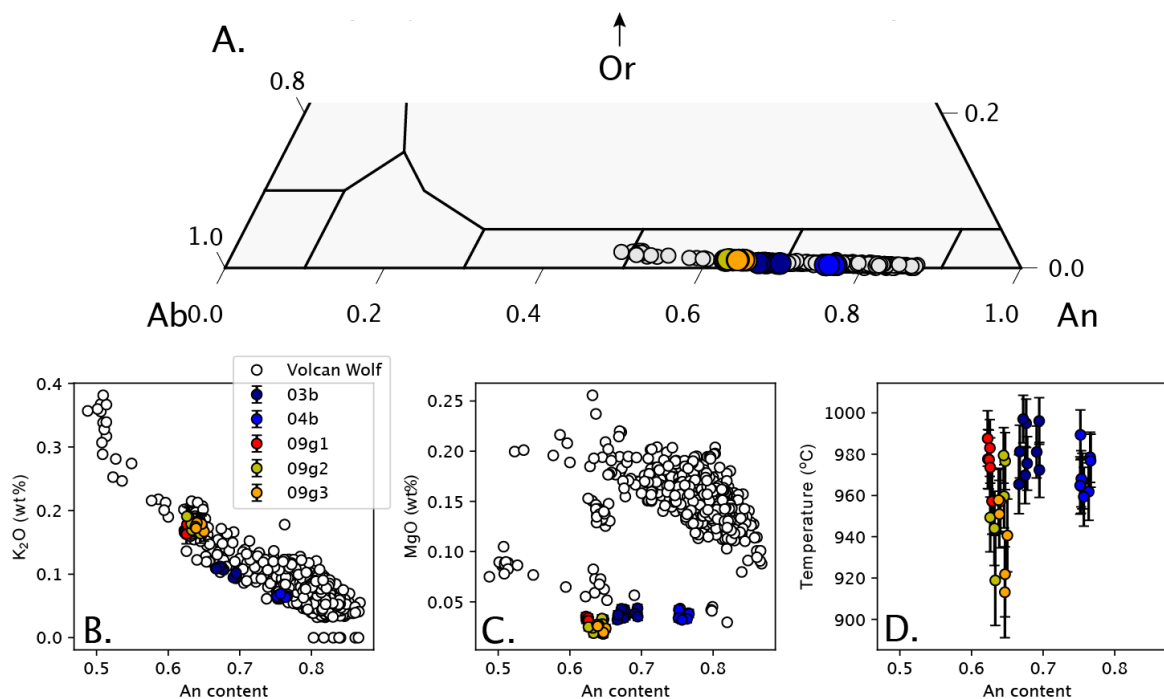
#### 626 4.3.1 Mineral Thermobarometry

627 Due to the presence of subsolidus equilibration in the Isla Floreana gabbros, widely used melt-based  
628 thermobarometers (e.g., Cpx-Liq) and single-phase thermobarometers that were calibrated on  
629 mineral-melt pairs (e.g. Cpx-only) cannot be applied to our samples (Gleeson et al., 2020a).

630 However, thermobarometers based on equilibria between different solid phases might provide first  
631 order constraints on the pressure of gabbro storage and, more importantly, constrain the system  
632 temperature (a necessary input parameter for converting fluid inclusion measurements into  
633 pressure estimates). For example, the temperature dependent exchange of Mg (coupled with Ca)  
634 between clinopyroxene and plagioclase holds a lot of promise for sub-solidus mafic systems and has  
635 previously been used to evaluate the cooling history of oceanic gabbros (Faak et al., 2014; Sun and  
636 Lissenberg, 2018).

637 Plagioclase compositions for the gabbroic xenoliths, determined through coupled EDS (Si, Al, Ca, Na,  
638 Fe) and WDS (Mg, K) analysis, show little variability within each sample (both core and rim analyses  
639 were performed for each crystal). Samples 09g1-3 have low Anorthite contents ( $An = Ca/(Ca+Na+K)$   
640 *molar*), ranging from 0.625 (09g1) to 0.643 (09g3), and moderately high K<sub>2</sub>O contents around 0.163 –  
641 0.191 wt% (Fig. 9). Sample 04b is characterized by substantially higher An values (~0.76), with  
642 sample 03b displaying intermediate compositions (~0.67 – 0.69; Fig. 9). Samples 03b and 04b also  
643 show lower K<sub>2</sub>O contents than the 09g gabbros 0.061 – 0.112 wt%. Notably, the An-K<sub>2</sub>O relationships  
644 in the gabbroic plagioclase closely mirrors that observed in plagioclase data from Volcán Wolf,  
645 chosen as a characteristic volcanic system in the western Galápagos due to the abundance of  
646 mineral data available (Fig. 9b; Stock et al., 2020). The MgO contents measured in the gabbroic

647 plagioclase are, however, substantially lower than the bulk of the data from Volcán Wolf. Samples  
 648 09g1, 09g2, and 09g3 contain average MgO contents of 0.032 wt%, 0.027 wt%, and 0.024 wt%,  
 649 respectively (Fig. 9c). These values are lower than those measured in samples 03b and 04b (0.038  
 650 wt% and 0.037 wt%, respectively).



**Figure 9** – Plagioclase compositions from the Isla Floreana gabbros compared to data from Volcán Wolf (Stock et al. 2020). **A.** Feldspar ternary plot produced using the Python3 package Thermobar (Wieser et al. 2022). **B.** K<sub>2</sub>O contents against plagioclase An content. **C.** Plagioclase MgO contents determined by WDS analysis. **D.** Temperature calculated from the thermometer of Sun and Lissenberg (2018). All error bars show the 1 sigma analytical uncertainty, with this uncertainty propagated through to the temperature calculations.

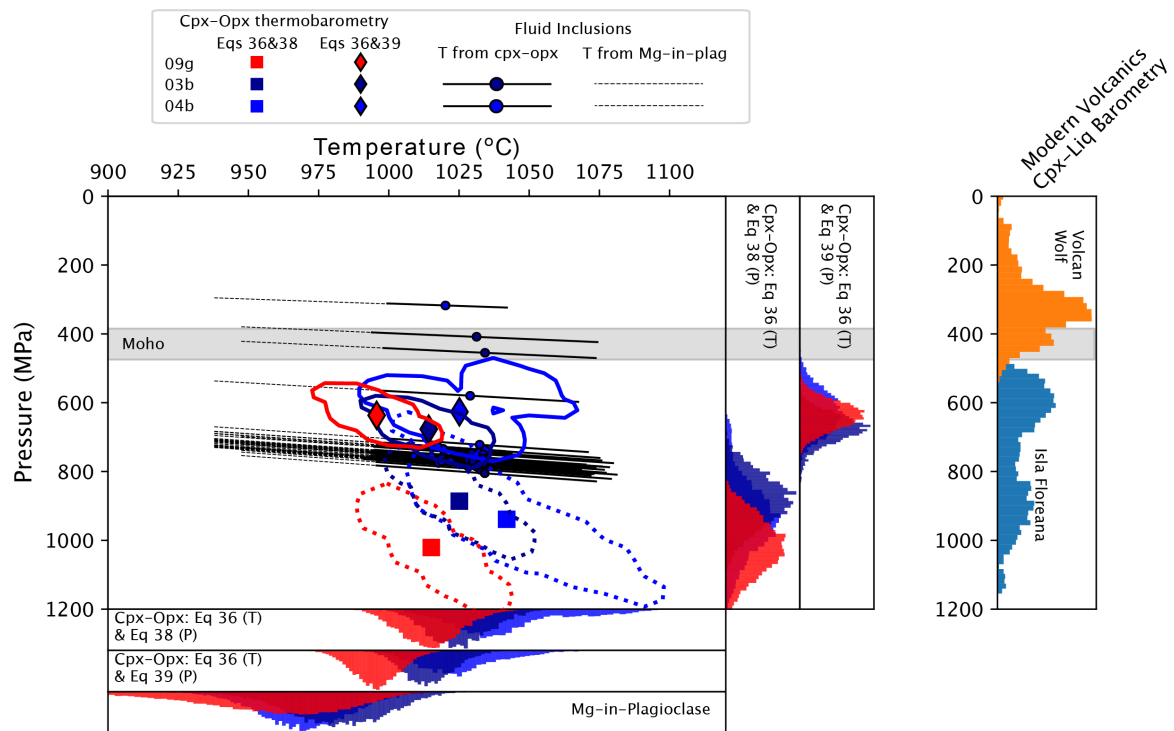
651 These plagioclase MgO contents, combined with EDS analysis of clinopyroxene crystals in the same  
 652 samples (to determine the Mg partition coefficient  $K_{Mg} = \frac{Mg_{Plag}}{Mg_{Cpx}}$ ), can be used to evaluate  
 653 the storage temperature of the Isla Floreana gabbros. To evaluate the uncertainty on our  
 654 temperature calculations, we use a Monte Carlo approach to propagate the analytical error  
 655 associated with the plagioclase MgO and An contents, and mean clinopyroxene MgO contents. For  
 656 each analysis and each parameter, we generate 2000 samples from a normal distribution with a  
 657 standard deviation defined by the either the 1 sigma analytical uncertainty (e.g., plagioclase MgO  
 658 contents) or the 1 sigma variation in measured composition (e.g., the mean clinopyroxene MgO

659 contents). As the parameterization is also pressure sensitive, we incorporate uncertainty in pressure  
660 using a uniform distribution spanning 0.2 and 0.8 GPa, the previously determined range in magma  
661 storage conditions for Galápagos basalts (Bell et al., 2021; Gleeson et al., 2020a; Stock et al., 2018).  
662 The mean temperature estimated for each individual plagioclase analysis varies from 913.2°C  
663 ( $\pm 22.0^\circ\text{C}$ ) to 997.0°C ( $\pm 11.5^\circ\text{C}$ ; Fig. 9d).

664 In addition to the Mg in plagioclase thermometer, the temperature and pressure of the system can  
665 be estimated using the composition of co-existing clinopyroxene and orthopyroxene crystals  
666 (Putirka, 2008). Importantly, EDS linescans reveal no compositional zoning in either clinopyroxene or  
667 orthopyroxene crystals near the grain boundaries, indicating complete chemical equilibration  
668 between these phases at their subsolidus storage conditions. To evaluate the role of analytical  
669 uncertainties on the precision of our 2-pyroxene thermobarometric calculations we generated 200  
670 hypothetical clinopyroxene/orthopyroxene compositions for each EDS analysis using Thermobar (3 –  
671 5 analyses taken per crystal Wieser et al., 2022). These compositions were determined by assuming  
672 a normal distribution of probabilities centered on the measured oxide concentration (i.e., the mean  
673 of the distribution) with a standard deviation set by the estimated error of analysis. For each  
674 clinopyroxene-orthopyroxene pair this process resulted in 600 – 1000 paired compositions that were  
675 used to calculate the temperature and pressure of storage. Calculations were performed by iterating  
676 the results of equations 36 & 38 and equations 36 & 39 from Putirka (2008), implemented in the  
677 Python3 package Thermobar (Wieser et al., 2022).

678 Temperatures determined through clinopyroxene-orthopyroxene thermobarometry, iterating  
679 equation 36 (temperature) with equation 38 (pressure) from Putirka (2008), range from 1042.3°C  
680 ( $\pm 18.9^\circ\text{C}$ ) for sample 04b to 1014.4°C ( $\pm 11.4^\circ\text{C}$ ) in sample 09g1 (Fig. 10). If equation 39 is used to  
681 determine the system pressure, the mean temperature estimated for each sample decreases by  
682 approximately 10-20°C. Overall, our results indicate that clinopyroxene-orthopyroxene equilibrium  
683 in the Isla Floreana gabbros records temperatures at the upper end of those estimated by Mg-in-

684 plagioclase thermometry, but with smaller uncertainties. At present, the cause of the offset between  
 685 the two methods is not clear, but the region of overlap between the two methods indicates that  
 686 storage at >900°C is likely for the Isla Floreana gabbros (Fig. 10).



**Figure 10** – Barometric results from the Isla Floreana gabbros (left hand panels) against the estimated Moho pressure and barometric analysis of modern day volcanics in the western (Volcán Wolf) and eastern (Isla Floreana) Galápagos. Modern day estimates are calculated using published clinopyroxene and liquid compositions from Stock et al. (2018) and Gleeson et al. (2020), for Volcán Wolf (orange) and Isla Floreana (blue), respectively. Calculations are performed in Thermobar, using the Neave and Putirka (2017) barometer solved iteratively with equation 33 from Putirka (2008). Melt matching algorithms were performed with the same limits as used in the original papers. Pressure and temperature estimates derived from the Isla Floreana gabbros, which include cpx-opx thermobarometry, Mg-in-plagioclase thermometry, and FI barometry (assuming pure CO<sub>2</sub> fluids), indicate storage of these gabbros at pressures below the Moho. FI pressures are determined by randomly sampling 1000 temperatures from the clinopyroxene-orthopyroxene thermometry results with the median (circles) and mean  $\pm 2$ \*standard deviation (solid lines) pressure and temperature results shown. Similar calculations are performed using the Mg-in-plagioclase determined temperatures, with the results displayed using the dashed black lines.

687 Application of clinopyroxene-orthopyroxene equilibria to determining the pressure of gabbro  
 688 storage is, however, associated with significant uncertainty due to the large variation in estimated  
 689 pressures associated with different equation pairs. If equation 39 is iterated with equation 36 to  
 690 simultaneously solve for pressure and temperature our data indicates storage around  $627 \pm 59$  MPa  
 691 (sample 04b) to  $678 \pm 42$  MPa (sample 03b). If we instead use equation 38 (which is independent of

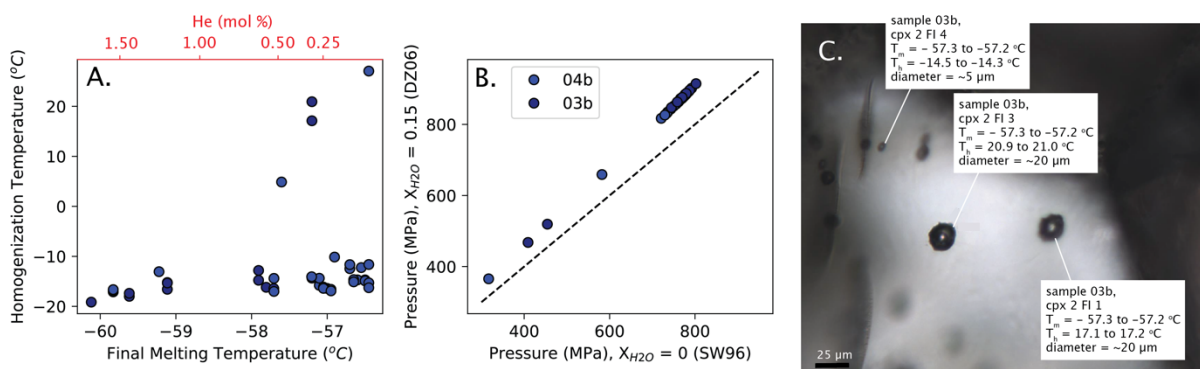
692 temperature) to determine the system pressure, we get estimates of  $940 \pm 102$  MPa and  $886 \pm 72$   
693 MPa for samples 04b and 03b (Fig. 10). The large difference between the two barometric equations  
694 indicates that there is a large degree of uncertainty in these pressure calculations. Considering the  
695 95% confidence 'envelope' on a P-T plot (calculated using the Python3 package *pyrolite*; Williams et  
696 al., 2020) reveals a small region of overlap between the two clinopyroxene-orthopyroxene pressure-  
697 temperature calculations at  $\sim 750$  MPa and  $1000 - 1025^\circ\text{C}$  (Fig. 10). However, given the difference  
698 between the two equations, and the uncertainties estimated for each approach, all we can  
699 realistically conclude from clinopyroxene-orthopyroxene barometry is that these gabbros were likely  
700 stored at pressures higher than  $\sim 450$  MPa, representing the maximum possible pressure at the base  
701 of the oceanic crust beneath the Galápagos Platform (calculated using the crustal thickness estimate  
702 of Feighner and Richards (1994a) and a constant crustal density of  $2800 \text{ kg/m}^3$ ).

#### 703 4.3.2 Fluid Inclusion Barometry

704 Both thermodynamic simulations and clinopyroxene-orthopyroxene barometry indicate that the  
705 gabbro xenoliths were stored at  $>450$  MPa, below the base of the Galápagos crust. Unlike mineral  
706 thermobarometers, which are relatively imprecise (e.g. Root Mean Square Error (RMSE)  $>3$  kbar for  
707 clinopyroxene-orthopyroxene; Wieser et al. 2025), the density of  $\text{CO}_2$  fluids trapped within crystals  
708 can provide a more sensitive barometer in  $\text{H}_2\text{O}$ -poor ocean island systems assuming the  
709 temperature of the system is somewhat constrained (and that inclusions are not modified during  
710 suberuptive ascent and/or cooling at the surface).

711 In sample 04b we identified and measured 30 apparent primary fluid inclusions (FIs) in 8  
712 clinopyroxene crystals, and in sample 03b we found 3 apparent primary FI in 1 clinopyroxene and 1  
713 plagioclases respectively, as well as 7 FIs across 2 secondary trails (1 plagioclase crystal and 1  
714 clinopyroxene crystal; Fig. 11). Of the 43 fluid inclusions, 37 display homogenization temperatures  
715 between  $-9.8^\circ\text{C}$  and  $-19.1^\circ\text{C}$ . Of the remaining FIs, 3 in sample 04b show clear signs of decrepitation  
716 (see Supplementary material) and display  $T_h$  values between  $27 - 27.1^\circ\text{C}$  (1 FI) and the critical point

717 of CO<sub>2</sub> (2 FIs). Other inclusions with higher homogenization temperatures include one FI in sample  
 718 04b ( $T_h = 4.8 - 4.9^\circ\text{C}$ ), and 2 neighboring, apparent primary fluid inclusions in sample 03b ( $T_h = 17.0 -$   
 719  $17.1^\circ\text{C}$  and  $T_h = 20.9 - 21.0^\circ\text{C}$ ). These 3 FIs are located near the crystal margins, and are among the  
 720 largest inclusions in our dataset, making them more susceptible to mechanical re-equilibration  
 721 during ascent and stalling (Wanamaker and Evans, 1989). As we are interested in determining the  
 722 long-term storage conditions of these xenoliths prior to ascent in their carrier liquid, we only  
 723 primarily consider inclusions with no clear signs of decrepitation/re-equilibration (and thus highest  
 724 densities) in the discussion below.



**Figure 11** – Analysis of fluid inclusions (FIs) from samples 03b and 04b. **A.** Comparison of mean homogenization temperature against the final melting temperature. Mixed CO<sub>2</sub>-He EOS calculations are used to convert the melting temperature of these inclusions into estimates of the possible He content of these fluid inclusions. **B.** Comparison of entrapment pressures (at the mean temperature determined by cpx-opx thermobarometry) for different  $X_{\text{H}_2\text{O}}$  values (EOS of Span and Wagner (1996) is used when  $X_{\text{H}_2\text{O}} = 0.0$ ; Duan and Zhang (2006) mixed EOS is used for  $X_{\text{H}_2\text{O}} = 0.15$ ). **C.** Example FIs from sample 03b, including the 2 larger inclusions that return anomalously high homogenization temperatures.

725 Assuming the trapped fluids are pure CO<sub>2</sub>, the homogenization temperatures indicate fluids densities  
 726 between 0.98 – 1.03 g/cm<sup>3</sup>. Using the median temperature from all clinopyroxene-orthopyroxene  
 727 thermobarometry calculations for each respective sample we can convert these pure-CO<sub>2</sub> densities  
 728 into storage pressures ranging from 719 – 802 MPa using the Span and Wagner (1996) EOS in DiadFit  
 729 (Wieser and DeVitre, 2024). A Monte Carlo approach, accounting for the uncertainty in the  
 730 measured homogenization temperatures and the equilibration temperatures determined through  
 731 clinopyroxene-orthopyroxene thermobarometry, suggests uncertainties of +/- 9 MPa for sample 04b  
 732 and +/- 10 MPa for sample 03b (implemented in DiadFit, Wieser and DeVitre, 2024). Consequently,

733 our data – assuming pure CO<sub>2</sub> fluids – strongly supports the thermobarometry and thermodynamic  
734 results discussed above: the Isla Floreana gabbros were stored below the base of the crust, within  
735 the lithospheric mantle, prior to entrainment and ascent in their carrier melt.

736 Converting homogenization temperatures into fluid densities and, ultimately,  
737 entrapment/equilibration pressures does, however, require greater examination of the fluid  
738 composition. Prior analysis of the H<sub>2</sub>O content of Galápagos magmas reveals erupted melt H<sub>2</sub>O  
739 contents around 0.5 – 1.0 wt% across the southwestern region of the archipelago, with reactive  
740 porous flow in crystal mush zones generating anomalously H<sub>2</sub>O-rich melt signatures (~2 wt%;  
741 Gleeson et al., 2022; Peterson et al., 2017). To account for the possible presence of H<sub>2</sub>O in the initial  
742 fluid phase we perform calculations using the mixed CO<sub>2</sub>-H<sub>2</sub>O EOS from Duan and Zhang (2006)  
743 implemented in DiadFit (Wieser and DeVitre, 2024). For these calculations we use a fluid X<sub>H<sub>2</sub>O</sub>  
744 composition of 0.15, representing the maximum X<sub>H<sub>2</sub>O</sub> value for any Galápagos magma at mantle  
745 depths (using estimates of melt H<sub>2</sub>O contents from the nominally anhydrous mineral analyses of  
746 Gleeson et al. 2022). Results demonstrate that incorporation of minor H<sub>2</sub>O into these fluids leads to  
747 increase in the estimated equilibration pressures, with the median pressure for calculations  
748 performed with a X<sub>H<sub>2</sub>O</sub> of 0.15 ranging from 821 to 916 MPa. Overall, our calculations indicate that  
749 the presence of water increases the estimated pressure by ~14%, further supporting the idea that  
750 these gabbros were formed, and were stored, well-below the base of the crust.

751 As H<sub>2</sub>O and CO<sub>2</sub> are immiscible at room temperature, the presence of H<sub>2</sub>O will not influence the  
752 melting point of a CO<sub>2</sub>-rich fluid inclusion. However, our microthermometry results reveal only 15  
753 inclusions – all in sample 04b – with final melting temperatures within 0.2°C of the melting  
754 temperature for pure CO<sub>2</sub>. Eight other FIs from sample 04b and 3 primary fluid inclusions from  
755 sample 03b have final melting temperatures within 1°C of the pure CO<sub>2</sub> melting point, with the  
756 remaining inclusions displaying final melting temperatures as low as -60.1°C. In addition, inclusions  
757 with low melting temperatures melt over a larger temperature range, with melting initiating as low

758 as  $-64.5^{\circ}\text{C}$  in some cases (although the point of melting initiation was difficult to constrain due to the  
759 small size of many of these inclusions;  $\sim 2\ \mu\text{m}$ ). This melting point depression indicates the presence  
760 of an additional component (aside from  $\text{H}_2\text{O}$  and  $\text{CO}_2$ ) within the FIs from sample 03b and 04b.  
761 Raman spectroscopy did not detect any other Raman-active gaseous species (ruling out  $\text{N}_2$ ,  $\text{H}_2\text{S}$ ,  $\text{SO}_2$   
762 etc.). Thus we suggest that the observed freezing point depression results from the presence of a  
763 Raman-inactive noble gas (e.g., He, Ne, Ar). Although not quite as low as our temperatures here, it is  
764 notable that Frezzotti et al. (1992) report melting temperatures of  $-58.2$ - $-56.5^{\circ}\text{C}$  in xenoliths from Salt  
765 Lake Crater, Oahu, which also do not show any Raman-active species other than  $\text{CO}_2$ .

766 Using mixed  $\text{CO}_2$ -He and  $\text{CO}_2$ -Ar EOS in CoolProp (Bell et al., 2014; Kunz and Wagner, 2012), we can  
767 calculate the predicted triple point temperature depression caused by the presence of He or Ar in  
768 the 03b and 04b FIs. These calculations suggest that the melting-point depression of the 03b and  
769 04b FIs could result from the presence of up to 1.7 mol% He or 2.7 mol% Ar (or some combination of  
770 both). He and Ar proportions up to  $\sim 2$  mol% have been identified in FIs from picritic lavas and mantle  
771 xenoliths collected from La Palma, Canary Islands (Sandoval-Velasquez et al., 2023), supporting this  
772 interpretation. We cannot examine the influence of He or Ar on the estimated FI equilibration  
773 pressures because converting mixed  $\text{CO}_2$ -He and  $\text{CO}_2$ -Ar fluid densities into high-temperature  
774 ( $\sim 1020^{\circ}\text{C}$ ) equilibration pressures is restricted by the temperature range of the mixed EOS  
775 calibrations, and the inability of CoolProp to converge on solutions for these mixed fluids above  $\sim 300$   
776  $- 400^{\circ}\text{C}$ . To place a first order assessment on the influence of He and Ar on the estimated pressures  
777 we use CoolProp to track the P-T trajectory of isochores for theoretical FIs with a homogenization  
778 temperature of  $-17^{\circ}\text{C}$  and a range of He and Ar proportions until the maximum convergence  
779 temperature for each fluid mixture is reached. The calculations reveal that, for a  $\text{CO}_2$ -Ar mixed fluid,  
780 the presence of 2 mol% Ar lowers the estimated equilibration pressure by  $<2\%$  (at  $\sim 450^{\circ}\text{C}$ ) relative  
781 to the isochore of a pure- $\text{CO}_2$  fluid). The presence of He, however, can increase the estimated  
782 equilibration pressure by 15% (at a temperature of  $\sim 270^{\circ}\text{C}$ ), with this relative offset decreasing with  
783 increasing temperature (see Supplementary Information). Therefore, while we cannot directly



784 estimate the equilibration pressure of CO<sub>2</sub>-He and CO<sub>2</sub>-Ar mixed fluids at the storage temperature of  
785 the Isla Floreana gabbros, we conclude that these impurities are likely to have a very minor negative  
786 (Ar) or moderate positive (He) influence on the estimated storage pressure derived from the FIs in  
787 samples 03b and 04b. As a result, although uncertainties in fluid composition undoubtedly influence  
788 the pressure estimates derived from FI measurements, we suggest that the FIs found in samples 03b  
789 and 04b confirm that the Isla Floreana gabbros were stored at pressures greater than ~700 MPa and  
790 less than ~1 GPa (maximum pressure estimated from CO<sub>2</sub>-H<sub>2</sub>O calculations combined with the  
791 potential influence of He), with a handful of fluid inclusions recording lower pressures due to  
792 decrepitation, leakage, re-equilibration, and/or stalling on ascent. Focusing on the fluid inclusions  
793 with melting temperatures within 0.1°C of pure CO<sub>2</sub> indicates that the final storage pressure is  
794 between 729 MPa ( $X_{\text{H}_2\text{O}} = 0.0$ ) to 873 MPa ( $X_{\text{H}_2\text{O}} = 0.15$ ). These pressure estimates closely align with  
795 the pressures determined by our thermodynamic calculations where the gabbro phase assemblage  
796 is recreated at 750 – 800 MPa and 850 – 1000 MPa (at 1020°C) for the pMELTS and Holland et al.  
797 (2018) thermodynamic models, respectively.

798 Converting these pressure estimates into depths using a constant crustal density of 2800 kg/m<sup>3</sup>  
799 above 16 km and an assumed mantle density of 3200 kg/m<sup>3</sup> below this depth, we constrain the  
800 storage depth of the Isla Floreana gabbros to 25.2 – 29.8 km depth ( $P = 730 - 870$  MPa). Critically,  
801 the depth of gabbro storage estimated here is significantly greater than the base of the oceanic crust  
802 beneath Isla Floreana (~16 km; Feighner and Richards, 1994), indicating that these samples likely  
803 formed, and were stored, within the mantle portion of the lithosphere. Therefore, we propose that  
804 during the plume-proximal stage of Isla Floreana's volcanic history magma storage was characterized  
805 by the presence of "crustal" intrusions (i.e., plagioclase-bearing cumulates) within the lherzolite –  
806 harzburgite mantle residue of the Nazca lithosphere.

807 Owing to the small number of samples considered in this study we cannot rule out the possibility  
808 that additional samples would reveal the presence of a wider (i.e. more vertically extensive) region

809 of magma storage beneath Isla Floreana. However, we note that a separate set of gabbroic samples  
810 analyzed by Lyons et al. (2007) also contain the opx-bearing mineralogy signature that is only stable  
811 at sub-Moho pressures (see Supplementary Material). Overall, we find no evidence for magma  
812 storage at shallower pressures, either during the plume-proximal or plume-distal period of Isla  
813 Floreana's magmatic history, indicating that crustal-level magma storage systems and/or intrusive  
814 bodies near the Moho either did not occur or were small and ephemeral and are not represented in  
815 the Isla Floreana xenolith populations.

## 816 5 IMPLICATIONS FOR VOLCANIC STRUCTURE AND EVOLUTION IN NEAR- 817 RIDGE SETTINGS.

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818 Traditional models for volcanic evolution in ocean island settings are largely based on observations  
819 from Hawai'i, and relate changes in the magma storage conditions and erupted compositions of  
820 lavas to the position of each volcano relative to the centre of plume upwelling (Clague and Sherrod,  
821 2014; Geist et al., 2014, 1995). As the pressure of magma storage beneath basaltic ocean island  
822 volcanoes and the flux of mantle derived magma into the lithosphere have recently been shown to  
823 be tightly linked (Baxter et al., 2023; Baxter and MacLennan, 2024; Gleeson et al., 2020a), we would  
824 expect to see a systematic increase in the pressure of magma storage as volcanic systems move  
825 away from the centre of plume upwelling; i.e., the shield to post-shield transition on Hawai'i (Clague  
826 and Sherrod, 2014).

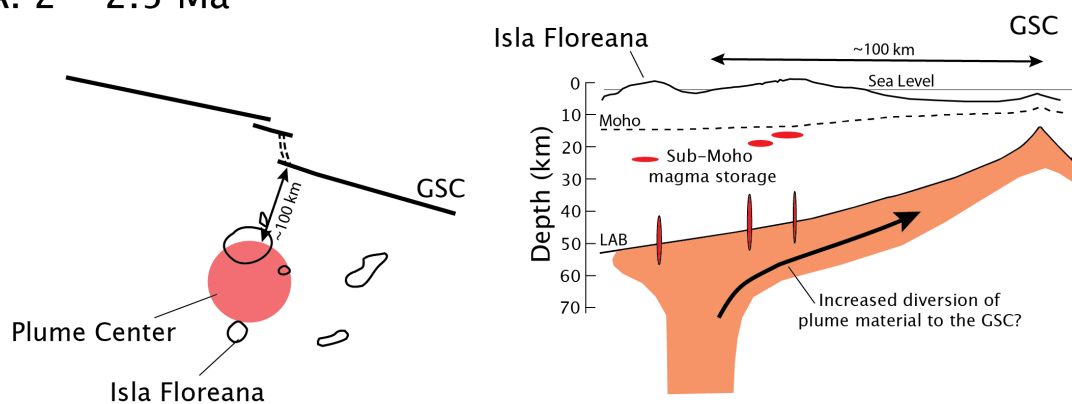
827 If we consider the modern-day volcanic activity in the Galápagos, the available data appears to fit  
828 this prediction very well. Plume-proximal systems that receive the largest flux of mantle-derived  
829 magma, such as Isla Fernandina, Volcán Wolf, Volcán Sierra Negra, or Volcán Cerro Azul, are  
830 characterized by mid-crustal magma storage regions that feed regular volcanic activity (Bell et al.,  
831 2021; Higgins and Stock, 2024; Stock et al., 2018). By contrast, the recent volcanic activity at Isla

832 Floreana, located ~100 – 150 km east-south-east of the centre of Galápagos plume upwelling, is  
833 characterized by the high-pressures of magma storage typical of a post-shield, or plume-distal, low  
834 melt-flux volcanic centre (Gleeson et al., 2022, 2020a; Fig. 12). From these observations we may  
835 have expected the Isla Floreana gabbros, associated with the plume-proximal period of Isla  
836 Floreana's magmatic history (Lyons et al., 2007), to record shallow, mid-crustal pressures (similar to  
837 the modern-day western Galápagos shield volcanoes). However, our new data indicates that the Isla  
838 Floreana gabbros were stored around 10 km below the base of the crust at pressures of ~750 MPa,  
839 similar to the pressure estimated for the recent Isla Floreana magmatic system (Gleeson et al., 2022,  
840 2020a).

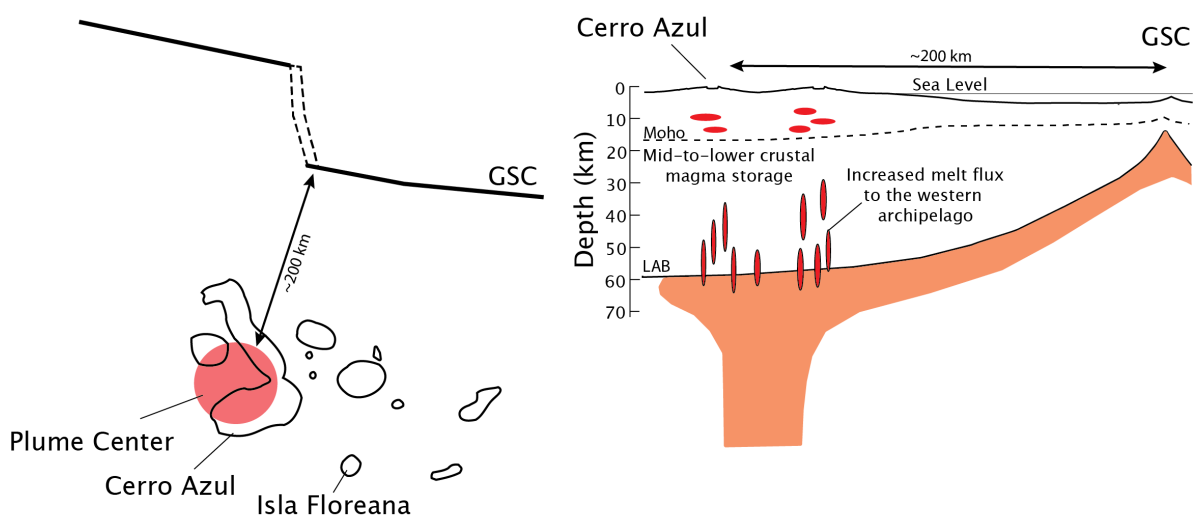
841 As indicated above, a wider region of magma storage – extending to the base of the oceanic crust –  
842 cannot be ruled out owing to the small number of samples analyzed. However, our data, combined  
843 with the mineralogical signature of other xenolith sample suites from Isla Floreana, clearly indicates  
844 that there has been persistent high-pressure magma storage at depths greater than the base of the  
845 crust throughout the ~2.5 Myrs of magmatic activity on Isla Floreana. In fact, there is no geochemical  
846 or petrological data currently available to indicate that Isla Floreana underwent any period of  
847 sustained magma storage within the crustal portion of the lithosphere. The absence of crustal  
848 storage beneath Isla Floreana, and other eastern Galápagos volcanic islands, is supported by the  
849 absence of shallow-level crustal cumulates detected by recent gravity surveys on other islands in the  
850 eastern archipelago (Cleary et al., 2020). We suggest that Isla Floreana has been characterized by a  
851 low magma flux (relative to the modern western Galápagos islands of Isabela and Fernandina)  
852 throughout its 2.5 Myr history, insufficient to sustain crustal-level magma storage (as links between  
853 magma storage depths and mantle derived melt flux have recently been established Baxter et al.,  
854 2023; Gleeson et al., 2020a). Importantly, this indicates that the eastern Galapagos volcanic islands  
855 cannot be considered as older 'analogues' for the present-day western Galapagos shield volcanoes.  
856 The different morphologies and chemical heterogeneity of erupted products, therefore, do not

857 relate to different stages of magmatic evolution, but are likely controlled by differences in the  
 858 magma flux rate through these volcanoes magmatic histories.  
 859 The origin of the anomalously low magma flux to the Isla Floreana magma system (and other eastern  
 860 Galapagos volcanic islands) during its plume proximal stage of volcanic activity could have several  
 861 origins. Temporal variations in the volume flux of mantle plume systems can occur – and have been  
 862 documented in Hawai’i – on similar timescales to that investigated here (Morrow and Mittelstaedt,  
 863 2021; Van Ark and Lin, 2004; Vidal and Bonneville, 2004; Wessel, 2016). However, we suggest that

### A. 2 – 2.5 Ma



### B. Present Day



**Figure 12** – Comparison of the tectonic setting of the Galápagos archipelago at 2 – 2.5 Ma (A.) and at the present day (B.). Reconstruction of the Galápagos Spreading Centre (GSC) is taken from Mittelstaedt et al. (2012), with the right hand panels adapted from Cleary et al. (2020). Schematic diagrams show the proposed increase in melt flux to the western Galápagos due to the increasing plume-ridge interaction distance with time and the smaller volume of plume material (and melt) transported north to the GSC. LAB – Lithosphere Asthenosphere Boundary.

864 our data supports the recent hypothesis of Cleary et al. (2020), that the persistent low melt flux  
865 supplied to the eastern Galapagos islands is related to the changing tectonic setting of the Galápagos  
866 mantle plume. Specifically, ~2-2.5 Myrs ago, when Isla Floreana and the other eastern Galápagos  
867 volcanic islands were located close to the centre of plume upwelling, the GSC was ~100 km closer to  
868 plume stem than it is at the present day. The increased proximity of the plume to the ridge at this  
869 time might have allowed more plume material and melt to be diverted northwards towards the  
870 ridge axis, causing more distributed volcanism across the Galápagos Platform and lowering the melt  
871 flux beneath the islands (Cleary et al., 2020; Gleeson and Gibson, 2021; Harpp and Geist, 2018), an  
872 interpretation which is supported by localized crustal thickness anomalies to the north of the GSC  
873 (Mittelstaedt et al., 2014). Furthermore, diversion of the plume material northwards, towards to  
874 GSC, might have the greatest effect on Isla Floreana owing to its location at the southern margin of  
875 the Galápagos Platform. As a result, when the eastern Galápagos islands were formed, Isla Floreana  
876 was characterized by a far lower flux of magma into the lithosphere than there is beneath the  
877 present-day western Galápagos shield volcanoes of Volcán Cerro Azul or Volcán Sierra Negra.

## 878 6 CONCLUSIONS

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879 Our new clinopyroxene trace element data, alongside published isotopic data of similar samples  
880 (Lyons et al., 2007), indicates that gabbroic xenoliths found in lava and scoria deposits on Isla  
881 Floreana in the southern Galápagos formed at ~2 – 2.5 Ma, when the island was located close to the  
882 centre of plume upwelling. Consequently, these samples provide an important insight into the  
883 temporal evolution of the Isla Floreana magmatic system, as well as the evolution of the Galápagos  
884 Archipelago in a rapidly evolving tectonic setting. Mineral compositions determined by EDS and  
885 WDS, alongside analysis of CO<sub>2</sub>-density in clinopyroxene- and plagioclase-hosted fluid inclusions, are  
886 used to estimate the final storage conditions of these gabbros prior to entrainment and ascent in  
887 their carrier melt. Our results indicate that the Isla Floreana gabbros were stored at ~900 – 1050 °C,

888 with pressures determined by clinopyroxene-orthopyroxene thermobarometry, fluid-inclusion CO<sub>2</sub>  
889 densities and thermodynamic predictions of phase stability ranging from ~700 MPa to 1 GPa.

890 The storage pressures estimated here indicate that the gabbroic nodules were stored below the  
891 base of the crust (~16 km), likely at depths between 25.2 – 29.8 km (although a wider, more  
892 vertically extensive magma storage zone extending to the base of the crust remains a possibility).

893 Consequently, our new data indicate that there has been persistent, high-pressure (i.e., lithospheric  
894 mantle) magma storage beneath Isla Floreana throughout its 2.5 Myrs of magmatic activity, with no  
895 petrological evidence indicating sustained crustal-level magma storage ever occurred, contrary to  
896 traditional models for ocean island evolution. In fact, our data indicates that there are no substantial  
897 differences between the magma storage conditions estimated for the plume-proximal and plume-  
898 distal magmatic systems beneath Isla Floreana (Gleeson et al., 2022, 2020a). In addition, our data  
899 indicates that the magma storage conditions beneath the plume-proximal stage of Isla Floreana's  
900 magmatic history are very different to the mid-crustal storage regions identified beneath present-  
901 day volcanic systems in the western Galápagos (Bell et al., 2021; Stock et al., 2018). As a result, our  
902 new data supports recent suggestions that the eastern Galápagos volcanic centres do not represent  
903 older analogs of the western Galápagos shield volcanoes (Cleary et al., 2020) but instead are  
904 characterized by persistent high-pressure magma storage, possibly resulting from the interaction of  
905 the Galápagos mantle plume with the nearby Galápagos Spreading Centre. Consequently, we  
906 suggest that the proximity of an ocean island to a nearby ridge segment has a substantial influence  
907 on the architecture of magma storage (as plume material and melt may be diverted away from the  
908 plume centre and towards the ridge axis) and the evolution of magmatic systems.

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## 915 DATA AVAILABILITY STATEMENT

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916 All data collected during the preparation of this project, including standard analyses, is available via  
917 [https://github.com/gleesonm1/GleesonEtAl\\_JPet\\_2024\\_supplement](https://github.com/gleesonm1/GleesonEtAl_JPet_2024_supplement) and has been archived using  
918 Zenodo (<https://doi.org/10.5281/zenodo.12746046>). In addition, all supplementary data can be  
919 found on EarthChem (<https://doi.org/10.60520/IEDA/113382>). All Python code used to create the  
920 figures presented as part of this manuscript can be found in the Zenodo repository listed above.  
921 Other code packages used in this study include: PetThermoTools v0.2.29  
922 ([github.com/gleesonm1/PetThermoTools](https://github.com/gleesonm1/PetThermoTools)); chemMap v0.0.4 ([github.com/gleesonm1/chemMap](https://github.com/gleesonm1/chemMap));  
923 Thermobar v1.0.43 ([github.com/PennyWieser/Thermobar](https://github.com/PennyWieser/Thermobar)); DiadFit v1.0.2  
924 ([github.com/PennyWieser/DiadFit](https://github.com/PennyWieser/DiadFit)); mineralML v0.0.0.8 ([github.com/sarahshi/mineralML](https://github.com/sarahshi/mineralML));  
925 laserTRAM ([github.com/jlubbersgeo/laserTRAM-DB](https://github.com/jlubbersgeo/laserTRAM-DB)); and pyrolite v0.3.5  
926 ([github.com/morganjwilliams/pyrolite/tree/develop](https://github.com/morganjwilliams/pyrolite/tree/develop)). Thermodynamic calculations were performed  
927 using alphaMELTS for Python v2.3.1 and MAGEMin\_C v1.4.9.

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# Supplementary Material: Persistent high-pressure magma storage beneath a near-ridge ocean island volcano (Isla Floreana, Galápagos)

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Supplementary Data and code associated with this project can be found in the associated Zenodo Repository: <https://doi.org/10.5281/zenodo.12746046> and EarthChem repository: <https://doi.org/10.60520/IEDA/113382>

# Volcanic and chemical divides

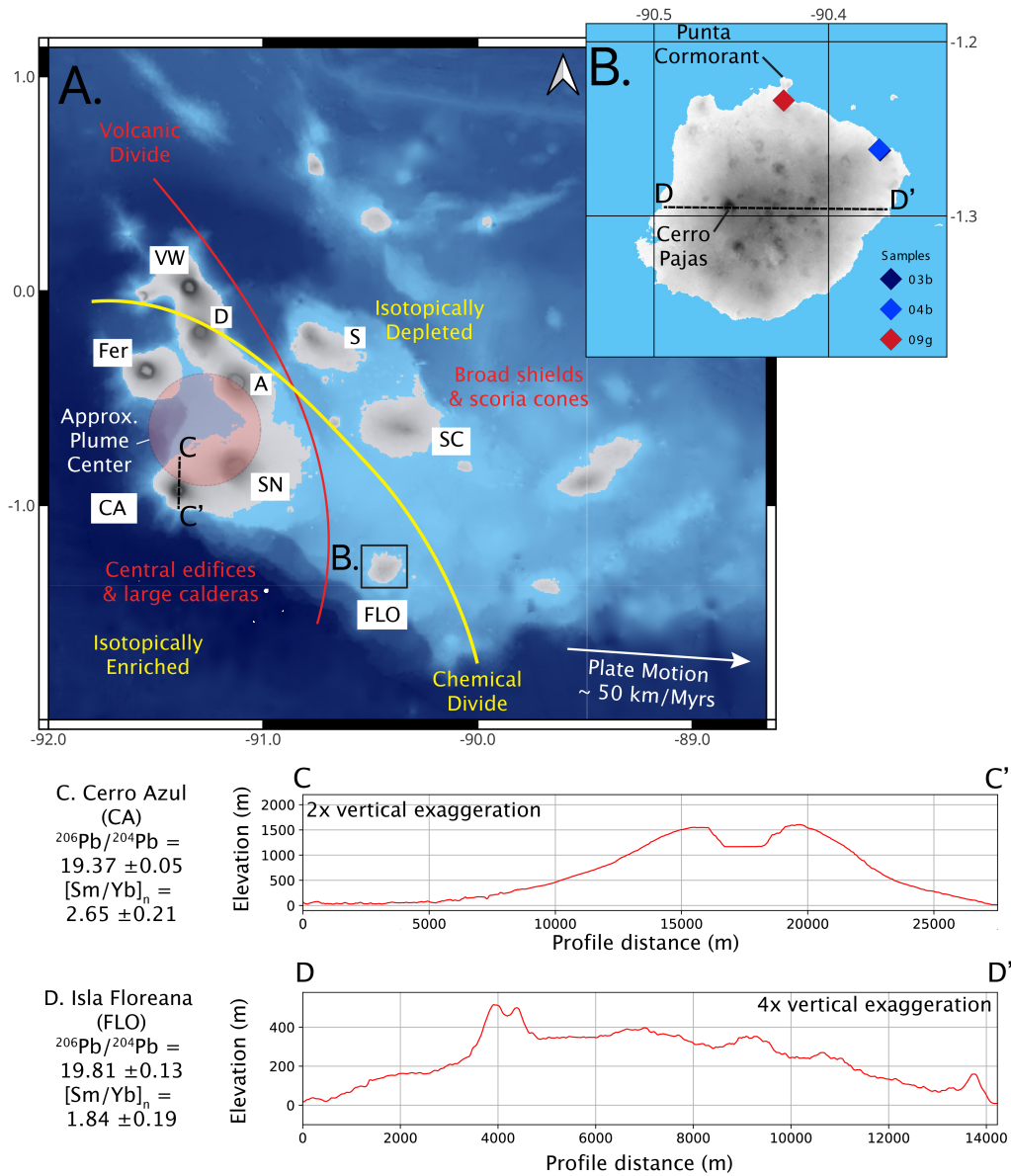


Fig. S.1: A copy of Figure 1 used in the main text. This figure here also contains topographic profiles across Volcán Cerro Azul and Isla Floreana, showing the different morphologies of volcanic systems in the western and eastern Galápagos, respectively.

## Standard analyses, precision and accuracy

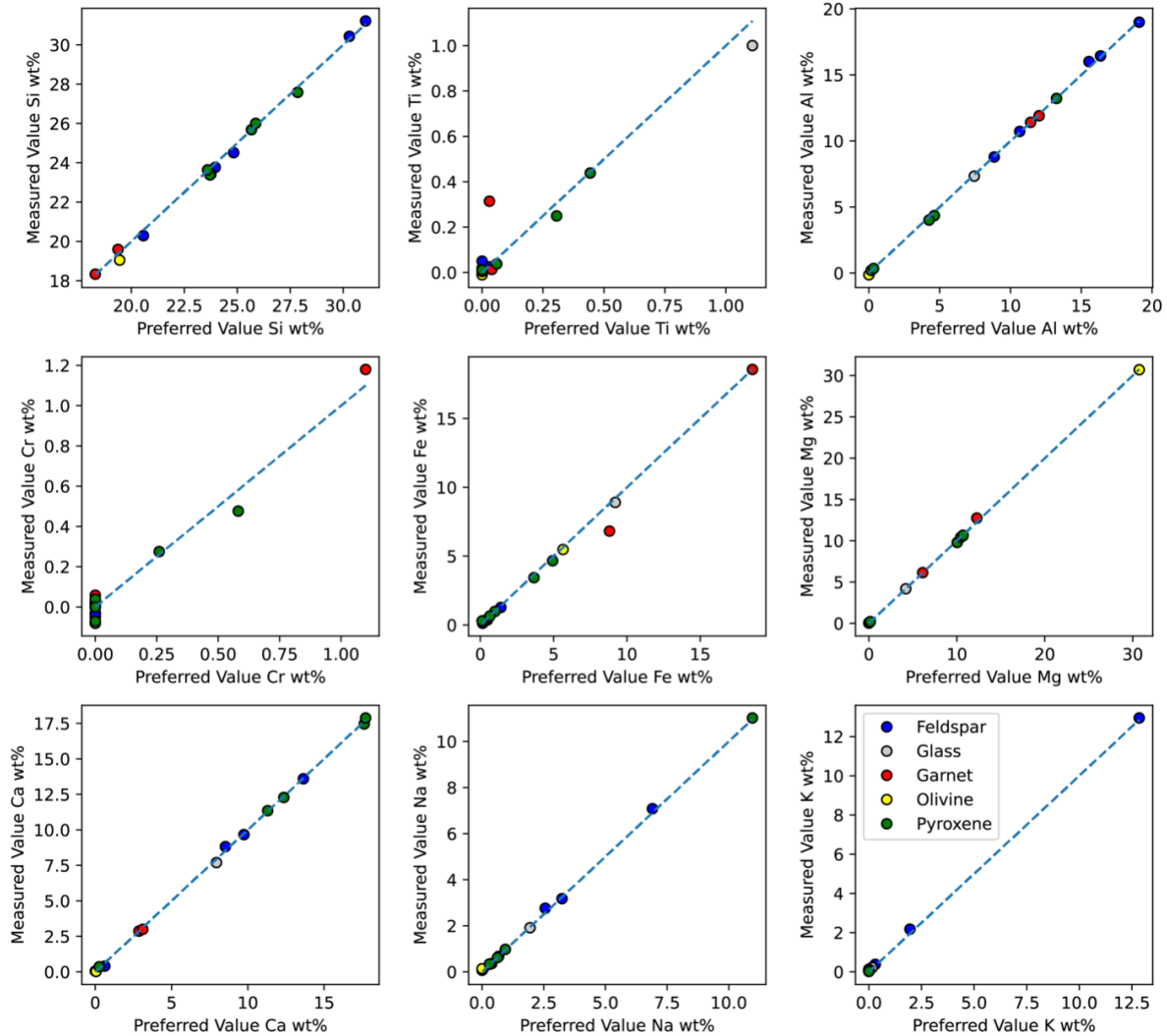


Fig. S.2: Comparison of measured and expected standard values for quantitative EDS mapping. Measured values are taken as the mean concentration of that element across the entire sample map. Preferred values represent the given value for each standard by Astimex or the Smithsonian. We note that our EDS maps detected significant TiO<sub>2</sub> in a Pyrope Garnet standard where only 0.04 wt% TiO<sub>2</sub> is reported by Astimex. This sample also reveals a slight offset in the measured and preferred FeO<sub>t</sub> values, which is not seen in our other standards.

Mineral major and minor element chemistry - across spatial scales ranging from full sample maps to individual point analyses - was collected by Energy Dispersive Spectrometry (EDS) using 2 X-Max 150 EDS detectors on the Zeiss Sigma HD Field Emission Gun Scanning Electron



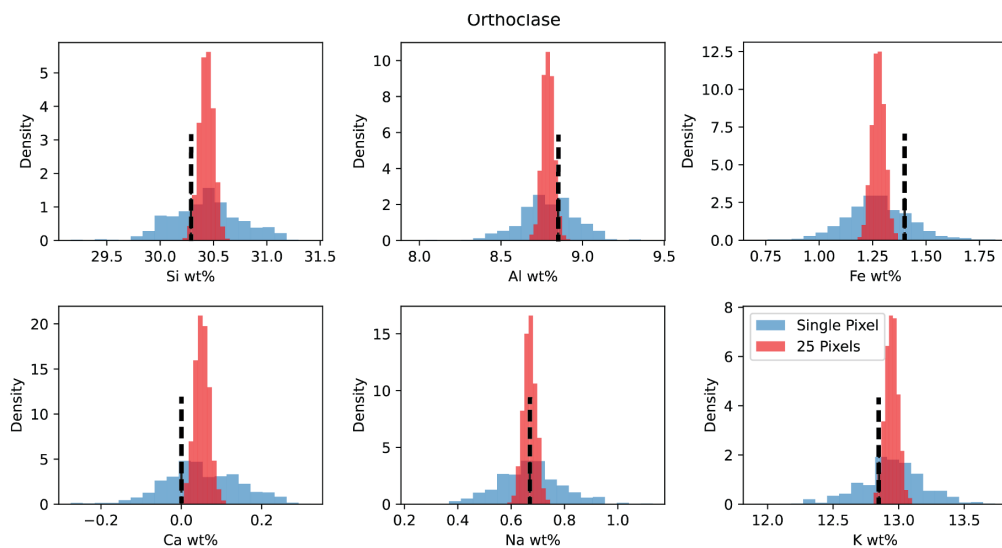


Fig. S.3: Comparison of standard map data to preferred standard values for Orthoclase. Blue histograms represent the concentrations determined from each individual pixel of the maps. Red histograms represent 1000 iterations of randomly sampled 5-by-5 pixel areas. Black dashed lines represent the preferred standard values.

Microscope (FEG-SEM) at Cardiff University. These data were then supplemented by EDS and Wavelength Dispersive Spectrometry (WDS) measurements of plagioclase crystals on the JEOL JSM IT800HL FEG-SEM at the University of California, Berkeley, using a single Ultim-Max 100 EDS detector for major elements, and a Wave WDS detector for Mg & K. Finally, Laser Ablation Inductively Coupled Plasma Mass Spectrometry (LA ICP-MS) analysis of clinopyroxene trace element concentrations was performed on a ASI RESOLUTION 193 nm Laser Ablation system coupled to an Agilent 8900 triple quadrupole ICP-MS hosted in the CELTIC laboratory at Cardiff University.

The primary calibration for EDS analyses performed at Cardiff University (whole-sample maps, linescans, and pyroxene point analyses) was constructed using a mix of mineral and metal standards: Albite (Si, Na), Almandine (Al, Fe), Diopside (Ca), Olivine (Mg), Rutile (Ti), Rhodonite (Mn), and Cr-metal (Cr). The accuracy of this calibration, for the different analysis types, was then assessed using a range of secondary standard materials sourced from either the Smithsonian Microbeam Standards Collection (Jarosewich et al., 1980) and Astimex Standards Ltd. The primary calibration used for EDS analysis of plagioclase crystals at UC Berkeley was constructed using Forsterite (Si), Diopside (Ca), and Nepheline (Na) standards, with the factory calibration used for all other elements. Primary and Secondary standard materials used at UC Berkeley originate from the California Division of Mines (Diopside; data from Charles Chesterman), Takei (1974; synthetic Forsterite), standards previously characterized at UC Berkeley by Ian Carmichael (Nepheline), and specimens (Albite and Labradorite) from Wenk and Kroll (1984).

Prior to and after collection of our whole-sample EDS map data we mapped a series of secondary standards, covering a range of different mineral groups, to assess the precision and accuracy of the map data. For each standard we collected a map,  $\sim 1$  mm by 1 mm, at the conditions used for our large-scale map analyses. Analytical accuracy was then calculated using

the mean concentration of each oxide in the EDS standard map and the preferred standard value (preferred values can be found in the associated Zenodo repository). Our data reveals that the analytical accuracy of all elements is almost always within 5%, and typically better than 2%, of the preferred value when the measured totals are >2 wt% (Fig. S.2).

To further assess the accuracy of our large-scale map data across a range of mineral groups we consider the measured composition of every pixel against the preferred standard value. This also allows us to assess the precision of EDS mapping for any particular pixel/spot and we also consider the influence of pixel 'binning', i.e., averaging the composition of pixels over a n-by-n area, on the precision of the standard map data by considering the compositions that would be returned by averaging a randomly selected 5-by-5 pixel grid (with the calculation performed 1000 times per standard). Fig. S.2 demonstrates the close agreement between the measured map data and preferred standard values for an Orthoclase standard. Additional graphs can be examined using the Jupyter Notebook supplied in the associated Zenodo repository. The relative precision of analysis is then calculated as the 1 sigma uncertainty in the standard map data, calculated both for individual pixel analyses and for 5-by-5 pixel bins (Fig. S.4).

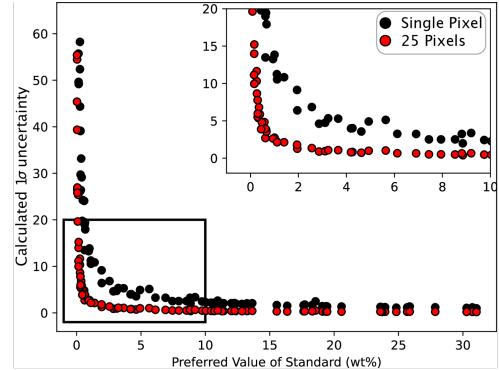


Fig. S.4: Comparison of the calculated uncertainty associated with the EDS map analysis for: (i) individual pixels; and (ii) a 5-by-5 pixel region.

Table 1: Recovery and Relative Standard Deviation for secondary standards analyzed alongside the orthopyroxene and clinopyroxene crystals found in the Galapagos Gabbros. Oxides with concentrations above 2 wt% (shown in bold) show analytical recovery between 98.9 and 100.3%, with the Relative Standard Deviation of these high concentration elements typically less than 0.4%.

Standard Material	Oxide (Preferred Value wt%)	Relative Standard Deviation	Recovery ( $\frac{Measured}{Preferred} \times 100$ )
Diopside	<b>SiO<sub>2</sub> (55.34)</b>	0.168	100.1
	<b>CaO (24.80)</b>	0.312	100.3
	<b>MgO (17.76)</b>	0.261	99.4
	Al <sub>2</sub> O <sub>3</sub> (0.62)	7.216	93.7
	FeO <sub>t</sub> (0.84)	2.664	107.9
	Na <sub>2</sub> O (0.40)	6.846	102
Cr-Diopside	<b>SiO<sub>2</sub> (54.91)</b>	0.076	100.2
	<b>CaO (24.670)</b>	0.198	99.5
	<b>MgO (17.73)</b>	0.205	99.2
	Al <sub>2</sub> O <sub>3</sub> (0.30)	5.862	95.7
	FeO <sub>t</sub> (1.29)	1.839	107
	Na <sub>2</sub> O (0.40)	4.611	101.2

Table 1: Recovery and Relative Standard Deviation for secondary standards analyzed alongside orthopyroxene and clinopyroxene crystals (continued).

Standard Material	Oxide (Preferred Value wt%)	Relative Standard Deviation (%)	Recovery (%)
	Cr <sub>2</sub> O <sub>3</sub> (0.30)	6.489	121.4
Jadeite	<b>SiO<sub>2</sub> (59.58)</b>	0.197	98.9
	<b>Al<sub>2</sub>O<sub>3</sub> (25.04)</b>	1.396	99.2
	<b>Na<sub>2</sub>O (14.77)</b>	1.094	99.9

Following full sample mapping the samples were characterized by point analyses and line-scans to examine the chemistry of specific crystals/regions. Full details of the secondary standard data for the point analyses performed alongside measurements of both pyroxene and plagioclase crystals can be found in Supplementary Datasets 1 & 2, with the main results presented in Tables S.1 and S.2 below. Analysis of secondary standard materials by LA ICP-MS can be found in Supplementary Dataset 3.

Table 2: Recovery and Relative Standard Deviation for secondary standards analyzed alongside the plagioclase crystals found in the Galapagos Gabbros. Results for MgO and K<sub>2</sub>O demonstrate the Recovery and Standards Deviation of WDS analysis, all other oxides were measured by EDS. When K<sub>2</sub>O contents are determined by EDS they are marked by a \*\*\*. Major elements for each standard (typically concentrations > 2 wt%) are shown in bold.

Standard Material	Oxide (Preferred Value wt%)	Relative Standard Deviation (%)	Recovery ( $\frac{Measured}{Preferred} \times 100$ )
Labradorite	<b>SiO<sub>2</sub> (51.25)</b>	0.193	99.8
	<b>Al<sub>2</sub>O<sub>3</sub> (30.91)</b>	0.11	99.7
	<b>CaO (13.40)</b>	0.262	100.1
	<b>Na<sub>2</sub>O (3.83)</b>	0.738	98.7
	FeO <sub>t</sub> (0.41)	5.763	100.3
	K <sub>2</sub> O (0.12)	1.33	97.6
		7.301***	98.9***
	MgO (0.14)	2.977	86.5
Anorthite	<b>SiO<sub>2</sub> (44.17)</b>	0.28	100.1
	<b>Al<sub>2</sub>O<sub>3</sub> (34.95)</b>	0.047	102.3
	<b>CaO (18.63)</b>	0.258	101.3
	Na <sub>2</sub> O (0.79)	3.288	84
Nepheline	<b>SiO<sub>2</sub> (43.49)</b>	0.074	100.9
	<b>Al<sub>2</sub>O<sub>3</sub> (33.76)</b>	0.172	100.6
	<b>Na<sub>2</sub>O (16.92)</b>	0.139	99.4

Table 2: Recovery and Relative Standard Deviation for secondary standards analyzed alongside the plagioclase crystals (continued).

Standard Material	Oxide (Preferred Value wt%)	Relative Standard Deviation	Recovery
	<b>K<sub>2</sub>O (5.61)</b>	0.448***	103.0***
Albite	<b>SiO<sub>2</sub> (68.241)</b>	0.07	101.4
	<b>Al<sub>2</sub>O<sub>3</sub> (19.90)</b>	0.374	99.5
	<b>Na<sub>2</sub>O (11.94)</b>	0.361	100.6
Orthoclase	<b>SiO<sub>2</sub> (64.793)</b>	0.035	101.5
	<b>Al<sub>2</sub>O<sub>3</sub> (16.72)</b>	0.203	98.9
	<b>K<sub>2</sub>O (15.49)</b>	0.161***	100.3***
	Na <sub>2</sub> O (0.91)	3.237	105.8
	FeO <sub>t</sub> (1.88)	2.044	94.2

The accuracy of the linescan analyses was tracked through linescans – carried out using identical analytical conditions to the unknowns – performed on Diopside and Cr-Augite secondary standard materials. Plotting up the recovery values for these linescans displays several things. First there is clear heterogeneity in the Diopside standard with one end of the linescan displaying a more Fe-rich and Mg-poor composition. This area was avoided for all other analysis of the Diopside standard performed in this study. Second, most oxides return recovery values centered on 100% (Fig. S.5)

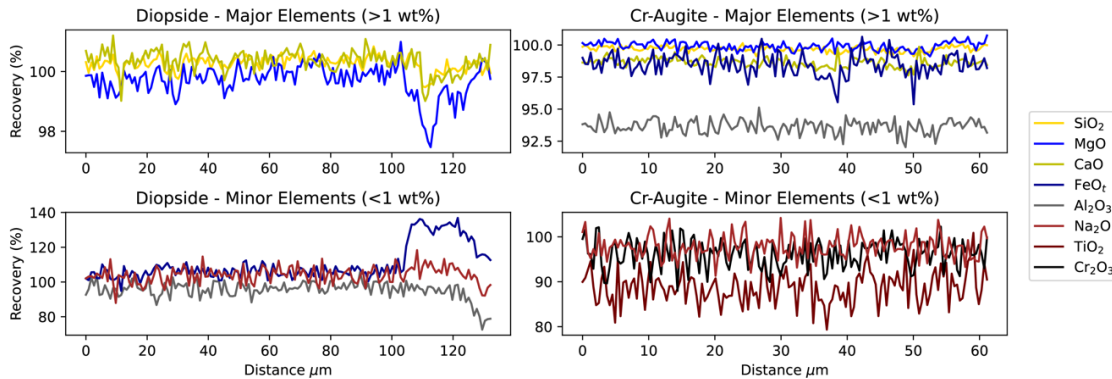


Fig. S.5: Linescan data from a Diopside and Cr-Augite standard relative to the preferred standard values.

## Whole-Rock Composition of the Isla Floreana Gabbros

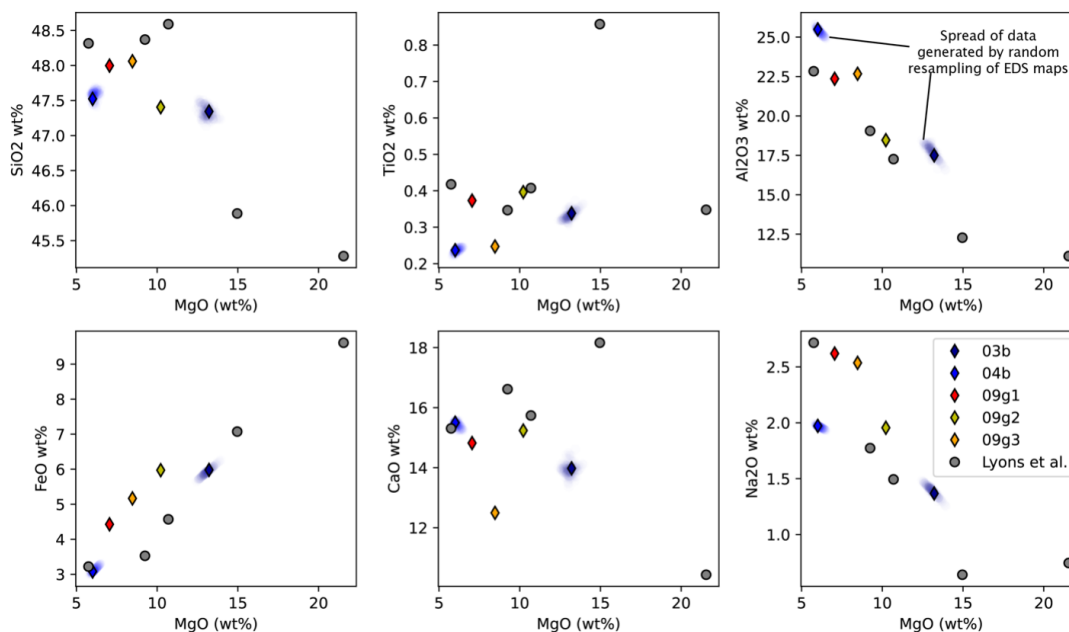


Fig. S.6: Comparison of the bulk-rock compositions generated by EDS mapping of the gabbro xenoliths analyzed in this study to XRF determined compositions of a separate suite of gabbro xenoliths that were also collected from the northern coast of Isla Floreana (data from Lyons et al. 2007). While these represent separate sample sets, we can see that the EDS-determined compositions lie within the range of composition determined by XRF on similar samples from the same island. Also shown is the spread of compositions generated by our bulk-rock error analysis (see Supplementary Text), where smaller subregions of samples 03b and 04b are randomly selected to calculate the uncertainty associated with our bulk-rock estimates.

As the EDS map data is shown to be accurate across a range of different mineral groups we use our EDS maps to estimate a ‘bulk-rock’ composition for each of the Galapagos gabbros following the methods outlined in the main text. For these calculations we used mineral densities of  $3.35 \text{ g/cm}^3$  for orthopyroxene,  $3.4 \text{ g/cm}^3$  for clinopyroxene,  $3.5 \text{ g/cm}^3$  for olivine,  $2.7 \text{ g/cm}^3$  for plagioclase, and  $4.05 \text{ g/cm}^3$  for spinel. Densities for all phases were taken from The Engineering Toolbox (2009), using linear interpolation between end-member components (based on mole proportions) to estimate the approximate density for each phase identified in the Galapagos gabbros. To assess the uncertainty generated by the random ‘cut’ of the sample analyzed by EDS mapping we ‘resampled’ the 03b and 04b maps 200 times, on each iteration a random set of coordinates was generated to extract an enclosed region covering at least 50% of the total sample surface and a bulk-rock composition was generated for this sub-area. This analysis reveals little variation in most elements, with the relative standard deviation of major elements  $<2\%$  (Fig. S.6).

The composition of the Isla Floreana gabbros also helps to confirm their origin as cumu-

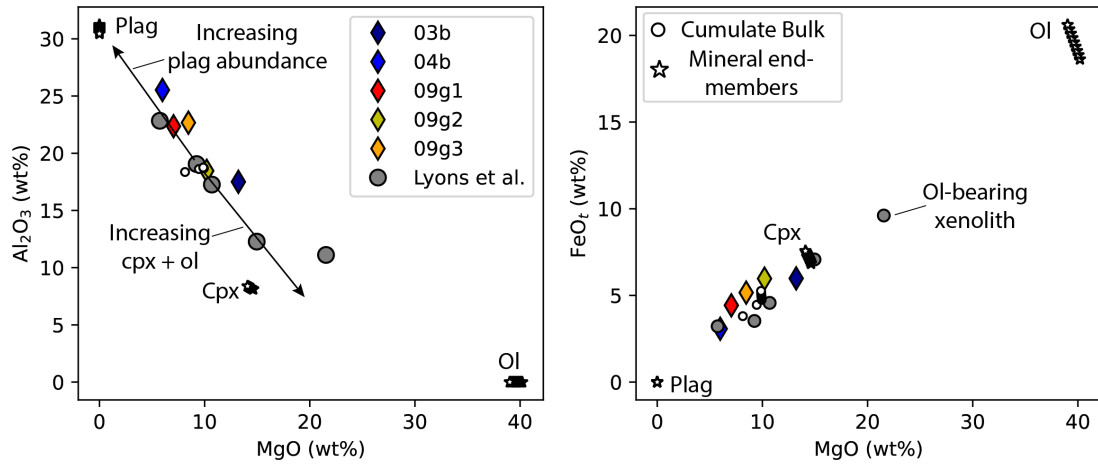


Fig. S.7: Gabbro bulk-rock compositions compared to the composition of cumulate residues and mineral compositions predicted by fractional crystallization models in pMELTS. Models demonstrate that the gabbro compositions plot along a mixing line between plagioclase and clinopyroxene/olivine, confirming that these samples likely originated as olivine-bearing gabbroic mushes, with orthopyroxene forming during subsolidus reactions.

late deposits, likely containing an initial mixture of olivine, clinopyroxene, plagioclase, and some residual melt (with olivine then being replaced by orthopyroxene during subsolidus reactions). Fractional crystallization calculations in pMELTS (performed using PetThermoTools) were used to track the composition of the cumulate residue (i.e., the bulk composition of the solid products) during gabbroic crystallization at 7000 bars. A high-MgO basalt from Cerro Azul, a  $\text{H}_2\text{O}$  content of 0.5 wt% and a  $\text{Fe}^{3+}/\text{Fe}_t$  ratio of 0.15 were used as the starting parameters. Results are shown in Fig. S.7, and demonstrate that the composition of all samples analyzed in this study (and by Lyons et al., 2007) can be explained as cumulate rocks with varying enrichment/depletion in plagioclase (potentially explained by density-driven variations in settling velocity).

## Phase Diagram Analysis

In the main text we use thermodynamic calculations to assess the conditions of subsolidus storage for the Isla Floreana gabbros. Here we show the phase diagrams calculated using PetThermoTools for samples 03b and 04b as well as gabbro FLX513 from Lyons et al. (2007). These calculations were performed using the Holland et al. (2018), pMELTS (Ghiorso et al. 2002), and rhyolite-MELTS (Gualda et al., 2012) thermodynamic models. In all figures below (Fig. S.8 – 16) we highlight the position of the olivine-out reaction and garnet-out reactions (with increasing and decreasing pressure, respectively), as well as the position of the orthopyroxene-in boundary (with decreasing pressure and temperature). These three reactions boarder the P-T regions where the mineral assemblage of the Isla Floreana gabbros are recreated by the thermodynamic models. Python code and Jupyter Notebooks used to create these phase diagrams are included in the Zenodo repository associated with this publication.

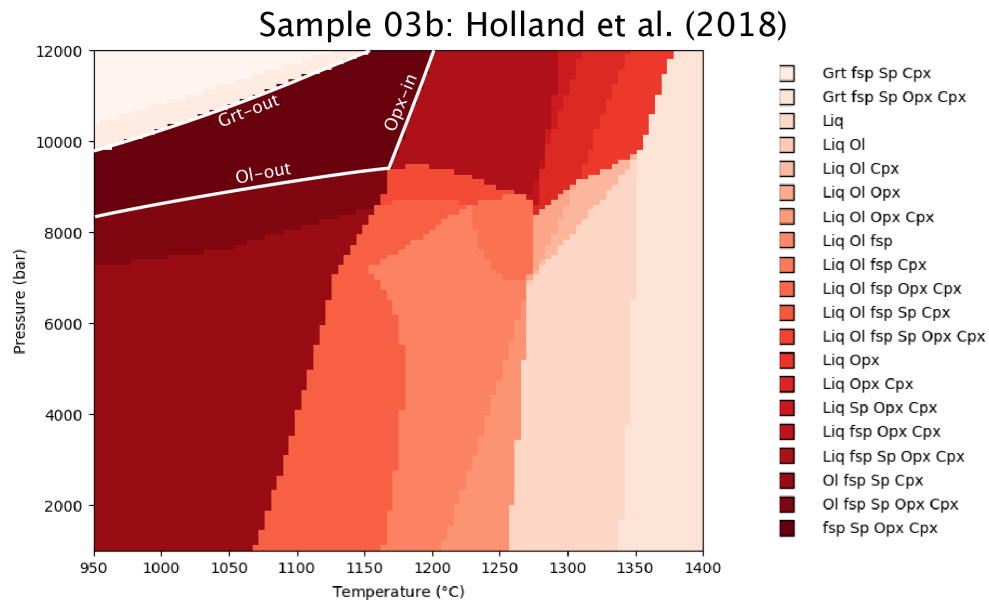


Fig. S.8: Phase diagram calculated using the EDS determined composition of sample 03b and the Holland et al. (2018) thermodynamic model. Calculations performed via PetThermoTools and MAGEMin (Gleeson & Wieser 2024; Riel et al. 2022; MAGEMin\_C v1.4.9). Position of the garnet, orthopyroxene, and olivine stability regions indicated by the white lines.

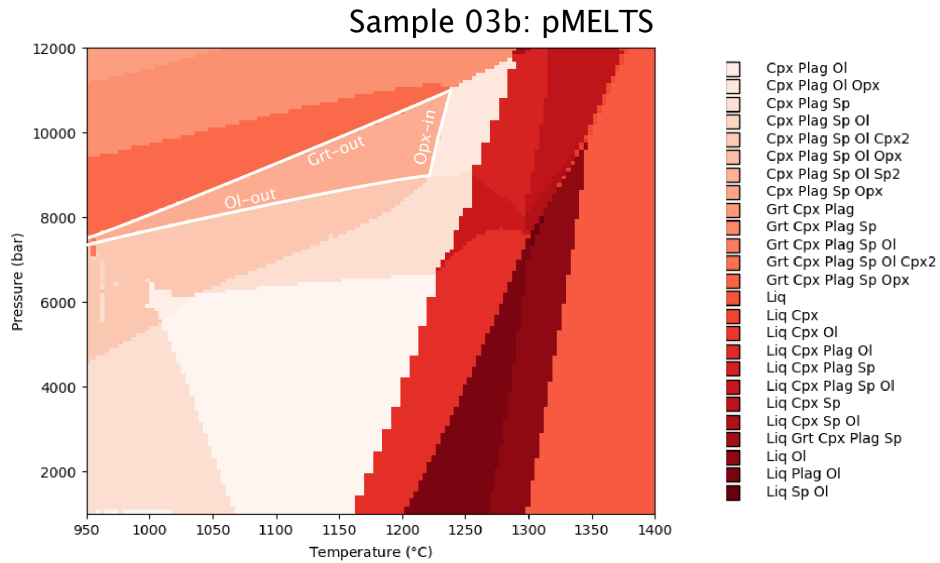


Fig. S.9: Phase diagram calculated using the EDS determined composition of sample 03b and the pMELTS thermodynamic model (Ghiorso et al. 2002). Calculations performed via PetThermoTools and alphaMELTS for Python (Gleeson & Wieser 2024). Position of the garnet, orthopyroxene, and olivine stability regions indicated by the white lines.

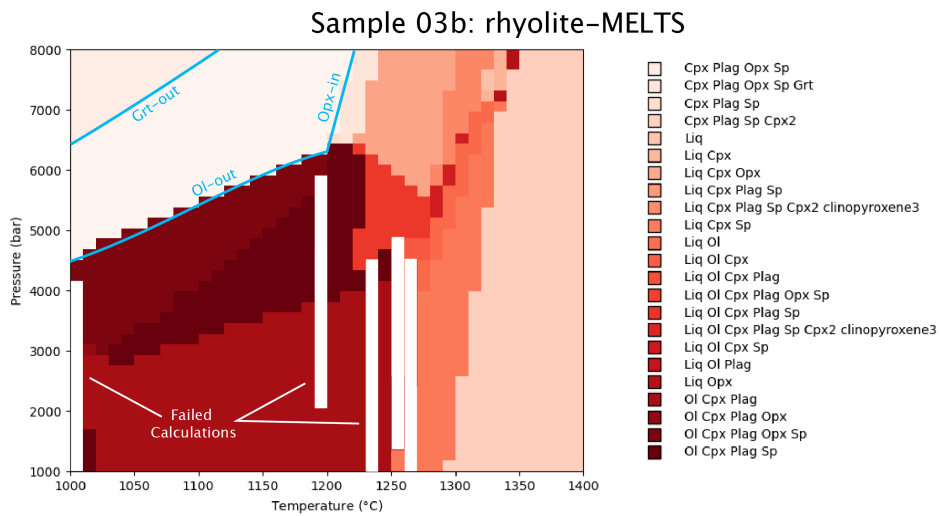


Fig. S.10: Phase diagram calculated using the EDS determined composition of sample 03b and the rhyolite-MELTS v1.0.2 thermodynamic model (Gualda et al. 2012). Calculations performed via PetThermoTools and alphaMELTS for Python (Gleeson & Wieser 2024). Position of the garnet, orthopyroxene, and olivine stability regions indicated by the blue lines.



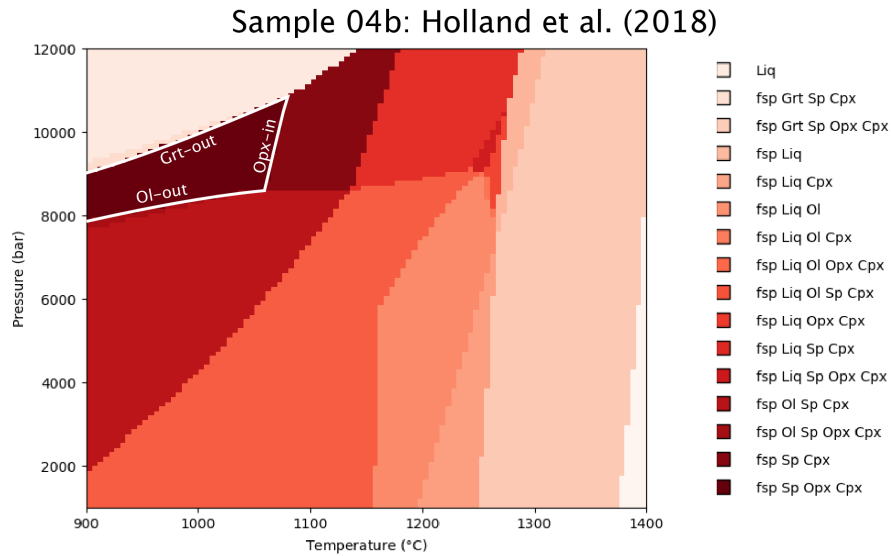


Fig. S.11: Phase diagram calculated using the EDS determined composition of sample 04b and the Holland et al. (2018) thermodynamic model. Calculations performed via PetThermoTools and MAGEMin (Gleeson & Wieser 2024; Riel et al. 2022; MAGEMin\_C v1.4.9). Garnet, orthopyroxene, and olivine stability regions indicated by the white lines.

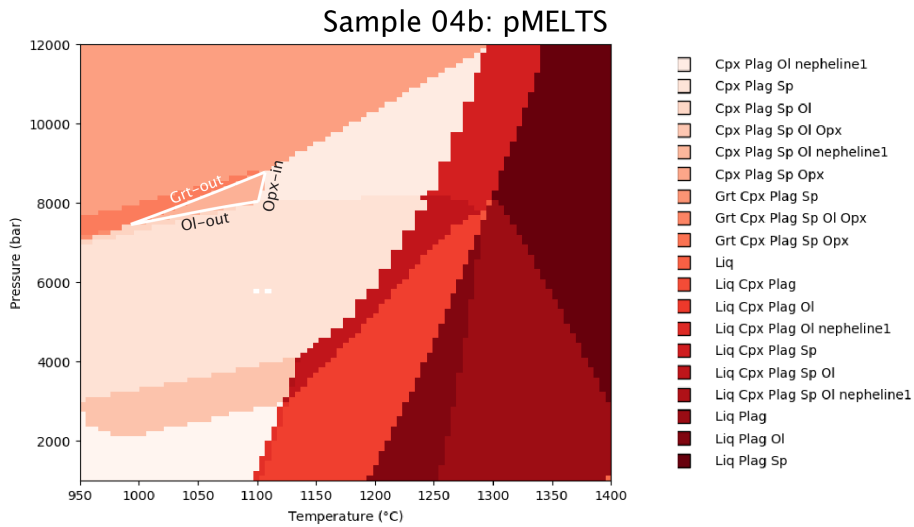


Fig. S.12: Phase diagram calculated using the EDS determined composition of sample 04b and the pMELTS thermodynamic model (Ghiorso et al. 2002). Calculations performed via PetThermoTools and alphaMELTS for Python (Gleeson & Wieser 2024). Position of the garnet, orthopyroxene, and olivine stability regions indicated by the white lines.

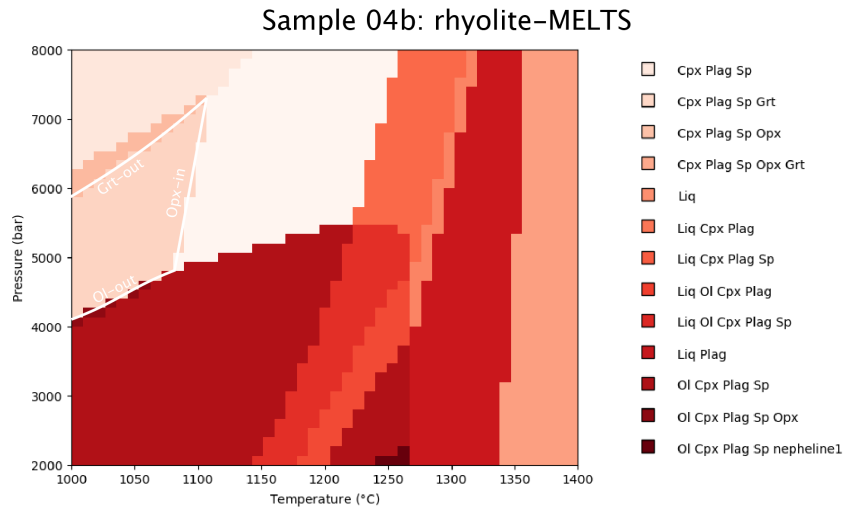


Fig. S.13: Phase diagram calculated using the EDS determined composition of sample 04b and the pMELTS thermodynamic model (Ghiorso et al. 2002). Calculations performed via PetThermoTools and alphaMELTS for Python (Gleeson & Wieser 2024). Position of the garnet, orthopyroxene, and olivine stability regions indicated by the white lines.

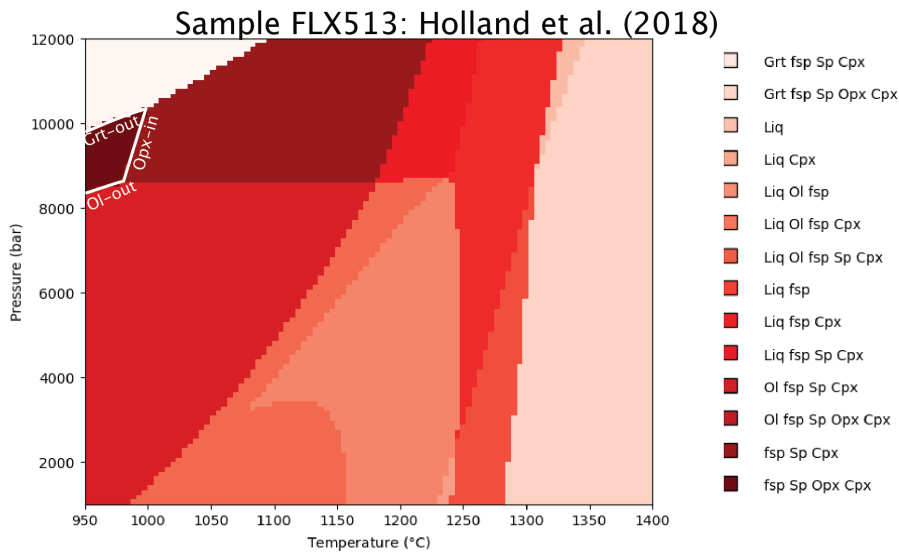


Fig. S.14: Phase diagram calculated using the XRF determined composition of sample FLX513 from Lyons et al. (2007) and the Holland et al. (2018) thermodynamic model. Calculations performed via PetThermoTools and MAGEMin (Gleeson & Wieser 2024; Riel et al. 2022; MAGEMin\_C v1.4.9). Garnet, orthopyroxene, and olivine stability regions indicated by the white lines.

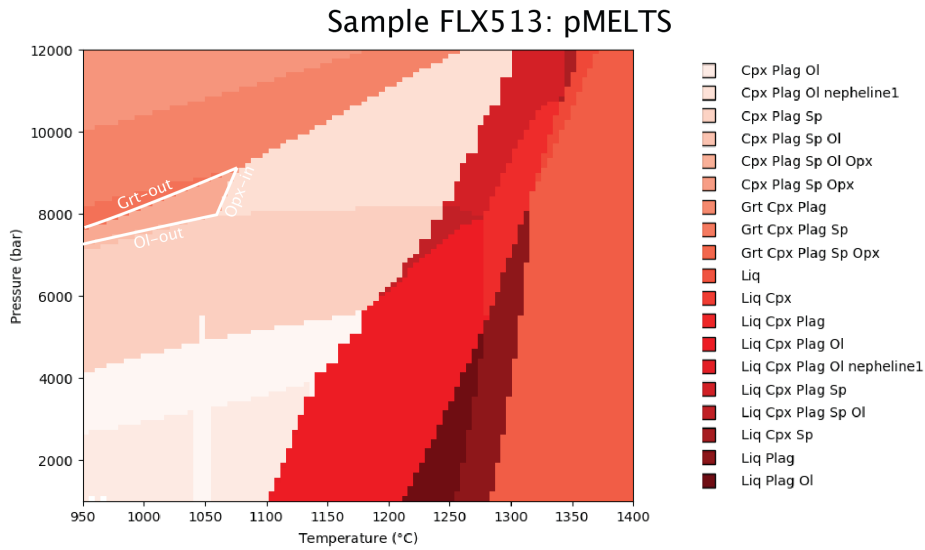


Fig. S.15: Phase diagram calculated using the XRF determined composition of sample FLX513 from Lyons et al. (2007) and the pMELTS thermodynamic model (Ghiorso et al. 2002). Calculations performed via PetThermoTools and alphaMELTS for Python (Gleeson & Wieser 2024). Position of the garnet, orthopyroxene, and olivine stability regions indicated by the white lines.

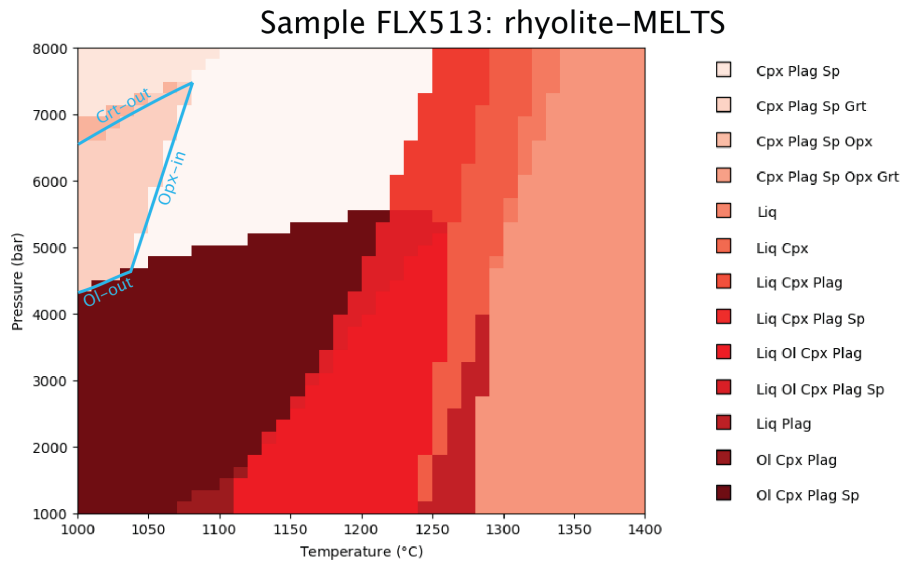


Fig. S.16: Phase diagram calculated using the XRF determined composition of sample FLX513 from Lyons et al. (2007) and the pMELTS thermodynamic model (Ghiorso et al. 2002). Calculations performed via PetThermoTools and alphaMELTS for Python (Gleeson & Wieser 2024). Position of the garnet, orthopyroxene, and olivine stability regions indicated by the blue lines.

## Phase Maps and Mineral Proportions

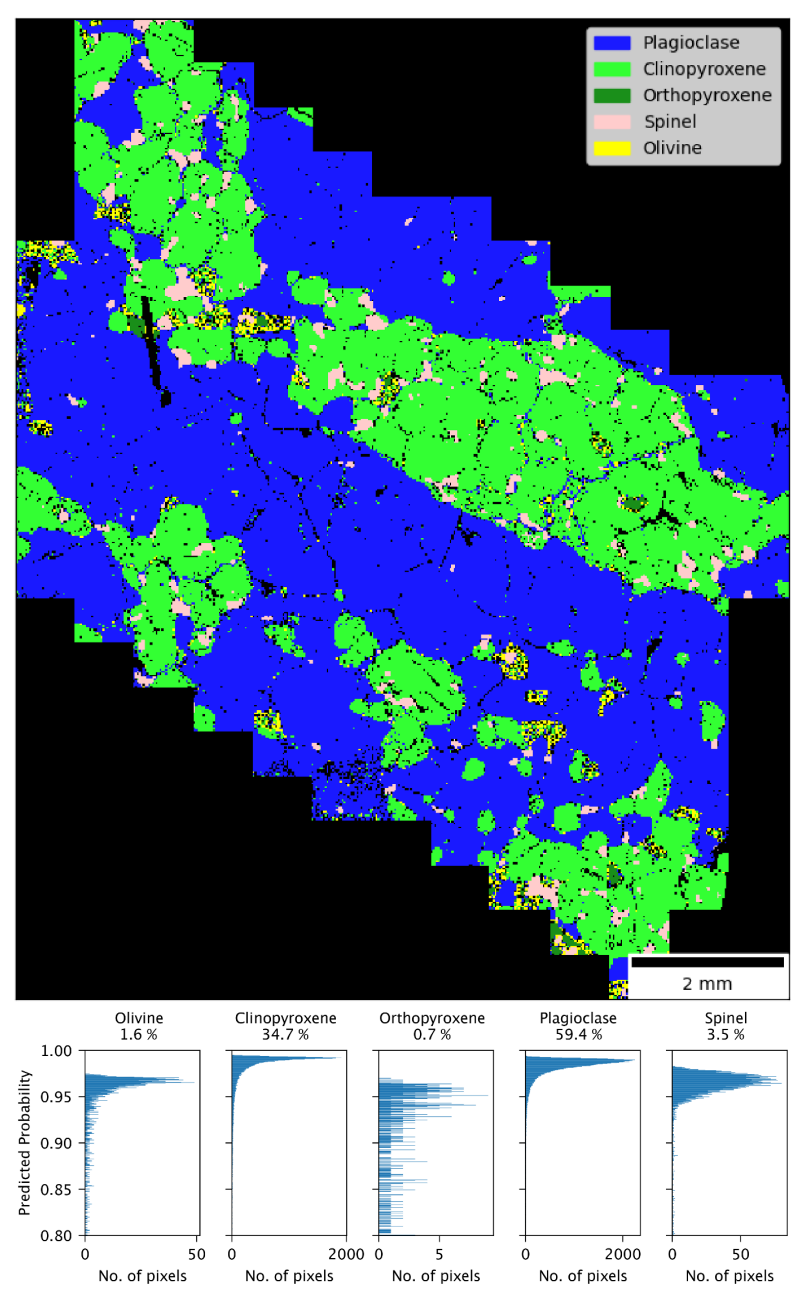


Fig. S.17: Phase map of sample 09g1 and associated phase proportions. As with all 09g samples olivine is identified at the boundaries of the orthopyroxene crystals

One of the main outcomes of our EDS maps, combined with the new capabilities of mineralML (Shi et al. 2023), is the ability to automatically convert EDS map data into phase maps and estimates of phase proportions. This is shown in Fig. 2 of the main text, with phase maps and associated phase proportions of the additional three samples shown here.

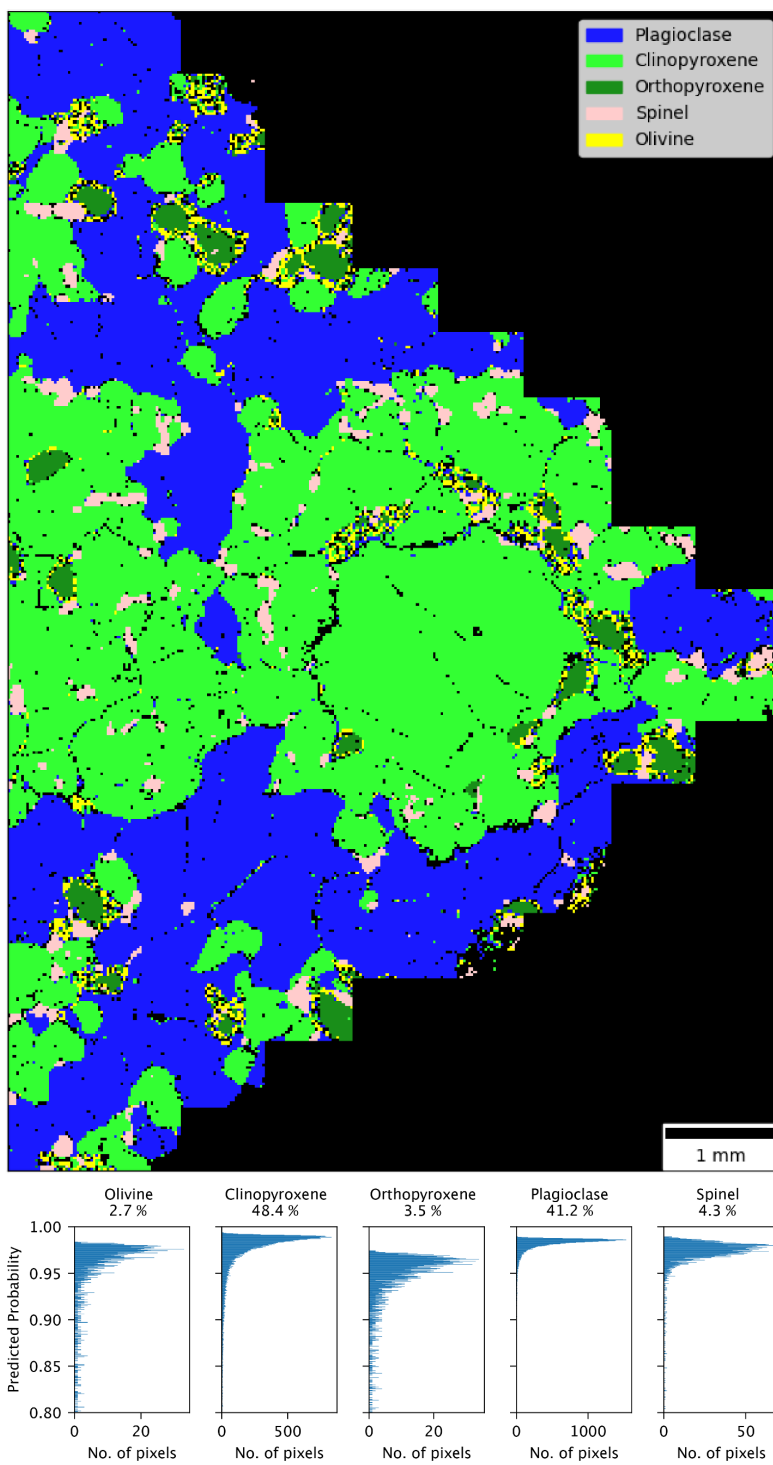


Fig. S.18: Phase map of sample 09g2 and associated phase proportions.

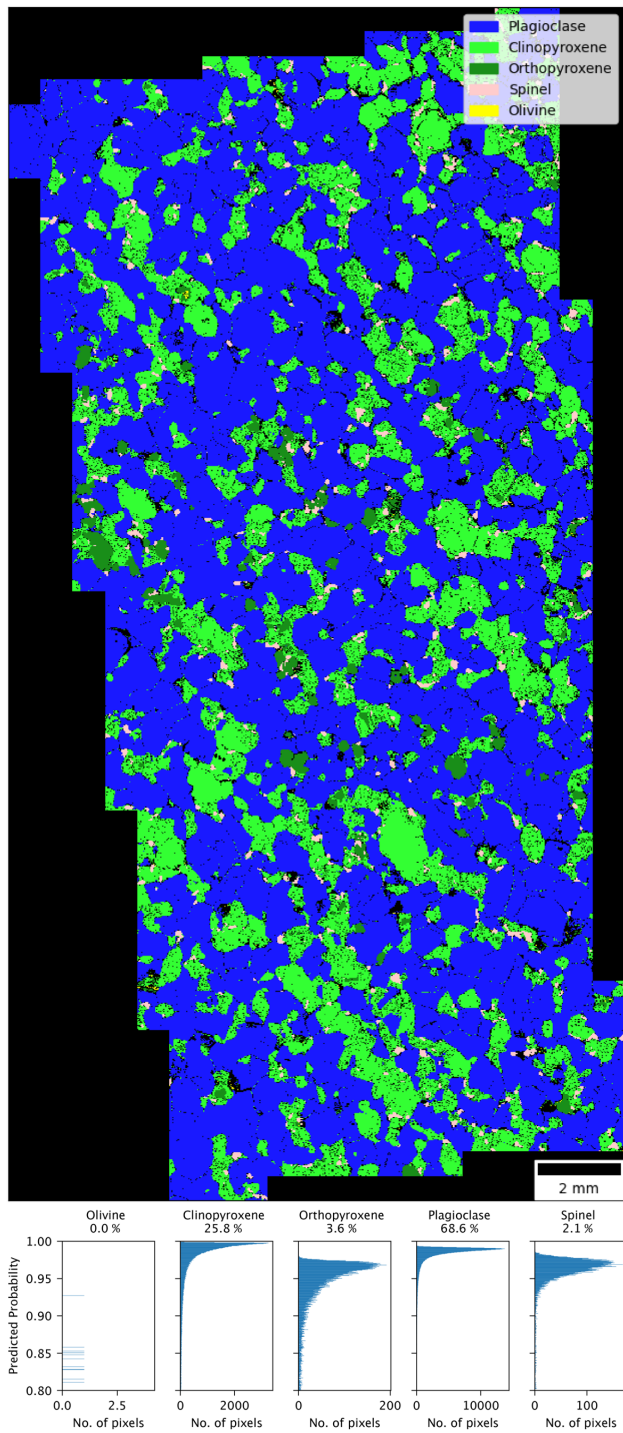


Fig. S.19: Phase map of sample 04b and associated phase proportions. This sample contains the highest proportion of plagioclase identified in any of the sample considered in this study.

## Compositional Variability

Data collected from EDS mapping provide insights into the compositional variability of the samples, with limited heterogeneity present with the exception of core-rim An zoning in sample 03b. Example compositional maps are shown below alongside histograms of the Plagioclase An content, Clinopyroxene Mg#, and Spinel Cr# calculated from the EDS map data. Code used to produce these figures is available in a Jupyter Notebook found within the associated Zenodo Repository.

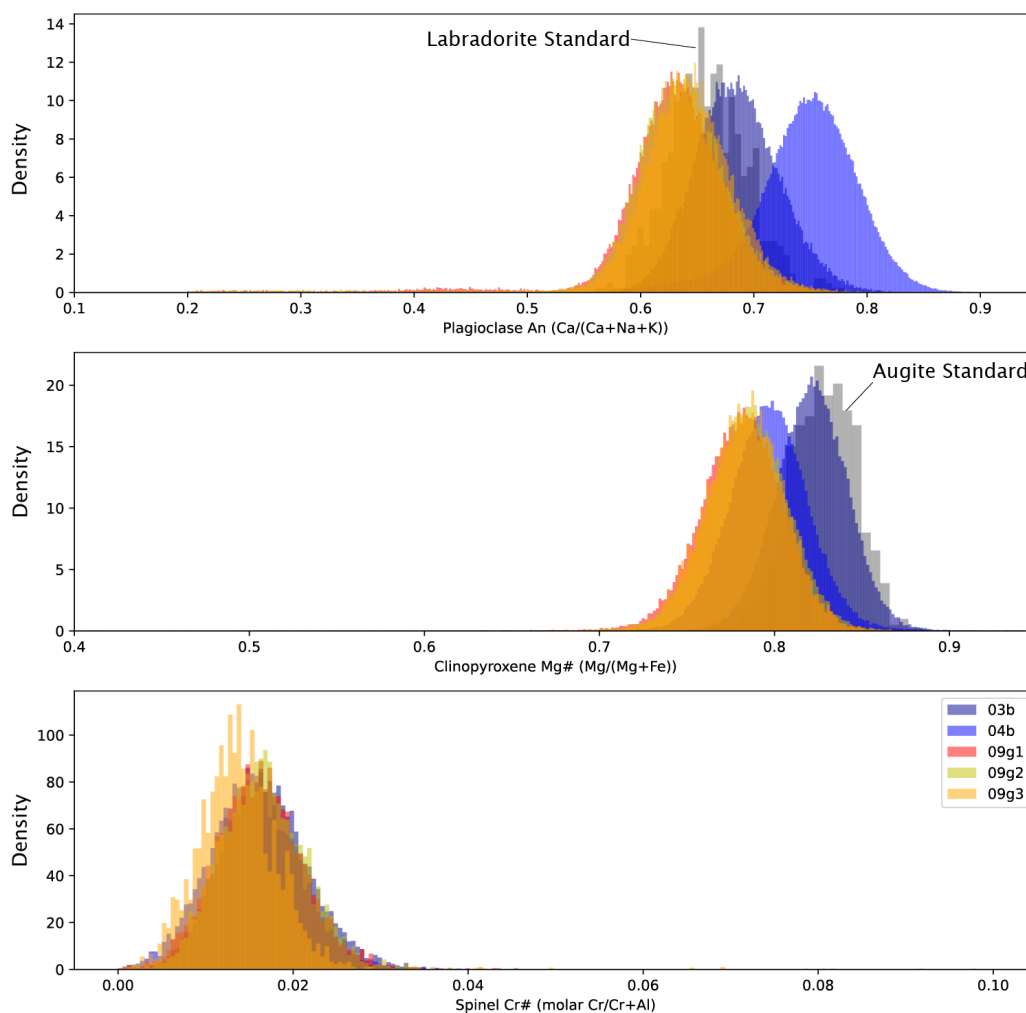


Fig. S.20: Histograms of key compositional parameters for the Isla Floreana gabbros. Plagioclase An contents reveals a tail to low An values for most samples, with these analyses typically found near or along grain boundaries. Labradorite and Augite standards indicate that the spread of compositions associated with the primary normal distribution of each sample is largely controlled by the analytical uncertainty of the EDS map analyses. Spinel crystals have extremely low Cr# values. Data for 09g1, 09g2, and 09g3 are near identical and overlap on all plots.

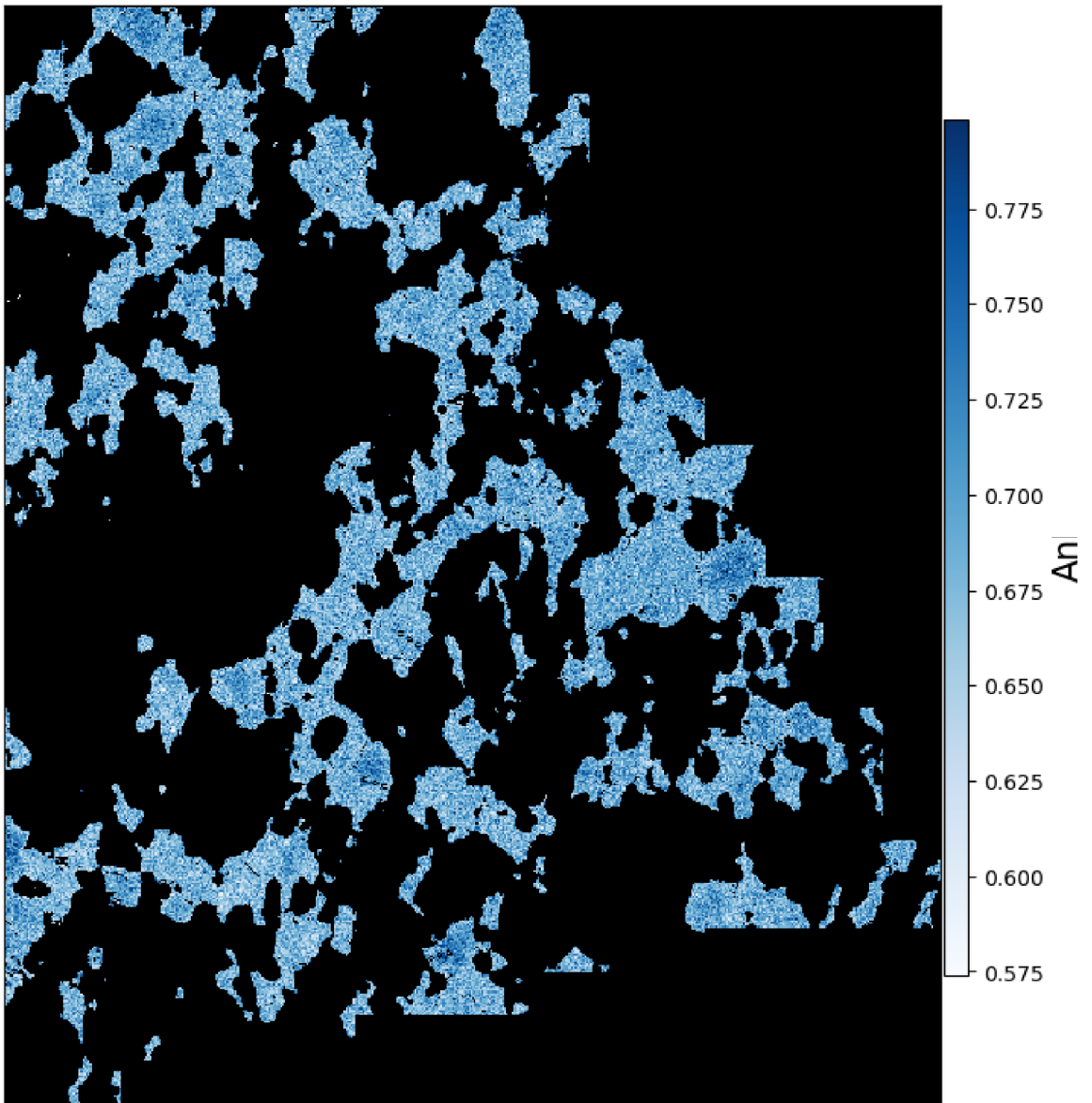


Fig. S.21: EDS map of An contents in plagioclase crystals from sample 03b. The darker colors in the core of some crystals indicate higher An contents, revealing the presence of core-rim normal zoning.



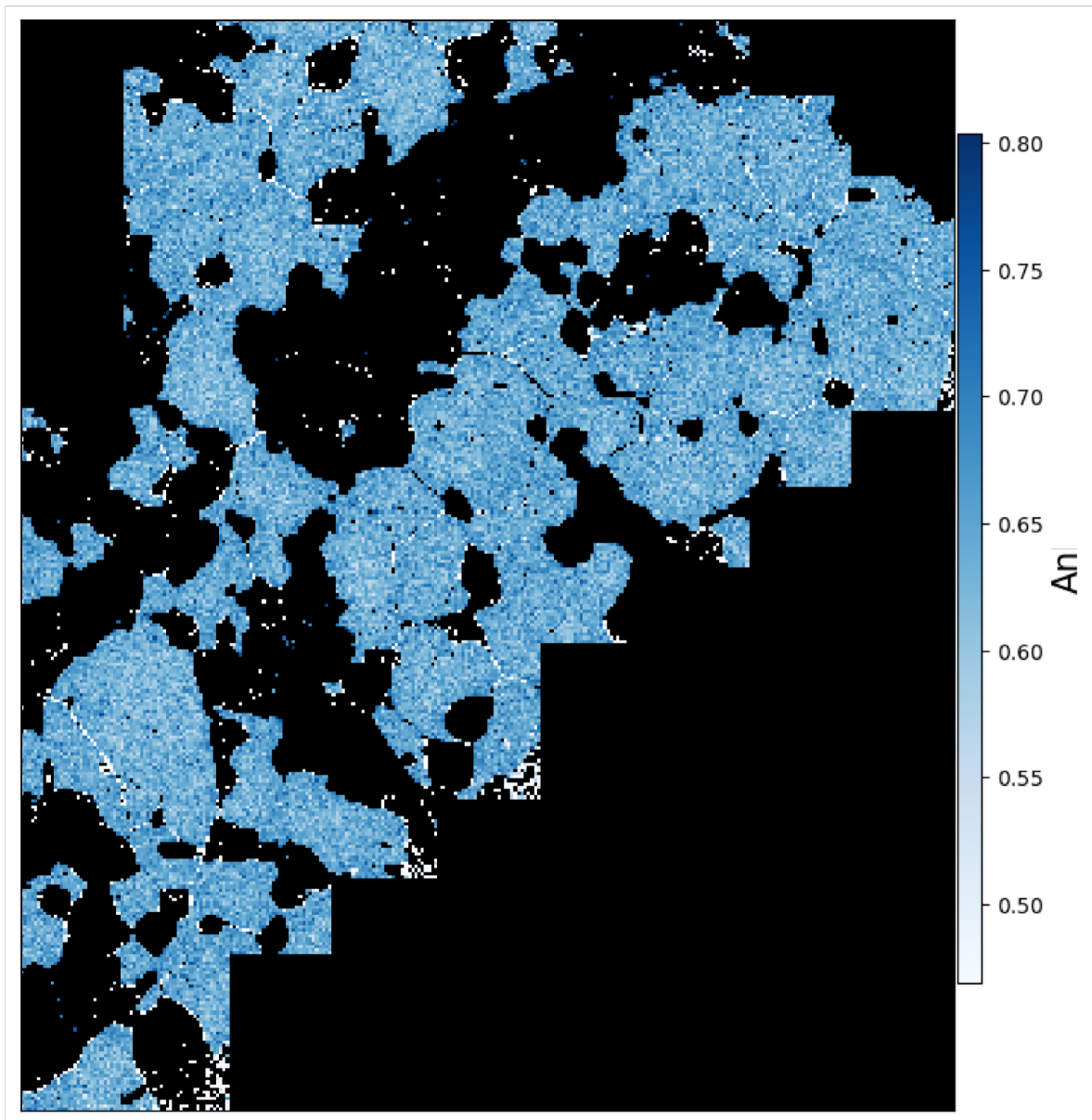


Fig. S.22: EDS map of An contents in plagioclase crystals from sample 09g3. Limited compositional variability is found in this sample compared to sample 03b.

In addition to the large-scale map data, high-resolution maps of orthopyroxene-clinopyroxene grain boundaries also reveal information about the compositional heterogeneity of the Isla Floreana gabbros, in particular, the presence of olivine in orthopyroxene breakdown rims.

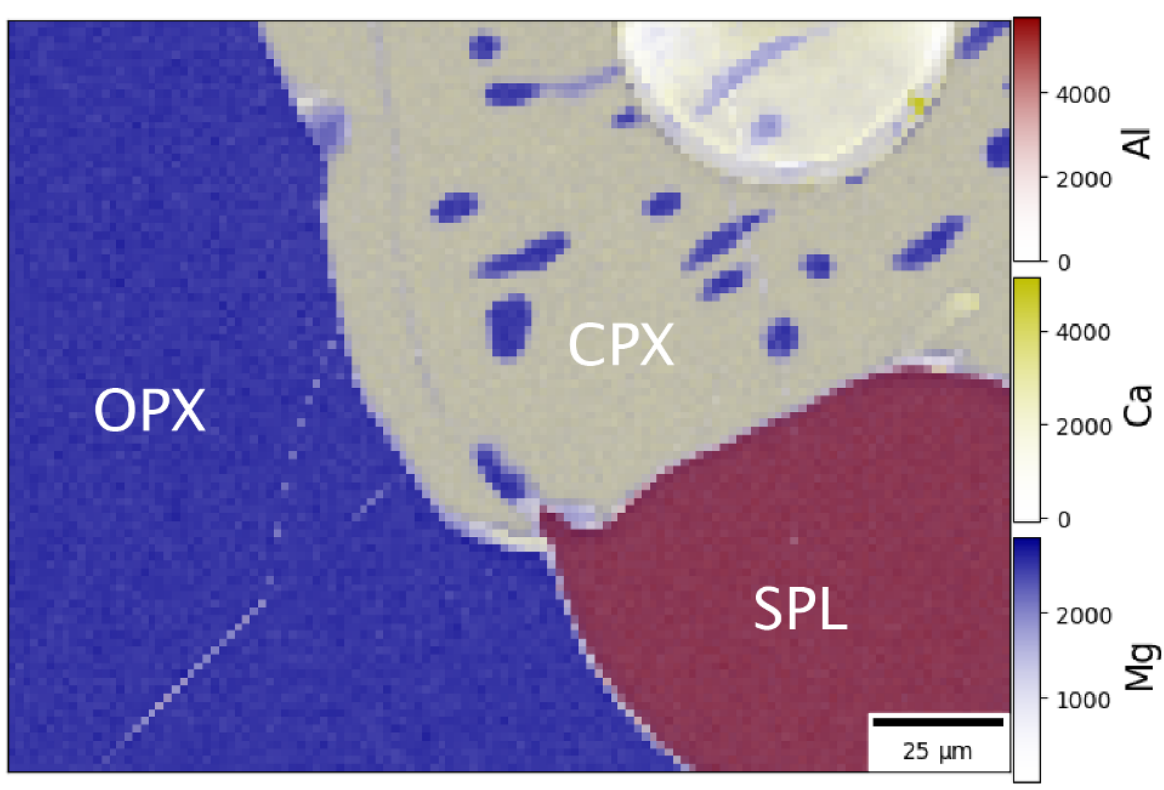


Fig. S.23: EDS chemical map of a clinopyroxene-orthopyroxene-spinel triple boundary in sample 04b. Colors are based on the total counts of Mg, Ca, and Al for each pixel. Exsolution of orthopyroxene is seen in the clinopyroxene crystals. This is a feature only seen in sample 04b.

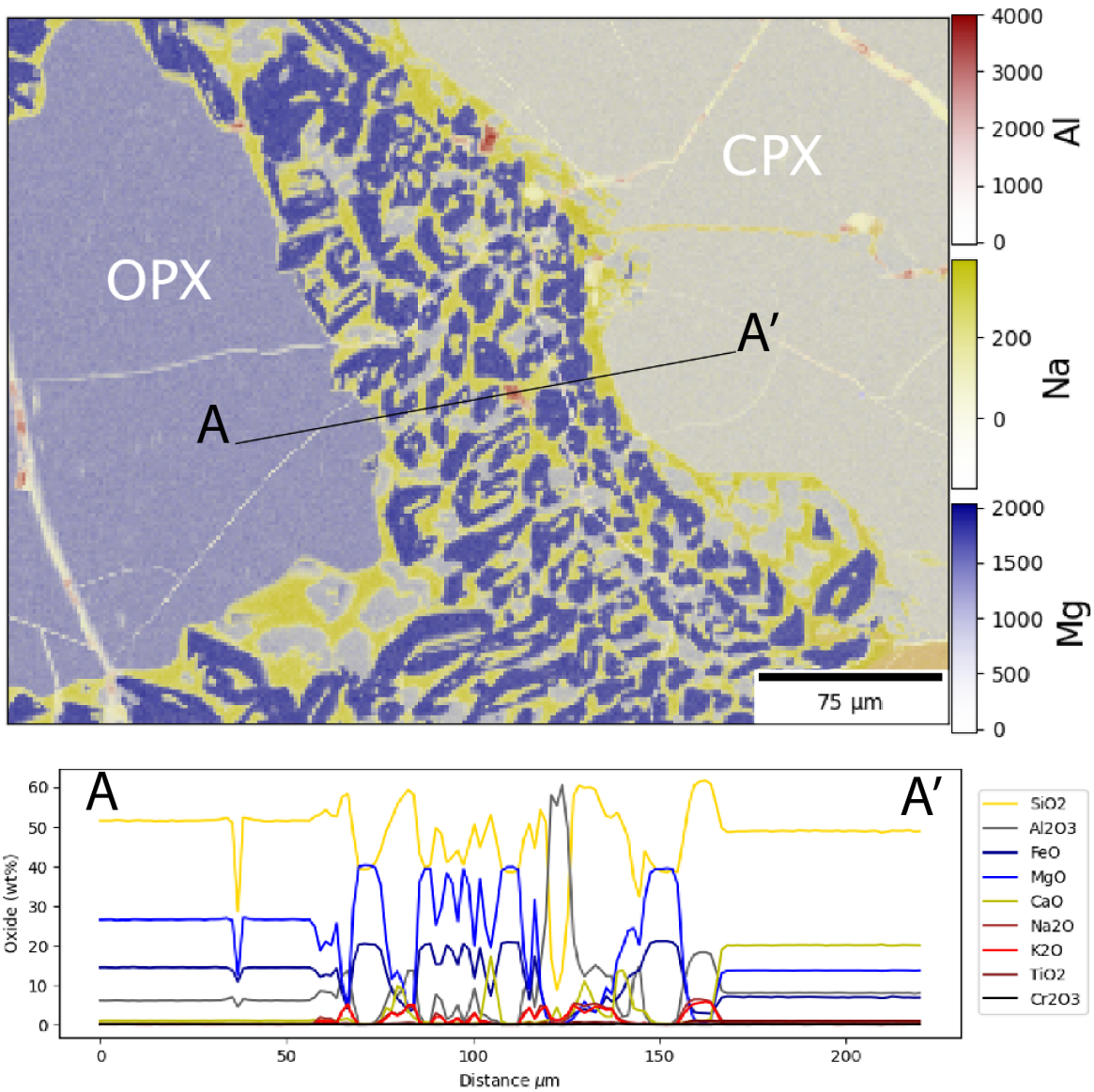


Fig. S.24: EDS chemical map of a clinopyroxene-orthopyroxene grain boundary in sample 09g1. Colors are based on the total counts of Mg, Ca, and Al for each pixel. A small spinel crystal, indicated by the high Al<sub>2</sub>O<sub>3</sub> contents in the linescan is also found in the breakdown zone.

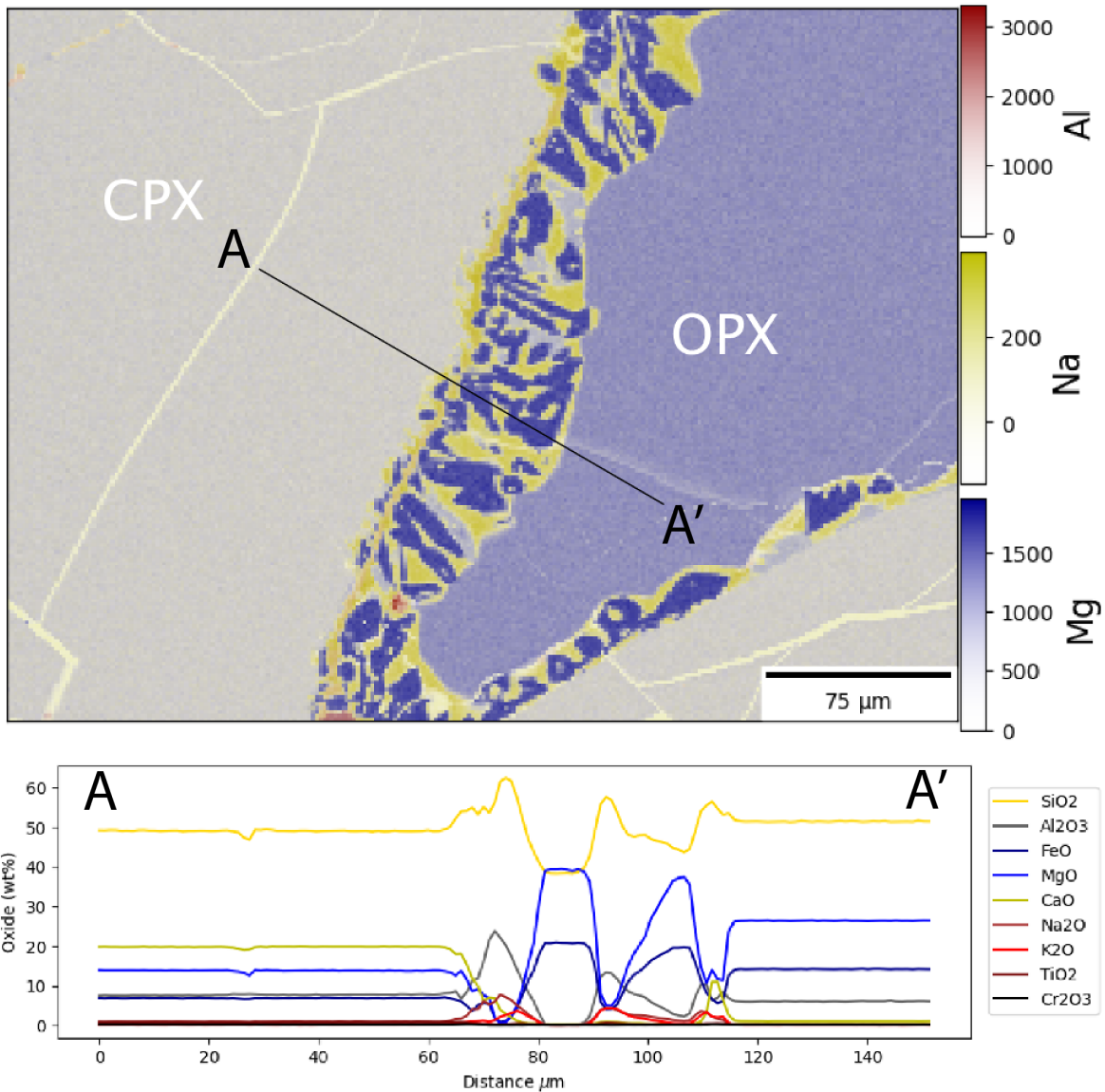


Fig. S.25: EDS chemical map of a clinopyroxene-orthopyroxene grain boundary in sample 09g2. Colors are based on the total counts of Mg, Ca, and Al for each pixel. The orthopyroxene grain is surrounded by a breakdown rim composed of olivine and silicate melt.

# Fluid Inclusions

## Fluid compositions

sample 04b  
cpx 3 FI1

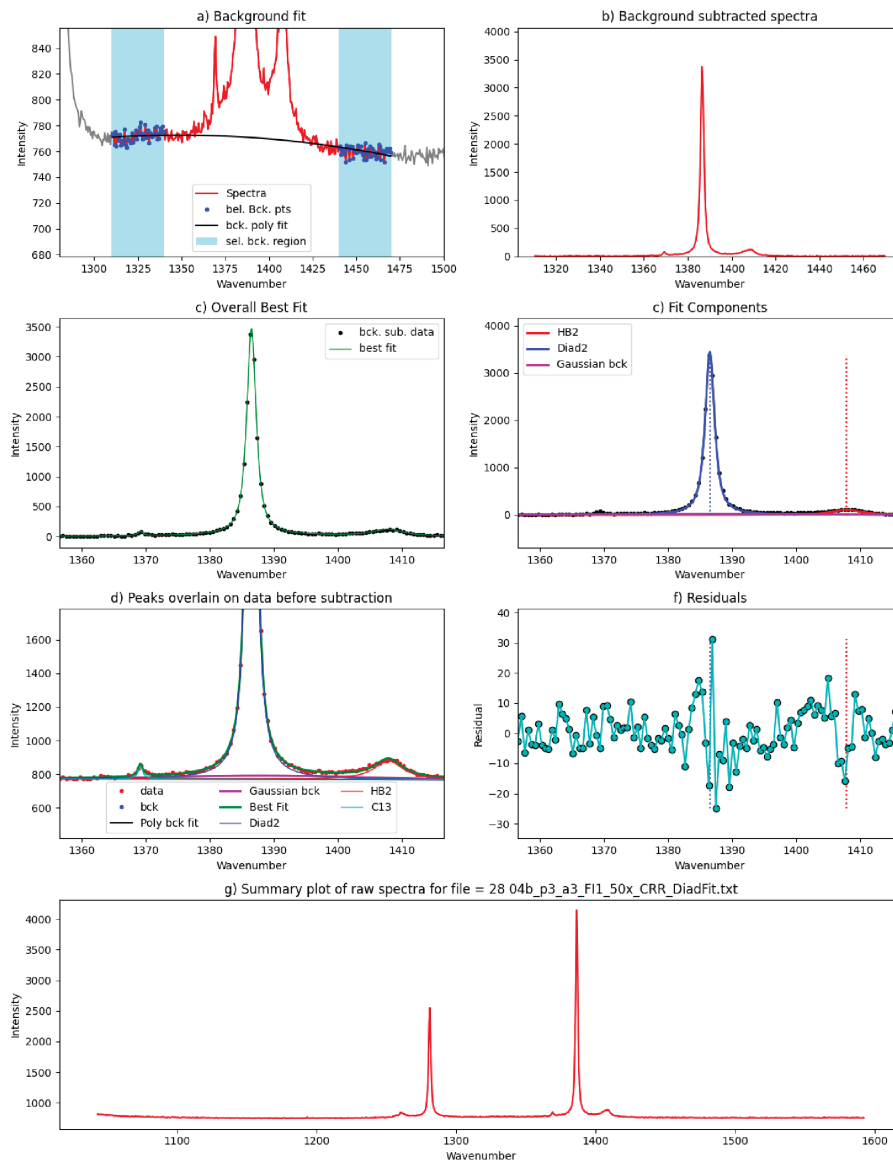


Fig. S.26: Raman spectrum for FI 1 in clinopyroxene 3 from sample 04b. Panels a. - f. demonstrate the way DiadFit determines the background position (a.) and finds the best fit to the second CO<sub>2</sub> diad and nearby secondary peaks. The raw spectra is then shown in panel g. with both the CO<sub>2</sub> diads clearly visible.

In this study we analyzed 43 fluid inclusions across 2 samples. Where possible we analyzed fluid inclusions by both microthermometry and Raman Spectroscopy to determine their composition and CO<sub>2</sub> density. Comparison of the two methods indicates close agreement in the estimated CO<sub>2</sub> densities, with the majority of data plotting between 0.98 - 1.02 g/cm<sup>3</sup> for

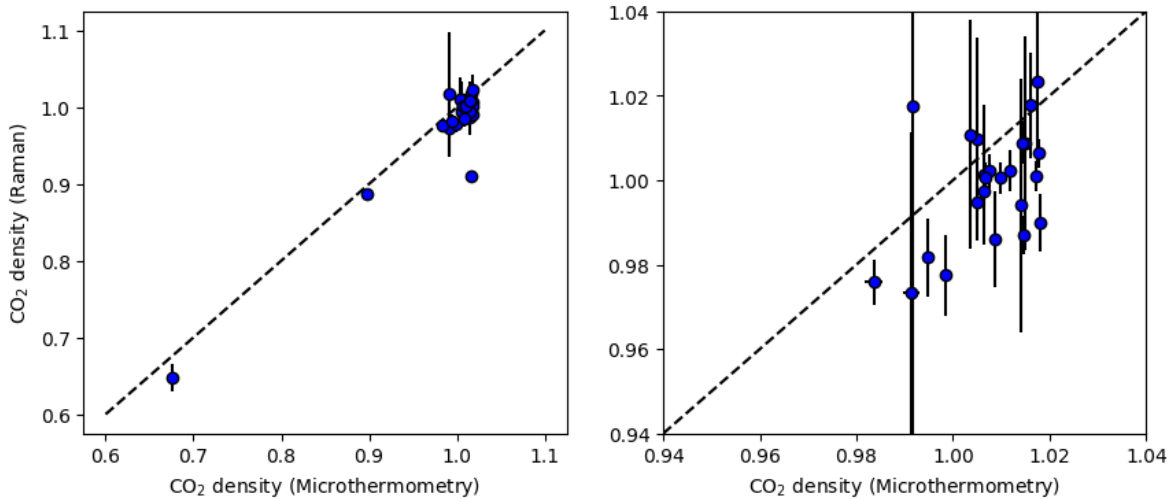


Fig. S.27: Comparison of the CO<sub>2</sub> density (in g/cm<sup>3</sup>) estimated using microthermometry and Raman Spectroscopy.

both approaches (Fig. S.27) Microthermometry revealed several inclusions that have melting temperatures  $> 1^{\circ}\text{C}$  below the theoretical melting point for pure CO<sub>2</sub>, indicating the presence of impurities within these inclusions. Notably, the inclusions with lower melting temperatures are not offset with respect to the estimated CO<sub>2</sub> densities from those with melting temperatures near the pure CO<sub>2</sub> theoretical melting temperature. This indicates that any impurity is likely to have little influence on our overall study (and calculated storage pressures). Nevertheless, both Raman Spectroscopy and COOLPROP modeling (Bell et al. 2014) were used to assess the cause of the observed melting point depression. Within our standard Raman analyses (6 mW laser power, 1800 grating), which targets wavenumbers between 1050 cm<sup>-1</sup> and 1580 cm<sup>-1</sup>, the CO<sub>2</sub> diads and associated minor peaks (hot bands, C-13) were the only peaks identified (Fig. S.26). In short, the Raman analyses provide no evidence for the presence of SO<sub>2</sub>, carbonates, or any other Raman-active species within this wavenumber region. Further examples of the raman spectra collected can be found in the associated Zenodo repository.

To confirm that our analyses did not miss the presence of impurities - owing to the choice of grating and/or low laser power - we also performed Raman analyses on a small number of fluid inclusions selected to encompass the entire range of melting temperatures identified in this study. For these secondary analyses we increased the laser power to 49 mW and the grating was changed to 300. This enabled us to access wavenumbers up to 3270 cm<sup>-1</sup>. Comparison of three spectra acquired using these conditions are shown in Fig. S.28 and demonstrate that there are no observable differences in the Raman spectra of fluid inclusions with melting temperatures between  $-56.5^{\circ}\text{C}$  and  $-60.1^{\circ}\text{C}$ . Consequently, we conclude that the melting point depression observed in our fluid inclusions does not result from the presence of a Raman active species (e.g., N<sub>2</sub>, CH<sub>4</sub>, CO, etc.), but is likely caused by the presence of a noble gas (e.g., He, Ar, Ne).

Using CoolProp (Bell et al., 2014) we assessed the potential He and Ar proportions in the fluid inclusions by calculating their influence on the triple-point temperature of mixed fluids (relative to the melting temperature of pure CO<sub>2</sub>) using CO<sub>2</sub> - He and CO<sub>2</sub> - Ar equations of state from (Fig. S29; Kunz and Wagner, 2012).

The EOS calculations also allow us to calculate the densities of CO<sub>2</sub>-He and CO<sub>2</sub>-Ar mixed

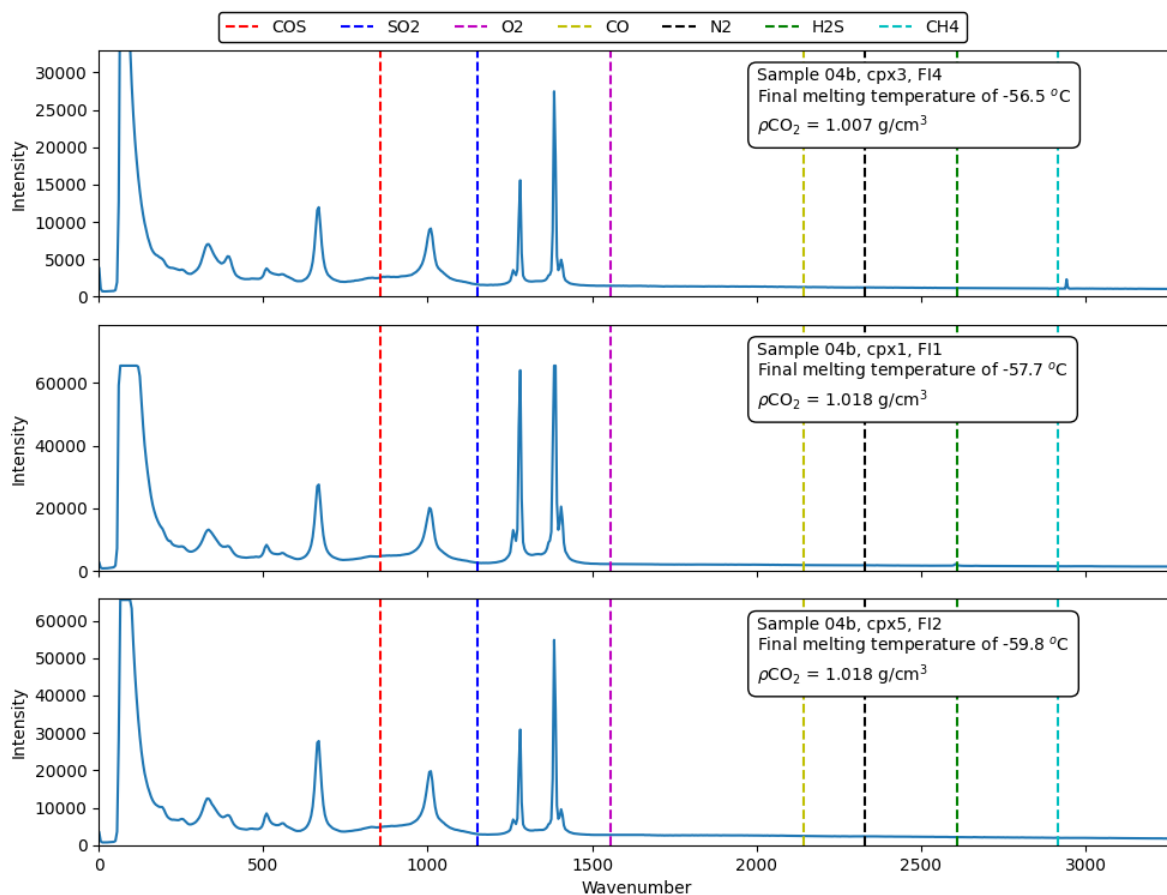


Fig. S.28: Comparison of Raman spectra collected at 49 mW for inclusions showing a range of melting temperatures. Comparison of the spectra demonstrates no notable difference between the different includes with no peaks overlapping with the positions of know Raman vibrations associated with common gaseous species (positions taken from Frezzotti et al., 2012.)

fluids, converting the measured homogenization temperatures (and He or Ar proportions) into these values. Results indicate that the presence of He increases the estimated density (i.e., a greater fluid density is required to match a specific homogenization temperature). In contrast, the presence of Ar in the FIs would lead to a very small reduction in the estimated density (relative to a pure-CO<sub>2</sub> fluid; Fig. S30).

Finally, to assess the influence of He and Ar on the estimated entrapment and/or equilibration pressures for the FIs considered in this study, we calculate the position of isochores for theoretical FIs with homogenization temperatures of -17°C and a range of He or Ar contents. Results indicate that the presence of He and Ar have opposite effects on the calculated pressures (Fig. S31). Calculations are performed with increasing temperature until CoolProp is no longer able to calculate the density of the mixed fluid. Typically this limit happens at <450°C for calculations involving the presence of He or Ar.

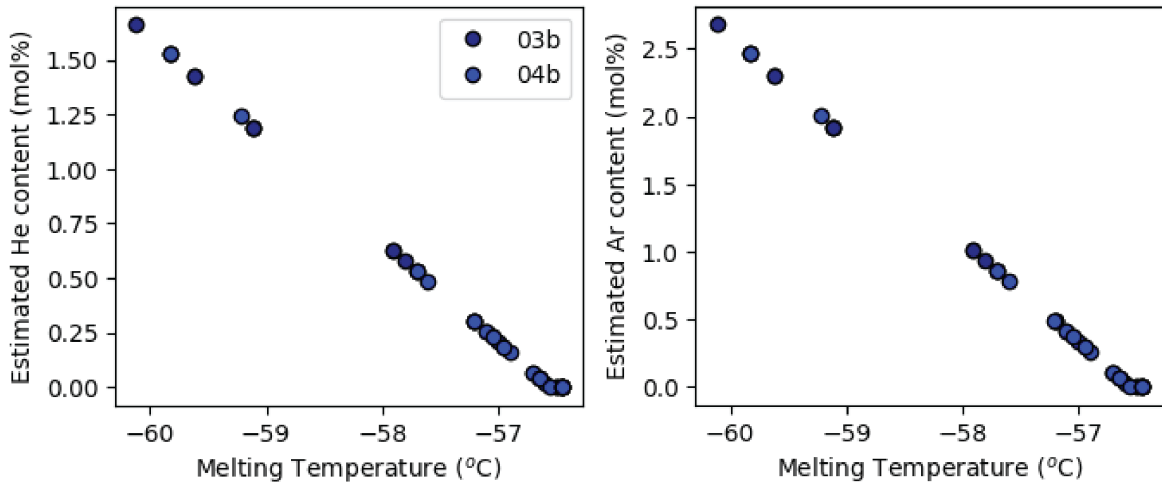


Fig. S.29: Melting temperature of the 03b and 04b fluid inclusions against the He (left) and Ar (right) contents required to cause these melting temperature decreases according to the mixed EOS of Kunz and Wagner (2012).

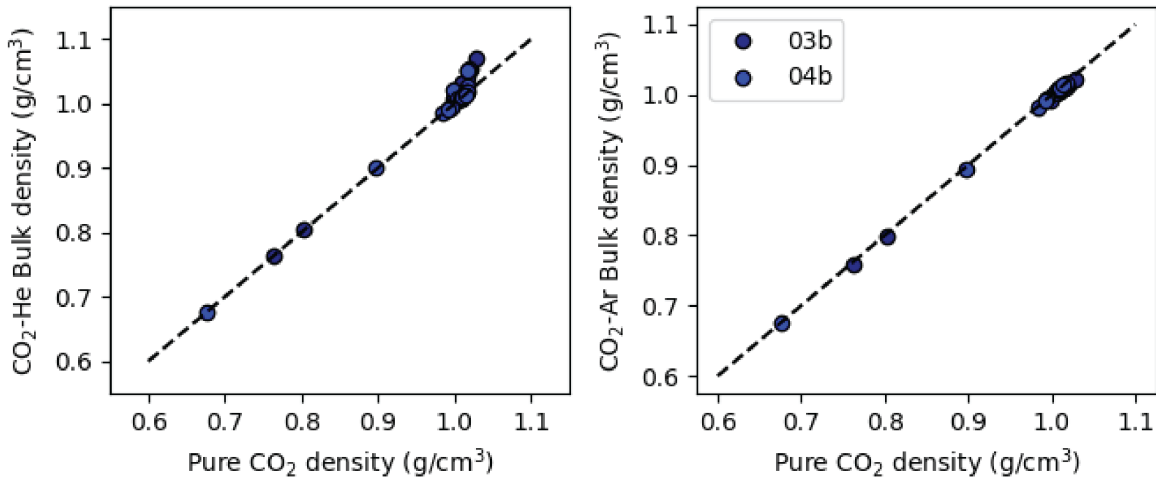


Fig. S.30: Calculated CO<sub>2</sub> densities (using the EOS of Span and Wagner (1996)) against estimated CO<sub>2</sub>-He (left) and CO<sub>2</sub>-Ar (right) mixed fluid densities using the mixed EOS of Kunz and Wagner (2012).

### Fluid Inclusion Textural Constraints

Within this study we were primarily interested in the inclusions that contain the highest CO<sub>2</sub> densities, as these are most likely to record the long-term storage pressure of the gabbro xenoliths/cumulates, rather than relecting processes happening on ascent. For that reason, we targeted fluid inclusions that were typically  $\mu$  15



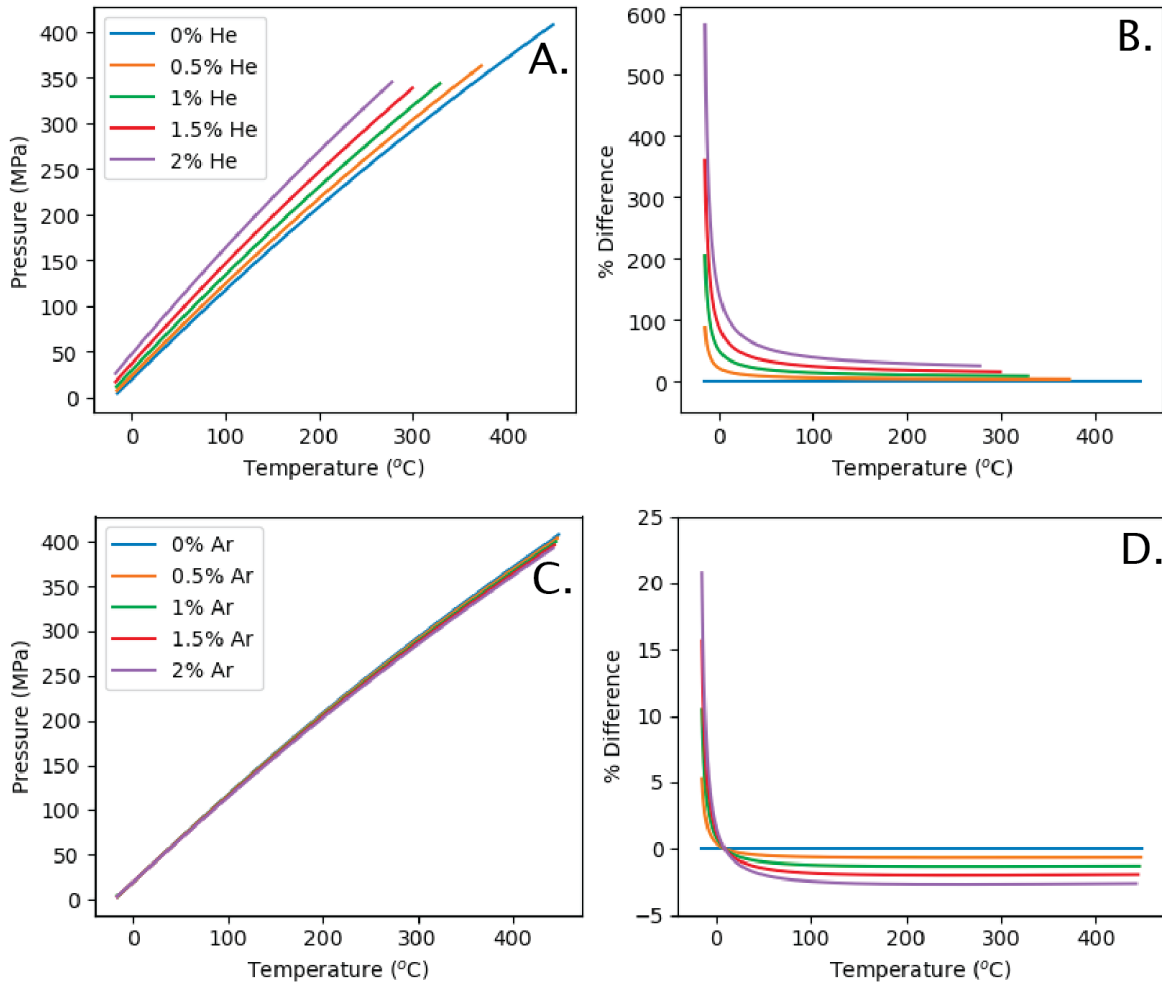


Fig. S.31: Isochores for FIs with different He (A.) and Ar (C.) contents. Results show an increase in pressure with increasing He, but a slight decrease with increasing Ar. Relative pressure variations against temperature are shown in B. (He) and D. (Ar)

*mm* and are not located on cracks. Some inclusions were identified on cracks/fractures and did not return any signal when analyzed by Raman Spectroscopy, indicating complete loss of CO<sub>2</sub> at low pressure (see Fig. S.38). Three inclusions in sample 04b display clear signs of decrepitation (irregular morphology or many smaller bubbles surrounding the main inclusion) and display systematically lower CO<sub>2</sub> densities than the rest of the inclusions considered in this study. Finally, the three other inclusions that contain CO<sub>2</sub> densities less than 0.9 g/cm<sup>3</sup> are larger than the majority of inclusions considered in this study (sample 03b, cpx 2, FI1 & 3, see Fig. 11 of the main text) or are located near a crack and grain boundary (sample 04b, cpx 1, FI3). Both factors are known to make fluid inclusions more susceptible to re-equilibration and could, potentially, explain the lower CO<sub>2</sub> densities in these inclusions.

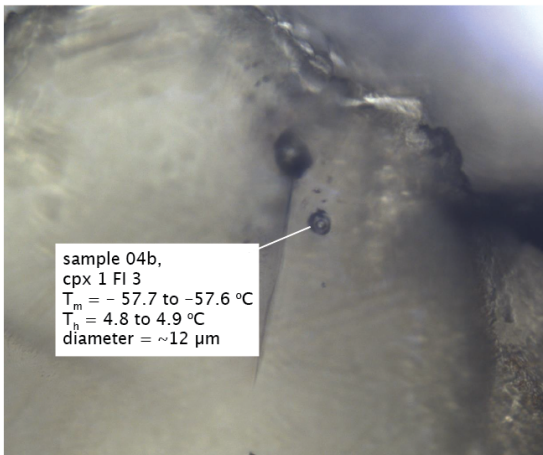
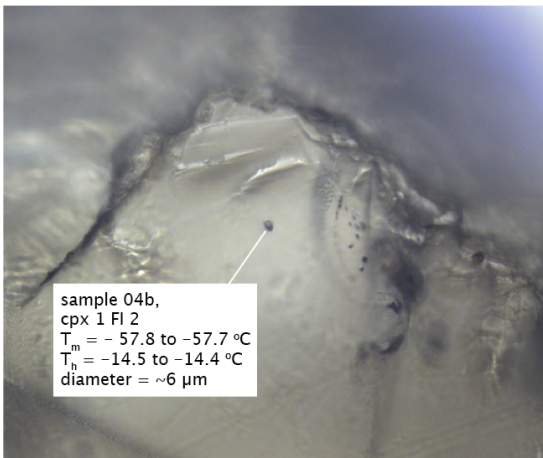
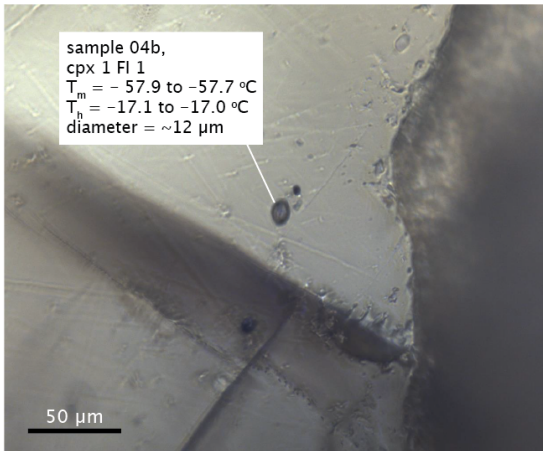


Fig. S.32: Microscope images of fluid inclusions in sample 04b, clinopyroxene 1. Melting and homogenization temperatures, alongside an estimate of the maximum inclusion diameter are given for each inclusion.

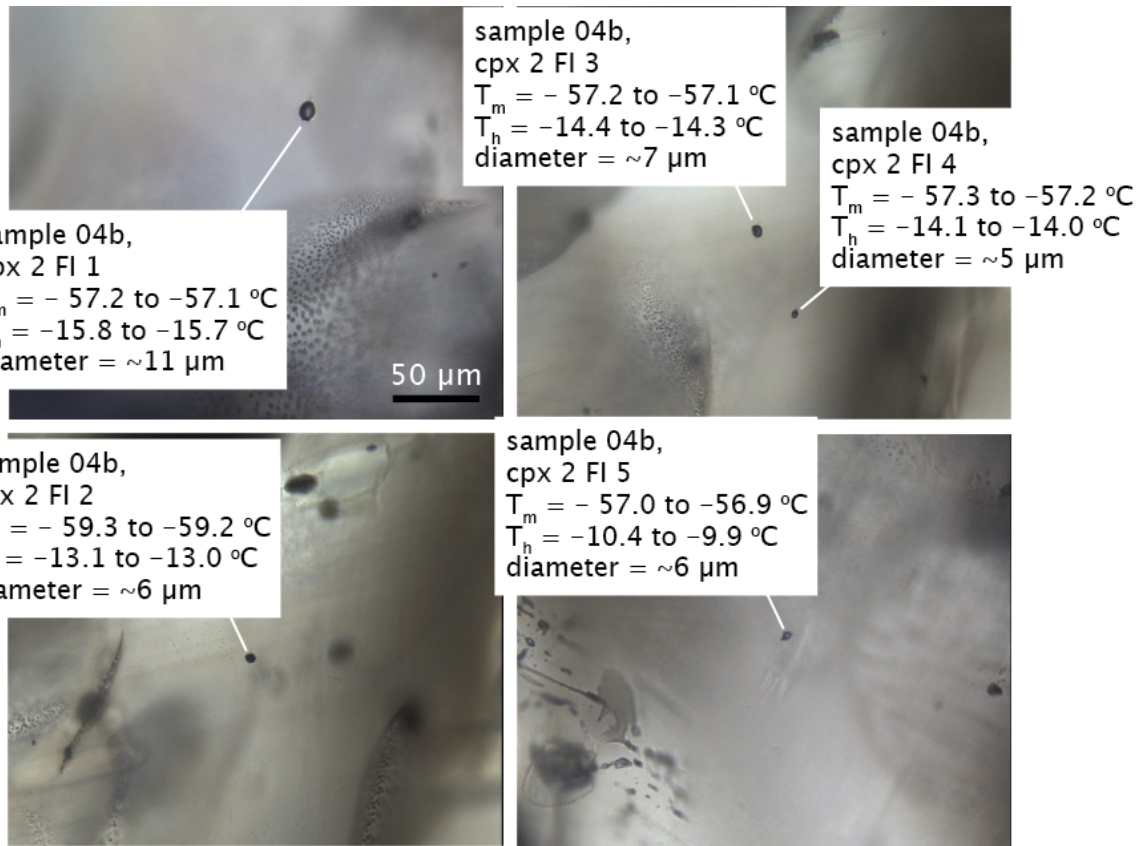


Fig. S.33: Microscope images of fluid inclusions in sample 04b, clinopyroxene 2. Melting and homogenization temperatures, alongside an estimate of the maximum inclusion diameter are given for each inclusion.

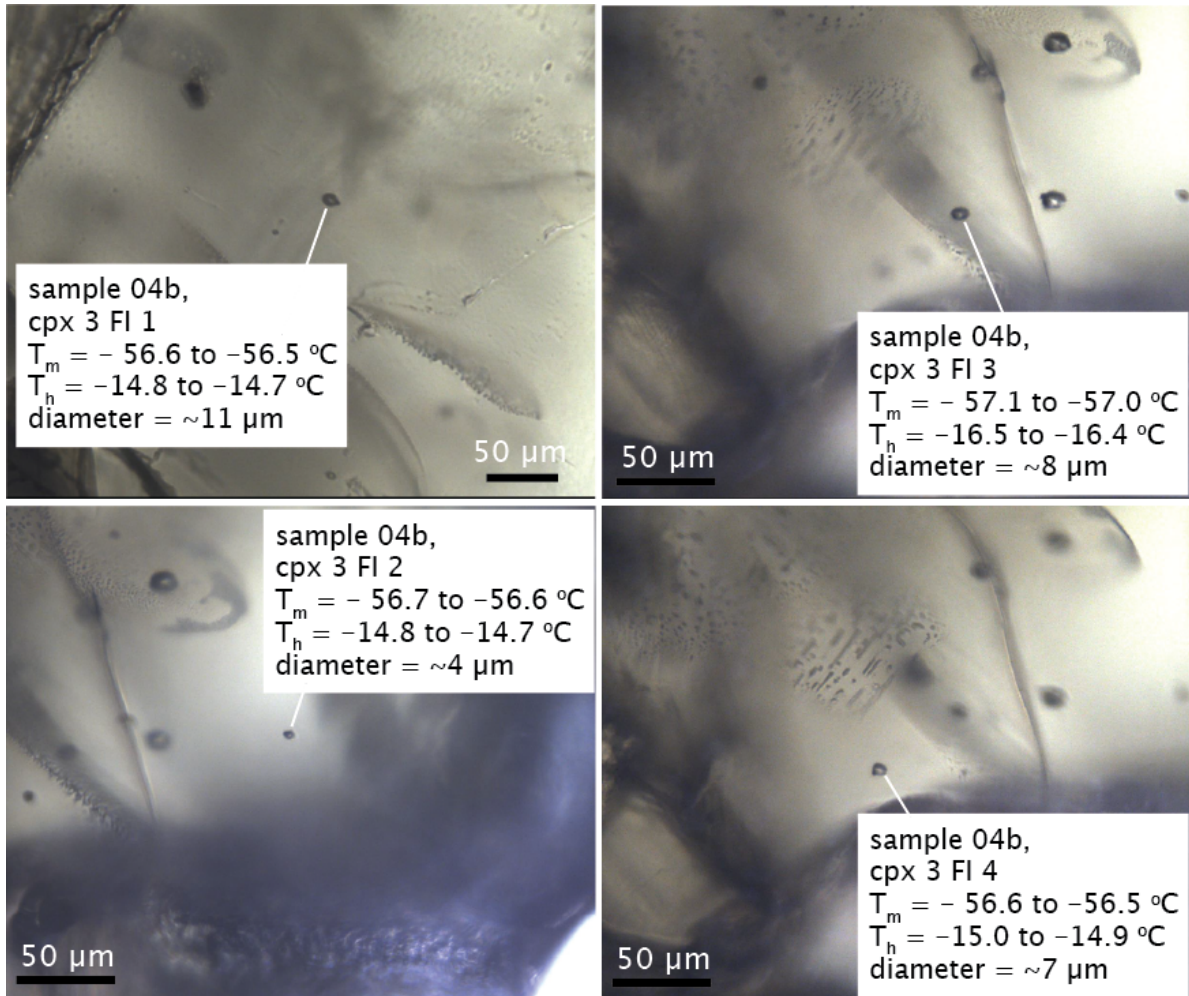


Fig. S.34: Microscope images of fluid inclusions in sample 04b, clinopyroxene 3. Melting and homogenization temperatures, alongside an estimate of the maximum inclusion diameter are given for each inclusion.

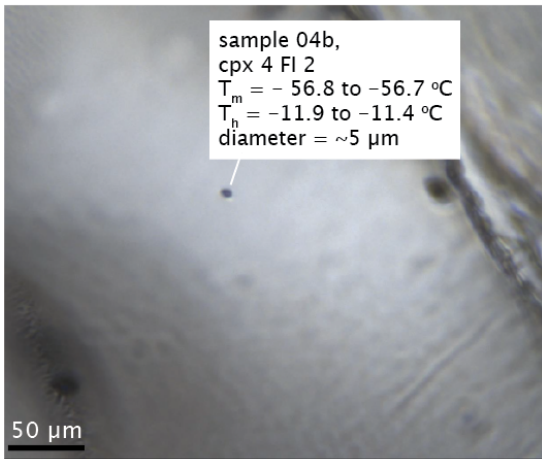
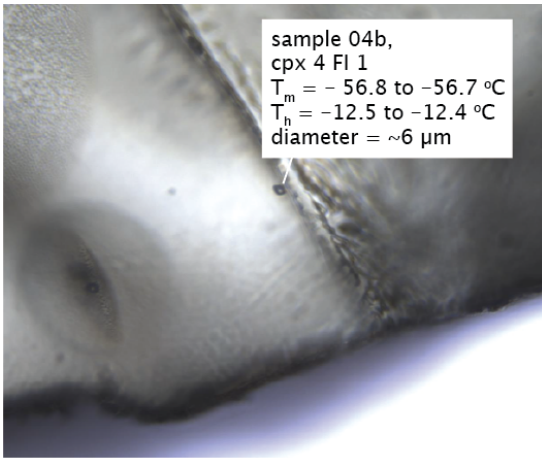


Fig. S.35: Microscope images of fluid inclusions in sample 04b, clinopyroxene 4. Melting and homogenization temperatures, alongside an estimate of the maximum inclusion diameter are given for each inclusion.

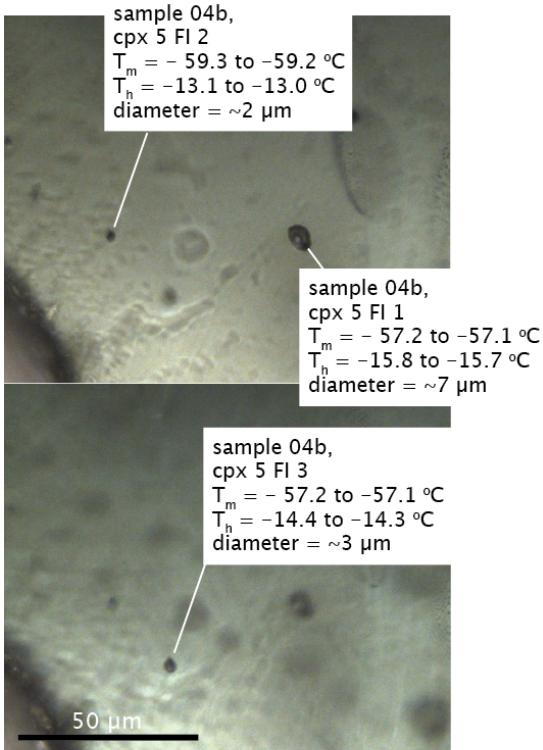


Fig. S.36: Microscope images of fluid inclusions in sample 04b, clinopyroxene 5. Melting and homogenization temperatures, alongside an estimate of the maximum inclusion diameter are given for each inclusion.

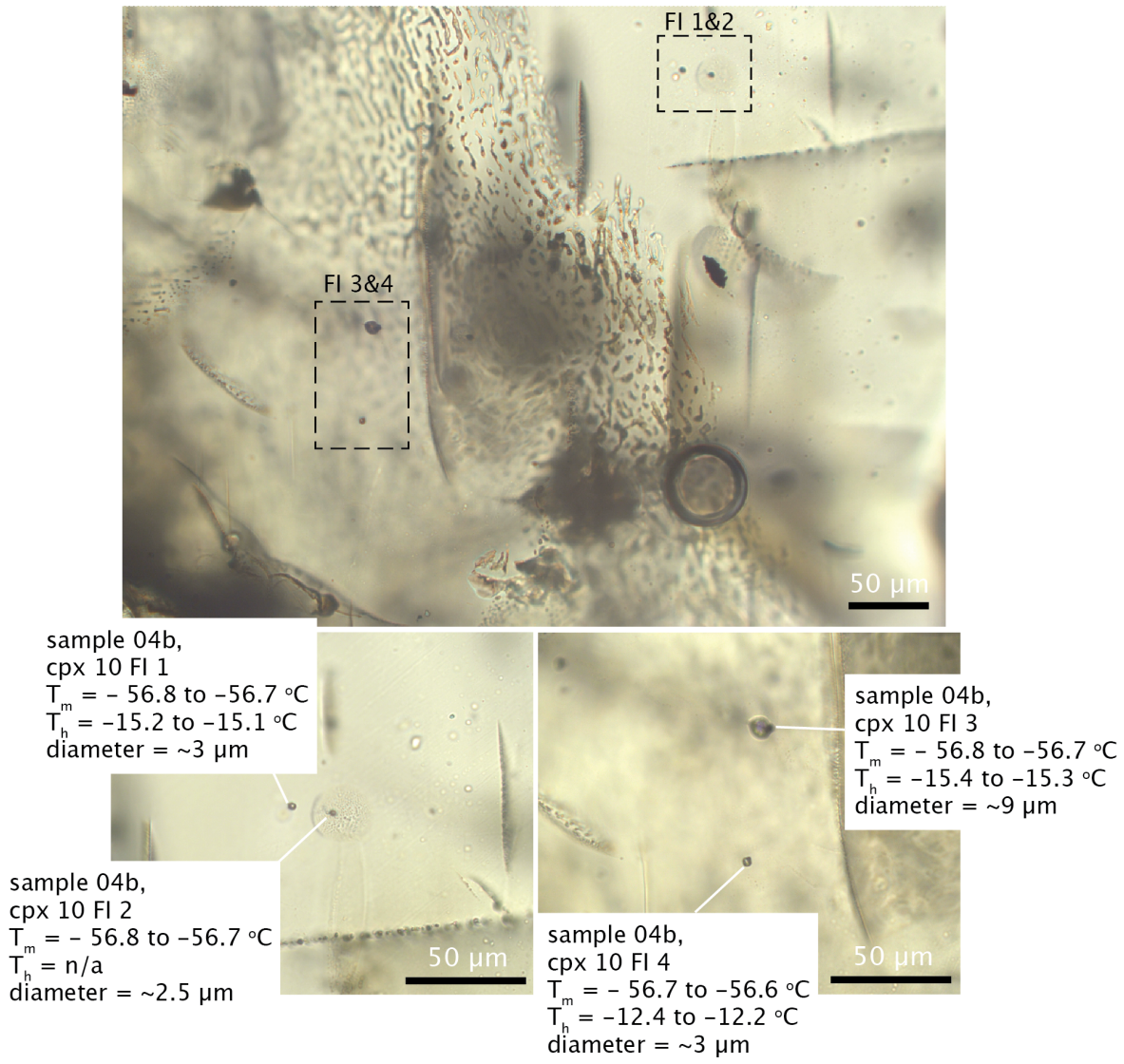


Fig. S.37: Microscope images of fluid inclusions in sample 04b clinopyroxene 10. This figure highlights 2 regions of a crystal with 2 fluid inclusions analyzed in each area. Fluid inclusion 2 displays clear signs of decrepitation and we were not able to determine the homogenization temperature by microthermometry. For other inclusions the melting and homogenization temperatures, alongside an estimate of the maximum inclusion diameter are given.

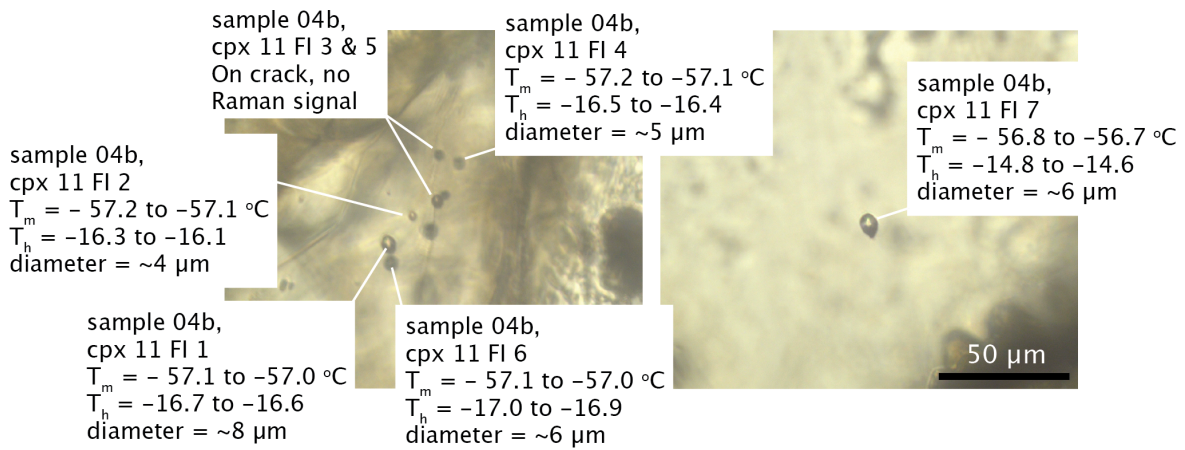


Fig. S.38: Microscope images of fluid inclusions in sample 04b clinopyroxene 11. This figure highlights 2 regions of a crystal with fluid inclusions analyzed in each area. Fluid inclusions 1 - 6 represent a trail of inclusions near a crack in the crystal. Inclusions 3 and 5 are located directly on the crack and we were not able to make any measurements of these inclusions by either Raman Spectroscopy or Microthermometry (indicating a complete loss of  $\text{CO}_2$  from these inclusions). For other inclusions the melting and homogenization temperatures, alongside an estimate of the maximum inclusion diameter are given.



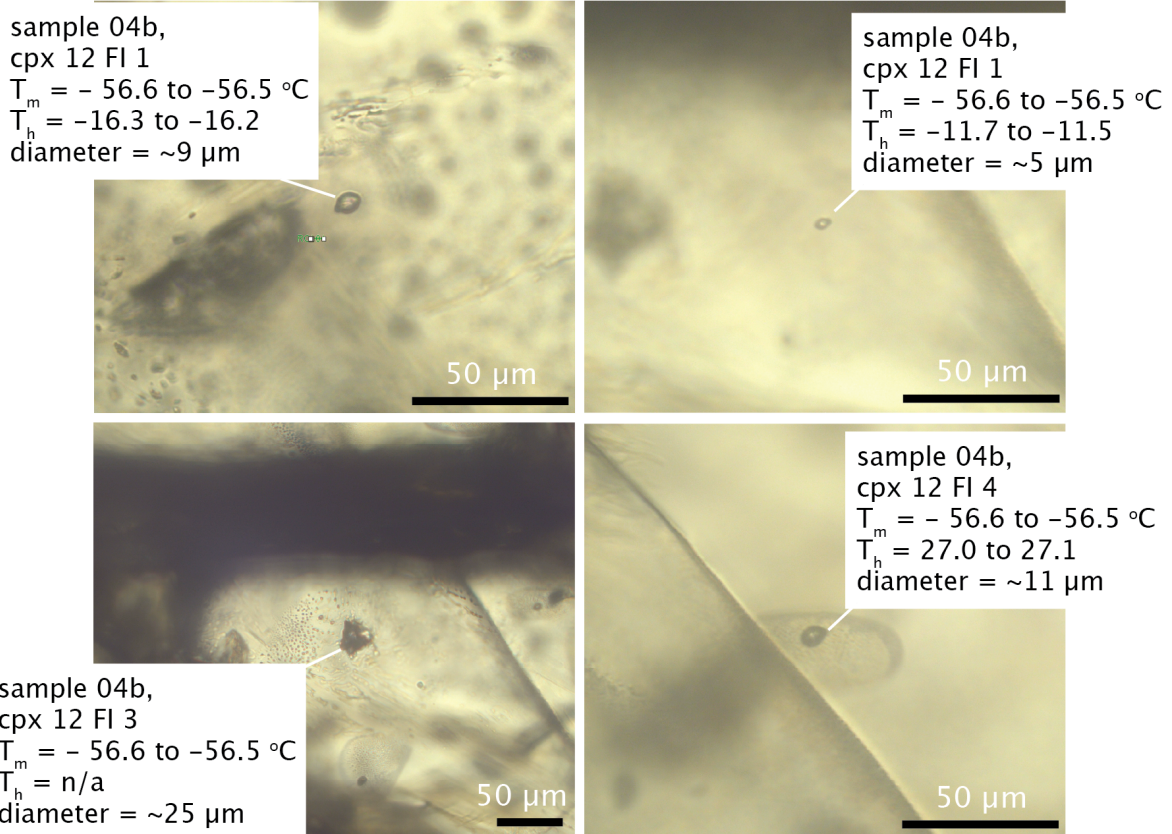


Fig. S.39: Microscope images of fluid inclusions in sample 04b clinopyroxene 12. Fluid inclusions 3 and 4 display clear signs of decrepitation and we were not able to determine the homogenization temperature of fluid inclusion 3 by microthermometry. For other inclusions the melting and homogenization temperatures, alongside an estimate of the maximum inclusion diameter are given.

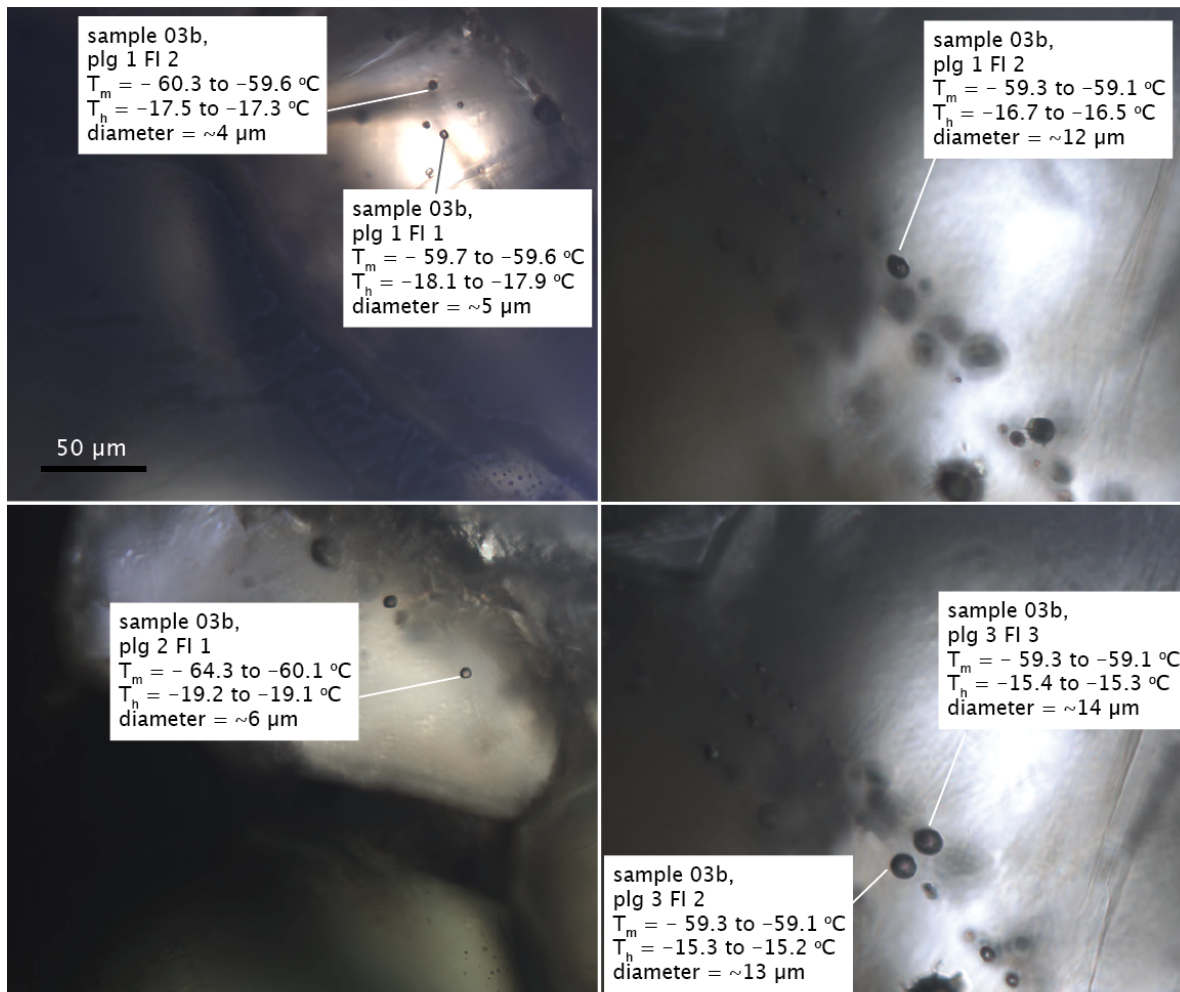


Fig. S.40: Microscope images of fluid inclusions in sample 03b with three separate plagioclase crystals shown (plagioclase 1 - top left, plagioclase 2 - bottom left, plagioclase 3 - right). Melting and homogenization temperatures, alongside an estimate of the maximum inclusion diameter are given for each inclusion.

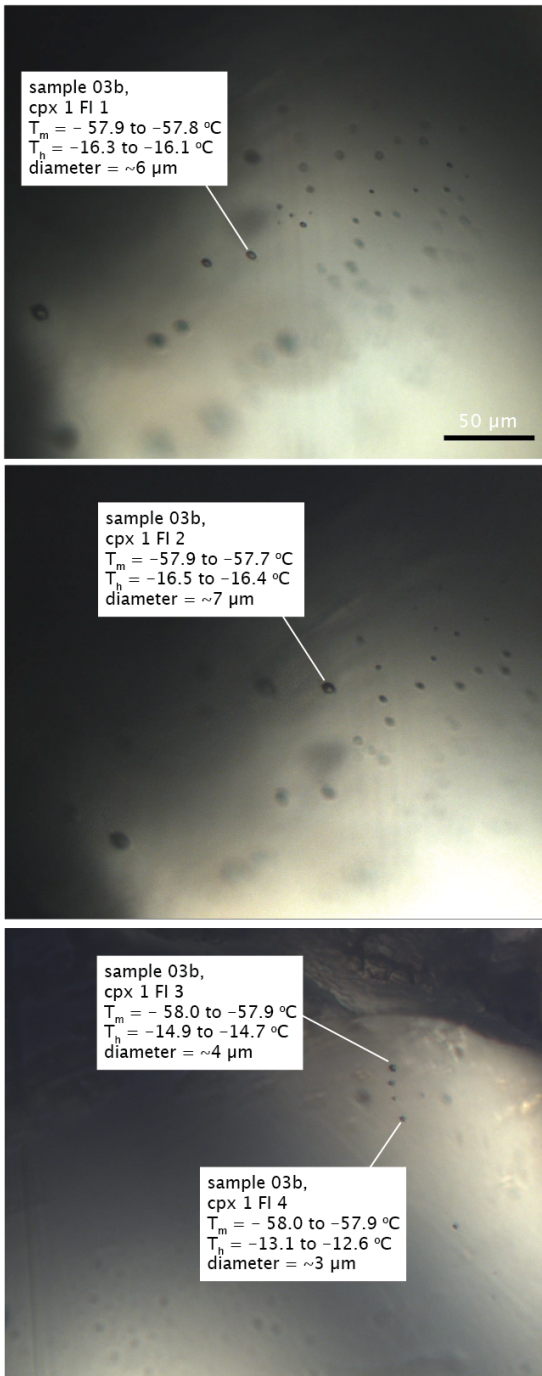


Fig. S.41: Microscope images of fluid inclusions in sample 03b, clinopyroxene 1. Melting and homogenization temperatures, alongside an estimate of the maximum inclusion diameter are given for each inclusion.

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